Domain flexibility modulates the heterogeneous assembly mechanism of anthrax toxin protective antigen

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Abstract

Individually, the three protein components of anthrax toxin are nontoxic, but when they assemble, they form active holotoxin complexes. The role of the protective antigen (PA) component of the toxin is to deliver the two other enzyme components, lethal factor (LF) and edema factor (EF), across the plasma membrane and into the cytoplasm of target cells. PA is produced as a proprotein, which must be proteolytically activated, and generally cell-surface activation is mediated by a furin-family protease. Activated PA can then assemble into one of two non-interconverting oligomers, a homoheptamer and homooctamer, which have unique properties. Herein we describe molecular determinants that influence the stoichiometry of PA in toxin complexes. By tethering PA Domain 4 to Domain 2 with two different length cross-links, we can control the relative proportions of PA heptamers and octamers. The longer cross-link favors octamer formation, whereas the shorter one favors formation of the heptamer. X-ray crystal structures of PA (up to 1.45 Å resolution), including these cross-linked PA constructs, reveal that a hinge-like movement of Domain 4 correlates with the relative preference for each oligomeric architecture. Furthermore, we report the conformation of the flexible loop containing the furin-cleavage site and show that for efficient processing, the furin site cannot be moved ~5 or 6 residues within the loop. We propose that there are different orientations of Domain 4 relative to the main body of PA that favor the formation of either the heptamer or the octamer.

Introduction

Anthrax toxin (Atx) is a virulence factor secreted by pathogenic strains of *Bacillus anthracis*. Atx consists of three nontoxic protein components,[3-5] namely protective antigen (PA, 83-kDa), lethal factor (LF, 90-kDa), and edema factor (EF, 89-kDa). The PA component assembles into a ring-shaped oligomer capable of forming a membrane-spanning translocase channel, which delivers the two enzyme components, LF and EF, into the

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Accession numbers. Coordinates and structure factors for PA$^{\Delta MIL}$ pH 8.5, PA$^{\Delta MIL}$ pH 6.5, PA$^{\Delta MIL}$ C$_{337}$-C$_{664}$, and PA$_{83}$ C$_{337}$-C$_{664}$-DCA have been deposited in the PDB with accession codes 3TEW, 3TEX, 3TEY, and 3TEZ, respectively.

cytosol of a host cell. LF is a Zn^{2+}-dependent protease,[6-8] which cleaves host-cell mitogen-activated protein kinase kinases. PA, LF, and EF are individually nontoxic; however, LF plus PA creates lethal toxin (LT), which can alter normal cell function and may even cause death.[9] EF is Ca^{2+}- and calmodulin-dependent adenylyl cyclase.[10-12] Similarly, PA plus EF produces edema toxin (ET), which induces tissue swelling and may also cause death.

The PA component contains four different domains (Domain 1-4), which are critical to various stages of toxin assembly and translocation. To form cytotoxic complexes, PA must first assemble with LF and/or EF. Two different types of assembly pathways have been described: a cell-surface pathway (for review, see ref. 1) and a plasma-based/extracellular pathway. On the surface of host cells, PA binds to one of two known Atx receptors: ANTXR1 (ref. 15) and ANTXR2 (ref. 16). The PA-ANTXR2 interaction is rather stable and dissociates with a half-life measured in days.[19] The interaction involves Domain 2 and Domain 4 in PA, such that the latter domain coordinates the receptor's Ca^{2+} or Mg^{2+} dependent metal ion adhesion site.[17-21] ANTXR2-bound PA is subsequently cleaved by a furin-family protease yielding the proteolytically-activated or nicked form, called $_nPA$. After a 20-kDa portion of $_nPA$ (PA_{20}) dissociates, the remaining 63-kDa, receptor-bound portion (PA_{63}) self-assembles into an ~2:1 mixture of heptameric (PA_7) and octameric (PA_8) ring-shaped oligomers. The complexes are endocytosed[25] and transferred to an acidic compartment,[26] where the lower pH conditions induce the PA oligomers to form transmembrane translocase channels.[27-29] LF and EF translocate through the PA channel to enter the cytosol of the host cell.

Recent studies investigating the molecular mechanism of PA oligomerization have revealed that the assembly of the octameric PA oligomer is largely dependent on the degree to which even-numbered PA intermediates (PA_2 and PA_4) are populated. Co-assembly factors such as LF, EF, or dimeric ANTXR2 (dANTXR2) presumably favor the formation of these PA_2 intermediates and, correspondingly, PA_8 formation[13] (Fig. 1a). Interestingly, when PA's membrane insertion loop (MIL) is deleted ($PA^{\Delta MIL}$), these mutant subunits can form 5-10 fold more octamer than heptamer—even in the absence of co-assembly factors. The structural explanation for the propensity of $PA^{\Delta MIL}$ to form the octamer is not well understood. Alignment of the PA subunits from the heptameric[23] and octameric[14] oligomer structures reveals that the conformation of Domain 4 (D4, residues 596-735) is quite flexible (Fig. 1b). Relative to the main body of the protein, PA D4 may be treated as a rigid body whose "pitch" may be described as a rotation around the "hinge" joining Domain 3 (D3, residues 458-595) and D4. While oligomerization may occur at a large contiguous surface containing residues from Domain 1 (D1, residues 1-258), Domain 2 (D2, residues 259-457) and D3, it is unknown how the orientation of D4 may modulate the oligomerization pathway and determine the stoichiometry of PA in the complex.

Mammalian furin-type proteases are members of the prohormone convertase (PC) family.[31-34] A close homolog of furin in yeast, kexin, functions analogously by cleaving prohormone proteins.[35-37] Furin, the most well-characterized PC (EC 3.4.21.75),[33] cleaves its substrates at a cationic, R-X-(R/K/X)-R↓, consensus site (where the scissile bond is indicated with an ↓).[38] Furin is involved in diverse cellular processes, such as homeostasis and embryogenesis, and it is expressed in all tissues and cells as a secreted membrane *N*-linked-type glycoprotein that matures in the trans-Golgi network. Furin and its yeast homolog, kexin, each anchor to the cell membrane via a homologous single-pass transmembrane helix located on its carboxy terminus. As a membrane protein, furin can cycle between the cell surface and the endosomal compartment. Furin functions to proteolytically process many types of membrane-bound and soluble factors, including: growth factors, extracellular matrix proteases, viral proteins and bacterial toxins.

The role of furin and related PCs in viral and bacterial pathogenesis is widespread. Many bacterial and viral proproteins require proteolytic activation prior to maturation into either active enzymes or assembled virulence factors. Generally, but not always, furin-activated proteolysis permits these pathogenic proteins to subsequently fuse with or penetrate the host cell membrane. Included amongst these virulence factors are the bacterial toxins,[40] anthrax,[41] shiga,[42] diphtheria,[43] tetanus, botulinum, *Clostridium septicum* α-toxin,[44] and aerolysin;[45] the influenza A virus fusion protein, hemagglutinin;[46] human immunodeficiency virus-1 envelope glycoprotein, gp160;[47] flaviviruses, such as tick-borne encephalitis and West Nile;[48] and filoviruses, such as Marburg and Ebola.[49]

In order for Atx to assemble, PA must first be cleaved by a furin-type protease. Cleavage occurs after the sequence, [164]RKKR↓S, in a solvent-exposed loop within D1.[41] Selective[50] and nonselective inhibition of furin and furin-like proteases strongly impairs Atx function. Furthermore, substrate-mapping studies of many related furin-type proteases show that PA's RKKR furin-recognition sequence (Fig. 1c) has maximized the motif for recognition by a broad number of these furin-type PC enzymes.[38] While NMR techniques have elucidated the structure of a furin cleavage site in a 19-residue peptide from gp160,[53] atomic-resolution structural information for furin cleavage sites in proviral and protoxin proteins have been challenging to obtain, because the sites are typically found on flexible loop structures. To gain an understanding on the structural basis of toxin assembly and furin activation we crystallized a variant of anthrax toxin PA and solved its structure to 1.45-Å resolution.

