Exploring the Role of a Conserved Class A Residue in the Ω-Loop of KPC-2 β-Lactamase

A MECHANISM FOR CEFTAZIDIME HYDROLYSIS[*]

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Background: The Ω-loop (Arg-164 to Asp-179) is a conserved region among class A β-lactamases.
Results: In KPC-2, a carbapenemase of significant clinical importance, Arg-164 substitutions in the Ω-loop selectively enhanced ceftazidime hydrolysis.
Conclusion: Ceftazidime resistance may proceed by a novel mechanism that uses covalent trapping and hydrolysis.
Significance: Future antibiotic design must consider the distinctive behavior of the Ω-loop of KPC-2.

Gram-negative bacteria harboring KPC-2, a class A β-lactamase, are resistant to all β-lactam antibiotics and pose a major public health threat. Arg-164 is a conserved residue in all class A β-lactamases and is located in the solvent-exposed Ω-loop of KPC-2. To probe the role of this amino acid in KPC-2, we performed site-saturation mutagenesis. When compared with wild type, 11 of 19 variants at position Arg-164 in KPC-2 conferred increased resistance to the oxyimino-cephalosporin, ceftazidime (minimum inhibitory concentration; 32→128 mg/liter) when expressed in *Escherichia coli*. Using the R164S variant of KPC-2 as a representative β-lactamase for more detailed analysis, we observed only a modest 25% increase in k_{cat}/K_m for ceftazidime (0.015→0.019 $\mu M^{-1} s^{-1}$). Employing pre-steady-state kinetics and mass spectrometry, we determined that acylation is rate-limiting for ceftazidime hydrolysis by KPC-2, whereas deacylation is rate-limiting in the R164S variant, leading to accumulation of acyl-enzyme at steady-state. CD spectroscopy revealed that a conformational change occurred in the turnover of ceftazidime by KPC-2, but not the R164S variant, providing evidence for a different form of the enzyme at steady state. Molecular models constructed to explain these findings suggest that ceftazidime adopts a unique conformation, despite preservation of Ω-loop structure. We propose that the R164S substitution in KPC-2 enhances ceftazidime resistance by proceeding through "covalent trapping" of the substrate by a deacylation impaired enzyme with a lower K_m. Future antibiotic design must consider the distinctive behavior of the Ω-loop of KPC-2.

Resistance to β-lactam antibiotics (*e.g.*, penicillins, cephalosporins, and carbapenems; Fig. 1) remains a major clinical challenge. In Gram-negative pathogens, β-lactamases (EC 3.5.2.6) represent the most common enzymatic mechanism of antibiotic resistance (1). As a consequence of facile transfer of β-lactamase (*bla*) genes among bacteria, the commercialization of novel classes of β-lactams has repeatedly prompted widespread resistance (2–5). Presently, carbapenemase-type β-lactamases are emerging that hydrolyze carbapenem antibiotics, which until recently were used as "last resort" drugs for treatment of life-threatening infections (3, 6).

Among carbapenemases, the Ambler class A *Klebsiella pneumoniae* carbapenemase (KPC)[5] family is the most widespread (1, 7). The bla_{KPC-2} gene is encoded on a plasmid and has been found in *Enterobacteriaceae* (*e.g.*, *K. pneumoniae* and *Escherichia coli*) and *Pseudomonas aeruginosa* (8–10). Moreover, pathogens harboring KPC-2 are resistant to all β-lactams and β-lactamase inhibitors, severely limiting treatment options and leading to high fatality rates (9, 11–14). Therefore, the biochemical characterization of KPC-2 is critical to inform the design of novel β-lactams with efficacy against KPC-2-producing pathogens.

In an effort to understand the structural basis for catalysis by class A carbapenemases, crystal structures of KPC-2, NmcA, and SME-1 have been solved (Fig. 2A) (15–18). Distinct from non-carbapenemase class A enzymes, KPC-2 β-lactamase exhibits a widened active site and a recessed position of the catalytic nucleophile, Ser-70. An additional notable feature is a Cys-69–Cys-238 disulfide bond, covalently linking the active site b3 β-strand and the helix encompassing Ser-70. Complementary mutagenesis studies have determined that Trp-105, Thr-237, and Arg-220 of KPC-2 define novel catalytic properties (19–21).

The Ω-loop is a "hot spot" for substitutions that extend the substrate spectrum of many class A enzymes (22, 23). This

