

# Isotope-Assisted Screening for Iron-Containing Metabolites Reveals a High Degree of Diversity among Known and Unknown Siderophores Produced by *Trichoderma* spp.

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Due to low iron availability under environmental conditions, many microorganisms excrete iron-chelating agents (siderophores) to cover their iron demands. A novel screening approach for the detection of siderophores using liquid chromatography coupled to high-resolution tandem mass spectrometry was developed to study the production of extracellular siderophores of 10 wild-type *Trichoderma* strains. For annotation of siderophores, an in-house library comprising 422 known microbial siderophores was established. After 96 h of cultivation, 18 different iron chelators were detected. Four of those (dimerum acid, fusigen, coprogen, and ferricrocin) were identified by measuring authentic standards. *cis*-Fusarinine, fusarinine A and B, and des-diserylglycylferrirhodin were annotated based on high-accuracy mass spectral analysis. In total, at least 10 novel iron-containing metabolites of the hydroxamate type were found. On average *Trichoderma* spp. produced 12 to 14 siderophores, with 6 common to all species tested. The highest number (15) of siderophores was detected for the most common environmental opportunistic and strongly fungicidic species, *Trichoderma harzianum*, which, however, did not have any unique compounds. The tropical species *T. reesei* had the most distinctive pattern, producing one unique siderophore (*cis*-fusarinine) and three others that were present only in *T. harzianum* and not in other species. The diversity of siderophores produced by *Trichoderma* spp. might be the result of further modifications of the nonribosomal peptide synthetase (NRPS) products and not due to diverse NRPS-encoding genes.

ron is the fourth most abundant element in earth's crust (1). It is an essential trace mineral required for a healthy diet by almost all organisms and is involved in electron transport in metabolic processes, such as respiration and photosynthesis (2). Despite its abundance in nature, the amount of bioavailable iron is limited, as atmospheric oxygen rapidly oxidizes iron to ferric oxyhydroxides (3) that are poorly soluble under neutral to alkaline pH (4). In the case of fungi, the minimum iron requirement for optimal growth was reported to be  $10^{-5}$  to  $10^{-7}$  M (corresponding to approximately 6 to 600  $\mu$ g/liter) (5). One strategy of plants and microbes for accessing iron is the production and excretion of siderophores. Siderophores (from the Greek sideros, "iron," and phorein, "to carry something") are ferric-iron-chelating, low-molecular-mass compounds (500 to 1,500 Da) (6) that are produced under iron depletion (7). Microbial siderophores show a very high binding constant for iron ( $>10^{30}$  M, depending on the pH), enabling them to extract iron even from stainless steel (8).

Interestingly, it has been shown that monohydroxamate and dihydroxamate siderophores of fungal origin increase iron uptake by plants. It was suggested that (hydrolysis products of) fungal siderophores can play an important role in increasing iron availability for plants in soils with small amounts of accessible iron (9). Most fungal siderophores belong to the group of hydroxamate siderophores. Hydroxamate siderophores share the structural unit  $N^5$ -acyl- $N^5$ -hydroxyornithine (2) and can be divided into three distinct groups: fusarinines, coprogens, and ferrichromes. They differ in the types of building blocks ( $N^5$ -acyl groups or amino acids) and how these building blocks are connected (ester/ peptide bonds and/or head-to-head or head-to-tail order) (2, 10).

Most of the fungal siderophores contain three hydroxamate groups and form hexadentate octahedral complexes with Fe<sup>3+</sup> (6). Moreover, they are usually synthesized with the aid of nonribosomal peptide synthetases (NRPSs) and are mostly derived from L-ornithine (2). The first step in the biosynthetic pathway is hydroxylation of L-ornithine by ornithine- $N^5$ -monooxygenases to  $N^5$ -hydroxy-L-ornithine. Second, the hydroxamate group is formed by transfer of an acyl group from acyl-coenzyme A (CoA) derivatives to  $N^5$ -hydroxyornithine. This results in  $N^5$ -acyl- $N^5$ hydroxy-L-ornithine with different possible acyl groups (Fig. 1). In a third step, NRPSs covalently link these units via ester or peptide bonds to linear or cyclic oligomeric iron chelators. Last, the NRPS products can be further modified by non-NRPS enzymes to yield a diversity of different siderophores originating from a single NRPS (8).

Many species of the filamentous mycotrophic fungus *Trichoderma* (teleomorph *Hypocrea*, Ascomycota, Dikarya) can be used

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FIG 1  $N^5$ -Acyl- $N^5$ -hydroxyornithine units observed for known fungal siderophores; R can be any of the known residues shown in diagrams a to i (10). An example structure (by adding the moiety f to the  $N^5$ -hydroxyornithine unit) is shown on the right. The asterisk indicates the polar acyl or ester/peptide bond, the cleavage of which results in either the moiety shown in capital letters or the moiety shown in capital letters and "‡." The cleavage of these moieties results in characteristic siderophore fragments found in tandem mass spectrometry (see Table S2 in the supplemental material).

as agents of biological control of plant-pathogenic fungi (biocontrol) and are known siderophore producers (11). It has been shown that competition for iron plays a key role in the biocontrol exerted by Trichoderma asperellum strain T34 against Fusarium oxysporum f. sp. lycopersici (12). Dutta et al. (4) found that Botrydiplodia theobromae and Fusarium spp. produced only small amounts of siderophores compared to Trichoderma spp., indicating a more efficient way for Trichoderma to access the available iron. However, our knowledge on the diversity of siderophores produced by the different Trichoderma species is slight. Anke et al. (13) investigated the siderophore production of nine morphologically defined Trichoderma strains and detected the siderophores coprogen, coprogen B, ferricrocin, and fusigen type in the culture broths and coprogen, ferricrocin, and palmitoylcoprogen from the mycelia. Other studies additionally reported on the occurrence of cis- and trans-fusarinine, dimerum acid, ferrichrome C, fusarinine B, fusigen (fusarinine C), and  $N^{\alpha}$ -dimethylisoneocoprogen II in various Trichoderma spp. (2, 14, 15).

Recent developments in mass spectrometry (MS) instrumentation give rise to new possibilities for the analysis of small molecules (e.g., metabolites present in biological samples) and exhibit both high mass resolution and good sensitivity. The acquisition of high-resolution full-scan mass spectral data allows for novel screening strategies by retrospective data analysis (16). Therefore, it can lead to novel insights into both the metabolic inventory of living systems and the underlying biological processes. These approaches are most powerful when combined with comprehensive database queries (17). Also, the diversity of *Trichoderma* species is well understood (reference 18 and references therein), and the genome sequences of three species (*T. atroviride, T. reesei*, and *T. virens*) are available (19, 20), enabling the interpretation of metabolites on the basis of the genes present.

In this study, liquid chromatography coupled with high-resolution mass spectrometry (LC-HRMS) was used to screen automatically for iron chelators/siderophores. The established method was applied to culture broths of 10 different *Trichoderma* strains belonging to eight phylogenetic species in order to evaluate siderophore production. Moreover, interpretation of siderophore diversity and its relationship to the genetic inventory (NRPS genes) and taxonomy has been carried out for the first time.

#### MATERIALS AND METHODS

In-house siderophore library and tandem-MS (MS/MS) neutral-loss list of fungal siderophores. An in-house library of published siderophores was established. Therefore, elemental formulas and the most common ion species observed in electrospray ionization (ESI)-MS (protonated molecule, ammonium and sodium adduct) were calculated. The structures of many siderophores are available at Siderophore Base (http: //bertrandsamuel.free.fr/siderophore\_base/siderophores.php) and in an excellent review article by Hider and Kong (6).

