

Structural Basis for Norovirus Inhibition and Fucose Mimicry by Citrate

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Human noroviruses bind with their capsid-protruding domains to histo-blood-group antigens (HBGAs), an interaction thought to direct their entry into cells. Although human noroviruses are the major cause of gastroenteritis outbreaks, development of antivirals has been lacking, mainly because human noroviruses cannot be cultivated. Here we use X-ray crystallography and saturation transfer difference nuclear magnetic resonance (STD NMR) to analyze the interaction of citrate with genogroup II (GII) noroviruses. Crystals of citrate in complex with the protruding domain from norovirus GII.10 Vietnam026 diffracted to 1.4 Å and showed a single citrate bound at the site of HBGA interaction. The citrate interaction was coordinated with a set of capsid interactions almost identical to that involved in recognizing the terminal HBGA fucose, the saccharide which forms the primary conserved interaction between HBGAs and GII noroviruses. Citrate and a water molecule formed a ring-like structure that mimicked the pyranoside ring of fucose. STD NMR showed the protruding domain to have weak affinity for citrate (460 μM). This affinity, however, was similar to the affinities of the protruding domain for fucose (460 μM) and H type 2 trisaccharide (390 μM), an HBGA shown previously to be specifically recognized by human noroviruses. Importantly, competition STD NMR showed that citrate could compete with HBGA for norovirus binding. Together, the results suggest that citrate and other glycomimetics have the potential to block human noroviruses from binding to HBGAs.

Human noroviruses, family *Caliciviridae*, are the dominant cause of outbreaks of gastroenteritis. Many aspects of human norovirus replication, however, remain unclear, mainly because these viruses cannot be grown in cell culture. Transmission predominately occurs through ingestion of contaminated foods, airborne transmission, and person-to-person contact. Medical treatment usually involves orally administered fluids and electrolyte replacement therapy. Currently, there is no effective vaccine.

Human noroviruses can be divided into 2 main genogroups (GI and GII), which can be further subdivided into at least 25 different genotypes (GI.1 to -8 and GII.1 to -17) (26, 57). The norovirus genome has three open reading frames (ORFs) that encode nonstructural, capsid, and small structural proteins, respectively. The capsid of human norovirus is composed of two domains, shell and protruding (P) domains. The shell forms a scaffold around the RNA, and the dimeric P domain contains determinants for both antigenicity and receptor binding (25, 43, 51). The P domain is further subdivided into P1 and P2 subdomains, where the P1 subdomain interacts with the shell domain and is buried under the outermost P2 subdomain.

Human noroviruses bind to histo-blood group antigens (HBGAs), with recognition occurring in the P domain. HBGAs are complex carbohydrates present on mucosal epithelial cells or free antigens in blood, saliva, and other fluids (32). X-ray crystal structures of norovirus P domains in complex with different HBGAs have defined distinct binding sites for GI and GII viruses (8, 11, 12, 21); in particular, the HBGA binding site of GII is located at the dimeric interface of two P domains, whereas the HBGA binding site in GI is located within a single P domain (8, 11, 12, 21).

A number of recent studies have shown that natural fruits or

their constituents, including orange juice, pomegranate juice, cranberry juice, and grape seed extract, can inhibit and/or reduce feline calicivirus and murine norovirus infectivity (23, 48–50, 54). Although there have been no studies to support the idea that natural fruits or their constituents can prevent human norovirus infections, and data on the mode of inhibition of fruits have been lacking, the stability of human norovirus virus-like particles over a pH range of 3 to 7 (3) suggested that the effect might be related to a specific interaction with compounds in fruits rather than a pH effect. In this study, we used X-ray crystallography and saturation transfer difference nuclear magnetic resonance (STD NMR) to provide atomic-level details on the interaction of citrate and GII human noroviruses. We show that citrate specifically binds at the HBGA recognition site of GII noroviruses, and this inhibits P domain binding of both fucose and HBGA.

MATERIALS AND METHODS

Protein expression, purification, and crystallization of the norovirus P domain. The norovirus Vietnam026 GII.10 P domain (GenBank accession no. AF504671) (22) was expressed in *Escherichia coli* as previously

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described (21). Briefly, a truncated form of the GII.10 P domain was optimized for *E. coli* expression, cloned in a modified pMal-c2x vector at the BamHI and NotI sites (New England BioLabs), and transformed into *E. coli* BL21 cells (Invitrogen), and expression was induced with 1 mM IPTG (isopropyl- β -D-thiogalactopyranoside) for 18 h at 22°C. After a series of purifications and cleavage steps, the P domain was concentrated to 2 mg/ml and stored in gel filtration buffer (0.35 M NaCl, 2.5 mM Tris [pH 7.0], 0.02% NaN₃) before crystallization. Crystals of the P domain were obtained by the hanging-drop vapor diffusion method, with the mother solution containing citric acid triammonium (0.66 M [pH 6.5]) and isopropanol (1.65% [vol/vol]).

