

# Substrate Transport Activation Is Mediated through Second Periplasmic Loop of Transmembrane Protein MalF in Maltose Transport Complex of *Escherichia coli*\*

Received for publication, January 8, 2012, and in revised form, March 2, 2012. Published, JBC Papers in Press, March 26, 2012, DOI 10.1074/jbc.M112.340679

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**Background:** ABC transporters are important substrate transport systems in all forms of life.

**Results:** Residual dipolar coupling (RDCs) and solvent-<sup>1</sup>H-NOE experiments are employed to characterize structural changes of MalE, induced upon addition of MalF-P2.

**Conclusion:** MalF-P2 induces an activated structural conformation of the substrate-binding protein MalE, which is required for substrate transport.

**Significance:** This work gives insight into the activation mechanism of ABC transporters.

In a recent study we described the second periplasmic loop P2 of the transmembrane protein MalF (MalF-P2) of the maltose ATP-binding cassette transporter (MalFGK<sub>2</sub>-E) as an important element in the recognition of substrate by the maltose-binding protein MalE. In this study, we focus on MalE and find that MalE undergoes a structural rearrangement after addition of MalF-P2. Analysis of residual dipolar couplings (RDCs) shows that binding of MalF-P2 induces a semiopen state of MalE in the presence and absence of maltose, whereas maltose is retained in the binding pocket. These data are in agreement with paramagnetic relaxation enhancement experiments. After addition of MalF-P2, an increased solvent accessibility for residues in the vicinity of the maltose-binding site of MalE is observed. MalF-P2 is thus not only responsible for substrate recognition, but also directly involved in activation of substrate transport. The observation that substrate-bound and substrate-free MalE in the presence of MalF-P2 adopts a similar semiopen state hints at the origin of the futile ATP hydrolysis of MalFGK<sub>2</sub>-E.

ATP-binding cassette (ABC)<sup>4</sup> transporters are active transporters of substrate that utilize the chemical energy stored in

\* This work was supported by the Center for Integrated Protein Science Munich.

<sup>1</sup> Supported by the Leibniz-Gemeinschaft, the Helmholtz-Gemeinschaft, Deutsche Forschungsgemeinschaft Grants Re1435, SFB449, SFB740, SFB610, and by access to research infrastructures activity in the 7th Framework Programme of the EC Project 261863, Bio-NMR.

<sup>2</sup> Supported by Deutsche Forschungsgemeinschaft Grants SFB 449, TP B14, and SCHN 274/9-3.

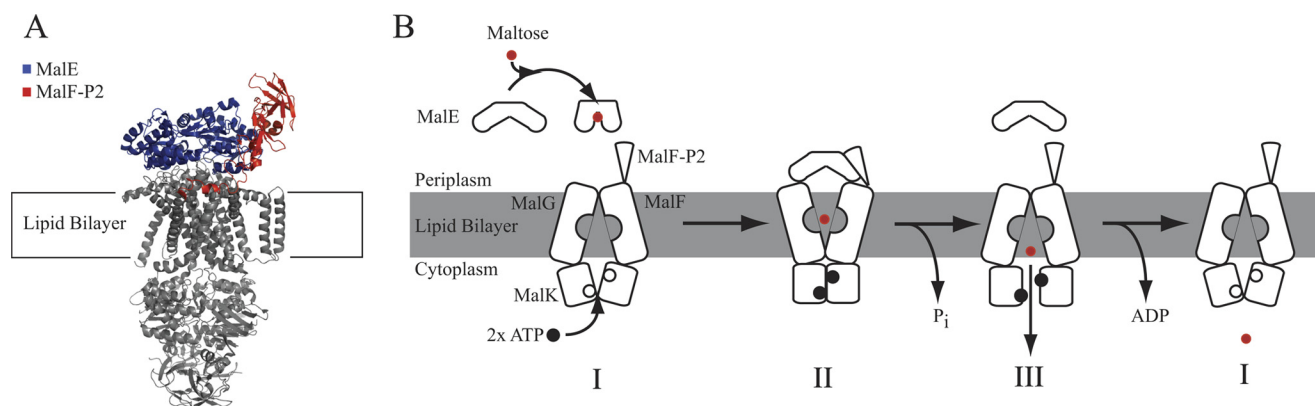
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<sup>4</sup> The abbreviations used are: ABC, ATP-binding cassette; HSQC, heteronuclear single-quantum coherence; ITC, isothermal titration calorimetry; PDB, Protein Data Bank; PRE, paramagnetic relaxation enhancement; RDC, residual dipolar coupling; r.m.s.d., root mean square deviation; TMD, transmembrane domain; TROSY, transverse relaxation optimized spectroscopy.

ATP to translocate the substrate across cellular membranes (1). Most ABC transporters are importers of substrate in bacteria and include a substrate-binding protein, two integral membrane components with low sequential homology, and two conserved nucleotide-binding domains (2). One of the best studied systems is the maltose transporter MalFGK<sub>2</sub>-E of *Escherichia coli*/*Salmonella* that consists of the periplasmic maltose-binding protein MalE, the two integral cytosolic membrane proteins MalF and MalG, and two copies of the nucleotide-binding domains MalK (3). A multitude of x-ray structures exists of the maltose transporter in its resting state, a catalytic-intermediate state and most recently in a pretranslocational state (4–7). It has been shown that the rigid-body rotation of the nucleotide-binding domains is confined to ATP binding and hydrolysis, which coincides with the conformational changes during substrate transport (7–9). From structural information together with biochemical data, a model for the substrate translocation process has been developed for the ABC import system, the alternating-access model (10). An adaptation of this model to the MalFGK<sub>2</sub>-E system is shown in Fig. 1. To date it is not known how the translocation process is initiated. Especially the highly dynamic region of the periplasmic side of the transporter in interaction with the substrate-binding protein MalE has eluded high-resolution information. In a recent study we described the second periplasmic loop P2 of MalF as a key player in the recognition of substrate via MalE (11). In this study we show that MalF-P2 not only is responsible for substrate recognition but also directly involved in activation of the MalFGK<sub>2</sub>-E complex by inducing structural changes in maltose-bound MalE.

## EXPERIMENTAL PROCEDURES

**Sample Preparation**—Unlabeled MalF-P2 and MalE were prepared as described (11) in LB medium. Uniformly <sup>15</sup>N,<sup>2</sup>H-labeled MalE (U-<sup>15</sup>N,<sup>2</sup>H MalE) was expressed using



**FIGURE 1. Substrate transport model for MalFGK<sub>2</sub>-E.** *A*, crystal structures of MalFGK<sub>2</sub>-E (6), MalE, and MalF-P2 color-coded in *blue* and *red*, respectively. *B*, transport model for ABC importer systems adapted to the MalFGK<sub>2</sub>-E system (10). The model contains three distinct steps. First, ATP is loaded into the MalK dimer, and the interface closes upon binding of substrate-loaded MalE at the periplasmic surface (*II*). Upon ATP hydrolysis substrate is released into the binding pocket of MalG,F and is transported into the cytoplasm (*III*). After substrate transport, ADP is released, and the transporter goes back to its resting-state (*I*). Amplitudes of motion for the TMDs are exaggerated in the representation to highlight changes. A TMD (MalF,G) rotation from the inward to the outward conformation was determined to be on the order of 22° (4). Fig. adapted from Locher (10).

