

Published in final edited form as:

Angew Chem Int Ed Engl. 2009 ; 48(6): 1097–1101. doi:10.1002/anie.200805009.

Cycloaddition-Promoted Self-Assembly of a Polymer into Well-Defined β -Sheets and Hierarchical Nanofibrils**

Ting-Bin Yu, Jane Z. Bai, and Prof. Dr. Zhibin Guan

Department of Chemistry, University of California, Irvine, Irvine, California 92697-2025 (USA),

Fax: (+1) 949-824-2210, Homepage: <http://chem.ps.uci.edu/~zguan>

Zhibin Guan: zguan@uci.edu

Keywords

cycloaddition; click chemistry; β -sheet; nanofibers; peptides; polymers; self-assembly; hierarchical nanostructures

Whereas biopolymers, such as proteins, are ubiquitous for well-defined secondary, tertiary and quaternary structures,^[1] it remains a fundamental challenge to design synthetic polymers that can fold into predictable high structures. One major research thrust in our laboratory is aimed at programming weak forces into polymers to guide their assembly into well-defined molecular and nano-structures.^[2] Previously we reported biomimetic multi-domain polymers following modular design of titin.^[3] In this study, our attention was drawn to β -sheet polymers. Despite significant progress has been made in the area of designing discrete short peptidomimetic oligomers with β -sheet structures,^[4] it remains largely illusive to materials chemists to design synthetic high polymers that can fold into well-defined β -sheets and hierarchical nanostructures. β -Sheet is not only a basic secondary structure in proteins, but also an important structural motif in many fibril biomaterials such as amyloids^[5] and silks.^[6] Their hierarchical nanostructures and excellent mechanical properties have inspired biomimetic material designs.^[7] Despite a number of peptide-related systems were reported to form β -sheet based fibrils, most of them are short peptides^[8] or peptide-polymer conjugates.^[9] The self-assembly in these systems usually proceeds *intermolecularly*, forming relatively weak structures. Genetically engineered polypeptides via recombinant DNA technology were reported to form β -sheets and various nanostructures,^[10] however, the efficiency and versatility are limited by the biosynthetic pathway. Both for fundamental interest and for advanced materials designs, it is highly desirable to develop efficient synthesis to access well-defined covalently bonded β -sheet polymers. Herein we describe a new strategy of constructing covalent synthetic polymers that fold into well-defined β -sheets and further assemble into hierarchical nanofibrils (Fig. 1).

** We acknowledge the financial support from the National Institutes of Health (R01EB004936) and the Department of Energy (DE-FG02-04ER46162). We thank Prof. Alexander McPherson in the Department of Molecular Biology & Biochemistry for assistance with AFM, Prof. James Nowick in the Department of Chemistry for assistance with HPLC, Dr. Jian-Guo Zheng at the Materials Characterization Center for assistance with TEM, and Drs. Wytze Van der Veer, Phil Dennison and John Greaves in the Department of Chemistry for assistance with CD, NMR and MS. We also thank Dr. Youli Li at the Materials Research Lab of UCSB for WXR instrument assistance.

Correspondence to: Zhibin Guan, zguan@uci.edu.

Supporting information for this article is available on the WWW under <http://www.angewandte.org> or from the author.

As illustrated in Fig. 1, we employed Cu(I)-catalyzed azide–alkyne cycloaddition (CuAAC or commonly referred to as “click” chemistry) for polymerization of a peptide monomer. CuAAC is a versatile methodology because of its efficiency, functional group tolerance and applicability to a broad range of substrates.^[11] Since the initial reports of this methodology,^[11a,b] this reaction has been employed in a wide range of applications including selective ligations,^[12] bioconjugation,^[13] molecular recognition,^[14] and material and polymer synthesis.^[15,16] Based on structural similarities, 1,4- and 1,5-disubstituted 1,2,3-triazole rings formed by azide–alkyne cycloaddition have been used as biomimetic surrogates of peptides^[17] in α -helical coil,^[18] β -strands,^[19] β -turn mimics,^[20,21] and prosthetic proteins.^[22] Despite these developments, it should be noted that the work described here represents the first example of applying this chemistry to induce high order structure formation in synthetic high polymers.

Our design is based on a convergent β -turn mimic our group has developed recently based on CuAAC reaction. We have shown that cycloaddition between two short peptide strands terminated with azide and alkyne forms 1,4-disubstituted 1,2,3-triazole ring that induces β -turn formation.^[21] ¹H NMR, FT-IR and molecular mechanics calculations reveal that three carbon linkers for 1,4-disubstituted triazole are optimal for the formation of β -turn structure in non-protic media. We reasoned that if an AB peptide monomer is prepared (A = azide, B = acetylene), [2+3] dipolar cycloaddition will not only efficiently polymerize the monomer, but should form β -turn to induce folding into antiparallel β -strands. Essentially, the cycloaddition polymerization induces folding of an encoded polymer into extensive β -sheets which further self-assemble into nanofibrils (Fig. 1).

