

# Structural basis for C-ribosylation in the alnumycin A biosynthetic pathway

Terhi Oja<sup>a</sup>, Laila Niiranen<sup>a</sup>, Tatyana Sandalova<sup>b</sup>, Karel D. Klika<sup>c</sup>, Jarmo Niemi<sup>a</sup>, Pekka Mäntsälä<sup>a</sup>, Gunter Schneider<sup>b,1</sup>, and Mikko Metsä-Ketelä<sup>a,b,1</sup>

Departments of <sup>a</sup>Biochemistry and Food Chemistry and <sup>c</sup>Chemistry, University of Turku, FIN-20014, Turku, Finland; and <sup>b</sup>Department of Medical Biochemistry and Biophysics, Karolinska Institutet, S-171 77 Stockholm, Sweden

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Alnumycin A is an exceptional aromatic polyketide that contains a carbohydrate-like 4'-hydroxy-5'-hydroxymethyl-2',7'-dioxane moiety attached to the aglycone via a carbon-carbon bond. Recently, we have identified the D-ribose-5-phosphate origin of the dioxane unit and demonstrated that AlnA and AlnB are responsible for the overall C-ribosylation reaction. Here, we provide direct evidence that AlnA is a natural C-glycosynthase, which catalyzes the attachment of D-ribose-5-phosphate to prealuminumycin by formation of the C<sub>8</sub>-C<sub>1'</sub> bond as demonstrated by the structure of the intermediate alnumycin P. This compound is subsequently dephosphorylated by AlnB, an enzyme of the haloacid dehalogenase superfamily. Structure determination of the native trimeric AlnA to 2.1-Å resolution revealed a highly globular fold encompassing an α/β/α sandwich. The crystal structure of the complex with D-ribose-5-phosphate indicated that the phosphosugar is bound in the open-chain configuration. Identification of residues E29, K86, and K159 near the C-1 carbonyl of the ligand led us to propose that the carbon-carbon bond formation proceeds through a Michael-type addition. Determination of the crystal structure of the monomeric AlnB in the open conformation to 1.25-Å resolution showed that the protein consists of core and cap domains. Modeling of alnumycin P inside the cap domain positioned the phosphate group next to a Mg<sup>2+</sup> ion present at the junction of the domains. Mutagenesis data were consistent with the canonical reaction mechanism for this enzyme family revealing the importance of residues D15 and D17 for catalysis. The characterization of the prealuminumycin C-ribosylation illustrates an alternative means for attachment of carbohydrates to natural products.

enzyme mechanism | natural product biosynthesis | protein structure

Many important natural products are decorated with carbohydrate moieties, and these include all of the major metabolite classes such as the polyketides, nonribosomal peptides, terpenoids, and flavonoids (1–3). The carbohydrate units are often essential for the biological activity and solubility of the compounds (1). For instance, many compounds in clinical use, such as the antibiotic erythromycin (4) and the anthracycline anticancer agent doxorubicin (5), owe their activities to deoxysugar units. These observations have generated considerable interest in the scientific community toward research on the biosynthesis, engineering, and attachment of deoxysugars onto natural products. Moreover, studies have revealed that the carbohydrate moieties are modified as 4-keto-6-deoxy-D-hexose nucleoside diphosphates before attachment by glycosyl transferases (GTs) (6, 7).

The 4'-hydroxy-5'-hydroxymethyl-2',7'-dioxane unit of the aromatic polyketide alnumycin A (1; Fig. 1), which is produced among others by *Streptomyces* sp. CM020, presents an unconventional solution for formation of carbohydrate-like structures in natural products. Intriguingly, the unit is connected via a C<sub>8</sub>-C<sub>1'</sub> bond to the prealuminumycin (2; Fig. 1) chromophore (8, 9). This mode of attachment is in contrast to the majority of GTs that most often attach carbohydrates through an O-glycosidic bond but that are then susceptible to hydrolysis in the acidic environment of the stomach or by the action of glycosidases in the small intestine (10). The more chemically stable C-glycosylated metabolites (11) are,

by comparison, rare and only found in selected natural products (12, 13), such as the polyketide antibiotic urdamycin (14).

