

REVIEW

Anthrax toxin protective antigen—Insights into molecular switching from prepore to pore

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Abstract: The protective antigen is a key component of the anthrax toxin, as it allows entry of the enzymatic components edema factor and lethal factor into the host cell, through the formation of a membrane spanning pore. This event is absolutely critical for the pathogenesis of anthrax, and although we have yet to understand the mechanism of pore formation, recent developments have provided key insights into how this process may occur. Based on the available data, a model is proposed for the kinetic steps for protective antigen conversion from prepore to pore. In this model, the driving force for pore formation is the formation of the phi (ϕ)-clamp, a region that forms a leak-free seal around the translocating polypeptide. Formation of the ϕ -clamp elicits movements within the prepore that provide steric freedom for the subsequent conformational changes required to form the membrane spanning pore.

Keywords: anthrax; toxin; protective antigen; pore; pH; membrane protein; review

Introduction

Anthrax disease is caused by the gram positive bacterium *Bacillus anthracis*, and this year marks the 10th anniversary of the anthrax attacks that occurred in September and October of 2001. There were 11 confirmed cases of inhalational anthrax due to exposure to aerosolized anthrax spores, resulting in the deaths of five people. Although administration of antibiotics during the 2001 attacks was effective in saving lives,¹ major efforts since then have been directed toward improving early diagnosis and treatment of the disease.² While anthrax has the potential

to be used as a biological weapon, it is also a threat to agriculture worldwide. Recently, animal outbreaks have occurred in Africa, Australia, Bangladesh, Brazil, Canada, China, Sweden, and the United States. Under more harsh environmental conditions (lack of nutrients, dry conditions), the bacterium forms a dormant spore which is very hardy and can last in the soil for an indefinite amount of time. To cause infection, however, the spore must germinate to an actively growing vegetative state, but requires the right physicochemical conditions (pH > 6, temperature > 15°C).^{3,4} The conversion from a dormant spore to a vegetative state was seen in the recent re-emergence of anthrax in Australia, where flooding unearthed anthrax spores that had been dormant for decades, eventually infecting and causing the deaths of over 53 cattle and 1 horse.⁵

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There are three primary routes of infection that *B. anthracis* uses—the skin (cutaneous), the gut (gastrointestinal) or the lung (inhalational), the former two the primary mechanisms of entry into animals from either contact with spores on the skin, or through ingesting contaminated soil.⁶ In inhalational anthrax, the inhaled spores are engulfed by macrophages and carried to the mediastinal lymph nodes, and eventually burst out of the cell and spread into the bloodstream.^{7–9} The incubation period for germination can be quite long, and is the reason for the 60-day long time course of antibiotic treatment for an anthrax infection. This allows a sufficient amount of time for spore germination and exposure of the bacterium to antibiotic.^{10,11}

Shortly after germination, the bacterium will secrete an antiphagocytic poly-D-glutamic acid capsule,¹² and a set of three proteins: edema factor (EF), lethal factor (LF), and protective antigen (PA).^{13,14} These three proteins are alone nontoxic, but can assemble together in the bloodstream or on host cell surfaces to form the anthrax toxin, which belongs to the broader class of toxins called AB toxins.¹⁵ Both EF and LF of the anthrax toxin correspond to the “A” components, and PA corresponds to the “B” component. PA attaches to host cells, recruiting EF and LF to the cell surface, and forms a pore inside the cell that allows entry of EF and LF into the cytosol. PA also provides protective immunity against anthrax infection in animals,¹⁶ and is the major component of the current licensed anthrax vaccine, anthrax vaccine absorbed (AVA), given to active military personnel in the United States.

In early studies of the anthrax toxin, the individual components EF, LF, and PA were purified from the supernatant of cultures of the Sterne strain of *B. anthracis*, a strain that lacks the pXO2 plasmid which encodes the poly-D-glutamic acid capsule required for virulence.¹⁷ In studies on primates, injection of PA, EF or LF alone, or the complex of PA + EF (called edema toxin, ETx) did not cause lethality (although negative effects were observed with the injection of PA alone¹⁸). However, injection of PA + LF (called lethal toxin, LTx) resulted in fairly rapid physiological changes, including anaphylaxis, respiratory difficulties, and changes in cortical brain activity, leading to the death of the animal within 30 hours.^{18,19} There are currently no antitoxin treatments available against the anthrax toxin, and therefore it is absolutely critical that antibiotic treatment be given as early as possible, before the secretion of these toxin components.

EF is an 89 kDa calcium-calmodulin-dependent adenylate cyclase, which raises intracellular cAMP levels²⁰ and has been shown to alter cellular chemotaxis.²¹ LF is a zinc-metalloproteinase that cleaves

mitogen activated protein kinase kinases (MAPKKs),²² and has been shown to inhibit the release of cytokines.^{23,24} However, recent studies using lung epithelial cells also indicate that LF disrupts the cytoskeletal and microtubule networks needed to maintain the integrity of the epithelial barrier.²⁵ PA is an 83 kDa, calcium binding, four-domain monomeric protein that interacts with host cells and undergoes several changes that lead ultimately to the formation of a membrane spanning pore, and is the main focus of this review.

