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Biology and therapeutic potential of hydrogen sulfide and hydrogen sulfide-releasing chimeras

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Abstract

Hydrogen sulfide, H₂S, is a colorless gas with a strong odor that until recently was only considered to be a toxic environmental pollutant with little or no physiological significance. However, the past few years have demonstrated its role in many biological systems and it is becoming increasingly clear that H₂S is likely to join nitric oxide (NO) and carbon monoxide (CO) as a major player in mammalian biology. In this review, we have provided an overview of the chemistry and biology of H₂S and have summarized the chemistry and biological activity of some natural and synthetic H₂S-donating compounds. The naturally occurring compounds discussed include, garlic, sulforaphane, erucin, and iberin. The synthetic H₂S donors reviewed include, GYY4137; cysteine analogs; S-propyl cysteine, S-allyl cysteine, S-propargyl cysteine, and N-acetyl cysteine. Dithiolethione and its NSAID and other chimeras such as, L-DOPA, sildenafil, aspirin, diclofenac, naproxen, ibuprofen, indomethacin, and mesalamine have also been reviewed in detail. The newly reported NOSH-aspirin that releases both NO and H₂S has also been discussed.

1. Introduction

The initial observations by Kimura's group [1] suggesting hydrogen sulfide (H₂S) is a biologically relevant signaling molecule have been followed by a myriad of studies demonstrating some effect of this gas on virtually every organ system and tissue. It is only logical that attention would soon turn to the therapeutic potential of this signaling system and a number of H₂S "releasing" compounds have been developed and are already undergoing intensive investigation. In this review we provide an overview of the chemistry and biochemistry of H₂S and a summary of the H₂S "donating" compounds that have been identified in nature or chemically synthesized. Emphasis is placed on areas that have received the most focus, namely the cardiovascular and gastrointestinal systems, immunology and cancer biology. As would be expected in a rapidly developing and exciting field, retrospection can take a back seat to prospecting. Our intent is to point out the former in order to better guide the latter.

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2. H₂S chemistry and biology

2.1. H₂S chemistry

H₂S is a malodorous gas with the smell of rotten eggs. H₂S is soluble in water up to ~117 mmol/L and when dissolved acts as a weak acid with the equilibrium: $\text{H}_2\text{S} \leftrightarrow \text{HS}^- + \text{H}^+ \leftrightarrow \text{S}^{2-} + \text{H}^+$. Only the first reaction is physiology relevant as the pK_{a1} is around 6.9, which is approximately equivalent to intracellular pH in many tissues and it is not far from the pH of mammalian blood (7.4). Although there is some disagreement over the second pK_{a2}, all estimates place it over 11 [2] and this for practical purposes renders the divalent S²⁻ insignificant in biological experiments. Both pKs are temperature-sensitive and this can have a significant effect on the relative concentrations of H₂S and HS⁻ when comparing experiments performed at room temperature (20°C), where the pK_{a1} is 6.98, to those performed at body temperature (37°C), where the pK_{a1} is 6.77. Failure to correct for this discrepancy can account for as much as a 30% error in estimating the amount of dissolved H₂S under physiological conditions [3]. Solvation of the commonly used sulfide salts, NaHS and Na₂S will alkalinize unbuffered solutions. Two protons are consumed in the hydration of Na₂S and 1 mM Na₂S can increase the pH of mammalian buffers by one-half pH unit; 10 mM H₂S increases pH over 2 units [3].

H₂S is also relatively unstable in solution. H₂S is spontaneously oxidized in the presence of oxygen and metal catalysts such as ferric iron and the concentration of H₂S can be halved in as little as three hours [2, 4]. Addition of an iron chelator such as diethylenetriaminepentaacetic acid (DTPA) can greatly delay spontaneous oxidation, however, this can also remove ions that are vital in most biological processes. The obnoxious smell of H₂S that is all too familiar to anyone working with sulfide salts, hints at an even greater threat to proper experimental design and that is the volatility of H₂S. Biological experiments are the worst-case scenario as opportunity for H₂S volatilization is exacerbated by the need to more or less continuously supply tissues with oxygen and remove carbon dioxide. Even in the absence of cells, half of a dose of H₂S can be lost from open tissue culture wells in five minutes. This time is reduced in bubbled tissue baths to around 3 minutes and it is even less in the Langendorff heart apparatus that is used to study cardiac function and the ability of H₂S to protect the heart from reperfusion injury [5].

The ability to measure H₂S concentration in biological samples is key to understanding its function and this appears to be one of the weakest links in the field of H₂S biology. Two methods, the methylene blue (MB) spectrophotometric assay and S²⁻ ion selective electrodes (ISE) are the most commonly employed methods despite reports that both artificially inflate H₂S concentrations in tissues and blood as much as one thousand-fold when compared to more recent methods such as the monobromobimane (MBB) method [6, 7], gas chromatography of head space gas [8, 9] or polarographic electrodes [10]; reviewed in Olson, [3, 11]; [12]. Despite the growing evidence that H₂S in blood and tissues is well under 1 μM, the MB and ISE methods are currently employed, even in ongoing clinical trials, and blood values from 20–300 μM are still being reported as “physiological.” It is also well known that H₂S is rapidly oxidized by tissues under normoxic conditions (see below). This not only supports very low tissue H₂S concentrations but suggests that H₂S may be restricted to an autocrine or paracrine signal [13]. The “next generation” of analytical techniques must be able to measure intracellular H₂S concentration and distribution. Several techniques, such as analyte filled carbon nano tubes [14] or fluorescent probes [15] show some promise but need further refinement.

2.2. H₂S biosynthesis and metabolism

Most, if not all of the biosynthesis of H₂S has been attributed to three enzymes, cystathionine β-synthase (CBS, EC 4.2.1.22), cystathionine γ-lyase (CSE aka CGL, EC 4.4.1.1) and the tandem enzymes cysteine aminotransferase (CAT, EC 2.6.1.3) and 3-mercaptopyruvate sulfurtransferase (3-MST, EC 2.8.1.2). The generally accepted pathways for H₂S synthesis in vertebrate tissues are shown in Fig. 1A. Initially CBS and CSE catalyzed reactions were thought to be the primary pathways for H₂S production. CBS catalyzes the β-replacement reaction of homocysteine with serine to form cystathionine, thereby irreversibly committing this reaction into the transsulfuration pathway [16]. CSE then catalyzes the α,γ-elimination of cystathionine to form cysteine, α-ketobutyrate and NH₃. Both CBS and CSE can then generate H₂S from cysteine via β-elimination reactions. CAT transfers the amine group from cysteine to a keto acid, typically α-ketoglutarate, forming 3-mercaptopyruvate. Subsequent desulfuration of 3-mercaptopyruvate by 3-MST forms the persulfide, 3-MST-SSH. Either thioredoxin (Trx) or dihydrolipoic acid (DHLA), both abundant in cells, then serves as the reductant, liberating H₂S from 3-MST [17–19].

Other pathways for H₂S synthesis have also been identified. Banerjee's group [20–22] showed that H₂S can be produced by CBS-catalyzed β-replacement of two molecules of cysteine (also forming lanthionine) and β-replacement of one molecule of homocysteine and one molecule of cysteine (also forming cystathionine). CSE can catalyze β-replacement of two molecules of cysteine (also forming lanthionine), γ-elimination of homocysteine (also forming homoserine), γ-replacement of two molecules of homocysteine (also forming homolanthionine) and β- or γ-replacement of homocysteine and cysteine (also forming cystathionine). CSE, but not CBS, activity is substantially increased by elevated homocysteine which led Chiku et al. [20] to propose that condensation of two homocysteine molecules and/or the condensation of homocysteine and cysteine are important clearance pathways in hyperhomocysteinemia and that the excess H₂S produced by these reactions may contribute to the cardiovascular pathology associated with severe hyperhomocysteinemia. On the other end of the spectrum, there also appears to be a mechanism to prevent an excess in intracellular cysteine, which would otherwise potentially favor excess H₂S production [16, 23]. Cysteine dioxygenase (CDO; EC1.13.11.20), prominent in the liver and to a lesser extent in the kidney, lung and brain, oxidizes cysteine sulfur to cysteinesulfinate which is ultimately converted to either taurine or sulfate. CDO activity is closely coupled to dietary cysteine or methionine. Recent developments in thiol metabolomics [24] provide considerable potential for further elucidation of these pathways.

Thiosulfate (S₂O₃²⁻) is an intermediate in the mitochondrial oxidation of H₂S to sulfate (SO₄²⁻) and this is generally considered to be a degradative pathway for sulfide excretion (see below). However, it has recently been shown that both 3-MST and rhodanase can regenerate H₂S from thiosulfate in the presence of physiological levels of DHLA, but not in the presence of cysteine, glutathione, NADPH or NADH (Fig. 1C; [17]). Thiosulfate reductase (TR) catalyzes the reduction of thiosulfate in the presence of glutathione and is found in mammalian liver and kidney, and to a lesser extent in brain, heart, intestine and testis; approximately 85–90% of the TR is found in the mitochondria [25]. It is intriguing to consider the possibility that there are physiologically relevant regenerative pathways within cells that can be used for sulfide recycling and presumably H₂S signaling.

