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## Highly Stereoselective C—C Bond Formation by Rhodium-Catalyzed Tandem Ylide Formation/[2,3]-Sigmatropic Rearrangement Between Donor/Acceptor Carbenoids and Chiral Allylic Alcohols

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## Abstract

The tandem ylide-formation/[2,3]-sigmatropic rearrangement between donor/acceptor rhodiumcarbenoids and chiral allyl alcohols is a convergent C—C bond forming process, which generates two vicinal stereogenic centers. Any of the four possible stereoisomers can be selectively synthesized by appropriate combination of the chiral catalyst  $Rh_2(DOSP)_4$  and the chiral alcohol.

## **1. INTRODUCTION**

Chiral allylic alcohols are readily available and have been widely used as versatile building blocks in organic synthesis.<sup>1</sup> Recently we discovered an unexpected reaction between rhodium carbenoids and allylic alcohols.<sup>2</sup> Normally alcohols react with carbenoids to form O-H insertion products.<sup>3-5</sup> However, we found that the reaction between donor/acceptorsubstituted carbenoids and racemic allylic alcohols bearing a 3,3-dimethyl functionality resulted in an enantioselective [2,3]-sigmatropic rearrangement.<sup>2a,6</sup> Homoallylic alcohols containing a single stereogenic center were formed in which the enantioselectivity was governed by the chirality of the catalyst rather than the chirality of the starting alcohol. As the resulting products can be used in extended domino sequences,<sup>2b</sup> we became interested in broadening the substrate scope and generality of the reaction. In particular, we wished to explore the possibility of generating products containing vicinal stereocenters in a stereoselective manner (Scheme 1). In this paper, we demonstrate that all four of the possible stereoisomers of the products can be selectively and predictably generated by using the appropriate combination of chiral allylic alcohol and chiral catalyst. The allylic stereocenter of the products is controlled by the chirality of the allylic alcohol and the alkene geometry, whereas the homoallylic stereocenter is dictated by the chirality of the catalyst.

## 2. RESULTS AND DISCUSSION

We began our investigations by studying the reaction of the stereoisomers of 3-penten-2-ol (1) with styryldiazoacetate 2, catalyzed by either  $Rh_2(R$ -DOSP)<sub>4</sub> or  $Rh_2(S$ -DOSP)<sub>4</sub> (Table 1). The reactions of the four possible combinations of (*E*)-1 and  $Rh_2(DOSP)_4$  revealed that all the stereoisomers of the products 3 could be obtained in a stereoselective manner (>9:1 dr and >99% ee<sup>7</sup>) (entries 1–4). A comparison of entries 1 and 2 (and 3 and 4) demonstrated

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Supporting Information. Synthetic details and spectral data. This material is available free of charge via the Internet at http://pubs.acs.org.

that the stereo-center at  $C_3$  of the product was governed by the configuration of the allyl alcohol. In contrast, a comparison of entries 1 and 3 (and 2 and 4) demonstrated that the chiral catalyst controled the configuration at  $C_2$ . The reactions of (S, Z)-1 with Rh<sub>2</sub>(R-DOSP)<sub>4</sub> and Rh<sub>2</sub>(S-DOSP)<sub>4</sub> were also examined (entries 5 and 6). Significant matched and miss-matched interactions between the chiral entities were displayed in these reactions.<sup>8</sup> The Rh<sub>2</sub>(R-DOSP)<sub>4</sub>-catalyzed reaction of (S,Z)-1 with **2** was very efficient, generating (2S,3R)-**3** in 69% yield and with 94:6 dr and >99% ee (entry 5). The stereochemical configuration of the product was the same as that of the product derived from the Rh<sub>2</sub>(R-DOSP)<sub>4</sub>-catalyzed reaction of (R,E)-1 (entry 4). However, the Rh<sub>2</sub>(R-DOSP)<sub>4</sub>-catalyzed reaction of (S,Z)-1 with **2** was a missmatched reaction. In this case a 3:1 mixture of diastereomers was produced in low overall yield (35% for the major diastereomer) (entry 6).

