

# Protein–Hydrogel Interactions in Tissue Engineering: Mechanisms and Applications

Silviya P. Zustiak, PhD,<sup>1</sup> Yunqian Wei, MS,<sup>2</sup> and Jennie B. Leach, PhD<sup>2</sup>

Recent advances in our understanding of the sophistication of the cellular microenvironment and the dynamics of tissue remodeling during development, disease, and regeneration have increased our appreciation of the current challenges facing tissue engineering. As this appreciation advances, we are better equipped to approach problems in the biology and therapeutics of even more complex fields, such as stem cells and cancer. To aid in these studies, as well as the established areas of tissue engineering, including cardiovascular, musculoskeletal, and neural applications, biomaterials scientists have developed an extensive array of materials with specifically designed chemical, mechanical, and biological properties. Herein, we highlight an important topic within this area of biomaterials research, protein–hydrogel interactions. Due to inherent advantages of hydrated scaffolds for soft tissue engineering as well as specialized bioactivity of proteins and peptides, this field is well-posed to tackle major needs within emerging areas of tissue engineering. We provide an overview of the major modes of interactions between hydrogels and proteins (e.g., weak forces, covalent binding, affinity binding), examples of applications within growth factor delivery and three-dimensional scaffolds, and finally future directions within the area of hydrogel–protein interactions that will advance our ability to control the cell–biomaterial interface.

## Overview of Protein–Hydrogel Interaction Mechanisms

AS WE BEGIN to understand more about the *in vivo* cellular milieu and strive to recapitulate more of its features, protein–hydrogel interactions are gaining increasing importance in tissue engineering. Protein-conjugated hydrogels are now routinely used as a simplified mimic of the extracellular matrix (ECM), where such cell populated constructs represent either platforms for basic discovery or fully functional tissue replacements. Whereas the value of hydrogel-based scaffolds has long been established for soft tissue engineering applications, the incorporation of various proteins in the scaffold structure has more recently been appreciated as beneficial toward eliciting a desired cell response, degradability, or manipulating the mechanical and physical properties of the scaffold itself. However, major concerns when incorporating proteins include the preservation of bioactivity as well as proper presentation for maximum desired effect.

This review provides an overview of strategies used to create protein–hydrogel conjugates, with emphasis on covalent and affinity interactions, but also touching upon weak interactions, such as adsorption. We focus on two applications of protein–hydrogel interactions in tissue engineering: immobilization of growth factors and conjugates for use as three-dimensional (3D) scaffolds for cell growth. Due to the

nature of engineered tissues, which require a favorable cell environment, the emphasis will be on chemistries that are either nontoxic to cells or lead to nontoxic final products. In addition, because preservation of protein activity is essential, only water-based chemistries will be considered. The necessity to employ only water-based, nontoxic chemistries for the design of cell-compatible biomaterials, particularly in the context of 3D scaffold preparations, poses rather rigid constraints on materials processing. Here, the cell compatibility of the materials extends beyond chemical versatility of the hydrogels; it encompasses the design of physical and bioactive environments that are capable of eliciting specific interactions from biological systems. The authors recommend several comprehensive reviews<sup>1–3</sup> stressing the importance and associated limitations of designing bioactive and bio-compatible hydrogels.

### *Adsorption, electrostatic, and other weak interactions*

Weak interactions, such as electrostatic binding or repulsion, hydrogen bonding, and van der Waals forces typically underline physisorption and chemisorption of biomolecules onto hydrogels. Adsorption has been extensively explored as means of incorporating biomolecules into scaffolds for release in tissue engineering applications. For example, a solution of vascular endothelial growth factor

<sup>1</sup>National Institute of Child Health and Human Development, National Institutes of Health, Bethesda, Maryland.

<sup>2</sup>Department of Chemical, Biochemical and Environmental Engineering, University of Maryland, Baltimore County, Baltimore, Maryland.

