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Gadolinium MRI Contrast Agents Based on Triazine Dendrimers: Relaxivity and In Vivo Pharmacokinetics

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Abstract

Four gadolinium (Gd)-based macromolecular contrast agents, **G3-(Gd-DOTA)24, G5-(Gd-DOTA)96, G3-(Gd-DTPA)24**, and **G5-(Gd-DTPA)96**, were prepared that varied in the size of dendrimer (generation three and five), the type of chelate group (DTPA or DOTA), and the theoretical number of metallated chelates (24 and 96). Synthesis relied on a dichlorotriazine derivatized with a DOTA or DTPA ligand that was incorporated into the dendrimer and ultimately metallated with Gd ions. Paramagnetic characteristics and in vivo pharmacokinetics of all four contrast agents were investigated. The DOTA-containing agents, **G3-(Gd-DOTA)24** and **G5-(Gd-DOTA)96**, demonstrated exceptionally high r1 relaxivity values at off peak magnetic fields. Additionally, **G5-(Gd-DOTA)96** showed increased r1 relaxivity in serum compared to that in PBS, which was consistent with in vivo images. While **G3-(Gd-DOTA)24** and **G3-(Gd-DTPA)²⁴** were rapidly excreted into the urine, **G5-(Gd-DOTA)96** and **G5-(Gd-DTPA)96** did not clear as quickly through the kidneys. Molecular simulation of the DOTA-containing dendrimers provides a single-molecular level characterization of the structures and suggests that a majority of the metallated ligands are accessible to water. These triazine dendrimer-based MRI contrast agents exhibit several promising features such as high in vivo r1 relaxivity, desirable pharmacokinetics, and well-defined structure.

Supporting Information.

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Details of synthesis and spectral data. This material is available free of charge via the Internet at [http://pubs.acs.org.](http://pubs.acs.org)

Keywords

magnetic resonance imaging; triazine dendrimer; relaxivity; pharmacokinetics; gadolinium

INTRODUCTION

Numerous nanomaterials—inorganic, organic, and hybrid—with unique chemical and physical characteristics are being explored for biologic application. Among them, synthetic dendritic macromolecules are versatile and promising for biological and medical applications because of their flexible, reproducible syntheses and biocompatible characteristics.¹ When used as macromolecular MRI contrast agents, several advantages over low molecular weight gadolinium-chelates emerge including enhanced r1 relaxivity^{2, 3} because of efficient water exchange, unique pharmacokinetics that can be adapted to target the blood pool,^{4, 5} liver,^{6, 7} kidney,^{8, 9} and lymphatic flows,^{10, 11} the ability to be a drug carrier, and flexibility for labeling due to their inherent multivalency.12, 13 The improved relaxivity and prolonged pharmokinetics help minimize the injected dose of gadolinium ions, thus, reducing the risk of nephrogenic systemic fibrosis (NSF) , $^{14-16}$ a recent concern in the design of gadolinium-based contrast agents. 17

Monodispersity, a feature of lower generation dendrimers, is a desirable characteristic of macromolecular contrast agents because it facilitates characterization and confidence in composition which lead to materials with batch-to-batch consistency vis a vis in vivo behavior. On increasing generation, however, the dispersity of most dendrimers increases. For instance, more than 10% of commercial PAMAM dendrimers larger than generation 4 appear to be defective, containing missing branches, molecular loops, and dimers^{18, 19} leading to inconsistent product quality that can be a potential regulatory concern when pursuing clinical application of dendrimers. These defects may be critical because in vivo pharmacokinetics of dendrimers can be influenced by the nature of the chemical core, interior, and surface of the dendrimer although this is still poorly understood. Even with the same surface modifications and similar sizes, dendrimers with different interiors such as PAMAM and PPI showed distinct in vivo pharmacokinetics.⁸ Moreover, the number of generations and nature of the conjugated chelates used to bind Gadolinium are also critical factors that alter relaxivity and pharmacokinetics.^{20, 21}

Triazine dendrimers can be reproducibly synthesized with well-defined structure.22-28 Their syntheses are straightforward. They are amenable to multiple characterization methods. They are stable. These characteristics offer distinct advantages over other classes of dendrimers, creating opportunities for drugs or diagnostic agents with more predictable and tunable pharmacokinetic profiles. Here, we describe a family of Gd-based MRI contrast agents that utilize monodisperse triazine dendrimers of two different generations²⁴ to yield

generation 3 and 5 targets upon conjugation with the DOTA or DTPA-containing ligands (Chart 1).

