

Published in final edited form as:

Free Radic Biol Med. 2012 November 1; 53(9): 1807–1817. doi:10.1016/j.freeradbiomed.2012.08.015.

Native Rates of Superoxide Production from Multiple Sites in Isolated Mitochondria Measured using Endogenous Reporters

Casey L. Quinlan¹, Jason R. Treberg^{*}, Irina V. Pervoshchikova, Adam L. Orr, and Martin D. Brand

The Buck Institute for Research on Aging, Novato, California 94945

Abstract

Individual sites of superoxide production in the mitochondrial respiratory chain have previously been defined and partially characterized using specific inhibitors, but the native contribution of each site to total superoxide production in the absence of inhibitors is unknown. We estimated rates of superoxide production (measured as H₂O₂) at different sites in rat muscle mitochondria using specific endogenous reporters. The rate of superoxide production by the complex I flavin (site I_F) was calibrated to the reduction state of endogenous NAD(P)H. Similarly, the rate of superoxide production by the complex III site of quinol oxidation (site III_{Q_o}) was calibrated to the reduction state of endogenous cytochrome *b*₅₆₆. We then measured the endogenous reporters in mitochondria oxidizing NADH-generating substrates, without added respiratory inhibitors, with and without ATP synthesis. We used the calibrated reporters to calculate the rates of superoxide production from sites I_F and III_{Q_o}. The calculated rates of superoxide production accounted for much of the measured overall rates. During ATP synthesis, site I_F was the dominant superoxide producer. Under non-phosphorylating conditions, overall rates were higher and sites I_F, III_{Q_o} and unidentified sites (perhaps the complex I site of quinone reduction, site I_Q) all made substantial contributions to measured H₂O₂ production.

Keywords

mitochondria; ROS; superoxide; complex I; complex III; NADH autofluorescence; cytochrome *b*.

Production of mitochondrial reactive oxygen species (ROS)² has been implicated in many detrimental or degenerative biological processes. A truncated list includes metabolic diseases [1;2], neurodegenerative diseases [3], and cancers [4]. Mitochondrial ROS production has been asserted to play a fundamental role in the aging process, although this remains contentious [5–9]. There is increasing evidence that ROS are also crucial signalling molecules in many important physiological pathways [10]. The increasing number of physiological and pathological hypotheses that cite ROS production as crucial to their

© 2012 Elsevier Inc. All rights reserved.

¹To whom correspondence should be addressed: 8001 Redwood Blvd, Novato, CA 94945, USA. Tel: (415) 209-2000;

cquinlan@buckinstitute.org.

Current address: The Departments of Biological Sciences and Human Nutritional Sciences, University of Manitoba, Winnipeg, MB R3T 2N2, Canada

Publisher's Disclaimer: This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final citable form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

²The abbreviations used are: ROS, reactive oxygen species; CDNB, 1-chloro-2,4-dinitrobenzene; I_F, flavin site of complex I; I_Q, quinone-binding site of complex I; III_{Q_o} or Q_o, quinol oxidation site of complex III; III_{Q_i} or Q_i, quinone reduction site of complex III; SOD, superoxide dismutase; FCCP, carbonylcyanide 4-(trifluoromethoxy)phenylhydrazone.

mechanism brings the question of physiological ROS production into focus. What are the native rates of mitochondrial ROS production in isolated mitochondria (i.e. the rates in the absence of added electron transport chain inhibitors)? What are these native rates in cells, or *in vivo*? What controls these rates? Is there one site in the electron transport chain of mitochondria that is responsible for most of the ROS production in cells? Do the rates of production differ with substrate and tissue type?

Chance and colleagues [11–13] established that isolated mitochondria can produce H_2O_2 *in vitro*. It is now appreciated that most of this H_2O_2 is formed as superoxide, and superoxide dismutase-2 [14] in the matrix converts it to H_2O_2 , which can escape and be assayed in the surrounding medium. The field has subsequently expanded considerably and many characteristics of H_2O_2 production by mitochondria have been revealed. Using respiratory chain inhibitors to manipulate the reduction state of sites under investigation and to prevent superoxide generation from other sites, at least eight specific sites of superoxide and H_2O_2 production in the Krebs cycle and the electron transport chain of mammalian mitochondria have been identified or suggested, and partially characterized. These sites are α -ketoglutarate dehydrogenase; pyruvate dehydrogenase; glycerol 3-phosphate dehydrogenase; the electron transferring flavoprotein:Q oxidoreductase (ETFQOR) of fatty acid β -oxidation; the flavin in complex I (site I_F); the ubiquinone-reducing site in complex I (site I_Q); the ubiquinol-oxidizing site in complex III (site III_{Q_0}) [15–17]; and complex II, which can generate large amounts of superoxide and/or H_2O_2 under appropriate conditions [18].

Despite the importance ascribed to mitochondria as a source of ROS, it is still unknown which enzymes or respiratory complexes are the main sources of superoxide or H_2O_2 production by mitochondria *in vivo*, *in situ* in cells, or *in vitro* [16;17]. This is because inhibiting or genetically modifying a candidate site interrupts normal electron flow and alters the redox states of remaining sites, and can dramatically alter their rates of superoxide or H_2O_2 production. This raises the question: how can the individual contributions from a complex suite of ROS-producing sites be assessed within intact mitochondria?

In the present paper we introduce and exploit a novel method of estimating the rates of superoxide generation from two specific sites by calibrating the reduction state of endogenous redox pools to the rate of superoxide generated from that site (the redox state of endogenous NAD(P)H to report site I_F , and the redox state of cytochrome b_{566} to report site III_{Q_0}), which we call endogenous reporters here. We describe the calibration of the reporters, then present the first quantitative estimate of the contributions of site I_F and site III_{Q_0} to total H_2O_2 production by isolated mitochondria under native conditions in the absence of respiratory chain inhibitors.

MATERIALS AND METHODS

Animals, reagents and mitochondrial preparation

Female Wistar rats (Harlan Laboratories), age 5–8 weeks, were fed chow *ad libitum* with free access to water. Skeletal muscle mitochondria were isolated at 4 °C in Chappell-Perry buffer (CP1; 100 mM KCl, 50 mM Tris, 2 mM EGTA, pH 7.1 at 25 °C) by standard procedures [19]. They had robust respiratory control ratios (3–4.5) for six hours after isolation, and we generally completed all assays in under four hours. The animal protocol was approved by the Buck Institute Animal Care and Use Committee, in accordance with IACUC standards. All reagents were from Sigma (St. Louis, MO) except Amplex UltraRed, which was from Invitrogen (Carlsbad, CA).

Superoxide Production

Rates of superoxide production were measured indirectly as rates of H_2O_2 production following dismutation of two superoxide molecules by endogenous or exogenous superoxide dismutase (SOD) to yield one H_2O_2 . H_2O_2 was detected using exogenous horseradish peroxidase and Amplex UltraRed [20]. Peroxidase activity associated with exogenous SOD was negligible compared to the added horseradish peroxidase activity, since SOD addition did not decrease the observed rate of H_2O_2 production when site I_F was driven by malate in the presence of rotenone. Where there is clear evidence that the initial species formed at a site is superoxide, as there is for sites I_F [21] and III_{Qo} [22], we refer to it throughout the present paper as ‘superoxide production’. Where it is not known or ambiguous whether a site generates superoxide or H_2O_2 , as it is for site I_Q , and α -ketoglutarate dehydrogenase and pyruvate dehydrogenase [23–25], and also where the sites involved are not fully defined, we refer to it as ‘superoxide/ H_2O_2 production’. Where we discuss purely the experimental measurement we refer to it as ‘ H_2O_2 production’. Where we discuss effects of the various species produced by mitochondria, including superoxide, H_2O_2 , and their downstream products such as hydroxyl radical and lipid peroxides, we refer to them collectively as ‘ROS’.

Mitochondria ($0.3 \text{ mg protein} \cdot \text{ml}^{-1}$) were suspended in medium at 37°C containing 120 mM KCl , 5 mM Hepes , $5 \text{ mM K}_2\text{PO}_4$, 2.5 mM MgCl_2 , 1 mM EGTA , and 0.3% (w/v) bovine serum albumin ($\text{pH } 7.0$ at 37°C), together with $5 \text{ U} \cdot \text{ml}^{-1}$ horseradish peroxidase, $25 \text{ U} \cdot \text{ml}^{-1}$ SOD and $50 \text{ } \mu\text{M}$ Amplex UltraRed. Where indicated, phosphorylating (“state 3”) conditions of ATP synthesis were established by adding an ADP regenerating system containing 0.1 mM ADP , $2.5 \text{ units} \cdot \text{ml}^{-1}$ hexokinase, and 20 mM glucose . Non-phosphorylating (“state 4”) conditions were established by adding $0.1 \text{ } \mu\text{g} \cdot \text{ml}^{-1}$ oligomycin. Reactions were monitored fluorometrically in a Shimadzu RF5301-PC or Varian Cary Eclipse spectrofluorometer ($\lambda_{\text{excitation}} = 560 \text{ nm}$, $\lambda_{\text{emission}} = 590 \text{ nm}$) with constant stirring, and calibrated with known amounts of H_2O_2 [19]. After addition of substrate at 5 minutes, linear rates of change were monitored between minutes 6.5 and 8; the background rate of change of Amplex UltraRed fluorescence was corrected for by subtracting the rate preceding substrate addition between minutes 4 and 5 (as illustrated in Fig. 4a).