PA-Receptor Binding

From early studies it was suggested that PA was the component that interacted with host cells,¹⁹ and later the receptor that bound to PA was identified as anthrax toxin receptor or “ATR,” previously identified as tumor endothelial marker 8 (TEM8).^{26,27} A second cell surface receptor was subsequently identified, called capillary morphogenesis protein 2 (CMG2) or anthrax toxin receptor 2 (ATR2).²⁸ More recently, heterodimeric complexes of integrin β_1 have also been shown to bind PA and promote internalization.²⁹ Natural ligands for CMG2 have been shown to include type IV collagen and laminin which comprise the basal lamina,³⁰ and for TEM8 the $\alpha 3$ subunit of collagen type VI.³¹ Exactly how PA effectively competes with these natural ligands for binding to these receptors is not well understood.

The receptors TEM8, CMG2, and integrin β_1 all contain an integrin I (inserted) domain or von-Willebrand factor A (vWA) domain, that contains acidic residues that coordinate a magnesium ion on the surface of the receptor, forming a metal-ion-dependent adhesion site (MIDAS).³² The three-dimensional structures of the full length PA,³³ and heptameric prepore,³⁴ bound to CMG2 [Figs. 1 and 2(A), respectively], confirmed earlier biochemical studies indicating that the receptor bound primarily to domain 4 of PA (residues 595-735).^{26,35} Domain 4 contains an essential aspartic acid (D683), which completes the coordination shell to the MIDAS site magnesium (Fig. 1). Mutation of this aspartic acid to asparagine (D683N) abrogates binding to TEM8, but surprisingly not to CMG2.³⁶ Domain 4 constitutes the majority of the binding interface, and buries $\sim 1300 \text{ \AA}^2$ of surface area, sufficient for binding to both TEM8 and CMG2.^{26,33,37} However, the crystal structures of PA or prepore bound to CMG2 also revealed that a small loop from domain 2 (the domain $2\beta_3$ - $2\beta_4$ loop, Fig. 1) is inserted into a groove on the surface of CMG2, and contributes another $\sim 600 \text{ \AA}^2$ of surface area to the binding interface. This additional surface area explains in part why the D683N mutation in PA does not prevent binding to CMG2, as well as the higher affinity of PA for CMG2 versus TEM8 ($K_D \sim 200 \text{ pM}$ vs. 200 nM , respectively).^{36,38}

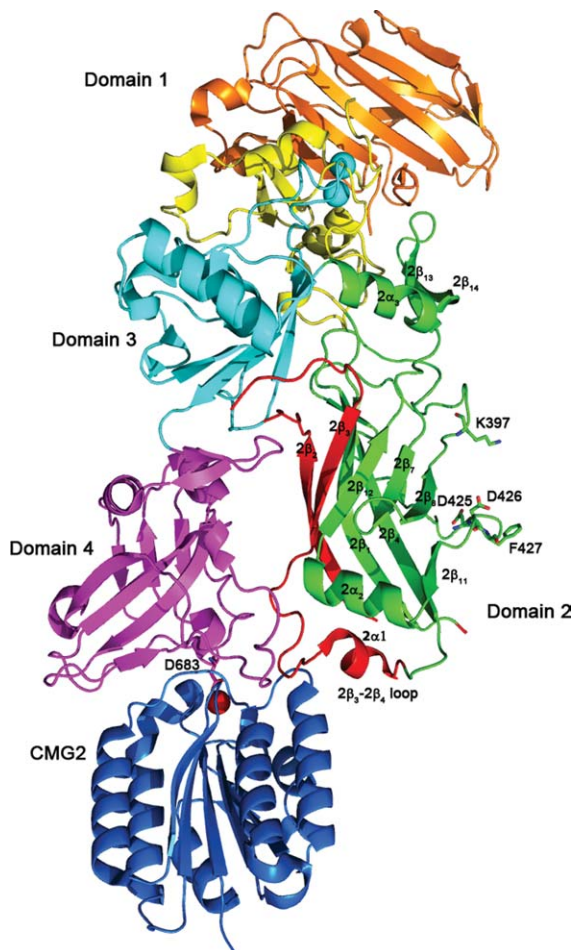


Figure 1. Structure of the PA bound to CMG2 (PDB: 1T6B).³³ The PA₂₀ region of domain 1 is colored orange, with domain 1' shown in yellow, and calcium ions shown as blue spheres; domain 2 is green, and the region which comprises the β -barrel portion of the pore (residues 275–352) is shown in red; domain 3 is shown in cyan, and domain 4 is shown in magenta. D683 is shown from domain 4 coordinating a magnesium ion shown in red, which lies in the MIDAS site of CMG2 (shown in blue). The residues that form the phi (ϕ) clamp – K397, D425, D426, and F427 are also shown.

PA—Formation of the Prepore

After PA has bound to the receptor on the cell surface, PA undergoes proteolytic processing by a furin-like protease, which cleaves off the first 167 amino acids of PA within domain 1, termed PA₂₀.³⁹ This leaves behind domain 1' (residues 168–258) and the remaining 3 domains, constituting a 63 kDa fragment (termed PA₆₃). PA₆₃ can assemble on the surface of cells into a heptameric^{40,41} or octameric⁴² donut shaped structure called the prepore.

