

Published in final edited form as:

FEBS Lett. 2012 February 17; 586(4): 330–336. doi:10.1016/j.febslet.2012.01.019.

Cataract-Linked γ D-Crystallin Mutants Have Weak Affinity to Lens Chaperones α -Crystallins

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Abstract

To test the hypothesis that α -crystallin chaperone activity plays a central role in maintenance of lens transparency, we investigated its interactions with γ -crystallin mutants that cause congenital cataract in mouse models. Although the two substitutions, I4F and V76D, stabilize a partially unfolded γ D-crystallin intermediate, their affinities to α -crystallin are marginal even at relatively high concentrations. Detectable binding required further reduction of γ D-crystallin stability which was achieved by combining the two mutations. Our results demonstrate that mutants and possibly age-damaged γ -crystallin can escape quality control by lens chaperones rationalizing the observation that they nucleate protein aggregation and lead to cataract.

Keywords

γ D-crystallin; denaturant unfolding; bimane fluorescence; α B-crystallin phosphorylation; α -crystallin; β -crystallin; chaperone; small heat-shock protein; cataract

1. Introduction

A unique cellular and molecular architecture enables the ocular lens to transmit and focus light on the retina. The lens consists of layered terminally differentiated cells, the fiber cells, which lose their organelles early in development and are unable to synthesize or turn over proteins [1]. Transparency and refractivity entail short-range order packing of highly water soluble proteins, α -, β - and γ -crystallins, which account for more than 90% of the lens protein weight [2]. A highly tuned balance of forces prevents long range order or crystallization at the high crystallin concentrations found in fiber cells [3,4].

Under assault by environmental factors, the optimized protein packing is eventually perturbed leading to protein aggregation and light scattering. The “wear and tear” of lens aging is recorded at the molecular level through accumulated modifications of the crystallins [5,6] such as truncations [7,8], glycation [9,10], oxidations [11] and deamidations [12]. In the absence of protein turn-over, loss of crystallins stability and/or solubility alters the balance of forces between proteins thereby compromising lens transparency and refractivity. Age-related cataract is an opacification of the lens which can result from protein unfolding and subsequent aggregation [1] and/or changes in the surface properties of molecules

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leading to loss of solubility [13–15]. One of the mechanisms hypothesized to delay aggregation in the lens is the activity of resident small heat-shock proteins (sHSP) [16] α A- and α B-crystallin. In his seminal work, Horwitz proposed [17,18] that by inhibiting lens proteins aggregation, α -crystallins delay the onset of cataract. The protein unfolding hypothesis for age-related cataract posits that progressive modifications of β - and γ -crystallins reduce their free energies of unfolding and promote their binding to α -crystallin. The disappearance of soluble α -crystallin from the center of normal human lenses by age 40 [19] is hypothesized to set the stage for aggregation of destabilized proteins. Aging is characterized by a progressive shift in the molecular weight distributions of fiber cells proteins [20]. Similar to late-onset aggregation diseases, the gradual accumulation of modified proteins in the lens overwhelms the chaperone machinery.

Despite their abundance, β - and γ -crystallins do not have a defined role in lens development. They are considered structural proteins whose packing and interactions are optimized for transparency and refractivity [1]. Both β - and γ -crystallins are members of the same protein superfamily [21]. The folds of the dimeric β - and monomeric γ -crystallins consist of four greek key motifs organized in two domains [22,23]. The main sequence difference between the two families is N-terminal extensions in the β -crystallins, with the basic isoforms having additional C-terminal extensions. Seven γ -crystallins are expressed in the human lens; the most abundant, γ D-crystallin, accounts for 11% of the total protein content in the fiber cells.

A number of mutations in γ -crystallins have been associated with congenital cataracts in humans and mouse models. These include “solubility mutants” where residue substitutions change the surface property of the molecule [24,25]. Another class of mutants is associated with changes in stability, i.e. shifts in the unfolding equilibrium [26,27]. Studies of two such mutants in mouse models suggest multiple mechanistic origins of cataract [26,27]. The substitution of I4F in γ B-crystallin results in large inter-fiber spaces and large aggregates in the inner fiber cells [26]. Although the mutant appears to associate with α -crystallin in mouse lens extract, temperatures higher than 37 °C were required to form the complex *in vitro*. γ D-crystallin mutant V76D forms intranuclear aggregates that disrupt denucleation of fiber cells [27].

