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Synthesis of methylphosphonic acid by marine microbes: a source for methane in the aerobic ocean

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Abstract

Relative to the atmosphere, much of the aerobic ocean is supersaturated with methane; however, the source of this important greenhouse gas remains enigmatic. Catabolism of methylphosphonic acid by phosphorus-starved marine microbes, with concomitant release of methane, has been suggested to explain this phenomenon, yet methylphosphonate is not a known natural product, nor has it been detected in natural systems. Further, its synthesis from known natural products would require unknown biochemistry. Here we show that the marine archaeon *Nitrosopumilus maritimus* encodes a pathway for methylphosphonate biosynthesis and that it produces cell-associated methylphosphonate esters. The abundance of a key gene in this pathway in metagenomic datasets suggests that methylphosphonate biosynthesis is relatively common in marine microbes, providing a plausible explanation for the methane paradox.

Methane plays a key role in the global carbon cycle and is a potent greenhouse gas. As such, knowledge of its sources and sinks is essential to climate change models and to understand the flow of carbon within the biosphere. An unsolved problem in this area is the observation that vast sections of the aerobic ocean are supersaturated with this gas, despite the fact that strictly anaerobic archaea are the only significant biological source of methane known (1). The amount of methane produced in these aerobic environments is substantial, comprising as much as 4% of the global methane budget (2). It has been suggested that anaerobic microenvironments within the aerobic ecosystem could allow production of methane by known methanogens; however, this is contested on a variety of grounds (for a discussion see (1, 3)). Recently, Karl *et al.* suggested a new model in which methane would be produced when aerobic marine microorganisms use methylphosphonic acid (MPn) as a source of phosphorus (2). The model is based on several observations: (i) a well-studied bacterial enzyme, carbon-phosphorus (C-P) lyase, produces methane from MPn (4), (ii), C-P lyase genes are abundant in marine microbes (5, 6), (iii) phosphonates comprise a significant fraction of the available phosphorus pool in marine systems (7, 8), and (iv) incubation of

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seawater microcosms with MPn leads to methane production (2). While this model is conceptually appealing, it has a significant missing link: MPn has never been detected in marine ecosystems, nor is it a known natural product. Moreover, based on known phosphonate biosynthetic pathways (9), it is difficult to see how MPn could be made without invoking unusual biochemistry.

With one exception, all known phosphonate biosynthetic pathways begin with formation of the C-P bond by the enzyme phosphoenolpyruvate mutase (Ppm) (9). We have used the *ppm* gene as a molecular marker to identify the genes responsible for synthesis of phosphonic acid antibiotics in numerous microorganisms (10-13). During the course of this work, we identified a putative phosphonate biosynthetic gene cluster in *Nitrosopumilus maritimus*, a member of the ubiquitous Group I marine Thaumarchaeota whose members are among the most abundant organisms in marine surface waters (14, 15). Based on the experimentally validated functions of homologous enzymes (10, 16, 17), it is very likely that *N. maritimus* has the capacity to synthesize 2-hydroxyethylphosphonate (HEP), which is a common intermediate in phosphonate biosynthetic pathways (Fig. S1A, Table S1). Immediately adjacent to the putative HEP biosynthetic genes is an operon encoding a putative oxidoreductase, two putative sulfatases and a protein of the cupin superfamily that we designated MpnS.

MpnS has weak homology to hydroxypropylphosphonate epoxidase (HppE) and hydroxyethylphosphonate dioxygenase (HepD), two enzymes that catalyze Fe(II)- and oxygen-dependent transformations of similar phosphonate substrates (Figs S1B & S2). Thus, we suspected that MpnS might be a similar phosphonate biosynthetic enzyme. To test this, we cloned and overexpressed the *mpnS* gene in *Escherichia coli* (18). Cell extracts containing MpnS stoichiometrically convert ¹³C-labelled HEP to a product whose retention time and molecular mass are identical to MPn in liquid chromatography-mass spectrometry (LC-MS) experiments (Figs. 1 & S3). Using purified MpnS protein and HEP labeled with ¹³C at either the 1- or 2- position, we conclusively showed that the products of the MpnS reaction are MPn and HCO₃⁻ (Fig. 1B & 1C). The MpnS-catalyzed reaction requires both Fe(II) and molecular oxygen, but does not require an exogenous electron donor. Thus, like HepD, MpnS is an Fe(II)-dependent oxygenase that cleaves the unactivated carbon-carbon bond of HEP. However, the two enzymes catalyze distinct reactions. In the HepD reaction, the reducing equivalents needed for incorporation of oxygen into the cleavage products are derived equally from the C-1 and C-2 carbons of HEP, whereas MpnS catalyzes the asymmetric oxidation of HEP, with all four electrons being derived from the C-2 carbon, affording the more reduced phosphonate product MPn.

