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Synthesis and evaluation of potent and selective human V1a receptor antagonists as potential ligands for PET or SPECT imaging

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Abstract

SRX246 is a potent, highly selective human vasopressin V1a antagonist that crosses the blood–brain barrier in rats. CNS penetration makes SRX246 an ideal candidate for potential radiolabeling and use in visualization and characterization of the role of the V1a receptor in multiple stress-related disorders. Before radiolabeling studies, cold reference analogs of SRX246 were prepared. This study describes the synthesis and in vitro screening for human V1a receptor binding and permeability of fluoro, iodo, and methyl reference compounds for SRX246 and the preparation of a tin precursor. For each compound, the potential utility of corresponding radiolabeled analogs for PET and SPECT imaging is discussed.

Keywords

Arginine vasopressin; Vasopressin V1a antagonist; Radiolabeling; PET; SPECT

1. Introduction

Arginine vasopressin (AVP) is a cyclic nonapeptide hormone that modulates a broad range of physiological and behavioral effects by binding to specific GPCRs in the central nervous system and certain peripheral tissues/sites.^{1,2} Three distinct AVP receptor subtypes have been identified—V1a, V1b, and V2. V1a is the predominant AVP receptor found in the brain. Especially high levels are found in the cerebral cortex, limbic system, hypothalamus, and brainstem.¹ The V1b receptor is located in the pituitary gland and it is less widespread in the brain than the V1a receptor. The V2 receptor is found in kidneys, where it mediates the antidiuretic effects of vasopressin. It is not generally thought to be expressed in the nervous systems of adult animals and humans. These findings have led to considerable interest in V1a and V1b receptors as potential targets for CNS therapeutics. The effects of AVP and its mechanism of action have been extensively studied and discussed elsewhere.^{3–6}

Selective V1a and V1b receptor antagonists that cross the blood–brain barrier have attracted considerable interest as potential therapeutics because the compounds represent a novel approach for the treatment of depression, anxiety, Post-Traumatic Stress Disorder (PTSD), and Intermittent Explosive Disorder.^{7–17}

Our interest is in the V1a receptor, where there is strong evidence from preclinical animal models as well as observations in humans indicating that this subtype may well play a central role in these indications.^{1,3,4,6,18–32} We were the first group to report the discovery of potent, highly selective, CNS-penetrating human V1a (hV1a) receptor antagonists.¹⁸ The technology places us in a strong position to build upon our earlier discoveries to develop imaging agents for the hV1a system based in particular on SRX246, one of our clinical candidates.

Both PET and SPECT protocols have proved very useful as tools to investigate in vivo drug behavior including a recent example for the vasopressinergic system focusing on PET and the human V1b (hV1b) receptor.^{33,34} The current study presents our progress generating reference compounds and a versatile tin intermediate that would allow for the preparation of both PET and SPECT tracers for the hV1a receptor, together with preliminary work on ¹²³I labeling.

2. Results and discussion

SRX246 is a mono beta-lactam derivative with an azetidone ring as a central pharmacophore core around which four distinct peripheral zones (Zone A, B, C and D) can be varied (Fig. 1). We intend to modify Zone B in SRX246 for radioisotope introduction (¹²³I, ¹⁸F or ¹¹C). This approach represents a less challenging synthetic pathway as compared to introducing the same tracers on the aryl group in either Zone A or C. We also know, based on previously described hV1a SARs,¹⁸ that Zone B is more tolerant of small structural modifications, especially at the *meta*-position of the phenyl ring.

In practice, the synthetic route (Scheme 1) leading to compounds **11a**, **11b** and **11c** was adapted from that previously published for the preparation of SRX246.¹⁸ The procedure is based on a chiral Staudinger 2+2 cyclo-addition reaction that allows the building of the azetidone ring. In this modified procedure, meta-substituted-*trans*-cinnamaldehydes were used instead of *trans*-cinnamaldehyde while the rest of the reaction sequence, reagents and conditions were kept the same. The three target compounds were obtained in consistent yield but required the preparation of the corresponding meta-substituted-*trans*-cinnamaldehydes. This was accomplished through a three step process (Scheme 2) from commercially available 3-fluoro or 3-methyl cinnamic acids.³⁵ Because of the lack of a reliable commercial source for 3-iodocinnamic acid, **1a** was obtained via a Horner–Wadsworth–Emmons reaction on the 3-iodobenzaldehyde instead.³⁶ Reduction of the ethyl cinnamate esters to the alcohols **2a**, **2b** and **2c** and further oxidation with pyridinium dichromate lead to the meta-substituted-*trans*-cinnamaldehyde **3a**, **3b** and **3c** in high yields. This three step process led to the target halogenated or alkyl derivatives in much higher overall yields compared to published procedures,³⁷ particularly for the iodo derivative. The absolute stereochemistry of the chiral methine carbons of the azetidone ring for compounds **11a**, **11b** and **11c** was assigned based on previous findings and confirmed by the ¹H and ¹³C NMR chemical shifts and coupling constants.¹⁸

After an earlier attempt to directly convert **11a** to **16**, we decided to evaluate the utility of our general synthetic pathway for the preparation of such a tin compound. It became quickly apparent that the preparation of the tin analog (compound **16**) represented a challenge because the conditions used for the synthesis of SRX246 were not directly transposable to

the synthesis of the tin compound. In particular, we had concerns about the potential detrimental cleavage of the tin substituent in the presence of an acid such as the one used (formic acid) to cleave the *t*-Butyl ester distal carboxyl acid protecting group off the D-aspartic acid and after the cyclo-addition step. A distinct synthetic pathway was designed (Scheme 3) that would circumvent this potential problem. In this new approach, we decided to introduce the two amide functions prior to the cyclo-addition step, which allowed us to avoid putting our tin group in jeopardy because the use of formic acid was now warranted in an earlier step of the process. In summary, it allowed the conservation of the alkyl stannyl group through the multi-step synthesis and also resulted in the 2+2 cyclo-addition being the final step of the reaction sequence. To achieve this, we also had to develop a method to prepare the new meta-trimethyltin-*trans*-cinnamaldehyde (**4**) needed for this approach. It was synthesized by direct modification of 3-iodocinnamaldehyde (**3a**) via a Stille reaction with hexamethylditin in the presence of a catalytic amount of tetrakis(triphenylphosphine)palladium.^{38,39} The desired trimethylstannane aldehyde was obtained in high yield (80%) and was easily purified by silica gel filtration. Following the preparation of the aldehyde and implementing this strategy, we were indeed able to obtain the target compound **16** in satisfactory overall yield.

Conversion to the cold and hot reference compounds, **11a** and [¹²³I]**11a**, via demetallation reaction was probed to further validate the potential use of **16** for ¹²³I radiolabeled ligands generation for SPECT imaging (Scheme 4). Preparation of **11a** and [¹²³I]**11a** from **16** was investigated following established methods.⁴⁰ In the presence of chloramine T (CAT) as the oxidant and sodium iodide as the iodine source, **16** was successfully converted to the cold iodo compound **11a** with a chemical yield superior to 98%. Radioiodination with [¹²³I]INa was conducted to afford [¹²³I]**11a** in 88% radiochemical yield. After RP-HPLC purification, the radiochemical purity of [¹²³I]**11a** was determined at 94% with a specific activity of 20 mCi/nmol. The potential use of **16** for the generation of ¹⁸F and ¹¹C radiolabeled ligands for PET imaging was also assayed in cold conditions only. Electrophilic fluorination of **16** with the well known fluorinated agent SelectFluor™, which has been highlighted recently as a promising fluorine source for ¹⁸F radiolabeling, afforded **11b** with a chemical yield of 63% (Scheme 4).⁴¹ Alternative approaches leading to high specific activity radioligands will also be investigated in future studies.^{42–44} Conversion of **16** to **11c** via a Stille coupling reaction used for ¹¹C radiolabeling⁴⁵ also showed promising results and confirmed the feasibility of ¹¹C radiolabeling (data not presented).

Having prepared compounds **11a**, **11b** and **11c** the next step was to evaluate their biological activity in vitro against the human vasopressin V1a receptor and in the Parallel Artificial Membrane Permeability Assays for the Blood–Brain Barrier (PAMPA-BBB) assay. As expected, the modifications in the cinnamyl part (Zone B) of the clinical candidate SRX246 did not affect dramatically the biological activity for the hV1a receptor (SRX246's hV1a K_i was reported previously at 0.3 nM).¹⁸ Compounds **11a**, **11b** and **11c** displayed potent nanomolar activity and extremely high affinity in vitro for the hV1a receptor (Table 1). Compound **11b** was the most active of the three potential radiolabeled ligands with a K_i of 0.61 nM. This is not surprising if one considers that fluorine is an excellent isostere for hydrogen,⁴⁶ creating one of the closest possible structural analogs of SRX246. In a screening assay for binding to hV1b and hV2 receptors, all the tested compounds were found with K_i greater than 1000 nM demonstrating excellent selectivity for the V1a receptor over other vasopressin receptor subtypes (Table 1). The lipophilicity was mildly affected by the addition of a fluorine atom or a methyl group with $c\text{Log}P$ increases of 0.16 and 0.49 units for **11b** and **11c**, respectively, compared to that of SRX246 ($c\text{Log}P = 4.26$, Spartan06). One can therefore reasonably predict that both compounds will exhibit bioavailability similar to SRX246, the first orally bioavailable, selective hV1a vasopressin antagonists with

CNS penetration.⁴⁷ A larger increase in lipophilicity is observed for **11a** (1.4 units for $c\text{Log}P$) which is generally the case with the addition of an iodine atom.