Recent studies have established that the biosynthesis of 1 proceeds through C-ribosylation of 2 by the concerted action of two enzymes: AlnA and AlnB (15, 16). When the two enzymes were incubated with 2 and D-ribose-5-phosphate, alnumycin C (3; Fig. 1) could be isolated from the reaction mixture (16). The formation of the C<sub>8</sub>-C<sub>1'</sub> bond was annotated to AlnA, which is homologous to pseudouridine C-glycosidases such as YeiN (17) and TM1464 (17, 18) from *Escherichia coli* and *Thermotoga maritima*, respectively, whereas AlnB, which is a predicted member of the haloacid dehalogenase (HAD) superfamily (19, 20), was thought to be responsible for dephosphorylation of the attached D-ribose-5-phosphate unit.

In this paper, we established the functions of the two enzymes through enzymatic synthesis of a phosphorylated intermediate and stepwise characterization of the reactions catalyzed by AlnA and AlnB. Furthermore, to understand the molecular basis for the C-ribosylation reaction, we determined the structures of the two enzymes by protein crystallography. Finally, structure/function studies of the enzymes enabled us to propose a mechanistic model for the unusual attachment of carbohydrates to natural products.

## Results and Discussion

**Identification of the Functions of AlnA and AlnB.** When AlnA was incubated together with 2 (0.5 mM) and a 150-fold excess of D-ribose-5-phosphate (75 mM), a previously undetected compound was observed by HPLC (Fig. S1). The product was confirmed as a true intermediate of the pathway, because it was readily converted in an independent reaction further to 3 by AlnB (Fig. S1). Despite the poor chromatographic behavior and instability of the compound, multiple enzymatic reactions provided sufficient material for structure elucidation by high resolution electrospray ionization mass spectrometry (HR-ESI-MS) and both <sup>1</sup>H and <sup>31</sup>P NMR (SI Text). The data ascertained that the product was in fact an epimeric pair of two unique compounds differing in their stereochemistry at C-1', denoted as alnumycin P1 (4a, major isomer 84%; Fig. 1) and P2 (4b, 16%; Fig. 1), which contained D-ribose-5-phosphate attached to the expected C-8 position of 2. The experiments therefore confirmed unequivocally the functions of the two enzymes: AlnA is a natural C-glycosynthase (21) responsible for formation of the C<sub>8</sub>-C<sub>1'</sub> bond, whereas AlnB is a phosphatase in accordance to the function predicted by the

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The authors declare no conflict of interest.

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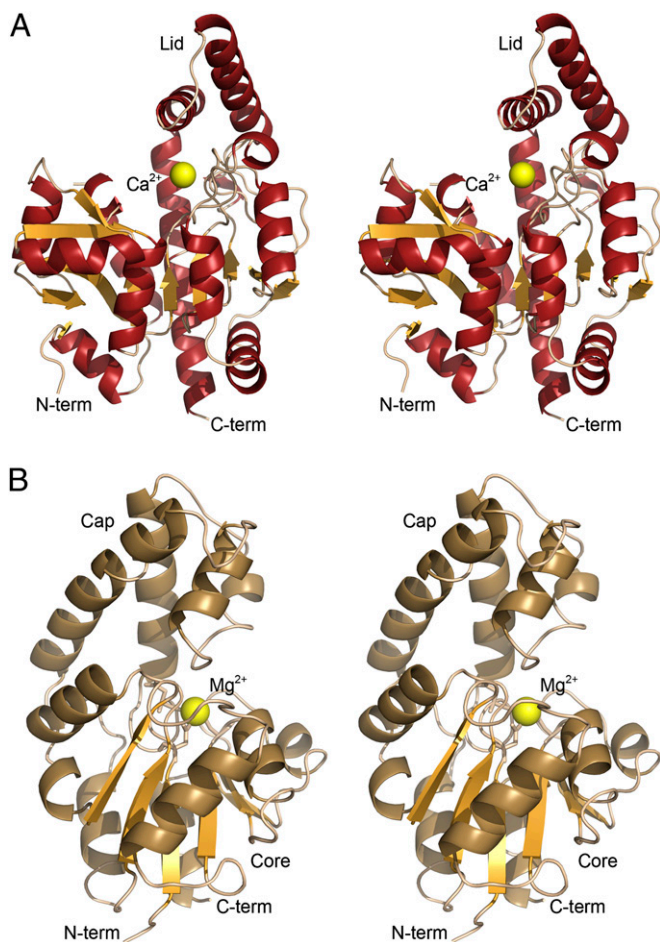
Data deposition: The atomic coordinates and structure factors have been deposited in the Protein Data Bank, [www.pdb.org](http://www.pdb.org) (PDB ID codes 4EX6–4EX9).