The existence of cataract-causing, destabilized mutants of lens proteins poses a conundrum for the chaperone hypothesis of α -crystallin function. α A-crystallin constitutes 30% of total lens proteins and its binding capacity could reach one substrate per subunit of equivalent molecular mass [16]. Thus, either the γ -crystallin mutants are not efficiently bound by α -crystallins or γ -crystallin may have a yet undefined cellular role in lens development that is inhibited by its destabilization and subsequent interaction with α -crystallin. To distinguish between these possibilities, we investigated the interaction between both α -crystallin subunits and destabilized mutants of γ D-crystallin. The experimental design follows our established approach that emphasizes a direct quantitation of binding in the absence of aggregation [28]. For this purpose, γ D-crystallin mutants associated with congenital cataract in mice were constructed and their unfolding equilibrium characterized. Binding to α -crystallins was detected by changes in the fluorescence intensity and anisotropy of a bimane probe site-specifically attached to γ D-crystallin. We find remarkably low affinities *in vitro* suggesting that these mutants may escape quality control by α -crystallin chaperone. These findings are interpreted in the context of the folding equilibrium of γ -crystallin and built-in repulsive interactions between crystallins critical for lens refractivity and transparency.

2. Material and methods

2.1 Expression and purification of crystallins

Human γ D-crystallin was a generous gift from Dr. C. Slingsby, Birbeck College. The I4F, V76D, and I4F/V76D mutants were generated by QuikChange procedure (Stratagene). Mutagenesis was confirmed by DNA sequencing.

The proteins were expressed and purified as described by Jung et al. [29] except that γ D-crystallin expression was at 25 °C for 10 h and the final gel filtration purification was carried out in Buffer A (9 mM MOPS, 6 mM Tris, 50 mM sodium chloride, 0.1 mM EDTA, 0.02% (w/v) sodium azide; pH 7.2) supplemented with 20% glycerol (v/v). Bimane was added after the second cation exchange column concurrently with pH adjustment to 7. The reactions were incubated for 2 hours at room temperature with 10 \times monobromobimane. The excess label was then removed by the gel filtration step. Stoichiometric labeling at cysteine, C111, was confirmed by MS/MS of intact protein as well as by MS/MS of tryptic protein fragments (data not shown). Predicted molar extinction coefficient of 43,235 M⁻¹cm⁻¹ was used to determine γ D concentrations. α A-, α B- and the triply phosphorylated analog (S19D/S45D/S59D) of α B-crystallin, referred to as α B-D3, were purified as previously described [30–32]. Protein concentrations for α A- and α B-D3 were determined from extinction coefficients [33]: 16,507 and 19,005 M⁻¹cm⁻¹, respectively.

2.2 Circular Dichroism (CD) spectroscopy

Spectra were collected on a Jasco J-810 Spectropolarimeter in 20 mM phosphate buffer, pH 7.1. Near-UV spectra (330–250 nm) were obtained using a 0.5 mg/ml protein solution at room temperature. Far-UV spectra (260–185 nm) were collected at room temperature using a 0.1 mg/ml protein solution.

2.3 Binding of γ D-crystallin to α -crystallin

25 μ M γ D-crystallin was mixed with α -crystallins at the appropriate molar ratio then incubated at 37 °C for 2 h. Bimane emission spectra were collected from 420 to 500 nm following excitation at 380 nm (Photon Technology International (PTI) L-format spectrophotometer). The fluorescence anisotropy of γ D-crystallin was measured on a T-format spectrophotometer (PTI). The fluorescence intensities parallel and perpendicular to the direction of polarized light were analyzed to determine steady state anisotropy values. The samples were excited at 380 nm, and the fluorescence signal was monitored at 465 nm. Binding isotherms were generated by plotting bimane emission or anisotropy at 465 nm as a function of the [α -crystallin/ γ D-crystallin] ratio.