Having shown that MpnS catalyzes the synthesis of MPn *in vitro*, we asked whether *N. maritimus* synthesizes phosphonic acids using ³¹P NMR spectroscopy (Fig. 2A). The ¹H-decoupled ³¹P spectrum of the soluble cell extract displayed two peaks in the 10-30 ppm range characteristic of phosphonic acids (19). The relative abundance of the two peaks varied with sample preparation and could be seen in both the soluble and cell debris fractions after sonication (Fig S4). Based on spiking of the sample with an authentic standard neither peak can be attributed to free methylphosphonate; however, because the phosphorus compounds are cell associated we expected them to be covalently linked to a larger, more complex molecule, thus changing the chemical shift in the ³¹P NMR spectrum. Accordingly, we conducted a series of ³¹P-¹H Heteronuclear Multiple Bond Correlation (HMBC) experiments to identify the atoms linked to the P nuclei seen in the NMR spectra (Fig 2B). Because the behavior of phosphonate esters in such experiments is not well documented, we also synthesized and characterized a series of phosphonate esters to support our assignments (Figs S5-S7). Based on these experiments the ³¹P NMR peak at 28.7 ppm can be confidently assigned as an ester of methylphosphonate. Further support for this

conclusion was provided by high-resolution mass spectrometry, which revealed the presence of free methylphosphonate after strong acid hydrolysis of *N. maritimus* cell debris (Fig 2C and S8). Based on these results and the gene context of the MpnS locus (Table S1), we suspect that *N. maritimus* synthesizes an exopolysaccharide decorated with methylphosphonate, similar to the HEP- and aminoethylphosphonate-modified polymers found in many bacteria and lower eukaryotes (20).

The data presented above suggest that *N. maritimus* produces a cell-associated methylphosphonate ester via an MpnS-dependent biosynthetic pathway. To link this finding to the larger marine environment we screened the Global Oceanic Survey (GOS) metagenomic dataset (21) for the presence of MpnS homologs. We also searched for homologs of the related HepD and HppE proteins. Initially we screened the assembled GOS scaffolds, finding forty-six MpnS and twenty HepD homologs using a BLASTP cutoff value of 10^{-10} (Table S2). No HppE homologs were observed. Importantly, none of the HepD homologs were identified using *N. maritimus* MpnS as the query sequence; likewise none of the MpnS homologs were identified using HepD as a query. Thus, BLASTP clearly differentiates between the two homologous groups, supporting the assignment of the recovered sequences as MpnS and HepD proteins, respectively. To independently support these functional assignments we constructed maximum likelihood phylogenetic trees including biochemically validated MpnS, HepD and HppE proteins (Fig 3A & S9). We also used a hierarchical clustering method to examine all putative and validated MpnS, HepD and HppE proteins (Fig S10). In both cases robust support for the functional assignments was obtained. Thus, we conclude that the recovered GOS MpnS homologs are likely to be methylphosphonate synthases.

Additional support for the function of the MpnS homologs was revealed by analysis of neighboring genes found in GOS DNA scaffolds (Fig 3B, Table S3). Many of the nearby ORFs are homologous to those found in the *N. maritimus* gene cluster, including the phosphonate biosynthetic genes *ppm*, *ppd*, and *pdh*, as well as homologs of the sulfatases and nucleotidyl transferase genes, suggesting that the GOS scaffolds encode genes for the synthesis of similar MPn esters. Several other genes found on the scaffolds provide evidence for the identity of the organisms in which they are found. One of the scaffolds includes a 23S rRNA gene that can be confidently placed within the SAR11 clade between *Pelagibacter* species (Fig S11), while two of the *manC* genes are nearly identical to ones found in *Pelagibacter* sp. HTCC7211. Although the *mpnS* gene is absent in sequenced *Pelagibacter* genomes, these data strongly support the conclusion that some members of this genus have the capacity to synthesize MPn.