In vitro PAMPA-BBB were also carried out as a predictive tool for CNS penetration (Table 1). Both the fluorine and the methyl derivative **11b** and **11c** exhibited high permeability values ($Pe > 4.0 \times 10^{-6} \text{ cm s}^{-1}$), which strongly suggest high predicted passive Blood–Brain Barrier (BBB) permeation.⁴⁸ Compound **11a** exhibited a lower value ($2.0 \times 10^{-6} < Pe < 4.0 \times 10^{-6} \text{ cm s}^{-1}$), suggesting an uncertain predicted BBB permeation. However, BOLD (Blood-Oxygen- Level Dependence) imaging studies in rodents demonstrated that this particular compound has a CNS modulatory effect, demonstrating that it is capable of brain penetration.⁴⁹

3. Conclusions

We have developed a reliable method to prepare a versatile precursor that should allow access to both PET (¹⁸F, ¹¹C) and SPECT (¹²³I) derivatives of SRX246, a highly selective and potent hV1a antagonist. Furthermore, we have synthesized the F, I, and Me-bearing reference compounds and have shown that they exhibit strong hV1a receptor affinity, do not bind to V1b and V2 receptors, and have a high likelihood of brain penetration based on PAMPABBB results. These findings offer the potential to begin a broader investigation into the role of the human vasopressin V1a receptor in various CNS functions using relevant imaging tracers.

4. Experimental

4.1. Chemistry

4.1.1. (E)-Ethyl 3-(3-iodophenyl)acrylate (1a)—Triethyl phosphonoacetate (9.21 g, 41.12 mmol) in tetrahydrofuran (65 mL) was treated with 2.5 M *n*-butyl lithium in hexanes (12.82 mL, 32.07 mmol) at -78°C . The resulting mixture was stirred for 10 min and added via a cannula to a pre-cooled solution at -78°C of 3-iodobenzaldehyde (4.77 g, 20.56 mmol) in tetrahydrofuran (20 mL). The resulting mixture was stirred at -78°C for 40 min and slowly warmed to ambient temperature for 45 min. The reaction was quenched with saturated aqueous NH_4Cl (200 mL). The aqueous solution was extracted with diethyl ether. The resulting organic layer was dried over magnesium sulfate and evaporated. The residue was purified by silica gel chromatography (90:10 hexanes/ethyl acetate) to give 5.71 g (92%) of compound **1a** as a yellow oil; ¹H NMR (CDCl_3) δ 1.32 (t, $J = 7.1$ Hz, 3H); 4.24 (q, $J = 7.1$ Hz, 2H); 6.39 (d, $J = 16.0$ Hz, 1H); 7.09 (dd, $J = J' = 7.8$ Hz, 1H); 7.43–7.46 (m, 1H); 7.54 (d, $J = 16.0$ Hz, 1H); 7.64–7.69 (m, 1H); 7.83–7.86 (m, 1H).

4.1.2. General procedure for the formation of meta-substituted cinnamaldehyde (3 steps)

4.1.2.1. Step 1. General procedure for the formation of meta-substituted ethyl cinnamate ester: A solution of 1 equiv of meta-substituted cinnamic acid in dichloromethane (2 mL dichloromethane/mmol acid) was treated by sequential addition of 0.1 equiv of 4-dimethylaminopyridine, 1.05 equiv of 1-[3-(dimethylamino) propyl]-3-ethylcarbodiimide hydrochloride and 10 equiv of ethyl alcohol. The reaction was stirred at ambient temperature until all of the reactants were consumed as measured by thin layer chromatography (CH_2Cl_2 100%). When complete (approximately 18 h), the reaction mixture was concentrated under reduced pressure. Water was added to the residue and the aqueous layer was extracted with dichloromethane. The combined organic layers were washed successively with 1 N hydrochloric acid, saturated aqueous sodium bicarbonate, and saturated sodium chloride. The organic layer was dried over magnesium sulfate, filtered, and concentrated under reduced pressure. The residue was used directly for further reactions.

4.1.2.2. Step 2. General procedure for the reduction of meta-substituted ethyl

cinnamate ester: A solution of 1 equiv of meta-substituted ethyl cinnamate ester in dichloromethane (2 mL dichloromethane/mmol ester) was treated slowly by the dropwise addition of 1 M diisobutylaluminium hydride (2.1 equiv) solution in dichloromethane at -78°C . The resulting mixture was stirred 2 h at -78°C and then quenched with 10% aqueous NaOH (25 mL/10 mmol ester). The resulting mixture was slowly warmup to ambient temperature and the layers separated. The aqueous layer was extracted with dichloromethane. The combined organic layers were washed successively with water, 1 N aqueous hydrochloric acid, and saturated aqueous sodium chloride. The organic layer was dried over magnesium sulfate, filtered, and concentrated under reduced pressure. The residue was used directly for further reactions or purified by chromatography from an appropriate solvent system before use.

4.1.2.3. Step 3. General procedure for the oxidation of meta-substituted allyl alcohol:

A solution of 1 equiv of meta-substituted allyl alcohol in dichloromethane (2 mL dichloromethane/mmol alcohol) was treated with 1.5 equiv of pyridinium dichromate under inert atmosphere. The reaction was stirred at ambient temperature until all of the reactants were consumed as measured by thin layer chromatography (CH_2Cl_2 100%). The resulting mixture was diluted with diethyl ether/hexanes (50:50) and filtered through silica gel. The silica gel was rinsed with hexanes/ethyl acetate (60:40) and the combined filtrate evaporated. The residue obtained was dissolved in dichloromethane, washed with 0.1 N aqueous HCl, water and dried over magnesium sulfate. The residue was used directly for further reactions or purified by chromatography from an appropriate solvent system before use. The following compounds were prepared according to these procedures.

4.1.2.4. (E)-3-(3-Iodophenyl)prop-2-en-1-ol (2a): Compound **1a** (5.71 g, 18.90 mmol) was treated according to the General Procedure Step 2 to give 4.88 g (99%) of compound **2a** as a yellow oil; $^1\text{H NMR}$ (CDCl_3) δ 1.55 (t, $J = 5.9$ Hz, 1H); 4.29–4.33 (m, 2H); 6.32 (dt, $J = 15.8$ Hz, $J = 5.5$ Hz, 1H); 6.49 (d, $J = 15.9$ Hz, 1H); 7.02 (dd, $J = J' = 7.8$ Hz, 1H); 7.29–7.32 (m, 1H); 7.53–7.56 (m, 1H); 7.71–7.72 (m, 1H).

4.1.2.5. (E)-3-(3-Iodophenyl)acrylaldehyde (3a): Compound **2a** (4.88 g, 18.76 mmol) was treated according to the General Procedure Step 3 to give 4.26 g (88%) of compound **3a** as a yellow solid. $^1\text{H NMR}$ (CDCl_3) δ 6.67 (dd, $J = 16.0$ Hz, $J = 7.6$ Hz, 1H); 7.15 (dd, $J = J' = 7.8$ Hz, 1H); 7.35 (d, $J = 16.0$ Hz, 1H); 7.49–7.53 (m, 1H); 7.72–7.77 (m, 1H); 7.87–7.91 (m, 1H); 9.68 (d, $J = 7.5$ Hz, 1H). $^{13}\text{C NMR}$ (CDCl_3) δ 94.81, 127.44, 129.54, 130.69, 136.10, 137.22, 139.89, 150.59, 193.22.

4.1.2.6. (E)-Ethyl 3-(3-fluorophenyl)acrylate (1b): 3-Fluoro cinnamic acid (2 g, 12.04 mmol) was treated according to the General Procedure Step 1 to give 2.22 g (95%) of compound **1b** as colorless oil. $^1\text{H NMR}$ (CDCl_3) δ 1.33 (t, $J = 7.1$ Hz, 3H); 4.27 (q, $J = 7.1$ Hz, 2H); 6.42 (d, $J = 16.0$ Hz, 1H); 7.07 (ddd, $J = J' = 8.2$ Hz, $J'' = 0.95$ Hz, 1H); 7.14–7.24 (m, 1H); 7.28–7.31 (m, 1H); 7.32–7.38 (1H); 7.63 (d, $J = 16.0$ Hz, 1H).

