

RESEARCH PAPER

# A pharmacological analysis of high-affinity sodium transport in barley (*Hordeum vulgare* L.): a $^{24}\text{Na}^+ / ^{42}\text{K}^+$ study

Lasse M. Schulze, Dev T. Britto, Mingyuan Li and Herbert J. Kronzucker\*

Department of Biological Sciences, University of Toronto, 1265 Military Trail, Toronto, ON, Canada M1C 1A4

\* To whom correspondence should be addressed. E-mail: herbertk@utsc.utoronto.ca

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## Abstract

Soil sodium, while toxic to most plants at high concentrations, can be beneficial at low concentrations, particularly when potassium is limiting. However, little is known about  $\text{Na}^+$  uptake in this 'high-affinity' range. New information is provided here with an insight into the transport characteristics, mechanism, and ecological significance of this phenomenon. High-affinity  $\text{Na}^+$  and  $\text{K}^+$  fluxes were investigated using the short-lived radiotracers  $^{24}\text{Na}$  and  $^{42}\text{K}$ , under an extensive range of measuring conditions (variations in external sodium, and in nutritional and pharmacological agents). This work was supported by electrophysiological, compartmental, and growth analyses.  $\text{Na}^+$  uptake was extremely sensitive to all treatments, displaying properties of high-affinity  $\text{K}^+$  transporters,  $\text{K}^+$  channels, animal  $\text{Na}^+$  channels, and non-selective cation channels.  $\text{K}^+$ ,  $\text{NH}_4^+$ , and  $\text{Ca}^{2+}$  suppressed  $\text{Na}^+$  transport biphasically, yielding  $\text{IC}_{50}$  values of 30, 10, and  $<5 \mu\text{M}$ , respectively. Reciprocal experiments showed that  $\text{K}^+$  influx is neither inhibited nor stimulated by  $\text{Na}^+$ . Sodium efflux constituted 65% of influx, indicating a futile cycle. The thermodynamic feasibility of passive channel mediation is supported by compartmentation and electrophysiological data. Our study complements recent advances in the molecular biology of high-affinity  $\text{Na}^+$  transport by uncovering new physiological foundations for this transport phenomenon, while questioning its ecological relevance.

**Key words:** Barley, compartmental analysis, electrophysiology, high-affinity  $\text{Na}^+$  uptake, ion transport, pharmacology, potassium, radiotracers, salinity, sodium.

## Introduction

The kinetics and mechanism of sodium transport into plant roots were first investigated in detail by Rains and Epstein (1965, 1967a, b), who demonstrated that transport of  $\text{Na}^+$  into roots of barley (*Hordeum vulgare* L.) has a biphasic character. Sodium fluxes at low external  $\text{Na}^+$  concentrations (typically,  $[\text{Na}^+]_{\text{ext}} < 1 \text{ mM}$ ) were shown to be mediated by a high-affinity transport system (HATS) that was suppressible by both  $\text{Ca}^{2+}$  and  $\text{K}^+$ , influenced by the counterion (e.g.  $\text{Cl}^-$ ,  $\text{F}^-$ ,  $\text{SO}_4^{2-}$ ), and affected by temperature and a small range of metabolic inhibitors. At higher  $[\text{Na}^+]_{\text{ext}}$ , including the range at which plants encounter sodium toxicity, a low-affinity system (LATS), was shown to operate. This type of biphasic flux pattern has been well documented in plants with respect to other important ions, including potassium (Szczerba *et al.*, 2009), ammonium (Jackson *et al.*, 2008), and nitrate (Jackson *et al.*, 2008).

Largely because of the urgent agronomic importance of salinity stress, the high-affinity system received little attention relative to the LATS for three decades following these discoveries (Rodríguez-Navarro and Rubio, 2006). Nevertheless, at low to moderate concentrations, sodium is commonly found to be a benign and even beneficial element (Marschner, 1995; Subbarao *et al.*, 2003), which can stimulate growth in many plant species, particularly when individuals are K-deprived (Hawker *et al.*, 1974; Marschner, 1995; Subbarao *et al.*, 2003; Horie *et al.*, 2007). In addition, sodium is a required micronutrient for the growth of many plant species possessing the  $\text{C}_4$  photosynthetic pathway (Brownell and Bielig, 1996). Moreover, the understanding of high-affinity  $\text{Na}^+$  transport may provide important clues as to the nature of  $\text{Na}^+$  fluxes in the toxic range (Kronzucker and Britto, 2011).

Interest in the sodium HATS has been substantially revived in recent years with the cloning and expression of genes belonging to the *HKT* family. Initially believed to represent high-affinity  $K^+$  transporters (Schachtman and Schroeder, 1994), HKTs now appear to be more closely involved with  $Na^+$  transport in both high- and low-affinity ranges, and may function either via a  $Na^+$  uniport mechanism or by  $K^+/Na^+$  symport (note: the high functional diversity of HKT transporters should be kept in mind; see reviews by Rodríguez-Navarro and Rubio, 2006; Huang *et al.*, 2008; Munns and Tester, 2008; Corratge-Faillie *et al.*, 2010; Kronzucker and Britto, 2011). Consistent with both the sodium–potassium inhibition studies of Rains and Epstein (1965, 1967b), and with the role proposed for  $Na^+$  as an ion that can partially substitute for  $K^+$ , it has been shown that expression of *HKT* genes, and HKT mediation of  $Na^+$  transport into plant roots, are suppressed by the presence of  $K^+$  in solution (Wang *et al.*, 1998; Garcíadeblás *et al.*, 2003; Horie *et al.*, 2007; Haro *et al.*, 2010). Conversely, in several plant species, the enhancement of *HKT* mRNA levels has been observed in response to  $K^+$  starvation (Wang *et al.*, 1998; Laurie *et al.*, 2002; Garcíadeblás *et al.*, 2003; Mian *et al.*, 2011; also see Buschmann *et al.*, 2000).

While ion channels of varying classes, especially non-selective cation channels (NSCCs) have been strongly implicated in the transport of  $Na^+$  into plant roots in the toxic range (Munns and Tester, 2008; Kronzucker and Britto, 2011), their contribution to high-affinity  $Na^+$  fluxes should not be ruled out. Although channels are generally considered to catalyse fluxes from millimolar concentrations, it has been demonstrated that the  $K^+$  channel AKT1 can transport  $K^+$  into root cells from external concentrations as low as 10  $\mu$ M (Hirsch *et al.*, 1998; Spalding *et al.*, 1999). Conversely, HKT transporters have been reported to display channel-like properties (Lan *et al.*, 2010; Xue *et al.*, 2011), and to have evolved from bacterial  $K^+$  channels (Durell *et al.*, 1999), which partially explains their ability to operate in the low-affinity range of  $Na^+$  transport, not just the high-affinity range (Munns and Tester, 2008).

Despite these and other recent advances, the physiological basis and significance of high-affinity  $Na^+$  transport system in higher plants is still not well understood (Rodríguez-Navarro and Rubio, 2006; Haro *et al.*, 2010), and some fundamental questions about the system's characteristics remain open. These questions fall into two major areas of interest: (i) the kinetics and mechanism of this transport system, and (ii) its nutritional and ecological significance. In the present work, existing knowledge of the physiological and ecophysiological foundations of high-affinity  $Na^+$  transport in these two areas has been extended, chiefly by use of tracer-flux, electrophysiological, and compartmental analyses in the model system barley.

