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Chiral Anion Dependent Inversion of Diastereo- and Enantioselectivity in Carbonyl Crotylation *via* Ruthenium Catalyzed Butadiene Hydrohydroxyalkylation

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Abstract

The ruthenium catalyst generated *in situ* from $H_2Ru(CO)(PPh_3)_3$, (*S*)-SEGPHOS and a TADDOL-derived phosphoric acid promotes butadiene hydrohydroxyalkylation to form enantiomerically enriched products. Notably, the observed diastereo- and enantioselectivity is opposite of that observed using BINOL-derived phosphate counterions in combination with (*S*)-SEGPHOS, the same enantiomer of chiral ligand. Match-mismatch effects between chiral ligand and chiral TADDOL-phosphate counterion are described. For the first time, single crystal X-ray diffraction data is reported for a ruthenium complex modified by a chiral phosphate counterion.

In the course of developing C-C bond forming hydrogenations beyond hydroformylation, we have found that hydrogen transfer between primary alcohols and π -unsaturated reactants generates organometal-aldehyde pairs that combine to form products of carbonyl addition, enabling a departure from stoichiometric organometallic reagents.¹ During this work, iridium catalysts for enantioselective carbonyl crotylation from the alcohol oxidation level that employ α -methyl allyl acetate as a crotyl donor were developed.^{2,3} Butadiene hydrohydroxyalkylation is an alternate strategy for carbonyl crotylation. In initial studies from our laboratory, iridium and ruthenium catalysts that displayed the essential reactivity and regioselectivity were identified, but poor stereoselectivity was observed.^{4a,b} While stereoselectivity can be enforced through the use of 2-silyl-butadienes,^{4c} direct regio- and stereoselective hydrohydroxyalkylations of butadiene itself remained elusive until ruthenium catalysts modified by chiral phosphate counterions were explored.^{4d,5,6,7} Using the indicated H_8 -BINOL-derived phosphate counterion, ruthenium catalyzed butadiene hydrohydroxyalkylation with primary benzylic alcohols delivered products of crotylation with good levels of *anti*-diastereo- and enantioselectivity. To achieve stereoselectivity in corresponding reactions of primary aliphatic alcohols, chiral phosphate counterions were assayed in combination with the chiral phosphine ligands (*R*)- and (*S*)-SEGPHOS.⁸ Here, we disclose that ruthenium catalysts modified by H_8 -BINOL and TADDOL-derived phosphate counterions enforce *opposite* diastereo- and enantioselectivity even when the same enantiomer of chiral phosphine ligand is employed. Based on these findings, a protocol for *syn*-diastereo- and enantioselective carbonyl crotylation *via* butadiene hydrohydroxyalkylation was achieved (Figure 1).

To develop stereoselective ruthenium catalyzed butadiene hydrohydroxyalkylations applicable to primary aliphatic alcohols, structural classes of phosphate counterions beyond BINOL-derived systems were explored. As demonstrated previously,^{4d,9} the acid-base

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Supporting Information Available: Experimental procedures and spectral data. This material is available free of charge *via* the internet at <http://pubs.acs.org>.

reaction of $\text{H}_2\text{Ru}(\text{CO})(\text{PPh}_3)_3$ with the chiral phosphoric acid HX conveniently attaches the chiral anion to the metal center. Remarkably, upon initial evaluation of TADDOL-derived phosphoric acids **A**₁–**A**₅ in the coupling of butadiene with heptanol **1a** using DPPF, an achiral phosphine ligand, a modest preference for the *syn*-diastereomer **2a** was observed (Table 1, entries 1–5). Enantioselectivities increased with increasing size of the TADDOL aryl substituent, and using the 3,5-xylyl substituted acid **A**₅ a 78% enantiomeric excess was observed for the *syn*-diastereomer **1a** (Table 1, entry 5). These data suggested the feasibility of developing a protocol for carbonyl *syn*-crotylation *via* butadiene hydrohydroxyalkylation. Toward this end, match-mismatch effects between the chiral phosphate counterion derived from **A**₅ and the chiral phosphine ligands (*R*)- and (*S*)-SEGPHOS⁸ were explored (Table 1, entries 6 and 7). For the matched case involving the combination of acid **A**₅ and (*S*)-SEGPHOS, an 89% enantiomeric excess was observed for the *syn*-diastereomer **1a**, although *syn*-diastereoselectivity remained modest (Table 1, entry 7). However, upon use of phosphoric acids **A**₆ accompanied by further variation of solvent, concentration and temperature, the *syn*-diastereomer **1a** was generated in 82% yield, 95% enantiomeric excess and 4.4:1 *syn*-diastereoselectivity (Table 1, entries 8–12).