M9 minimal medium supplemented with D<sub>2</sub>O and <sup>15</sup>NH<sub>4</sub>Cl as the sole deuterium and nitrogen source. Protein was stored in 20 mM phosphate buffer, pH 7.4, 100 mM NaCl at -80 °C for later use.

**NMR Experiments and Analysis**—Experiments were performed with 0.3–1 mM U-<sup>15</sup>N,<sup>2</sup>H MalE samples at 310 K in 20 mM NaH<sub>2</sub>PO<sub>4</sub>/Na<sub>2</sub>HPO<sub>4</sub>, pH 7.4, 100 mM NaCl, in 95% H<sub>2</sub>O/5% D<sub>2</sub>O. Data were recorded using Bruker 750 and 900 MHz spectrometers (Bruker Biospin, Karlsruhe, Germany) equipped with triple resonance cryogenic probes. Resonance assignments were transferred from earlier studies (12, 13).

NMR experiments were carried out at equimolar ratio of U-<sup>15</sup>N,<sup>2</sup>H MalE and maltose, 500 μM. To investigate the MalE/MalF-P2 complex, a molar ratio of 1:1.2 of U-<sup>15</sup>N,<sup>2</sup>H MalE (500 μM) and nonlabeled MalF-P2 (600 μM) in the absence and presence of 500 μM maltose was employed. A 20% molar excess of MalF-P2 was used to assure that the MalE/MalF-P2 complex was observed. The interaction of MalE to maltose in the presence and absence of MalF-P2 was observed by mapping chemical shift perturbations of backbone resonances. Residual dipolar coupling measurements were carried out using liquid-crystalline Pf1 filamentous phages (Profos AG) as an alignment medium. A concentration of 7 mg/ml Pf1 yielded a residual quadrupolar coupling of the D<sub>2</sub>O resonance of 7.8 Hz (14, 15). <sup>1</sup>H-<sup>15</sup>N HSQC and <sup>1</sup>H-<sup>15</sup>N HSQC TROSY experiments were recorded to obtain the <sup>15</sup>N *J*-coupling for the alignment tensor analysis of MalE (16).

RDC experiments were measured for the maltose-bound MalE in the absence (reference experiment) and presence of MalF-P2, and for the MalE/MalF-P2 complex in the absence of maltose. RDC values were fitted in Module v1.0 (17) using the structures of substrate-free MalE (PDB ID code 1PEB), substrate-bound MalE (PDB ID code 1MPD), and MalE as observed in the MalFGK<sub>2</sub>-E complex (PDB ID code 2R6G). Alignment tensors of the N- and C-terminal domains of MalE in the presence and absence of maltose were fitted with a 100-step Monte Carlo error analysis in Module (17). Consequently, the alignment tensor was fitted again using a com-

mon alignment frame. After rigid-body modulation to obtain the orientations of the N- and C-terminal domains in the common alignment frame, the resulting MalE structure in the MalE/MalF-P2 complex was acquired (17). The procedure was carried out for MalE in the presence and absence of maltose.

Boundaries of the N- and C-terminal domains were set accordingly, N: residues 1–110; 260–313 and C: residues 115–257; 335–370, and the linking domain: residues 314–334. The linking domain was omitted from the RDC fitting. Resulting PDBs and visualization of alignment tensors from the rigid-body modeling were derived from Module v1.0 (17).

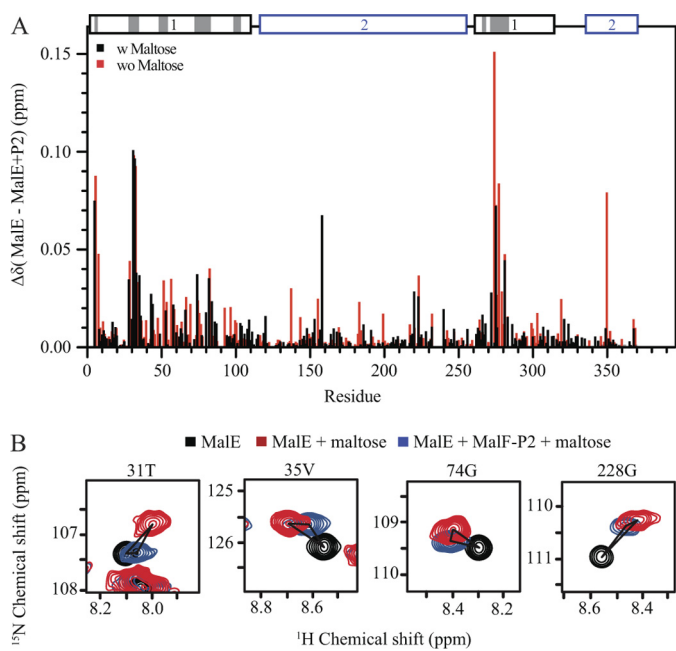
**Solvent-PRE Measurements**—Paramagnetic relaxation enhancement (PRE) experiments were obtained by quantification of the HSQC peak intensities for maltose-bound MalE in the presence and absence of MalF-P2. Solvent-PREs were measured by the addition of chelated Gd<sup>3+</sup> (OmniScan) to the sample buffer. Stock solutions of Gd<sup>3+</sup> were prepared in sample buffer, which was titrated into the protein solution to yield a concentration of 0.5–50 mM Gd<sup>3+</sup>. Spectra were acquired with 0.5, 1.0, 2.0, 3.0, 5.0, 10.0, 25.0, and 50.0 mM Gd<sup>3+</sup>. Concentration-dependent HSQC cross-peak intensities *I<sub>i</sub>* were fitted according to Equation 1,

$$\frac{I_i(C)}{I_{ref}} = \exp(-\epsilon_i T c) \quad (\text{Eq. 1})$$

and evaluated according to Hilty *et al.* (18).  $\epsilon_i$ , *T*, and *c* refer to the relative solvation rate, the INEPT, delay and the Gd<sup>3+</sup> concentration, respectively. No significant changes in <sup>15</sup>N line widths of MalE due to chemical exchange upon addition of MalF-P2 in the presence and absence of maltose were observed.

All NMR spectra were processed with TopSpin v2.0 (Bruker Biospin, Karlsruhe, Germany) and analyzed with CCPN analysis v2.4 (19). Solvent PREs and correlation factors for RDCs were fitted using Prism v5.0b (GraphPad). RDC alignment tensor fitting and rigid-body modulations were accomplished in Module v1.0 (17).

