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## Determinants of Ligand Affinity and Heme Reactivity in H-NOX Domains\*\*

Emily E. Weinert<sup>1</sup>, Lars Plate<sup>2</sup>, Charlotte A. Whited, Charles Olea Jr.<sup>2</sup>, and Michael A. Marletta<sup>1,2,3,4</sup>

Michael A. Marletta: marletta@berkeley.edu

<sup>1</sup>California Institute for Quantitative Biosciences, University of California, Berkeley, CA 94720

<sup>2</sup>Department of Molecular and Cell Biology, University of California, Berkeley, CA 94720

<sup>3</sup>Department of Chemistry, University of California, Berkeley, CA 94720

<sup>4</sup>Division of Physical Biosciences, Lawrence Berkeley National Laboratory, Berkeley, CA 94720

Department of Chemistry, California Institute of Technology, Pasadena, CA 91125

Investigations into the mechanisms by which heme proteins control ligand affinity and reactivity have been studied for decades, with the globins serving as model systems for histidyl-ligated proteins.<sup>[1]</sup> The recent discovery of a novel family of heme proteins, Heme Nitric oxide/OXygen (H-NOX) binding domains,<sup>[2–4]</sup> has provided an opportunity to investigate if factors previously found to control ligand affinity and reactivity in the globins can be generalized between protein folds or if there are additional determinants.

The H-NOX family includes the heme domain of soluble guanylate cyclase (sGC), the mammalian NO receptor,<sup>[5]</sup> as well as heme proteins from bacteria, such as *Thermoanaerobacter tengcongensis* (*Tt*).<sup>[3]</sup> H-NOX domains are a potentially illuminating choice for further investigations into ligand affinity because sGC does not bind O<sub>2</sub>, even under ambient conditions, whereas other members of the family, such as *Tt*H-NOX, bind O<sub>2</sub> with very high affinity ( $K_d = 90$  nM).<sup>[3, 6]</sup> A distinctive feature of the wild type (WT) *Tt* H-NOX crystal structure was a distal pocket hydrogen bonding network, where a key H-bond of Y140 to the bound O<sub>2</sub> was observed and the H-bonds of N74 and W9 to Y140.<sup>[4]</sup> This hydrogen bonding triad is absent in non-O<sub>2</sub> binding H-NOX domains, highlighting hydrogen bonding as a major component of O<sub>2</sub> binding. A distal H-bond to O<sub>2</sub> has been noted as important for the globins. However, introduction of a tyrosine into sGC does not result in O<sub>2</sub> binding,<sup>[7]</sup> suggesting that there are additional determinants of ligand affinity in H-NOX domains.

To further investigate how the H-NOX fold regulates ligand binding, and how this regulation is similar or dissimilar to that of the globins, site-directed mutagenesis was performed on *Tt*H-NOX. It has been found that the introduction of phenylalanine mutations into the distal pocket of myoglobin (Mb), at positions such as L29 and V68, dramatically alters the oxygen affinity by decreasing off-heme ligand binding sites and reducing ligand

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Correspondence to: Michael A. Marletta, marletta@berkeley.edu.

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"trapping" (Table 1).<sup>[8–10]</sup> In addition to altering O<sub>2</sub> affinity, reaction of ferrous-oxy proteins with NO, or NO dioxygenation (Scheme 1), in Mb was decreased by the addition of distal pocket bulk (Table 1) due to the blocking off-heme binding sites, preventing NO from readily residing in the heme pocket near the bound O<sub>2</sub>.<sup>[8, 11]</sup> Therefore, to probe the effects of distal pocket bulk on O<sub>2</sub> affinity and NO reactivity of H-NOX proteins, as compared to the globins, isoleucine 75 and leucine 144 were mutated to phenylalanine since both residues are in the back of the distal pocket (Figure 1), analogous to positions L29 and V68 of Mb.

