

NIH Public Access **Author Manuscript**

Chem Res Toxicol. Author manuscript; available in PMC 2013 October 15.

Published in final edited form as: Chem Res Toxicol. 2012 October 15; 25(10): 2179–2193. doi:10.1021/tx300253j.

Mapping Serum Albumin Adducts of the Food-borne Carcinogen 2-Amino-1-methyl-6-phenylimidazo[4,5-*b***]pyridine by Data-Dependent Tandem Mass Spectrometry**

Lijuan Peng†, **Surendra Dasari**\$, **David L. Tabb**\$, and **Robert J. Turesky**†,*

†Division of Environmental Health Sciences, Wadsworth Center, New York State Department of Health, Albany, New York 12201

\$Department of Biomedical Informatics, Vanderbilt University, Nashville, TN 37232-8575

Abstract

2-Amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PhIP) is a heterocyclic aromatic amine that is formed during the cooking of meats. PhIP is a potential human carcinogen: it undergoes metabolic activation to form electrophilic metabolites that bind to DNA and proteins, including serum albumin (SA). The structures of PhIP-SA adducts formed in vivo are unknown and require elucidation before PhIP protein adducts can be implemented as biomarkers in human studies. We previously examined the reaction of genotoxic N-oxidized metabolites of PhIP with human SA in vitro and identified covalent adducts formed at cysteine³⁴ (Cys³⁴); however, other adduction products were thought to occur. We have now identified adducts of PhIP formed at multiple sites of SA reacted with isotopic mixtures of electrophilic metabolites of PhIP and 2-amino-1-methyl-6- $[{}^{2}H_{5}]$ -phenylimidazo $[4,5-b]$ pyridine ($[{}^{2}H_{5}]$ -PhIP). The metabolites used for study were: 2-nitro-1methyl-6-phenylimidazo[4,5-b]pyridine (NO₂-PhIP), 2-hydroxyamino-1-methyl-6phenylimidazo[4,5-b]pyridine (HONH-PhIP), or N-acetyloxy-2-amino-1-methyl-6 phenylimidazo[4,5-b]pyridine (N-acetoxy-PhIP). Following proteolytic digestion, PhIP-adducted peptides were separated by ultra performance liquid chromatography and characterized by ion trap mass spectrometry, employing isotopic data-dependent scanning. Analysis of the tryptic or tryptic/ chymotryptic digests of SA modified with NO_2 -PhIP revealed that adduction occurred at Cys³⁴, Lys¹⁹⁵, Lys¹⁹⁹, Lys³⁵¹, Lys⁵⁴¹, Tyr¹³⁸, Tyr¹⁵⁰, Tyr⁴⁰¹, and Tyr⁴¹¹, whereas the only site of HONH-PhIP adduction was detected at Cys³⁴. N-Acetoxy-PhIP, a penultimate metabolite of PhIP that reacts with DNA to form covalent adducts, did not appear to form stable adducts with SA; instead, PhIP and 2-amino-1-methyl-6-(5-hydroxy)-phenylimidazo[4,5-b]pyridine, an aqueous reaction product of the proposed nitrenium ion of PhIP, were recovered during the proteolysis of ^N-acetoxy-PhIP-modified SA. Some of these SA adduction products of PhIP may be implemented in molecular epidemiology studies to assess the role of well-done cooked meat, PhIP, and the risk of cancer.

Introduction

2-Amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PhIP) is a carcinogenic heterocyclic aromatic amine (HAA) that is formed, by reaction of creatinine with phenylalanine, during the cooking of meats, poultry, and fish.¹ The concentrations of PhIP can reach up to 480 parts per billion in well-done cooked poultry.² PhIP comprises about 70% of the daily mean

^{*}Address correspondence to: Robert J. Turesky, Phone: 518-474-4151, Fax: 518-473-2095, Rturesky@wadsworth.org. Supporting Information Available: Additional information as noted in text (Figure S-1 - Figure S-11). This material is available free of charge via the Internet at [http://pubs.acs.org.](http://pubs.acs.org)

Peng et al. Page 2

intake of HAAs in the United States.³ Many epidemiological studies have reported a positive association between frequent consumption of well-done cooked meats containing PhIP and an increased risk of stomach, colon, pancreas, prostate, and breast cancers, although some studies have failed to find associations between well-done meat and cancer risk.4–6 A major limiting factor in most epidemiological studies is the uncertainty in quantitative estimates of chronic exposure to PhIP or other HAAs, and thus, the association of HAAs formed in cooked meat and cancer risk has been difficult to establish. In the $11th$ Report on Carcinogens, PhIP and several other HAAs were classified as 'reasonably anticipated to be human carcinogens'.⁷ However, there is a critical need to establish longterm biomarkers of HAAs that can be implemented in molecular epidemiology studies to firmly evaluate the health risk of these genotoxicants.

Long-term biomarkers of carcinogens, such as DNA or protein adducts, are a measure of the biologically effective dose and have been used for human risk assessment of environmental and dietary genotoxicants.8,9 Electrophilic genotoxic metabolites of PhIP react with DNA and proteins.¹⁰ PhIP undergoes N-oxidation by cytochrome P450 enzymes to form 2hydroxyamino-1-methyl-6-phenylimidazo[4,5-b]pyridine (HNOH-PhIP) which can directly react with DNA, or HONH-PhIP can undergo conjugation reactions with phase II enzymes to form highly reactive esters that covalently bind to DNA.¹⁰ PhIP-DNA adducts have been detected in human tissues and fluids, by immunohistochemistry methods, $11,12$ accelerator mass spectrometry,13 or by liquid chromatography/mass spectrometry.14 However, DNA adduct measurements are often precluded by the unavailability of biopsy samples in large scale human studies. Moreover, DNA adducts are generally repaired, and the adduct levels can be below the limit of detection, even when measured by sensitive tandem mass spectrometry techniques.

Hemoglobin (Hb) and serum albumin (SA) carcinogen adducts have been used as an alternative to DNA adducts for biomarkers of several different classes of carcinogens.^{9,15} Stable carcinogen protein adducts are expected to accumulate and follow the kinetics of the lifetime of Hb or half-life of SA, during chronic exposure, and augment the sensitivity of adduct detection.¹⁶ The N-hydroxylated metabolites of many aromatic amines, carcinogens that are structurally related to HAAs, bind to human hemoglobin (Hb) at appreciable levels.17 The arylhydroxylamines penetrate the erythrocyte and undergo a co-oxidation reaction with oxy-hemoglobin (oxy-Hb), to form the arylnitroso intermediates and methemoglobin (met-Hb).¹⁸ The arylnitroso compounds can react with the Cys^{93} residue of the human β-Hb chain to form a sulfinamide adduct.^{9,19} However, the covalent binding of HONH-PhIP and other N-hydroxylated-HAAs to Hb in rodents and humans is very low, and the HAA-Hb sulfinamide adduct does not appear to be a promising biomarker to assess human exposure.¹⁰

Many genotoxicants and toxic electrophiles also bind covalently to human SA.^{9,15,16,20} Among the 585 amino acids of the mature SA sequence, the thiol group of Cys^{34} , the imidazole nitrogen atoms of histidine, the amino and guanidine groups of the side chain of lysine and arginine, the carboxyl groups of the side chain of aspartic and glutamic acid, and the phenolic group of tyrosine are most frequently involved in the formation of SA adducts.20 PhIP was reported to bind to human SA in vivo at levels that may be sufficient to establish mass spectrometry techniques for biomonitoring.13 At least one of the PhIP-SA adducts undergoes hydrolysis under acidic pH conditions to regenerate PhIP.²¹ However, the reactive species responsible for adduct formation and the structures of the intact PhIP-SA adduct(s) are unknown. N-Oxidation products of PhIP covalently bind to rat or human SA in vitro.22–24 2-Nitroso-1-methyl-6-phenylimidazo[4,5-b]pyridine (NO-PhIP) reacted with SA in vitro to produce the N^2 -[cystein-S-yl-PhIP]-S-oxide at Cys^{34.24} This sulfinamide adduct is labile towards acid and some portion of the acid-labile adduct(s) of PhIP formed

with SA in vivo may exist as the sulfinamide. The Cys³⁴ of SA also reacted with 2-nitro-1methyl-6-phenylimidazo $[4,5-b]$ pyridine (NO₂-PhIP) in vitro to form a stable sulfur-carbon linked adduct with PhIP.24 However, other PhIP-SA adduction products are likely to have been formed and remain to be characterized.

Isotope pattern-dependent mass spectrometric scanning methods have been employed to analyze the formation of reactive metabolites from a number of toxicants.²⁵ Isotopic datadependent scanning has also been used to identify proteins in liver microsomal preparations chemically modified with unlabeled and ¹⁴C-radiolabeled furan containing drugs.²⁶ The goal of our current study was to investigate the reactivity of human SA with electrophilic Noxidation products of PhIP in vitro and to characterize the adduction products by mass spectrometric techniques. We have exploited the isotopic data-dependent scanning technique to map the sites of PhIP binding to human SA modified with a mixture of unlabeled and $[{}^{2}H_{5}]$ -labeled *N*-oxidized metabolites of PhIP.

Materials and Methods

Caution: PhIP is a carcinogen and should be handled in a well-ventilated fume hood with the appropriate protective clothing.

Chemicals and Materials

PhIP was purchased from Toronto Research Chemicals (Toronto, ON, Canada). 2-Amino-1 methyl-6- $[^2H_5]$ -phenylimidazo $[4,5$ -*b*]pyridine ($[^2H_5]$ -PhIP, 99% isotopic purity) was a gift from Dr. Mark Knize and Dr. Kristen Kulp, (Lawrence Livermore National Laboratory, Livermore, CA). Human SA, cysteine, tyrosine, lysine, tryptophan, β-mercaptoethanol, iodoacetamide, dithiothreitol, N-ethylmaleimide (NEM) and 4-chloromercuribenzoic acid (4-CMB) were obtained from Sigma (St. Louis, MO). The human plasma was purchased from Bioreclamation LLC (Hicksville, NY). Trypsin gold, sequencing grade trypsin, and chymotrypsin were purchased from Promega (Madison, WI). Pronase E, leucine aminopeptidase, and prolidase were purchased from Sigma (St. Louis, MO). All solvents were high-purity B & J Brand from Honeywell Burdick and Jackson (Muskegon, MI). ACS reagent-grade formic acid (88%) was purchased from J.T. Baker (Phillipsburg, NJ). Isolute C18 SPE column (25 mg) was from Biotage (Charlotte, NC). HiTrap Blue affinity and GE P10 columns were obtained from GE Healthcare (Piscataway, NJ). Spire PEP tips were a gift from Thermo (Bellefonte, PA). All other chemical reagents were ACS grade and purchased from Sigma- Aldrich.

Synthesis of *N***-Oxidized Metabolites of PhIP**

Unlabeled PhIP and $[^{2}H_{5}]$ -PhIP were mixed at a molar ratio of 1:1. NO₂-PhIP and NO₂- $[^2H_5]$ -PhIP were synthesized by diazotization of PhIP with NaNO₂, as described previously.²⁷ Subsequently, the $NO₂$ -PhIP analogues were reduced with hydrazine, using Pd/C as a catalyst, to produce HONH-PhIP and HONH- $[^2H_5]$ -PhIP.²⁷ The method of synthesis of N -acetoxy-PhIP is a modification of procedures reported in the literature.^{28,29} An equimolar mixture of HONH-PhIP and HONH- $[^2H_5]$ -PhIP (20 μ g, 83 nmol) in C₂H₅OH (50 μ L) at −5 °C was reacted with acetic anhydride in acetic acid (10% v/v, 2 μ L) for 16 min. The solution was then diluted with chilled, deionized $H_2O(500 \mu L)$ and applied to the Spire PEP tip (12 mg) under a gentle vacuum. The resin was washed with chilled deionized H_2O (500 μ L), and the mixture of N-acetoxy-PhIP and N-acetoxy-[²H₅]-PhIP was eluted with chilled C_2H_5OH (~100 μ L). The product was characterized by UPLC-ESI/MS² with a triple stage quadrupole mass spectrometer (vide infra). The N-acetoxy-PhIP derivatives were prepared immediately prior to their reaction with peptides or SA.

