

Cation-exchange chromatography of monoclonal antibodies

Characterization of a novel stationary phase designed for production-scale purification

Marina Urmann,¹ Heiner Graalfs,² Matthias Joehncck,² Lothar R. Jacob² and Christian Frech^{1,*}

¹Institute for Biochemistry; University of Applied Sciences Mannheim; Mannheim, Germany; ²Performance & Life Science Chemicals; Merck KGaA; Darmstadt, Germany

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Abbreviations: CLSM, confocal laser scanning microscopy; DBC, dynamic binding capacity [mg/mL gravity settled resin]; hcp, host cell protein; LGE, linear gradient elution; A, parameter from the GH- I_R -plot; A includes the equilibrium constant, the ionic capacity of the resin and B [M^B]; B, number of charges involved in protein adsorption; C, concentration of unbound protein [mg/mL]; C_0 , initial protein concentration in the sample [mg/mL]; GH, normalized gradient slope [M]; I_R , ionic strength at peak position [M]; q, amount of bound protein [mg/mL gravity settled resin]; q_0 , maximum adsorption capacity [mg/mL gravity settled resin]; V, volume of protein sample [mL]; V_M , sample volume of gravity settled resin [mL]

A novel cation-exchange resin, Eshmuno™ S, was compared to Fractogel® SO₃⁻ (M) and Toyopearl GigaCap S-650M. The stationary phases have different base matrices and carry specific types of polymeric surface modifications. Three monoclonal antibodies (mAbs) were used as model proteins to characterize these chromatographic resins. Results from gradient elutions, stirred batch adsorptions and confocal laser scanning microscopic investigations were used to elucidate binding behavior of mAbs onto Eshmuno™ S and Fractogel® SO₃⁻ and the corresponding transport mechanisms on these two resins. The number of charges involved in mAb binding for Eshmuno™ S is lower than for Fractogel® SO₃⁻, indicating a slightly weaker electrostatic interaction. Kinetics from batch uptake experiments are compared to kinetic data obtained from confocal laser scanning microscopy images. Both experimental approaches show an accelerated protein adsorption for the novel stationary phase. The influence of pH, salt concentrations and residence times on dynamic binding capacities was determined. A higher dynamic binding capacity for Eshmuno™ S over a wider range of pH values and residence times was found compared to Fractogel® SO₃⁻ and Toyopearl GigaCap S-650M. The capture of antibodies from cell culture supernatant, as well as post-protein A eluates, were analyzed with respect to their host cell protein (hcp) removal capabilities. Comparable or even better hcp clearance was observed at much higher protein loading for Eshmuno™ S than Fractogel® SO₃⁻ or Toyopearl GigaCap S-650M.

Introduction

Recently concerns regarding potential bottlenecks in the downstream processing of monoclonal antibodies (mAbs) on an industrial scale have been heatedly discussed. Current antibody purification platforms seem to be capable of processing large fermentation volumes. However, the first adsorptive column, rather than clarification operations such as filtration and centrifugation, has been identified as the true bottleneck in antibody manufacturing.^{1,2} Increasing cell culture mAb titers are proving to be problematic for downstream purification processes because the relatively low binding capacities of protein A resins for capture limit overall throughput.

Although production costs have gone down due to improvements in fermentation, the relative proportion of cost of goods correlated to downstream purification is increasing sharply; downstream processing might comprise up to 80% of the entire production costs. Due to continuous pressure on manufacturing costs, more and more resources are now being dedicated to process optimization. In particular, the first downstream operation step affects the overall throughput and economics. Both cross-flow microfiltration and centrifugation unit operations are routinely employed to remove the solid components of the fermentation process. The objective of the subsequent step, the capture step, is to recover as much product as possible and provide a product pool that is suitable for subsequent chromatographic

*Correspondence to: Christian Frech; Email: c.frech@hs-mannheim.de

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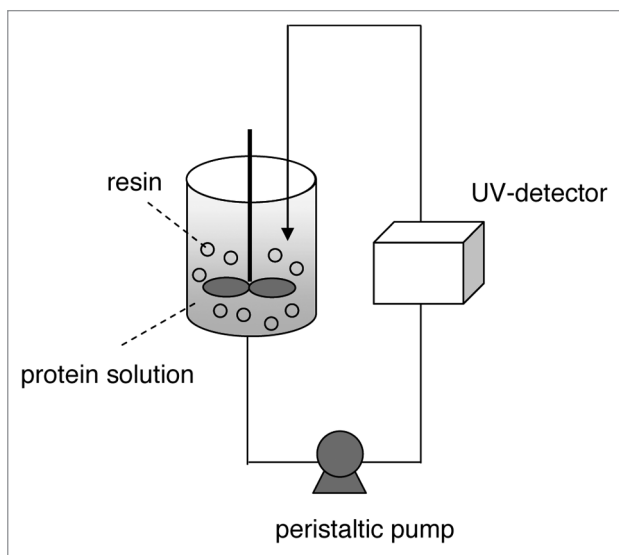


Figure 1. Experimental set-up for stirred batch experiments.

