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Industrial toxicants and Parkinson's disease

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Abstract

The exposure of the human population to environmental contaminants is recognized as a significant contributing factor for the development of Parkinson's disease (PD) and other forms of parkinsonism. While pesticides have repeatedly been identified as risk factors for PD, these compounds represent only a subset of environmental toxicants that we are exposed to on a regular basis. Thus, non-pesticide contaminants, such as metals, solvents, and other organohalogen compounds have also been implicated in the clinical and pathological manifestations of these movement disorders and it is these non-pesticide compounds that are the subject of this review. As toxic exposures to these classes of compounds can result in a spectrum of PD or PD-related disorders, it is imperative to appreciate shared clinico-pathological characteristics or mechanisms of action of these compounds in order to further delineate the resultant disorders as well as identify improved preventive strategies or therapeutic interventions.

Introduction

Parkinson disease (PD) is a progressive neurodegenerative movement disorder that affects 1% of people over the age of 55, increasing to 5% by 85 years of age. Although the average age of onset is 70 years old, a significant number of patients (~4%) will develop early-onset PD, which can occur before they are 50 years old (Farrer, 2006). Clinically, PD is predominantly defined as a movement disorder, characterized by an alteration in the ability to initiate and maintain normal movement, manifesting as slowness of movement, a resting tremor, and postural instability. These symptoms are a direct result of the loss of dopamine-producing neurons located in the substantia nigra pars compacta (SNpc) of the midbrain and a concomitant reduction of the neurotransmitter dopamine and dopaminergic terminals in the caudate and putamen of the striatum (Fahn, 2003). These deficits can be readily assessed in the awake patient through the use of several imaging techniques, including positron emission tomography (PET) and single photon emission computer tomography (SPECT), which both incorporate the use of ligands that are specific for dopaminergic transporters or receptors. Importantly, many of these motor symptoms can be alleviated, albeit not completely or permanently, through dopamine replacement therapies, such as L-DOPA (Fahn et al., 2004).

In addition to motor abnormalities, PD patients can also present with a suite of non-motor problems that accompany or may even precede the motor issues and can range from

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gastrointestinal and cognitive deficits to olfactory and sleep disturbances. While damage to the nigrostriatal dopamine system underlies the motor perturbations, many of the peripheral alterations are thought to arise from damage to other neurotransmitter systems, including loss of noradrenergic neurons and projections from the locus coeruleus as well as cholinergic deficits in the nucleus basalis of Meynert (Fahn, 2003).

It should be kept in mind that PD is only a singular syndrome that is a part of a much larger clinico-pathological definition of movement disorders, defined as parkinsonism (Tuite and Krawczewski, 2007). Parkinsonism encompasses multiple different movement disorders that all seem to share a similar tetrad of movement deficits, including rigidity, tremor, slowness of movement, and postural instability, although these may be present in different combinations. The etiology of these deficits is extremely varied, ranging from multiple systems atrophy and progressive supranuclear palsy to drug- or toxicant-induced syndromes, such as carbon monoxide and manganese. Furthermore, the cause of parkinsonism will be further defined by the presence or absence of pathological signs and symptoms. For instance, while PD shows a discrete loss of dopaminergic neurons in the SNpc and a favorable response to dopamine replacement therapy, other conditions such as parkinsonism as a result of manganese toxicity do not (Guilarte, 2010). It can be appreciated that this spectrum of parkinsonian clinico-pathological presentations can significantly complicate the differential diagnosis of PD and other parkinsonian disorders.

Although PD is primarily viewed as a disease of aging the signs and symptoms of PD can be accelerated through a genetic predisposition to the disease or exposure to an environmental risk factor (Farrer, 2006). To date, mutations to several genes have been identified as genetic risk factors for the disease, yet these genetic alterations are only able to account for 5–10% of the cases of PD. This would suggest that there are exogenous or environmental factors that influence the risk of development of PD, that either work independently or in conjunction with genetic predisposition to facilitate the onset of the disease (Gao and Hong, 2011).

Indeed, work over the last several decades has provided extensive support for the idea that exposure to different environmental factors could be a significant risk factor for the development of PD (Wirdefeldt et al., 2011). The first indication that an exogenous insult could be responsible came in the form of I.V. drug users who had injected a synthetic meperidine compound that was contaminated with the neurotoxic species, 1-methyl-4-phenyl-1,2,3,6-tetrahydropyridine (MPTP), causing an acute onset of an irreversible parkinsonian state that resembled idiopathic PD (Langston et al., 1983, Markey et al., 1984). MPTP is rapidly taken up into the brain and quickly converted to its neurotoxic metabolite, MPP⁺ by astrocytes. It is then extruded from the astrocytes by the organic cation transporter 3 (Cui et al., 2009). Once in the extracellular space MPP⁺ is shuttled through the dopamine transporter located on the presynaptic terminal of dopamine neurons and is transported into the mitochondria where it inhibits mitochondrial respiration, resulting in the death of dopaminergic neurons and their projections in the SNpc and striatum, respectively. The role of an environmental factor or factors was further bolstered through a series of epidemiological studies that examined the incidence of PD in mono- and dizygotic twins (Tanner et al., 1999). Through this work it was uncovered that the incidence of PD in each of these groups was virtually identical suggesting the heritability of PD was low. More importantly, it provided evidence that an exogenous factor was significantly influencing the relative risk of PD.

While multiple studies have provided extensive support for an environmental role in the etiopathogenesis of PD, the identification of certain settings or toxicants has remained relatively elusive. However, in the last several years a significant effort has been made to

identify particular toxicants or classes of toxicants, which may promote the development of the disorder. While exposure to pesticides and other agricultural products has received a considerable amount of attention and has been demonstrated to be a substantial contributor to the incidence of PD, they represent only a single class of environmental toxicants to which the human population is routinely exposed (Priyadarshi et al., 2000, Ascherio et al., 2006). The reader is referred to other reviews on the subject (Dick et al., 2007, Hatcher et al., 2008). In addition to agricultural products, industrial contaminants are beginning to receive recognition as potential risk factors for the development of PD (Steenland et al., 2006, Goldman, 2010, Seegal et al., 2010). Industrial toxicants are a broad and diverse class of compounds that are utilized in the manufacture and production of various commercial and household products, ranging from the use of carbon disulfide in the vulcanization of rubber to brominated flame retardants in the insulation of electrical components in a computer. As with pesticides, the human population is routinely exposed to industrial toxicants, either through an occupational setting or via contaminated food or their presence in everyday household products.