The crystallographic structure of the heptameric prepore [Fig. 2(A)] revealed the monomer–monomer contact interfaces which involve interactions that stretch between domain 1' to domain 4.^{34,41} Surprisingly, aside from the loss of PA₂₀, there is little change in the backbone conformation

between monomeric PA and the heptameric prepore. There is also little change in the structure of the prepore upon binding CMG2.³⁴ Consistent with the lack of change in backbone conformation in the free and bound states, studies on the kinetics of association of PA₆₃ into heptamers showed that the rate constants for association do not change upon binding to CMG2, indicating that receptor binding does not significantly influence the structure of the prepore.³⁸

Through mutagenesis studies, domain 3 has been shown to be particularly important for the formation of the prepore, where a single point mutation at position 512, D512A, largely blocked heptameric prepore formation.⁴³ Although the resolution of the crystal structure of the heptameric prepore is low (~ 3.6 Å),³⁴ the structure indicates that D512 lies at the beginning of a short α -helix, where the side-chain carboxyl of D512 is positioned to form an intramolecular hydrogen bond with the backbone amide NH of L514. The L514 NH also forms a hydrogen bond with the amide carbonyl of T240 in an adjacent α -helix from a neighboring PA₆₃ monomer, and this intermolecular helix-to-helix interaction may be an important determinant in stabilizing the prepore structure.

Although the form most well studied is the heptameric prepore ((PA₆₃)₇), in an elegant series of experiments PA was shown to be capable of forming octameric prepore ((PA₆₃)₈) complexes. Kintzer *et al.* used a combination of electron microscopy, mass spectrometry, crystallography, electrophysiology and other biochemical methods to show the presence of octamers both in solution and on cell surfaces.⁴² At 20°C the octamer species is only present at a very small percentage, with the predominate form being the heptamer. However, shifting the temperature to 37°C causes the heptamers to aggregate, while the octamers remain soluble.

Once the prepore has formed, binding sites are created for EF and LF, which bind within the region of domain 1'.^{44–46} In studies by Ezzell, and later by Panchal, cleavage of PA to PA₆₃ has been shown to occur in the bloodstream, where presumably the prepore forms and binds to either the EF or LF components.^{47,48} In fact, their studies indicate that very little to no monomeric PA (83 kDa) exists in the bloodstream after infection, and only the PA₆₃ fragment is present. In contrast, studies by Moayeri and coworkers showed that in rats injected with PA, very little cleavage of PA occurs after 6 hours within the bloodstream, but in mice injected with PA, cleavage of PA to PA₆₃ was fast (within 5 min).⁴⁹ Whether the PA₆₃ observed represents the heptameric or octameric prepore is not known. However, since the heptameric structure forms SDS-resistant oligomers at pH values below 7.5 and is unstable at 37°C,^{34,50,51} conditions which favor the octameric prepore, the