Results

Crystal structure of the PA$^{\Delta MIL}$ monomer

For our initial crystallographic studies, we chose a PA construct (PA$^{\Delta MIL}$) in which the MIL in D2 (residues 303-324) was replaced with the Pro-Gly dipeptide (which generally favors the formation of a Type II turn). The highly dynamic MIL is absent from previous structures of the PA monomer; therefore, we assumed its absence would improve crystal quality. Furthermore, we had previously solved the structure of PA$_8$ to 3.2-Å resolution[14] and the PA$_8$(LF$_N$)$_4$ complex to 3.1-Å resolution[24] using the PA$^{\Delta MIL}$ construct. These PA$^{\Delta MIL}$ crystal forms diffracted to higher resolution than the PA$_7$ crystal forms using the wild-type (WT) PA construct, which previously diffracted to 3.6 Å[18] and 4.5 Å.[23]

The PA$^{\Delta MIL}$ construct yielded large, well-diffracting orthorhombic crystals belonging to the $P2_12_12_1$ space group and diffracted X-rays to 1.45 Å. The structure was solved using molecular replacement; one PA protein was identified in the asymmetric unit. Interestingly, the unit cell dimensions and space group are identical to what Petosa *et al.* referred to as "Crystal form 1". From their study, Crystal form 1 did not diffract as well as "Crystal form 2," the basis for the PA structure with Protein Data Bank (PDB) accession code 1ACC.[23] Coordinates for Crystal form 1 were not deposited in the PDB, however. Contiguous electron density was identified for residues 15-735 of PA$^{\Delta MIL}$ (Fig. 2a). Loop regions lacking in the 1ACC search model were built manually in the experimental electron density using standard methods.

The resolution of our structure allowed us to refine all of PA's atomic coordinates individually, and the atomic displacement parameters (ADP) were refined anisotropically. Numerous solvent molecules were resolved, and their ADP values were also refined anisotropically. Overall, the model is in good agreement with the electron density, with an R value of 19.4% and an R_{free} value of 21.6% (Table 1). The resulting structure closely resembles previous PA monomer structures. D1 contains a jelly-roll motif with peripheral small helices and loops, including the furin-cleavage site-containing L$_{162-174}$, as well as a twin-Ca^{2+} coordination site resembling the classical EF hand motif (Fig. 2b). D2 contains a

Greek-key motif implicated in ultimately forming the 14- or 16-stranded β-barrel channel, through which LF and EF translocate. Additionally, a loop containing $2\alpha_1$ was resolved, which has been implicated in binding to ANTXR2. D3 contains a mixed β sheet and associated helices, including $3\alpha_1$, which was not present in 1ACC. D4 contains an immunoglobulin-like β-sandwich and is the most flexible of the PA domains.

Ordered loop regions

We resolved continuous electron density for PA$^{\Delta MIL}$ from residues 15 to 735 at the carboxy-terminus. While acknowledging that the membrane insertion loop (residues 305-324) is missing by design, this is the most complete PA model to date. (N.B. the numbering convention of 1ACC is retained.) While our structure and the three reported monomeric structures, PA (PDB 1ACC),[23] the ANTXR2-bound PA (PDB 1T6B),[17] and the 2-fluoro-histidine substituted PA (PDB 3MHZ),[54] align with close agreement with a root mean square (RMS) < 1, A number of surface-loop regions not modeled in 1ACC are present in the PA$^{\Delta MIL}$ structure. Residues 99-102 and 512-515 are not modeled in 1ACC; however, they are present in 1T6B, while the latter is present in 3MHZ. Residues 163 and 168-173, a region that includes the furin proteolytic-cleavage site, are not present in any deposited structure of PA, both monomeric and oligomeric. Interestingly, residues 159-161 are in a different conformation in 1ACC relative to our model, indicating that the loop containing the furin-cleavage site can adopt multiple conformations. Residues 162 and 164-167 are found in 3MHZ in approximately the same conformation as reported here. Also D2 residues 275-287 are present, modeled here as an extended loop whose atoms have ADP values comparable to the average protein atom. Residues 343-350, which are not present in 1ACC, are modeled in both 1T6B and 3MHZ.

Receptor-binding loop-helix 2α₁

In the published PA monomer structure (1ACC),[23] residues 343-350 are not modeled. In the co-crystal structure of PA and ANTXR2 (1T6B)[17] and the structure of the 2-fluoro-histidine substituted PA (3MHZ),[54] these residues are well-ordered in a loop-helix motif (Fig. 2c). Presumably, this ordering is due to the strong van der Waals interaction of PA residue L320 with ANTXR2 L154, which contributes to the strong energetics of PA-receptor binding. This loop-helix is also well-ordered for PA$^{\Delta MIL}$; the degree of order in this structure may reasonably be explained by the crystal contacts between $2\alpha_1$ and $3\alpha_1$. PA$^{\Delta MIL}$ and ANTXR2-bound PA from 1T6B align well and are in good agreement (RMS <0.3 Å). Alignment of the structures by Cα of residues 16-340 results in a slight shift of $2\alpha_1$ in the PA$^{\Delta MIL}$ structure relative to 1T6B. This movement frees up potential steric clashes of PA $2\alpha_1$ with the extracellular PA-binding domain of ANTXR2; therefore, the positioning of $2\alpha_1$ in the structure presented here likely represents the biological conformation in the absence of receptor and agrees well with the 3MHZ structure.[54]

Changes in orientation of PA D4

In the structure of PA in complex with ANTXR2 (1T6B), D4 undergoes a rigid-body conformational change, resulting in a slight tilting of the domain relative to the unbound structure of PA (1ACC). This movement is actually required so that ANTXR2 may bind PA at two discontinuous contact points, i.e., at $2\alpha_1$ in D2 and around D683 in D4. Interestingly, a backbone alignment of the PA$^{\Delta MIL}$ structure with the ANTXR2-bound PA structure from D1-D3 reveals that the orientation of D4 relative to the main body (D1-D3) does not significantly deviate. Even though both the PA$^{\Delta MIL}$ and 1ACC structures are not bound to receptor, the orientation of D4 in PA$^{\Delta MIL}$ more closely resembles that of the receptor-bound structure, 1T6B (Fig. 2d). Thus the orientation of D4 is the most significant structural change we can identify in the PA$^{\Delta MIL}$ structure.

Crystal structures of cross-linked PA monomers

In order to test the structural consequence of the orientation of D4 relative to the main body of PA, we engineered a pair of Cys mutations into the D2-D4 interface. We selected the residues, S337C and N664C, and incorporated the mutations into both the WT PA and PA$^{\Delta MIL}$ backgrounds. We then produced two different length cross-links. For the shorter-length cross-link, we allowed a disulfide bond to form between the two Cys residues, and we refer to this cross-link as C_{337}-C_{664}. For the longer-length cross-link, we used the small, thiol-specific cross-linker, 1,3-dichloroacetone (DCA),[55] and we refer to this DCA link as C_{337}-C_{664}-DCA (Fig. 3a). We solved the crystal structures of both PA$^{\Delta MIL}$ C_{337}-C_{664} and PA C_{337}-C_{664}-DCA (Table 1). We found that C_{337}-C_{664} contains a disulfide bond (Fig. 3b), while PA C_{337}-C_{664}-DCA contains a dithioether cross-link consistent with the DCA modification (Fig. 3c). The two structures agree well with other PA monomer structures. We then performed a global, Cα alignment of PA$^{\Delta MIL}$ C_{337}-C_{664} to PA$^{\Delta MIL}$ to assess whether the orientation of D4 was impacted by the cross-links. We find that domains D2 and D4 showed the largest displacement, resulting in maximum RMS displacements greater than 0.8 Å (Fig. 3d). This deviation increases with distance from the hinge between D3 and D4, indicating that the deviation is the result of a change in the hinge angle relating D4 with the main body of the protein. When we performed a similar RMS-displacement analysis, comparing PA C_{337}-C_{664}-DCA to PA$^{\Delta MIL}$. We found that the backbone positions did not move appreciably (Fig. 3d). Finally, for the purpose of comparison, we aligned the Cα's of PA from 1ACC to PA$^{\Delta MIL}$ to assess the RMS deviation contribution of the MIL. While surface loops and other protein periphery contributed to large RMS-displacements in excess of 11 Å, the backbone atoms comprising the major secondary structures in D1, D2 and D3 show relatively strong agreement. We found that most of the major deviations of backbone positions (> 1.2 Å) were localized in D4 (Fig. 3d).