For our initial concept demonstration, we installed alkyne and azide moieties onto a simple alanyl-glycine (AG)₃ hexapeptide as the monomer unit (**6**) because AG repeats are common motifs for antiparallel β -sheet formation in silk^[6] and biosynthetic poly-peptides.^[10a,c] Based on our previous model studies,^[21] three-carbon linkers were chosen to attach alkyne and azide to the C- and N- termini of the monomer **6** to maximize β -turn formation. One challenge in the syntheses and studies of well-defined β -sheet systems is their general poor solubility. Protecting strategy and switch-peptide concept have been utilized to circumvent this problem.^[9a] In our design, we introduced an acid-cleavable 2,4-dimethoxy-benzyl (DMB) protecting group on one amide to prevent premature aggregation during the polymerization and facilitate polymer processing and characterizations. 2,4-dimethoxy-benzyl (DMB) has widely been used in peptide synthesis to inhibit excessive H-bonding.^[23] In the end, deprotection of DMB triggers intramolecular folding and intermolecularly self-assembly (Fig. 1).

The synthesis of the monomer and polymer are illustrated in Scheme 1. The peptide monomer was prepared by combining standard solution and solid phase peptide coupling reactions. The preparation of the acetylene half of the monomer (**4**) began with Gabriel synthesis of 5-amino-1-pentyne followed by solution coupling to Boc-glycine to give Boc-Gly-pentyne **2**. Boc deprotection of **2** followed by coupling with Boc-alanine afforded **3**. Following Boc removal, DMB group was installed by reductive amination to give the DMB protected Ala-Gly-pentyne **4**. The azide half of the monomer (azido-Ala-Gly-Ala-Gly-OH, **5**) was prepared by solid phase peptide synthesis using 2-chlorotrityl chloride resin and standard Fmoc protocols.^[24] The final step of segment coupling between the secondary amine in **4** and the terminal carboxylic acid in **5** was effected by using HATU as the coupling reagent in DMF (with 5% DMSO). The final monomer **6** was purified by column chromatography and characterized by ¹H, ¹³C NMR, FTIR, HRMS and analytical HPLC. The full synthesis and characterization details for the monomer and polymers can be found in the Supporting Information (Scheme S1).

With the monomer **6** in hand, [2+3] cycloaddition polymerization was then carried out using a modified procedure^[16] to afford polymer **7** with M_n of 11500 g/mol, M_w of 21800 g/mol and PDI of 1.89, determined by gel permeation chromatography (GPC) using poly(ethylene glycol) (PEG) as the molecular weight standards. The structure of polymer **7** was confirmed by NMR and FTIR. FTIR spectrum of **7** showed a sharp signal at $\sim 2100\text{ cm}^{-1}$ corresponding to the azide functionality, indicating that polymer **7** still carries active azide end groups amenable for further reactions. DMB protecting groups successfully prevent premature aggregation to render polymer **7** fully soluble in polar organic solvent like DMF and 2,2,2-trifluoroethanol (TFE). This enables the preparation of a soluble and processable prepolymer with encoded information for subsequent folding and self-assembly. Upon complete removal of the DMB groups in a 1/5 (v/v) TFA/methanol mixture (confirmed by ^1H NMR, see Fig. S1 in Supporting Information), the resulting polymer **8** started folding and self-assembling into nanofibrils.

The β -sheet structure of polymer **8** was subsequently confirmed by FTIR, far-UV circular dichroism (CD) spectroscopy, and wide-angle powder X-ray diffraction (WXR). FTIR spectrum (Fig. 2a) of the protected polymer **7** shows strong amide I band at $\sim 1655\text{ cm}^{-1}$ and amide II band $\sim 1541\text{ cm}^{-1}$, characteristic of random coil conformation.^[25] Upon DMB deprotection and nanofibril formation, polypeptide **8** exhibits a significant enhancement at amide I band $\sim 1628\text{ cm}^{-1}$ and amide II band $\sim 1533\text{ cm}^{-1}$, supporting a β -sheet conformation.^[25] Additionally, the weak component at $\sim 1695\text{ cm}^{-1}$ is indicative of antiparallel β -sheets.^[25] CD spectrum of **8** in hexafluoroisopropanol (HFIP) solution exhibits a minimum at 206 nm and maximum at 195 nm (see Fig. S2 in Supporting Information), confirming the β -sheet conformation because similar CD spectra have also been observed in β -sheet forming poly(AG)₃YG or poly(AG)₃HG made via genetic process.^[10] In addition, wide-angle powder X-ray diffraction (WXR) pattern of polymer **8** reveals major reflections at d spacings of 4.57, 4.22, and 3.64 Å, respectively (see Fig. 2b), which are similar to those observed for antiparallel β -sheet d spacings reported for the unoriented silk fibroin film^[26] and the biosynthetic poly(AG)_n.^[10a]