To evaluate scope, optimal conditions identified for the *syn*-crotylation of **1a** were applied to primary alcohols **1b**–**1j**. The desired products of *syn*-crotylation **2b**–**2j** were obtained in good yield with *syn*-diastereoselectivities ranging between 4–5:1 and uniformly high levels of enantioselectivity (Table 2). Interestingly, the observed diastereo- and enantioselectivity is opposite to that observed using BINOL-derived phosphate counterions in combination with DPPF or even (*S*)-SEGPHOS (Scheme 1). To gain insight into the origins of stereoselectivity, a crystal of the ruthenium complex modified by (*S*)-SEGPHOS and the phosphate counterion derived from the acid **A**₂ was subjected to X-ray diffraction analysis. As anticipated, the phosphate counterion exists in a *trans*-relationship with respect to the carbonyl ligand.⁹ Based on the connectivity revealed in the crystal structure and the observed stereoselectivities, a preliminary stereochemical model was formulated and reconciled with the indicated catalytic mechanism (Scheme 2). The unusual *syn*-diastereoselectivity may arise as a consequence of kinetically preferred hydrometallation of the *s-cis* conformer of butadiene to furnish the *anti*- π -crotylruthenium isomer.¹⁰ It is postulated that the steric demand of the TADDOL-based phosphate counterion retards the rate of isomerization to the *syn*- π -crotylruthenium isomer, which would mandate intervention of an even more sterically congested secondary σ -crotylruthenium species. In this way, the kinetic stereoselectivity of the hydrometallation event is preserved, and carbonyl addition occurs by way of the (*Z*)- σ -crotylruthenium haptomer by way of closed Zimmerman-Traxler type transition structure¹¹ to furnish the *syn*-diastereomer.

In summary, we report the chiral anion dependent inversion of diastereo- and enantioselectivity in butadiene hydrohydroxyalkylation to form products of carbonyl *syn*-crotylation, as well as the first X-ray crystal structure of a ruthenium complex modified by a chiral phosphate counterion. These studies provide important insight into the structural and interactional features of the catalyst, which should accelerate the design of improved second generation protocols. More broadly, the merged redox-construction events described herein minimize the degree of separation between reagent and feedstock, and represent a departure from premetallated reagents in chemistry carbonyl addition; a cornerstone of synthetic organic chemistry.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

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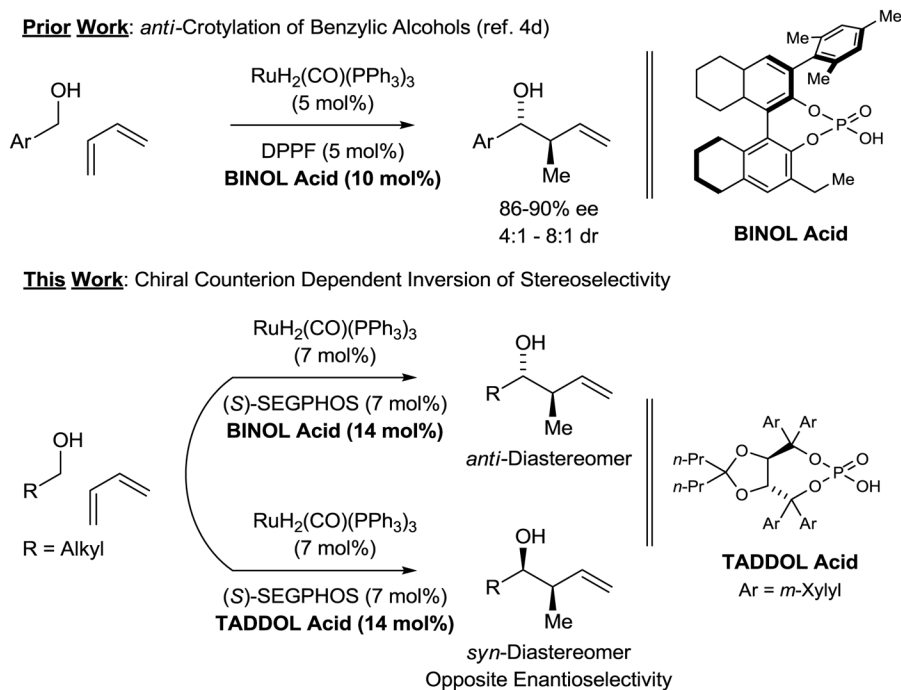