Relatives of *Nitrosopumilus* and *Pelagibacter* are among the most abundant organisms in the sea, with global populations estimated at 10^{28} for both ammonia-oxidizing Thaumarchaeota (14) and members of the SAR11 clade (22). Thus, the observation of *mpnS* in some members of these genera is consistent with the idea that MPn synthesis is prevalent in marine systems. To provide direct support for this notion, we measured the abundance of the *mpnS* gene relative to the abundance of typical single-copy genes as previously described (23). We also quantified the occurrence of the *ppm* gene to provide an estimate the relative occurrence of phosphonate synthesis in general (Table S4). Based on these data, we estimate that *ca.* 16% of marine microbes are capable of phosphonate biosynthesis, while 0.6% have the capacity to synthesize MPn. Because the GOS samples are confined to the upper few meters of the ocean, extrapolation of this analysis to the deeper ocean should be viewed with some skepticism. Nevertheless, the upper 200 m of the world's oceans are thought to contain *ca.* 3.6×10^{28} microbial cells with an average generation time of *ca.* two weeks (24). Thus, even with the relatively modest abundance of MPn biosynthesis suggested by our data, it seems quite possible that these cells could provide sufficient MPn precursor to account for

the observed methane production in the aerobic ocean via the C-P lyase dependent scenario suggested by Karl *et al* (2).

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

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References and Notes

1. Rogers, JE.; Whitman, WB., editors. Microbial production and consumption of greenhouse gases : methane, nitrogen oxides, and halomethanes. American Society for Microbiology; Washington, D.C.: 1991.
2. Karl DM, et al. Nat. Geosci. 2008; 1:473.
3. Reeburgh WS. Chem. Rev. 2007; 107:486. [PubMed: 17261072]
4. Daughton CG, Cook AM, Alexander M. FEMS Microbiol. Lett. 1979; 5:91.
5. Martinez A, Tyson GW, DeLong EF. Environ. Microbiol. 2010; 12:222. [PubMed: 19788654]
6. Ilikchyan IN, McKay RML, Zehr JP, Dyhrman ST, Bullerjahn GS. Environ. Microbiol. 2009; 11:1314. [PubMed: 19220397]
7. Clark LL, Ingall ED, Benner R. Am. J. Sci. 1999; 299:724.
8. Clark LL, Ingall ED, Benner R. Nature. 1998; 393:426.
9. Metcalf WW, van der Donk WA. Annu. Rev. Biochem. 2009; 78:65. [PubMed: 19489722]
10. Borisova SA, Circello BT, Zhang JK, van der Donk WA, Metcalf WW. Chem. Biol. Jan 29.2010 17:28. [PubMed: 20142038]
11. Blodgett JA, Zhang JK, Metcalf WW. Antimicrob. Agents and Ch. 2005; 49:230.
12. Eliot AC, et al. Chem Biol. 2008; 15:765. [PubMed: 18721747]
13. Circello BT, Eliot AC, Lee JH, van der Donk WA, Metcalf WW. Chem Biol. 2010; 17:402. [PubMed: 20416511]
14. Karner MB, DeLong EF, Karl DM. Nature. 2001; 409:507. [PubMed: 11206545]
15. Konneke M, et al. Nature. 2005; 437:543. [PubMed: 16177789]
16. Shao Z, et al. J. Biol. Chem. 2008; 283:23161. [PubMed: 18544530]
17. Seidel HM, Freeman S, Seto H, Knowles JR. Nature. 1988; 335:457. [PubMed: 3138545]
18. Materials and methods are available as supplementary material on Science Online.
19. Tebby, J.C., editor. CRC handbook of phosphorus-31 nuclear magnetic resonance data. CRC Press; Boca Raton: 1991.
20. Hilderbrand, R.L., editor. The Role of Phosphonates in Living Systems. CRC Press; Boca Raton: 1983.
21. Yooseph S, et al. PLoS. Biol. 2007; 5:432.
22. Morris RM, et al. Nature. 2002; 420:806. [PubMed: 12490947]
23. Howard EC, Sun S, Biers EJ, Moran MA. Environ. Microbiol. 2008; 10:2397. [PubMed: 18510552]
24. Whitman WB, Coleman DC, Wiebe WJ. Proc. Natl. Acad. Sci. USA. 1998; 95:6578. [PubMed: 9618454]
25. Dunwell JM, Culham A, Carter CE, Sosa-Aguirre CR, Goodenough PW. Trends Biochem. Sci. 2001; 26:740. [PubMed: 11738598]
26. Liu PH, et al. J. Am. Chem. Soc. 2001; 123:4619. [PubMed: 11457256]