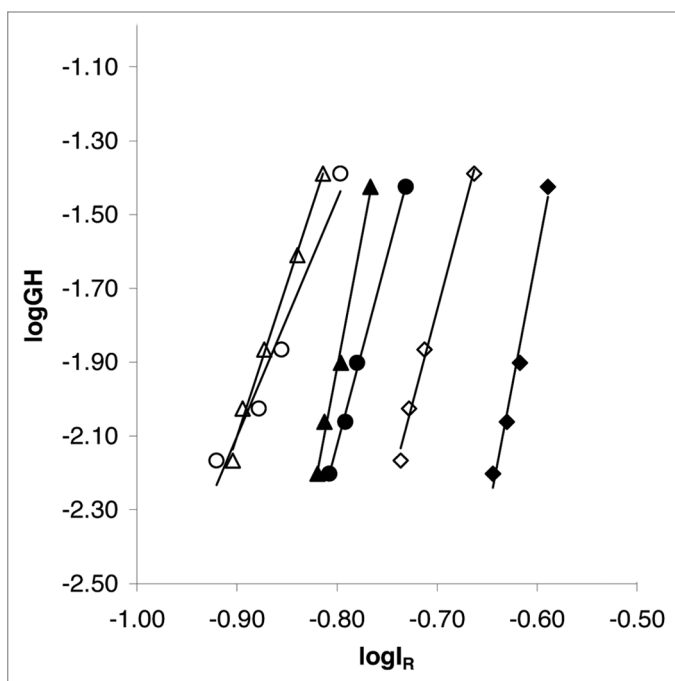


Figure 2. GH- I_R -plots for the monoclonal antibody mAb01 and the resins Eshmuno™ S and Fractogel™ SO₃⁻ (M) at different pH values. Linear gradient elutions were performed with 3.93 mL Superformance columns (10 mm i.d. x 50 mm L) at a linear flow rate of 119 cm/h and gradient lengths of 15–90 CV from 0–1 M NaCl. The protein load was 0.44 mg/mL packed resin. Filled symbols represent Fractogel™ SO₃⁻, blank symbols Eshmuno™ S. Rhombes represent pH 5, circles pH 5.5 and triangles pH 6.

column steps, i.e., with volumes small enough to be handled quickly and sample compositions that do not interfere with the next chromatographic mode.

The first chromatographic step is usually performed on a Protein A affinity column. The dynamic binding capacity of

Protein A resins is usually ≤ 40 g/L and depends strongly on residence time. The binding capacity of the used resin together with the column size dictates the number of cycles that must be run to process the entire cell culture supernatant.^{3,4} Affinity chromatography on Protein A is widely accepted and is a part of platform purification strategies because it can be implemented in almost all antibody production schemes.⁵ However, the high price (US \$9,000–12,000 per liter) for the corresponding stationary phases contributes significantly to the manufacturing costs.

Currently, discussions about improvements in large scale mAb purification quite often revolve on the re-design of the capture step. Alternative techniques for capture, such as precipitation, liquid/liquid partitioning and crystallization, are also under investigation so that the protein purification toolbox can be expanded.^{6–8} However large scale crystallization is still far away from being a routine method for production of mAbs.

Another way to address improvement of the first step is to develop alternative capture techniques such as high-capacity capture steps on cation exchange media.^{9,10} Successful use of appropriate cation exchange columns has been reported to resolve downstream bottlenecks, sustain cost-effective production, and manage large quantities.^{11–14}

Cation exchangers have proven to give high step yields and reduce the level of host cell proteins (hcp) very effectively. Another advantage of ion exchangers is their considerably higher chemical stability, which minimizes ligand leaching. Since a large number of different stationary phases for cation exchange chromatography are commercially available from different manufacturers, a well-designed evaluation strategy is necessary to identify the best cation exchanger for a given separation. Although in many cases the same functional groups have been utilized to create attractive binding sites for the proteins, all stationary phases vary significantly in a number of chemical and physical properties. In practice, not only does the base bead structure of the stationary phase have an impact on the binding and elution kinetics, but the surface modification chemistry also influences the protein binding mechanism.¹⁵

Design and optimization of ion-exchange chromatography unit operations require consideration of many operating and chromatographic parameters. In ion exchange chromatography, protein adsorption depends on (1) the composition and concentration of the protein sample, (2) operating conditions such as buffer composition and pH, flow rate and sample load and (3) physical properties of the adsorbent matrix.¹⁶ Several mathematical models describe the retention behavior of a protein depending on the relevant separation parameters. The linear gradient elution model developed by Yamamoto delivers data for the prediction of protein elution behavior, as well as information about stationary phase properties and electrostatic and non-electrostatic protein-matrix-interactions.^{17,18} Linear gradient elutions were performed and analysed as described by Ishihara and Yamamoto to obtain information about the strength of protein-matrix-interactions.¹⁶

Other important properties of chromatographic media are adsorption capacities and mass transfer characteristics. Numerous studies deal with the investigation of binding capacities and protein transport and the explanation of adsorption and diffusion

mechanisms inside the chromatographic media using different methods.¹⁹⁻²⁷ In this study, information about protein adsorption and mass transfer was obtained by stirred batch experiments and confocal laser scanning microscopy (CLSM).

For antibody capture in particular, novel Eshmuno™ S tentacle media were developed to provide high capacities and faster protein binding properties. The stationary phase is a surface-grafted, rigid, hydrophilic polyvinylether polymer bead. As the tentacles are highly flexible, the accessibility of the ionic groups without steric hindrance is improved, resulting in tighter binding of antibodies. All tentacles are covalently attached to the polyvinylether backbone and are chemically stable under conditions applied during chromatography, regeneration and sanitization. Binding strength, adsorption capacities and mass transfer rates were investigated to characterize the novel cation exchange resin. Results were discussed and compared to properties of Fractogel® EMD SO₃⁻ (M) and Toyopearl GigaCap S-650M.

Results and Discussion

Binding strength, transport mechanisms and dynamic binding capacities. Linear gradient elutions were performed with mAb01 and the resins Fractogel® EMD SO₃⁻ and Eshmuno™ S at pH 5, 5.5 and 6 to determine the characteristic charge B, which is essentially a measure of the number of sites involved in binding, and A, a measure of the strength of the ionic interaction between the protein and the stationary phase. The resulting GH-I_R-curves are shown in Figure 2 and the calculated parameters B and A are listed in Table 2.

The GH-I_R-plots and the number of charges involved in protein binding B indicate that mAb01 binds more weakly to the functional groups of Eshmuno™ S compared with Fractogel® EMD SO₃⁻. B-values derived from the slope of the GH-I_R-curves with Eshmuno™ S are lower than B-values for Fractogel® EMD SO₃⁻ at the same pH. Furthermore, mAb01 elutes at lower salt concentrations in the case of Eshmuno™ S indicated by the shift of the GH-I_R-curves to more negative log I_R-values and therefore lower I_R-values.