Data collection, structure solution, and refinement. X-ray diffraction data at a 1.000-Å wavelength were collected at the Southeast Regional Collaborative Access Team (SER-CAT) beamline 22-BM at the Advanced Photon Source, Argonne National Laboratory, Argonne, IL, and processed with HKL2000 (41). Prior to data collection, crystals were transferred to a cryoprotectant solution consisting of the mother liquor in 30% ethylene glycol, loop mounted, and flash-cooled in a nitrogen cryostat to 100°K. Structures were solved by molecular replacement with PHASER (35) by using Protein Data Bank (PDB) code 2OBR (11) as a search model. Structures were refined in multiple rounds of manual model building in COOT (16) and positional together with TLS refinement in REFMAC (13) and PHENIX (1).

Structure analysis and figures. Citrate and H type 2 interactions were determined using Discovery Studio (Accelrys, version v2.5.5.9350). Figures were rendered using PyMOL (Schrodinger, LLC, version 1.2r3) and ChemDraw Ultra (Cambridgesoft, version 12.0.2.1076).

STD NMR. All NMR data were recorded at 298°K on a Bruker Avance 600 NMR spectrometer equipped with a cryogenically cooled z-shielded gradient probe. One-dimensional (1D) STD NMR spectra were acquired with selective irradiation at -1 and +40 ppm (on and off resonance, respectively) using a train of 50-ms Gaussian-shaped radio frequency pulses separated by 1-ms delays and an optimized power level of 57 db. Water suppression was achieved with a binomial 3-9-19 pulse sequence. Samples were prepared in 20 mM sodium phosphate buffer containing 50 mM sodium chloride at pH 6.8. The NMR data were processed and analyzed with Topspin 2.1. STD enhancements were expressed as the STD amplification factor, A_{STD} , defined as $A_{STD} = (I_0 - I_{SAT}) I_0^{-1} ([L_t]/[P])$, where L_t and P are the total ligand and protein concentrations, respectively (34). HBGAs, H type 2 disaccharide [α -L-fucose-(1-2)- β -D-galactose], and H type 2 trisaccharide [α -L-fucose-(1-2)- β -D-galactose-(1-4)-2-N-acetyl- β -D-glucosamine] were purchased from V-labs, and L-fucose was purchased from Sigma-Aldrich. For the citrate experiments, sodium citrate dihydrate (Sigma-Aldrich) was added to sodium phosphate buffer and then titrated at pH 6.85 \pm 0.1.

Computational citrate docking studies of other saccharide-binding proteins. Citrate docking analyses were performed against six different saccharide-binding proteins, including *Anguilla anguilla* agglutinin (PDB identification no. 1K12) (5), *Aleuria aurantia* lectin (PDB identification no. 1IUC) (19), *Streptococcus pneumoniae* virulence factor SpGH98 (PDB identification no. 2J1S) (7), *Pseudomonas aeruginosa* PA-III lectin (PDB identification no. 2JDH) (33), parainfluenza virus 5 hemagglutinin-neuraminidase (PDB identification no. 1Z4X) (56), and porcine adenovirus type 4 galectin domain (PDB identification no. 2WSV) (20). Water molecules and ligands were removed from the PDB files, with the exception of one water molecule (HOH 935) in 1Z4X, which is present in both ligand-free and sialyllactose-bound hemagglutinin-neuraminidase structures. For 2JDH, the two calcium ions in the fucose binding site were kept, and the partial charges for the calcium ions were assigned to 1.5 as suggested by previous studies (38). AutoDock4.2 (39) was used as the docking engine, with the grid files generated by Autogrid4.2 using default parameters and centered on the cocrystallized ligands. The citrate molecule was docked to the three structures using default parameters (ga_pop_size = 150, ga_num_evals = 2,500,000, and ga_run = 50). For each structure, the docking pose with the lowest estimated free energy of binding among

the 50 docking runs was selected as the predicted binding pose. For comparison, for each complex, the cocrystallized ligand (or the terminal monosaccharide having the largest contact area with the binding site, if the cocrystallized ligand was not a monosaccharide) was docked in the saccharide binding site using the same procedure. Fucose and citrate molecules were also docked to the fucose-bound GII.10 P domain (PDB identification no. 3ONY) and the citrate-bound GII.10 P domain, respectively, for comparison. The water molecule (HOH 135) mediating the interaction between citrate and the protein was present during the citrate docking analysis.

Protein structure accession number. Atomic coordinate and structure factors for the citrate-bound GII.10 P domain have been deposited in the Protein Data Bank under accession no. 3RY8.

RESULTS

X-ray crystal structure of citrate bound to the GII.10 P domain.

The GII.10 P domain protein could be expressed in *E. coli* to 2 mg/liter and was purified and prepared for crystallization as previously described (21). To obtain a GII.10 P domain-citrate complex, we chose a crystallization condition that was similar to our previously reported GII.10 P domain-HBGA complex conditions (21), though with the addition of citrate. The GII.10 P domain-citrate complex formed rectangular plate crystals, and X-ray diffraction data revealed a space group of P2₁, the same as the previous GII.10 P domain-HBGA complexes (21), and strong diffraction to 1.4 Å. Structure solution by molecular replacement revealed one dimer per asymmetrical unit (Fig. 1A), and refinement led to an R_{value} of 0.139 ($R_{\text{free}} = 0.151$), with well-defined density for most of the P domain dimer (Table 1). Electron density for residues 296 to 299 (chain A) and 296 to 300 and 344 to 351 (chain B) was poor, and these residues were not modeled. Extra electron density was observed at the HBGA binding site, where a single citrate molecule was clearly distinguished and refined (Fig. 1B; see Fig. S1 in the supplemental material). The structure of the GII.10 P domain in complex with citrate was highly reminiscent of the other known structures (GI.1, GII.4, and GII.12), where the P1 subdomain contains a single α -helix and the P2 subdomain contains six anti-parallel β -strands that form two anti-parallel β -sheets (21).