NAD(P)H measurements

Experiments were performed using $0.3 \text{ mg mitochondrial protein} \cdot \text{ml}^{-1}$ at 37°C in parallel with measurements of H_2O_2 and cytochrome b_{566} in the same medium with the same additions. The reduction state of endogenous NAD(P)H was determined by autofluorescence [26] using a Shimadzu RF5301-PC or Varian Cary Eclipse spectrofluorometer at $\lambda_{\text{excitation}} = 365 \text{ nm}$, $\lambda_{\text{emission}} = 450 \text{ nm}$. NAD(P)H was assumed to be 0% reduced after 5 min without added substrate (the small apparent further oxidation of NAD(P)H in the absence of added substrate was ignored; this caused a small overestimate of NAD(P)H reduction state, but did not affect the reported rates of H_2O_2 production by site I_F since the overestimate affected both assay and calibration equally) and 100% reduced with 5 mM malate and $4 \text{ } \mu\text{M}$ rotenone (Fig. 4b). Intermediate values were determined as %NAD(P)H relative to the 0% and 100% values.

Cytochrome b_{566} measurements

Experiments were performed at $1.5 \text{ mg mitochondrial protein} \cdot \text{ml}^{-1}$ in parallel with measurements of H_2O_2 and NAD(P)H in the same medium. The reduction state of endogenous cytochrome b_{566} was measured with constant stirring at 37°C in an Olis DW-2 dual wavelength spectrophotometer at $566\text{--}575 \text{ nm}$. The signal at this wavelength pair reports $\sim 75\%$ cytochrome b_{566} and $\sim 25\%$ cytochrome b_{562} [20;27]. We did not correct for the b_{562} contribution because the combined signal led to the same interpretations as the

corrected signal, and less protein was required for each assay. Cytochrome b_{566} was assumed to be 0% reduced after 5 min without added substrate and 100% reduced with saturating substrates plus antimycin A (Fig. 4c). Intermediate values were determined as % b_{566} reduced relative to the 0% and 100% values. At least 15 consecutive data points were used to calculate the average % reduction in each condition. Reduction of cytochrome b_{562} was measured at 561–569 nm [27].

Protonmotive force and respiration measurements

Protonmotive force was assayed as mitochondrial membrane potential (in the presence of nigericin to abolish pH gradients), using an electrode sensitive to the membrane-permeant cationic probe, methyltriphenylphosphonium [19]. This gave minimum values for the normal steady state, because nigericin slightly inhibited substrate oxidation. Mitochondrial respiration was measured in parallel in a Clark-type oxygen electrode (in the absence of nigericin). Mitochondria were incubated under identical conditions to the H_2O_2 assays at 37 °C. The four experimental conditions in Fig. 8 generated the following protonmotive force values and respiration rates (mean \pm SEM of three biological replicates). (a) 5 mM malate, phosphorylating: 140 ± 5 mV, 89 ± 40 nmol O $\text{min}^{-1} \cdot \text{mg protein}^{-1}$; (b) 5 mM malate, non-phosphorylating: 165 ± 1 mV, 28 ± 3 nmol O $\text{min}^{-1} \cdot \text{mg protein}^{-1}$; (c) 5 mM glutamate + 5 mM malate, phosphorylating: 158 ± 3 mV, 222 ± 29 nmol O $\text{min}^{-1} \cdot \text{mg protein}^{-1}$; (d) 5 mM glutamate + 5 mM malate, non-phosphorylating: 183 ± 2 mV, 48 ± 10 nmol O $\text{min}^{-1} \cdot \text{mg protein}^{-1}$.

CDNB treatment and calibration

To allow correction for losses of H_2O_2 caused by peroxidase activity in the matrix, mitochondria were treated where stated with 1-chloro-2,4-dinitrobenzene (CDNB) to deplete glutathione and decrease glutathione peroxidase and peroxiredoxin activity [26]. In a previous publication [26], we performed the appropriate controls in skeletal muscle mitochondria to show that this protocol does not damage respiratory chain components and introduce artifactual H_2O_2 production, nor does it have measurable effects on the redox state of the NAD(P)H pool. Mitochondria ($5 \text{ mg protein} \cdot \text{ml}^{-1}$) were treated with $35 \mu\text{M}$ CDNB or ethanol control in CP1 medium for 5 min at room temperature, mixed with an equal volume of ice-cold CP1 and centrifuged for 5 min at $15\,000 g$ (at 2–4 °C). The pellet was washed twice in ice-cold CP1 and resuspended to approximately $30 \text{ mg protein} \cdot \text{ml}^{-1}$.

The CDNB correction curve (Fig. 1) was determined as described [26], but with improved accuracy at low H_2O_2 production rates. The previous correction curve [26] utilized rates from 0.25 – $1.5 \text{ nmol } H_2O_2 \cdot \text{min}^{-1} \cdot \text{mg protein}^{-1}$ in a medium lacking phosphate and magnesium. We generated a new CDNB-correction curve in the current medium, at rates below $0.1 \text{ nmol } H_2O_2 \cdot \text{min}^{-1} \cdot \text{mg protein}^{-1}$, by titrating malate into CDNB and ethanol-control treated mitochondria. The rates of H_2O_2 production were measured after the addition of $4 \mu\text{M}$ rotenone and assumed to be purely matrix-directed. The empirically-derived hyperbolic equation that described this relationship was:

$$v_{CDNB} = v_{control} + (100 * v_{control}) / (72.6 + v_{control}) \quad (\text{Eq. 1})$$

(rates in $\text{pmol } H_2O_2 \cdot \text{min}^{-1} \cdot \text{mg protein}^{-1}$). We emphasize that this relationship should be empirically determined for any new experimental condition and within the relevant data range.

Curve fitting

The control data for the I_F calibration curve (Fig. 5c) were fitted by non-linear regression to a single exponential, to give the parameter values in Eq. 2:

$$v_{H_2O_2 (\%NAD(P)H)} = 0.29 * e^{(0.054 * \%NAD(P)H)} + 11.2 \quad (\text{Eq. 2})$$

where $v_{H_2O_2}$ is the rate of H₂O₂ production.

The control data for the III_{Q₀} calibration curve (Fig. 7c) were fitted in the same way to give the parameter values in Eq. 3.

$$v_{H_2O_2 (\%b_{566})} = 0.68 * e^{(0.096 * \%b_{566})} - 0.68 \quad (\text{Eq. 3})$$

Data for CDNB-treated mitochondria were fitted in the same way (parameter values not shown).

Statistics

When using the calibration curves in Figs 5c and 7c to calculate rates of H₂O₂ production, the error in the measurements during calibration was taken into account. This error was calculated by error propagation using Eq. 4.

$$SE_{H_2O_2 (\%reporter)}^{Fit} = \sqrt{SE_{\%reporter}^2 * (dv_{H_2O_2 (\%reporter)} / d(\%reporter))^2 + SE_{H_2O_2}^2} \quad (\text{Eq. 4})$$

For the data in Fig. 8, this error was combined with the error in the level of the measured reporter (Table 1) using Eq. 5:

$$SE_{H_2O_2 (\%reporter)}^{reported} = \sqrt{SE_{\%reporter}^2 * (dv_{H_2O_2 (\%reporter)} / d(\%reporter))^2 + SE_{H_2O_2 (\%reporter)}^{Fit}^2} \quad (\text{Eq. 5})$$

The sums of the reported rates of H₂O₂ production were calculated with Eq. 6:

$$v_{H_2O_2 (total)}^{reported} \pm SE_{H_2O_2 (total)}^{reported} = (v_{H_2O_2 (\%NAD(P)H)}^{reported} + v_{H_2O_2 (\%b_{566})}^{reported}) \pm \sqrt{(SE_{H_2O_2 (\%NAD(P)H)}^{reported})^2 + (SE_{H_2O_2 (\%b_{566})}^{reported})^2} \quad (\text{Eq. 6})$$

The significance of differences between reported and experimentally measured rates of H₂O₂ production in each experimental condition was tested using Welch's t-test. Because error propagation was used to include uncertainty in the calibration curve, we could not enter individual data-points for statistical analysis. Instead we used the traits describing the population of data (mean, SEM based on error propagation and number of observations) to calculate if differences were significant ($p < 0.05$).

RESULTS AND DISCUSSION

Endogenous reporters of the rates of mitochondrial superoxide and H₂O₂ production

The rate of superoxide/H₂O₂ production at each individual site in the electron transport chain will be a function of the concentrations of the electron donor (i.e. the reduction state of the specific centre that donates its electron to oxygen), and electron acceptor (oxygen), and the appropriate rate constants for the reaction.

Despite earlier assumptions to the contrary, e.g. [29], all sites of superoxide/H₂O₂ production have similar hyperbolic responses to increasing oxygen concentrations and exhibit saturation kinetics, with little change in production rate above about 10 μM O₂ at standard pressure [30]. In our air-saturated, open-cuvette experiments, the O₂ concentration was about 200 μM, so the rate of production at any site will be a function of the concentration of the reduced electron donor and not rate limited by [O₂]. For our analysis, the function relating the redox state of the donor to the rate of superoxide/H₂O₂ production has to be unique, but its other properties do not matter.

The redox states of the particular electron donors at each site may not be readily measurable. However, any electron carrier that is close to redox equilibrium with the donor will be a reporter of the donor's redox state. Such a reporter can be used to construct an empirical calibration curve that describes how the rate of superoxide/H₂O₂ production from a site is related to the reduction state of the reporter. For our analysis, the relationship between the redox states of the reporter and the donor needs to be unique, but its other properties do not matter. However, if the reporter is close to equilibrium with the donor, their redox states will be related by the Nernst equation and can be modelled using exponentials. Even if there are complications (such as significant disequilibrium between donor and reporter or more complex kinetic interactions within a redox site), as long as the reporter has a unique relationship to the superoxide/H₂O₂ production rate in the range of conditions to be investigated, it can still be used.

Therefore, if there are unique relationships between the redox states of the reporter and the donor, and between the redox state of the donor and the rate of superoxide/H₂O₂ production, an empirical calibration of H₂O₂ production as a function of the redox state of the reporter can be used to predict the rate of superoxide/H₂O₂ production from any site with a suitable reporter. This is the basis of the method we introduce in the present paper to measure the native rates of superoxide production from different sites in the electron transport chain of isolated mitochondria in the absence of added inhibitors of electron transport. Fig. 2 depicts the principles and assumptions employed in using endogenous reporters of superoxide production rate in this way.