2-Hydroxy-1-methyl-6-phenylimidazo[4,5-b]pyridine (2-HO-PhIP) was prepared by incubation of NO₂-PhIP (1.7 μ g) in 90 mM NH₄OH (500 μ L) at 40 °C for 1 h. 2-Amino-1methyl-6-(5-hydroxy)-phenylimidazo[4,5-b]pyridine (5-HO-PhIP) was obtained by the solvolysis of N-acetoxy-PhIP as described previously.^{22,30} The identities of 2-HO-PhIP and 5-HO-PhIP were confirmed by their characteristic product ion spectra.^{24,30}

Amino acid adducts of NO₂-PhIP were synthesized by the reaction of cysteine, tryptophan, tyrosine or lysine (0.1 μ mol) with NO₂-PhIP (0.1 μ mol) in 50 mM ammonium bicarbonate buffer, pH 8.5 (1 mL) at 37 °C for 18 h. The adducts were isolated by enrichment with C18 SPE resins that were prewashed with $CH₃OH$ and $H₂O$. The reaction products were applied to the resin, and the resin was washed with $H_2O(2 \text{ mL})$. The adducts were eluted with CH3OH (1 mL) and concentrated to dryness by vacuum centrifugation. The modified amino acids were purified by HPLC with Agilent 1100 HPLC system (Palo Alto, CA) and an Agilent Eclipse XDB-C18 column $(4.6 \times 250 \text{ mm})$. A linear gradient was employed, starting from 100% A solvent (0.1% HCO₂H) and reaching 100% B solvent (95% CH₃CN containing, 4.9% H₂O, and 0.1% HCO₂H) at 20 min, at a flow rate of 1 mL/min. The wavelength was monitored at 210 and 320 nm. The structures of the PhIP-modified amino acids were determined by LC -ESI/MSⁿ with a linear quadrupole ion trap MS (LTQ MS, Thermo Fisher, San Jose, CA). The approximate yields of the $NO₂$ -PhIP-modified cysteine and tyrosine adducts were \sim 50%, and the yield of the NO₂-PhIP-modified lysine adduct was \sim 5% on the basis of UV measurements. NO₂-PhIP did not form an adduct with tryptophan, under these reaction conditions.

Modification of Human SA and Plasma with *N***-Oxidized Metabolites of PhIP**

Commerical human SA was pretreated with β-mercaptoethanol to reduce mixed disulfides formed at $Cys^{34.31}$ For some studies, the Cys^{34} of SA was blocked by titration of the thiol residue to its end point with 4-CMB , $31,32$ or by reaction with a 5-fold mol excess of NEM for 5 h at room temperature, followed by gel filtration. The sulfhydryl content of SA was determined using Ellman's reagent.33 The SA reduced with β-mercaptoethanol contained a sulfhydryl content of 0.95 mol -SH/mol SA, whereas the sulfhydryl content of 4-CMB or NEM modified SA was 0.02 mol -SH/mol SA.

A solution of NO₂-PhIP and NO₂-[²H₅]-PhIP (90 nmol in 6.6 μ L DMSO), or HONH-PhIP and HONH- $[^2H_5]$ -PhIP (90 nmol in 13 μ L C₂H₅OH), or N-acetoxy-PhIP and N-acetoxy- $[{}^{2}H_{5}]$ -PhIP mixture (90 nmol in 40 μ L C₂H₅OH) was reacted with reduced SA (2 mg, 30) nmol) in 1 mL 10 mM potassium phosphate buffer (pH 7.4). The reactions of $NO₂$ -PhIP and HONH-PhIP with SA were performed at 37°C for 18 h, whereas the reaction of N-acetoxy-PhIP with SA was performed at 37 °C for 2 h. Unbound PhIP derivatives were removed from SA by solvent extraction with ethyl acetate (3 mL, 3 times), followed by gel-filtration chromatography of the SA with the PD-10 column containing 10 mM potassium phosphate buffer (pH 7.4).

Human plasma was diluted with PBS by 5-fold. A 1 mL solution of diluted plasma was then treated with a 1:1 isotopic mixture of NO₂-PhIP and NO₂-[²H₅]-PhIP (450 nmol in 33 μ L DMSO) at 37 °C for 18 h. The purification of SA from plasma matrix was done with a HiTrap Blue affinity column.²⁴ In brief, the treated plasma was diluted with buffer A (50 mM KH₂PO₄, pH 7.0) and centrifuged to remove particulates, before it was applied to a HiTrap Blue affinity column. Buffer B (50 mM KH_2PO_4 buffer (pH 7.0) containing, 1.5 M KCl (3 mL) was used to elute SA from the affinity column. Thereafter, any remaining unbound PhIP derivatives were removed from SA by solvent extraction, followed by gelfiltration chromatography of the SA in 10 mM potassium phosphate buffer as described above.

Trypsin Digestion

Enzymes were prepared as described previously.²⁴ Three different proteolytic conditions were employed to digest SA adducts modified by N-oxidized-PhIP derivatives.

- **1.** Trypsin digestion: PhIP-modified SA (0.5 mg) was concentrated to dryness by vacuum centrifugation and dissolved in 0.25 M Tris buffer containing 6 M guanidine-HCl (pH 7.4, 0.5 mL). Dithiothreitol (6.5 mg, 20 mM) was added to the solution, and the mixture was incubated at 55 \degree C for 1 h. After the solution was cooled to room temperature, excess iodoacetamide (6 mg, 65 mM) was added, and the mixture was incubated in the dark for 30 min. The denatured and alkylated SA was then subjected to gel-filtration chromatography with the PD-10 column in 50 mM ammonium bicarbonate buffer (pH 8.5), to remove excess iodoacetamide. The SA solution (6 μ g in 35 μ L) was mixed with CaCl₂ (1 μ L of 50 mM stock solution), followed by addition of trypsin at a protease:SA ratio of 1:50 (w/w). The digestion was performed for 20 h at 37 °C. Thereafter, the protein digest was diluted with H₂O (1 mL) and applied to a C18 SPE. The CH₃OH eluents were concentrated to dryness by vacuum centrifugation and resuspended in $1:1 H₂O$: DMSO for UPLC-ESI/MSⁿ analysis.
- **2.** Trypsin/chymotrypsin digestion: The denatured and alkylated PhIP-modified SA (5 μ g, 75 pmol, in 35 μ L 50 mM ammonium bicarbonate buffer (pH 8.5) containing $CaCl₂$ (1 mM) was digested with trypsin at a protease: protein ratio of 1:50 (w/w) and chymotrypsin at a protease:SA ratio of 1:12.5 (w/w). The digestion was performed at 37° C for 20 h, followed by the same SPE purification procedure described above.
- **3.** Pronase E/leucine aminopeptidase/prolidase digestion: This enzymatic digestion was performed as described previously.²⁴ In brief, the PhIP-modified SA (5 μ g, 75 pmol) in 50 mM ammonium bicarbonate buffer (pH 8.5) containing 1 mM MnCl₂ was treated with Pronase E at a protease:protein ratio of 1:2 (w/w), leucine aminopeptidase at a protease:protein ratio of 1:30 (w/w), and prolidase at a protease:SA ratio of 1:8 (w/w). The digested was conducted at 37° C for 20 h, followed by the same SPE purification procedure described above.

Mass Spectrometry Methods

Product ion spectra of N-oxidized PhIP derivatives were acquired by infusion with a Finnigan Quantum Ultra triple stage quadrupole mass spectrometer (TSQ MS, Thermo Fisher, San Jose, CA) interfaced with a CaptiveSpray™ source from Michrom Bioresource Inc. (Auburn, CA). Typical instrument tuning parameters were: capillary temperature 200 °C, source spray voltage 1.4 kV, tube lens offset 95 V, capillary offset 35 V, and in-source fragmentation 5 V. Argon, set at 1.5 mTorr, was used as the collision gas, and the collision energy was variable and set between 15 to 35 eV. In some instances, metabolites were characterized online with a Waters NanoAcquity UPLC (New Milford, MA) interfaced with the TSQ MS. LC separation of peptides was performed with a Magic C18AQ column (0.3 \times 150 mm) (Michrom Bioresource Inc.). Metabolites were resolved with a linear gradient starting from 90% A solvent (0.01% HCO₂H) to 100% B solvent (95% CH₃CN containing, 4.99% H₂O, and 0.01% HCO₂H) over 20 min, at a flow rate of 5 μ L/min.

Accurate mass measurements of the synthesized amino acid and peptide adducts of NO₂-PhIP were acquired on the LTQ Orbitrap XL (Thermo Fisher Scientific, Bremen, Germany) at the Proteomics Core Facility in the Center for Biotechnology and Interdisciplinary Studies, Rensselaer Polytechnic Institute, Troy, NY. The MS was interfaced with the Michrom CaptiveSpray™ ion source. Full scan mass spectra were acquired from a scan range of m/z 100 to 700 at a resolution of R = 60000 at m/z 400. The injection time was 500

ms and data were acquired with a μ scan count of 1. The detected ions were recalibrated on the fly using phthalate as the lock mass at m/z 149.02332. The spray voltage was 2.2 kV, capillary temperature was 200 °C, and the tube lens was 100 V. The isolation width was 2 amu, and the collision-induced dissociation (CID) in the linear ion trap was set at a normalized collision energy of 35%.

UPLC-ESI/MSⁿ

The separation of peptides was performed with a NanoAcquity UPLC system (Waters Corp., Milford, MA) equipped with a Magic C18AQ column $(0.3 \times 150 \text{ mm})$ from Michrom Bioresource Inc. The Digests $(3 \mu L)$ were injected and peptides were resolved with a linear gradient starting from 90% A solvent (0.01% HCO₂H) to 100% B solvent (95% CH₃CN containing, 4.99% H₂O, and 0.01% HCO₂H) over 60 min, at a flow rate of 5 μ L/min. MS spectra were acquired with a linear quadrupole ion trap mass spectrometer (LTQ, Thermo Fisher, San Jose, CA). Xcalibur version 2.07 software was used for data manipulations. All analyses were conducted in the positive ionization mode and employed an Advance CaptiveSpray[™] source from Michrom Bioresource Inc. The temperature of capillary tube was set at 200 °C; the spray voltage was 1.5 kV; and the in-source fragmentation was 10 V. There was no sheath or auxiliary gas. Helium was used as the collision and damping gas in the ion trap and was set at a pressure of 1 mTorr. One μ scan was used for data acquisition. The automatic gain control (AGC) settings were full MS target $30,000$ and MSⁿ target 10,000, and the maximum injection time was 10 ms.

Data-Dependent MS/MS

A full scan was obtained for eluting peptides in the range of 300–2000 amu, followed by three data-dependent MS/MS scans. The MS/MS spectra were recorded using dynamic exclusion of previously analyzed precursors for 180 s with a repeat of 3 and a repeat duration of 60 s. MS/MS spectra were generated by CID of the peptide ions at a normalized collision energy of 35% to generate a series of b- and y-ions as major fragments. For isotopically labeled experiments, the mass spectrometer was programmed to switch from the full survey scan MS to the MS/MS scan mode, which was triggered by the characteristic isotopic pattern of unlabelled PhIP/ $[^2H_5]$ -PhIP at a partner intensity ratio of 65 – 100%, employing the mass tags scanning option. The mass to charge differences were set at m/z 5.00, 2.50, or 1.67, respectively, for singly, doubly, or triply charged peptide species. The mass tag of singly charged species $(m/z 5)$ was employed to scan for amino acid adducts recovered from SA digested with Pronase E/leucine aminopeptidase/prolidase, whereas the mass tags of doubly and triply charged ions (m/z 2.50 and 1.67) were employed to scan for SA peptide adducts recovered from tryptic or tryptic/chymotryptic digests. The four most abundant ions above 1000 counts that displayed the isotopic pattern were selected for CID with a normalized collision energy of 35%, employing an isolation width of 2 amu.