The I75F, L144F, and I75F/L144F mutants were spectrally similar to WT *Tt*H-NOX (Figure S1), however, kinetic characterization revealed dramatic changes upon the introduction of distal pocket bulk. The I75F mutation resulted in an approximately 9-fold increase in the O<sub>2</sub> off-rate, while the L144F mutation resulted in an approximately 13-fold increase in the O<sub>2</sub> off-rate (11.1 9 s<sup>-1</sup> and 16.06 s<sup>-1</sup>, respectively, Table 1). The two mutations had a synergistic effect in the I75F/L144F double mutant, resulting in an O<sub>2</sub> off-rate of 45.7 s<sup>-1</sup>. This represents a 37-fold increase in the O<sub>2</sub> off-rate from the introduction of only two mutations, neither of which involved a direct change in the hydrogen bonding network. Typically such large changes in O<sub>2</sub> off-rates occur when the primary hydrogen bond donor to the O<sub>2</sub> is altered or removed.<sup>[9]</sup>

There were also significant changes in the O<sub>2</sub> on-rate upon introduction of the mutations (Table 1), which, combined with the changes in O<sub>2</sub> off-rates, resulted in large decreases in the dissociation constants (*K<sub>d</sub>*) for O<sub>2</sub>. WT *Tt*H-NOX has an O<sub>2</sub> *K<sub>d</sub>* of ~90 nM, while the I75F mutant was found to have a *K<sub>d</sub>* of 497 nM. The increases in *K<sub>d</sub>* were even more dramatic for the L144F and I75F/L144F mutants (2.36 μM and 11.15 μM, respectively). These represent 6, 27, and 126-fold increases over WT for the I75F, L144F, and I75F/L144F mutations, respectively. To date, these are the largest changes in O<sub>2</sub> affinity for *Tt*H-NOX mutants that stably bind O<sub>2</sub>. Similar mutations in Mb actually resulted in significantly decreased O<sub>2</sub> dissociation rates and very tight binding constants,<sup>[8]</sup> suggesting that the introduction of distal pocket bulk in *Tt* results in different changes to the protein structure as compared to Mb.

To understand the cause of the changes in O<sub>2</sub> affinity upon introduction of distal pocket bulk, the structure of *Tt*I75F/L144F was solved (PDB ID 3IQB). In Mb, the mutation of distal pocket residues to phenylalanine resulted in the increased bulk filling cavities in the heme pocket.<sup>[11]</sup> However, introduction of phenylalanines into *Tt*H-NOX rearranged the heme pocket, resulting in a more elongated protein structure (~2.5 Å increase in distance from α-helix F to α-helix B, Figure 2A). While an extremely distorted heme was observed in WT *Tt*H-NOX, the I75F/L144F heme was found to be much flatter (rmsd from planarity of 0.45 vs. 0.20 Å, respectively, Figure 2B) and tilted within the pocket to accommodate the increased bulk (Figure 3A). The addition of the two phenylalanines also altered the conformation of F78, which resulted in a van der Waals contact with α-helix A and a translation of that helix away from the heme (Figure 3B). The combined changes in heme positioning and protein scaffold resulted in a longer Fe(II)-Y140 distance in I75F/L144F, as compared to WT (6.4 Å vs. 5.1 Å, Figure 2B).

Besides the changes in distal pocket hydrogen bonding, there were also differences in the proximal pocket of I75F/L144F, as compared to WT *Tt*H-NOX. The proximal histidine was found to have rotated around the Fe-imidazole bond into a more eclipsed conformation (Figure 2B). An eclipsed histidine conformation has previously been shown to result in decreased O<sub>2</sub> binding energy and contribute to lower O<sub>2</sub> affinity by decreasing Fe-histidine bond strength.<sup>[12]</sup> The weaker Fe(II)-histidine bond also likely resulted in the spectrally observed mixture of 5- and 6-coordinate Fe(II)-NO for I75F/L144F. In addition, although