Finally, transmission electron microscopy (TEM) and atomic force microscopy (AFM) were employed to directly visualize the self-assembled nanostructures. For TEM studies, polymer **8** nanofibril suspensions were deposited on carbon-coated grid and a 2% uranyl acetate stain was used to increase edge contrast of the nanofibrils. Indeed, TEM images show assemblies of polymer **8** into long linear nanofibrils (Fig. 3c). Examination of the micrograph at higher magnification indicates the nanofibrils are composed of a stack of molecular fibrils formed from the β -sheet polymer. The average width of the molecular fibril was determined to be $3.8 \pm 0.4\text{ nm}$ (see Fig. S4 in Supporting Information), which is in good agreement with the estimated width of the single β -sheet (see Fig. 4). The average width of the nanofibrils was found to be around 32 nm, indicating in average ~ 8 β -sheets stack laterally in one nanofibril. In order to investigate further the fibrillar morphology and texture of β -sheet fibrils, we analyzed the fibril dimensions by AFM for samples spin-coated on mica surfaces. AFM images (Fig. 3a and S6) revealed a similar morphology for the nanofibrils as shown by TEM. In addition, AFM analysis provides the height values of nanofibrils in the range of 1.7 to 7 nm with the most probable height at $5.0 \pm 0.5\text{ nm}$ (Fig. 3b and Fig. S8 in Supporting Information). Both TEM and AFM images clearly demonstrated that the intramolecularly folded β -sheets of polymer **8** further self-assemble intermolecularly into amyloid-like nanofibrils.

On the basis of the TEM and AFM results and previous models,^[10a,c] we propose here a model for the hierarchical self-assembly of polymer **8** into extensive β -sheets and nanofibrils (Fig. 4). In previous studies of poly(AG)_n systems,^[10a,c] it was shown that all alanine methyl groups orient toward the same face and the β -sheets arrange in the way that two like

surfaces are in contact. We assume this occurs the same way in our system. The polymer **8** first folds into individual antiparallel β -sheets (Fig. 4a,c), then stack face-to-face into bilayers (Fig. 4b,d), and finally assemble both horizontally and vertically into nanofibrils (Fig. 4e). The β -strands run perpendicular to the fibril axis, resembling the cross- β structure of amyloid fibrils.^[5] This model agrees with our experimental data. The observed width of individual molecular fibril (3.8 ± 0.4 nm) is consistent with the width of the β -strand estimated from the model (~ 3.7 nm). The nanofibril height measured by AFM falls in the range of 1.7 – 7 nm, with most probable height at 5.0 ± 0.5 nm. This agrees with the stacking of 1 – 4 layers of β -sheet bilayers with the most common ones of 3 layers ($1.7\text{nm} \times 3$). Due to polydispersity of polymer **8**, the longitudinal length of β -sheets varies so different β -sheet bilayers are interdigitated (Fig. 4d,e).

In summary, we describe here the first example of a synthetic polymer that can fold into well-defined β -sheets and further self-assemble into hierarchical nanostructures. The polymer with DMB protecting group was efficiently synthesized via Cu(I)-catalyzed azide-alkyne cycloaddition. Upon deprotection of DMB, the polymer was triggered to folds into well-defined β -sheets structure. The β -sheet structure of **8** was confirmed by FTIR, CD and WXR data. TEM and AFM images show that the β -sheets further assemble into hierarchical amyloid-like nanofibrils. A key design element here is that the [2+3] dipolar cycloaddition not merely serves as the polymerization method but also induces the folding and self-assembly of the formed polymer. This demonstrates a unique example in which a polymerization leads to intramolecular folding to a secondary structure (β -sheet) and further intermolecular organization into hierarchical nanostructures. The efficiency and versatility of the click chemistry should allow for further design of more complex polymer materials. In our continuing studies, β -sheet motif will be combined with other folding motifs for the design of novel hierarchical biomaterials with advanced physical properties and specific functions on the nanometer scale.