4.1.2.7. (E)-Ethyl 3-(*m*-tolyl)acrylate (1c): 3-Methyl cinnamic acid (1.58 g, 9.77 mmol) was treated according to the General Procedure Step 1 to give 1.85 g (93%) of compound **1c** as a white solid. $^1\text{H NMR}$ (CDCl_3) δ 1.32 (t, $J = 7.1$ Hz, 3H); 2.35 (s, 3H); 4.24 (q, $J = 7.1$ Hz, 2H); 6.40 (d, $J = 16.0$ Hz, 1H); 7.15–7.19 (m, 1H); 7.22–7.28 (m, 1H); 7.28–7.37 (2H); 7.66 (d, $J = 16.0$ Hz, 1H).

4.1.2.8. (E)-3-(3-Fluorophenyl)prop-2-en-1-ol (2b): Compound **1b** (2.2 g, 11.4 mmol) was treated according to the General Procedure Step 2 to give after flash column

chromatography purification (hexanes/dichloromethane 50:50 to dichloromethane 100%) 1.4 g (82%) of compound **2b** as colorless oil. $^1\text{H NMR}$ (CDCl_3) δ 1.48 (t, $J = 5.9$ Hz, 1H); 4.30–4.34 (m, 2H); 6.35 (dt, $J = 15.8$ Hz, $J = 5.5$ Hz, 1H); 6.57 (d, $J = 15.9$ Hz, 1H); 6.91 (ddd, $J = J' = 8.0$ Hz, $J'' = 2.2$ Hz, 1H); 7.04–7.09 (m, 1H); 7.11–7.14 (m, 1H); 7.23–7.29 (m, 1H).

4.1.2.9. (E)-3-(m-Tolyl)prop-2-en-1-ol (2c): Compound **1c** (1.73 g, 9.09 mmol) was treated according to the General Procedure Step 2 to give 1.24 g (92%) of compound **2c** as colorless oil. $^1\text{H NMR}$ (CDCl_3) δ 1.39 (t, $J = 5.9$ Hz, 1H); 2.33 (s, 3H); 4.28–4.32 (m, 2H); 6.34 (dt, $J = 15.9$ Hz, $J = 5.7$ Hz, 1H); 6.56 (d, $J = 15.9$ Hz, 1H); 7.03–7.08 (m, 1H); 7.15–7.25 (m, 3H).

4.1.2.10. (E)-3-(3-Fluorophenyl)acrylaldehyde (3b): Compound **2b** (1.4 g, 9.29 mmol) was treated according to the General Procedure Step 3 to give 1.3 g (96%) of compound **3b** as a colorless oil. $^1\text{H NMR}$ (CDCl_3) δ 6.68 (dd, $J = 16.0$ Hz, $J = 7.5$ Hz, 1H); 7.12 (ddd, $J = J' = 8.2$ Hz, $J'' = 0.95$ Hz, 1H); 7.22–7.26 (m, 1H); 7.31–7.34 (m, 1H); 7.37–7.42 (m, 1H); 7.42 (d, $J = 16.0$ Hz, 1H); 9.69 (d, $J = 7.6$ Hz, 1H). $^{13}\text{C NMR}$ (CDCl_3) δ 114.50 (d, $^2J_{\text{C-F}} = 22.6$ Hz, 1C), 117.87 (d, $^2J_{\text{C-F}} = 21.3$ Hz, 1C), 124.19 (d, $^4J_{\text{C-F}} = 2.5$ Hz, 1C), 129.39, 130.48 (d, $^3J_{\text{C-F}} = 8.8$ Hz, 1C), 150.79 (d, $^4J_{\text{C-F}} = 2.5$ Hz, 1C), 162.78 (d, $^1J_{\text{C-F}} = 247.7$ Hz, 1C), 193.13.

4.1.2.11. (E)-3-(m-Tolyl)acrylaldehyde (3c): Compound **2c** (1.24 g, 8.36 mmol) was treated according to the General Procedure Step 3 to give after flash column chromatography purification (hexanes 100% to hexanes/ethyl acetate 90:10) 0.68 g (55%) of compound **3c** as a colorless oil. $^1\text{H NMR}$ (CDCl_3) δ 2.37 (s, 3H); 6.69 (dd, $J = 15.9$ Hz, $J = 7.7$ Hz, 1H); 7.22–7.40 (m, 3H); 7.43 (d, $J = 16.0$ Hz, 1H); 9.68 (d, $J = 7.7$ Hz, 1H). $^{13}\text{C NMR}$ (CDCl_3) δ 21.25, 125.69, 128.36, 128.93, 129.08, 132.10, 133.90, 138.78, 153.10, 193.81.

4.1.3. (E)-3-(3-(Trimethylstannyl)phenyl)acrylaldehyde (4)—Compound **3** (0.5 g, 1.93 mmol) dissolved in dioxane (20 mL) was treated with hexamethylditin (0.476 g, 1.45 mmol) and tetrakis(triphenylphosphine)palladium (0) (0.006 g, 0.0387 mmol). The reaction was refluxed for 150 min and shielded from light. The reaction mixture was then cooled down to ambient temperature and filtered through celite, the celite was washed with dichloromethane and the solvent concentrated. The resulting oil was filtered through silica gel (40 g) pre-equilibrated with a mixture of light ether/dichloromethane (90:10), rinsed with this mixture and dichloromethane and eluted with ethyl acetate/dichloromethane (30:70). The fractions collected were then evaporated to give 0.464 g (80%) of compound **4** as bright yellow oil. $^1\text{H NMR}$ (CDCl_3) δ 0.36 (s, with Sn satellites, d, $J = 54.3$ Hz, 9H); 6.73 (dd, $J = 15.9$ Hz, $J = 7.7$ Hz, 1H); 7.44–7.56 (m, 3H); 7.64 (br s, 1H); 9.69 (d, $J = 7.7$ Hz, 1H). $^{13}\text{C NMR}$ (CDCl_3) δ -9.54, 128.09, 128.48, 128.54, 133.41, 135.98, 138.78, 143.67, 153.24, 193.74.

4.1.4. General procedure for amide formation from a carboxylic acid—(R)-tert-Butyl 3-(((benzyloxy)carbonyl)amino)-4-oxo-4-(((R)-1-phenylethyl)amino)butanoate (5).¹⁸ A solution of 0.6g (1.75mmol) of *N*-benzyloxycarbonyl-D-aspartic acid β -*t*-butyl ester monohydrate in 10 mL dichloromethane was treated by sequential addition of 0.238mL (1.84 mmol) of (*R*)- α -methylbenzylamine, 0.237 g (1.84 mmol) of 1-hydroxy-7-benzotriazole, and 0.337 g (1.84 mmol) of 1-[3-(dimethylamino)propyl]-3-ethylcarbodiimide hydrochloride. After 18 h stirring at ambient temperature, the reaction mixture was washed sequentially with a saturated sodium bicarbonate solution and with saturated sodium chloride solution. The organic layer was evaporated to give 0.726 g (97%)

of compound **5** as an off-white solid. $^1\text{H NMR}$ (CDCl_3) δ 1.38 (s, 9H); 1.43 (d, $J = 6.9$ Hz, 3H); 2.54 (dd, $J = 17.2$ Hz, $J = 7.2$ Hz, 1H); 2.87 (dd, $J = 17.3$ Hz, $J = 4.0$ Hz, 1H); 4.46–4.50 (m, 1H); 4.99–5.15 (m, 3H); 5.92–5.96 (m, 1H); 6.78–6.82 (m, 1H); 7.21–7.33 (m, 10 H).

4.1.5. General procedure for hydrolysis of a *tert*-butyl ester—A solution of *tert*-butyl ester derivative in formic acid, typically 1 g in 10 mL, is stirred at ambient temperature until no more ester is detected by thin layer chromatography (dichloromethane 95%/methanol 5%), a typical reaction time being around 3 h. The formic acid is then evaporated under reduced pressure to yield the corresponding carboxylic acid. The following compound was prepared according to this procedure.

4.1.5.1. (*R*)-3-(((Benzyloxy)carbonyl)amino)-4-oxo-4-(((*R*)-1-phenylethyl)amino)butanoic acid (12**):** Compound **5** (0.726 g, 1.70 mmol) was hydrolyzed to give 0.614 g (97%) of compound **12** as a white solid; $^1\text{HNMR}$ ($\text{MeOH-}d_4$) δ 1.40 (br s, 3H); 2.56–2.81 (m, 2H); 5.0–5.15 (m, 3H); 7.11–7.41 (m, 10H).

4.1.6. Benzyl ((*R*)-4-([1,4'-bipiperidin]-1'-yl)-1,4-dioxo-1-(((*R*)-1-phenylethyl)amino)butan-2-yl)carbamate (13**)**—Compound **13** was prepared using the General procedure for amide formation from a carboxylic acid except that *N*-benzyloxycarbonyl-D-aspartic acid β -*t*-butyl ester was replaced with compound **12** (0.614 g, 1.65 mmol) and (*R*)- α -methylbenzylamine was replaced with 4-(1-piperidiny)piperidine. Compound **13** was obtained as a pink solid (0.718 g, 83%). $^1\text{H NMR}$ (CDCl_3) δ 1.22–1.34 (m, 1H); 1.34–1.47 (m, 7H); 1.47–1.62 (m, 4H); 1.72–1.90 (m, 2H); 2.33–2.60 (m, 6H); 2.80–3.01 (m, 1H); 3.06–3.20 (m, 1H); 3.66–3.83 (m, 1H); 4.46–4.67 (m, 2H); 4.95–5.17 (m, 3H); 6.22–6.38 (m, 1H); 7.11–7.40 (m, 10H).

