# **Probing the antibody-catalyzed water-oxidation pathway at atomic resolution**

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#### Contributed by Richard A. Lerner, December 19, 2003

**Antibodies can catalyze the generation of hydrogen peroxide (H2O2) from singlet dioxygen (1O\*2) and water via the postulated** intermediacy of dihydrogen trioxide (H<sub>2</sub>O<sub>3</sub>) and other trioxygen **species. Nine different crystal structures were determined to elucidate the chemical consequences to the antibody molecule itself of exposure to such reactive intermediates and to provide insights into the location on the antibody where these species could be generated. Herein, we report structural evidence for modifications of two specific antibody residues within the interfacial region of the variable and constant domains of different murine antibody antigen-binding fragments (Fabs) by reactive species generated during the antibody-catalyzed water oxidation process. Crystal structure analyses of murine Fabs 4C6 and 13G5 after UV-irradiation revealed complex oxidative modifications to tryptophan L163** and, in 4C6, hydroxylation of the C<sup> $\gamma$ </sup> of glutamine H6. These discrete **modifications of specific residues add further support for the ''active site'' of the water-oxidation pathway being located within the interfacial region of the constant and variable domains and highlight the general resistance of the antibody molecule to oxidation by reactive oxygen species generated during the wateroxidation process.**

crystal structure  $\vert$  amino acid modification  $\vert$  reactive oxygen species  $\vert$  $oxidative$  damage  $|$  UV-irradiation

Antibodies, regardless of source or antigenic specificity, can<br>catalyze the generation of multimolar equivalents of hydrogen peroxide  $(H_2O_2)$  from singlet dioxygen  $(^1O_2^*)$  and water (1) via the postulated intermediacy of dihydrogen trioxide  $(H_2O_3)$ (2). Activation of this process by photochemical sensitizers and visible light leads to highly efficient killing of bacteria, wherein the antibody-catalyzed water-oxidation pathway appears to generate an additional oxidant with a chemical signature similar to that of ozone  $(O_3)$  (3). More recently, we reported that this process regioselectively converted antibody antigen-binding fragment (Fab) 4C6-bound benzoic acid into *para*-hydroxybenzoic acid, as well as regioselectively hydroxylating the 4-position of the indole of a single tryptophan residue, L163, within the 4C6 Fab structure (4).

A number of structural questions remain to be answered regarding the ability of antibodies to catalyze this biologically unique oxidation pathway. First, where is the location of the active site in which  ${}^{1}O_{2}^{*}$  and H<sub>2</sub>O react, and second, what are any oxidative consequences to the antibody molecule itself as a result of carrying out this reaction? To understand these issues, we determined a series of x-ray structures from Fab fragments of antibodies that were previously used to document the wateroxidation pathway.

#### **Materials and Methods**

PNAS P

**UV-Irradiation of Fab Fragments.** Murine Fab fragments were generated and purified as previously described for 4C6 (5) and 13G5 (6). To obtain UV-irradiated Fabs, purified 4C6 Fab (200  $\mu$ l, 26 mg/ml in 0.1 M sodium acetate buffer, pH 5.5) or 13G5 Fab (200  $\mu$ l, 15 mg/ml in PBS, pH 7.4) were irradiated on a transilluminator with UV light  $(312 \text{ nm}, 0.8 \text{ mW} \cdot \text{cm}^{-2})$  for 30 min at room temperature.

**Hydrogen Peroxide Soaking Experiments.** An aqueous solution of  $H<sub>2</sub>O<sub>2</sub>$  (final concentration 3 mM) was added to a crystal of the native 4C6 Fab grown from  $15\%$  (wt/vol) polyethylene glycol  $(PEG)$  4000/0.2 M sodium acetate/0.1 M Tris $\cdot$ HCl, pH 8.5. The crystal was allowed to react for 3 min at room temperature before being flash-cooled and stored in liquid nitrogen.