Citrate was highly coordinated by the GII.10 P domain. At a 1.4-Å resolution, detailed interactions between citrate and the P domain could be defined. Seven residues of the P domain, many of which are conserved and located at the dimer interface, are involved in hydrogen bonding interactions with citrate (Fig. 1B and C). These include the side chain of Tyr452 and main chain of Gly451 from one P domain subunit as well as side chains Arg356 and Asp385 and the main chain of Asn355 of a second P domain subunit. Unique to citrate binding, side chains of Asn342 and Ser387 make a water-mediated hydrogen bond with the C-5 CO group of citrate. Superposition of citrate-bound and apo GII.10 P domain structures indicated that the citrate interaction did not cause any conformation changes in the GII.10 P domain.

Comparisons of citrate and HBGA interaction with the GII.10 P domain. Compared with GII.10 P domains in complex with HBGAs, we found that citrate essentially mimics the fucose unit of HBGAs. By using H type 2 di- and trisaccharides as examples, superposition of the P domains revealed that three carbon atoms, including C-2, C-3, and the C-3 carboxy carbon, and three oxygen atoms, including the C-1 and C-3 carboxy oxygens and the C-3 hydroxyl group of citrate closely overlapped with C-5/C-4/C-3/O-5/O-4/O-3 of the terminal α -fucose ring (Fig. 2). In addition,

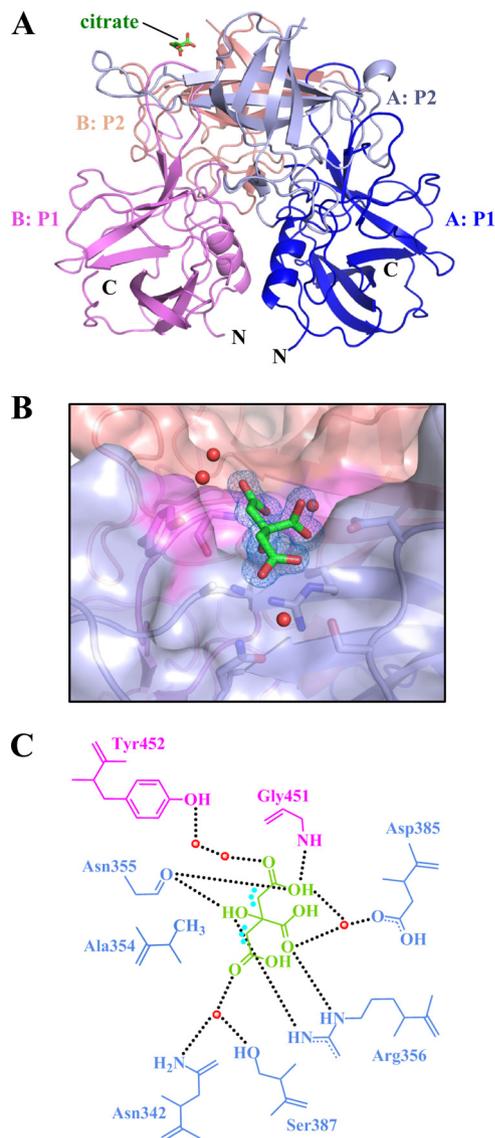


FIG 1 Citrate binding to the GII.10 P domain. (A) X-ray crystal structure of the GII.10 P domain dimer (ribbon structure) and the bound citrate (green sticks). Each P subdomain is colored differently: i.e., chain A, P1, blue; chain A, P2, light blue; chain B, P1, violet; and chain B, P2, salmon. (B) Surface representation of the GII.10 P domain (colored as in panel A) showing the residues (sticks) and water molecules (red spheres) interacting with the citrate molecule (green sticks). The $2F_o - F_c$ density was contoured at 1.0σ . (C) Residues interacting with the citrate molecule were contributed by both monomers (colored as in panel A), where the black dotted lines represent the hydrogen bonds, the cyan dots near the citrate represent the hydrophobic interactions with Ala354, and the red spheres represent water molecules. For simplicity, only the backbone is shown for residues that were backbone mediated. Hydrogen bond distances were less than 3.1 \AA , though the majority were $\sim 2.8 \text{ \AA}$.

a water molecule, present in the citrate-bound structure but absent from the HBGA-bound structures, occupied the site of the C-2 hydroxyl of fucose (Fig. 2). In this configuration, the citrate and associated water molecule formed a ring-like structure, mimicking the pyranoside ring of fucose. Finally, the comparisons showed that of the seven residues involved in hydrogen bonding interactions with citrate, five made almost identical interactions with their comparable atoms in fucose.

Characterization of binding of citrate, H type 2 trisaccharide, and fucose to GII.10 P domain by STD NMR. Given the remarkable similarities observed for citrate and fucose binding to the GII.10 P domain by crystallography, we sought to characterize in solution by NMR the binding of GII.10 P domain with citrate, HBGAs, and fucose and ultimately to determine their relative binding affinities and whether they bind competitively.