In this review we will direct our focus to relevant clinical and research findings, that provide support for the role of industrial toxicants in the etiology of PD and other parkinsonian disorders. Our decisions for and discussion of a particular compound was based upon its adherence to one or more criteria: 1). Has this compound been associated with PD in the human population? 2). Does exposure to this compound produce a movement disorder associated with PD or parkinsonism in the human population? 3). Does the compound target similar mechanisms and produce a similar pathology as that seen in PD?

Organohalogen Industrial Contaminants

Organohalogenated compounds (OHC) are a class of carbon-containing chemicals with varying degrees of halogen substitution on carbon atoms. The most prevalent halogen substitutions are comprised of chlorine, bromine, and fluorine to produce organochlorine, organobromine, and organofluorine compounds, respectively. The use of these compounds in an industrial setting is varied and can range from electrical insulating coatings, flame-retardant oils and in adhesives and plastics. The same properties that have made these compounds so attractive to industrial use, including being highly stable and resistant to degradation, have also made them extremely dangerous to the environment and the human population. Although there are several different compounds that are classified as being organohalogens, we will focus our discussion on the two that have received the most attention given their effects on the brain and as a risk factor for PD.

Polychlorinated Biphenyls

Polychlorinated biphenyls (PCBs) were first introduced into industrial and commercial use around 1930 and are composed of biphenyl rings with varying degrees and positions of chlorine substitutions present on each ring, allowing for the production of 209 different conformations or congeners. In the United States PCBs were predominantly manufactured as mixtures of various congeners and sold under the trade name of Aroclor. Of these mixtures, Aroclor 1242, 1254 and 1260, which contain 42%, 54% and 60% chlorine, respectively, were the major mixtures produced. Because of their thermal stability, PCBs have been widely used as coolants and lubricants in transformers and capacitors, in hydraulic fluids, as well as other electrical equipment (Erickson, 1986). Although the manufacturing and use of PCBs was discontinued in 1977 in the United States, their physiochemical properties, allow them to persist and bioaccumulate in the environment, increasing the risk for human exposure (Safe, 1993, Kamrin et al., 1994).

One danger of PCBs is their high degree of lipophilicity, allowing them to preferentially deposit in lipid-rich regions of the body, such as adipose tissue and the brain. This characteristic of PCBs coupled with the difficulty in degrading or metabolizing and eliminating them from the body, allows them to accumulate within the body and create a toxic setting of constant exposure to these compounds over several years. This type of exposure is of serious concern as PCBs have been shown to be detrimental to the central nervous system (Faroon et al., 2001). Both epidemiological as well as laboratory studies have routinely demonstrated PCB exposure to be a key mediator of several neurological deficits including alterations in cognitive function and motor development. Whether the exposure was *in utero* or later in adulthood was inconsequential as alterations in these neurobehavioral parameters were observed. In addition to these neurobehavioral deficits, exposure to PCBs has been associated with the incidence of PD. The brain has an extremely high concentration of lipids and PCBs are found to deposit ubiquitously within the brain. Indeed, Corrigan et al., (1998, 2000) identified significant levels of total PCBs and several of their major congeners in human brain. Interestingly, when stratified between patients with PD and pathological controls the total concentration of PCB as well as the concentrations of specific PCB congeners were found to be elevated in the PD specimens. Furthermore, a recent epidemiological study demonstrated an increased incidence of PD in females who had been occupationally exposed to PCBs while working in an electrical capacitor plant (Steenland et al., 2006). The relevance of gender in this finding is unclear.

More recently, the potential cellular and molecular pathways and mechanisms responsible for the association between PCB exposure and PD are being uncovered and have identified several neuronal targets, including alterations in calcium homeostasis (Kodavanti et al., 1998), generation of oxidative stress, and disruption to neurotransmitter systems (Fonnum et al., 2006, Lyng et al., 2007, Lyng and Seegal, 2008, Lee et al., 2011). Of these, alterations to dopamine neurotransmission have been the most extensively studied, both in *in vitro* and *in vivo* model systems. Of these studies, the consensus finding is that exposure to PCBs results in a reduction of dopamine, either in dopaminergic cells lines or slices as well as from the nigrostriatal system of mice, rats, and nonhuman primates (Seegal et al., 1986, Seegal et al., 1989, Seegal et al., 1990, Seegal et al., 1991, Shain et al., 1991, Seegal et al., 1994, Seegal et al., 1998, Lee and Opanashuk, 2004, Richardson and Miller, 2004). The mechanisms leading to the reduction in dopamine levels are varied as exposure has been demonstrated to inhibit tyrosine hydroxylase and aromatic acid decarboxylase, which are both enzymes involved in the synthesis of dopamine. Additionally, PCBs have been shown to alter the expression and function of the plasmalemmal dopamine transporter (DAT) as well as the vesicular monoamine transporter (VMAT2) to sequester dopamine (Bemis and Seegal, 2004, Richardson and Miller, 2004, Caudle et al., 2006, Fonnum et al., 2006). Interestingly, a recent SPECT imaging study of capacitor workers who had high occupational exposures to PCBs revealed a reduction in DAT density in the striatum of patients diagnosed with PD compared with control, similar to the results demonstrated by Caudle et al., (2006) in mice.