## Materials and methods

### Plant culture

Barley seeds (*Hordeum vulgare* L. cv. 'A.C. Metcalfe') were surface-sterilized for 10 min in 1% sodium hypochlorite followed

by a 3 h rinse with distilled water. Seeds were germinated for 3 d in acid-washed sand prior to placement for 4 d in 12 l aerated hydroponic vessels, containing either: simple solutions of NaCl (100  $\mu$ M) or CaSO<sub>4</sub> (200  $\mu$ M, removed prior to influx measurement), or a modified, complete Hoagland's solution: 5 mM KNO<sub>3</sub>, 0.25 mM KH<sub>2</sub>PO<sub>4</sub>, 2 mM MgSO<sub>4</sub>, 0.1 mM Ca(NO<sub>3</sub>)<sub>2</sub>, 0.2  $\mu$ M Na<sub>2</sub>MoO<sub>4</sub>, 50  $\mu$ M FeSO<sub>4</sub>, 70  $\mu$ M H<sub>3</sub>BO<sub>3</sub>, 14  $\mu$ M MnCl<sub>2</sub>, 1  $\mu$ M ZnSO<sub>4</sub>, 0.5  $\mu$ M CuSO<sub>4</sub>, and 100  $\mu$ M NaCl. The pH was adjusted to 6.3–6.4 by adding KOH or H<sub>2</sub>SO<sub>4</sub>. To prevent nutrient depletion, solutions were exchanged on days 3 and 5. Plants were cultured in climate-controlled walk-in growth chambers under fluorescent lights with an irradiation, at plant height, of 200  $\mu$ mol photons m<sup>-2</sup> s<sup>-1</sup> (Philips Silhouette High Output F54T5/850HO; Philips Electronics Ltd., Markham, ON, Canada). A 16 h photoperiod with day and night temperatures of 20 °C and 15 °C, respectively, was used. The relative humidity was 60%. For flux experiments, plants were bundled in groups of 6–8 at the base of the shoot with a plastic collar, 0.5 cm in height, one day prior to measurement.

### Flux analysis

Unidirectional influx was determined over a 5 min uptake period, as described in detail previously (Kronzucker *et al.*, 1999; Szczerba *et al.*, 2006). In brief, roots of intact, bundled seedlings were immersed in 240 ml labelling vessels containing growth solution and one of two  $\gamma$ -emitting radioisotopes, <sup>24</sup>Na<sup>+</sup> or <sup>42</sup>K<sup>+</sup> (McMaster University Nuclear Reactor, Hamilton, ON, Canada), followed by a 5 s dip and 5 min desorption in non-radioactive growth solution. Unless otherwise noted, flux experiments were conducted with plants grown in simple solutions of 100  $\mu$ M NaCl. For the experiment presented in Table 2, inhibitors were applied 10 min prior to the uptake step, and over its duration, at the following concentrations: KCN+SHAM (10 mM each); KCl (5 mM); (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (5 mM); CaCl<sub>2</sub> (10 mM); tetraethylammonium chloride (TEACl, 10 mM); BaCl<sub>2</sub> (5 mM); CsCl (10 mM); LaCl<sub>3</sub> (50  $\mu$ M); GdCl<sub>3</sub> (50  $\mu$ M and 5 mM); quinine (1 mM), lidocaine (as lidocaine hydrochloride monohydrate, 10 mM), and dopamine (as dopamine hydrochloride, 5 mM). Chemicals were obtained from Sigma-Aldrich Co. (St Louis, MO, USA) and Alfa Aesar (Ward Hill, MA, USA). In the pH 4.2 treatment, H<sub>2</sub>SO<sub>4</sub> was used to lower the pH, while the 4 °C condition was maintained by immersion of uptake vessels in an ice water bath. Concentration of inhibitors and exposure times were chosen according to published protocols (Demidchik *et al.*, 2002; Essah *et al.*, 2003).

Compartmental analysis by tracer efflux was conducted as previously published (Siddiqi *et al.*, 1991; Malagoli *et al.*, 2008), with minor modifications. In brief, roots of intact, bundled seedlings were first immersed for 1 h in growth medium containing the radiotracer <sup>24</sup>Na<sup>+</sup>. Seedlings were then attached to efflux funnels, in which roots were eluted of radioactivity by a series of 36 aliquots of non-labelled growth solution. The desorption series was timed as follows: 15 s ( $\times$ 4), 20 s ( $\times$ 3), 30 s ( $\times$ 2), 40 s ( $\times$ 2), and 60 s ( $\times$ 25). Solutions were continuously mixed and aerated by a fine stream of air bubbles. In efflux protocols involving amiloride treatment, the amiloride was applied 19.5 min into the desorption series, as amiloride hydrochloride (0.1 mM; Cuin *et al.*, 2011).

In all flux protocols, plant roots were detached from shoots immediately after labelling and desorption, and spun in a low-speed centrifuge for 30 s to remove surface water prior to weighing. Radioactivity in all plant parts and eluates was measured by gamma counting using one of two  $\gamma$ -counters (Canberra-Packard Quantum Cobra Series II, model 5003, Packard Instrument Co., Meriden, CT, USA; PerkinElmer Wallac 1480 Wizard 3<sup>™</sup>, Turku, Finland), and corrected for isotopic decay.

### Tissue Na<sup>+</sup> content

To determine Na<sup>+</sup> tissue content, roots of 7-d-old barley seedlings, grown in simple solutions of 0.01, 0.1, 1, 5, 10, 25, or 50 mM

NaCl, were desorbed for 5 min in 10 mM CaSO<sub>4</sub>, to remove excess Na<sup>+</sup> from the apoplast. Shoots and roots were separated, weighed, and oven-dried for a minimum of 3 d in a drying oven at 80–85 °C, then reweighed, pulverized, and acid-digested with 30% nitric acid for three more days. Na<sup>+</sup> tissue content was measured with a single-channel flame photometer (Digital Flame Analyzer model 2655-00; Cole Parmer; Cole-Parmer, Anjou, QC, Canada).

#### Electrophysiological measurements

All electrical measurements were performed in a grounded Faraday cage, using intact roots of plants grown in a simple solution of 100 μM NaCl. A single seminal root of a seedling (aged 7–8 d) was isolated and positioned over platinum pins in a narrow Plexiglas chamber initially containing 100 μM NaCl. The seedling's other roots were immersed in a sub-chamber adjacent to and continuous with the chamber containing the isolated root. This apparatus (total volume=125 ml) was installed on the stage of an inverted Leica DME microscope (Leica Microsystems Inc., Concord, Ontario, Canada). Impalements were made with a micromanipulator (SD instruments; MX310R, Siskiyou Corporation, Grants Pass, Oregon, USA) under 100× magnification. The root was secured by small foam-rubber plugs to avoid movement during impalement or solution exchange. Impalements were made with an electrode in a grounded, aerated bathing solution, 2–3 cm from the root tip. A reference electrode was placed adjacent to the impalement. Borosilicate glass pipettes (I.D.=0.75 mm; O.D.=1.0 mm) from World Precision Instruments Inc. (World Precision Instruments Inc., Sarasota, Florida, USA) were used to make microelectrodes (tip diameter <1 μm) with an electrode puller (Sutter Instrument Co.; P-30). The microelectrodes ( $R=14.6\pm 0.5$  MΩ) were filled with 3 M KCl solution (pH=2 to reduce tip potential). Both the impaling and the reference electrodes were made in this manner. A solenoid coil (World Precision Instruments Inc. Sarasota, Florida, USA) was positioned around the reference electrode to reduce electrical interference. Membrane potential measurements were made with an electrometer (World Precision Instruments Inc.; Duo 773) and recorded on an oscilloscope (Tektronix; TDS2002B, Tektronix Inc., Beaverton, Oregon, USA). Once a steady reading was obtained, treatment solution was perfused through 1/16-inch Tygon tubing via a peristaltic pump at a rate of approximately 7.5 ml min<sup>-1</sup>. Each treatment solution consisted of a single salt in a background of 0.1 mM NaCl. The salt concentrations of individual treatments were 30 μM KCl, 50 μM KCl, 5 μM (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, 30 μM (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, 5 μM CaCl<sub>2</sub>, or 30 μM CaCl<sub>2</sub>.