MATERIALS AND METHODS

Synthesis

The macromolecular Gd^{3+} chelates were synthesized using the triazine dendrimers of generation 3 and 5 prepared as previously reported.24 To incorporate either DOTA or DTPA groups, a conjugation strategy using a functionalized monomer was adopted (Scheme 1). That is, the t-butyl ester-protected ligands were incorporated into the dichlorotriazine **5** or **8** in 88% yield. Reactions with the dendrimer platforms afforded the four poly(monochlorotriazine) dendrimers, **6, 7, 9**, and **10**, respectively. The surface group HEEP, 1-[2-(2-hydroxyethoxy)ethyl]piperazine, was chosen due to high reactivity of the cyclic secondary amine towards monochlorotriazines and the hydrophilic auxiliary groups. Following an acidic deprotection, the desired macromolecular ligands, **1, 2, 3**, and **4** were obtained. From the G3 and G5 dendritic platforms, these agents were synthesized in four steps in 70-80% overall yield.

Preparation of Gd(III) Complexes

The dendrimers (1-4) were dissolved in deionized water. $Gd(OAc)$ ₃ (24 equiv. for 1 and 3, 96 equiv. for **2** and **4**) was added to the solution at room temperature. The pH of the solution was adjusted to 5.5-6 by adding dilute NaOH (aq). Xylenol orange solution was used to monitor the presence of free gadolinium ions. The reaction solution was stirred at 50 °C for 2 days. In order to remove nonspecifically bound Gd^{3+} ions, the solution was stirred with 3 mM of EDTA at room temperature for 2 hrs. The resulting solution was purified by diafiltration using Amicon stirred ultrafiltration cell equipment (PLCC membrane, MWCO 5 kDa for 1 and 3, 10 kDa for 2 and 4) in deionized water over 3 days. No free Gd^{3+} ions were observed from the purified solution with xylenol orange test. Each solution was lyophilized to provide the dendrimer-Gd complexes, **G3-(Gd-DOTA)24** (from **1**), **G5-(Gd-DOTA)⁹⁶** (from 2), $G3-(Gd-DTPA)_{24}$ (from 3), and $G5-(Gd-DTPA)_{96}$ (from 4), respectively.

Computational Procedure

The protocol used for the creation and the parametrization of the **G3-(Gd-DOTA)24** and **G5-(Gd-DOTA)96** dendrimers has been previously used by us for the simulation of other dendritic molecules.29 The dendritic scaffolds were constructed as composed by different residues: a central (CEN) unit, the flexible (FLX) units composing the branches, the rigid triazine residues (RIG) and the surface units, namely the DOTA (DOT) groups and the protective terminal units (END). The entire modeling work was carried out using the AMBER 11 suite of programs.³⁰ The DOTA groups were parameterized as previously reported.31 The force field parameters for the other non-standard residues composing the dendrimers were assigned consistently with the "general AMBER force field (GAFF)" (gaff.dat).³² The partial charges for these residues were obtained using the AM1-BCC^{33,34} calculation method within the *antechamber*³⁵ module of AmberTools 1.4 (AMBER 11). The models for **G3-(Gd-DOTA)24** and **G5-(Gd-DOTA)96** were immerged in a periodic box containing TIP3P water molecules³⁶ extending 14 Å from the solute atoms. A suitable number of ions were added in solution using the standard *addIons* utility of the *leap* module (AMBER 11) to guarantee the systems neutrality and the ionic concentration of 150 mM [NaCl]. Water molecules were eventually replaced with ions when superposition occurred. The parm99 all-atom force field³⁷ was used to parametrize the standard residues in the system. The two solvated dendrimers were initially minimized. Two first steps of molecular dynamics simulations (MD) followed, lasting 50 ps in NVT and NPT conditions

respectively, to reach the simulation temperature of 300 K and to equilibrate the solution density inside the periodic box. The systems were then equilibrated for 100 nanoseconds (ns) in NPT periodic boundary condition at 300 K and 1 atm, using a time step of 2 femtoseconds, the Langevin thermostat and a 10 Å cutoff. During this time, both dendrimers reached the equilibrium with good stability. The SHAKE algorithm was used during the simulations on the bonds involving Hydrogen atoms^{38,39} and the particle mesh Ewald⁴⁰ (PME) approach was adopted to treat the long-range electrostatic effects. The equilibrated phase of the MD trajectories was further processed with the ptraj module of AMBER 11 to the extract structural data (i.e., the radius of gyration, R_g , and the radial distribution functions, RDF).

