

# Effects of Carbonyl Sulfide and Carbonic Anhydrase on Stomatal Conductance<sup>1[OA]</sup>

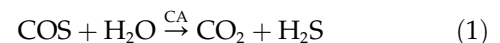
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The potential use of carbonyl sulfide (COS) as tracer of CO<sub>2</sub> flux into the land biosphere stimulated research on COS interactions with leaves during gas exchange. We carried out leaf gas-exchange measurements of COS and CO<sub>2</sub> in 22 plant species representing deciduous and evergreen trees, grasses, and shrubs, under a range of light intensities, using mid-infrared laser spectroscopy. A narrow range in the normalized ratio of the net uptake rates of COS ( $A^s$ ) and CO<sub>2</sub> ( $A^c$ ), leaf relative uptake ( $A^s/A^c \times [\text{CO}_2]/[\text{COS}]$ ), was observed, with a mean value of  $1.61 \pm 0.26$ , which is advantageous to the use of COS in photosynthesis research. Notably, increasing COS concentrations between 250 and 2,800 pmol mol<sup>-1</sup> (enveloping atmospheric levels) enhanced stomatal conductance ( $g_s$ ) to a variable extent in most plants examined (up to a normalized enhancement factor [ $f_e = (g_{s-\text{max}} - g_{s-\text{min}})/g_{s-\text{min}}$ ] of 1). This enhancement was completely abolished in carbonic anhydrase (CA)-deficient antisense lines of both C3 and C4 plants. We suggest that the stomatal response is mediated by CA and may involve hydrogen sulfide formed in the reaction of COS and water with CA. In all species examined, the uptake rates of COS and CO<sub>2</sub> were highly correlated, but there was no relationship between the sensitivity of stomata to COS and the rate of COS uptake (or, by inference, hydrogen sulfide production). The basis for the observed stomatal sensitivity and its variations is still to be determined.

Carbonyl sulfide (COS) is a ubiquitous constituent of the atmosphere. Its concentration in the background atmosphere is  $500 \pm 100$  pmol mol<sup>-1</sup> (i.e. about a factor of 1 million less abundant than CO<sub>2</sub>; Montzka et al., 2007), but its concentration near vegetation may vary over a much wider range depending on proximity to sources, such as biomass burning or urban pollution, or sinks, such as leaves and soils (Montzka et al., 2007; Blake et al., 2008; Campbell et al., 2008). We recently published studies that focused on the physiology of COS uptake by leaves, advancing the goal of using measurements of this trace gas to help quantify the contributions of gross primary productivity and ecosystem respiration to net carbon exchange at local and regional scales (Campbell et al., 2008; Stimler et al., 2010a, 2011). This approach relies on the knowledge of the relative ratio of the COS/CO<sub>2</sub> uptake rates at the leaf level [leaf relative uptake {LRU} =  $(A^s/A^c) \times (C_a^c/C_a^s)$ , where  $A$  is uptake rate,  $C_a$  is ambient concentration, and superscripts  $s$  and  $c$  denote COS or CO<sub>2</sub>, respectively]. Both COS and CO<sub>2</sub> fluxes into leaves are

influenced by physical limitations along the diffusion pathway (Kluczewski et al., 1985; Goldan et al., 1988; Stimler et al., 2010a), followed by hydration reactions. COS reaction with the enzyme carbonic anhydrase (CA) in the presence of water results in the production of CO<sub>2</sub> and hydrogen sulfide (H<sub>2</sub>S) in an exergonic, one-way, reaction (Protoschill-Krebs et al., 1996; Yonemura et al., 2005; Liu et al., 2010):



Equation 1 indicates the important role of CA in COS uptake and its potential significance in modifying the concentrations of H<sub>2</sub>S inside leaves. H<sub>2</sub>S, in turn, has been implicated in a range of possible biological effects (for a recent review, see Wang, 2010). Limited information is available, however, on the variations in the rate of this process and its relation to the rate of CO<sub>2</sub> uptake among plant species (Sandoval-Soto et al., 2005; Yonemura et al., 2005; Stimler et al., 2010a, 2011).

In this study, we focus on the effect of COS concentration on stomatal conductance ( $g_s$ ). This effect was first noted in an early study of COS uptake by plants, which showed that COS within the range of natural variation appeared to influence  $g_s$  (Goldan et al., 1988). Recently, we confirmed this observation, showing that increasing COS concentration from 0 to 2,500 pmol mol<sup>-1</sup> (parts per trillion) resulted in a large increase in  $g_s$  in leaves of three species of C3 plants under otherwise constant and optimal conditions (Stimler et al., 2010a). Given these observations, we suggest that COS concentration may be a significant and hitherto unrecognized variable in studies of  $g_s$ . For example, gases