#### Statistics

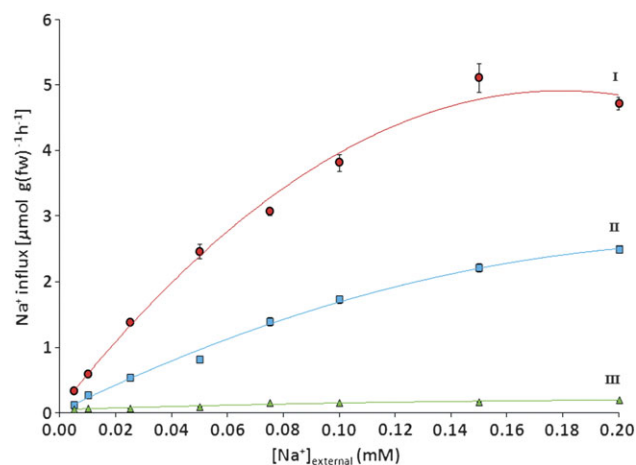
For flux measurements, each plant bundle was treated as a single replicate. Direct influx measurements were compared by ANOVA followed by the Tukey *post-hoc* test. Traces of efflux runs following amiloride application were compared with control traces using a Student's *t* test. Statistical values and Michaelis–Menten parameters were estimated by the use of GraphPad Prism statistical software (v. 5.01).

## Results

The sodium-concentration dependence of unidirectional Na<sup>+</sup> influx was measured in plants grown under three conditions, using a 5 min <sup>24</sup>Na uptake protocol (Fig. 1). Maximal influxes were observed in plants that were grown only with CaSO<sub>4</sub> (200 μM), which was withdrawn 10 min prior to tracer application. By contrast, the fluxes were suppressed by more than 90% when Ca<sup>2+</sup> was not withdrawn (not shown). The suppressed fluxes were considerably lower than those observed by Rains and Epstein (1967a) in the presence of

higher (500 μM) Ca<sup>2+</sup>, which only reduced Na<sup>+</sup> fluxes by about half compared to Ca<sup>2+</sup>-free conditions, at a representative [Na<sup>+</sup>]<sub>ext</sub> of 100 μM. Plants grown and measured with full nutrient solution showed the lowest fluxes, which were barely detectable, while plants grown at 100 μM NaCl transported sodium at an intermediate rate. For the three growth/measurement conditions, influx showed saturating patterns with increasing [Na<sup>+</sup>]<sub>ext</sub>, and Michaelis–Menten constants were determined (Table 1). In general, Na<sup>+</sup> fluxes shown in the upper two curves of Fig. 1 ('I' and 'II') were of a magnitude similar to those measured by other workers (Rains and Epstein, 1967a; Jeschke, 1982; Wrona and Epstein, 1985), and also to those of high-affinity K<sup>+</sup> fluxes into barley roots (Szczzerba *et al.*, 2006). In addition, the half-maximal saturation ( $K_m$ ) values in Table 1 were in good general agreement with those provided in Haro *et al.* (2010) for multiple species, including barley.

More detailed experiments were conducted with plants grown at 100 μM NaCl (curve 'II'), in the absence of inhibitory calcium. Table 2 illustrates the high malleability of unidirectional Na<sup>+</sup> influx in these plants, when exposed to 16 pharmacological, nutritional, and physicochemical treatments. In all cases, the flux of Na<sup>+</sup> was profoundly reduced, with inhibition ranging from 74% to 95%. The



**Fig. 1.** Concentration dependence of unidirectional, high-affinity Na<sup>+</sup> influx, measured in 7-d-old barley seedlings grown with 200 μM Ca<sup>2+</sup> (I, circles), 100 μM Na<sup>+</sup> (II, squares), or full nutrient medium (III, triangles). Values represent means ±SEM ( $n=4-6$ ).

**Table 1.** Michaelis–Menten parameters for high-affinity Na<sup>+</sup> influx, derived from concentration-dependence curves in Fig. 1: values represent mean ±SEM ( $n=4-6$ )

Concentration-dependence curve	Growth condition	$V_{max}$ (μmol g <sup>-1</sup> fw h <sup>-1</sup> )	$K_m$ (μM)
I	200 μM Ca <sup>2+</sup>	7.91	110
II	100 μM Na <sup>+</sup>	5.34	219
III	Full medium	0.22	36



**Table 2.** Effects of pharmacological, physical, and nutritional treatments on high-affinity Na<sup>+</sup> influx in intact barley seedlings grown at 100 μM, and measured at 10 and 100 μM Na<sup>+</sup>. Treatments were applied 10 min prior to, and during, the 5 min labelling step, dip (5 s), and desorption (5 min). Bottom of table: Na<sup>+</sup> influx measurements at 10 μM Na<sup>+</sup> external in plant grown at 0.2 mM Ca<sup>2+</sup>, where the Ca<sup>2+</sup> is withdrawn prior to the uptake. Measurements significantly different from the control are designated with a \*, *P* < 0.05. Values represent mean ± SEM (*n* = 4–19).

Treatment (measured at 100 μM Na <sup>+</sup> )	Na <sup>+</sup> influx (nmol g <sup>-1</sup> fw h <sup>-1</sup> )	Suppression (%)
Control	1124.7 ± 18.3	0
CN <sup>-</sup> +SHAM (10 mM)	61.8* ± 2.2	94.5
pH 4.2	238.7* ± 11.9	78.8
4 °C	218.7* ± 9.0	80.6
K <sup>+</sup> (5 mM)	87.4* ± 3.7	92.2
NH <sub>4</sub> <sup>+</sup> (5 mM)	71.2* ± 4.7	93.7
Ca <sup>2+</sup> (5 mM)	91.1* ± 4.4	91.9
TEA <sup>+</sup> (10 mM)	150.3* ± 5.5	86.6
Ba <sup>2+</sup> (5 mM)	118.1* ± 4.9	89.5
Cs <sup>+</sup> (10 mM)	88.6* ± 3.9	92.1
La <sup>3+</sup> (50 μM)	293.3* ± 17.1	73.9
Gd <sup>3+</sup> (50 μM)	211.0* ± 6.3	81.2
Gd <sup>3+</sup> (5 mM)	213.3* ± 12.6	81.0
Zn <sup>2+</sup> (10 mM)	85.8* ± 3.9	92.4
Quinine (1 mM)	134.1* ± 14.4	88.1
Lidocaine (10 mM)	101.9* ± 3.2	90.9
Dopamine (5 mM)	158.1* ± 11.6	85.9

Treatment (measured at 10 μM Na <sup>+</sup> )	Na <sup>+</sup> influx (nmol g <sup>-1</sup> fw h <sup>-1</sup> )	Suppression (%)
Control	593.4 ± 17.0	0
Gd <sup>3+</sup> (50 μM)	334.0* ± 12.7	43.7
TEA <sup>+</sup> (10 mM)	259.1* ± 8.6	56.3
Cs <sup>+</sup> (10 mM)	102.4* ± 4.0	82.7
Lidocaine (10 mM)	90.5* ± 2.2	84.7

