

# Geraniol and Geranial Dehydrogenases Induced in Anaerobic Monoterpene Degradation by *Castellaniella defragrans*

Frauke Lüddecke,<sup>a</sup> Annika Wülfing,<sup>a</sup> Markus Timke,<sup>a</sup> Frauke Germer,<sup>a</sup> Johanna Weber,<sup>a</sup> Aytac Dikfidan,<sup>a</sup> Tobias Rahnfeld,<sup>a</sup> Dietmar Linder,<sup>c</sup> Anke Meyerdierks,<sup>b</sup> and Jens Harder<sup>a</sup>

Department of Microbiology<sup>a</sup> and Department of Molecular Ecology,<sup>b</sup> Max Planck Institute for Marine Microbiology, Bremen, Germany, and Biochemisches Institut, Fachbereich Medizin, Justus-Liebig-Universität, Giessen, Germany<sup>c</sup>

*Castellaniella defragrans* is a *Betaproteobacterium* capable of coupling the oxidation of monoterpenes with denitrification. Geraniol dehydrogenase (GeDH) activity was induced during growth with limonene in comparison to growth with acetate. The N-terminal sequence of the purified enzyme directed the cloning of the corresponding open reading frame (ORF), the first bacterial gene for a GeDH (*geoA*, for geraniol oxidation pathway). The *C. defragrans* geraniol dehydrogenase is a homodimeric enzyme that affiliates with the zinc-containing benzyl alcohol dehydrogenases in the superfamily of medium-chain-length dehydrogenases/reductases (MDR). The purified enzyme most efficiently catalyzes the oxidation of perillyl alcohol ( $k_{\text{cat}}/K_m = 2.02 \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$ ), followed by geraniol ( $k_{\text{cat}}/K_m = 1.57 \times 10^6 \text{ M}^{-1} \text{ s}^{-1}$ ). Apparent  $K_m$  values of  $<10 \mu\text{M}$  are consistent with an *in vivo* toxicity of geraniol above  $5 \mu\text{M}$ . In the genetic vicinity of *geoA* is a putative aldehyde dehydrogenase that was named *geoB* and identified as a highly abundant protein during growth with phellandrene. Extracts of *Escherichia coli* expressing *geoB* demonstrated *in vitro* a geranial dehydrogenase (GaDH) activity. GaDH activity was independent of coenzyme A. The irreversible formation of geranic acid allows for a metabolic flux from  $\beta$ -myrcene via linalool, geraniol, and geranial to geranic acid.

A basic reaction in biochemistry is the oxidation of an alcohol to an aldehyde by an oxidoreductase. Alcohol dehydrogenases (ADHs; EC 1.1.1.x) are grouped into long-chain, medium-chain, and short-chain ADHs, according to their sequence length. Members of the medium-chain dehydrogenase/reductase superfamily (41, 45) are characterized by a Rossmann dinucleotide-binding domain (5) and two zinc ions as the structurally and catalytically acting transition metal in the active site (2). The medium-chain alcohol dehydrogenases evolved from a common ancestor into several families, and there is good evidence for different evolutionary rates within the families (45). The catalytic process is well understood (42). In bioconversion reactions, an ADH is usually followed by an aldehyde dehydrogenase (ALDH; EC 1.2.1.x) catalyzing the oxidation of the aldehyde to the corresponding carboxylic acid. The ALDH superfamily is ubiquitous in nature, oxidizing a wide range of aliphatic and aromatic aldehydes (34, 39, 56).

A number of oxidoreductases acting on alcohols with an adjacent carbon-carbon double bond, i.e., the allyl alcohol, have been isolated from different sources and characterized: allyl ADH (EC 1.1.1.54), retinol DH (EC 1.1.1.105), geraniol DH (GeDH; EC 1.1.1.183), coniferyl ADH (EC 1.1.1.194), cinnamyl ADH (EC 1.1.1.195), and farnesol DH (EC 1.1.1.216). The allyl alcohol motif is also present in benzyl alcohol, and therefore many benzyl ADHs, known as aryl ADHs (EC 1.1.1.90), can act on allyl alcohols.

In a *Pseudomonas putida* isolate, a 3-methyl-2-buten-1-ol dehydrogenase was found to be a benzyl ADH with broad specificity toward allyl and benzyl alcohols (36). Geraniol (3,7-dimethyl-*trans*-2,6-octadien-1-ol) is a  $\text{C}_{10}$  homologue of 3-methyl-2-buten-1-ol and is known to be an intermediate in the anaerobic degradation of  $\beta$ -myrcene by *Castellaniella defragrans* (7). This betaproteobacterium, originally named *Alcaligenes defragrans* (29), was isolated with various monoterpenes, natural unsaturated hydrocarbons ( $\text{C}_{10}\text{H}_{16}$ ) that can be simply differentiated by

their acyclic, monocyclic, or bicyclic structure (Fig. 1) (17). Plants synthesize monoterpenes for thermotolerance or other plant-environment interactions in amounts of over  $100 \text{ Tg C year}^{-1}$ ; thus, they represent an important component in the carbon cycle on earth (33, 51, 54). In insects, monoterpenes are synthesized as pheromones (4, 37, 50). Furthermore, these substances are widely used in the food, flavor, and fragrance industries due to their odorous properties (8). Geraniol exudes a sweet, rose-like scent and is commercially synthesized in amounts of  $4,000 \text{ Mg year}^{-1}$  (3).

In *C. defragrans*, the linalool dehydratase-isomerase catalyzes the hydration of the acyclic  $\beta$ -myrcene to (S)-(+)-linalool as well as the isomerization to geraniol (7, 35). The formation of geranic acid was observed *in vivo* and *in vitro* (24), indicating the presence of dehydrogenases catalyzing the oxidation of the allyl alcohol geraniol to geranic acid, most probably via geranial. So far, the only geraniol dehydrogenases (GeDHs) characterized on the molecular level come from sweet basil, *Ocimum basilicum* (25), and the astigmatid mite *Carpoglyphus lactis* (37). An aldehyde dehydrogenase acting on geranial has never been reported. Therefore, we identified and characterized two relevant enzymes: an alcohol dehydrogenase with a remarkably high geraniol affinity and a geranial dehydrogenase, which specifically oxidizes geranial (Fig. 2). Both enzymes were induced in monoterpene-grown cells, and together with the linalool dehydratase-isomerase, they provide the

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Address correspondence to Jens Harder, jharder@mpi-bremen.de.