In addition to the binding strength and the elution behavior of the antibody, mass transfer properties of the two resins were examined by stirred batch experiments with mAb01. Uptake-curves were measured at pH 5, 5.5 and 6 with 0.5 mL settled resin in 100 mL protein solution. The initial antibody concentration was 1 mg/mL. Curves in Figure 3 show differences in the rate of protein binding and adsorption capacities of the two resins. Uptake of the antibody with Eshmuno™ S is enhanced compared to the uptake with Fractogel® EMD SO₃⁻. In contrast to Fractogel® EMD SO₃⁻, Eshmuno™ S particles are saturated to 50% after less than 30 min, whereas Fractogel® EMD SO₃⁻ beads needed nearly 50 min for 50% saturation. Furthermore, Eshmuno™ S showed higher maximum binding capacities for mAb01 than Fractogel® EMD SO₃⁻ for all examined pH values (Fig. 4). For Fractogel® EMD SO₃⁻ the maximum binding capacity q₀ was observed at pH 5.5, while q₀ is almost independent of pH for Eshmuno™ S.

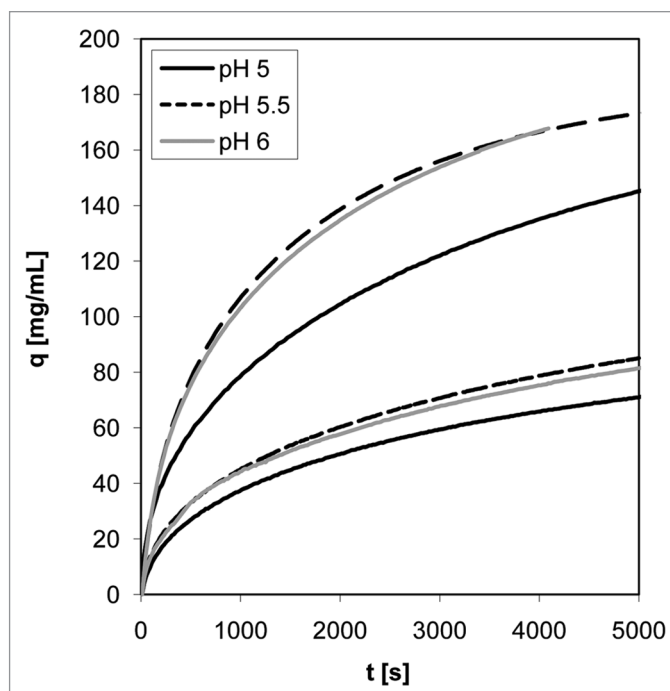


Figure 3. Uptake-curves from stirred batch experiments with mAb01 for Eshmuno™ S and Fractogel® SO₃⁻ (M). The three lower curves belong to Fractogel® SO₃⁻; the three upper curves to Eshmuno™ S. Experiments were performed with 100 mL protein solution and 0.5 mL sedimented resin. The initial protein concentration was 1 mg/mL.

Adsorption capacities fitted from the uptake-curves were compared to DBC determined by breakthrough experiments (Fig. 4). DBC were determined with 1 mL scout columns and a residence time of 2 min. The same buffers as for the batch experiments were used with the sample concentration at 1 mg/mL. The highest DBC was determined at pH 6 for both resins. For Fractogel® EMD SO₃⁻ only a slight difference between the capacities obtained at pH 5.5 and 6 is observed, with the lowest capacity at pH 5. Eshmuno™ S shows almost identical capacities at pH 5 and 5.5.

The maximum binding capacity q₀ for protein adsorption to the stationary phase is determined at equilibrium. Mass transfer rates thus are not a determining factor for q₀. Ligand density can affect protein capacities in different ways. At low ligand density, capacities ascend with increasing density until the ligand spacing is comparable to the diameter of the protein that is adsorbed.³⁰ At ligand densities well above this critical value additional binding sites can have a negative influence on the capacity because proteins will only pack into the pore space to a certain density as electrostatic effects repel adjacent protein molecules. Once that packing density has been achieved, the repulsive forces between the proteins dominate and additional ligands are not beneficial.³¹ To increase the efficiency with which the pore space is utilized, a more efficient distribution of the ligand within the pore space and a higher flexibility regarding binding orientation is beneficial. For the Eshmuno™ S resin the weaker binding of the antibody allows a more flexible and faster re-orientation of the adsorbed mAb at a high protein load.

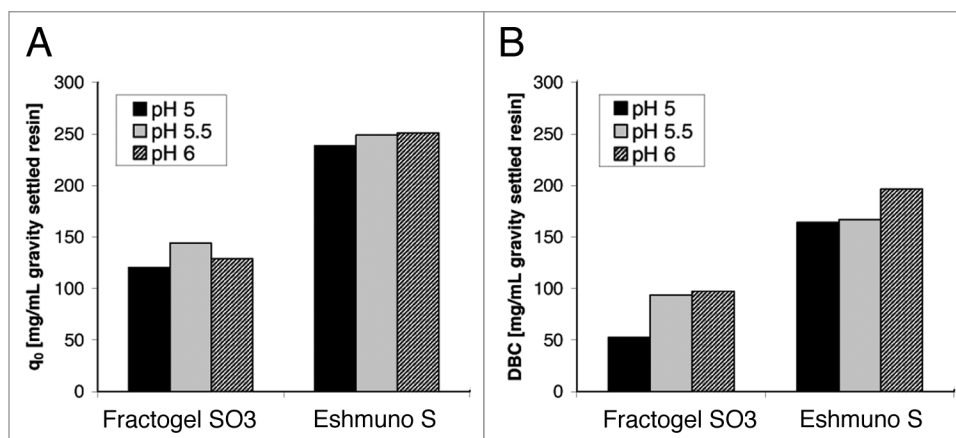


Figure 4. Maximum binding capacities determined by stirred batch experiments (A) and dynamic binding capacities (B) for Fractogel[®] SO₃⁻ and Eshmuno[™] S with mAb01. Stirred batch experiments were performed with 0.5 mL sedimented resin and 100 mL protein solution with an initial protein concentration of 1 mg/mL. 10% breakthrough experiments to determine DBC were performed with 1 mL Scout columns and a residence time of 2 min. The sample protein concentration was 1 mg/mL.