[5] The abbreviations used are: KPC, *K. pneumoniae* carbapenemase; MIC, minimum inhibitory concentration; ESI, electrospray ionization.
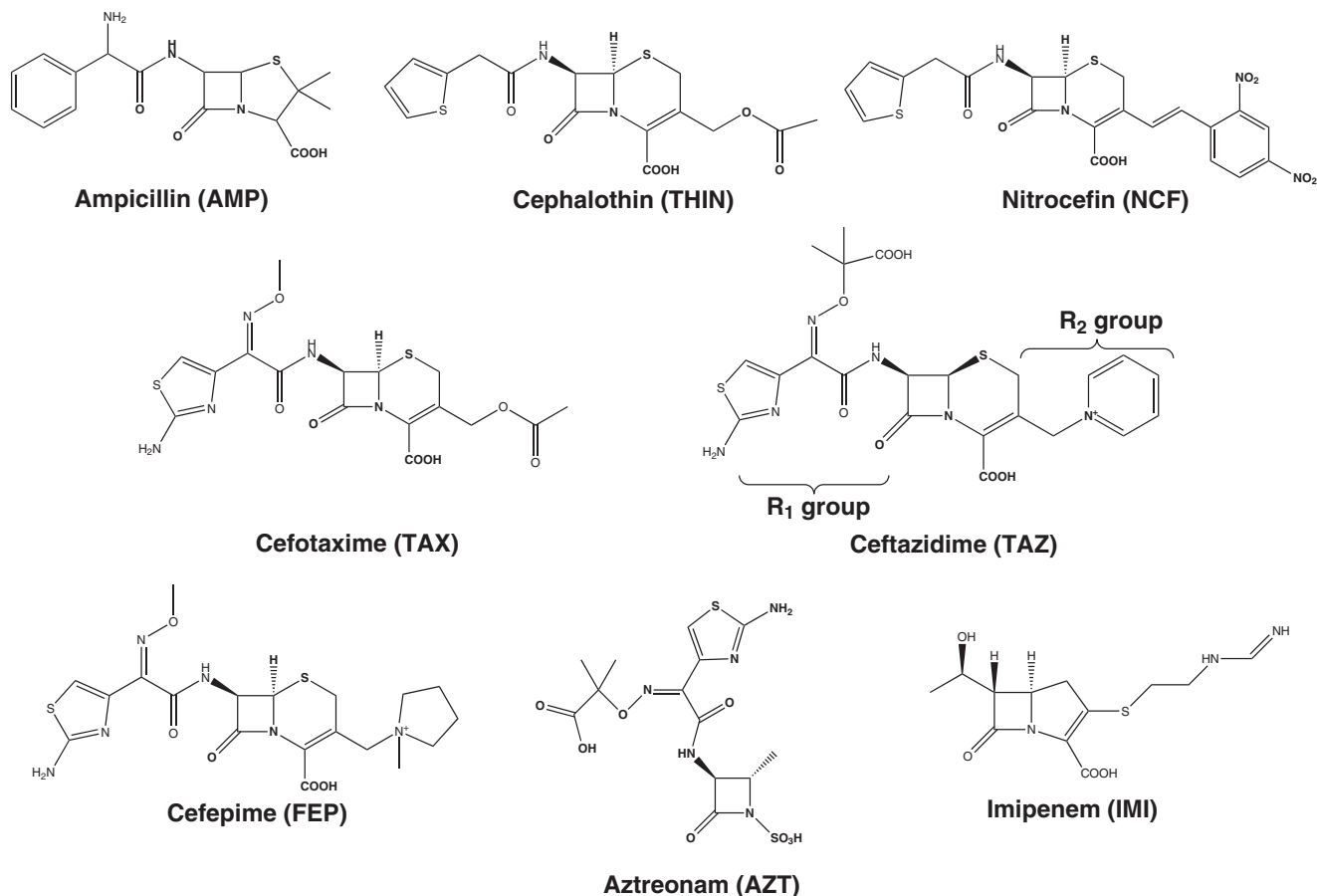
FIGURE 1. **Chemical structures of β-lactams used in this study.** On ceftazidime, R_1 and R_2 side chains are indicated. Note the bulky, oxyimino R_1 side chains characteristic of the extended spectrum cephalosporins cefotaxime, ceftazidime, and cefepime, and the monobactam aztreonam.
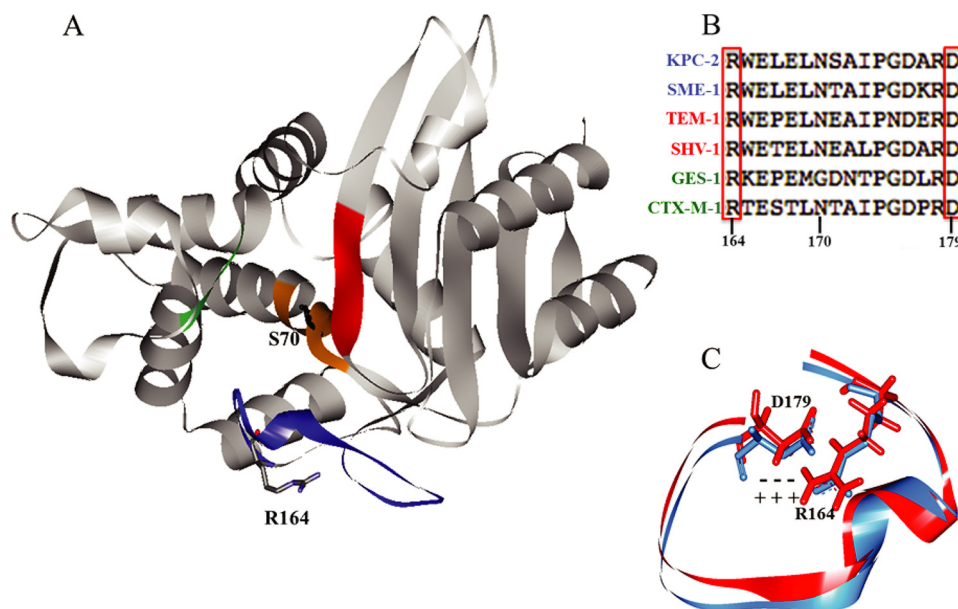

FIGURE 2. *A*, structure of KPC-2 (Protein Data Bank code 2OV1). The nucleophile Ser-70 and residue of focus in this study, Arg-164, are indicated. Active site regions are color-coded as follows: *orange*, S*XX*K motif (residues 70–73); *green*, SDN loop (130–132); *blue*, Ω-loop (164–179); *red*, b3 β-strand (234–242). *B*, sequence alignment of the Ω-loops of KPC-2 with other class A carbapenemases (*blue*), penicillinases (*red*), and cephalosporinases (*green*). Note the strict conservation of Arg-164 and Asp-179. *C*, overlay of Ω-loops from TEM-1 (*red*) and KPC-2 (*blue*), showing the salt bridge between Arg-164 and Asp-179.

region (residues 164–179 in class A β-lactamases) forms the "floor" of the active site and contains Glu-166 and Asn-170, two residues that participate in priming a water molecule for deacy-

lation of the β-lactam (Fig. 2*B*) (15, 24–26). In the class A penicillinases TEM-1 and SHV-1, the flexibility of the Ω-loop is restricted by a network of hydrogen and ionic bonds, one of

which connects Arg-164 and Asp-179 at opposite poles of the "neck" of the Ω-loop (Fig. 2*C*) (27, 28). Disruption of the Arg-164/Asp-179 salt bridge by substitution at either residue enhances the "flexibility" of the Ω-loop, thereby allowing for hydrolysis of extended-spectrum cephalosporins (29, 30). In some variants, this occurs in the apo enzyme (*e.g.*, the R164S variant of TEM-1), whereas in others, it requires ligand binding (*e.g.*, SHV R164H or -S) (31, 32).

There exists a high degree of both sequence and structural conservation in the Ω-loop region of KPC-2 and TEM-1, including conservation of the salt bridge between Arg-164 and Asp-179 (Fig. 2, *B* and *C*). Therefore, we were compelled to generate a full library of substitutions at Arg-164 in KPC-2 and perform a complete biochemical characterization of the Arg-164 variant enzymes. The data reveal that Ω-loop substitutions at this conserved residue distinctly impact catalysis of ceftazidime an important advanced generation antibiotic and argue for a "covalent trapping" mechanism of resistance in addition to hydrolysis.

MATERIALS AND METHODS

Plasmids, PCR Primers, and Mutagenesis—For antimicrobial susceptibility testing, all KPC-2 variants were generated in the pBR322-*catI* plasmid harboring the *bla*$_{KPC-2}$ gene as described previously (9). For protein purification the R164S substitution was generated in pET24a(+) harboring *bla*$_{KPC-2}$ without the sequence encoding the leader peptide, a plasmid generated previously (21). The QuikChange XL site-directed mutagenesis kit (Agilent, Santa Clara, CA) was used to generate all variants, following the manufacturer's protocol. For PCR mutagenesis, seven degenerate oligonucleotide sets covering all 19 amino acid substitutions (Thermo Fisher Scientific, Hampton, NH) at Arg-164 were generated and yielded 15 of the 19 variants. The R164H, R164S, R164I, and R164F variants were constructed by site-directed mutagenesis. All primers were designed using Agilent genomics software. All of the mutations were verified by sequencing.

Antimicrobial Susceptibility Testing—Lysogeny broth (LB) minimal inhibitory concentration (MICs) determinations for β-lactams were performed on single clones of *E. coli* DH10B (Invitrogen) harboring pBR322-*catI-bla*$_{KPC-2}$ or a variant. We performed our experiments according to the criteria of the Clinical and Laboratory Standards Institute (33). The MICs shown in Table 1 are the mode of at least three independent experiments.

Immunoblotting—*E. coli* DH10B harboring pBR322-*catI-bla*$_{KPC-2}$ or a variant were grown to stationary phase in LB containing 20 mg/liter chloramphenicol selection. The cells were pelleted and boiled for 10 min in Laemmli buffer to induce lysis. The details of the immunoblotting procedure were described previously (21). The membranes were probed with a rabbit anti-KPC-2 polyclonal antibody (Sigma-Genosys) and, as a loading control, an anti-DNA-K rabbit polyclonal antibody (Stressgene, San Diego, CA).