To facilitate the interpretation of MS/MS spectra, common fragments of siderophore MS/MS spectra were collected from the literature, measurements of standard compounds, and theoretical considerations. Mass increments between two typical fragments in the MS/MS spectrum correspond to characteristic structural units of the intact siderophores (neutral losses [Fig. 1]) and were collected to form an MS/MS neutral-loss list.

**Strains and reagents.** The species and strains used in this study were selected based on their known mycoparasitic activity (I. S. Druzhinina and L. Espino de Ramer, unpublished data) (Table 1). We used the strongly opportunistic species *Hypocrea atroviridis/T. atroviride, T. asperellum, T. gamsii*, and *T. hamatum* from the section *Trichoderma; Hypocrea virens/T. virens, T. harzianum*, and the weak antagonist *T. pachybasioides/T. polysporum* from the section *Pachybasium*; and the moderate mycoparasite *H. jecorina/T. reesei* strain QM6a from the section *Longibrachiatum*. The strains are maintained at the Collection of Industrially Important Microorganisms (TUCIM) at Vienna University of Technology, Vienna, Austria. The origins of the respective strains and the GenBank accession numbers for their DNA barcodes are listed in Table 1.

Methanol (MeOH) (LiChrosolv; LC gradient grade), iron standard solution [1 g/liter Fe,  $Fe(NO_3)_3$  in 0.5 mol/l HNO\_3],  $Mg(NO_3)_2 \cdot 6H_2O$ ,  $MnSO_4 \cdot 4H_2O$ , and  $CaCl_2 \cdot 2H_2O$  were purchased from Merck (Darmstadt, Germany); acetonitrile (ACN) (HiPerSolv Chromanorm; high-performance liquid chromatography [HPLC] gradient grade), Tween 80, and

#### TABLE 1 Strains used in the study

Tayon	Codo	TUCIM	Other collection	Origin	Feelow	Distribution	Mycoparasitic	DNA barrada markar
	Code	110.	code(s)	Origin	Ecology	Distribution	activity	
Section Trichoderma Hypocrea atroviridis/ Trichoderma atroviride	at1680	1680	IMI 206040	Slovenia	Soil, dead wood	Cosmopolitan in temperate zones	Strong	http://genome.jgi-psf.org/Triat2/Triat2 .home.html
	at2108 atP1	2108 4241	P1	Hungary	Soil Hybrid strain	Strong	Strong ITS1 and -2 rRNA, GU197852	ITS1 and -2 rRNA, TUCIM unpublished
T. asperellum T. gamsii T. hamatum	asp gam ham	2128 2323 2689		Russia Italy Ethiopia	Soil Soil Rhizosphere of <i>Coffea</i> arabica	Cosmopolitan Cosmopolitan Cosmopolitan	Strong Strong Strong	<i>tef1</i> °, TUCIM unpublished <i>tef1</i> , EF488129 ITS1 and -2 rRNA, TUCIM unpublished
Section Pachybasium								
H. virens/T. virens	vir	3530	CBS 249.59, Gv29-8 <sup>a</sup>	USA	Soil	Cosmopolitan	Strong	http://genome.jgi-psf.org/TriviGv29_8_2 /TriviGv29_8_2.home.html
H. pachybasioides/T. polysporum	poly	462		Australia	Bark	Cosmopolitan, rare	Weak <sup>b</sup>	ITS1 and -2 rRNA, AY240169
T. harzianum	harz	916	CBS 226.95 <sup>a</sup>	UK	Soil	Cosmopolitan in temperate zones	Strong	tef1, AY605833
Section Longibrachiatum H. jecorina/T. reesei	rees	917	QM6a <sup>a</sup>	Solomon Islands	Military tents	Rare, tropical	Moderate	http://genome.jgi-psf.org/Trire2/Trire2. home.html

<sup>a</sup> Type strain for the species.

<sup>b</sup> I. S. Druzhinina and L. Espino de Rammer, unpublished data.

<sup>c</sup> tef1, translation elongation factor 1 alpha gene.

ethanol were purchased from VWR (Vienna, Austria); formic acid (FA) (MS grade), nitric acid TraceSelect (69%), chloramphenicol, MgSO<sub>4</sub> · 7H<sub>2</sub>O, ZnSO<sub>4</sub> · 7H<sub>2</sub>O, K<sub>2</sub>HPO<sub>4</sub>, KNO<sub>3</sub>, L-asparagine, D-glucose, CuSO<sub>4</sub> · 5H<sub>2</sub>O, and reserpine were obtained from Sigma-Aldrich (Vienna, Austria). Ultrapure water (18.2 MΩcm) was prepared successively by reverse osmosis with an ELGA Purelab Ultra-AN-MK2 system (Veolia Water, Vienna, Austria) and used throughout the study. Nitric acid was diluted 1:200 with water (0.5% [vol/vol] HNO<sub>3</sub>). Hydrochloric acid (HCl) (*pro analysi*; J. T. Baker) was purchased from Bartelt (Vienna, Austria).

Standards for siderophores (HPLC Calibration Kit Coprogens and Fusarinines and HPLC Calibration Kit Ferrichromes) were purchased from EMC (Tübingen, Germany).

**Graphite furnace atomic absorption spectroscopy (GF-AAS).** Fe determination of the cultivation medium was performed on a PerkinElmer 4100 ZL atomic absorption spectrometer equipped with a THGA graphite furnace and an AS-70 autosampler. Background correction was performed by longitudinal Zeeman effect. Platform atomization from a transversally heated pyrolytically coated graphite tube was applied. Argon was used as the purge gas, and  $Mg(NO_3)_2$  (1.5 g/liter in 0.5% HNO<sub>3</sub> [vol/vol]) was used as the matrix modifier. Twenty microliters of sample and 10 µl modifier were injected (direct injection without further pretreatment). The temperature program consisted of 100°C for 6 s, 110°C for 30 s, 130°C for 45 s, 1,300°C for 30 s, 2,100°C for 4 s, and 2,400°C for 4 s. The working range was 5 µg/liter to 50 µg/liter Fe, and the limit of detection was estimated to be 2 µg/liter Fe by measuring blanks (0.5% [vol/vol] HNO<sub>3</sub>) and low-Fe-matrix samples. Standard addition revealed no significant sensitivity change due to the matrix.

**Cultivation conditions and sampling.** All glassware was rinsed twice with 6 M HCl, followed by five times with ultrapure water prior to use in order to remove traces of iron.

The cultivation medium consisted of glucose (0.5% [mass/vol]), L-asparagine (0.5% [mass/vol]),  $K_2HPO_4$  (0.08% [mass/vol]),  $KNO_3$  (0.07% [mass/vol]),  $MgSO_4 \cdot 7H_2O$  (0.05% [mass/vol]),  $CaCl_2 \cdot 2H_2O$  (0.02% [mass/vol]),  $MnSO_4 \cdot 4H_2O$  (0.001% [mass/vol]),  $ZnSO_4 \cdot 7H_2O$  (0.001% [mass/vol]), and  $CuSO_4 \cdot 5H_2O$  (0.0005% [mass/vol]). Chloramphenicol was prepared as a 30-mg/ml stock solution in 70%

(vol/vol) ethanol and added to the liquid medium to a final concentration of 30 µg/ml to prevent bacterial contamination.

