

# PERMEABILITY AND INTERNAL CONCENTRATION OF IONS DURING DEPOLARIZATION OF THE ELECTROPLAX\*

BY ARTHUR KARLIN

DEPARTMENTS OF NEUROLOGY AND PHYSIOLOGY, COLLEGE OF PHYSICIANS AND  
SURGEONS, COLUMBIA UNIVERSITY

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Electroplax from the electric organs of *Electrophorus electricus* are large, sheet-like cells derived from muscle, whose function, the generation of transcellular current, is a consequence of their asymmetry: the membrane on one side of the cell is innervated and electrically and chemically excitable, while that on the other side is noninnervated and inexcitable.<sup>1, 2</sup> The application of acetylcholine and related compounds to the innervated membrane results in a depolarization of the cell that is most likely initially due, as in muscle,<sup>3</sup> to changes in the permeability of the subsynaptic regions of the membrane to  $\text{Na}^+$  and  $\text{K}^+$ .<sup>4-6</sup> This depolarization, measured across the innervated membrane after a 1-20-minute application of a depolarizing agent, has been taken as a measure of the primary response of the electroplax to such agents, and inferences regarding the acetylcholine receptor have been drawn from dose-response data so obtained.<sup>7-9</sup> This depolarization is not, however, simply and solely a function of the initial permeability changes. During these depolarizations, which are relatively long compared with the neurally evoked postsynaptic potential of 1-2-millisecond duration, there appear to be appreciable changes in the intracellular concentrations of  $\text{Na}^+$  and  $\text{K}^+$ . These concentration changes, and probably in addition the compensating process of active transport, partially determine the potential difference across the membrane. First, evidence will be presented indicating that the concentration changes are appreciable, and, second, these changes and the permeability changes will be estimated. The estimate of the permeability change should provide a better measure of the primary response of the electroplax to acetylcholine and its congeners.

*Methods.*—Electroplax were isolated<sup>10</sup> and were mounted in a Lucite holder in which the cell separates two pools of a modified Ringer's solution (containing 165 mM NaCl, 2.3 mM KCl, 2 mM  $\text{CaCl}_2$ , 2 mM  $\text{MgCl}_2$ , 1.2 mM  $\text{K}_2\text{HPO}_4$ , 0.3 mM  $\text{KH}_2\text{PO}_4$ , 10 mM glucose, pH 7.1). The holder used (details will appear elsewhere) was a modification of that described previously<sup>11</sup> in that the electroplax is placed horizontally, innervated side down, and part of the innervated membrane, exposed through a window ( $3 \times 0.75$  mm), is in contact with a stream of fresh solution. The rate of flow was such that the solution bathing the innervated side could be changed in a few seconds. Potential differences across either membrane were measured between an intracellular glass microelectrode filled with 3 M KCl and agar bridges in the outside solutions.

*Results and Discussion.*—*Effect of ouabain on the repolarization:* Normal depolarizations of both the innervated and the noninnervated membranes in response to 40  $\mu\text{M}$  carbamylcholine on the innervated side are shown in Figure 1A. Washing the carbamylcholine out with Ringer's solution causes a rapid repolarization (with an overshoot). In contrast, the repolarization following prior exposure of the electroplax to ouabain is strongly inhibited (Fig. 1B). Ouabain is believed to be a

specific inhibitor of the energized extrusion of  $\text{Na}^+$ , to which is coupled, either directly or indirectly, the uptake of  $\text{K}^+$ .<sup>12</sup> Whereas carbamylcholine acts only on the innervated membrane, ouabain inhibits the repolarization more strongly when added to the noninnervated side than to the innervated side. After exposure to ouabain, the repolarization following small as well as large depolarizations is inhibited, and the extent of inhibition increases with the number of depolarizations and the time of exposure to ouabain. Twenty minutes after the application of  $10^{-4}$  M ouabain to the noninnervated membrane, the repolarization following only 30 seconds of  $20 \mu\text{M}$  carbamylcholine was inhibited. In contrast, after 60 minutes,  $10^{-4}$  M ouabain had, in the absence of induced depolarizations, only a slight effect on the resting potential and action potential. The effects of ouabain implicate active  $\text{Na}^+$  transport in the repolarization process and, hence, indicate a significant accumulation of  $\text{Na}^+$  and loss of  $\text{K}^+$  (see below) during the depolarization. Previously, N-ethylmaleimide was found to inhibit the repolarization of the electroplax, and this effect was also probably due to inhibition of ATP formation or of its utilization by the  $\text{Na}^+$  pump.<sup>9</sup>