Crystal structure of PA$^{\Delta MIL}$ at pH 6.5

During our analysis of structural changes in the relative orientation of D4, we realized that certain structures were obtained at slightly alkaline pH values of ~8.5 and other crystals were obtained at slightly acidic pH values of ~6.5. In order to rule out the possibility that the D4 movements we observe are simply due to differences in the pH of the crystallization condition, we crystallized PA$^{\Delta MIL}$ at pH 6.5. The structure was of similar quality to the others described thus far (Table 1). We then compared the structure of PA$^{\Delta MIL}$ at pH 6.5 to that obtained at pH 8.5 by computing the RMS displacement of the Cα backbone positions between the two different pH conditions. We found that the orientation of D4 is identical in the pH 8 and pH 6.5 crystal forms, and this change in pH does not dictate the relative orientation of D4 in PA (Fig. 3d).

The D2-D4 interface modulates the propensity of PA to form octameric oligomers

In order to study the contribution of PA's D2-D4 interface to its oligomeric preference, we co-assembled native and cross-linked PA and PA$^{\Delta MIL}$ constructs with WT LF$_N$ and analyzed the assembly products by negative-stain electron microscopy (Fig. 4). We obtained ~1500-5000 images of ring-shaped particles in each sample; these images were processed using single-particle alignment, classification, and class-averaging image analysis. From these analyses, we determine the proportion of heptameric and octameric oligomers in each sample.

Our data show that two separate processes control oligomeric preference. First, we find that when we co-assemble PA$^{\Delta MIL}$ with LF$_N$, the resulting PA oligomer products are enriched with ~55% octamers (Fig. 4). Under identical solution conditions, when we co-assemble WT PA with LF$_N$ we find only ~35% of the resulting complexes are octameric.[14] By further comparison, when WT and PA$^{\Delta MIL}$ assemble on their own, each produces ~2% and 24%

octamer, respectively.[14] Taken together these results imply two different mechanisms lead to octamer formation, and those mechanisms are additive. Second, the length of the cross-link between D4 and D2 changes the propensity of PA to form octameric complexes (Fig. 4). The shorter C_{337}-C_{664} disulfide cross-link produces ~24% and ~30% octamer in the PA and PA$^{\Delta MIL}$ backgrounds, respectively, which is slightly less than non-cross-linked PA or PA$^{\Delta MIL}$ when co-assembled with LF$_N$ (Fig. 4). Interestingly, extending the length of the cross-link by 3 carbon atoms in the C_{337}-C_{664}-DCA construct resulted in complexes that were highly enriched with octamers (>85%) for both the WT PA and PA$^{\Delta MIL}$ backgrounds (Fig. 4). Control constructs in which a single Cys residue was introduced showed no preference for the formation of the octameric complex. We surmise that cross-linking D4 to D2 alters the degree of conformational restriction in D4. Therefore, the interface between D2 and D4 plays an important role in determining the oligomeric preference of PA.

Correlation of the conformation and D4 and oligomeric heterogeneity

Since D4 appears to have the largest global RMS deviation when comparing the various PA monomer structures (Fig. 3d), we hypothesize that its position may provide the structural basis for determining PA oligomeric stoichiometry. To correlate D4 conformations with oligomeric stoichiometry, we aligned the Cα atoms of D1-D3 residues for PA (derived from 1ACC), acidic PA$^{\Delta MIL}$, PA$^{\Delta MIL}$ C_{337}-C_{664}, and PA$^{\Delta MIL}$ C_{337}-C_{664}-DCA to the corresponding residues from PA$^{\Delta MIL}$. The center-of-mass (COM) for each construct's D2 and D4 was determined, accounting for differences in sequence. We then defined the hinge angle (θ) for D4 movements. θ is defined by the center of the oligomer's lumen (COL), the COM of D2, and the COM of D4. These centers were projected to a plane orthogonal to the central pore axis through the COL, and the COM of D2 is the angle vertex (Fig. 5a). We find that θ increases relative to PA$^{\Delta MIL}$ for the WT PA and PA$^{\Delta MIL}$ C_{337}-C_{664}, which gave rise to a lower fraction of octamers, increasing by 2.5 degrees and 0.8 degrees, respectively. In contrast, θ for PA$^{\Delta MIL}$ at pH 6.5 and PA$^{\Delta MIL}$ C_{337}-C_{664}-DCA, the constructs that form an excess of octamer, increases by only 0.15 and 0.3 degrees, respectively (Fig. 5b). Furthermore, the distance displaced (Δd) between the COM of D4, relative to PA$^{\Delta MIL}$, changes by 1.4 Å and 0.5 Å for WT PA and PA$^{\Delta MIL}$ C_{337}-C_{664}, respectively, but it changes by only 0.2 Å for both PA$^{\Delta MIL}$ at pH 6.5 and PA C_{337}-C_{664}-DCA (Fig. 5b). Therefore, we conclude that there are two D4 conformations, termed Pro-PA$_7$ and Pro-PA$_8$, which differ in the linearity of D4 with respect to D2 and the oligomer lumen. These different conformations can give rise to lower and higher fractions of PA$_8$, respectively (Fig. 5c).

The furin-cleavage site

The loop containing the furin-cleavage site is ordered in our PA$^{\Delta MIL}$ structure (Fig. 6a). The structure of the loop begins as a β hairpin with the preceding loop, as R164 hydrogen bonds back to S161. Notably, none of the residues make contact with residues in the PA$_{63}$ domains, although the loop is in relatively close in proximity to the D2 loop containing F464. This result is expected given the fact that cleavage at the site functions to release PA$_{20}$ from the rest of PA; and previous studies reported that the presence of the furin-site loop does not affect the dissociation kinetics of the PA$_{20}$ fragment.[57] Several residues in the furin-cleavage loop make contact with PA$_{20}$. Most prominent is R164, whose guanidino group forms a salt bridge with E155, as well as ionic interactions with the ε-hydroxyl of Q152 and the backbone carbonyl of L153. The γ-hydroxyl of S160 forms a hydrogen bond with the amide nitrogen of K166, and the ε-amino group of K166 hydrogen bonds to the backbone of Q158. The side chain for R167, which is the P1 site residue for furin, is not well ordered; however, its backbone carbonyl forms a hydrogen bond with Q115. Finally, K117 enters a hydrogen-bond network with the carbonyl oxygens of T169 and P173 and loosely approaches one of the conformers for S168. We conclude that the furin-site loop is

freely accessible to solvent, makes numerous contacts with PA_{20} residues, and yet forms few interactions with PA_{63}.

We then asked whether protoxin processing by furin is dependent on the positioning of the site along the extended loop region. Using WT PA, we produced constructs in which the RKKR sequence was moved either six residues toward the N-terminus of PA (PA^{+6}) or five residues toward to the C-terminus (PA^{-5}). Samples were incubated with furin at various time points and analyzed by SDS-PAGE. While furin-dependent processing of WT PA is evident after 30 minutes and is essentially complete after 3 hours, virtually no processing was observed for either PA^{+6} or PA^{-5} (Fig. 6b). Therefore, we conclude that the positioning of the furin-cleavage site is important to proper protoxin processing, and all sites within the loop are not equally accessible to furin.

Furin-dependent PA processing is not ANTXR2-dependent

We were intrigued by the degree to which the furin-cleavage site was ordered in these crystal structures. Is ordering of the ANTXR2 binding site linked to the ordering of the furin-cleavage site in PA? We suspected that crystal packing contacts resulted in the stabilization of 3α1, which is a structure implicated in ANTXR2-binding. Also it is known that anthrax toxin assembly is coordinated on cell surfaces by means of cholesterol-rich microdomains.[25] Could receptor binding alter the dynamics of the furin-cleavage site to either stimulate or inhibit cleavage to allow PA to assemble within microdomains? We tested this hypothesis by monitoring furin-dependent processing of both PA and PA in complex with ANTXR2 using SDS-PAGE (Fig. 6b). We used both monomeric and dimeric ANTXR2 (dANTXR2) constructs to also test whether clustering of PA monomers affected the rate of furin processing (data from dANTXR2 not shown). The disappearance of the full-length PA band, and corresponding appearance of the PA_{63} band occurs at similar rates, that is, either when PA is free of receptor or it is complexed with dimeric or monomer versions of ANTXR2. Therefore, we conclude that the binding of PA to ANTXR2 has no effect on the rate of furin processing.

