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Management of 3-aminopyridine-2-carboxaldehyde thiosemicarbazone-induced methemoglobinemia

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Abstract

The anticancer agent 3-aminopyridine-2-carboxaldehyde thiosemicarbazone is a ribonucleotide reductase inhibitor. It inactivates ribonucleotide reductase by disrupting an iron-stabilized radical in ribonucleotide reductase's small subunits, M2 and M2b (p53R2). Unfortunately, 3-aminopyridine-2-carboxaldehyde thiosemicarbazone also alters iron II (Fe²⁺) in hemoglobin. This creates Fe³⁺ methemoglobin that does not deliver oxygen. Fe²⁺ in hemoglobin normally auto-oxidizes to inactive Fe³⁺ methemoglobin at a rate of nearly 3% per day and this is counterbalanced by a reductase system that normally limits methemoglobin concentrations to less than 1% of hemoglobin. This balance may be perturbed by symptomatic toxicity levels during 3-aminopyridine-2-carboxaldehyde thiosemicarbazone therapy. Indications of 3-aminopyridine-2-carboxaldehyde thiosemicarbazone sequelae attributable to methemoglobinemia include resting dyspnea, headaches and altered cognition. Management of methemoglobinemia includes supplemental oxygen, ascorbate and, most importantly, intravenously administered methylene blue as a therapeutic antidote.

Keywords

methemoglobinemia; ribonucleotide reductase; triapine

Ribonucleotide reductase (RNR) provides cells with deoxyribonucleoside triphosphates demanded for DNA synthesis and repair [1–5]. RNR consists of two large M1 subunits and either two small M2 [6] or M2b (p53R2) [7,8] subunits. The small subunits house an iron-stabilized free radical that shuttles to and from the enzyme's active site in the large subunit [9]. Well-known inhibitors of RNR include hydroxyurea [10], which annihilates the radical, and gemcitabine [11], which becomes a cytidine diphosphate analog that covalently destroys

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RNR's substrate binding site [12]. These and other RNR inhibitors in cancer therapy were recently reviewed [13,14].

The RNR inhibitor 3-aminopyridine-2-carboxyaldehyde thiosemicarbazone (3-AP; NSC #663249) holds promise as an anticancer agent [15,16]. Timed after DNA damage (e.g., damage promulgated by ionizing radiation), 3-AP's cell death-provoking effect may be due to a cell's protracted inability to supply on-the-spot deoxyribonucleoside triphosphates, which are needed for DNA damage repair [2,3]. This idea led to anticancer clinical trials that tested RNR inhibition by 3-AP when administered alone compared with coadministration with DNA-damaging chemotherapy or radiation [15–20]. A major dose-limiting toxicity in early clinical trials was symptomatic dyspnea due to treatment-related methemoglobinemia. Methemoglobinemia is a reversible condition in which greater than 15% of a patient's hemoglobin is incapable of carrying oxygen because its iron is oxidized. Because 3-AP efficacy presumably depends on its disruption of the iron-stabilized tyrosyl free-radical site of RNR's small subunit (M2 or M2b), its hemoglobin iron toxicity may be inseparable from its efficacy. In this review, we discuss methemoglobin metabolism, the pharmacodynamics of RNR inhibitor methemoglobinemia and its treatment.

Hemoglobin & methemoglobin metabolism

Hemoglobin in erythrocytes carries oxygen in reversible association with iron in a reduced, ferrous Fe^{2+} state. Oxygenated Fe^{2+} hemoglobin iron oxidizes to Fe^{3+} methemoglobin and superoxide at a rate of approximately 3% per day. Methemoglobin returns to hemoglobin by action of cytochrome b5 reductase and cytochrome b5 (upper path in Figure 1) [21]. This pathway accounts for 94% of the conversion of methemoglobin to hemoglobin [22] and normally maintains methemoglobin levels below 1% of total hemoglobin. Dyspnea is observed when methemoglobin blood levels reach 25% [23].