The fungal strains were pregrown on 2% (mass/vol) malt extract agar at 25°C with a 12-h light cycle. The inoculum was prepared after conidial maturation (2 to 3 days) by rolling a sterile, wetted cotton swab over conidiating areas. Conidia were suspended in sterile ultrapure water containing 0.03% (vol/vol) Tween 80 in disposable borosilicate test tubes. The concentration of the spore suspension was adjusted to a transmission of 0.31 using a Biolog turbidimeter (Biolog Inc., Hayward, CA) at an optical density at 590 nm (OD<sub>590</sub>) corresponding to  $6 \times 10^6$  conidia per ml. Then, 100 µl of the conidial suspension was dispensed into each well of 24-well plates (each well contained 2 ml of liquid medium). The microplates (Greiner, Germany) were incubated under controlled conditions (25°C; 12-h light/dark cycle). Mycelial growth was measured at OD<sub>750</sub> after 24, 48, 72, and 96 h.

Culture broth from the wells was harvested after 48, 72, and 96 h and filtered through disposable syringe filters (0.45-µm cellulose; Asahi Glass Co., Ltd., Japan). Four technical replicates (inoculated from the same spore suspension) were taken for all strains. The three strains for which genome sequences are available (*T. atroviride* IMI 206040, *T. virens* Gv29-8, and *T. reesei* QM6a) were tested in three independent biological replicates.

Liquid chromatography–high-resolution (tandem) mass spectrometry analysis. To detect the ferriforms of the siderophores, 980  $\mu$ l of filtered broth sample was mixed with 10  $\mu$ l of aqueous FeCl<sub>3</sub> solution (2% [mass/vol] FeCl<sub>3</sub>, 10% [vol/vol] FA) prior to LC-MS measurements.

Ten microliters and 20  $\mu$ l of sample (full-scan MS and MS/MS measurements, respectively) were injected into the HPLC system (Accela; Thermo Fisher Scientific, San Jose, CA) equipped with a reversed-phase Atlantis dC<sub>18</sub> analytical column, 150- by 2.1-mm inside diameter (i.d.), 3- $\mu$ m particle size (Waters, Vienna, Austria), and a C<sub>18</sub> 4- by 3-mm i.d. security cartridge (Phenomenex, Torrance, CA). The column temperature was maintained at 25°C. Eluent A was ultrapure water and eluent B was MeOH, both containing 0.1% (vol/vol) FA. For chromatographic separation, the initial mobile-phase composition (100% eluent A) was held constant for 1 min, followed by a linear gradient to 100% eluent B in



FIG 2 Work flow for the screening of siderophores by LC-HRMS(/MS).

35 min. This final condition was held for 4.5 min, followed by 4-min column reequilibration at 100% eluent A. The flow rate was 200  $\mu$ l/min.

The HPLC system was coupled to an Accela PDA (scan wavelength, 200 to 600 nm; bandwidth, 1 nm; scan rate, 20 Hz) and subsequently to an LTQ Orbitrap XL (both Thermo Fisher Scientific) equipped with an ESI interface, which was operated in positive ionization mode using the following settings: electrospray voltage, 4 kV; sheath gas, 40 arbitrary units; auxiliary gas, 5 arbitrary units; capillary temperature, 300°C. All other source parameters were automatically tuned for a maximum MS signal intensity of reserpine solution. To this end, 10  $\mu$ l/min of reserpine solution (10 mg/liter, dissolved in 8:2 [vol/vol] ACN-ultrapure water) was infused via syringe pump into the mobile phase (1:1 eluent A-eluent B) at a flow rate of 200  $\mu$ l/min.

For the FT-Orbitrap full-scan and MS/MS measurements, the automatic gain control was set to a target value of  $5 \times 10^5$ , and a maximum injection time of 500 ms was chosen.

For the initial full-scan measurements, the mass spectrometer was used with a resolving power setting of 100,000 full width at half-maximum (FWHM) (at a mass-to-charge ratio [m/z] of 400) and a scan range of m/z 200 to 2,000. For subsequent MS/MS measurements, a survey full scan was followed by two data-dependent MS/MS measurements of the 1st and 2nd most intense ions from a parent mass list (obtained by initial screening of full-scan MS measurements at a parent mass width of 20 ppm and an isolation width of 2), all with a resolving power setting of 30,000 FWHM (at m/z 400). One MS/MS measurement was done using collision-induced dissociation (CID) (35% normalized collision energy), and another MS/MS measurement was done using higher-energy collision dissociation (HCD) (38% normalized collision energy). Data were generated using Xcalibur 2.1.0 (Thermo Fisher Scientific). Samples were measured in a randomized manner.

**Evaluation of LC-MS**(/**MS**) **data.** The analytical workflow is shown in Fig. 2. In the first step, raw data were automatically screened for the natural iron isotopic pattern in order to detect (putative novel) iron chelators. Therefore, the mass spectra were investigated; to be regarded as a possible iron chelator, a principal ion containing <sup>56</sup>Fe had to be accompanied by the corresponding <sup>54</sup>Fe isotopic signal at -1.99533 m/z with a maximum relative mass deviation of  $\pm 5 ppm$  (from the calculated mass of the <sup>54</sup>Fe signal) in the same spectrum. Also, the relative intensity of the <sup>54</sup>Fe-containing from the <sup>56</sup>Fe-containing ion. A minimum intensity of 10,000 counts was required for the monoisotopic ion.

The extracted ion chromatograms (EICs) of possible identified iron chelators (<sup>56</sup>Fe- and <sup>54</sup>Fe-containing ion species) were extracted from raw data, and the Pearson's correlation coefficients of those EICs were calculated; as a threshold, a correlation coefficient of 0.75 had to be exceeded. To achieve peak annotation, the *m*/*z* values obtained were compared with the calculated *m*/*z* values of the protonated molecule, the ammonium and the sodium adduct of siderophores present in the in-house siderophore library (maximum relative mass deviation,  $\pm 5$  ppm). In order to relate the peaks found to each other, for all putative iron chelators, the exact masses of ammonium and sodium adducts of each putative ion chelator were calculated and searched for in the spectra (maximum relative mass deviation,  $\pm 5$  ppm). Additionally, manual curation and annotation of uncommon adducts and in-source fragments were performed.

For *m/z* values of iron chelators found via the screening approach, an Xcalibur processing method (Genesis peak detection algorithm) was generated to determine peak areas in the Xcalibur Quan Browser (Thermo Fisher Scientific). In this way, it was also possible to find some of the iron chelators in samples where the automated screening originally had failed to detect them (strict criteria).

The putative siderophore m/z values that were detected in all replicates were included in a parent mass list for further MS/MS investigation. If available, MS/MS spectra were compared with the MS/MS spectra of authentic standard compounds in order to identify known siderophores. If they were not available, further confidence criteria were applied to annotated siderophores (an m/z value was present in the in-house library, but no commercial standard was available) and "putative novel siderophores" (an m/z was not present in the in-house library). UV/visible (UV/Vis) chromatograms were evaluated at 420 to 450 nm, since most siderophores show absorption maxima between 420 nm and 450 nm (21). Between two full-scan measurements (first time, 10-µl injection volume; second time, 20-µl injection volume), the peak heights had to increase by approximately a factor of 2 in order to ensure that it was a reliable signal. Additionally, m/z values that also occurred in the blank control (culture medium plus ultrapure water, Tween 80, and aqueous FeCl<sub>3</sub> solution) were eliminated and not further considered. Finally, MS/MS spectra were searched for characteristic mass increments from the MS/MS neutral-loss list of fungal siderophores (see Table S2 in the supplemental material). First, the parent masses that were selected for further MS/MS investigation were tested if they exhibited the natural iron isotopic pattern. Base peaks of MS/MS spectra had to show a minimum intensity of 100 counts. Only fragment ions with a minimum relative intensity of 15% relative to the base peak were considered. For each fragment ion observed in an MS/MS spectrum, potential masses of related fragments were calculated based on the MS/MS neutral-loss list. The corresponding calculated m/zvalues were compared to the observed ions in a spectrum. To be correctly assigned, a maximum relative mass deviation of  $\pm 10$  ppm had to be achieved.