CBS and CSE are cytosolic enzymes. CBS was originally considered to be the predominant enzyme for H₂S production in the brain, whereas H₂S synthesis in the heart and vasculature was attributed to CSE (reviewed in Kimura, [26]). Recent studies with improved markers have provided a broader picture of enzyme distribution, e.g., CBS in vascular endothelium, CAT and 3-MST in vascular endothelium and brain and MST, but not CAT, in vascular smooth muscle [26, 27]. CAT and 3-MST have been found in both mitochondria and cytosol

[25], although 3-MST may be more localized to the mitochondrial matrix [17]; see also Whiteman et al., [28] for a general summary. Few studies, however, have examined H₂S production under the more stringent conditions of physiological levels of tissue enzymes and substrates. Kabil et al. [29] found that CSE predominates over CBS in both kidney and liver and at physiological levels of substrate and enzymes CSE may account for 97 % of H₂S production by transsulfuration in the liver, where as at saturating substrate concentrations CBS is the major source of H₂S in the kidney and brain. Recently, CBS and CSE were identified in human plasma where they were proposed to reduce circulating homocysteine and protect the endothelium from oxidative stress by generating H₂S [30].

CBS, CSE and CAT are pyridoxal 5' phosphate (PLP)-dependent, enzymes. *S*-adenosylmethionine (AdoMet), is an allosteric activator of CBS that regulates sulfur flow into the transsulfuration pathway when methionine levels are low [16]. However, the ability of CSE to generate H₂S from homocysteine, independent of CBS activity [20–22] suggests this is not the case and further studies are called for. CBS contains a heme group and is surprisingly inhibited by CO, but not by NO binding to the heme [31]. The effects of other potential inhibitors/activators of CBS, such as redox potential and Ca-calmodulin are equivocal. Recently, Mikami et al. [32] have shown that both cytosolic and mitochondrial CAT activity is greatest in calcium-free solutions and completely inhibited by 2.9 μM calcium. These results are strongly suggestive of an effective physiological mechanism for regulating H₂S production via the CAT/3-MST pathway.

Inhibitors of H₂S biosynthesis are relatively non-specific and may be poorly absorbed by tissues [28, 33]. Propargyl glycine (PPG) is an irreversible inhibitor, and β-cyanoalanine (BCA) a reversible inhibitor of CSE. Aminooxyacetate (AOA) is a irreversible inhibitor of CBS and hydroxylamine (HA) inhibits PLP-dependent enzymes including CBS, CSE and CAT (although a number of studies claim this is a specific inhibitor of CBS). Pyruvate acid has been used as a competitive inhibitor of 3-MST with limited success. Additional inhibitors, although rarely used in the context of H₂S physiology, can be found in Whiteman et al. [28].

2.2.1. Other potential biosynthetic pathways—Other H₂S-generating pathways have been described in invertebrates [34] and may in time be identified in mammals. Perhaps not surprisingly, some of the biological effects attributed to H₂S are mimicked by other sulfur donating molecules, perhaps because of common structural features, or the ability of cells to generate a common signaling molecule from them. Carbonyl sulfide (O=C=S; [3]), sulfur dioxide (SO₂; [35]), hydrosulfite (a.k.a. dithionite, S₂O₄²⁻; [36]) and sulfenic acids (R-S=O; [37]) are a few possibilities.

2.2.2. Metabolism (inactivation)—Much of the inactivation of H₂S occurs via mitochondrial oxidation [38]. Two membrane-bound sulfide:quinone oxidoreductases (SQR) oxidize sulfide to elemental sulfur while simultaneously reducing their cysteine disulfides. This produces a persulfide group at each of the SQR cysteines (SQR-SSH). Sulfur dioxygenase (SDO) then oxidizes one of the persulfides to sulfite (H₂SO₃). Sulfur from the second persulfide is transferred from the SQR to sulfite by sulfur transferase (ST) producing thiosulfate (H₂S₂O₃). Most thiosulfate is further metabolized to sulfate by thiosulfate reductase and sulfite oxidase. One electron from each of the two H₂S are fed into the respiratory chain via the quinone pool (Q), and finally transferred to oxygen at complex IV, thereby consuming molecular oxygen and water in the process. Oxygen consumption is obligatory during H₂S metabolism and one mole of oxygen is consumed for every mole of H₂S oxidized along the electron transport chain [38]. Oxidation of H₂S to thiosulfate requires additional oxygen at the level of SDO resulting in a net utilization of 1.5 moles of oxygen per mole of H₂S (or 0.75 moles of O₂ per mole H₂S [39]. Metabolism of H₂S

through SQR appears ubiquitous in tissues although the brain may be an exception [39]. It has been reported that sulfide oxidation in the mitochondria takes priority over oxidation of carbon-based substrates and that the capacity of cells to oxidize sulfide appears to be considerably greater than the estimated rate of sulfide production [8, 40]. This maintains very low intracellular H₂S concentrations.

The relationship between H₂S and O₂ consumption has been described as classical hormesis; at low concentrations H₂S stimulates oxygen consumption (and may even result in net ATP production), whereas it is inhibited by elevated H₂S [3]. It has been assumed that the turning point for this effect occurs around 50 μM H₂S. However, a recent study has shown that the mitochondrial metabolite, dehydroascorbic acid (DHA), can decrease H₂S toxicity [41]. Although the physiological concentrations of DHA are not known, DHA is recycled with ascorbic acid which is commonly found in low millimolar concentrations. It remains to be determined if this is a physiologically relevant detoxification mechanism and, perhaps more importantly, if DHA concentration can be regulated in response to a toxic load of H₂S.

3. H₂S “donating” compounds

With the myriad of purported biological actions of H₂S there is growing interest in more precise delivery of this volatile gas to target tissues in the form of H₂S “donating” compounds. H₂S donating compounds can be divided into three general classes; sulfide salts, naturally occurring compounds (mainly in foods) and synthetic moieties. These donating compounds are summarized in the following paragraphs and representative compounds are shown in Figs. 2 and 3. Additional drugs can be found in recent reviews [42–44]. The progress of these drugs in clinical trials can be accessed from the website “www.clinicaltrials.gov” and the reader is encouraged to evaluate these trials from the perspective provided in the present review. A somewhat dated list of patent applications on H₂S-releasing molecules and dosages can be found in Bannenberg and La Vieira [45]. It should also be noted that the sulfide moieties involved in H₂S biosynthesis described above, i.e., cysteine, homocysteine, etc, are also excellent H₂S donors, but are not considered here. The various aspects of H₂S effects on biological systems can be found in a number of recent reviews [46–52].

3.1. Sulfide salts

Sulfide salts, namely NaHS, Na₂S and CaS, form HS⁻ and H₂S immediately upon solvation in physiological buffers and are probably not truly H₂S “donating” compounds. These have the advantage of rapidly increasing H₂S concentration, but because H₂S is rapidly lost from solution by volatilization in laboratory conditions [5], or transferred across respiratory membranes [53, 54], their effective residence time in tissues is relatively short. This somewhat limits their therapeutic potential. A number of studies have been conducted with Na₂S in sterile solution (IK-1001) but these have been largely limited to experimental rather than clinical application. For example, Insko et al. [54] demonstrated in a rat model that detectable amounts of exhaled H₂S could be measured after intravenous administration of IK-1001. The same group has extended these observations into humans, showing that administration of IK-1001 increased blood sulfide and thiosulfate levels and that exhaled H₂S could be detected [55].

3.2. Naturally occurring compounds

3.2.1. Garlic

3.2.1.1. Chemistry: While garlic has long been felt beneficial as an antioxidant, recent evidence suggests that a number of beneficial effects of garlic are derived from H₂S production. Thus far the best characterized naturally occurring H₂S-donating compound

from garlic (*Allium sativum*) is allicin (diallyl thiosulfinate) which decomposes in water to a number of compounds [56]. Benavides et al. [57] measured H₂S production in real time with a polarographic sensor and observed that under anoxic conditions and in the presence of GSH, red blood cells rapidly (within minutes) produced H₂S from garlic extract as well as from two of the decomposition products of allicin, diallyl disulfide (DADS) and diallyl trisulfide (DATS; Fig. 2A), with approximately 3 times the yield from the latter. H₂S could also be generated in the presence of GSH, without red blood cells, and by reduced thiols on the red blood cell membrane. Infusion of diallyl disulfide 1.8 mg/kg/min in rats increased exhaled H₂S [54] providing additional support for H₂S production from garlic in vivo. Other molecules that can be extracted from garlic include a number of analogs of cysteine that are also readily synthesized, these are considered in the section on synthetic H₂S donors below. Ajoenes (Fig. 2B) are also potential H₂S donors, although H₂S production from these molecules has not yet been measured.

3.2.1.2. Biology: The beneficial effects of garlic in the prevention of renal, hepatic, cardiac, cerebral and pulmonary ischemia-reperfusion injury are well-known (reviewed in [58]), as are their potential as anti-cancer [59] and anti-diabetes [60] agents. Garlic extracts DATS and DADS vasodilate rat aortas via H₂S production [57].

3.2.2. Sulforaphane, erucin and iberin

3.2.2.1. Chemistry: Sulforaphane (Fig. 2C), the isothiocyanate compound from broccoli (*Brassica oleracea*), is rapidly absorbed by humans, reaching peak concentrations at 1 h and declining thereafter with a half-life of 1.8 hours [61]. A related isothiocyanate compound, erucin (Fig. 2D), is found in high levels in rocket salad species (*Eruca sativa*). Allyl isothiocyanate (Fig. 2E) is derived from wasabi, mustard and horseradish. It has not yet been determined if any of the isothiocyanates are metabolized to produce H₂S, although their purported health benefits are quite similar to those that are attributed to H₂S or other H₂S-donating drugs, suggesting a role for H₂S, although this needs to be evaluated further.

3.2.2.2. Biology: Sulforaphane, Erucin and a related isothiocyanate, iberin (Fig. 2F), upregulate thioredoxin reductase 1 (EC 1.8.1.9) in human breast cancer MCF-7 cells. Because thioredoxin reductase is the only enzyme that reduces thioredoxin and reduced thioredoxin is required to release H₂S from 3-MST (see above) it is quite possible that many of the health benefits of cruciferous vegetables involve H₂S. Sulforaphane protects vascular smooth muscle cells and endothelial cells from oxidative and inflammatory stress and suppresses angiogenesis [62–64]. Sulphoraphane has neuroprotective and anti-inflammatory actions mediated in part through activation of heme oxygenase-1 (HO-1) and it provides some protection against ischemia reperfusion injury, hemorrhage and serotonin-induced toxicity [65, 66].