Figure 1. Ruthenium catalyzed diastereo- and enantioselective crotylation *via* butadiene hydrohydroxyalkylation.

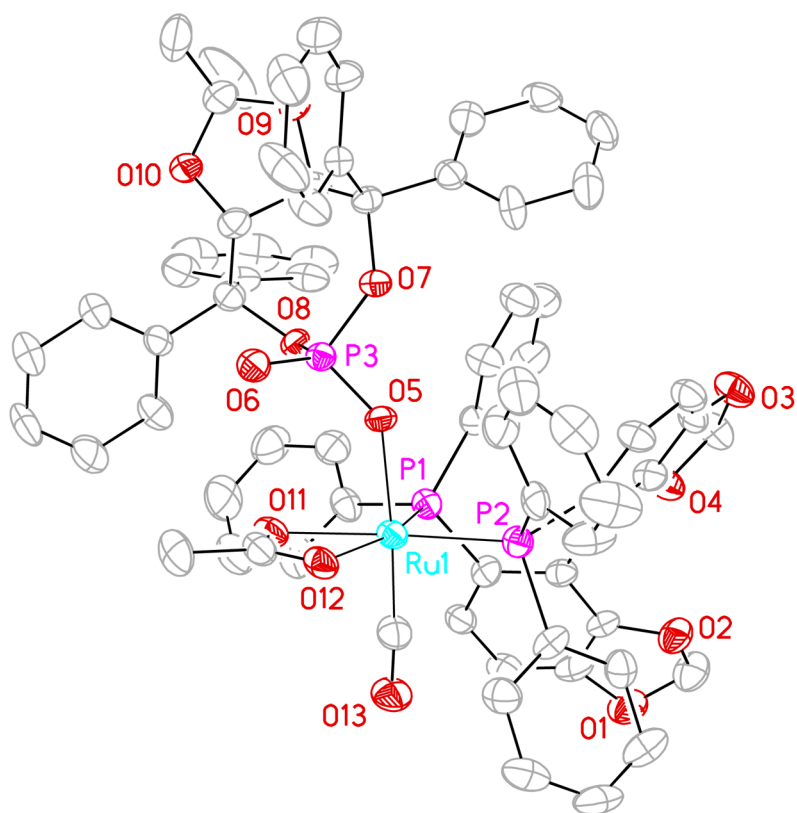
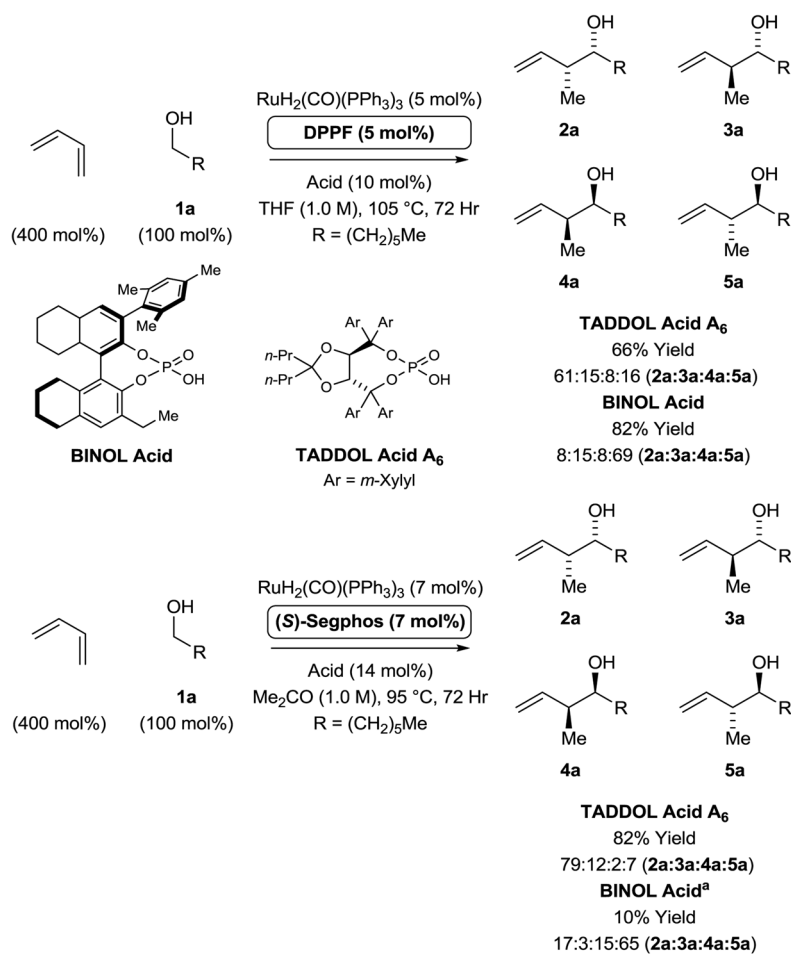
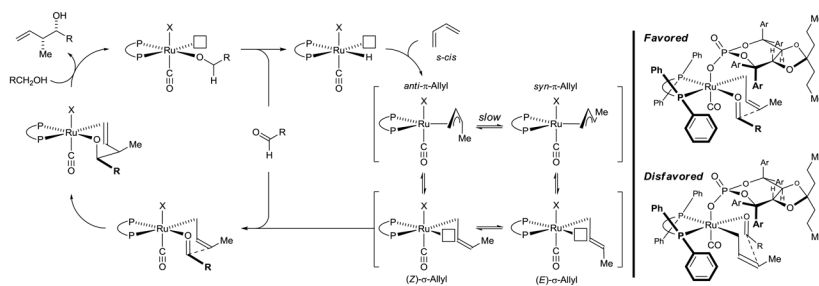