The mechanism of RNR inhibition by 3-AP is via inactivation of the tyrosyl free radical within the M2 or M2b (p53R2) small subunits [24,25], which in effect, is a molecular interaction of an Fe^{2+} -3-AP chelate and of oxygen generating local reactive oxygen species capable of annihilating the nearby tyrosyl free radical. In a similar manner, an Fe^{2+} -3-AP chelate impairs methemoglobin-hemoglobin cycling (Table 1) wherein 3-AP-induced methemoglobinemia occurs in 23% of the treated patients on clinical trials [15,20]. Other RNR inhibitors do not cause methemoglobinemia because their mechanisms of action are different: hydroxyurea, as a one-electron reductant, disrupts the free radical in RNR M2 and M2b subunits but does not associate directly with molecular iron; and gemcitabine blocks RNR's M1 subunit but does not interact with iron in the M2 and M2b subunits. In contrast to 3-AP, chemicals, such as parabactin and desferrioxamine, chelate intracellular iron pools. By creating low intracellular iron levels, these chemicals interfere with activation and reactivation of iron moieties in RNR after spontaneous loss of iron from the native enzyme [26].

Pharmacodynamics of 3-AP methemoglobinemia

Two clinical trials monitored methemoglobin after 3-AP infusion (Table 1). In the first Phase I dose-escalation clinical trial in patients with advanced solid cancers [20], 3-AP was administered intravenously over 2–4 h at dose levels of 105, 140 or 185 mg/m^2 on days 1, 8 and 15 of each 28-day cycle. In this study, gemcitabine was also given over a 30-min intravenous infusion 1–4 h after 3-AP infusion at a dose level of 0–1000 mg/m^2 . Of 26 patients, a total of six manifested dyspnea at approximately the time of 3-AP infusion. In three patients, methemoglobin levels were determined and were 12, 12 and 11%. A suggested association of 3-AP-related dyspnea was claimed.

Because of this, a second Phase I dose-escalation clinical trial in patients with advanced-stage cervical cancer tracked serum levels of 3-AP and methemoglobin [15]. In this trial, 3-AP (25 or 50 mg/m²) was given as a three-times weekly intravenous 2-h infusion during once-weekly cisplatin (40 mg/m²) and daily radiation. Here, 3-AP serum concentrations were measured in heparinized blood samples by high-pressure liquid chromatography tandem mass spectrometry with parallel online turbulent flow extraction and positive ion selected reaction monitoring [15]. Samples were taken on day 1 and day 10 of treatment before, and at 2, 4, 6 and 24 h after the start of the 2-h 3-AP infusion (Figure 2). For procedural detail, serum concentrations of 3-AP were determined according to a modified internally standardized high-pressure liquid chromatography UV procedure provided by the manufacturer [JD TALTON, NANOTHERAPEUTICS, INC., FL, USA, PERS. COMM.]. Linear calibration curves of 3-AP:internal standard peak height ratios versus 3-AP concentration were established over a concentration range of 0.02–10 µg/ml of 3-AP in serum for each analyzed batch of serum specimens. Quality-control specimens at low (0.05 µg/ml), medium (0.8 µg/ml), and high (8.5 µg/ml) concentrations of 3-AP were determined concurrently with each analytical specimen batch. Permissible control ranges were ± 20, ± 15 and ± 15% of nominal concentration for the low-, medium- and high-concentration quality-control specimens, respectively. Typical within-day precision for replicate determinations of the low-, medium- and high-concentration quality-control specimens was 6–21, 5–8 and 2–6%, respectively (relative standard deviation [SD], n = 6 within-day replicates, 3 days of results compared). Interday precision for six replicate determinations of the low-, medium- and high-quality-control specimens during method validation studies was 21, 7 and 5%, respectively (relative SD, n = 3). Corresponding time-point serum methemoglobin concentrations were determined by direct spectrophotometry with a verified detection range of 0–100% [27,28]. Methemoglobin specimens were drawn into heparinized arterial blood-gas syringes on wet ice and measured within 10 min of procurement.