Experimental Section

Monomer 6

In a 250 mL round bottom flask, compound **4** (0.365 g, 1.01 mmol) and peptide **5** (0.39 g, 1.01 mmol) were dissolved in 15 mL of DMF:DMSO (95:5) mixed solvent containing 0.265 mL DIPEA. Following addition of HATU (0.461g, 1.2 mmol), the mixture was stirred at room temperature for 48 hours. After addition of 100 mL of H₂O to the completed reaction solution, the mixture was extracted with EtOAc (50 mL \times 2). The combined organic layer was washed with H₂O (100 mL \times 1), 1M HCl (100 mL \times 1), sat. NaHCO₃ (100 mL \times 1), and brine, then dried over anhydrous MgSO₄, filtered, and finally concentrated on a rotary evaporator. The residue was purified by flash chromatography with MeOH/DCM (1:9) to give **6** as white solid (349 mg, 46%). ¹H NMR (500 MHz, MeOD-*d*₄) δ 7.16 (d, *J* = 8.3, 1H), 6.53 (s, 1H), 6.48 (d, *J* = 8.3, 1H), 4.62 (d, *J* = 15.7, 1H), 4.40–4.37 (m, 2H), 4.22 (s, 2H), 4.17 (d, *J* = 7.2, 1H), 4.05 (d, *J* = 7.2, 1H), 3.89–3.58 (m, 10H), 3.43 (q, *J* = 7.0, 1H), 3.32–3.27 (m, 2H), 3.19–3.14 (m, 1H), 2.28 (t, *J* = 7.2, 2H), 2.16–2.10 (m, 3H), 1.86–1.81 (m, 2H), 1.72–1.70 (m, 2H), 1.36–1.10 (m, 9H); ¹³C NMR (125MHz, DMSO-*d*₆) δ 173.3, 172.7, 172.4, 172.0, 171.5, 171.1, 169.9, 169.2, 168.9, 168.6, 161.8, 160.6, 159.2, 158.2, 132.8, 129.3, 117.4, 105.2, 104.9, 98.8, 98.7, 84.6, 84.5, 71.9, 71.8, 56.1, 55.9, 55.8, 55.7, 55.3, 50.7, 39.0, 48.5, 48.5, 45.6, 44.0, 43.0, 42.4, 41.5, 38.1, 32.4, 31.2, 28.5, 24.8, 18.7, 18.2, 17.1, 15.83, 15.77, 15.0; HRMS (ESI), *m/z* calcd for [C₃₃H₄₈N₁₀O₉ + Na]⁺ = 751.3503; found 751.3494.

Polymer 7

7 was prepared according to a modified literature procedure.^[25] Peptide monomer **6** (182 mg, 0.25 mmol), copper acetate (4 mg, 0.03 mmol), and 0.25 mL of N₂-degassed DMF were introduced into a small vial. Under stirring the mixture was heated at 80°C in an oil bath for 2 hours. The initial clear solution turned into a dark green gel. After cooling down the polymerization mixture with an ice bath, the gel was dissolved with additional DMF and then polymer was precipitated into 0.1 N HCl (20 mL). The precipitate was purified by three consecutive times of centrifugation and re-dispersion with 0.1 N HCl. A white solid precipitate was finally isolated and dried under vacuum to give 155 mg of product **7** (yield: 85%). The molecular weight of the polymer was measured with GPC using poly(ethylene glycol) as standards: $M_n = 11500$, and $M_w = 21750$. ¹H NMR (500 MHz, *d*₆-DMSO and *d*-TFA) δ 8.37-8.26 (m, 1H), 7.02 (s, 1H), 6.39-6.33(m, 2H), 4.42-4.07 (m, 7H), 3.83-3.20 (m, 11H), 2.73 (s, 2H), 2.17-2.08 (m, 4H), 1.80 (s, 4H), 1.19-1.10 (m, 9H).

Full experimental details, including the syntheses and characterization, NMR, HRMS, GPC, FTIR, CD, WXR, TEM and AFM experiments, can be found in the Supporting Information.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

References

1. Stryer, L. *Biochemistry*. 4. Spektrum; Heidelberg, Germany: 1996.
2. Guan Z. *Polym Int*. 2007; 56:467-473.
3. a) Guan Z, Roland JT, Bai J, Ma S, McIntire T, Nguyen M. *J Am Chem Soc*. 2004; 126:2058-2065. [PubMed: 14971940] b) Roland JT, Guan Z. *J Am Chem Soc*. 2004; 126:14328-14329. [PubMed: 15521732] c) Kushner AM, Gabuchian V, Johnson EG, Guan Z. *J Am Chem Soc*. 2007; 129:14110-14111. [PubMed: 17973379]
4. a) Nowick JS. *Acc Chem Res*. 2008 ASAP. b) Hughes RM, Waters ML. *Curr Opin Struct Biol*. 2006; 16:514-524. [PubMed: 16837192]
5. Chiti F, Dobson CM. *Annu Rev Biochem*. 2006; 75:333-366. [PubMed: 16756495]
6. Guerette PA, Ginzinger DG, Weber BHF, Gosline JM. *Science*. 1996; 272:112-115. [PubMed: 8600519]
7. a) Cherny I, Gazit E. *Angew Chem*. 2008; 120:4128-4136. *Angew Chem Int Ed*. 2008; 47:2-10. b) König HM, Kilbinger AFM. *Angew Chem*. 2007; 119:8484-8490. *Angew Chem Int Ed*. 2007; 46:8334-8340.
8. a) Lashuel HA, LaBrenz SR, Woo L, Serpell LC, Kelly JW. *J Am Chem Soc*. 2000; 122:5262-5277. [PubMed: 22339465] b) Lopez de la Paz M, Goldie K, Zurdo J, Lacroix E, Dobson CM, Hoenger A, Serrano L. *Proc Natl Acad Sci USA*. 2002; 99:16052-16057. [PubMed: 12456886] c) Schneider JP, Pochan DJ, Ozbas B, Rajagopal K, Pakstis L, Kretsinger J. *J Am Chem Soc*. 2002; 124:15030-15037. [PubMed: 12475347] d) Reches M, Porat Y, Gazit E. *J Biol Chem*. 2002; 277:35475-35480. [PubMed: 12095997] e) Jung JP, Jones JL, Cronier SA, Collier JH. *Biomaterials*. 2008; 29:2143-2151. [PubMed: 18261790]
9. a) Hentschel J, Krause E, Börner HG. *J Am Chem Soc*. 2006; 128:7722-7723. [PubMed: 16771470] b) Boerner HG, Smarsly BM, Hentschel J, Rank A, Schubert R, Geng Y, Discher DE, Hellweg T, Brandt A. *Macromolecules*. 2008; 41:1430-1437. c) Jahnke E, Lieberwirth I, Severin N, Rabe JP, Frauenrath H. *Angew Chem*. 2006; 118:5510-5513. *Angew Chem, Int Ed*. 2006; 45:5383-5386. d) Klok HAJ. *J Polym Sci Part A*. 2005; 43:1-17. e) Hartgerink JD, Beniash E, Stupp SI. *Science*. 2001; 294:1684-1688. [PubMed: 11721046] f) Burkoth TS, Benzinger TLS, Urban V, Lynn DG, Meredith SC, Thiyagarajan P. *J Am Chem Soc*. 1999; 121:7429-7430. g) Qu Y, Payne SC, Apkarian RP, Conticello VP. *J Am Chem Soc*. 2000; 122:5014-5015. h) Rathore O, Sogah DY. *J Am Chem Soc*. 2001; 123:5231-5239. [PubMed: 11457385] i) Smeenk JM, Otten MJB, Thies J, Tirrell DA,