4.1.7. General procedure for hydrogenolysis of a benzyloxycarbonyl amine

4.1.7.1. (*R*)-4-([1,4'-Bipiperidin]-1'-yl)-2-amino-4-oxo-*N*-((*R*)-1-phenylethyl)butanamide (14**):** A suspension of 0.718 g (1.37 mmol) of compound **13** and palladium (5 wt % on activated carbon, 0.35 g) in 20 mL methanol was held under an atmosphere of hydrogen for 18 h. The reaction was filtered to remove the palladium over carbon and the filtrate was evaporated to give 0.484 g (87%) of compound **14** as a colorless oil. $^1\text{H NMR}$ ($\text{MeOH-}d_4$) δ 1.23–1.39 (m, 1H); 1.39–1.52 (m, 7H); 1.52–1.76 (m, 4H); 1.80–1.92 (m, 2H); 2.42–2.61 (m, 7H); 2.61–2.69 (m, 1H); 2.69–2.80 (m, 1H); 2.87–3.03 (m, 1H); 2.64–3.74 (m, 1H); 3.88–4.00 (m, 1H); 4.49–4.59 (m, 1H); 4.92–5.10 (m, 1H); 7.16–7.36 (m, 5H). The following compound was obtained according to this procedure.

4.1.7.2. (*R*)-*tert*-Butyl 3-amino-4-oxo-4-(((*R*)-1-phenylethyl) amino)butanoate (6**):** Compound **5** (0.618 g, 1.45 m mol) was hydrogenolized to give 0.407 g (96%) of **6** as an off-white solid; $^1\text{H NMR}$ (CDCl_3) δ 1.40 (s, 9H); 1.47 (d, $J = 6.9$ Hz, 3H); 1.98 (br s, 2H); 2.49 (dd, $J = 7.9$ Hz, $J = 17.7$ Hz, 1H); 2.83 (dd, $J = 3.6$ Hz, $J = 16.7$ Hz, 1H); 3.69 (br s, 1H); 4.99–5.10 (m, 1H); 7.19–7.33 (m, 5H); 7.65–7.68 (m, 1H).

4.1.8. General procedure for formation of a 2-azetidinone from an imine and an acetyl chloride

4.1.8.1. Step 1: General procedure for formation of an imine from an amino acid derivative: A solution of 1 equiv of an α -amino acid ester or amide in dichloromethane was treated sequentially with 1 equiv of an appropriate aldehyde, and a desiccating agent, such as magnesium sulfate or sodium sulfate, in the amount of about 14 equiv (2 g) of desiccating agent per gram of starting α -amino acid ester or amide. The reaction was stirred at ambient temperature until all of the reactants were consumed as measured by thin layer

chromatography (CH₂Cl₂ 95%/MeOH 5%). When complete, the reaction mixture was then filtered, the filter cake was washed with dichloromethane, and the filtrate concentrated under reduced pressure to provide the desired imine that was used directly in the subsequent step.

4.1.8.2. Step 2: General procedure for the [2+2] cyclo-addition of an imine and an acetyl chloride: A dichloromethane solution of the imine (10mL dichloromethane/1 g imine) was cooled to 0 °C. To this cooled solution was added 1.5 equiv triethylamine, followed by the dropwise addition of a dichloromethane solution of 1.1 equiv of an acetyl chloride **8** (10mL of dichloromethane/1 g of **8**). The reaction mixture was allowed to warm to ambient temperature over 1 h and was then quenched by the addition of a saturated aqueous solution of ammonium chloride. The resulting mixture was partitioned between water and dichloromethane. The layers were separated and the organic layer was washed successively with 1 N hydrochloric acid, saturated aqueous sodium bicarbonate, and saturated aqueous sodium chloride. The organic layer was dried over magnesium sulfate, filtered and concentrated under reduced pressure. The residue was purified by chromatography (hexanes/ethyl acetate or dichloromethane/methanol) or by crystallization from dichloromethane/methanol. The following compounds were obtained according to this sequence of procedures.

4.1.8.3. (R)-4-([1,4'-Bipiperidin]-1'-yl)-4-oxo-2-((3S,4R)-2-oxo-3-((S)-2-oxo-4-phenyloxazolidin-3-yl)-4-((E)-3-(trimethylstannyl) styryl)azetidin-1-yl)-N-((R)-1-phenylethyl)butanamide (16): Imine **15** prepared from 0.445 g (1.15 mmol) of **14** and (E)-3-(3-(trimethylstannyl)phenyl)acrylaldehyde **4** was combined with 2-(4(S)-phenyloxazolidin-2-on-3-yl) acetyl chloride **8** to give 0.639 g (65%) of compound **16** as a yellow solid after flash column chromatography (CH₂Cl₂ 99% to 90% gradient/MeOH 1% to 10% gradient, NH₄OH <1%). ¹H NMR (CDCl₃) δ 0.29 (s, with Sn satellites, d, J = 54.2 Hz, 9H); 1.22–1.50 (m, 4H); 1.50–1.65 (m, 7H); 1.65–1.84 (m, 2H); 2.30–2.55 (m, 6H); 2.85–2.96 (m, 1H); 3.14–3.21 (m, 1H); 3.34–3.43 (m, 1H); 3.75–3.86 (m, 1H); 3.96–4.08 (m, 1H); 4.15 (t, J = 8.5 Hz); 4.18–4.26 (m, 1H); 4.41–4.55 (m, 1H); 4.61–4.67 (m, 1H); 4.67–4.77 (m, 2H); 5.0–5.13 (m, 1H); 6.33–6.41 (m, 1H); 6.76–6.84 (m, 1H); 7.15–7.49 (m, 15H); 8.21–8.29 (m, 1H). ¹³C NMR (CD₃CN) δ -9.63, 25.38/25.42, 26.96/27.01, 28.32/28.40, 28.69/28.90, 34.13/34.19, 42.09/42.19, 45.61/45.47, 50.25/50.26, 50.78/50.83, 55.60, 55.80, 61.27, 63.06, 63.09/63.14, 63.88/63.92, 71.90/71.92, 123.43/123.52, 126.76, 127.04, 127.63, 128.31/128.36, 129.34, 130.09, 130.21, 135.76, 136.56/136.58, 136.96, 138.52/138.55, 138.97/139.04, 143.98, 145.75, 159.35/159.38, 164.62/164.62, 168.60/168.68, 169.15/169.21. HRMS (FAB) calcd for C₄₅H₅₈N₅O₅Sn 868.3464, found 868.3466 (M+H)⁺.

4.1.8.4. (R)-tert-Butyl 3-((2R,3S)-2-((E)-3-iodostyryl)-4-oxo-3-((S)-2-oxo-4-phenyloxazolidin-3-yl)azetidin-1-yl)-4-oxo-4-(((R)-1-phenylethyl)amino)butanoate (9a): Imine **7a** prepared from 0.815 g (2.78 mmol) of **6** and of **3a** was combined with 2-(4(S)-phenyloxazolidin-2-on-3-yl) acetyl chloride **8** to give 1.462 g (71%) of compound **9a** as a white solid after flash column chromatography (hexanes 95% to 25% gradient/ethyl acetate 5% to 75% gradient). ¹H NMR (CDCl₃) δ 1.35 (s, 9H); 1.58 (d, J = 7.1 Hz, 3H); 3.09 (dd, J = 3.6 Hz, J = 17.8 Hz, 1H); 3.17 (dd, J = 10.7 Hz, J = 17.8 Hz, 1H); 3.92 (dd, J = 3.5 Hz, J = 10.7 Hz, 1H); 4.16 (dd, J = 7.6 Hz, J = 8.5 Hz, 1H); 4.25 (d, J = 5.0 Hz, 1H); 4.58 (dd, J = 5.0 Hz, J = 9.3 Hz, 1H); 4.67 (t, J = 8.7 Hz, 1H); 4.73 (t, J = 8.2 Hz, 1H); 5.04–5.13 (m, 1H); 6.29 (dd, J = 9.3 Hz, J = 15.8 Hz, 1H); 6.62 (d, J = 15.8 Hz, 1H); 7.02 (t, J = 7.8 Hz, 1H); 7.11–7.28 (m, 4H); 7.34–7.48 (m, 7H); 7.58–7.65 (m, 2H); 8.05–8.11 (m, 1H).

