

Stereospecific conversion of prostaglandin D₂ to (5Z,13E)-(15S)-9 α ,-11 β ,15-trihydroxyprosta-5,13-dien-1-oic acid (9 α ,11 β -prostaglandin F₂) and of prostaglandin H₂ to prostaglandin F_{2 α} by bovine lung prostaglandin F synthase

(prostaglandin-D₂ 11-ketoreductase/11-epi-prostaglandin F_{2 α} /bovine lung)

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ABSTRACT A prostaglandin F (PGF) synthase was recently purified from bovine lung that catalyzed the reduction of both PGH₂ and PGD₂ but at different active sites on the enzyme. In view of the recent finding that PGD₂ is stereospecifically reduced to a unique biologically active compound, (5Z,13E)-(15S)-9 α ,11 β ,15-trihydroxyprosta-5,13-dien-1-oic acid (9 α ,11 β -PGF₂ or 11-epi-PGF_{2 α}), by a human liver cytosolic enzyme, detailed characterization of the products formed from PGH₂ and PGD₂ by the bovine lung PGF synthase was carried out. Chromatographic characteristics of the products formed and stereochemical analysis procedures using mass spectrometry indicated that the enzyme stereospecifically reduces PGH₂ to PGF_{2 α} , whereas PGD₂ is stereospecifically converted to 9 α ,11 β -PGF₂. The finding that this enzyme catalyzes the formation of both C-11 hydroxy epimers of PGF₂, albeit from different substrates, is of interest in that these two compounds may exert different biological actions.

Prostaglandin F_{2 α} (PGF_{2 α} ; also called 9 α ,11 α -PGF₂) is produced by a number of mammalian organs (1, 2) and exerts a variety of biological actions (3-6). Three different biosynthetic pathways have been described for the formation of PGF_{2 α} : (i) 9,11-endoperoxide reduction of PGH₂ (7-9), (ii) 11-ketoreduction of PGD₂ (10, 11), and (iii) 9-ketoreduction of PGE₂ (12, 13). Recently a cytosolic enzyme with 11-ketoreductase activity was purified to apparent homogeneity from bovine lung (14). Interestingly, it was found that this enzyme catalyzed the reduction of both PGD₂ and PGH₂ but at two different active sites on the enzyme. Thus, this enzyme has been called "PGF synthetase" (14), herein called "PGF synthase."

Recently it was reported that PGD₂ is stereospecifically transformed *in vitro* to a unique biologically active compound, (5Z,13E)-(15S)-9 α ,11 β ,15-trihydroxyprosta-5,13-dien-1-oic acid (9 α ,11 β -PGF₂; also called 11-epi-PGF_{2 α}) by a cytosolic enzyme present in human liver (15). Furthermore, evidence was presented that the formation of 9 α ,11 β -PGF₂ also occurs *in vivo* in humans. The PGF₂ products formed from PGD₂ and PGH₂ by the bovine lung PGF synthase had been tentatively identified as PGF_{2 α} . However, in view of the above finding that PGD₂ is stereospecifically reduced to 9 α ,11 β -PGF₂ by a human liver enzyme, detailed characterization of the products formed from PGD₂ and PGH₂ by the bovine lung PGF synthase was performed. We report the

finding that this enzyme stereospecifically reduces PGH₂ to PGF_{2 α} , whereas PGD₂ is stereospecifically converted to 9 α ,11 β -PGF₂.

MATERIALS AND METHODS

Materials. [5,6,8,9,12,14,15-³H₇(N)]PGD₂ (100 Ci/mmol) and [1-¹⁴C]arachidonic acid (60 mCi/mmol; 1 Ci = 37 GBq) were obtained from New England Nuclear. [1-¹⁴C]PGH₂ was prepared from [1-¹⁴C]arachidonic acid as described (14). 9 α ,11 β -PGF₂ was synthesized as described by Iguchi *et al.* (16). [3,3,4,4-²H₄]PGF_{2 α} was a gift of John Pike (Upjohn Company). Other PGs were kindly donated by Ono Pharmaceutical Company. Other materials and commercial sources were as follows: NADP, NADPH, glucose 6-phosphate, and glucose-6-phosphate dehydrogenase from Baker's yeast (type IX), from Sigma; precoated silica gel glass plates (F254), from Merck. The boric acid-impregnated silica gel plates were prepared by the method of Lincoln *et al.* (17). PGF synthetase was purified from bovine lung to apparent homogeneity as described (14).