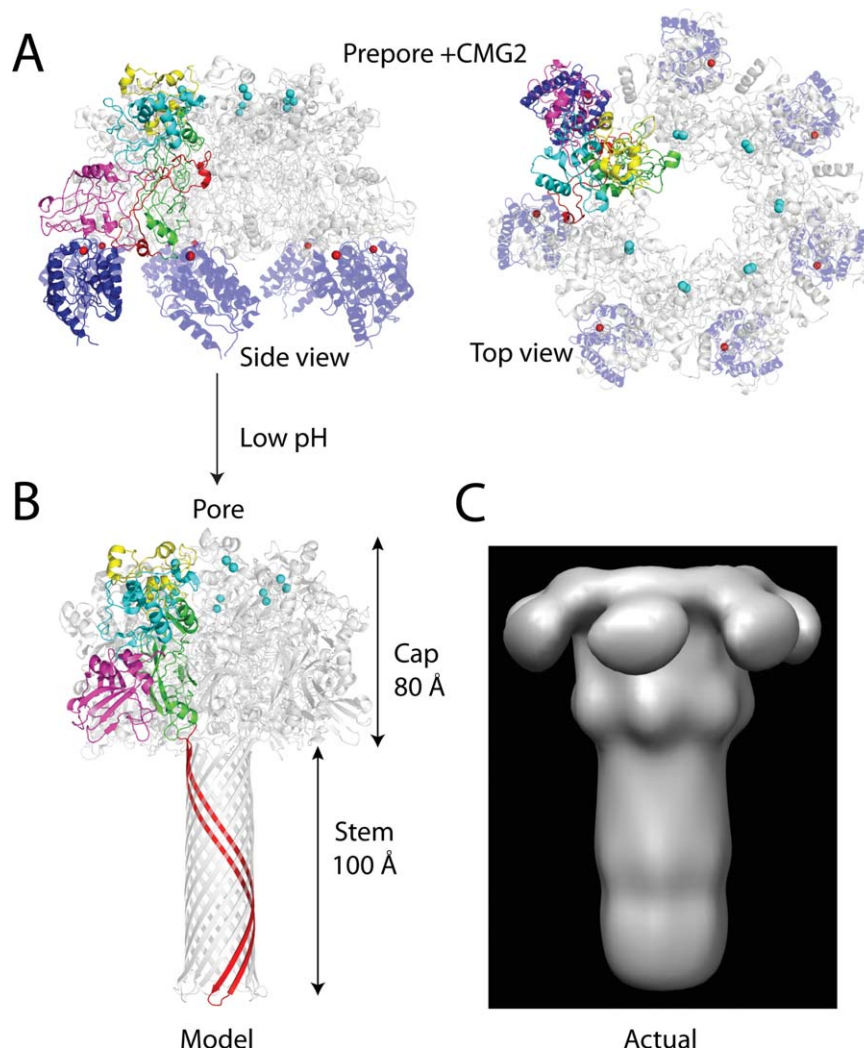


Figure 2. (A) Prepore and pore structures. Heptameric prepore bound to CMG2 (PDB:1TZN),³⁴ with color coding identical to Figure 1 (but without PA₂₀). Only one subunit of the heptamer is colored, the remaining subunits are shown in gray. CMG2 is in blue. Shown are both a side view and a top view of the prepore. (B) The final pore structure as modeled by Nguyen,⁶⁵ with the long β-barrel stem from a single subunit stretching from residues 275-352 is shown in red. (C) EM structure of the pore.⁶⁶ Note the absence of domain 4, and the differences in the β-barrel stem from the model of Nguyen. Figure was generated using Chimera.⁶⁷

latter is more likely to be the form of PA₆₃ encountered in the studies by Ezzell, Panchal and Moayeri.⁵² Therefore, it may be that there are two potential mechanisms for cell entry of lethal and edema toxin. In one mechanism, a cell surface receptor binds to PA (which is stable at physiological pH and temperature), and is then cleaved by a furin-like protease to PA₆₃, which allows the heptameric or octameric prepore structure to form and subsequent binding of EF and LF; in a second mechanism, the prepore (either heptameric or octameric) is formed in the bloodstream, and the prepore, either alone or bound to EF or LF, binds to the receptor.⁵³ It has been suggested that the heptameric form of PA may bind cells more local to the site of infection, while the octameric form, which is more stable, may act at sites more distant from the point of infection.⁵²

Endocytosis

Once the prepore has formed and bound to the cell surface, the toxin is internalized by receptor-mediated endocytosis. Internalization occurs through regions of the plasma membrane called lipid rafts, which are rich in cholesterol and sphingomyelin.^{54–56} Endocytosis has also been shown to be mediated by clathrin, which is required for the uptake of other cellular proteins, most notably the low-density lipoprotein (LDL) by the LDL receptor.⁵⁷ However, a key difference is that while the LDL receptor is naturally present in lipid rafts, the toxin receptor only associates with lipid rafts after PA has bound to the receptor, and after proteolytic processing and formation of the prepore.^{54,55,58} Recent studies have also shown that the intracellular portion of the receptor TEM8 may interact directly with actin, where it changes the conformation of the vWA domain on the

extracellular side from a high affinity state to a low affinity state.⁵⁹ Conversely, interaction of PA with TEM8 dissociates the receptor from actin, presumably to allow oligomerization into the prepore and clustering into clathrin-coated pits.⁵⁶ Recent studies by Zornetta and coworkers also suggest a role for caveolae, also present in lipid rafts, to be important in cellular uptake of the toxin.⁶⁰ Once the toxin is endocytosed, it is trafficked to late endosomes, which become acidic.^{54,61}

Formation of the Membrane Spanning Pore—Identification of Domain 2 as the Major Contributor to the Structure of the Pore

Within the acidified endosome (pH ~5-6), the prepore undergoes a major conformational change to form a membrane spanning pore, but exactly how this process occurs is the subject of much study. The structure that is ultimately formed is an extended, 14-stranded β -barrel pore which is similar to that of the α -hemolysin pore from *Staphylococcal aureus*.⁶² Studies by Benson, and later extended by Nassi, using cysteine scanning mutagenesis and labeling with the membrane impermeable probe methanethiosulfonate ethyltrimethylammonium (MTS-ET), showed that the residues which comprise the “stem” of the pore extend from residue 275 to 352 (the red colored ribbons in Fig. 1).^{63,64} This includes the domain $2\beta_2$ - $2\beta_3$ strands, the $2\beta_2$ - $2\beta_3$ loop that includes the transmembrane region (residues 303-324), and the $2\beta_3$ - $2\beta_4$ loop. Based on these biochemical experiments and the structure of the α -hemolysin pore, Nguyen generated a model of the pore which was 180 Å in length (cap and β -barrel stem included), with the 14-stranded β -barrel stem extending 100 Å away from the 80 Å cap region [Fig. 2(B)].⁶⁵ Using negative stain electron microscopy (EM), Katayama and coworkers have provided the first direct images of the pore, bound to the chaperone GroEL or in lipid nanodiscs.^{66,68} The pore dimensions are similar to that predicted from the biochemical data of Nassi and Benson, which again shows the pore forming a ~100 Å long β -barrel stem [Fig. 2(C)]. However, the pore structure as observed by the EM analysis shows several differences from the model, including a constriction in the cap region, that is, near to where F427 is predicted to be located. This residue is conserved among PA homologs, and forms a key structure in the lumen of the pore called the phi (ϕ)-clamp. This residue is critical for translocation of EF and LF into the cell, and forms a leak-free seal around a translocating polypeptide chain.^{69,70} In addition, domain 4 was not resolved in the EM analysis, and is likely to be highly dynamic in the pore structure. In any case, domain 2 and possibly other regions within PA must undergo a major conformational change to form the extended β -barrel struc-

ture, and this is most evident from the EM studies.⁶⁸

To form the pore, the residues that comprise the domain $2\beta_2$ - $2\beta_3$ strands as well as the $2\beta_2$ - $2\beta_3$ and $2\beta_3$ - $2\beta_4$ loops linking these strands must peel away from the core of domain 2 (Fig. 1). As pointed out by Santelli and coworkers, there are many steric factors that impede this process. The domain $2\beta_3$ - $2\beta_4$ loop, which is bound in a groove on the surface of CMG2, must dissociate from the receptor to allow movement of the $2\beta_2$ - $2\beta_3$ strands away from the core of domain 2. Also inhibiting this process is domain 4, which is buttressed against domain 2. The crystal structure indicates that in order for pore formation to occur, domain 4 must at least partially move away from domain 2.^{33,50}