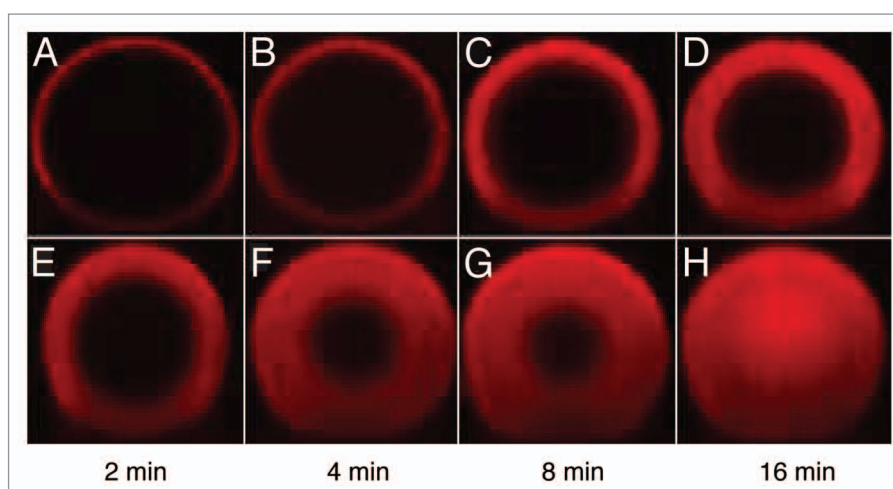


Figure 5. Confocal microscopy images: mAb03 uptake to Fractogel[®] SO₃⁻ (A–D) and Eshmuno[™] S (E–H) during finite bath adsorption at pH 5 and 4 mS/cm. Images recorded at 2, 4, 8 and 16 min illustrate significant differences in mass transfer with Eshmuno[™] S showing much faster uptake kinetics.

DBC is strongly influenced by the time allowed for protein adsorption, i.e., residence time and mass transfer rates. The faster protein uptake of Eshmuno[™] S, as compared with Fractogel[®] EMD SO₃⁻, very likely contributes to the higher DBC of the former. Diffusion of the antibody into the Fractogel[®] EMD SO₃⁻ particles might be hindered by other antibody molecules already bound to the bead. Due to the strong protein-matrix interaction, as found by linear gradient elutions, a blockage of pore's entrances is most likely the reason. If the diffusion of protein into the particle pores is obstructed, mass transfer rates are decelerated and DBC are decreased. Weaker adsorption of the antibody to the stationary phase, as observed for Eshmuno[™] S, might allow bound protein to detach and rebind, and therefore might prevent diffusional hindrance. Thus, to some extent weaker protein-matrix interactions can be attended by better protein transport characteristics and better dispersion of the protein across the whole particle, resulting in faster mass transport and

higher binding capacities as found for Eshmuno[™] S when compared with Fractogel[®] EMD SO₃⁻.

These considerations are supported by results from confocal laser scanning microscopy (Fig. 5). Examinations showed differences in the dispersion of mAb03 across Fractogel[®] EMD SO₃⁻ (M) and Eshmuno[™] S particles. In these experiments, mAb03 was labeled to visualize the movement of the antibodies across the particle. On Fractogel[®] EMD SO₃⁻ beads, the front of labelled protein showed a clear boundary while the labelled antibodies formed a diffuse ring on Eshmuno[™] S particles. This indicates weaker binding for Eshmuno[™] S; the proteins seem to stick to the particle surface of Fractogel[®] EMD SO₃⁻ due to the stronger interaction. Further, in the case of Eshmuno[™] S, mAb03 is spread over the particle much faster compared with Fractogel[®] EMD SO₃⁻. This indicates differences in the transport mechanisms of Fractogel[®] EMD SO₃⁻ and Eshmuno[™] S that seem to correlate with the strength of

Table 1. Properties of strong cation exchangers

Trade name	Base matrix	Mean particle size (μm)	Ion capacity ($\mu\text{equ/mL}$)	Lysozyme dynamic binding capacity at 10% breakthrough (mg/mL settled resin)	Polyclonal hlgG dynamic binding capacity at 10% breakthrough (mg/mL settled resin)
Eshmuno™ S	cross-linked hydrophilic vinyl ether with "tentacle" polymer functional groups	85	50–100	100 ¹	80 ²
Fractogel® EMD SO ₃ ⁻ (M)	Methacrylate with "tentacle" polymer functional groups	65	70–110	90 ³	60 ²
Toyopearl® GigaCap S-650 M	Methacrylate with proprietary polymer functional groups	75	100–200	167 ⁴	100 ²

¹residence time 1 min; pH 7.0. ²residence time 2 min; pH 5.0. ³residence time 2 min; pH 7.0. ⁴residence time 0.85 min; pH 4.7.²⁸

protein-matrix-interactions examined by linear gradient elutions. As with the results from LGE for mAb01, the ionic strength at elution of mAb03 was higher for Fractogel® EMD SO₃⁻ compared with Eshmuno™ S (results not shown).

Also, the faster uptake kinetics for Eshmuno™ S resin has an influence on the dependency of DBC on residence time. DBC of Toyopearl GigaCap S-650M, Fractogel® EMD SO₃⁻ and Eshmuno™ S for mAb03 at 2 and 5 min residence times are shown in **Figure 6**. All resins show an increased DBC with lower flow rates, i.e., longer residence times. The capacities obtained for Eshmuno™ S are about 1.5 times higher than those obtained for Toyopearl GigaCap S-650M, which was reported as a material for high-capacity antibody capture.⁹ This increased DBC is also observed at 2 min residence time and allows high-capacity purification of the antibody at high linear flow rate (445 cm/h for 15 cm column height).

For a number of ion exchange resins unexpected behavior of the DBC for antibodies on pH values and conductivity was observed.^{31–33} In these materials, dextran surface extenders lead to an exclusion mechanism due to strong antibody binding to the outer part of the resin beads and exclusion of additional antibody from entering the bead. Weakening of the interaction by increasing conductivity and pH leads to an increase in DBC until a maximum is achieved, where DBC decreases with increasing conductivity and pH.