<sup>1</sup>To whom correspondence may be addressed. E-mail: Gunter.Schneider@ki.se or mikko.mk@gmail.com.

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**Fig. 2.** Schematic views of AlnA and AlnB. Stereoview images of the overall structures of (A) AlnA and (B) AlnB.

electron density observed for a metal ion, which is likely to be  $Mg^{2+}$  taken up from the crystallization media, at the junction of the cap and core domains was in line specifically with findings from related HAD phosphatases (23).

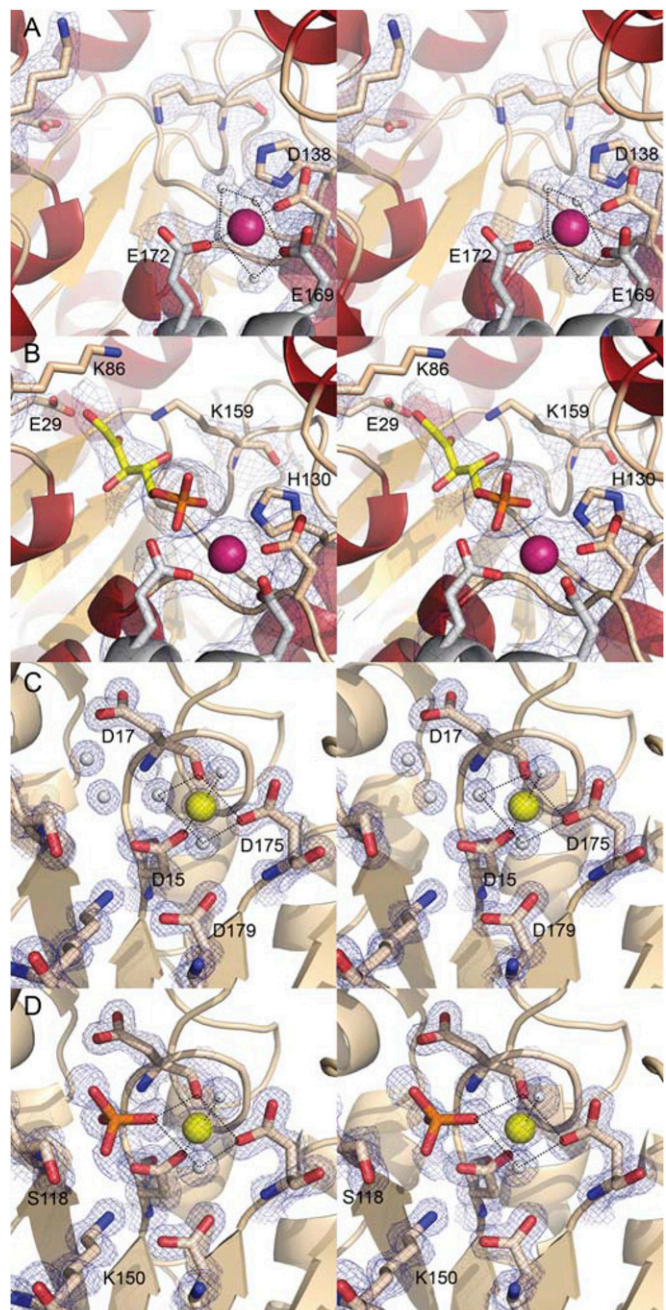
The canonical active site in HAD enzymes is formed between the core and cap domain interphase, where four loop regions from the core domain protrude providing a platform for binding of the  $Mg^{2+}$  ion and the phosphate group of the ligand (19). In AlnB, the octahedral coordination geometry to the metal cofactor is generated via three water molecules, the carboxylate groups of D15 and D175, and the carbonyl oxygen of D17 (Fig. 3C). The complex structure of AlnB with free phosphate at 1.5-Å resolution revealed the expulsion of three water molecules on ligand binding and hydrogen bonds of the ligand phosphate group to S118 and K150 in addition to coordination to the  $Mg^{2+}$  ion (Fig. 3D).