3.3. Synthetic H₂S donors

3.3.1. GYY4137

3.3.1.1. Chemistry: GYY4137 (morpholin-4-ium 4 methoxyphenyl(morpholino) phosphinodithioate, Cayman Chemical; Fig. 2G) is a water-soluble molecule. Formation of H₂S from GYY 4137 was first reported by [67]. One mmol/l of GYY4137 releases ~40 μmoles of H₂S in the first 10 min and then releases approximately 40 μmoles of H₂S per hour for the ensuing 80 min when dissolved in acidic (pH 3.0) phosphate buffer; relatively little (<3 μmoles) H₂S appears to be released over this same period when dissolved in buffer at pH 7.4 or 8.5. Injection of 133 μmol/kg (approximately 665 μM if distributed throughout the extracellular compartment) either intraperitoneally (ip) or intravenously (iv) into rats, increases plasma H₂S from ~32 μmol/l to ~80 μmol/l in 30 min and this remains elevated

(50 $\mu\text{mol/l}$) for 3 hours. It should be noted that plasma H_2S was measured in these studies by DTMB (5,5'-dithiobis-(2-nitrobenzoic acid; Ellman's reagent) which reacts with thiol groups (R-SH) including glutathione and protein thiols, therefore, the effects of GYY4137 on plasma H_2S per se remain to be determined. Lee et al., [68] observed a near constant 15–20 $\mu\text{mol/l}$ H_2S for up to 7 days following addition of GYY4137 to culture media containing human breast adenocarcinoma MCF-7 cells (methylene blue method). The in vivo study of Li et al. [67] contrasts with that of Yu et al., [69] who reported that after i.p. injection of 200 $\mu\text{mol/l}$ the H_2S concentration in heart and liver increased, but only for 20 min.

3.3.1.2. Biology: In rats, GYY4137 relaxes aortas, mediated in part by K_{ATP} channels, vasodilates the perfused, pre-constricted kidney and exhibits antihypertensive activity but does not affect cardiac contractility or heart rate in vitro, furthermore, GYY4137 was not cytotoxic to cultured smooth muscle cells [67]. However, it has been reported that GYY4137 interacts with endogenous NO generated from l-arginine to stimulate heart contraction [70]. The effects of GYY4137 were slower in onset and longer lasting than those of NaHS. Paradoxically, the same dose of GYY4137 ip (133 $\mu\text{mol/kg}$) significantly increased blood pressure in rats rendered hypotensive by lipopolysaccharide (LPS)-induced endotoxic shock [71]. GYY4137 also decreased secretion of many of the markers of LPS-induced inflammation in cultured RAW 264.7 cells including nitrate/nitrite, PGE_2 , $\text{TNF-}\alpha$, $\text{IL-1}\beta$, IL-6 , $\text{NF-}\kappa\text{B}$, expression of inducible nitric oxide synthase (iNOS), and cyclooxygenase-2. One hour pre-treatment with GYY4137 prior to LPS did not provide a significant anti-inflammatory effect in rats, whereas GYY4137 administered 1 or 2 hours after LPS significantly reduced nitrite/nitrate, C-reactive protein, L-selectin and lung myeloperoxidase activity and decreased plasma creatinine/alanine amino transferase.

Anti-cancer effects of GYY4137 have been observed in vitro and in vivo [68]. Five-day treatment with GYY4137 (400 $\mu\text{mol/l}$), but not similar concentrations of the phosphate-substituted homolog, ZYJ1122 (morpholin-4-inum diphenylphosphinic acid), or NaHS significantly reduced proliferation of breast adenocarcinoma (MCF-7), acute promyelocytic leukemia (MV4-11) and myelomonocytic leukemia (HL-60) cells. At 800 $\mu\text{mol/l}$ GYY4137 killed 90–97% of these cells and human cervical carcinoma (HeLa), colorectal carcinoma (HCT-116), hepatocellular carcinoma (Hep-G2), osteosarcoma (U2OS), but did not affect survival of non-cancer human diploid lung fibroblasts (IMR90 and WI-38). The apparent IC_{50} for GYY4137 in MCF-7, HL60 and MV4-11 cells was around 350 $\mu\text{mol/l}$. GYY4137 promoted apoptosis as seen by presence of cleaved PARP and cleaved caspase 9, it also triggered cell-cycle arrest in the G_2/M phase in MCF-7 cancer cells without affecting non-cancerous IMR90 cells. Daily i.p. injection of GYY4137 also decreased the growth of HL-60 and MV4-11 tumors subcutaneously transplanted in immunodeficient mice. GYY4137 up to 2 mmol/l did not affect viability in non-tumorigenic HepG2 cells [69]. GYY4137 is a derivative of Lawesson's reagent (Fig. 2H), another H_2S donor which has been used with some success as an anti-inflammatory drug in the stomach [72].

3.2.2.3. Questions remaining: In the study by Li et al. [67] GYY4137 was reported to be more potent than NaHS in relaxing rat aortas (EC_{50} s 115.7 vs 274.1 $\mu\text{mol/l}$). However, when dissolved in buffer at pH 7.4, 1 mmol/l GYY4137 produced less than 3 $\mu\text{mol/l}$ H_2S . Therefore, 115 $\mu\text{mol/l}$ GYY4137 wouldn't be expected to produce more than 0.3 $\mu\text{mol/l}$ of H_2S . Because rat aortas did not respond to 100 $\mu\text{mol/l}$ H_2S (as NaHS), it is not clear how so little H_2S from GYY4137 could be vasoactive, unless the GYY4137 was internalized and the H_2S generated intracellularly. As described above, there is also some confusion surrounding the time-course of H_2S production from GYY4137. In addition, H_2S formation from GYY4137 in buffer was measured with an amperometric sensor, whereas H_2S formation in plasma was not determined amperometrically but was measured with the

methylene blue method. It is unclear why the latter was used as it is associated with considerable artifact [11].

3.3.2. Cysteine analogs

3.3.2.1. Chemistry: The well-known ability of cysteine to mimic H₂S effects, presumably by providing additional H₂S has led to the evaluation of a number of cysteine analogs as potential substrates for endogenous cysteine-metabolizing enzymes. Some of the more common molecules are: S-propyl cysteine (SPC), S-allyl cysteine (SAC, also a derivative of garlic), S-propargyl cysteine (SPRC; Fig. 2I) and N-acetyl cysteine (NAC; Fig. 2J). Wang et al. [73] measured H₂S production by homogenized rat ventricles with the direct methylene blue method and observed that SPC, SAC and SPRC all increased H₂S production by at least two-fold and the rate of H₂S production increased as the terminal carbon became progressively less saturated. This effect was mediated by CSE. SPRC administered *i.p.*, dose-dependently increases H₂S in LPS-treated rats, using direct methylene blue, control H₂S in hippocampus was 50 μM [74].

SAC has been shown not only to serve as a substrate for H₂S by CSE but also to up-regulate the enzyme in an acute myocardial infarction rat model. Conversely in sham operated rats, SAC did not significantly increase plasma H₂S [75]. SPRC and SAC have been conjugated with leonurine, an alkaloid in Chinese motherwort (Fig. 2N) [50, 76]. Leonurine-SPRC is somewhat more effective and it increased H₂S content in hypoxic neonatal rat ventricular myocytes. A derivative of diallyl disulfide ACS 81 (Fig. 2K), (3-(prop-2-en-1-yl)disulfanyl)propanoic acid) has been conjugated with L-DOPA and has essentially identical effects as the dithiolethiol-L-DOPA conjugates discussed below.

3.3.2.2. Biology

3.3.2.2.1. Cardiovascular: Daily *i.p.* injection of cysteine analogs, SPC, SAC and SPRC for seven days prior to and two days after coronary artery ligation of rat hearts significantly protected the hearts from ischemia reperfusion injury and was associated with preserved superoxide dismutase (SOD), glutathione peroxidase activity, and tissue GSH levels, while reducing lipid peroxidation products [73].

In an acute myocardial infarction rat model, pre-treatment with SAC (50 mg/kg/day for 7 days) significantly lowered mortality and reduced infarct size but did not affect blood pressure, SAC also increased left ventricular CSE; inhibition of CSE with PPG eliminated the beneficial effects of SAC, increased infarct size, and significantly elevated blood pressure. Although SAC is a substrate for H₂S production, it also augments H₂S by up-regulating CSE [75].

Leonurine-SPRC provides a cardioprotective effect in hypoxic neonatal rat ventricular myocytes by increasing cell viability, decreasing LDH leakage, decreasing MDA and ROS while increasing SOD, catalase activity, and have a general anti-apoptotic effect through an increase in bcl-2 and an decrease in bax and caspase-3 [50].

3.3.2.2.2. Neuromodulation: SPRC administered *i.p.* attenuated lipopolysaccharide (LPS)-induced deficits in spatial learning and memory in rats and restored the LPS deficit in hippocampal H₂S levels. SPRC also reduced the LPS-elevation of TNF-α, TNFR1, AβPP and Aβ_{1-40/42}, increased Iκβ-α degradation and NF-κβ p65 phosphorylation, and decreased TNF-α, TNFR1 and AβPP mRNA [74].