Data Analysis

The mass spectral data PhIP- and $[^{2}H_{5}]$ -PhIP-modified amino acid and peptide adducts were acquired by mass tags with the Xcalibur software. Scan filters were employed to extract the protonated ions and MS/MS spectral data of the isotopic pairs. The tandem mass spectra of peptides were interpreted manually and facilitated by online software (Protein Prospector, Univ. of California, San Francisco, [http://us.expasy.org/proteomics\)](http://us.expasy.org/proteomics). Fully automated data analysis on peptide adducts was conducted by converting RAW data to mz5 format in the ProteoWizard msConvert tool.³⁴ Spectra were identified by the MyriMatch algorithm,³⁵ version 2.1.111, using a 31 protein subset database containing SA from an initial search against the RefSeq human protein database, version 37.3. Searching for modifications in a protein database containing only a handful of sequences can inappropriately force spectra to match to modified peptides. Therefore, we generated a database of 31 proteins by a fully

tryptic search of the LC-MS/MS data generated here in combination with a separate LC-MS/ MS experiment from unrelated major proteins frequently observed in pull-downs with immunoprecipitation to add protein distractors to the database. The sequence database was reversed so that each protein sequence appeared in both normal and reversed orientations (totaling 62 sequences), enabling false discovery rate estimation. MyriMatch was configured to have cysteines to contain carbamidomethyl $(+57.021$ Da) as a dynamic modification and to allow for the possibility of oxidation (+15.996) on methionines, and deamidation (−17.03 Da) of N-terminal glutamines. Peptides modified with $NO₂-PhIP$ and $NO₂-[^2H₅]-PhIP$ were searched with [C,K,Y] allowed to have dynamic modifications of 207.1 and 212.1 amu; peptides modified with HONH-PhIP and HONH $-[^2H_5]$ -PhIP or their N-acetoxy derivatives were allowed to have dynamic modifications at [C,K,Y,S,T,W,H] of 222.1 and 227.1 amu, for adduction with PhIP, or dynamic modifications at [C] of 238.1 and 243.1 amu for adduction with NO-PhIP (sulfinamide adducts); and 254.1 and 259.1 amu (sulfonamide adducts). Candidate peptides were allowed to have trypsin cleavages or protein termini at one or both termini (semi-tryptic search), and up to 2 missed cleavages were permitted. The precursor error was set at 1.25 m/z , but fragment ions were required to match within 0.5 $m/$ ^z. Analyses of modified SA were also performed with trypsin/chymotrypsin digests, employing the same configurations. The IDPicker algorithm³⁶ v3.0.420 filtered the identifications for each spectrum with a 5% identification false discovery rate at the peptidespectrum match level.

Results

Mapping of Human SA Modified with NO2-PhIP and NO2-[2H5]-PhIP

Data-Dependent Analysis of Human SA Modified with NO2-PhIP Followed by Digestion with Pronase E, Leucine aminopeptidase and Prolidase—This three enzyme mixture efficiently digests many proteins to amino acids.^{37,38} The full scan mode was monitored from the mass range of 300 to 600 Da, a range that encompasses the molecular masses $[M+H]^+$ of all possible amino acid adducts formed with NO_2 -PhIP. The mass chromatograms of the enzymatic digests of unmodified and NO₂-PhIP-modified SA acquired by the data-dependent MS/MS with mass tags enabled are presented in Supporting Information, Figure S-1.

Three pairs of adducts were detected ($t_R = 6.7, 13.9,$ and 17.8 min). The first adduct was characterized as desamino-PhIP-K. The product ion spectra of the adduct ($[M+H]^+$ at m/z 354.2) and its isotopic labeled species ($[M+H]^+$ at m/z 359.2) are shown in Figure 1. The masses are consistent with the molecular weight of a reaction product formed between lysine (146.2 Da) and NO_2 -PhIP or NO_2 -[²H₅]-PhIP (254.1 or 259.1 Da), where the amine group of the side chain of lysine displaced the nitro moiety of PhIP. The product ion spectra of desamino-PhIP-K ($[M+H]^+$ at m/z 354.2) and $[^2H_{5}]$ -desamino-PhIP-K ($[M+H]^+$ at m/z 359.2) exhibited a 'twin' pattern of ions that differ by 5 Da. The product ion spectrum of desamino-PhIP-K contained 3 major fragment ions at m/z 337.2, 309.2 and 291.2, corresponding to the losses of NH_3 , NH_3 +CO, and NH_3 +HCO₂H, respectively. The peaks observed at m/z 225.1 and 230.1, respectively, in the spectra of desamino-PhIP-K and desamino- $[^2H_5]$ -PhIP-K are attributed to protonated PhIP and $[^2H_5]$ -PhIP, which were produced by cleavage of the bond between the ε-carbon and the side chain amine group of Lys. The product ion spectra acquired at the MS³ scan stage on m/z 225.1 and 230.1 are identical to the spectra of PhIP and $[^2H_5]$ -PhIP (data not shown). The synthetic desamino-PhIP-K adducts exhibit the identical t_R and product ion spectra (data not shown). The assignments of these fragment ions were supported by exact-mass measurements (Table 1).

The product ion spectra of adducts ($[M+H]^+$ at m/z 329.2 and 334.2) of the second pair of adducts ($t_R = 13.9$ min are identical to the spectra that we reported previously for desamino-PhIP-C adduct, where the thiol group of Cys³⁴ displaced the nitro moiety of PhIP.²⁴

The most abundant amino acid adducts ($t_R = 17.8$ min) have molecular weights of 388.2 and 393.2 Da; these masses are consistent with masses of adducts formed between tyrosine (181.1 Da) and 209.1 (desamino-PhIP) or 214.1 Da (desamino- $[^2H_5]$ -PhIP), less two protons. Major fragment ions are observed in the product ion spectrum of desamino-PhIPmodified tyrosine ($[M+H]^+$ at m/z 389.3) (Figure 2A) at m/z 344.2 $[M+H-NH_3-CO]^+$, 343.2 [M+H-HCO₂H]⁺ (tyrosine immonium ion plus desamino-PhIP), 328.2 [M+H–NH₃-CO₂]⁺, 315.2 $[M+H-C_2H_4NO_2\bullet]^+$, and at m/z 209.2 $[M+H-C_2H_10NO_3\bullet]^+$, which is attributed to the radical cation of desamino-PhIP. The product ion spectrum of the desamino- $[^2H_5]$ -PhIP-Y $([M+H]^+$ at m/z 394.1) displays the corresponding fragment ions at m/z 349.2, 348.2, 333.2, 320.2, and 214.2 (Figure 2B). Consecutive reaction monitoring was performed at the $MS³$ and MS⁴ scan stages for several fragment ions of desamino-PhIP-Y to determine the site of bond formation between NO₂-PhIP and tyrosine (Figures 2C – 2E). The radical cation at m/z 315.2 $[M+H-C_2H_4NO_2\bullet]^+$ (m/z 320.2 for desamino- $[^2H_5]$ -PhIP-Y) is proposed to arise by homolytic cleavage of the bond between the α and β -carbon atoms of tyrosine; this is an uncommon mechanism of fragmentation of amino acids by ESI-MS under low-energy CID conditions (Scheme 1).³⁹ The second generation product ion spectrum $(MS³)$ of the ion at m/z 315.2 (m/z 320.2 for desamino- $[^2H_5]$ -PhIP-Y) contains the ion at m/z 209.2 (m/z 214.2 for desamino- $[^{2}H_{5}]$ -PhIP-Y) as the base peak (Figure 2C). This ion is proposed to occur by cleavage of the bond of the C-2 atom of the imidazole moiety of PhIP and the phenolic oxygen of tyrosine, to produce the radical cation of desamino-PhIP and the quinone methide (106.1 Da) as a neutral species (Scheme 1). The second generation product ion spectrum $(MS³)$ of the m/z 343.2 ([M+H-HCO₂H]⁺ (tyrosine immonium ion containing desamino-PhIP) (Figure 3D) contains a major fragment at m/z 226.2, which is the m/z of the protonated 2-HO-PhIP. The third generation product ion spectrum (MS⁴) of m/z 226.2 produced ions at $m/z 211.2$, 198.2 and 183.2; these ions correspond to, respectively, the loss of the CH₃•, CO, and the losses of CH₃• and CO from 2-HO-PhIP (Figure 2E). The mass spectrum is in excellent agreement to the spectrum of synthetic 2-HO-PhIP (data not shown). The mass spectral data demonstrate that bond formation adduct occurred between the 4-HO group of tyrosine and the C-2 imidazole atom of PhIP. The assignments of these fragment ions were supported by exact-mass measurements (Table 1).

Data-Dependent Analysis of Human SA Modified with NO2-PhIP Following Protein Digestion with Trypsin—Data mining of the peptide chemical features by MyriMatch revealed that the sequence coverage of unmodified SA exceeded 80% in the data dependent MS/MS mode (data not shown). Cys³⁴ is the only unpaired cysteine residue in SA that is available for reaction with carcinogenic metabolites and toxic electrophiles.^{16,20} However, human SA contains 59 lysine and 18 tyrosine residues⁴⁰ that can potentially react with NO₂-PhIP to form adducts. The specific lysine and tyrosine residues within SA that reacted with NO2-PhIP were identified by tandem MS sequencing of the proteolytic digests of the modified SA.⁴¹

The mass chromatograms of unmodified SA peptides, and NO_2 -PhIP- and NO_2 - $[^2H_5]$ -PhIP adducted peptides of SA are shown in Figure 3. A deuterium isotope effect caused the $[^{2}H_{5}]$ -PhIP-peptide adducts to elute about 2 s earlier than the unlabeled adducts. These slight differences in t_R values skewed the 1/1 ratio of the unlabelled PhIP/ $[^2H_5]$ -PhIP-adduct signals. Therefore, the partner intensity ratio for $PhIP/[²H₅]$ -PhIP was expanded to 65 – 100%, to efficiently trigger the acquisition of MS/MS spectra on both unlabeled and labeled peptide adducts. This "relaxed" ratio resulted in elevated background signals of some

peptides in non-modified SA. Ten NO2-PhIP-modified peptides were identified (Table 2), and several product ion spectra of representative peptide adducts are presented Figure 4.

The first pair of modified peptides ($t_R = 4.1$ min) were present as doubly charged ions [M $+2H$ ²⁺ at m/z 477.8 and 480.3. This molecular mass corresponds to the mass of the amino acid sequence 191–197 of SA (ASSAKQR, $[M+H]$ ⁺ at m/z 747.4) plus desamino-PhIP or desamino- $[^2H_5]$ -PhIP, respectively (addition of 207 or 212 Da). The product ion spectra of the proposed ASSAK*QR adducts ($[M+2H]^{2+}$ at $m \times 477.8$ and 480.3) are shown in Figure 4A and 4B. The ions observed at m/z 291.1 and m/z 296.1, respectively, in the spectrum of $NO₂$ -PhIP- and $NO₂$ - $[²H₅]$ -PhIP-modified ASSAKQR are at the same m/z observed in the spectra of desamino-PhIP-K ($[K^*$ -NH₃-HCO₂H]⁺ (Figure 1), suggesting that NO₂-PhIP had bound to lysine. The peptide fragment ions $(y_2, y_2\text{-}NH_3, b_3 \text{ and } b_4)$ were detected, respectively, at m/z 303.1, 286.2, 246.2 and 317.2 for both NO₂-PhIP- and NO₂-[²H₅]-PhIP modified peptides, as would be expected for fragment ions in the unmodified peptide ASSAKQR. The peptide fragment ions (y₃*, y₄*, y₅*, y₆*, b₅* and b₆*) were shifted by 207 and 212 Da, respectively for the NO_2 -PhIP- and the NO_2 -[²H₅]-PhIP-modified peptides.