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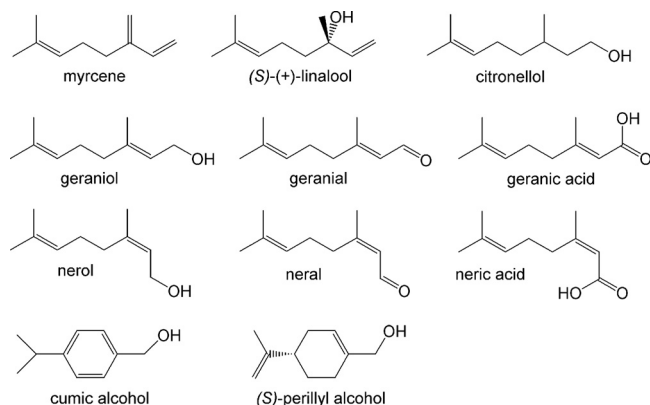


FIG 1 Monoterpene and monoterpenoid structures.

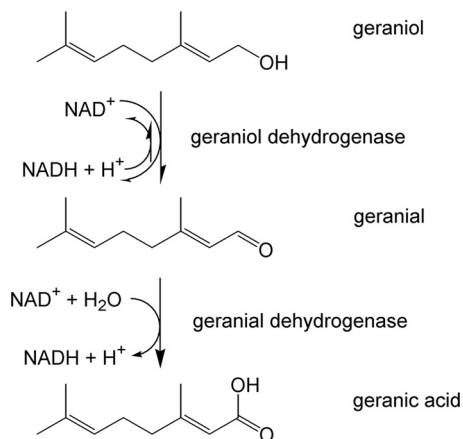


FIG 2 Geraniol oxidation pathway.

molecular basis for our previous observation, the formation of geranic acid from  $\beta$ -myrcene in cell extracts (24).

## MATERIALS AND METHODS

**Cultivation and biomass production.** Since its isolation in 1994, *C. de-fragrans* strain 65Phen has been maintained in the laboratory (17, 21) by four transfers per year, with 10% (vol/vol) inoculum. Anaerobic growth experiments were performed in 21-ml culture tubes with 15 ml aqueous medium, and the monoterpene was provided in 0.6 ml of the organic carrier phase 2,2,4,6,6,8,8-heptamethylnonane (HMN) (17). Biomass production yielded approximately 40 g (wet weight) cells grown on 15 mM limonene,  $\alpha$ -phellandrene, or 100 mM acetate and 100 mM nitrate in a 10l fermentor (24). For large-scale, heterologous production of proteins, *Escherichia coli* BL21 Star (DE3) (Invitrogen, Darmstadt, Germany) carrying the overexpression plasmid was grown in terrific broth medium (7).

**Enzyme assays and inhibition studies.** The NADH formation rate at 21°C in the standard assay for GeDH activity was photometrically measured at 340 nm. The assay mixture contained 100 mM glycine-NaOH, pH 9.4, 0.8 mM geraniol, and 1 mM NAD<sup>+</sup> (final concentration). Rate constants were calculated based on a molar extinction coefficient for NADH of 6,220 M<sup>-1</sup> cm<sup>-1</sup>. The catalytic properties (apparent  $K_m$  and maximum rate of the enzyme system [ $V_{max}$ ]) were determined by the Hanes-Woolf algorithm (10). A dimer with a molecular mass of 76,544 Da (based on the deduced protein sequence) was used for the calculation of  $k_{cat}$ . For inhibition studies with *N*-ethylmaleimide (NEM), diethylpyrocarbonate (DEPC), and 1-cyclohexyl-*N*-(2-morpholinoethyl)carbodiimide (CMC), the purified recombinant GeDH was incubated together with the inhibitor for 1 h before the reaction was started by addition of geraniol and NAD<sup>+</sup>. Fifty and eighty percent inhibitory concentrations ( $I_{50}$  and  $I_{80}$ , respectively) were graphically determined. The specificity of inhibition was proven by addition of a free amino acid (cysteine, histidine, or aspartate) at a concentration equimolar to the inhibitor at the  $I_{80}$ .

Geraniol is commercially available only as citral, a mixture of geranial and neral. Thus, GeDH activity was determined photometrically with 0.3 mM citral and 0.5 mM NAD<sup>+</sup> in the aforementioned buffer at 30°C.

**Chemical analyses.** Monoterpenes and monoterpenoids were analyzed in 1- $\mu$ l volumes by gas chromatography (GC) with flame ionization detection (FID) (XL auto system; PerkinElmer, Überlingen, Germany). Separation was performed on an Optima-5 column (0.25- $\mu$ m film thickness, 0.32-mm inside diameter [i.d.] by 50 m; Macherey-Nagel, Düren, Germany). The following temperature program was applied. The injection port temperature was 300°C, and the column start temperature was 85°C (held for 1 min). This was increased to 120°C at a rate of 15°C min<sup>-1</sup> and held at 120°C for 0.1 min; the temperature was then increased to 320°C at a rate of 45°C min<sup>-1</sup> and held at 320°C for 0.5 min. The detection temperature was 350°C. The split ratio was set to 1:10. For neral and geranial in citral, equal FID sensitivities were assumed and concentrations

were calculated from the citral concentration and relative areas of geranial and neral. Other GC-FID and GC-mass spectrometry (MS) analyses were performed as previously reported (17, 21, 22).

Organic acids, i.e., geranic acid and neric acid, were analyzed on a reverse-phase high-performance liquid chromatography (HPLC) system using the method described by Heyen and Harder (24). For detection of geranic acid formation, a 10-fold-diluted sample of the assay mixture was injected on a Nucleodur C<sub>18</sub> Isis column (Macherey-Nagel, Düren, Germany). Separation was performed with 1 mM H<sub>3</sub>PO<sub>4</sub> at 1 ml min<sup>-1</sup> in a water-acetonitrile gradient from 20 to 90% (vol/vol) acetonitrile at 25°C. UV detection was performed at 215 nm.