Fungal siderophores mostly possess a mass of  $\geq$ 500 Da. Therefore, iron chelators found with our screening approach exhibiting a molecular mass below 500 Da were excluded from this study, since they might be the result of degradation processes.

Genome-wide screening for siderophore synthetase genes in Trichoderma. The publicly available genome databases (JGI; http: //genome.jgi-psf.org) for T. atroviride, T. reesei, and T. virens were screened for NRPS genes that might be involved in the biosynthesis of siderophores. A similarity search (BLASTP) of all NRPS protein sequences found in Trichoderma genomes (19) as queries against all available fungal genomes revealed 353 homolog NRPS protein sequences. The multiple-sequence alignment was assembled in ClustalX (22), and the maximum-parsimony phylograms were constructed as implemented in PAUP\*4.0b10 (heuristic search with 1,000 replicates; Maxtrees in effect). The resulting tree was screened for annotated siderophore synthetases. All orthologous NRPSs that are putatively involved in siderophore biosynthesis, as well as other orthologous genes found in the literature that contribute to siderophore synthesis, were independently analyzed by similarity searchs (BLASTP) of their protein sequences in the NCBI database (http://blast.ncbi.nlm.nih.gov/Blast.cgi).

#### RESULTS

In-house siderophore library and MS/MS neutral-loss list of fungal siderophores. An in-house library of microbial siderophores was generated (as a Microsoft Excel spreadsheet), including siderophores described in the literature (6, 10, 23–40). The library contains the elemental formulas and theoretical masses of the protonated molecules and the ammonium and sodium adducts of ferriforms of published siderophores. In bi- and tetradentate iron chelators, the protonated molecule corresponds to the dimer  $[Fe^{3+}SID_{2}-3H^{+}+H]^{+}$  and in tetradentate and hexadentate chelators to  $[Fe^{3+}SID-3H^{+}+H]^{+}$ . Currently, the in-house library consists of 422 entries of bacterial (ca. 90%) and fungal (ca. 10%) siderophores and is freely available on request from the laboratory of the corresponding author.

A list of m/z values originating from neutral losses caused by



**FIG 3** Growth of *Trichoderma* strains used in this study measured as mycelial density at  $OD_{750}$ . The strains were cultivated with supplementation by asparagine and zinc sulfate. The error bars show standard deviations; n = 4. ham, *T. hamatum*; asp, *T. asperellum*; rees, *T. reesei*; at1680, *T. atroviride* IMI 206040; at2108, *T. atroviride* TUCIM 2108; atP1, *T. atroviride* P1; harz, *T. harzianum*; poly, *T. polysporum*; gam, *T. gamsii*.

MS/MS fragmentation of known fungal siderophores was generated based on the literature (41–44) and MS/MS measurements of authentic siderophore standards (see Table S2 in the supplemental material). Additionally, the potential mass increments originating from neutral losses after rupture of the acyl bonds of  $N^5$ -acyl- $N^5$ hydroxyornithine units or the ester/peptide bonds between such building blocks (for known acyl groups a to i [10]) (Fig. 1) were added to the list of neutral losses. Not all of those neutral losses have been reported in the literature. Nevertheless, in analogy to earlier reports, fragmentation at these bonds can be expected due to the high polarity of the acyl bond (resulting in mass increments indicated as capital letters in Fig. 1). Moreover, the cleavage of an ester/peptide bond results in mass increments that are indicated as capital letters and "‡" in Fig. 1.

Trichoderma spp. are able to grow under conditions of Fe depletion. We used a medium with neutral pH containing extra asparagine and zinc sulfate, because these conditions have been reported to be optimal for siderophore production (4, 13, 45). Chloramphenicol was added to the cultivation medium to prevent false-positive findings due to bacterial growth and thereby formation of bacterial siderophores. Prior to cultivation, growth media were measured by GF-AAS; the iron concentration was below 3  $\mu$ g/liter Fe (corresponding to approximately 5  $\times$  10<sup>-8</sup> M Fe, conditions of iron depletion). Under these low-iron conditions, no significant differences were detected in growth when medium with or without iron or asparagine was used; however, siderophore production was increased by asparagine and zinc sulfate supplementation (data not shown). Trichoderma strains were in the exponential growth phase at all time points when the samples were taken (Fig. 3). No conidiation was observed during the experiment.

Interestingly, under the conditions of the experiment, the 10

strains showed different growth rates, with *T. hamatum*, *T. asperellum*, and *T. reesei* producing the highest biomass after 96 h. *T. gamsii*, which is a fast-growing fungus on complex media (46), had an extended lag phase up to 72 h and started to develop mycelial growth only close to the end of experiment. At the first sampling point (48 h), only *T. hamatum* and *T. reesei* had produced considerable biomass, while all other species had only begun to grow. However, nearly equal amounts and numbers of siderophores were detected at this time point for at least six tested strains: *T. hamatum*, *T. asperellum*, and *T. virens*, seven.

Trichoderma spp. produce a high diversity of extracellular iron chelators. Figure 4 shows that in total, 18 siderophores were detected for the tested Trichoderma species when following the established workflow (Fig. 2). Examples of EIC and UV/Vis chromatrograms are shown in Fig. S1 in the supplemental material. Based on absolute measured signal intensities in the supernatants, the highest concentration of most extracellular (putative) siderophores was observed after a cultivation duration of 96 h. Therefore, these supernatants were used for MS/MS measurements. Under the tested conditions, characteristic neutral losses for hydroxamate siderophores (see Table S2 in the supplemental material) were found in the MS/MS spectra (Table 2) of all Fe-containing metabolites. For identified siderophores, the m/z in Table 2 corresponds to the protonated molecules. In cases of annotated/ putative novel siderophores, the protonated molecule was annotated based on the presence of the most common adduct ions (if present in the mass spectra).