4.1.8.5. (R)-tert-Butyl 3-((2R,3S)-2-((E)-3-fluorostyryl)-4-oxo-3-((S)-2-oxo-4-phenyloxazolidin-3-yl)azetidin-1-yl)-4-oxo-4-(((R)-1-phenylethyl)amino)butanoate (9b):

Imine **7b** prepared from 0.330 g (1.13 mmol) of **6** and of **3b** was combined with 2-(4(*S*)-phenyloxazolidin-2-on-3-yl) acetyl chloride **8** to give 0.539 g (76%) of compound **9b** as a white solid after flash column chromatography (hexanes 100% to 70% gradient/ethyl acetate 0% to 30% gradient). ¹H NMR (CDCl₃) δ 1.35 (s, 9H); 1.59 (d, *J* = 7.0 Hz, 3H); 3.10 (dd, *J* = 3.5 Hz, *J* = 17.5 Hz, 1H); 3.19 (dd, *J* = 9.5 Hz, *J* = 18.0 Hz, 1H); 3.92 (dd, *J* = 3.2 Hz, *J* = 10.8 Hz, 1H); 4.15 (t, *J* = 8.2 Hz, 1H); 4.25 (d, *J* = 5.0 Hz, 1H); 4.58 (dd, *J* = 5.0 Hz, *J* = 9.5 Hz, 1H); 4.66 (t, *J* = 8.7 Hz, 1H); 4.73 (t, *J* = 8.2 Hz, 1H); 5.06–5.14 (m, 1H); 6.31 (dd, *J* = 9.2 Hz, *J* = 15.7 Hz, 1H); 6.68 (d, *J* = 16.0 Hz, 1H); 6.92–7.02 (m, 3H); 7.15–7.30 (m, 4H); 7.35–7.47 (m, 7H); 8.07–8.14 (m, 1H).

4.1.8.6. (R)-tert-Butyl 3-((2R,3S)-2-((E)-3-methylstyryl)-4-oxo-3-((S)-2-oxo-4-phenyloxazolidin-3-yl)azetidid-1-yl)-4-oxo-4-(((R)-1-phenylethyl)amino)butanoate (9c):

Imine **7c** prepared from 0.40 g (1.37 mmol) of **6** and of **3c** was combined with 2-(4(*S*)-phenyloxazolidin-2-on-3-yl) acetyl chloride **8** to give 0.469 g (55%) of compound **9c** as a white solid after flash column chromatography (hexanes 95% to 25% gradient/ethyl acetate 5% to 75% gradient) ¹H NMR (CDCl₃) δ 1.35 (s, 9H); 1.59 (d, *J* = 7.1 Hz, 3H); 2.34 (s, 3H); 3.10 (dd, *J* = 3.5 Hz, *J* = 17.7 Hz, 1H); 3.19 (dd, *J* = 10.9 Hz, *J* = 17.8 Hz, 1H); 3.94 (dd, *J* = 3.5 Hz, *J* = 10.8 Hz, 1H); 4.14 (t, *J* = 8.5 Hz, 1H); 4.24 (d, *J* = 5.0 Hz, 1H); 4.58 (dd, *J* = 5.0 Hz, *J* = 9.4 Hz, 1H); 4.65 (t, *J* = 8.8 Hz, 1H); 4.73 (t, *J* = 8.4 Hz, 1H); 5.03–5.12 (m, 1H); 6.32 (dd, *J* = 9.5 Hz, *J* = 15.8 Hz, 1H); 6.72 (d, *J* = 15.8 Hz, 1H); 7.0–7.23 (m, 7H); 7.33–7.48 (m, 7H); 8.10–8.16 (m, 1H).

4.1.9. The following compounds were prepared according to the procedure 'general procedure for hydrolysis of a tert-butyl ester'

4.1.9.1. (R)-3-((2R,3S)-2-((E)-3-Iodostyryl)-4-oxo-3-((S)-2-oxo-4-phenyloxazolidin-3-yl)azetidid-1-yl)-4-oxo-4-(((R)-1-phenylethyl)amino)butanoic acid (10a): Compound **9a** (1.46 g, 1.98 mmol) was hydrolyzed to give 1.35 g (quantitative yield) of **10a** as an off-white solid; ¹H NMR (CDCl₃) δ 1.57 (d, *J* = 7.1 Hz, 3H); 3.25 (d, *J* = 7.6 Hz, 2H); 3.92 (t, *J* = 7.1 Hz, 1H); 4.16 (dd, *J* = 8.4 Hz, *J* = 7.5 Hz, 1H); 4.27 (d, *J* = 5.0 Hz, 1H); 4.52 (dd, *J* = 9.2 Hz, *J* = 5.0 Hz, 1H); 4.66 (t, *J* = 8.7 Hz, 1H); 4.72 (t, *J* = 8.0 Hz, 1H); 5.03–5.11 (m, 1H); 6.25 (dd, *J* = 15.8 Hz, *J* = 9.3 Hz, 1H); 6.47 (d, *J* = 15.8 Hz, 1H); 7.02 (t, *J* = 7.8 Hz, 1H); 7.03–7.06 (m, 1H); 7.08–7.28 (m, 3H); 7.30–7.46 (m, 7H); 7.58–7.63 (m, 2H), 8.23–8.28 (m, 1H).

4.1.9.2. (R)-3-((2R,3S)-2-((E)-3-Fluorostyryl)-4-oxo-3-((S)-2-oxo-4-phenyloxazolidin-3-yl)azetidid-1-yl)-4-oxo-4-(((R)-1-phenylethyl)amino)butanoic acid (10b): Compound **9b** (0.535 g, 0.85 mmol) was hydrolyzed to give 0.487 g (quantitative yield) of compound **10b** as an off-white solid; ¹H NMR (CDCl₃) δ 1.59 (d, *J* = 7.0 Hz, 3H); 3.25 (d, *J* = 7.0 Hz, 2H); 3.92 (t, *J* = 7.2 Hz, 1H); 4.16 (t, *J* = 8.0 Hz, 1H); 4.27 (d, *J* = 5.0 Hz, 1H); 4.52 (dd, *J* = 5.0 Hz, *J* = 9.5 Hz, 1H); 4.66 (t, *J* = 8.7 Hz, 1H); 4.72 (t, *J* = 8.0 Hz, 1H); 5.05–5.14 (m, 1H); 6.28 (dd, *J* = 15.5 Hz, *J* = 9.2 Hz, 1H); 6.66 (d, *J* = 16.0 Hz, 1H); 6.92–7.05 (m, 3H); 7.19–7.50 (m, 11H); 8.25–8.31 (m, 1H). ¹³C NMR (CDCl₃) δ 21.66, 34.40, 49.83, 54.41, 61.02, 62.02, 63.44, 71.15, 113.29 (d, ²*J*_{C-F} = 21.3 Hz, 1C), 115.76 (d, ²*J*_{CF} = 21.3 Hz, 1C), 122.29, 122.83, 126.28, 127.19, 128.52, 129.76, 129.80, 130.33 (d, ³*J*_{C-F} = 7.5 Hz, 1C), 135.82, 137.34 (d, ³*J*_{CF} = 7.5 Hz, 1C), 138.10, 143.59, 158.04, 162.97 (d, ¹*J*_{C-F} = 246.5 Hz, 1C), 163.32, 167.28, 174.48.

4.1.9.3. (R)-3-((2R,3S)-2-((E)-3-Methylstyryl)-4-oxo-3-((S)-2-oxo-4-phenyloxazolidin-3-yl)azetidid-1-yl)-4-oxo-4-(((R)-1-phenylethyl)amino)butanoic acid (10c): Compound **9c** (0.126 g, 0.20 mmol) was hydrolyzed to give 0.116 g (quantitative yield) of compound **10c** as an off-white solid; ¹H NMR (CDCl₃) δ 1.59 (d, *J* = 7.1 Hz, 3H); 2.34 (s, 3H); 3.22 (dd, *J* = 17.8 Hz, *J* = 8.1 Hz, 1H); 3.30 (dd, *J* = 17.8 Hz, *J* = 4.60 Hz, 1H); 3.92 (dd, *J* = 9.6 Hz, *J* =

4.6 Hz, 1H); 4.15 (t, $J = 8.2$ Hz, 1H); 4.26 (d, $J = 5.0$ Hz, 1H); 4.52 (dd, $J = 9.4$ Hz, $J = 5.0$ Hz, 1H); 4.66 (t, $J = 8.7$ Hz, 1H); 4.73 (t, $J = 8.0$ Hz, 1H); 5.04–5.14 (m, 1H); 6.29 (dd, $J = 15.8$ Hz, $J = 9.6$ Hz, 1H); 6.69 (d, $J = 15.7$ Hz, 1H); 7.01–7.47 (m, 14H); 8.29–8.34 (m, 1H). ^{13}C NMR (CDCl_3) δ 21.36, 21.98, 34.45, 49.86, 54.29, 60.99, 62.00, 63.68, 71.07, 120.37, 126.19, 126.91, 127.19, 127.78, 128.44, 128.96, 129.40, 129.71, 135.10, 135.82, 138.37, 139.60, 143.69, 158.03, 163.43, 167.39, 174.64.