How does this process work? What is the initiating step? Because the prepore to pore conversion process can occur within the pH range of ~7-7.5, the earliest hypotheses suggested that pore formation occurred as a consequence of the protonation of histidine residues.⁷¹ Histidine has a side-chain pKa of ~6-6.5, which is dependent on local environment. There are 10 histidine residues in PA, five of which are located in domain 2, and four of these five reside in the β strands that comprise the β -barrel of the pore. Thus, protonation of one or more of these histidines was postulated to initiate the conformational change to a pore. However, in an effort to determine the role of histidine protonation in the process of pore formation, experiments have been carried out in which PA was labeled biosynthetically with the histidine analog 2-fluorohistidine (2-FHis), which has a side-chain pKa of ~1⁷² and should resist protonation at pH values required for pore formation (~5). In these studies, no difference was observed in the pH values required to form a pore between the WT and 2-FHis-labeled proteins.⁷³ In addition to these experiments, Mourez and coworkers used cysteine scanning mutagenesis to determine which residues in PA were critical for function, making 568 individual mutations to cysteine. The majority of the mutations that resulted in defects in cytotoxicity were localized to domain 2. While many of these mutations failed to convert the prepore to an SDS-resistant pore, none of the histidine mutations showed defects in cytotoxicity (except for H304C, which was not expressed).⁷⁴ Furthermore, before the mutagenesis study by Mourez, Miller and coworkers had shown that deletion of residues 302-325 from PA did not affect the pH required to form an SDS-resistant pore, suggesting that H304 and H310 are not part of the pH sensing mechanism.⁵⁰

The study by Mourez was carried out to identify key residues that specifically prevent the prepore to pore conversion process and translocation and that are dominant-negative, in which co-oligomerization of potentially a single defective mutant with WT