within coordinate error of most WT structures,<sup>[4, 13]</sup> a slight increase in the iron-histidine bond distances was observed that may contribute to the decreased O<sub>2</sub> affinity. Calculations previously done on an imidazole-ligated model system found that increasing the Fe-imidazole bond distance by as little as ~0.1 Å, which is difficult to resolve crystallographically, resulted in an approximately 3 kcal/mol decrease in binding energy.<sup>[12]</sup> The combination of changes to the distal and proximal pockets provides an explanation for the observed 126-fold decrease in O<sub>2</sub> affinity for *Tt*I75F/L144F, as compared to WT. Previous work has found that these factors are important for controlling the energetics of O<sub>2</sub> binding in Mb and leghemoglobin,<sup>[12]</sup> suggesting that these are key determinants of O<sub>2</sub> affinity in widely varying protein architectures.

The effects of distal pocket bulk on reactivity of ferrous-oxy *Tt*H-NOX were further investigated using NO as a probe. The reaction of Fe(II)-O<sub>2</sub> protein with NO, (sometimes called NO dioxygenation) (Scheme 1), appears to be an inherent property of the globins and may provide insight as to how the reactivity of the Fe-O<sub>2</sub> bond in *Tt*H-NOX has been altered by the mutations. WT *Tt*H-NOX was found to have a NO dioxygenation rate of 0.051 μM<sup>-1</sup>s<sup>-1</sup> (Table 1), which is the slowest rate found in the literature.<sup>[14, 15]</sup> The I75F, L144F, and I75F/L144F mutants were found to have rates of 0.19 μM<sup>-1</sup>s<sup>-1</sup>, 0.64 μM<sup>-1</sup>s<sup>-1</sup>, and 2.0 μM<sup>-1</sup>s<sup>-1</sup>, respectively. All of these rates are among the slowest reported.<sup>[14, 15]</sup>

It is interesting that the dioxygenation rates for the *Tt*H-NOX proteins described above are orders of magnitude slower than the globins, even though they both utilize protoporphyrin IX hemes for O<sub>2</sub> binding and have similar O<sub>2</sub> affinities. Previous work has suggested that the ferric-superoxide-like character of the Fe(II)-O<sub>2</sub> complexes and the half life of the O<sub>2</sub> on the iron, as the Fe(II)-O<sub>2</sub> protein is a required reactant, are important for controlling NO dioxygenation rates.<sup>[8, 16]</sup>