(VEGF) loaded onto freeze-dried collagen gels allowed the growth factor to adsorb onto the hydrogel during hydration and swelling. Adsorption had also been used for loading bone morphogenic protein-2 (BMP-2) into various particulate vehicles<sup>4,5</sup> and basic fibroblast growth factor into hydrogel matrices.<sup>6</sup> In some cases, such as the adsorption of human serum albumin onto chitosan/alginate systems, electrostatic and hydrophobic forces, as well as surface topography, were found to affect protein adsorption synergistically.<sup>7</sup>

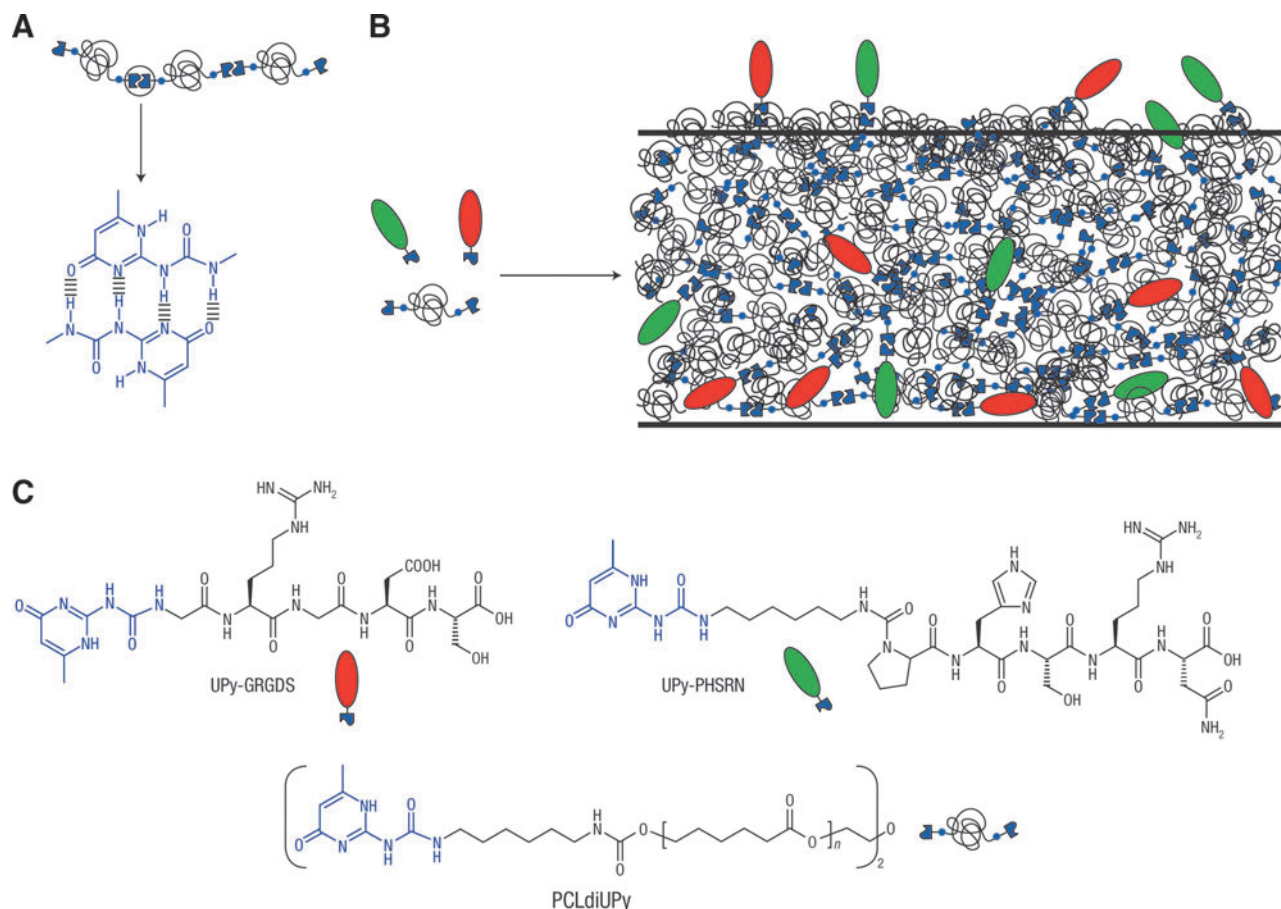
Hydrogen bonding is another weak interaction used in protein-polymer modifications, where plasma surface treatment can be used to introduce carboxylic functional groups to the polymer.<sup>8</sup> Due to its reversible nature, hydrogen bonding is amenable to the creation of degradable polymer scaffolds. For example, the self-complimentary hydrogen-bonding moieties (ureido-pyrimidinone or UPy groups) were utilized as the basis to assemble polymers and biomolecules (Fig. 1).<sup>9</sup> In gels with polar and charged functionality, hydrogen bonding, electrostatic interactions, and van der Waals forces can all influence protein-hydrogel interactions. For example, pH-sensitive insulin release was achieved with carboxylated chitosan-grafted nanoparticles.<sup>10</sup>