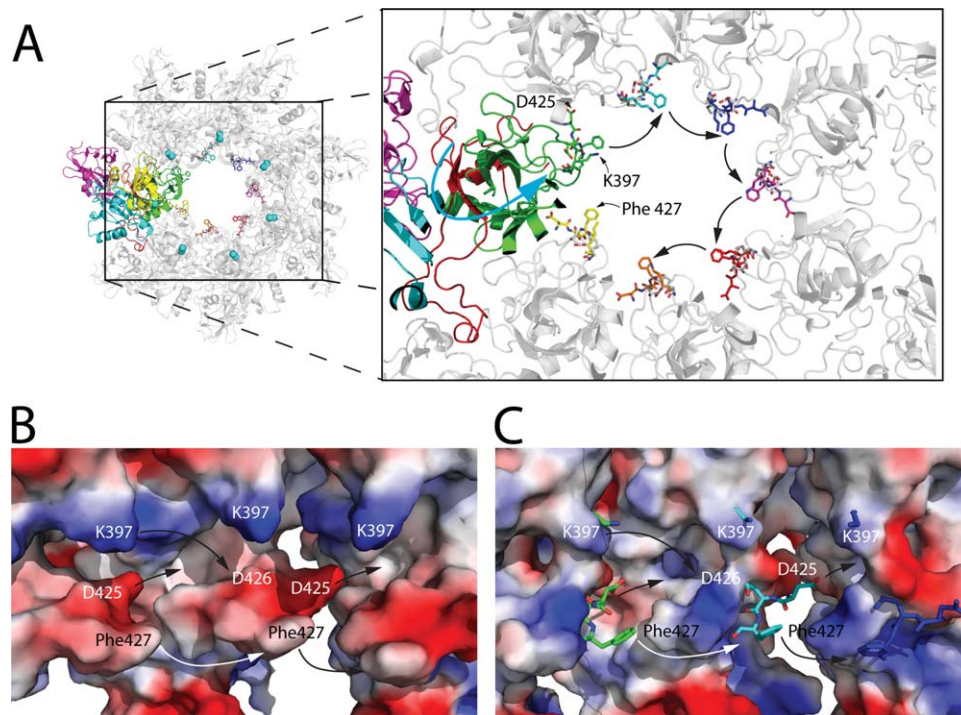


Figure 3. (A) Formation of the ϕ -clamp. Structure of the heptameric prepore (PDB:1TZO),³⁴ with color coding according to Figure 2(A), except that CMG2 is absent. Notice that there are additional residues represented as colored sticks that are within the lumen of the prepore. These residues are from the structure of the monomer of PA bound to CMG2 (PDB: 1T6B), and was generated by creating a structural alignment between PA bound to CMG2 and heptameric prepore. The alignment was carried out using the SSM superpose algorithm in Coot (v 0.1.6).⁷⁷ Additionally, we show K397 and D425 from 1TZO, represented as gray sticks. Note that the domain 2 β_2 -2 β_3 strands are on the opposite side of where K397, D425, D426, and F427 are located. The counter-clockwise movement of the subunit proposed here is indicated by the blue arrow. Residues that correspond to 1T6B and 1TZO are indicated. (B) Vacuum electrostatic potential surface of a heptamer comprised of solely 1T6B molecules, generated by the Coot alignment. This is purely a hypothetical model showing the potentials created by K397, D425, D426, and F427 together. In C, the vacuum electrostatic potential generated from the crystal structure of the heptameric prepore 1TZO is shown. For contrast, residues K397, D425, D426, and F427 from the electrostatic potential map in (B) are shown as sticks in C. In both (B) and (C), movement of the subunits together is indicated by the arrows.

PA₆₃ monomers (ratio of 1 mutant: 6 WT) prevents one or both of these processes from occurring.⁷⁴ Prior to that study, Sellman and coworkers had identified three key conserved residues (K397, D425, and F427), that upon mutation led specifically to either a block in pore formation (K397A and K397D; D425A, D425N, D425E, and D425K mutations) or a block in translocation (F427A).^{75,76} Also, if these mutant proteins are allowed to co-oligomerize with the WT PA₆₃, the resulting prepores are rendered non-functional, indicating that the mutations have a dominant negative effect on the prepore structure.⁷³ K397 is located in a loop linking domain 2 β -strands 7 and 8 (2 β_7 -2 β_8 loop), while D425 and F427 are located in a loop linking domain 2 β -strands 10 and 11 (2 β_{10} -2 β_{11} loop). Importantly, these strands are *not* part of the domain 2 β_2 -2 β_3 strands that form the β -barrel portion of the pore, and in fact are on the opposite side of domain 2 facing the lumen of the prepore (see Figs. 1 and 3). Conservative mutations of D425 to either asparagine or glutamate (isosteric and isoelectric changes) block low pH-induced pore

formation, and as a consequence translocation of LF_N, the N-terminal PA binding domain of LF, through the pore. Importantly, studies by Sellman and coworkers showed that mutations at D425 and K397 did not change the ability of PA to bind to cells or form prepores, or alter the cleavage pattern by trypsin to form PA₆₃, strongly suggesting that these mutations did not affect proper folding (although local effects on structure could not be resolved).

These studies have been extended recently by Janowiak and coworkers, who developed a clever strategy to isolate single-subunit mutant containing heptameric prepore molecules (1:6, mutant:WT).⁷⁸ In that study, a biotinylated form of the D425A or F427A mutant was co-oligomerized with an excess (~20-fold) of WT PA, and purified using avidin affinity chromatography. Under these conditions, the ratio of 1 mutant to 6 WT monomers per prepore was confirmed, and the D425A single mutation was shown to block low pH (pH 5.5) induced pore formation, highlighting again the importance of this residue in the prepore to pore conversion process. The

F427A mutant had a marginal effect on pore formation, but largely blocked translocation of LF_N.

F427 has been shown to be particularly important in protein translocation as mentioned above, forming the ϕ -clamp.⁷⁰ However, F427 seems to also play an important role in the formation of the pore, if the right mutation at this site is made. For instance, mutation of F427 to Gly, Asp or Arg strongly inhibited low-pH induced pore formation.⁷⁹ While the specific mechanism for how F427 may facilitate pore formation has yet to be resolved, these authors suggest that pi-stacking interactions between adjacent F427 rings may be a driving force for the formation of the pore. Alternatively, Melnyk and Collier showed that K397 and D426 are important for translocation of LF_N across the membrane, and postulated that these residues form a conserved intermolecular salt bridge in the pore state that correctly orients F427 into a functional ϕ -clamp conformation.⁸⁰ Since K397 was identified in previous experiments by Sellman and coworkers to be critical for pore formation, perhaps the formation of this salt bridge is an additional driving force for the formation of the pore. In Figure 3(B,C), we have created a vacuum electrostatic map of the region including K397, D425, D426, and F427, showing the potential positioning of F427 in the lumen of the prepore. Since D426 and F427 are missing electron density in the structures of the heptamer (PDB:1TZO (without CMG2) and 1TZN³⁴), we have added these residues as sticks by overlaying the structure of the monomeric form of PA bound to CMG2 (PDB:1T6B, which has electron density for D426 and F427) and 1TZO. Clearly, there are cavities and room for movement of the side-chains closer together in the pore state.

pH Changes in the Structure of the Monomer as Clues to Regions Sensitive to pH

We know that the consequence of lowering the pH on the prepore structure is pore formation, but what is the consequence of lowering the pH on the monomeric PA structure? Again, aside from PA₂₀, the structures of PA and the prepore overlay quite well, so it may be that there are regions that are sensitive to pH in the monomer structure that provide clues as to which regions in the prepore are also sensitive to pH. Interestingly, regions within the monomeric form of PA have been identified as sensitive to pH using X-ray crystallography, and include the domain 2 β_3 -2 β_4 loop that binds to the receptor.⁴¹ The X-ray structure of PA at pH 7.5 and 6 showed that the 2 β_3 -2 β_4 loop (Fig. 1) becomes disordered at pH 6. The crystal structure of a homolog of PA, C2-II toxin, solved at pH 4.3, showed little change in the overall structure compared to PA (at pH 6), except that the electron density for domain 4 and for the domain 2 β_3 -2 β_4 loop was missing.⁸¹ These studies

indicate that while the overall structure of the monomer is relatively stable to variations in pH, the structure of the domain 2 β_3 -2 β_4 loop is sensitive to pH, and may aid in facilitating the prepore to pore conversion.