Fraction 2 was found to contain another adducted peptide of $NO₂$ -PhIP at a lysine residue $(t_R = 5.7$ min). The sequence was identified as ATK*EQLK, which covers the amino acid sequence 539–545 of SA. The product ion spectra of the NO_2 -PhIP- and NO_2 - $[^2H_5]$ -PhIPadducted ATK*EQLK are presented in Supporting Information, Figure S-2.

Two more adducted peptides containing NO2-PhIP modified lysine residues were detected in Fraction 3 ($t_R = 5.9$ min) and Fraction 4 ($t_R = 8.3$ min) at relatively low abundance, which encompasses the amino acid sequences 198–205 (LK*CASLQK) and 349–359 (LAK*TYETTLEK) of SA, respectively. (Table 1)

The fifth NO₂-PhIP-modified peptide displayed a strong signal ($t_R = 9.7$ min), and both singly charged $[M+H]^+$ (*m*/z 618.4 and 623.4) and doubly charged species $[M+2H]^{2+}$ (*m*/z 309.9 and 312.4) were observed. This modified peptide was assigned to the amino acid sequence 411–413 of SA (YTK ([M+H]⁺ at m/z 411.2), and contained a mass increment of 207 and 212 Da, consistent with adduction by desamino-PhIP and desamino- $[^2H_5]$ -PhIP. Either Tyr^{411} or Lys^{413} of SA could be the site of modification. The product ion spectra of the modified YTK peptide contained several prominent fragment ions (Figures 4C and 4D). The y_1 ion observed at m/z 147.1 corresponds to lysine, and the fragment ions observed at m/z 230.1 and 248.2 (y₂ and y₂-H₂O) in the spectra for both NO₂-PhIP-and NO₂-[²H₅]-PhIP modified peptides are also seen in the spectrum of YTK (data not shown). The fragment ions $(a_2^*, b_2^*$ and b_2^* -H₂O) contain mass increments of desamino-PhIP (207 Da) and desamino-[²H₅]-PhIP (212 Da), respectively. The b₂*-CO₂ ion (C₂₅H₂₆N₅O₂) at *m*/z 428.3 and 433.3 for unlabeled and desamino- $[^2H_5]$ -PhIP adducts is proposed to occur by a rearrangement involving the HO-group of the threonine side chain to the electrophilic carbonyl moiety, followed by loss of CO_2 .⁴² The ions at m/z 343.2 and 348.2 (y*) for desamino-PhIP- and desamino- $[{}^{2}H_{5}]$ -PhIP adducts are assigned to the desamino-PhIP-Y immonium ions. These spectral data prove that adduction of $NO₂$ -PhIP occurred at tyrosine and not lysine. The second generation product ion spectrum of the ion at m/z 343.2 for desamino-PhIP-Y*TK and at m/z 348.2 for desamino[²H₅]-PhIP-Y*TK ([M+H-HCO₂H]⁺) contained the prominent fragment ion at m/z 226.1, attributed to 2-HO-PhIP (m/z 231.1 for 2-HO-[²H₅]-PhIP), as was observed in the spectra of desamino-PhIP-Y (Figure 1). Accurate mass measurement of the monoisotopic elemental mass of synthetic Y^*TK and its product ion spectra confirmed the assignment (Table 1). Thus, the structure of the modified peptide in Fraction 5 is assigned Y*TK with the site of adduction occurring between the 4-HO-tyrosine group and the C-2 imidazole atom of $NO₂$ -PhIP.

Three more fractions containing peptides modified with $NO₂$ -PhIP at tyrosine residues were observed in Fraction 8 (t_R =14.9 min), Fraction 9 (t_R =16.8 min), and Fraction 10 (t_R =21.6 min) (Table 1). The modified peptides in Fraction 8 were doubly charged species ([M +2H]²⁺ at m/z 568.1 and 570.6), corresponding to the peptide containing amino acid residues 138–144 of SA (Y*LYEIAR [M+H]⁺ at m/z 927.5) plus the desamino-PhIP and desamino- $[{}^{2}H_{5}]$ -PhIP moieties, respectively. The product ion spectra of this pair of modified peptides contained almost the complete series of b- and y-ions and the immonium ions for the desamino-PhIP modified tyrosine (Figures 4E and 4F), which confirmed that Tyr^{138} was the amino acid modified in this sequence. The peptide adduct observed in Fraction 9 corresponded to the amino acid residues $390 - 412$ of SA (QNCELFEQLGEY*K, [M +2H]²⁺at m/z 933.2) with NO₂-PhIP adducted to Y, based on the interpretation of the product ion spectra (Table 1, and Supporting Information, Figure S-3).

The pair of peptide adducts detected in Fraction 10 were consistent with the amino acid residues 149–159 of SA (FY*APELLFFAK, $[M+2H]^{2+}$ at m/z 777.1 and 779.6) having adductions with desamino-PhIP and desamino- $[^2H_5]$ -PhIP moieties at Y, respectively.

NO₂-PhIP-modified peptides of the Cys³⁴ residue were detected in fractions 6 and 7 (t_R = 11.9 and 12.2 min). Fraction 6 contained triply charged desamino-PhIP- and desamino- $[^2H_5]$ -PhIP-labeled peptide adducts $[M+3H]^{3+}$ at m/z 572.4 and 574.0 (Supporting Information, Figures S-4A and S-4B). The peptide sequence is ascribed to amino acid residues 30–41 of SA, with adduct formation occurring at Cys³⁴ (YLQQC*PFEDHVK, 1505.7 Da). The ions observed at m/z 242.1 in the mass spectrum of NO₂-PhIP-modified peptide and at $m/z 247.1$ in the spectrum of $NO₂-[^2H₅]$ -PhIP-modified peptide were identified as the protonated 2-thioimdazole derivatives of PhIP and $[^2H_5]$ -PhIP, by their spectra at the MS³ scan stage as reported previously.²⁴ The y and b ion series supported the proposed assignment of the adduct structure. The b_2 (m/z 277.1), b_3 (m/z 404.6), b_4 (m/z 533.2), b_5 (m/z 843.1), b_6 (m/z 940.4), y_2 (m/z 246.2), y_3 (m/z 383.3), y_4 (m/z 498.3), y_5 $(m/z 627.3)$, y_6 ($m/z 774.6$) and y_7 ($m/z 871.3$) observed in both product ion spectra are consistent with predicted sequence for unmodified YLQQCPFEDHVK. A number of ions display the characteristic isotopic pattern with a mass difference of 5 Da and encompass the modified amino acid residue, b_6 *-NH₃, y₉*-NH₃, y₉*-NH₃-HS-PhIP, y₁₀*-NH₃, y₁₀*-NH₃-HS-PhIP and y_{11} ^{*}.

A pair of PhIP and $[{}^{2}H_{5}]$ -PhIP labeled peptide adducts are observed as triply charged species $[M+3H]^{3+}$ at m/z at 638.5 and 640.2 in Fraction 7. The product ion spectrum displayed many of the peptide sequence series of y and b ions in common to that observed for YLQQC*PFEDHVK, but also contained ions attributed to 3 additional amino acid residues. The peptide was identified as AQYLQQC*PFEDHVK (1704.8 Da) modified by desamino-PhIP (207 Da) or desamino- $[^2H_5]$ -PhIP (212 Da); this pair of peptide adducts occurred at amino acid residues $28 - 41$ of SA (Supporting Information, Figure S-4C and S-4D).

Data-Dependent Analysis Human SA Modified with NO2-PhIP Following

Protein Digestion with Trypsin/Chymotrypsin—Data-dependent scanning of modified SA digested with trypsin/chymotrypsin resulted in the discovery of six pairs of modified peptide adducts (Supporting Information, Figure S-5 and Table 2). Consistent with the trypsin digestion, Y*TK adducts derived from trypsin/chymotrypsin of modified SA were identified as doubly charged ion $[M+2H]^{2+}$ at m/z 309.9 and 312.4. The NO₂-PhIPmodified LQQC*PF adducts (amino acid residues 31–36 of SA) were observed as doubly charged species $[M+2H]^{2+}$ (m/z 472.0 and 474.4) (t_R =17.5 min). The product ion spectra were consistent with results from our previous study that showed the formation of LQQC*PF (C-desamino-PhIP or C- desamino- $[^2H_5]$ -PhIP) adducts.²⁴

Peptide adducts observed at 11.4 min displayed triply charged ions $[M+3H]^{3+}$ at m/z 517.8 and 519.7, which are an addition mass of 207 and 212 Da to the protonated peptide LQQCPFEDHVK $(m/z 1343.6,$ residues 31–41 of SA) with a missed cleavage at C-terminal of phenylalanine. The appearance of b₂ (m/z 242.1), b₂ – NH₃ (m/z 225.1), y₃ (m/z 383.2), $b_4*(m\angle 340.8 \text{ or } 343.1,$ doubly charged), $y_7(m\angle 871.3)$, $y_8*(m\angle 591.4 \text{ or } 594.1,$ doubly charged), y₉*(m/z 655.5 or 657.9, doubly charged), y₁₀*-NH₃ (m/z 710.7 or 713.2, doubly charged), as well as protonated HS-PhIP and HS- $[^2H_5]$ -PhIP ions (m/z 242.1 and 247.1) provided the evidence for PhIP adduction at LQQC*PFEDHVK. (Supporting Information, Figure S-6).

Another site of NO_2 -PhP modification of human SA was identified at Tyr¹⁵⁰. Three peptides containing the FY*AP, due to missed cleavages with chymotrypsin, were detected at t_R 22.8, 25.6 and 28.0 min (Table 2, Supporting Information, Figure S-7).

Data Dependent Analyses of SA in Human Plasma Reacted with a Mixture of NO2-PhIP and NO2-[2H5]-PhIP—The formation of PhIP-SA adduction products was also examined in human plasma treated with NO_2 -PhIP and NO_2 - $[^2H_5]$ -PhIP, because plasma proteins, fatty acids, drugs, and other organic compounds that are associated with SA can induce structural changes in the conformation of SA and affect the reactivity of the amino acid residues.^{40,43} Data-dependent scanning of the proteolytic digest with trypsin or trypsin/ chymotrypsin revealed that NO_2 -PhIP-modified peptides at Cys^{34} , Lys¹⁹⁵, Lys¹⁹⁹, Lys³⁵¹, Lys⁵⁴¹, Tyr⁴⁰¹, Tyr⁴¹¹ and Tyr¹⁵⁰ of SA, results consistent with the adduction products of NO2-PhIP formed with commercial human SA (Supporting Information, Figure S-8).

Targeted UPLC-ESI/MS2 Analysis of Human SA Modified with NO2-PhIP Following Protein Digestion with Trypsin—The sites of NO₂-PhIP binding to human SA reported above were characterized with SA that had been modified with a 3 mol excess of NO₂-PhIP. However, when the reaction of NO₂-PhIP with SA was performed with 10fold less carcinogen (0.3 mol NO₂-PhIP: 1 mol SA), targeted UPLC-ESI/MS² analysis revealed that the desamino-PhIP adducts were hardly formed at Lys^{195} , Lys^{199} , Lys^{541} , Cys^{34} , Tyr¹³⁸, and Tyr⁴⁰¹, and desamino-PhIP adduct formation at Tyr⁴¹¹ accounted for about 99% of the total ion counts of SA adducts (Figure 5). Thus, $Ty⁴¹¹$ is a primary binding site of $NO₂$ -PhIP.