- Stunnenberg HG, van Hest JCM. *Angew Chem.* 2005; 117:2004–2007. *Angew Chem, Int Ed.* 2005; 44:1968–1971. j) Zhang S, Marini DM, Hwang W, Santoso S. *Curr Opin Chem Biol.* 2002; 6:865–871. [PubMed: 12470743]
10. a) Krejchi MT, Atkins EDT, Waddon AJ, Fournier MJ, Mason TL, Tirrell DA. *Science.* 1994; 265:1427–1432. [PubMed: 8073284] b) West MW, Wang W, Patterson J, Mancias JD, Beasley JR, Hecht MH. *Proc Natl Acad Sci USA.* 1999; 96:11211–11216. [PubMed: 10500156] c) Topilina NI, Higashiya S, Rana N, Ermolenkov VV, Kossow C, Carlsen A, Ngo SC, Wells CC, Eisenbraun ET, Dunn KA, Lednev IK, Geer RE, Kaloyeros AE, Welch JT. *Biomacromolecules.* 2006; 7:1104–1111. [PubMed: 16602727]
11. a) Tornøe CW, Christensen C, Meldal M. *J Org Chem.* 2002; 67:3057–3064. [PubMed: 11975567] b) Rostovtsev VV, Green LG, Fokin VV, Sharpless KB. *Angew Chem.* 2002; 114:2708–2711. *Angew Chem, Int Ed.* 2002; 41:2596–2599. c) Kolb HC, Finn MG, Sharpless KB. *Angew Chem.* 2001; 113:2056–2075. *Angew Chem, Int Ed.* 2001; 40:2004–2021. d) Meldal M, Tornøe CW. *Chem Rev.* 2008; 108:2952–3015. [PubMed: 18698735]
12. Bock VD, Hiemstra H, van Maarseveen JH. *Eur, J Org Chem.* 2006:51–68.
13. Moses JE, Moorhouse AD. *Chem Soc Rev.* 2007; 36:1249–1262. [PubMed: 17619685]
14. Meudtner RM, Hecht S. *Angew Chem.* 2008; 120:5004–5008. *Angew Chem, Int Ed.* 2008; 47:4926–4930.
15. Meudtner RM, Hecht S. *Macromol Rapid Comm.* 2008; 29:347–351. Lutz J-F. *Angew Chem.* 2007; 119:1036–1043. *Angew Chem, Int Ed.* 2007; 46:1018–1025. Lutz J-F, Boerner HG, Weichenhan K. *Australian J Chem.* 2007; 60:410–413. Lutz J-F, Boerner HG. *Progress Polym Sci.* 2008; 33:1–39. Wu P, Feldman AK, Nugent AK, Hawker CJ, Scheel A, Voit B, Pyun J, Frechet JMJ, Sharpless KB, Fokin VV. *Angew Chem.* 2004; 116:4018–4022. *Angew Chem, Int Ed.* 2004; 43:3928–3932. also see *Macromol Rapid Comm.* 2008; 29(12–13) for a series of reviews on applications of CuAAC to polymer synthesis.
16. a) Liu Y, Diaz DD, Accurso AA, Sharpless KB, Fokin VV, Finn MG. *J Polym Sci Part A: Polym Chem.* 2007; 45:5182–5189. b) Geng J, Lindqvist J, Mantovani G, Haddleton DM. *Angew Chem.* 2008; 120:4248–4251. *Angew Chem Int Ed.* 2008; 47:4180–4183. c) Hilf S, Hanik N, Kilbinger AFM. *J Polym Sci Part A: Polym Chem.* 2008; 46:2913–2921. d) Fournier D, Hooogenboom R, Schubert US. *Chem Soc Rev.* 2007; 36:1369–1380. [PubMed: 17619693] e) van Dijk M, Mustafa K, Dechesne AC, van Nostrum CF, Hennink WE, Rijkers DTS, Liskamp RMJ. *Biomacromolecules.* 2007; 8:327–330. [PubMed: 17291054] d) van Dijk M, Nollet ML, Weijers P, Dechesne AC, van Nostrum CF, Hennink WE, Rijkers DTS, Liskamp RMJ. *Biomacromolecules.* 2008 ASAP. f) Riva R, Schmeits S, Jerome C, Jerome R, Lecomte P. *Macromolecules.* 2007; 40:796–803. g) Lutz J-F, Boerner HG, Weichenhan K. *Macromolecules.* 2006; 39:6376–6383.
17. Angell YL, Burgess K. *Chem Soc Rev.* 2007; 36:1674–1689. [PubMed: 17721589]
18. Horne WS, Yadav MK, Stout CD, Ghadiri MRM. *J Am Chem Soc.* 2004; 126:15366–15367. [PubMed: 15563148]
19. Angelo NG, Arora PS. *J Am Chem Soc.* 2005; 127:17134–17135. [PubMed: 16332031]
20. a) Angell YL, Burgess K. *J Org Chem.* 2005; 70:9595–9598. [PubMed: 16268639] b) Angell Y, Chen D, Brahimi F, Uri Saragovi H, Burgess K. *J Am Chem Soc.* 2008; 130:556–565. [PubMed: 18088119]
21. Oh K, Guan Z. *Chem Comm.* 2006:3069–3071. [PubMed: 16855688]
22. Tam A, Arnold U, Soellner MB, Raines RT. *J Am Chem Soc.* 2007; 129:12670–12671. [PubMed: 17914828]
23. Zahariev S, Guarnaccia C, Zanuttin F, Pintar A, Esposito G, Maravic G, Krust B, Hovanesian AG, Pongor S. *J Pept Sci.* 2005; 11:17–28. [PubMed: 15635723]
24. *NovaBiochem Catalog.* 2003
25. a) Miyazawa T, Blout ER. *J Am Chem Soc.* 1961; 83:712–719. b) Haris PI, Chapman D. *Biopolymers.* 1995; 37:251–263. [PubMed: 7540054] c) Safar J, Roller PP, Ruben GC, Gajdusek DC Jr, Gibbs CJ. *Biopolymers.* 1993; 33:1461–1476. [PubMed: 8400035]