β-Lactamase Purification—KPC-2 and R164S proteins were purified from *E. coli* Origami2 DE3 (Novagen, Darmstadt, Germany) harboring pET24a(+) *bla*$_{KPC-2}$ or the variant as described previously (21). Briefly, bacteria grown in super optimal broth were induced to express β-lactamase with 500 μM isopropyl-β-D-thiogalactoside for 2 h at 37 °C, at a starting A_{600} of 0.6. Bacterial cells were pelleted and frozen at −20 °C for at least 12 h. The pellets were lysed in a buffer containing 40 mg/ml lysozyme, and the cleared supernatant was subjected to preparative isoelectric focusing. Gel segments containing β-lactamase were identified by assaying for activity against nitrocefin. The purity of active fractions was assessed by SDS-PAGE followed by staining with Coomassie Brilliant Blue R250. Fractions that were qualitatively determined to be more than ~80% pure were pooled and concentrated in preparation for anion exchange chromatography, which was performed using a HiTrap Q HP column (GE Healthcare). Purity was reassessed, and fractions containing more than ~90% purity were pooled, exchanged into 10 mM PBS (pH 7.4), and concentrated. Protein concentration was calculated by measuring absorbance at λ_{280} and using the extinction coefficient of the protein. The identity of each protein preparation was confirmed by mass spectrometry (described below). The proteins were stored at either 4 °C for ~1 week or, for longer periods, at −20 °C in 50% glycerol.

Kinetics—β-Lactamases hydrolyze β-lactam substrates through a two-step mechanism, which is commonly represented as follows.

$$E + S \underset{k_{-1}}{\overset{k_1}{\rightleftharpoons}} E{:}S \xrightarrow{k_2} E - S \xrightarrow{k_3} E + P \qquad \text{(Eq. 1)}$$
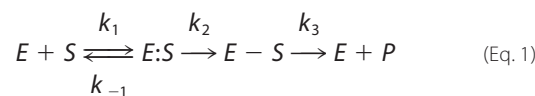
In this scheme, k_1 is a second order rate constant governing the formation of the precovalent Henri-Michaelis complex, *E*:*S*, and k_{-1} is a first order rate governing its dissociation. k_2 represents the first order rate constant for formation of the acyl-enzyme, $E - S$. k_3 represents the first order rate constant for product deacylation and release.

The relationship between reaction velocity (*v*) and substrate concentration can be represented by the Henri-Michaelis-Menten equation.

$$v = (V_{max}{*}[S])/(K_m + [S]) \qquad \text{(Eq. 2)}$$

$$k_{cat} = V_{max}/[E] \qquad \text{(Eq. 3)}$$

$$k_{cat} = k_2{*}k_3/(k_2 + k_3) \qquad \text{(Eq. 4)}$$

Qualitatively, k_{cat} represents the first order rate constant for reaction of the *E*:*S* to generate product and regenerate free enzyme, K_m is the concentration of substrate required to reach ½ V_{max}, and k_{cat}/K_m is the second order rate constant for reaction of the free *E* and *S* to form free *E* and *P*. To measure the steady-state parameters k_{cat}, K_m, k_{cat}/K_m, reactions were initiated at room temperature in 10 mM PBS (pH 7.4) containing variable amounts of enzyme depending on the enzyme-substrate pair, where conditions always conform to the steady-state assumption (<10% of the substrate is consumed in the reaction). Initial velocities were obtained on an Agilent 8453 diode array spectrophotometer by monitoring changes in absorption upon β-lactam ring opening, using the following extinction coefficients: ampicillin, $\Delta\epsilon_{235} = -900$ M^{-1} cm^{-1}; cephalothin, $\Delta\epsilon_{262} = -7660$ M^{-1} cm^{-1}; nitrocefin, $\Delta\epsilon_{482} = 17,400$ M^{-1} cm^{-1}; cefoxitin, $\Delta\epsilon_{260} = -6225$ M^{-1} cm^{-1}; cefo-

taxime, $\Delta\epsilon_{262} = -7250$ M^{-1} cm^{-1}; ceftazidime, $\Delta\epsilon_{256} = -7600$ M^{-1} cm^{-1}; cefepime, $\Delta\epsilon_{260} = -750$ M^{-1} cm^{-1}; aztreonam, $\Delta\epsilon_{318} = -640$ M^{-1} cm^{-1}; and imipenem, $\Delta\epsilon_{299} = -9000$ M^{-1} cm^{-1}. For velocity determinations for ampicillin, cephalothin, nitrocefin, and imipenem, a 1-cm path length quartz cuvette was used. For cefoxitin, cefotaxime, ceftazidime, cefepime, and aztreonam, a 0.2-cm path length quartz cuvette was used.

Steady-state kinetic parameters were derived by a nonlinear least squares fit of the data to Equation 2 using the program Origin 8.1 (OriginLab, Northampton, MA). The velocities for cefepime hydrolysis by KPC-2 did not approach saturation at testable concentrations. In this case, the slope of the velocity *versus* cefepime concentration line is the second order rate constant for hydrolysis at steady state, k_{cat}/K_m.

Pre-steady-state Kinetics—Pre-steady-state kinetics were carried out on an Applied Photophysics π*-180 stopped flow spectrophotometer. 100 μM ceftazidime was reacted with either 2 or 4 μM KPC-2 or the R164S variant at 25 °C for times ranging from 2 to 20 s. At least four reactions were performed under each condition. The data were analyzed using Pro K software. To determine the burst, π, the linear portion of the curve corresponding to the steady-state velocity was extrapolated back to the *y*-intercept at time = 0 s. The following equation describes the burst.

$$\pi = [E]*[a/(a + b)]^2 \qquad \text{(Eq. 5)}$$

Based on the reaction in Equation 1, *a* is the net rate constant for formation of *E-S* from free *E* and *S* ($a = k_2[S]/(K' + [S])$ where $K' = (k_{-1} + k_2)/k_1$), and *b* is k_3 (34).

Electrospray Ionization (ESI)-MS—In these experiments, 19 μM (10 μg) β-lactamase was incubated with ceftazidime at a molar ratio of 1:100 at room temperature in 10 mM PBS (pH 7.4). For KPC-2 and the R164S variant, the reactions proceeded for 5, 15, and 60 s. The reactions were quenched by equilibration to 0.2% formic acid. The samples were desalted and concentrated using a C_{18} ZipTip (Millipore, Bedford, MA) according to the manufacturer's protocol. ESI-MS of the eluted sample was performed on a Q-STAR Elite quadrupole time of flight mass spectrometer equipped with a TurboIon spray source (Applied Biosystems, Framingham, MA), and spectra were deconvoluted using the Applied Biosystems Analyst program.

CD and Thermal Denaturation—CD experiments were carried out in a Jasco (Easton, MD) J-815 spectrometer with a Peltier effect temperature controller. Quartz cells with a 0.1-cm path length were used for all experiments. For thermal denaturation, 19 μM KPC-2 or the R164S variant with or without 250 μM ceftazidime were monitored for helical content by CD at λ_{220} between 25 and 65 °C with a heating rate of 2 °C/min. Two-state, reversible denaturation was verified by the return of the original CD signal of the protein upon cooling and by comparison of the enzyme activity prior to denaturation and then after cooling.

For CD spectra, 19 μM KPC-2 or the R164S variant was incubated with 250 μM ceftazidime in PBS (pH 7.4). CD spectra were obtained at room temperature, and data points were recorded every 0.1 nm between λ_{200} and λ_{260} with a scan rate of 20 nm/min. CD spectra for ceftazidime alone were taken and subtracted from the spectra of enzyme and ceftazidime.