Though adsorption may be sufficient for *in vitro* applications, this approach is associated with challenges *in vivo* due to desorption and exchange with physiological fluid. Further, inherently adsorptive scaffolds may cause a long-term inflammation or fibrosis. Conversely, nonionic hydrophilic polymers, such as poly(ethylene glycol) (PEG), poly(vinyl alcohol), and polyacrylamide, are considered inert to protein adsorption and other methods for efficient immobilization are generally required.

*Affinity interactions*

Inspired by the mechanism by which bioactive molecules are stored and presented in native ECM, interactions based on affinity between two biomolecules pose the advantage of tailoring the biomolecule release rate and mechanism. In hydrogels, affinity molecules or peptides are used to incorporate functionalized biomolecules, such as growth factors, into the gel structure.

**Naturally derived affinity molecules.** Sulfated glycosaminoglycans, such as heparin sulfate, are one class of ECM molecules that bind growth factors *in vivo* via electrostatic



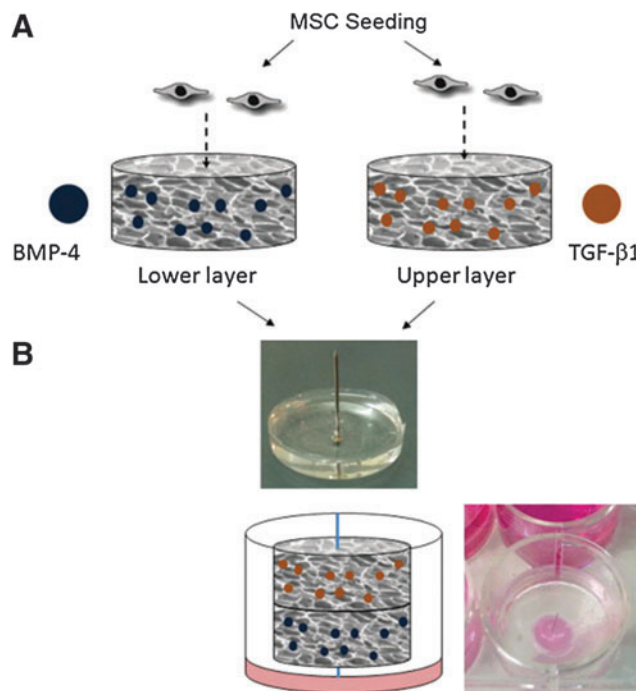
**FIG. 1.** (A) The self-complementary hydrogen-bonding ureido-pyrimidinone (UPy) moiety in a supramolecular polymer. (B) The modular approach to constructing bioactive materials with various properties by mixing different UPy-functionalized biomolecules (green and red moieties) with UPy-polymers. (C) As building blocks for these materials, two UPy-functionalized peptides (UPy-GRGDS and UPy-PHSRN) were used in combination with an oligocaprolactone functionalized on both ends with UPy-units (PCLdiUPy). Image reproduced from Dankers *et al.*<sup>9</sup> Color images available online at [www.liebertpub.com/teb](http://www.liebertpub.com/teb)

interactions. Due to their high content of functional groups available for modification with crosslinking sites, glycosaminoglycans have been utilized for several decades to modulate growth factor release in hydrogels. In 1991, Edelman *et al.* developed heparin-modified beads to achieve a controlled release of basic fibroblast growth factor by affinity binding.<sup>11</sup> Since this work, heparin has been extensively explored in biomaterials for affinity interactions with a number of growth factors,<sup>12,13</sup> including hepatocyte growth factor (HGF),<sup>14,15</sup> neurotrophin-3,<sup>16</sup> BMP-2,<sup>17</sup> and VEGF.<sup>18</sup> Other uses of heparin-based affinity include assembly of non-covalent 3D networks for enhancing cellular affinity<sup>19</sup> or inducing stem cell differentiation.<sup>20</sup> Another glycosaminoglycan, the negatively charged chondroitin sulfate, has demonstrated affinity for several positively charged growth factors, including nerve growth factor (NGF).<sup>21</sup>