Candidate sites of native superoxide and H₂O₂ production

In this paper we use the term “native rate” to mean the rate in the absence of added respiratory chain inhibitors. The obvious candidates for sites of native mitochondrial superoxide and H₂O₂ production during oxidation of NAD-linked substrates are respiratory chain complexes I and III [12;16;29;31;32]. These two complexes contain three potentially important sites: site I_F [21;31], site I_Q [33;34]; and site III_{Q_o} [12;20]. The black bars in Fig. 3 show the maximum rate of H₂O₂ production observed from each of these sites under optimal conditions of electron supply, in the presence of appropriate inhibitors to prevent escape of electrons, with suppression of matrix peroxidase-catalyzed losses of hydrogen peroxide.

The grey bars of Fig. 3 show the observed native rates of H₂O₂ production by mitochondria oxidizing NAD-linked substrates in the presence and absence of ATP synthesis. The maximum rate of H₂O₂ production from each site greatly exceeds any of the observed native rates, and the sum of the maximum rates is 50-fold greater than any of the native rates, illustrating vividly that any one site, or any combination of the sites, could, in principle, account for the native rates. In the following sections we will dissect the native H₂O₂ production under each condition into its components using reporter-based assays.

This study can be divided into two parts: (A) identification and calibration of endogenous reporters using respiratory inhibitors to define the individual sites, and (B) measurement of

the reporters in a complex system in which the sites of superoxide production are unknown, prediction of the contributions of each of the reported sites, and comparison of the sum of predicted rates to the total observed native H_2O_2 production rates.

Experimentally, both parts of this study were performed in the same way (Fig. 4). For each condition, during either calibration or measurement, we measured Amplex UltraRed oxidation to determine the H_2O_2 production rate (Fig. 4a), and the redox states of two endogenous reporters, NAD(P)H (Fig. 4b) and cytochrome b_{566} (Fig. 4c).

The reduction state of NAD(P)H as a site-specific endogenous reporter of superoxide production at site I_F

The immediate electron donor to oxygen during superoxide production at site I_F (the flavin mononucleotide in the NADH oxidation site of complex I) is thought to be the fully-reduced flavin [21]. Although the reduction state of the flavin can be detected by its absorbance [35], and the “semiflavin” product can be detected using EPR [36], the signals are small and therefore unsuitable for routine use as reporters of the rate of superoxide production from site I_F .

NADH, the reductant of the flavin, is more suitable as a reporter. The observation in isolated mitochondria that complex I is readily reduced by NADH, and in turn can readily reduce NAD^+ during reverse electron transport, indicates that NADH and flavin may be close to equilibrium [37]. A potential problem is that NAD^+ competes with NADH for the binding site, which can kinetically limit the reduction of the flavin and make it sensitive to NAD-pool size [21]. However, in isolated mitochondria the NAD-pool size is effectively constant, and any kinetic limitation by NAD^+ does not appear to prevent easy reversibility. NADH may therefore be a suitable endogenous reporter of the redox state of the flavin and of I_F superoxide production [26;38]. The reduction state of endogenous mitochondrial NAD(P)H can be measured by autofluorescence with excitation at 365 nm and emission at 450 nm. It is generally referred to as NAD(P)H to acknowledge that some of the signal comes from NADPH, although that contribution is expected to be small in skeletal muscle mitochondria [26]. Much of the fluorescence signal may come from NADH bound to the active site of complex I [39], making it particularly specific and suitable as an endogenous reporter of site I_F . Following these considerations, we have previously found that the autofluorescence of NAD(P)H can report the rate of superoxide production at site I_F under different conditions [26;38].

Calibration of NAD(P)H redox state as a reporter of superoxide production from site I_F

Site I_F generates superoxide at maximal rate when the NAD(P)H pool is highly reduced; this can be achieved by the addition NADH to complex I, or NAD-linked substrates to isolated mitochondria, in the presence of rotenone (a Q-binding site inhibitor of complex I) to block electron escape from the complex [26;40–42].

The site can be titrated by adding different sub-maximal concentrations of substrate. We chose to calibrate with different concentrations of malate in the presence of rotenone. This was to minimize contributions from downstream electron transport chain complexes, which will remain oxidized under these conditions. The matrix enzymes α -ketoglutarate dehydrogenase and pyruvate dehydrogenase can generate superoxide/ H_2O_2 under some conditions [23–25]. If these matrix dehydrogenases do cause any H_2O_2 production in these calibrations with no obvious source of α -ketoglutarate or pyruvate, using electrons from NAD(P)H, it will be correctly accounted for but wrongly attributed to site I_F .

Fig. 5a shows the relationship of superoxide production rate from site I_F (measured as H_2O_2) to the concentration of malate during titration with malate in the presence of

rotenone. Fig. 5b shows the relationship of NAD(P)H reduction state to the concentration of malate during parallel titrations. Fig. 5c is the replot of superoxide production rate from site I_F as a function of NAD(P)H reduction in control mitochondria (filled symbols). This is the critical calibration curve to be used to report the rate of superoxide production from site I_F .

For completeness, Fig. 5c also shows the relationship between superoxide production rate from site I_F and NAD(P)H reduction in CDNB-treated mitochondria (open symbols). CDNB treatment, by decreasing matrix glutathione concentration and slowing matrix peroxidase-dependent consumption of H_2O_2 , allows a more realistic approximation of actual rates of superoxide production by site I_F compared to the measurements under standard conditions [26].

The reduction state of cytochrome b_{566} as a site-specific endogenous reporter of superoxide production at site III_{Q_0}

The immediate electron donor to oxygen during superoxide production at site III_{Q_0} (the quinol oxidation site located on the outer side of complex III, facing the intermembrane space) is thought to be the semiquinone in the Q_0 site [12;43;44]. Although this semiquinone can be detected using EPR [45], the signal is very difficult to work with and therefore unsuitable for routine use as a reporter of the rate of superoxide production from site III_{Q_0} .

Cytochrome b_{566} is the immediate oxidant of this semiquinone in the Q_0 site [46;47], and may be more suitable as a reporter, since its reduction state can be measured by dual-wavelength absorbance spectroscopy [20;43].

In the presence of the Q_i site inhibitor, antimycin A, superoxide production at site III_{Q_0} is not a unique function of cytochrome b_{566} redox state but depends on the redox states of both cytochrome b_{566} and cytochrome b_{562} [20]. However, the redox relationship between these cytochromes depends on the membrane potential. At a protonmotive force of about 140 mV (applied by ATP hydrolysis) in the presence of antimycin A, cytochrome b_{562} has a similar apparent mid-point potential to cytochrome b_{566} , and the dependence of superoxide production rate approximates to a function of cytochrome b_{566} reduction alone [20]. In the present study all measurements of native rates were made in the absence of antimycin A and the presence of high protonmotive force (140 – 183 mV). Under these conditions, the relationship between the superoxide production rate at site III_{Q_0} and the reduction state of cytochrome b_{566} (Fig. 7c) was similar to that found previously when a protonmotive force of 140 mV was imposed in the presence of antimycin A [20]. Therefore, as proton motive force rises higher than 140 mV, we assume that further oxidation of cytochrome b_{562} relative to cytochrome b_{566} has no effect on the relationship between superoxide production and cytochrome b_{566} redox state. These considerations allow the use of cytochrome b_{566} redox state as a reporter of site III_{Q_0} .

Calibration of cytochrome b_{566} redox state as a reporter of superoxide production from site III_{Q_0}

The superoxide production rate of site III_{Q_0} was assayed as the rate with succinate as substrate, in the presence of rotenone, that was sensitive to the Q_0 site inhibitor myxothiazol (Fig. 6a). We initially assumed that this protocol would avoid reduction of NADH, but found that NAD(P)H was significantly reduced by succinate, and further reduced when myxothiazol was added (Fig. 6b). The residual rate of superoxide production after addition of myxothiazol in Fig. 6a was correctly reported by the I_F calibration in Fig. 5c (not shown), identifying its source as site I_F .

The superoxide production rate of site III_{Q_0} was calibrated at different succinate:malonate ratios. Fig. 6c shows the reduction of NAD(P)H at each succinate:malonate ratio. During

reporter calibration, the contribution of site I_F to the signal before and after addition of myxothiazol was corrected for during calculation of the rate from site III_{Q₀}.

Fig. 7a shows the relationship of the superoxide production rate from site III_{Q₀} (measured as H₂O₂) to the concentration of succinate during titration with different succinate:malonate ratios. Fig. 7b shows the relationship of cytochrome *b*₅₆₆ reduction state to the concentration of succinate during parallel titrations. Fig. 7c is the replot of the corrected superoxide production rate from site III_{Q₀} as a function of cytochrome *b*₅₆₆ reduction in control mitochondria (filled symbols). This is the critical calibration curve to be used to report the rate of superoxide production from site III_{Q₀}.

For completeness, Fig. 7c also shows the relationship between superoxide production rate from site III_{Q₀} and cytochrome *b*₅₆₆ reduction that would be predicted in CDNB-treated mitochondria (open symbols) to allow a more realistic approximation of total superoxide production by site III_{Q₀}. The prediction was made by correcting the observed rates in control mitochondria to the rates in CDNB-treated mitochondria using the relationship in Fig. 1, assuming that 50% of the superoxide from site III_{Q₀} is generated in the matrix [26;29;48;49].

Use of the reporters to predict native rates of mitochondrial superoxide production

The calibration curves shown in Figs 5 and 7 were constructed to report the rates of superoxide production from site I_F and site III_{Q₀} under different applied conditions in these isolated skeletal muscle mitochondria. The applied conditions could include different substrates, different inhibitors, and different respiratory states, making these reporter-based assays powerful and general tools for the characterization of mitochondrial superoxide/H₂O₂ production. For the initial application and validation of the method, we chose to measure the native rates of superoxide production by sites I_F and III_{Q₀} in a relatively simple system: mitochondria oxidizing NAD-linked substrates (5 mM malate, or the combination of 5 mM malate plus 5 mM glutamate), in the absence of inhibitors, during maximum rates of ATP synthesis (phosphorylating; state 3), or in the presence of oligomycin to establish non-phosphorylating conditions (state 4).