Discussion

Elucidating the thermodynamic mechanisms governing the formation of specific macromolecular architectures in the cell has remained a significant barrier to understanding the assembly and function of molecular machinery. Macromolecular assembly is a ubiquitous cellular process central to normal physiology as well as the mechanisms of microbial pathogenesis. Viruses and bacterial virulence factors, including anthrax toxin, must properly assemble for function. Because each PA subunit contains four folded domains, a number of functions including host-receptor recognition, oligomerization, channel formation, enzyme subunit binding, and enzyme activity may be attributed to each individual domain and/or inter-domain cross-talk. The fact that anthrax toxin assembles heterogeneously is a testament to the inherent complexity and flexibility involving the interactions of these multidomain subunits during assembly. Heterogenous assembly mechanisms can lead to new functions for the macromolecular machine, such as altered complex stability, dynamics and function. Furthermore, the complexity of the intra-domain and inter-domain interactions allow for allosteric regulators to bind and alter either the assembly or the function of the complex.

Due to the primacy of virus and virulence-factor assembly to pathogenesis, small-molecule inhibitors that disrupt or alter assembly may seem at first glance to be reasonable drug-development strategies; however, these types of therapeutics have lagged considerably relative to those that target metabolic pathways and traditional enzyme activities. The lack of assembly-targeted drugs is, in part, due to the difficulty of assembly assays compared to

enzyme assays or binary-component binding reactions. But more likely, the interfacial surface areas that define the thermodynamic interactions between subunits in these molecular machines are too extensive to impair or inhibit with a small-molecule drug by means of direct interference. Thus, the most-likely path to developing small molecules that inhibit virulence factor assembly would likely involve the triggering of a cryptic and/or native site that allosterically controls assembly. The allosteric-modulation strategy enables the development of a more practically-sized molecule to modulate the assembly pathway. It is in this spirit that we investigate the molecular mechanism of anthrax toxin assembly.

Protoxin maturation by furin

Since many viruses and toxins are secreted as proproteins, they must be proteolytically processed or activated prior to assembly. This mechanism ensures that these virulence factors can localize properly and infect targeted cells. In our structure of PA$^{\Delta \text{MIL}}$, we modeled residues 162-173, which contain the site for furin-dependent cleavage and subsequent activation of PA (Fig. 6a). Furin has been implicated as the protease responsible for the proprotein processing of a myriad of disease pathogens.[40-49] The crystal structure of furin, in complex with an inhibitor peptide, has been solved; thus the structural basis for furin specificity has been reported.[33] Aligning the PA sequence containing the furin-cleavage motif (S_{163}-R-K-K-R) to the dec-R-K-V-R-cmk inhibitor (Fig. 6c), where the amino and carboxy termini of the inhibitor peptide were modified with the decyl (dec) and carboxymethyl ketone (cmk) groups, respectively, shows the backbone atoms are in close agreement, with the exception of the P1 site, R167. In order to align well with the inhibitor, the peptide bond between K166 and R167 must be rotated. We find that residues remain well within the β-strand region of the Ramachandran plot upon rotation.[58] Given that R167 is freely accessible to solvent and its movement is not restricted in our structure of PA$^{\Delta \text{MIL}}$, we expect that this movement is not unreasonable. Successful rotation of R167 positions the scissile bond site in proximity to the catalytic Ser and His residues of furin and allows the residues carboxy to 167 to exit the furin catalytic site (Fig. 6d). Furthermore, we show that moving the furin-cleavage site either upstream or downstream results in a loss of protoxin processing in vitro (Fig. 6b). We propose that steric clashes with PA$_{20}$ residues accounts for the specific requirement of the furin-cleavage site positioning on L$_{158-173}$. The atomic model of PA presented here provides a starting point for understanding the structural basis for furin-dependent cleavage of PA. Furthermore, being a protease-sensitive site, the ordering of the furin loop in the PA$^{\Delta \text{MIL}}$ structure suggested that there was some level of cross talk between the furin loop, D2, and D4. We tested this hypothesis and found that receptor binding across D2 and D4 does not alter the rate of furin processing. Furin processing and PA assembly on cell surfaces, which appears to be coordinated with cholesterol-rich microdomains,[25] may be controlled through other means.

Furin is a critical housekeeping enzyme involved in proprotein and prohormone maturation. Protoxins have taken advantage of the ubiquitous convertase, using it to activate virulence factors and infect a plethora of cell types. While furin-specific inhibitors have emerged as attractive therapeutics for preventing protoxin activation, such treatments may also obstruct normal cell processes that rely on furin processing. A recent study found that the efficacy of furin targets transforming growth factor beta (TGFβ1) and matrix metalloprotein (MT1-MMP) were not affected by furin inhibition, and it was proposed that the inherent redundancy of cell-surface convertases might compensate for the loss in furin activity.[50] However, other furin targets have not been studied, and a broader picture of the side effects associated with inhibiting furin activity remains unclear. Here, we report for the first time the atomic structure of a furin-cleavage site, which may lead to the development of drugs that are specifically tailored to inhibiting the furin-protoxin interaction, rather than furin itself. Indeed, one such inhibitor that binds to the viral envelope protein gp160 in a region

proximal to the furin-cleavage site has been implicated as a possible treatment for HIV-1.[59] In this manner, the inhibitor interacts exclusively with the target antigen, preventing furin processing; thus, potential side effects arising from furin inhibition activity would be minimized. This avenue may be worth pursuing with anthrax toxin, as furin-loop accessibility can affect the rate of processing (Fig. 6b) and drug binding may lead to altered conformational dynamics.

Molecular determinants of PA's heterogeneous oligomerization mechanism

Another goal of this study is to obtain a molecular-level understanding of anthrax toxin assembly. How does the PA subunit assemble into a heterogeneous mixture of two different oligomeric architectures? From this effort we expect to draw broader conclusions about macromolecular assembly paradigms and to inspire novel therapies for anthrax disease. Previous studies indicate that PA subunits form mixtures of PA_7 and PA_8 complexes, yielding of a PA_8 content of ~5% to 30%. The PA_8 complex in particular is interesting because it is more pH-stable and thermostable than the more abundant PA_7 complex.[13] PA_8 complexes can survive in the plasma fraction of bovine blood with an ~30-minute half life, while the PA_7 complexes precipitate and inactivate in a few minutes.[13] Our present model is that allosteric control of the level of PA_8 complexes during pathogenesis may alter the level of toxicity achieved by the toxin. An alternative, non-mutually-exclusive model is that the heterogeneity allows for more efficient assembly on cell surfaces, preventing situations where the chance occurrence of odd- or even-numbered subcomplex stoichiometries would impede proper assembly if a strict oligomeric architecture were prerequisite for assembly.[14] In either case, it is likely that some level of control is present in the system, and this control determines the ultimate stoichiometry of the oligomeric mixture produced.

In previous work, we demonstrated that two separate phenomena produce relatively heterogeneous PA oligomer populations: (i) Addition of protein assembly co-factors that facilitate the formation of dimeric PA intermediates can drive PA_8 formation up to as high as ~30%; these proteins include the dANTXR2, LF, and EF. (ii) The $PA^{\Delta MIL}$ construct tends to form ~30% PA_8 complexes in the absence of dimerization-enhancing protein co-factors.[14] Mechanism (i) appears to have a straightforward intermolecular explanation, where dimerization favors the formation of the less prevalent even-numbered PA_8. However, mechanism (ii), albeit an artificial one, is not clear. Does the presence of the MIL trigger an allosteric site that prevents octamer formation? Or does the MIL simply block octamerization by a steric mechanism, where it would interfere with D4 from an adjacent PA in the oligomer? In this report, we explored the mechanism (ii), which appears allosteric in nature. These results describe for the first time that a complex network of interactions exist within the PA monomer, which can coordinate oligomeric architecture. Interestingly, when $PA^{\Delta MIL}$ is co-assembled with LF_N, PA_8 complexes are enriched >50% (Fig. 4), or 24% more than when $PA^{\Delta MIL}$ is assembled in isolation.[14] This result indicates that the mechanisms (i) and (ii) are in fact additive. Furthermore, we produced a PA construct whose D4 is loosely tethered to the main body of the protein via a dithioacetone cross-link. Independent of the presence of the MIL, this construct yields a highly enriched PA_8 stoichiometry—in excess of 85% (Fig. 4). Therefore, we find that removing the MIL and loosely tethering D4 to D2 have similar effects on PA_8 enrichment, albeit the latter is a noticeably stronger effect. On the other hand, tightly cross-linking D4 and D2 with a disulfide bond appears to inhibit the formation of the PA_8 complex relative to WT (Fig. 4). Thus the conformation of individual PA subunits influences the outcome of an assembly reaction.