**Crystallization Conditions.** Crystallization experiments were performed by using the sitting-drop vapor-diffusion method at 295 K with a drop consisting of 1  $\mu$  of reservoir solution and 1  $\mu$  of protein solution. For 4C6 Fab, UV-irradiated 4C6 Fab crystal **1** (see Table 1) and the native 4C6 Fab crystal **2** were grown from  $15\%$  (wt/vol) PEG  $4000/0.2$  M ammonium acetate/0.1 M sodium acetate, pH 4.6. The UV-irradiated 4C6 crystal **3** and the native 4C6 crystal 4 (5) were grown from 15% (wt/vol) PEG  $4000/0.2$  M ammonium acetate/0.1 M trisodium citrate, pH 5.6, whereas native  $4C6$  crystals **5** and **6** arose from  $15\%$  (wt/vol) PEG  $4000/0.2$  M sodium acetate/0.1 M Tris $\cdot$ HCl, pH 8.5. For 13G5 Fab, UV-irradiated 13G5 crystal **7** and native 13G5 crystal **8** were grown from  $20\%$  (wt/vol) PEG  $3000/0.2$  M zinc acetate/  $5-7\%$  (vol/vol) isopropyl alcohol/0.1 M imidazole, pH 8.0.

**Data Collection and Processing.** Crystals of 4C6 Fab for data sets A–G were cryocooled in liquid nitrogen after soaking in mother liquid substituted with  $25\%$  (vol/vol) glycerol. Crystals of  $13G5$ Fab for data sets H and I were flash-cooled in liquid nitrogen with  $20\%$  (vol/vol) glycerol as cryoprotectant. All of the data sets were integrated and scaled with HKL2000 (7) (see Tables 1 and 2).

For the UV-irradiated 4C6 crystal **1**, data set A was collected in-house on a MAR Research image plate detector (diameter 300 mm) mounted on a Rigaku x-ray generator (50 kV, 70 mA) as described (4). A second data set, B, was also collected from this crystal **1** at the Stanford Synchrotron Radiation Laboratory (SSRL) beamline 11-1. A native 4C6 crystal **2**, data set C, was collected as a control for data sets A and B at SSRL beamline 9-1. For the UV-irradiated 4C6 crystal **3**, data set D was collected at SSRL beamline 11-1, whereas the control native 4C6 crystal **4** data set E was collected at the Advanced Light Source (ALS) as described (5). For the  $H_2O_2$ -soaked 4C6 crystal 5, data

Abbreviations: Fab, antibody antigen-binding fragment; PEG, polyethylene glycol; SSRL, Stanford Synchrotron Radiation Laboratory.

Data deposition: The six structures and structure factors of 4C6 Fab and two structures and structure factors of 13G5 Fab have been deposited at the Protein Data Bank, www.rcsb.org (PDB ID codes 1ru9, 1rua, 1ruk, 1rul, 1rum, 1rup, 1ruq, and 1rur, respectively).

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### Table 1. Data collection and refinement statistics for UV-irradiated, H<sub>2</sub>O<sub>2</sub>-soaked, and native 4C6 Fab crystals



Data set

ALS, Advanced Light Source; APS, Advanced Photon Source; rmsd, rms deviation.

\*The data statistics for published data set E (5) are shown here for comparison.

†Parentheses denote outer-shell statistics.

PNAS PN

 ${}^{\ddagger}R_{\text{sym}} = \Sigma_h \Sigma_l |I_l(\mathsf{h}) - \langle l(\mathsf{h}) \rangle|/\Sigma_h \Sigma_l |I_l(\mathsf{h})$ , where  $\langle l(\mathsf{h}) \rangle$  is the average intensity of  $i$  symmetry-related observations of reflections with Bragg index  $h$ .

 ${}^5R_{\text{cryst}} = \Sigma_{hk}$   $|F_{\text{o}} - F_{\text{c}}/\Sigma_{hk}$   $|F_{\text{o}}|$ , where  $F_{\text{o}}$  and  $F_{\text{o}}$  are the observed and calculated structure factors.