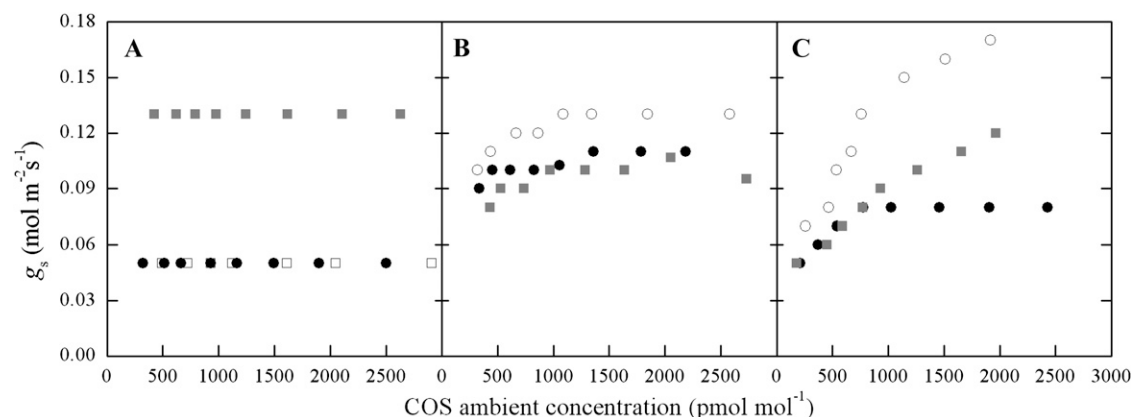
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**Figure 1.**  $g_s$  ( $\text{mol m}^{-2} \text{s}^{-1}$ ) during COS response experiments in representative plants from each group reported in Table I, with normalized  $g_s$  of  $f_e < 0.1$  (A),  $f_e < 0.3$  (B), and  $f_e > 0.3$  (C). For the variations in  $f_e$  in each group, see Table I. Experiments were conducted under light intensity of  $1,189 \mu\text{mol photons m}^{-2} \text{s}^{-1}$ , atmospheric concentration of  $\text{CO}_2$  (approximately  $400 \mu\text{mol mol}^{-1}$ ), temperature of approximately  $23^\circ\text{C}$ , and RH of approximately 75%. A, *Citrus maxima* (white squares), *Citrus madurensis* (black circles), and *Quercus robur pedunculiflora* (gray squares). B, *Crocoshmia*  $\times$  *crocoshmiiflora* [*aurea*  $\times$  *pottsii*] (white circles), *Diospyros virginiana* (black circles), and *Cestrum nocturnum* (gray squares). C, *Eucalyptus camaldulensis* (white circles), *Antigonon leptopus* (black circles), and *Diospyros digyna* (gray squares).

used in laboratory studies may or may not contain COS. Artificial air mixed from standard grades of  $\text{N}_2$ , oxygen, and  $\text{CO}_2$  in our laboratory is free of COS, while urban air might contain  $2,000 \text{ pmol mol}^{-1}$  COS. Uncontrolled variation in COS concentration could complicate the interpretation of studies of  $g_s$ . We also

note that this effect may provide insights into the mechanisms regulating  $g_s$ .

The objective was, first, to examine the stimulation of  $g_s$  by COS in a range of species including major functional groups (deciduous and evergreen trees, shrubs, and grasses) and both C3 and C4 photosyn-

**Table I.** Minimum and maximum rates of COS flux ( $A^s$ ;  $\text{pmol m}^{-2} \text{s}^{-1}$ ),  $g_s$  ( $\text{mmol m}^{-2} \text{s}^{-1}$ ), and the enhancement factor ( $f_e$ ) during COS response experiments

$f_e = (g_{s\text{-max}} - g_{s\text{-min}}) / g_{s\text{-min}}$ . SD values for measurements on different leaves are indicated ( $n = 4-6$ ) as well as means and SD of individual  $f_e$  values. Experiments were conducted under atmospheric concentrations of  $\text{CO}_2$  (approximately  $400 \mu\text{mol mol}^{-1}$ ), temperature of approximately  $23^\circ\text{C}$ , RH of approximately 75%, and minimum and maximum concentrations of COS were approximately 250 and  $2,000 \text{ pmol mol}^{-1}$ .

Species	Type	$A^s$		$g_s$		$f_e$
		Minimum	Maximum	Minimum	Maximum	
$f_e < 0.1$						
<i>Agapanthus africanus</i>	Grass	5.0 (0.3)	24.0 (0.6)	40	44 (5.7)	0.10 (0.14)
<i>Citrus madurensis</i>	Evergreen	2.2 (0.1)	18.2 (3.7)	40 (14)	43 (9.9)	0.10 (0.14)
<i>Ficus neriifolia</i>	Evergreen	3.5	45.1	20	20	0.00
<i>Macadamia</i>	Evergreen	2.8 (1.8)	31.8 (0.5)	65 (21)	70 (28.3)	0.06 (0.09)
<i>Quercus robur pedunculiflora</i>	Deciduous	2.8 (2.7)	41.0 (1.1)	120 (14)	129 (2.12)	0.08 (0.11)
$f_e < 0.3$						
<i>Cestrum nocturnum</i>	Shrub	2.9 (2.5)	27.8 (0.3)	60 (28)	70 (28.3)	0.19 (0.09)
<i>Citrus maxima</i>	Evergreen	1.1 (1.3)	33.4 (11.6)	50	60 (14.1)	0.20 (0.28)
<i>Diospyros virginiana</i>	Deciduous	2.8 (0.8)	21.1 (0.3)	110 (28)	125 (21.2)	0.15 (0.10)
<i>Jasminum sambac</i>	Shrub	12.4 (10.3)	37.6 (5.7)	95 (21)	105 (35.4)	0.09 (0.13)
<i>Passiflora edulis</i>	Shrub	8.4 (3.7)	32.7 (3.7)	90 (42)	110 (42.4)	0.25 (0.38)
<i>Quisqualis indica</i>	Deciduous	1.0 (0.4)	33.2 (7.6)	85 (35)	95 (35.4)	0.13 (0.05)
<i>Viburnum tinus</i>	Shrub	2.1 (0.8)	23.7 (14.2)	50 (20)	57 (11.5)	0.22 (0.38)
$f_e > 0.3$						
<i>Antigonon leptopus</i>	Deciduous	11.3 (7.3)	36.8 (0.7)	115 (92)	155 (106)	0.44 (0.23)
<i>Belamcanda chinensis</i>	Grass	17.7	44.6	50	100	1.00
<i>Crocoshmia</i> $\times$ <i>crocoshmiiflora</i> [ <i>aurea</i> $\times$ <i>pottsii</i> ]	Grass	9.6 (0.8)	80.0 (8.2)	115 (21)	150 (28.3)	0.30 (0.01)
<i>Eucalyptus camaldulensis</i>	Evergreen	4.6 (4.9)	56.0 (3.5)	70	135 (75)	0.93 (0.71)
<i>Flaveria bidentis</i>	Herbaceous	5.5 (0.03)	53.8 (0.03)	191 (31)	266 (34)	0.44 (0.32)
<i>Limonium perezii</i>	Shrub	3.0 (0.4)	27.9 (5.5)	55 (7)	75 (7.0)	0.37 (0.05)
<i>Nicotiana tabacum</i>	Herbaceous	4.5 (3.8)	51.6 (12.4)	125 (25)	175 (7.1)	0.44 (0.17)
<i>Salvia longispicata</i> $\times$ <i>Salvia farinacea</i>	Evergreen	5.3 (4.4)	95.8 (6.8)	166 (42)	220 (72.1)	0.30 (0.12)
<i>Diospyros digyna</i>	Evergreen	1.8 (1.5)	34.0 (5.4)	60 (14)	105 (21.2)	0.84 (0.79)