STD enhancements were observed for methylene protons H2A and H2B of citrate, consistent with their close proximity to the protein in the bound state (Fig. 3A). In the crystal structure, these hydrogens are within van der Waals contact of the methyl of Ala354 (Fig. 1C). With H type 2 trisaccharide, the most prominent STD signals that could be assigned corresponded to H-1, H-2, and H-4 of α -fucose; H-3 of galactose; and H-1, H-2, and *N*-acetyl of glucosamine (Fig. 3B). We also characterized binding of monosaccharide α/β -fucopyranose, as it also would be used in competition STD NMR experiments. As seen in Fig. 3C, binding of both anomers was observed, with H-1, H-2, and H-4 of α -fucose versus H-2, H-4, and H-5 of β -fucose showing the strongest enhancements. Although natural H type 2 HBGAs contain α -Fuc(1-2)Gal and not β -Fuc, it is interesting that the HBGA binding site of norovirus can bind both. By NMR, we observed binding of α and β forms of the monosaccharide (Fig. 3C) as well as synthetic H type 2 trisaccharide α/β -Fuc(1-2) β -Gal(1-4) β -GlcN (Fig. 3B) and H type 2 disaccharide α -Fuc(1-2) β -Gal(1-4) (data not shown), and by crystallography, binding of synthetic H type 2

TABLE 1 Data collection and refinement statistics for structures of the GII.10 Vietnam026 norovirus P domain^a

Parameter	Value(s) for citrate (026_citrate; PDB accession no. 3RY8) ^b
Data collection	
Space group	P2 ₁
Cell dimensions	
<i>a</i> , <i>b</i> , <i>c</i> (Å)	63.76, 79.81, 69.60
α , β , γ (°)	90, 96.84, 90
Resolution (Å)	50-1.40 (1.45-1.40)
<i>R</i> _{sym}	7.3 (30.5)
<i>I</i> / σ <i>I</i>	18.7 (3.2)
Completeness (%)	99.9 (99.6)
Redundancy	3.7 (3.2)
Refinement	
Resolution range (Å)	31.98-1.399
No. of reflections	131,576
<i>R</i> _{work} / <i>R</i> _{free}	0.1388/0.1506
No. of atoms:	
Total	5,587
Protein	4,722
Ligand/ion	57
Water	808
<i>B</i> -factors	
Protein	18.3
Ligand/ion	19.2
Water	30.2
RMSD	
Bond length (Å)	0.011
Bond angle (°)	1.393

^a Each data set was collected from a single crystal.

^b Values in parentheses are for the highest-resolution shell.

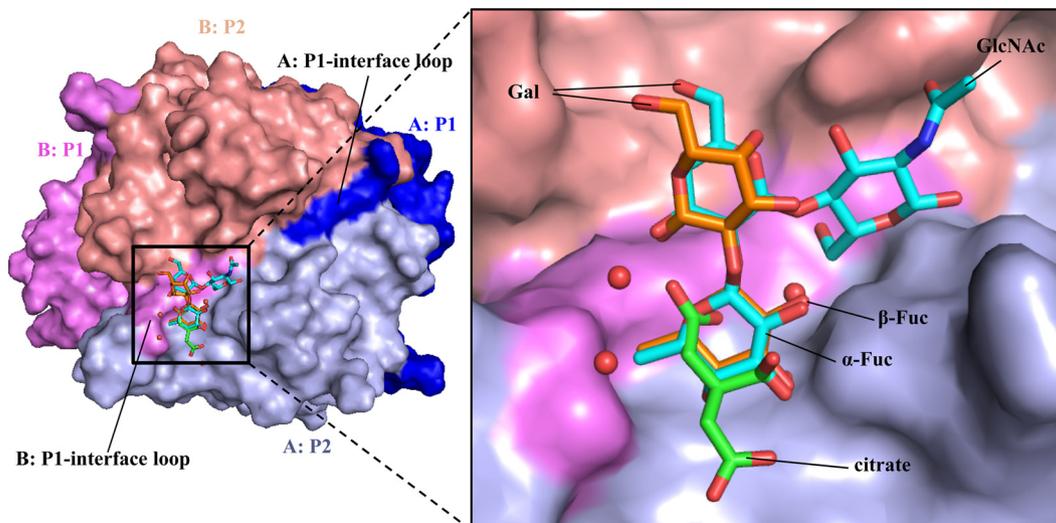


FIG 2 The HBGA and citrate binding site overlapped on the top of the GII.10 P domain. (A) The citrate molecule (green sticks) bound at the HBGA binding site; shown here are the bound H type 2 di- and trisaccharides (orange and cyan sticks, respectively). (B) Close-up of the black square in panel A, showing the H type 2 disaccharide [α -L-fucose(1-2)- β -D-galactose] and H type 2 trisaccharide [β -L-fucose(1-2)- β -D-galactose(1-4)-2-N-acetyl- β -D-glucosamine].

trisaccharide β -Fuc(1-2) β -Gal(1-4) β -GlcN was observed, in addition to binding of other HBGAs containing the α -Fuc(1-2) β -Gal linkage (21). Finally, it is interesting to note that a similar mode of citrate binding was observed for the soluble GII.12 P domain (see Fig. S2 in the supplemental material).