Molecular Modeling—The KPC-2 β-lactamase crystal coordinates (Protein Data Bank accession code 2OV5) were used to generate the R164S variant by substituting serine for arginine at position 164. The KPC-2 structure and the generated R164S model were optimized by energy minimization using Discovery Studio 3.1 software (Accelrys, San Diego, CA) (19). The minimization was performed in several steps, using steepest descent and conjugate gradient algorithms to reach the minimum convergence (0.01 kcal/mol*Å). The protein was immersed in a water box, 7 Å from any face of the box, and the solvation model employed was with periodic boundary conditions. The force field parameters of CHARMm were used for minimization, and the particle mesh Ewald method addressed long range electrostatics. The minimized structures were utilized for constructing the Michaelis-Menten and acyl complexes with ceftazidime. The ligand structure was built using D.S. Fragment Builder tools. The CHARMm force field was applied; the molecule was solvated with periodic boundary conditions and minimized using a Standard Dynamics Cascade protocol (one minimization using the steepest descent algorithm, followed by adopted basis Newton-Raphson algorithm and three subsequent dynamics stages at a constant pressure and 300 K). The minimized ligand was docked in the active site of the enzyme/model using flexible docking algorithm (19). The generated conformations were visually inspected, the most favorable were solvated, and energy was minimized using the conjugate gradient algorithm with periodic boundary conditions to 0.001 minimum derivatives. To check the stability and look for possible conformational changes of the complex, molecular dynamics simulation was conducted (19). The molecular dynamics simulation was run for 6 ps.

RESULTS AND DISCUSSION

Consequences of the Arg-164 Substitution on the Phenotype, Protein Expression, and Stability of KPC-2 Variants Expressed in E. coli—Site-saturation and site-directed mutagenesis were performed in the pBR322-*catI* vector harboring bla_{KPC-2} yielding all 19 variants at position Arg-164 (Ambler numbering scheme). To evaluate how substitutions at this site impact bacterial susceptibility to β-lactams, we transformed pBR322-*catI-bla*$_{KPC-2-R164X}$ into the uniform *E. coli* DH10B background and measured MICs. Penicillins (ampicillin) narrow and extended spectrum cephalosporins (cephalothin, cefoxitin, cefotaxime, ceftazidime, and cefepime), aztreonam, and carbapenems (imipenem, meropenem, ertapenem, and doripenem) were tested. The results are presented in Table 1.

KPC-2 expressed in *E. coli* conferred high level resistance to ampicillin (MIC = 4096 mg/liter) and cephalothin (1024 mg/liter). In general, all the engineered Arg-164 substitutions resulted in reduced MICs to ampicillin, cephalothin, cefotaxime, cefepime, aztreonam, and imipenem. In contrast, 11 of 19 substitutions conferred elevated MICs for ceftazidime *versus* KPC-2 (32 to 128 mg/liter). The R164S variant of KPC-2 expressed in *E. coli* was chosen for further study (see below) because this substitution demonstrates the highest level of

TABLE 1

MIC values for β-lactams (mg/liter)

AMP, ampicillin; THIN, cephalothin; TAX, cefotaxime; TAZ, ceftazidime; FEP, cefepime; AZT, aztreonam; IMI, imipenem.

	AMP	THIN	TAX	TAZ	FEP	AZT	IMI
E. coli DH10B	1	4	.06	0.25	0.06	0.25	0.12
pBR322-*catI-bla*$_{KPC-2}$	4096	1024	8	32	8	256	4
Hydrophobic							
R164G	2048	256	4	128	4	128	1
R164A	2048	256	4	128	4	64	0.5
R164L	1024	256	0.12	64	1	16	0.12
R164V	256	64	0.06	64	0.5	2	0.25
R164I	256	16	0.12	128	0.25	1	0.12
R164M	1024	256	0.12	64	1	16	0.25
Positive							
R164K	2048	256	4	128	4	128	1
R164H	2048	512	4	64	8	128	2
Negative							
R164D	2048	256	4	128	4	128	1
R164E	2048	256	4	128	4	64	0.5
Aromatic							
R164F	1024	128	0.12	64	1	16	0.25
R164Y	1	8	.06	4	0.12	0.5	0.12
R164W	1	4	.06	0.25	0.06	0.25	0.12
Polar							
R164S	2048	256	4	128	4	64	1
R164T	2048	256	4	128	4	64	1
R164N	2048	512	4	128	4	256	2
R164Q	2048	512	4	128	4	128	1
Unique							
R164P	1	8	0.5	128	0.5	2	0.12
R164C	1	8	0.06	0.25	0.06	0.25	0.12
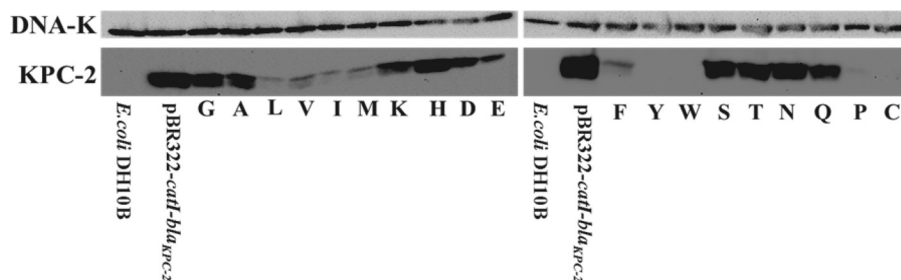
FIGURE 3. **Immunoblot for KPC-2 and, as a loading control, DNA-K, probing *E. coli* DH10B cells harboring pBR322-*catI-bla*$_{KPC-2\ R164X}$ grown in LB.**

resistance to ceftazidime and possesses clinical correlates in other class A enzymes (35, 36).

This pattern of susceptibility findings prompted us to examine expression levels of the Arg-164 variants grown in *E. coli* DH10B. Here we asked whether loss or gain of resistance resulting from altering the amino acid at Arg-164 was a consequence of changes in protein expression or catalytic properties. With some exceptions (R164L, R164I, R164M, R164E, R164V, and R164F), our immunoblotting experiments showed that steady-state expression levels paralleled the resistance to ampicillin and cephalothin (Fig. 3). These observations indicated that, in general, the phenotype of the KPC variants expressed in bacteria may be partly governed by the expression level of the enzyme.

As a biochemical correlate, we next determined whether the change in the nature of the amino acid across the "neck" of the Ω-loop had any "cost" on protein thermal stability. We purified the R164S variant for analysis because this substitution in other class A β-lactamases is predicted to disrupt the key salt bridge with Asp-179. Melting temperature (T_m) was determined by monitoring CD difference at λ_{220} as the temperature was raised

from 25 to 65 °C, and affirmed a two-state, reversible denaturation. The T_m was 56 °C for KPC-2 and 52 °C for the R164S variant (Fig. 4). These observations support the notion that loss of the ionic bond between Ω-loop "neck" residues Arg-164 and Asp-179 destabilizes class A enzymes while enhancing their activity against ceftazidime. This "activity/stability tradeoff" is common among β-lactamases (37, 38).