Patient 3-AP and methemoglobinemia levels were measured in blood samples collected for pharmacokinetic evaluation from ten patients with advanced-stage cervical cancer. A total of 20 treatment courses (ten on day 1 and ten on day 10) were evaluated for each of two dose levels of 3-AP (25 mg/m² [n = 6] or 50 mg/m² [n = 4]) studied. Maximal plasma concentrations (C_{max}) of 3-AP were measured at the termination of the 2-h intravenous 3-AP infusion. The mean 3-AP peak plasma concentration was 262 ng/ml (SD = 51 ng/ml) at the 25 mg/m² dose level and 560 ng/ml (SD = 59 ng/ml) at the 50 mg/m² dose level. The mean elimination half-life was 2 h, with no change in elimination half-life between day 1 and day 10 for either the 25 mg/m² (p = 0.18) or 50 mg/m² (p = 0.35) dose levels. The plasma concentration of 3-AP 6 h after the start of the 2-h infusion fell to 2% C_{max} (25 mg/m²) and 13% C_{max} (50 mg/m²). The corresponding peak methemoglobin level was 1% (SD = 0.6%) at the 3-AP 25 mg/m² dose level and was 6% (SD = 2.8%) at the 3-AP 50 mg/m² dose level.

Although patient numbers were small in the two clinical trials performed, these two trials suggest that higher doses of 3-AP lead to decreased mean oxygen saturation and higher peak methemoglobin levels (Table 1). For example, the peak methemoglobin levels recorded in one of these studies follow peak serum 3-AP levels by a 2 h delay (Figure 2). The elevated methemoglobin levels usually resolved within 2 h of the peak (Figure 2). Repeated 3-AP dosing had no substantial additive effect (3-AP at 25 mg/m²: day 1 mean [SD] = 1% [0.6%], day 10 mean [SD] = 0.9% [0.4%]; 3-AP at 50 mg/m²: day 1 mean [SD] = 6% [2.8%] day 10 mean [SD] = 4.7% [2.2%]), suggesting that mechanisms responsible for recycling methemoglobin were not irreversibly impaired.

For clinical and interpretative context, chemically-induced methemoglobinemia occurs most commonly from inhalation or tactile exposure to oxidizing chemical agents. Agents known

to induce methemoglobinemia up to 15% after prolonged exposure, include acetonitrile (e.g., nail varnish remover), anesthetics (e.g., lidocaine, prilocaine), aniline dyes, chlorates (e.g., matches, explosives, weed killers), naphthalene (e.g., moth balls), volatile nitrites, phenazopyridine (e.g., pyridium), quinones (e.g., chloroquine and primaquine), sulfonamides (e.g., sulfamethoxazole) and dapsone. Mechanisms of methemoglobin induction involve direct oxidizing effects on the hemoglobin or generation of oxygen and peroxide free-radicals capable of oxidizing hemoglobin.

Clinical signs & symptoms of methemoglobinemia

Relative to hemoglobin, the oxygen dissociation curve of methemoglobin is left-shifted [29]. This means that methemoglobin binds tighter to oxygen, and thus, less oxygen is released for tissue perfusion. While healthy individuals have few symptoms when methemoglobin constitutes up to 15% of total hemoglobin, concentrations of 20–30% may lead to symptomatic dyspnea, headache and cognitive changes. Patients with anemia, pulmonary disease, abnormal hemoglobins and the elderly may have symptoms at lower methemoglobin levels. When methemoglobins rise above 30%, adverse sequelae may become more severe and lead to cardiac dysrhythmias, seizures, coma and, if devastating organ function ensues, patient death.

Inhibitors of RNR, such as 3-AP, are likely to induce low-level (+1–+10%) methemoglobinemia in all patients administered with the drug. 3-AP may also promote metabolic acidosis and hypoxia independent of its iron-chelating properties. Treating physicians and staff should be aware of these phenomena, and should consider implementing an antidote therapy (e.g., methylene blue) when patients self-report symptoms such as shortness of breath, headache or altered cognition.