Hyaluronic acid (HA) and alginate are also naturally derived biomolecules and have structural similarities to heparin. Sulfated HA-alginate biomaterials have exhibited affinity binding to VEGF, platelet-derived growth factor-BB, and transforming growth factor- $\beta$ 1 (TGF- $\beta$ 1).<sup>22</sup> Alginate hydrogels are simple injectable hydrogels, capable of controlled delivery of multiple proteins, while preserving the properties of the alginate.<sup>23</sup> For example, alginate hydrogels were used to achieve dual delivery of insulin-like growth factor-1 and HGF to promote myocardial repair in a rat model of acute myocardial infarction,<sup>24</sup> as well as for dual delivery of TGF- $\beta$ 1 and BMP-4 to induce articular cartilage regeneration in a rabbit model of subchondral defect (Fig. 2).<sup>25</sup>

Protein components of the ECM also have affinity to growth factors and have been explored in affinity binding biomaterials. For example, collagens I, III, V, and VI have affinity to HGF<sup>26</sup> and fibronectin and vitronectin domains can bind several growth factors, including HGF.<sup>27</sup>

**Affinity peptides and other synthetic affinity molecules.** A variety of affinity peptides have been identified which are derived from ligand–receptor binding motifs and ECM structural molecules. In comparison to large naturally derived biomolecules, short peptide sequences are generally easier to synthesize and incorporate into hydrogels, and offer more control over material properties. For example, Maynard and Hubbell identified a sulfated heparin-mimetic tetrapeptide that binds to VEGF<sup>28</sup>; Anseth and coworkers, identified affinity peptides that bind to TGF- $\beta$ 1,<sup>29</sup> and Huddalla and Murphy used a peptide derived from the heparin-binding domain of FGF-2 to sequester heparin proteoglycans.<sup>30,31</sup> The general approach to incorporate peptides into hydrogels is to utilize their various functional groups (i.e., amines, carboxyls, alcohols, thiols) to bind to the hydrogel structure (see also section Covalent interactions). For example, peptides containing Cys can react via Michael-type addition of thiol to vinyl sulfone or acrylate groups present in hydrogel polymers.<sup>32</sup> Due to their versatility and relatively simple structure, the kinetics of a variety of peptide-based affinity interactions have been explored. Notably, Lin and Anseth determined the relationship between the peptide structure and affinity binding to the basic fibroblast growth factor within PEG hydrogels.<sup>33</sup> Sakiyama-Elbert and coworkers identified peptides with affinity to NGF and also developed a mathematical model to describe the kinetics of the affinity binding reaction.<sup>34</sup> Vulic and Shoichet uncovered



**FIG. 2.** A scheme describing the construction of a bilayer affinity-binding alginate scaffold. Transforming growth factor- $\beta$  (TGF- $\beta$ 1) and bone morphogenic protein-4 (BMP-4) were affinity-bound to alginate sulfate in macroporous alginate scaffolds. After cell seeding and a culture of 2 days (A), the seeded scaffolds were combined together by their assembly on a stainless steel pin placed perpendicular to a supporting poly(dimethyl siloxane) layer (B). Image reproduced from Re'em *et al.*<sup>25</sup> Color images available online at [www.liebertpub.com/teb](http://www.liebertpub.com/teb)

the specific binding of the Src homology 3 domain expressed in growth factors to proline-rich peptides,<sup>35</sup> and Wang *et al.* developed a new collagen mimetic peptide that binds to collagen type I via ionic interactions and shows specific affinity to VEGF.<sup>36</sup>