thetic types, and second, to take advantage of existing antisense constructs to the enzyme CA (Price et al., 1994; Cousins et al., 2006) to examine the importance of this enzyme for both the uptake of COS and the enhancement of  $g_s$  by COS. Since CA catalyzes the conversion of COS to  $\text{CO}_2$  and  $\text{H}_2\text{S}$ , the involvement of CA in the stomatal response to COS may also indicate the participation of  $\text{H}_2\text{S}$  produced in the mesophyll.

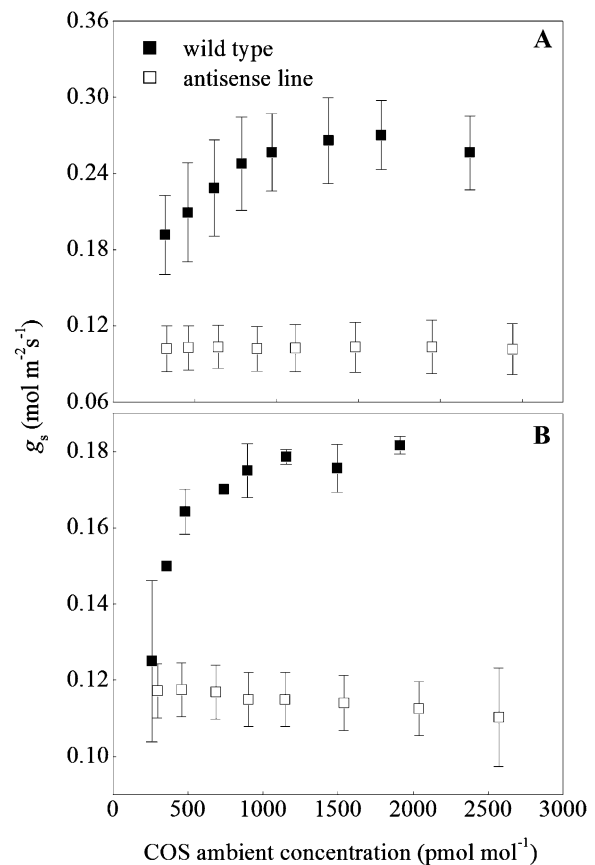
## RESULTS AND DISCUSSION

### $g_s$ Response to COS

We recently reported (Stimler et al., 2010a) increasing  $g_s$  in response to increasing ambient COS concentrations within the range observed under natural conditions (Montzka et al., 2007). Here, we extend this study to examine the variations in the  $g_s$  response to COS among 22 plant species exposed to ambient COS concentrations in the range of 250 to 2,800  $\text{pmol mol}^{-1}$  (enveloping the mean atmospheric concentration of approximately 500  $\text{pmol mol}^{-1}$ ). The response of  $g_s$  with increasing COS was quite variable and could not be easily characterized by vegetation or functional type (Fig. 1). We calculated the relative enhancement ( $f_e$ ) for each species across the COS range used [ $f_e = (g_{s\text{-max}} - g_{s\text{-min}})/g_{s\text{-min}}$ ] and grouped the species in Table I and Figure 1 as follows: (1) no effect ( $f_e < 0.1$ ); (2) moderate effect ( $0.1 > f_e > 0.3$ ); and (3) high effect ( $f_e > 0.3$ ). The  $g_s$  enhancement observed here showed different characteristics from those reported by Goldan et al. (1988) that indicated a sharp reduction in leaf resistance, mainly at low, subambient COS concentrations. Stimler et al. (2010a) reported large enhancements of up to  $f_e$  of about 2 in *Rosa sinensis*, *Salvia officinalis*, and *Capsicum annuum*, with a linear response across a range of ambient COS concentrations. At present, we do not understand the basis for these differences, but the results presented here indicate that  $g_s$  response to COS is prevalent, can be highly variable, and cannot be readily predicted at present.

While a strong correlation between  $A^s$  and both  $g_s$  and ambient COS concentrations was observed in all leaves ( $r^2$  of the linear best fit line = 0.63–0.97 for different species; for a more detailed discussion of this aspect, see Stimler et al., 2010a), there was no correlation between  $A^s$  and  $f_e$  among the plant species examined. For example, *Eucalyptus camaldulensis*, which had among the highest  $f_e$  values (0.93), had a  $A^s = 4.5 \text{ pmol m}^{-2} \text{ s}^{-1}$ , while *Ficus carica*, which showed no enhancement, had an  $A^s = 3.5 \text{ pmol m}^{-2} \text{ s}^{-1}$  at 350  $\text{pmol mol}^{-1}$  COS. Maximum observed  $A^s$  values at high COS (approximately 2,500  $\text{pmol mol}^{-1}$ ) ranged among species between 18.2 and 95.8  $\text{pmol m}^{-2} \text{ s}^{-1}$ , consistent with previously reported values (Taylor et al., 1983; Kesselmeier and Merk, 1993; Kesselmeier et al., 1999; Geng and Mu, 2004; Stimler et al., 2010a). Therefore, there are no clear relationships between the sensitivity of stomata to COS and the rate of COS uptake.