Although the core domain provides catalytically important residues, the cap domain is mainly responsible for substrate recognition and is more varied in HAD enzymes. In AlnB, the putative substrate binding pocket within the four-helix cap domain is surprisingly small, considering the structure of the substrate **4**. However, the cavity is lined mostly with hydrophobic residues such as I27, I47, L48, V51, and L55, which are suitable for binding of the polyaromatic substrate. The only nonhydrophobic exception that may be important for substrate recognition is K119, which protrudes inside the four-helix bundle from the core domain. The ligand could be computationally docked inside this four-helix bundle (Fig. 4B), which positioned the phosphate group of **4a** next

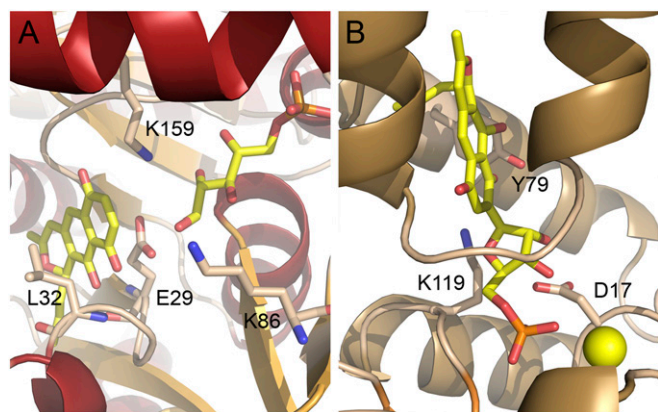
to the  $Mg^{2+}$  ion in the same position as in the ternary complex with free phosphate.

#### In Vitro Mutagenesis and Implications for the Catalytic Mechanisms.

AlnA catalyzes an unusual reaction: the attachment of D-ribose-5-phosphate to the polyketide aglycone via a  $C_8-C_{1'}$  bond. The protein bears no similarity in sequence or fold to glycosyl transferases, and moreover, it does not use nucleoside diphosphates



**Fig. 3.** Molecular architectures of the active sites of AlnA and AlnB. Stereoview images of (A) native AlnA at 2.1-Å resolution ( $\sigma = 1.2$ ), (B) AlnA in complex with D-ribose-5-phosphate (yellow) at 3.15-Å resolution ( $\sigma = 1.2$ ), (C) native AlnB in the open conformation at 1.25-Å resolution ( $\sigma = 2.0$ ), and (D) AlnB in complex with phosphate (orange) at 1.5-Å resolution ( $\sigma = 1.5$ ). The ligands were omitted from calculation of the electron density maps (2Fo-Fc) to reduce model bias. The metal ions  $Ca^{2+}$  (purple) and  $Mg^{2+}$  (yellow) and water molecules (white) are shown as spheres.



**Fig. 4.** Docking of (A) prealnumycin (**2**) into the active site of AlnA containing bound *D*-ribose-5-phosphate and (B) alnumycin P1 (**4a**) inside the four-helix bundle of AlnB.

for catalysis, which indicates that the reaction mechanism is likely to differ from C-glycosyl transferases (7). To date, no mechanistic studies have been conducted for this enzyme family, and therefore several residues in the active site were probed by mutagenesis (Table 1) to gain insight into the C-glycosynthase reaction.