## Substrate Transport Activation of Maltose Transport Complex

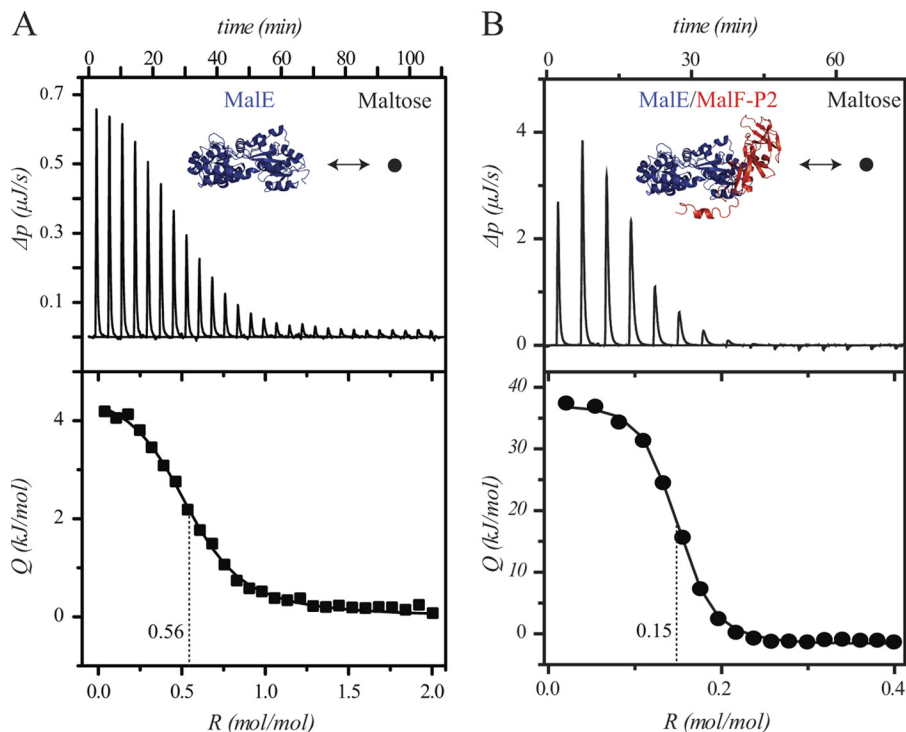


**FIGURE 2. Chemical shift titration experiments of MalE and MalF-P2 in the presence and absence of maltose.** *A*, resonance-specific  $^{15}\text{N}$ - $^1\text{H}$  chemical shift changes upon titration of MalF-P2 to  $^{15}\text{N}$ - $^1\text{H}$  MalE, in the presence (black) and absence (red) of maltose. Two distinct binding regions in the N-terminal domain of MalE are observed. These regions match the contacts seen in the crystal structure of MalFGK<sub>2</sub>-E (PDB ID code 2R6G) (6). MalE and MalF-P2 contacts (with  $\Delta(\text{N}, \text{C}, \text{O}) < 5 \text{ \AA}$ ) in the MalFGK<sub>2</sub>-E crystal structure are highlighted in gray on the primary sequence (top). In addition, small chemical shift changes are observed for residues not directly involved in the interaction. *B*,  $^{15}\text{N}$ - $^1\text{H}$  correlation spectra for selected residues, not located in the binding interface. Resonances are color coded, MalE (black), MalE/maltose (blue), and MalE/MalF-P2/maltose (red). Chemical shift analysis shows resonances distinct from the maltose-bound and maltose-free MalE, implying that MalE in the presence of MalF-P2 adopts a different conformation.

**Isothermal Titration Calorimetry (ITC) Experiments**—MalE and MalF-P2 ITC samples were extensively dialyzed overnight at 7 °C (Spectral/Por, Spectrum Laboratories) either separately or mixed at a 1:1.2 (MalE:MalF-P2) molar ratio, with concentrations ranging from 0.5 to 1 mM in 20 mM phosphate buffer, pH 7.4, 100 mM NaCl, with light stirring. Maltose stocks were prepared from the dialysis buffer at concentrations ranging from 0.5 to 2 mM. MalE, MalF-P2, MalE/MalF-P2, and maltose samples were filtered and degassed before insertion to the injection syringe and calorimeter cell.

High-sensitivity microcalorimetry was performed on a VP-ITC instrument (MicroCal). Experiments of maltose binding to MalE were performed at 5 °C with either MalE only or a 1:1.2 molar ratio of MalE:MalF-P2 in the calorimeter cell and maltose solution in the injection syringe. Time spacings between injections were chosen long enough to allow for complete reequilibration (600 s). Typically, 20–40 injections were performed. Base-line subtraction and peak integration were performed as described by the manufacturer (MicroCal) with Origin 5.0 (OriginLab). Association curves were fitted with a nonlinear one-site binding model, yielding the binding constant and molar ratio of the interaction.

Determination of the dissociation constant for binding of maltose to MalE was performed employing 40  $\mu\text{M}$  MalE in the calorimetry cell and 500  $\mu\text{M}$  maltose in the syringe. For measurement of binding of maltose to the MalE/MalF-P2 complex a solution of 75  $\mu\text{M}$  MalE and 90  $\mu\text{M}$  MalF-P2 was provided in the calorimetry cell and a 500  $\mu\text{M}$  maltose solution in the syringe. Control experiments were performed to find out whether maltose interacts with MalF-P2 alone.



**FIGURE 3. ITC experiments.** *A*, maltose titrated into MalE;  $K_D = 2.18 \pm 0.091 \mu\text{M}$ ,  $\Delta H = 4.467 \pm 0.037 \text{ kJ}$ ,  $\Delta S = 41.8 \text{ J/K}$ . *B*, maltose titrated into MalE and MalF-P2 (1:1.2 molar ratio);  $K_D = 0.121 \pm 0.043 \mu\text{M}$ ,  $\Delta H = 37.59 \pm 0.849 \text{ kJ}$ ,  $\Delta S = 166 \text{ J/K}$ . All experiments were carried out at 5 °C. In the presence of MalF-P2 the apparent stoichiometry is reduced to ~25%, indicating that one quarter of all MalE/MalF-P2 complexes are able to coordinate a maltose molecule.

Experiments were performed with MalF-P2 (50  $\mu\text{M}$ ) in the calorimetry cell and a 500  $\mu\text{M}$  maltose solution in the syringe. No interaction could be detected between maltose and MalF-P2 alone.

**Maltose Binding Assay**—Binding of [ $^{14}\text{C}$ ]maltose to MalE in the absence and presence of MalF-P2 was monitored by a filtration assay according to Richarme and Kepes (20). Briefly, purified MalE (2.5  $\mu\text{M}$ ) was incubated with [ $^{14}\text{C}$ ]maltose (12  $\mu\text{M}$ ) in the absence or presence of MalF-P2 (2.5 and 5  $\mu\text{M}$ , respectively) in 20 mM phosphate buffer, pH 7.4, 100 mM NaCl at 25  $^{\circ}\text{C}$  for 15 min. The reaction was terminated by adding 2 ml of ice-cold saturated  $(\text{NH}_4)_2\text{SO}_4$ , and the mixture was immediately passed through a nitrocellulose filter (0.45- $\mu\text{m}$  pore size). After two washes with 5 ml of the  $(\text{NH}_4)_2\text{SO}_4$  solution followed by 5 ml of distilled water, the radioactivity retained on the filters was determined in a liquid scintillation counter.