3.3.2.2.3. Cancer: SPRC (1 μM-30 mM) produced a dose dependent inhibition of growth of cultured SGC-7901 human gastric cancer cells. 1 μM SPRC produced 18% inhibition of

viability and suppressed colony forming and migration ability (~18% inhibition with 1 μM) significantly increased CSE protein expression and H_2S concentration in culture media; PPG significantly decreased SPRC efficacy. SPRC effects were more pronounced than SAC. SPRC stimulated apoptosis, and induced cell cycle arrest at the G_1/S phase [77]. SPRC increased Bax and p53 but not Bcl-2 protein and mRNA expression. SAC increased p53 protein expression and Bax mRNA but did not affect other parameters. In male nude mice with SGC-7901 tumors implanted in the flank, SPRC and SAC (50 and 100 mg/kg) increased plasma H_2S concentration, reduced tumor volume and increased apoptotic tumor cell number and increased protein expression of Bax and p53 and decreased Bcl-2. Both SPRC and SAC increased Bax and p35 mRNA expression but did not affect Bcl-2 [77].

3.3.2.2.4. Inflammation: In the lipopolysaccharide-induced inflammatory response of rat embryonic ventricular myocardial cells (H9c2 cells) Pan et al. [78] observed that SPRC prevented NF- κB activation, suppressed LPS- induced extracellular signal-regulated kinase 1/2 (ERK1/2) phosphorylation, and intracellular ROS production. SPRC induced phosphorylation of Akt and reduced mRNA and protein expression of tumor necrosis factor- α (TNF- α) and these effects were abolished by PPG. SPRC (1 μM) inhibited iNOS and ICAM mRNA and increased CSE and supernatant H_2S concentration [77].

Other: Cystamine (Fig. 2L) dose-dependently increased cell proliferation of HIV-infected CEM-6 human lymphocytes [79]. Methyl disulfides (R-S-S- CH_3) increase growth of murine lymphoma L1210 cells [80].

3.3.2.2.5. Questions remaining: In the study by Ma et al. [77], H_2S was measured in the supernatant above cells 24 hours after addition of SPRC or SAC and in the study by Pan et al. [78], H_2S was measured in the supernatant above cells 6.5 hours after addition of SPRC. Given the volatility of H_2S , it is unclear how much H_2S was actually present at these late dates.

3.3.3. Cysteine-activated H_2S donors

3.3.3.1. Chemistry: Zhao et al. [81] developed a novel class of H_2S donors based on the lability of nitrosothiol (S-N) bonds. In their scheme a *N*-(benzoylthio)benzamide moiety (Fig. 2M) slowly releases H_2S in the presence of excess cysteine. A peak of 25.4 μM H_2S was observed 18 minutes after 40 μM of *N*-(Benzoylthio)benzamide was incubated with 4 mM cysteine. The rate and quantity of H_2S released could be altered by adding side groups on the benzoylthio ring and 11 such substitutions were examined in. Adding - CF_3 to the fourth carbon decreased the peak time to 13 minutes and increased the concentration at the peak to 35.6 μM , indicating nearly complete metabolism of the compound to H_2S . An - OCH_3 group in both the ortho and para-positions increased the peak time to 50 minutes and decreased the concentration at the peak to 23.0 μM . These reactions were most efficient when the cysteine:*N*-(Benzoylthio)benzamide ratio was at least 10, further increasing this ratio did not appreciably increase yield. All of these compounds were soluble in phosphate buffer and in the absence of cysteine did not generate H_2S . The authors also showed that H_2S could be generated from bovine calf plasma that contained 500 μM cysteine and that adding *N*-methylmaleimide to the plasma to block free cysteine inhibited H_2S formation. This provided additional evidence that cysteine is the regulator of this H_2S -generating reaction and can be used in biological systems.

3.3.3.2. Biology: To date no biological studies have been reported.

3.3.3.2.1. Questions remaining: As with the study of H_2S formation from GYY4137 (above), the formation of H_2S in buffer was measured amperometrically, whereas H_2S

generation in plasma was measured using the methylene blue method. The actual rate of H₂S formation from cysteine and *N*-(Benzoylthio)benzamide in plasma needs to be verified. Furthermore, as cysteine alone can contribute to H₂S generation, additional, cysteine-only, control studies are necessary.

3.3.4. Dithiolethione and its NSAID chimeras

3.3.4.1. Chemistry: Anethole dithiolethione (ADT; Fig 3A) and its main metabolite (ADT-OH; 5-(4-hydroxyphenyl)-3*H*-1,2-dithiole-3-thione; Fig. 3B) have been used extensively as a donor of H₂S and its ease of esterification with other therapeutics has led to a considerable variety of H₂S “donating” drugs. ADT-OH can also be modified prior to esterification, i.e., ACS 5, ACS 48, ACS 50 (Figs. 3C-E). A number of the ADT-OH-H₂S molecules are shown in Fig. 3F.

Lee et al. [82] conjugated L-DOPA with ACS 5, ACS 48 and ACS 50 to produce ACS 85, ACS 83 and ACS 84, respectively. When ACS 84 was injected i.v. into rats essentially all of the dithiolethione (ACS 50) was cleaved from the L-DOPA within one hour and dopamine, L-DOPA and ACS 50 could be found in the brain. Intracellular concentration of H₂S in human monocyte (THP-1), astrocytoma (U373) and neuroblastoma (SH-SY5Y) cells increased 12 hours after treatment with 10 μM ACS 50 to 6.5, 0.8 and 0.8 μmol/l, respectively and by 48 hours post-treatment these were 6, 1.8 and 1.7 μmol/l, respectively. At 48 hours post-treatment, H₂S concentration in the extracellular fluid was between 2 and 3 μmol/l. Comparatively little (10%) H₂S was released from either PBS buffer or culture media at 48 h (0.2 μmol/l). H₂S production from ADT-OH, ACS 5, ACS 48, ACS 50, ACS 81 by isolated mitochondria was rapid, over 80% within 1 h [82].

ADT-OH conjugated with aspirin (S-ASP, ACS 14) injected i.v. into rats was rapidly deacetylated then hydrolyzed at a somewhat slower rate freeing ADT-OH; plasma H₂S peaked within five minutes of injection and fell to less than 20% of the peak value by 40 minutes [83]. I.p. injection of ACS 14 still resulted in rapid disappearance of the parent compound and formation of ADT-OH, however, the latter remained significantly elevated two hours after injection, plasma H₂S peaked in about one hour after either ACS 14 or ADT-OH injection then dipped to half the peak value in about four hours but then slowly increased over the remaining 24 hour experimental period [84]. When THP-1 or U118 cells were incubated with NaHS, ADT-OH, S-ASP, or ADT-OH conjugated with diclofenac (S-DI) they steadily accumulated H₂S for 12 h [85].

ADT-OH conjugated with sildenafil (ACS 6) released H₂S slowly with peak concentrations at 120 minutes, approximately 4 times as much H₂S was released in the presence of pulmonary artery endothelial cells compared to buffer [86].

When either ADT-OH or ADT-OH conjugated with diclofenac (S-diclofenac, S-diclo, ACS 15, ATB-337) were added to rat liver homogenate there was a slow release of H₂S (2.6% of the S-diclo) that persisted for at least 75 minutes; only one third of this amount of H₂S was released by these compounds in buffer. Inhibition of carboxylesterase with sodium fluoride reduced H₂S production from S-diclo in homogenized liver by nearly two-thirds. When incubated in plasma, release of H₂S from S-diclo slowly increased reaching a maximum in 90 minutes, declining thereafter. IP injection of 47.2 μmol/kg of S-diclo increased plasma H₂S from 30 to 37 μM at 45 minutes and plasma H₂S remained elevated (33 μM) after 6 hours; injection of equal molar amount of NaHS increased plasma H₂S from 25 to 34 μM at 45 minutes but plasma H₂S returned to control values by 6 hours. Plasma H₂S was unaffected by IP injection of diclofenac, [87].

Recently, Lazzarato et al. [88] synthesized a new class of ADT-OH esters of aspirin, ((4-(3-Thioxo-3*H*-1,2-dithiol-5-yl)phenoxy)carbonyloxy)-methyl 2- (Acetyloxy)benzoate and ((2-(4-(3-Thioxo-3*H*-1,2-dithiol-5-yl)phenoxy)ethoxy)carbonyloxy)-methyl 2- (Acetyloxy)benzoate (Fig. 3G). These compounds are reported to have the advantage of releasing aspirin when incubated in human serum. The compounds are relatively stable in buffered solutions (20% or less decomposition at three hours at pH 7.4) but relatively quickly are metabolized in human serum (halftime of aspirin release between two and three minutes) and aspirin accounts for approximately 1/3 of the yield.

3.3.4.2. Biology: Combining ADT-OH with a variety of compounds has frequently shown to enhance their biological activity or, especially in the case of the nonsteroidal anti-inflammatory drugs (NSAIDs), to counter their adverse effects.

3.3.4.2.1. Cardiovascular: ACS 14 given orally (50 mg/kg) for seven days did not affect body weight, systolic blood pressure, or heart rate in rats; ACS 14 and aspirin lowered thromboxane B₂ and 6-keto-PGF_{1 α} , whereas these variables were unaffected by ADT-OH or salicylic acid. ACS 14 but not ADT-OH lowered plasma oxidative stress marker, 8-isoprostane, and both compounds increased plasma, cardiac and aortic tissue glutathione. ACS 14 ameliorated hemorrhagic lesions in the stomach due to oral aspirin and increased heme oxygenase-1 (HO-1) promoter activity in NIH3T3-HO-1-*luc* cells [83]. I.p. injection of ACS 14 initially lowered, but then increased, plasma levels of reduced glutathione, cysteine and the glutathione metabolite cysteinylglycine but produced a steady fall in plasma homocysteine, the initial fall in reduced thiols was not due to a higher pro-oxidant burden as oxidative stress in blood continued to fall throughout the experimental period [84]. ACS 14, salicylic acid conjugated with ADT-OH (ACS 21) and ADT-OH reduced the hypertension brought about by glutathione depletion with buthionine sulfoximine (BSO) and lowered plasma thromboxane B₂, 8-isoprostane and insulin and restored plasma glutathione without creating gastric lesions. In addition, they improved endothelial function in isolated aortas and improved outcome in cardiac ischemia reperfusion experiments, similar benefits were not experienced with either aspirin or salicylic acid strongly suggesting the benefits of the H₂S-releasing moiety [89].