Recent studies have shown that AlnA may use both *D*-ribose-5-phosphate and *D*-ribulose-5-phosphate as cosubstrates (16). We propose that the reaction catalyzed by AlnA might proceed through a Michael-type 1,4-addition and that the reactive phosphosugar species would be the linear ene-diol intermediate, which is deprotonated in the initial step at the C-2' hydroxyl group by E29 and stabilized by the nearby K159 (Fig. 5A). In effect, incubation of AlnA in the presence of *D*-ribose-5-phosphate alone led to conversion of the substrate to *D*-ribulose-5-phosphate (Fig. S6), which is consistent with our mechanistic model as such isomerization has been proposed to occur via an ene-diol species (24). The slow rate of the conversion suggests that the riboisomerase activity of AlnA is not a true catalytic activity, but rather a secondary moonlighting effect caused by the release of the ene-diol species in the absence of the cosubstrate **2**. The next step would be a nucleophilic attack of the enolate onto the aromatic carbon at C-8. The developing negative charge at the quinone oxygen in the transition state could be stabilized by K86. Residues E29, K86, and K159 appear to be critical for the reaction as demonstrated by the mutagenesis data (Table 1) and suggested by the conservation of equivalent residues in other members of this enzyme family (Fig. S7).

After formation of the C<sub>8</sub>–C<sub>1'</sub> bond, an intermediate suitable for formation of a cyclic hemiketal may be obtained via tautomerization. Subsequent dehydration of the hemiketal and restoration of the polyaromatic chromophore would then lead to the endproducts of the reaction (Fig. 5A). However, the discovery of the two C-1' epimeric products **4a** and **4b** of the AlnA reaction suggests that the final cyclization might also occur nonenzymatically. The narrow shape of the active site yields support for this, because it is unlikely that the attached *D*-ribose-5-phosphate could fold onto itself in the confined space. The benefit of this proposal is that it may also be used to explain the formation of the pathway shunt product alnumycin D (Fig. S8), which contains ribose attached in pyranose form. The considerable enzymatic activity of the H130A (coordination to *D*-ribose-5-phosphate) and D138A (coordination to the Ca<sup>2+</sup> ion) implies that the metal ion is mainly used for binding of the phosphosugar cosubstrate and not in the actual catalytic reaction. This observation is supported by the tolerance of the pseudouridine synthase YeiN toward various divalent metal ions (17).

In contrast to the complex reaction catalyzed by AlnA, the formation of **3** from **4** by AlnB is more straightforward to decipher. The loss of activity through mutagenesis of the conserved

aspartic acid residues D15 and D17 imply that the reaction involves canonical nucleophilic chemistry combined with acid/base catalysis (19, 25). In the initial step, the phosphoryl group is likely to be transferred from **4** to D15 of AlnB, producing a labile phospho-aspartyl-enzyme intermediate (Fig. 5B). The dual role of D17 would be to initially act as a general acid to facilitate the release of the product **3** by protonation. In the second step, the same aspartic acid is thought to act as a general base activating a nucleophilic water molecule, which is the ultimate acceptor of the phosphotransfer reaction (19). The decreased activities of K119A and K119R support our docking experiments and suggest that the four-helix bundle is used for substrate recognition. The moderate decrease in activity on the Y79A replacement, however, indicates that this residue is not critical in substrate binding. In general, the overall structure (Fig. 2A), the mutagenesis data (Table 1), and the conservation of characteristic sequence motifs (Fig. S9) provide compelling evidence that AlnB is likely to function in a manner similar to the classical phosphatases of the HAD enzyme family (19, 25).