Methylene blue as an antidote for RNR inhibitor-induced methemoglobinemia

Three antidotes for 3-AP-induced methemoglobinemia include methylene blue, ascorbate, and glutathione. Methylene blue infusion has been advocated for treatment of symptomatic methemoglobinemia owing to the dye's ability to donate an electron for nonenzymatic reduction of methemoglobin. In this reaction, NADPH methemoglobin reductase converts the oxidized form of the dye (methylene blue) to the reduced form (leukomethylene blue) using the cofactor NADPH (Figure 2). The reduced dye biochemically reduces Fe^{3+} iron in methemoglobin to Fe^{2+} hemoglobin, with the dye being recycled. Of note, oximeter readings of blood oxygen saturation may be obscured by methylene blue [30]. Also, NADPH methemoglobin reductase accounts for 6% of restoration of hemoglobin from methemoglobin and is a distinct enzyme from its counterpart NADH-cytochrome b5 reductase. Chemical restoration of hemoglobin from methemoglobin by ascorbate and glutathione are beyond the scope of this review [31].

It is important for treating physicians to recognize that patients receiving 3-AP, who have a known glucose-6-phosphate dehydrogenase (G6PD) deficiency, are more susceptible to cellular oxidative stress and related adverse events. Patients with G6PD deficiency are unable to produce the NADPH needed to maintain glutathione levels. Since the methylene blue antidote system requires NADPH as a cofactor in the reaction, methylene blue will be ineffective in these individuals. Indeed, methylene blue administration to patients with G6PD deficiency could actually worsen methemoglobinemia due to depletion of cellular reserves of NADPH. Under comorbid conditions of profound cell hemolysis resulting from chemotherapeutics, erythrocyte enzymes that are needed to reduce methylene blue are released and the antidote is ineffective. If methylene blue cannot be reduced, it can act as an

oxidizing agent converting hemoglobin to methemoglobin, thus exacerbating the condition it is intended to treat.

Expert commentary & conclusion

Encouraging early results of 3-AP radiochemotherapy in the management of cervical cancer [15,16] raises the possibility of future cancer therapy clinical trials, utilizing RNR inhibitors paired with DNA-damaging therapies. RNR inhibitors, such as 3-AP, are associated with excellent targeted disease control, but are also linked to dose-limiting methemoglobinemia. Further study of the manipulation of hemoglobin–methemoglobin metabolism is warranted prior to widespread clinical use of RNR inhibitors in anticancer management strategies. The present literature on intravenous 3-AP in cervical cancer management suggests that the biologically effective 25-mg/m² 3-AP dose level is safe and not associated with burdensome treatment-related methemoglobinemia and dyspnea. Our findings in this study confirm this sentiment. Finally, while it is important to investigate the manipulation of iron in erythrocyte hemoglobin during intravenous administration of 3-AP, it is also equally critical to evaluate the effects of oral administration of 3-AP on cellular iron. Both excitement and caution are appropriate in interpreting available properties of 3-AP for cancer therapies.

Future perspective

Clinical trials incorporating RNR inhibitor radiochemotherapy are eagerly awaited over the next 5 years. Understanding of iron-chelating RNR inhibitors, such as 3-AP, and their resultant anticipated low levels of methemoglobinemia are critical to the gradual acceptance of this combined approach to anticancer management. RNR inhibitors, such as 3-AP, seemingly disrupt iron at the free radical site of RNR's small subunit (M2 or M2b), making its hemoglobin iron toxicity inseparable from its efficacy. It is hoped that pharmacodynamic results from clinical trials of 3-AP radiochemotherapy will, at least in part, evaluate the methemoglobin-inducing effect of oral 3-AP, as this form of the drug may be expected to be more widely used in the worldwide management of cervical cancer due to its logistical advantage of convenient cost-effective administration.

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Executive summary

Ribonucleotide reductase rate-limits the conversion of ribonucleotide to deoxyribonucleotides

- To catalyze this reaction, an iron-stabilized free radical in the small subunit of ribonucleotide reductase shuttles to and from the enzyme's large subunit active site.

3-aminopyridine-2-carboxaldehyde thiosemicarbazone interacts with iron II

- 3-aminopyridine-2-carboxaldehyde thiosemicarbazone (3-AP)–Fe²⁺ interactions disrupt a free radical in the small subunits of ribonucleotide reductase to inactivate the enzyme.

