

## Isolation and Characterization of Chromium(VI)-Reducing Bacteria from Tannery Effluents

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**Abstract** Two chromium-resistant bacteria (IFR-2 and IFR-3) capable of reducing/transforming Cr(VI) to Cr(III) were isolated from tannery effluents. Isolates IFR-2 and IFR-3 were identified as *Staphylococcus aureus* and *Pediococcus pentosaceus* respectively by 16S rRNA gene sequence analyses. Both isolates can grow well on 2,000 mg/l Cr(VI) (as  $K_2Cr_2O_7$ ) in Luria-Bertani (LB) medium. Reduction of Cr(VI) was found to be growth-associated in both isolates and IFR-2 and IFR-3 reduced 20 mg/l Cr(VI) completely in 6 and 24 h respectively. The Cr(VI) reduction due to chromate reductase activity was detected in the culture supernatant and cell lysate but not at all in the cell extract supernatant of both isolates. Whole cells of IFR-2 and IFR-3 converted 24 and 30% of the initial Cr(VI) concentration (1 mg/l) in 45 min respectively at 37°C.  $NiCl_2$  stimulated the growth of IFR-2 whereas  $HgCl_2$  and  $CdCl_2$  significantly inhibited the growth of both isolates. Optimum temperature and pH for growth of and Cr(VI) reduction by both isolates were found to be between 35 and 40°C and pH 7.0 to 8.0. The two bacterial isolates can be good candidates for detoxification of Cr(VI) in industrial effluents.

**Keywords** Bioremediation · Chromium · Tannery effluents

### Introduction

Chromium is used in different industrial processes and released into the environments. Leather processing industries (LPI) use chromium material (chrome liqueur or chrome powder) for tanning of leather. Residual chromium thus is discharged in solid or liquid effluents. Soluble hexavalent chromium [Cr(VI)] is extremely toxic and shows mutagenic and carcinogenic effects on biological systems due to its strong oxidizing nature [1]. Cr(III) is less toxic and bioavailable than Cr(VI), as it readily forms insoluble oxides and hydroxides above pH 5 [2].

In Bangladesh, about 250 leather processing industries at Hazaribagh, South Western part of the Dhaka city, discharge both liquid and solid wastes into canals and rivers with residual chromium. This causes chromium contamination not only of water but also aquatic lives, land, vegetables farming and crops thus posing a serious health hazard. The most direct and severe effect comes out with the leather-cut wastes, which are conventionally being processed into feed ingredient (as a protein source). This chromium (tetravalent or hexavalent form) content produce thus might be a good source of chromium migration into the food chain.

A wide variety of bacteria have been reported for reducing/transforming Cr(VI) to Cr(III), under aerobic and anaerobic condition, e.g. *Intrasporangium* Sp. Q5-1, *Bacillus* sp. ES29, *Escherichia coli*, *Enterobacter cloacae*, *Pseudomonas fluorescens* LB300 [3–7]. Application of Cr-resistant bacteria for detoxification of Cr(VI) has been considered as an economical, effective and safe procedure over conventional physical and chemical methods [8].

We have isolated two chromium resistant bacteria from tannery effluents collected from drain water nearby

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tannery industries at Hazaribagh area. These bacteria are capable of reducing Cr(VI) to Cr(III). We are working to develop a bioremediation process for treatment of chromium-polluted tannery effluents as well as leather-cut wastes for conversion of the latter into a safe poultry feed ingredient.

## Materials and Methods

### Sample Collection and Processing

Tannery effluent/drain water was collected from Hazaribagh Tannery area of Dhaka city in sterile bottles and immediately transferred to the laboratory. Samples were diluted in normal saline and inoculated (0.1 ml) on Luria-Bertani (LB) agar plate containing 500 mg/l of potassium dichromate ( $K_2Cr_2O_7$ ) as Cr(VI) by spread plate method. Plates were incubated at 37°C for 4 days. Useful single colonies were selected and sub-cultured onto fresh LB agar medium and preserved at 4°C or in 10% glycerol stock for further studies.