The implication of alterations to dopamine handling via disruption of DAT and VMAT2 is still under investigation. It is important to point out that deficits in the ability of VMAT2 to sequester cytosolic dopamine can result in an increase in oxidative stress in the dopamine neuron and subsequent loss of dopamine in the striatum, degeneration of dopamine neurons in the SNpc as well as behavioral alterations reminiscent of those found in PD (Caudle et al., 2007, Taylor et al., 2009). It has been postulated that this dopaminergic degeneration is a result of the accumulation of dopamine in the cytosol and its subsequent oxidation to neurotoxic species and the formation of reactive oxygen and nitrogen species (Caudle et al., 2008).

Polybrominated Diphenyl Ethers

Polybrominated diphenyl ethers (PBDE) are a class of brominated compounds chemically similar to PCBs, composed of varying degrees and positions of bromine substitutions to give 209 different structures or congeners. As PCB manufacture and use were being phased-out in the late 1970's due to their public health concerns, PBDEs were quickly introduced to replace them. Similar to PCBs, PBDEs are additive flame retardants primarily used in the manufacturing and insulation of electronic equipment. In addition, they are also found in polyurethane foam used in carpeting, furniture cushions and home insulation (Darnerud et al., 2001, de Wit, 2002). The major commercial PBDE products produced were marketed as one of three mixtures, known as pentabrominated BDE, octabrominated BDE, and decabrominated BDE. Although the manufacture of penta and octa BDEs has been discontinued since 2004, Deca BDEs are the most extensively produced and widely used PBDEs. The additive binding of these compounds, rather than the chemical incorporation into the products allows PBDEs to migrate out of the plastic and enter the environment. Furthermore, like PCBs their lipophilicity, resistance to degradation, and the ability to biomagnify make PBDEs extremely persistent in the environment (Norstrom et al., 2002). However, in contrast to PCBs and other organochlorines, whose environmental levels have been decreasing, levels of PBDEs have significantly increased, raising concern over the safety of PBDEs and their potential adverse health effects. Like PCBs, the predominant route of human exposure to PBDEs is through the ingestion of contaminated food as well as occupational exposure (Alaee et al., 2001, Luross et al., 2002, Norstrom et al., 2002). Exposure to PBDEs, whether through diet or occupational exposure has resulted in a considerable rise in PBDE body burden in the United States population, especially in serum and breast milk (Sjodin et al., 2004).

While data concerning the health effects of PBDEs on the human population is lacking, several studies have characterized the toxicological outcomes of exposure to PBDEs in *in vitro* and *in vivo* systems. Interestingly, the neuronal components disrupted are strikingly similar to those altered following exposure to PCBs. Indeed, several *in vitro* studies have demonstrated the ability of PBDEs and other brominated flame retardant compounds to elicit an increase in oxidative stress (Madia et al., 2004, Reistad et al., 2005, Reistad et al., 2006, Reistad et al., 2007), disruption of the calcium signaling pathway, (Kodavanti and Derr-Yellin, 2002, Kodavanti and Ward, 2005, Reistad et al., 2005), as well as inhibition of DAT and VMAT2 function at low micromolar concentrations. Furthermore, as seen with PCBs, these effects appear to be specific for the dopamine system, as PBDEs demonstrated low inhibitory potential for other neurotransmitter systems. As with PCBs, this specificity for the dopamine system and VMAT2, in particular raises concerns about the neurotoxic potential of these compounds to the nigrostriatal dopamine system, as mentioned above. Further studies will need to be conducted in order to parcel out the cumulative effects of PBDEs on the dopamine system and their possible role in PD.

Metals

Metals have been utilized in biological processes since the beginning of cellular life on Earth as humans require a plethora of metals for enzymes that make use of iron, copper, magnesium, manganese, zinc, selenium, cobalt and molybdenum. The utility of these metals in biology stem from their catalytic properties as they vastly increase the rates of enzyme reactions due to their facility in transferring electrons. In addition to their biological use, metals have been used for over 7,000 years in commerce and industry. As a result, metals are recognized as some of the earliest toxicants in history, with reports of metal toxicity by Hippocrates around 370 BC. Since then, the use and potential for human exposure to metals has evolved as metals are being used in new and diverse ways, including as reactive metal

surfaces of nanoparticles. Thus, it is imperative to understand the toxicological processes involved in metal toxicity and the implications of their exposure to the human population.

While the toxic effects of metal exposure on the human body has been well documented, it has only been in the last 20–30 years that the neurotoxicological outcomes of metal exposure have been more appreciated. Indeed, the brain appears to be exquisitely vulnerable to metal toxicity as metals target various aspects of the neuron, including the generation of oxidative stress and disruption of neurotransmission. These effects are especially detrimental when they occur in the basal ganglia. As with other environmental factors, metal exposure has been suggested to be a risk factor for the development of Parkinson disease and other parkinsonian-related movement disorders. This section will present recent findings concerning the potential role and mechanisms of action of metal toxicity in Parkinson disease and parkinsonism.

Iron

Iron (Fe) is the most abundant element on earth and an essential metal for life as it is used extensively by proteins involved in the electron transport chain and oxygen transport. Normal exposure to iron occurs through our diet, especially meat, poultry, and fish, as well as iron supplementation. Further iron exposure can occur through occupational exposures, predominantly from metal fumes or metal dust, as would be generated during welding and in iron and steel production. Iron that has entered the body is absorbed across cell membranes by the divalent metal transporter 1 (DMT1) before it is bound by proteins, such as transferrin, which transports iron into tissues, and ferritin (H- and L-ferritin), which serves as a storage depot for free iron. These mechanisms are important for the cell as they function to regulate and maintain iron homeostasis, as an increase in free iron is highly susceptible to free radical generation via the Fenton reaction, which catalyzes the conversion of hydrogen peroxide to the highly reactive hydroxyl radical. These hydroxyl radicals readily react with multiple components of the cell, including DNA, membrane lipids, and protein, leading to their dysfunction. Within the dopamine neurons in the substantia nigra pars compacta, a readily available supply of hydrogen peroxide is present as this is a normal byproduct of the production and metabolism of dopamine within the neuron, suggesting that a decrement in iron homeostasis could facilitate the further production of reactive species that could subsequently damage various aspects of the neuron.