**Protein purification and analyses.** For the wild-type GeDH, 40 g (wet weight) of frozen cells was thawed in 100 ml 100 mM potassium phosphate, pH 7.0, 2 mM dithiothreitol (DTT) (buffer A) and then homogenized and disintegrated in three passages through a French pressure cell press (Aminco, Rochester, NY) at 10.3 MPa. Ultracentrifugation for 90 min at 150,000  $\times$  g at 4°C yielded a soluble extract. The enzyme was purified at 4°C on a Pharmacia LC system (GE Healthcare, Freiburg, Germany). The extract was applied at a flow rate of 2.5 ml min<sup>-1</sup> onto a DEAE fast-flow column (3-cm i.d., 200-ml column volume [CV]) and separated with a linear gradient of 0 to 1 M KCl in buffer A. Active fractions eluted early in the gradient and were directly applied to a phenyl-Sepharose 6 fast-flow column (2.6-cm i.d., 100-ml CV). Activity eluted at the end of a gradient of buffer A to 5 mM potassium phosphate, pH 7.0, 2 mM DTT, and 80% (vol/vol) ethylene glycol. After dialysis against 100 mM Na-HEPES, pH 7.0, 2 mM DTT (buffer B), the active fractions were purified on a DEAE fast-flow column (1.1-cm i.d., 10-ml CV) with a 0 to 1 M KCl gradient in buffer B. Molecular sieve chromatography was performed on a Superose 6 column (1.0-cm i.d., 47-ml CV) with buffer B containing 100 mM KCl.

The GeDH expressed in *E. coli* pET42a(+)*geoA* was purified at 4°C on an Äkta LC system (GE Healthcare, Freiburg, Germany) with filtered (0.2- $\mu$ m pore size) and degassed buffers. Soluble extract was prepared as aforementioned and diluted to 10 mg protein ml<sup>-1</sup>. The extract contained 130 mg protein and was loaded at 4 ml min<sup>-1</sup> on a butyl Sepharose 4 fast-flow column (2.6-cm i.d., 50-ml CV) that was equilibrated with 1.5 M (NH<sub>4</sub>)<sub>2</sub>(SO<sub>4</sub>) in 50 mM potassium phosphate buffer, pH 7.0, 2 mM DTT. Separation occurred in a gradient varying in both salt content and solvent polarity with 0 to 100% of 50% (vol/vol) ethylene glycol in 50 mM potassium phosphate buffer, pH 7.0, 2 mM DTT. Enzyme activity eluted at 42.5 vol% ethylene glycol. Active fractions were dialyzed against 10 mM potassium phosphate buffer, pH 7.0, 2 mM DTT (buffer C). Anion-exchange chromatography on a Source 15Q column (1.6-cm i.d., 20-ml CV) removed residual ethylene glycol. The enzyme activity eluted with 200 mM KCl in buffer C at 2 ml min<sup>-1</sup>. After concentration on a 10-Da membrane, gel filtration was performed on a Superdex 200 column (1.6-cm i.d.,

TABLE 1 Strains and plasmids used in this study

Strain or plasmid	Genotype, markers, and further characteristics	Source or reference
<b>Strains</b>		
<i>E. coli</i>		
One Shot TOP10	F <sup>-</sup> <i>mcrA</i> Δ( <i>mrr-hsdRMS-mcrBC</i> ) φ80 <i>lacZ</i> ΔM15 Δ <i>lacX74 recA1 araD139</i> Δ( <i>ara leu</i> )7697 <i>galU galK rpsL</i> (Str <sup>r</sup> ) <i>endA1 nupG</i>	Invitrogen
BL21 Star(DE3)	F <sup>-</sup> <i>ompT hsdSB</i> (r <sub>B</sub> <sup>-</sup> m <sub>B</sub> <sup>-</sup> ) <i>gal dcm</i> (DE3)	Invitrogen
DH5 α	<i>supE44</i> Δ <i>lacU169</i> φ80 <i>lacZ</i> ΔM15 <i>hsdR17 recA1 endA1 gyrA96 thi-1 relA1</i>	Invitrogen
EPI300-T1R	F <sup>-</sup> <i>mcrA</i> Δ( <i>mrr-hsdRMS-mcrBC</i> ) (Str <sup>r</sup> ) φ80d <i>lacZ</i> ΔM15 Δ <i>lacX74 recA1 endA1 araD139</i> Δ( <i>ara-leu</i> )7697 <i>galU galkλ<sup>-</sup> rpsL nupG trfA tonA dhfr</i>	Epicentre
<i>C. defragrans</i> 65Phen	Wild type	17
<b>Plasmids</b>		
pCR2.1-TOPO	TOPO TA cloning vector; Am <sup>r</sup> Km <sup>r</sup> <i>lacZα</i>	Invitrogen
pCR4-TOPO	TOPO TA cloning vector; Am <sup>r</sup> Km <sup>r</sup> <i>lacZα</i>	Invitrogen
pBluescript SK(+/-)	Cloning vector; Am <sup>r</sup> Km <sup>r</sup> <i>lacZα-ccdB</i>	Stratagene
pCC1FOS	Cloning vector fosmid prepn, Chl <sup>r</sup>	Epicentre
pET42a(+)	Expression vector; Km <sup>r</sup>	Novagen
pET42a(+) <i>geoA</i>	Expression vector; Km <sup>r</sup> <i>geoA</i>	This study
pET42a(+) <i>geoB</i>	Expression vector; Km <sup>r</sup> <i>geoB</i>	This study

120-ml CV) equilibrated in buffer C at 1 ml min<sup>-1</sup>. Standard proteins for the size determination were thyroglobulin (669 kDa), ferritin (440 kDa), catalase (232 kDa), aldolase (158 kDa), bovine serum albumin (67 kDa), ovalbumin (43 kDa), chymotrypsin (25 kDa), and RNase A (13.7 kDa).

Proteins were quantified according to the protocol of Bradford (6) with bovine serum albumin as a standard and visualized by SDS-PAGE according to the protocol of Laemmli (32). Native polyacrylamide gel electrophoresis was applied to determine independently the molecular mass of the GeDH. After separation in an 8% SDS-free polyacrylamide gel at 4°C, the gel was divided; one half was stained with Coomassie G250, and the other half was stained in 24.5 mM nitroblue tetrazolium chloride, 12 mM phenazine ethosulfate, 1 mM NAD<sup>+</sup>, and 0.4 mM geraniol in 100 mM glycine, pH 9.4, for 45 min in the dark. Afterwards, the gel was fixed with 7.5% acetic acid (modified from the protocol of Collins and Hege-man [9]).