Mass Spectrometric Characterization of Human SA Modified with HNOH-PhIP

Data-Dependent Analysis Human SA Modified with HONH-PhIP Following Protein Digestion with Pronase E, leucine aminopeptidase, and prolidase— The reaction of human SA with HONH-PhIP in vitro produced the N^2 -cysteinesulfinyl-PhIP; this sulfinamide linkage accounted for greater than 95% of the HONH-PhIP bound to $SA.²⁴$ The enzymatic digestion of HONH-PhIP- and HONH- $[²H₅]$ -PhIP-modified SA with Pronase E, leucine aminopeptidase, and prolidase resulted in extensive hydrolysis of the sulfinamide linkage, and targeted MS/MS analyses of the protein digest showed that PhIP was recovered in high yields.²⁴ The amino acid adduct N^2 -cysteinesulfinyl-PhIP was not detected in the 3-enzyme digest; however, data-dependent scanning identified two pairs of HONH-PhIP-tripeptide adducts ($t_R = 17.8$ and 19.3 min) (Supporting Information, Figure S-9, Table 3). The first pair of adducts was detected as singly charged species $[M+H]^+$ at m/ z 604.2 ($m/z = 609.2$ for $[^2H_5]$ -PhIP-peptide). The product ion spectra (Figures 6A and 6B) were in excellent agreement with the mass spectra previously reported for the N^2 cysteinesulfinyl-PhIP tripeptide C*PF (C-[S=O]-PhIP) and (C-[S=O]- $[^2H_5]$ -PhIP) adducts.²⁴ The second set of adducts ($t_R = 19.3$ min) was detected as singly charged species [M+H]⁺ at m/z 620.2 and 625.2, respectively, for the HONH-PhIP- and HONH- $[^2H_5]$ -PhIP-modified peptides. The mass of this adduct pair is 16 Da greater than the molecular mass of the C*PF

 $(C-[S=O]-PhIP$ and $C-[S=O]-[^2H₅]-PhIP$). The structure of this adduct pair is proposed to be the sulfonamide C*PF (C-[SO₂]-PhIP) and (C-[SO₂]- $[^2H_5]$ -PhIP). The assignment of the structure is supported by the product ion spectra (Figures 6C and 6D). The fragment ion observed at m/z 396.2 in the spectra of both modified peptides is attributed to the loss of PhIP (224.1 Da) and $[^{2}H_{5}]$ -PhIP (229.1 Da). Thus, the 2 oxygen atoms are associated with the C*PF peptide and not the PhIP moiety. The mass difference of 30 Da observed between the fragment ion at m/z 396.2 and the protonated CPF ([M+H]⁺ at m/z 366.1) is consistent with the presence two sulfur-oxygen double bonds on Cys^{34} . The ions at m/z 263.2, 313.2, 330.1, 358.1, 427.1 and 455.1 correspond to, respectively, y_2 , the C* immonium ion-NH₃, the C* immonium ion, and the b_1^* , a_2^* and b_2^* ions.

Data-Dependent Analysis Human SA Modified with HONH-PhIP Following

Protein Digestion with Trypsin/chymotrypsin—The PhIP sulfinamide linkage at Cys34 of SA undergoes hydrolysis to produce PhIP, by heat and treatment with DTT during the denaturation of SA (Peng, L., unpublished observations). However, the hexapeptide N^2 cysteinesulfinyl-PhIP LQQC*PF (C-[S=O]-PhIP, $t_R = 16.8$ min, $[M+2H]^{2+}$ m/z 487.4 and 489.7 for PhIP and $[{}^{2}H_{5}]$ -PhIP-modified peptides) was recovered from HONH-PhIPmodified SA, following digestion with trypsin/chymotrypsin without heat denaturation and DTT treatment. The missed cleavage adducts LQQC*PFEDHVK (C-[S=O]-PhIP) and LQQC*PFEDHVK (C-[SO₂]-PhIP) were also observed, respectively at $t_R = 11.3$ min and 13.4 min, occurring as triply protonated species $[M+3H]^{3+}$ (Supporting Information, Figure S-10 and Figure S-11).

Mass Spectrometric Characterization of *N***-Acetoxy-PhIP and its Adduction Products with Human SA**

^N-Acetoxy-PhIP is a penultimate metabolite of PhIP that reacts with DNA to form stable covalent adducts.^{28,29,44} The conditions of synthesis of N-acetoxy-PhIP were optimized to produce this reactive intermediate in high yield with minimal side-reaction products (Figure 7A). The amount of N-acetoxy-PhIP was about 100-fold greater than the amount of its inactive isomer, the hydroxamic acid, N -hydroxy- N -(1-methyl-6-phenylimidazo[4,5^b]pyridin-2-yl)-acetamide (N-hydroxy-N-acetyl-PhIP), assuming comparable ionization efficiencies of the compounds by ESI/MS. The structures of these acetylated isomers of HONH-PhIP ($[M+H]^+$ at m/z 283.1) are readily distinguished by their product ion spectra (Figures 7B and 7C). The product ion spectrum of N-acetoxy-PhIP contains major fragment ions at $m/z 223.0$ and 224.0, corresponding, respectively, to losses of $CH_3CO_2^{\bullet}$ and $CH₃CO₂H$, whereas the product ion spectrum of N-hydroxy-N-acetyl-PhIP contains a base peak ion at $m/z 241.1$, which is attributed to the loss of ketene (CH₂CO).

Data-Dependent Analysis Human SA Modified with N-Acetoxy-PhIP Following Protein Digestion with Pronase E, Leucine Aminopeptidase, and Prolidase—

The UV spectrum of human SA modified with N-acetoxy-PhIP displays a maximum at 320 – 325 nm, whereas non-modified SA does not possess a chromophore around these wavelengths. The level of chemical modification of SA by N-acetoxy-PhIP was estimated at \sim 170 pmol PhIP/nmol SA, based on the assumption that N-acetoxy-PhIP-SA adducts possess molar absorption coefficients that are comparable to that of PhIP (315 nm; ε (M⁻¹) cm−1) at 22,220). However, the isotopic data-dependent scanning was not successful in detecting SA adducts of N-acetoxy-PhIP, irrespective of the enzymatic digestion conditions. Nucleophilic amino acids of SA were expected to react with N-acetoxy-PhIP or the proposed nitrenium ion⁴⁵ to form adducts with amino acids or peptides with mass increments of 223 or 228 Da, attributed respectively, to reactive PhIP or $[^2H_5]$ -PhIP species, less one proton from the amino acid residue. Although data-dependent scanning analyses failed to detect stable covalent adducts, targeted UPLC-ESI/MS² analysis of the N-acetoxy-

PhIP-modified SA digested with the Pronase E, amino leucine peptidase, and prolidase revealed the presence of high levels of PhIP and 5-HO-PhIP. The identities of PhIP (data not shown), and 5-HO-PhIP and 5-HO- $[^2H_5]$ -PhIP were corroborated by their product ion spectra (Figure 8). Pretreatment of SA with the selective thiol reagents 4-CMB or NEM prior to reaction with N-acetoxy-PhIP, decreased the amounts of PhIP and 5-HO-PhIP recovered from the proteolytic digest by 3-fold. This finding suggests that a substantial portion of the N-acetoxy-PhIP had bound to Cys^{34} as unstable adducts that underwent hydrolysis to form PhIP and 5-HO-PhIP, during proteolytic digestion.

Discussion

NO2-PhIP, HONH-PhIP, and N-acetoxy-PhIP are reactive species that contribute to the deleterious biological effects of PhIP through DNA and protein adduct formation.^{10,46} Metabolites of PhIP are known to bind to human SA in vivo;¹³ however, the structures of the adducts remain to be elucidated. Human SA is comprised of three homologous domains and each domain contains two subdomains.⁴⁷ Studies have shown that micro environments of pH within the different domains of SA can lower the pKa values of some nucleophilic side chain groups of amino acids and enhance their reactivity with electrophiles $43,48$ Moreover, the tertiary structure of SA can exert an influence on noncovalent proteinelectrophile interactions and direct ensuing covalent adduct formation between specific amino acids and electrophiles.49 Thus, we had anticipated that nucleophilic amino acids of SA would display different degrees of reactivity towards these distinct electrophilic metabolites of PhIP, resulting in an array of covalent adducts of PhIP within different sequence locations of SA.

We employed a two tier approach to map the sites of human SA modification with $NO₂$ -PhIP, HONH-PhIP and N-acetoxy-PhIP. In the first approach, SA samples modified with metabolites of PhIP were enzymatically digested with a mixture of Pronase E, leucine aminopeptidase, and prolidase, to produce amino acid containing adducts.37,38 Datadependent scanning of $NO₂$ -PhIP-modified SA revealed that adducts were formed with Cys, Lys, and Tyr. This 3-enzyme mixture did not generate amino acid adducts of SA modified with HONH-PhIP, but produced tripeptide adducts of PhIP formed at Cys^{34} . The Cys^{34} residue of human SA is known to react with many genotoxicants and toxic electrophiles to form adducts with acrylamide,⁵⁰ nitrogen mustard,⁵¹ α , β -unsaturated aldehydes,⁵² the neurotoxin brevetoxin B,⁵³ acetaminophen,⁵⁴ benzene,⁵⁵ 2-amino-3-methylimidazo[4,5flquinoline (IQ),³¹ as well as PhIP.^{22–24} Many of these adducts do not undergo digestion beyond the tripeptide stage with Pronase. Thus, the efficacy of proteoloytic digestion of chemically-modified SA is highly dependent upon structure of the toxicant and the nature of the bond formed between the toxicant and the sulfhydryl group of Cys^{34} .

With knowledge about the primary amino acids that had formed adducts with PhIP metabolites, the sequence locations of PhIP adducts were mapped in tryptic or tryptic/ chymotrypic digests obtained from SA modified with a 3-fold excess of an equimolar mixture of unlabeled and $[^{2}H_{5}]$ -labeled PhIP metabolites. The dual isotope label provided a characteristic fingerprint for the unambiguous identification of PhIP-modified peptides. Data-dependent scanning of $NO₂$ -PhIP-modified SA enabled us to identify adducts at nine sequence locations: Lys¹⁹⁵, Lys¹⁹⁹, Lys³⁵¹, Lys⁵⁴¹, Cys³⁴, Tyr¹³⁸, Tyr¹⁵⁰, Tyr⁴⁰¹ and $Ty⁴¹¹$. The identification of SA modifications by N-oxidized derivatives of PhIP was greatly accelerated, by the MyriMatch algorithm over manual interpretation. However, the major adduct formed at Tyr^{411} , Y*TK, escaped detection by MyriMatch, probably because there was an insufficient number of y and b ions to identify the modified peptide by the algorithm (Figure 3C). Similarly, the C*PF adducts formed with NO-PhIP that were recovered from the 3-enzyme digest also escaped detection by MyriMatch. The employment

of the isotopic mixture of the N-oxidized PhIP metabolites was critical for the successful identification of these tripeptide adducts by manual inspection.