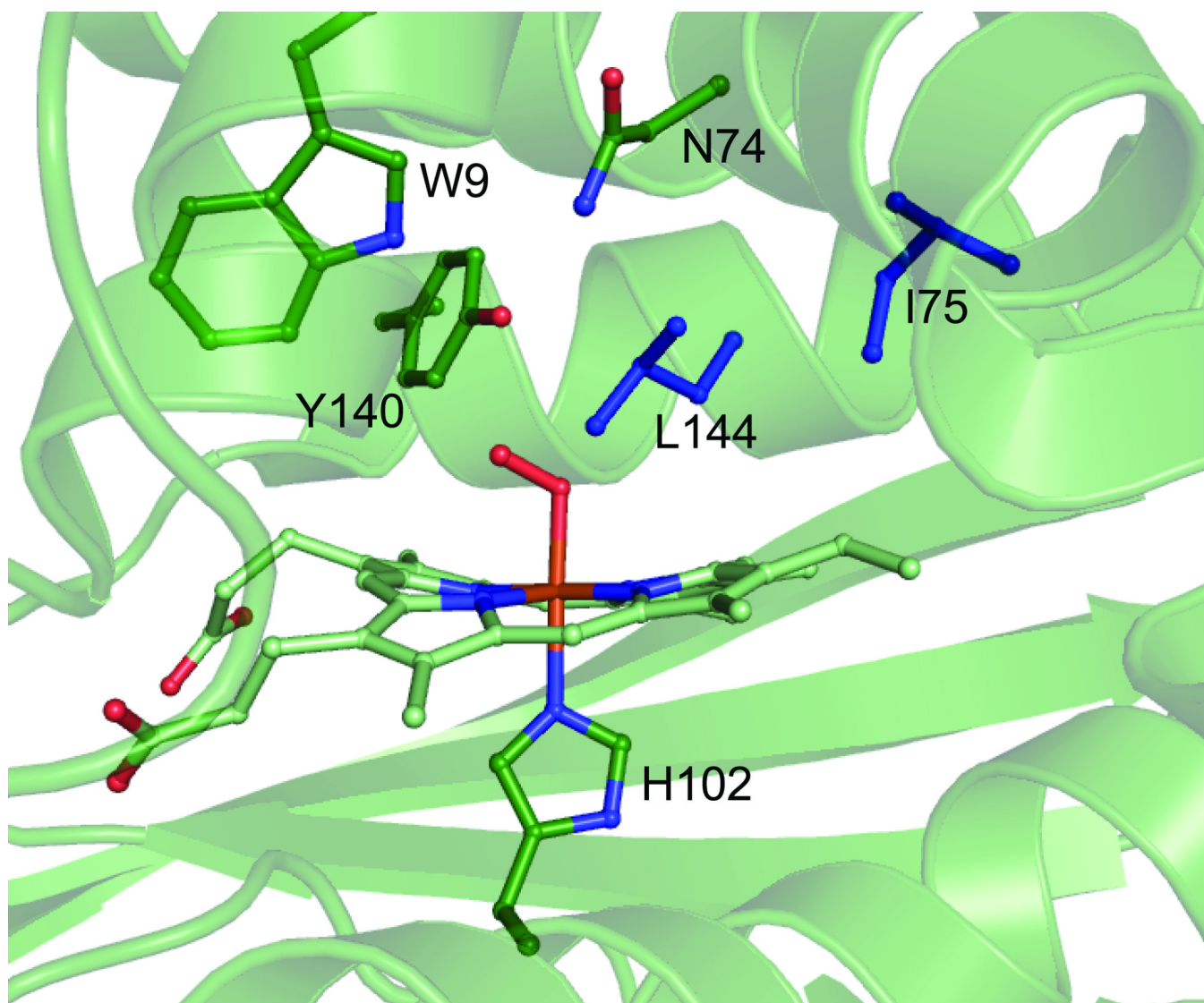
The O-O stretch, which can be measured by resonance Raman spectroscopy, provides a measurement of the bond order for the bound O<sub>2</sub>. *Tt*H-NOX has been found to have an O-O stretch of 1131 ν<sup>-1</sup> (vs. O=O stretch of 1556 ν<sup>-1</sup>),<sup>[17]</sup> which is very similar to the globins (1122–1155 ν<sup>-1</sup>) and indicates a metal-superoxide species.<sup>[18]</sup> Therefore, differences in superoxide character of the bound O<sub>2</sub> likely are not fully responsible for the dramatic differences in NO dioxygenation rates between the globins and *Tt*H-NOX. In addition, long residence time of the O<sub>2</sub> at the heme clearly does not dictate NO reactivity in this family of mutants, as WT *Tt*H-NOX shows the slowest NO dioxygenation rate while having the slowest *k*<sub>off</sub>.

Within this family of *Tt*H-NOX mutants, a possible difference in NO dioxygenation rates is access to the heme-bound O<sub>2</sub>. The I75F/L144F structure has a small cavity near the bound O<sub>2</sub> that could be occupied by NO, while no cavities can be found in the WT structure. Studies with Mb have shown that blocking cavities in the distal pocket can decrease NO reactivity, suggesting that the reverse may be happening upon introduction of distal pocket bulk in *Tt*H-NOX. In addition, previous work has found that hydrogen bonding of the bound O<sub>2</sub> by a tyrosine decreases the NO reactivity of a protein,<sup>[15]</sup> likely by changing the electronics of the heme iron. WT *Tt*H-NOX, with a hydrogen bonding tyrosine, has been found to have a reduction potential of 167 mV vs. the standard hydrogen electrode (SHE),<sup>[19]</sup> as compared to a reduction potential of 59 mV vs. SHE for Mb,<sup>[20]</sup> which has a hydrogen bonding histidine. The more difficult one electron oxidation of WT *Tt*H-NOX likely makes formation of the ferric product of NO dioxygenation more unfavorable. Investigations into the heme electronics of the I75F, L144F, and I75F/L144F *Tt*H-NOX distal pocket mutants are currently underway.