Enzyme Incubation Conditions and Isolation of Products Formed. The reaction mixture for PGD₂ contained 0.1 M potassium phosphate (pH 6.5), 1.5 mM [³H]PGD₂ (0.1 μ Ci), 0.5 mM NADP, 5 mM glucose 6-phosphate, 1 unit of glucose-6-phosphate dehydrogenase, and purified enzyme in a total volume of 0.1 ml. Incubation was carried out at 37°C for 1 hr. Approximately 80% of PGD₂ was converted to a single more polar product. The reaction mixture for PGH₂ contained 0.1 M potassium phosphate (pH 6.5), 70 μ M [1-¹⁴C]PGH₂ (0.37 μ Ci), 0.5 mM NADPH, and purified enzyme in a total volume of 1 ml. Incubations were carried out at 37°C for 2 min. About 30% of PGH₂ was metabolized by the enzyme during a 2-min incubation. Under these conditions, PGH₂ was degraded nonenzymatically to PGE₂ (22%), PGD₂ (16%), 12-L-hydroxy-5,8,10-heptadecatrienoic acid (HHT) (2%), and PGF_{2 α} (<1%). Reactions were terminated, and products were extracted with 3 vol of diethyl ether/methanol/0.2 M citric acid, 30:4:1 (vol/vol), to the reaction mixture. The organic phase was evaporated, and the residue was subjected to TLC with the solvent system A (see below). The major enzymatic product formed from both PGH₂ and PGD₂ that migrated in the region of PGF_{2 α} (R_f

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Abbreviations: PG, prostaglandin; 9 α ,11 β -PGF₂ (11-epi-PGF_{2 α}), (5Z,13E)-(15S)-9 α ,11 β ,15-trihydroxyprosta-5,13-dien-1-oic acid; RP, reversed phase; Me₃Si, trimethylsilyl; BA-TLC, boric acid-impregnated TLC; BSTFA, N,O-bis(trimethylsilyl)trifluoroacetamide.

approximately 0.27) was scraped and extracted from the silica gel with ethyl acetate/acetic acid, 100:1 (vol/vol).

Identification of Products Formed by PGF Synthase. Further analysis of the PGF₂ products formed from PGD₂ and PGH₂ was carried out by using boric acid-impregnated TLC (BA-TLC), reversed-phase (RP)-HPLC, and GC/MS.

Three different solvent systems (A, B, and C) were used for BA-TLC: A, benzene/dioxane/acetic acid, 20:20:1 (vol/vol); B, diethyl ether/methanol/acetic acid, 90:2:0.1 (vol/vol); and C, ethyl acetate/acetic acid, 99:1 (vol/vol). TLC plates subjected to chromatography with solvent systems A and B were developed twice, and the plate chromatographed with solvent system C was developed three times.

RP-HPLC was performed on a 5- μ m Alltech C₁₈ column with the solvent system acetonitrile/water/acetic acid, 27:73:0.1 (vol/vol) at 1-ml/min in 1-ml fractions. Characteristic elution volumes for 9 α ,11 β -PGF₂ and PGF_{2 α} were approximately 41–43 ml and 53–56 ml, respectively.

Methyl esters of compounds were formed by treatment with excess ethereal diazomethane. Trimethylsilyl (Me₃Si) ethers were formed by treatment with 20 μ l of *N,O*-bis(trimethylsilyl)trifluoroacetamide (BSTFA) (Regis, Morton Grove, IL) and 20 μ l of dimethylformamide at room temperature for 15 min. Butylboronate derivatives were formed by treatment with 50 μ l of butylboronic acid (Aldrich) in pyridine (10 mg/ml) at 60°C for 45 min.