### Concluding Remarks

The vast majority of natural product glycosylations performed by GTs proceed through an *O*-glycosidic linkage (1). In selected cases, the enzymes have been reported to be capable of forming a hydrolytically resistant C-glycosidic bond (14), but this has been rare by comparison. Here we clarified the structural basis for an alternative means for C-ribosylation found on the biosynthetic pathway of alnumycin A that is mechanistically distinct from GTs. We have shown that the two-step process involves a Michael-type 1,4-addition of *D*-ribose-5-phosphate to the polyketide aglycone catalyzed by AlnA, which is followed by dephosphorylation of the intermediate alnumycin P by AlnB. The study suggests further that the source of the stereochemical diversity of alnumycins may originate from nonenzymatic cyclization of the AlnA reaction product. The identification of the amino acid residues important for catalysis and substrate recognition for both enzymes paves the way for structure-based protein engineering attempts for expansion of the substrate specificity of the enzymes and the generation of novel C-ribosylated metabolites.

### Materials and Methods

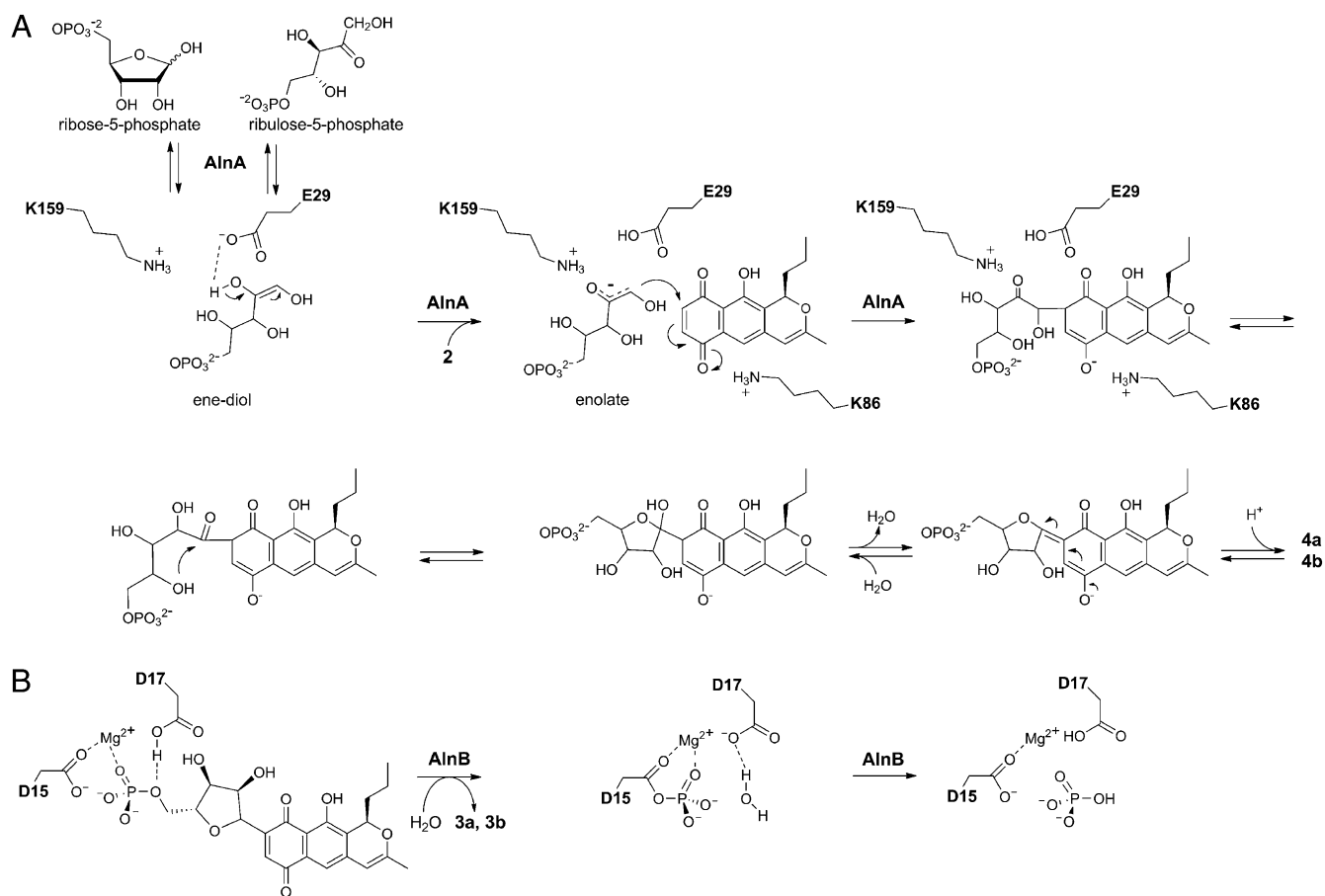
**Crystallography.** The crystallization of the proteins is described in *SI Text*. The X-ray data obtained at the European Synchrotron Radiation Facility