### Identification by 16S rRNA Gene Sequence

Genomic DNA of bacterial isolates were isolated according to the method described by Sambrook et al. [9]. Gene fragments specific for the highly variable V3 region of the bacterial 16S rRNA gene was amplified by PCR as described by Ovreås et al. [10] using universal bacterial primers L340GCF and K517R (Sigma, USA) in a thermal cycler (MJ Research Inc., Watertown, USA). The sequence of the L340GCF primer, which included a 40 base pair GC-clamp attached to the 5'-end, was 5'-CGCCCCGCCGC CGCGCGGGCGGGCGGGCACGGGGGGACTCCTA CGGGAGGCAGCAG-3' and the sequence of the K517R primer was 5'-ATTACCGCGGCTGCTGG-3'. The PCR products were subjected to 2% agarose gel electrophoresis, stained with ethidium bromide and visualized on a UV transilluminator for the presence of about 200 bp PCR products. Amplified 16S rRNA gene PCR products were purified using StrataPrep® PCR purification kit (Stratagene, USA) according to the manufacturers protocol. Sequencing reactions were carried out using ABI-Prism Big dye terminator cycle sequencing ready reaction kit and the PCR products were purified by a standard protocol. The purified cycle sequenced products were analyzed with an ABI-Prism 310 genetic analyzer. The chromatogram sequencing files were edited using Chromas 2.32. The homology of the 16S rRNA gene sequences was checked with the 16S rRNA gene sequences of other organisms that had already been submitted to GenBank database using the BLASTN [11] (<http://www.ncbi.nih.gov/BLAST/>) algorithm.

### Effect of Chromium(VI) Concentration on Growth

The effect of varying concentrations (0–2,000 mg/l) of Cr(VI) (as  $K_2Cr_2O_7$ ) on the tolerance of the microbial growth was examined in duplicate in LB broth (5 ml) at 37°C under 150 rpm in a water bath (N-Biotek, Inc. Korea). The growth was monitored measuring OD at 600 nm.

### Chromium(VI) Assay

Hexavalent chromium was determined colorimetrically with a spectrophotometer (Genesys 5, USA) using the S-diphenylcarbazide (DPC) (Nacalai Tesque, Inc., Japan) method [12]. The DPC reagent was prepared by adding 24 ml of 85%  $H_3PO_4$  to 56 ml distilled water. This solution was mixed with 0.076 g DPC previously dissolved in 20 ml of 95% ethanol. The reagent was stored in dark at 4°C. Cr(VI) in the sample was assayed by adding 125 µl of the DPC reagent to 1 ml of chromium samples, mixed gently and kept at room temperature for 20 min. The absorbance of the color produced was measured at 540 nm using a spectrophotometer. Cr(VI) concentration in the sample was calculated from a standard curve using  $K_2Cr_2O_7$  as standard.

### Determination of Chromate Reductase Activity

The reaction mixture for the chromate reductase activity contained 2 mg/l Cr(VI) as  $K_2Cr_2O_7$  in 0.5 ml of 100 mM phosphate buffer, pH 7.0 held at 37°C in a water bath. The reaction was initiated by the addition of 0.5 ml of the enzyme (as culture supernatant, cell extract supernatant or cell lysate) and residual Cr(VI) concentration was measured after 1 h following the method described above [12]. One unit of enzyme activity was defined as 1 µmol of Cr(VI) reduced/min/ml at 37°C.

### Time Course of Growth and Cr(VI) Reduction

The time course of growth and Cr(VI) reduction/transformation was evaluated with 20 mg/l  $K_2Cr_2O_7$  in shake flask culture (N-Biotek, Inc. Korea) containing 100 ml LB broth at 37°C, 150 rpm. Samples (1.5 ml) were withdrawn at 0.5 or 2 h intervals. The growth was monitored measuring OD at 600 nm. Samples were centrifuged (Eppendorf, Germany) at 10,000 rpm for 5 min and the supernatants were assayed for residual Cr(VI) concentration.