2.4 Analytical size exclusion chromatography (SEC)

Binding between bimane-labeled γ D-crystallins and α -crystallins was analyzed by gel filtration on a Superose 6 10/300 GL Column (GE Healthcare Life Sciences) equilibrated in Buffer A. Samples were incubated at 37 °C for two hours prior to injection. Protein elution was monitored by tandem absorbance (280nm) and fluorescence (excitation 380nm; emission 475nm).

2.5 Thermodynamic analysis of γ D-crystallin mutants

Denaturant-unfolding curves were constructed by monitoring tryptophan fluorescence of γ D as a function of guanidine hydrochloride concentration. Bimane-labeled γ D-crystallin constructs (5 μ M final concentration) were mixed with 1–5 M guanidine hydrochloride (8M stock solution in Buffer A supplemented with 1mM DTT, pH 7.2) and were incubated at 25 °C or at 37 °C for 10 h. The denaturation curves did not change upon further incubation for

up to 24 hours. The curves were fit to a three-state unfolding model [34] by nonlinear least-squares methods using the software Origin (Origin Lab).

3. Results

3.1 Methodology

The topological location of the γ D-crystallin mutants, which cause congenital cataract in mice, is shown in Figure 1. I4 and V76 are located in the N-terminal domain of the molecule in a buried environment. I4F substitutes a conserved isoleucine in the first β -strand of the greek key motif by a phenylalanine. Because the mutation preserves the hydrophobic nature of the side chain, its effects on stability must be mediated by steric repacking. In contrast, the substitution of valine 76, which is located in a loop between β -strands 5 and 6, by an aspartate introduces a charged residue in the packed protein interior. Although Wang et al. [27] noted that the aspartate's carboxylate group could hydrogen bond to nearby arginine residues, this does not appear sufficient to neutralize its destabilizing effect. Finally, we have combined both mutations with the goal of further destabilizing the protein. Our model of α -crystallin chaperone activity (Scheme 1) [16] links its apparent affinity to the fractions of: 1) activated α -crystallin, (sHSP)_a and [35,36] 2) partially or globally unfolded substrate (I and U in Equation 1). Thus, in comparing multiple mutants with different free energies of unfolding, their relative affinities to α -crystallin are expected to parallel reduction in the stability of the native state N.

To investigate γ D-crystallin binding to α -crystallin, we selectively attached a bimane label at cysteine 111 (Figure 1). Bimane labeling marginally reduces the stability of the WT without significant changes in the shape and apparent cooperativity of the unfolding curve (Supplementary Figure 1).

3.2 Structure and unfolding equilibrium of γ D-crystallin mutants

Figure 2A shows that, at room temperature, I4F and V76D substitutions do not change the overall secondary structure of γ D-crystallin. The minimum at 218 nm suggests that γ D-crystallin mutants maintain their β -sheet content although it is somewhat diminished for I4F/V76D. In contrast, the near-UV CD spectrum (Figure 2B) reports significant alterations in the tertiary structure specifically in the environment of aromatic amino acids. These are more accentuated for the V76D and the double mutant I4F/V76D consistent with the expected disruptive effects of a charged residue in the protein hydrophobic core.

That these mutations destabilize γ D-crystallin is evident in denaturant unfolding curves (Figure 3). At 25 °C, left shifts relative to the WT are observed in the low denaturant region leading to biphasic curves. The presence of an unfolding intermediate, hereafter referred to as I, is manifested by a plateau in the 0.5–2.5 M GdnHCl range. Mutations in γ -crystallin N-terminal domain, including V76D, have been shown to stabilize an intermediate with an unfolded N-terminal domain [37]. The midpoint of the transition from the native state, N, to I shifts to lower denaturant concentrations in the mutants relative to the WT with the largest shift observed for the double mutant. The second transition in the unfolding curves obtained at higher denaturant concentrations reflects destabilization of the C-terminal domain leading to a globally unfolded state, U. Thus, Figure 3A demonstrates that I4F and V76D destabilize the native state (N) relative to an intermediate with an unfolded N-terminus. The combination of the two mutations which are in close proximity in the three dimensional structure (Figure 1) further destabilizes N. Table 1 reports the free energy, ΔG^0 , associated with each transition and the dependence of ΔG^0 on denaturant concentration, m. Because the WT unfolding curve does not have two explicit transitions, the fit errors in ΔG^0 are

considerable. In contrast, the parameters for the mutants are more narrowly defined revealing that the V76D substitution is more destabilizing than the I4F substitution.