Affinity of citrate, H type 2 tri- and disaccharides, and L-fucose to the GII.10 P domain. We used single-ligand titration STD NMR experiments to determine the K_D (equilibrium dissociation constant) of citrate binding to the GII.10 P domain (Fig. 4A) (2). STD amplification factors (A_{STD}) (34) were calculated by integrating the signal at δ_H 2.54 ppm in difference and corresponding reference spectra. Initial growth rates (A_{0STD}) were obtained by measuring the effect on A_{STD} as a function of various saturation time (t_{sat}) and fitting the data to the equation $A_{STD} = A_{max\ STD} [1 - \exp(-kt_{sat})]$ for each concentration (300, 600, 900, 1,200, and 1,500 μ M) of the ligand. The K_D of citrate was in turn measured as $460 \pm 80 \mu$ M by fitting A_{0STD} values as a function of ligand concentration using the equation $y = B_{max}/(K_D + x)$, where x is the ligand concentration and B_{max} represents the plateau of the curve (Fig. 4) (2, 37). For H type 2 trisaccharide, the STD enhancements for the N-acetyl signal were sufficiently strong to allow for accurate integration, even at very short saturation times (0.1 s); thus, a K_D value of $390 \pm 70 \mu$ M could be determined directly by fitting A_{STD} values as a function of ligand concentration (40) (Fig. 5). K_D values for fucose and H type 2 disaccharides were in turn obtained from single point competition STD experiments as described previously (36) to give values of 460 ± 10 and $420 \pm 40 \mu$ M, respectively (see Fig. S3 in the supplemental material).

Competition of HBGAs and citrate with the GII.10 P domain. To confirm the overlapping mode of binding observed in the crystal structure of citrate and fucose of H type 2 ligands, A_{STD} values of L-fucopyranose and H type 2 trisaccharide were monitored while titrating citrate to the samples. As seen in Fig. 4E, addition of citrate to a sample of P domain-H type 2 trisaccharide diminishes the trisaccharide signals in a concentration-dependent manner, indicating that citrate directly competes with the trisaccharide for P domain binding, giving a K_i of $600 \pm 20 \mu$ M. Upon addition of

citrate, the pH of the solutions was found to remain constant (pH 7.2 ± 0.1), indicating that the competition was a direct result of citrate binding rather than pH. The same effect was observed in STD competition experiments with L-fucose (data not shown). Importantly, the reverse set of experiments showed that HBGAs can compete with citrate for P domain binding (data not shown), indicating that the P domain is unaffected by the presence of citrate. Together, these results conclusively demonstrate molecular mimicry between citrate and fucose of HBGAs.

DISCUSSION

Despite the discovery of human norovirus nearly 40 years ago (27), little is known about the capsid interaction with ligands (18, 44) other than HBGAs (8, 11, 12, 15, 21, 45). Our finding that citrate binds at the terminal fucose binding site was somewhat unexpected, given that the structure of citrate is unlike the structure of fucose and considering that the GII.10 P domain could not bind HBGAs having an α -fucose1-3/4 saccharide (21). In an earlier enzyme immune assay study, Feng et al. screened \sim 5,000 compounds (the Diversity screening set; Timtec, Inc.) for their ability to block GI and GII norovirus virus-like particles (VLPs) from binding to saliva samples of known HBGA type (18). They found 14 compounds that had strong inhibition; however, the mode of action was not determined. In a more recent NMR study, Rademacher et al. screened \sim 500 compounds (the Maybridge Ro5 fragment library; Thermo Fisher Scientific, Inc.) for their ability to bind to a GII.4 VLP HBGA binding site (44). They showed that both univalent and multivalent compounds were capable of binding to the HBGA binding site. Interestingly, for both studies, the compounds that showed the highest affinities included compounds with at least one ring component. Taken together, these studies indicated that the HBGA binding site was capable of binding numerous compounds other than HBGAs, ranging from the small (smallest) citrate molecule to larger multivalent compounds.

For over a decade, the GII.4 noroviruses have remained as the dominant genotype of outbreaks of gastroenteritis around the