The highest number of iron chelators, 15, was found for T. harzianum, followed by 14 in cultures of two strains of T. atroviride (IMI 206040 and P1). T. gamsii and T. reesei displayed the lowest number, with only 12 iron chelators detected, while all other strains/species exhibited 13 siderophores (Fig. 4). Dimerum acid, coprogren, and fusigen (identified) were produced by all Trichoderma strains; ferricrocin (identified) was produced by all Trichoderma strains but T. gamsii and T. polysporum and was detected at relatively low abundance compared to the other siderophores. Also, fusarinine A was produced by all strains (annotated), although it was significantly more abundant in T. harzianum and T. reesei than in all other strains. T. reesei was the only strain that produced cis-fusarinine (annotated), which occurred as both a monomer and a dimer, the latter being more abundant. The EIC at m/z 798.309 showed three distinct characteristic peaks at different retention times: 11.7 min (T. reesei and T. harzianum), 12.7 min, and 13.2 min (all strains but T. reesei). This mass signal corresponds to the protonated molecules of the known isomeric siderophores fusarinine B and des-diserylglycylferrirhodin (DDF). The MS/MS spectra of all three of these iron chelators exhibit characteristic mass increments for hydroxamate siderophores. Both, fusarinine B and DDF consist of the same structural subunits (three N<sup>5</sup>-acyl-N<sup>5</sup>-hydroxyornithine units with three "F" residues [Fig. 1]) and corresponding neutral losses ( $\Delta m/z$ , 242.127, corresponding to F‡-H<sub>2</sub>O [see Table S2 in the supplemental material]) were found in the MS/MS spectra of all three compounds. The chromatographic peak at 11.7 min exhibited extensive in-source fragmentation. Nevertheless, unambiguous assignment of the three MS/MS spectra to the two known siderophores was not possible. Moreover, seven putative novel siderophores were produced by T. atroviride strains IMI 206040 and P1 (similar patterns) and T. harzianum; six by T. asperellum,

*T. atroviride* TUCIM 2108, *T. gamsii*, *T. hamatum*, *T. polysporum*, and *T. virens* (similar patterns); and two by *T. reesei* (*m/z* 798.309 was not considered a novel siderophore). The most distinctive profile of siderophores was found in *T. reesei*, which produces one unique compound (*cis*-fusarinine) and three other iron-chelating metabolites present only in *T. harzianum* (Fig. 4). The siderophore production patterns of all other strains were relatively similar, with *T. harzianum* displaying the highest siderophore diversity. To see whether siderophore production and diversity correspond to the phylogeny of the genus, we constructed a cladogram, which shows the reasonable groupings within the genus (Fig. 4).

The three biological replicates of the sequenced *Trichoderma* strains revealed consistent qualitative siderophore profiles and only minor differences in relative siderophore abundances (see Table S1 in the supplemental material).

Although the three *T. atroviride* strains were isolated from different ecosystems and locations, they show similar siderophore production patterns. IMI 206040 and P1 produced 14 and TUCIM 2108 13 (putative) siderophores. They produced the same siderophores, the only exception being m/z 902.333 (18.9 min), which was not found in TUCIM 2108 (Fig. 4). The signal with m/z 902.333 showed very low intensity, which might explain why it was not found in the other strain.

**Siderophore production in relation to genomic inventory.** In order to understand the diversity of *Trichoderma* siderophores found in this study, we performed a phylogenetic analysis of all NRPS protein sequences found in the genomes of *T. atroviride*, *T. reesei*, and *T. virens*, as well as all their fungal orthologues obtained from the JGI genome database. Furthermore, we screened the literature for genes involved in siderophore production in other fungal genera and searched for homologs in *Trichoderma* genomes (Table 3).

Our phylogenetic analysis resulted in three clades containing orthologous genes related to siderophore biosynthesis (data not shown). (i) T. atroviride (gene ID 318290), T. reesei (gene ID 69946), and T. virens (gene ID 85582) orthologues coding for putative protein-containing domains consistent with A. fumigatus SidA (JGI genome annotation of *T. reesei*, gene ID 69946 [http: //genome.jgi-psf.org]); (ii) T. reesei and T. virens orthologues (gene ID 71005 and 70206, respectively) related to nonribosomal peptide synthetase SidD in A. fumigatus and A. clavatus; (iii) a Trichoderma homolog of nonribosomal peptide synthetase, related to NPS6 (siderophore) of Cochliobolus heterostrophus (T. atroviride, T. reesei, and T. virens; gene ID 44273, 67189, and 39887, respectively), putatively involved in fusarinine synthesis in Aspergillus nidulans, Aspergillus oryzae, and Fusarium graminearum (47, 48). Furthermore, homologs of sidL and sidC genes involved in A. fumigatus biosynthesis of ferricrocin (49) and all homologs of sidI, sidH, and sidF genes, together with orthologues *NPS6* and *sidD* stepwise involved in the biosynthesis of fusarinine C (49–51), were found in all three *Trichoderma* spp. (Table 3). However, the homolog of acetyltransferase SidG (51), which is required for conversion of fusarinine C (fusigen) into triacetylfusarinine C in A. fumigatus, was found only in T. virens.

### DISCUSSION

In this study, we established an LC-HRMS/MS screening approach to study the production of extracellular iron-containing metabolites in microbial samples.



FIG 4 Siderophore production in *Trichoderma* species detected using full-scan and MS/MS measurements (signal intensities were normalized to mycelial production): *T. atroviride* (at2108, at1680, and atP1 for TUCIM 2108, IMI 206040, and P1, respectively), *T. hamatum* (ham), *T. virens* (vir), *T. harzianum* (harz), *T. polysporum* (poly), *T. asperellum* (asp), *T. gamsii* (gam), and *T. reesei* (rees). Dimerum acid, coprogen, fusigen, and ferricrocin were identified; fusarinine A, fusarinine B/DDF, and *cis*-fusarinine were annotated (no unambiguous annotation of fusarinine B/DDF was possible). The strains with sequenced genomes are marked with asterisks. The different colors represent different concentration ranges. The vertical cladogram was obtained based on a complete linkage rule using 1- Pearson's *R* distance. The vertical lines indicate standard deviations; n = 4.

To facilitate the analysis of siderophores in microbial cultures, an in-house siderophore library was established containing 422 entries for bacterial (ca. 90%) and fungal (ca. 10%) siderophores. Fungal siderophores show higher structural similarity than bacterial siderophores, which allowed the generation of an MS/MS neutral-loss list of fungal siderophores. Subsequently, a novel screening method for siderophores was established using LC-HRMS/ MS. The screening approach was applied to investigate the