As expected, stabilities of WT and mutants γ -crystallin are further reduced at the physiological temperature of 37 °C (Table 1). The midpoints for both transitions, from N to I and I to U, shift to lower denaturant concentrations. At this temperature, the native state of the double mutant is marginally stable. The equilibrium constant for the first transition, $[I]/[N]$, is 0.014 suggesting that I is relatively populated even in the absence of denaturant (by comparison the fraction of U is on the order of 10^{-7}).

3.3 α -crystallins interactions with γ D-crystallin mutants

Neither α A-crystallin nor α B-crystallin chaperone binds WT γ D-crystallin at detectable level; even at high concentrations and molar ratios (data not shown). Furthermore, only marginal binding by α A-crystallin was detected for each of the single mutants (Figure 4A). Thus, the I4F and V76D mutations which destabilize the native state relative to the intermediate do not cross the energetic threshold for stable binding to the α -crystallin subunits.

Further destabilization of γ D-crystallin native state by the combination of I4F and V76D is sufficient to promote stable binding to α A-crystallin. As previously described [28], complex formation between α -crystallin and bimane-labeled substrates changes the intensity and/or wavelength of the bimane group. Figure 4A compares the binding levels of the three mutants. Although the fluorescence of I4F and V76D changes upon addition of α -crystallin, the data is scattered and does not trace the shape of a binding isotherm indicating weak binding despite the high concentrations of γ D-crystallin single mutants (50 μ M compared to 25 μ M for the double mutant) and large molar excess of α A-crystallin. In contrast, large and consistent changes in bimane intensity are observed for the double mutant as a function of increasing α A-crystallin molar excess. The binding isotherm can be fit assuming that α -crystallin binds only through its high affinity mode [28], i.e. with a ratio of 4 α A subunits to each γ -crystallin subunit. The complexes between α -crystallin and γ D-crystallin formed within two hours of incubation at 37 °C and binding did not increase with longer incubation times (Supplementary Figure 2).

While the double mutant did not effectively bind α B-crystallin, it formed stable complexes with its phosphorylation mimic (Figure 4B). Phosphorylation of α B-crystallin at serines 15, 45 and 59 [38] enhances binding by shifting the equilibrium of Equation 2 (Scheme 1) towards an activated chaperone form [30]. Substitution of these residues by aspartates to generate α B-D3 mimics phosphorylation's effects on α B-crystallin oligomer equilibrium and chaperone activity. Binding of γ D-crystallin mutants to α B-D3 produced a marginal change in intensity. Therefore, complex formation was detected through changes in bimane anisotropy. Molar excess of α B-D3 leads to a progressively larger anisotropy of bimane-labeled γ D-crystallin due to the expected larger size of the complex between the two proteins. Similar to α A-crystallin, binding of the double mutant was more robust than the single mutants (data not shown) although the dissociation constant was smaller than that obtained with α A-crystallin indicating higher affinity.

3.4 Analysis of the α -crystallin/ γ -crystallin complex by size exclusion chromatography

The binding trends described above were verified by size exclusion chromatography. Binding of monomeric γ D-crystallin to the oligomeric α -crystallin shifts the bimane fluorescence to shorter retention times due to the larger mass of the complex [39]. For high affinity binding, the retention times are close to those of α -crystallin while for low affinity

binding, the chaperone/substrate complex runs at the void volume of a superose 6 column [16].

Figure 5A shows that incubation of α A-crystallin with WT γ D-crystallin or the single mutants I4F (data not shown) and V76D does not result in detectable binding. In contrast, incubation with the double I4F/V76D leads to the concomitant reduction in the γ -crystallin peak intensity (both absorbance and fluorescence) and appearance of a fluorescence peak migrating with a retention time similar to the chaperone. This indicates the formation of a stable complex between the two proteins. At large molar excess of α A-crystallin, the binding is manifested by a distinct change in the shape of its UV absorbance peak.