Because Arg-164 substitutions both decreased protein expression and enhanced resistance to ceftazidime, we hypothesized that ceftazidime may increase the stability of the R164S variant, leading to a greater intracellular abundance of the enzyme and, perhaps, higher ceftazidime resistance. Incubation with ceftazidime raised the $T_m = \sim1$ °C for both KPC-2 and R164S, suggesting that ceftazidime did not substantially stabilize the R164S variant (Fig. 4). We interpret from this finding that the more ceftazidime resistance in the variant stems not from enhanced protein stability but is likely a result of changes in the catalytic properties conferred by the R164S substitution.

β-Lactam Kinetics—To understand how disruption of the salt bridge between Arg-164 and Asp-179 impacts catalysis by
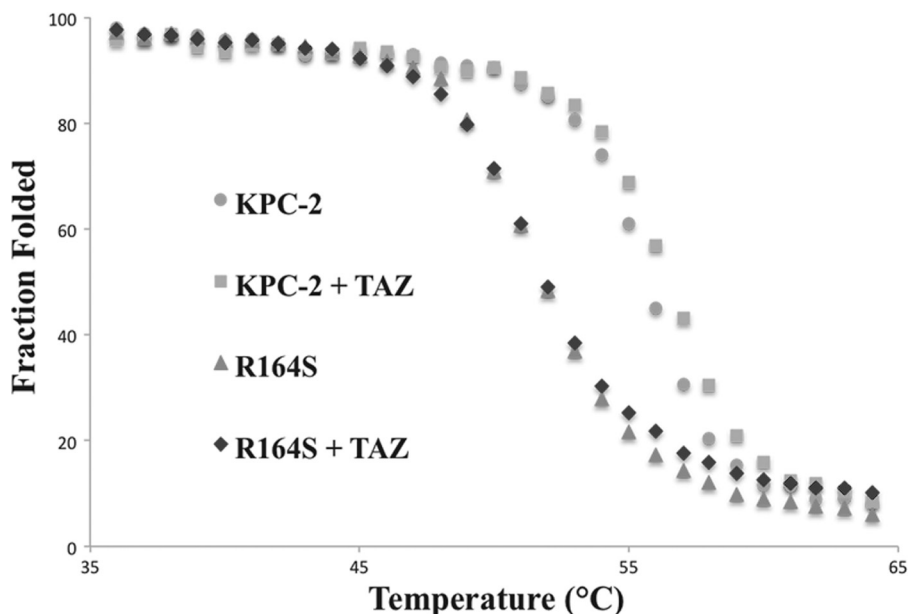
FIGURE 4. **Thermal denaturation experiment.** 19 μM KPC-2 or the R164S variant was incubated with or without 250 μM ceftazidime and tested by circular dichroism at λ_{220}. Mean residue molar ellipticity values (MRE) were converted to a fraction of folded protein at each temperature by the formula: Fraction folded = $MRE_{time\ point}$/$MRE_{minimum}$. The calculated T_m values for KPC-2 and the R164S variant were 56 and 52 °C, respectively, whereas ceftazidime (*TAZ*) stabilized both proteins by 1 °C.

TABLE 2

Kinetics with β-lactams

The units of measure are μM for K_m, s^{-1} for k_{cat}, and μM^{-1} s^{-1} for k_{cat}/K_m.

	KPC-2	R164S	Fold change (R164S/KPC-2)
Ampicillin			
K_m	51 ± 7	93 ± 10	1.8
k_{cat}	26 ± 1	26 ± 1	1
k_{cat}/K_m	0.5	0.3	0.6
Cephalothin			
K_m	30 ± 6	40 ± 8	1.3
k_{cat}	250 ± 20	280 ± 20	1.1
k_{cat}/K_m	8.3	7	0.8
Nitrocefin			
K_m	7 ± 0.7	8 ± 1	1.1
k_{cat}	95 ± 10	66 ± 7	0.7
k_{cat}/K_m	13	8	0.6
Cefoxitin			
K_m	340 ± 100	260 ± 50	0.8
k_{cat}	8 ± 1	3.2 ± 0.5	0.4
k_{cat}/K_m	0.024	0.012	0.5
Cefotaxime			
K_m	515 ± 160	260 ± 70	0.5
k_{cat}	180 ± 20	90 ± 10	0.5
k_{cat}/K_m	0.35	0.35	1
Ceftazidime			
K_m	375 ± 60	180 ± 30	0.5
k_{cat}	5.8 ± 0.6	3.5 ± 0.3	0.6
k_{cat}/K_m	0.015	0.019	1.3
Cefepime			
K_m	>1000	900 ± 150	NA
k_{cat}	>6	6.1 ± 0.6	NA
k_{cat}/K_m	0.0067	0.0067	1
Aztreonam			
K_m	920 ± 160	700 ± 115	0.8
k_{cat}	280 ± 30	120 ± 10	0.4
k_{cat}/K_m	0.30	0.17	0.6
Imipenem			
K_m	14 ± 2	11 ± 2	0.8
k_{cat}	13 ± 1	5.5 ± 0.5	0.4
k_{cat}/K_m	1	0.5	0.5

KPC-2, we performed steady-state kinetic analysis on KPC-2 and the R164S variant. In concert with the relatively moderate MIC changes, the R164S variant did not exhibit more than a 3-fold change in any steady-state kinetic parameter for all of the substrates tested (Table 2). For ampicillin and cephalothin, the substitution slightly decreased k_{cat}/K_m because of an increased
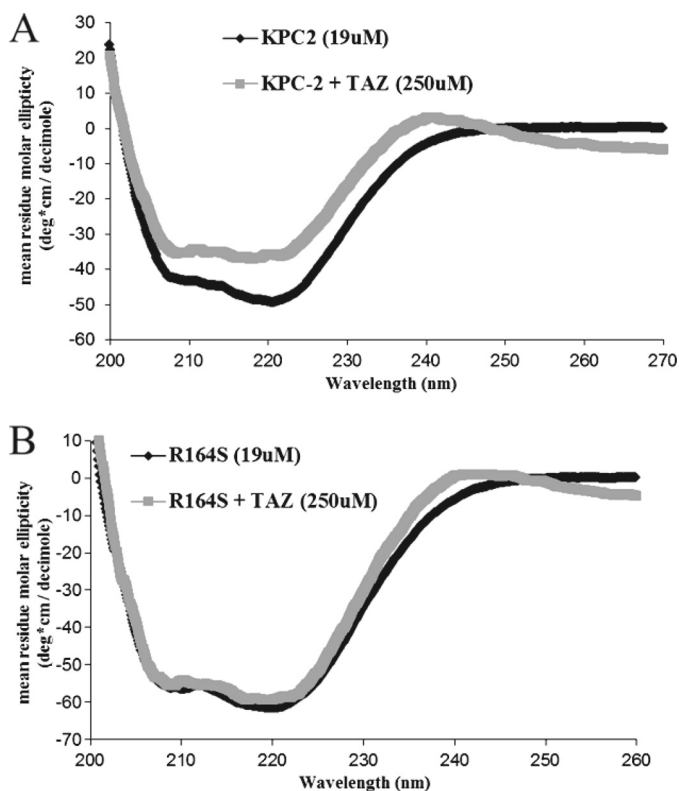
FIGURE 5. Determination of protein secondary structure by circular dichroism difference spectrum. 19 μM KPC-2 (*A*) or the R164S variant (*B*) was incubated either alone or with 250 μM ceftazidime (*TAZ*), and CD spectra were gathered and plotted as described under "Materials and Methods."