Figure 2. Structure of Ru(CO)(OAc)(A₂-phosphate)[(S)-SEGPPOS] as determined by single crystal X-ray diffraction analysis. Displacement ellipsoids are scaled to the 50% probability level.

**Scheme 1.**

Chiral phosphate counterion dependent inversion of diastereo- and enantioselectivity using the chiral catalyst modified by (*S*)-SEGPHOS.

^aReaction performed at 105 °C.



Scheme 2.
Catalytic mechanism for ruthenium catalyzed crotylation *via* butadiene hydrohydroxyalkylation.

Selected optimization experiments in the ruthenium catalyzed *syn*-diastereo- and enantioselective hydrohydroxyalkylation of butadiene with heptanol **1a**.^a

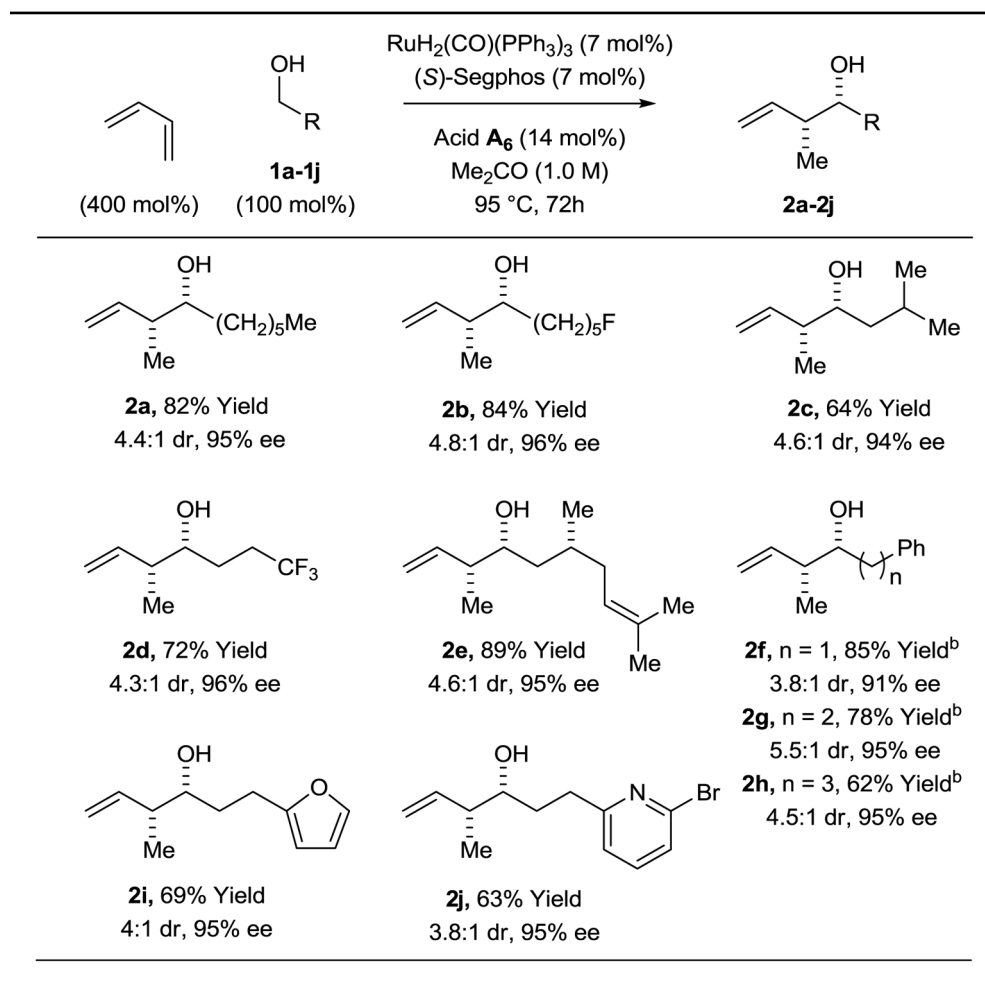
Table 1

Entry	Acid	Ligand	Ru (mol%)	T (°C)	Solvent (M)	Yield (%)	ee (%)	2a:3a
1	A ₁	DPPF	5	105	THF (2.0)	64	1	1:1
2	A ₂	DPPF	5	105	THF (2.0)	82	18	1.6:1
3	A ₃	DPPF	5	105	THF (2.0)	50	29	1.4:1
4	A ₄	DPPF	5	105	THF (2.0)	58	63	1.8:1
5	A ₅	DPPF	5	105	THF (2.0)	85	78	1.4:1
6	A ₅	(<i>R</i>)-Segphos	5	105	THF (2.0)	48	31	1.1:1
7	A ₅	(<i>S</i>)-Segphos	5	105	THF (2.0)	57	89	1.7:1