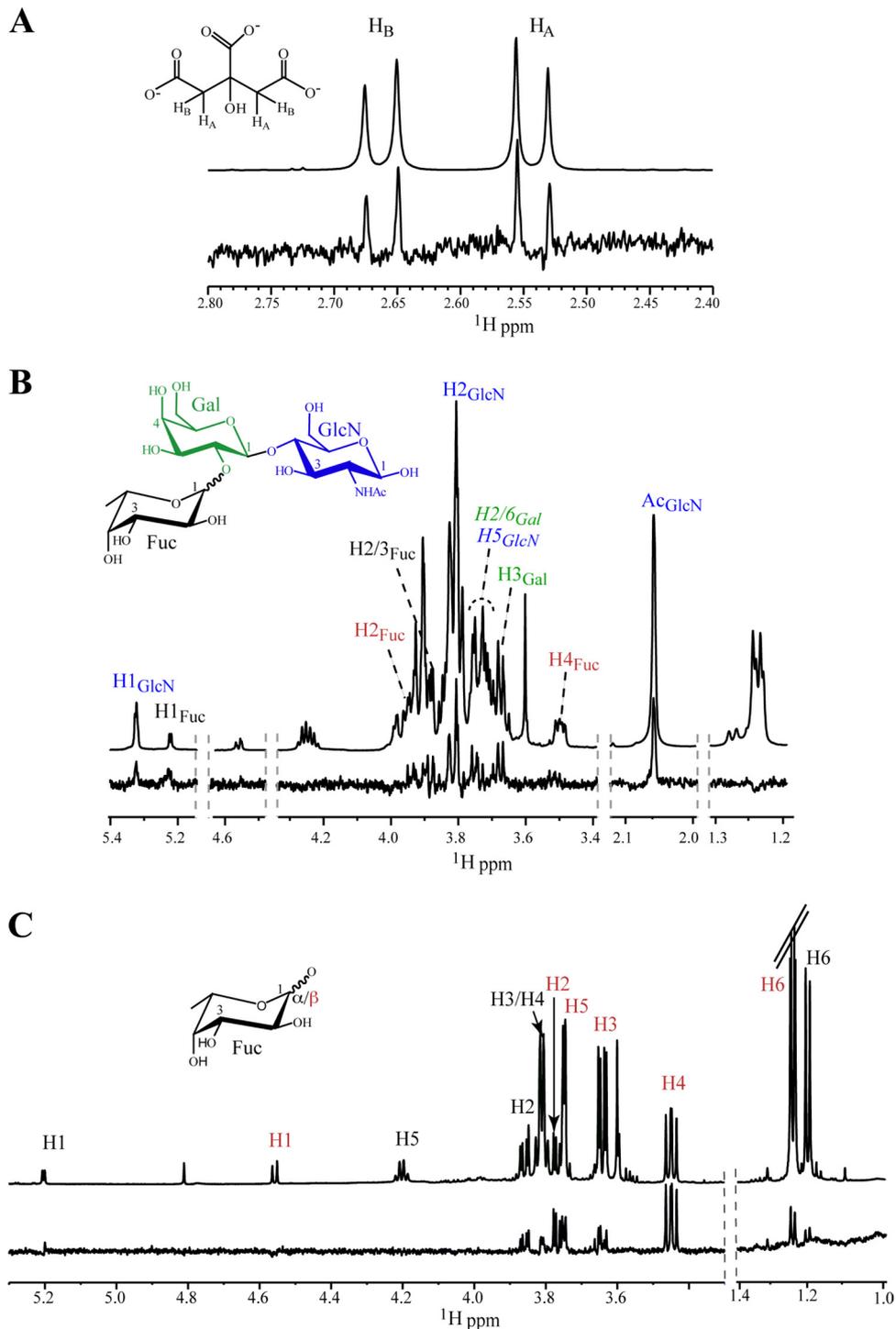


FIG 3 STD NMR spectra for citrate, H type 2 trisaccharide, and L-fucose bound to the GII.10 P domain. STD (lower) and reference (upper) spectra of (A) citrate (1.2 mM), (B) H type 2 trisaccharide (1.2 mM), and (C) L-fucose (mixture of α and β anomers) (1.2 mM) in the presence of the GII.10 P domain (15 μ M). Nonoverlapping protons that exhibit STD enhancements are labeled and color coded by sugar residue, and signals for β -Fuc are red. One group of overlapping signals appears in italics.

world and as such the most well studied. Most studies agreed that a dominant GII.4 norovirus was replaced the following year or next by a new GII.4 “variant” norovirus that had $\sim 5\%$ amino acid change in the capsid gene (6, 9, 10, 30, 31, 47). The reason that the GII.4 variants dominated and not some other genotype was un-

known, but studies have shown specific mutations at or surrounding the HBGA binding site were capable of altering the HBGA binding patterns (15, 30, 31, 52). These small changes were thought to lead to new GII.4 variants capable of causing pandemics, analogous to influenza A virus evolution (14, 29). Despite