To better understand the basis of the  $g_s$  enhancement, we examined CA-deficient antisense lines of both C3 and C4 plants (Fig. 2; Stimler et al., 2011). Wild-type plants of both the C3 *Nicotiana tabacum* and the C4 *Flaveria bidentis* exhibited strong  $g_s$  enhancement in response to increasing COS, with  $f_e = 0.44$  on average (Table I). This enhancement was completely abolished in the CA-deficient plants. These plants showed constant  $g_s$  values across the wide range of ambient COS concentrations (Fig. 2), with  $g_s$  values of  $0.10 \pm 0.0001 \text{ mol m}^{-2} \text{ s}^{-1}$  for *F. bidentis* and  $0.11 \pm 0.002 \text{ mol m}^{-2} \text{ s}^{-1}$  for *N. tabacum*. Clearly, increasing ambient COS concentration in itself could not influence  $g_s$ . As also shown by Stimler et al. (2011), COS uptake was also abolished in the antisense plants, supporting the hypothesis that COS uptake is critically dependent on the catalysis of the hydrolysis of COS to  $\text{CO}_2$  and  $\text{H}_2\text{S}$ . Only when CA was active, converting the COS to  $\text{H}_2\text{S}$ , did the  $g_s$  enhancement occur.



**Figure 2.**  $g_s$  ( $\text{mol m}^{-2} \text{ s}^{-1}$ ) to COS during COS response experiments in the wild type (WT) and CA-deficient antisense lines of *F. bidentis* (A) and *N. tabacum* (B). Antisense lines were characterized with 2% and 10% of CA activity compared with the wild-type plants in A and B, respectively (Cousins et al., 2006). Conditions are as indicated in Figure 1. Error bars represent SD of four to six leaves.

### Variability in LRU among Plant Species

As part of our survey, we also examined the coupling between the rate of COS uptake ( $A^s$ ) and that of  $\text{CO}_2$  uptake ( $A^c$ ). This is necessary, first, to check to what extent stomatal sensitivity to COS influences variations in the COS/ $\text{CO}_2$  uptake ratios. This could provide indications of whether the COS effect is only on  $g_s$  or also on other (e.g. metabolic) processes. Second, estimating the range of variation in the COS/ $\text{CO}_2$  uptake ratio across species and functional types is critical for assessing the effectiveness of COS as a tracer of  $\text{CO}_2$  fluxes. Recent studies show the potential of using COS as a tracer of photosynthetic  $\text{CO}_2$  uptake by land plants (Montzka et al., 2007; Blake et al., 2008; Campbell et al., 2008). This is supported by the close links between the seasonal dynamics of atmospheric COS and  $\text{CO}_2$  at regional and atmospheric boundary layer scales (Montzka et al., 2004, 2007; Blake et al., 2008; Campbell et al., 2008) as well as

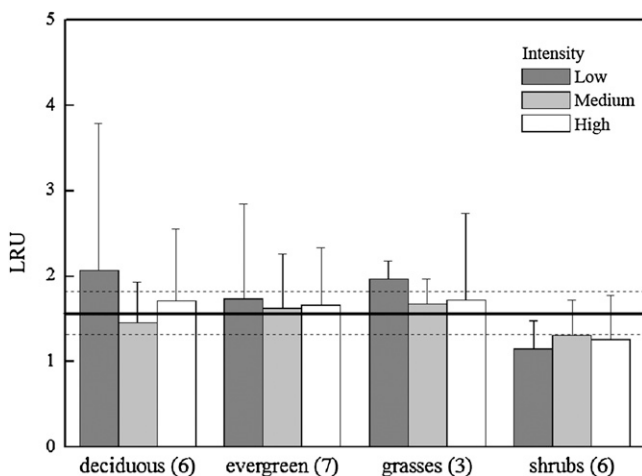
at the leaf level (Kesselmeier and Merk, 1993; Sandoval Soto et al., 2005; Yonemura et al., 2005; Stimler et al., 2010a, 2011). The application of COS as a tracer in this context relies on knowledge of the relative COS/ $\text{CO}_2$  uptake rates at the leaf level:  $\text{LRU} = (A^s/A^c) \times ([\text{CO}_2]/[\text{COS}])$ , where  $A^c$  and  $A^s$  are the uptake rates of COS and  $\text{CO}_2$ , respectively, and the square brackets indicate the respective ambient concentrations.

Information on the variations in LRU among plant species is limited, and available studies (Sandoval-Soto et al., 2005; Yonemura et al., 2005; Stimler et al., 2010a, 2011) cover only a limited number of species. Our survey included 22 plant species covering different functional types (Table II; summarized in Fig. 3). We observed a relatively narrow range of LRU values across the 22 plant species examined. Under near-ambient concentrations of COS and  $\text{CO}_2$  and at room temperature, and across a  $10\times$  range in light intensities, an overall average LRU value of  $1.61 \pm 0.26$  ( $n = 125$ ) was observed. No inherent differences were

**Table II.** Mean LRU values of the plant species used in this study and measured under low, moderate, and high light intensities (179, 352, and  $1,889 \mu\text{mol m}^{-2} \text{s}^{-1}$ , respectively), approximately  $500 \text{ pmol mol}^{-1}$  COS and approximately  $400 \mu\text{mol mol}^{-1} \text{CO}_2$ , temperature of approximately  $23^\circ\text{C}$ , and RH of approximately 75%

Values represent means ( $\pm$ SD) of three to four measurements on different leaves.