4.1.10. Compounds 11a, 11b, and 11c were prepared according to the 'general procedure for amide formation from a carboxylic acid', except that *N*-benzyloxycarbonyl-D-aspartic acid α -*t*-butyl ester monohydrate was replaced with 10a, 10b and 10c and (*R*)- α -methylbenzylamine replaced by 4-(1-piperidinyl) piperidine, all compounds exhibited an ^1H and ^{13}C NMR spectrum consistent with the assigned structure

4.1.10.1. (*R*)-4-([1,4'-Bipiperidin]-1'-yl)-2-((2*R*,3*S*)-2-((*E*)-3-iodostyryl)-4-oxo-3-((*S*)-2-oxo-4-phenyloxazolidin-3-yl)azetid in-1-yl)-4-oxo-*N*-((*R*)-1-phenylethyl)butanamide (11a): Compound **11a** was prepared using the 'General procedure for amide formation from a carboxylic acid', except that *N*-benzyloxycarbonyl-D-aspartic acid α -*t*-butyl ester monohydrate was replaced with **10a** (0.60 g, 0.88 mmol) and 3-(trifluoromethyl)benzylamine was replaced with 4-(1-piperidinyl)piperidine. Compound **11a** (0.72 g, 98%) was obtained as an off-white solid after flash silica gel column chromatography (CH_2Cl_2 99.5% to 88% gradient/MeOH 0.5% to 12% gradient, NH_4OH <1%). ^1H NMR (CDCl_3) δ 1.24–1.49 (m, 4H); 1.52–1.65 (m, 7H); 1.74–1.85 (m, 2H); 2.30–2.60 (m, 6H); 2.84–2.96 (m, 1H); 3.12–3.21 (m, 1H); 3.31–3.40 (m, 1H); 3.75–3.83 (m, 1H); 3.98–4.04 (m, 1H); 4.16 (t, $J = 7.8$ Hz, 1H); 4.20–4.26 (m, 1H); 4.40–4.52 (m, 1H); 4.63–4.77 (m, 3H); 5.05–5.14 (m, 1H); 6.26–6.34 (m, 1H); 6.61–6.69 (m, 1H); 6.99–7.06 (m, 1H); 7.09–7.14 (m, 1H); 7.17–28 (m, 4H); 7.32–7.49 (m, 1H); 7.57–7.65 (m, 2H); 8.16–8.25 (m, 1H). ^{13}C NMR (CD_3CN) δ 22.88, 25.39/25.42, 26.95/27.00, 28.33/28.38, 28.69/28.89, 34.21/34.27, 42.11/42.19, 45.49/45.62, 50.10/50.11, 50.77/50.82, 55.52/55.71, 61.19, 63.05/63.08, 63.14/63.18, 63.62/63.64, 71.93/71.95, 95.23, 125.24/125.31, 126.85, 126.93, 127.82, 128.35/128.41, 129.46, 130.11, 130.22, 131.70, 136.47, 136.81/136.86, 138.20, 138.62/138.64, 139.46, 145.67, 159.27, 164.64/164.66, 168.59/168.66, 169.10/169.58. HRMS (FAB) calcd for $\text{C}_{42}\text{H}_{49}\text{N}_5\text{O}_5$ 830.2773, found 830.2782 (M+H) $^+$.

4.1.10.2. (*R*)-4-([1,4'-Bipiperidin]-1'-yl)-2-((2*R*,3*S*)-2-((*E*)-3-fluorostyryl)-4-oxo-3-((*S*)-2-oxo-4-phenyloxazolidin-3-yl)azetid in-1-yl)-4-oxo-*N*-((*R*)-1-phenylethyl)butanamide (11b): Compound **11b** was prepared using the 'General procedure for amide formation from a carboxylic acid', except that *N*-benzyloxycarbonyl-D-aspartic acid α -*t*-butyl ester monohydrate was replaced with **10b** (0.030 g, 0.052 mmol) and 3-(trifluoromethyl)benzylamine was replaced with 4-(1-piperidinyl)piperidine. Compound **11b** (0.037 g, quantitative yield) was obtained as an off-white solid after flash silica gel column chromatography (CH_2Cl_2 99% to 90% gradient/MeOH 1% to 10% gradient, NH_4OH <1%). ^1H NMR (CDCl_3) δ 1.26–1.47 (m, 4H); 1.49–1.65 (m, 7H); 1.70–1.82 (m, 2H); 2.33–2.55 (m, 6H); 2.83–2.95 (m, 1H); 3.11–3.22 (m, 1H); 3.33–3.42 (m, 1H); 3.74–3.82 (m, 1H); 3.95–4.02 (m, 1H); 4.15 (t, $J = 8.1$ Hz); 4.19–4.25 (m, 1H); 4.40–4.50 (m, 1H); 4.62–4.75 (m, 3H); 5.05–5.14 (m, 1H); 6.29–6.38 (m, 1H); 6.69–6.75 (m, 1H); 6.90–7.01 (m, 3H); 7.15–7.29 (m, 4H); 7.31–7.47 (m, 7H); 8.20–8.28 (m, 1H). ^{13}C NMR (CD_3CN) δ 22.79, 25.59, 27.21/27.18, 28.47/28.50, 28.81/29.03, 34.18/34.25, 42.20/42.28, 45.57/45.70, 50.09, 50.78/50.84, 55.64/55.84, 61.23, 62.96/62.98, 63.12/63.16, 63.63/63.67, 71.94/71.96, 114.00 (d, $^2J_{\text{C-F}} = 21.3$ Hz, 1C), 115.99 (d, $^2J_{\text{C-F}} = 21.3$ Hz, 1C), 123.72 (d, $^4J_{\text{C-F}} = 2.5$ Hz, 1C), 125.19/125.27, 126.92, 127.83, 128.31/128.37, 129.38, 130.10, 130.22, 131.60 (d, $^3J_{\text{CF}} = 7.5$ Hz, 1C), 137.21/137.25, 138.56/138.60, 139.53 (d, $^3J_{\text{CF}} = 6.3$ Hz, 1C), 145.66, 159.30/159.33, 163.94 ($^1J_{\text{C-F}} = 243.9$ Hz, 1C), 164.62/164.65,

168.59/168.66, 169.06/169.12. HRMS (FAB) calcd for C₄₂H₄₉FN₅O₅ 722.3712, found 722.3703 (M+H)⁺.

4.1.10.3. (R)-4-([1,4'-Bipiperidin]-1'-yl)-2-((2R,3S)-2-((E)-3-methylstyryl)-4-oxo-3-((S)-2-oxo-4-phenyloxazolidin-3-yl)azetididin-1-yl)-4-oxo-N-((R)-1-

phenylethyl)butanamide (11c): Compound **11c** was prepared using the 'General procedure for amide formation from a carboxylic acid', except that *N*-benzyloxycarbonyl- D-aspartic acid α -*t*-butyl ester monohydrate was replaced with **10c** (0.03 g, 0.053 mmol) and 3-(trifluoromethyl)benzylamine was replaced with 4 (1-piperidinyl)piperidine. Compound **11c** (0.038 g, quantitative yield) was obtained as an off-white solid after flash silica gel column chromatography (CH₂Cl₂ 99% to 90% gradient/MeOH 1% to 10% gradient, NH₄OH <1%). ¹H NMR (CDCl₃) δ 1.29–1.53 (m, 4H); 1.56–1.76 (m, 7H); 1.83–1.94 (m, 2H); 2.33 (m, 3H); 2.41–2.69 (m, 6H); 2.85–2.98 (m, 1H); 3.11–3.22 (m, 1H); 3.32–3.42 (m, 1H); 3.78–3.89 (m, 1H); 3.97–4.06 (m, 1H); 4.13 (t, *J* = 8.3 Hz, 1H); 4.19–4.25 (m, 1H); 4.43–4.58 (m, 1H); 4.61–4.78 (m, 3H); 5.02–5.12 (m, 1H); 6.28–6.38 (m, 1H); 6.70–6.79 (m, 1H); 6.99–7.23 (m, 1H); 7.29–7.47 (m, 7H); 8.21–8.30 (m, 1H). ¹³C NMR (CD₃CN) δ 21.48, 22.92, 25.62, 27.23/27.25, 28.49/28.53, 28.83, 29.06, 34.18/34.25, 42.21/42.29, 45.59/45.29, 50.14, 50.80/50.86, 55.58/55.79, 61.25, 62.98/63.00, 63.09/63.14, 63.92/63.96, 71.91/71.94, 123.12/123.18, 124.67, 126.89, 127.75, 128.32/128.38, 128.49, 129.39, 129.71, 130.11, 130.19, 130.22, 137.04, 138.56/138.60, 138.82/138.87, 139.41, 145.70, 159.33, 164.65/164.68, 168.61/168.68, 169.12/169.18. HRMS (FAB) calcd for C₄₃H₅₂N₅O₅ 717.3963, found 718.3954 (M+H)⁺.