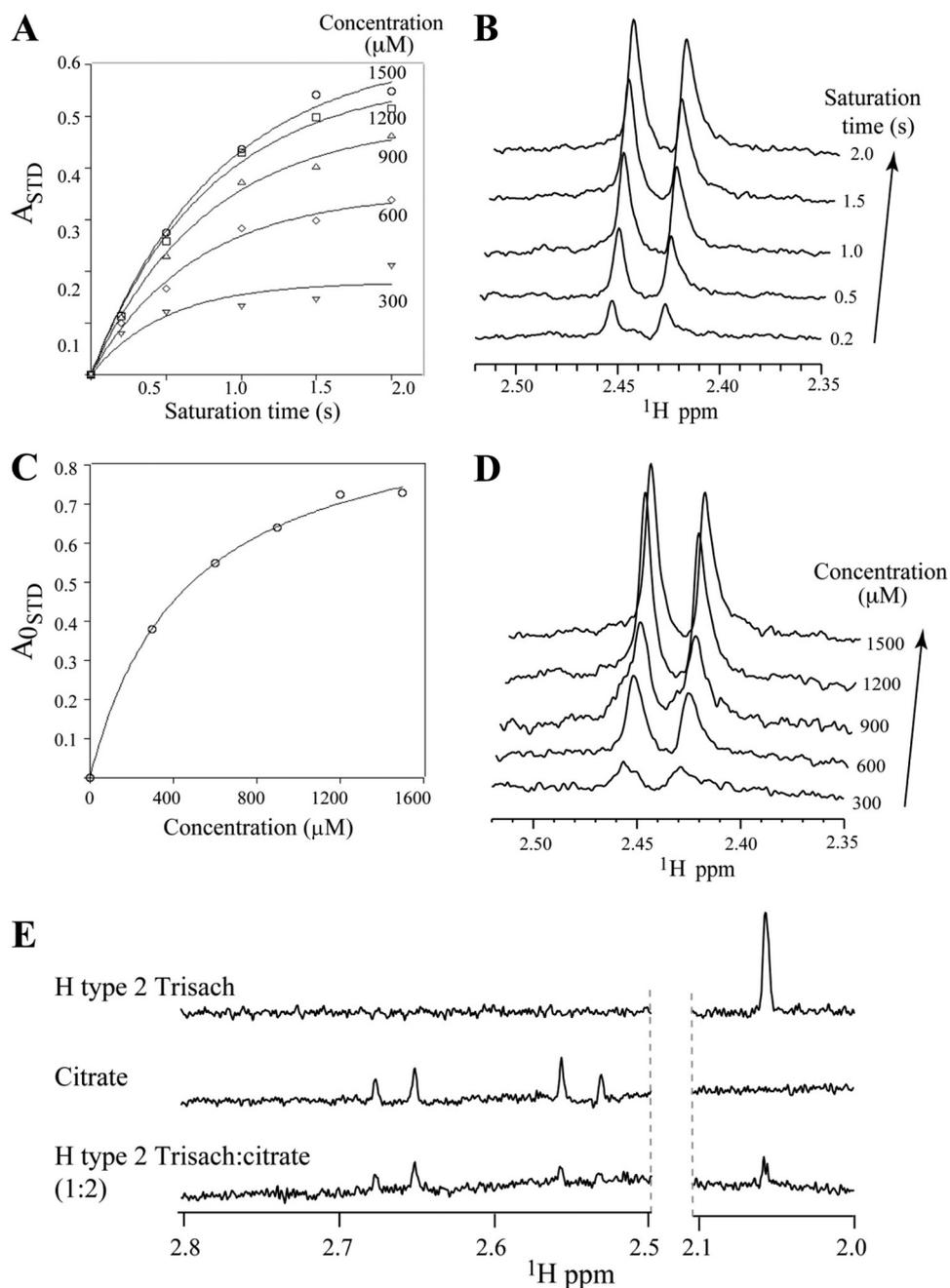


FIG 4 Binding affinity of citrate and HBGAs to GII.10 P domain by STD NMR. Data were used to obtain the K_D for citrate and H type 2 trisaccharide (Trisach) binding to GII.10 P by single-ligand titration STD NMR experiments (2). (A) Effect on STD enhancement (expressed as A_{STD}) (34, 37) as a function of saturation time (t_{sat}) and ligand concentration; (B) stacked plots of spectra for 1.5 mM citrate as a function of t_{sat} (y axis); (C) Langmuir binding curve used to obtain the K_D from the initial slope of A_{STD} as a function of citrate concentration. (D) Stacked plot of various citrate concentrations (t_{sat} 2 s, 15 μ M protein); (E) competition STD spectra of H type 2 trisaccharide (top), citrate (middle), and 1:2 H type 2 trisaccharide-citrate (0.75:1.5 mM; bottom) used to calculate the K_i of citrate (36).

these amino acid changes, few if any occurred at the fucose-binding site, thus highlighting the common site of vulnerability for GII noroviruses, especially for the pandemic GII.4 variant noroviruses. It is not known if the GI noroviruses will bind citrate given that the GI and GII P domain interactions with HBGAs were different, but since GI.1 P domain interacted with α -fucose1-2 and it was reported that the HBGA binding site

was conserved among GI noroviruses (12), we suspect that GI noroviruses may also bind citrate, although further structural studies are needed.

Our unexpected finding that citrate and fucose have similar binding modes to the norovirus GII.10 P domain raises the question of whether such citrate mimicry of monosaccharide binding could be a general phenomenon or whether it is spe-