27. Cicchillo RM, et al. *Nature*. 2009; 459:871. [PubMed: 19516340]
28. Higgins LJ, Yan F, Liu PH, Liu HW, Drennan CL. *Nature*. 2005; 437:838. [PubMed: 16015285]
29. Nair SK, van der Donk WA. *Arch. Biochem. Biophys.* 2011; 505:13. [PubMed: 20854789]
30. Kuemin M, van der Donk WA. *Chem. Commun.* 2010; 46:7694.
31. Witteck JT, et al. *Angew. Chem., Int. Ed. Engl.* 2007; 46:9089. [PubMed: 17990255]
32. Martens-Habbena W, Berube PM, Urakawa H, de la Torre JR, Stahl DA. *Nature*. 2009; 461:976. [PubMed: 19794413]
33. Papadopoulos JS, Agarwala R. *Bioinformatics*. 2007; 23:1073. [PubMed: 17332019]
34. Stamatakis A. *Bioinformatics*. 2006; 22:2688. [PubMed: 16928733]
35. Olson SA. *Brief Bioinform.* 2002; 3:87. [PubMed: 12002227]
36. de Hoon MJ, Imoto S, Nolan J, Miyano S. *Bioinformatics*. 2004; 20:1453. [PubMed: 14871861]
37. Saldanha AJ. *Bioinformatics*. 2004; 20:3246. [PubMed: 15180930]
38. Desper R, Gascuel O. *Mol. Biol. Evol.* 2004; 21:587. [PubMed: 14694080]
39. Markowitz VM, et al. *Nuc Acids Res.* 2010; 38:D382.

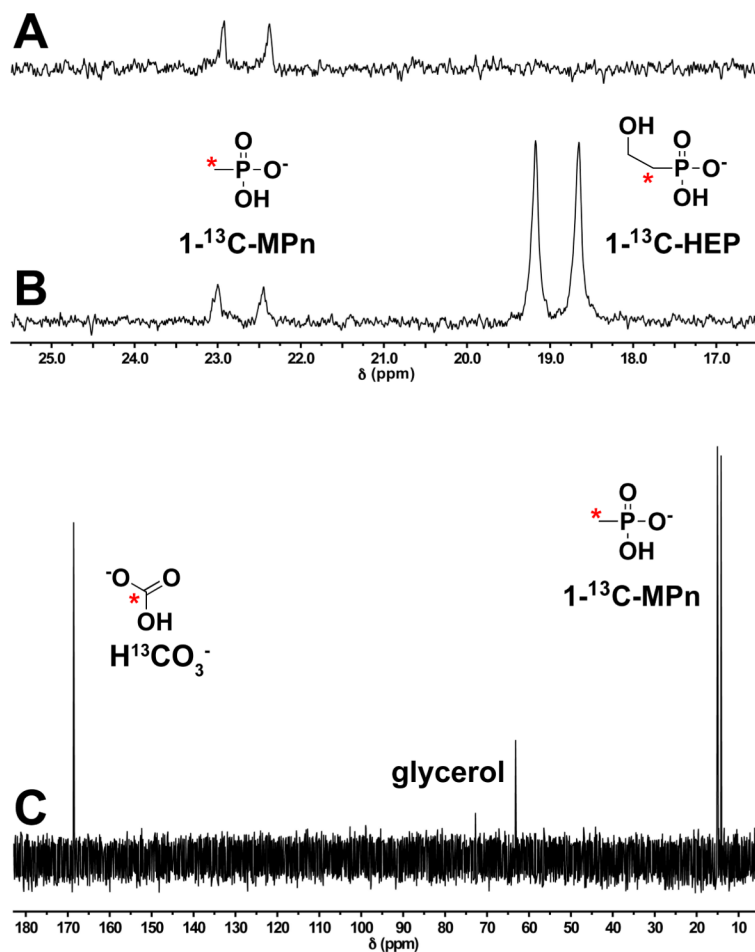


Fig. 1.