GC/MS analysis was performed by using a Nermag R10-10C gas chromatograph/mass spectrometer interfaced with a DEC PDP-11/23 plus computer system. GC was performed as a six-meter DB-1 fused silica capillary column programmed from 190–325°C at 25°C/min. Helium was used as the carrier gas at a flow rate of 1 ml/min. Ion-source temperature was 200°C, electron energy was 87 eV, and filament current was 0.25 mA.

RESULTS

After incubation of PGH₂ and PGD₂ with the purified bovine lung PGF synthase, products formed were extracted and subjected to TLC using solvent system A. The major enzymatic products formed from both PGH₂ and PGD₂, which migrated in the region of PGF_{2 α} (*R_f* approximately 0.27), were isolated and further analyzed by BA-TLC, GC/MS, and HPLC.

To assess the stereochemical configuration of the C-9 and C-11 PGF-ring hydroxyl groups, the ability of the products to form a butylboronate derivative when treated with *n*-butylboronic acid was determined. A butylboronate derivative will form bridging the oxygens at C-9 and C-11 only if both hydroxyls are *cis* in the prostane ring (18).

The enzymatic products formed from PGD₂ and PGH₂ were initially converted to a methyl ester and then sequentially treated with *n*-butylboronic acid and BSTFA and subsequently analyzed by GC/MS. [²H₄]PGF_{2 α} was coterivatized with each sample to provide a means to determine the completion of the boronation reaction when analyzed by GC/MS.

Identification of the Product Formed from PGD₂. Figs. 1 and 2 show the mass spectra obtained from analysis of the product formed from PGD₂. The mass spectrum in Fig. 1 is characteristic for the methyl ester–Me₃Si ether derivative of both 9 α ,11 β -PGF₂ (15) and PGF_{2 α} (19), both of which have essentially identical mass spectra except for minor variation in relative ion abundances. The mass spectrum in Fig. 2 is essentially identical to that previously reported for the methyl ester–butylboronate–Me₃Si ether derivative of [²H₄]PGF_{2 α} (15). Importantly, there are no doublet ion peaks in either mass spectrum four atomic mass units apart, indicating that the [²H₄]PGF_{2 α} that was added to the sample was quantitatively converted to a butylboronate derivative, whereas none