Vegetation Type	Light Intensity					
	Low		Moderate		High	
Trees						
Deciduous						
<i>Antigonon leptopus</i>	0.87	(0.99)	1.40	(0.26)	1.09	(0.06)
<i>Quisqualis indica</i>	0.94	(0.45)	0.63	(0.01)	0.95	(0.06)
<i>Quercus robur pedunculiflora</i>	1.51	(0.54)	1.45	(2.04)	3.22	(1.83)
<i>Diospyros virginiana</i>	2.00	(0.36)	1.90	(0.70)	1.40	(0.12)
<i>Psidium cattleianum</i>	5.04	(2.82)	1.99	(0.55)	1.52	(0.77)
<i>Ficus carica</i> L.			1.36	(0.12)	2.12	(0.04)
Average	2.07	(1.72)	1.45	(0.48)	1.72	(0.84)
Evergreen						
<i>Citrus madurensis</i>	1.36	(0.27)	1.56	(0.35)	2.91	(1.86)
<i>Citrus maxima</i>	0.77	(0.05)	1.32	(0.13)	1.94	(0.87)
<i>Diospyros digyna</i>	3.60	(0.47)	2.91	(0.98)	0.73	(0.02)
<i>Ficus neriifolia</i>	2.22	(0.14)	1.65	(0.94)	1.72	(1.26)
<i>Macadamia</i>	1.92	(0.55)	1.19		1.46	(0.95)
<i>Passiflora edulis</i>			1.84	(0.68)	1.68	(0.70)
<i>Persea</i>	0.56	(0.24)	0.94	(0.86)	1.23	(0.15)
Average	1.74	(1.11)	1.63	(0.64)	1.67	(0.67)
Nontrees						
Grasses						
<i>Agapanthus africanus</i>	2.12	(0.13)	1.76	(0.65)	2.90	(0.79)
<i>Belamcanda chinensis</i>			1.34	(0.49)	1.11	(0.27)
<i>Crocospia</i> $\times$ <i>crocospmiiflora</i> [aurea $\times$ pottsii]	1.82	(0.80)	1.91	(0.67)	1.15	(0.35)
Average	1.97	(0.21)	1.67	(0.30)	1.72	(1.02)
Shrubs						
<i>Abutilon pictum</i>	0.92	(0.98)	1.01	(0.40)	1.00	(0.35)
<i>Cestrum nocturnum</i>			1.20	(0.12)	1.93	(0.79)
<i>Jasminum sambac</i>	1.59	(1.01)	1.81	(1.56)	1.49	(0.79)
<i>Limonium perezii</i>	1.22	(0.10)	1.11	(0.08)	0.71	(0.09)
<i>Passiflora edulis</i>			1.84	(0.68)	1.68	(0.70)
<i>Viburnum tinus</i>	0.85	(0.05)	0.88	(0.13)	0.75	(0.15)
Average	1.15	(0.34)	1.31	(0.42)	1.26	(0.51)
Total average	1.76	(0.44)	1.51	(0.44)	1.60	(0.64)



**Figure 3.** Mean LRU ratios across 22 plant species, grouped into vegetation types, measured under three levels of light intensity (low, medium, and high refer to 179, 352, and 1,889  $\mu\text{mol photons m}^{-2} \text{s}^{-1}$ , respectively). The number of species sampled for each group is indicated in parentheses. The overall mean value for all plants was  $1.61 \pm 0.26$  (indicated by black and dashed lines). Conditions during measurements are as indicated in Figure 1.

apparent among the vegetation groups, with deciduous trees, evergreen trees, and grasses showing LRU of  $1.75 \pm 0.3$ ,  $1.68 \pm 0.05$ , and  $1.79 \pm 0.16$ , respectively. Individual measurements of LRU ranged from  $0.56 \pm 0.24$  in the evergreen *Persea americana* to  $5.04 \pm 2.82$  in the deciduous tree species *Psidium cattleianum*. These LRU values under low light levels were generally more variable, possibly due to the sensitivity to light of photosynthesis, but not of COS uptake, as also noted by Stimler et al. (2010a) in light response measurements. Separating the data between trees and nontrees indicated mean LRU values of  $1.7 \pm 0.9$  and  $1.42 \pm 0.53$ , respectively. Shrubs had a mean LRU value of  $1.24 \pm 0.08$ . These values are generally consistent with those reported previously (Kesselmeier and Merk, 1993; Sandoval-Soto et al., 2005; Yonemura et al., 2005; Stimler et al., 2010a). The result of a narrow LRU range across plant species is clearly advantageous to the use of COS in photosynthetic  $\text{CO}_2$ -uptake studies. The lack of correlation between the LRU of a species and the sensitivity of its stomata to COS may indicate that the COS effect is largely limited to  $g_s$ , influencing both COS and  $\text{CO}_2$  diffusional fluxes into the leaf with little effect on the ratio (LRU; for a discussion of the codiffusion of COS and  $\text{CO}_2$ , see Stimler et al., 2010a).

#### Does the CA-Mediated $g_s$ Response Involve $\text{H}_2\text{S}$ ?

Stimler et al. (2010a) hypothesized that the apparent enhancement of  $g_s$  by COS could be a stomatal response to  $\text{H}_2\text{S}$ , which is quantitatively produced from COS in its reaction with water and CA (Liu et al., 2010). The product,  $\text{H}_2\text{S}$ , can lead to the synthesis of