Figure 5B compares the binding of γ D-crystallin mutants to α B-crystallin and its phosphorylation mimic α B-D3. A 50-fold excess of α B-crystallin is required to detect fluorescence of the complex migrating at retention times similar to the α B-crystallin peak. In agreement with the conclusion of Figure 4, SEC analysis indicates that α B-D3 has the highest affinity to the γ D-crystallin mutants showing detectable binding even for the two single mutants (Figure 5B). The α B-D3/ γ D complex is detectable by absorbance at 280 nm through a distinct peak running as a shoulder to the α B-D3 peak indicating an apparent larger size for the complex as expected (Arrows in Figures 5B and 5C).

4. Discussion

Although previous studies investigated α -crystallin suppression of γ -crystallin thermally-induced aggregation, the data presented here is the first direct quantitation of binding between the two proteins under conditions where γ -crystallin native state is more energetically favorable than its intermediate or unfolded states. We find that destabilization of γ D-crystallin induces binding to α -crystallin. In addition, the stability of the α -crystallin/ γ -crystallin complex evidenced by size exclusion chromatography confirms that the interaction between the two proteins is driven by α -crystallin chaperone activity. However, despite the perturbation of the native structure, binding levels for the single mutants I4F and V76D are low suggesting that destabilization by these mutations is not sufficient to trigger binding. This is unexpected considering the disruption of the tertiary structure by V76D, manifested by the near-UV CD spectrum, and the marginal stability of its native state at 37 °C (Table 1).

To frame these results into a more general context, it is instructive to compare the binding characteristics of the γ D-crystallin mutants to the only other protein to which α -crystallin apparent affinity was extensively characterized, T4 Lysozyme (T4L). Similar to γ D-crystallin, T4L is a highly stable protein with ΔG_{unf}^0 of 15 kcal/mol [31]. Mutations that reduce stability were found to trigger binding to sHSP despite limited structural perturbation of the native state [31]. Apparent dissociation constants of T4L binding to α -crystallin are at least an order of magnitude smaller than those measured for the most destabilized γ D-crystallin mutant investigated here, I4F/V76D [28,40]. Weak affinities to α -crystallin were also observed for β B1-crystallin mutants and attributed to the population of a dimeric intermediate with low affinity to α -crystallin [41]. Together, these results suggest that interaction between lens proteins and α -crystallin in the “chaperone mode” departs from the mechanistic principles deduced from studies of model substrates.

The γ D-crystallin mutants studied here shift the unfolding equilibrium towards I. Table 1 shows that to a first approximation $\Delta G_{\text{N} \rightarrow \text{I}}^0$ decreases while $\Delta G_{\text{I} \rightarrow \text{U}}^0$ remains relatively constant. To illustrate the effects of changes in $\Delta G_{\text{N} \rightarrow \text{I}}^0$ on the fractional population of N, I and U, we carried out simulations based on a three-state equilibrium (Figure 6). Unlike a two-state folding equilibrium, the increase in the population U upon destabilization of N is

scaled by the equilibrium constant of the second transition, $K_{I \rightarrow U}$, which for γ -crystallin is much smaller than 1. Thus, the fractional population of U remains negligible (Figure 6A). Figure 6B demonstrates that significant population of U requires a reduction in the value of $\Delta G^0_{I \rightarrow U}$ which for γ D-crystallin describes the unfolding of the C-terminal domain.

The coupled model of sHSP chaperone activity (Scheme 1) predicts that shift in the substrate folding equilibrium triggers binding if it increases the population of a substrate state recognized by the sHSP. Thus, the relatively large dissociation constants suggest that the γ D-crystallin intermediate, I, has low affinity to α -crystallin. We conclude that similar to T4L [40], α -crystallin high affinity binding requires global unfolding of γ D-crystallin.