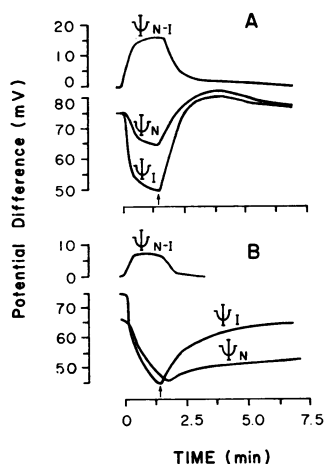
*Analysis of the depolarization:* Both the potential difference measured across the innervated membrane,  $\psi_I$  (outside minus inside), and that measured across the noninnervated membrane,  $\psi_N$  (outside minus inside), decrease following application of carbamylcholine to the innervated membrane (Fig. 1). The decrease in

FIG. 1.—Response of the electroplax to  $40 \mu\text{M}$  carbamylcholine.

The effect of carbamylcholine on the transcellular potential difference ( $\psi_{N-I}$ ), the p.d. across the innervated membrane ( $\psi_I$ ), and the p.d. across the noninnervated membrane ( $\psi_N$ ) were measured consecutively. The traces have been superimposed on the time axis; carbamylcholine was added to the innervated membrane in each case at 0 min and washed out at the arrow.

(A) Control.

(B)  $10^{-4}$  M ouabain was present on the noninnervated side of the cell for 10 min before  $\psi_I$  was measured, 25 min before  $\psi_N$  was measured, and 40 min before  $\psi_{N-I}$  was measured.



$\psi_I$  may be due, in part, to a permeability change *per se*, but the decrease in  $\psi_N$  is unlikely to be so caused, since presumably the permeability of the noninnervated membrane does not change: carbamylcholine does not reach it, and, in any case, carbamylcholine does not act on the noninnervated side. Furthermore, transcellular current driven by the change in  $\psi_I$ , and returning around the cell past the window, could have only a negligible effect on  $\psi_N$  since the resistance of the noninnervated membrane is only  $1/30$  of the resistance of the innervated membrane.<sup>2</sup> Therefore, the depolarization of the noninnervated membrane is most likely due to a change in intracellular ion concentration, namely, a decrease in ( $\text{K}^+$ ).  $\psi_N$  decreases to a lesser extent than  $\psi_I$ , and hence, in the presence of carbamylcholine the transcellular potential,  $\psi_{N-I} = \psi_N - \psi_I$ , is nonzero (Fig. 1).

Expressions for the changes in  $\psi_I$ ,  $\psi_N$ , and  $\psi_{N-I}$  may be derived from the Goldman equation,<sup>13</sup>

$$\psi = \frac{RT}{F} \ln \frac{K_i + \rho Na_i + \sigma Cl_o}{K_o + \rho Na_o + \sigma Cl_i}, \quad (1)$$

where  $\psi$  is outside potential *minus* inside potential, *i* and *o* designate intra- and extracellular concentrations,  $\rho$  is the ratio of Na<sup>+</sup> permeability to K<sup>+</sup> permeability, and  $\sigma$  is the ratio of Cl<sup>-</sup> permeability to K<sup>+</sup> permeability. The Cl<sup>-</sup> permeability has been said to be low in the electroplax.<sup>6, 14, 15</sup> I have observed that the immediate effect of changing from the usual outside solution (given above), containing 175 mM Cl<sup>-</sup>, to solutions in which most of the Cl<sup>-</sup> had been replaced by SO<sub>4</sub><sup>2-</sup> or by glutamate ((Cl<sup>-</sup>) = 10–20 mM) was a transient depolarization of the innervated membrane by 1–2 mV and of the noninnervated membrane by 3–5 mV. These results imply that  $\sigma$  is less than 0.1 for the innervated membrane and less than 0.2 for the noninnervated membrane.<sup>16</sup> Because of the relative magnitude of the terms in equation (1), Cl<sup>-</sup> may be neglected with a resulting error of less than 10 per cent. Equation (1) reduces thereby to

$$\psi = \frac{RT}{F} \ln \frac{K_i + \rho Na_i}{K_o + \rho Na_o}, \quad (2)$$

which I will assume applies to both membranes with the same resting value of  $\rho$ . Now let primed variables represent values at a given time after addition of a depolarizer and unprimed variables the values before, and let  $\Delta\psi = \psi' - \psi$ , then the positive extents of depolarization of the two membranes are

$$-\Delta\psi_I = \frac{RT}{F} \ln \frac{(K_i + \rho Na_i)(K_o + \rho' Na_o)}{(K_i' + \rho' Na_i')(K_o + \rho Na_o)} \quad (3)$$

and

$$-\Delta\psi_N = \frac{RT}{F} \ln \frac{K_i + \rho Na_i}{K_i' + \rho Na_i'}. \quad (4)$$