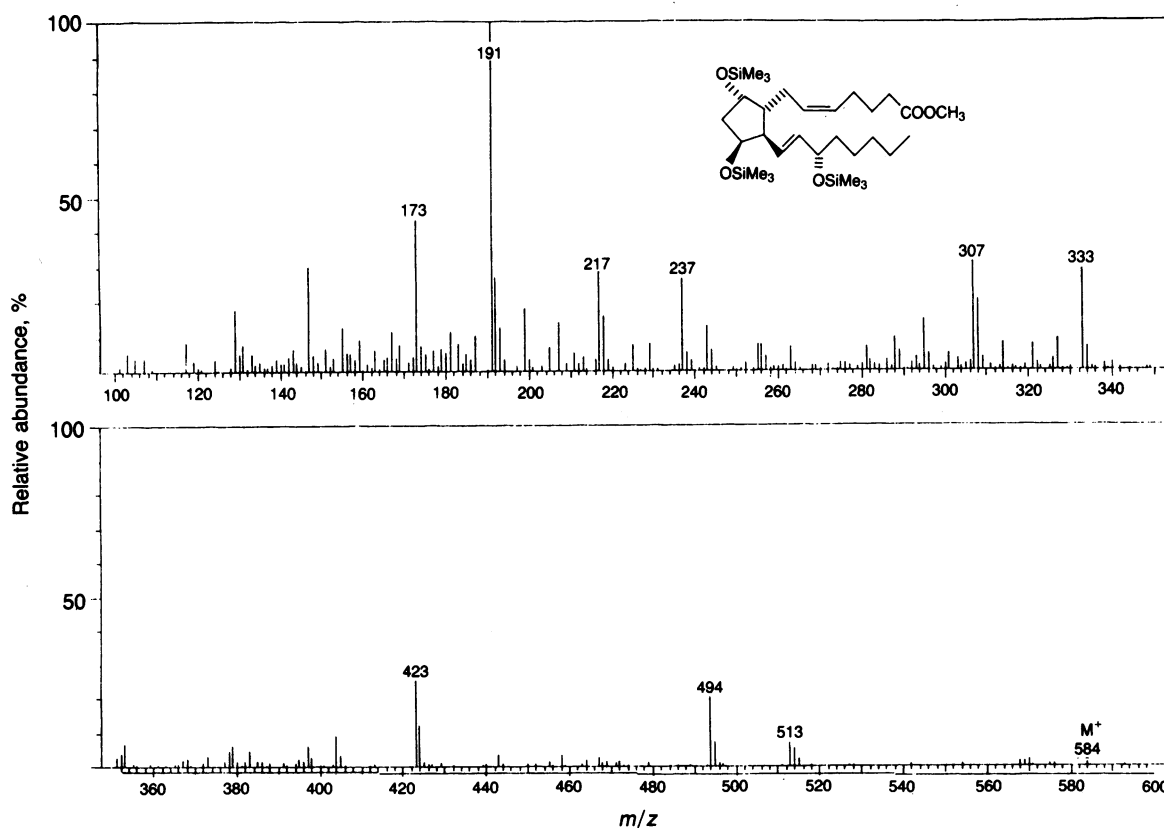


FIG. 1. Mass spectrum of the PGF₂ product formed from PGD₂ by the PGF synthase after coterivatization with [²H₄]PGF_{2 α} by treatment sequentially with diazomethane, *n*-butylboronic acid, and BSTFA.