Paramagnetic Characteristics

Relaxivity was measured by a custom designed variable field T1-T2 analyzer (Southwest Research Institute, San Antonio, TX) and a 3T MR system (Signa LX, GE Healthcare, Waukesha, WI, USA). For the variable field analyzer, the field strength was varied from 0.035 to 1.5 T (1.5-63 MHz) and the sample temperature was set at 23 °C. Solutions of **G3- (Gd-DOTA)24, G5-(Gd-DOTA)96, G3-(Gd-DTPA)24, G5-(Gd-DTPA)96, Gd-DOTA** (Prohance), and **Gd-DTPA** (Magnevist, Berlex, Princeton, NJ) (1 mMGd, PBS, pH 7.4) were prepared, and T1 was measured using a saturation recovery pulse sequence with 32 incremental recovery times. T2 was measured using a Carr-Purcel-Meiboom-Gill pulse sequence of 500 echoes and an interecho time of 2 msec. To investigate the temperature dependency, additional measurements were performed at 0.8-T field strength.

Relaxivity was measured at 23 °C using a 3.0-T clinical scanner (Signa Excite, GE Healthcare, Waukesha, WI, USA) equipped with a rectangular single loop receiver coil (84 \times 126 \times 6 mm). Phantom solutions in 1x; PBS consisting of 0, 0.1, 0.25, 0.5, 0.75 and 1 mM in Gd(III) together with Gd-DTPA as reference controls were prepared. A series of singleslice 2D inversion recovery (IR) fast-spin echo images of all solutions were obtained at an echo time (TE) of about 7.54 ms using different inversion recovery times (TI) of 50, 100, 350, 750, 1250, 2500, and 5000 ms. The r1 values for each dilution were determined by fitting ROI intensity values from variable IR images using Igor Pro ([http://](http://www.wavemetrics.com) www.wavemetrics.com). r2 values were measured from ROI values from T2 maps, which were calculated from the multiecho images in $ImageJ$ ([http://rsb.info.nih.gov/ij\)](http://rsb.info.nih.gov/ij) using the MRI analysis plugin ([http://rsb.info.nih.gov/ij/plugins/mri-analysis.html\)](http://rsb.info.nih.gov/ij/plugins/mri-analysis.html). The molar relaxivities, r1 and r2, were obtained from the slope of 1/T1 or 1/T2 vs [Gd(III)] plots determined from region of interest measurements.

Contrast-Enhanced Dynamic 3D-Micro MRI of Mice

All dynamic micro-MRI images were obtained at 3-T with a 1-in. round receive-only modified Alderman-Grant resonators. Mice were chemically restrained with 2% isoflurane (Abbott Laboratories, NJ) in O2 delivered using a Summit Anesthesia Solutions vaporizer (Bend, OR) at a flow rate of 0.8 L/min. Respiration rate was maintained at 25-30 respirations per min and monitored using a Biopac System MP150 (Biopac Inc., Goleta, CA). The temperature of the mouse was maintained at 32 ± 1 °C using a Polyscience Model 210 heating recirculator with 3M Fluorinert Electronic Liquid FC-77 with body temperature monitored using FOTM fiber optic sensors (Fiso Technologies Inc., San Jose, CA) and a UMI-8 multichannel instrument (Fiso Technologies Inc). A tail vein cannula consisting of a 30-gauge needle attached to Tygon tubing (0.01 in i.d) was then established. Prior to injection of the contrast agent, a T1 map was obtained by using a 3D-fast spoiled gradient echo image (3D-fSPGR) sequence at two different flip angles (repetition time/echo time 10.888/3.86 ms; flip angles 8° and 24° ; bandwidth 31.25 kHz; matrix size $512\times128\times40$; voxel resolution $156\times156\times600$ µm; slice thickness 0.6 mm; two averages; scan time \sim 2

min). One hundred microliter (100 μL) of the **G3-(Gd-DOTA)24, G5-(Gd-DOTA)96, G3- (Gd-DTPA)24, G5-(Gd-DTPA)96**, or **PAMAM-G4-Gd-DTPA** was injected (0.03 mmolGd/kg) from the tail vein. The region of interest was placed on the liver, the cortex and medulla of the kidney, the muscle at the femoral region and the jugular vein, and the timeintensity curves were made. In addition, the slice data were processed into 3D images with the maximum intensity protection (MIP) method.