The “pure” aspirin and H₂S releasing drugs synthesized by [88] are reported to have good antiaggregatory effects on platelets (IC₅₀ of 150 and 16 μ M).

The phosphodiesterase inhibitor, sildenafil conjugated with ADT-OH (ACS 6) relaxes the rabbit corpus cavernosum with greater sensitivity and efficacy than NaHS, and the relaxation is due solely to the sildenafil moiety downregulating phosphodiesterase 5. However, while sildenafil acts through a PKG-dependent mechanism, the H₂S acts through a PKA-dependent mechanism and together they have a greater inhibitory effect on p47^{phox} (and therefore NADPH oxidase) than either sildenafil or H₂S alone. ACS 6 has the added benefit of antioxidant activity both in vitro and in rats in vivo [90]. Similar observations in porcine pulmonary endothelial cells suggests that ACS 6 may also be beneficial in treating adult respiratory distress syndrome [86].

IP injection of 47.2 μ mol/kg S-diclofenac, sufficient to raise rat plasma H₂S from 30 to 37 μ M did not affect arterial blood pressure [87]. In the isolated rabbit heart subjected to low-flow ischemia-reperfusion damage, S-diclofenac and NaHS produced a dose dependent normalization of coronary perfusion pressure, reduced left ventricular contracture during ischemia, and at reperfusion they improved left ventricular pressure and dp/dt while reducing creatinine kinase and LDH and increasing GSH. These effects appear to be mediated in part by K_{ATP} channels. Inhibition of NO biosynthesis exacerbated ischemia-

reperfusion damage and this was offset by both S-diclofenac and NaHS. Cardiac parameters in control hearts were unaffected by S-diclofenac or NaHS [91].

In normal rat aortic (A-10) and immortalized (CRL-2018) smooth muscle cells S-diclofenac exhibited a concentration-dependent reduction in cell survival (~5% survival at 100 μ M). Both NaHS and S-diclofenac decrease the percentage of A-10 cells in the G₁ phase and increased apoptosis. S-diclofenac increased pro-apoptotic proteins p53, AIP1 and Bax but did not increase the anti-apoptotic protein, Bcl-2. However, S-diclofenac did not affect cells synchronized by starvation where G₁ cells accumulate. These results suggest that S-diclofenac may be useful for prevention of smooth muscle proliferation [92].

3.3.4.2.1.1. Questions remaining: There are a number of concerns regarding the amount, duration and stability of H₂S released by these compounds and these are most likely due to the use of methylene blue in the H₂S assays. Lee et al. [82] reported that intracellular H₂S remained constant at 6 μ M (60 % of total ACS 50 dose) from 12 to 48 h in cultured cells and continued to increase in media for 48 h; even H₂S from NaHS only declined by 10% over 48 hours. In the study by Muzaffar et al. [86] ADT-OH conjugated with sildenafil (ACS 6) released more H₂S from ACS 6 in buffer than from equal molar (10 μ M) concentrations of NaHS; peak H₂S concentration from NaHS (<6 μ M) did not occur until 30 minutes, whereas with ACS 6 H₂S peaked at 120 minutes (8 μ M) and remained elevated (6 μ M) for 840 minutes thereafter. Giustarini et al. [84] showed that ACS 14 or ADT-OH increased plasma H₂S from ~0.35 μ M to 6 μ M within one hour after i.p. injection, plasma levels then fell to half this at four hours but then continued to rise for the remaining 20 hour experimental period. H₂S production by homogenized rat liver at 30 min in the presence of 10 mM cysteine was 29 nmol/mg protein, equal to 4.1 mM/l cytosol, LPS increased this to 42 nmol/mg protein equal to 6 mmol/l cytosol. These experiments will need to be re-visited using improved methodologies to accurately evaluate H₂S release.

3.3.4.2.2. Neuromodulation: Both the H₂S-donating moiety, ACS 5, ACS 48 and ACS 50 and their L-DOPA-conjugated H₂S donors, ACS 85, ACS 83 and ACS 84 (S-DOPAs) increase intracellular glutathione in SH-SY5Y cells and reduce the activity of monoamine oxidase-B (MOA-B) but not MOA-A (this effect could not be attributed to the L-DOPA moiety). They also increase viability in models of LPS neurotoxicity and by lowering release rates of TNF- α , IL-6 and nitrite [85]. β -Amyloid (A β) induced toxicity and decreased cell viability in microglia is reduced by ACS 84. ACS 84 reduces release of nitrite (an index of NO release) and TNF- α , preserves mitochondrial membrane potential, inhibits activation of JNK and p38 MAPK but does not affect COX-2 expression or ERK [93].

ACS 14 specifically increased cell viability in BV-2 microglial cells, decreased LDH release in A β treated BV-2 cells and suppressed COX-2 protein expression and growth arrest DNA damage. It also protected mitochondria from A β -induced loss of membrane potential and reduced A β -induced activation of p38-mitogen activated protein kinase (MAPK), and release of NO and TNF- α [94].

S-ASP, S-DI and ADT-OH protect human neuroblastoma SH-SY5Y from toxic substances released from LPS and γ -interferon activated human microglia and THP-1 cells but do not affect untreated SH-SY5Y cells. S-ASP and S-DI are particularly efficacious in suppressing release of TNF- α , IL-6 and nitrite from LPS and interferon- γ stimulated human microglia and THP-1 cells [85].

3.3.4.2.3. Anti-inflammatory: The role of H₂S in inflammation and immunity is controversial, with some studies suggesting an anti-inflammatory effect, whereas others suggest a contribution to immune-mediated tissue injury. In the carrageenan-induced rat

hindpaw model of inflammation, CSE and myeloperoxidase (MPO) activity were increased. Pretreatment with DL-propargylglycine, an CSE inhibitor, reduced carrageenan-induced hindpaw edema in a dose-dependent manner [95].

Mice given endotoxin had increased plasma, liver, and kidney H₂S levels, this was accompanied by increases in CSE gene expression in both liver and kidney. There was also histological evidence of lung, liver, and kidney tissue inflammatory damage. Administration of NaHS resulted in histological signs of lung inflammation, increased lung and liver MPO activity, and increased plasma TNF- α levels [96]. Elevated H₂S levels have also been observed in the plasma of septic shock patients [96]. Surprisingly, administration of NO-releasing flurbiprofen to LPS treated rats resulted in a dose-dependent inhibition of liver H₂S synthesis and CSE mRNA levels [97, 98]. Together, these observations suggest a proinflammatory role for H₂S. A proinflammatory role for H₂S has also been suggested in an animal model of pancreatitis [99, 100].

On the other hand, potent anti-inflammatory effects of H₂S have also been reported. In an air pouch model of inflammation, carrageenan caused infiltration of large amounts of leukocytes and neutrophils. Pretreatment with a number of different H₂S donors, (NaHS, N-acetylcysteine, Lawesson's reagent) reduced the number of leukocytes in a dose-dependent manner [101]. This reduction was comparable to that seen by using diclofenac, L-NAME (NOS inhibitor); or dexamethasone. Pretreatment with the CSE inhibitor, β -cyanoalanine, reversed the inhibitory effects of N-acetylcysteine. Pretreatment with diclofenac, NaHS, or Na₂S similarly reduced carrageenan-induced rat paw edema while β -cyanoalanine showed a significant increase in paw swelling in response to carrageenan [101]. H₂S has been shown to induce apoptosis in neutrophils [102], which could contribute to resolution of inflammation [103].

It therefore appears that like NO, physiological concentrations of H₂S produce anti-inflammatory effects; where as higher concentrations are pro-inflammatory. The mechanisms and pathways affected by S-NSAIDs in reducing inflammation are depicted in Fig 4.

3.3.4.2.4. Gastrointestinal sparing effects: NSAIDs can have significant toxicity, which includes gastrointestinal side effects, ranging from dyspepsia to gastrointestinal bleeding, obstruction, and perforation; renal side effects; and other side effects, including hypersensitivity reactions. Among patients using NSAIDs, up to 4% per year suffer serious gastrointestinal complications resulting in about 8000 deaths [104]. The gastric damage is caused through two distinct mechanisms: (1) direct epithelial damage as a result of their acidic properties; (2) breakdown of mucosal defense mechanisms (leukocyte adherence, decreases in blood flow, bicarbonate and mucus secretions) due to a reduction of mucosal prostaglandin (PG) synthesis [105].

In an animal model of chronic ulcer, which highly resemble human ulcers in terms of pathological features and healing mechanisms [106], induction of gastric ulceration was associated with a increase in expression of CSE and CBS enzymes as well as H₂S levels, suggesting a defensive response [72]. Ulcer healing was observed following administration of three chemically distinct (Lawesson's reagent, 4-hydroxythiobenzamide, H₂S-5-ASA) H₂S donors, and administration L-cysteine, a precursor of endogenous H₂S synthesis, also enhanced ulcer healing [72]. Since these H₂S donors did not affect gastric pH or inhibit acid secretion; and K_{ATP} channel agonists or antagonists had no effect on ulcer healing; and since H₂S is a vasodilator, it is possible that this may have contributed to healing process observed in this study [72].