DBC for Eshmuno™ S and mAb03 depending on pH and conductivity are shown in **Figure 7**. For all pH values, the DBC at 4 mS/cm is almost the same, except at pH 5.75 which shows a slightly lower capacity. At the lowest pH (4.5), almost no dependency on the conductivity was found and high-capacity capture of antibodies at 12 mS/cm is possible. With increasing pH, the influence of the salt concentration in the buffer increases, although the differences between DBC at 4 or 6 mS/cm are rather marginal at pH 5.0 and 5.25. This indicates robustness of the process with respect to buffer ionic strength at low pH values. At pH values above pH 5.0, classical ion exchange behavior is observed and exclusion and pore blockage mechanisms do not play a significant role.

Antibody capture with high capacity resins. In addition to high binding capacities for pure antibodies, Eshmuno™ S has higher capacities than the other tested resins for capture of mAbs from diluted cell culture supernatant (**Table 3**). Obviously

Table 2. B and A values determined from GH-I_n-plots with mAb01 for Fractogel SO₃⁻ and Eshmuno S

Resin	pH 5		pH 5.5		pH 6	
	B	A	B	A	B	A
Fractogel SO ₃ ⁻	13.37	6.77E-09	9.22	8.78E-08	13.39	1.78E-11
Eshmuno S	9.25	3.76E-07	5.43	3.18E-05	7.27	5.48E-07

binding capacity data obtained with purified antibodies do not necessarily reflect binding properties determined using complex samples like clarified cell culture harvests.

The chromatogram for direct capture of mAb02 from diluted cell culture supernatant on Eshmuno™ S is shown in **Figure 8**. The column was slightly overloaded (5% breakthrough) and 107 mg protein per mL of resin was applied to the column. The antibody was eluted with a linear salt gradient in a narrow peak. The recovery rate for the antibody was 99% and the yield 94%.

For the purification of 10 kg antibody (2,000 L bioreactor, 5 g/L titer) a protein A capture column of 250 L (US \$250,000 resin costs, DBC 40 g/L, US \$10,000 per liter) is required, while the same quantity of antibody can be bound to 100 L of high capacity ion exchange resins (less than 1/10 of the protein resin costs, DBC 100 g/L). Additional cost savings can be achieved due to shorter process times and a longer resin life span.

To evaluate the purity of the mAb, the eluate peak was pooled and analyzed by SDS-PAGE using a NuPAGE 4–12% BisTris gel with MES running buffer (**Fig. 8**). The protein bands were stained with Coomassie. Only traces of intact antibody could be detected in the flowthrough fractions, and the amount of impurities seen in the load and flowthrough is significantly reduced in the eluate pool.

The collected fractions were further analyzed by SEC and ELISA. The total amount of hcp in fraction E1 to E9 and the antibody concentration is shown in **Figure 9**. The amount of hcp/mg of eluted antibody increases notably in later fractions, indicating that hcp and antibody are not co-eluting. The overall hcp reduction factors are summarized in **Table 4**.

When used as a capture step Eshmuno™ S provides 150 fold hcp removal, which is three times more than Fractogel® EMD SO₃⁻ and 25% more than Toyopearl GigaCapS-650M for

Table 3. Dynamic binding capacities¹ of Toyopearl GigaCap S, Fractogel SO₃⁻ and Eshmuno S for the capture of mAbs from diluted cell culture supernatant (CCS)

	mAb03 CCS (mg/mL resin)	mAb03 purified (mg/mL resin)	mAb02 CCS (mg/mL resin)	mAb02 purified (mg/mL resin)
Eshmuno S	89	90	96	109
Fractogel SO ₃ ⁻	69	77	70	76
Toyopearl GigaCap S	64	83	64	88

¹Capacities were determined by breakthrough analysis with buffer containing 20 mM phosphate and 20 mM NaCl (conductivity 4.3 mS/cm). Experiments were performed at a residence time of 5 min (mAb02; 0.62 mg/mL; 5% breakthrough) or 2 min (mAb03; 1.5 mg/mL; 10% breakthrough) at pH 6 or pH 5.5, respectively.

Table 4. Removal of hcps during capture and post protein A purification using different cation exchangers¹

	Toyopearl GigaCap S	Fractogel SO ₃ ⁻	Eshmuno S
hcp clearance factor (capture)	121 ² 28 ⁴	52 ² 108 ⁴	150 ² 77 ⁴
hcp clearance factor (post Protein A)	n.d.	3.7 ³	5 ³

¹for experimental details see Table 3. ²mAb02; ³mAb03; ⁴mAb04.

mAb02. On the other hand, results obtained for mAb04 indicate better removal of hcps when using Fractogel[®] EMD SO₃⁻ compared to Eshmuno[™] S (77 fold compared to 108 fold). This might be caused by the higher total mAb load for Eshmuno[™] S compared to Fractogel[®] EMD SO₃⁻.

For the post protein A pool, Eshmuno[™] S reduces hcp levels about 5- to 7-fold, whereas Fractogel[®] EMD SO₃⁻ shows nearly 4-fold clearance at 4 mS/cm and pH 5.5. There was no difference in removal between experiments performed at 2 min or 5 min residence time (data not shown).

Removal of hcp can vary quite widely, and experiments have indicated that these variations are linked to differences in surface chemistry of the chromatographic resin, as well as cell culture and harvest conditions.⁵ Data presented here indicate a contribution to hcp removal in a range that can be normally achieved by a cation-exchange chromatography step; however, the influence of the post load wash conditions were not studied, and thus further investigations are required.

Mechanical stability of Eshmuno[™] S. In order to save time and effort later in development, large scale manufacturing considerations should be taken into account at the beginning of process development. Compared to packing on a laboratory scale, larger column diameters can become problematical due to increasing back pressures. The best resin and separation conditions must be identified and additionally a test has to be conducted to see whether the chosen chromatographic media can be operated at the desired flow rates independent of column sizes. Scale-up of ion-exchange chromatography operations is usually achieved by increasing the column diameter while keeping the resin bed height and linear flow rate constant. This ensures that the residence time is always the same.