Based upon our structural observations that D4 can occupy a Pro-PA_7 and Pro-PA_8 conformation, we propose that these two conformations predispose the PA subunit to form either respective oligomeric architecture. The structure of $PA^{\Delta MIL}$ and various cross-linked

variants reveal that the major conformational change in these monomers surrounds the hinge-like movement of D4 (Fig. 5b-c). D4 can adopt two different conformations, Pro-PA$_7$ and Pro-PA$_8$, which affect PA oligomer stoichiometry. Interestingly, the two conformations can be populated with different length cross-links tethering D4 to D2. The Pro-PA$_7$ conformation is favored with the shorter disulfide bond cross-link, while the Pro-PA$_8$ conformation is favored by the 3-atom extension inherent to the dithioacetone cross-link. We suspect that restricting the conformational space of the D2-D4 pitch angle with the dithioacetone cross-link essentially lowers the kinetic barrier for PA to adopt the Pro-PA$_8$ conformation. Furthermore, the propensity of a particular PA monomer to adopt either D4 conformation is not limited to the presence of the MIL, as the cross-link phenomenon is consistent with both WT PA and PA$^{\Delta MIL}$ backgrounds (Fig. 4). Thus steric interference by the MIL may not be the only factor influencing the oligomeric architecture, since its presence can be overridden by the D2-D4 dithioacetone cross-link (Fig. 4). The Pro-PA$_8$ conformation populated by the dithioacetone cross-link, therefore, also controls the PA oligomerization interface in a way that favors the PA$_8$ species over the PA$_7$ species. In conclusion, we propose that the flexibility and relative orientation of D4 holds an important key in deciding the outcome of PA's heterogeneous oligomerization pathway. We suspect that other heterogeneous oligomeric systems may also have similar mechanisms that control product stoichiometry. Future work should address how small-molecule, drug-like molecules can influence virulence factor assembly, stoichiometry, and function.

Materials and Methods

Protein expression and purification

An expression plasmid encoding the PA deletion mutant, PA$^{\Delta MIL}$, was produced by removing residues 305-324 while simultaneously introducing the mutations V303P and H304G.[14] The resulting PA construct replaces the membrane insertion loop (residues 303-324), leaving in its place a type-II turn; residue numbering is consistent with 1ACC.[23] Site-directed mutations of PA and PA$^{\Delta MIL}$ were produced using the QuikChange procedure (Agilent Technologies, Santa Clara, CA). PA and subsequent PA mutants were over-expressed in the periplasm of *Escherichia coli* BL21(DE3) and purified as PA monomers over Q-sepharose anion exchange (GE Biosciences), as described.[14] PA constructs used for crystallography were further purified over S200 gel filtration chromatography (GE Biosciences) in buffer A (20 mM Tris-Cl, 150 mM NaCl pH 8), and concentrated to 25 mg/mL.

For PA S337C N664C and PA$^{\Delta MIL}$ S337C N664C mutants, monomers were split into two fractions prior to storage at −80°C. The first fraction was allowed to form a disulfide bond under oxidizing conditions while 10 mM dithiothriotol (DTT, Gold Biotechnology, St. Louis, MO) was added to the second fraction. DCA-modified constructs were produced following a prior method.[55] Cys-containing PA proteins were immobilized on Q-sepharose anion exchange using N$_2$-purged buffers. ~5 mg/mL fractions of DTT-free S337C N664C protein were incubated with 1 mM 1,3-dichloroacetone (Sigma-Aldrich, St. Louis, MO) on ice for 30 min, and the subsequent labeled protein was purified by S200 gel filtration in buffer A.

LF$_N$ (LF residues 1-263) was over-expressed from a pET15b construct[60] in the cytoplasm of *E. coli* BL21(DE3) and purified as described.[14] The hexahistidine tags were removed by incubation with 0.5 units of bovine α-thrombin (Enzyme Research, South Bend, IN) per mg of LF$_N$ for 30 minutes at room temperature in buffer A supplemented with 2.5 mM CaCl$_2$ and 1 M dextrose.

Dimeric ANTXR2, corresponding to the soluble domain residues 40-217, was over-expressed from a pGEX vector (GE Healthcare) as a glutathione-S-transferase fusion protein in the cytoplasm of *E. coli* and purified on a glutathione-sepharose affinity column (GE Healthcare), as described.[19] Monomeric ANTXR2 was produced by transferring residues 40-217 from pGEX into pET15b using the NdeI and BamHI restriction sites, yielding a N-terminal His-tag construct. Site-directed mutagenesis yielded the mutant C175A. The protein was over-expressed in *E. coli* and purified by Ni^{2+}-affinity chromatography, as described.[17]

The PA^{+6} and PA^{-5} mutants were made using a three-step, gene-synthesis procedure, in a manner similar to a previous procedure.[24] Overlapping oligonucleotides encoding the desired sequences were synthesized (Elim Biopharmaceuticals, Inc., Hayward, CA) and amplified by two rounds of polymerase chain reaction (PCR). In Round I, 20 nM of nested oligonucleotides with consistent annealing temperatures of ~55 °C were amplified in a standard PCR reaction. In Round II, 1 μL of the PCR product made in Round I was amplified with the two outermost PCR primers (1 μM each) to make the synthetic double-stranded DNA fragment. These synthetic DNA fragments were ligated via a 5′ Hind III site and 3′ Kpn I site into the PA reading frame from a pET22b vector containing an in-frame, silent Kpn I restriction site in PA at V175.

Protein crystallization

Initial crystallization trials were carried out by screening ~400 sparse-matrix[61] conditions using a Mosquito nanoliter robot (TTP Labtech, Cambridge, MA) to set 200 nL drops by the hanging-drop vapor-diffusion method[62] in 96-well format. Crystal conditions were further investigated using manual 24-well format trays using a 1:1 ratio of 1 or 2 μL drops of protein and reservoir solution. For $PA^{\Delta MIL}$ pH 8.5 and PA C_{337}-C_{664}-DCA the reservoir solution was composed of 15-25% (w/v) polyethylene glycol monomethyl ether (PEG ME) with an average molecular weight of 2000 Da, 100 mM Tris-Cl, 200 mM trimethylamine *N*-oxide, pH 8.3-8.7. These proteins formed trapezoidal prism-shaped crystals overnight, maturing to dimensions 300-800 μm. Crystals were harvested in an antifreeze solution containing a 1:1 mixture of 50% (v/v) PEG ME with an average molecular weight of 550 Da, and reservoir solution supplemented with 150 mM NaCl. $PA^{\Delta MIL}$ pH 6.5 and $PA^{\Delta MIL}$ C_{337}-C_{664} reservoir solution contained 29-32% (v/v) pentaerythritol ethoxylate (15/4 EO/OH), 50 mM bis-tris Cl, 50 mM ammonium sulfate, pH 6.5-6.7. These proteins formed coffin-shaped crystals within 24 hours, maturing to dimensions 200-400 μm, and the crystals were harvested directly out of the drop. All crystal forms were flash-frozen by rapidly plunging the crystal into liquid N_2.