A number of H₂S-releasing NSAIDs have been shown to exhibit a significant protective effect in the stomach and intestine compared to their traditional counterparts. For example, S-diclofenac had significant GI sparing effects compared to diclofenac [87, 107] and was equieffective in inhibiting both Cyclooxygenase (COX)-1 and COX-2 enzymatic activity [107]. It also reduced plasma IL-1 β and TNF- α levels, while it elevated the anti-inflammatory, IL-10 [87]. Similar GI sparing effects have been observed with S-indomethacin [108] and S-mesalamine [109]. S-naproxen has also been shown to be more effective than naproxen in reducing inflammation, while producing significantly less damage to the stomach [110]. Inflammatory bowel disease (IBD; Crohn's disease and ulcerative colitis) is a chronic disorder characterized by extensive ulceration and inflammation. It is generally regarded as being driven by Th1 cytokines, including IL-1, IL-2 TNF α , IFN γ and IL-12 [111, 112]. The chemokine, RANTES has also been implicated in the pathogenesis of colitis in this model [113]. The animal model used for studying IBD uses trinitrobenzene sulfonic acid (TNBS). This model is well characterized, and exhibits responsiveness to various therapies similar to those used for human IBD and shares many features with IBD in humans, particularly Crohn's colitis [114, 115]. Using this model in mice, S-mesalamine was more effective than mesalamine in reducing mucosal injury and disease activity (body weight loss, fecal blood, diarrhea) and colonic granulocyte infiltration [109]. S-mesalamine also reduced the expression of mRNA for TNF α , IFN γ , IL-1, IL-2, IL-12 and RANTES [109]. Using the same model in rats, S-mesalamine modulated expression of colonic pro-inflammatory mediators, COX-2 and IL-1 β [116].

3.3.4.2.5. Chemopreventive properties: NSAIDs in general and aspirin in particular represent the prototypical chemopreventive agents against cancer. Epidemiological studies describing results on >1 million subjects have pointed out the protective effect of NSAIDs against colon, esophageal, gastric, breast, pancreatic and ovarian cancers [117]. The epidemiological findings are in agreement with a large body of in vitro and animal data [118]. However, regular NSAID use may lead to serious side effects as described above. Hydrogen sulfide-releasing NSAID (HS-NSAIDs or S-NSAIDs) are a novel group of compounds that appear to overcome the shortcomings of regular NSAIDs although human safety data on this class of compounds are not yet available. Recently, the effects of four H₂S-releasing NSAIDs, (HS-aspirin, HS-sulindac, HS-ibuprofen, HS-naproxen, these are the same as S-aspirin, S-sulindac, S-ibuprofen, S-naproxen) have been described on the growth properties of eleven different human cancer cell lines of six different tissue origins [119]. These cell lines were of adenomatous (colon, pancreatic, lung, prostate), epithelial (breast), and lymphocytic (leukemia) origin. In all of these HS-NSAIDs, the H₂S-releasing moiety was ADT-OH, which was conjugated to the parent NSAID through an ester bond [119]. These HS-NSAIDs inhibited the growth of these cancer cell lines with potencies of 28- to >3000-fold greater than that of their traditional counterparts. HS-NSAIDs inhibited cell proliferation, induced apoptosis, and caused G₀/G₁ cell cycle block [119]. HS-aspirin preferentially inhibited the growth of estrogen receptor (ER)-negative breast cancer cells compared with a normal epithelial mammary cell line [120]. This effect was H₂S-dependent as pretreatment with NaF, a carboxylesterase enzyme inhibitor, reduced the potency of HS-aspirin for cell growth inhibition by about 60-fold [120].

The transcription factor NF- κ B is constitutively activated in ER(-) breast cancer cell lines and primary tumors and therefore, constitutes a valid therapeutic target [121]. HS-ASA significantly inhibited DNA binding activity of NF- κ B (p65) which was associated with a decrease in NF- κ B protein p65 levels in the nucleus. This was mediated by blockade of I κ B α phosphorylation through inhibition of IKK activity [120]. Redox balance of the cells may influence the function of NF- κ B by regulating its ability to bind to DNA, HS-ASA strongly inhibited thioredoxin reductase (TrxR) activity in ER(-) breast cancer cells and also increased intracellular reactive oxygen species (ROS) [120]. In a xenograft model of ER(-)

breast cancer, HS-ASA significantly reduced tumor volume and tumor mass, through induction of apoptosis and inhibition of both proliferation and NF- κ B [120]. Recently, using ER(-) breast cancer cells, it has been reported that dithiolethiones inhibit NF- κ B transcriptional activity not through H₂S release but rather via a covalent reaction with the NF- κ B p50 and p65 subunits [122]. The same group has also reported that ACS-2, (4-(3-thioxo-3*H*-1,2-dithiol-5-yl)phenyl 2-propylpentanoate), an analog of ADT-OH which is more stable *in vivo*, had no effect on tumor volume in a murine model of ER(-) breast cancer, however, it did significantly reduce tumor burden in these mice [122].

The balance between phase-I carcinogen-activating and phase-II detoxifying xenobiotic metabolizing enzymes is critical to determining an individual's risk for cancer. HS-ASA induced the activities of glutathione S-transferase (GST) and NAD(P) H:quinone oxidoreductase (NQO1), two phase-II enzymes, in HT-29 human colorectal cancer cells and mouse hepatoma Hepa 1c1c7 cells [123]. This induction was H₂S-dependent since pretreatment of the cells with the esterase inhibitor, NaF, reversed HS-ASA-mediated induction of NQO1 activity to near basal levels [123]. Treatment of both cell lines with HS-ASA resulted in increases in NQO1, GST and UDP-glucuronyltransferase (UGT1, another phase-II enzyme) protein levels as well as increases in the phase-I enzymes CYP1A1 and CYP2E1 [123]. However, *in vivo* studies showed that HS-ASA increased the hepatic protein levels of phase-II enzymes, GST, NQO1, and UGT but had no effect on the protein levels of CYP1A1 and CYP2E1 [123]. These findings strongly suggest that *in vivo* HS-ASA is a monofunctional inducer of phase-II metabolizing enzymes.

To our knowledge and according to the website, "ClinicalTrials.gov", none of the agents that are documented to be H₂S donors *in vivo* have entered any clinical trials. However, the Antibe Therapeutics website states that ATB-429, a hydrogen sulfide-releasing derivative of mesalamine, has successfully completed animal studies and is entering the first stage of clinical trials. This is for inflammatory bowel disease, i.e., Crohn's Disease and Ulcerative Colitis. All other H₂S-releasing hybrid compounds are in preclinical development.

4. Nitric oxide and hydrogen sulfide releasing aspirin (NOSH-aspirin)

4.1. Chemistry

Another approach in reducing the gastrointestinal side effects of NSAIDs has been to join nitric oxide (NO) to an NSAID through an aromatic or aliphatic linker, NO-NSAIDs (reviewed in [118]). The development of NO-NSAIDs was based on the observation that NO has some of the same properties as PGs within the gastric mucosa, thus modulating some components of the mucosal defense systems [124]. Preclinical studies have shown that NO-NSAIDs in general and NO-aspirin in particular, have chemopreventive properties and are much more potent than their traditional counterparts (reviewed in [118]). However, NO-NSAIDs possessing an aromatic linker can generate quinone methide intermediates, questioning the role of NO [125–127], or may have potential genotoxicity [128]. NO-aspirin with an aliphatic linker has a very high IC₅₀ for cell growth inhibition [129], limiting its usefulness.

Based on the chemistry of NO and H₂S, and the structural components of NO-NSAIDs and S-NSAIDs, recently it was postulated that an NSAID possessing moieties that could release both of these gasotransmitters might be more potent and effective than either one alone [130]. This led to the synthesis and characterization of a new class of anti-inflammatory and anti-cancer compounds, NOSH-aspirin, Fig 5, [130, 131]. Four NOSH compounds have been reported, in all of these aspirin was used as a scaffold to which NO and H₂S releasing moieties were coupled with one of the 1, 2 positions. Nitrate (-ONO₂) was used for NO release and this was attached to the aspirin through an aliphatic spacer, while one of the

following H₂S-releasing moieties, 5-(4-hydroxyphenyl)-3H-1, 2-dithiole-3-thione (ADT-OH), or 4-hydroxy benzothiazamide (TBZ) or lipoic acid were directly coupled to aspirin [130].

The NOSH-compounds described above were designed to release both NO and H₂S. Hence, using the methylene blue method, it was shown that NOSH-1 (NOSH-aspirin, NBS-1120) released H₂S both in vitro [131] and in vivo [130] settings. In order to show the actual release of H₂S gas from NOSH-aspirin, homogenized mouse liver was incubated with NOSH-ASA and examined in real time with a polarographic sensor [131]. The rate of H₂S production under these conditions was significantly less in media than when incubated with tissue, suggesting that significantly more H₂S was formed inside the cell, which may contribute to the high potency of NOSH-aspirin. Treatment of cells or animals with NOSH-aspirin showed dose-dependent increases in NO levels [130, 131].

4.2. Biology

4.2.1. Anti-inflammatory—NOSH-1 (NOSH-aspirin, NBS-1120) was shown to have anti-inflammatory properties similar to its parent compound, aspirin [130]. Using the carrageenan rat's paw edema model, NOSH-1 showed a reduction in paw inflammation, which was dose-dependent. Comparison of PGE₂ content of paw exudates from control, ASA-treated, and NOSH-1-treated animals showed a clear and significant COX inhibition by aspirin and NOSH-1. Evaluation COX expression in paw exudates showed that COX-2, which produces inflammatory PGE₂, was barely detectable in the controls, was significantly induced by carrageenan, and dose-dependently inhibited by NOSH-1 [130]. Administration of ASA to rats increased the proinflammatory cytokine tumor necrosis factor- α (TNF- α) in plasma by about 20-fold; however, this rise was considerably lower in the NOSH-1 treated animals [130].