Since pressure drop across chromatography beds employing soft media can be a significant problem in the operation of

large-scale preparative chromatography columns, the mechanical stability of Eshmuno[™] S was examined in 10 and 76 cm i.d. columns. With respect to rigidity and compressibility, vinyl ether-based Eshmuno[™] S behaved similarly to the semi-rigid poly-methacrylate-based resins Toyopearl and Fractogel[®] EMD SO₃⁻. In order to achieve a properly packed column, the percent compression for Eshmuno[™] S is in the range of 8–10.5% with respect to settled resin. The linear relationship between back-pressure and velocity is shown in **Figure 10**. Pressure-flow profiles for 10.5% are in a linear range up to flow rates of about 800 cm/h, experimental data using the 76 cm i.d. column were limited by the maximum flow rate of the pump at 420 cm/h. Due to linear pressure versus flow curves for Eshmuno[™] S, pressure drops and flow velocities can be predicted for industrial-scale columns by just a few experiments on a laboratory scale.

Materials and Methods

Resins and columns. The strong cation exchangers Fractogel[®] EMD SO₃⁻ (M), Eshmuno[™] S (Merck KGaA Darmstadt) and Toyopearl GigaCap S-650 (Tosoh Bioscience GmbH) were used in this study. Resins were packed in 1 mL Scout Columns (Merck KGaA Darmstadt, Germany), Superformance columns (Götec-Labortechnik GmbH, Germany) or MediaScout[®] MiniChrom columns (Atoll, Germany). Properties of the stationary phases used in this work are summarized in **Table 1**.

Buffers and samples. Different buffers were used for the experimental work: phosphate/acetate buffer consisting of 21.5 mM NaH₂PO₄, 3.4 mM Na₂HPO₄, 16.5 mM sodium acetate and 0.48 g acetic acid for pH 5 and 5.5, 30 mM Na-citrate buffer at pH 6 and buffer containing 20 mM NaH₂PO₄ for pH values between 4.5 and 6. PH values were adjusted with NaOH or HCl. All buffers were prepared both without sodium chloride (buffer A) as described above and with additional 1 M NaCl (buffer B). Salt concentrations between 0 and 1 M NaCl could be prepared by mixing these two buffers.

Experiments were performed with the monoclonal antibodies mAb01, mAb02 and mAb03, which were either diluted in buffer or diafiltered.

Chromatography systems. Experiments were carried out with the liquid chromatography systems ÄKTA UPC 100, ÄKTApurifier 100 or ÄKTApurifier UPC 100 (GE Healthcare). A 76 cm i.d. Eastern Rivers, Inc. column equipped with hydraulic pistons to lower the head was used for measuring pressure versus flow curves.

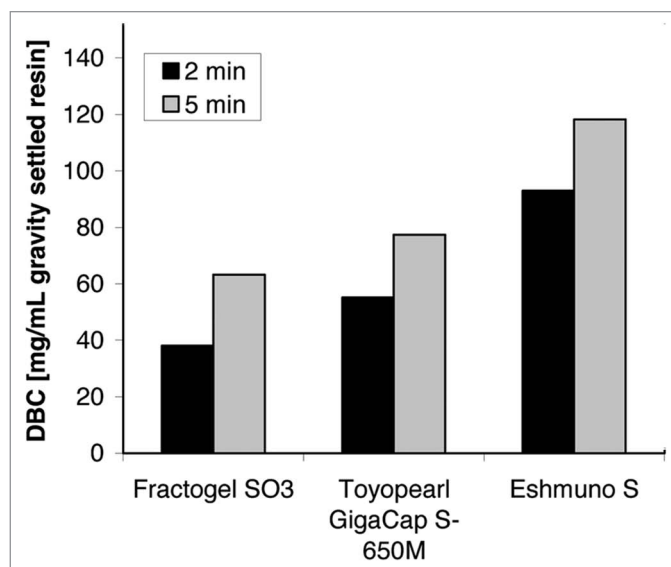


Figure 6. Dynamic binding capacities of Toyopearl GigaCap S, Fractogel[®] SO₃⁻ (M) and Eshmuno[™] S for mAb03 in dependence on the residence time. Capacities were determined at 10% breakthrough with buffer pH 5.5 containing 20 mM phosphate and NaCl (conductivity 4 mS/cm). Experiments were performed with an antibody concentration of 3 mg/mL and MediaScout[®] MiniChrom columns (50 mm L x 8 mm i.d.).

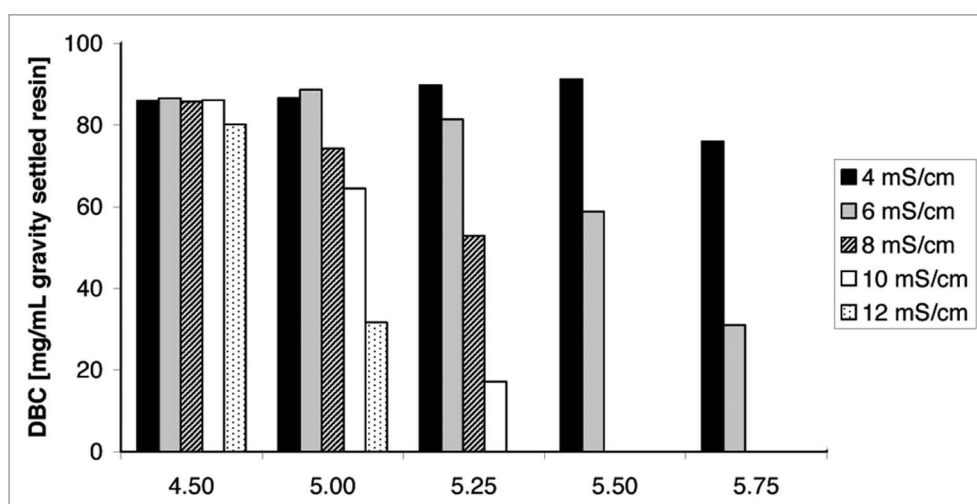


Figure 7. Dynamic binding capacities of mAb03 on Eshmuno[™] S for different pH values and conductivities. Capacities were determined at 10% breakthrough with buffers containing 20 mM phosphate and NaCl. Experiments were performed at a residence time of 2 min, and an antibody concentration of 3 mg/mL on a Superformance column (50 mm L x 10 mm i.d.).