Soluble extracts from cells grown on α-phellandrene or acetate were separated by anion-exchange chromatography on a Mono Q column (1-ml CV) with a NaCl gradient (1 to 400 mM, 40 fractions in a 10 mM range) in 50 mM Tris-HCl, pH 7.8, 2 mM DTT. Proteins in each fraction were separated by SDS-PAGE. The N termini of induced proteins as well as of native, purified GeDH were sequenced by Edman degradation after separation by SDS-PAGE and transfer by blotting onto a polyvinylidene difluoride (PVDF) membrane.

Soluble protein extracts for the detection of GaDH activity were prepared from *E. coli* pET42a(+)*geoB* as aforementioned with French pressure cell disintegration and ultracentrifugation, followed by removal of small molecules in a dialysis against 50 mM Na-HEPES, pH 7.0, 2 mM DTT (Visking dialysis tubing; Serva, Heidelberg, Germany).

**Molecular biology and data deposition.** Standard techniques for molecular cloning and sequencing were applied (48) using the listed strains and plasmids (Table 1) and primers (Table 2). Fosmid libraries were prepared with the pCC1FOS vector using a CopyControl fosmid library production kit according to the manufacturer's instructions (Epicentre, Madison, WI) with the following modifications. Genomic DNA was embedded in low-melting-point agarose and equilibrated in 0.5× TE (48) and in end repair mix without enzyme. DNA strand ends were filled by incubation of 6 μl end repair enzyme mix together with the 40-μl agarose plug for 50 min at room temperature in 120 μl end repair mix containing a 0.5 mM concentration of the deoxynucleoside triphosphates (dNTPs). A transfer of the agarose plug in 500 μl 0.5 M EDTA stopped the reaction. DNA of about 25 to 48 kb was obtained on a preparative pulsed-field gel electrophoresis (PFGE) gel using 1% SeaPlaque GTG agarose (FMC Bio-Products) and a Bio-Rad contour-clamped homogeneous electric field (CHEF) DRIII system (Bio-Rad) applying 0.5× TBE (48), a switch time of 1 to 10 s, a reorientation angle of 120°, and 6 V/cm at 14°C for 16 h. The gel section of interest was excised, equilibrated in 1× TE, and digested with beta-agarase (New England BioLabs). The DNA was concentrated by drop dialysis on mixed cellulose ester membrane discs (Millipore; 0.025-μm pore size) against 30% polyethylene glycol 8000 in bidistilled water.

Repetitive elements in the fosmid sequences required a semimanual assembly and contig closure by primer walking and gapping PCRs according to standard protocols. Sequence analyses and assembly were performed with Sequence Analysis 5.2 (Applied Biosystems, Foster City, CA), Sequencher 4.5 (Gene Codes, Ann Arbor, MI), and Lasergene (DNASar, Madison, WI). The expression system used the vector pET42a(+) (No-

TABLE 2 List of primers

Name	Purpose	Sequence <sup>a</sup>
GDHfd1	Nested PCR on N-terminal protein sequence	ATGAACTGTAC(GC)CA(AG)GA(CT)TT
GDHfd2		AC(GC)CA(AG)GA(CT)TTCAT(CT)(AT)(GC)(GC)GC
GDHRev		AC(GC)GG(CT)TC(GC)AC(GC)GC(GC)A(AG)(GC)GG
GDH-F1	PCR-mediated synthesis of DIG-labeled DNA probe	ACGCAGGATTTTCATCAGG
GDH-R1		TACGGGTTCGACGGCGAA
<i>geoA</i> _BgIII_R	Construction of expression vector pET42a(+) <i>geoA</i>	<u>AGATCTT</u> CAGAACACCAGCACCGGCTTG
<i>geoA</i> _NdeI_F		<u>CATATGA</u> ACGACACCCAGGATTTTCATTTC
<i>geoB</i> _NdeI_F		<u>CATATGACCA</u> TGCATCACCAGCACATCTTC
<i>geoB</i> _SalI_R	Construction of expression vector pET42a(+) <i>geoB</i>	<u>GTCGACCTAGCCA</u> AGCAGGTACACTGAC

<sup>a</sup> Restriction sites are underlined.

TABLE 3 Purification of GeDH from *C. defragrans* 65Phen

Purification step	Total protein (mg)	Total activity (U)	Sp act (U/mg)	Yield (%)	Purification (fold)
Cell extract	3,160	175	0.056	100	1
1st DEAE fast flow	269	111	0.41	71	12
Phenyl-Sepharose 6 fast flow	46	101	2.19	58	69
2nd DEAE fast flow	36	156	4.33	89	114

vagen, Darmstadt, Germany) in the host *E. coli* BL21 Star(DE3) (Invitrogen, Darmstadt, Germany).

**Nucleotide sequence accession number.** The 50-kb genomic contig including *geoA* and *geoB* as well as the protein sequences were deposited within the EMBL nucleotide sequence database (accession number FR669447.2).

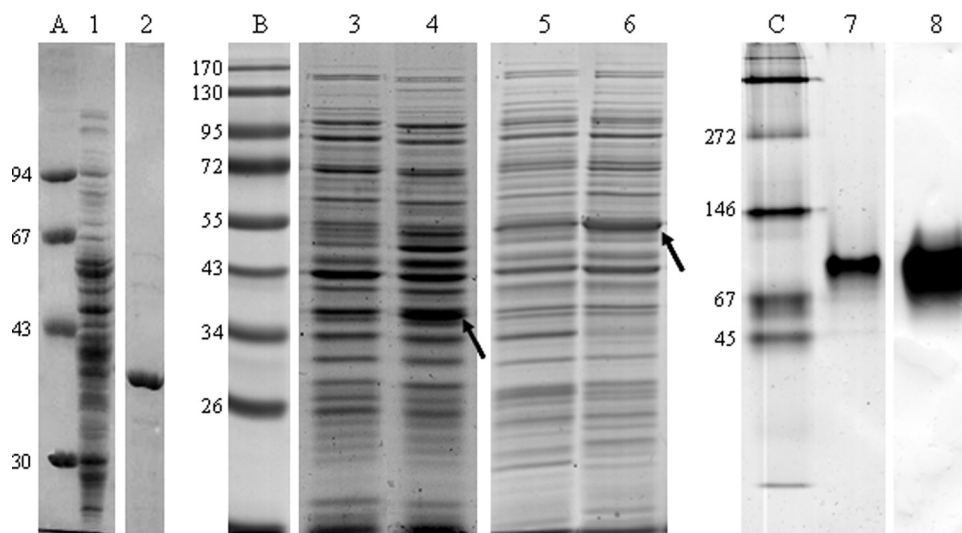
## RESULTS

**Geraniol dehydrogenase activity induced in monoterpene-grown cultures.** The identification of geranic acid as an intermediate in *C. defragrans* cells during the anaerobic mineralization of monoterpenes (24) suggested the presence of geraniol and geraniol dehydrogenase activities (7). We had reported that geraniol did not support the growth of *C. defragrans* (17). In those experiments with HMN as an organic carrier phase, a calculated geraniol concentration of 4 mM corresponded to actual concentrations of 80 mM relative to that of the HMN phase and of 70  $\mu$ M geraniol in the aqueous phase, according to the partial pressure of geraniol in the organic phase and a water solubility of geraniol of 2.62 mM (57). We then tested lower geraniol concentrations and observed geraniol utilization at aqueous concentrations of 5  $\mu$ M geraniol. Higher concentrations of geraniol inhibited microbial growth, and nerol, geranial, and neral accumulated in the organic