The major site of binding of NO_2 -PhIP to SA occurred at Tyr⁴¹¹, this amino acid is present in the center of a binding pocket with Arg⁴¹⁰ in subdomain IIIA of human SA.⁴⁸ The phenolic HO group of tyrosine has a pKa of \sim 10, whereas the pKa of phenolic HO group of Tyr⁴¹¹ is lowered to 7.9 due to the electrostatic interaction with guanidinium ω -NH of Arg^{410.56} Tyr⁴¹¹ is a major site for covalent adduct with nitrophenyl acetate, 48 organophosphorous pesticides,57 and nerve agents (tabun, sarin, cyclosarin, VX); secondary adduction sites with many of these toxicants occurs at Tyr^{148} , Tyr^{150} and Tyr^{161} of SA.^{57–60}

 $NO₂$ -PhIP also formed adducts with SA at Lys¹⁹⁵, Lys¹⁹⁹, and Lys⁵⁴¹. The Lys¹⁹⁵ and Lys^{199} residues occur at the interface of subdomains IB and IIA, and Lys^{541} resides within subdomain III B.47 These lysine residues are known to form acyl-linked adducts with glucuronide conjugates of tolmetin and permethrin.^{61,62} Lys¹⁹⁵ or Lys¹⁹⁹ was also reported to form an adduct with $2,4$ -dinitro-1-chlorobenzene,⁶³ and both lysine residues are reactive sites in SA and form adducts with 4-hydroxy-trans-2-nonenal.⁵²

 Cys^{34} is another reactive site in human SA that forms adducts with NO₂-PhIP. The Cys³⁴ resides in a shallow crevice of SA and predominantly exists in the thiolate form due to its interaction with three ionizable residues, Asp^{38} , His³⁹ and Tyr⁸⁴, which are within close vicinity because of the tertiary structure of SA.^{43,48,64} As a result, the thiol of Cys^{34} has an unusually low pKa value of 6.5 compared to about $8.0 - 8.5$ in many other proteins or peptides at physiological pH.^{43,52} These molecular and structural features of SA can help to explain the high reactivity of Cys^{34} towards many low molecular weight electrophiles of diverse structure.^{20,65}

The sites of $NO₂$ -PhIP binding to human SA were characterized with SA that had been modified with a 3 mol excess of $NO₂$ -PhIP. However, when the reaction of $NO₂$ -PhIP with SA was performed with a limiting amount of carcinogen, desamino-PhIP-Y*TK adduct formation at Tyr411 accounted for about 99% of the total ion counts of SA adducts. These binding data indicate that Tyr^{411} is a primary binding site of NO₂-PhIP and a likely adduction site of $NO₂$ -PhIP in vivo.

The adduction of HONH-PhIP to SA was found to occur only with Cys^{34} ; this data is in agreement with our previous finding.²⁴ The adduct exists as the sulfur-nitrogen linked N^2 -[cystein-S-yl-PhIP]-S-oxide. Therefore, the reactivity of HONH-PhIP with SA was poor and protein adduct formation only occurred following the oxidation of HONH-PhIP to NO-PhIP. The sulfhydryl SH of Cys^{34} reacted with the N=O bond of NO-PhIP to form a semimercaptal, which underwent rearrangement to the more stable sulfinamide structure.⁶⁶ We also identified the sulfonamide adduct of PhIP formed at Cys³⁴; this adduct was not previously reported and its mechanism of formation is uncertain.24 The reaction of HONH-PhIP with oxygen results in the production of NO-PhIP and superoxide anion.⁶⁷ The superoxide anion may have oxidized a portion of the $Cys³⁴$ to the sulfinic acid prior to its reaction with N -oxidized-PhIP, resulting in formation of the sulfonamide adduct.⁶⁸ Alternatively, the superoxide anion generated over the time, by HONH-PhIP, may have oxidized a portion of the N^2 -[cystein-S-yl-PhIP]-S-oxide linkage to the sulfonamide structure; the oxidation of N^2 -[cystein-S-yl-PhIP]-S-oxide to the sulfonamide adduct during enzymatic digestion also cannot be excluded.⁶⁹

^N-Acetoxy-PhIP, a penultimate metabolite of PhIP that forms stable covalent adducts at appreciable levels with DNA, 29,44 was the least efficient of the N-oxidized metabolites of PhIP to react with human SA and form stable covalent adducts. Both HONH-PhIP and Nacetoxy-PhIP bind to DNA, almost exclusively at $dG^{29,44,70}$ DNA adduct formation with N-

acetoxy-PhIP is believed to occur through the proposed nitrenium ion⁴⁵ via an S_N1 or an intermediate S_N 2 mechanism.⁷¹ N-Acetoxy-PhIP undergoes solvolysis more facilely than HONH-PhIP, to produce the nitrenium ion, and explains the superior reactivity of Nacetoxy-PhIP towards DNA.45 Adduct formation of N-acetoxy-PhIP with SA was expected to occur by a similar mechanism. Although stable adducts of N-acetoxy-PhIP were not detected, appreciable levels of PhIP and 5-HO-PhIP were recovered from the N-acetoxy-PhIP-treated SA digested with Pronase E, leucine aminopeptidase, and prolidase. The pretreatment of SA with the selective thiol reagents 4-CMB or NEM decreased the amounts of PhIP and 5-HO-PhIP recovered from the digests of N-acetoxy-PhIP-modified SA by about 3-fold. These findings point to the formation of unstable adducts of PhIP at Cys³⁴. A previous study reported that certain adducts formed between N-acetoxy-PhIP and rat SA were unstable and underwent hydrolysis to produce 5-HO-PhIP.²² The reaction of N acetoxy-PhIP with cysteine or glutathione also produced labile adducts, tentatively assigned as sulfenamide conjugates, which underwent hydrolysis to form 5-HO-PhIP.²² Thus, a large portion of the PhIP and 5-HO-PhIP recovered from N-acetoxy-PhIP bound to rat or human SA may be attributed to an unstable sulfenamide linked adduct formed at Cys^{34} (Scheme 2).

The Tyr⁴¹¹ adduct formed with NO_2 -PhIP and the Cys³⁴ sulfinamide adduct formed with NO-PhIP are distinct structures derived from different N-oxidized metabolites of PhIP; however, both adducts are biomarkers of the genotoxic metabolite, HONH-PhIP. The N-Hydroxy metabolites of aromatic amines and HAAs are well known to undergo oxidation, by enzymatic and non-enzymatic chemistry, to form their corresponding nitroso derivatives.^{67,72,73} We have also observed that NO_2 -PhIP is formed from HONH-PhIP, through the NO-PhIP intermediate, under aerobic conditions in phosphate buffered saline (pH 7.4) (Turesky R, unpublished observations). Moreover, $NO₂$ -PhIP is formed during the metabolism of PhIP in hepatocytes of rats pretreated with polychlorinated biphenyls, and the desamino-PhIP-gluathione conjugate, glutathionyl-1-methyl-6-phenylimidazo[4,5 ^b]pyridine has been detected in bile of rats exposed to PhIP.74 The pathway by which NO2- PhIP is formed in hepatocytes or in liver of rodents is not clear, but the findings indicate that NO2-PhIP occurs during the N-oxidation of PhIP.

In summary, our mapping study of N-oxidized metabolites of PhIP adducted to human SA point to the Tyr⁴¹¹ and Cys³⁴ residues as major sites of binding, respectively, for NO₂-PhIP and NO-PhIP. Sensitive tandem MS methods are under development to determine if these SA adducts of PhIP are present in humans. Our goal is to implement stable covalent SA protein adducts of PhIP as biomarkers in molecular epidemiology studies to measure exposure and assess the role of this HAA in diet-related cancers.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

Comments and discussion of the data with Dr. Paul Vouros, Northeastern University; Dr. Paul Skipper, MIT; and Dr. Dan Liebler, Vanderbilt University, are greatly appreciated.

Funding Source: This research was supported by grant R01 CA122320 (L.P. and R.J.T.), and R01 CA126479 (D.T. and S.D.) from the National Cancer Institute.

Abbreviations

PhIP 2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine

Reference List

- 1. Felton, JS.; Jagerstad, M.; Knize, MG.; Skog, K.; Wakabayashi, K. Contents in foods, beverages and tobacco. In: Nagao, M.; Sugimura, T., editors. Food Borne Carcinogens Heterocyclic Amines. John Wiley & Sons Ltd; Chichester, England: 2000. p. 31-71.
- 2. Sinha R, Rothman N, Brown ED, Salmon CP, Knize MG, Swanson CS, Rossi SC, Mark SD, Levander OA, Felton JS. High concentrations of the carcinogen 2-amino-1-methyl-6 phenylimidazo[4,5-b]pyridine (PhIP) occur in chicken but are dependent on the cooking method. Cancer Res. 1995; 55:4516–4519. [PubMed: 7553619]
- 3. Keating GA, Bogen KT. Estimates of heterocyclic amine intake in the US population. J Chromatogr B Analyt Technol Biomed Life Sci. 2004; 802:127–133.
- 4. Sinha R. An epidemiologic approach to studying heterocyclic amines. Mutat Res. 2002; 506– 507:197–204.
- 5. Knize MG, Felton JS. Formation and human risk of carcinogenic heterocyclic amines formed from natural precursors in meat. Nutr Rev. 2005; 63:158–165. [PubMed: 15971410]
- 6. Zheng W, Lee SA. Well-done meat intake, heterocyclic amine exposure, and cancer risk. Nutr Cancer. 2009; 61:437–446. [PubMed: 19838915]
- 7. National Toxicology Program. Report on Carcinogenesis. 11. U.S. Department of Health and Human Services, Public Health Service; Research Triangle Park, N.C: 2005.
- 8. Jarabek AM, Pottenger LH, Andrews LS, Casciano D, Embry MR, Kim JH, Preston RJ, Reddy MV, Schoeny R, Shuker D, Skare J, Swenberg J, Williams GM, Zeiger E. Creating context for the use of

DNA adduct data in cancer risk assessment: I. Data organization. Crit Rev Toxicol. 2009; 39:659– 678. [PubMed: 19743944]