For these experiments, NAD(P)H reduction state, cytochrome *b*₅₆₆ reduction state, and the overall rate of H₂O₂ production were measured in parallel on the same batch of skeletal muscle mitochondria (in different cuvettes), as exemplified in Fig. 4. The observed reduction levels of the two reporters are listed in Table 1. These values were used to report the rates of superoxide generation from sites I_F and III_{Q₀} using the calibration curves in Fig. 5c and Fig. 7c. The predicted rates of superoxide production from each site for control and CDNB-treated mitochondria are reported in Table 2 and Fig. 8. The measured total rates of H₂O₂ production under the same conditions are also shown, allowing comparison of the sum of the reported rates from each site with the experimentally-observed total rates.

The results for the four experimental conditions are shown in Fig. 8. Figs 8a and 8b show H₂O₂ production by mitochondria oxidizing malate; Figs 8c and 8d show production by mitochondria oxidizing glutamate plus malate.

When mitochondria oxidized malate whilst generating ATP (Fig. 8a), NAD(P)H and cytochrome *b*₅₆₆ were relatively oxidized (Table 1), as would be expected since malate is a relatively poor substrate for supply of electrons, and ATP synthesis keeps the protonmotive force relatively low (140 mV) and the electron transport chain relatively oxidized. Under these conditions, the reported rates of superoxide production were low (Table 2, Fig. 8a). Most of the superoxide (~80%) was reported to arise from site I_F. The sum of the reported rates was not different from the measured rate of H₂O₂ production (second bar in Fig. 8a)

($p > 0.8$; t-test). This good agreement between the sum of the reported rates and the measured rate is consistent with the reporters working well, and with little or no contribution to H_2O_2 production under these conditions from sites other than I_F and III_{Q_0} . The third bar in Fig. 8a shows the rates reported using the CDNB-corrected calibration curves in Figs 5c and 7c. Because the correction was applied to the whole of the superoxide production from site I_F , but only to the 50% of superoxide production from site III_{Q_0} assumed to be in the matrix, and because the correction is different at different rates, the proportion of superoxide production reported from site I_F was slightly changed. Again, there was good agreement between the sum of the reported rates and the measured H_2O_2 production rate (fourth bar in Fig. 8a); there was no statistical difference between the total reported and measured H_2O_2 production rates ($p > 0.9$; t-test).

Fig. 8b shows the same analysis, but this time of mitochondria oxidizing malate under non-phosphorylating conditions (not generating ATP), in which the protonmotive force was higher (165 mV) and the electron transport chain was consequently less oxidized (Table 1). The reported contribution of site I_F was little changed, but the reported contribution of site III_{Q_0} increased ~3-fold (Table 2). The reported superoxide production rate was shared between sites I_F and III_{Q_0} (~36% I_F and ~20% III_{Q_0}). In this case the sum of the reported rates was significantly less than the measured H_2O_2 production rate. In the CDNB treated mitochondria, the reporters accounted for only about half of the observed rate, with ~30 $\text{pmol H}_2\text{O}_2 \cdot \text{min}^{-1} \cdot \text{mg protein}^{-1}$ remaining unassigned to either site.

Figs 8c and 8d show the second experimental condition, mitochondria oxidizing glutamate and malate. Addition of glutamate removes matrix oxaloacetate by transamination and makes malate a more effective substrate for supply of electrons. Fig. 8c shows the phosphorylating state. Compared to Fig. 8a, addition of glutamate increased protonmotive force from 140 to 158 mV, appeared to cause slightly more reduction of cytochrome b_{566} , but not of NAD(P)H, and increased the reported contribution of III_{Q_0} , but not of site I_F (Table 2). The reported and measured rates in CDNB-treated mitochondria are also shown. Both sites I_F and III_{Q_0} contributed to superoxide production, with site I_F tending to dominate (~70% I_F). There was no significant difference between the sum of the reported rates and the observed rate of H_2O_2 production in either set of data.

Fig. 8d shows the analysis of H_2O_2 production by mitochondria oxidizing glutamate plus malate, but under non-phosphorylating conditions. These conditions generated the highest rates of H_2O_2 production observed in this study. Because of the good electron supply and high protonmotive force (183 mV), there was relatively high reduction of both NAD(P)H and cytochrome b_{566} (Table 1). This led to relatively high reported rates from both sites. Compared to the phosphorylating state, the rate of superoxide production by site III_{Q_0} increased ~4–5-fold (Table 2). The reported rate of superoxide production was shared between sites I_F and III_{Q_0} (~45% I_F and ~22% III_{Q_0}). Similarly to the situation with malate in the non-phosphorylating condition, the sum of the reported rates was significantly less than the observed H_2O_2 production rate. In the CDNB-treated mitochondria ~60 $\text{pmol H}_2\text{O}_2 \cdot \text{min}^{-1} \cdot \text{mg protein}^{-1}$ were unassigned, accounting for ~33% of the observed signal.

CONCLUSIONS

The present study is the first to dissect the individual sites of superoxide production by mitochondria under native conditions of oxidation of NAD-linked substrates with no added respiratory chain inhibitors.

The two conditions that most favour formation of H_2O_2 by isolated mitochondria are a highly reduced NAD(P)H pool [21;26;50], and a high protonmotive force together with a

reduced ubiquinone pool [33;38;51;52]. From studies using respiratory chain inhibitors, we know which sites are most active under specific contrived conditions, such as site I_Q during oxidation of succinate before addition of rotenone [33;51;52], site I_F during oxidation of NAD-linked substrates in the presence of rotenone [40–42], or site III_{Q₀} during oxidation of succinate in the presence of rotenone and antimycin A [12;20;43;44]. Although efforts have been made to characterize the mitochondrial sites that generate superoxide/H₂O₂ in complex systems [53–56], such studies using inhibitors do not unambiguously reveal which sites are most active under native conditions in the absence of added respiratory chain inhibitors with more physiologically-relevant substrates.

Previous studies that have attempted to identify specific sites used pharmacological or genetic manipulation of putative sites of superoxide/H₂O₂ production and inferred the sites from the resulting changes in ROS production. For example, it is sometimes reported that addition of antimycin A to cells increases ROS production, with the implied or explicit inference that the native ROS production must have been from complex III (e.g. [57]). This inference is invalid for two reasons. First, the observation that a site produces ROS after inhibition does not show that it produced them before the addition of the inhibitor. Second, the addition of an inhibitor will increase superoxide/H₂O₂ production from upstream sites as they become more reduced, and decrease superoxide/H₂O₂ production from any downstream sites as they become more oxidized, distorting the pattern of production so much that few valid inferences can be made about the native sites of production before inhibitor addition. Similarly, addition of inhibitors of site III_{Q₀}, such as myxothiazol or stigmatellin, or genetic ablation of III_{Q₀} activity [58] may decrease ROS production in cells. However, such inhibition will tend to increase superoxide/H₂O₂ production from site I_F and complex II by causing reduction of ubiquinone and NADH, and to decrease production from site I_Q by lowering the protonmotive force. These complications make the assignment of production to site III_{Q₀} using these methods unreliable.

Previous studies recognized that the reduction states of the NADH and ubiquinone pools affect or determine superoxide production rates, but did not calibrate these relationships or use them for quantitative predictions. Such studies related the observed rate of superoxide production by site I_F to NAD(P)H reduction state [21;31;38], and the observed rate of superoxide production by site III_{Q₀} to cytochrome *b*₅₆₆ reduction state [20].

In the present paper we introduce a novel general solution to the problem of identifying and quantifying individual sites of superoxide/H₂O₂ production: the calibration of endogenous reporters using inhibitors to define the sites that they report on, and the measurement of those reporters in the absence of added respiratory chain inhibitors to predict native rates of superoxide production from individual sites and their contribution to the total rate of superoxide production. We show that NAD(P)H reduction state can be used to report the rate of superoxide production from site I_F, and cytochrome *b*₅₆₆ reduction state can be used to report the superoxide production rate from site III_{Q₀}. Using these calibrated reporters, we estimated the contributions of sites I_F and III_{Q₀} to H₂O₂ production in mitochondria isolated from rat skeletal muscle oxidizing malate or glutamate plus malate as substrates under two different conditions.

Fig. 8 shows that under phosphorylating conditions the reported rates of superoxide production from sites I_F and III_{Q₀} accounted fully for the observed rates, with site I_F generating much of the total. We conclude that these two sites are the only significant contributors to mitochondrial H₂O₂ production during ATP synthesis under the conditions used in the present paper.

Under non-phosphorylating conditions superoxide production was ~40% from site I_F, ~20% from site III_{Q_o} and ~40% from unidentified sites. It is possible that this difference between reported and observed rates reflects systematic errors in our assumptions and calibrations, or it may suggest that there were contributions to H₂O₂ production by other sites in this system. The best candidates for other site(s) are the lipoic acid-containing enzyme complexes (α -ketoglutarate dehydrogenase and pyruvate dehydrogenase) and site I_Q. Complex II is not a good candidate, since its superoxide/H₂O₂ production is strongly inhibited by malate [18], which was added in all the current experiments.

When malate alone is oxidized, α -ketoglutarate dehydrogenase is not expected to be active. However, in the presence of glutamate it is possible that α -ketoglutarate dehydrogenase was provided sufficient substrate to generate superoxide/H₂O₂ [23–25], and this could have contributed some of the unassigned H₂O₂ generation during oxidation of glutamate plus malate. However, when we used 5 mM glutamate plus 5 mM malate instead of 5 mM malate alone in the I_F calibration of Fig. 5c (data not shown) there was no significant difference in the rate of H₂O₂ production. This suggests that α -ketoglutarate dehydrogenase does not contribute significant H₂O₂ to the total even when glutamate is present under our conditions.