carrier phase with concomitant formation of traces of nitrite (see Tables S1 and S2 in the supplemental material).

We investigated geraniol biotransformations by soluble extracts of *C. defragrans*. Geranic acid was formed not only from  $\beta$ -myrcene, as previously reported (24), but also from geraniol and nerol. Nerol was also transformed to neric acid (see Table S3 in the supplemental material). Alcohol dehydrogenase activities were determined in soluble extracts by the reduction of NAD<sup>+</sup>. Geraniol dehydrogenase and benzyl alcohol dehydrogenase activities were 1.39  $\pm$  0.10 and 24.6  $\pm$  0.2 mU mg<sup>-1</sup> protein ( $n$  = 2) in extracts from limonene-grown cells, in comparison to 0.14  $\pm$  0.00 and 1.64  $\pm$  0.09 mU mg<sup>-1</sup> protein ( $n$  = 2) in extracts from acetate-grown cells. The 10- to 15-fold-higher rates of NADH formation indicate an inducible GeDH activity.

**Identification of a GeDH and induced proteins.** GeDH activity was purified with the guidance of the purification protocol for a dehydrogenase from *P. putida* acting on 3-methyl-2-buten-1-ol, a homologue of geraniol (36). The 114-fold purification (Table 3) yielded a nearly homogenous protein with an apparent molecular mass of 39 kDa as determined by SDS-PAGE (Fig. 3A) and 52 kDa as determined by molecular sieve chromatography. The determination of the N-terminal protein sequence (MN-TQDFISAQA-VL-QVGGPLAVEPVI) by Edman degradation enabled the design of a degenerate primer and the amplification of a small DNA fragment 73 bp in length (Table 2). DNA sequencing of the cloned 73-bp fragment verified an open reading frame for the N-terminal protein sequence. Application of the 73-bp fragment as probe in Southern blot analysis revealed a location on a BamHI restriction fragment. A plasmid library with BamHI fragments of the *C. defragrans* genome was screened with the digoxigenin (DIG)-labeled 73-bp probe. A positive plasmid contained a genomic fragment 5,631 bp in length with the complete gene for the geraniol dehydrogenase at the 5' end of the insert. We named the gene *geoA*, the first gene on the geraniol oxidation pathway.



**FIG 3** (A) SDS-PAGE of the native GeDH purified from soluble extract of *C. defragrans* 65Phen (lane 1, soluble extract; lane 2, GeDH). (B) Induction of expression of GeDH (39 kDa) (lanes 3 and 4) in *E. coli* pET42a(+ )*geoA* and GaDH (53 kDa) (lanes 5 and 6) in *E. coli* pET42a(+ )*geoB*. The induced proteins are indicated by an arrow. Protein samples were taken just before (lanes 5 and 7) and after (lanes 6 and 8) 2 h of isopropyl- $\beta$ -D-thiogalactopyranoside (IPTG) induction. (C) Native PAGE of GeDH. One microgram of purified recombinant GeDH was applied per lane; lane 7 is Coomassie G250 stained, and lane 8 is activity stained with geraniol as a substrate. Lane A, low-molecular-weight marker (Pharmacia); lane B, Page Ruler prestained protein ladder (Fermentas); lane C, native protein marker (Serva). Numbers at the left are molecular weights (in thousands).



**TABLE 4** N-terminal amino acid sequences and identified open reading frames of proteins found to be present during growth with phellandrene but not with acetate

Position in contig	N-terminal amino acid sequence	N-terminal amino acid sequence of ORF in contig (GenBank accession no. FR669447)	Annotation
ORF4	ND-TPPGQWTP(P)VV	MNDRTPPGQWTPPPVD	Hypothetical protein
ORF11	MANPKSEYDVIIIVGGGLNGLA	MANPKSEYDVIIIVGGGLNGLA	Phytoene dehydrogenase-like oxidoreductase
ORF12	MSEVKQ-DVVVIGAG-	MSEVKQCDVVVIGAGH	Phytoene dehydrogenase-like oxidoreductase
ORF18	(TA)I(D)(T)Q(H)IFVGGQWIAP(K)	MTIDHQHIFVGGQWIAPK	Geranial dehydrogenase <i>geoB</i>
ORF44	MIE-LFGPE-(F)M(F)-(D)TV-K	MIERRLFGPEHEMFRDTRK	Acyl-CoA dehydrogenase

For the analysis of the genomic neighborhood, we prepared a fosmid library and identified by PCR screening two fosmids that carried *geoA*. A continuous sequence of 50 kb was obtained by semimanual sequence assembly and used to interpret the results of a differential proteomic analysis. *C. defragrans* was grown with acetate or phellandrene. Proteins present in soluble extracts of these cells were compared by anion-exchange chromatography in combination with one-dimensional SDS-PAGE (see Fig. S1 in the supplemental material). The N termini of proteins specifically expressed in monoterpene-utilizing cells were sequenced by Edman degradation. Some amino acid sequences were coded by open reading frames annotated on the 50-kb contig. One of these proteins corresponded to a predicted aldehyde dehydrogenase (ALDH), initially named *geoB*. Other expressed genes located on the contig include oxidoreductases and acyl coenzyme A (acyl-CoA) dehydrogenases (Table 4), which may play a role in the further degradation of geranic acid.