A variety of synthetic affinity molecules have been developed for affinity chromatography and are readily applied to biomaterial systems as well. For example, metal-ion chelation chemistry has been utilized to functionalize hydrogels with affinity motifs. Lin and Metters modified gels with iminodiacetic acid groups that chelate with hexa-histidine tagged protein.<sup>37</sup> This approach may provide a specific advantage for controlled release as the growth factor can be dissociated and released on demand by the addition of metal ions. Further, multiple ligand–metal pairs may be implemented to provide cascades of multiple growth factors.<sup>38</sup>

#### Covalent interactions

Covalent tethering of peptides and proteins onto hydrogels is desirable in a variety of tissue engineering and controlled release applications and has been utilized with natural as well as synthetic hydrogels.<sup>39,40</sup> Covalent attachment of biomolecules is generally more stable, provides greater control over biomolecule presentation and release, and is able to generate superior hydrogel mechanical properties as compared to physical entrapment, adsorption, or affinity interactions.<sup>41,42</sup>

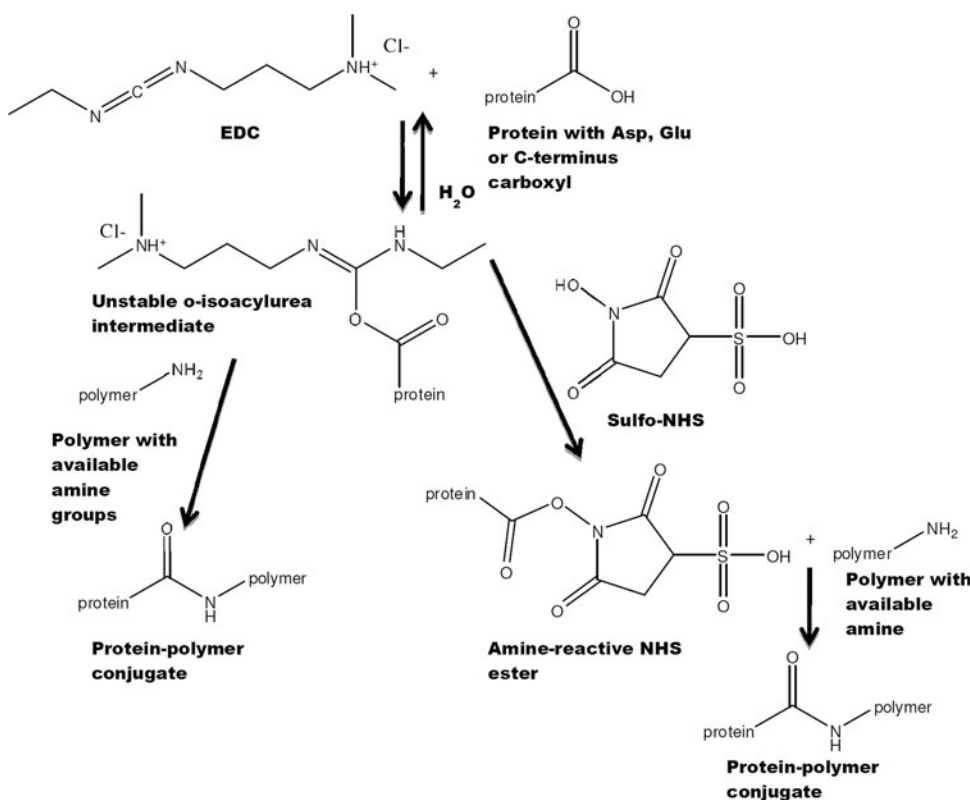
For the covalent functionalization of hydrogels with proteins, one must consider reaction conditions that sustain the protein structure and activity. Protein immobilization reactions generally require aqueous conditions, since organic solvents may cause denaturation and degradation. Many of the widely used protein chemistries are nonspecific in that they target amines (N-terminus, Lys), carboxyls (C-terminus, Asp, Glu), and hydroxyls (Ser, Thr, Tyr). The presence of multiple available residues for modification makes these covalent interactions attractive for a wide range of applications, such as fluorescent labeling. However, in other applications, such as tethering of growth factors or adhesive motifs, random modifications can affect the proper biomolecule presentation by either altering protein conformation or by sterically blocking the protein-active site. Thus, random modification can lead to heterogeneous and poorly reproducible substrates. We highlight several immobilization strategies that have shown particular advantage for reaction specificity, patterning, or preserving bioactivity and biocompatibility. For an excellent resource on these and other covalent tethering chemistries, we recommend Hermanson's "Bioconjugate Techniques".<sup>43</sup>