Site I_Q generates superoxide/H₂O₂ at high rates in the presence of succinate and absence of rotenone [33] (Fig. 3), and in the presence of rotenone and NAD-linked substrates when a protonmotive force is generated by ATP hydrolysis [54]. The site is very sensitive to protonmotive force (particularly the transmembrane pH gradient) [33] and to the redox state of the ubiquinone pool [38], both of which increase under non-phosphorylating compared to phosphorylating conditions, and during oxidation of glutamate plus malate compared to malate alone. Therefore, site I_Q is an attractive candidate for the unassigned H₂O₂ production during non-phosphorylating respiration with malate alone and with glutamate plus malate as substrates.

Our results show that the rate of H₂O₂ production by mitochondria depends strongly on their state (i.e. on the protonmotive force): in mitochondria treated with CDNB the rate of H₂O₂ production increased 2–3-fold (malate) or ~4-fold (glutamate plus malate) between the phosphorylating and non-phosphorylating states (Table 2). The rate of superoxide/H₂O₂ production by the electron transport chain cannot depend simply on the rate of electron flow through the chain (as implied when rates are reported as a percentage of respiration rate), since there is no obvious mechanism for such a relationship, and rates are increased when electron transport is stimulated by addition of substrate, yet decreased when electron transport is increased by addition of uncouplers [28]. However, in this context in which no inhibitors were added and respiration was not manipulated, reporting ROS production as a percentage of respiration can give a general sense of the scale of the electron leak. In non-phosphorylating conditions, electron leak to oxygen was 0.25% (malate) and 0.4% (glutamate plus malate) of total respiration rate. Under phosphorylating conditions it was considerably less, 0.03% (malate) and 0.02% (glutamate plus malate) of total respiration rate.

An interesting aspect of our results is the prediction that the rate of direct superoxide production (as opposed to H₂O₂ production) in the extramitochondrial compartment depends even more strongly on the respiratory state of the mitochondria: the rate of external superoxide production increased ~3-fold (malate) or 4–5-fold (glutamate plus malate) between phosphorylating and non-phosphorylating states (Table 2). External superoxide production is expected to arise only from site III_{Q_o} in this system. Superoxide from site I_F, site I_Q and α -ketoglutarate dehydrogenase is produced exclusively in the matrix [29], but ~50% of site III_{Q_o} superoxide production is thought to be directed to the extramitochondrial compartment [26;29;48;49]. In principle, these large changes in extramitochondrial H₂O₂ or

superoxide production rate could be used to signal the respiratory state of the mitochondria to the cytosolic compartment through H₂O₂ or superoxide-sensitive components in the cytosol, as proposed by others [16;17;58].

As a proof of concept and illustration of its application, the experimental design in the present work was necessarily relatively simple. However, the method can readily be expanded to other substrates, conditions, and sites of interest. For example, site I_Q has one of the highest measured V_{max} values of all the mitochondrial H₂O₂-producing sites [17;33], (Fig. 3), but its contribution to H₂O₂ production during oxidation of physiologically-relevant substrates such as fatty acids or glycerol 3-phosphate in cells, or *in vivo* is unknown. Similarly, complex II, α-glycerophosphate dehydrogenase, ETFQOR and lipoic acid containing dehydrogenases may each be capable of generating superoxide/H₂O₂ at significant rates under specific conditions. Through careful experimental design, the strategy introduced here should enable elucidation of the specific H₂O₂-producing behaviour of isolated mitochondria under many experimental conditions. If technical issues can be resolved, extension of these principles may also prove useful with intact cells as well as whole tissues and organisms. An improved understanding of which mitochondrial superoxide-and H₂O₂-producing sites are active physiologically and pathologically should greatly assist evaluation of the mechanisms and roles of mitochondrial ROS production in physiology and pathology.

Acknowledgments

FUNDING This work was supported by National Institutes of Health grants P01 AG025901, PL1 AG032118 and R01 AG033542 and by The Ellison Medical Foundation, grant AG-SS-2288-09. JRT is supported by the Canada Research Chairs Program.

We thank Akos A. Gerencser for contributing his expertise in mathematical analysis.

Reference List

1. Begrich K, Igoudjil A, Pessayre D, Fromenty B. Mitochondrial dysfunction in NASH: causes, consequences and possible means to prevent it. *Mitochondrion*. 2006; 6:1–28. [PubMed: 16406828]
2. Green K, Brand MD, Murphy MP. Prevention of mitochondrial oxidative damage as a therapeutic strategy in diabetes. *Diabetes*. 2004; 53(Suppl 1):S110–S118. [PubMed: 14749275]
3. Witte ME, Geurts JGG, de Vries HE, van der Valk P, van Horsen J. Mitochondrial dysfunction: a potential link between neuroinflammation and neurodegeneration? *Mitochondrion*. 2010; 10:411–418. [PubMed: 20573557]
4. Rodriguez-Enriquez S, Neuzil J, Saavedra E, Moreno-Sanchez R. The causes of cancer revisited: “mitochondrial malignancy” and ROS-induced oncogenic transformation - why mitochondria are targets for cancer therapy. *Mol Aspects Med*. 2010; 31:145–170. [PubMed: 20206201]
5. Barja G. Free radicals and aging. *Trends Neurosci*. 2004; 27:595–600. [PubMed: 15374670]
6. Muller FL, Lustgarten MS, Jang Y, Richardson A, Van Remmen H. Trends in oxidative aging theories. *Free Radic Biol Med*. 2007; 43:477–503. [PubMed: 17640558]
7. Ladiges W, Van Remmen H, Strong R, Ikeno Y, Treuting P, Rabinovitch P, Richardson A. Lifespan extension in genetically modified mice. *Aging Cell*. 2009; 8:346–352. [PubMed: 19485964]
8. Mookerjee SA, Divakaruni AS, Jastroch M, Brand MD. Mitochondrial uncoupling and lifespan. *Mech Ageing Dev*. 2010; 131:463–472. [PubMed: 20363244]
9. Quinlan, CL.; Treberg, JR.; Brand, MD. Mechanisms of Mitochondrial Free Radical Production and their Relationship to the Aging Process. In: Masoro, E.; Austad, SN., editors. *The Handbook of the Biology of Aging*. Academic Press; San Diego, USA: 2010. p. 45-55.
10. Finkel T. Signal transduction by reactive oxygen species. *J Cell Biol*. 2011; 194:7–15. [PubMed: 21746850]

11. Boveris A, Oshino N, Chance B. The cellular production of hydrogen peroxide. *Biochem J.* 1972; 128:617–630. [PubMed: 4404507]
12. Boveris A, Chance B. The mitochondrial generation of hydrogen peroxide. General properties and effect of hyperbaric oxygen. *Biochem J.* 1973; 134:707–716. [PubMed: 4749271]
13. Loschen G, Flohe L, Chance B. Respiratory chain linked H₂O₂ production in pigeon heart mitochondria. *FEBS Lett.* 1971; 18:261–264. [PubMed: 11946135]
14. McCord JM, Fridovich I. The biology and pathology of oxygen radicals. *Ann Intern Med.* 1978; 89:122–127. [PubMed: 208444]
15. Adam-Vizi V. Production of reactive oxygen species in brain mitochondria: contribution by electron transport chain and non-electron transport chain sources. *Antioxid Redox Signal.* 2005; 7:1140–1149. [PubMed: 16115017]
16. Murphy MP. How mitochondria produce reactive oxygen species. *Biochem J.* 2009; 417:1–13. [PubMed: 19061483]
17. Brand MD. The sites and topology of mitochondrial superoxide production. *Exp Gerontol.* 2010; 45:466–472. [PubMed: 20064600]
18. Quinlan CL, Orr AL, Treberg JR, Perevoshchikova IV, Brand MD. Mitochondrial complex II generates superoxide in the forward and reverse reactions. *J Biol Chem.* 2012 Jun 11. [Epub ahead of print].
19. Affourtit, C.; Quinlan, CL.; Brand, MD. Measurement of proton leak and electron leak in isolated mitochondria. In: Palmeira, CM.; Moreno, AJ., editors. *Mitochondrial Bioenergetics.* Humana Press; New York, NY: 2012. p. 165-182.
20. Quinlan CL, Gerencser AA, Treberg JR, Brand MD. The mechanism of superoxide production by the antimycin-inhibited mitochondrial Q-cycle. *J Biol Chem.* 2011; 286:31361–31372. [PubMed: 21708945]
21. Sun J, Trumppower BL. Superoxide anion generation by the cytochrome bc₁ complex. *Arch Biochem Biophys.* 2003; 419:198–206. [PubMed: 14592463]
22. Kussmaul L, Hirst J. The mechanism of superoxide production by NADH:ubiquinone oxidoreductase (complex I) from bovine heart mitochondria. *Proc Natl Acad Sci USA.* 2006; 103:7607–7612. [PubMed: 16682634]
23. Bunik VI, Sievers C. Inactivation of the 2-oxo acid dehydrogenase complexes upon generation of intrinsic radical species. *Eur J Biochem.* 2002;5004–5015. [PubMed: 12383259]
24. Starkov AA, Fiskum G, Chinopoulos C, Lorenzo BJ, Browne SE, Patel MS, Beal MF. Mitochondrial alpha-ketoglutarate dehydrogenase complex generates reactive oxygen species. *J Neurosci.* 2004; 24:7779–7788. [PubMed: 15356189]
25. Tretter L, Adam-Vizi V. Generation of Reactive Oxygen Species in the Reaction Catalyzed by alpha-Ketoglutarate Dehydrogenase. *J Neurosci.* 2004; 24:7771–7778. [PubMed: 15356188]
26. Treberg JR, Quinlan CL, Brand MD. Hydrogen peroxide efflux from muscle mitochondria underestimates matrix superoxide production - a correction using glutathione depletion. *FEBS J.* 2010; 277:2766–2778. [PubMed: 20491900]
27. Meinhardt SW, Crofts AR. The role of cytochrome b₅₆₆ in the electron-transfer chain of *Rhodospseudomonas sphaeroides*. *Biochim Biophys Acta, Bioenerg.* 1983; 723:219–230.
28. Brand MD, Affourtit C, Esteves TC, Green K, Lambert AJ, Miwa S, Pakay JL, Parker N. Mitochondrial superoxide: production, biological effects, and activation of uncoupling proteins. *Free Radic Biol Med.* 2004; 37:755–767. [PubMed: 15304252]
29. St-Pierre J, Buckingham JA, Roebuck SJ, Brand MD. Topology of superoxide production from different sites in the mitochondrial electron transport chain. *J Biol Chem.* 2002; 277:44784–44790. [PubMed: 12237311]
30. Hoffman DL, Brookes PS. Oxygen sensitivity of mitochondrial reactive oxygen species generation depends on metabolic conditions. *J Biol Chem.* 2009; 284:16236–45. [PubMed: 19366681]
31. Hansford RG, Hogue BA, Mildaziene V. Dependence of H₂O₂ formation by rat heart mitochondria on substrate availability and donor age. *J Bioenerg Biomembr.* 1997; 29:89–95. [PubMed: 9067806]
32. Liu Y, Fiskum G, Schubert D. Generation of reactive oxygen species by the mitochondrial electron transport chain. *J Neurochem.* 2002; 80:780–787. [PubMed: 11948241]