The genes *geoA* and *geoB* were introduced into the overexpression vector pET42a(+) and functionally expressed in *E. coli* BL21 Star(DE3) (Table 1). Induction of gene expression yielded proteins with the expected molecular masses (Fig. 3B). Soluble extracts of *E. coli* carrying *geoA* showed a NADH formation rate of  $0.26 \pm 0.03$  mU mg<sup>-1</sup> protein ( $n = 3$ ) with geraniol. Soluble extracts of *E. coli* carrying *geoB* had an NADH formation rate of  $3.7 \pm 0.9$  mU mg<sup>-1</sup> protein ( $n = 3$ ) with citral, the commercially available mixture of geranial and neral. Soluble extracts of *E. coli* pET42a(+) oxidizes neither geraniol nor citral, and therefore the inserted genes must be responsible for the monoterpene oxidation. The genes were named geraniol dehydrogenase (*geoA*/GeDH) and geranial dehydrogenase (*geoB*/GaDH).

**Characterization of the geraniol dehydrogenase.** The identified open reading frame for the GeDH was 1,122 bp in length, had a GC content of 71.21% and a molecular mass of 38,272 Da, and coded for a protein of 373 amino acids (aa) (see Fig. S2 in the supplemental material). It displays specific motifs of medium-chain-length, Zn-containing, NAD<sup>+</sup>-dependent alcohol dehydrogenases (28) and affiliates within the *mdr19* family, which comprises benzyl/aryl ADHs with a long quaternary structure-determining loop (QSDL) of more than 31 aa (30). The amino acid identity to eukaryotic geraniol dehydrogenases was low: 25.3% to the GeDH from sweet basil, *Ocimum basilicum* (25), and 27.1% to the GeDH from the astigmatid mite *Carpoglyphus lactis* (38). The *C. defragrans* GeDH was compared in a multiple-sequence alignment with ADHs of high similarity, namely, plant GeDHs and well-characterized ADHs, such as the horse liver ADH (HLADH), which is representative of Zn-containing, medium-chain ADHs of the MDR type (14, 27, 44) (Fig. S4). The coenzyme binding domain, which contains the Rossmann fold (46) with the glycine-rich phosphate binding loop (GXGXXG) as well as the

catalytic zinc binding motif (GHXXGXXXXXGXXV), was found to be conserved in *C. defragrans* GeDH (Gly200, Gly 202, Gly205, and Gly66-Val80). The structural zinc atom is coordinated by four conserved Cys residues (Cys96, Cys99, Cys102, Cys110).

The recombinant GeDH was purified to homogeneity. The native molecular mass was determined on different molecular sieve columns and varied between 52 and 91 kDa, likely indicating an equilibrium between the monomeric and dimeric state. Independently, native PAGE with subsequent GeDH activity staining revealed an apparent molecular mass of  $85 \pm 7$  kDa, based on the size of marker proteins (Fig. 3C). These observations suggest that a dimer represents the native conformation of the active enzyme. Also the other members of the *mdr19* family of the MDR ADH are active as dimers, with the exception of the benzyl ADH from *Acinetobacter calcoaceticus* (Protein Data Bank [PDB] accession number 1F8F) (30). Consequently, a dimer was used in the calculation of the catalytic efficiency.

According to gas chromatography analyses, the purified GeDH catalyzed the formation of geranial from geraniol. The *cis* isomer neral was formed in glycine buffer, pH 9.4, but not at pH 7.0 in 100 mM potassium phosphate, 2 mM DTT. Geraniol was not isomerized to nerol or linalool at pH 7.0. These results suggest the retention of the *trans* configuration of the alkene during biological oxidation and a chemical isomerization of geranial to neral under alkaline conditions. Citral, the commercially available mixture of geranial and neral, was not further oxidized.

Because GeDH is classified among the benzyl and aryl ADHs, the kinetic properties of the native and the recombinant purified GeDHs were determined for benzyl alcohol, cumic alcohol (*p*-isopropyl-benzyl alcohol), and (*S*)-(-)-perillyl alcohol in comparison with geraniol, nerol, and citronellol (Fig. 1). The purified GeDH exhibited typical Michaelis-Menten kinetics with all substrates (Table 5). The apparent  $K_m$  values for geraniol and perillyl alcohol were around 5  $\mu$ M, indicating a high affinity for these substrates. The affinities for nerol and citronellol were significantly lower. Benzyl alcohol had the highest  $V_{max}$  value. The catalytic efficiency calculation identified perillyl alcohol as the best substrate for the enzyme, followed by geraniol and cumic alcohol.

The enzyme reduced one molecule of NAD<sup>+</sup> to NADH per geraniol molecule provided in geraniol-limited assays. NADP<sup>+</sup> was ineffective as a cosubstrate. The cofactor specificity for NAD<sup>+</sup> was likely defined by the negative charge of Glu224, which is known to repel the additional phosphate of NADP<sup>+</sup> (31, 44, 55). The pH optimum was, as expected for MDR ADHs, in the alkaline range at pH 9.4. Dichlorophenolindophenol (DCPIP), but not phenazine methosulfate (PMS), was accepted as an alternative electron acceptor in the enzyme reaction. GeDH is sensitive to chelating reagents, as expected for zinc-containing ADHs; 7 mM EDTA inhibited the enzyme by 96%. Inhibition by *N*-ethylma-

TABLE 5 Enzyme kinetics of native and recombinant GeDH<sup>a</sup>

Substrate	Native GeDH			Recombinant GeDH		
	$K_m$ ( $\mu\text{M}$ )	$V_{\text{max}}$ ( $\text{U mg}^{-1}$ )	$k_{\text{cat}}/K_m$ ( $1 \cdot 10^6 \text{ s}^{-1} \text{ M}^{-1}$ )	$K_m$ ( $\mu\text{M}$ )	$V_{\text{max}}$ ( $\text{U mg}^{-1}$ )	$k_{\text{cat}}/K_m$ ( $1 \cdot 10^6 \text{ s}^{-1} \text{ M}^{-1}$ )
Geraniol	5	10	1.57	3.3	2.6	0.62
Nerol	45	18	0.31	23.2	9.1	0.31
Citronellol	86	11	0.10	57.5	3.7	0.05
(S)-(-)-Perillyl alcohol	7	18	2.02	4.4	19.9	3.55
Cumic alcohol	21	14	0.52	6.0	7.2	0.94
Benzyl alcohol	170	47	0.22	115.7	16.8	0.11

<sup>a</sup> Values are apparent values.

leimide ( $I_{80} = <0.5 \text{ mM}$ ), diethylpyrocarbonate ( $I_{80} = 17 \text{ mM}$ ), and 1-cyclohexyl-*N*-(2-morpholinoethyl)carbodiimide ( $I_{80} = 34 \text{ mM}$ ) was partly suppressed by addition of equimolar amounts of cysteine, histidine, and aspartate, respectively. Inhibition was reduced from 80% to values between 43 and 58%. This indicates a participation of these amino acids in GeDH activity.