¶*R*free was calculated as for *R*cryst, but on 5% of data excluded before refinement.

set F was collected at SSRL beamline 9-1, and its control native

soaked 4C6 Fab structure F. Model rebuilding was performed by using  $\sigma$  (10).

## **Results**

**Modification of TrpL163 of 4C6 Fab.** Comparison of the UVirradiated (A) and native (C) 4C6 Fab structures at 2.50- and 1.40-Å resolution, respectively, showed no discernible changes in their main-chain conformations (rms deviation 0.25 Å) (Fig. 1). However, closer inspection revealed that one particular tryptophan,  $Trp^{L163}$ , and one glutamine,  $Gln^{H6}$ , were consistently modified upon UV-irradiation of 4C6 Fab (Fig. 1, compare data sets A and C). The structural modifications were interpreted as regioselective hydroxylation of the 4-position as well as reduced definition at the 1-position of the indole of Trp<sup>L163</sup>, and as  $\gamma$ -hydroxylation of Gln<sup>H6</sup> (Fig. 2 *aA*, *bA*, *cA*, and *dA* and Fig. 3 *aA*, *bA*, *cA*, and *dA*, respectively). Analysis of the higher 1.75-Å resolution structure B collected at a synchrotron source from the same crystal **1** as structure A revealed that TrpL163 had been modified more severely (Fig. 2 *aB*, *bB*, *cB*, and *dB*). The modification involved an apparent fragmentation at a locus between the 4- and 7-positions of the phenyl ring of the tryptophan indole; no modification of TrpL163 (Fig. 2 *aC*, *bC*, *cC*, and *dC*) was detected in the control native data set C with crystal **2** that was collected at the same synchrotron source (Table 1).

This extensive modification of TrpL163 was also apparent in electron density maps from the 1.88-Å resolution data set D

4C6 crystal **6** data set G was collected on an attenuated beamline 19-ID at the Advanced Photon Source (APS).

The UV-irradiated 13G5 crystal **7** data set H was collected in-house on a Mar detector (diameter 345 mm) mounted on a Siemens x-ray generator (50 kV, 100 mA), and its control native 13G5 crystal **8** data set I was collected at SSRL beamline 9-1.

**Crystallographic Refinement.** For all data sets of 4C6 Fab, CNS (8) was used for rigid-body refinement, using the published native crystal structure E (PDB ID code 1ncw) as the initial model (5). Data set A was refined only with CNS, whereas refinements of all other structures were completed with SHELXL (9) (see Table 1). Anisotropic *B*-value refinement was applied to high-resolution data sets C, E, F, and G. The 13G5 Fab-inhibitor complex (PDB ID code 1a3l) (6) was used as the initial model for rigid-body refinement of the 13G5 data sets by using CNS (8). Data set H was refined only with CNS (8) (see Table 2), whereas high-resolution structure I was completed with SHELXL (9), using anisotropic *B*-value refinement (see Table 2). In all 4C6 and 13G5 Fab structures, Trp<sup>L163</sup> was initially modeled and refined as alanine to avoid any potential bias in the electron-density maps. In UV-irradiated 4C6 Fab structures A, B, and D and  $H_2O_2$ -soaked 4C6 Fab structure F, TrpL163 was also refined as 4-hydroxytryptophan, and Gln<sup>H6</sup> as  $\gamma$ -hydroxyglutamine except in the H<sub>2</sub>O<sub>2</sub>-

# **Table 2. Data collection and refinement statistics for UV-irradiated and native 13G5 Fab**



See footnotes to Table 1.

(Fig. 2 *aD*, *bD*, *cD*, and *dD*), also collected at the SSRL beamline 11-1 from a second UV-irradiated 4C6 Fab crystal **3** that had been obtained under different crystallization conditions. It should be noted that, in contrast to crystal **1**, crystal **3** was not previously irradiated on the in-house x-ray facility. Again, no structural modifications were observed with the control native 4C6 Fab crystal **4** (Fig. 2 *aE*, *bE*, *cE*, and *dE*).