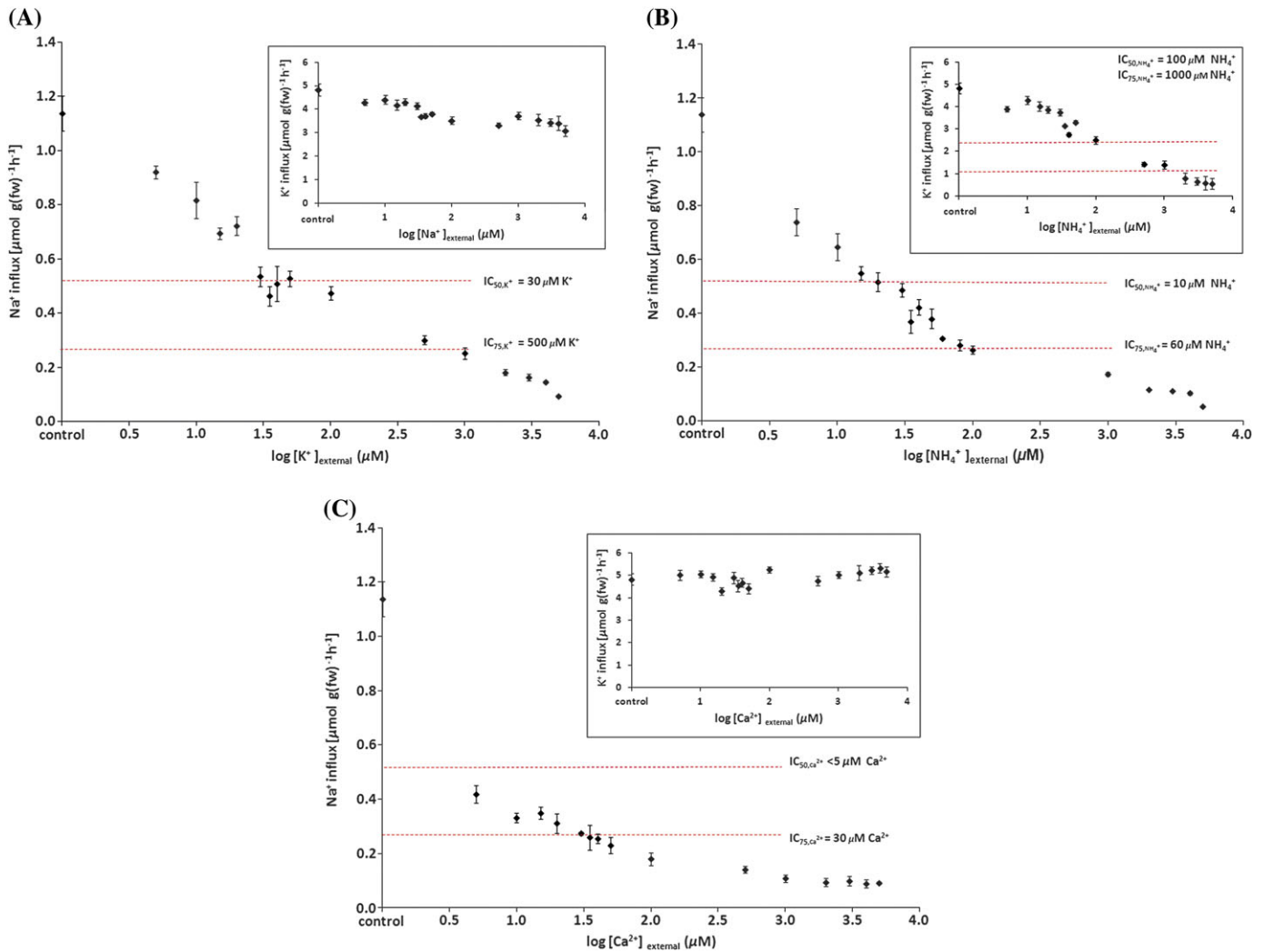
strongest suppression was seen when both respiratory electron transport pathways were blocked using cyanide and SHAM. The presence of Ca<sup>2+</sup> in solution also abolished almost the entire flux, as did the presence of K<sup>+</sup>, confirming the observations of others (Rains and Epstein, 1967a, b; Garcíadeblás *et al.*, 2003; Horie *et al.*, 2007; Haro *et al.*, 2010), but, in addition, it was found that another common nutrient, ammonium (NH<sub>4</sub><sup>+</sup>), brought Na<sup>+</sup> influx to very low values, for which no precedent could be found in the literature. Na<sup>+</sup> influx responded negatively to the application of a wide range of channel inhibitors: both the NSCC blockers Gd<sup>3+</sup>, La<sup>3+</sup>, Zn<sup>2+</sup>, and quinine, and the K<sup>+</sup> channel blockers TEA<sup>+</sup>, Ba<sup>2+</sup>, and Cs<sup>+</sup>, powerfully suppressed Na<sup>+</sup> influx (Demidchik and Tester, 2002; Kader and Lindberg, 2005; Volkov and Amtmann, 2006). In addition, it was found that the anaesthetic lidocaine and the neurotransmitter dopamine, two compounds known to block Na<sup>+</sup>-specific channels in animal systems (Ragsdale *et al.*, 1996; Cantrell and Catterall, 2001), both greatly reduced Na<sup>+</sup> influx, a novel finding with respect to plant Na<sup>+</sup> transport.

Effects of selected inhibitors were further investigated in plants measured at 10 μM [Na<sup>+</sup>]<sub>ext</sub>. Fluxes in these plants were also suppressed by the channel blockers Gd<sup>3+</sup>, TEA<sup>+</sup>, Cs<sup>+</sup>, and lidocaine, although to a slightly lesser extent than plants measured at 100 μM (Table 2).

Of the inhibitors given in Table 2, the effects on Na<sup>+</sup> influx of those that are commonly regarded as plant nutrients (i.e. K<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, and Ca<sup>2+</sup>) were investigated more thoroughly. Figure 2A shows how strongly the presence of K<sup>+</sup> reduces Na<sup>+</sup> influx: half-maximal inhibition of the control flux occurs at only 30 μM [K<sup>+</sup>]<sub>ext</sub> (=IC<sub>50,K<sup>+</sup></sub>), and is suppressed by another 50% at 500 μM [K<sup>+</sup>]<sub>ext</sub> (=IC<sub>75,K<sup>+</sup></sub>). By contrast, the reciprocal effect, i.e. suppression of K<sup>+</sup> influx (as measured using <sup>42</sup>K) by Na<sup>+</sup> was much less pronounced; an IC<sub>50</sub> was not found, even with Na<sup>+</sup> application of up to 5 mM (Fig. 2A, inset). A somewhat different picture emerged when NH<sub>4</sub><sup>+</sup> was present during uptake (Fig. 2B). Both Na<sup>+</sup> and K<sup>+</sup> influxes were greatly reduced by NH<sub>4</sub><sup>+</sup>, particularly in the case of Na<sup>+</sup>, which was suppressed with IC<sub>50</sub> and IC<sub>75</sub> values of 10 and 60 μM, respectively, while those for K<sup>+</sup> influx were 100 and 1000 μM (inset). By contrast, Na<sup>+</sup> influx was highly sensitive to extremely small amounts of Ca<sup>2+</sup>, (IC<sub>50,Ca<sup>2+</sup></sub> < 5 μM and IC<sub>75,Ca<sup>2+</sup></sub> = 30 μM; Fig. 2C), whereas K<sup>+</sup> influx showed no signs of suppression by Ca<sup>2+</sup> supply (Fig. 2C, inset).