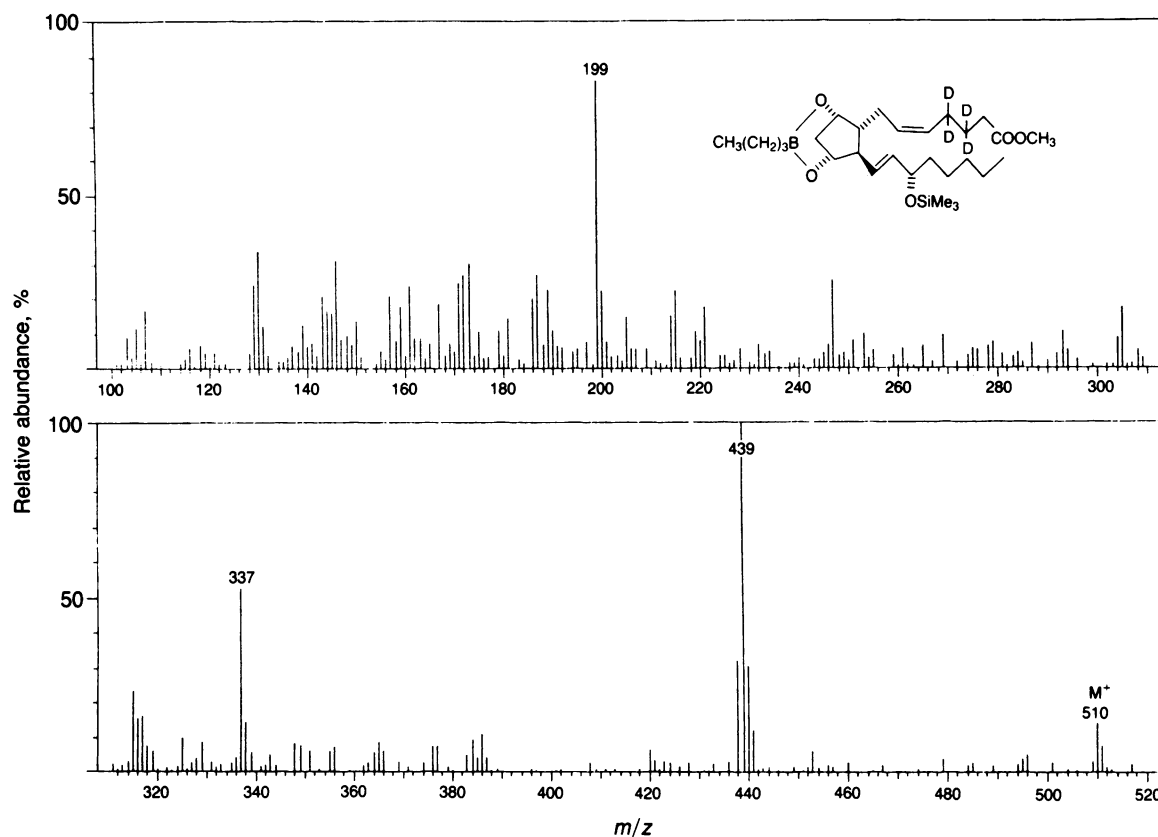


FIG. 2. Mass spectrum of $[^2\text{H}_4]\text{PGF}_{2\alpha}$ coderivatized with the PGF_2 product formed from PGD_2 by the PGF synthase after treatment sequentially with diazomethane, *n*-butylboronic acid, and BSTFA.

of the enzymatic product formed from PGD_2 reacted with *n*-butylboronic acid.

The failure of the PGF_2 product formed from PGD_2 to form a butylboronate derivative provided evidence that PGD_2 was stereospecifically converted by the enzyme to $9\alpha,11\beta\text{-PGF}_2$. This was further supported by BA-TLC (data not shown) and RP-HPLC analysis of the product. $\text{PGF}_{2\alpha}$ and $9\alpha,11\beta\text{-PGF}_2$ are widely separated on RP-HPLC (15). RP-HPLC analysis of the product formed from PGD_2 revealed only a single radioactive peak with an elution volume characteristic of that of $9\alpha,11\beta\text{-PGF}_2$, 40–42 ml (Fig. 3). Virtually no radioactive peak was detected at the elution volume characteristic for $\text{PGF}_{2\alpha}$ (approximately 53–56 ml).

Identification of the Product Formed from PGH_2 . When the methyl esters of the product formed from PGH_2 and added $[^2\text{H}_4]\text{PGF}_{2\alpha}$ were sequentially treated with *n*-butylboronic acid and BSTFA and were analyzed by GC/MS, a mass spectrum essentially identical to that shown in Fig. 1 was obtained. Again, there were no doublet ion peaks in this mass spectrum four atomic mass units apart, indicating that the compound did not react with *n*-butylboronic acid and that $[^2\text{H}_4]\text{PGF}_{2\alpha}$ was quantitatively converted to a butylboronate derivative. However, analysis of the mass spectrum obtained for the $[^2\text{H}_4]\text{PGF}_{2\alpha}$ (Fig. 4) revealed doublet ion peaks four atomic mass units apart, indicating the presence of a mixture of unlabeled $\text{PGF}_{2\alpha}$ and $[^2\text{H}_4]\text{PGF}_{2\alpha}$. Doublet ion peaks were present for ions retaining the deuterium atoms on C-3 and C-4 of the $[^2\text{H}_4]\text{PGF}_{2\alpha}$, such as the molecular ions at m/z 506 and 510 and fragmentation ions at m/z 435 and 439 [$M^+ - 71$, loss of $\cdot(\text{CH}_2)_4\text{CH}_3$] and at m/z 333 and 337 [$M^+ - 173$, loss of $\cdot\text{CH}(\text{OSiMe}_3)(\text{CH}_2)_4\text{CH}_3$]. However, the ion at m/z 199, $[\text{CH}=\text{CHCH}(\text{OSiMe}_3)(\text{CH}_2)_4\text{CH}_3]^+$, comprising the lower side chain (C-13 to C-20) did not retain the deuterium atoms at C-3 and C-4 and, thus, was present in the mass spectrum of both unlabeled $\text{PGF}_{2\alpha}$ and $[^2\text{H}_4]\text{PGF}_{2\alpha}$. For this reason,

also, the base ion of this mixed mass spectrum is m/z 199 instead of m/z 439 (compare Fig. 2).