Similar observations were made for the  $H_2O_2$ -soaked 4C6 crystals. Fab crystal **5** at 1.48 Å (data set F) also revealed that the overall backbone structure of the  $H_2O_2$ -soaked 4C6 Fab had no discernible changes relative to the native structure (data set G). However, the side chain of TrpL163 was also modified in the  $H_2O_2$ -soaked crystal 5 (Fig. 2  $aF$ ,  $bF$ ,  $cF$ , and  $dF$ ) relative to the control native 4C6 Fab (Fig. 2 *aG*, *bG*, *cG*, and *dG*). The  $H<sub>2</sub>O<sub>2</sub>$ -induced modifications to Trp<sup>L163</sup> (data set F) were similar to those observed with UV-irradiated 4C6 structures B (Fig. 2 *aB*, *bB*, *cB*, and *dB*) and D (Fig. 2 *aD*, *bD*, *cD*, and *dD*). An additional modification, unique to data set F, occurred with the antigen-binding site  $Trp^{H97}$  (see Fig. 1) that also appeared to be oxidized with low occupancy but at the 7-position of the indole (data not shown). No other discernible chemical changes in the remaining tryptophan residues were detected in any 4C6 Fab data sets. However, several other solvent-exposed residues had possible chemical modifications in the  $H_2O_2$ -soaked 4C6 crystals, including the side chains of Gln<sup>L38</sup>, Gln<sup>L42</sup>, Asp<sup>L60</sup>, Tyr<sup>L87</sup>, Asp<sup>H72</sup>, and AspH99, although this effect appeared to be only partial (data not shown) with some correlation between solvent accessibility and susceptibility of the individual amino acid to modification.

**Modification of GlnH6 of 4C6 Fab.** In the UV-irradiated 4C6 Fab crystal structure A, additional spherical density proximal to the side-chain  $C^{\gamma}$  of Gln<sup>H6</sup> was revealed from a difference Fourier map (Fig. 3*dA*). This density was not present in maps from several other data sets collected from native 4C6 Fab crystals (such as in Fig. 3 *dC* and *dE*). This additional electron density was also found in the maps calculated from data sets B (Fig. 3*dB*) and D (Fig. 3*dD*). A hydroxyl group was modeled into the density and refined to an occupancy factor of  $\approx 60-80\%$  in data sets A, B, and D. Interestingly, no hydroxylation of  $Gln<sup>H6</sup>$  was observed from the  $H_2O_2$ -soaked 4C6 Fab data set F.

**Modification of TrpL163 of 13G5 Fab.** Comparison of the UVirradiated (H) and native (I) 13G5 Fab structures at 1.86- and 1.50-Å resolution, respectively, showed no discernible difference in their main-chain structures (data not shown). Only





Fig. 1. Stereoview of the crystal structure of 4C6 Fab, with the C<sup>«</sup> trace of the light (L) and heavy (H) chains colored in light and dark gray, respectively. The modified tryptophan Trp<sup>L163</sup> is highlighted in red, and other tryptophan residues (such as Trp<sup>H97</sup>) are colored green. The modified glutamine residue Gln<sup>H6</sup> is also colored red. All of the figures were generated in BOBSCRIPT (12) and rendered in RASTER3D (13).