Figure 3 shows the results of <sup>24</sup>Na<sup>+</sup>-efflux experiments with plants grown and measured at 100 μM Na<sup>+</sup> (i.e. steady-state conditions). It was found that applying amiloride, a well-known blocker of Na<sup>+</sup>/H<sup>+</sup> antiporters (Canessa *et al.*, 1994; Cuin *et al.*, 2011), midway through the elution period, did not affect the efflux of <sup>24</sup>Na<sup>+</sup> (Fig. 3). By contrast, application of high (5 mM) K<sup>+</sup> in the same manner did produce a small but detectable increase in Na<sup>+</sup> efflux (not shown), a phenomenon previously observed by Jeschke (1970). Compartmental analysis of <sup>24</sup>Na<sup>+</sup>-efflux and -retention data yielded key flux and pool-size parameters (Fig. 3, tabular inset). The value of Na<sup>+</sup> influx was similar to that given in Fig. 1, but a surprising result was that 65% of the absorbed <sup>24</sup>Na<sup>+</sup> was subsequently released by root cells, in a futile-cycling scenario not usually observed at such low substrate concentrations (Britto and Kronzucker, 2006; Coskun *et al.*, 2010), but very commonly at high concentrations, especially for Na<sup>+</sup> (Essah *et al.*, 2003; Britto and Kronzucker, 2006; Malagoli *et al.*, 2008; Britto and Kronzucker, 2009). The other parameter of high interest in this analysis was the Na<sup>+</sup> pool size estimated for the cytosolic compartment (see below for discussion about phase identification). Its value was found to be 3.37 mM, within the range of [Na<sup>+</sup>]<sub>cytosol</sub> previously determined, under saline conditions, by Carden *et al.* (2003) in two cultivars of barley by use of Na-selective microelectrodes (2–28 mM), but well below normal cytosolic [K<sup>+</sup>] values near 100 mM (Kronzucker *et al.*, 2003b).

Figure 4 shows the strong (*R*<sup>2</sup> = 0.99) semi-logarithmic correlation between Na<sup>+</sup> influx and the corresponding depolarization of the electrical potential (ΔΨ) across the plasma membrane of barley roots cells. As with the influx of sodium, membrane potentials responded immediately to the



**Fig. 2.** Influences of (A)  $K^+$ , (B)  $NH_4^+$ , and (C)  $Ca^{2+}$  on high-affinity  $Na^+$  influx (and of  $Na^+$ ,  $NH_4^+$ , and  $Ca^{2+}$  on high-affinity  $K^+$  influx, respectively; insets). Fluxes were measured at  $100 \mu\text{M } Na^+$  or  $K^+$ . Dashed lines represent  $IC_{50}$  and  $IC_{75}$  values for the inhibition of control fluxes. Values represent means  $\pm$  SEM ( $n=3-14$ ).

addition of  $IC_{50}$  and  $IC_{75}$  concentrations of  $K^+$  and  $NH_4^+$  (but not  $Ca^{2+}$ , not shown), in the presence of  $0.1 \text{ mM } Na^+$ . Membrane depolarizations brought about by  $K^+$  and  $NH_4^+$  are well known to occur in plant root cells, even at such low concentrations (Cheeseman and Hanson, 1979; Ayling, 1993; Wang *et al.*, 1994).

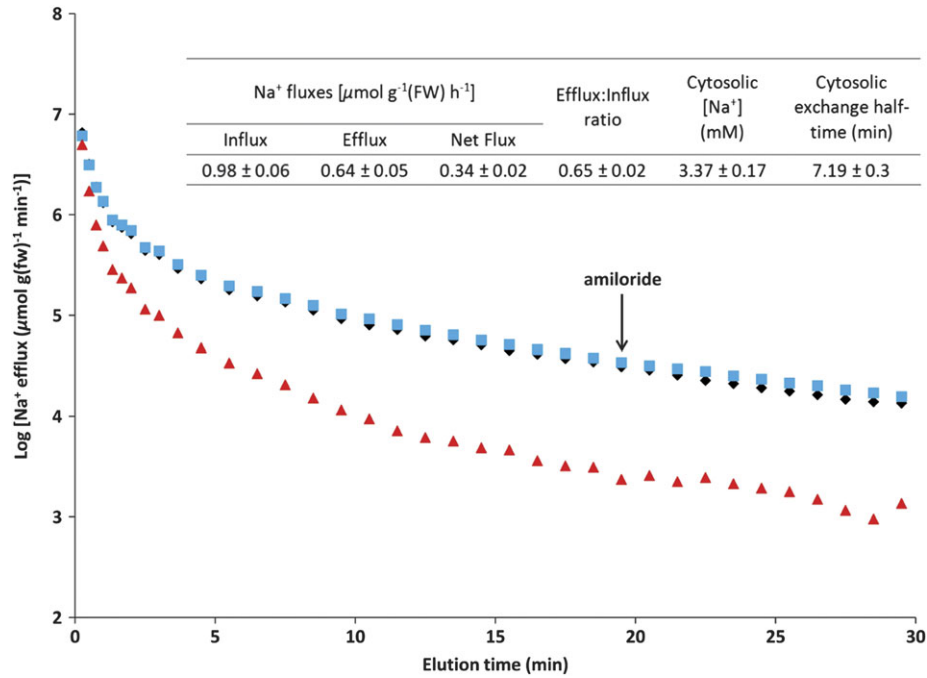
Figure 5 depicts  $Na^+$  shoot fresh weight and total  $Na^+$  content of 7-d-old barley seedlings grown under seven sodium regimes, ranging from  $0.01$  to  $50 \text{ mM } [Na^+]_{\text{ext}}$ . Initially, rising tissue  $Na^+$  was positively associated with an increase in fresh weight. However, once  $[Na^+]_{\text{ext}}$  reached or exceeded  $10 \text{ mM}$ , fresh weight declined rapidly, while  $Na^+$  content continued to rise. This decline in biomass was more severe than previously observed for barley plants growing in complete solution, probably due to a lack of essential nutrients (Kronzucker *et al.*, 2006). Interestingly, even under steady-state growth on  $0.01 \text{ mM } Na^+$ , barley seedlings accumulated almost half of what was accumulated by seedlings growing at  $5000$  times the  $Na^+$  supply.

## Discussion

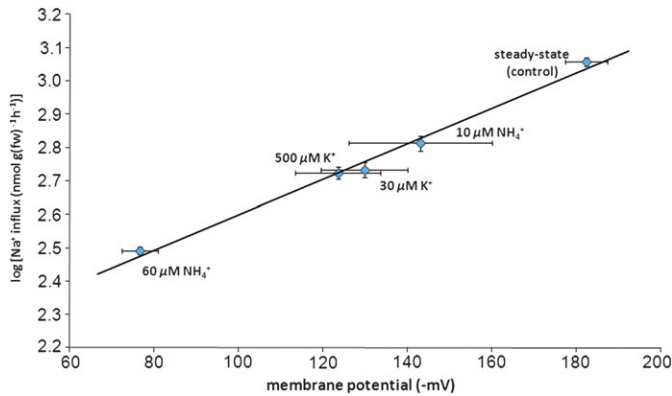
This study investigates high-affinity  $Na^+$  transport in plant roots from a variety of perspectives. Sodium influx values were determined by use of a  $5 \text{ min } ^{24}\text{Na}$ -absorption protocol, which agreed well with those determined by the alternative method of compartmental analysis by  $^{24}\text{Na}$  efflux. While these procedures resulted in basal values consistent with those reported in the literature, the inhibitory profiles, electrophysiological data, compartmental analyses, and growth and tissue data presented here provide new information about this transport phenomenon, and the possible mechanisms by which it occurs, as well as its ecological significance.