**Characterization of the geranial dehydrogenase.** The *geoB* gene has 1,437 bp and codes for a protein of 478 aa (see Fig. S3 in the supplemental material) with a molecular mass of 50,637 Da, which coincides with the apparent molecular mass of 53 kDa from SDS-PAGE (Fig. 3B). ALDHs with related sequences originated from the Gram-positive order *Actinomycetales*, i.e., from *Rhodococcus opacus* (59% identity, GenBank accession number YP\_002781874.1), *Rhodococcus rhodochrous* (58% identity, AAC15840.1), *Rhodococcus jostii* (59% identity, ABG99066.1), and “*Streptomyces bingchengensis*” (57% identity, ADI11766.1). The *C. defragrans* GaDH is affiliated with the ALDH superfamily. A Clustal W alignment with ALDHs from different organisms revealed amino acid residues that are conserved in more than 95% of ALDHs (Fig. S5) (40). In *C. defragrans* GaDH, they are the following: Arg69, Gly146, Asn155, Pro157, Gly172, Lys178, Gly230, Gly254, Gly283, Cys286, Glu383, Phe385, Pro387, Gly433, Asn438, and Gly450. The Rossmann fold motif was shortened by one amino acid and only partially covered; Gly209 and Gly214 were present, but the second Gly residue was replaced by an Asp.

HPLC analyses of free medium-chain fatty acids revealed that only the *trans*-isomer geranic acid, and not the *cis* isomer neralic acid, was the product of the oxidation catalyzed by soluble extracts of *E. coli* pET42a(+)*geoB*. Geranial disappeared faster than neral, suggesting that geranial is biologically oxidized and that neral is chemically isomerized to geranial and subsequently oxidized biologically (Table 6). The formation of geranic acid and NADH correlated in a 1:1 ratio (Fig. 4). Thus, the geranial dehydrogenase acted specifically on geranial.

## DISCUSSION

In this study, we identified a geraniol dehydrogenase and a geranial dehydrogenase of the  $\beta$ -myrcene degradation pathway present in denitrifying *Castellaniella defragrans*. The activities of both enzyme activities were specifically induced during growth with monoterpenes, and these enzymes were expressed in *E. coli* in their active forms. In contrast to the linalool dehydratase isomerase (7), the GeDH and the GaDH were not oxygen sensitive and were located in the cytoplasm. The genes did not code for a signal peptide for a periplasmic location. The involvement of alcohol and aldehyde dehydrogenase *per se* in degradation pathways is common, but the detailed characterization revealed the particular properties of GeDH and GaDH and emphasized their singularity.

The GeDH of *C. defragrans* has, among the GeDHs so far reported, the highest affinity for geraniol (Table 7). A number of other GeDHs from plant sources (25, 43, 49) and the insect pest *Carpoglyphus lactis* (37) are involved in geraniol synthesis and therefore may not have evolved a higher affinity for geraniol than enzymes in degradation pathways. The transformation of  $\beta$ -myrcene to geraniol by the linalool dehydratase isomerase is thermodynamically unfavorable and results in a steady state at low geraniol concentrations (7, 35), thus requiring a high-affinity GeDH for an efficient metabolic flux. Furthermore, geraniol was found to inhibit monoterpene metabolism at aqueous concentrations above 5  $\mu\text{M}$  in the two-liquid-phase system with geraniol in the organic carrier phase. The molecular target for this specific inhibition remains unclear, but incorporation of the hydrophobic substances in membranes resulting in disruption of the proton motive force has often been reported (13, 53).

The enzyme possesses a higher affinity for the allyl alcohols geraniol and nerol than for the nonallylic citronellol. This may have a chemical explanation: the alkene bond donates overlapping  $\pi$  electron density to the carbon of the alcohol, thereby stabilizing

TABLE 6 Citral conversion by soluble extracts of *E. coli* pET42a(+) and pET42a(+)*geoB*<sup>a</sup>

Plasmid	Time (h)	Geranial (mM)	Neral (mM)	Geraniol (mM)	Nerol (mM)	Geranic acid (mM)
pET42a(+) <i>geoB</i>	0	11.60 $\pm$ 0.15	7.20 $\pm$ 0.20	0	0	0
	2	2.95 $\pm$ 0.15	3.25 $\pm$ 0.05	2.3 $\pm$ 0.1	1.4 $\pm$ 0.1	7.65 $\pm$ 0.15
	24	0.32 $\pm$ 0.08	0.25 $\pm$ 0.05	0	0	8.90 $\pm$ 1.10
pET42a(+)	0	11.50 $\pm$ 0.10	6.75 $\pm$ 0.05	0	0	0
	2	11.32 $\pm$ 0.28	6.95 $\pm$ 0.35	0	0	0
	24	8.15 $\pm$ 0.55	5.20 $\pm$ 0.40	0	0	0

<sup>a</sup> Soluble extracts were assayed in duplicate with 0.5 mM NAD<sup>+</sup> in a two-phase system with 20 mM citral in HMN. The organic phase serves as reservoir for the substrate as well as for the products and was analyzed by gas chromatography.