4.2. Radiochemistry and conversions to cold analogs by destannylation

4.2.1. Cold conversion to 11a via iododestannylation—A solution of **16** (1.27 μ mol) in ethanol (200 μ L) was treated with 1 equiv NaI (1.27 μ mol), 10 equiv of chloramine T trihydrate (12.7 μ mol) and 100 μ L aqueous HCl (1 M). The reaction was stirred for 5 min and quenched by the addition of sodium thiosulfate (100 μ L) and sodium bicarbonate (200 μ L). The mixture was loaded onto a conditioned Oasis HLB cartridge. The cartridge was washed three times with 250 μ L water/methanol (95%/5%), eluted with acetonitrile (2 \times 500 μ L) and concentrated under reduced pressure. The mixture was analyzed by HPLC (**method A**) and compared to the previously prepared iodinated standard **11a**. An estimated yield of 98% was determined by RP-HPLC. MS (ESI), calcd for C₄₂H₄₉IN₅O₅ 830.27, found 830.29 (M+H)⁺. **HPLC method A:** instrument, Agilent 1120; mobile phase A, methanol 10%/water 90%/TFA 0.05%; mobile phase B, methanol 90%/water 10%/TFA 0.05%; linear gradient method, 20/80% A/B to 0/100% A/B in 3 min, hold at 100% B for 4 min; injection volume, 5 μ L; flow rate: 1 ml/min; column, Eclipse XDB-C18 4.6 \times 150 mm 5 μ m; UV detection, 257 nm.

4.2.2. Cold conversion to 11b via electrophilic fluorodestannylation—

Compound **16** (5 mg, 5.77 μ mol) was treated with a solution of SelectFluor tetrafluoroborate and silver triflate in anhydrous acetone. The reaction was stirred 30 min at room temperature under inert atmosphere (argon). The resulting mixture was concentrated under reduced pressure and reconstituted in a mixture 1:1 methanol/water prior to loading onto a conditioned Oasis HLB cartridge. The cartridge was washed three times with 500 μ L water/methanol (95%/5%), eluted with acetonitrile (2 \times 500 μ L) and concentrated under reduced pressure. The mixture was analyzed by HPLC (**method A**) and compared to the previously prepared fluorinated standard **11b**. An estimated yield of 63% was determined by RP-HPLC. MS (ESI), calcd for C₄₂H₄₉FN₅O₅ 722.36, found 722.38 (M+H)⁺.

4.2.3. Hot conversion to [¹²³I]11a via iododestannylation—A solution of **16** (50 μ g, 0.057 μ mol) in ethanol (500 μ L) was added to 500 μ L [¹²³I]NaI (11.0 mCi) in aqueous

NaOH (0.01 N), 26 μg (0.11 μmol) chloramine T trihydrate in 25 μL H_2O and 20 μL aqueous HCl (1 M). The reaction was stirred for 5 min, then 500 μL mobile phase A from HPLC was added and the resulting solution purified by HPLC (**method B**). [^{123}I]11a was obtained with an estimated radiochemical yield of 88% and a specific activity of 9.7 mCi (94% radiochemical purity of 94%). **HPLC method B:** instrument, Agilent 1120; mobile phase A, 10 mM ammonium acetate in water; mobile phase B, 10 mM ammonium acetate in acetonitrile; gradient method, 100/0% A/B to 20/80% A/B in 30 min, hold for 10 min; injection volume, 10 μL ; flow rate: 1 ml/min; column, Eclipse Plus C18 4.6 \times 250 mm 5 μm ; UV detection, 257 nm.

4.3. Biological evaluation

4.3.1. V1a receptor binding assay—The hV1a expressing cell line, cell culture conditions, and the cell-based receptor binding assay procedures were performed according to the methods described by Thibonnier et al.,⁵⁰ with modifications as reported earlier.¹⁸

4.4. Permeability assay

4.4.1. General procedure for the PAMPA-BBB assay—Parallel Artificial Membrane Permeability Assays (PAMPA) to model the Blood–Brain Barrier (BBB) was run according to a published procedure.⁴⁸ Stock solutions of test compounds and standards (5 mg/mL) were prepared in DMSO. Stock solution in DMSO (5 mg/mL) was diluted 200-fold (10 μL in 1.99 mL) in universal buffer pH 7.4.⁵¹ Three hundred microliters of secondary stock solution (25 $\mu\text{g}/\text{mL}$) were transferred to wells in donor plate. The donor plate used was a PTFE 96-well plate (catalog No. MSSACCEPTOR, Millipore). A PVDF filter membrane plate (catalog No. MAIPN4550, Millipore) was coated with 4 μL porcine brain lipid (Avanti Polar) in dodecane (20 mg/mL). A PVDF filter membrane plate served as the acceptor plate and was immediately filled with 300 μL universal buffer pH 7.4. The plate was left undisturbed at room temperature for 18 h. Verapamil and theophylline were used as standard for high and low permeability, respectively. Effective permeability (P_e) was calculated according to the equation reported by Faller.⁵²

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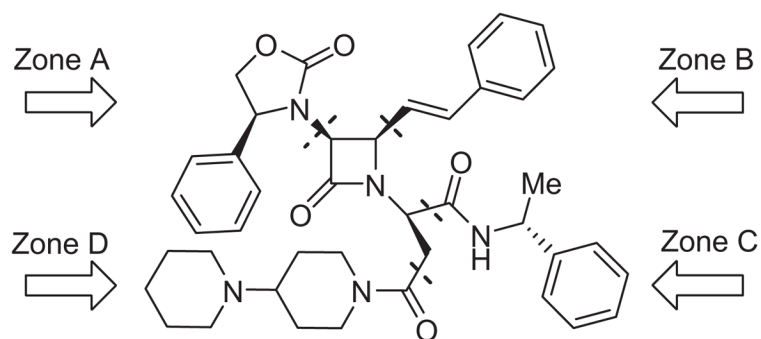
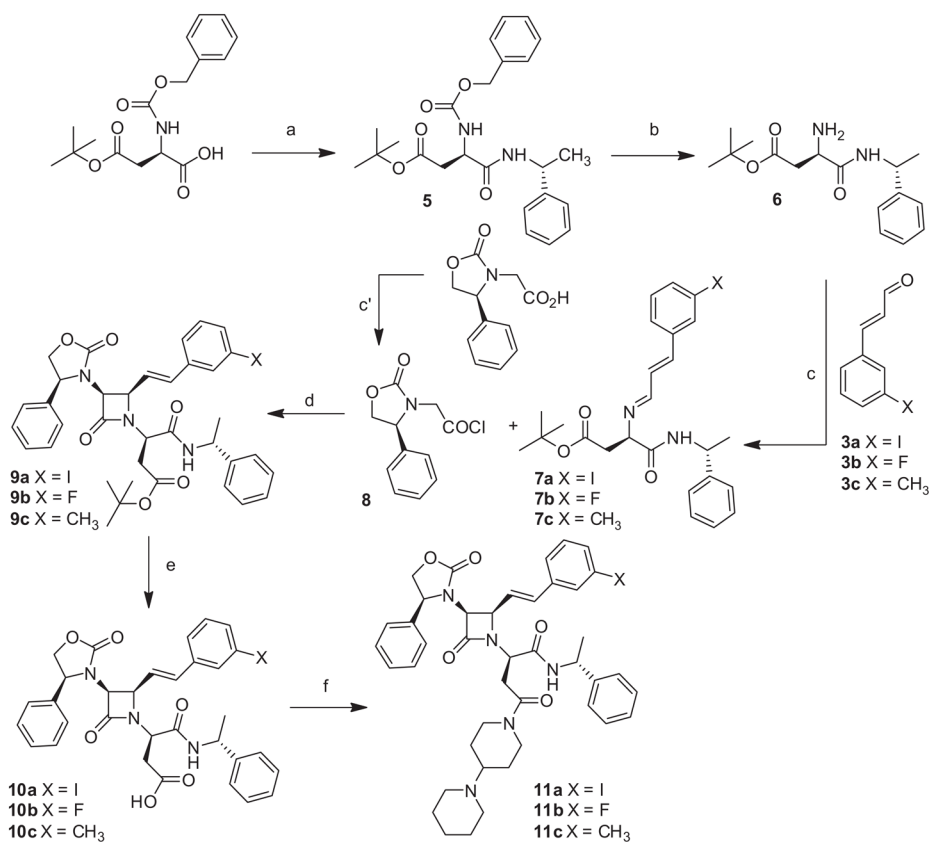
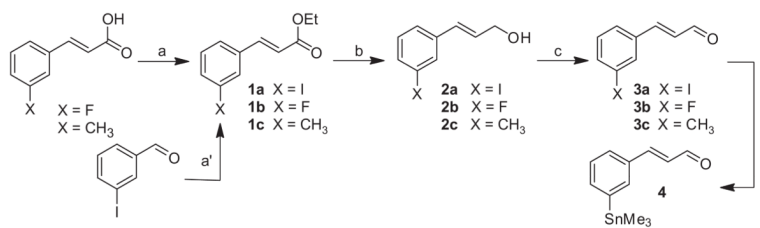