Cys (De Kok et al., 1998; Stuiver and De Kok, 2001) and can be oxidized to  $\text{SO}_3^{2-}$ ,  $\text{S}_2\text{O}_3^{2-}$ , and eventually sulfate. These compounds are not likely to have major signaling effects.  $\text{H}_2\text{S}$ , however, is a reactive gas with a wide range of activities, including effects on membrane ion channels, and was suggested to be a third biosignaling compound together with nitric oxide and carbon monoxide (for a recent review, see Wang, 2010). It was recently argued that  $\text{H}_2\text{S}$  could cause both opening (Lisjak et al., 2010) and closing (García-Mata and Lamattina, 2010) of stomata (Coyne and Bingham, 1978; Unsworth and Black, 1981; Gonzales, 1983). Note, however, that these studies rely on chemical compounds that are expected to produce intracellular  $\text{H}_2\text{S}$ , with limited controls on its concentrations, or on application of external concentrations of  $\text{H}_2\text{S}$  that are difficult to relate to concentrations inside the leaf. For example, the studies of Coyne and Bingham (1978) used parts per million levels of  $\text{H}_2\text{S}$ . At such high concentrations,  $\text{H}_2\text{S}$  is toxic to plants (Thompson and Kats, 1978; De Kok et al., 1998, 2002). It is difficult to estimate what concentrations may have occurred inside the leaves, but it seems likely that these concentrations are well above the expected levels around leaves in nature. Background atmospheric concentrations of  $\text{H}_2\text{S}$  are only approximately 7 to 14  $\text{pmol mol}^{-1}$  (parts per trillion level; Watts, 2000). Therefore, it may be more relevant to consider internal sources of  $\text{H}_2\text{S}$ .  $\text{H}_2\text{S}$  can be produced, de novo, in the leaves, but fluxes into sulfur-free air (i.e. enhanced fluxes) measured from untreated spruce (*Picea* sp.) leaves were only in the range of 0.2 to 0.5  $\text{pmol m}^{-2} \text{s}^{-1}$  (Rennenberg et al., 1990). Under steady-state conditions, the COS inflow into leaves must be nearly balanced by  $\text{H}_2\text{S}$  outflow (assuming that the metabolic consumption of  $\text{H}_2\text{S}$  is negligible). Given a flux of, say, 20  $\text{pmol m}^{-2} \text{s}^{-1}$  (a modest rate observed by Stimler et al. [2010a, 2011]), the rate of production of  $\text{H}_2\text{S}$  from COS may be 1 order of magnitude larger than the observed rate of endogenous synthesis. Furthermore,  $\text{H}_2\text{S}$  produced in the mesophyll must diffuse out through the stomata, and the intercellular concentration must be well above the ambient level. We calculate that the internal  $\text{H}_2\text{S}$  concentration in a leaf during steady-state photosynthesis might be 100 to 300  $\text{pmol mol}^{-1}$  higher than ambient concentrations.

The possible mechanism of the  $\text{H}_2\text{S}$  effect on  $g_s$  is not known at present. But, as noted above,  $\text{H}_2\text{S}$  is an active gas and is known to activate specific anion channels in mammalian cells and to specifically influence the flow of calcium ions across the cell membranes (Wang, 2010) as well as stimulate  $\text{K}^+$  channels (Zhao et al., 2001; Jiang et al., 2010). It is possible that similar effects also exist in plants. It is also not yet clear why the effect is so variable among plant species. But this may reflect variable sensitivity to  $\text{H}_2\text{S}$ , internal gradients in  $\text{H}_2\text{S}$  concentrations that would depend on internal conductance to COS and  $\text{H}_2\text{S}$ , as well as the type and location of CA involved and its activity (Fabre et al., 2007; Furne et al., 2008). The

plants might also differ in their capacity to consume H<sub>2</sub>S in sulfur metabolism (Rennenberg, 1984).

### The Role of CA

Using a different line of research, CA was also implicated as a “sensoenzyme” through influencing the production of HCO<sub>3</sub><sup>-</sup> (Frommer, 2010; Hu et al., 2010). However, under natural conditions, CA action on CO<sub>2</sub> and COS, to produce HCO<sub>3</sub><sup>-</sup> and H<sub>2</sub>S, respectively, cannot be separated without control of the COS concentration. Furthermore, the COS effect on *g*<sub>s</sub> reported here and by Stimler et al. (2010a) was observed at concentrations likely to occur under natural or experimental conditions. The studies reported here clearly implicate CA as a plausible source of H<sub>2</sub>S within the leaf. This report is, to our knowledge, the first to demonstrate a possible alternative mechanism whereby CA could function as a sensoenzyme.

In this study, we examined the stimulation of conductance by COS in a range of species and show that there is a large variation, with some species showing almost no response while others are highly responsive. Using C3 and C4 plants with antisense constructs to the enzyme CA, we show that the activity of this enzyme is essential for both the uptake of COS and the enhancement of *g*<sub>s</sub> by COS. Since CA catalyzes the conversion of COS to CO<sub>2</sub> and H<sub>2</sub>S, it seems likely that H<sub>2</sub>S produced in the mesophyll is involved in the stomatal response. In all plant species examined, the uptake of COS and CO<sub>2</sub> was highly correlated, and there was no relationship between the sensitivity of stomata and the rate of COS uptake (or, by inference, H<sub>2</sub>S production). The basis for the stomatal sensitivity and the variation in sensitivity is still to be determined, but the results evoke a possible new role for CA in plant response to the environment.

## MATERIALS AND METHODS

### Plant Material

To cover a diverse range of plant species and functional types, we used plants of 22 species that include six deciduous trees, seven evergreen trees, three grasses, and six shrubs. All plants were purchased in local nurseries and were grown under standard greenhouse conditions. Seeds of antisense lines and wild-type *Nicotiana tabacum* (C3) and *Flaveria bidentis* (C4) were contributed by Susanne von Caemmerer (Australian National University) and grown in pots in the greenhouse. Various levels of CA activity were achieved in each plant using suppression methods as described (Price et al., 1994; Cousins et al., 2006). Plants were kept under ambient light and temperature during the experimental period.

### Gas-Exchange Measurements

The experimental system consisted of a flow-through leaf cuvette made of Teflon-coated stainless steel with a magnetically operated fan and a glass window at the top. A whole leaf was sealed in the cuvette (O-ring seal except around the petiole, which was sealed with high-vacuum putty). Measurements on intact leaves sealed into the leaf chamber were performed under a relative humidity (RH) of approximately 70% and an air temperature of approximately 24°C. Two types of measurements were conducted for each species: first, exposing the plants to three light intensities (135, 352, and 1,889

μmol photons m<sup>-2</sup> s<sup>-1</sup>), regulated with layers of Miracloth and filtered through 5 cm of water; second, conducting COS response curves by mixing purified synthetic air that contains approximately 500 μmol mol<sup>-1</sup> CO<sub>2</sub> with compressed air from a calibrated high-concentration COS tank (550 nmol mol<sup>-1</sup>). Outflow from the leaf cuvette was split into two streams for COS and CO<sub>2</sub>/water analysis. All flow rates were regulated and measured with mass-flow controllers (MKS Instruments).