### Cell Fractionation for Localization of Cr(VI) Reduction and Chromate Reductase Activity

Bacteria were grown overnight in 100 ml of LB medium with 20 mg/l  $K_2Cr_2O_7$  at 37°C with orbital shaking

(150 rpm). Thereafter, cells of each isolate were harvested by centrifugation (Biofuge Primo, Heraeus, Germany) of 30 ml culture at 5,000 rpm for 10 min. Culture supernatant was collected and the cell pellet was resuspended in 30 ml phosphate buffer (100 mM, pH 7.0). Cells in an ice bath were disrupted with an ultrasonic probe (550 Sonic Dismembranator, Fisher Scientific, Pittsburgh). Power was applied ten times in 30 s pulses with 30 s intervals. The sonicate was centrifuged at 16,000 rpm at 4°C for 20 min. Cell extract supernatant was transferred in a fresh tube and was kept in ice. Cell lysate was also resuspended in 30 ml phosphate buffer. Total chromium and hexavalent chromium in three fractions (culture supernatant, cell extract supernatant and cell lysate) were determined by Graphite Furnace Atomic Absorption (GFAA) spectrometer (AAAnalyst 800, Perkin Elmer, USA) and diphenylcarbazide method [12], respectively.

#### Chromium Reduction by Whole Cell in Buffer

Bacterial cells were grown in 50 ml LB broth in a 250 ml Erlenmeyer flask. After 20 h of incubation at 37°C and 150 rpm agitation, cells were harvested from 1.5 ml culture by centrifugation at 10,000 rpm for 5 min. The cell pellet in an eppendorf tube was washed twice with phosphate buffer and resuspended in 0.5 ml phosphate buffer (100 mM, pH 7.0). Cr(VI) reduction was initiated by adding 0.5 ml of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> (2 mg/l) in phosphate buffer and incubated at 37°C in a water bath for 45 min. The reaction mixture was centrifuged at 10,000 rpm for 5 min and the supernatant was analysed for residual Cr(VI) determination by the method as described above. A reaction blank was prepared by adding 0.5 ml phosphate buffer to 0.5 ml of K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> (2 mg/l) and considered as 100% Cr(VI). Phosphate buffer was used to set the spectrophotometer to zero.

#### Effect of Different Metals on the Growth of Bacterial Isolates

The effect of different salts on the growth of bacterial isolates was examined in LB broth (5 ml) supplemented with 200 mg/l of CuCl<sub>2</sub>, ZnCl<sub>2</sub>, NiCl<sub>2</sub>, CdSO<sub>4</sub>, and HgCl<sub>2</sub>. The tubes were inoculated with 25 µl of the inoculum and incubated at 37°C and 150 strokes/min in a water bath for 20 h. The growth of the bacterial isolates was measured spectrophotometrically at 600 nm.

#### Effect of Temperature and pH on the Growth and Cr(VI) Reduction

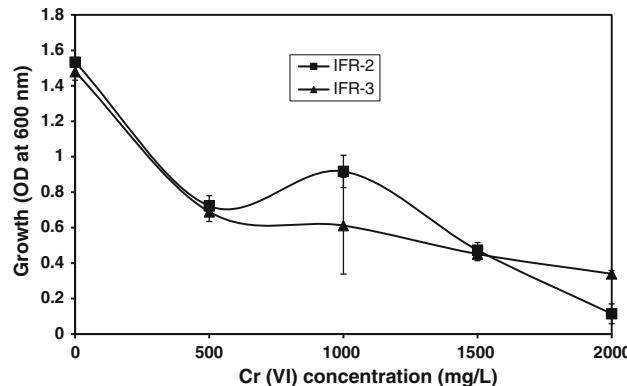
The effect of temperature and pH on the growth and Cr(VI) reduction by bacterial isolates were investigated using LB medium (5 ml) containing 10 mg/l Cr(VI) as K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub>.