In vitro assay of MpnS activity. (A) Crude cell extract from an *E. coli* MpnS overexpression strain was incubated aerobically with $1\text{-}^{13}\text{C}$ -HEP in the presence of Fe(II) and the phosphorus-containing products were examined using ^{31}P NMR spectroscopy. After incubation for 1 hour a single product was observed as a doublet centered at 23.5 ppm. The mass and retention time of this product determined by LC-MS is consistent with this product being $1\text{-}^{13}\text{C-MPn}$ (Fig. S3). (B) Spiking of this reaction with the substrate, $1\text{-}^{13}\text{C}$ -HEP produced a second doublet centered at 19 ppm, showing that the substrate was completely consumed in the initial reaction. (C) The identity of the reaction products was determined using ^{13}C NMR after repeating the assay in a sealed vial using purified MpnS with a mixture of $1\text{-}^{13}\text{C}$ -HEP and $2\text{-}^{13}\text{C}$ -HEP as substrates. The C-2 labeled carbon of HEP is converted to ^{13}C bicarbonate ($\text{H}^{13}\text{CO}_3^-$), while the C-1-labelled carbon is converted to $1\text{-}^{13}\text{C-MPn}$. Bonding to phosphorus splits the ^{13}C peak in the NMR spectrum. Thus, the C-1 peak is split and the C-2 peak is not. Glycerol, a component of the assay mixture, is also observed in the ^{13}C spectrum. The ^{13}C label is indicated by a red asterisk.

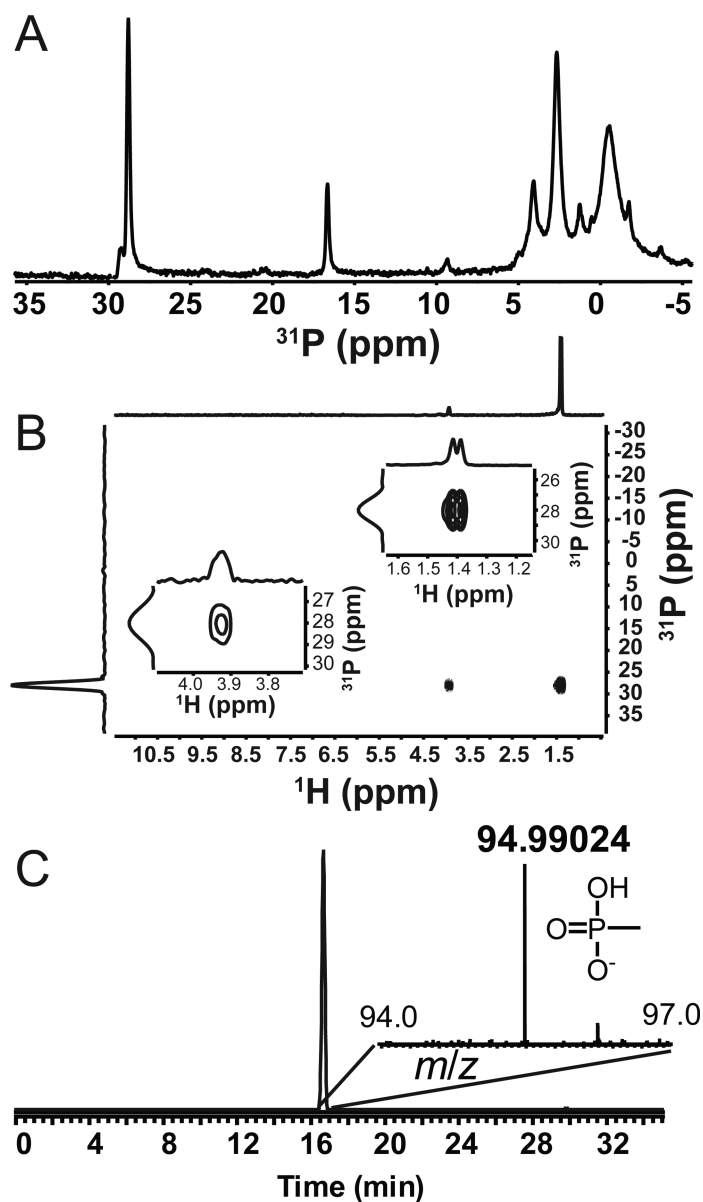


Fig. 2.