TABLE 2 Characteristic neutral losses of MS/MS spectra found for observed (putative novel) sideropho
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Name		Elemental formula(s) corresponding to experimentally	determined neutral losses in spectra <sup>n</sup>
(annotated)	m/z (retention time [min])	CID	HCD
Fusarinine A cis-Fusarinine (dimer)	$556.183^{p} (8.5) 574.193^{q} (10.0)$	$C_{6}H_{8}O_{2}^{a,b,c}, C_{6}H_{10}O_{3}^{c} C_{6}H_{8}O_{2}^{a,b,c}, C_{6}H_{10}O_{3}^{c} (2\times)$	$\begin{array}{c} C_4H_4^{\ d}, C_5H_6^{\ d}, C_6H_8O_2^{\ a,b,c}, C_{11}H_{19}N_2O_5^{\ e} \\ C_3H_5NO_3^{\ d}, C_5H_{10}N_2O_2^{\ a}, C_5H_8N_4O_3^{\ f}, \\ C_{11}H_{19}N_2O_5^{\ c}, C_{11}H_{20}N_2O_5^{\ e}, \\ C_{11}H_{21}N_2O_6^{\ e} \end{array}$
Fusarinine B/DDF <sup>o</sup>	798.309 <sup><i>q</i></sup> (11.7)	$\begin{array}{l}C_{6}H_{8}O_{2}{}^{a,b,c}\left(3\times\right),C_{6}H_{10}O_{3}{}^{c},C_{5}H_{10}N_{2}O_{2}{}^{a},\\C_{11}H_{18}N_{2}O_{4}{}^{g},C_{11}H_{20}N_{2}O_{5}{}^{e}\left(2\times\right)\end{array}$	11 21 2 0
Fusarinine B/DDF <sup>o</sup>	798.309 <sup><i>q</i></sup> (12.7)	$\begin{array}{c} C_{4}H_{4}^{\ d}\left(2\times\right), C_{6}H_{10}O_{3}^{\ c}, C_{5}H_{10}N_{2}O_{2}^{\ a}, C_{7}H_{12}N_{2}O_{3}^{\ ag}\\ (2x), C_{11}H_{18}N_{2}O_{4}^{\ g}\left(3\times\right), C_{11}H_{20}N_{2}O_{5}^{\ e} \end{array}$	$C_6H_8O_2^{a,b,c}, C_7H_{12}N_2O_3^{a,g}, C_{11}H_{18}N_2O_4^{g}, C_{11}H_{20}N_2O_5^{e}, C_{12}H_{20}N_3O_2^{e,a,c,h}$
Fusarinine B/DDF <sup>o</sup>	798.309 <sup><i>q</i></sup> (13.2)	$C_4H_4^{\ d}, C_6H_{10}O_3^{\ c}, C_{11}H_{18}N_2O_4^{\ g}, C_{11}H_{20}N_2O_5^{\ e}$	$C_2H_2O^{c,i}, C_9H_{16}N_2O_4^{e}$
Unknown	$510.175^{p}$ (10.7)	$C_6 H_8 O_2^{a,b,c} (2 \times), C_{12} H_{19} N O_4^{b}$	$C_{11}H_{17}N_2O_5^{e}, C_{11}H_{19}N_2O_5^{e}$
Unknown	$555.151^{p}$ (11.5)	$C_6 H_8 O_2^{a,b,c}$	$C_5H_6^d$
Unknown Unknown	651.220 <sup><i>p</i></sup> (11.8) 651.220 <sup><i>p</i></sup> (12.1)	$ \begin{array}{c} C_{2}H_{4}O^{\tilde{b}}, C_{4}H_{4}^{d}, C_{3}H_{5}O^{i}, C_{5}H_{6}^{d}, C_{6}H_{8}O_{2}^{a,b,c}\left(2\times\right), \\ C_{4}H_{7}NO_{3}^{d}, C_{5}H_{10}N_{2}O_{2}^{a}, C_{5}H_{9}NO_{3}^{j}, \\ C_{9}H_{17}N_{2}O_{3}^{j,k}\left(2\times\right), C_{8}H_{12}N_{2}O_{5}^{e}, C_{9}H_{17}N_{2}O_{4}^{e}, \\ C_{11}H_{17}N_{2}O_{5}^{e}, (2\times), C_{11}H_{18}N_{2}O_{5}^{e}, C_{11}H_{19}N_{2}O_{5}^{e}, \\ C_{11}H_{17}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{18}N_{2}O_{5}^{e}, C_{11}H_{19}N_{2}O_{5}^{e}, \\ (3\times), C_{11}H_{20}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{21}N_{2}O_{5}^{e}\left(2\times\right), \\ C_{11}H_{17}N_{2}O_{6}^{e}, C_{11}H_{19}N_{2}O_{6}^{e}, C_{11}H_{21}N_{2}O_{6}^{e}, \\ C_{2}H_{2}O^{c,i}\left(2\times\right), C_{4}H_{4}^{d}, C_{5}H_{6}^{d}\left(2\times\right), C_{6}H_{8}O_{2}^{a,b,c}, \\ (2\times), C_{6}H_{11}N_{2}O_{4}^{a,l}, C_{9}H_{17}N_{2}O_{3}^{j,k}\left(2\times\right), \\ C_{11}H_{17}N_{2}O_{4}^{e}\left(2\times\right), C_{11}H_{18}N_{2}O_{5}^{e}, C_{11}H_{17}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{18}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{19}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{18}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{19}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{18}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{19}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{18}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{19}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{20}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{21}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{20}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{19}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{20}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{19}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{20}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{20}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{20}N_{2}O_{5}^{e}\left(2\times\right), C_{11}H_{21}N_{2}O_{5}^{e}, \\ (2\times), C_{11}H_{20}N_{2}O_{5}^{e}\left(2\times\right), \\ C_{11}H_{20}N_{2}O_{5}^{e}\left(2\times\right)$	
Unknown	878.370 <sup>q</sup> (25.2)	$C_{6}H_{8}O_{2}^{a,b,c}, C_{6}H_{9}O_{3}^{c}, C_{6}H_{10}O_{3}^{c}$ (4×), $C_{12}H_{19}NO_{4}^{b}$	$C_5H_6^{d}, C_6H_8O_2^{a,b,c}, C_6H_{10}O_3^{c}$
Unknown	892.387 <sup><i>q</i></sup> (27.3)	$C_{6}H_{8}O_{2}^{a,b,c}(2\times), C_{6}H_{8}O_{3}^{c}, C_{6}H_{10}O_{3}^{c}(3\times), C_{12}H_{19}NO_{4}^{b}$	$C_{2}H_{4}O^{b}, C_{4}H_{4}^{d}, C_{5}H_{6}^{d'}(2\times), C_{6}H_{8}O_{2}^{a,b,c}(3\times), C_{6}H_{8}O_{3}^{c}, C_{7}H_{16}N_{2}O_{2}^{k}$
Unknown	$902.333^{q}(18.9)$	$C_{3}H_{5}O^{i}, C_{4}H_{7}O_{2}^{c}, C_{6}H_{8}O_{2}^{a,b,c}, C_{11}H_{18}N_{2}O_{4}^{g}, C_{9}H_{16}N_{2}O_{4}^{e}, C_{5}H_{12}N_{2}O^{i}, C_{5}H_{10}N_{2}O_{2}^{a}$	
Unknown	906.403 <sup><i>q</i></sup> (29.3)	$C_6H_8O_2^{a,b,c}$ , $C_6H_{10}O_3^{c}(3\times)$ , $C_{12}H_{19}NO_4^{b}$	
Unknown	934.397 <sup><i>q</i></sup> (24.3)	$C_{6}H_{8}O_{2}^{a,o,c}, C_{5}H_{14}N_{2}O^{k}, C_{6}H_{10}O_{3}^{c} (4\times), \\C_{8}H_{16}O_{3}^{m}, C_{8}H_{13}N_{2}O_{5}^{e}, C_{9}H_{17}N_{2}O_{4}^{e}, \\C_{9}H_{17}N_{2}O_{5}^{e}, C_{12}H_{19}NO_{4}^{b}$	$C_4H_4^{a}, C_5H_6^{a}, C_6H_8O_2^{a,b,c}, C_6H_{11}O_3^{c}$

<sup>a</sup> Reported for fusarinines (20).

<sup>b</sup> Reported for coprogens (20).

<sup>c</sup> Expected/typical for acyl loss (e.g., F-H).

<sup>*d*</sup> Reported for cyclic ferrioxamines (20).

<sup>e</sup> Expected, typical for loss of N<sup>5</sup>-acyl-N<sup>5</sup>-hydroxyornithine unit (e.g., F<sup>‡</sup>-H).

<sup>*f*</sup> Reported for ferrichromes (41).

<sup>*i*</sup> Reported for Fe-rhodotoluate (20).

<sup>*j*</sup> Expected, in analogy to observed neutral losses (20).

<sup>k</sup> Reported for desferrioxamine B (42).

<sup>1</sup> Observed for MS/MS measurement of fusigen standard.

<sup>m</sup> Reported for dimerum acid (44) and observed for MS/MS measurement of dimerum acid standard.