Determination of dynamic binding capacities. Dynamic binding capacities (DBC) for different antibodies and resins were determined by breakthrough curves. Proteins were diluted with binding buffer and applied on the column until either 5 or 10% breakthrough was reached. After removal of unbound protein by washing, bound protein was eluted by increasing the salt concentration. DBC were determined with varying residence time, pH value, and conductivity of the binding buffer.

Linear gradient elutions. Linear gradient elutions to determine GH-I_R-plots were performed with mAb01 and mAb03 and the cation exchangers Fractogel[®] EMD SO₃⁻ (M) and Eshmuno[™] S. The antibodies were diluted in buffer A and applied to the columns equilibrated with the same buffer to a protein load of

0.44 mg/mL of packed resin. After a washing step a linear salt gradient from buffer A to buffer B was applied. The lengths of the gradients were in the range of 15–90 column volumes. The linear flow rate F/A_c was 119 cm/h. Gradient elutions were performed at pH 5.0, 5.5 and 6.0.

The peak position was determined by fitting the elution curve with an EMG-Fit using the program TableCurve 2D (Systat Software Inc.). The ionic strength I_R [M] was calculated from the conductivity at this position. GH-I_R-curves were determined as shown by Ishihara and Yamamoto.¹⁶

Stirred batch experiments. Uptake curves for Fractogel[®] EMD SO₃⁻ (M) and Eshmuno[™] S with mAb01 were obtained by stirred batch experiments. In this method, gel particles are suspended in

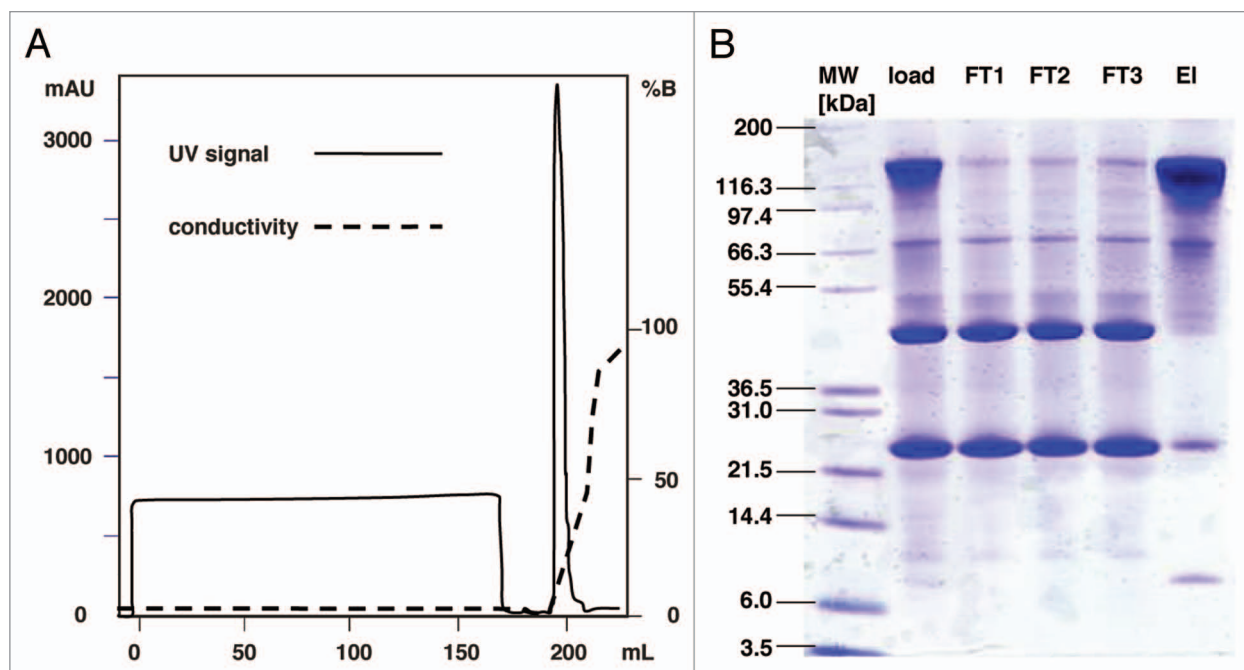


Figure 8. Chromatogram of direct capture of mAb02 (0.62 mg/mL) from diluted cell culture supernatant on Eshmuno™ S (A). A 1 mL Scout column was loaded to 5% breakthrough. Experiments were performed with binding buffer pH 6 containing 20 mM phosphate and 20 mM NaCl (conductivity 4.3 mS/cm) at a residence time of 5 min. Bound protein was eluted by a linear gradient from binding buffer to elution buffer containing 20 mM phosphate and 0.8 M NaCl in 20 column volumes. Sample, flowthrough fractions (FT) and the pooled elution fractions (EI) were analysed by SDS page under non-reducing conditions (B).

a vessel containing the protein solution. The amount of bound protein is determined by recirculation of the protein solution through a UV-detector (Fig. 1).

The gel particles were suspended by agitation with a stirrer and held back in the vessel with a frit while protein solution was continuously recirculated through a UV flow cell back into the vessel with a peristaltic pump. Experiments were performed by injection of 0.5 mL sedimented resin (V_M) in 100 mL (V) antibody solution with a protein concentration C_0 of 1 mg/mL. The absorbance at 280 nm was measured to determine the concentration of unbound protein C and thereby calculate the amount of bound protein per mL resin q with the following equation:

$$q = [V/V_M] (C_0 - C) \quad (1)$$

The maximum adsorption capacities based on the volume of resin q_0 were fitted from the uptake-curves with equations able to describe mass transfer processes in stationary phases.¹⁹⁻²¹

Confocal laser scanning microscopy. Confocal laser scanning microscopy (CLSM) using a mAb labelled with ALEXA® Fluor 546 (Molecular Probes, Eugene Oregon, USA) was conducted as described by Stanislawski.²⁹ Fluorophore-labelled mAb03 was scanned under Leica CLSM TCS SL using oil immersion objective (40 fold magnification).