RESULTS AND DISCUSSION

Characterization of Compounds

Characterization of compounds rests on NMR spectroscopy and mass spectrometry. After conjugation reaction of the dendrimer platform with the dichlorotriazine monomer containing ligands, the resulting poly(monochlorotriazines), **6, 7, 9**, and **10**, were monitored by 1 H NMR and MALDI-TOF MS. While the peaks derived from 1 H NMR spectroscopy are somewhat broad in Figure 1, integration yields an estimate of the number of chelate groups. That is, signals from the chelate groups (a, b, and d) can be compared to those from the dendrimer platform (c). The integrated intensity of signal c was used as a standard for the comparison as the dendrimer platforms. We assume from the reasonable match, that the reaction to install chelate on the dendrimer proceeded exceedingly well given the small deviations between theoretical expectations and experimental observations. Analyzing extent of reaction is more difficult with MALDI-TOF MS (Figure 1), but the results are consistent. Table 1 represents chemical summary of the macromolecular ligands (**1-4**). All the compounds contain around 24 (for G3) or 96 (for G5) ligands, the maximum number that can be theoretically attached. After metalation with Gd^{3+} , these materials comprise approximately 13 wt% of gadolinium as determined by a titration assay: A known amount of dendrimer stock solution was titrated with a 10 mM of $Gd(OAc)_{3}$ (aq) at pH 5.5-6 until the solution tested positive for free gadolinium ions using xylenol orange solution. The solution was stirred at 50 °C for 2 days and then titrated with a 1 mM EDTA (aq) until no free Gd^{3+} was observed from xylenol orange test. By this method, the molar ratio of Gd^{3+} per dendrimer was determined, which was consistent with the theoretical number of ligands of each dendrimer (Table 1).

Computational Modeling

Computational models with explicit solvent and ions have provided a unique insight into the structure of triazine dendrimers. The simulations shown in Chart 2 are obtained from 100 ns of molecular dynamics (MD) simulations at 300 K and 1 atm of pressure. Both the equilibrated structures of DOTA chelates **G3-(Gd-DOTA)24** and **G5-(Gd-DOTA)96** are presented. From these computational models, we can calculate a number of interesting parameters. At the equilibrium, **G3-(Gd-DOTA)24** and **G5-(Gd-DOTA)96** showed a radius of gyration (R_g) of 22.6 +/- 0.2 Å and 35.1 +/- 0.3 Å, respectively.

These computations provide insights into interactions between these macromolecular chelates and water as well as the accessibility of the chelate to solvent as shown by radial distribution functions (RDFs). An RDF represents the density and the distribution of atoms in space with respect to the center of a given system of reference (the center of mass of the dendrimer). The RDFs shown in Figure 2 show the dendrimer platform (black continuous line), the DOTA chelates complexed with Gd ions (black dotted line), and water molecules (blue profile) with respect to the center. This plot gives indications about the presence of solute or solvent atoms in a certain zone of the system (spatial density). Since these curves are calculated at each step of the simulation, and they are reported in the plots as averaged over the equilibrated phases of the dynamic trajectories, they give information also on the dynamics of the system. These plots provide an indication on the time period in which a

certain atom is present in a certain area in the space (dynamic density). In RDF plots, high and narrow peaks in a small area of these graphs mean not only high density of atoms in a certain zone, but also high localization and low mobility of these atoms whether dendrimer, DOTA or water. Conversely, broad and low intensity peaks indicate low density and high mobility.