K_m. For oxyimino cephalosporins cefotaxime, ceftazidime, and cefepime, we observed decreased k_{cat} values and K_m values ($\sim$2-fold), indicating that the substitution lowers K_m but inhibits catalysis for these substrates (Table 2). The k_{cat}/K_m was altered only for ceftazidime, with the variant exhibiting a modest increase of 25% in catalytic efficiency.

Mechanistic Insights into Ceftazidime Hydrolysis and Resistance—Given that the modest increase of the R164S enzyme in k_{cat}/K_m for ceftazidime was not completely sufficient to explain the 4-fold enhanced resistance phenotype when expressed in bacteria, we addressed the reaction mechanism in more detail.

To determine whether conformational changes accompany ceftazidime hydrolysis, we obtained CD spectra during the course of the reaction. KPC-2 underwent a subtle loss of helicity throughout $\lambda_{200-240}$ upon ceftazidime hydrolysis (Fig. 5*A*). In contrast, the R164S variant CD spectrum did not change appreciably upon ceftazidime hydrolysis (Fig. 5*B*). Previously, Taibi-Tronche *et al.* (39) observed a loss of helical content of TEM-1 during the course of cefepime hydrolysis. This similarity between KPC-2 and TEM-1 raises the possibility that conformational changes may be an important feature of extended spectrum cephalosporin hydrolysis by class A enzymes. Our findings suggest that the CD change observed may be a "slow" relaxation of KPC-2 required to accommodate the bulky oxyimino R_1 substituent of the substrate. In stark contrast, the CD spectrum of the R164S variant did not change in the reaction, suggesting that the variant may be "primed" to accommodate

ceftazidime and does not require the conformational change, resulting in an enhanced K_m. The differences exhibited in this experiment may be further evidence for a lower K_m of the R164S variant for ceftazidime. Importantly, these data raise the possibility that a different form of the enzyme is populated at steady-state for the Arg-164 variant.

To understand how the KPC-2 R164S substitution impacts the kinetics of acylation and deacylation, we performed timed ESI-MS and pre-steady-state kinetic experiments. By ESI-MS (t = 5, 15, or 60 s) and a ceftazidime:enzyme molar ratio (100:1), we determined the proportion of acyl-enzyme to Michaelis-Menten complex and used this ratio as an approximation of the relative rate constants for acylation (k_2) and deacylation (k_3). We observed that acyl-enzyme species were not captured with the KPC-2 enzyme at any time point, suggesting that k_3 is faster relative to k_2 (Fig. 6*A*). Surprisingly, for the R164S variant, a species corresponding to an acyl-enzyme adduct appeared at 5 s, diminished at 15 s, and disappeared by 60 s, indicative of a slower k_3 (Fig. 6*B*) and accumulation of the acyl-enzyme at steady-state.

The m/z of the R164S acyl-enzyme adducts at 5 and 15 s were $+\Delta468$ (Fig. 6*C*). The $+\Delta468$ adduct corresponds to a ceftazidime adduct in which the R_2 group is eliminated (R_2 elimination has been described previously and occurs rapidly with ceftazidime upon β-lactam ring opening) (31). An unexpected mass of 29,264 Da, which we are unable to identify, was also observed.

Stimulated by the results of the ESI-MS analysis and to verify the results independently, pre-steady-state reactions were performed with ceftazidime. Here 100 μM ceftazidime was reacted with either 2 or 4 μM of KPC-2 or the R164S variant for 2–20 s; a burst of substrate hydrolysis was detected with both enzymes. For KPC-2, the burst, π, was low (0.20 for 2 μM KPC-2) and slightly increased with increasing enzyme concentration (Fig. 7). These values suggest that k_3 is fast relative to k_2. Therefore, k_3 has a negligible impact on k_{cat}, and k_2 is the rate-limiting step in ceftazidime hydrolysis. On the other hand, the π value of the R164S variant was much higher (1.43 for 2 μM R164S), and when the concentration of enzyme was raised, the π value increased by 0.3 (Fig. 7). Because of the difficulty achieving saturating substrate concentrations as a result of weak binding, under these conditions the observed burst will not quantitatively reflect the maximal amplitude for either enzyme. However, the mutant clearly shows a burst of product formation occurring pre-steady-state, providing evidence that k_3 is slow relative to the net rate constant for formation of $E:S$ and $E - S$. These results are perfectly consistent with the direct observation of acyl-enzyme by MS and the CD data showing distinct enzyme forms for the mutant and wild type at steady-state.

Overall, we propose that the R164S substitution in KPC-2 augments ceftazidime resistance by enhancing k_{cat}/K_m (only 25%) and also through a change in the kinetic mechanism of the reaction. The data suggest that a mechanism favoring slow hydrolysis relative to acylation (*i.e.*, enhanced accumulation of acyl-enzyme) as observed in the variant is dominant over hydrolysis, as seen in KPC-2, in determining phenotype. Although the R164S substitution has only a small effect on the measured steady-state kinetic parameters, it likely results in a
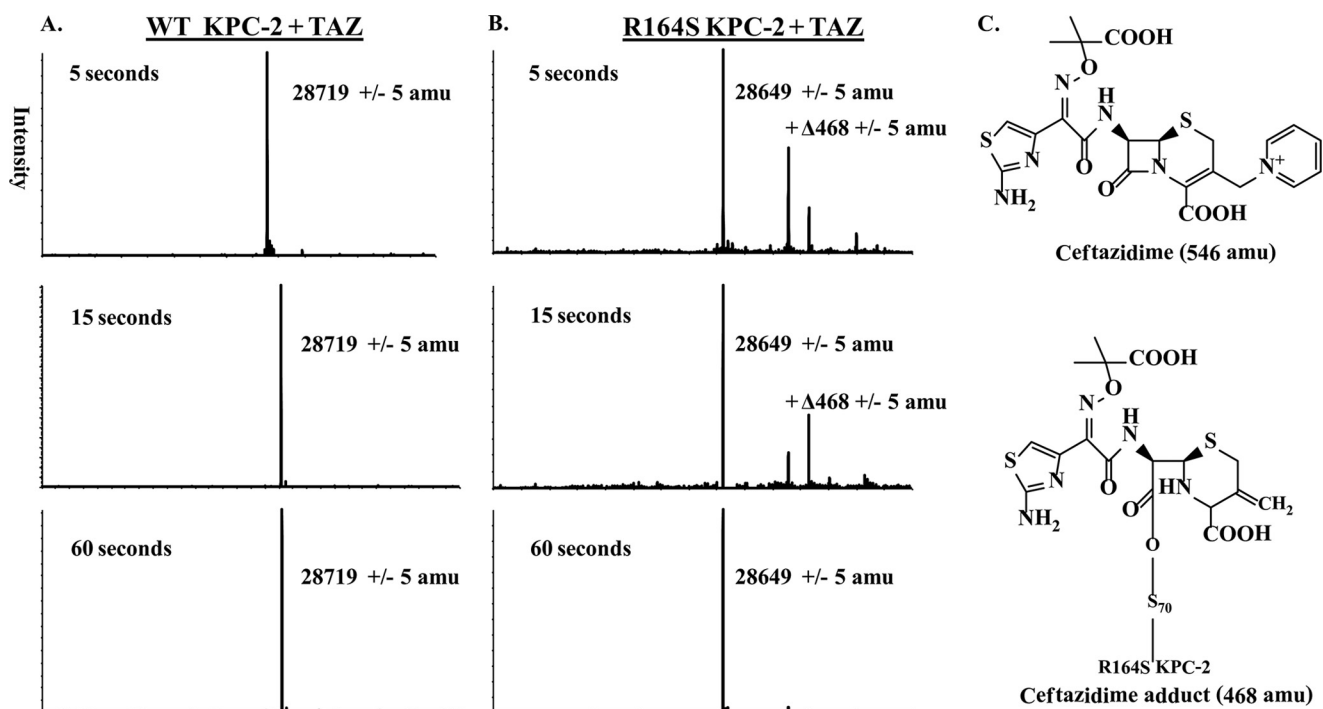
FIGURE 6. **Deconvoluted ESI-MS spectra of ceftazidime reaction coordinate.** *A* and *B*, 19 μM KPC-2 (*A*) or the R164S variant (*B*) was incubated with ceftazidime (*TAZ*) at a 100:1 substrate to enzyme ratio, and reactions were quenched at 5, 15, or 60 s in 0.2% formic acid, processed, and then analyzed as described under "Materials and Methods." All of the measurements have an error of $\pm$ 5 atomic mass units. *C*, ceftazidime undergoing R_2 elimination to obtain the $+\Delta468$ adduct.
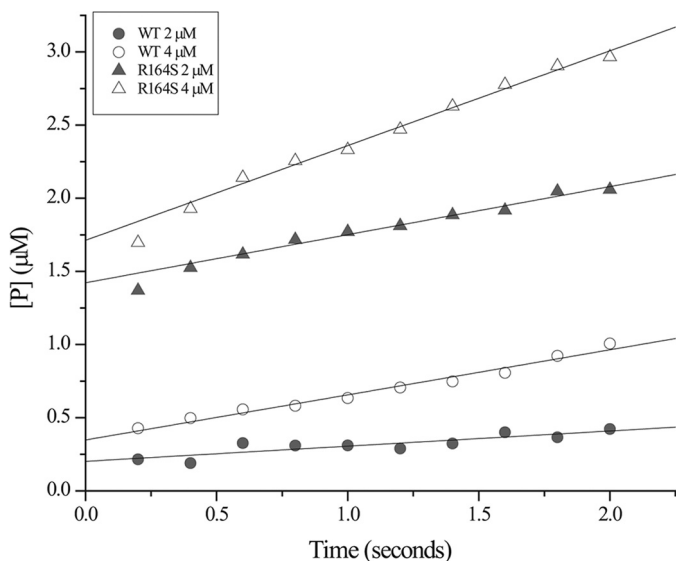