X-ray diffraction data collection, solution and refinement

X-ray diffraction data were collected at a wavelength of 1.1159 Å (11111 eV) and a temperature of 100 K at the Lawrence Berkeley National Lab's Advanced Light Source, Beamline 8.3.1 using a Quantum 315r CCD area detector (ADSC, Poway, CA).[63] Crystals containing $PA^{\Delta MIL}$ pH 8.5 diffracted X-rays to 1.45 Å in the orthorhombic space group $P2_12_12_1$ with unit cell dimensions of $a = 72.36$ Å, $b = 93.39$, $c = 118.08$, and angles, α, β, and γ, of 90°. Crystals containing PA C_{337}-C_{664}-DCA diffracted X-rays to 1.83 Å in the orthorhombic space group $P2_12_12_1$ with unit cell dimensions of $a = 71.29$ Å, $b = 93.69$, $c = 117.85$, and angles, α, β, and γ, of 90°. For both samples, a second data set was collected from the same crystal with a shorter exposure time to obtain the overloaded, low angle reflections. Crystals containing $PA^{\Delta MIL}$ C_{337}-C_{664} diffracted X-rays to 2.09 Å in the orthorhombic space group $P2_12_12_1$ with unit cell dimensions of $a = 71.56$ Å, $b = 94.60$, $c = 119.80$, and angles, α, β, and γ, of 90°. Crystals containing $PA^{\Delta MIL}$ pH 6.5 diffracted X-rays to 1.70 Å in the orthorhombic space group $P2_12_12_1$ with unit cell dimensions of $a = 71.40$

Å, b = 93.93, c = 117.85, and angles, α, β, and γ, of 90°. The diffraction data were indexed, scaled, and merged in HKL2000.[64] Molecular replacement for PA$^{\Delta MIL}$ pH 8.5 was carried out in PHASER,[65] using a loop-stripped model of PA (1ACC)[23] as the search model. Missing loops with significant (σ > 2.5) F_o-F_c difference density were built manually in COOT,[66] and the electron density was corroborated with composite simulated-annealing omit maps generated in PHENIX.[67] The coordinates and ADP values for all atoms were refined individually and anisotropically, respectively, in PHENIX, followed by further rounds of model building and solvent picking in COOT. Some of the alternate side-chain conformations were identified using RINGER,[68] and occupancy refinement was carried out in PHENIX. Structure solution and refinement for PA C_{337}-C_{664}-DCA and PA$^{\Delta MIL}$ pH 6.5 was carried out similarly, accept that PA$^{\Delta MIL}$ was used as the search model, and ADP values were refined isotropically. A similar strategy was also employed for PA$^{\Delta MIL}$ C_{337}-C_{664} except that atomic coordinates were refined both individually and with rigid body groups, and ADP values were refined isotropically. Two molecules of 2-methoxyethanol were identified in the density for PA$^{\Delta MIL}$ pH 8.5 and PA C_{337}-C_{664}-DCA, as well as an acetone molecule linking Cys337 and Cys664 for PA C_{337}-C_{664}-DCA. The oxygen atom on the ketone was not resolved; therefore, its occupancy was set to 0. The final model geometry for all structures was validated using MOLPROBITY[69] and PROCHECK[70]. Least-squares quotient (LSQ) Cα alignment of protein backbones was carried out in COOT.[66] RMS deviations of protein backbone were computed with COMPAR in the CCP4 program suite.[71] All molecular graphics were generated using CHIMERA.[72]

Furin proteolysis of PA

WT PA and PA mutants (1 mg/mL) were incubated with 10 units of furin (New England Biolabs, Ipswitch, MA) per mg of PA in buffer A supplemented with 1 mM $CaCl_2$ at room temperature, as described.[73] For the ANTXR2-PA proteolysis experiments, monomeric ANTXR2 C175A or dimeric GST-linked ANTXR2 was incubated with PA in a 4:1 molar ratio in buffer A supplemented with 1 mM $CaCl_2$ and 1 mM $MgCl_2$ for 10 minutes at room temperature. The co-complex was then processed with 20 units of furin per mg of PA. Samples were separated on a 10% SDS-PAGE gel and stained with Coomassie Brilliant Blue (Bio-RAD Laboratories, Hercules, CA), and the Pageruler protein ladder (Fermentas Life Sciences, Glen Burnie, MD) was used to determine the molecular weights of the products.

LF_N-driven assembly of $_nPA$ constructs

PA constructs (1 mg/mL) were nicked with trypsin (Sigma-Aldrich) at a ratio of 1:1000 (w/w) for 15 minutes at room temperature and then treated with soybean tryspin inhibitor (Worthington Biochemical, Lakewood, NJ) at 1:100 (w/w). The resulting $_nPA$ mixture was co-assembled with LF_N in a 1:1 molar ratio for 1 hour at room temperature. Samples were purified over S200 gel filtration in buffer A.

Electron microscopy

Grid preparation, data collection, and image processing were performed as described.[14] Briefly, each PA oligomer was diluted to 100 nM (with respect to PA monomer concentration) in buffer A supplemented with 0.001% (w/v) n-dodecyl-β-D-maltopyranoside (Affymetrix, Maumee, OH). PA complexes containing Cys mutations were diluted in buffer containing 5 mM DTT to improve image quality. 400 mesh copper grids were successively covered by continuous carbon film via a formvar support. 4 μl of PA sample was applied to the grid for 1 minute, washed in 3 successive drops of water, and then stained with 2% uranyl acetate (Sigma-Aldrich). Negative-stain EM images were recorded with a Tecnai 12 (FEI Company, Hillsboro, OR) operated at 120 kV at 49,000× magnification. Images were taken using a CCD camera at a 2.13-Å/pixel specimen scale. Particle images were selected

for each data set using automatic or manual particle picking using boxer in EMAN.[74] Reference-free processing was computed using SPIDER.[75] Images were subjected to three successive cycles of multi-reference alignment, multivariate statistical analysis, and classification, such that the last classification was done using only the lowest order eigenvectors.[78] Following the last classification, we separated the images by size and by the heptameric and octameric oligomerization states. Also we used a second method of image processing, whereby crystal-structure-reference images were made from two-dimensional projections of low resolution density maps generated from the crystal structures of the PA heptamer[18] and octamer[14] in SPIDER.[75] Crystal-structure-referenced images were aligned and classified using the lowest order eigenvectors. Final class-average images were manually inspected as being heptameric, octameric, or unclassifiable "junk" (<5% of the total particles). The number of particles per identifiable classification was used to determine the percentages of heptamers and octamers. Approximately 1500-5000 total particles were analyzed per sample. Either classification procedure produces similar results for the analysis of the composition of PA oligomer samples, agreeing with other biophysical measures of oligomeric heterogeneity.

Acknowledgments

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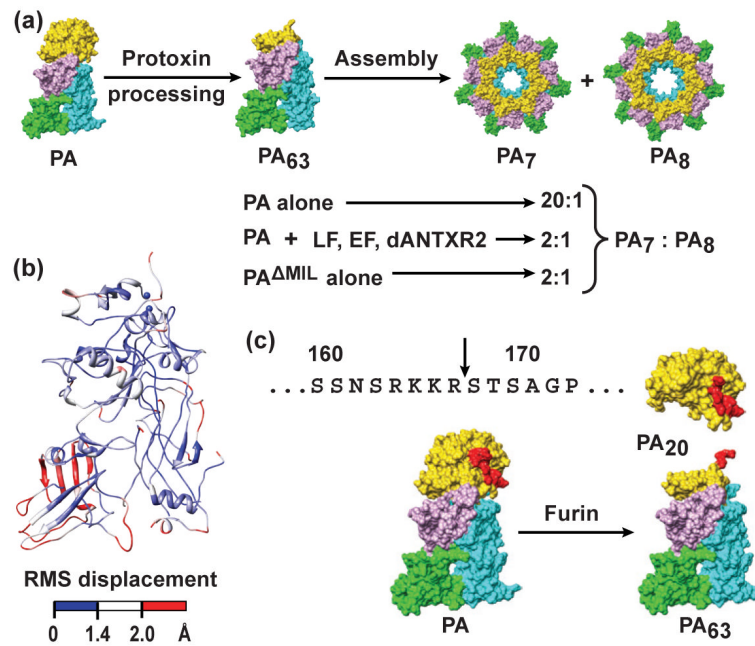
Figure 1. Anthrax toxin assembly is heterogeneous and requires proteolytic processing
(a) Cartoon depiction of anthrax toxin assembly. PA domains are colored as follows: D1 (gold), D2 (cyan), D3 (violet), and D4 (green). Upon oligomerization, PA forms both PA_7 and PA_8 complexes. The oligomeric ratio of PA complexes (PA_7:PA_8) decreases from about 20:1 for unliganded PA assembly to 2:1 in the presence of ligand (LF, EF, or dANTXR2). The $PA^{\Delta MIL}$ construct can form complexes at a 2:1 oligomeric ratio; however, it does so in the absence of ligand. **(b)** Global Cα alignment of a PA subunit (chain G) from the PA_7 crystal structure (PDB 1TZO)[18] onto a PA subunit (chain G) from the PA_8 crystal structure (PDB 3HVD).[14] The ribbons are colored based on the RMS displacement of the Cα positions. Small, intermediate and large RMS displacements are colored blue, white and red, respectively. **(c)** (Above) The PA sequence containing the furin-cleavage site, where the scissile bond is indicated with an arrow. (Below) The furin-dependent cleavage reaction of the PA proprotein releases the 20-kDa fragment (PA_{20}) and yields the 63-kD portion, called PA_{63}. PA_{63} ultimately assembles into PA_7 and PA_8 oligomers.
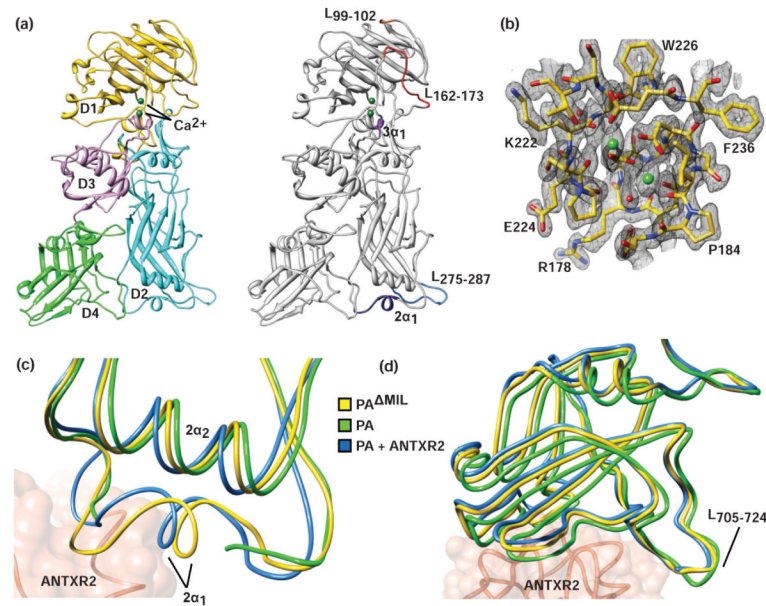
Figure 2. The structure of PA$^{\Delta MIL}$

