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Asymmetric Synthesis of Highly Functionalized Cyclopentanes by a Rhodium- and Scandium-Catalyzed Five-Step Domino Sequence

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Abstract

A domino sequence has been developed between vinyldiazoacetates and racemic allyl alcohols, involving five distinct steps. The sequence generates highly functionalized cyclopentanes with four new stereogenic centers as single diastereomers in 64–92% ee. The first step is a rhodium-catalyzed oxygen ylide formation, which is then followed by a [2,3]-sigmatropic rearrangement, an oxy-Cope rearrangement, a keto/enol tautomerization, and then finally a carbonyl ene reaction. With appropriate substrates, a further silyl deprotection and a 6-*exo-trig* cyclization can be added to the domino process.

Introduction

Stereoselective methods for the formation of five- to seven-membered carbocycles by means of coupling acyclic fragments are important strategic reactions for the synthesis of complex targets.¹ The Diels-Alder reaction is the classic transformation for the construction of cyclohexenes containing multiple stereocenters.² Novel strategies have been developed for the formation of five- and seven-membered carbocycles, and many of these methods emulate the synthetic power of the Diels-Alder reaction.^{1,3} In this paper, we describe a highly diastereo- and enantioselective reaction between vinyldiazoacetates and racemic allyl alcohols for the synthesis of functionalized cyclopentanes, bearing four new stereogenic centers (Scheme 1).

The discovery of the new cyclopentane synthesis originated from our studies on rhodiumcarbenoid chemistry.⁴ The metal-catalyzed reaction of diazo compounds is an especially useful method for the generation of highly reactive metal-carbenoids, which are then capable of initiating a cascade sequence of reactions.^{5,6} We have been exploring the development of new cascade sequences that start from vinyldiazoacetates as substrates.⁷ Recently, we

[†]Electronic Supplementary Information (ESI) available: [Full experimental data and spectral data]. See DOI: 10.1039/b000000x/

[‡]General Procedure for One-Pot Cyclopentane Synthesis. An oven-dried, 25 mL round-bottomed flask, equipped with a stir bar, was capped with a rubber septum and placed under a dry argon atmosphere. The reaction vessel was charged with Rh₂(*S*-DOSP)4 (19 mg, 0.01 mmol, 0.01 equiv) and the allyl alcohol (1.0 mmol, 1.0 equiv) in heptane (1.0 mL). The solution was cooled to 0 °C in an ice bath before adding a heptane solution (10 mL) of the diazo compound (1.1 mmol, 1.1 equiv) drop-wise over 30 min. Following addition, the reaction was stirred at 0 °C for 2 h before warming to rt for 30 min. The rubber septum was removed and the reaction flask was fixed with a reflux condenser and heated to 80 °C for 24 h or until TLC indicated complete conversion of the [2,3]-rearrangement product to a mixture of oxy-Cope and ene products. Scandium triflate (98 mg, 0.20 mmol, 0.20 equiv) was then added in a single portion and the reaction was then cooled to ambient temperature and concentrated *in vacuo*. The product was purified by flash chromatography.

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reported the tandem oxygen ylide formation/[2,3]-sigmatropic rearrangement between styryldiazoacetate **1** and allyl alcohols **2**, which generated dienols **3** in 92–98% ee (Scheme 2).⁸ As the tandem oxygen ylide formation/[2,3]-sigmatropic rearrangement is broadly effective, we became intrigued with the possibility that the products of this reaction would be prone to further rearrangement. The results of a systematic study to explore the possibility of generating an extended domino sequence⁹ is described herein.

Results and discussion

Initial studies and reaction optimization

In order to determine the plausibility of a domino reaction, we examined the reactions of the styryl derivative **3** (R=Me), which is formed in 98% ee using the tandem oxygen ylide/[2,3]-sigmatropic rearrangement.⁸ Since **4** (Scheme 3) is a 3-hydroxy-1,5-hexadiene containing two adjacent quaternary centers, it was anticipated that this product would be liable to undergo an oxy-Cope rearrangement.¹⁰ Indeed, on heating in cyclohexane, **4** underwent a quantitative rearrangement to the enol **5**, as a single diastereomer. The reaction did not proceed with complete retention of the optical purity, as the product was formed in 82% ee.