The above data indicated that both $9\alpha,11\beta\text{-PGF}_2$ and $\text{PGF}_{2\alpha}$ were formed during incubation of PGH_2 with the enzyme. This was further supported by RP-HPLC analysis (Fig. 5). A radioactive peak representing approximately 23%

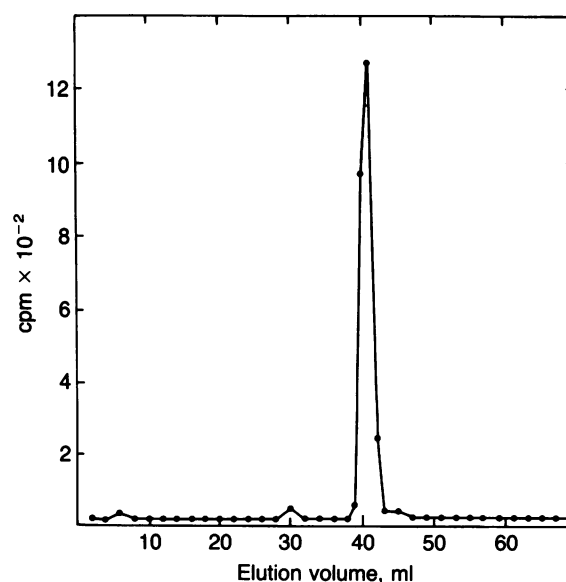


FIG. 3. RP-HPLC analysis of the product formed from incubation of $[^3\text{H}]\text{PGD}_2$ with the PGF synthase. The solvent system was acetonitrile/water/acetic acid, 27:73:0.1, run isocratically at 1 ml/min; 1-ml fractions were collected. $9\alpha,11\beta\text{-PGF}_2$ is characteristically eluted at approximately 40–43 ml and $\text{PGF}_{2\alpha}$ at 53–56 ml.