### *Interactions between $Na^+$ uptake and nutrient provision*

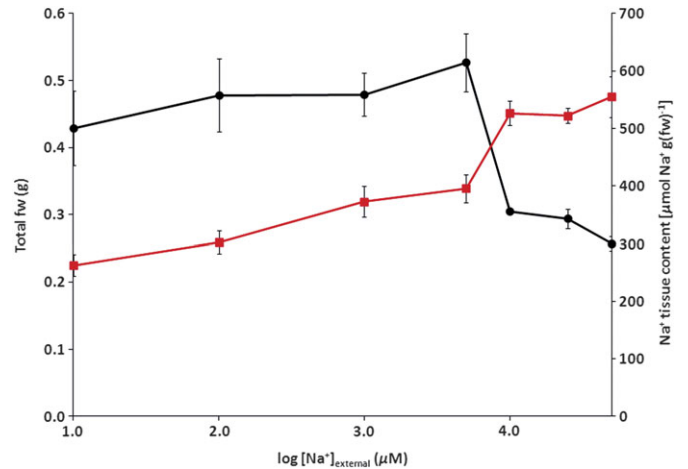
One of the most striking findings here is the degree to which three nutrients commonly found in soils can inhibit the influx of sodium under high-affinity conditions (Table 2;



**Fig. 3.** Traces of <sup>24</sup>Na<sup>+</sup> efflux from roots of intact barley seedlings grown and measured at 100 μM Na<sup>+</sup>. Diamonds, steady-state conditions with no inhibitors; squares, sudden application of 0.1 mM amiloride as shown by the arrow (at 19.5 min); triangles, 5 mM K<sup>+</sup> present in the labelling medium and during subsequent elution. Each point on the efflux traces represents the mean of nine replicates for control and amiloride treatments, and the mean of four replicates for the K<sup>+</sup> treatment. Tabular inset: flux parameters derived from compartmental analysis by tracer efflux, under steady-state conditions (mean ± SEM; n=9).



**Fig. 4.** Relationship between membrane electrical potential and unidirectional Na<sup>+</sup> influx, and in response to NH<sub>4</sub><sup>+</sup> and K<sup>+</sup>. NH<sub>4</sub><sup>+</sup> and K<sup>+</sup> concentrations represent IC<sub>50</sub> and IC<sub>75</sub> values for the suppression of Na<sup>+</sup> influx by these ions (see Fig. 2). Na<sup>+</sup> fluxes were measured at 100 μM [Na<sup>+</sup>]<sub>ext</sub>. The data represent means of 4–23 replicates ± SEM.



**Fig. 5.** Tissue Na<sup>+</sup> content (squares) and fresh weight (circles) of barley seedlings was determined in plants grown at 0.01, 0.1, 1, 5, 10, 25, and 50 mM Na<sup>+</sup> steady state. Values represent means ± SEM (n=6–10).

Fig. 2): external concentrations of K<sup>+</sup>, NH<sub>4</sub><sup>+</sup>, and Ca<sup>2+</sup> as low as 500, 60, and 30 μM, respectively, were sufficient to reduce Na<sup>+</sup> influx by 75% (Fig. 2). While other studies have shown K<sup>+</sup> and Ca<sup>2+</sup> to significantly reduce Na<sup>+</sup> uptake (Rains and Epstein, 1967a, b; Garcíadeblás et al., 2003; Haro et al., 2010), none has shown that NH<sub>4</sub><sup>+</sup> can also accomplish this, despite the well-known inhibition by NH<sub>4</sub><sup>+</sup> of high-affinity K<sup>+</sup> transport (Santa-Maria et al., 2000, Kronzucker et al., 2003b). Moreover, it is shown just how strong these effects can be, by application of a wide concentration range

of inhibitory nutrients and using a short-term labelling procedure that yields unidirectional influx values of high precision. Most studies in this area have used depletion or longer-term tracer-uptake protocols, which do not provide unidirectional influx measurements; for this reason, they may have underestimated the Na<sup>+</sup> influx capacity and detail (Britto and Kronzucker, 2001), particularly if the high ratio (0.65) of efflux to influx seen here is a general feature of Na<sup>+</sup> HATS in higher plants, as appears to be the case with

low-affinity  $\text{Na}^+$  influx (Essah *et al.*, 2003; Malagoli *et al.*, 2008; Britto and Kronzucker, 2009).

Ammonium, while typically not the major source of N for plants, is often found in soils well above the  $\text{IC}_{75}$  of 60  $\mu\text{M}$  (Glass *et al.*, 2002; Kronzucker *et al.*, 2003a, and references therein). In addition, 500  $\mu\text{M}$   $\text{K}^+$  is not uncommon in soils (let alone the  $\text{IC}_{50}$  of 30  $\mu\text{M}$ ; Marschner, 1995; Szczerba *et al.*, 2006), and soil  $[\text{Ca}^{2+}]$  values, almost universally, far exceed the  $\text{IC}_{75}$  of 30  $\mu\text{M}$  reported here (Zidan *et al.*, 1991; Hirschi, 2004). This suggests that influx of  $\text{Na}^+$  in the high-affinity range is appreciable only when measured under rather unrealistic conditions, even when considering that soils can have considerable chemical heterogeneity over small spatial scales (Jackson and Caldwell, 1993). The nutritionally suppressed influx of  $\text{Na}^+$ , which is far lower than normal uptake rates of  $\text{K}^+$  (Kronzucker *et al.*, 2003b), calls into question the significance of sodium uptake as a K-replacement strategy under K limitation (see Introduction), at least in this range of external  $[\text{Na}^+]$  for this cultivar of barley. It is also, however, at odds with Horie *et al.* (2007), who found, in rice seedlings, that both OsHKT2;1-mediated high-affinity  $\text{Na}^+$  uptake and  $\text{Na}^+$ -stimulated growth under  $\text{K}^+$  deprivation, occurred in the presence of 1 mM  $\text{Ca}^{2+}$ . These issues are at least partially related to the expression and functioning of  $\text{Na}^+$  HATS being highly species- (and cultivar-) dependent, as is the degree to which  $\text{Na}^+$  may substitute for  $\text{K}^+$ . In another contrasting finding, net  $\text{Na}^+$  fluxes in the HATS range were not suppressed by  $\text{K}^+$  in the case of sunflower (Garcia-deblás *et al.*, 2003), which is regarded as a salt-tolerant sodium accumulator (Bhatt and Indirakutty, 1973). Thus, in future work, it will be interesting to investigate nutritional sensitivities (or lack thereof) of the sodium HATS in plant species, such as sugar beet, that are able beneficially to accumulate large amounts of  $\text{Na}^+$ , even in complete growth media, and also in  $\text{C}_4$  plants that require  $\text{Na}^+$  as a micronutrient.

#### Possible mechanisms of high-affinity $\text{Na}^+$ uptake

It is worth noting that, even while the flux of  $\text{Na}^+$  may be highly suppressed under normal nutritional conditions, a very high substrate concentration, as is found in the toxic range, may yet drive  $\text{Na}^+$  into the plant via this same mechanism, at rates sufficient to cause detrimental  $\text{Na}^+$  accumulation (Kronzucker and Britto, 2011). This may be particularly important given that HKT transporters have characteristics of both carriers and ion channels (see Introduction). In this vein, Laurie *et al.* (2002) showed that TaHKT2;1-mediated  $\text{Na}^+$  influx reduces growth in wheat roots, when exposed to high (200 mM)  $\text{Na}^+$ , and deprived of  $\text{K}^+$ . Interestingly, the authors concluded that  $\text{Na}^+$  transport to the stele was a major control point at which HKT appeared to operate, consistent with the foliar accumulation of  $\text{Na}^+$  as a significant aspect of  $\text{Na}^+$  stress (Munns and Tester, 2008; cf. Kronzucker and Britto, 2011). Additional support for the possibility that high-affinity  $\text{Na}^+$  transporters catalyse fluxes in the toxic range can be seen in the electrophysiological work of Buschmann *et al.* (2000), who

showed that  $\text{Na}^+$  currents were enhanced in response to  $\text{K}^+$  deprivation (also see Mian *et al.*, 2011).