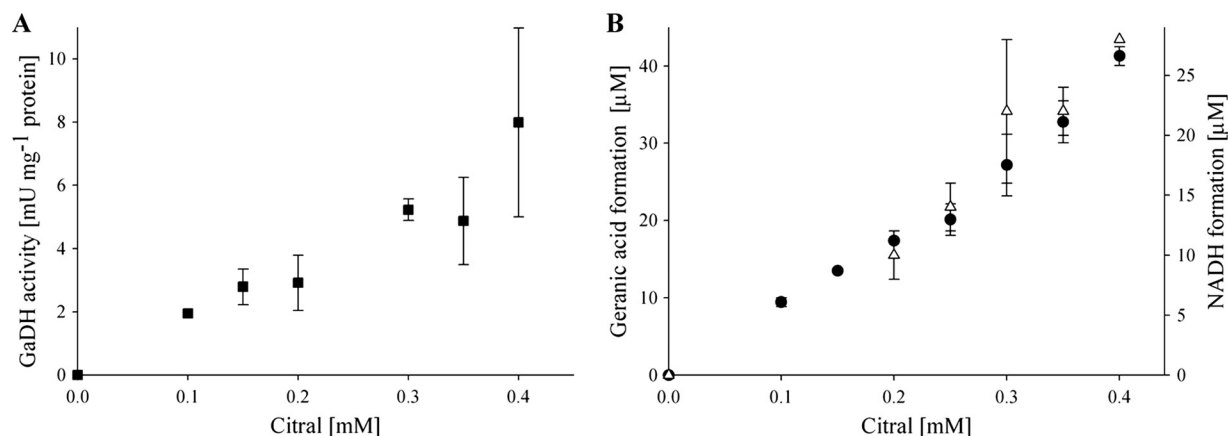


FIG 4 (A) Geranyl dehydrogenase activity (NADH formation) at different citral concentrations, determined in triplicate measurements. (B) Geranic acid formation (●) was measured via RP-HPLC, and NADH formation (Δ) was determined spectrophotometrically.

the transient positive charge during hydride transfer of  $\text{NAD}^+$  and enhancing substrate binding as well as catalysis (12, 44). Alternatively, the binding pocket may be narrow and favor substrates with  $\text{sp}^2$ -hybridized C-2 and C-3 atoms. Citronellol, with  $\text{sp}^3$ -hybridized C-2 and C-3 atoms, is more space filling and may not fit perfectly into the binding pocket (15).

The preference for allyl alcohols over benzyl alcohols remains an open question. A multiple alignment with highly similar ADHs, i.e., plant GeDH and well-characterized ADHs, revealed some potentially important amino acids. His48, Arg229, and Lys368 are the positive charges interacting with the cofactor  $\text{NAD}^+$ . Interestingly, mutation of Arg50 to His50 in benzyl ADH from *A. calcoaceticus* (18) decreased the catalytic activity of the perillyl-ADH activity 10-fold but that of the benzyl ADH activity 1,000-fold. This suggests a crucial role for His48 in the activity of *C. defragrans* GeDH. His51 of HLADH is replaced by Val52 in *C. defragrans* GeDH. The positive charge is not conserved among many benzyl ADHs and is not necessary for the reaction (18). The substrate specificity for MDR ADHs is defined by the hydrophobic cleft formed by residues of both domains and the cofactor itself. HLADH catalyzes the oxidation of many alcohols; however, an increase in the spatial volume of the alcohol correlates with a lower activity (44). The substrate binding site can be divided: the inner part is close to the catalytic zinc ion and formed by Ser48, Phe93, Phe140, and Leu141, the middle part by Leu57, Leu116, Val294, and Ile318, and the outer part by Phe110, Met306, and Leu309

(HLADH [14]). The *C. defragrans* GeDH may have a smaller, more polar, pocket with Thr49, Phe93, Phe141, and Phe142 in the interior and Phe57, Phe118, Leu293, and Ile317 in the middle part. Threonine replaces serine and reduces the available space for the alcohol, thus favoring smaller alcohols (11, 16, 58) or the less space-filling allyl and benzyl alcohols. The increase in phenylalanines enables interaction with  $\pi$  electrons of the substrate. The crystal structure of a Phe93Ala mutant revealed a loss of the ability to bind a benzyl alcohol in the perfect position for the catalysis (47). In fact, Thr49 and Phe93 are present in most of the bacterial benzyl ADHs (18, 52). Overall, the alignment together with our knowledge of ADHs suggests that in the *C. defragrans* GeDH, the amino acids His48, Thr49, and Phe93 play a crucial role in the substrate specificity of the enzyme.

In contrast to the often-observed operon organization of genes involved in metabolic pathways, the genes of the initial  $\beta$ -myrcene pathway, *ldi*, *geoA*, and *geoB*, are well separated on the genome but physically close in distance. The *C. defragrans* GaDH is not a citral dehydrogenase; it acts only on the *trans*-isomer geranial and affiliates with the ALDH superfamily. So far, the presence of a geranial dehydrogenase has not been reported. Future studies will deal with the characterization of this enzyme in more detail.

Thermodynamically, oxidation of the aldehyde to geranic acid yields sufficient free energy for ATP synthesis from inorganic phosphate and ADP. However, the hydration of  $\beta$ -myrcene, the isomerization to geraniol, and the coupling of geraniol oxidation

TABLE 7 Apparent  $K_m$  values, pH optima, and cofactor dependence for enzymes acting on geraniol

Organism	$K_m$ ( $\mu\text{M}$ )	pH optimum	Cofactor	EC no. (enzyme)	Reference
<i>C. defragrans</i>	3.3	10	$\text{NAD}^+$	1.1.1.183	This study
<i>Carpoglyphus lactis</i> (prune mite)	51	9	$\text{NAD}^+$	1.1.1.183	38
<i>Citrus</i> sp. (orange)	46.5	9	$\text{NADP}^+$	1.1.1.183	43
<i>Cymbopogon flexuosus</i> (lemongrass)	100	ND <sup>a</sup>	$\text{NADP}^+$	1.1.1.183	49
<i>Ocimum basilicum</i> (basil)	30	9.5	$\text{NADP}^+$	1.1.1.183	25
<i>Homo sapiens</i> (human)	25	9	$\text{NADPH}$	1.1.1.21 (aldehyde reductase)	16
<i>Ipomoea batatas</i> (sweet potato)	729	ND	$\text{NADP}^+$	1.1.1.216 (farnesol dehydrogenase)	26
<i>Arabidopsis thaliana</i>	800	10	$\text{NAD}^+$	1.1.1.284 [(S)-(hydroxymethyl) glutathione dehydrogenase]	1
Rosa hybrid	2,783	ND	ND	2.3.1.84 (alcohol <i>o</i> -acetyltransferase)	19
<i>Sorghum bicolor</i> (sorghum)	140	ND	ND	2.4.1.85 (cyanohydrin beta-glucosyltransferase)	20
<i>Thea sinensis</i> (tea)	6,250	ND	NAD	1.1.1.1 (alcohol:NAD oxidoreductase)	23

<sup>a</sup> ND, not determined.

to the reduction of NAD<sup>+</sup> are three thermodynamically unfavorable reactions. Thus, geraniol oxidation shifts the overall reaction from  $\beta$ -myrcene to geranic acid into a favorable reaction and allows the process to occur even at low myrcene concentrations.

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