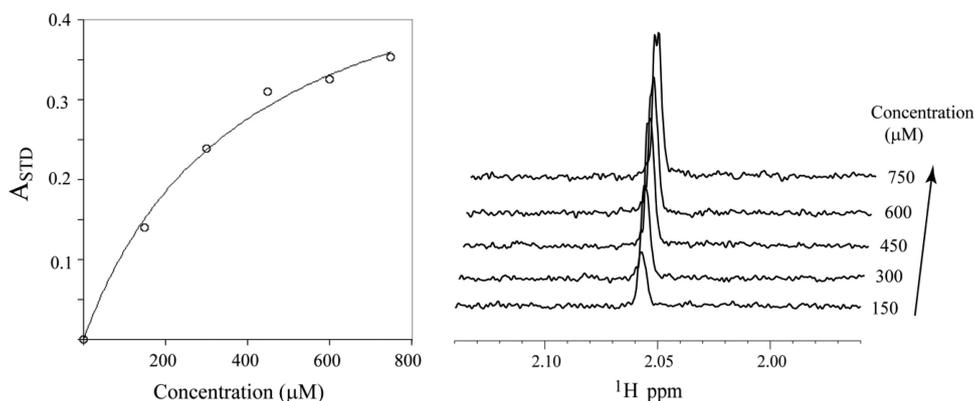


FIG 5 Binding affinity of H type 2 trisaccharide to the GII.10 P domain (left) effect on STD enhancement, expressed as A_{STD} , as a function of trisaccharide concentration in the presence of 15 μM GII.10 P domain. $t_{\text{sat}} = 0.1$ s. Curve fitting (described in the text) provides the K_D value. (Right) Stacked STD NMR spectra showing the change in enhancement of the nonoverlapped *N*-acetyl proton signals as a function of increasing concentration of H type 2 trisaccharide (40).

cific to norovirus and other caliciviruses. To investigate this, we performed *in silico* docking studies of citrate against four different fucose-binding proteins (*Anguilla anguilla* agglutinin, *Aleuria aurantia* lectin, *Streptococcus pneumoniae* virulence factor SpGH98, and *Pseudomonas aeruginosa* PA-IIL lectin) and two other saccharide-binding proteins (parainfluenza virus 5 hemagglutinin-neuraminidase and porcine adenovirus type 4 galectin domain), for which fucose or other saccharide-bound crystal structures were available (see Table S1 in the supplemental material). Computational docking analyses reveal different levels of citrate mimicry of monosaccharide binding for other saccharide-binding proteins. For *Anguilla anguilla* agglutinin, citrate, in its predicted binding pose, overlapped with the C-5, C-4, C-3, O-5, O-4, and O-3 atoms of fucose in a similar way to what was observed in the GII.10 P domain (Table S1), while forming hydrogen bonds with the same sets of protein residues as fucose (see Fig. S4 in the supplemental material). Citrate was thus predicted to show a high degree of mimicry to fucose, similarly to our experimental findings for the GII.10 P domain. For the other three fucose-binding proteins, citrate, in its predicted binding poses, did not overlap with the cocrystallized fucose, although it still formed the same sets of polar interactions as the cocrystallized fucose (see Fig. S5 to S7 in the supplemental material). Hence, our docking studies suggest that the mimicry between citrate and fucose binding observed for the GII.10 P domain could be a common, although not universal, phenomenon across other fucose-binding proteins. For all six fucose- and other saccharide-bound proteins for which docking was performed, the predicted citrate binding poses were able to form polar interactions with the same sets of protein residues as the cocrystallized ligand see (Fig. S4 to S9 in the supplemental material), indicating that citrate might be generally useful as a scaffold for designing glycomimetic inhibitors against these and other saccharide-interacting pathogens. Furthermore, a search of the ZINC database (4) revealed that there are more than three thousand compounds with at least 50% similarity to citrate. Thus, *in silico* screening of this database may present a promising approach for identifying small molecules that bind to saccharide-binding proteins. We note, however, that the predicted binding pose of citrate docked to fucose-bound

GII.10 P domain had a root mean square deviation (RMSD) of 3.60 Å, while the predicted binding pose of citrate docked to citrate-bound GII.10 P domain with the cocrystallized water molecule had an RMSD of 1.87 Å. This indicates that the resulting docking modes could be error prone. Given that calculating small molecule-receptor binding energies is a difficult and error-prone task (24, 46), ultimately experimental validation would be necessary to confirm the generality of the citrate-saccharide mimicry predicted here.

The STD NMR data provided strong evidence that the integrity of the GII.10 P domain remained unchanged in the presence of different concentrations of citrate buffer and since the pH of the citrate buffer remained more or less the same during the titration, a specific effect of citrate was responsible for the reduction in HBGA attachment. Although the K_D values of citrate and H type 2 trisaccharide for the GII.10 P domain are in the range of 360 to 490 μM , these relatively weak affinities are typical for univalent protein-carbohydrate interactions (17, 28). Given that 90 copies of dimeric P domains are present on norovirus capsid, it is plausible that a multivalent version of citrate- or fucose-like ligands would greatly enhance affinities and provide a starting point for norovirus inhibitors. Indeed, Rademacher et al. show that multivalent fucose-like compounds have increased avidity over their univalent counterparts (44).

In conclusion, we have described the structural basis by which citrate binds to the HBGA binding site of the norovirus GII.10 P domain and can in turn inhibit HBGA binding. Natural compounds, such as juice from lemons and limes, which contain ~ 300 mM citric acid (42), may already reduce or inhibit norovirus infections, as suggested by a number of recent studies (23, 48–50, 54). In regard to this, it is tempting to speculate that a few drops of lemon juice with one's oysters might reduce norovirus infection. Epidemiological studies on the ingestion of foods high in citrate and norovirus infection may be illuminating, as may be correlations with related glycomimetics—e.g., with ascorbic acid (vitamin C). Controlled possibly volunteer studies should also provide an accurate assessment of norovirus inhibition. Additional compound screening will likely be required to identify a universal norovirus inhibitor with high potency and broad reactivity, and the structural basis for norovirus interaction with citrate as revealed here may be helpful in such efforts.

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