4.2.2. Chemopreventive properties—The cell growth inhibitory properties of the four NOSH-compounds were evaluated in eleven different human cancer cell lines of six different histological subtypes. The cell lines were that of colon (HT-29: COX-1 and COX-2 positive, HCT 15: COX null, and SW480: COX-1 positive, low levels of endogenous COX-2), breast (MCF7: [ER(+)], MDA MB-231 and SKBR3: [ER(-)]); T-cell leukemia (Jurkat), pancreatic (BxPC3: both COX-1 and COX-2 positive, MIAPaCa-2: COX-null), prostate (LNCaP), and lung (A549). All four NOSH compounds were extremely effective in inhibiting the growth of these cell lines, however, NOSH-1 was the most potent, with IC₅₀s for cell growth inhibition ranging from 48–280 nM at 24h [130]. In a fold comparison study of the IC₅₀ values (ASA/NOSH 1–4), NOSH-1 was at least 100,000-fold more potent than ASA in HT-29 colon cancer cells. The increase in potency for NOSH-2, -3, and -4 in the same cell line after 24h of treatment were >60,000-fold, >600-fold, and >16,000-fold, respectively [130]. In HT-29 colon cancer cells, the enhanced potency of NOSH-1 increased to ~125,000-fold and ~250,000-fold after 48h and 72h of treatment [131].

NOSH-aspirin (NBS-1120) treatment of athymic nude bearing a human colon cancer xenograft, caused an 85% reduction in tumor volume after 18 days of treatment [131]. There were no overt signs of toxicity, as the mice in vehicle and NOSH-aspirin treated groups did not vary in their growth rate or behavior [131].

4.3. Outstanding questions

Research in the field of nitric oxide-hydrogen sulfide-releasing NSAIDs (NOSH-NSAIDs) is in its infancy. To date only a series of compounds based on aspirin have been reported. How other NOSH-NSAIDs such as NOSH-naproxen, NOSH-sulindac, NOSH-ibuprofen etc would behave has yet to be reported. The effect of these novel compounds on molecular

targets relevant to carcinogenesis should prove very interesting indeed. NO and H₂S share many similar actions, including modulation of leukocyte adherence to the vascular endothelium, both are “gasotransmitters”, and both bind avidly to hemoglobin. Do these similarities act to potentiate each other? Does *S*-nitrosylation of the transcription factor NF- κ B, which is central to the inflammatory process, by NO affect its *S*-sulfhydration by H₂S and visa versa? The answer to these questions and many others will prove to be most interesting.

5. Other questions remaining and future directions

While there is ever increasing literature on the effects of a variety of H₂S-donating compounds on physiological systems there have been relatively few studies of the effects of these drugs either with, or without, the H₂S-donating moiety on endogenous H₂S production. This is further complicated by the fact that in most instances tissue H₂S production or tissue H₂S concentration has been measured using methods of questionable accuracy. In this regard, perhaps the most outstanding issue that must be resolved is technical in nature; we must be able to measure and localize H₂S concentration at the cellular and intracellular level. This will allow us to specifically correlate the therapeutic effects of manipulation of endogenous H₂S metabolism on tissue function and to determine the efficacy of exogenous H₂S “donors” in delivery of H₂S to intracellular targets. We must also be able to identify the enigmatic artifact that appears in many assays, especially the methylene blue method and sulfide specific ion selective electrodes, that allude to perturbations in H₂S/sulfide metabolism. That said, there are several studies that may suggest that some of these anti-inflammatory drugs do themselves affect tissue H₂S production. Bilaska et al. [132] use the direct methylene blue method and reported that 10 mg aspirin per day (i.p.) for five days increased H₂S in brain (from 0.23 to 0.37 nmol/g wet weight; equivalent to 0.33 and 0.53 μ mol/l cytosol) and decreased it in the liver from 1.52 to 1.15 nmol/g wet weight; equivalent to 2.17 and 1.64 μ mol/l cytosol). Wili ski et al. [133] show that acetaminophen (*N*-acetyl-*p*-aminophenol) decreased H₂S content in the mouse brain but increased it in the heart liver and kidney. The authors use the direct methylene blue method and report control values of 1.5, 6.9, 3.9 and 7.1 μ g/g tissue forebrain heart liver and kidney, respectively, which if expressed as μ mol/l cytosol (assuming 70% of the cell weight is liquid) would be approximately 61, 290, 165, and 300 μ mol/l. Clearly these are unrealistically high, however, they are nevertheless suggestive of drug perturbation of intracellular sulfide. Using the same methods, this group also observed that the statin, atorvastatin, substantially increased kidney H₂S and slightly increased H₂S in brain heart and liver [134]. Wójcicka et al. [135] also observed that atorvastatin increased H₂S production in perivascular adipose tissue by producing coenzyme Q₉ deficiency and thereby inhibiting mitochondrial oxidation. Salloum et al. [136] measured H₂S production in a tadalafil-treated myocardial ischemia/reperfusion mouse heart model and reported that the H₂S concentration went from 0.6 to 3.2 μ M/mg protein. H₂S was measured amperometrically after homogenized tissue was incubated in zinc acetate (1%) for 30 minutes followed by TCA (10%) for 60 minutes, all at 37°C. These are equivalent to H₂S concentrations of 85.8 and 457.6 μ mol/l in the cytosol, respectively and at those levels are lethal to cells.

In our efforts to better understand the role of H₂S in health and disease, we need to think about genetically modified animals. This will allow us to delve into the mechanism(s) of cell signaling that involve H₂S. It would be important to understand the role of redox mechanisms and with the NOSH compounds the relative intracellular protein *S*-nitrosylation and *S*-sulfhydration and their interrelationships. In order to aim for targeted therapeutic applications, we also need a wider range of compounds that can release H₂S at different rates. This will go a long way in our efforts to better understand the precise cellular and

intracellular sites of action of H₂S. It appears that there will be a number of “gaseous solutions” in harnessing a wide array of human maladies.

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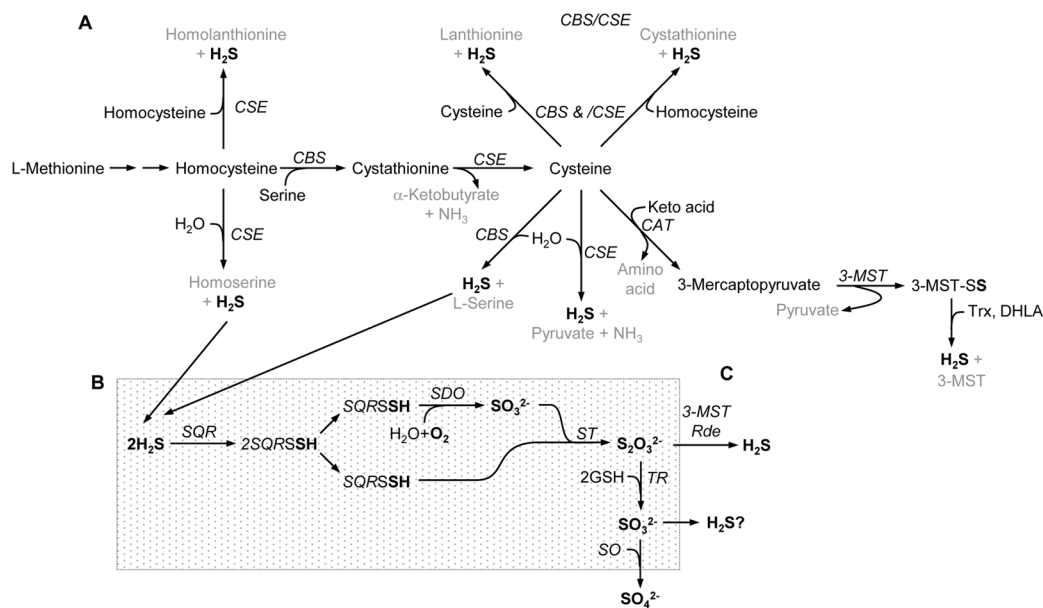


Figure 1. Metabolic pathways for H₂S biosynthesis (A) and degradation (B). Abbreviations: DHLA, dihydrolipoic acid; CAT, cysteine aminotransferase; CBS, cystathionine β-synthase; CSE, cystathionine γ-ligase; GSH, reduced glutathione; Rde, rhodanase; SDO, sulfur dioxygenase; SO, sulfite oxidase; SQR, sulfur:quinone oxidoreductase; ST, sulfur transferase; TR, thiosulfate reductase; Trx, thioredoxin; 3-MST, 3-mercaptopyruvate sulfur transferase.