26. a) Fraser, RDB.; MacRae, TP. *Conformation in Fibrous Proteins*. Academic Press; New York: 1973. b) Asakura T, Sugino R, Okumura T, Nakazawa Y. *Protein Sci.* 2002; 11:1873–1877. [PubMed: 12142441]

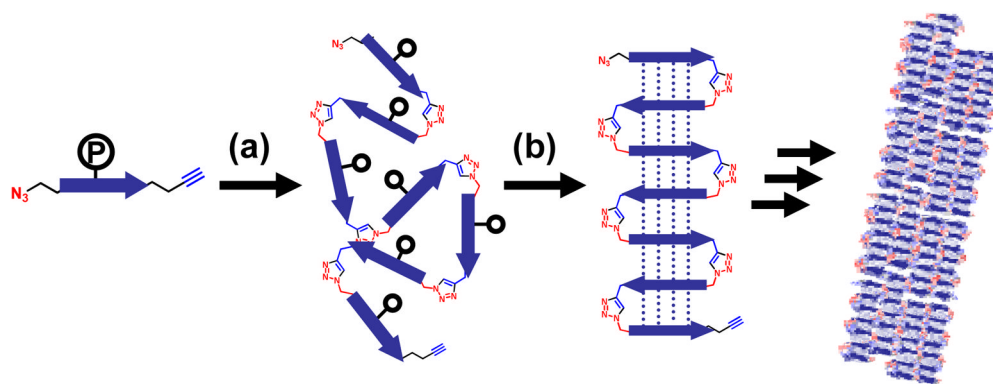


Figure 1. Concept of cycloaddition-induced folding and self-assembly: (a) [2+3] cycloaddition polymerization of a protected peptide monomer; (b) upon deprotection polypeptides fold into well-defined antiparallel β -strands; (c) self-assembly of multiple β -sheets forms hierarchical nanofibrils.