Media were inoculated with 25 µl of overnight cultures of the bacterial isolates (OD<sub>600</sub> 1.2–1.7). Chromium reduction and growth was studied at various incubation temperatures (25–45°C). For the effect of pH, sterilized culture medium was adjusted to pH 5.0, 6.0, 7.0, 8.0, 9.0 and 10.0 with predetermined amounts of sterilized 0.1 N HCl or 1 N NaOH and the tubes were incubated at 37°C in a water bath. Growth was measured after 24 h of incubation by taking absorbance at 600 nm. The culture was centrifuged (10,000 rpm, 5 min) and the supernatant was used to determine the residual Cr(VI) concentration.

#### Results and Discussion

Eight unique colonies were selected from Cr(VI) containing LB plates, inoculated with effluent/drain water collected from Hazaribagh Tannery area, Dhaka. These isolates were further inoculated in LB broth supplemented with different concentrations of Cr(VI) to determine their Cr(VI) tolerance between 500 and 2,000 mg/l. Isolates IFR-2 and IFR-3 showed highest level of tolerance, growing well in presence of 2,000 mg/l Cr(VI) (Fig. 1) whereas other isolates did not grow well above 1,000 mg/l Cr(VI). Therefore isolates IFR-2 and IFR-3 were selected for further studies. Camargo et al. [4] isolated some chromium resistant bacteria that can tolerate or reduce Cr(VI) at concentrations of 1,500–2,500 mg/l.

The isolates IFR-2 and IFR-3 were identified by both morphological and the 16S rRNA gene sequence analyses technique (Table 1). Isolate IFR-2 showed 98% identity with *Staphylococcus aureus* subsp. *aureus* and isolate IFR-3 was 100% identical to *Pediococcus pentosaceus* ATCC 25745. According to the scientific literatures, these species of bacteria have never been reported as chromium-resistant bacteria.



**Fig. 1** Effect of Cr(VI) as K<sub>2</sub>Cr<sub>2</sub>O<sub>7</sub> concentrations on the growth of bacterial isolates (IFR-2, IFR-3)

**Table 1** Identification of chromium(VI)-reducing bacteria by 16S rRNA gene sequence analyses

Isolate	Organism	Accession no.	E value	Identity (%)
IFR-2	<i>S. aureus</i> subsp. <i>aureus</i>	NC_009641.1	1e <sup>-66</sup>	98
IFR-3	<i>P. pentosaceus</i> ATCC 25745	NC_008525.1	2e <sup>-69</sup>	100

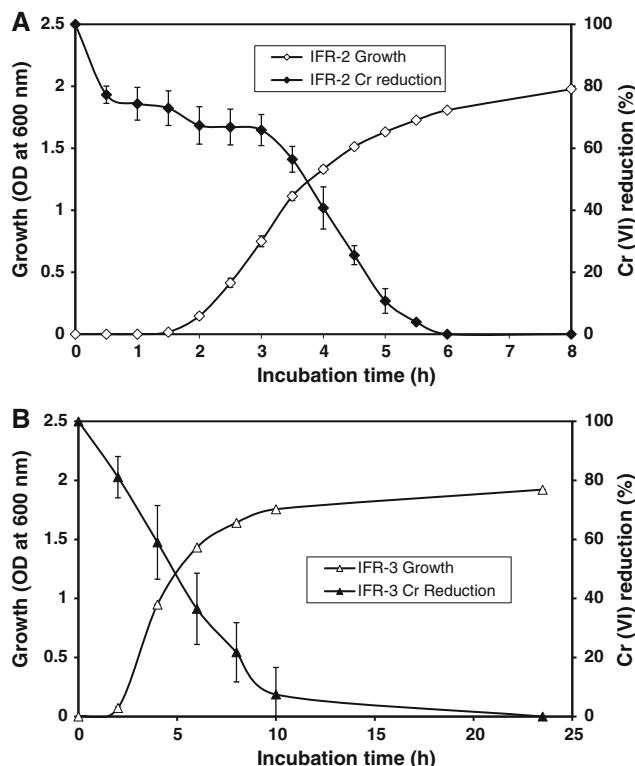
Time courses of growth and Cr(VI) reduction or detoxification by chromate reductase of isolates IFR-2 and IFR-3 were investigated in presence of an initial Cr(VI) concentration of 20 mg/l in shake flask culture. Cr(VI) reduction by both isolates were found to be growth-associated (Fig. 2). Isolate IFR-2 reduced 90% of Cr(VI) in 5 h (Fig. 2A) while IFR-3 reduced 90% of Cr(VI) in 10 h of incubation (Fig. 2B). Cr(VI) was completely reduced after 6 and 24 h respectively by isolates IFR-2 and IFR-3. It has been reported that *Bacillus* Sp. ES29 reduced 90% of Cr(VI) added (2 mg/l) to the medium in less than 6 h [13]. Thacker and Madamwar [14] have shown that the bacterial isolate DM1 reduces 50 ppm of chromium to 0 ppm in 54 h. They added a second aliquot of chromium which was reduced to 0 ppm in 99 h, and the third aliquot was reduced to 21.8 ppm in 126 h.