In vivo production of methylphosphonate esters by *N. maritimus*. (A) A cell extract of *N. maritimus* was prepared by sonication of whole cells as described. After removal of the cell debris by centrifugation, the supernatant was examined by ^{31}P NMR spectroscopy revealing at least two compounds with chemical shifts in the range typical of phosphonic acids. (B) The two-dimensional HMBC NMR spectrum of *N. maritimus* cell extract. Comparison of the proton splitting patterns (shown in the insets) to those of model compounds (Figs S6 & S7) clearly shows that the P compound at 28 ppm in the ^{31}P dimension is a methylphosphonate ester. The doublet of the proton at 1.4 ppm coupled to the phosphorus is diagnostic for a methyl group bonded directly to phosphorus, *i.e.* a methylphosphonate moiety. (C) High resolution LC-MS analysis showing the presence of free methylphosphonate after strong acid hydrolysis of *N. maritimus* cell debris. The extracted ion chromatogram centered around m/z 94.99035 (exact monoisotopic mass of methylphosphonate $[\text{M}-\text{H}]^-$) with Fourier-transform mass spectrum and ion structure is

shown in the inset. The chromatographic and MS fragmentation pattern is identical to an authentic MPn standard (Fig S8).

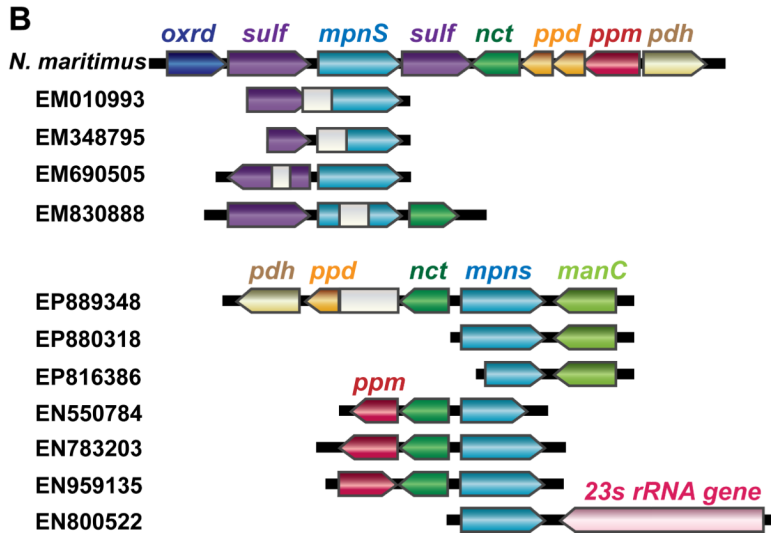
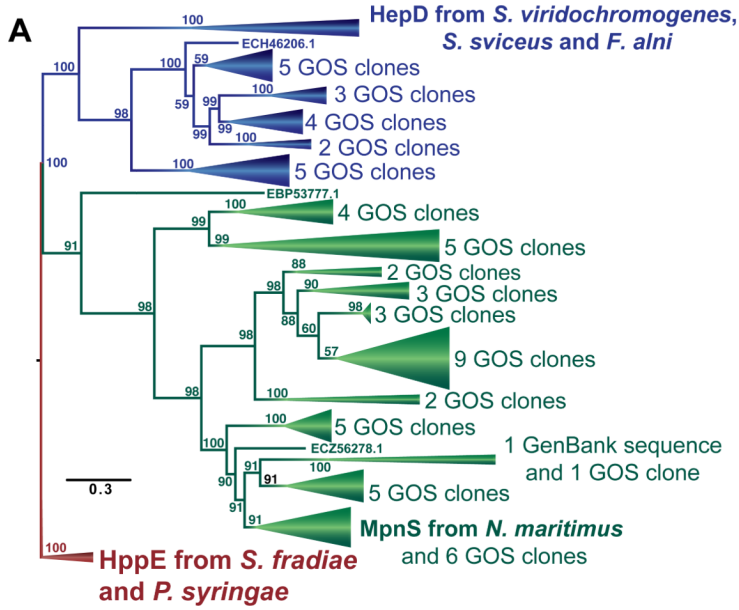


Fig. 3. (A) The evolutionary relationships of biochemically characterized MpnS, HepD and HppE proteins (shown in bold) and homologs recovered from Genbank and the GOS metagenomic dataset was inferred using maximum likelihood analysis as described. Bootstrap values from 100 replicates are shown at the nodes. Robust bootstrap support for the tree shows that the method clearly differentiates MpnS (green), HepD (blue) and HppE (red) proteins. The full tree with all individual homologs shown is presented in Fig S9. (B) The gene content of large scaffolds containing the GOS MpnS homologs is compared to the *mpnS* locus of *N. maritimus*. The grey boxes represent sequencing gaps between paired-end reads.