(a) (left) Ribbons depiction of PA$^{\Delta MIL}$. D1 (gold), D2 (cyan), D3 (violet), D4 (green), and calcium ions (dark green). (right) Loop structures not present in 1ACC[23] are indicated as follows: L$_{99-102}$ (orange), L$_{162-173}$ (red) containing the furin-cleavage site, L$_{275-287}$ (blue), the loop containing 2α$_1$ (navy blue), and 3α$_1$ (purple). **(b)** Sample 2F_o-F_c electron density calculated at the end of refinement to 1.45 Å for the twin Ca^{2+}-coordination site contoured at σ=2.5. **(c)** LSQ Cα alignment of residues 16-340 for PA monomer (green, 1ACC),[23] ANTXR2-bound PA (blue, 1T6B)[17] and PA$^{\Delta MIL}$ (yellow). ANTXR2 (coral) from IT6B is included for reference. **(d)** Relative D4 orientation for PA monomer, ANTXR2-bound PA, and PA$^{\Delta MIL}$ (colored as in panel c) LSQ Cα superposition for residues 25-595.
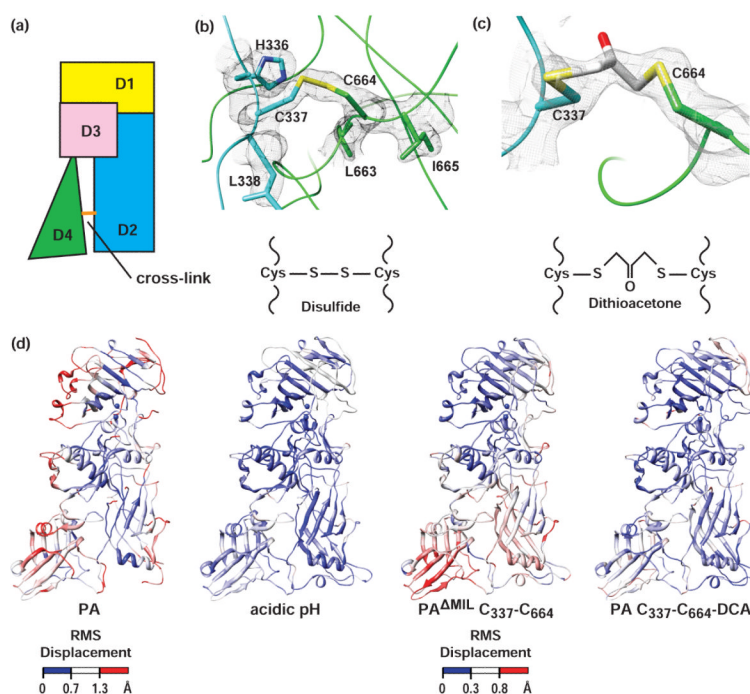
Figure 3. Cysteine cross-linking of PA D2 and D4
(a) Cartoon showing the location of the Cys-Cys cross-link introduced into PA, where domains are colored as in Fig. 1a. Two different cross-links were introduced via either a disulfide tether or the bifunctional cross-linking reagent, 1,3-dichloroactetone. Simulated-annealing $2F_o$-F_c electron density map calculated at the end of refinement for **(b)** PA$^{\Delta MIL}$ C$_{337}$-C$_{664}$ and **(c)** PA C$_{337}$-C$_{664}$-DCA, contoured at σ=2.0 and σ=1.5, respectively, for which Cys337 and Cys664 and the bridging acetone cross-linker have been omitted. Sulfur and oxygen atoms are colored yellow and red, respectively. **(d)** Global Cα alignment of PA$^{\Delta MIL}$ to (left to right) PA from PDB accession code 1ACC, PA$^{\Delta MIL}$ acidic pH, PA$^{\Delta MIL}$ C$_{337}$-C$_{664}$ and (right) PA C$_{337}$-C$_{664}$-DCA and the corresponding RMS deviations colored from low (blue) to high (red).
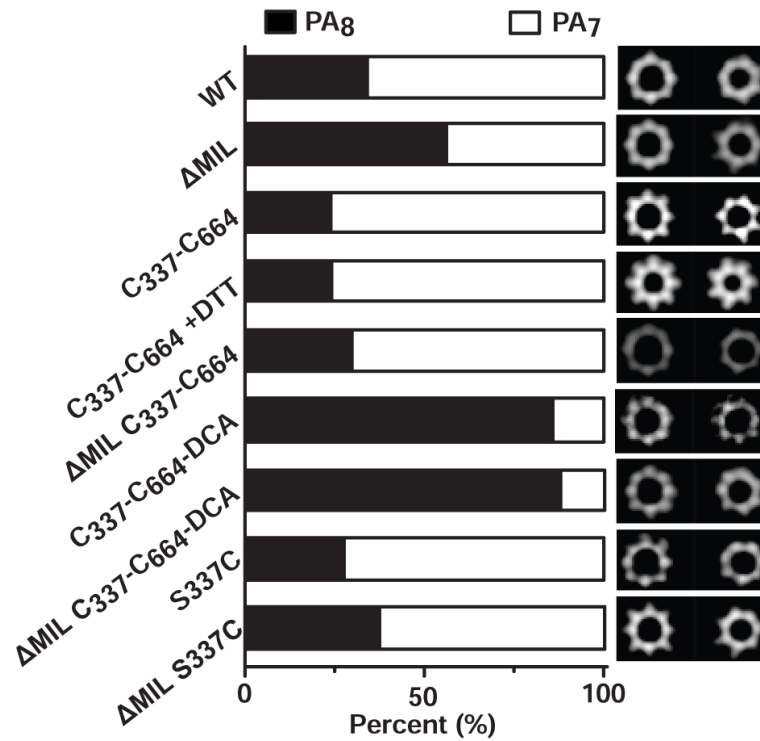
Figure 4. The interface of PA D2 and D4 controls oligomeric stoichiometry
The percentages of PA_7 (white) and PA_8 (black) complexes are given following the assembly of various PA constructs in the presence of LF_N. Representative class-average images for octamers (left) and heptamers (right) are shown at the right. The number (*N*) of particles analyzed are 2218, 3241, 2534, 2147, 2054, 1972, 1535, 1750 and 4845 for PA WT, $PA^{\Delta MIL}$, PA C_{337}-C_{664}, PA C_{337}-C_{664}+DTT, $PA^{\Delta MIL}$ C_{337}-C_{664}, PA C_{337}-C_{664}-DCA, $PA^{\Delta MIL}$ C_{337}-C_{664}-DCA, PA S337C and $PA^{\Delta MIL}$ S337C, respectively. Note: "+DTT" indicates that 10 mM of DTT was included in the oligomerization reaction.
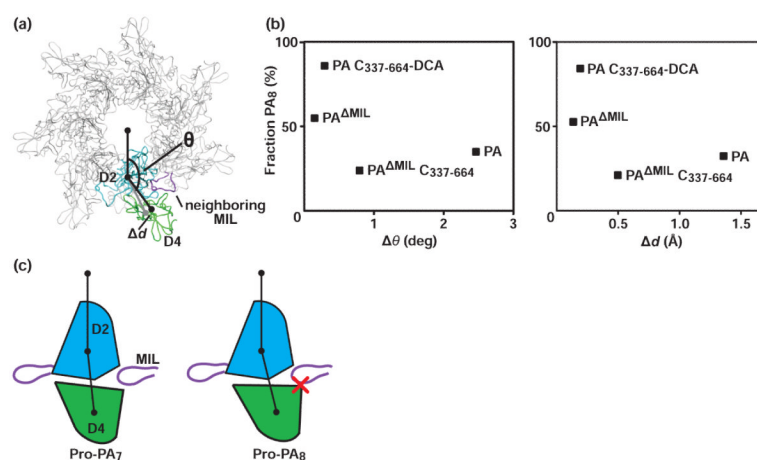
Figure 5. Molecular basis for PA oligomeric stoichiometry