The oxidative stress caused by free iron can directly lead to neuronal damage and neurotoxicity, as evidenced in two genetic forms of iron mishandling. Friedrich's ataxia in which the iron-regulating gene, frataxin is silenced, results in cytosolic iron accumulation and motor deficits. In addition, neurodegeneration with brain iron accumulation 1 (NBIA1), also called pantothenate kinase-associated neurodegeneration (PKAN), leads to iron buildup in the basal ganglia and subsequent death of neurons. This neuronal death leads to parkinsonian motor symptoms as well as dementia and dystonia. In addition to these two disorders, several epidemiological studies have determined a significant positive correlation between plasma levels of iron and the incidence of PD. Related to these findings, an increased accumulation of iron has been found in the SNpc of PD patients compared with controls (Sofic et al., 1988). While the cause for this accumulation is not clear, studies have demonstrated an increase in the expression of the lactoferrin receptors and the DMT1, on dopaminergic neurons in the SNpc of patients with PD, which could facilitate the transport of iron into the dopamine neurons. In a similar study, Friedman et al (2011) found a reduction in the amount of L-ferritin, which functions to sequester and store free cytosolic iron, in PD patients. Taken in concert, an increase in iron transport mechanisms as well as a reduction in the ability of the dopamine neurons to store free iron could provide an advantageous environment for the generation of oxidative stress.

The deleterious effects of iron mishandling have also been demonstrated in animal models of PD. Exposure of neonatal mice to levels of iron routinely found in baby formula resulted in an age-dependent reduction in TH+ neurons in the SNpc as well as a loss of dopamine in the striatum of aged animals (Kaur et al., 2007). Conversely, treatment of mice with an iron chelator attenuated the parkinsonian pathology induced by MPTP. Similarly, overexpression of H-ferritin in nigral neurons rendered them resistant to the dopaminergic effects of MPTP (Kaur et al., 2003). Finally, unilateral injection of FeCl₃ into the SNpc of adult rats resulted in a significant reduction in striatal dopamine as well as other dopaminergic markers and a concomitant deficit in dopamine-related behaviors (Sengstock et al., 1994, Junxia et al., 2003). While a significant amount of attention has been paid to the potential role of iron-mediated generation of oxidative stress as a causative factor in damage to the nigrostriatal dopamine system, several studies have focused on the ability of iron, as well as other metals to facilitate the fibrillization and aggregation of the PD-related protein, alpha-synuclein (Uversky et al., 2001). It is this aggregated and fibrillar conformation that is considered to be the predominant toxic form of alpha-synuclein and involved in the degenerative process of dopaminergic neurons. While iron in and of itself is believed to contribute to the change in conformation of alpha-synuclein, additional studies have demonstrated the ability of oxidative species to also participate in the formation of neurotoxic alpha-synuclein species (Giasson et al., 2000).

Copper

Copper (Cu) is another essential metal that has numerous biological roles in the body, including formation of red blood cells, iron transport, mitochondrial respiration, as well as being an integral part of key enzymes, such as superoxide dismutase and dopamine β-hydroxylase (Harris, 2000). However, like iron, copper is highly reactive and can undergo the Fenton reaction to generate hydroxyl radicals from hydrogen peroxide, which can have detrimental effects on many facets of normal neuronal function. Thus, a tight homeostatic control of copper needs to be in place. Exposure to copper primarily occurs via ingestion of copper-containing food and beverages in addition to occupational exposure through welding fumes and metal smelting and mining. Following exposure, copper is absorbed from the gut by the copper membrane transporter 1 (CMT1) and is predominantly found bound to ceruloplasmin in the blood before it is transported into cells (Hellman and Gitlin, 2002).

The role of altered copper homeostasis in neurological disorders is well documented, especially as it relates to Wilson's disease. Wilson's disease is a genetic disorder that results in the accumulation of copper in tissues and damages specific regions of the brain, most notably the basal ganglia nuclei and substantia nigra. This accumulation appears to cause a form of parkinsonism characterized by movement deficits that are moderately responsive to L-DOPA, as well as pathological changes in pre- and postsynaptic dopamine markers in the striatum (Barbeau and Friesen, 1970, Hitoshi et al., 1991, Oder et al., 1993, Barthel et al., 2003). In addition to Wilson's disease, there is some evidence that provides support for a role of copper exposure and PD pathology. Previous epidemiological studies by Gorell et al., (Gorell et al., 1997, 1999a) have suggested an association between copper exposure and PD, showing an almost 2.5-fold increase in risk for PD following a 20 year occupational exposure to copper. More recently, further evidence for copper exposure as a risk factor for PD has been demonstrated following intranigral injection of copper into rats. Investigators noted a significant reduction in several indices of dopaminergic neurodegeneration, including reduced striatal dopamine and loss of tyrosine hydroxylase projections and neurons in the striatum and substantia nigra, respectively (Yu et al., 2008).

The mechanisms of action that underlie copper-induced neuropathology are still unclear. However, given the propensity for the accumulation of free copper to generate highly

neurotoxic reactive species, it can be speculated that oxidative stress plays a substantial part in mediating the neurodegenerative process. As mentioned above, the effects of oxidative stress on neuronal function are promiscuous, causing damage to multiple aspects of the cell. A major target of oxidative damage appears to be the mitochondria (Rossi et al., 2004). Indeed, recent reports have suggested that mitochondria are preferentially damaged following exposure to copper (Paris et al., 2009). Alteration to mitochondria function has long been appreciated as a potential pathogenic cascade in PD, most notably, through the aberrant generation of oxidative species following damage (Schapira, 2008). Additionally, like iron, copper has been shown to accelerate the oligomerization of alpha-synuclein monomers into neurotoxic fibrils, which can impair mitochondrial function (Uversky et al., 2001). Whether or not this interaction underlies dopaminergic loss is still unclear.