Further extension of the domino sequence was examined. Enol 5 was not expected to be a stable product, and indeed, on attempted purification by silica gel chromatography, it tautomerized to the ketone 6 as a single diastereomer in 82% ee. As the ketone in 6 is adjacent to an ester carbonyl, $\mathbf{6}$ would be anticipated to be quite reactive, and it would be reasonable to expect that it would be prone to an intramolecular carbonyl ene reaction¹¹ to form the cyclopentane 7a. Previous examples of tandem oxy-Cope rearrangement/ene reaction sequences are known^{10a-b} The intramolecular ene reaction was readily catalyzed by scandium(III) triflate¹² at elevated temperatures, which allowed for the conversion of $\mathbf{6}$ to the cyclopentane 7a. The new stereogenic centers were formed during the concerted transformations, the oxy-Cope rearrangement and the ene reaction, and consequently, 7a was formed in a highly diastereoselective manner. The absolute configuration of 1,5hexadiene 4 was assigned in the earlier [2,3]-sigmatropic rearrangement studies,⁸ and the relative configuration of cyclopentane **7a** was determined by X-ray crystallography.^{13,14} The stereochemistry for the conversion of 4 to 7a is consistent with the oxy-Cope rearrangement of **4** to **5** proceeding via a chair transition state.^{10,15}, Epimerization of the stereocenters in 5 would not be expected in the course of tautomerization or the ene reaction, and so, consistency in the configuration of 7a was anticipated.

Optimization studies were then conducted such that the cascade sequence between the vinyldiazoacetate 1 and allyl alcohol 2a to form the cyclopentane 7a could be conducted as a one-pot process. It was envisioned that after the initial enantioselective rhodium-catalyzed step, the rest of the sequence should be feasible by using the appropriate combination of solvent, temperature and/or Lewis acid catalyst. Table 1 summarizes the optimization of the reaction between 1 and 2a. Prior studies had shown that the optimum temperature for the tandem oxygen ylide formation/[2,3]-sigmatropic rearrangement was 0 °C.8 After completion of the [2,3]-rearrangement, the crude product mixture was heated in the parent solvent for 20 h before treating with a catalytic amount of scandium(III) triflate for 4 h. At temperatures less than ~75 °C, only the [2,3]-product was observed in the ¹H NMR of the crude reaction mixture (entry 1). By a modest increase in temperature, however, the desired product 7a was isolated in generally good yields (entries 2, 4–9). In a more polar solvent system, such as ethyl acetate, attenuated yields and enantioselectivities were observed (entry 3). The optimum conditions were found to be 1 mol % of the rhodium catalyst in heptane at 80 °C. Under these conditions 7a was formed in 95% yield and 82% ee. (entry 7). These conditions were used as the standard for the remainder of the study, although in some cases

high yield of product could be obtained without the need of adding scandium(III) triflate. In these cases, however, extended reaction times were required as the ene reaction was rather sluggish. It should be noted that much lower catalyst loading can be used if desired,¹⁶ because on decreasing the catalyst loading to 0.01 mol % (entry 9) only a slight change in yield and enantioselectivity (75% yield, 79% ee) was observed.

Substrate scope

The next series of experiments explored the general scope of the reaction with respect to allyl alcohols bearing a 3,3'-dimethyl moiety (Table 2). The steric tolerance of the reaction was probed through substrates with various linear and branched alkyl chains (entries 1–4). All aliphatic substituents examined proved efficient substrates for the transformation, affording the cyclopentanes **7a–i** in 67–95% yield. Substituents bearing a functionalized R-group, including an olefin, ketal, silyl ether, phenyl, and silyl (entries 5–9) were all compatible, indicative of the diverse functionalization one could install at the C_3 -position of **7**. In the case of the ketal-protected alcohol **2f** (entry 6), however, the product was obtained as the deprotected ketone due to the acidic reaction medium. Similarly, under the prescribed conditions, the silyl ether **2g** underwent deprotection and subsequent side reactions upon exposure to scandium(III) triflate. By increasing the temperature, however, to 98 °C after the 20 h period, the corresponding cyclopentane **7g** was obtained in excellent yield (65%) in the absence of Lewis acid catalyst.

The reaction of various vinyldiazoacetates **8** was then explored, which introduced different functionality at the C_4 -position of the cyclopentane **9** (Table 3). The effect of substituents on the aryl ring was probed (entries 1–3). The best result was obtained for the electron rich 4-methoxy group in **8b**; however, substrates **8a** and **8c** with electron poor groups (entries 1 and 3) also afforded the desired products. Moreover, for the *p*-methoxy and *p*-bromostyryldiazoacetates, **8b** and **8c**, an improvement in enantioselectivity was observed. The relatively low yield for **8c** was associated with competing O—H insertion versus the [2,3]-sigmatropic rearrangement.⁸ The *E*-methyl hexenoate **8d** (entry 4) was an efficient substrate; however, the enantioselectivity of the [2,3]-rearrangement in this case was modest, and this translated to low enantioselectvity (64% ee) in the corresponding cyclopentane **9d**. The absolute configuration of **9c** was determined by X-ray crystallography.¹³ Since equivalent transitions states are presumed to be involved in the formation of **7a–i**, **9a**, **9b**, and **9d**, the absolute configuration of those products was assigned by analogy.