- 9. Skipper PL, Tannenbaum SR. Protein adducts in the molecular dosimetry of chemical carcinogens. Carcinogenesis. 1990; 11:507–518. [PubMed: 2182215]
- 10. Turesky RJ, Le Marchand L. Metabolism and biomarkers of heterocyclic aromatic amines in molecular epidemiology studies: lessons learned from aromatic amines. Chem Res Toxicol. 2011; 24:1169–1214. [PubMed: 21688801]
- 11. Zhu J, Chang P, Bondy ML, Sahin AA, Singletary SE, Takahashi S, Shirai T, Li D. Detection of 2 amino-1-methyl-6-phenylimidazo[4,5-b]-pyridine-DNA adducts in normal breast tissues and risk of breast cancer. Cancer Epidemiol Biomarkers Prev. 2003; 12:830–837. [PubMed: 14504191]
- 12. Zhu J, Rashid A, Cleary K, Abbruzzese JL, Friess H, Takahashi S, Shirai T, Li D. Detection of 2 amino-1-methyl-6-phenylimidazo [4,5-b]-pyridine (PhIP)-DNA adducts in human pancreatic tissues. Biomarkers. 2006; 11:319–328. [PubMed: 16908439]
- 13. Dingley KH, Curtis KD, Nowell S, Felton JS, Lang NP, Turteltaub KW. DNA and protein adduct formation in the colon and blood of humans after exposure to a dietary-relevant dose of 2 amino-1-methyl-6-phenylimidazo[4,5-b]pyridine. Cancer Epidemiol Biomarkers Prev. 1999; 8:507–512. [PubMed: 10385140]
- 14. Bessette EE, Spivack SD, Goodenough AK, Wang T, Pinto S, Kadlubar FF, Turesky RJ. Identification of carcinogen DNA adducts in human saliva by linear quadrupole ion trap/ multistage tandem mass spectrometry. Chem Res Toxicol. 2010; 23:1234–1244. [PubMed: 20443584]
- 15. Liebler DC. Proteomic approaches to characterize protein modifications: new tools to study the effects of environmental exposures. Environ Health Perspect. 2002; 110(Suppl 1):3–9. [PubMed: 11834459]
- 16. Tornqvist M, Fred C, Haglund J, Helleberg H, Paulsson B, Rydberg P. Protein adducts: quantitative and qualitative aspects of their formation, analysis and applications. J Chromatogr B Analyt Technol Biomed Life Sci. 2002; 778:279–308.
- 17. Bryant MS, Vineis P, Skipper PL, Tannenbaum SR. Hemoglobin adducts of aromatic amines: associations with smoking status and type of tobacco. Proc Natl Acad Sci U S A. 1988; 85:9788– 9791. [PubMed: 3200858]
- 18. Kiese M, Taeger K. The fate of phenylhydroxylamine in human red cells. Naunyn Schmiedebergs Arch Pharmacol. 1976; 292:59–66. [PubMed: 934354]
- 19. Ringe D, Turesky RJ, Skipper PL, Tannenbaum SR. Structure of the single stable hemoglobin adduct formed by 4-aminobiphenyl in vivo. Chem Res Toxicol. 1988; 1:22–24. [PubMed: 2979706]
- 20. Rubino FM, Pitton M, Di FD, Colombi A. Toward an "omic" physiopathology of reactive chemicals: thirty years of mass spectrometric study of the protein adducts with endogenous and xenobiotic compounds. Mass Spectrom Rev. 2009; 28:725–784. [PubMed: 19127566]
- 21. Magagnotti C, Orsi F, Bagnati R, Celli N, Rotilio D, Fanelli R, Airoldi L. Effect of diet on serum albumin and hemoglobin adducts of 2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PhIP) in humans. Int J Cancer. 2000; 88:1–6. [PubMed: 10962432]
- 22. Reistad R, Frandsen H, Grivas S, Alexander J. In vitro formation and degradation of 2-amino-1 methyl-6- phenylimidazo[4,5-b]pyridine (PhIP) protein adducts. Carcinogenesis. 1994; 15:2547– 2552. [PubMed: 7955104]
- 23. Chepanoske CL, Brown K, Turteltaub KW, Dingley KH. Characterization of a peptide adduct formed by N-acetoxy-2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PhIP), a reactive intermediate of the food carcinogen PhIP. Food Chem Toxicol. 2004; 42:1367–1372. [PubMed: 15207388]
- 24. Peng L, Turesky RJ. Mass spectrometric characterization of 2-amino-1-methyl-6 phenylimidazo[4,5-b]pyridine N-oxidized metabolites bound at Cys34 of human serum albumin. Chem Res Toxicol. 2011; 24:2004–2017. [PubMed: 21916490]
- 25. Ma S, Subramanian R. Detecting and characterizing reactive metabolites by liquid chromatography/tandem mass spectrometry. J Mass Spectrom. 2006; 41:1121–1139. [PubMed: 16967439]

- 26. Tzouros M, Pahler A. A targeted proteomics approach to the identification of peptides modified by reactive metabolites. Chem Res Toxicol. 2009; 22:853–862. [PubMed: 19317514]
- 27. Turesky RJ, Lang NP, Butler MA, Teitel CH, Kadlubar FF. Metabolic activation of carcinogenic heterocyclic aromatic amines by human liver and colon. Carcinogenesis. 1991; 12:1839–1845. [PubMed: 1934265]
- 28. Goodenough AK, Schut HA, Turesky RJ. Novel LC-ESI/MS/MSⁿ method for the characterization and quantification of 2′-deoxyguanosine adducts of the dietary carcinogen 2-amino-1-methyl-6 phenylimidazo[4,5-b]pyridine by 2-D linear quadrupole ion trap mass spectrometry. Chem Res Toxicol. 2007; 20:263–276. [PubMed: 17305409]
- 29. Lin D, Kaderlik KR, Turesky RJ, Miller DW, Lay JO Jr, Kadlubar FF. Identification of N- (deoxyguanosin-8-yl)-2-amino-1-methyl-6-phenylimidazo [4,5-b]pyridine as the major adduct formed by the food-borne carcinogen, 2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine, with DNA. Chem Res Toxicol. 1992; 5:691–697. [PubMed: 1446011]
- 30. Langouët S, Paehler A, Welti DH, Kerriguy N, Guillouzo A, Turesky RJ. Differential metabolism of 2-amino-1-methyl-6-phenylimidazo[4,5- b]pyridine in rat and human hepatocytes. Carcinogenesis. 2002; 23:115–122. [PubMed: 11756232]
- 31. Turesky RJ, Skipper PL, Tannenbaum SR. Binding of 2-amino-3-methylimidazo[4,5-f]quinoline to hemoglobin and albumin in vivo in the rat. Identification of an adduct suitable for dosimetry. Carcinogenesis. 1987; 8:1537–1542. [PubMed: 3652389]
- 32. Boyer PD. Spectrophotometric Study of the Reaction of Protein Sulfhydryl Groups with Organic Mercurials. J Am Chem Soc. 1954; 76:4331–4337.
- 33. Riener CK, Kada G, Gruber HJ. Quick measurement of protein sulfhydryls with Ellman's reagent and with 4,4′-dithiodipyridine. Anal Bioanal Chem. 2002; 373:266–276. [PubMed: 12110978]
- 34. Kessner D, Chambers M, Burke R, Agus D, Mallick P. ProteoWizard: open source software for rapid proteomics tools development. Bioinformatics. 2008; 24:2534–2536. [PubMed: 18606607]
- 35. Tabb DL, Fernando CG, Chambers MC. MyriMatch: highly accurate tandem mass spectral peptide identification by multivariate hypergeometric analysis. J Proteome Res. 2007; 6:654–661. [PubMed: 17269722]
- 36. Ma ZQ, Dasari S, Chambers MC, Litton MD, Sobecki SM, Zimmerman LJ, Halvey PJ, Schilling B, Drake PM, Gibson BW, Tabb DL. IDPicker 2.0: Improved protein assembly with high discrimination peptide identification filtering. J Proteome Res. 2009; 8:3872–3881. [PubMed: 19522537]
- 37. Tsao M, Otter DE. Quantification of glutamine in proteins and peptides using enzymatic hydrolysis and reverse-phase high-performance liquid chromatography. Anal Biochem. 1999; 269:143–148. [PubMed: 10094785]
- 38. Baxter JH, Lai CS, Phillips RR, Dowlati L, Chio JJ, Luebbers ST, Dimler SR, Johns PW. Direct determination of methionine sulfoxide in milk proteins by enzyme hydrolysis/high-performance liquid chromatography. J Chromatogr A. 2007; 1157:10–16. [PubMed: 17467723]
- 39. Papayannopoulos IA. The interpretation of collision-induced dissociation tandem mass spectra of peptides. Mass Spectrom Rev. 1995; 14:49–73.
- 40. Peters T Jr. Serum albumin. Adv Protein Chem. 1985; 37:161–245. [PubMed: 3904348]
- 41. Steen H, Mann M. The ABC's (and XYZ's) of peptide sequencing. Nat Rev Mol Cell Biol. 2004; 5:699–711. [PubMed: 15340378]
- 42. Delatour T, Richoz J, Vouros P, Turesky RJ. Simultaneous determination of 3-nitrotyrosine and tyrosine in plasma proteins of rats and assessment of artifactual tyrosine nitration. J Chromatogr B. 2002; 779:189–199.
- 43. Stewart AJ, Blindauer CA, Berezenko S, Sleep D, Tooth D, Sadler PJ. Role of Tyr84 in controlling the reactivity of Cys34 of human albumin. FEBS J. 2005; 272:353–362. [PubMed: 15654874]
- 44. Frandsen H, Grivas S, Andersson R, Dragsted L, Larsen JC. Reaction of the N^2 -acetoxy derivative of 2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PhIP) with 2′-deoxyguanosine and DNA. Synthesis and identification of N^2 -(2′-deoxyguanosin-8-yl)-PhIP. Carcinogenesis. 1992; 13:629– 635. [PubMed: 1576716]

- 45. Nguyen TM, Novak M. Synthesis and decomposition of an ester derivative of the procarcinogen and promutagen, PhIP, 2-amino-1-methyl-6-phenyl-1H-imidazo[4,5-b]pyridine: unusual nitrenium ion chemistry. J Org Chem. 2007; 72:4698–4706. [PubMed: 17542636]
- 46. Glatt, H. Metabolic factors affecting the mutagenicity of heteroyclic amines. In: Skog, K.; Alexander, J., editors. Acrylamide and Other Hazardous Compounds in Heat-Treated Foods. Woodhead Publishing Ltd; Cambridge, England: 2006. p. 358-404.
- 47. Sugio S, Kashima A, Mochizuki S, Noda M, Kobayashi K. Crystal structure of human serum albumin at 2.5 A resolution. Protein Eng. 1999; 12:439–446. [PubMed: 10388840]
- 48. Carter DC, Ho JX. Structure of serum albumin. Adv Protein Chem. 1994; 45:153–203. [PubMed: 8154369]
- 49. Skipper PL. Influence of tertiary structure on nucleophilic substitution reactions of proteins. Chem Res Toxicol. 1996; 9:918–923. [PubMed: 8870977]
- 50. Noort D, Fidder A, Hulst AG. Modification of human serum albumin by acrylamide at cysteine-34: a basis for a rapid biomonitoring procedure. Arch Toxicol. 2003; 77:543–545. [PubMed: 12819856]
- 51. Noort D, Hulst AG, Jansen R. Covalent binding of nitrogen mustards to the cysteine-34 residue in human serum albumin. Arch Toxicol. 2002; 76:83–88. [PubMed: 11914777]
- 52. Aldini G, Regazzoni L, Orioli M, Rimoldi I, Facino RM, Carini M. A tandem MS precursor-ion scan approach to identify variable covalent modification of albumin Cys34: a new tool for studying vascular carbonylation. J Mass Spectrom. 2008; 43:1470–1481. [PubMed: 18457351]
- 53. Wang Z, Ramsdell JS. Analysis of interactions of brevetoxin-B and human serum albumin by liquid chromatography/mass spectrometry. Chem Res Toxicol. 2011; 24:54–64. [PubMed: 21142195]
- 54. Damsten MC, Commandeur JN, Fidder A, Hulst AG, Touw D, Noort D, Vermeulen NP. Liquid chromatography/tandem mass spectrometry detection of covalent binding of acetaminophen to human serum albumin. Drug Metab Dispos. 2007; 35:1408–1417. [PubMed: 17510247]
- 55. Bechtold WE, Willis JK, Sun JD, Griffith WC, Reddy TV. Biological markers of exposure to benzene: S-phenylcysteine in albumin. Carcinogenesis. 1992; 13:1217–1220. [PubMed: 1638689]
- 56. Ahmed N, Dobler D, Dean M, Thornalley PJ. Peptide mapping identifies hotspot site of modification in human serum albumin by methylglyoxal involved in ligand binding and esterase activity. J Biol Chem. 2005; 280:5724–5732. [PubMed: 15557329]
- 57. John H, Breyer F, Thumfart JO, Hochstetter H, Thiermann H. Matrix-assisted laser desorption/ ionization time-of-flight mass spectrometry (MALDI-TOF MS) for detection and identification of albumin phosphylation by organophosphorus pesticides and G- and V-type nerve agents. Anal Bioanal Chem. 2010; 398:2677–2691. [PubMed: 20730528]
- 58. Li B, Schopfer LM, Hinrichs SH, Masson P, Lockridge O. Matrix-assisted laser desorption/ ionization time-of-flight mass spectrometry assay for organophosphorus toxicants bound to human albumin at Tyr411. Anal Biochem. 2007; 361:263–272. [PubMed: 17188226]
- 59. Noort D, Hulst AG, van ZA, van RE, van der Schans MJ. Covalent binding of organophosphorothioates to albumin: a new perspective for OP-pesticide biomonitoring? Arch Toxicol. 2009; 83:1031–1036. [PubMed: 19575182]
- 60. Li B, Nachon F, Froment MT, Verdier L, Debouzy JC, Brasme B, Gillon E, Schopfer LM, Lockridge O, Masson P. Binding and hydrolysis of soman by human serum albumin. Chem Res Toxicol. 2008; 21:421–431. [PubMed: 18163544]
- 61. Ding A, Zia-Amirhosseini P, McDonagh AF, Burlingame AL, Benet LZ. Reactivity of tolmetin glucuronide with human serum albumin. Identification of binding sites and mechanisms of reaction by tandem mass spectrometry. Drug Metab Dispos. 1995; 23:369–376. [PubMed: 7628303]
- 62. Noort D, van ZA, Fidder A, van OB, Hulst AG. Protein adduct formation by glucuronide metabolites of permethrin. Chem Res Toxicol. 2008; 21:1396–1406. [PubMed: 18549292]
- 63. Aleksic M, Pease CK, Basketter DA, Panico M, Morris HR, Dell A. Investigating protein haptenation mechanisms of skin sensitisers using human serum albumin as a model protein. Toxicol In Vitro. 2007; 21:723–733. [PubMed: 17317089]