### CO<sub>2</sub> and COS Analysis

CO<sub>2</sub> and water vapor concentrations in the air entering and leaving the leaf cuvette were measured with an infrared gas analyzer (Li-6262; Li-Cor) at precision better than 0.5 μmol mol<sup>-1</sup> for CO<sub>2</sub> and 0.1 mmol mol<sup>-1</sup> for water vapor.

COS concentration was measured using a mid-infrared dual-quantum cascade laser at a wavelength of 2,056 cm<sup>-1</sup> using an LN<sub>2</sub>-cooled HgCdTe detector (Kolmar Technologies) as described by Stimler et al. (2010b). Briefly, the measurement method is direct detection of the absorption spectrum followed by quantitative spectral fitting combined with the measured pressure, temperature, and path length of the absorption cell and the laser spectral line width using TDL WINTEL software, as described by Nelson et al. (2004). The concentrations of COS and the laser line widths are real-time determined from the spectra through a nonlinear least-squares fittings algorithm that uses spectral parameters from HITRAN (Rothman et al., 2003). The data analysis procedure includes pulse normalization reduction of the sample and automatic background correction (N<sub>2</sub>). Pulse normalization corrects for variations in pulse-to-pulse amplitude in pulsed laser systems by normalizing the signal pulse train to a reference pulse train. The automatic background correction uses the dry nitrogen spectrum and divides the sample spectra by it. Corrections were carried out every 300 s. Maximum precision of the COS measurements was ±10 pmol mol<sup>-1</sup> in a 138-s integration time, reducing to 50 pmol mol<sup>-1</sup> in fast 1-Hz measurements (Stimler et al., 2010b).

As for CO<sub>2</sub>, COS uptake rates were calculated based on the concentration difference between the inlet and outlet of the leaf cuvette, the flow rate, and the leaf area. *g*<sub>s</sub> was estimated from conventional gas-exchange measurements (von Caemmerer and Farquhar, 1981).

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## LITERATURE CITED

- Blake NJ, Campbell JE, Vay SA, Fuelberg HE, Huey LG, Sachse G, Meinardi S, Rowland FS, Blake DR (2008) Carbonyl sulfide (OCS): large scale distributions over North America during INTEX-NA and relationship to CO<sub>2</sub>. *J Geophys Res Atmos* **113**: D09S90
- Campbell JE, Carmichael GR, Chai T, Mena-Carrasco M, Tang Y, Blake DR, Blake NJ, Vay SA, Collatz GJ, Baker I, et al (2008) Photosynthetic control of atmospheric carbonyl sulfide during the growing season. *Science* **322**: 1085–1088
- Cousins AB, Badger MR, von Caemmerer S (2006) A transgenic approach to understanding the influence of carbonic anhydrase on C<sup>18</sup>O discrimination during C<sub>4</sub> photosynthesis. *Plant Physiol* **142**: 662–672
- Coyne PI, Bingham GE (1978) Photosynthesis and stomatal light responses in snap beans exposed to hydrogen sulfide and ozone. *J Air Pollut Control Assoc* **28**: 1119–1123
- De Kok LJ, Stuijver CEE, Stulen I (1998) The impact of elevated levels of atmospheric H<sub>2</sub>S on plants. In LJ De Kok, I Stulen, eds, Responses of Plant Metabolism to Air Pollution and Global Change. Backhuys Publishers, Leiden, The Netherlands, pp 51–63
- De Kok LJ, Stuijver CEE, Westerman S, Stulen I (2002) Elevated levels of hydrogen sulfide in the plant environment: nutrient or toxin. In K Omasa, H Saji, S Youssefian, N Kondon, eds, Air Pollution and Biotechnology in Plants. Springer-Verlag, Tokyo, pp 201–213
- Fabre N, Reiter IM, Becuwe-linka N, Genty B, Rumeau D (2007) Characterization and expression analysis of genes encoding α and β carbonic anhydrases in *Arabidopsis*. *Plant Cell Environ* **30**: 617–629
- Frommer WB (2010) Biochemistry: CO<sub>2</sub> sense. *Science* **327**: 275–276
- Furne J, Saeed A, Levitt MD (2008) Whole tissue hydrogen sulfide concentrations are orders of magnitude lower than presently accepted values. *Am J Physiol Regul Integr Comp Physiol* **295**: R1479–R1485