33. Lambert AJ, Brand MD. Superoxide production by NADH:ubiquinone oxidoreductase (complex I) depends on the pH gradient across the mitochondrial inner membrane. *Biochem J.* 2004; 382:511–517. [PubMed: 15175007]
34. Treberg JR, Brand MD. A model of the proton translocation mechanism of complex I. *J Biol Chem.* 2011; 286:17579–17584. [PubMed: 21454533]
35. Scholz R, Thurman RG, Williamson JR, Chance B, Bücher T. Flavin and pyridine nucleotide oxidation-reduction changes in perfused rat liver. *J Biol Chem.* 1969; 244:2317–2324. [PubMed: 4306507]
36. Ohnishi ST, Shinzawa-Itoh K, Ohta K, Yoshikawa S, Ohnishi T. New insights into the superoxide generation sites in bovine heart NADH-ubiquinone oxidoreductase (Complex I): The significance of protein-associated ubiquinone and the dynamic shifting of generation sites between semiquinone and semiquinone radicals. *Biochim Biophys Acta, Bioenerg.* 2010; 1797:1901–1909.
37. Brown GC, Brand MD. Proton/electron stoichiometry of mitochondrial complex I estimated from the equilibrium thermodynamic force ratio. *Biochem J.* 1988; 252:473–479. [PubMed: 2843170]
38. Treberg JR, Quinlan CL, Brand MD. Evidence for two sites of superoxide production by mitochondrial NADH-ubiquinone oxidoreductase (complex I). *J Biol Chem.* 2011; 286:27103–27110. [PubMed: 21659507]
39. Blinova K, Levine RL, Boja ES, Griffiths GL, Shi-Zhen D, Ruddy B, Balaban RS. Mitochondrial NADH Fluorescence Is Enhanced by Complex I Binding. *Biochemistry.* 2008; 47:9636–9645. [PubMed: 18702505]
40. Herrero A, Barja G. Sites and mechanisms responsible for the low rate of free radical production of heart mitochondria in the long-lived pigeon. *Mech Ageing Dev.* 1997; 98:95–111. [PubMed: 9379714]
41. Hirst J, King MS, Pryde KR. The production of reactive oxygen species by complex I. *Biochem Soc Trans.* 2008; 36:976–980. [PubMed: 18793173]
42. Fato R, Bergamini C, Bortolus M, Maniero AL, Leoni S, Ohnishi T, Lenaz G. Differential effects of mitochondrial Complex I inhibitors on production of reactive oxygen species. *Biochim Biophys Acta, Bioenerg.* 2009; 1787:384–392.
43. Erecinska M, Chance B, Wilson DF, Dutton PL. Aerobic reduction of cytochrome b 566 in pigeon-heart mitochondria (succinate-cytochrome C₁ reductase-stopped-flow kinetics). *Proc Natl Acad Sci USA.* 1972; 69:50–54. [PubMed: 4550509]
44. Muller F, Crofts AR, Kramer DM. Multiple Q-cycle bypass reactions at the Q_o site of the cytochrome bc₁ complex. *Biochemistry.* 2002; 41:7866–7874. [PubMed: 12069575]
45. Cape JL, Bowman MK, Kramer DM. A semiquinone intermediate generated at the Q_o site of the cytochrome bc₁ complex: Importance for the Q-cycle and superoxide production. *Proc Natl Acad Sci USA.* 2007; 104:7887–7892. [PubMed: 17470780]
46. Trumpower BL. Evidence for a protonmotive Q cycle mechanism of electron transfer through the cytochrome b–c₁ complex. *Biochem Biophys Res Commun.* 1976; 70:73–80. [PubMed: 179550]
47. Crofts AR. The cytochrome bc₁ complex: function in the context of structure. *Annu Rev Physiol.* 2004; 66:689–733. [PubMed: 14977419]
48. Muller FL, Liu Y, Van Remmen H. Complex III releases superoxide to both sides of the inner mitochondrial membrane. *J Biol Chem.* 2004; 279:49064–49073. [PubMed: 15317809]
49. Miwa S, Brand MD. The topology of superoxide production by complex III and glycerol 3-phosphate dehydrogenase in *Drosophila* mitochondria. *Biochim Biophys Acta Bioenerg.* 2005; 1709:214–219.
50. Kudin AP, Bimpong-Buta NY, Vielhaber S, Elger CE, Kunz WS. Characterization of superoxide-producing sites in isolated brain mitochondria. *J Biol Chem.* 2004; 279:4127–4135. [PubMed: 14625276]
51. Korshunov SS, Skulachev VP, Starkov AA. High protonic potential actuates a mechanism of production of reactive oxygen species in mitochondria. *FEBS Lett.* 1997; 416:15–18. [PubMed: 9369223]
52. Muller FL, Liu Y, Abdul-Ghani MA, Lustgarten MS, Bhattacharya A, Jang YC, Van Remmen H. High rates of superoxide production in skeletal-muscle mitochondria respiring on both complex I- and complex II-linked substrates. *Biochem J.* 2008; 409:491–499. [PubMed: 17916065]

53. Turrens JF, Alexandre A, Lehninger AL. Ubisemiquinone is the electron donor for superoxide formation by complex III of heart mitochondria. *Arch Biochem Biophys.* 1985; 237:408–414. [PubMed: 2983613]
54. Lambert AJ, Brand MD. Inhibitors of the quinone-binding site allow rapid superoxide production from mitochondrial NADH:ubiquinone oxidoreductase (complex I). *J Biol Chem.* 2004; 279:39414–39420. [PubMed: 15262965]
55. Gyulkhandanyan AV, Pennefather PS. Shift in the localization of sites of hydrogen peroxide production in brain mitochondria by mitochondrial stress. *J Neurochem.* 2004; 90:405–421. [PubMed: 15228597]
56. Kudin AP, Malinska D, Kunz WS. Sites of generation of reactive oxygen species in homogenates of brain tissue determined with the use of respiratory substrates and inhibitors. *Biochim Biophys Acta Bioenerg.* 2008; 1777:689–695.
57. Klimova T, Chandel NS. Mitochondrial complex III regulates hypoxic activation of HIF. *Cell Death Differ.* 2008; 15:660–666. [PubMed: 18219320]
58. Bell EL, Klimova TA, Eisenbart J, Moraes CT, Murphy MP, Budinger GRS, Chandel NS. The Qo site of the mitochondrial complex III is required for the transduction of hypoxic signaling via reactive oxygen species production. *J Cell Biol.* 2007; 177:1029–1036. [PubMed: 17562787]

Highlights

- Redox state of endogenous reporters was calibrated to rates of $\text{H}_2\text{O}_2/\text{O}_2^{\cdot-}$ production
- NADH and cytochrome b reported $\text{H}_2\text{O}_2/\text{O}_2^{\cdot-}$ production from complexes I and III
- Site-specific rates of $\text{H}_2\text{O}_2/\text{O}_2^{\cdot-}$ production were quantified under native conditions
- This approach can quantify site-specific $\text{H}_2\text{O}_2/\text{O}_2^{\cdot-}$ production in complex situations

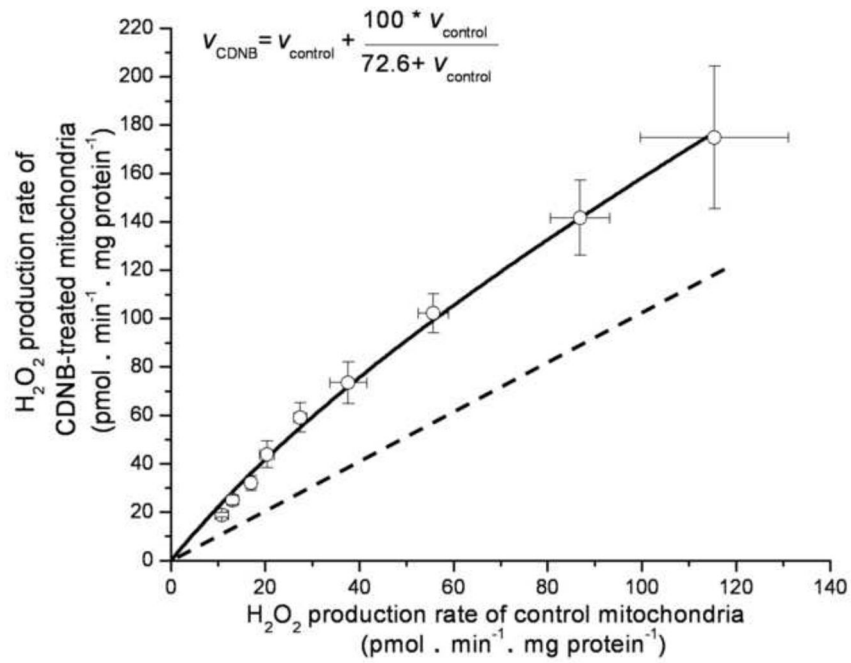
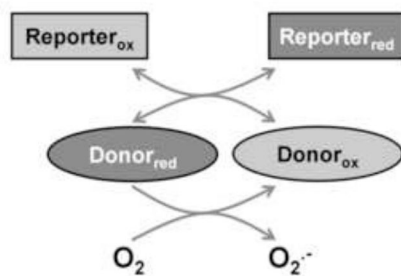


FIGURE 1. Comparison of rates of superoxide production (measured as H₂O₂ production) by site I_F in CDNB-pretreated and control mitochondria
 Site I_F was titrated by adding malate from 0.01 mM to 5 mM in separate runs followed by addition of rotenone. Points were fitted to give the parameter values in Eq. 1 (inset). The dashed line indicates a 1:1 relationship. Values are means ± SEM (n = 5).