The inhibitor data presented in Table 2 provide additional clues about the mechanism underlying high-affinity  $\text{Na}^+$  influx. Numerous treatments known to block several types of ion channels (NSCCs,  $\text{K}^+$ -specific channels, and  $\text{Na}^+$ -specific channels in animals) were able to diminish the  $\text{Na}^+$  flux profoundly in the present study. Seven of these inhibitory treatments ( $\text{Ca}^{2+}$ ,  $\text{Gd}^{3+}$ ,  $\text{La}^{3+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Ba}^{2+}$ , low pH, and quinine) are known to affect NSCCs (Kronzucker and Britto, 2011), which have been strongly implicated in  $\text{Na}^+$  uptake in the toxic range (Munns and Tester, 2008). However,  $\text{Cs}^+$  and tetraethylammonium ( $\text{TEA}^+$ ), two agents to which most NSCCs have been reported to be insensitive (Kronzucker and Britto, 2011), also strongly blocked  $\text{Na}^+$  transport. Nevertheless, it has been shown that NSCCs can transport both  $\text{Cs}^+$  and  $\text{TEA}^+$  (Davenport and Tester, 2000; Demidchik and Tester, 2002), which could cause an electrical depolarization of the plasma membrane and, therefore, a reduction in the driving force for channel-mediated  $\text{Na}^+$  uptake (see below); this may in itself explain the inhibition of the flux observed here. Similarly, low temperature (4 °C) and cyanide/SHAM applications can also cause depolarizations (Lew, 1991) and thus reduce the flux, if it is driven by the membrane potential.

Interestingly, HKT-type transporters are thought to have evolved from prokaryotic  $\text{K}^+$  channels (Shafir *et al.*, 2008). They have, in addition, been suggested to be a sub-class of NSCCs, for which no genetic identities have yet been unequivocally ascribed (Horie *et al.*, 2009) and, indeed, rice OsHKT2;4 has been shown to be sensitive to both  $\text{Gd}^{3+}$  and  $\text{La}^{3+}$  (Lan *et al.*, 2010). Catalysis via HKT-type transporters of the high-affinity fluxes presented here is suggested by these studies, as well as by demonstrations that HKT-mediated  $\text{Na}^+$  transport is highly sensitive to external  $\text{K}^+$  (Horie *et al.*, 2007; Haro *et al.*, 2010; also see Introduction). Curiously, the *Arabidopsis* genome encodes only one HKT-type transporter, AtHKT1.1, which had been suggested to play a role in low-affinity  $\text{Na}^+$  transport in the toxic range (Rus *et al.*, 2001), but more recent evidence indicates that its major role is in the internal redistribution of  $\text{Na}^+$ , rather than its primary acquisition (Berthomieu *et al.*, 2003; Møller *et al.*, 2009).

#### Thermodynamics of high-affinity $\text{Na}^+$ transport

If it is indeed correct that ion channels (or channel-like HKTs) mediate high-affinity  $\text{Na}^+$  transport, possibly in a manner similar to that proposed for high-affinity  $\text{K}^+$  transport by Hirsch *et al.* (1998), a thermodynamic scenario would have to exist that permits a 'passive' influx of  $\text{Na}^+$ . Thermodynamic evaluation necessitates information about the concentration gradient of  $\text{Na}^+$  across the membrane, as well as its state of electrical polarization. To determine the internal (cytosolic)  $\text{Na}^+$  pool, by use of efflux trials as shown in Fig. 3, the intracellular origin of the released tracer was first verified (Britto and Kronzucker, 2003; Coskun *et al.*, 2010); this was done in several ways. First,



the influx values determined from tracer-efflux and -retention data were very close to those determined by the 'direct-influx' procedure at 100  $\mu\text{M}$   $[\text{Na}^+]_{\text{ext}}$  (Table 2; Fig. 3, tabular inset). Second, the presence of 5 mM  $\text{K}^+$  in the external medium during the 1 h labelling period resulted in a dramatic reduction in tracer release during elution (Fig. 3, lower trace); this is readily explained by the blockage of  $^{24}\text{Na}^+$  influx during labelling by  $\text{K}^+$  (Fig. 2); such a powerful effect would not be expected had the origin of the  $^{24}\text{Na}^+$  trace been extracellular spaces of the root. A third line of evidence comes from comparing the most suppressed influx in Table 2 (i.e. in response to the cyanide+SHAM treatment) to the efflux determined in Fig. 3. The small influx that remains after this powerful metabolic treatment ( $0.06 \mu\text{mol g}^{-1} \text{fw h}^{-1}$ ) represents the maximum contribution to influx that the apoplastic component can make. Because the measured efflux of  $0.65 \mu\text{mol g}^{-1} \text{fw h}^{-1}$  far exceeds this quantity, it can be concluded that >90% of released tracer originated intracellularly. Finally, it was of interest to determine which intracellular compartment(s) was/were the source of released tracer. Prior work indicates that half-times for vacuolar  $\text{Na}^+$  exchange in barley seedlings can be 77–231 h (Jeschke, 1982), far longer than the 1 h labelling period employed in the present study. In addition, the quantity of  $\text{Na}^+$  found in the traced pool ( $0.169 \mu\text{mol g}^{-1} \text{fw}$ ) was far less than can account for the tissue accumulation of  $\text{Na}^+$  ( $300 \mu\text{mol g}^{-1} \text{fw}$ ; Fig. 5), which is likely to be mostly vacuolar. Thus, very little of the released tracer is likely to have originated in the vacuole, suggesting that, instead, its origin was the cytosolic compartment of root cells.

This information, in conjunction with the finding that root cell membranes of plants grown at 100  $\mu\text{M}$   $\text{Na}^+$  are highly electrically polarized, with a steady-state potential of  $-182 \text{ mV}$  (Fig. 4), a condition not uncommon in K-deprived plants (Cheeseman and Hanson, 1979), allows the Nernst equation to be used to indicate the direction of passive flux. Given the small  $\text{Na}^+$  pool and the highly polarized membrane, this direction is clearly inward, even at 100  $\mu\text{M}$   $[\text{Na}^+]_{\text{ext}}$  (i.e.  $\Delta\Psi$  is negative of the calculated Nernst potential of  $-92 \text{ mV}$ ), supporting the idea of a passive, possibly channel-mediated influx of  $\text{Na}^+$ . This possibility is further supported by the strong correlation between  $\Delta\Psi$  and  $\text{Na}^+$  influx (Fig. 4). Interestingly, this relationship is not linear but log-linear, indicating that the behaviour of the transport system is not Ohmic but may involve voltage gating, or be influenced by asymmetries in  $\text{Na}^+$  concentrations or channel structure (Hille, 2001).