Further examples

The domino sequence could be extended to include two additional steps when the monosilyl protected diol **10** was used as substrate. (Scheme 4). When the standard reaction was conducted, but now using an equivalent of scandium(III) triflate in refluxing heptane, the reaction did not stop at the cyclopentane **11**. Instead, the fused pyran **12** was isolated as a single diastereomer in 80% ee. The formation of **12** is likely to proceed by silyl deprotection and a 6-*exo-trig* cyclization. The order of the last two steps in not known. The relative configuration of the *trans*-fused ring system was determined by X-ray crystallography.^{12,13}

The synthetic utility of the domino sequence was showcased in the reaction with the readily available diterpenoid, (–)-pulegol (13). Two diastereomeric hydrindane products 14 and 15 are possible, containing five stereogenic centers, two of which are quaternary. We have already demonstrated that the diastereoselectivity of the ylide formation/2,3-sigmatropic rearrangement with (–)-pulegol (13) is controlled by the chiral catalyst, with $Rh_2(R-DOSP)_4$ as catalyst giving a 10 : 1 diastereomeric mixture and $Rh_2(S-DOSP)_4$ giving a 4 : 1 diastereomeric mixture, favoring the other diastereomer.⁸ In contrast, the enantiomers of the

catalyst show very different levels of diastereocontrol in the domino sequence. In the Rh₂(R-DOSP)₄-catalyzed reaction between **13** and **1**, the single diastereomer **14** of the hydrindane was formed in 69% yield. In the Rh₂(S-DOSP)₄-catalyzed reaction between **13** and **1**, a 1 : 2 mixture of the diastereomers **14** and **15** was formed.¹⁷ The reason for the difference in diastereocontrol will be discussed later.

Stereochemical rationale

One of the most impressive features of the domino sequence for the synthesis of cyclopentanes is the high level of diastereocontrol possible in the sequence. Four new stereogenic centers are generated and the monocyclic products are produced as single diastereomers with 64–92% ee. The enantiomeric purity of the cyclopentanes is slightly degraded compared to the enantiomeric purity of the initial 2,3-sigmatropic rearrangement products. The enantioselectivity of the tandem ylide formation/2,3-sigmatropic rearrangement is already established.⁸ The highly stereoselective nature of the oxy-Cope and the anionic oxy-Cope has been documented in detail.^{15,18} In general, in the absence of significant 1,3-diaxial interactions, a chair-like transition state is strongly favored.¹⁸ Two additional stereogenic centers are generated in the ene reaction, in which the relative stereochemistry is presumably controlled by the already established stereocenters.

From the exploratory studies in Scheme 3, it is clear that the slight loss of enantiomeric purity occurs during the oxy-Cope rearrangement. This can be explained by considering the two possible chair-like transition states **TS-C1** and **TS-C2** for the [3,3]-rearrangement portrayed in Figure 1. Enantiocontrol will depend on the axial/equatorial preference of the hydroxyl or the carbomethoxy groups. The cyclohexane *A*-values of the hydroxyl and carbomethoxy substituents are 0.60-1.04 and 1.2-1.3 kcal mol⁻¹, respectively.¹⁹ Therefore the **TS-C1**, with the hydroxyl group in an axial position should be favored, forming (**4***S*, **5***R*)-**5***a* as the predominant product, but the difference in A-values is not sufficient to eliminate some of the product formed via **TS-C2**. Both transition states generate the same diastereomer but opposite enantiomers of the product.

For the reaction with (–)-pulegol, the major diastereomer produced in the $Rh_2(R-DOSP)_4$ catalyzed reaction with **1** has been shown to be the [2,3]-sigmatropic rearrangement product **16**.⁸ Compound **16** is ideally suited to undergo a stereoselective oxy-Cope rearrangement as illustrated in Figure 2. In the chair transition state **TS-C3**, the less sterically demanding hydroxyl group is adopting the axial position and the remote C_9 -methyl group is oriented away from the site of bond formation, minimizing any unfavorable steric interactions. Transition-state **TS-C3** leads to the formation of the observed diastereomer **14**. In the case of the major diastereomer of the [2,3]-sigmatropic rearrangement product from the $Rh_2(S-$ DOSP)₄-catalyzed reaction, neither of the two possible chair forms is strongly favored because one would have the ester group in the axial position and the other would have bond formation occurring on the same face as the C_9 -methyl group.