In the culture supernatants of final samples of the above experiment, Cr(VI) added in the culture (20 mg/l) medium was not detected at all (Table 2) while AAS results

revealed a total Cr of 19.81 ( $\pm 0.37$ ) and 19.27 ( $\pm 0.33$ ) mg/l respectively in the culture supernatants of isolates IFR-2 and IFR-3. These results suggest that Cr(VI) was almost completely reduced by both isolates in the extracellular environment as very negligible amount of Cr(VI) and total Cr were detected in the cell extract supernatants of isolates IFR-2 and IFR-3 (Table 2). To investigate the location of enzymatic reduction, the chromate reductase activity was measured in the culture supernatant, cell lysate and cell extract supernatant. Chromate reductase activity was detected in the culture supernatant and cell lysate but not at all in the cell extract supernatant of both isolates. The residual Cr(VI) concentration in the cell lysate and culture supernatants of IFR-2 and IFR-3 ranged between 30 and 60% (Fig. 3) while the cell extract supernatant retained 100% of Cr(VI) (1 mg/l). The specific activity of chromate reductase was calculated from this data and ranged between 0.033 to 0.053  $\mu\text{mol}$  Cr(VI) reduced/min/mg protein in the cell lysate and culture supernatants respectively (Table 3). We therefore presume that chromate reductase is a membrane-bound enzyme in both isolates and is also secreted in the extracellular environment. Camargo et al. [13] observed chromate reductase activity in the cell-free extract and soluble fraction but very low activity in the membrane fraction of *Bacillus* sp. ES29. Wang et al. [15] reported that chromate reductase activity is preferentially associated in the membrane fraction of *E. cloacae* HO1. Membrane bound chromium reduction activity has also been reported in *P. fluorescens* LB300 [7].

Chromate reduction activities of whole cells of bacterial isolates were determined at 37°C. Whole cells of IFR-2 and IFR-3 reduced 24 and 30% of the initial Cr(VI) concentration (1 mg/l) in 45 min respectively. Initial cell density has been shown to have a dramatic effect on Cr(VI) reduction [16]. A number of factors may be contributing to the observed trends, including Cr(VI) toxicity acclimation time needed to readjust cells to low nutrient loads before initiation of reduction.