8	A ₅	(<i>S</i>)-Segphos	5	95	<i>t</i> -BuOH (2.0)	23	90	2.7:1
9	A ₅	(<i>S</i>)-Segphos	5	95	Me ₂ CO (2.0)	46	90	3.0:1
10	A ₆	(<i>S</i>)-Segphos	5	95	Me ₂ CO (2.0)	55	94	4.1:1
11	A ₆	(<i>S</i>)-Segphos	5	95	Me ₂ CO (1.0)	57	94	4.6:1
⇒12	A ₆	(<i>S</i>)-Segphos	7	95	Me ₂ CO (1.0)	82	95	4.4:1

$A_1: R^1 = Me, R^2 = Me, R^3 = Me$
 $A_2: R^1 = Me, R^2 = Me, R^3 = Ph$
 $A_3: R^1 = Me, R^2 = Me, R^3 = t-Bu$
 $A_4: R^1 = Me, R^2 = Me, R^3 = i-Pr$
 $A_5: R^1 = Me, R^2 = Me, R^3 = t-Bu$
 $A_6: R^1 = Me, R^2 = Me, R^3 = t-Bu$

^aYields are of material isolated by silica gel chromatography. DPPF = 1,1'-bis(diphenylphosphino)ferrocene; SEGPHOS = 5,5'-bis(diphenylphosphino)-4,4'-bi-1,3-benzodioxole. See Supporting Information for further details.

Table 2

Ruthenium catalyzed crotylation *via* butadiene hydrohydroxyalkylation using aliphatic alcohols **1a-1j**^a^aYields are of material isolated by silica gel chromatography. See Supporting Information for further details.^b5 mol% catalyst-ligand and 10 mol% acid.