- 64. Christodoulou J, Sadler PJ. 1H NMR of albumin in human blood plasma: drug binding and redox reactions at Cys34. FEBS Lett. 1995; 376:1–5. [PubMed: 8521951]
- 65. Rappaport SM, Li H, Grigoryan H, Funk WE, Williams ER. Adductomics: Characterizing exposures to reactive electrophiles. Toxicol Lett. In Press.
- 66. Dolle B, Topner W, Neumann HG. Reaction of arylnitroso compounds with mercaptans. Xenobiotica. 1980; 10:527–536. [PubMed: 7445520]
- 67. Hiramoto K, Negishi K, Namba T, Katsu T, Hayatsu H. Superoxide dismutase-mediated reversible conversion of 3-hydroxyamino-1-methyl-5H-pyrido[4,3-b]indole, the N-hydroxy derivative of Trp-P-2, into its nitroso derivative. Carcinogenesis. 1988; 9:2003–2008. [PubMed: 2846195]
- 68. Umemoto A, Grivas S, Yamaizumi Z, Sato S, Sugimura T. Non-enzymatic glutathione conjugation of 2-nitroso-6-methyldipyrido [1,2-a: 3′,2′-d] imidazole (NO-Glu-P-1) in vitro: N-hydroxysulfonamide, a new binding form of arylnitroso compounds and thiols. Chem -Biol Interact. 1988; 68:57–69. [PubMed: 3203408]
- 69. Finch JW, Crouch RK, Knapp DR, Schey KL. Mass spectrometric identification of modifications to human serum albumin treated with hydrogen peroxide. Arch Biochem Biophys. 1993; 305:595– 599. [PubMed: 8373198]
- 70. Tang Y, LeMaster DM, Nauwelaers G, Gu D, Langouet S, Turesky RJ. UDP-Glucuronosyltransferase-mediated metabolic activation of the tobacco carcinogen 2-amino-9Hpyrido[2,3-b]indole. J Biol Chem. 2012; 287:14960–14972. [PubMed: 22393056]
- 71. Bentley TW, Schleyer PR. Role of nucleophilic solvent assistance and nucleophilically solvated ion pair intermediates in solvolysea of primary and secondary arenesulfonates. J Am Chem Soc. 1976; 98:7658–7666.
- 72. Kim D, Kadlubar FF, Teitel CH, Guengerich FP. Formation and reduction of aryl and heterocyclic nitroso compounds and significance in the flux of hydroxylamines. Chem Res Toxicol. 2004; 17:529–536. [PubMed: 15089095]
- 73. Lindeke B. The Non- and postenzymatic chemistry of N-oxygenated molecules. Drug Metab Rev. 1982; 13:71–121. [PubMed: 7044734]
- 74. Alexander J, Wallin H, Rossland OJ, Solberg KE, Holme JA, Becher G, Andersson R, Grivas S. Formation of a glutathione conjugate and a semistable transportable glucuronide conjugate of N^2 oxidized species of 2-amino-1-methyl-6-phenylimidazo[4,5-b]pyridine (PhIP) in rat liver. Carcinogenesis. 1991; 12:2239–2245. [PubMed: 1747923]

Figure 1.

Product ion mass spectra of desamino-PhIP-K ([M+H]⁺, m/z 354.2) recovered from NO₂-PhIP-modified SA (upper panel) and desamino- $[^2H_5]$ -PhIP-K ([M+H]⁺, m/z 359.2) recovered from $NO₂$ -[²H₅]-PhIP modified SA (bottom panel), following digestion of with Pronase E, leucine aminopeptidase, and prolidase.

Figure 2.

Product ion mass spectra of A) desamino-PhIP-Y ($[M+H]^+$ at m/z 389.2) recovered from NO₂-PhIP-modified SA; B) desamino- $[^2H_5]$ -PhIP-Y ($[M+H]^+$ at m/z 394.2) recovered from NO2-[2H5]-PhIP-modified SA, following digestion with Pronase E, leucine aminopeptidase, and prolidase; C) second generation product ion spectrum of the ion at m/z 315.2; D) second generation product ion spectrum of ion at m/z 343.2; and E) third generation product ion spectrum of m/z 226.2 (2-HO-PhIP) from desamino-PhIP-Y.

Figure 3.

Data-dependent MS/MS scanning with mass tags of NO_2 -PhIP and NO_2 -[²H₅]-PhIPmodified SA following trypsin digestion, either unmodified (upper panel) or SA modified with a 3 mol excess of an equimolar mixture of NO_2 -PhIP and NO_2 - $[{}^{2}H_5]$ -PhIP modified (bottom panel). The chromatograms of the MS/MS data were acquired on the ions exhibiting a difference of m/z 2.5 in doubly charged form and a difference of m/z 1.67 in triply charged form, employing the mass tags scanning option with a partner intensity ratio of 65 to 100% for PhIP/[2H5]-PhIP. The ion intensities are normalized to the same scale for untreated and NO2-PhIP-modified SA.

Peng et al. Page 24

Figure 4.

Product ion mass spectra of A) Fraction 1, ASSAK*QR (desamino-PhIP-K) adduct ([M +2H]²⁺ at m/z 477.8) recovered from NO₂-PhIP-modified SA; B) ASSAK*QR (desamino-[²H₅]-PhIP-K) adduct ([M+2H]²⁺ at *m/z* 480.3) recovered from NO₂-[²H₅]-PhIP modified SA; C) Fraction 5, Y*TK (desamino-PhIP-Y) adduct ($[M+2H]^{2+}$ at m/z 309.9) recovered from NO₂-PhIP-modified; D) Y*TK (desamino-[²H₅]-PhIP-Y) adduct ([M+2H]²⁺ at m/z 312.4); E) Fraction 8, Y*LYEIAR (desamino-PhIP-Y) adduct ([M+2H]²⁺ at m/z 568.1) recovered from NO₂-PhIP-modified SA; and F) Y*LYEIAR (desamino- $[^2H_5]$ -PhIP-Y) adduct ($[M+2H]^2$ ⁺ at m/z 570.6) recovered from NO₂-[²H₅]-PhIP modified SA by trypsin digestion.

Figure 5.

UPLC-ESI/MS² mass chromatograms of adducts of NO₂-PhIP-modified SA formed by reaction of the carcinogen with human SA at a molar ratio of 3:1 (left panel) and 0.3:1 (right panel); the SA was digested with trypsin. Targeted ESI/MS² was done on the following modified peptides: Fraction 1, ASSAK*QR as doubly charged species ($[M+2H]^{2+}$ at m/z 477.8); Fraction 2, ATK*EQLK ($[M+2H]^{2+}$ at m/z 512.9); Fraction 3, LK*CASLQK ([M +2H]²⁺ at m/z 577.8); Fraction 4, LAK*TYETTLEK ([M+3H]³⁺ at m/z 501.9, extracted ions: m/z 378.8, 660.6 and 720.4); Fraction 5, Y*TK ([M+2H]²⁺ at m/z 309.2); Fraction 7, AQYLQQC*PFEDHVK ($[M+3H]^{3+}$ at m/z 638.5); Fraction 8, Y*LYEIAR ($[M+2H]^{2+}$ at m/z 568.1); and Fraction 9, QNCELFEQLGEY*K ([M+2H]²⁺ at m/z 933.2).

Figure 6.

Product ion mass spectra of A) C*PF (C-[S=O]-PhIP) sulfinamide adduct ($[M+H]^+$ at m/z 604.2) recovered from HONH-PhIP-modified SA; B) $C*PF$ (C-[S=O]-[²H₅]-PhIP) sulfinamide adduct ($[M+H]^+$ at m/z 609.2) recovered from HONH- $[^2H_5]$ -PhIP modified SA; C) C*PF (C-[SO₂]-PhIP) sulfonamide adduct ([M+H]⁺ at m/z 620.2) recovered from HONH-PhIP-modified SA, and C*PF (C- $[SO_2]$ - $[^{2}H_5]$ -PhIP) sulfonamide adduct ($[M+H]$ ⁺ at m/z 625.2) recovered from HONH- $[{}^{2}H_{5}]$ -PhIP modified SA digested with Pronase E, leucine aminopeptidase, and prolidase.

Peng et al. Page 27

Figure 7.

(A) UPLC-ESI/MS analysis of reaction products formed by reaction of HONH-PhIP with acetic anhydride, following SPE. The peaks in the chromatogram are PhIP ($t_R = 10.0$), 5-HO-PhIP ($t_R = 10.2$), N-hydroxy-N-acetyl-PhIP ($t_R = 10.9$), and N-acetoxy-PhIP ($t_R = 13.7$). Product ion spectra were acquired on acetylated isomers of HONH-PhIP ([M+H]⁺ at m/z 283.1): (B) product ion spectrum of N -acetoxy-PhIP and (C) product ion spectrum N hydroxy-N-acetyl-PhIP.

Peng et al. Page 28

Figure 8.

Product ion mass spectra of A) 5-HO-PhIP ($[M+H]^+$ at m/z 241.2) recovered from Nacetoxy-PhIP-modified SA (upper panel) and 5-HO- $[^2H_5]$ -PhIP (([M+H]⁺ at m/z 246.2) recovered from N-acetoxy- $[^2H_5]$ -PhIP-modified SA (bottom panel) digested with Pronase E, leucine aminopeptidase, and prolidase. The ions observed at m/z 227.2 and m/z 226.2 in the product ion spectrum of 5-HO- $[^2H_5]$ -PhIP are attributed to losses of HDO and D₂O.

Peng et al. Page 29

Scheme 1. Proposed mechanisms of fragmentation of desamino-PhIP-Y

Scheme 2.

Proposed mechanisms of hydrolysis of Cys³⁴ PhIP-sulfenamide of SA to produce 5-HO-PhIP and PhIP.

\$watermark-text

\$watermark-text

Chem Res Toxicol. Author manuscript; available in PMC 2013 October 15.

b roduction at the MS³ scan stage (389.16 > 315.14 \rightarrow 209.09)

 3 scan stage (389.16 > 315.14 \rightarrow 209.09)

Product ion at the MS

Table 2

Assignment of NO₂-PhIP modified amino acid residues from SA that were identified by data-dependent MS/ MS analysis

Table 3

Assignment of HONH-PhIP modified amino acid residues from SA that were identified by data-dependent MS/MS analyses