The central finding of this paper is that a subclass of destabilizing, cataract-linked mutants of γ -crystallin can escape detection, binding and sequestration by α -crystallin. While α -crystallin binding to destabilized proteins has been extensively studied using model substrates, the results reported here suggest that conclusions based on these studies are not directly extendable to lens proteins. Each mutation or modification has to be investigated in the context of the proper unfolding equilibrium and a quantitative analysis of binding carried out. Another critical and confounding factor not addressed here is molecular crowding [42] which has been invariably neglected regardless of substrate identity or the method of its destabilization. Crowding in lens fiber cells is more peculiar involving the three molecules whose interaction is to be studied. The results presented here suggest that a detailed understanding of these molecular interactions may not be readily derived from *in vitro* studies. Thus analysis of these interactions would require adequate cellular and/or organism models that better capture the intricacies of the protein environment in lens fiber cells. Finally, our findings may have direct relevance to age-related cataract. It is possible that age-related modifications of γ -crystallins, that likewise selectively stabilize I, fail to interact with α -crystallin allowing them to nucleate aggregation.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

The authors thank Edward Monson for technical assistance and Hanane Koteiche for critically reading the manuscript. This work was supported by Grants EY12018 and P30EY008126 from the National Institutes of Health.

Abbreviations

αB-D3	α B-crystallin S19D/S45D/S59D
GdnHCl	guanidine hydrochloride
sHSP	small heat-shock proteins
T4L	T4 lysozyme
WT	wild type

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- How can destabilizing mutations of lens γ -crystallin cause cataract?
- We characterize the unfolding equilibrium of two cataract-causing mutations and investigate their interactions with α -crystallins
- Cataract-causing mutations stabilize an unfolding intermediate of γ D-crystallin
- Cataract-causing mutants of γ D-crystallin are not efficiently bound by lens α -crystallin
- Destabilized variants of γ D-crystallin, due to mutations or age-related modifications, can nucleate aggregation leading to lens opacity and cataract

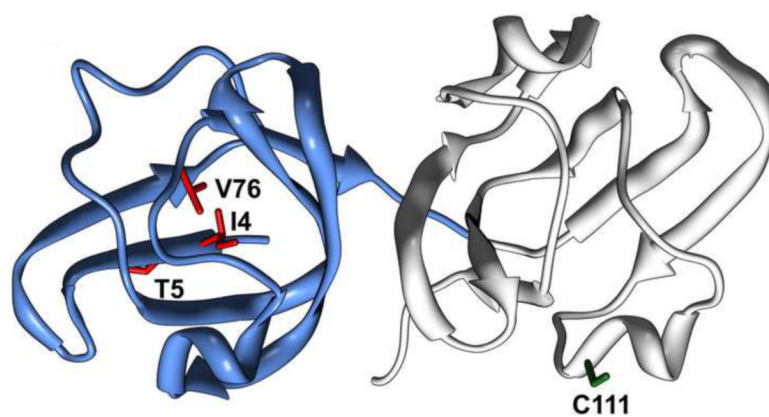


Figure 1. Ribbon representation of γ D-crystallin structure (pdbID 1HK0). The N-terminal domain is colored in blue; sites of cataract-causing mutations are highlighted in red and cysteine 111 in green.

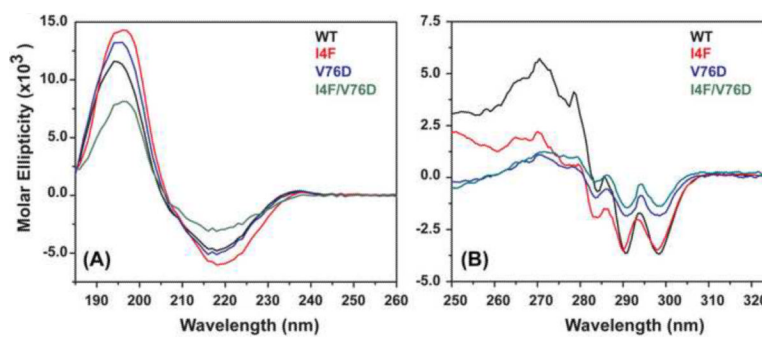


Figure 2. Room temperature CD analysis of γ D-crystallin mutants. (A) Far-UV spectra showing a pronounced minimum at 218 nm. (B) Near-UV spectra reporting changes in the tertiary fold as a consequence of the mutations.