 $^{n}$  2×/3×, neutral loss two/three times that observable in the MS/MS spectrum.

<sup>o</sup> Unambiguous annotation not possible.

<sup>*p*</sup> Ion species not identified.

<sup>q</sup> Protonated molecule.

<sup>r</sup> Siderophores for which standards were available are not included.

production of extracellular siderophores of 10 wild-type *Trichoderma* strains attributed to eight species covering the three major sections of the genus. A number of strict screening criteria were defined in order to ensure that compounds were indeed iron chelators produced by the fungi under investigation. They in-

cluded verification of low-iron conditions by GF-AAS, addition of chloramphenicol to the culture medium in order to exclude bacterial contamination (production of bacterial siderophores), a realistic mass range allowing putative novel siderophores known from former publications, a characteristic iron isotopic pattern

<sup>&</sup>lt;sup>g</sup> Reported for ferrichromes (20).

<sup>&</sup>lt;sup>h</sup> Observed for MS/MS measurement of triacetylfusigen standard.

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		Trichoderma ξ	țenes involved in sideroph	ore biosynthesis						
			T. virens		T. atroviride		T. reesei			
Suppose name $(m/z)$ ; retention time $[min])$	Type	Gene name(s)	GenBank no.	E value <sup>b</sup>	GenBank no.	E value	GenBank no.	E value	Annotation	Reference(s)
All	All	sidA	EHK26838	2E-162	EHK44669	2E-144	EGR44632	1E-161	The initial biosynthetic step, shared by pathways of both intra- and extracellular siderophore biosynthesis, is catalyzed by the ornithine-N <sup>5</sup> - monoxygenase SidA (in A funioatus	20
Ferricrocin (771.248; 14.6)	Ferichrome	sidL	EHK18514	5E-145	EHK46975	2E-144	EGR46972	9E-135	$XP_{-755}$ (193). Biosynthesis of ferricrocin and hydroxyferricrocin involves acetylation of $N^5$ -hydroxyornithine to $N^5$ -acetyl- $N^5$ . SidL in A. <i>furnigatus</i> SidL in A. <i>furnigatus</i>	49
		sidC, sid2, NPS2	EHK26839	0E+00	EHK44670	0E + 00	EGR44663	0E+00	(XP_750193). In A. fumigatus, NRPS SidC (XP_753088, Sid2 in U. maydis) couples 3 acetylhydroxyornithine molecules to 1 serine and 2 glycine residues to yield	
Coprogen (822.309; 18.2) Fusigen (780.299; 14.3)	Fusarinine/coprogen	Ibis	EHK18683	0E+00	EHK45574	0E + 00	EGR46098	0E+00	cyclic ferricrocin. Acyl-CoA ligase Sidl in A. <i>fumigatus</i> (XP_753087) converts mevalonate to	51
cis-fusarin (574.193; 10.0) Fusarinine A <sup>a</sup> (556.183; 8.5) Fusarinine B <sup>a</sup>		sidH	EHK21207, EHK20720, EHK20161	2E-102, 3E-89, 8E-85	EHK46085	3E-87	EGR44125, EGR44857, EGR49665	4E-98, 2E-91, 3E-84	mevalonyl-CoA. Enoyl-CoA hydratase SidH in A. Jiunigaus (XP_748661) converts mevalonyl-CoA to	51
(798.305, 12.7) Dimerum acid (538.172; 10.7)		sidF	EHK21208	0E + 00	EHK45575	3E-129	EGR44134	0E+00	anhydromevalonyl-CoA. In A. fumigatus, anhydromevalonyl-CoA is transferred to N <sup>5</sup> - hydroxyornithine by the transacylase SidF (XP 748660) that is required for biosynthesis of fusarinine C and	20
		NPS6	EHK18682	0E + 00	EHK46196	0E + 00	EGR46022	0E+00	triacetylfusarinine C. NRPS related to NPS6 siderophore of <i>Cochliobolus</i> <i>hercostrophus</i> <i>(hAX09988)</i> ; NPS6 might produce a siderophore corresponding to fusarinine of <i>A. idulans</i> , <i>A. oryzata</i> , and <i>F.</i> <i>graminaerum</i> and corpesen in <i>Neurospora</i> <i>crassa</i> based on the domain structure of the predicted protein and protein and consure homology	4 <sup>4</sup>

(3576) <sup>2</sup> (A)
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T

E values refer to the query accession Unambiguous anno

Also *Trichoderma* genome annotation (http://genome.jgi-psf.org/pages/home.jsf?query=Trichoderma&searchType=Keyword).

Diversity of Siderophores Produced by Trichoderma

present in the full-scan mass spectra, characteristic UV/Vis absorption of ferrisiderophores at 420 to 450 nm, increasing peak heights of putative novel siderophores in different measurements due to increased sample injection, measurement of a blank control to exclude false positives, and characteristic mass increments corresponding to neutral losses in MS/MS spectra of putative novel siderophores (Fig. 1).

Minor differences between the three biological replicates of the sequenced Trichoderma strains occurred and are explicable due to the already low intensity values (peak area, ca.  $10^4$  counts  $\cdot$  s) in the replicates where they were found; their absence in the other replicates is most likely due to concentration variations (see Table S1 in the supplemental material). Three iron chelators were found in two of three biological replicates in T. reesei; they were removed only from the T. reesei results (Fig. 4 and Table 2), since they were consistently found in T. atroviride IMI 206040 and T. virens.

The putative siderophore with m/z 1121.344 was consistently found in all Trichoderma samples. Since it showed perfect coelution with dimerum acid, it was excluded from the results because it was likely being generated during the ionization process in the ESI ion source.

The screening approach applied suggested high diversity in siderophore production by Trichoderma spp. In total, 18 different siderophores were detected in the culture filtrates. Ferricrocin plays an important role in intracellular iron storage (52) and is usually described as an intracellular siderophore. Nevertheless, it has been previously reported to be found as a minor component in Trichoderma culture filtrates (13). Possibly, small amounts of ferricrocin are washed out from the mycelium during filtration of the culture broths. Dimerum acid has so far been described only for T. virens (15). The possibility that its presence, as well as the presence of fusarinine A, in all samples is due to hydrolysis of larger siderophores cannot be ruled out, since they constitute subunits of many other, larger siderophores. In total, at least 10 putative novel siderophores were found using our screening approach.

Our findings demonstrate the potential of LC-HRMS(/MS) screening methods using selected criteria for the elucidation of (putative novel) metabolites of special interest, e.g., siderophores, relevant to the molecular mechanisms regarding their beneficial use, e.g., their use as biocontrol agents.