**Table 1.** Activity of AlnA and AlnB variants determined with the Malachite green assay

Variant	Released PO <sub>4</sub> (nmol)	Activity (%)
<b>AlnA</b>		
Native	9.2 ± 0.4	100
E29A	0.39 ± 0.51	4.2
E29Q	0.05 ± 0.49	0.5
K86A	0.48 ± 0.51	5.2
H130A	1.1 ± 0.5	12
D138A	1.1 ± 0.3	12
K159A	0.35 ± 0.52	3.8
K159R	0.39 ± 0.51	4.2
<b>AlnB</b>		
Native	1.2 ± 0.1	100
D15A	0.04 ± 0.02	3.3
D15N	0	0
D17A	0.02 ± 0.01	1.7
Y79A	0.66 ± 0.06	55
K119A	0.05 ± 0.03	4.2
K119R	0.02 ± 0.01	1.7

The coupled AlnA–AlnB reactions were performed in triplicate, with each individual sample assayed in duplicate, and the value for the blank reaction was subtracted. SD for each assay result is shown in *SI Text*.





**Fig. 5.** Schematic representations of the reactions catalyzed by (A) AlnA and (B) AlnB.

(Grenoble, France) was processed with the program MOSFLM (26), and the data were scaled using the program SCALA from the CCP4 package (27). Five percent of the reflections, with the same set selected for the native and complex structures, were excluded for calculation of  $R_{\text{free}}$ . Statistics for the data collection are given in Table S1. The structures of AlnA and AlnB were solved by molecular replacement using the program MOLREP (28) in the CCP4 package. For native AlnA, a homologous protein from *T. maritima* (PDB access code 1VKM) was used as a search model, whereas for native AlnB, the BALBES (29) server was used to construct several molecular replacement templates, from which the core domain of a phosphatase of unknown function from *Haemophilus somnus* (PDB ID code 2HSZ) yielded the correct solution. For the complex structures, the corresponding native structure was used as a search model. The native data for AlnA and AlnB were submitted for automated model building by ARP/wARP (30), and all structures were finalized by several rounds of manual building in the program COOT (31) and refinement using REFMAC (32). Structure validation was done with the MolProbity web server (33). The figures depicting protein structures were prepared using PyMol (The PyMOL Molecular Graphics System, Version 1.3; Schrödinger).

**Docking Calculations.** The PRODRG-server (34) was used to prepare models of the substrates. The ligands were docked in place using the program GOLD (35) integrated into the SOMA2 (36) modeling environment at the CSC-IT Center for Science Ltd (Espoo, Finland). Water molecules were removed from the active sites of the proteins, and several alternative side-chain conformations were manually tested to improve the computational calculations. For docking of **2** into the active site of AlnA in complex with *D*-ribose-5-phosphate, the conformation of the side-chain of L32 was altered to enable positioning of the propyl side chain of **2** into a hydrophobic pocket at the back of the active site. For docking of **4a** into the active site of AlnB, the side-chains of I27, V51, and L59 were altered to allow sufficient space for the ligand inside the cap domain.

**In Vitro Mutagenesis and Enzyme Activity Assays.** The AlnA and AlnB mutants were generated by PCR mutagenesis using the four-primer method as described in *SI Text*. The oligonucleotides used are listed in Table S2. The constructs were cloned into the modified pBADHisB (Invitrogen) vector (37) and verified by DNA sequencing. All variants were produced and purified to at least 95% purity as described previously for AlnB (16), except the D15A variant of AlnB, which required an additional gel filtration step in the storage buffer with a HiLoad 26/60 Superdex 200 column as part of an ÄKTA FPLC system (GE Healthcare), resulting in ~85% purity.