**Carbodiimide and succinimide chemistries.** The carboxyl-activating agent 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide (EDC) is a water soluble carbodiimide widely used for coupling to primary amines, yielding stable amide bonds (Fig. 3). Carbodiimide chemistry has been applied to modify amine-terminated polycaprolactone-PEG nanofibers with epidermal growth factor,<sup>44</sup> to immobilize VEGF onto collagen scaffolds,<sup>45</sup> and to immobilize NGF onto gelatin-tricalcium phosphate membranes.<sup>46</sup> EDC reacts with a carboxyl to form

an amine-reactive O-acylisourea intermediate. Since this intermediate rapidly hydrolyzes in an aqueous environment, EDC is often used in combination with N-hydroxysuccinimide (NHS) or sulfo-NHS to increase coupling efficiency by creating a more stable intermediate.<sup>43</sup> Reaction conditions, such as pH, are key determinants of reaction outcome, with optimal pH from 4.7 to 6.0, even though reaction can occur at mildly alkaline conditions, but with a slower rate and lower yields. Potential side reactions to note when considering EDC/NHS chemistry with proteins include the reaction of EDC with Cys sulfhydryls and Tyr phenols as well as the reaction of sulfo-NHS with the imidazolyl group in His.<sup>43</sup>

One challenge to EDC chemistry stems from the fact that the carbodiimide, due to its nonspecific nature, may cause inter- or intraprotein reactions and alter biomolecule activity or cause aggregation. To address this drawback, step immobilization can be used to first activate polymer carboxyls, and then react with protein amines; this approach was effective for modifying collagen gels to present uniform or gradient patterns of immobilized VEGF.<sup>47,48</sup>

**Photo-immobilization.** In many cases, it is beneficial to use photoreactive heterobifunctional crosslinkers, where one terminus is tailored to react with the desired functional groups (usually amines on proteins) and the other terminus is photoreactive (e.g., phenyl azide, benzophenone or diazo compound). These linkers react nonspecifically with ultraviolet (UV) light by creating singlet nitrenes that undergo insertion into C-H, N-H, and other bonds. Such crosslinkers are commercially available and their chemistry is simple, effective, and inexpensive. N-sulfosuccinimidyl-6-(4'-azido-



**FIG. 3.** Schematic representation of 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide (EDC)/sulfo-N-hydroxysuccinimide (NHS) chemistry. In this example, a stable amide bond is formed between the carboxylate group of a protein and the amine groups of a polymer, but due to the modular nature of the reaction, protein amines can also be reacted with polymer carboxylates. Sulfo-NHS is added to improve reaction yield by forming a semistable reaction intermediate.

2'-nitrophenylamino) hexanoate (sulfo-SANPAH, Table 1), which contains a photoactivatable phenyl azide and an amine-reactive NHS is one example of this class of crosslinkers. Sulfo-SANPAH has been widely used to conjugate biomolecules onto polyacrylamide gels for studying cellular response to substrate stiffness.<sup>49–51</sup>

Alternatively, photochemistry can be combined with EDC chemistry, as in the case with the azidophenyl-derivatized polyallylamine method developed by Matsuda and Sugawara<sup>52</sup> and modified by Ito *et al.*,<sup>53,54</sup> where the protein is rendered photoreactive by modifying carboxyls with polyallylamine using EDC chemistry. This chemistry has been used for the immobilization of growth factors onto poly(dimethyl siloxane)<sup>55</sup> and glass.<sup>56</sup> Another type of photoreactive heterobifunctional crosslinkers contains an NHS or a succinimidyl carbonate group for conjugation to protein amines and a photoreactive acrylate or methacrylate group. Acrylates undergo free radical polymerization in the presence of a photoinitiator and UV irradiation. This chemistry is widely applied for growth factor immobilization, where the crosslinkers may contain a spacer arm to improve growth factor accessibility to the cells.<sup>57–60</sup> Similar conjugations have been used to create protein-polymer and peptide-polymer conjugates to yield photopolymerizable biomimetic hydrogels.<sup>61–63</sup>