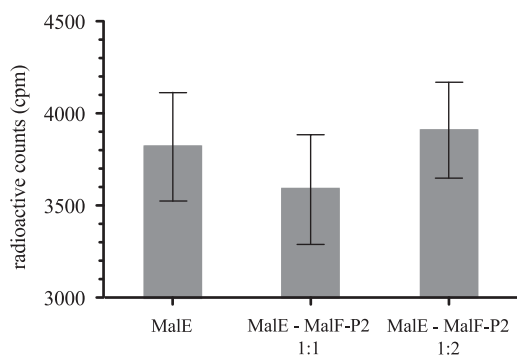
**Molecular Dynamics Simulation**—The MalFGK<sub>2</sub>-E crystal structure (PDB ID code 2R6G) was taken as the starting point to perform a 50-ns molecular dynamics simulation in GROMACS using the GROMOS force field (ffG53a6) (21) of MalE and MalF-P2. The proteins were placed into a triclinic box. The simulations were done under physiological conditions. For this purpose the proteins and the complex structure were calculated in a 0.1 M NaCl (*i.e.*  $\sim$ 15,600 H<sub>2</sub>O molecules, 45 Na<sup>+</sup> and

33 Cl<sup>-</sup> ions, respectively) solution at 310 K and periodic boundary conditions for the box. Short-range nonbonded interactions were calculated with the Cut-off Lennard-Jones potential up to a distance of 1.2 nm between the interacting atoms. For long-range electrostatic interactions the Particle Mesh Ewald (PME) option was used with a grid spacing of 0.12 nm. The bond length of a heavy atom (C, N, O) to an H-atom in protein or water was constrained using a LINCS and SETTLE algorithm, respectively. Every simulation ensemble was energy minimized in a two-step minimization strategy. In a first step steepest descent and in a second a conjugate gradient minimizing routine were used. After equilibration over a period of 500 ps using a positional restraint of 1000 kJmol<sup>-1</sup> nm<sup>-2</sup> on the backbone of the two proteins molecule, a free simulation was performed for a period of 50 ns for the MalE/MalF-P2 complex. The temperature and pressure (1 bar) were controlled using the Berendsen coupling with a relaxation time of 0.1 ps for temperature and 1 ps for pressure. During the course of the simulation, the actual frame was stored every 5 ps. Visualization of trajectories and arrangement of figures was made with VMD (22).

## RESULTS

In an earlier report it was shown that the periplasmic loop P2 of MalF in the MalFGK<sub>2</sub> transporter folds independently in solution and can interact with MalE in the presence and absence of maltose (11). Here, the interaction of MalE to MalF-P2 was investigated.

**NMR Titration of MalF-P2 to U-<sup>15</sup>N,<sup>2</sup>H MalE in Presence and Absence of Maltose**—<sup>1</sup>H-<sup>15</sup>N correlation spectra were recorded with 500  $\mu\text{M}$  U-<sup>15</sup>N,<sup>2</sup>H MalE and 600  $\mu\text{M}$  unlabeled MalF-P2 in the presence and absence of 500  $\mu\text{M}$  maltose. Chemical shift changes are mapped onto the primary sequence in Fig. 2. The largest chemical shift changes ( $\Delta\delta > 0.05$  ppm) are located predominantly in the N-terminal lobe of MalE. Surprisingly small but significant chemical shift changes are observed along the full primary sequence of MalE ( $< 0.02$  ppm), regardless of proximity to the substrate (Fig. 2). Spectra of maltose-bound and maltose-free MalE were superimposed with spectra of maltose-bound MalE in the presence of MalF-P2. A large number of MalE resonances from the MalE/MalF-P2/maltose complex spectra show chemical shifts that are distinct to either the maltose-free or the maltose-bound state of MalE. The



**FIGURE 4. Radioactive maltose precipitation experiments.** All experiments were performed with equimolar amounts of MalE and maltose in the presence or absence of MalF-P2 at a 1:1 or 1:2 molar ratio. Radioactive counts from filter assays (20) of [ $^{14}\text{C}$ ]maltose were measured by a liquid scintillation counter. Experiments were performed in duplicate for each sample. No significant difference of free maltose in the supernatant is observed after precipitation of the protein.