PNAS PN



Fig. 2. Fourier electron density maps for Trp<sup>L163</sup> in 4C6 Fab: the UV-irradiated data set A, the UV-irradiated data set B, the native control data set C, the UV-irradiated data set D, the native control data set E, the H<sub>2</sub>O<sub>2</sub>-soaked data set F, and the native control data set G. For *a* and *b*, the tryptophan was modeled and refined as 4-hydroxytryptophan for data sets A, B, D, and F and was modeled and refined as tryptophan for control data sets C, E, and G. As a control (*c* and *d*), the tryptophan residue was modeled and refined as alanine to avoid model bias. (a)  $2F_0 - F_c$  maps (blue), contoured at 1.0  $\sigma$ . (b)  $F_0 - F_c$ maps, contoured at 2.5  $\sigma$  (green) and  $-2.5 \sigma$  (red) for data sets A, B, and D and contoured at 3.0  $\sigma$  (green) and  $-3.0$   $\sigma$  (red) for data sets C, E, F, and G. (c) 2 $F_0$  –  $F_c$  maps (blue), contoured at 0.8  $\sigma$  for data sets A, B, C, D, and E, and contoured at 0.9  $\sigma$  for data sets F and G. (d)  $F_o - F_c$  maps, contoured at 2.5  $\sigma$  (green) for data sets A, B, and D, and contoured at 3.0  $\sigma$  (green) for data sets C, E, F, and G.



Fig. 3. Fourier electron density maps around Gln<sup>H6</sup> in 4C6 Fab: for data sets A to E as described for Fig. 2. For *a* and *b*, the glutamine residue was modeled and refined as  $\gamma$ -hydroxyglutamine for data sets A, B, and D and modeled and refined as glutamine for control data sets C and E. As a control (*c* and *d*), the glutamine residue was modeled and refined as glutamine for all data sets. (*a*)  $2F_{\rm o} - F_{\rm c}$  maps (blue), contoured at 1.0  $\sigma$ . (b)  $F_{\rm o} - F_{\rm c}$  maps, contoured at 3.0  $\sigma$ (green) and  $-3.0$   $\sigma$  (red). (c)  $2F_o - F_c$  maps (blue), contoured at 1.0  $\sigma$ . (d)  $F_o$  - $F_c$  maps, contoured at 3.0  $\sigma$  (green).

oxidative modifications to TrpL163 were apparent in structure H, which was determined from an in-house data set (Fig. 4 *aH*, *bH*, *cH*, and *dH*). The 1-, 2-, and 8-positions of the indole of the TrpL163 have reduced definition in the electron density maps relative to that of the control structure I (Fig. 4 *aI*, *bI*,  $cI$ , and  $dI$ ). Trp<sup>L163</sup> is located in the constant region of 13G5 Fab light chain, and is in a position similar to that in the 4C6 Fab. No modifications were observed for the H6 residue in UV-irradiated 13G5 Fab, which is, in this case, glutamate instead of glutamine as in 4C6 Fab.

#### **Discussion**

The key conclusion from this study is that only very few modifications occur in antibodies upon irradiation with UV light and, by assumption, during processing of  ${}^{1}O_{2}^{*}$  into  $H_{2}O_{2}$ . This structural observation supports our kinetic investigations, which show that antibodies continue to catalyze the water-oxidation pathway with no apparent reduction in activity even after several



Fig. 4. Fourier electron density maps showing Trp<sup>L163</sup> in 13G5 Fab for UV-irradiated data set H and native control data set I. For *a* and *b*, the tryptophan residue was refined as tryptophan, whereas for control (*c* and *d*), the tryptophan residue was refined as alanine to avoid model bias. (*a*) 2*F*<sup>o</sup>  $F_c$  maps (blue), contoured at 1.0  $\sigma$ . (b)  $F_o - F_c$  maps, contoured at 3.0  $\sigma$  (green) and  $-3.0 \sigma$  (red). (*c*)  $2F_0 - F_c$  maps (blue), contoured at 0.8  $\sigma$ . (*d*)  $F_0 - F_c$  maps, contoured at 3.0  $\sigma$  (green).

hours of UV-irradiation (2). However, one residue,  $Trp^{L163}$ , is consistently modified during UV-irradiation and is a highly conserved in murine antibodies. The TrpL163 residue is located in the constant domain of the light chain, and its side chain projects into the interfacial region between the variable and constant domains. Of the 11 tryptophan residues within 4C6 Fab, TrpL163 is the only one modified in the water-oxidation process. TrpL163 is the most solvent-accessible of all of the 4C6 tryptophan residues, with a solvent-accessible surface area of  $113 \text{ Å}^2$ (1.4-Å probe radius). However, it is unlikely that exposure to solvent is the sole reason for its modification, given that most of the other tryptophan residues have some solvent exposure. For example, no measurable modifications occurred on TypH97, which is located atop the antigen-binding site and is only slightly less solvent accessible  $(100 \text{ Å}^2)$  (Fig. 1).