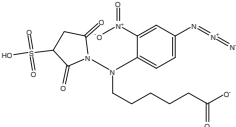

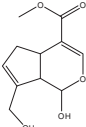
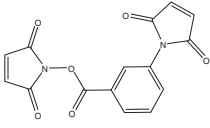
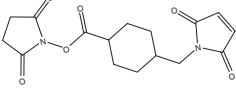
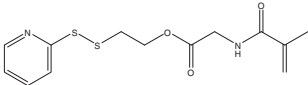
Photoimmobilization of proteins is an attractive strategy with many advantages, such as facile spatial and temporal control over reaction kinetics as well as rapid reaction conditions. However, to avoid the concern of generating active

radical species during UV irradiation, which may affect protein bioactivity and cell viability, reaction conditions, such as UV exposure time and photoinitiator type and concentration must be optimized.<sup>64</sup>

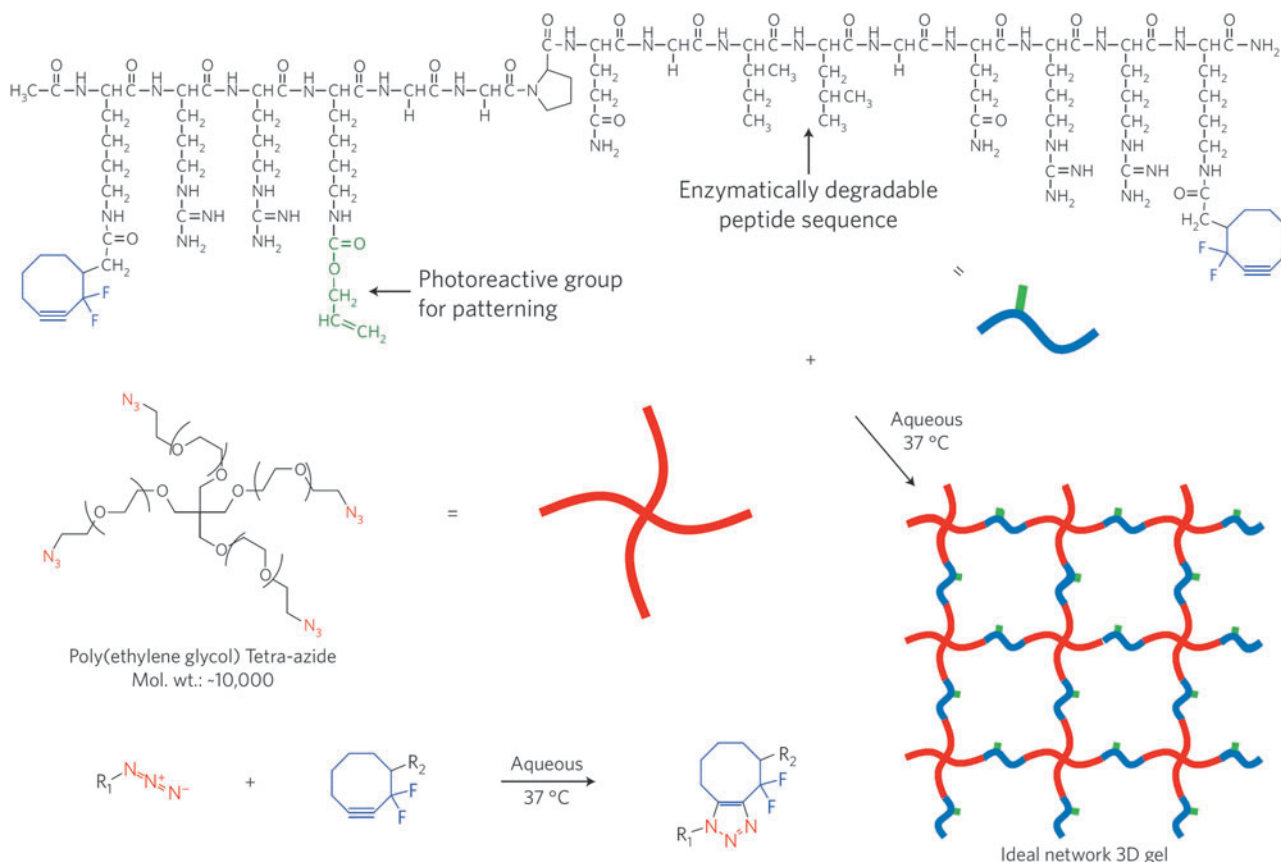
**Click chemistry.** Click chemistry is a general concept that describes reactions that are high yielding, stereospecific, highly versatile, simple to perform and purify, and occur in benign solvents.<sup>65</sup> Originally, click chemistry was introduced to accelerate drug discovery and testing, but it is increasingly becoming utilized in tissue engineering. To address a major drawback of conventional click chemistries, namely, the use of cytotoxic copper or ruthenium catalysts, copper-free click chemistry has been developed.<sup>66,67</sup> For example, Nimmo *et al.* utilized click chemistry to design a one-step, aqueous-based method to crosslink HA hydrogels for use in tissue engineering.<sup>68</sup> Anseth and coworkers adapted a click chemistry to perform spatial and temporal patterning of polypeptides onto PEG hydrogels to pattern 3D cell microenvironments (Fig. 4).<sup>69,70</sup>