Effect of the Receptor on the Prepore to Pore Conversion

Aside from being a part of the pore, the domain 2 β_3 -2 β_4 loop has been shown to play a key role in determining the pH requisite for pore formation. Previous studies by Miller and coworkers have shown that if pore formation is monitored on the surface of CHO-K1 cells rather than in solution, the pH requisite for pore formation is ~ 1 pH unit less (pH ~ 6).⁵⁰ Lacy and coworkers showed that binding of the vWA domain of the receptor CMG2 to the heptameric prepore lowers the pH requisite for pore formation, from a pH of 7.5 in the absence of the receptor to a pH of ~ 5 -6 when bound to the receptor.³⁴ Importantly, mutation of residues in CMG2 which destabilize intermolecular interactions with the domain 2 β_3 -2 β_4 loop result in higher pH values required for pore formation, indicating that this loop plays an important role in dictating the pH requisite for pore formation.^{36,82} Using histidine hydrogen-deuterium exchange (HDX), binding of the receptor reduces the rate of hydrogen exchange for most of the histidine residues in monomeric PA, in particular domain 2, including one (H299), that is, distant (~ 40 Å) from the binding interface.⁸³ This suggests that receptor binding reduces the dynamics of PA, and may strengthen hydrogen bonding interactions throughout the protein, aiding to stabilize the protein against variations in pH.

As mentioned above, the prepore needs to at least partially dissociate from the receptor to sterically allow pore formation to occur. Studies by Rainey and coworkers showed using immunoprecipitation that both CMG2 and TEM8 dissociate from the prepore at pH values concomitant with pore formation (pH ~ 5 for CMG2, and pH ~ 6 for TEM8).⁸⁴ Receptor dissociation at low pH was also supported by NMR studies, but interestingly the receptor did not release from the monomeric form of PA at low pH, only the heptameric prepore form. It was argued that since the binding interfaces between monomer and heptameric prepore change very little upon receptor binding, it is unlikely that the protonation of residues within the receptor is the cause of low-pH induced receptor dissociation.⁸⁵ Furthermore, studies on the isolated domain 4 indicate that the receptor remains bound at pH values that are required for pore formation.³⁷ Taking into account all of these observations, it was argued that receptor release may *only* occur as a consequence of pore formation.⁸⁵

Additional evidence that receptor release occurs as a consequence of pore formation has come from

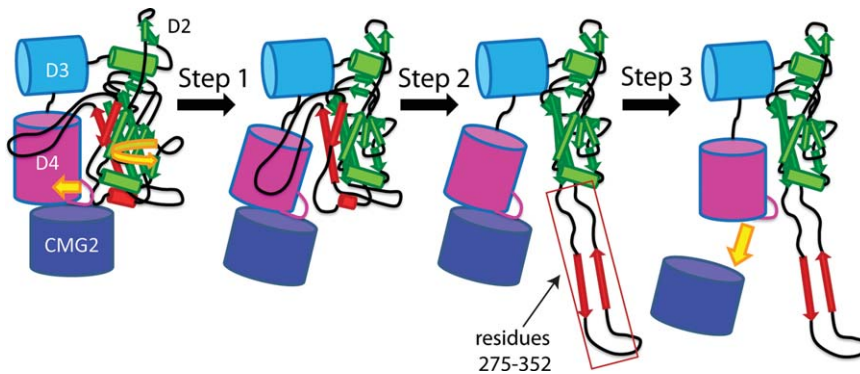


Figure 4. Model of pore formation. In STEP 1, low pH triggers the formation of the ϕ -clamp, which forces a counter-clockwise movement of the subunit and partial receptor dissociation. In STEP 2, the domain $2\beta_2$ - $2\beta_3$ strands (comprising residues 275 to 352) peel away from the core of domain 2. In STEP 3, the receptor dissociates.

recent studies using the 2-FHis-labeled PA, but conducting experiments in the presence of the vWA domain of CMG2. As mentioned, the 2-FHis-labeled prepore is able to form pores at pH values identical to the WT protein. However, when the 2-FHis-labeled prepore is bound to the vWA domain of CMG2, pore formation is largely blocked (a small percentage of pores could form, at pH values similar to that of the WT prepore bound to CMG2).⁸³ Using NMR, it was shown that the receptor did not release at low pH, suggesting that the inability to form a pore was related to an inability to release the receptor. It was argued that there is a structural tightening of the prepore (reduced dynamics) when the receptor binds,⁸⁶ and that this tightening strengthens hydrogen bonds to the imidazole nitrogens of the histidine residues. Because the 2-FHis residues have a low pKa (~ 1), protonation is more difficult, and thus hydrogen bonds to the 2-FHis imidazole nitrogens are likely to be more stable.⁸³ It was suggested, based on that study, that histidine protonation is a trigger for pore formation, but only when bound to the receptor. However, given the studies by Sellman and others, it is unlikely to be the initiating step (see below). Thus, it is hypothesized that the receptor prevents conformational changes that destabilize the prepore to pH, including the conformational changes observed in the domain $2\beta_3$ - $2\beta_4$ loop, which, as mentioned is a region that is sensitive to pH.