(a) General principles of the reporter-based ROS assay



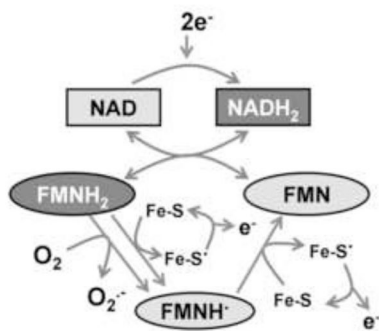
Assumptions:

1. $V_{O_2^{\bullet-}} = f(\text{Donor}_{red})$
2. $\frac{\text{Reporter}_{red}}{\text{Reporter}_{ox}} = f\left(\frac{\text{Donor}_{red}}{\text{Donor}_{ox}}\right)$

Therefore,

$$V_{O_2^{\bullet-}} = f(\text{Reporter}_{red})$$

(b) Site I_F donor and reporter



(c) Site III_{Qo} donor and reporter

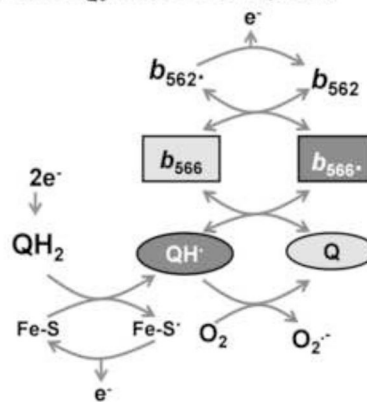


FIGURE 2. The theory and assumptions of the reporter-based assay of superoxide production rate

(a) Theory. When the donor species in a superoxide-producing site (ovals) is reduced (darker shading) it donates electrons to oxygen to generate superoxide. The redox state of the donor is reported by an adjacent redox centre (rectangles). Two primary assumptions are made. First, the rate of superoxide production is a unique function (f) of the reduction state of the donor (in the simplest case, the rate of superoxide production is the product of a pseudo first-order rate constant and the concentration or % reduction of the donor). Second, the reduction state of a relevant, nearby reporter is a unique function of the reduction state of the donor (in the simplest case the two centres are at equilibrium and their redox states are related by the Nernst equation). It follows that the rate of superoxide production will be a unique function of the reduction state of the reporter (in the simplest case, that function can be derived from the two simplest cases above). These assumptions allow the rate of superoxide production by a particular donor to be calibrated to the reduction state of the appropriate reporter. In the present study we calibrated the rate of superoxide production from site I_F to the reduction state of NAD(P)H, and the rate of superoxide production from site III_{Qo} to the reduction state of cytochrome b_{566} . (b) Reactions of donor (FMNH₂) and reporter (NADH₂) at site I_F. (c) Reactions of donor (QH•) and reporter (reduced cytochrome b_{566}) at site III_{Qo}.

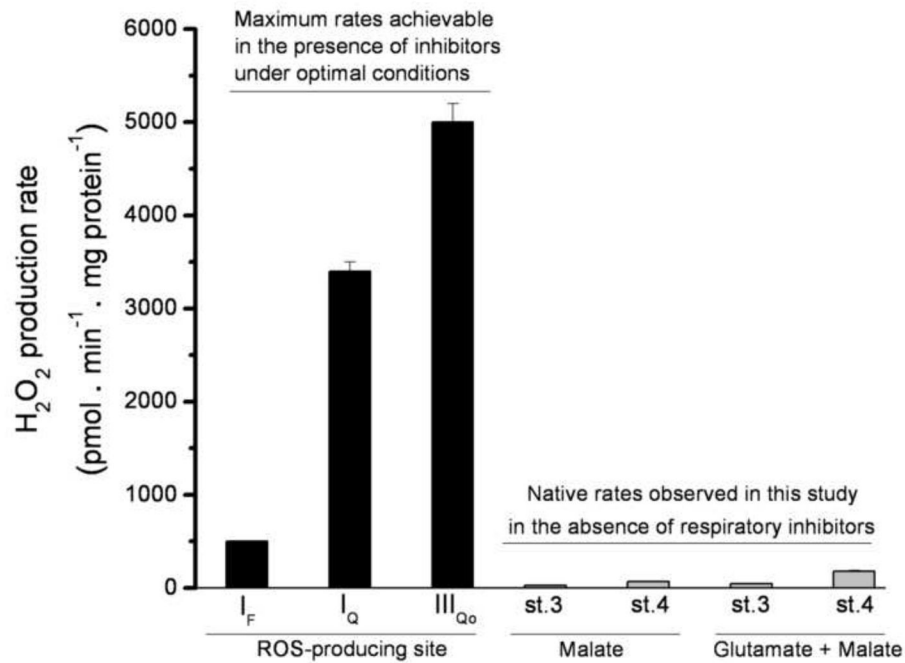


FIGURE 3. Maximum observed rates of H₂O₂ production from different sites compared to the native rates observed in this study

Black bars indicate the approximate maximum rates of H₂O₂ production from different sites in CDNB-treated rat skeletal muscle mitochondria, to establish a basis for comparison with the native rates. Data were obtained in standard KCl buffer (see MATERIALS AND METHODS), but lacking phosphate and magnesium (phosphate lowers the rate of H₂O₂ production from both site I_Q and site III_{Q₀}). Site I_F was assayed in the presence of 5 mM malate, 4 μM rotenone, and 4 μM FCCP. Site I_Q was assayed in the presence of 5 mM succinate as the rotenone-sensitive portion of the signal (no correction was made for the changes in rate of H₂O₂ production by other sites on addition of rotenone). Site I_Q data were adapted from [21]. Site III_{Q₀} was assayed with 5 mM succinate and 2.5 mM malonate, in the presence of 2 μM antimycin A, as the myxothiazol-sensitive portion of the signal (no correction was made for the changes in rate of H₂O₂ production by other sites on addition of myxothiazol). Site III_{Q₀} data were adapted from [20]. The final four bars (grey) show the current assay conditions (data from Fig. 8): CDNB-treated mitochondria oxidizing 5 mM malate or 5 mM glutamate plus 5 mM malate in the absence of inhibitors during ATP synthesis (st. 3) or in non-phosphorylating conditions (st. 4); the sites generating H₂O₂ in each condition will be determined in this study (Fig. 8). Data are means ± SEM (n = 4).

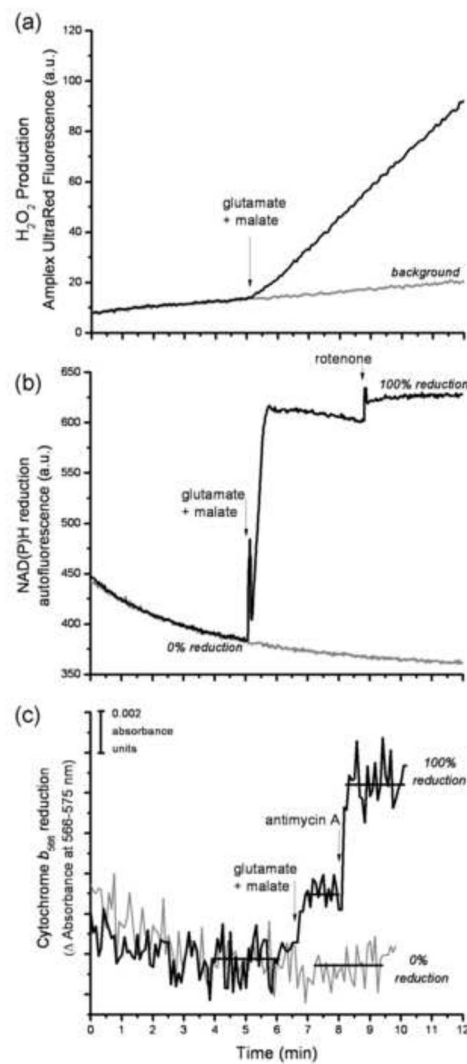


FIGURE 4. Design of the reporter-based superoxide assay

The assay was designed to measure the rate of H₂O₂ production using Amplex UltraRed, and the steady-state reduction levels of NAD(P)H and cytochrome *b*₅₆₆ under closely similar conditions. The timing of all additions was synchronized between the three assays. The addition of substrate, in this case 5 mM glutamate plus 5 mM malate, led to an increased rate of change of Amplex UltraRed fluorescence (a), and increased steady-state reduction levels of both NAD(P)H (b) and cytochrome *b*₅₆₆ (c). The gray traces in each graph show the control in the absence of substrates or inhibitors. The 100% value for each reporter in (b) and (c) was established by addition of its relevant downstream inhibitor (4 μM rotenone or 2 μM antimycin A), as described in MATERIALS AND METHODS. The horizontal bars in (c) indicate the regions of data that were averaged to give the mean value used to calculate the % reduction of cytochrome *b*₅₆₆. All of the data for the calibration curves (Figs 5 and 7) and measurements of native rates (Fig. 8) were collected in essentially this way.