The results reported here show that distal pocket bulk significantly decreases *Tt*H-NOX O<sub>2</sub> affinity. However, in contrast to Mb, the addition of distal pocket bulk results in a more elongated structure and changes to the Y140-Fe bond distance and histidine orientation. The dramatic decrease in O<sub>2</sub> affinity in *Tt*I75F/L144F suggests that these structural features may provide a general method of controlling O<sub>2</sub> affinity.

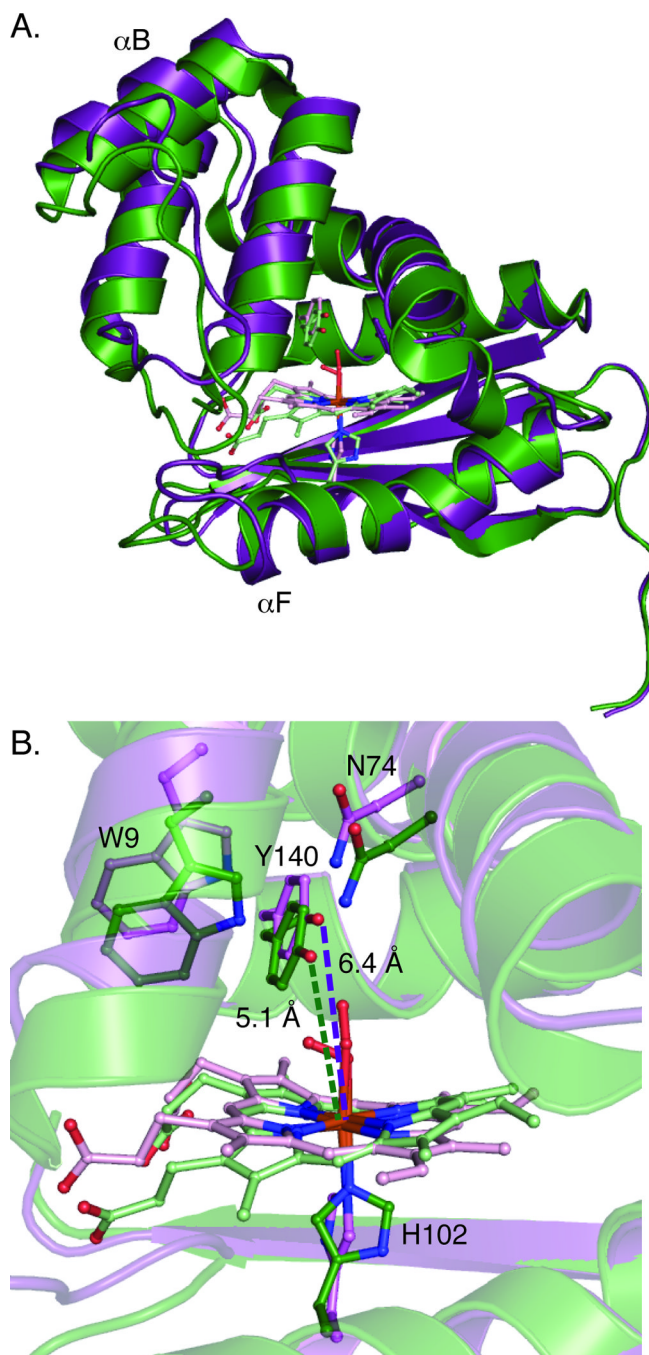
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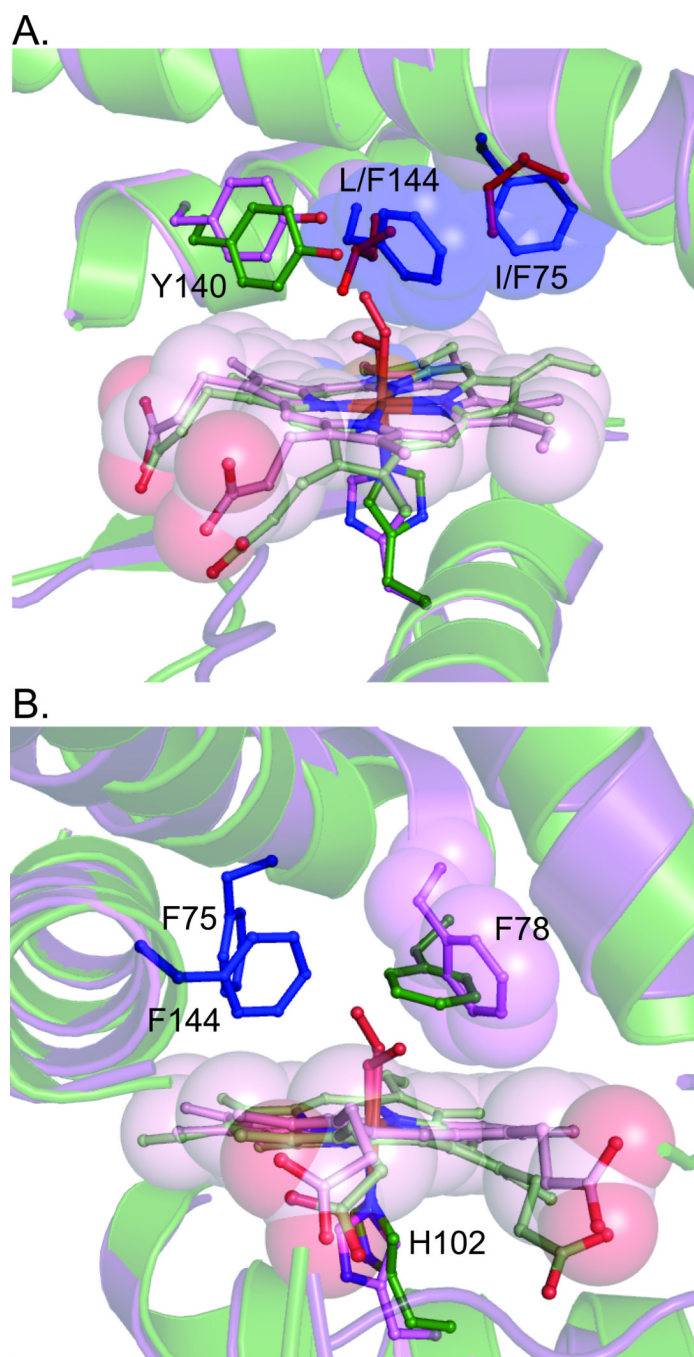
**Figure 1.** Structure of WT *TrH*-NOX heme pocket.<sup>[4]</sup> The heme and key residues are shown in green. L144 and I75 are highlighted in blue.