A Proposed Model for Pore Formation

Based on these cumulative observations, we propose the following kinetic steps for anthrax prepore to pore conversion, which we show in Figure 4, beginning with a subunit of the heptameric prepore (for descriptive reasons) bound to CMG2. STEP 1, at low pH, residues which include K397, D425, D426, and F427, form a narrow iris which becomes the ϕ -clamp. This is based largely on the observations by Sellman, Mourez, Sun and Janowiak and coworkers, that mutation of D425, K397 or, under certain instances F427, blocked pore formation in the ab-

sence of the receptor, and in the absence of folding defects, indicating that independent of receptor binding, changes in structure surrounding these residues must occur to initiate pore formation. This structure is stabilized by the formation of favorable intermolecular salt bridges,^{76,80} or pi-pi interactions between adjacent Phe427 rings.⁷⁹ The movement toward the formation of the ϕ -clamp would cause rotation of the individual subunits in a counter-clockwise rotation (Fig. 3), and provides a driving force such that the domain 2-domain 4 interface is shifted away from one another, allowing the domain $2\beta_3$ - $2\beta_4$ loop to partially dissociate from the receptor. Both processes (formation of the ϕ -clamp and dissociation of the domain 2 $2\beta_3$ - $2\beta_4$ loop) would likely occur concomitantly.

Early formation of the ϕ -clamp would play an important functional role, since the ϕ -clamp is expected to form a leak-free seal around the N-terminal regions of EF and LF. Formation of the ϕ -clamp as an initial step, followed by insertion of the N-terminal domain of either EF or LF,⁸⁷ but before the formation of the transmembrane β -barrel, would prevent the flow of ions (H^+) from the acidic endosome into the more basic cytosol, preventing dissipation of the gradient required for translocation.⁶⁹ In addition, it is known that elimination of domain 4 from domain 2 changes the pH requisite for pore formation (in the absence of receptor) to pH > 8 , indicating that domain 4 stabilizes the prepore to variations in pH.⁶⁸ Partial dissociation from domain 4 and from the receptor allows more steric freedom of movement such that the domain $2\beta_2$ - $2\beta_3$ strands can dissociate from the core of domain 2.

STEP 2, the domain $2\beta_2$ - $2\beta_3$ strands peel away from the core of domain 2 like a banana peeling from the fruit. Based on our studies using the 2-FHis-labeled prepore, this process is likely to be facilitated by histidine protonation, but only when bound to the receptor. Preliminary mutagenesis experiments indicate that at least four of the five histidine residues in domain 2 must be mutated

together to prevent pore formation, and as with the 2-FHis-labeled prepore, these combined mutations only prevent pore formation when bound to the receptor (unpublished observations).

STEP 3, the receptor dissociates. In our NMR studies, CMG2 can remain bound to the isolated domain 4 at pH values concomitant with pore formation,³⁷ and thus it may be that pore formation induces a structural change in domain 4 that causes receptor release. Alternatively, it may be that once the pore has formed, domain 4 is sterically prevented from associating with the receptor, perhaps by partially filling the void left by the movement of the domain 2 β_3 -2 β_4 strands. Once the β strands have peeled away from domain 2, they come together spontaneously to form the β -barrel stem and transmembrane pore [Fig. 2(C)], the barrel inserts into the membrane, and the pore is complete.

Although experimental evidence for the stepwise formation of the pore is lacking, the fact that mutation of residues that form the ϕ -clamp prevents pore formation even when the receptor is not bound would indicate that the formation of the ϕ -clamp is an initiating step in the process. Alternatively, it may be that the domain 2 β_3 -2 β_4 loop, which is also sensitive to pH, dissociates from the surface of CMG2 and this allows freedom of movement such that the ϕ -clamp may form. In either case, if the ϕ -clamp does form before the unfurling of the β -strands from domain 2, it would support a more general mechanism in which the ion-selectivity filter region of a pore forms initially before acquisition of the ion-conducting pore state. Such is the case for the voltage-gated potassium channel Kv1.3, for instance, where the reentrant pore conformation is attained before oligomerization into the tetramer and formation of a folded functional channel.⁸⁸

Once the pore has formed, EF and LF must unfold their three-dimensional structures for passage through the pore into the cell.^{69,89} The pore can form independent of binding to EF or LF, and is the likely cause of cytotoxicity in the absence of the enzymatic components.^{90,91} Although as yet there is no direct evidence that EF and LF translocate through the pore into the cytosol, introduction of disulfide bonds into LF blocks translocation.⁹² In addition, recent studies by Zornetta and coworkers utilizing GFP fusions between EF or LF, have shown that the movement of the fusion protein into the cytosol is dependent on the ability of the fusion protein to efficiently unfold.⁶⁰ The recent structure of the prepore form of the octamer, bound to LF_N, has provided key insight into how the prepore interacts with the enzymatic components⁴⁶ (for a recent excellent review, see Thoren and Krantz⁹³). Advancements along this line continue to be made in deciphering not only the mechanisms of translocation of EF and LF, but for providing much needed knowledge on the general

mechanisms of protein unfolding and translocation through membranes.

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