Protein analysis. SDS-PAGE was performed using NuPAGE 4–12% BisTris gels (Invitrogen, Carlsbad, CA, USA) run with MES buffer according to the manufacturer's instructions and

visualized with Coomassie staining. Mark12 Unstained Standard (Invitrogen) was used as protein standard.

Analytical size-exclusion chromatography (SEC) was conducted using a G3000SWXL column (Tosoh Bioscience GmbH) with a GFC-3000 guard cartridge (Phenomenex). The mobile phase was a 25 mM $\text{NaH}_2\text{PO}_4 \times \text{H}_2\text{O}$ buffer solution (pH adjusted to 7.0 with 5 M NaOH) containing 150 mM Na_2SO_4 and 0.05% NaN_3 . The column was operated isocratically at a constant flow rate (1 mL/min) using a LaChromHPLCsystem (Merck-Hitachi, Darmstadt, Germany).

Hcp were detected using commercial microtitre plate ELISA methods provided by Cygnus Technologies, NC, USA according to the manual.

Conclusions

Cation-exchange chromatography is currently part of all mAb manufacturing processes and has been proven to give high step yields and to reduce the level of hcp very effectively. Its potential as capture step is still underestimated, although cost savings can be realized through use of ion-exchange chromatography as the first column step. However, detailed knowledge of the relevant resin's properties and optimal process parameters is required.

Therefore many parameters, including binding strength, adsorption capacities, mass transfer rates as well as the influence of salt concentrations, pH, and flow rate have to be assessed to evaluate the strengths and weaknesses of ion exchange resins.

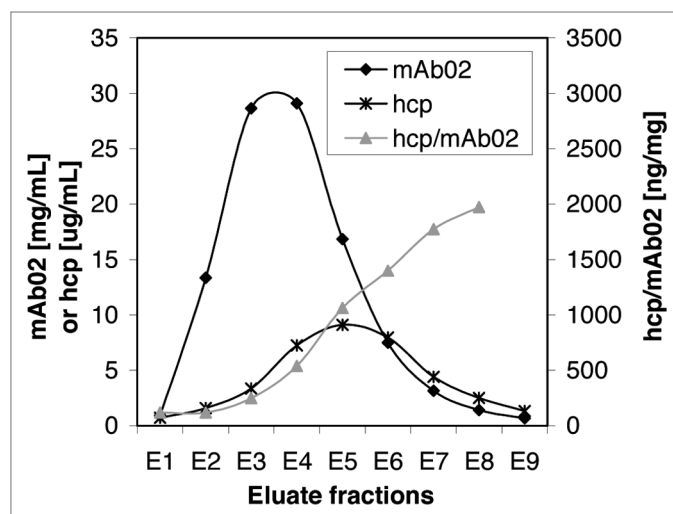


Figure 9. mAb02 and hcp concentrations in the eluate fractions. The antibody concentration was determined by analytical SEC. Hcp concentration was measured with an ELISA assay specific for the host cells. Most of the hcp elute after the main antibody peak is desorbed from the column.

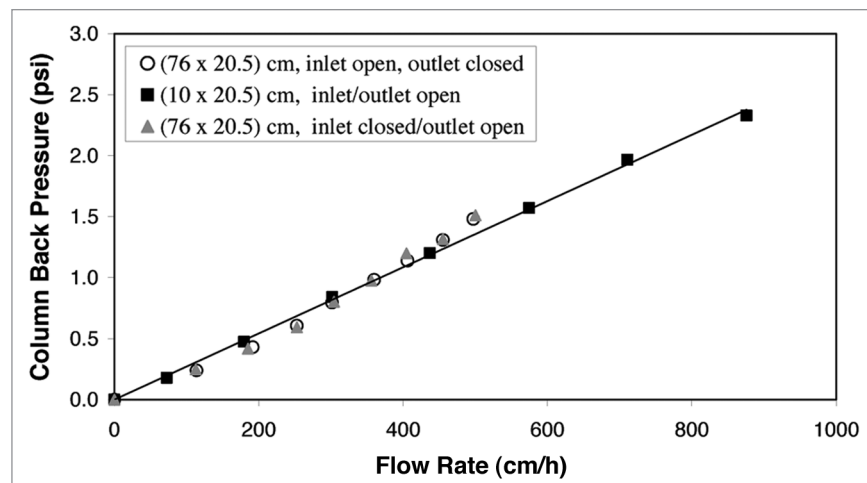


Figure 10. Pressure-flow-data for Eshmuno™ S. Eshmuno™ S was flow packed into columns of 20.5 cm bed height in 0.01 M NaOH with 10.5% bed compression. The resin showed similar pressure-flow curves with 10 cm or 76 cm i.d. column.

In this study higher binding capacities were determined for Eshmuno™ S compared with the other tested resins Fractogel® EMD SO₃⁻ and Toyopearl GigaCap S-650M. Increased DBC were obtained for different mAbs, pH-values and conductivities. Fast mass transfer as found by stirred batch experiments allowed high DBC even at high flow rates.

Superior DBC of Eshmuno™ S obtained for pure mAbs were also found for direct capture of mAbs from cell culture supernatant. The removal of hcp was in the same range as for Fractogel® EMD SO₃⁻ and Toyopearl GigaCap S-650M. Results of this study show that the increase in binding capacity is not at the expense of specificity, and indicate the applicability of this novel cation-exchanger for high capacity antibody capture.

With respect to production processes the excellent pressure flow behavior of Eshmuno™ S allows the resin to be operated

in process scale columns. Fast mass transfer rates enable small elution volumes, which directly influences the duration and performance of a chromatographic step. The results achieved in this study showed that novel cation-exchange resins can be implemented as an alternative antibody capture step. It combines high capacities, yields and the ability to significantly reduce host cell impurities and allows high flow rates, and therefore high throughput.

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