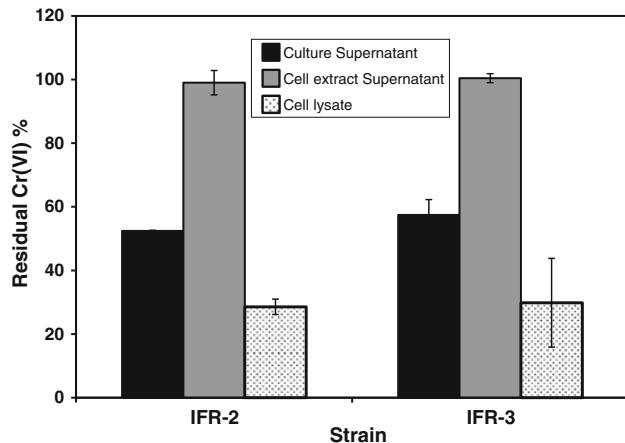
The effects of different salts on the growth of bacterial isolates were also investigated. The growth of isolate IFR-2 was stimulated in presence of  $\text{NiCl}_2$  and very mild growth was observed in presence of  $\text{CuCl}_2$  and  $\text{ZnCl}_2$  whereas growth was almost completely inhibited in presence of  $\text{CdCl}_2$  and  $\text{HgCl}_2$  (Fig. 4). Isolate IFR-3 showed good growth in presence of  $\text{ZnCl}_2$ ,  $\text{NiCl}_2$  and  $\text{CuCl}_2$ , mild growth in presence of  $\text{CdCl}_2$  and almost no growth in



**Fig. 2** Time course of growth and Cr(VI) (20 mg/l) reduction by bacterial isolates IFR-2 (A) and IFR-3 (B)

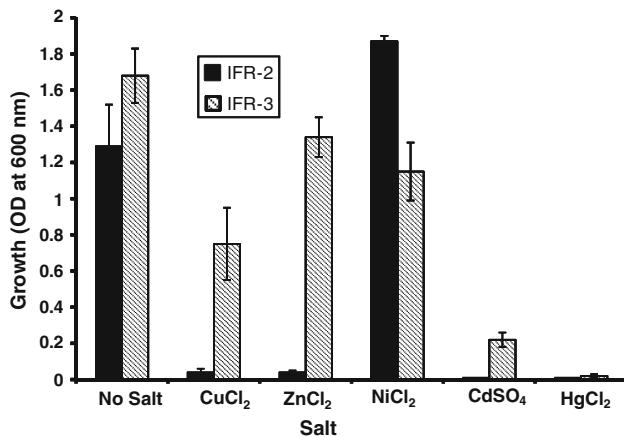
**Table 2** Determination of Cr(VI) and total Cr as  $K_2Cr_2O_7$  concentrations in the culture supernatants and cell extract supernatants of bacterial isolates by DPC method and Atomic Absorption Spectrometer, respectively

Fraction	Cr(VI) concentration (mg/l)		Total Cr concentration (mg/l)	
	IFR-2	IFR-3	IFR-2	IFR-3
Culture supernatant	0.0	0.0	19.81 ( $\pm 0.37$ )	19.27 ( $\pm 0.33$ )
Cell extract supernatant	0.64 ( $\pm 0.01$ )	0.53 ( $\pm 0.04$ )	0.021 ( $\pm 0.01$ )	0.032 ( $\pm 0.01$ )

**Fig. 3** Cr(VI) reduction by different fractions of bacterial isolates

presence of  $HgCl_2$  (Fig. 4). Basu et al. [17] have reported that  $Hg^{2+}$  and  $Pb^{2+}$  are toxic to bacterial isolates and inhibited their growth.

The optimum temperatures for growth and Cr(VI) reduction was found between 35 and 40°C for both isolates IFR-2 and IFR-3 (Fig. 5). The growth of both isolates increased with the increase of temperature up to 35°C and decreased at 45°C. Cr(VI) reduction by both isolates also increased with the increase of temperature up to 35°C and decreased above 40°C (Fig. 5). However, more than 95% of Cr(VI) was reduced by both isolates between temperatures 25–45°C. No significant difference was observed in the presence or absence of Cr(VI) on the growth of both isolates at different temperatures. Thacker and Madamwar [14] reported a maximum reduction of Cr(VI) at 35°C by the bacterial isolate DM1. An NAD(P)H dependent hexavalent chromium reductase was purified from *Pseudomonas ambigua* which showed activity over a wide range of temperature from 40 to 70°C [18].