Furthermore, since  $\Delta\psi_{N-I} = \Delta\psi_N - \Delta\psi_I$  (which is positive),

$$\Delta\psi_{N-I} = \frac{RT}{F} \ln \frac{(K_i' + \rho Na_i')(K_o + \rho' Na_o)}{(K_i' + \rho' Na_i')(K_o + \rho Na_o)}. \quad (5)$$

Equation (4) expresses mathematically the argument given above that the depolarization of the noninnervated membrane implies a significant change in intracellular Na<sup>+</sup> and K<sup>+</sup> concentrations. Also, consistent with this interpretation is the fact that, after ouabain,  $-\Delta\psi_N$  increases (Fig. 1); i.e., there is greater net accumulation of Na<sup>+</sup> and net loss of K<sup>+</sup> when the Na<sup>+</sup> pump is inhibited. Prior exposure to ouabain also increases  $-\Delta\psi_I$ , but less than  $-\Delta\psi_N$ , and decreases  $\Delta\psi_{N-I}$ . These effects are in agreement with equations (3) and (5).

Expressions relating  $\rho'$ ,  $Na_i'$ , and  $K_i'$  to the measured potential differences and and to  $\rho$ ,  $Na_i$ ,  $K_i$ ,  $Na_o$ , and  $K_o$  may now be derived from the above equations.  $\rho$  is not known but may be estimated from equation (2). Mean values for  $K_i$  and  $Na_i$  have been determined to be 170 mM and 10 mM, respectively.<sup>4, 14</sup> The mean

resting potential of 25 cells was 81 mV. Inserting these values, and  $RT/F = 25$  mV,  $K_o = 5$  mM,  $Na_o = 165$  mM, into equation (2) and solving for  $\rho$  yields  $\rho = 0.01$ , a value reasonably similar to estimated values of  $\rho$  for other cells.<sup>17-19</sup> Assuming that  $\rho$  is constant, we may calculate  $K_i$  for each particular cell from equation (2) in the form

$$K_i = \xi(K_o + \rho Na_o), \quad (6)$$

where  $\rho Na_i$  has been neglected and  $\xi = e^{(F\psi/RT)}$  ( $\psi = \psi_I = \psi_N$ ). As a consequence of the relatively low permeability of  $Cl^-$ , it may be assumed that the sum of the intracellular concentrations of  $Na^+$  and  $K^+$  is constant (this also follows if nearly all the impermeable intracellular solutes are univalent anions);<sup>18</sup> i.e.,

$$Na_i + K_i = Na_i' + K_i' = M \quad (7)$$

and inserting the mean values given above,  $M = 180$  mM. Combining equations (4) and (7) we get an expression for  $K_i'$ ,

$$K_i' = \frac{K_i(1 - \rho) + \rho M(1 - \Delta\xi_N)}{\Delta\xi_N(1 - \rho)}, \quad (8)$$

where  $\Delta\xi_N = e^{(-F\Delta\psi_N/RT)}$ .

Finally, solving equation (5) for  $\rho'$ , we obtain

$$\rho' = \frac{\Delta\xi_{N-I}K_i'(K_o + \rho Na_o) - K_o(K_i' + \rho Na_i')}{Na_o(K_i' + \rho Na_i') - \Delta\xi_{N-I}Na_i'(K_o + \rho Na_o)}, \quad (9)$$

where  $\Delta\xi_{N-I} = e^{(F\Delta\psi_{N-I}/RT)}$ . Equations (6), (7), (8), and (9) may be combined to yield  $\rho'$  as a function of fixed parameters and measured variables,

$$\rho' = \frac{A\Delta\xi_{N-I}[A(1 - \rho)\xi + \rho M(1 - \Delta\xi_N)] - K_o(1 - \rho)[A(1 - \rho)\xi + \rho M]}{Na_o(1 - \rho)[A(1 - \rho)\xi + \rho M] - A\Delta\xi_{N-I}[M(\Delta\xi_N - \rho) - A(1 - \rho)\xi]} \quad (10)$$

where  $A = K_o + \rho Na_o$ . An approximate form of the above equation shows clearly the principal dependence of  $\rho'$  on  $\Delta\xi_{N-I}$  (i.e., on  $\Delta\psi_{N-I}$ ), and vice versa,

$$\rho' - \rho \cong \left( \frac{K_o}{Na_o} + \rho \right) (\Delta\xi_{N-I} - 1), \quad (11)$$

which is valid for  $\rho Na_i'/K_i'$  much less than 1 and  $Na_o/(K_o + \rho Na_o)$  much greater than  $\Delta\xi_{N-I}Na_i'/K_i'$ . The first condition is satisfied over the entire range of responses, but the second is well satisfied for responses about half maximal and less and only during the initial part of larger responses.