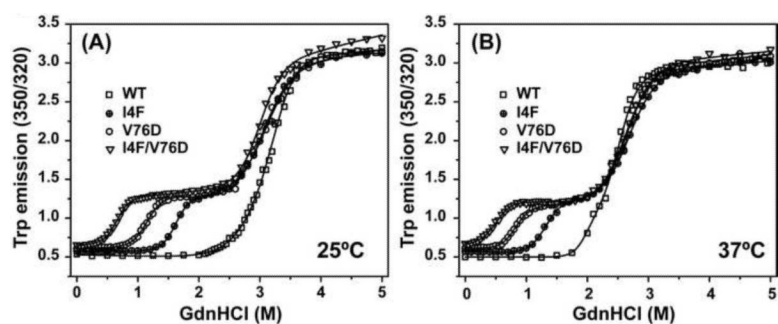


Figure 3. Denaturant unfolding curves of γ D-crystallin mutants at (A) 25 °C and (B) 37 °C. The substitutions lead to the appearance of two explicit transitions compared with the WT monophasic curve. The first transition corresponds to the population of an intermediate with an unfolded N-terminal domain. The second transition corresponds to unfolding of the C-terminal domain. The solid lines represent non-linear least-squares fit yielding the parameters reported in Table 1.

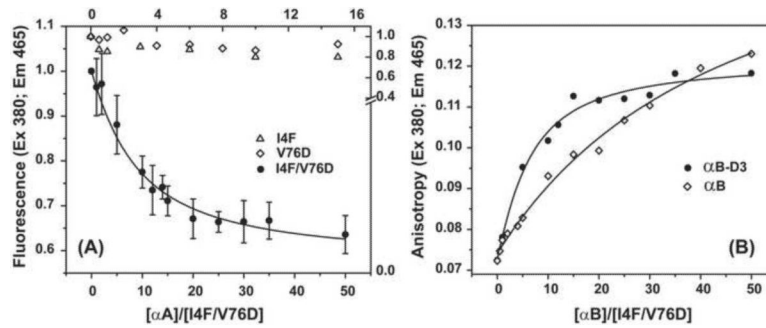


Figure 4. Binding isotherms of γ D-crystallin mutant to (A) α A-crystallin and (B) α B-crystallin and its phosphorylation mimic α B-D3. The concentration of the single mutants was 50 μ M while the double mutant was 25 μ M. Samples were incubated at 37 $^{\circ}$ C for two hours at the appropriate ratio as described in the methods section. The solid lines for the double mutant I4F/V76D in (A) and (B) are non-linear least-squares fit assuming high affinity binding where 4 α -crystallin subunits bind one γ -crystallin. The dissociation constants are 43 ± 7 and 27 ± 6 μ M for α A-crystallin and α B-D3, respectively. For α B-crystallin, the dissociation constant was larger than 300 μ M.

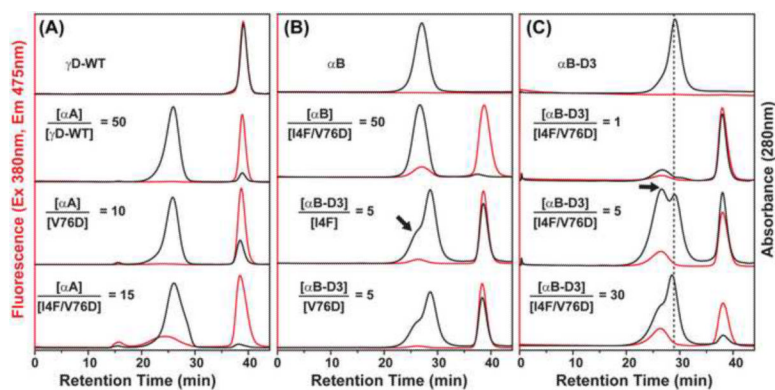


Figure 5.