Manganese

Metals, especially transition metals, play an essential role in the normal functioning of numerous biological processes, particularly as cofactors necessary for enzyme function. For instance, manganese is an essential cofactor for several enzymes, including superoxide dismutase (SOD), and plays a role in the synthesis and metabolism of neurotransmitters (Schroeder et al., 1966, Hurley et al., 1984, Golub et al., 2005). In contrast, as with other metals, manganese can have detrimental effects on this system through its accumulation and generation of reactive species among other mechanisms. Although manganese exposure can occur through several different forms, including ingestion of food, exposure to manganese-containing fuel additives, the predominant route of exposure is via occupational inhalation of manganese fumes in an industrial setting, most notably from volatilization of manganese containing substrates used during welding (Huang et al., 1989, Hudnell, 1999). Manganese is also present in flux agents, such as electrodes as well as being added to consumables containing between 2–15% manganese (Villaume, 1979, Burgess, 1995). Inhalation of particulate manganese is able to bypass the blood-brain barrier (BBB) where it is taken up directly by presynaptic nerve endings in the olfactory bulb. However, these results have mostly been obtained through a rodent model of exposure, thus the relevance of this pathway is unclear in humans, given the interspecies difference in the olfactory system and brain between rodents and humans (Brenneman et al., 2000, Aschner et al., 2005). After being taken up, manganese is retrogradely transported to the cell body where it can be released into the interstitial space (Tjalve and Henriksson, 1999, Vitarella et al., 2000, Fechter et al., 2002, Normandin et al., 2004). Exposure to high levels of airborne manganese has been associated with several neurologic symptoms, including reduced neurobehavioral performance, neuropsychological impairment, and disruption of motor function (Huang et al., 1993, Gibbs et al., 1999). However, the most significant consequence of manganese toxicity appears to be impaired motor function, which culminates in a disorder referred to as manganism. Thus, welding and manganese have received considerable attention due to the presence of motor symptoms that are shared with many clinical features of PD.

The mechanisms by which manganese mediates manganism are not wholly understood. Manganese, when in excess, can inhibit mitochondrial function, reduce glutathione levels, increase NMDA-mediated neurotoxicity and alter calcium homeostasis, all which culminate in cellular dysfunction (Maynard and Cotzias, 1955, Brouillet et al., 1993, Gavin et al., 1999). Furthermore, manganese is a potent mediator of pro-oxidant activities, through its production of reactive species, particularly superoxides, peroxide, and hydroxyl radical (Graham et al., 1978, Cohen, 1984). In addition, it can catalyze the oxidation of dopamine to reactive quinones (Graham, 1978, Halliwell and Gutteridge, 1984). Significant evidence has been collected demonstrating the toxicity of these oxidative species in dopaminergic neurodegeneration (Jenner, 2003). The controversy related to the contribution of manganese to PD revolves around two main issues.

First, epidemiologically (reviewed by Santamaria et al., 2007 (Santamaria et al., 2007)), while several reports have identified occupational exposure to manganese to significantly increase the risk of PD (Gorell et al., 1999a, Gorell et al., 1999b), others have failed to establish such an association (Semchuk et al., 1993, Seidler et al., 1996, Marsh and Gula, 2006). The fundamental problems appear to relate to the fact that some of the studies have relied heavily upon self-reported neurological symptoms, lack of exposure data, small sample size, and inadequate control groups. These factors make it difficult to reach a substantial conclusion as to the influence of welding and/or manganese on the development of neurological deficits.

The clinical and pathological evidence also raises doubts on the contribution of manganese to the risk of PD. The pathology observed with PD is largely attributable to the progressive degeneration of dopamine-containing neurons in the SNpc and loss of the dopaminergic projections to the striatum (Ehringer and Hornykiewicz, 1960, Albin et al., 1989, Crossman, 1989, DeLong, 1990, Fahn, 2003). Another important pathological feature of PD is the presence of eosinophilic, cytoplasmic inclusions of fibrillar, misfolded proteins, including α -synuclein, termed Lewy bodies and Lewy neurites, located in the perikarya and cellular processes, respectively, of remaining neurons in the SNpc and other brain regions (Braak et al., 1995, Forno, 1996). In contrast, symptoms related to manganism are attributed primarily to damage to neurons in the globus pallidus (internal and external segments), while largely sparing the caudate, putamen, and SNpc. Furthermore, the lack of Lewy body inclusions differentiates the pathology of manganism from PD, in addition to magnetic resonance imaging (MRI) detection of manganese accumulation in the pallidum following exposure, compared to a normal MRI in PD. However, while most report the preservation of dopaminergic regions involved in PD pathology (Shinotoh et al., 1995, Olanow et al., 1996, Pal et al., 1999, Olanow, 2004), others suggest a mild damage to these regions as well as a general alteration in their function in manganism (Suzuki et al., 1975, Eriksson et al., 1992, Kim et al., 2002, Wright et al., 2004, Chen et al., 2006, Guilarte et al., 2006).

Similarly, laboratory studies using nonhuman primates largely confirm the almost exclusive role of the globus pallidus in Mn pathology, while leaving the dopaminergic inputs relatively untouched. In general, a lack of dopaminergic pathology, as evidenced by no change in DAT expression, striatal dopamine levels, some reductions in dopamine receptor expression, and nonresponse to L-DOPA treatment following an exposure paradigm that closely mimics that seen in humans. Taken together, while manganese exposure is widely studied in relation to PD, many believe that manganism is a separate entity as it predominantly involves the globus pallidus rather than the SNpc, and the seemingly overlapping clinical signs may just be a result of the involvement of basal ganglia dysfunction and damage to common output pathways in both disorders (Dobson et al., 2004).

Lead

Lead is a non-essential metal that can replace other metals in biological processes to the detriment of an organism. The predominant route of exposure within the human population is through ingestion of contaminated food products. However, higher than normal exposures to lead can also occur via lead-based paint, lead in dust, lead contaminated drinking water, and lead in the air from combustion of leaded fuels (Manton et al., 2005). Although the environmental levels of lead have been drastically reduced over the last several decades, lead exposure still represents a significant health concern, especially for infants and young children who are still progressing through critical periods of neurodevelopment.