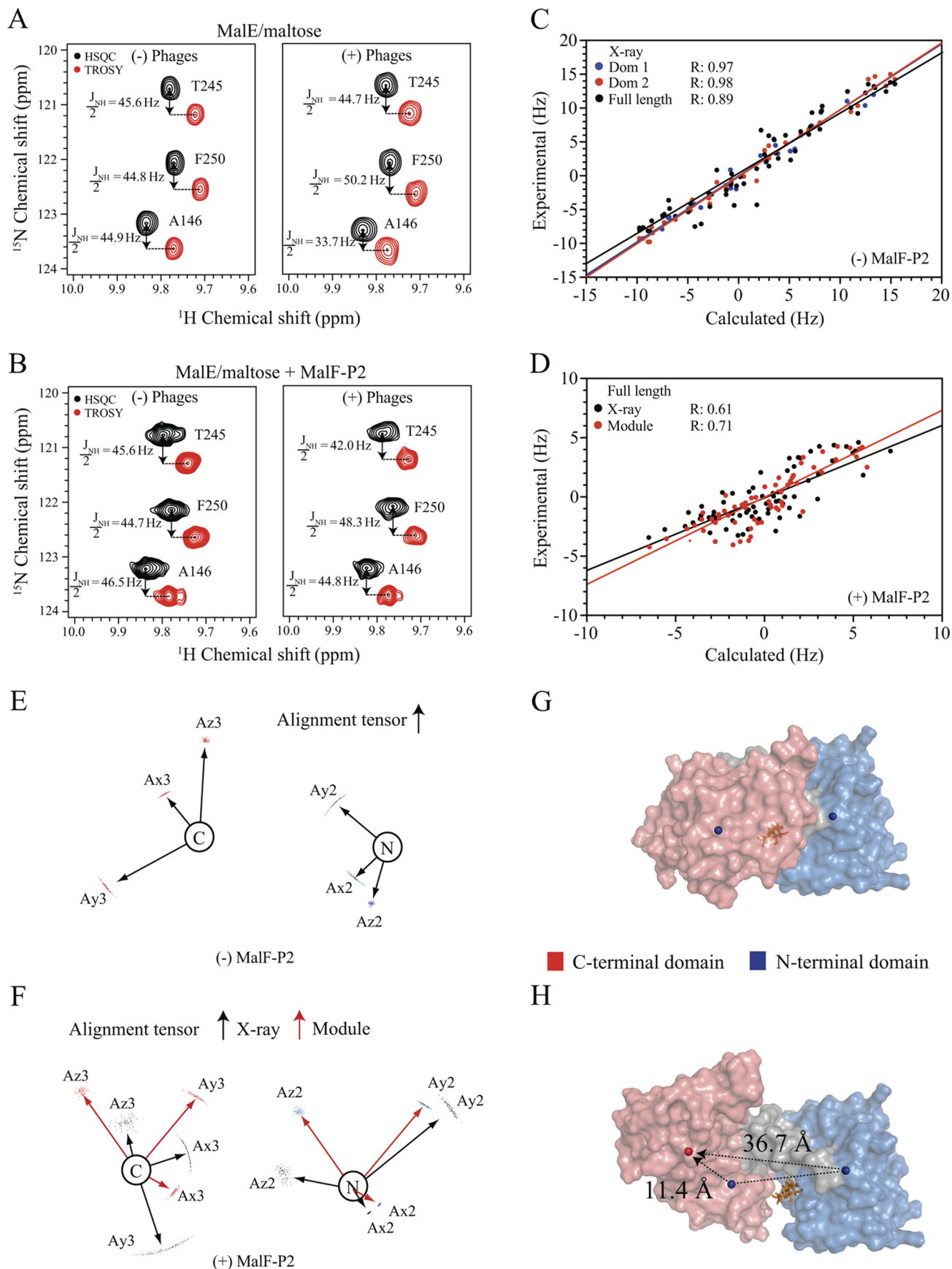
**TABLE 1**

**R values of fitted RDC data of substrate-bound MalE/MalF-P2**

Correlation factors are calculated with respect to the full structure, the two individual domains for substrate-free and -bound MalE, and the intermediate state in the MalFGK<sub>2</sub>-E complex. A good correlation is observed for the full-length structure and the individual domains in the reference experiment (MalE/maltose). Fitted H<sup>N</sup>-N RDCs from the MalE/MalF-P2/maltose complex to known structures of MalE show a decreased correlation factor for the individual domains and the full-length structure. MalE/MalF-P2/maltose-derived RDCs show a better correlation with respect to the full-length structure of MalE with corrected relative orientations of the individual domains by Module (17). A similar dependence is observed for RDCs derived from MalE/MalF-P2 in the absence of maltose.

Complex	Domain 1	Domain 2	Full-length
H <sup>N</sup> -N RDC MalE/maltose			
MalE with maltose (PDB: 1MPD)	0.97	0.98	0.89
MalE without maltose (PDB: 1PEB)	0.94	0.90	0.63
H <sup>N</sup> -N RDC MalE/MalF-P2/maltose			
MalE with maltose (PDB: 1MPD)	0.65	0.76	0.61
MalE without maltose (PDB: 1PEB)	0.61	0.68	0.56
MalE (MalFGK <sub>2</sub> -E) (PDB: 2r6g)	0.68	0.56	0.47
MalE/MalF-P2/maltose (Module)	0.65	0.76	0.71
H <sup>N</sup> -N RDC MalE/MalF-P2			
MalE without maltose (PDB: 1PEB)	0.62	0.78	0.63
MalE/MalF-P2/maltose (Module)	0.62	0.78	0.68

# Substrate Transport Activation of Maltose Transport Complex



cross-peaks are also not positioned in between the open and closed state of MalE (Fig. 2). Fast exchange between the two states can therefore be excluded. The distinct chemical shifts rather hint at a third conformational state of MalE when bound to MalF-P2.

**Effect of Maltose Binding to MalE after Complex Formation with MalF-P2**—ITC experiments were performed to evaluate whether MalE can bind maltose when complexed to MalF-P2. Reference ITC experiments of maltose binding to MalE show values similar to those earlier reported (23). When titrating maltose into the measurement cell filled with a 1:1.2 molar ratio of MalE:MalF-P2, the fitted binding curve yields comparable binding affinity. However, the stoichiometry,  $R$ , is reduced to a fourth, indicating a lower activity of MalE relative to the reference experiment (Fig. 3) and indicating that release and rebinding of maltose to MalE only occur at negligible rates compared with the uncomplexed state.

To evaluate whether MalF-P2 can stimulate release of maltose from maltose-bound MalE, radioactive maltose precipitation experiments were performed. Maltose-bound MalE was precipitated in the absence or presence of either equimolar or 2-fold excess of MalF-P2 in solution. Radioactive counts were measured before and after precipitation of proteins in solution, and all experiments were performed in duplicate (Fig. 4). No significant difference in maltose counts was found for the maltose-bound MalE in the absence or presence of MalF-P2 in solution, implying that maltose is maintained bound in the substrate-pocket of MalE in the presence of MalF-P2.

**RDC Measurements Elucidate Structural Changes of MalE**— $H^N$ -N RDCs were extracted from spectra of maltose-bound MalE in the absence and presence of MalF-P2, 182 and 131 respectively. RDCs were analyzed employing the substrate-free, the maltose-bound MalE and the MalE structure in a catalytic intermediate state of MalFGK<sub>2</sub>-E (PDB ID codes 1PEB, 1MPD, and 2R6G, respectively). Further, the structures were separated into the N-terminal (residues 1–110; 260–313) and C-terminal (residues 115–257; 335–370) domains to further evaluate the relative orientation of the two domains to one another. The linker regions between the N- and C-terminal domains were omitted in the analysis. Reference RDCs of MalE/maltose show a high correlation coefficient for the individual N- and C-terminal domains,  $R \sim 0.97$ , and their relative orientation,  $r = 0.89$  (Table 1), with respect to the x-ray structure (PDB ID code 1MPD).  $H^N$ -N RDCs were then extracted from the MalE/MalF-P2/maltose complex. The extracted RDCs were fitted to the

**TABLE 2**

**Ratio of solvent-PRE relaxation rates for substrate-bound MalE in presence and absence of MalF-P2**

Relaxation rates of residues lining the binding pocket and/or directly associated with substrate binding were extracted and compared relatively with each other. A >5-fold increase in relaxation rates of residues located in the vicinity of the maltose-binding pocket is observed after addition of MalF-P2.