#### 'Can channels do it all?'

The finding that high-affinity  $\text{Na}^+$  transport shares many characteristics with ion-channel behaviour provides new evidence in the ongoing debate summarized in the title of an essay by Kochian and Lucas (1993), 'Can  $\text{K}^+$  channels do it all?' Although that paper was specifically about  $\text{K}^+$ , its conclusion that  $\text{K}^+$  channels cannot catalyse active, high-affinity  $\text{K}^+$  uptake is relevant here. The term 'high affinity'

refers to the substrate binding characteristics of the enzyme or transporter (Marschner, 1995), as identified by the Michaelis constant ( $K_m$ ). Here,  $\text{Na}^+$  transport from low external concentrations does display saturable Michaelis–Menten properties with low  $K_m$  values, characteristic of high-affinity systems (Table 1), while at the same time showing pharmacological sensitivities typical of ion channels, even at external  $[\text{Na}^+]$  of only 10  $\mu\text{M}$  (Table 2). However, thermodynamic analysis indicates that high-affinity  $\text{Na}^+$  transport is not an example of 'active transport', but appears to be driven by the membrane electrical potential, as is the case with channels (Fig. 4; Britto and Kronzucker, 2006). In addition, the strong inhibition of the flux by low temperature ( $4 \text{ }^\circ\text{C}$ ) and cyanide/SHAM application is not necessarily a hallmark of active transport, but these treatments can act on channels via alterations in membrane potential (Lew, 1991). Thus, our study suggests that either channels, or channel-like transporters (such as HKTs) may indeed 'do it all', with respect to sodium uptake. However, this scenario differs from the proposed channel-mediated uptake of  $\text{K}^+$  from low external concentrations (Hirsch *et al.*, 1998; Spalding *et al.*, 1999), in that many  $\text{K}^+$ -specific, high-affinity carrier-type transporters are known, which are likely to operate in concert with channel mediation; this is not so with  $\text{Na}^+$ . A remaining question is whether ion channels, which are known to be saturable in some instances (Hille, 2001), can have  $K_m$  values as low as those presented here.

#### The question of $\text{Na}^+$ - $\text{K}^+$ symport

Another controversy surrounding high-affinity  $\text{Na}^+$  uptake concerns the possibility that it is linked to symport with  $\text{K}^+$  (Rubio *et al.*, 1995; Maathuis *et al.*, 1996; Corratgé-Faillie *et al.*, 2010; Kronzucker and Britto, 2011). No evidence that this occurs was found; on the contrary,  $\text{Na}^+$  uptake was profoundly reduced by low  $\text{K}^+$ , while  $\text{Na}^+$  (up to 5 mM) suppressed  $\text{K}^+$  uptake only moderately (Fig. 2A). This is in agreement with Maathuis *et al.* (1996), who found no *in planta* evidence for  $\text{Na}^+/\text{K}^+$  symport. In this context, it is also worthwhile to ask whether the system that transports  $\text{Na}^+$  also transports other ions. At 100  $\mu\text{M}$  substrate concentrations,  $\text{K}^+$  influx was not affected by either  $\text{Ca}^{2+}$  (Fig. 2C, inset),  $\text{Gd}^{3+}$  or  $\text{Ba}^{2+}$  (not shown), nor was  $\text{NH}_4^+$  influx altered by  $\text{Ba}^{2+}$  or  $\text{Cs}^+$  (not shown); the contrast of this finding with the inhibitory profile of  $\text{Na}^+$  influx (Table 2) indicates that neither  $\text{K}^+$  nor  $\text{NH}_4^+$  is a substrate for this transport system.

Electrophysiological analysis (Fig. 4) suggests that  $\text{K}^+$  and  $\text{NH}_4^+$  inhibit high-affinity  $\text{Na}^+$  influx, in the short term, by depolarizing root cell membranes (Cheeseman and Hanson, 1979; Ayling, 1993; Wang *et al.*, 1994). By contrast,  $\text{Ca}^{2+}$  did not depolarize (not shown); thus, its suppression of the  $\text{Na}^+$  flux appears to occur via a different means, presumably a direct action upon the transporter.

#### An unknown mechanism of $\text{Na}^+$ efflux?

In our thermodynamic analysis, an inwardly passive mechanism would imply an active extrusion of  $\text{Na}^+$ . Various



mechanisms of outward  $\text{Na}^+$  pumping from plant roots have been proposed, including a sodium ATPase (Lunde *et al.*, 2007; Jacobs *et al.*, 2011) and a  $\text{Na}^+/\text{H}^+$  antiport (Mennen *et al.*, 1990). The former has been identified in algae and bryophytes, but not in higher plants, while the latter appears to be catalysed by the SOS1 protein, and is strongly tied to salt tolerance in many higher plants (Shi *et al.*, 2000; Cuin *et al.*, 2011; see Kronzucker and Britto, 2011, for a review). However, SOS1 has been reported to be directly or indirectly sensitive to the  $\text{Na}^+/\text{H}^+$  antiport-inhibitor amiloride (Cuin *et al.*, 2011), which yielded no change to the tracer-efflux line in Fig. 3. Thus, it appears that the transporter mediating  $\text{Na}^+$  efflux under high-affinity conditions is a previously unidentified component of this cellular sodium cycle.

#### *Na<sup>+</sup> accumulation and compartmentation*

Another significant outcome of the compartmental analysis shown here is that the cytosolic pool of  $\text{Na}^+$  was estimated at only 3.37 mM, at an extracellular concentration of 100  $\mu\text{M}$ , and in the absence of  $\text{K}^+$  (Fig. 3, tabular inset). This value is within the range measured by Carden *et al.* (2003) using  $\text{Na}^+$ -selective intracellular electrodes, albeit under salinity stress (to our knowledge, our values are the first cytosolic  $[\text{Na}^+]$  estimates to be determined in the high-affinity range). Nevertheless, this value is far lower than typical values of cytosolic  $[\text{K}^+]$ , which are about 100 mM (Kronzucker *et al.*, 2003b). On the other hand, the tissue accumulation of  $\text{Na}^+$  under these conditions is substantial, at 300  $\mu\text{mol g}^{-1}$  fw (Fig. 5), and is similar in magnitude to that of  $\text{K}^+$  accumulation at 100  $\mu\text{M}$   $[\text{K}^+]_{\text{ext}}$  (Szczzerba *et al.*, 2006). This suggests that, under conditions that optimize  $\text{Na}^+$  uptake in the high-affinity range, and under growth conditions where nutrients inhibitory to its flux are not present,  $\text{Na}^+$  can substitute for the osmolytic function of  $\text{K}^+$  on a total-tissue basis, probably accumulating in the vacuoles, but cannot replace the critical requirement of high  $[\text{K}^+]$  in the cytosol. It is likely that this osmolytic function is responsible for the positive relationship between  $\text{Na}^+$  accumulation and growth between 0.01 and 5 mM  $[\text{Na}^+]_{\text{ext}}$ , depicted in Fig. 5.

#### *Concluding remarks*

It was found that high-affinity, unidirectional influx of  $\text{Na}^+$  exhibits values comparable to those of high-affinity  $\text{K}^+$  uptake, but can only do so under special conditions; i.e. in the near absence of the major soil nutrients  $\text{K}^+$ ,  $\text{Ca}^{2+}$ , and  $\text{NH}_4^+$ . Ecologically, therefore, the relevance of high-affinity  $\text{Na}^+$  transport should be viewed with caution. However, the study of this phenomenon, even under ecologically unrealistic conditions, reveals surprising results which may have implications for  $\text{Na}^+$  acquisition in the toxic range. Chief amongst these is the dual display of both channel and carrier characteristics, in terms of inhibitory profiles, saturability with low  $K_m$ , and thermodynamic conditions. The possible involvement of HKT-type transporters, for which channel-like behaviour (and descent from bacterial channels) has

recently been shown, is supported for high-affinity  $\text{Na}^+$  influx. No evidence, however, was seen for the coupling of  $\text{K}^+$  influx to that of  $\text{Na}^+$ . In addition, it is shown that  $\text{Na}^+$  influx is part of a futile cycle even at low  $[\text{Na}^+]_{\text{ext}}$ , and results in only small cytosolic accumulation (<5 mM), but total tissue accumulation that can rival that of  $\text{K}^+$ .

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