Exposure to lead can manifest in several different neurological deficits, depending on amount of exposure as well as age of the subject, with children demonstrating a substantially more drastic neurotoxicological response compared with adults (Goyer, 1996). In general, neuropathological symptoms include severe edema and loss of neurons in the central nervous system as well as demyelination and axonal degeneration in the periphery, which manifests as a motor-based peripheral neuropathy and ataxia. In addition to these motor abnormalities, recent epidemiological studies have indicated a potential association between exposure to lead and an increased risk of PD, suggesting a 2–3-fold increase in risk for PD following lead exposure (Coon et al., 2006, Weisskopf et al., 2010). Animal studies have also demonstrated a significant alteration in the function of the nigrostriatal dopamine system following exposure to lead, including a reduction in the number of dopamine neurons in the SNpc as well as a reduction in the firing rate of dopaminergic neurons, which is independent of neuronal loss (Scortegagna and Hanbauer, 1997, Tavakoli-Nezhad et al., 2001).

The precise mechanisms of action leading to these effects remain to be determined and could arise from several different avenues, as lead-mediated neuronal dysfunction is somewhat promiscuous, affecting multiple aspects of neuronal function (Gmerek et al., 1981, Lasley and Gilbert, 1996, Ferguson et al., 2000). For example, lead can disrupt calcium handling of the neuron by inhibiting uptake of calcium through calcium channels in the presynaptic terminal as well as the mitochondria. In addition, lead can dissipate the proton gradient across synaptic vesicles, possibly leading to decreased DA sequestration into vesicles (Borisova et al., 2011). Indeed, recent work from our lab has clearly demonstrated the detrimental effects mishandling of cytosolic dopamine by synaptic vesicles can have on the nigrostriatal dopamine system, including reduction in dopamine neurons in the SNpc and loss of dopamine in the striatum, as well as formation of neurotoxic reactive species (Caudle et al., 2007).

Mercury

Mercury exists in several different forms or species. These forms include inorganic, such as mercury chloride, elemental mercury, or organic mercury, such as methylmercury. Of these species, elemental mercury and methylmercury are the most toxicologically important as they elicit the most detrimental effects following human exposure. Exposure to these compounds usually occurs in an occupational setting, as is the case with inhalation of vaporized mercury, while exposure to methylmercury predominantly occurs through ingestion of contaminated fish. Elemental mercury can be vaporized leading to an increased incidence of inhalational exposure to mercury, especially in an occupational setting. Mercury has a very high affinity for sulfur and will readily bind to thiol containing molecules, such as cysteine. The binding of methylmercury to cysteine allows it to be transported across the blood brain barrier by the neutral amino acid transporter (Yin et al., 2008).

The neurological effects of mercury exposure have been well documented. Identification of mercury exposure mediating neurological alterations was in the 1800's in the milliner's trade where hat makers routinely inhaled vaporized mercuric nitrate, which was used to cure the felt of hats. Those who had become intoxicated exhibited signs of movement disorders, including tremors and polyneuropathy. Widespread intoxication by methylmercury has also occurred as with the mass contamination of Minimata Bay, Japan for several decades in the mid 1900's. Consumption of contaminated fish, especially by women who were pregnant resulted in severe neurodevelopmental deficits, including incapacitating alterations in movement and cognition. Adult exposure was similarly debilitating, causing ataxia, tremor, and sensory deficits, most likely the result of cerebral edema and atrophy. While these data

provide a very clear indication that mercury intoxication causes significant motor symptoms, to date, an association between mercury exposure and PD has not been established (Gorell et al., 1999a). However, a significant amount of data suggests that exposure to mercury can have substantial impact on the normal functioning of the dopamine nigrostriatal dopamine system. Most notably, Lin et al., (2011) recently reported a significant reduction in DAT function in the striatum of workers occupationally exposed to mercury vapor. This dysfunction in DAT was similarly demonstrated *in vitro* with a reduction in dopamine uptake in synaptosomes treated with mercury (Hare et al., 1990, Dreiem et al., 2009). In addition, mercury exposure reduced neurites of dopaminergic neurons isolated from the ventral mesencephalon and caused a reduction of striatal dopamine in mice exposed to methylmercury (Gotz et al., 2002, Bourdineaud et al., 2011). The mechanisms of action related to these deficits are not clearly defined, however, methylmercury has been shown to be involved in the generation of oxidative stress through various routes, including acceleration of lipid peroxidation, mitochondrial damage, and stimulate superoxide production (Yee and Choi, 1996).

Nanoparticles

Nanoparticles can be grouped into two distinct, yet similar categories, based upon their origin. Combustible nanoparticles are those, which arise from environmental sources, such as diesel exhaust or welding fumes. On the hand, engineered or manufactured nanoparticles refer to compounds that are synthesized such as titanium oxide, zinc oxide, or carbon nanotubes, among others (Oberdorster et al., 2005). Our discussion will be focused on engineered nanoparticles as these are seeing an emerging utility in various biological and biomedical avenues, such as cosmetics, adjuncts for medical imaging, vehicles for drug delivery, and biosensors. More specifically, they are currently being used as therapeutic avenues in several neurodegenerative disease, including PD (Huang et al., 2009, Modi et al., 2009). While these capabilities have provided new and exciting opportunities for their use their physiochemical properties have increased concern for the potential health effects that increased exposure to nanoparticles may have. Nanoparticles vary in size from 1–100 nm and can be covered with metallic coatings, ranging from titanium (Ti), aluminum (Al), iron (Fe), manganese (Mn), copper, (Cu), and gold (Au), among others (Win-Shwe and Fujimaki, 2011). Thus, given our previous discussion of the neurotoxic effects of metals, especially in terms of oxidative stress, nanoparticles represent a new risk category for metals exposure. Furthermore, their small size introduces new problems with how the body handles absorption, distribution and excretion of these materials. Indeed, nanoparticles in the blood can gain access to the brain through the blood brain barrier. And once inside the brain they can interact with neurons and glia, affecting their function and expression.