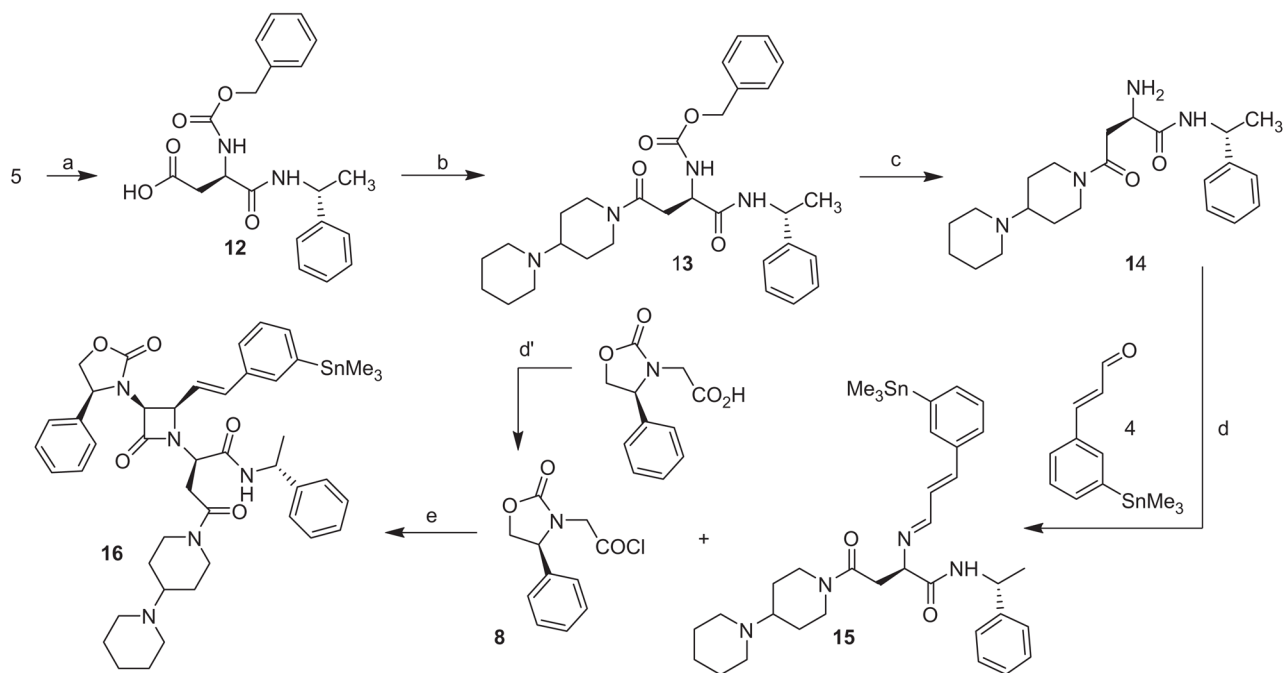
Figure 1.
Key structural features of SRX246.

**Scheme 1.**

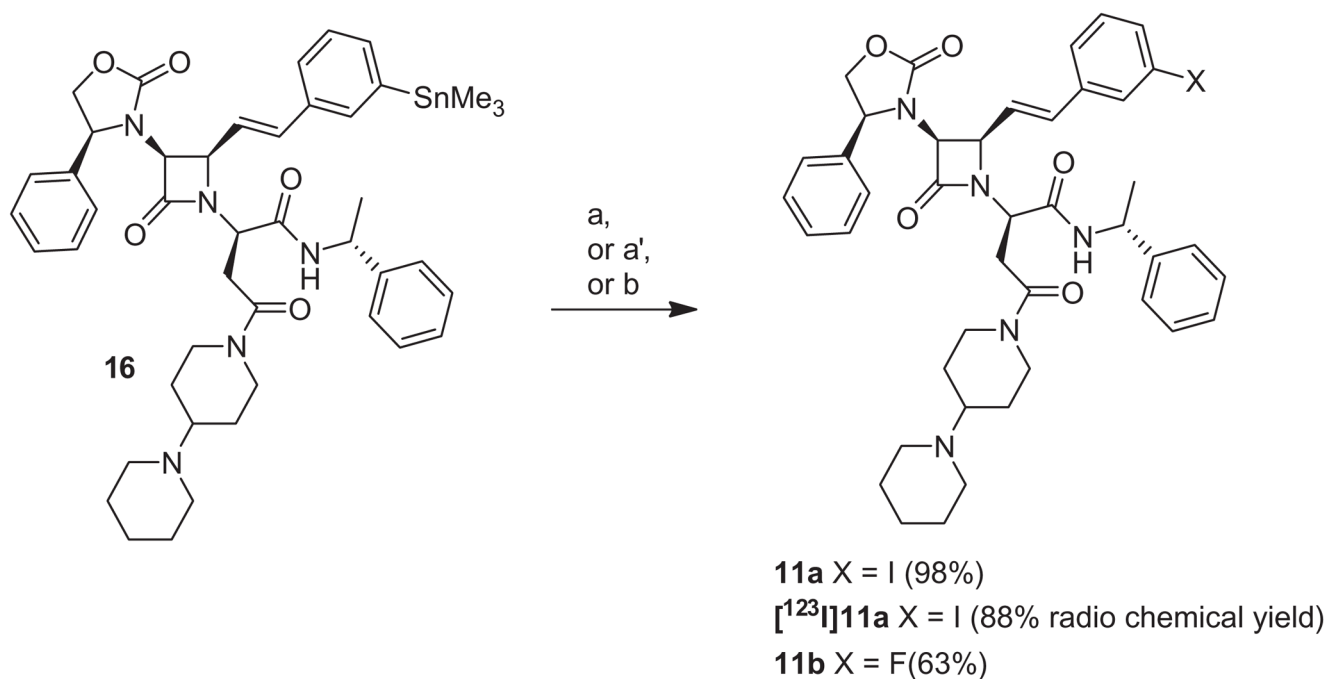
Syntheses of fluoro, iodo and methyl analogs of SRX246. Reagents and conditions: (a) (*R*)-methylbenzylamine, HOBt, EDC·HCl, CH₂Cl₂, rt, 18 h; (b) H₂, Pd/C 5%, MeOH, 18 h; (c) Na₂SO₄ anhydrous, CH₂Cl₂; (c') oxalyl chloride, DMF cat., CH₂Cl₂; (d) NEt₃, CH₂Cl₂, 0 °C to rt, 1 h; (e) HCO₂H; (f) 4-(1-piperidinyl) piperidine, HOBt, EDC, HCl, CH₂Cl₂, rt, 18 h.

**Scheme 2.**

Preparation of meta-substituted-*trans*-cinnamaldehydes. Reagents and conditions: (a) EDC·HCl, DMAP, EtOH, CH₂Cl₂, rt, 18 h; (a') triethyl phosphonoacetate, *n*BuLi in hexanes, THF, -78 °C, 10 min; 3-iodobenzaldehyde, THF, -78 °C, 40 min, to rt 45 min; (b) DIBAL, CH₂Cl₂, -78 °C, 2 h; 10% aqueous NaOH, -78 °C to rt, 15 min; (c) pyridinium dichromate, CH₂Cl₂; (d) hexamethylditin, Pd(PPh₃)₄, dioxane, 150 min.

**Scheme 3.**

Synthesis of radiolabeling precursor **16**. Reagents and conditions: (a) HCO_2H ; (b) 4-(1-piperidiny) piperidine, HOBT, EDC·HCl, CH_2Cl_2 , rt, 18 h; (c) H_2 , Pd/C 5%, MeOH, 18 h; (d) Na_2SO_4 anhydrous, CH_2Cl_2 ; (d') oxalyl chloride, DMF cat., CH_2Cl_2 ; (e) NEt_3 , CH_2Cl_2 , 0 °C to rt, 1 h.

**Scheme 4.**

Conversion of trimethyltin derivative **16** to **11a**, [¹²³I]**11a** and **11b**. Reagents and conditions:

(a) NaI, chloramine T, 1 M HCl, ethanol, rt, 5 min; (a') [¹²³I]NaI, chloramine T, 1 M HCl, ethanol, rt, 5 min; (b) SelectfluorTM (BF₄⁻), AgOTf, acetone, rt, 30 min.

Table 1

Vasopressin receptor affinity and PAMPA-BBB results for potential SRX246 based PET and SPECT ligands

Compound	hV1a K_i (nM)	hV1b K_i (nM)	hV2 K_i (nM)	cLogP	Pe (10^{-6} cm s $^{-1}$)	Predicted CNS penetration
11a	1.1	>1000	>1000	5.62	2.2	CNS±
11b	0.61	1154	>10,000	4.42	5.3	CNS+
11c	0.63	1300	>10,000	4.75	4.5	CNS+