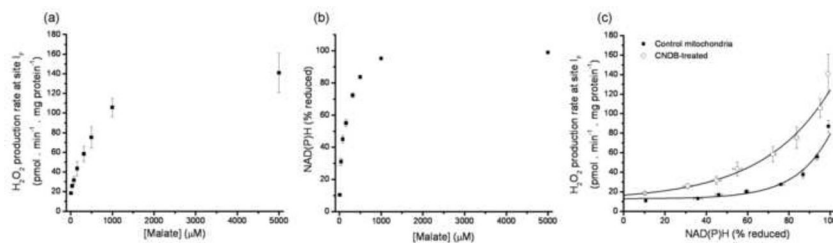


FIGURE 5. Relationship between the rate of superoxide production by site I_F and the reduction state of NAD(P)H

(a) Dependence of the rate of H₂O₂ production on malate concentration after addition of 4 μM rotenone. (b) Dependence of %NAD(P)H reduction on malate concentration after addition of rotenone (100% reduction was subsequently established by addition of 5 mM malate). (c) Final calibration of the relationship between the rate of superoxide production from site I_F and NAD(P)H reduction state, obtained by combining panels (a) and (b). Filled symbols: control mitochondria; open symbols: CDNB-treated mitochondria (underlying data for CDNB-treated mitochondria is not shown). Where not visible, error bars are contained within the points. Lines show exponential relationships (for simplicity), fitted by non-linear regression to give the parameter values in Eq. 2. See MATERIALS AND METHODS. Data are means ± SEM (n = 6).

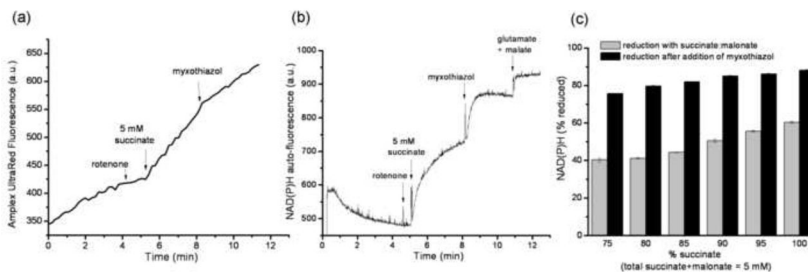


FIGURE 6. Contribution of site I_F during calibration of site III_{Q0}.

For the calibration curve in Fig. 7c, the rate of superoxide production by site III_{Q0} was measured as myxothiazol-sensitive H₂O₂ production at different succinate:malonate ratios in the presence of rotenone. (a) Typical Amplex UltraRed fluorescence trace. 2 μM myxothiazol was added where indicated. (b) Corresponding NAD(P)H autofluorescence trace indicating that NAD(P)H becomes reduced under these conditions (5 mM glutamate plus 5 mM malate were added at the end to establish 100% reduction of the NAD(P)H pool. Therefore, correction for the changes in superoxide production from site I_F before and after addition of myxothiazol was required. (c) The mean ± SEM reduction state of NAD(P)H at each succinate:malonate ratio in the absence and presence of 2 μM myxothiazol used to generate the correction using the I_F calibration curve in Fig. 5c (n = 3).

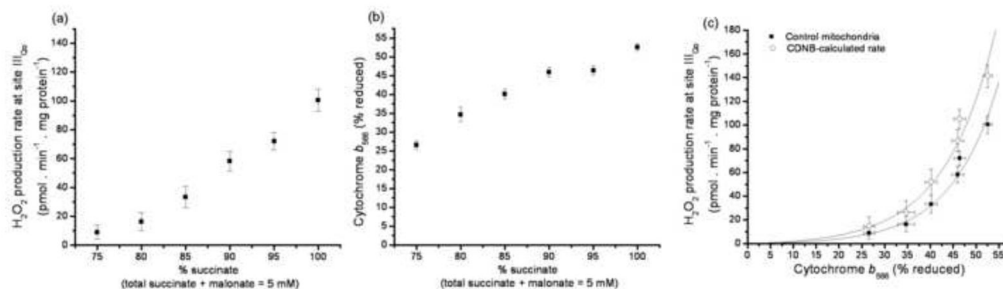


FIGURE 7. Relationship between the rate of superoxide production by site III_{Q0} and the reduction state of cytochrome b_{566}

(a) Dependence of the rate of myxothiazol-sensitive H₂O₂ production on succinate concentration at fixed succinate+malonate concentration in the presence of rotenone. Data were corrected for the contribution of site I_F (Fig. 6c and Fig. 5c). (b) Dependence of cytochrome b_{566} reduction on succinate concentration in parallel incubations (100% reduction was subsequently established by addition of 2 μM Antimycin A). (c) Final calibration of the relationship between the rate of superoxide production from site III_{Q0} and cytochrome b_{566} reduction state, obtained by combining panels (a) and (b). Filled symbols: control mitochondria; open symbols: control values after correction to CDNB-treated mitochondria using Eq. 1, assuming that 50% of superoxide from site III_{Q0} was produced in the matrix (see [26]). Where not visible, error bars are contained within the points. Lines show exponential relationships (for simplicity), fitted by non-linear regression to give the parameter values in Eq. 3. See MATERIALS AND METHODS. Data are means ± SEM (n = 9).

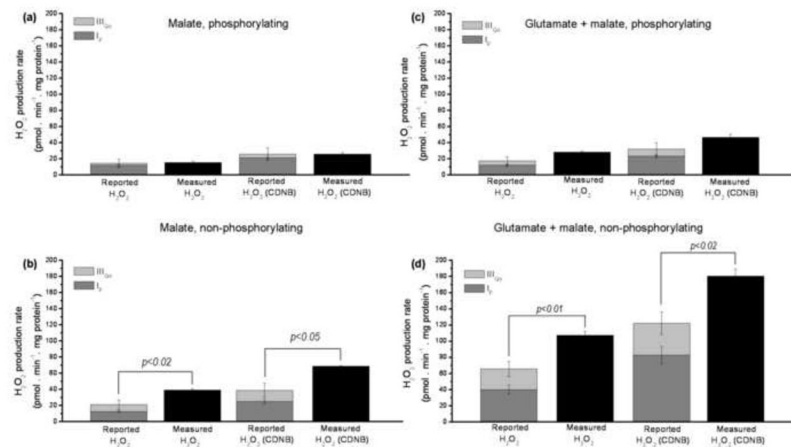


FIGURE 8. Reported and measured native rates of superoxide and H₂O₂ production by mitochondria oxidizing NAD-linked substrates in the absence of electron transport chain inhibitors

Data from Table 2. (a) 5 mM malate as substrate under phosphorylating conditions. (b) 5 mM malate under non-phosphorylating conditions. (c) 5 mM glutamate + 5 mM malate under phosphorylating conditions. (d) 5 mM glutamate + 5 mM malate under non-phosphorylating conditions. The two left-hand bars in each panel show results for control mitochondria; the two right hand bars show results for CDNB-treated mitochondria. Reported rates from site I_F are in dark grey; those from site III_Q₀ are in light grey. Black bars represent measured rates. Data are means ± SEM (n = 6). SEM values for reported rates were determined by error propagation; significance was tested using Welch's t-test, see MATERIALS AND METHODS.

TABLE 1

Reduction level of the reporters for sites I_F and III_{Q_o} in four different experimental conditions. The redox states of NAD(P)H and cytochrome *b*₅₆₆ were determined in parallel as described in MATERIALS AND METHODS.

Substrate, experimental condition	I _F reporter (%reduced NAD(P)H)	III _{Q_o} reporter (%reduced cytochrome <i>b</i> ₅₆₆)
5 mM malate, phosphorylating	20±1	16±2
5 mM malate, non-phosphorylating	30±2	27±7
5 mM glutamate + 5 mM malate, phosphorylating	26±1	23±3
5 mM glutamate + 5 mM malate, non-phosphorylating	85±7	38±5

Values are means ± SEM, n = 6.

TABLE 2

Reported and measured native rates of superoxide and H₂O₂ production by mitochondria oxidizing NAD-linked substrates in the absence of electron transport chain inhibitors. Reported rates were calculated from the data in Table 1 using the calibration curves for control or CDNB-treated mitochondria as appropriate in Fig. 5c for site I_F and Fig. 7c for site III_{Q₀}. Measured rates refer to control or CDNB-treated mitochondria as shown.

	Rate of superoxide or H ₂ O ₂ production (pmol H ₂ O ₂ ·min ⁻¹ ·mg protein ⁻¹)							
	Estimated rate from site I _F		Estimated rate from site III _{Q₀}		Sum of estimated rates		Total measured rate	
	Control	CDNB	Control	CDNB	Control	CDNB	Control	CDNB
5 mM malate, phosphorylating	12 ± 1	22 ± 2	3 ± 5	4 ± 8	15 ± 5	26 ± 8	16 ± 2	26 ± 2
5 mM malate, non-phosphorylating	13 ± 1	25 ± 2	9 ± 6	14 ± 9	21 ± 6	39 ± 9	39 ^{**} ± 2	69 [*] ± 1
5 mM glutamate + 5 mM malate, phosphorylating	12 ± 1	23 ± 2	5 ± 5	9 ± 8	18 ± 5	32 ± 8	29 ± 1	47 ± 4
5 mM glutamate + 5 mM malate, non-phosphorylating	40 ± 6	83 ± 11	26 ± 9	40 ± 14	66 ± 11	123 ± 17	108 ^{***} ± 5	182 ^{***} ± 9

Values are means ± SEM, n = 6 for control conditions, and n = 4 for CDNB. SEM values for reported rates were determined by error propagation and significance was tested using Welch's t-test, see MATERIALS AND METHODS.

* p<0.05,

** p<0.02,

*** p<0.01 compared to the appropriate sum of reported rates.