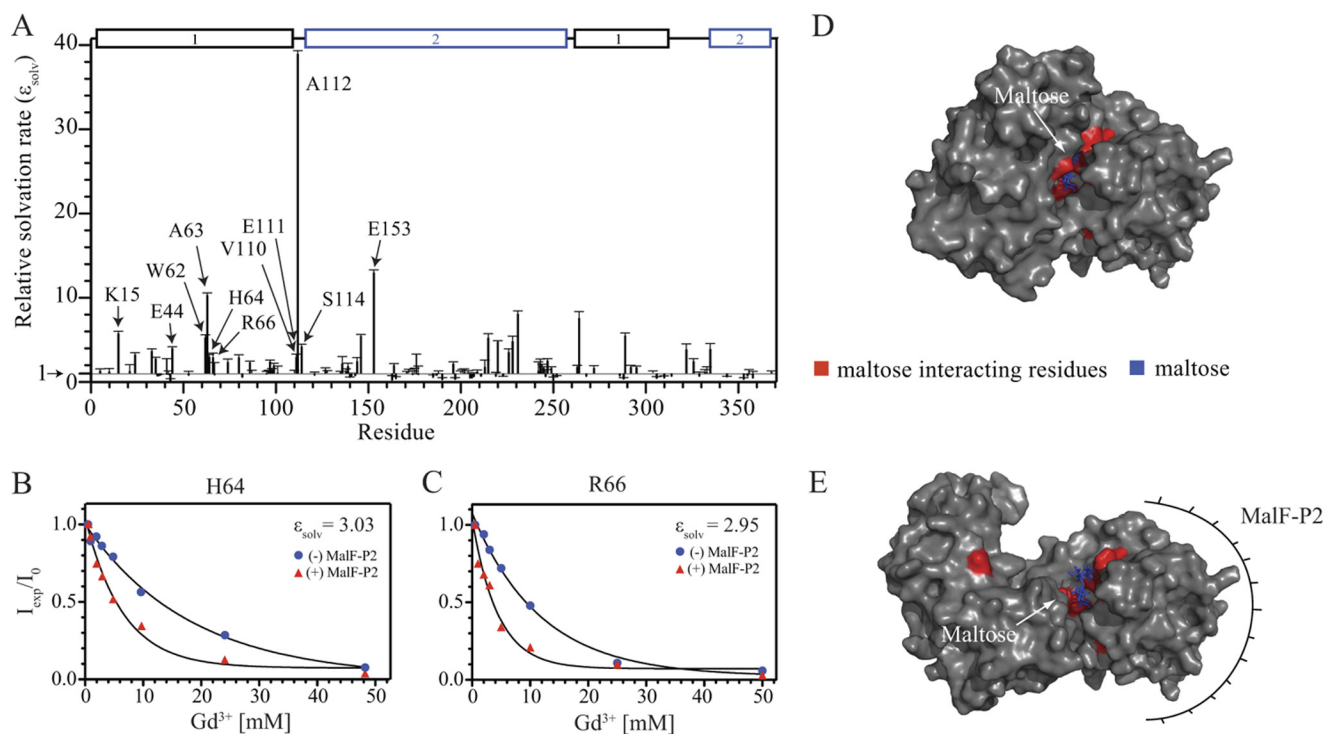
Residue	$\epsilon_{\text{MalE/MalF-P2}}/\epsilon_{\text{MalE}}$	Residue	$\epsilon_{\text{MalE/MalF-P2}}/\epsilon_{\text{MalE}}$
Lys <sup>15</sup>	5.78 ± 0.28	Val <sup>110</sup>	1.02 ± 0.43
Glu <sup>44</sup>	4.02 ± 0.20	Glu <sup>111</sup>	2.96 ± 0.38
Trp <sup>62</sup>	5.33 ± 0.30	Ala <sup>112</sup>	38.9 ± 0.40
Ala <sup>63</sup>	10.3 ± 0.28	Ser <sup>114</sup>	4.22 ± 0.31
His <sup>64</sup>	3.03 ± 0.29	Glu <sup>153</sup>	13.0 ± 0.34
Arg <sup>66</sup>	2.95 ± 0.54		

maltose-free and maltose-bound structures of MalE as well as the MalE structure from the MalFGK<sub>2</sub>-E vanadate-trapped intermediate state (PDB ID codes 1PEB, 1MPD, and 2R6G, respectively) (Table 1). A correlation coefficient of  $r = 0.65$ , 0.76, and 0.61 for the N- and C-terminal domain and the full-length structure, respectively, is observed. The lower correlation coefficient for the individual N- and C-terminal domains can be directly interpreted as a consequence of a conformational rearrangement in the binding interface and the C-terminal domain. Minor structural changes are corroborated by chemical shift changes in the C-terminal domain in the HSQC titration experiments (Fig. 2). The reduced correlation coefficient for the full structure, relative to the individual domains, shows that the relative orientation of the individual domains deviates from all of the previously solved x-ray structures (PDB ID codes 1PEB, 1MPD, and 2R6G).

Fitted alignment tensors from the MalE/MalF-P2/maltose complex to the individual domains used a 100-step Monte Carlo analysis in Module v1.0 (17) yielding a good convergence. The obtained alignment tensors for the individual N- and C-terminal domains are on the same order of magnitude, which allows for rigid-body modulation (Fig. 5) (17). The alignment tensors from the N- and C-terminal domains of MalE/MalF-P2/maltose are then set to a common alignment frame and allowed to undergo rigid-body rotations to accommodate the common alignment frame. MalE/MalF-P2/maltose-derived RDCs were then fitted to the resulting full-length structure after rigid-body modulation. This yields the relative orientation of the two domains in the MalE/MalF-P2/maltose complex (Fig. 5). An increase in the correlation factor is now observed ( $r = 0.71$ ), indicating that an improved relative domain orientation is found (Table 1). The reorientation of the relative domains to each other

**FIGURE 5. RDC experiments with maltose-bound U-<sup>15</sup>N, <sup>2</sup>H MalE in the presence and absence of MalF-P2.** A and B, HSQC and TROSY spectra yielding the  $J_{NH}/2$ -coupling for selected residues in isotropic (left) and anisotropic (right) buffer for MalE/maltose and MalE/MalF-P2/maltose, respectively. C, experimental and calculated residual dipolar couplings for MalE/maltose and their correlation factors, using the N- (blue) or C-terminal (red) domain or full-length (black) crystal structure of maltose-bound MalE (PDB ID code 1MPD). D, experimental and calculated residual dipolar couplings for MalE/MalF-P2/maltose with respect to the full-length MalE/maltose crystal structure (black) and subsequently to the Module (17)-derived MalE structure. Module uses RDC-derived alignment tensors for individual domains and applies rigid-body modulation to optimize their relative domain orientation. The increased correlation factor in the Module-derived MalE structure from MalE/MalF-P2/maltose RDCs shows that the relative orientation of the individual domains changes in the presence of MalF-P2. E, alignment tensors for the N- and C-terminal domains from MalE/maltose obtained from residual dipolar couplings. F, alignment tensors of the N- and C-terminal domains fitted to the MalE/maltose crystal structure (black) and alignment tensors after correction of the relative domain orientation in Module (red). G and H, surface representation of the MalE/maltose crystal structure and the Module-obtained MalE in the MalE/MalF-P2/maltose complex. The N- and C-terminal domains and their center-of-masses are indicated as blue spheres for the MalE/maltose structure (G). The corrected center-of-mass C-terminal domain is represented by a red sphere (H). Changes in center-of-mass distances from the reference domain orientation to the corrected relative orientation are marked with black dashed lines. Maltose is represented in stick model (orange) in the binding pocket.

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**FIGURE 6. Residue-specific solvent accessibility of MalE in the absence and presence of MalF-P2.** *A*, relative solvent accessibility rates  $\epsilon_{\text{solv}}$  determined by solvent-PRE measurements. Conformational changes due to the complex formation of MalE/MalF-P2/maltose induce small changes of solvation rates over the full primary sequence. Residues in the vicinity of the maltose-binding pocket (marked with *arrows*) show similar or significantly higher solvation ratios in the presence of MalF-P2. *B* and *C*, relaxation rates for residues His<sup>64</sup> and Arg<sup>66</sup>, interacting directly with maltose, in the absence (*blue spheres*) and presence (*red triangles*) of MalF-P2, respectively. *D* and *E*, surface representations of the MalE/maltose structures in the absence (PDB ID code 1MPD) and presence of MalF-P2 (Module-derived), respectively. Maltose (*blue stick model*) and interacting residues (*red*) are labeled in the structures.

results in exposure of the binding pocket of MalE to the solvent in the MalE/MalF-P2/maltose complex.

Moreover, RDCs extracted for MalE in the MalE/MalF-P2 complex in the absence of maltose yield a similar relative orientation of the N- and C-terminal domains as found in the MalE/MalF-P2/maltose complex. Thus, the domain reorientation of MalE in complex with MalF-P2 seems independent of the substrate.