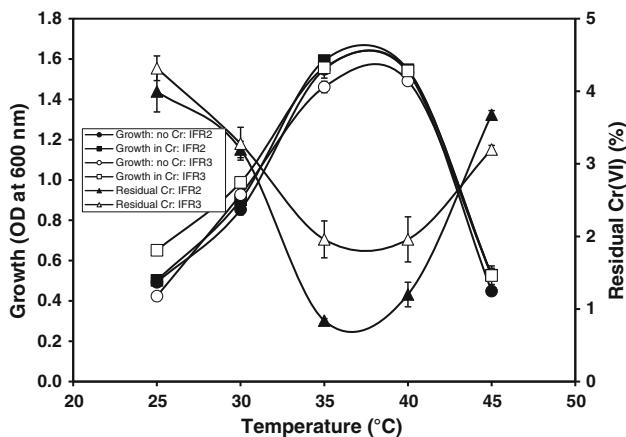
**Fig. 4** Effect of different metals (200 mg/l) on the growth of bacterial isolates

The bacterial isolates IFR-2 and IFR-3 can grow and reduce more than 95% Cr(VI) in a wide range of pH between 5.0 to 8.0 as seen in Fig. 6. The optimum pH for growth and Cr(VI) reduction by both isolates was found between pH 7.0 and 8.0. Growth and Cr(VI) reduction by both isolates was significantly reduced at pH above 8.0 (Fig. 6). Optimal Cr(VI) reduction was shown to be directly related to the optimum pH (8.0) for the growth of *Bacillus* isolates [4]. McLean et al. [16] reported a wide range of pH between 6.0 to 9.0 for Cr(VI) reduction by *P. synxantha*.

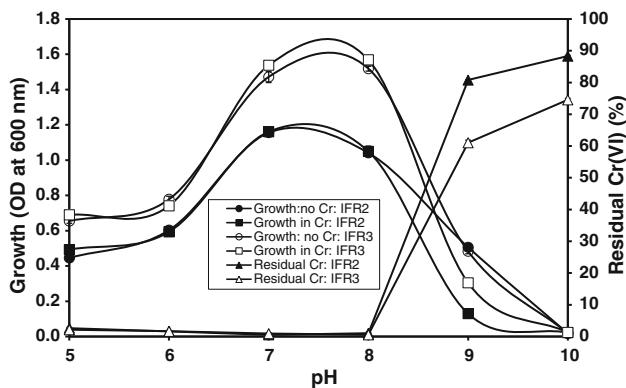
Chromate tolerance mechanisms in bacteria have been reported to include reduction, methylation, precipitation at the cell surface, blocking cellular uptake by altering the uptake pathway and removal from the cytoplasm by efflux pumps [19–21]. In the present study the mechanism of chromate tolerance was not investigated. However, our results in this study provide a basis for assessing the potential of using indigenous Cr(VI)-reducing novel

**Table 3** Determination of protein concentration and chromate reductase specific activity in the culture supernatant and cell lysate of bacterial isolates

Fraction	Protein concentration (mg/ml)		Chromate reductase specific activity ( $\mu\text{mol Cr(VI)}$ reduced/min/mg protein)	
	IFR-2	IFR-3	IFR-2	IFR-3
Culture supernatant	0.067	0.061	0.051 ( $\pm 0.01$ )	0.053 ( $\pm 0.02$ )
Cell lysate	0.13 ( $\pm 0.01$ )	0.153 ( $\pm 0.04$ )	0.04 ( $\pm 0.01$ )	0.033 ( $\pm 0.01$ )



**Fig. 5** Effect of temperatures on the growth and Cr(VI) (10 mg/l) reduction by bacterial isolates



**Fig. 6** Effect of pH on the growth and Cr(VI) (10 mg/l) reduction by bacterial isolates

bacteria for bioremediation application. Development of economic and efficient bioremediation technique based on immobilized whole cell system for the reduction and removal of toxic Cr from tannery effluents and solid wastes may help to cope with the extensive use of Cr in leather tanning.

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