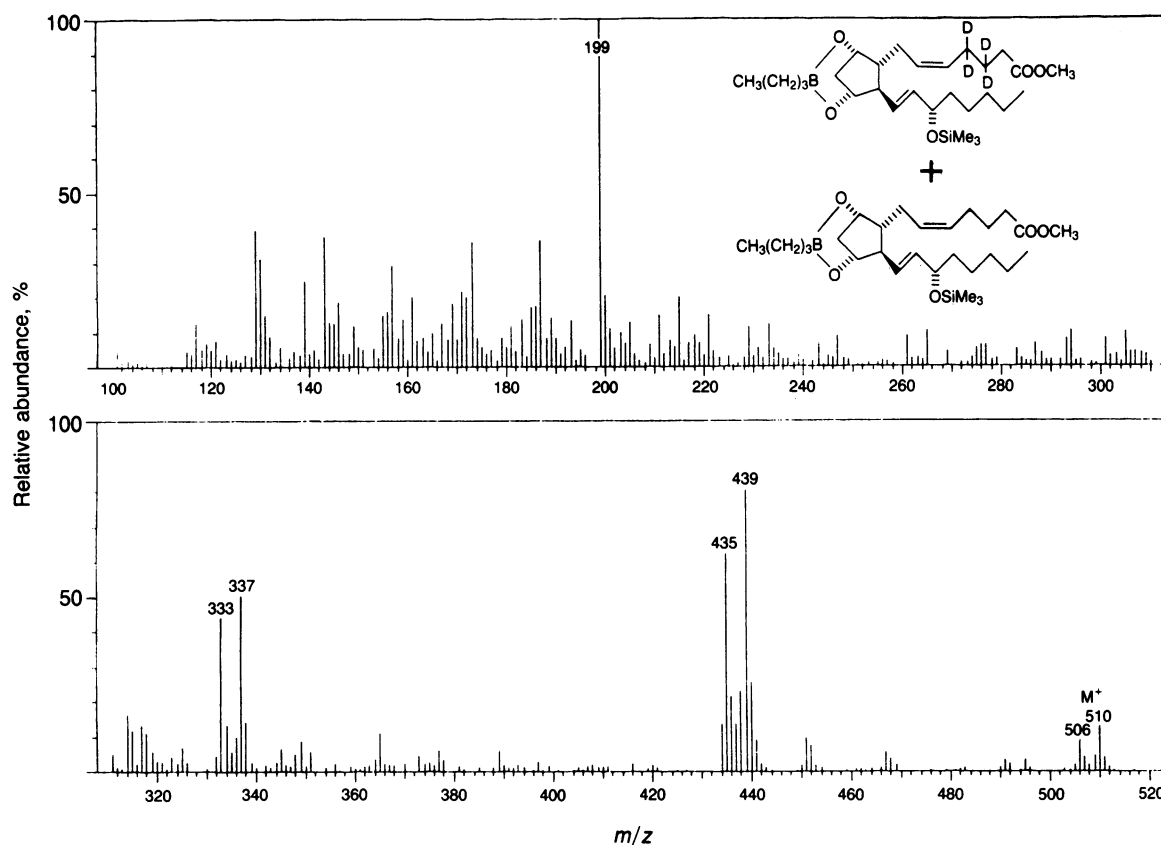


FIG. 4. Mass spectrum obtained for $[^2\text{H}_4]\text{PGF}_{2\alpha}$ coderivatized with the PGF_2 products formed from PGH_2 during incubation with the PGF synthase after treatment sequentially with diazomethane, *n*-butylboronic acid, and BSTFA.

of the total recovered radioactivity was eluted with a retention volume characteristic of $9\alpha,11\beta\text{-PGF}_2$ at 41–43 ml in addition to a less-polar peak (77%), which was eluted with a retention volume characteristic for $\text{PGF}_{2\alpha}$ at 53–56 ml.

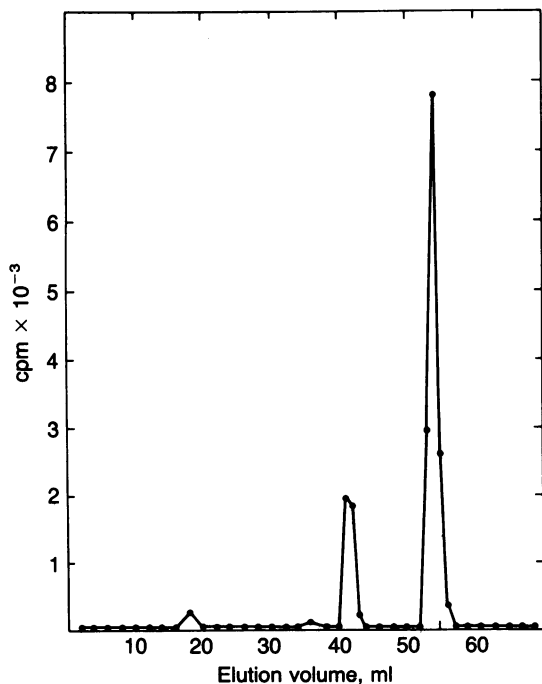


FIG. 5. RP-HPLC analysis of the products formed during incubation of $[^{14}\text{C}]\text{PGH}_2$ with the PGF synthase. The solvent system was acetonitrile/water/acetic acid, 27:73:0.1, run isocratically at 1 ml/min; 1-ml fractions were collected. $9\alpha,11\beta\text{-PGF}_2$ is characteristically eluted at approximately 40–43 ml and $\text{PGF}_{2\alpha}$ at 53–56 ml.

Origin of the Formation of $9\alpha,11\beta\text{-PGF}_2$ During Incubation of PGH_2 with PGF Synthase. Since the C-11 oxygen of PGH_2 is oriented α , it is difficult to envision how the enzyme could reduce PGH_2 to $9\alpha,11\beta\text{-PGF}_2$ with inversion of the configuration of C-11. However, as previously reported, PGH_2 is nonenzymatically degraded in part to PGD_2 during the incubation of PGH_2 with the PGF synthase (14). Therefore, an attractive explanation for the formation of $9\alpha,11\beta\text{-PGF}_2$ during incubation of PGH_2 with this enzyme is that PGH_2 is not directly transformed by the enzyme to $9\alpha,11\beta\text{-PGF}_2$ but that the PGD_2 which is formed from nonenzymatic degradation of PGH_2 during the incubation is reduced by the enzyme to $9\alpha,11\beta\text{-PGF}_2$.

The above hypothesis was confirmed by demonstrating that incubation of radiolabeled PGH_2 with the enzyme in the presence of a concentration of unlabeled PGD_2 that was approximately 40-fold greater than the K_m of the enzyme for PGD_2 prevented the formation of radiolabeled $9\alpha,11\beta\text{-PGF}_2$ (Table 1). Likewise, phenanthrenequinone, which competitively inhibits the reduction of PGD_2 by this enzyme, almost completely abolished the formation of $9\alpha,11\beta\text{-PGF}_2$.