Unfortunately, manufactured nanoparticle technology is too new to have identified cohorts of exposed workers with health problems and specific brain disorders, thus there is no definitive proof that exposure to engineered nanoparticles increases the risk of neurodegenerative diseases such as PD (Oberdorster 2009). However, recent laboratory studies have suggested that exposure to engineered nanoparticles can have deleterious effects on the dopamine system. Several *in vitro* studies using PC12 cells have demonstrated reductions in dopamine, increases in oxidative stress, as well as alterations to dopamine- and PD associated genes, including tyrosine hydroxylase, alpha-synuclein, and parkin (Hussain et al., 2006, Wang et al., 2009). Similarly, *in vivo* exposures have also demonstrated reductions in striatal dopamine, in addition to increased oxidative stress and neuroinflammation following exposure to metal nanoparticles (Hu et al., 2010, Wu et al., 2011). While the precise cause for these changes is unclear, increase in oxidative stress may be a major contributor.

Solvents

Solvents represent a broad range of chemicals with the common utility of dissolving one substance into another. The most common solvents in use today include trichloroethylene (TCE), toluene, acetone, hexane, carbon disulfide, which all serve multiple purposes in industrial and home uses. The human population is exposed to solvents through numerous different routes. The most prevalent route is occupational exposure via inhalation or dermal exposure. However, non-occupational exposure can also occur through ingestion of contaminated water as well as inhalation exposure, both following improper waste disposal or accidental release of solvents. In general, solvents are lipophilic, allowing them to be amenable to quick and easy absorption into the body and to target organs following exposure. Over the years, several neurological deficits have been associated with solvent exposure. In general, these could be classified as exhibiting movement disorders, such as tremor as well as other motor deficits. However, it is unclear whether these symptoms represent PD or a parkinsonian-related disorder. Although many epidemiological studies have attempted to elucidate an association between solvent exposure and PD, a solid link has yet to be determined.

Trichloroethylene

Trichloroethylene is a chlorinated hydrocarbon that has proved to be very versatile in its uses and applications, including its early use as an anesthetic giving its properties as a central nervous system depressant. However, TCE has a broader use as a solvent in many different settings, including the rubber industry, adhesive formulations, dyeing and finishing operations, printing inks, paints, lacquers, varnishes, adhesives, and paint strippers. It is applied prior to plating, anodizing, and painting. While TCE use has been greatly reduced in most of the aforementioned applications, it is still commonly used in the degreasing of metals and as an intermediate for hydrofluorocarbon production. In addition to a high occupational exposure the human population are also routinely exposed to TCE via contact or consumption of contaminated water, food or contact with consumer goods containing TCE.

The overall toxicity of TCE appears to result from its ability to disrupt cellular membrane integrity, formation of oxidative free radicals resulting in lipid peroxidation, as well as disruption of calcium transport. This membrane disruption leads to demyelination with TCE exposure having been shown to result in loss of sensory nerve function (Huber, 1969, Feldman et al., 1970, Feldman et al., 1985). The main metabolites of TCE are trichloroethanol, trichloroethanol-glucuronide and trichloroacetic acid (TCA), the former excreted rapidly in the urine and latter appearing slower, suggesting TCA to be a major source of TCE toxicity (Cole et al., 1975, Yoshida et al., 1996). With growing reports of TCE contamination of drinking water, understanding the possible long-term effects of TCE exposure has become more important. There has been increasing interest into the link between chronic exposure to TCE and Parkinson's disease and a few case reports and a population based study have provided more evidence to suggest that a potential association between long term exposure to TCE and PD pathology (Bringmann et al., 1992, Guehl et al., 1999, Kochen et al., 2003, Gash et al., 2008, Goldman, 2010). Most recently a study of twins discordant for PD identified exposure to TCE as having a 6-fold increased risk for PD (Goldman et al., 2011). These results are largely supported by the experimental literature through the use of animal models of TCE exposure. Two reports in the last few years have demonstrated significant damage to the nigrostriatal dopamine system in rats following TCE administration. Following up on case study data, Gash et al., (2008) found that TCE exposure resulted in a loss of dopaminergic neurons in the SNpc, as well as a concomitant reduction in dopamine in the striatum and midbrain. While the exact mechanisms of action of TCE on these neurons is not known, they were able to show that TCE inhibits

mitochondrial complex I, as seen with other parkinsonian mimetics, such as MPTP and rotenone. Another study demonstrated similar findings of loss of dopamine neurons in the SNpc, again in rats exposed to TCE, in addition to other pathological hallmarks of PD, such as accumulation of alpha-synuclein and an increase in reactive oxygen species (Liu et al., 2010). They were also able to show that the effects of TCE were preferential to the dopamine neurons in the SNpc as no loss of GABAergic or cholinergic neurons was observed in this area and the dopaminergic neurons of the ventral tegmental area were similarly spared, as seen in PD.

Methanol

The primary use of methanol is as a fuel additive and as a precursor in the production of plastics, formaldehyde, acetic acid and explosives. Methanol is also an additive in paint strippers, aerosol spray paints and car windshield washer products. Methanol exposure can occur via inhalation of methanol vapors, dermal exposure to aqueous solutions containing methanol, however, the most common route of methanol intoxication involves the deliberate or accidental ingestion. Like n-hexane, the most toxic compound is not methanol, rather metabolic intermediates of methanol, specifically formic acid, give rise to the toxic nature of methanol exposure. Methanol is oxidized by alcohol dehydrogenase in the liver to yield formaldehyde which is then rapidly oxidized into formic acid (Teng et al., 2001). Symptoms of methanol poisoning resemble that of ethanol intoxication, such as lethargy, confusion, headache, nausea, ataxia and vision impairment. Following the initial symptoms, neurological deficits can occur that can be classified as parkinsonian or dystonic with reports of lesions to the basal ganglia, putamen and other white matter regions (McLean et al., 1980, Ley and Gali, 1983, Verslegers et al., 1988, Carcaba et al., 2002, Finkelstein and Vardi, 2002, Reddy et al., 2007). In these cases, the appearance of motor deficits is slow to present and may occur weeks after initial exposure. The rigidity and bradykinesia may be partially relieved with levodopa treatment but the white matter lesions may extend beyond the basal ganglia (Reddy et al., 2010).