Nevertheless, the nature of the structural modifications to TrpL163 in the Fabs 4C6 and 13G5 is quite different. In the 4C6 Fab, the 4-position of the indole appears to be hydroxylated and, upon synchrotron x-ray irradiation, an apparent fragmentation occurs at a locus between the 4- and 7-positions of the 4 hydroxyphenyl ring of the indole. In the 13G5 Fab structure, it is the indole's pyrrole ring that is disrupted.

 $Gln<sup>H6</sup>$  is located in the variable domain of the heavy chain of the 4C6 Fab with its side chain buried in a hydrophobic core, but close to the interfacial region of the constant and variable domains (Fig. 1). Gln<sup>H6</sup> appears to be hydroxylated at the  $\gamma$ position in the UV-irradiated 4C6 crystal structures (A, B, and D). For 13G5 Fab, however, H6 is a glutamate instead of glutamine residue, and no modification was found.

Chemical modification of tryptophan in aqueous systems by the hydroxyl radical (HO•), generated either by pulsed radiolysis of water or by Fenton chemistry, leads to a complex mixture of products that includes *N*-formylkynurenine, multiple hydroxylation products, and fragmentation products (11). Much less is known about the chemical products of glutamine and HO<sup>\*</sup>; however,  $\gamma$ -hydroxylation is an entirely reasonable modification.

Thus, the combined analyses of the structural modifications to TrpL163 in both the 4C6 and 13G5 Fabs and the regioselective hydroxylation of Gln<sup>H6</sup> in the 4C6 Fab strengthen the evidence for generation of a hydroxyl radical (or a surrogate such as the hydrotrioxy radical) in the antibody-catalyzed water-oxidation pathway.

The hydrogen peroxide soaking experiment was undertaken to investigate the possible location of the active site for the wateroxidation pathway in antibodies. Hydrogen peroxide is the stable product of the antibody-catalyzed water-oxidation process and is known to inhibit (IC<sub>50</sub>  $\approx$  500  $\mu$ M) the overall process by presumed binding within the putative catalytic site (2). Therefore, it was surmised that, if a discrete binding site existed,  $H_2O_2$ would be sequestered within the active site by soaking the crystal of 4C6 Fab in an  $H_2O_2$  solution at a concentration well above its  $IC_{50}$ . Given that x-ray irradiation of  $H_2O_2$ , even at liquid nitrogen temperatures, generates hydroxyl radicals, it was anticipated that during the in-house/synchrotron x-ray analysis of the  $H_2O_2$ soaked crystals that residues in or close to the active site could become hydroxylated.

The  $H_2O_2$  crystal soaking experiment revealed clear evidence that TrpL163 was indeed modified (with a high occupancy,  $>80-90\%$ ) within the 4C6 Fab structure. The nature of the modification was very similar to that observed when the 4C6 Fab was treated with UV-irradiation, i.e., when the water-oxidation pathway was activated, the phenyl ring of the indole side chain was fragmented.

Thus, evidence for structural modifications of Trp<sup>L163</sup> in different Fabs by UV-irradiation supports the proposal that TrpL163 is located close to the site of reactive oxygen species generated during the antibody-catalyzed water-oxidation pathway and that the precise locus and nature of the modification to the side chain is entirely dependent upon the indole side chain's orientation and localized environment with respect to the attacking species. Furthermore, the  $H_2O_2$  crystal soaking experiment supports the notion that the active site for the antibodycatalyzed water-oxidation pathway is situated at or near the interfacial region of the variable and constant domains and that its locus is in close proximity to TrpL163.