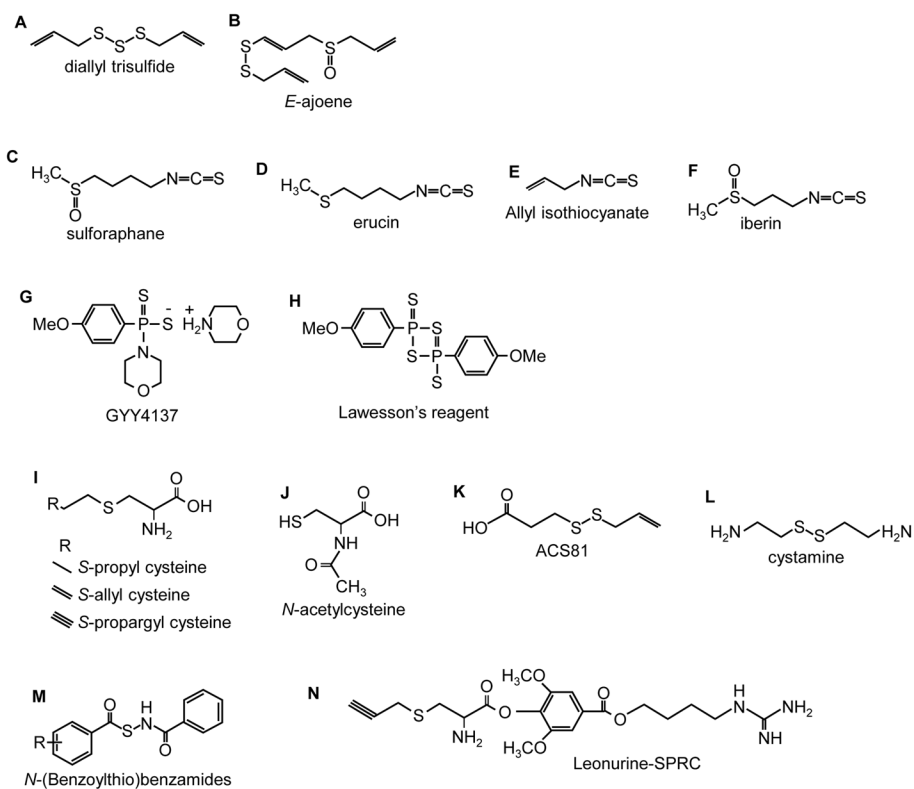


Figure 2.
Naturally occurring H₂S “donating” compounds.

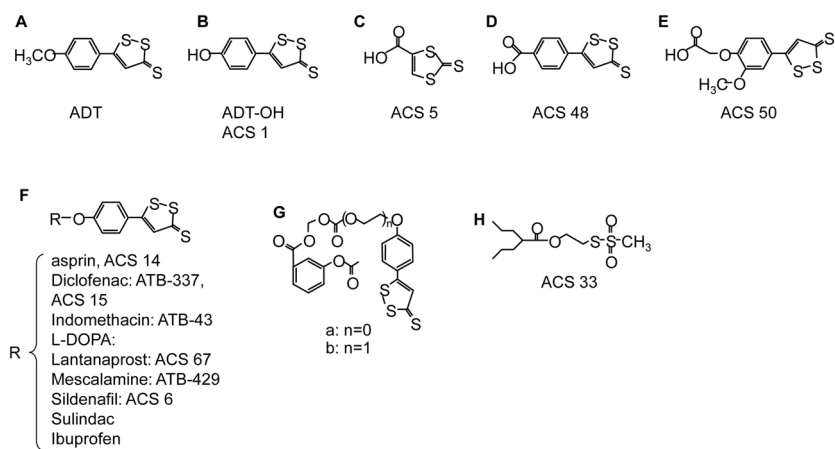


Figure 3.
Synthetic H₂S donating compounds.

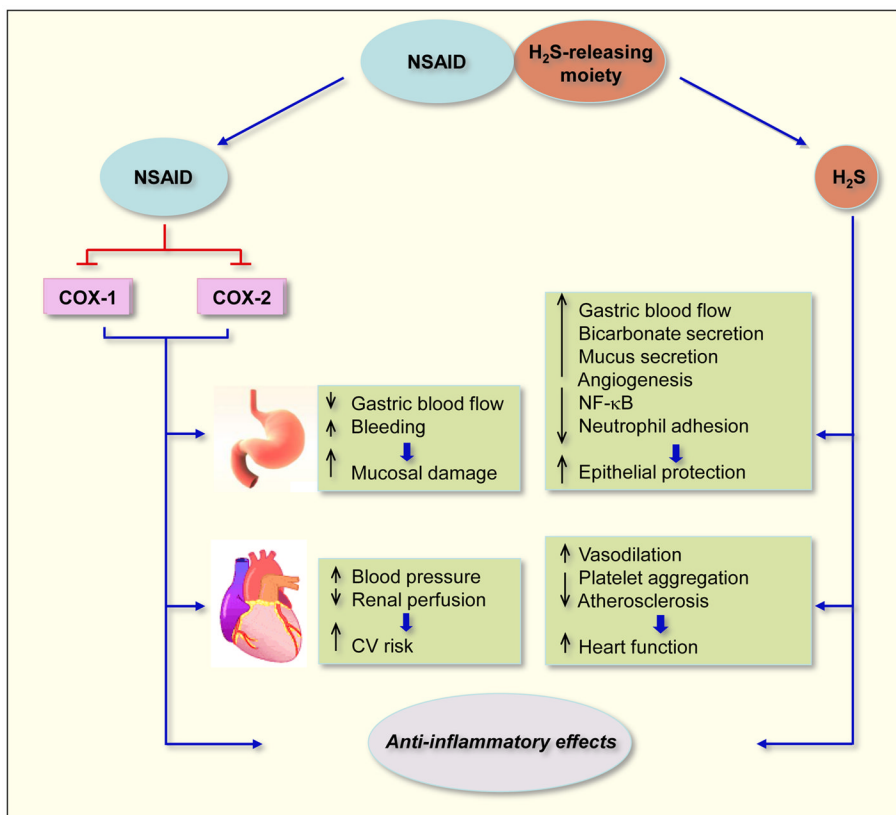
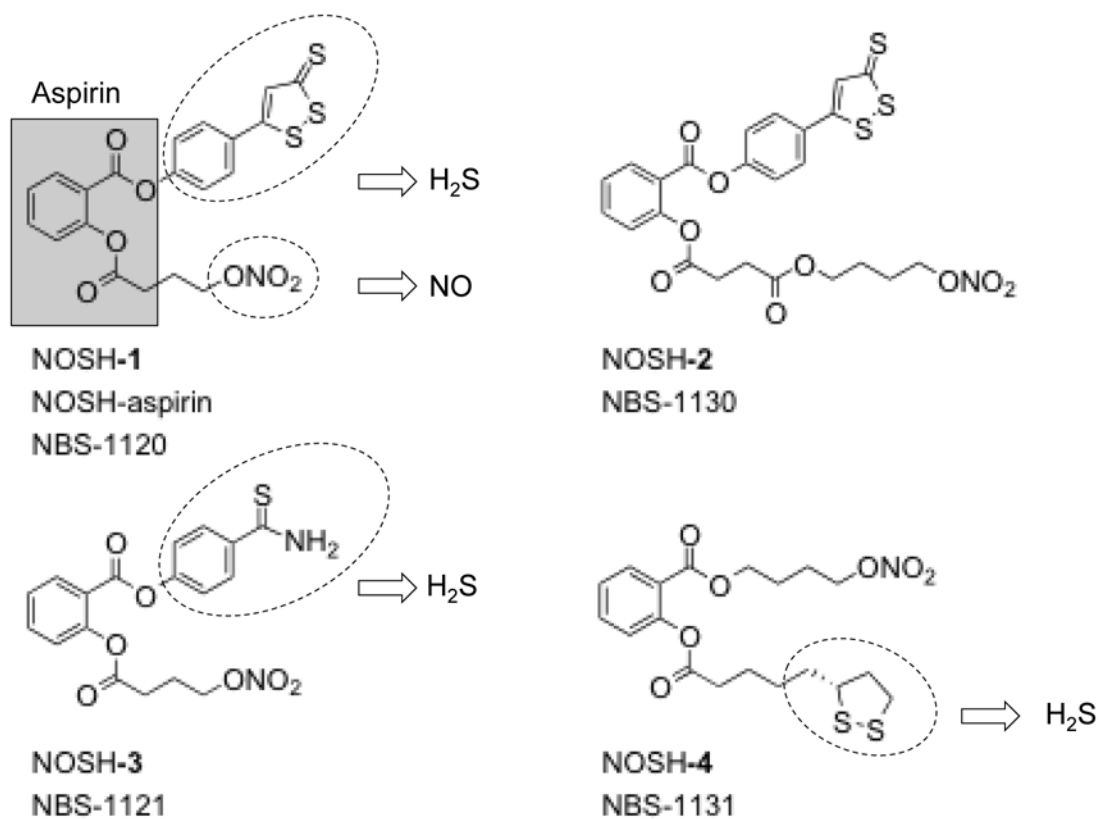


Figure 4. Mechanisms of action of H₂S-releasing NSAIDs. When hydrolyzed, H₂S-releasing NSAIDs produce the parent NSAID and the H₂S-releasing moiety from which H₂S is released. The NSAID component inhibits COX-1 and COX-2 resulting in compromised mucosal defense mechanisms, which may lead to ulcers. NSAIDs can reduce renal perfusion, which can lead to increases in blood pressure (BP) leading to cardiovascular (CV) damage. The released H₂S counteracts many of the detrimental effects of NSAIDs. These protective effects appear to be mediated through activation of K_{ATP} channels. H₂S enhances the mucosal defense mechanisms; causes vasodilation thus reducing BP leading to cardioprotective effects. Both the NSAID and H₂S have anti-inflammatory effects, the former through inhibition of COX and latter through inhibition of nuclear transcription factor κB (NF-κB).

**Figure 5.**

The chemical structures of NOSH compounds. NOSH-1: (4-(3-thioxo-3H-1,2-dithiol-5-yl)phenyl 2-((4-(nitrooxy)butanoyl)oxy)benzoate); NOSH-2: (4-(nitrooxy)butyl 2-((4-(3-thioxo-3H-1,2-dithiol-5-yl)phenoxy)carbonyl)phenyl)); NOSH-3: (4-carbamothioylphenyl 2-((4-(nitrooxy)butanoyl)oxy)benzoate); and NOSH-4: (4-(nitrooxy)butyl 2-(5-((R)-1,2-dithiolan-3-yl)pentanoyloxy)benzoate).