3-AP disrupts hemoglobin & methemoglobin metabolism

- 3-AP also interacts with Fe²⁺ in hemoglobin, interrupting recycling of spontaneously generated methemoglobin (Fe³⁺) back to Fe²⁺ hemoglobin.

Anticancer inhibitors of ribonucleotide reductase manifest methemoglobinemia in human clinical trials

- Shortness of breath, headaches and altered cognition are observed with these anticancer agents.
- Methemoglobinemia may be managed by supplemental oxygen, ascorbate and, most importantly, intravenously administered methylene blue as a therapeutic antidote.

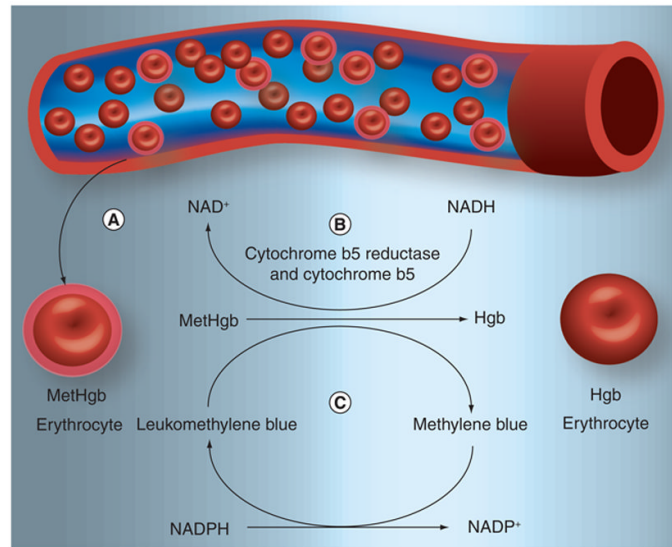


Figure 1. Recycling of methemoglobin to hemoglobin

Normal erythrocyte Hgb carries oxygen in a reversible association with reduced or ferrous iron (Fe^{2+}). Oxygenated Fe^{2+} Hgb oxidizes to Fe^{3+} MetHgb and superoxide at a rate of approximately 3% per day. **(A)** In the presence of 3-aminopyridine-2-carboxaldehyde thiosemicarbazone (3-AP), 3-AP interacts with the Fe^{2+} of Hgb in order to form 3-AP- Fe^{3+} , which, in effect, creates MetHgb. **(B)** MetHgb may be reduced to Hgb by a combination of cytochrome b5 reductase and cytochrome b5. **(C)** The MetHgb antidote, methylene blue, can also facilitate this reaction. Hgb: Hemoglobin; MetHgb: Methemoglobin.

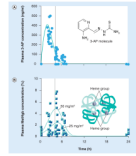


Figure 2. 3-aminopyridine-2-carboxaldehyde thiosemicarbazone pharmacokinetics and methemoglobin pharmacodynamics

Plasma concentration of 3-AP was determined in ten women with cervical cancer undergoing 3-AP radiochemotherapy. **(A)** Peak 3-AP concentration was observed at the end of a 2-h infusion. **(B)** Peak methemoglobin levels were detected 2 h after discontinuation of 3-AP infusion. **(A & B)** A vertical bar denotes the 4-h time point after the start of infusion. 3-AP: 3-aminopyridine-2-carboxaldehyde thiosemicarbazone; MetHgb: Methemoglobin.

Table 1

Methemoglobinemia induced by 3-aminopyridine-2-carboxaldehyde thiosemicarbazone.

Number of patients	3-AP dose (mg/m ²)	Infusion length (h)	Reaction	Mean pulse O ₂ saturation (%)	Mean peak methemoglobin (%)	Ref.
1	105	2	Dyspnea, hypertension	75	12	[20]
2	105	4	Dyspnea, pallor	88	11	
6	25	2	None	96	1	[15]
4	50	2	None	94	6	

3-AP: 3-aminopyridine-2-carboxaldehyde thiosemicarbazone.