The antibody-catalyzed water-oxidation pathway is a complex cascade initiated by singlet oxygen, ultimately generating hydrogen peroxide. A cascade of intermediates have been postulated to include dihydrogen trioxide, ozone, and even hydroxyl radical surrogates, such as the hydrotrioxy radical (HO<sub>3</sub>). This study provides the strongest evidence yet for the structural location of the active site of this highly conserved process as being in the interfacial region of the variable and constant domains. Furthermore, the relatively small number of structural modifications to side-chain residues in the antibody after activation of the water-oxidation pathway reinforces the remarkable ability of these proteins to handle reactive oxygen species in a manner that does not lead to their rapid destruction.

We thank the staffs of the Stanford Synchrotron Radiation Laboratory beamlines BL9-1 and BL11-1, the Advanced Photon Source beamline 19-ID, and the Advanced Light Source beamline 5.0.2, and several lab members, especially Dr. Xiaoping Dai, for help with data collection. We are grateful to Dr. Kim D. Janda for antibody samples. This work was supported by National Institutes of Health Grant CA27489 (to I.A.W. and R.A.L.) and The Skaggs Institute for Chemical Biology. This is manuscript no. 16245-MB of The Scripps Research Institute.

Jones, T., Bautista, C., Ruedi, J. M., Gutierrez, A., Janda, K. D., *et al*. (2002) *Science* **298,** 2195–2199.

- 4. Wentworth,P.,Wentworth,A.D.,Zhu,X.Y.,Wilson,I.A.,Janda,K.D.,Eschenmoser, A. & Lerner, R. A. (2003) *Proc. Natl. Acad. Sci. USA* **100,** 1490–1493.
- 5. Zhu, X. Y., Heine, A., Monnat, F., Houk, K. N., Janda, K. D. & Wilson, I. A. (2003) *J. Mol. Biol.* **329,** 69–83.

<sup>1.</sup> Wentworth, A. D., Jones, L. H., Wentworth, P., Janda, K. D. & Lerner, R. A. (2000) *Proc. Natl. Acad. Sci. USA* **97,** 10930–10935.

<sup>2.</sup> Wentworth, P., Jones, L. H., Wentworth, A. D., Zhu, X. Y., Larsen, N. A., Wilson, I. A., Xu, X., Goddard, W. A., Janda, K. D., Eschenmoser, A. & Lerner, R. A. (2001) *Science* **293,** 1806–1811.

<sup>3.</sup> Wentworth, P., McDunn, J. E., Wentworth, A. D., Takeuchi, C., Nieva, J.,

- 6. Heine, A., Stura, E. A., Yli-Kauhaluoma, J. T., Gao, C. S., Deng, Q. L., Beno, B. R., Houk, K. N., Janda, K. D. & Wilson, I. A. (1998) *Science* **279,** 1934–1940.
- 7. Otwinowski, Z. & Minor, W. (1997) *Methods Enzymol.* **276,** 307–326.
- 8. Brünger, A. T., Adams, P. D., Clore, G. M., DeLano, W. L., Gros, P., Grosse-Kunstleve, R. W., Jiang, J. S., Kuszewski, J., Nilges, M., Pannu, N. S., *et al*. (1998) *Acta Crystallogr. D* **54,** 905–921.
- 9. Sheldrick, G. M. & Schneider, T. R. (1997) *Methods Enzymol.* **277,** 319–343.
- 10. Jones, T. A., Zou, J. Y., Cowan, S. W. & Kjeldgaard, M. (1991) *Acta Crystallogr. A* **47,** 110–119.
- 11. Maskos, Z., Rush, J. D. & Koppenol, W. H. (1992) *Arch. Biochem. Biophys.* **296,** 514–520.
- 12. Esnouf, R. M. (1999) *Acta Crystallogr. D* **55,** 938–940.
- 13. Merritt, E. A. & Murphy, M. E. P. (1994) *Acta Crystallogr. D* **50,** 869– 873.

PNAS PNAS PNAS