DISCUSSION

We recently reported that PGF synthase purified from bovine lung catalyzed the reduction of both PGH_2 and PGD_2 but at different active sites on the enzyme (14). The products formed by the enzyme from PGH_2 and PGD_2 were tentatively identified as $\text{PGF}_{2\alpha}$. However, more detailed characterization of these products has now revealed that the enzyme stereospecifically reduces PGH_2 to $\text{PGF}_{2\alpha}$ but stereospecifically converts PGD_2 to $9\alpha,11\beta\text{-PGF}_2$.

Therefore, the 11-ketoreductase activity of the purified bovine lung PGF synthase is similar to the cytosolic enzyme with 11-ketoreductase activity in human liver, which stereospecifically reduces PGD_2 to $9\alpha,11\beta\text{-PGF}_2$ (15).

Table 1. Effects of PGD₂ and phenanthrenequinone (PQ) on the product formation from PGH₂ by PGF synthase

Addition	Products from PGH ₂ , nmol/min	
	9 α ,11 β -PGF ₂	PGF _{2α}
None	0.14	0.20
5 mM PGD ₂	0.02	0.18
28 μ M PQ	0.03	0.31

The reaction mixture contained 0.1 M potassium phosphate (pH 6.5), 80 μ M [1-¹⁴C]PGH₂ (0.42 μ Ci), 0.5 mM NADP, 5 mM glucose 6-phosphate, 1 unit of glucose-6-phosphate dehydrogenase and enzyme (6 μ g), in a total volume of 0.1 ml in the absence or presence of 5 mM PGD₂ or 28 μ M PQ. Incubation was carried out at 37°C for 2 min. Reactions were terminated by addition of 3 vol of diethyl ether/methanol/0.2 M citric acid as described. The organic phase was evaporated, and the sample was analyzed by RP-HPLC with the solvent system acetonitrile/water/acetic acid, 30:70:0.1 (vol/vol). In the absence of enzyme, 0.018 and 0.044 nmol of 9 α ,11 β -PGF₂ and PGF_{2 α} , respectively, were produced under these conditions. Numbers are corrected for the nonenzymatic formation of each product.

Whether the human liver enzyme with 11-ketoreductase activity also catalyzes the reduction of PGH₂ has not been examined.

Thus, at this time it is not possible to generalize whether 11-ketoreductase and 9,11-endoperoxide reductase activities are linked to a single enzyme in all organs and tissues and in all species. Although the bovine lung PGF synthase and the human liver enzyme with 11-ketoreductase activity stereospecifically reduce PGD₂ to 9 α ,11 β -PGF₂, 11-ketoreductase activity has been reported in sheep blood that stereospecifically reduces PGD₂ to PGF_{2 α} (20). Although 11-ketoreductase activity also has been described in rabbit liver (11, 21) and a variety of organs in the rat (10), stereochemical analysis of the C-11 hydroxyl in the PGF₂ products formed was not carried out. Therefore, it is uncertain whether the sheep blood enzyme with 11-ketoreductase activity is unique in that it converts PGD₂ to PGF_{2 α} , whether there may be species differences in the 11-ketoreductase, or whether different 11-ketoreductases are distributed in certain organs within a species.

The full spectrum of biological activity of 9 α ,11 β -PGF₂ remains to be elucidated. PGF_{2 α} and 9 α ,11 β -PGF₂ in some systems may exert similar biological effects, such as raising blood pressure in the rat (15) and contraction of human coronary arteries *in vitro* (22). However, 9 α ,11 β -PGF₂ has been shown to inhibit ADP-induced human platelet aggregation, whereas PGF_{2 α} is devoid of an inhibitory effect on

ADP-induced platelet aggregation (23). Thus, the finding that this single enzyme purified from bovine lung catalyzes the formation of both PGF_{2 α} and 9 α ,11 β -PGF₂, albeit from different substrates, is of interest in that these two PGs may in some systems exert different biological actions.

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