RDF plots show reduced surface DOTA backfolding most likely due to the large size of the DOTA indicators. From these data we infer that the metallated ligands are accessible to water, with more groups available on **G5-(Gd-DOTA)96** than **G3-(Gd-DOTA)24**. The larger dendrimer presents a higher surface density of Gd-DOTA groups than smaller dendrimer (the 2nd peak at 20 angstroms). Thus, the 96 Gd-DOTA groups of the generation 5 dendrimer may induce a higher surface density through structural crowding. This crowding is consistent with a decrease of water in that zone. In addition, some of these DOTA groups appear backfolded toward the dendrimer's core as indicated by peaks between c.a. 15 and 25 angstroms from the dendrimer's center. **G3-(Gd-DOTA)24** shows a more uniform structure and a more flexible surface. After the 1st high intensity peak, the decrease in RDF is continous and uniform.

Computation suggests that both dendrimers allow for consistent water penetration—even with waters found in close proximity to the cores. Whether these waters can exchange with bulk solvent is less certain, as peaks could also be generated by trapped molecules close to the dendrimer's center.

Relaxivity

Relaxivity profiles of triazine dendrimers conjugated with Gd-DOTA were different from those conjugated with DTPA-Gd or similar dendrimer agents that use PAMAM as a core instead of triazines. Both **G5-(Gd-DOTA)96** and **G3-(Gd-DOTA)24** showed higher r1 and r2 relaxivity compared with **G5-(Gd-DTPA)96** and **G3-(Gd-DTPA)24** especially at lower magnetic field than 30 MHz (Figure 3), where all dendrimer-based macromolecular MRI contrast agents showed the peak r1 relaxivity. The r1 relaxivity of all agents decreased at high magnetic field such as 128 MHz (3-Tesla) and 300 MHz (7-Tesla). Furthermore, triazine dendrimer-based agents, especially when conjugated with Gd-DOTA showed higher r1 relaxivity in serum than in PBS (Table 2), which can be advantageous *in vivo* imaging.

A striking finding of this study is the differences in relaxivity profiles between triazine dendrimers and PAMAM dendrimers based on the chelate used. DOTA conjugated to G3 and G5 triazine dendrimers showed much higher r1 and r2 relaxivity at low magnetic field (12 MHz or lower) and higher r1 relaxivity at high magnetic field (128 MHz) than comparable PAMAM dendrimers even in the PBS buffer. The peak r1 relaxivity values of both dendrimers of the same generation were similar at 30 MHz. This increased relaxivity may be caused by the chelate conformation, which allows either fast or slow water exchange. DOTA is known to form a variety of conformations, when conjugated to different moieties via a side chain.^{31,32} Therefore, benzyl-DOTA conjugated with triazine dendrimer can exchange water rapidly leading to high r1 relaxivity at off peak magnetic fields.³³

Another advantage of triazine G5 agents is that r1 increased dramatically in serum vs. PBS. Although the r1 of both triazine G5s and PAMAM-G4 in PBS were around 10 /mM/s, the r1 of triazine G5s in serum increased more than 2-fold up to above 20/mM/s at 128 MHz, a finding that did not occur with PAMAM-based MRI-CAs. In particular, triazine G5-DOTA demonstrated an r1 of 24 /mM/s, which is 85% higher than triazine G5-DTPA (13 /mM/s). This high relaxivity is likely related to loose serum protein binding which further slows the molecular tumbling rate.

In Vivo Dynamic Imaging Studies

In vivo pharmacokinetics using dynamic MRI was different for G3-based and G5-based agents. **G3-(Gd-DTPA)24** and **G3-(Gd-DOTA)24** quickly cleared from the kidneys. Dynamic MRI showed a quick decrease in blood signal and a temporally rapid increase of signal in the kidneys, followed by a gradual signal increase in the bladder (Figure 4a and 4b, Supplemental video 1 and 2, and Figure 5a and 5b). In contrast, **G5-(Gd-DTPA)96** and **G5- (Gd-DOTA)96** did not show clearance through the kidney, but demonstrated steady signal in the blood and the kidneys, and no or minimal signal increase in the bladder (Figure 4c and 4d, Supplemental video 3 and 4, and Figure 5a and 5b). In addition, **G5-(Gd-DOTA)⁹⁶** showed higher signal than **G5-(Gd-DTPA)⁹⁶** in vivo, which is consistent with its high serum relaxivity (Figure 4c and 4d) that has never seen, when employed PAMAM dendrimer as a platform (Supplemental figure 22). No significant difference of T1 signals in the liver was observed among the macromolecular chelates regardless of the generation and ligand (Figure 5c).