Examples of the characterization of novel siderophores. The putative siderophore with m/z 878.371 was produced by all Trichoderma strains but T. reesei (Fig. 4). Figure 5 shows the EIC, the MS spectrum, and the MS/MS spectrum of m/z 878.371. Investigation of the MS/MS spectra revealed the mass increment between two signals in the MS/MS spectrum corresponding to the neutral losses (calculated values)  $\Delta$ 130.063, which is typical for the loss of  $C_6H_{10}O_3$  caused by cleavage of the acyl bond of the  $N^5$ -acyl- $N^5$ -hydroxyornithine unit (corresponding to "G-H" in Fig. 1 and Table S2 in the supplemental material);  $\Delta$ 112.052, indicating a loss of C<sub>6</sub>H<sub>8</sub>O<sub>2</sub> caused by either the rupture of the acyl bond of the  $N^5$ -acyl- $N^5$ -hydroxyornithine unit (corresponding to "B-H"/ "F-H" in Fig. 1 and in Table S2 in the supplemental material), but also described by Mawji et al. (43) for fusarinines and coprogens; and  $\Delta 241.131$  (C<sub>12</sub>H<sub>19</sub>NO<sub>4</sub>), described by Mawji et al. (43) for coprogens. Interestingly, there were two more similar (putative) siderophores that eluted one after another with a mass difference of ca. Δ14.016 (CH<sub>2</sub>), namely, *m/z* 892.387 and 906.403 (also present in all Trichoderma strains but T. reesei), and showing similar characteristic mass increments in the MS/MS spectra that might



FIG 5 EIC of *m*/z 878.370 (±5 ppm), MS spectrum at a retention time of 25.13 min, and MS/MS spectrum (using CID) of *m*/z 878.372 at 25.14 min in *T. polysporum*. Elemental formulas corresponding to characteristic neutral losses of siderophores are indicated in the MS/MS spectrum.

indicate that they correspond to a homologous siderophore differing in a single CH<sub>2</sub> group.

The diversity of siderophores corresponds to the ecology of Trichoderma species. In this study, we tested one relatively (in relation to the average for the genus [I. S. Druzhinina and L. Espino de Rammer, unpublished data]) weak mycoparasitic species (*T. polysporum*) and one moderate (*T. reesei*, [53]) and six strong antifungal agents. The results show that the diversity of siderophores is not reflected in the degree of antagonistic activity of Trichoderma spp., as T. polysporum and the strongly mycoparasitic T. gamsii both had 12 siderophores detected. However, comparison of the ecologies of the tested species suggests a habitat bias: T. *reesei*, the species with the most distinctive siderophore profile, is a rare tropical species that is known only from several specimens isolated from wood from low levels of tropical rain forests. Unlike all other Trichoderma species, T. reesei has never been isolated from soil. Moreover, the species has an outstanding capacity to secrete cellulose-degrading enzymes. All other species are powerful environmental opportunists that are cosmopolitan and frequently isolated from soil, the rhizosphere, dead wood, and other fungi. Analysis of opportunistic species related to T. reesei (e.g., T. longibrachiatum) will be necessary to verify this hypothesis.

Low diversity of genes coding for siderophore synthetases in *Trichoderma* might indicate post-synthetic modifications as a reason for high compound diversity. As in other filamentous ascomycetes, *Trichoderma* siderophores are mostly produced nonribosomally by large multifunctional peptide synthetases, which are organized into repetitive synthase units. Each of the repetitive synthase units has functions required to complete a different single amino acid elongation step in the synthesis of the peptide product (15). The siderophores of *Trichoderma* spp. belong to the fusigen, ferrichrome, and coprogen families (54), and their orthologous NRPS gene clusters involved in siderophore synthesis (SidD and NPS6) have been identified in *T. atroviride, T. reesei*, and *T. virens* (19, 54).

Our phylogenetic analyses identified three clades involved in siderophore biosynthesis in three *Trichoderma* species: a putative ornithine- $N^5$ -monooxygenase, SidA; an NPS6 siderophore synthetase; and a putative nonribosomal peptide synthetase, SidD. The last two are orthologous enzymes, yet *T. atroviride* contains the SidD homolog, which is more closely related to NPS6 of *C. heterostrophus*, whereas the other two species share orthologues closely related to both SidD and NPS6.

The ornithine-N<sup>5</sup>-monooxygenase SidA is known to catalyze the initial step in siderophore biosynthesis in both intra- and ex-

Numerous publications on *Aspergillus* spp. and other siderophore-producing fungi have reported several other genes that might be involved in siderophore production. Recently, it was reported that the biosynthesis of ferricrocin and hydroxyferricrocin involves acetylation of  $N^5$ -hydroxyornithine to  $N^5$ -acetyl- $N^5$ -hydroxyornithine by the *sidL* gene (49). *sidL* was not found to be genomically clustered with other siderophore-biosynthetic genes, and it is not regulated by iron availability (49). We found that all three *Trichoderma* spp. possess a *sidL*-homologous gene that has 47 to 49% identity to *A. fumigatus*, with significant E values (Table 3).

Furthermore, Tobiasen et al. (47) recently reported that NPS2 encodes an NRPS with a composition analogous to the structure of SidC from *A. nidulans* (45) and Sid2 from *Ustilago maydis* (59), whose genes are known to produce the siderophores ferricrocin and ferrichrome, respectively. Schwecke et al. (60) concluded that NPS2 produces ferricrocin, based on the domain structure and architecture of the predicted enzyme. Additionally, it was shown that NPS2 in *F. graminearum* produces a siderophore corresponding to ferricrocin from *A. nidulans* (47). Corresponding to our detection of ferricrocin in *T. virens*, *T. atroviride*, and *T. reesei*, we found the highly similar orthologous genes *sidL* and *sidC* in the genomes of all three species (Table 3).

Moreover, it was recently shown that the biosynthesis of fusarinine- and coprogen-type siderophores in *A. fumigatus* requires five genes corresponding to *sidI*, *sidH*, *sidF*, *sidD*, and *sidF* (49– 51), starting with the hydroxylation of ornithine, catalyzed by the monooxygenase SidA. Based on finding highly similar orthologues for all the genes involved in fusarinine C synthesis in *Trichoderma* spp. (Table 3), we propose that the production of fusarinine-type siderophores is similar to the well-understood process in *Aspergillus* spp. However, the nonribosomal siderophore synthetase SidD, which is involved in the last step of fusarinine C synthesis in *A. fumigatus* (51) (Table 3), or its orthologue NPS6 was present in all three *Trichoderma* spp., suggesting their important roles in fusarinine C production in *Trichoderma*.

We detected dimerum acid in all three *Trichoderma* spp., but until now, its production has been reported only for *T. virens* (15). Wilhite et al. (15) found that Psy1 disruptants produced normal amounts of gliotoxin in *T. virens* but grew poorly under low-iron conditions, suggesting that Psy1 plays a role in siderophore production. The disruptants could not produce the major *T. virens* siderophore dimerum acid, a dipetide of acylated  $N^5$ -hydroxyornithine (15). However, Wiest et al. (61) described Psy1 as a fragment of peptaibol synthase, which is unrelated to siderophore biosynthesis, and furthermore, stressed that it is unlikely that it is involved in dimerum acid biosynthesis because of the numerous dimerum acid/coprogen-type siderophore producers that lack Psy1 orthologues (61).

However, the diversity of siderophores excreted by *Trichoderma* spp. is much higher than the number of NRPS genes present in their genomes (Table 3). Nevertheless, the siderophore production pattern is not reflected in the phylogenetic diversity of NRPS siderophore synthetases. However, further biosynthetic modification of NRPS products by non-NRPS enzymes, such as transacetylases and oxygenases, has been reported (8). Our data suggest that the great diversity of siderophores found in this study might be the result of even more enzymes involved in the modification of siderophores. Modification by external enzymes (in *trans*-modifications) or enzymes that work postassembly could also support the diversity of siderophores found in our study (62). Such known modifications include glycosylation, halogenation, and oxidation/reduction.

Since the competition for iron can play a key role in the biological control exerted by *Trichoderma* spp., the new insights into the productivity and diversity of extracellular iron-containing metabolites that have been gained in our studies demonstrate the great value of the LC-HRMS/MS method developed for this research area.

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