FIGURE 7. **Pre-steady-state kinetic traces of ceftazidime hydrolysis.** In a stopped flow spectrophotometer, 100 μM ceftazidime was incubated with 2 and 4 μM KPC-2 or 2 and 4 μM the R164S variant, and the rate of product (*P*) formation (converted from absorbance readings, $\Delta\epsilon_{256} = -7600$ M^{-1} cm^{-1}) was followed for 2 s. Representative *traces* are shown.

very large decrease in the hydrolysis rate, such that it becomes at least partially rate-limiting for k_{cat}. These characteristics are reminiscent of those of the TEM-1 M182T/E166R variant, which binds ceftazidime with low K_m but is completely deacylation-deficient yet still confers a high level resistance (40). This study in TEM advanced a "covalent trapping" mechanism of resistance, in which the enzyme serves as a "sink" for substrate rather than as a catalyst. We infer from this that regeneration of existing apo-enzyme via deacylation, which is faster for KPC-2, is less important for determining resistance levels compared with the translation of nascent enzyme.

We next considered whether "covalent trapping" of ceftazidime by KPC-2 is substrate-specific in *E. coli*. In answering this question, we point out that resistance to cefotaxime, a similar substrate, is not enhanced by the R164S substitution. The substitution causes a similar ~2-fold reduction in k_{cat} and K_m for cefotaxime, resulting in an unchanged catalytic efficiency. Unlike ceftazidime, however, cefotaxime is a "fast" substrate ($k_{cat} = $ ~200 s^{-1}), such that the absolute reduction in k_{cat} observed in the variant for cefotaxime (~100 s^{-1}) is much greater than for ceftazidime (~3 s^{-1}). On the basis of this key difference, we posit that under our experimental conditions, deacylation is dominant in clearing "fast" substrates *in vivo*, whereas commitment to catalysis may be the dominant determinant of resistance for "slow" substrates.

Modeling of Ceftazidime Michaelis-Menten Complexes—To anticipate the structural basis for the enhanced ceftazidime resistance, we constructed molecular models of the Michaelis-Menten complex of ceftazidime with KPC-2 and the R164S variant to help explain our findings. To start, our model predicts that the disposition of the Ω-loop is similar, but not identical (Fig. 8*A*). Our representations did not suggest that there is disorder in R164S, as observed in other structures (30, 32). The mini helix encompassing Glu-166 and Asn-170 is slightly more mobile in the R164S variant compared with KPC-2, but we do not observe a major reconfiguration of this segment as observed in the R164S-substituted TEM-64 enzyme. We suspect that the Ω-loop of KPC-2 is likely stabilized by as yet unidentified inter-
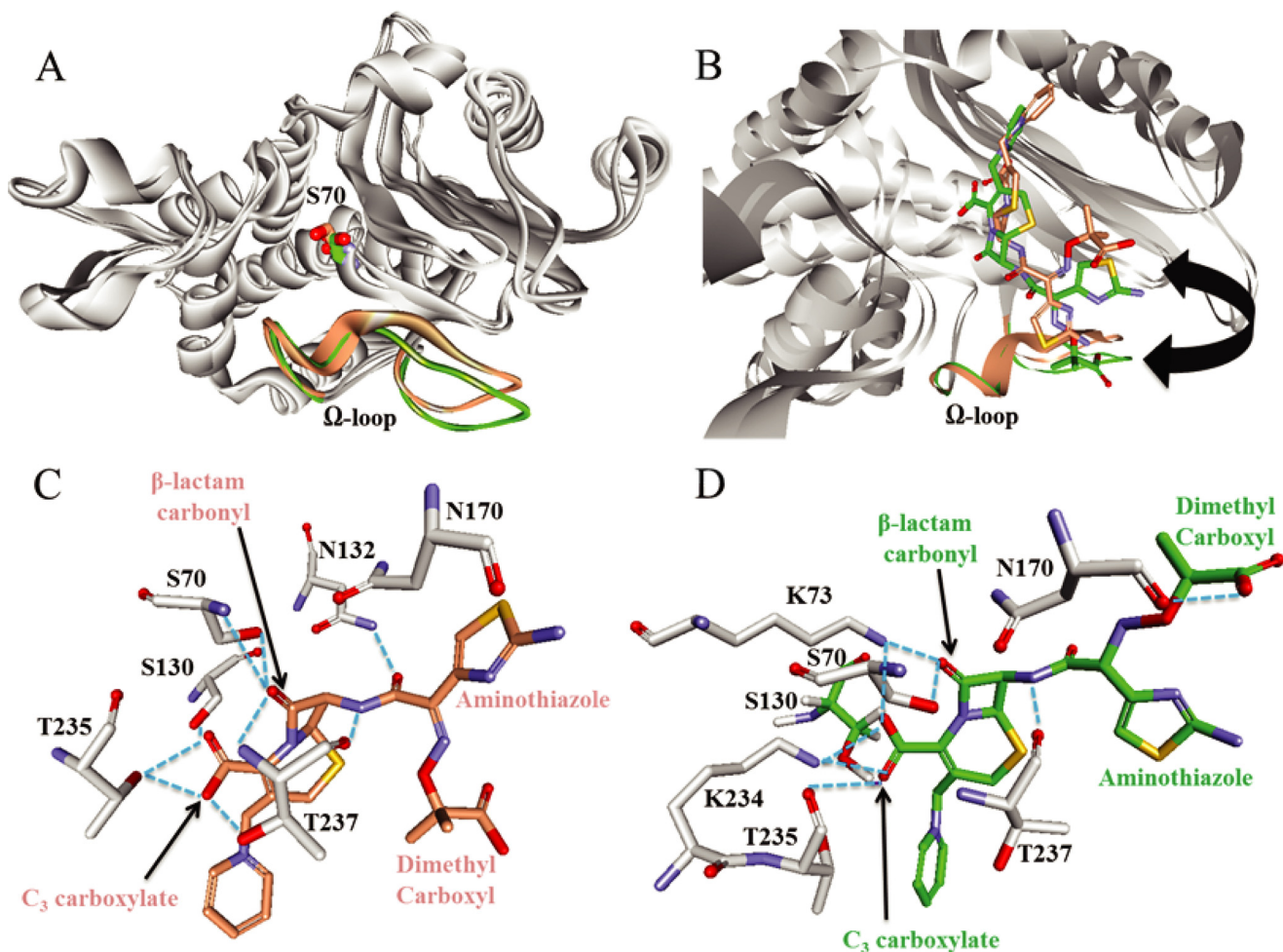
FIGURE 8. **Molecular representations of the Michaelis-Menten complexes of ceftazidime with KPC-2 or the R164S variant.** *A*, superimposition of the KPC-2 (*beige*) and R164S (*green*) KPC-2 models, with the Ω-loops and positions of Ser-70 highlighted. The Ω-loop demonstrates slightly increased flexibility in the R164S variant, but the spatial positions of all side chains are within 1 Å in the two models. Conversely, the Oγ of Ser-70 has shifted 2.3 Å in the model for KPC-2. *B*, superimposition of the models, highlighting the conformations of ceftazidime in the KPC-2 (*beige*) and R164S (*green*) models. The *arrow* indicates the flipping of the ceftazidime aminothiazole and di-methyl carboxyl groups in the two models. *C* and *D*, representation of the predicted interactions by DS Studio 3.1 between ceftazidime and indicated active site residues in KPC-2 (*C*) and the R164S variant (*D*), indicated by *blue dotted lines*. In particular, note the hydrogen bond between the main chain oxygen of Asn-170 and the R₁ carboxyl group of ceftazidime in the R164S, but not KPC-2 model. Also note the more extensive stabilization of the β-lactam carbonyl in the KPC-2, but C₃ carboxylate in the R164S variant.