CONCLUSIONS

Gd-based macromolecular contrast agents were synthesized using a well-characterized G3 or G5 triazine platform. NMR spectroscopy and mass spectrometry confirmed that both G3 and G5 contrast agents conjugated with either DTPA or DOTA contained approximately 24 and 96 chelates respectively, which represents the theoretical limit which is also unusual for such conjugations with other dendrimers. Both G3 dendrimers, as expected, showed a shorter half life in blood and more rapid renal excretion than G5 dendrimers. Although triazine G5 agents are smaller than PAMAM G4 agents, they show reduced their renal excretion.²³ The r1 relaxivity of dendrimers chelated with DOTA was higher than that of dendrimers chelated with DTPA. **G5-(Gd-DOTA)96** had a markedly higher relaxivity in serum than in PBS. Thus, this agent had a 4-5 fold improvement in relaxivity compared to low molecular weight Gd-DOTA or Gd-DTPA. The pharmacokinetics of dendrimer chelated with either DTPA or DOTA was similar. Taken together, the DOTA version of the triazine dendrimer has advantages in relaxivity in comparison to DTPA chelates at widely accessible MRI frequencies/field strengths (64-128 MHz; 1.5T-3T) especially in vivo, and has the added advantage of higher association constants for gadolinium, making it more attractive for clinical translation. Such aspirations benefit from the ease of characterization associated with these materials.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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¹H NMR spectra and molecular weights of the dendrimers, **6, 7, 9**, and **10**, containing DOTA or DTPA ligands.

Figure 2.

Radial distribution functions of dendrimer (black solid line) and metallated chelates (black dotted line) and water molecules (blue line) of **G3-(Gd-DOTA)24** (left) and **G5-(Gd-DOTA)96** (right).

Figure 3.

r1 (a) and r2 (b) relaxivity profiles of triazine dendrimer-based conjugates compared with low molecular weight Gd-chelates (Gd-DTPA and Gd-DOTA) below 63 MHz magnetic field at 23°C.

Figure 4.

Early (4 min) and late (60 min) whole body MIP MR images of mice injected with 0.03 mmol/kg of **G3-(Gd-DOTA)24** (a), **G3-(Gd-DTPA)24** (b), **G5-(Gd-DOTA)96** (c), and **G5- (Gd-DTPA)96** (d) are shown. Entire dynamic MR images, which were taken every 2 min, are shown as Supplemental movies 1 to 4. (1: **G3-(Gd-DOTA)24**; 2: **G3-(Gd-DTPA)24**; 3: **G5-(Gd-DOTA)96**; 4: **G5-(Gd-DTPA)96**). Both G3 agents show enhanced kidneys (arrows) as well as blood pool in image 4 min after injection, however, show only the urine in the bladder 60 min after injection. In contrast, both G5 agents show blood pool up to 60 min after injection with minimal enhancement in the kidney and the bladder.

Figure 5.

Serial dynamic T1 signals of the blood pool (a), the kidney (b) and the liver (c) are plotted. T1 signals are consistent with the dynamic image shown in Figure 4 and Supplemental videos. Image 1 is Pre-contrast and image 2 is obtained after contrast agent administration. Images were obtained every 2 min after injection (images 2-32).

Scheme 1.

Synthesis of the Triazine Dendrimers (1—4) Containing DOTA or DTPA Ligands (G3 and G5 Dendrimer Platforms Depicted as Black Dot Spheres)

Chart 1.

Representation of the two triazine dendrimers (left) and the metallated chelates that appear on each peripheral piperazine.

Chart 2.

Equilibrated configurations of **G3-(Gd-DOTA)24** (a) and **G5-(Gd-DOTA)96** (b). The dendritic scaffold is colored in black, DOTA chelates are colored by atoms (C: grey, N: blue, O: red and H: white) and Gd atoms are colored in green. The center of the dendrimer is colored in pink. Explicit water molecules and Cl⁻ and Na^+ atoms are not shown for clarity.

Table 1

Chemical Summary of the Macromolecular Ligands $(1-4)$ and the Corresponding Gd³⁺ Complexes Chemical Summary of the Macromolecular Ligands $(1-4)$ and the Corresponding Gd³⁺ Complexes

determined by titration assay

Table 2

Relaxivities of Gd Contrast Agents at 128 MHz