The tandem ylide formation/[2,3]-sigmatropic rearrangement was examined with a series of donor/acceptor-substituted diazoacetates with a variety of aryl and alkenyl substituents. In all cases, the major diastereomer was produced with very high asymmetric induction (>99% ee), but the diastereoselectivity was variable. In the case of the aryldiazoacetates, **4a** and **4b**, the diastereo-selectivity was 9:1 (Table 2, entries 1–2). The *p*-bromostyryl derivative **4c** (entry 3) was comparable to the unsubstituted phenyl system (Table 1, entry 1). The butenyl-and propenyl-substituted diazo compounds (**4d** and **4e**, respectively) underwent the rhodium-catalyzed transformation with high levels of asymmetric induction (entries 4 and 5). These results are consistent with previous examples of the high enantioselectivity exhibited in the Rh<sub>2</sub>(S-DOSP)4 catalyzed reactions of diazoacetates **4a-e**.<sup>2</sup> In entry 6, the unsubstituted vinyldiazoacetate **4f** was obtained in modest yield (43%) and with poor diastereoselectivity (79:21 dr) It is well established that Rh<sub>2</sub>(*S*-DOSP)<sub>4</sub>–catalyzed cyclopropanations with **4f** proceed with moderate enantiocontrol<sup>9</sup> and the moderate diastereoselectivity observed in entry 6 is consistent with a low level of stereocontrol by Rh<sub>2</sub>(*S*-DOSP)<sub>4</sub> in this case.

The tolerance of the reaction to various substituents on the alcohols was then studied, and these results are summarized in Table 3. In general, extended aliphatic and aryl substituents at the  $C_3$  position of the alcohol 6 were well tolerated (entries 1–2) including the 3,3disubstituted substrate (entry 3). This substrate exemplified the utility of the metal-carbenoid transformation, facilitating the high yielding preparation of a product bearing two contiguous quaternary stereogenic centers with high levels of enantioselectvity and diastereoselectivity. Allyl alcohols with relatively bulky substituents, such as isopropyl and trimethylsilyl (6d and 6e) also afforded the corresponding rearrangement products, but the yields were modest (60% and 42%, respectively). An array of alcohols bearing  $C_2$ substitution (6f-h) were evaluated, and they were also amenable to this transformation (entries 6–8). It was expected that in the metal-bound oxonium-ylide intermediate formed any functionality at  $C_2$  would be oriented away from the catalyst and thus, would have little consequence on the reactivity. Finally, the effect of various functional groups at the carbinol position was explored in entries 9-11 and in all cases the desired products were formed. Of particular significance is the reaction of the mono-benzyl-protected 1,2-diol 6k, which was capable of selective reaction at the allylic alcohol over the benzyl ether functionality.

The synthetic utility of the rhodium-catalyzed sigmatropic rearrangement with the chiral alcohols lies in the ability to generate two adjacent stereogenic centers in a controlled and predictable manner. A distinctive feature of the transformation is the generation of a quaternary hydroxyl carbonyl moiety bearing a vicinal stereocenter, which is a structural feature embedded in a number of natural products.<sup>10</sup> We also decided to demonstrate the broader synthetic potential of the reaction by illustrating a two-step conversion of the products to enones, containing a chiral center  $\alpha$  to the carbonyl (equations 1 and 2). Enones containing quaternary (**10a**) and tertiary (**10b**) stereocenters  $\alpha$  to the carbonyl were readily

prepared in excellent yields. A particularly appealing feature of this approach to chiral enones is the likelihood that a chiral catalyst would not be required because the stereogenic center a to the carbonyl is controlled by the chirality of the starting alcohol.