A stereochemical analysis of the highly diastereoselective carbonyl ene reaction is shown in Figure 3. The stereoelectronic requirement for transposition of the ene and enophile for the carbonyl ene reaction limits the possible transition states to the two boat-like **TS-B1** would be expected to be strongly favored, because **TS-B2** would have substantial *pseudo* 1,3-diaxial interactions between the phenyl and carbomethoxy moieties.

Conclusions

In summary, we have developed a novel, multi-step, domino sequence for the asymmetric synthesis of highly decorated cyclopentanes. The domino sequence proceeds in high yield and with well-defined stereochemistry. These studies demonstrate the synthetic utility of

highly reactive metal-carbenoids to initiate an elaborate sequence of reactions, resulting in the rapid generation of synthetic complexity.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Figure 3. Transition states for the carbonyl ene reaction.

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Scheme 1. Stereoselective synthesis of cyclopentanes







Scheme 3. Cyclopentane synthesis from 1 and 2.



Scheme 4.

Domino ylide formation/[2,3]-sigmatropic rearrangement/Oxy-Cope rearrangement/ tautomerization/Ene reaction/silyl deprotection/6-*exo-trig* cyclization.



Rh₂(*R*-DOSP)₄: 69% yield, >20 : 1 dr Rh₂(*S*-DOSP)₄: 57% combined yield, 1 : 2.1 dr

Scheme 5. Derivitization of (–)-pulogeol 13.

Optimization of the one-pot cyclopentane synthesis.



73					<i>i</i> a	
entry	Rh ₂ (S-DOSP) ₄	solvent	temp (°C)	yield ^a (%)	dr^{b}	ее ^с (%)
	1 mol %	<i>n</i> -Hex	67	0		
2	1 mol %	DCE	83	43	> 20:1	54
ю	1 mol %	EtOAc	LL	18	> 20 : 1	51
4	1 mol %	PhCH ₃	110	71	> 20:1	70
5	1 mol %	c-Hex	81	91	> 20 : 1	74
9	1 mol %	Hep	98	88	> 20 : 1	80
7	1 mol %	Hep	80	95	> 20:1	82
×	0.1 mol %	Hep	80	85	> 20:1	79
6	0.01 mol %	Hep	80	75	> 20 : 1	79
^a Isolated	yield.					
Determ	ined from ¹ H NMF	s of crude r	eaction r	nixture.		

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 c Determined by chiral HPLC.

Table 2

Scope of the one-pot cyclopentane synthesis with alcohols 2.

	ر ج ب		Rh ₂ (S-DOSP) _k Sc(OTf) ₃ (20 n heptane, 0 - 80	(1 mol %) nol %) 0 °C	MeO2C OH
2a-i	1	1			Ph ^é R 7a-i
entry	comp.	R	yield ^d (%)	dr^{b}	ee ^c (%)
-	а	Me	95	> 20 : 1	82
2	þ	<i>i</i> -Pr	67	> 20 : 1	80
ю	c	<i>i</i> -Bu	73	> 20 : 1	80
4	q	<i>n</i> -Hex	80	> 20 : 1	78
5	9	22	86	> 20 : 1	76
9	f		45 <i>d</i>	> 20 : 1	90 (96) 6
٢	ac) ~OTBS	65^{f}	> 20 : 1	78
8	Ч	Bn	42	> 20 : 1	87
6	i	SMT View	59	> 20 : 1	84
^a Isolated	yield.				
b Determin	ned from ¹ H	NMR of crud	le reaction n	nixture.	
$c_{ ext{Determin}}$	ned by chiral	HPLC.			
$^{d}_{\mathrm{Yield}\mathrm{of}}$	the deprotec	ted ketone.			
$^{e}_{\mathrm{Number}}$	in parenthes	es indicates et	e of recrysta	llized prod	uct.

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 $f_{\rm E}$ reaction conducted in refluxing heptane in the absence of Sc(OTf)3.

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Table 3

Scope of the one-pot cyclopentane synthesis with diazoacetates 8.

HO~~	لاً ۲	N2 CO,Me	DOSP)₄ (1 mc f)₃ (20 mol %) e, 0 - 80 °C	1%) MeO ₂ C	HO
2a		8a-d		ັ⊻ິ	a-d
entry	comp.	R	yield ^a (%)	dr^{b}	ee ^c (%)
1	а	p-CF ₃ C ₆ H ₄	63	> 20 : 1	78
2	q	p-(OMe)C ₆ H ₄	94	> 20:1	87
б	с	$p ext{-BrC}_{6} ext{H}_{4}$	48	> 20 : 1	92
4	q	Et	63	> 20 : 1	64
^a Isolated yi	eld.				
b Determine	d from ¹ H	NMR of crude rea	ction mixtur	ف	
$c_{\mathrm{Determine}}$	d by chiral	HPLC.			