(a) A model showing the angle, θ, which is used as a metric relating the observed movement of D4 relative to the amino-terminal domains of the PA_{63} moiety. θ is defined by the center of the oligomeric lumen (COL) and the centers of mass (COM) of D2 and D4. Also indicated in the model is the metric for the displacement of the D4 COM, called Δd. The PA_7 coordinates, containing a model of the MIL, are from 1TZO.[18] D2, D4, and the MIL of the adjacent PA subunit are colored cyan, green, and purple, respectively. **(b)** Deviations in θ ($\Delta\theta$) (left) and Δd (right) are plotted against the percentage of PA_8 oligomer produced (**as shown in** Fig. 4). The deviations are computed relative to the reference model from $PA^{\Delta MIL}$: $\Delta\theta = \theta(PA^{\Delta MIL}) - \theta(PA_i)$, and $\Delta d = d(PA^{\Delta MIL}) - d(PA_i)$, where PA_i is the PA monomer measured. **(c)** A molecular mechanism for PA oligomerization, where the hinge-like movement of D4 can lead to the population of two different population of PA that favor the formation of PA_7 and PA_8, called Pro-PA_7 (left) and Pro-PA_8 (right), respectively. In Pro-PA_8 D4 is rotated and translated relative to D4 for Pro-PA_7. The Pro-PA_8 conformation is sterically inhibited by the presence of the adjacent PA subunit's MIL during assembly. Presumably, the intramolecular dithioacetone bridge linking D4 to D2 favors the Pro-PA_8 conformation in both the presence and absence of a neighboring MIL.
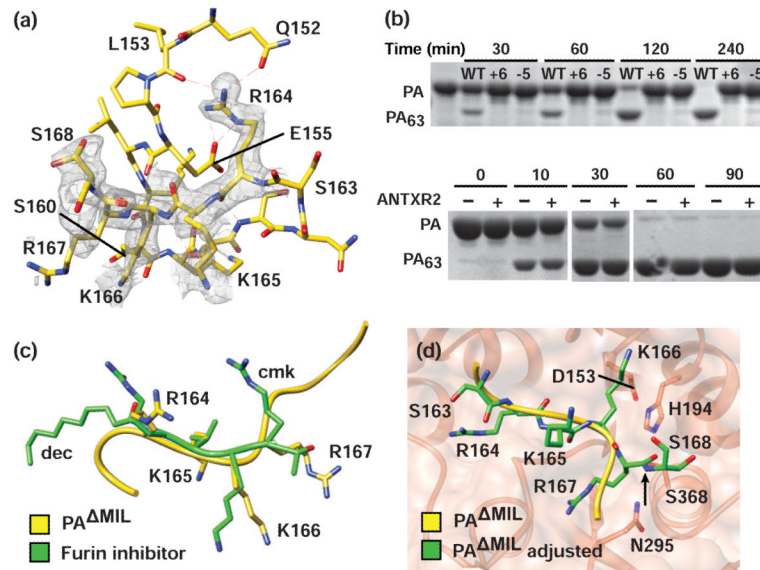
Figure 6. Structural basis for furin-dependent cleavage of PA
(a) Positioning of the furin-cleavage site loop relative to the rest of PA_{20}. Electrostatic interactions are represented as dashed red lines; carbon (yellow), oxygen (red), and nitrogen (blue) atoms are indicated. Sample composite simulated-annealing electron density (gray mesh) calculated at 1.45 Å at the end of refinement for PA residues 164-168 is contoured at σ=1. **(b)** SDS-PAGE assay for furin processing of (top) PA_{83}, PA^{+6}, and PA^{-5} and (bottom) PA and ANTXR2-bound PA. Proteins and incubation times as well as molecular weights corresponding to PA and PA_{63} are indicated above and to the left, respectively. **(c)** LSQ all-atom alignment of PA residues 164-167 with the furin-inhibitor peptide from 1P8J.[33] **(d)** Placement of the aligned (yellow) PA furin-cleavage site in the active site of furin (1P8J).[33] Rotation around the peptide bond linking residues K166 and R167 (green) relieves steric clashes and positions the scissile bond in proximity to the catalytic S368 of furin.

Table 1
Data collection and refinement statistics

Data collection	PA$^{\Delta MIL}$ pH 8.5	PA$^{\Delta MIL}$ C$_{337}$-C$_{664}$	PA C$_{337}$-C$_{664}$-DCA	PA$_{\Delta MIL}$ pH 6.5
Space group	$P2_12_12_1$	$P2_12_12_1$	$P2_12_12_1$	$P2_12_12_1$
Cell dimensions				
a, b, c (Å)	72.36, 93.39, 118.08	71.56, 94.60, 119.80	71.29, 93.69, 117.85	71.40, 93.93, 117.85
α, β, γ(°)	90, 90, 90	90, 90, 90	90, 90, 90	90, 90, 90
Wavelength (Å)	1.1159	1.1159	1.1159	1.1159
Resolution (Å)	23.0-1.45 (1.49-1.45)[a]	25.06-2.06 (2.09-2.06)[b]	24.9-1.83 (1.87-1.83)[c]	9.99-1.70 (1.73-1.70)[d]
R_{merge} (%)	9.6(74.8)	11.4(86.8)	12.0(75.7)	8.5(85.5)
$I/\sigma I$	13.9(1.9)	15.7(2.5)	20.9(4.3)	22.1(2.1)
Completeness (%)	90.3(81.0)	99.9(99.2)	99.7(96.0)	99.8(98.0)
Redundancy	6.5(3.4)	7.6(7.7)	14.2(10.5)	8.1(7.4)
Wilson B	27.000	36.331	26.490	27.278
Refinement				
Resolution (Å)	23.0-1.45	23.86-2.06	24.9-1.83	9.99-1.70
No. reflections	126,263	50,702	70,160	87,171
R_{work}, R_{free} (%)	19.4/ 21.6	20.7 / 23.4	19.3 / 22.6	19.4 / 21.7
No. atoms				
Protein	5647	5589	5616	5590
Ligand/ion	17	2	13	2
Water	348	208	422	398
Average ADP-values				
Protein	21.7	47.8	36.6	40.3
Ligand/ion	29.5	31.0	33.4	27.0
Water	29.9	42.9	39.3	43.0
RMS deviations				
Bond lengths (Å)	0.008	0.008	0.016	0.007
Bond angles (°)	1.151	1.041	1.087	1.053
Ramachandran (%)				
Favored	98.1	97.9	98.1	97.1
Outliers	0	0	0	0
MolProbity score	1.51	1.93	1.58	1.86

[a] Values in parenthesis refer to data in the highest resolution shell (1.49-1.45 Å).

[b] Values in parenthesis refer to data in the highest resolution shell (2.13-2.09 Å).

[c] Values in parenthesis refer to data in the highest resolution shell (1.87-1.83 Å).

[d] Values in parenthesis refer to data in the highest resolution shell (1.87-1.83 Å).