As an example, these equations will be applied to the data represented in Figure 1A. The effect of carbamylcholine on each of the potential differences was measured consecutively on the same cell with a 10-20-minute recovery between responses; ideally, the changes should be measured simultaneously (only two of the three have to be measured). However, normally (and in this case) the cell repolarizes to the initial resting potential following a depolarization, and the curve obtained by subtracting  $\psi_I$  from  $\psi_N$  is superimposable upon the measured trace of  $\psi_{N-I}$ . (Neither condition applies to the data in Figure 1B obtained in the presence

of ouabain, which prevents the cell from returning to its initial state.) The assumed fixed values are  $\rho = 0.01$  and  $M = 180$  mM; also,  $\text{Na}_o = 165$  mM and  $\text{K}_o = 5$  mM. Initially,  $\psi = \psi_I = \psi_N = 77$  mV, and equations (6) and (7) yield the initial values,  $\text{K}_i = 145$  mM,  $\text{Na}_i = 35$  mM. Seventy-five seconds after the addition of  $40 \mu\text{M}$  carbamylcholine,  $-\Delta\psi_N = 12.5$  mV and  $\Delta\psi_{N-I} = 16$  mV. Equation (8) yields  $\text{K}_i' = 88$  mM, and then equation (7),  $\text{Na}_i' = 92$  mM; i.e.,  $\text{K}_i - \text{K}_i' = \text{Na}_i' - \text{Na}_i = 57$  mM. With these values for  $\Delta\psi_{N-I}$ ,  $\text{Na}_i'$ , and  $\text{K}_i'$  equation (9) then yields  $\rho' = 0.049$ . (The approximate equation (11) gives  $\rho' = 0.046$ .)  $\rho'$ , it should be pointed out, is a characteristic of the whole innervated membrane; the contributions of the junctional and extrajunctional regions to  $\rho$  and  $\rho'$  will depend on the relative permeabilities of the two areas at rest and during a depolarization.

The results of the analysis are that in the first 75 seconds of an approximately half-maximal response 40 per cent of the intracellular  $\text{K}^+$  exchanges for  $\text{Na}^+$  and that the ratio of the permeability of  $\text{Na}^+$  to that of  $\text{K}^+$  increases fivefold. While the latter estimate seems quite reasonable, the former, although qualitatively consistent with the effect of ouabain on the repolarization, may be somewhat high. The net outward  $\text{K}^+$  flux required for such an exchange is 5- to 50-fold greater than a value for the unidirectional  $\text{K}^+$  flux previously reported.<sup>6</sup> The estimates of  $\text{K}_i$  and  $\text{K}_i'$ , and hence of the flux, are subject to the errors inherent in the measurement of the resting potential with internal glass microelectrodes. The estimate of  $\rho'$ , however, which depends most strongly on the transcellular potential, should be more reliable, and should provide, despite the simplifications involved, a better measure than the extent of depolarization of the primary response of the electroplax to acetylcholine and its congeners.

*Summary.*—Carbamylcholine-induced depolarizations of the isolated electroplax appear to involve initial changes in ion permeability and subsequent changes in intracellular ion concentrations. The significance of the concentration changes is indicated by the inhibition by ouabain of the repolarization and by the depolarization of the noninnervated and inexcitable membrane, as well as the innervated membrane, when carbamylcholine is applied to the innervated side. The changes in concentration and in permeability are estimated by an analysis based on the Goldman equation and on the functional asymmetry of the electroplax. The estimate of the permeability change is proposed as a measure of the primary response of the electroplax to acetylcholine and its congeners.

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<sup>16</sup> Equation (1) yields:  $\sigma \cong \frac{(e^{F\Delta\psi/RT} - 1)K_i}{Cl_o' - Cl_o e^{F\Delta\psi/RT}}$ , where  $\Delta\psi = \psi' - \psi$ , the potential after minus the potential before changing the solution,  $Cl_o'$  is the final chloride concentration and  $Cl_o$  the initial concentration, and  $K_i + \rho Na_i$  has been approximated by  $K_i$  (= 170 mM).

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