- García-Mata C, Lamattina L** (2010) Hydrogen sulphide, a novel gas transmitter involved in guard cell signalling. *New Phytol* **188**: 977–984
- Geng C, Mu Y** (2004) Carbonyl sulfide and dimethyl sulfide exchange between lawn and the atmosphere. *J Geophys Res* **109**: 1–9
- Goldan PD, Fall R, Kuster WC, Feshenfeld FC** (1988) Uptake of COS by growing: a major trophospheric sink. *J Geophys Res* **93**: 14186–14192
- Gonzales GJ** (1983) Potential effects of hydrogen sulfide gas from geothermal energy conversion on two plant species native to northern New Mexico. PhD thesis. New Mexico State University, Los Alamos
- Hu H, Boisson-Dernier A, Israelsson-Nordström M, Böhmer M, Xue S, Ries A, Godoski J, Kuhn JM, Schroeder JI** (2010) Carbonic anhydrases are upstream regulators of CO<sub>2</sub>-controlled stomatal movements in guard cells. *Nat Cell Biol* **12**: 87–93
- Jiang B, Tang G, Cao K, Wu L, Wang R** (2010) Molecular mechanism for H<sub>2</sub>S-induced activation of K<sub>ATP</sub> channels. *Antioxid Redox Signal* **12**: 1167–1178
- Kesselmeier J, Merk L** (1993) Exchange of carbonyl sulfide (COS) between agricultural plants and the atmosphere: studies on the deposition of COS to peas, corn and rapeseeds. *Biogeochemistry* **23**: 47–59
- Kesselmeier J, Teusch N, Kuhn U** (1999) Controlling variables for the uptake of atmospheric carbonyl sulfide by soil. *J Geophys Res* **104**: 11577–11584
- Kluczewski SM, Brown KW, Bell JNB** (1985) Deposition of [<sup>35</sup>S]-carbonyl sulphide to vegetable crops. *Radiat Prot Dosimetry* **11**: 173–177
- Lisjak M, Srivastava N, Teklic T, Civalle L, Lewandowski K, Wilson I, Wood ME, Whiteman M, Hancock JT** (2010) A novel hydrogen sulfide donor causes stomatal opening and reduces nitric oxide accumulation. *Plant Physiol Biochem* **48**: 931–935
- Liu Y, Ma J, He H** (2010) Heterogeneous reactions of carbonyl sulfide on mineral oxides: mechanism and kinetics study. *Atmos Chem Phys* **10**: 10335–10344
- Montzka S, Calvert P, Hall BD, Elkins JW, Conway TJ, Tans PP, Sweeny C** (2007) On the global distribution, seasonality and budget of atmospheric carbonyl sulfide (COS) and some similarities to CO<sub>2</sub>. *J Geophys Res* **112**: D09302
- Montzka SA, Aydin M, Battle M, Butler JH, Saltzman ES, Hall BD, Clarke AD, Mondeel D, Elkins JW** (2004) A 350-year atmospheric history for carbonyl sulfide inferred from Antarctic firn air and air trapped in ice. *J Geophys Res* **109**: D22302
- Nelson DD, McManus B, Urbanski S, Herndon S, Zahniser MS** (2004) High precision measurements of atmospheric nitrous oxide and methane using thermoelectrically cooled mid-infrared quantum cascade lasers and detectors. *Spectrochim Acta A Mol Biomol Spectrosc* **60**: 3325–3335
- Price GD, von Caemmerer S, Evans JR, Yu JW, Lloyd J, Oja V, Kell P, Harrison K, Gallagher A, Badger MR** (1994) Specific reduction of chloroplast carbonic anhydrase activity by antisense RNA in transgenic tobacco plants has a minor effect on photosynthetic CO<sub>2</sub> assimilation. *Planta* **193**: 331–340
- Protoschill-Krebs G, Wilhelm C, Kesselmeier J** (1996) Consumption of carbonyl sulfide (COS) by higher plant carbonic anhydrase (CA). *Atmos Environ* **30**: 3151–3156
- Renzenberg H** (1984) The fate of excess sulfur in higher plants. *Annu Rev Plant Physiol* **35**: 121–153
- Renzenberg H, Huber B, Schröder P, Stahl K, Haunold W, Georgii HW, Slovik S, Pfanz H** (1990) Emission of volatile sulfur compounds from spruce trees. *Plant Physiol* **92**: 560–564
- Rothman LS, Barbe A, Benner DC, Brown L, Camy-Payret C, Carleer MR, Chance K, Clerbaux C, Dana V, Devi VM, et al** (2003) The HITRAN molecular spectroscopic database: edition of 2000 including updates through 2001. *J Quant Spectrosc Radiat Transfer* **82**: 5–44
- Sandoval-Soto L, Stanimirov M, Von Hobe M, Schmitt V, Valdes J, Wild A, Kesselmeier J** (2005) Global uptake of carbonyl sulfide (COS) by terrestrial vegetation: estimates corrected by deposition velocities normalized to the uptake of carbon dioxide (CO<sub>2</sub>). *Biogeosciences* **2**: 183–201
- Stimler K, Berry JA, Montzka SA, Yakir D** (2011) Association between carbonyl sulfide uptake and <sup>18</sup>Δ during gas exchange in C<sub>3</sub> and C<sub>4</sub> leaves. *Plant Physiol* **157**: 509–517
- Stimler K, Montzka SA, Berry JA, Rudich Y, Yakir D** (2010a) Relationships between carbonyl sulfide (COS) and CO<sub>2</sub> during leaf gas exchange. *New Phytol* **186**: 869–878
- Stimler K, Nelson D, Yakir D** (2010b) High precision measurements of atmospheric concentrations and plant exchange rates of carbonyl sulfide (COS) using mid-IR quantum cascade laser. *Global Change Biol* **16**: 2496–2503
- Stuiver CEE, De Kok LJ** (2001) Atmospheric H<sub>2</sub>S as sulfur source for Brassica oleracea: kinetics of H<sub>2</sub>S uptake and activity of O-acetylserine (thiol)lyase as affected by sulfur nutrition. *Environ Exp Bot* **46**: 29–36
- Taylor GE, Mclaughlin JSB, Shriner JDS, Selvidge WJ** (1983) The flux of sulfur-containing gases to vegetation. *Atmos Environ* **17**: 789–796
- Thompson CR, Kats G** (1978) Effects of continuous H<sub>2</sub>S fumigation crop and forest plants. *Environ Sci Technol* **12**: 550–553
- Unsworth MH, Black VJ** (1981) Response to pollutants. In PG Jarvis, TA Mansfield, eds, *Stomatal Physiology*. Cambridge University Press, New York, pp 187–203
- von Caemmerer S, Farquhar GD** (1981) Some relationship between biochemistry of photosynthesis and the gas exchange of leaves. *Planta* **153**: 376–387
- Wang R** (2010) Toxic gas, lifesaver. *Sci Am* **302**: 66–71
- Watts SF** (2000) The mass budget of carbonyl sulfide, dimethylsulfide, carbon disulfide and hydrogen sulfide. *Atmos Environ* **34**: 761–779
- Yonemura S, Sandoval-Soto L, Kesselmeier J, Kuhn U, Von Hobe M, Yakir D, Kawashima S** (2005) Uptake of carbonyl sulfide (COS) and emission of dimethyl sulfide (DMS) by plants. *Phyton* **45**: 17–24
- Zhao W, Zhang J, Lu Y, Wang R** (2001) The vasorelaxant effect of H<sub>2</sub>S as a novel endogenous gaseous K<sub>ATP</sub> channel opener. *EMBO J* **20**: 6008–6016