Conclusion

Toxic exposure to non-pesticide compounds, such as organohalogenes, metals, and solvents have received strong support as risk factors for PD and other parkinsonian disorders. Unfortunately, making a clear delineation as to the contribution of a class of compounds or a specific compound to a particular suite of pathological and clinical symptoms remains to be achieved. While several compounds and classes of compounds may appear to share similar features in their neurotoxicity we simply do not have enough relevant data to make that assessment. The reasons for this are varied and could be due to lack of a large enough cohort in the human population or a clinical heterogeneity within the cohort, a lack of a systematic molecular and behavioral appraisal of damage to the nigrostriatal and other relevant neural circuits following exposure. It also must be considered that these compounds, in and of themselves, do not explicitly cause PD or parkinsonism. Rather, they serve to facilitate the expression of the disorder through their interaction with other underlying genetic or environmental factors. And without these factors in place, the neurotoxicity of these exposures is less pronounced, making it more difficult to identify a particular compound as the singular causative factor in the disorder.

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Introduction, use, and persistence of industrial toxicants in the U.S.

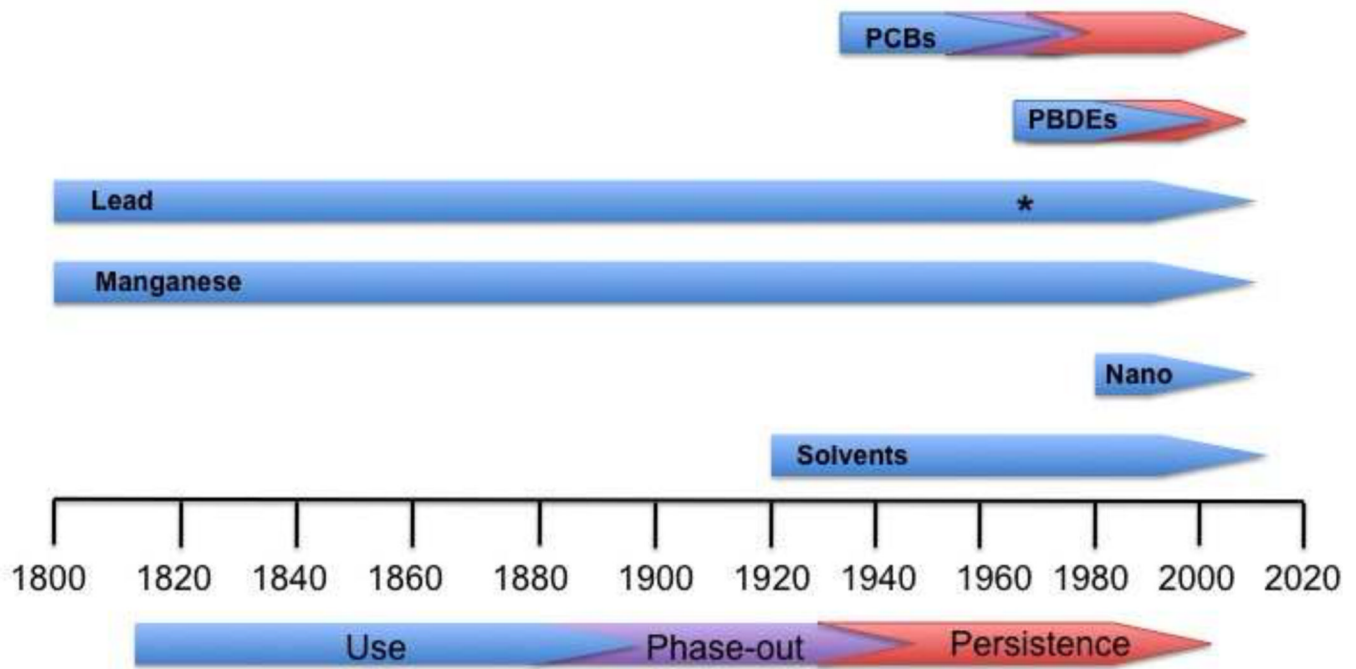


Figure 1.

Introduction and use of various industrial toxicants in the U.S. The commercial introduction and use of several widely used industrial contaminants are shown for the last two centuries, beginning around the Industrial Revolution. While some of these compounds, such as PCBs have had their use phased out, others such as many solvents have only had their use restricted, yet are still in high use in the U.S. In contrast, most metals are still widely used and have seen little or no restriction placed on their use. An exception to this is lead, which saw a removal from use as a gasoline additive in the 1970's (denoted by an asterisk), resulting in a drastic reduction in the blood lead levels of the human population.

Table 1

Association between exposure to industrial toxicants and Parkinson's disease and potential mechanisms of pathology.

Chemical	Strength of Association		Mechanisms of action
	Human	Animal	
Organohalogenes			
PCBs	++	++	Oxidative stress, dopamine homeostasis, calcium homeostasis
PBDEs	0	+	Oxidative stress, dopamine homeostasis, calcium homeostasis
Metals			
Iron	+	++	Oxidative stress, alpha-synuclein fibrillization
Copper	+	+	Oxidative stress, alpha-synuclein fibrillization, mitochondrial dysfunction
Manganese	+	+	Oxidative stress, mitochondrial dysfunction, calcium homeostasis
Lead	+	+	Dopamine homeostasis, calcium homeostasis
Mercury	0	+	Oxidative stress, dopamine homeostasis, mitochondrial dysfunction
Nanoparticles	0	+	Oxidative stress, neuroinflammation
Solvent			
TCE	++	++	Oxidative stress, mitochondria dysfunction, calcium homeostasis, alpha-synuclein accumulation

++ strong association, i.e. multiple reports.

+ some association, i.e. individual reports, small sample size. 0 = no association or insufficient evidence.

There is evidence for all of these compounds to cause some level of parkinsonism in either humans or animal models. Additionally, the major pathogenic mechanisms of action, as suggested by the literature, have been provided.