actions between Ω-loop residues and other active site motifs or a network of water molecules that reside in the Ω-loop cavity (27).

A second notable feature of the model is the 2.3 Å offset of Oγ of Ser-70 between the two structures (Fig. 8*A*). By superimposing our models on the KPC-2 crystal structure (Protein Data Bank code 2OV1), we find that a shift occurs in KPC-2, not the variant (data not shown). The Ser-70 shift recalls the crystal structure of Ke *et al.* (15), who found evidence of enhanced flexibility of Ser-70, likely imparting the range of motion necessary to position the nucleophile for hydrolysis of many β-lactams.

Most importantly, ceftazidime adopts different conformations in the two models (Fig. 8*B*). In the KPC-2 β-lactamase, the aminothiazole ring in the R₁ group of ceftazidime is oriented toward the Ω-loop, whereas the di-methyl carboxyl group is facing away, exposed to solvent. This conformation is reminiscent of previous crystal structures of D,D-peptidase/transpeptidase with cephalothin and cefotaxime (31, 41). In the R164S variant, however, the position of the aminothiazole ring and the

dimethyl carboxyl group are reversed, with the di-methyl carboxyl oriented toward the Ω-loop.

Because of these possible conformational differences, the stabilizing interactions between ceftazidime and the two enzymes are quite different. In KPC-2, the β-lactam carbonyl is situated in the oxyanion hole created by the main chain nitrogens of Ser-70 and Thr-237 (Fig. 8*C*). In contrast, in the R164S variant, the carbonyl oxygen is not in the oxyanion hole, coordinated by Lys-73 Nζ and the nucleophile Ser-70 Oγ (Fig. 8*D*). These differences in the position of the β-lactam carbonyl oxygen likely have important effects on catalysis, which can only proceed if the incipient negative charge on the oxygen in the tetrahedral transition state is properly stabilized by the oxyanion hole. Furthermore, in the R164S variant, the interaction between the ceftazidime carbonyl oxygen and the nucleophile (Oγ) would likely inhibit its activation for attack on the carbon of the β-lactam carbonyl, slowing catalysis.

Lastly, steric crowding between the side chain of Asn-170 and the aminothiazole group of ceftazidime contributes to the poor affinity of class A non-carbapenemases for extended spec-

trum cephalosporins (31). Similarly, in the KPC-2 model, the terminal amino group of the aminothiazole ring resides within 3.2 Å of the main chain oxygen of Asn-170 (Fig. 8*C*). This interaction imposes steric and possibly electrostatic constraints. In contrast, in the R164S variant, the ceftazidime carboxyl group is located 2.8 Å from the main chain oxygen of Asn-170 and is predicted to hydrogen bond with the ceftazidime moiety (Fig. 8*D*). This "flipping" of the ceftazidime carboxyl and aminothiazole in the active site likely facilitates an electrostatically favorable interaction in the R164S variant, contributing to the enhanced affinity for ceftazidime.

CONCLUSION

In this work we have evaluated how substitutions at residue Arg-164, a conserved active site Ω-loop "neck" residue commonly mutated in TEM-type (ESBLs), impact catalysis and stability of the class A carbapenemase KPC-2. The "ceftazidime-selective" resistance compelled us to investigate the kinetic mechanism for ceftazidime hydrolysis by KPC-2 and the highly resistant R164S variant. We feel that two important points emerge from the findings in this paper.

The first relates to structure/function; Ω-loop flexibility is directly coupled to catalytic properties in class A β-lactamases. The microbiological, kinetic, and modeling data for KPC-2 R164S, considered in concert with similar data for SHV-1 and TEM-1, suggest that greater enhancement in Ω-loop flexibility in the penicillinases is accompanied by greater catalytic and phenotypic changes than for KPC-2, where we see more moderate catalytic effects. In contrast, despite relative preservation of Ω-loop structure, the thermal stability of the R164S variant is decreased. TEM-1 Arg-164 and SHV-1 Asp-179 variants display similarly decreased stability, but concomitantly display increased Ω-loop flexibility and a much more dramatic gain in ceftazidime activity. Why this "activity/stability tradeoff" is more favorable to TEM-1 and SHV-1 penicillinases than to the KPC-2 carbapenemase is unknown.

The second important point is that the kinetic mechanism also argues for a "covalent trapping" ceftazidime resistance mechanism by R164S. The kinetic data suggest that for substrates such as ceftazidime, slow hydrolysis and accumulation of the acyl-enzyme is more critical than regeneration of native enzyme via deacylation in determining resistance phenotypes in bacteria. Design of "next generation" oxyimino cephalosporins and monobactams should pay special attention to β-lactamase affinity and acylation kinetics to preserve their clinical longevity.

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