The activity measurements were conducted in coupled assays that were set up in 12.5 mM Hepes (pH 7.2), 37.5 mM NaCl, 1.25 mM  $\text{MgCl}_2$ , and 27% (vol/vol) glycerol as described above, with the exception that the substrate concentrations were 1 mM *D*-ribose-5-phosphate and 0.4 mM prealnumycin (**2**) in 4% (vol/vol) DMSO. The 60  $\mu\text{M}$  AlnA variants and 30  $\mu\text{M}$  AlnA were incubated with 3.75  $\mu\text{M}$  AlnB for 4 h, and the 300 nM AlnB and variants with 40  $\mu\text{M}$  AlnA for 90 min at 288 K, followed by extraction with one to two volumes of  $\text{CHCl}_3$ . The reactions were monitored as endpoint assays through formation of free phosphate in the aqueous phase as detected by use of the Malachite green dye and formation of alnumycin C (**3**) in the organic phase by HPLC. Because both methods yielded comparable results, final measurements were conducted in triplicate and were based on detection of inorganic phosphate (Table 1). Determination of the initial velocities of the reactions for kinetic characterization of AlnA was accomplished through time-course analysis of formation of inorganic phosphate using endpoint measurements. The kinetic measurements were conducted at 288 K in solutions composed of 50 mM Trizma-HCl (pH 7.2), 50 mM NaCl, 10% (vol/vol) glycerol, 4% (vol/vol) DMSO, 40  $\mu\text{M}$  AlnA, and 1  $\mu\text{M}$  AlnB. The titrations were conducted in 7.5–1,000  $\mu\text{M}$  *D*-ribose-5-phosphate in 100  $\mu\text{M}$  prealnumycin and 10–500  $\mu\text{M}$  prealnumycin in 1 mM *D*-ribose-5-phosphate. The oxygen concentration was reduced as described above.

**Analysis of Alnumycin P1 and P2.** The AlnA reactions were conducted in solutions composed of 10 mM Trizma-HCl (pH 6.8), 25 mM NaCl, 25 mM KCl,  $\geq 10\%$  (vol/vol) glycerol, 75 mM *D*-ribose-5-phosphate, 0.5 mM prealnumycin

(2) in 5% (vol/vol) DMSO, and 23  $\mu\text{M}$  AlnA and incubated for 3 h at 288 K. The oxygen concentration was reduced using the glucose oxidase (100 nm)-catalase (1.5  $\mu\text{M}$ ) system in 60 mM D-glucose. The products **4a** and **4b** were recovered using Sep-Pak PLUS C18 cartridges (Millipore, Waters) and purified by a SCL-10Avp HPLC system equipped with a SPD-M10Avp diode array detector (Shimadzu) and a Discovery HSC18 column (5  $\mu\text{m}$ , 5 cm  $\times$  4.6 mm; Supelco), eluted with 20 mM aqueous ammonium acetate (pH 3.6) and a 70–100% (vol/vol) methanol gradient. The same HPLC conditions were applied for acquiring high-resolution HPLC-ESI-MS data using a MicroTOF-Q mass spectrometer (Bruker) with 4-kV capillary voltage, 573 K dry heater temperature, and a nebulizer pressure of 1.6 bar coupled to a 1200 Series HPLC system equipped with a diode array detector (Agilent Technologies). NMR spectra were acquired using a Bruker Avance NMR spectrometer operating at 500 and 200 MHz for  $^1\text{H}$  and  $^{31}\text{P}$  nuclei, respectively. A full

description on the production, isolation, and structure determination is presented in *SI Text* and in *Table S3*.

**Protein Data Bank Accession Codes.** The crystallographic data have been deposited in the RCSB Protein Data Bank with the accession codes 4EX8, 4EX9, 4EX6, and 4EX7 for native AlnA, AlnA ligand complex, native AlnB, and AlnB in complex with phosphate, respectively.

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