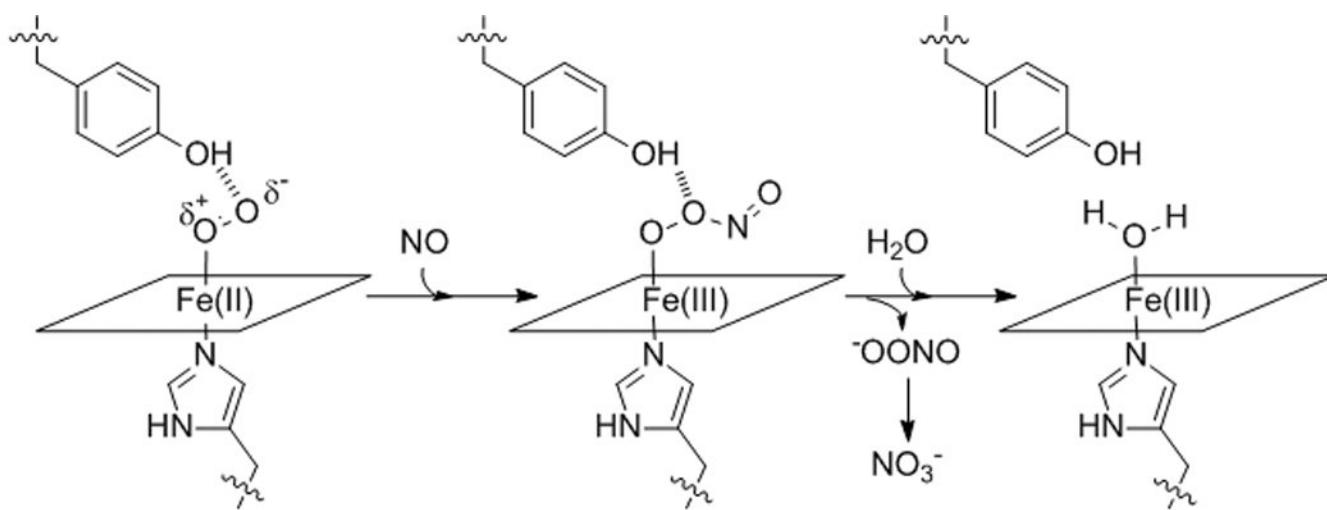


**Figure 2.**

**A.** Overlay of WT *Tt* (green) and I75F/L144F (purple) structures. The structures have been aligned along the C-terminal  $\beta$ -sheets since this alignment resulted in the hemes being approximately overlaid. A calculated distance difference matrix (Figure S2) shows that the C-terminal  $\beta$ -sheets change very little between the structures. **B.** Heme environments of WT and I75F/L144F highlighting changes in inter-atom distances and heme planarity and tilt.



**Figure 3.** Heme environment of the I75F/L144F mutant (purple) as compared to WT *Tt*H-NOX (green). **A.** Arrangement of F75 and F144 (blue) in the distal pocket. **B.** Changes in positioning of adjacent residue F78 upon introduction of the phenylalanine mutations.



**Scheme 1.**  
NO dioxygenation reaction.



Table 1

Kinetic parameters for TtH-NOX and sperm whale myoglobin.

Protein	$K_d$ (nM)	$k_{off}$ ( $s^{-1}$ )	$k_{on}$ ( $\mu M^{-1} s^{-1}$ )	$k_{ox}$ ( $s^{-1}$ , $\times 10^{-5}$ )	$k'_{NO,ox}$ ( $\mu M^{-1} s^{-1}$ )
Tt wt	88.2 $\pm$ 0.67	1.20 $\pm$ 0.02	13.6 $\pm$ 1.0	stable	0.051 $\pm$ 0.002
Tt I75F	497 $\pm$ 16	11.19 $\pm$ 0.12	22.5 $\pm$ 0.7	stable	0.19 $\pm$ 0.01
Tt L144F	2360 $\pm$ 50	16.06 $\pm$ 0.21	6.8 $\pm$ 0.1	1.67 $\pm$ 0.03	0.64 $\pm$ 0.02
Tt I75F/L144F	11150 $\pm$ 330	45.7 $\pm$ 0.9	4.1 $\pm$ 0.1	17.5 $\pm$ 1.7	2.0 $\pm$ 0.1
Mb	910 <sup>a</sup>	15 <sup>a</sup>	17 <sup>a</sup>	1.53 <sup>a</sup>	34 <sup>a</sup>
Mb V68F	2100 <sup>a</sup>	2.5 <sup>b</sup>	1.2 <sup>b</sup>	1.94 <sup>a</sup>	9.4 <sup>a</sup>
Mb L29F	67 <sup>b</sup>	1.4 <sup>b</sup>	21 <sup>b</sup>	0.014 <sup>b</sup>	8.1 <sup>a</sup>
Mb L29F/V68F	13 <sup>a</sup>	0.17 <sup>a</sup>	13 <sup>a</sup>		2.9 <sup>a</sup>

<sup>a</sup>[9]

<sup>b</sup>[10]