Analysis of γ D-crystallin mutant binding to α -crystallin by size exclusion chromatography (SEC) detected by in-line absorbance at 280 nm and fluorescence at 475 nm. (A) Samples of α A-crystallin and γ D-crystallin were incubated at the indicated molar ratios and injected on a Superose 6 column. The binding of γ D-crystallin I4F/V76D is manifested by a fluorescence peak with retention time similar to that of α A-crystallin. (B) Comparative SEC analysis of binding to α B-crystallin and α B-D3. A 50-fold molar excess of the former is required for detectable binding of the double mutant while α B-D3 binds the single mutants. The y axes were scaled to show details of the complex peak. (C) The complex between α B-D3 and γ D-crystallin is detected as a distinct peak, indicated by the arrow, migrating at a larger mass than α B-D3 alone.

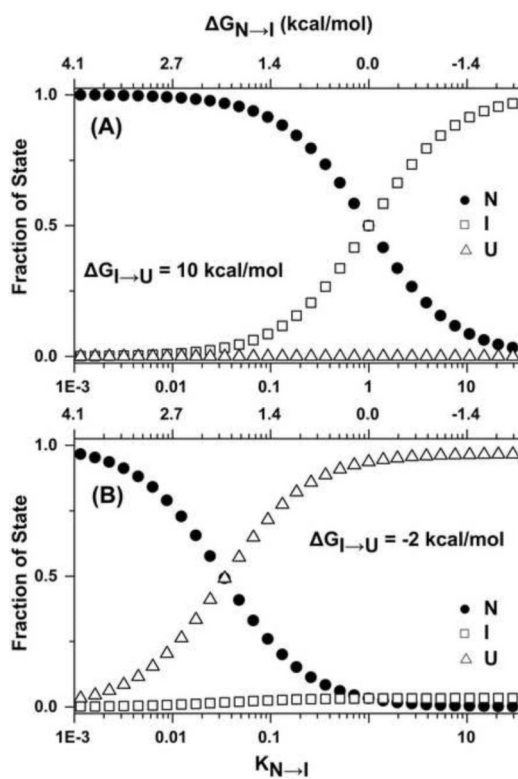


Figure 6. Fraction of N, I and U as a function of changes in $\Delta G_{N \rightarrow I}^0$ for (A) $\Delta G_{I \rightarrow U}^0 = 10$ kcal/mol and (B) $\Delta G_{I \rightarrow U}^0 = -2$ kcal/mol. For (B) shifts in the $\Delta G_{N \rightarrow I}^0$ lead to a large increase in the fraction of U. These simulations apply to three-state unfolding equilibria.



scheme 1.

Table 1
Equilibrium unfolding/refolding parameters for γ D-crystallin wild-type and mutants at pH 7.2

γ D-crystallin	Temperature °C	$N \rightleftharpoons I$		$I \rightleftharpoons U$	
		$\Delta G_{N \rightarrow I}^0$ kcal/mol	m	$\Delta G_{I \rightarrow U}^0$ kcal/mol	m
WT	25	6.23 ± 17.9	2.08 ± 3.90	10.4 ± 24.1	3.12 ± 6.77
	37	10.7 ± 18.7	5.46 ± 9.46	10.3 ± 3.88	4.08 ± 1.60
I4F	25	8.50 ± 2.18	5.26 ± 1.39	7.67 ± 1.30	2.48 ± 0.40
	37	5.99 ± 0.78	4.52 ± 0.62	7.56 ± 0.62	2.83 ± 0.23
V76D	25	5.48 ± 0.66	4.71 ± 0.58	7.85 ± 0.77	2.59 ± 0.25
	37	4.36 ± 0.92	5.38 ± 1.08	8.14 ± 0.84	3.07 ± 0.31
I4F/V76D	25	3.96 ± 0.62	5.99 ± 0.83	9.68 ± 1.04	3.24 ± 0.35
	37	2.67 ± 0.54	5.07 ± 0.74	7.37 ± 0.36	2.81 ± 0.13

ΔG^0 , m, and $C_{1/2}$ are the free energy, the denaturant dependence of ΔG^0 , and the concentration of urea at the transition midpoint, respectively, for the two transitions of γ D-crystallin unfolding. ΔG^0 , m are reported with the associated fit error.