**Solvent Exposure of Binding Cleft of MalE by Solvent-PRE Measurements**—To assess the solvent accessibility of the binding cleft in maltose-bound MalE qualitatively, in the presence and absence of MalF-P2 (Table 2), solvent-PRE measurements with chelated  $\text{Gd}^{3+}$  (OmniScan) in solution were performed.  $\text{Gd}^{3+}$  is a paramagnetic ion that drastically increases the relaxation rates of hydrogen atoms in its vicinity. Relaxation rates were measured via  $^1\text{H}$ - $^{15}\text{N}$  HSQC experiments recorded with increasing concentrations of  $\text{Gd}^{3+}$  (0.5–50 mM) (Fig. 6). MalE amide  $^1\text{H}$ ,  $^{15}\text{N}$  cross-peak intensities were fitted with respect to the  $\text{Gd}^{3+}$  concentration. In principle, a reduced solvent accessibility is expected for residues in MalE that directly interact with MalF-P2. Residues not interacting with MalF-P2 should not experience any changes. However, the opposite effect was observed in our experiments. We found strongly increased solvation rates for residues Lys<sup>15</sup>, Glu<sup>61</sup>, Trp<sup>62</sup>, Ala<sup>63</sup>, Arg<sup>66</sup>, Val<sup>110</sup>, Glu<sup>111</sup>, Ala<sup>112</sup>, and Glu<sup>153</sup>, located directly in the substrate-binding pocket. As these residues are not directly involved in MalF-P2 binding, a change in the solvent accessibility of the binding cleft must occur. This is corroborated by the

conformational changes in the N- and C-terminal domain of MalE, as seen in the changes of the RDC values.

Surface representations of the MalE/maltose structures, in the absence (PDB ID code 1PEB) and presence of MalF-P2 (Module-derived), highlight the increased solvent exposure of the binding pocket of the MalE/MalF-P2 complex. The increased solvent-accessibility of specific amino acids in the substrate-binding pocket is readily appreciated (Fig. 6).

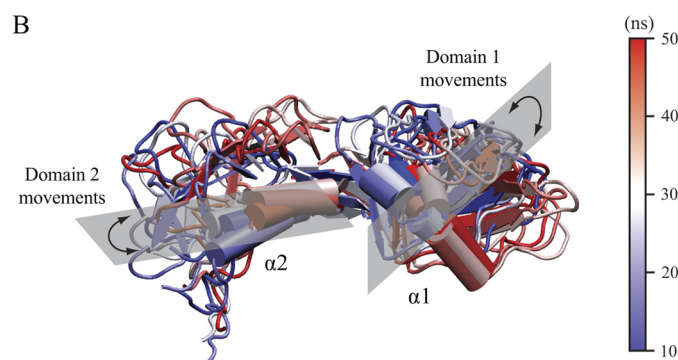
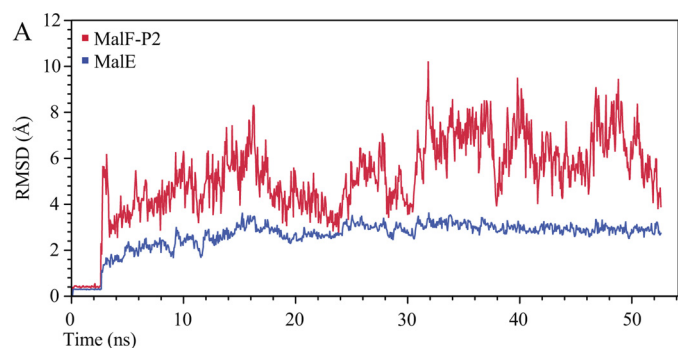
**Molecular Dynamics Simulation of MalE/MalF-P2 Complex**—To further assess the MalE and MalF-P2 binding interface a GROMACS molecular dynamics simulation (21) was performed. As starting points for the simulated annealing the crystal structures of MalE and MalF-P2 in the MalFGK<sub>2</sub> complex were used. The full-length sequence of MalE was calculated for each simulation ensemble. The C-terminal part of MalF-P2, residues 266–275, was discarded as earlier data show no interaction between MalE and MalF-P2 in this region (11). Further, the C-terminal  $\alpha$ -helix  $\alpha 3$  cannot fold up independently in solution (24). Root mean square deviation (r.m.s.d.) trajectories were plotted for MalE, MalF-P2, and the MalF-P2/MalE complex (Fig. 7). The molecular dynamics trajectory shows that MalE maintains its initial structure during the course of the simulation. Dynamics is only observed for surface-exposed loops in the C-terminal lobe of MalE. MalE thus behaves as a rigid body with a r.m.s.d. amplitude of 3 Å throughout the simulation. In contrast to MalE, MalF-P2 shows large dynamic variations throughout the simulation. The backbone r.m.s.d. initially

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risers to accommodate the binding surface to MalE properly. After 25 ns, MalF-P2 is properly positioned on the binding surface of MalE (Fig. 7). The second half of the simulation yields a higher r.m.s.d. trace, as the two domains of MalF-P2 are dynamic in their interaction with MalE. From NMR

interaction studies it is known that the interaction between MalF-P2 and MalE particularly involves the two  $\alpha$ -helices  $\alpha 1$  and  $\alpha 2$  (11) (Fig. 8). These  $\alpha$ -helices are located in each of the two domains of MalF-P2. During the simulated annealing both domains of MalF-P2 are reorienting to accommodate MalE. After 40–50 ns of the simulated annealing  $\alpha$ -helix  $\alpha 1$  in domain 1 reorients and moves toward domain 2 of MalF-P2 on the binding surface of MalE (Fig. 7). Whereas the angle between the  $\alpha$ -helices at the beginning of the simulation is on the order of  $30^\circ$ , the two domains are oriented almost perpendicular to each other at the end of the trajectory.

A second 50-ns molecular dynamics simulation was performed to monitor the r.m.s.d. traces from the respective calculations. The two calculations converged and showed similar dependence on the dynamic behavior of the MalF-P2/MalE interaction.