(1)

(2)

Due to the uniformly high levels of asymmetric induction for the tandem ylide-formation/ [2,3]-sigmatropic rearrangement, we sought a general transition-state model which would rationalize the observed stereochemical results.<sup>6</sup> It has been well established that the Rh<sub>2</sub>(*S*-DOSP)<sub>4</sub>-catalyzed reactions of vinyldiazoacetates results in attack at the *Re* face of the vinylcarbenoid.<sup>11</sup> The [2,3]-sigmatropic rearrangement would be expected to proceed through an envelope-like transition state, in which  $A_{1,3}$ -strain is minimized.<sup>12</sup> A reasonable model, which takes into account the established stereochemical understanding of these reactions is shown in Figure 1. *Re* face attack of the carbenoid by (*S*,*E*)-1 would generate an intermediate that would preferentially undergo a 2,3-sigmatropic rearrangement through TS-A, in which the  $A_{1,3}$  strain is minimized. This transition state would lead to the formation of the observed (2*R*,3*R*) isomer. Likewise, the reaction of (*R*,*E*)-1 would proceed through TS-B, which would generate the (2*R*,3*S*) isomer. The Re face attack on the carbenoid controls the stereochemistry at C<sub>2</sub> in the product and at least in the case of (*E*)-1, the carbenoidinduced stereogenic center does not have a significant influence on the stereochemistry of the [2,3]-sigmatropic rearrangement.

### 3. CONCLUSION

In summary, the tandem ylide-formation/[2,3]-sigmatropic rearrangement between donor/ acceptor rhodium-carbenoids and chiral allyl alcohols is a convergent C—C bond forming process, which generates two vicinal stereogenic centers. Any of the four possible stereoisomers can be selectively synthesized by appropriate combination of the chiral catalyst Rh<sub>2</sub>(DOSP)<sub>4</sub> and the chiral alcohol. Only traces of O—H insertion products are observed in these reactions, which further illustrates the difference in reactivity of donor/ acceptor carbenoids compared to conventional carbenoids, lacking a donor group.

#### Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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**Figure 1.** Transition-state analysis for the formation of **3**.



**Scheme 1.** Rhodium(II)-catalyzed [2,3]-sigmatropic rearrangement of allyl alcohols

Stereocontrolling elements of the tandem ylide formation/[2,3]-sigmatropic rearrangement										
	Ph CO2Me		ee, %d	66<	66<	66<	66<	66<	66<	
		Me <sup>xxx</sup> Me	dr <sup>c</sup>	92:8	91:9	92:8	95:5	94:6	75:25	
			Yield, $\%b$	70	64	54	78	69	35	
			Product	MeO <sub>2</sub> C OH Ph 2R,3R-3	Phoece of Me	Ph HQ CO <sub>2</sub> Me Me 2.5.3.8-3	Phylophylophylophylophylophylophylophylop	Ph HQ CO2Me	Phoec of the phoece of the pho	
			Rh <sub>2</sub> (DOSP) <sub>4</sub>	S	S	К	R	R	S	
			Substrate	Me An Me	$\underset{R,E-1}{\overset{OH}{\underset{R,E-1}{\overset{OH}{\underset{R}}}}}$	$\underset{S,E-1}{Me}_{Me}$	$\underset{R,E-1}{\overset{OH}{\underset{R,E-1}{\bigvee}}}_{ME}$	Me OH	Me OH S,Z-1	
			entry <sup>a</sup>	-	0	ω	4	Ś	Ś	6



 $^{a}$ Reaction conditions: To a pentane solution of the allyl alcohol (1 equiv) and Rh<sub>2</sub>(S-DOSP)<sub>4</sub> (0.01 equiv)at 0  $^{\circ}$ C under an atmosphere of Ar was added a solution of the diazo compound (1.1 equiv) in pentane solution over 1.5 h. The reaction was stirred for a further 1 h at 0 °C and then concentrated under reduced pressure.

 $b_{\rm Isolated}$  yield of the major diastereomer.

Table 1

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 $^{\mathcal{C}} \text{Determined by }^{1}\text{H}$  NMR analysis of the crude reaction mixture.

 $d_{\rm Determined}$  by chiral HPLC.

Table 2



dDetermined by chiral HPLC.

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Scope of the allyl alcohol  $6^a$ 





 $^{a}$ Same reaction conditions as described in Table 1.

b Isolated yield of the major diastereomer.

 $^{C}$ Determined by <sup>1</sup>H NMR analysis of the crude reaction mixture.

dDetermined by chiral HPLC.

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