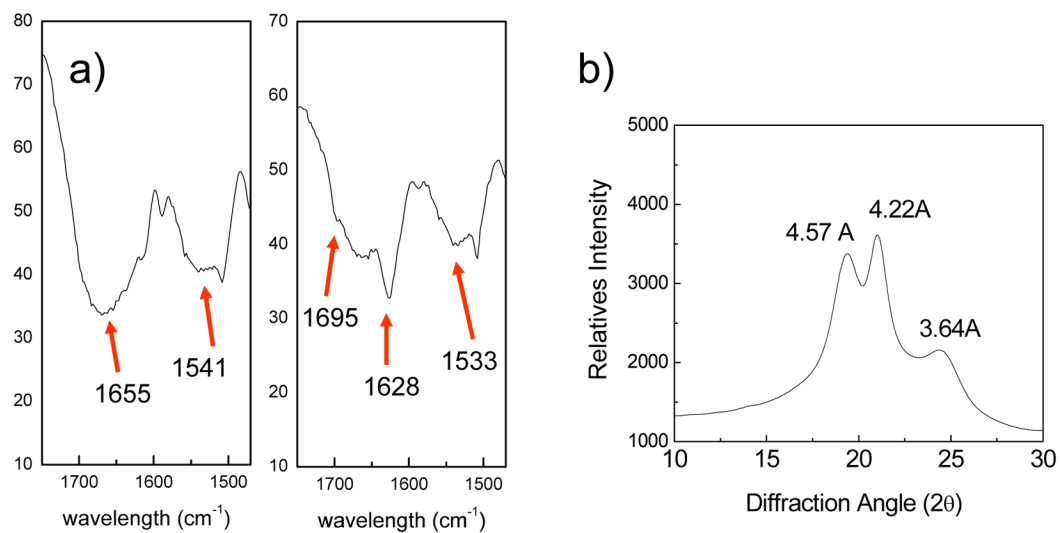


Figure 2. (a) FTIR spectra of **7** (left) and **8** (right) in amide I, II band region; (b) Wide angle X-ray diffraction of polymer **8**.

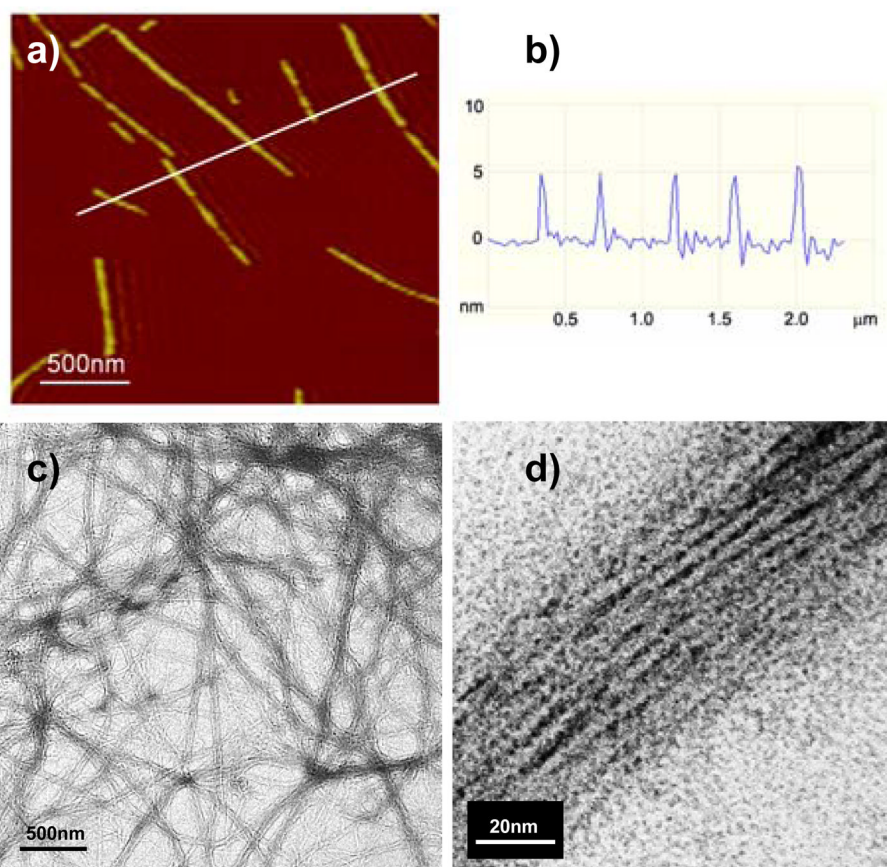


Figure 3.

(a) A representative AFM micrograph for nanofibrils; (b) Height profile for 5 nanofibrils across the white line; (c) A representative TEM image of nanofibrils; (d) A zoom up view of one nanofibril.