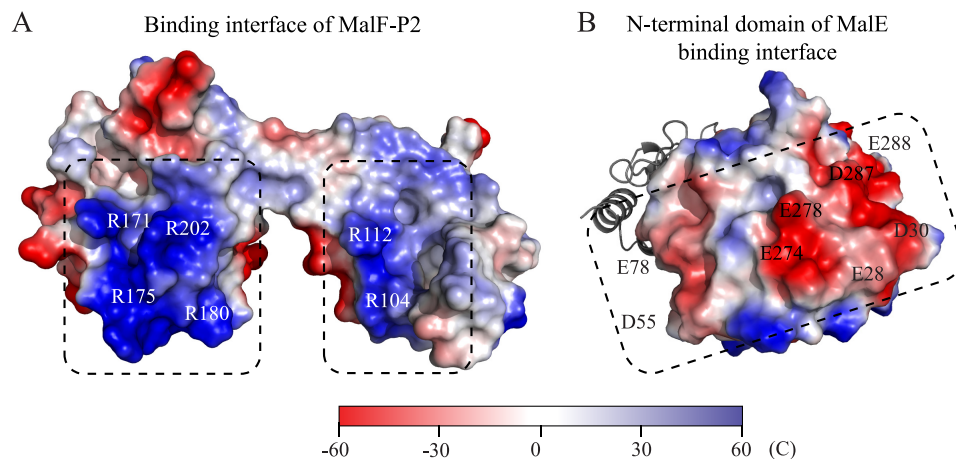


**FIGURE 7. 50-ns molecular dynamics simulation of the MalE/MalF-P2 complex.** *A*, backbone r.m.s.d. trajectory of the MalF-P2/MalE complex. The first 2.5 ns comprise a waterbox refinement before starting the simulated annealing. The backbone r.m.s.d. trajectory of MalE is represented in *blue* and MalF-P2 in *red*. MalE keeps a rigid structure with little dynamics (r.m.s.d.  $\sim 3$  Å) in the course of the full simulation. A much larger dynamic behavior is observed for MalF-P2 (r.m.s.d. = 4–9 Å). MalF-P2 undergoes large conformational rearrangements initially until the correct conformation with respect to MalE is found. *B*, representations of MalF-P2 at different time steps of the simulation. Structures are shown at every 10 ns of the simulation with colors shifting from *blue* to *white* to *red*. The two  $\alpha$ -helices of MalF-P2,  $\alpha 1$  and  $\alpha 2$ , make up the main interaction with MalE (11). During the course of the simulation they undergo large rearrangements while still maintaining the interaction to MalE.

## DISCUSSION

In an earlier report we have demonstrated that the periplasmic loop P2 of MalF acts as a receptor that recognizes MalE (11). Here we extend this study and show that MalF-P2 induces an activated conformational state of MalE that precedes substrate transport in MalFGK<sub>2</sub>. Although x-ray structures have been able to give valuable snapshots of the maltose transporter in three different conformational states (4–6), none could so far explain how the initial activation of MalFGK<sub>2</sub> at the periplasmic side is accomplished. A recent molecular dynamics study hints at the dynamic behavior of the association process, but experimental data are so far missing (25).

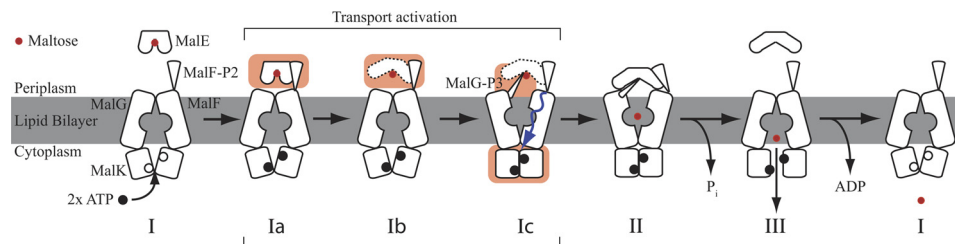
We found that substrate recognition is mediated by MalF-P2, which causes an alternative active conformation of the two domains of MalE in the presence of MalF-P2. From these new data we propose an adapted activation mechanism for the existing transport model (26) which is represented in (Fig. 9). Here, MalF-P2 first acts as a receptor for MalE (Fig. 9, *Ia*). In the presence of MalF-P2, a conformational change of MalE is induced. The substrate becomes solvent-exposed in the binding pocket, and concomitantly space is provided for insertion of



**FIGURE 8. Electrostatic surface representation of the MalE/MalF-P2 binding interface.** Highly charged patches with complementary charges (*dashed boxes*) are located directly at the binding interface of the N-terminal lobe of MalE and the two domains of MalF-P2. Important residues are indicated in MalF-P2 (*A*) and MalE (*B*), respectively.



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**FIGURE 9. Activation model of substrate transport in MalFGK<sub>2</sub>.** Based on the alternative-access model (Fig. 1), three additional intermediates are proposed to account for our findings. *Ia*, MalK subunits in an open state containing bound ATP. Substrate recognition is mediated through binding of MalF-P2. *Ib*, conformational change of MalE, exposing the substrate to the solvent and allowing insertion of MalG-P3 into the binding pocket (dashed lines MalE). *Ic*, substrate availability signaled reciprocally across the membrane inducing a closure of the MalK dimer. Subsequently, the transmembrane MalF/G domains switch from an inward-facing to an outward-facing conformation allowing the released maltose to bind into the binding pocket of the transmembrane domains of MalF/G. Amplitudes of motion for the TMDs are exaggerated to highlight changes. A TMD rotation from the inward to the outward conformation was determined to be on the order of 22° (4).

the periplasmic loop P3 of MalG (Fig. 9, *Ib*). In the x-ray structure (6), the loop MalG-P3 is fully inserted into the binding pocket, effectively “scooping out” the substrate. So far, however, it is not understood how the MalG-P3 loop gets inserted into the binding cleft of MalE. MalE binds maltose with micromolar affinity (23) and creates a highly stable complex burying the substrate between the two domains (20). In the substrate-bound conformation, solvent does not have access to the binding cleft (23). Conformational changes of MalE triggered by MalF-P2 allow MalG-P3 either to interact directly with the substrate or to insert into the binding pocket, stabilizing the complex during the substrate import cycle (Fig. 9, *Ic*). ATP hydrolysis drives substrate transport, and much effort has been made to identify the interactions that cause hydrolysis (26). In an earlier study by Grote *et al.* (27) it was demonstrated that MalF-P2 signals across the lipid bilayer to the cytosolic part of MalFGK<sub>2</sub>, effectively communicating reciprocally over the cellular membrane during the substrate transport cycle (Fig. 9, *Ic*).

The MalE/MalF-P2 interaction is mediated through complementary electrostatic forces (Fig. 8). MalE has two large negatively charged patches at the binding interface to MalF-P2. Complementarily, MalF-P2 has one positively charged patch at each of its two domains around  $\alpha$ -helices  $\alpha 1$  and  $\alpha 2$ . In the molecular dynamics simulation, the two domains of MalF-P2 turn out to be highly dynamic. We speculate that the complementary electrostatic patches stabilize the complex, inducing structural changes in MalE, which are required to dock to the MalFGK<sub>2</sub>-E complex. The molecular dynamics simulations suggest that the interaction of MalF-P2 and MalE is highly dynamic with MalF-P2 “gliding” on the interaction surface of MalE. The dynamics is a necessity to facilitate the binding of MalE when docking to the transmembrane proteins MalF/G. Dynamics enables different kinds of binding modes of MalF-P2 with respect to MalE, which reflect the different states during substrate transport in the MalFGK<sub>2</sub>-E complex.

As shown here, MalF-P2 also affects the substrate-free MalE, inducing the activated conformation of MalE. This activated state would enable the insertion of MalG-P3 and the stabilization of the MalFGK<sub>2</sub>-E complex in the absence of substrate. In turn, this would cause ATP hydrolysis and explain the observed futile ATP hydrolysis of the transporter in the absence of substrate (28, 29).

*Acknowledgment—We thank Dr. Oksana Krylova for help with the experimental setup of ITC measurements and fruitful discussions.*

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