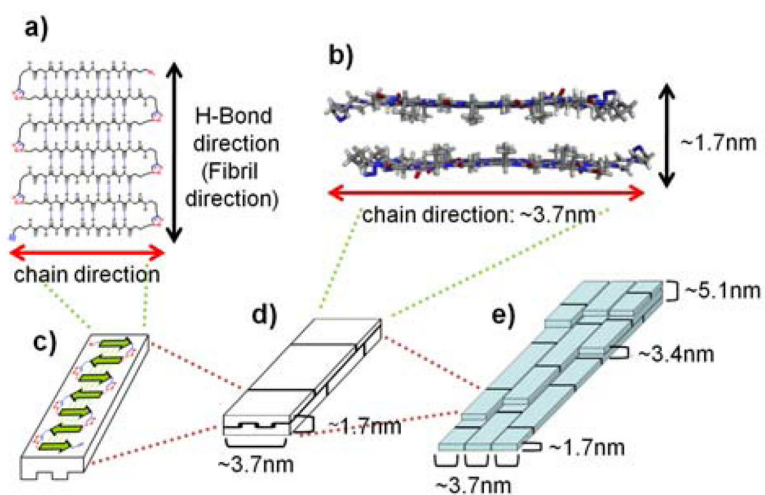
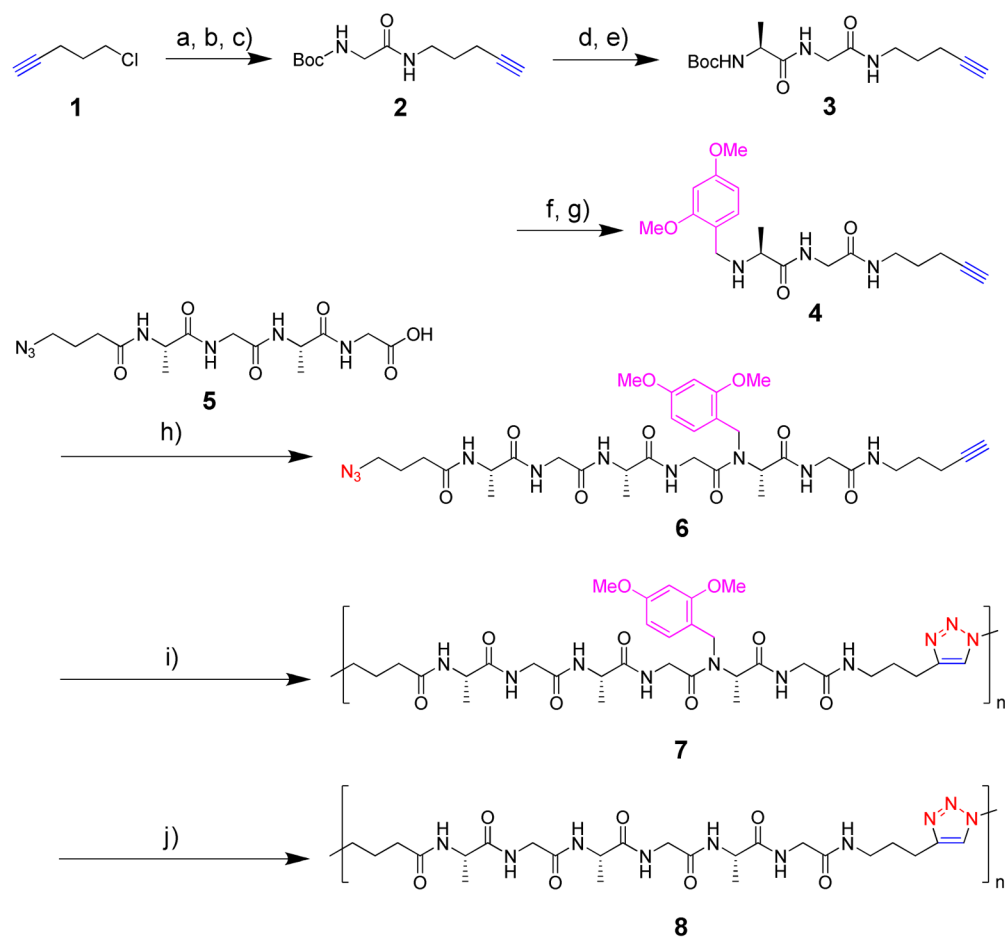


Figure 4.

Proposed model for hierarchical self-assembly of **8** to form nanofibrils: (a) Top-view of an antiparallel single β -sheet; (b) Side-view of a face-to-face stacked double layer of two β -sheets; (c) A single β -sheet; (d) Face-to-face stacked double layer of two β -sheets; (e) Stacking of many double layers forms the hierarchical nanofibrils.

**Scheme 1.**

Synthesis of the β -sheet mimic polypeptide **8**. Reaction conditions: (a) phthalimide, K_2CO_3 , DMF, 70 °C, 24 h, (96%); (b) hydrazine hydrate, EtOH, 70 °C, 2 h, (74%); (c) Boc-glycine, EDC, HOBt, iPr_2EtN , DCM, rt, 12 h, (98%); (d) TFA, DCM, rt, 3 h; (e) Boc-alanine, EDC, HOBt, iPr_2EtN , DCM, rt, 12 h, (90%); (f) TFA, DCM, rt, 3 h; (g) 2,4-dimethoxybenzaldehyde, $NaCNBH_3$, MeOH, rt, 12 h, (83%); (h) **7**, HATU, iPr_2EtN , DMF (with 5% DMSO), 48 h, (46%); (i) 2 mol% $CuOAc$, DMF, 80 °C, 2 h, (85%); (j) TFA, DCM, 2 h.