**Other chemistries.** There are a large number of other chemistries used to create protein-hydrogel conjugates of which we will only attempt to describe a few. One such reaction is the Michael-type addition pioneered for use in tissue engineering by Elbert and Hubbell<sup>71</sup> and Lutolf and Hubbell.<sup>72</sup> In this reaction, unsaturated double bonds of chemical moieties, such as vinyl sulfone, acrylate, or methacrylate react under basic conditions with a thiol typically

TABLE 1. CROSSLINKERS USED FOR PROTEIN-HYDROGEL CONJUGATION

Crosslinker name	Crosslinker structure	Comments
Sulfo-SANPAH <sup>123,124</sup>		Water-soluble, amine-reactive and light activatable crosslinker based on aryl azide chemistry; noncleavable
Glutaraldehyde <sup>125–128</sup>		Amine-to-amine crosslinker; highly toxic to cells; water-soluble
Genipin <sup>129–132</sup>		Natural nontoxic amine-to-amine crosslinker; derived from gardenia fruit; stable; water-soluble; improves mechanical properties of naturally occurring hydrogels, such as collagen Type I
MBS <sup>133</sup>		Nontoxic amine-to-sulphydryl crosslinker; reactive at physiological conditions; not soluble in water
SMCC <sup>87,134</sup>		Nontoxic amine-to-sulphydryl crosslinker; stable maleimide groups; reactive at physiological conditions; rapid reaction kinetics; not soluble in water
2-(2-pyridin-2-yl)disulfanyl)ethyl 2-(methacrylamido)acetate (PDMA) <sup>81,82</sup>		Modify protein amines to bear methacrylamide groups; link is reducible due to presence of disulfide bond

SANPAH, N-sulfosuccinimidyl-6-(4'-azido-2'-nitrophenylamino) hexanoate; MBS, m-maleimidobenzoyl-N-hydroxysuccinimide ester; SMCC, succinimidyl-4-(N-maleimidomethyl)cyclohexane-1-carboxylate.



**FIG. 4.** Click-functionalized macromolecular precursors undergo the Cu-free [3+2] Huisgen cycloaddition to form a 3D ideal network hydrogel through a step-growth polymerization mechanism. Here, two orthogonal click chemistries are used: one for hydrogel formation and another for biochemical patterning within the preformed material. The modular aspect of these reactions allows for independent control of the network structure and chemistry, and facile incorporation of biological molecules. Image reproduced from DeForest *et al.*<sup>70</sup> Color images available online at [www.liebertpub.com/teb](http://www.liebertpub.com/teb)

provided by Cys. For peptides, the reactivity of the thiol can be modulated by the presence of charged amino acids in the direct vicinity of the Cys.<sup>73</sup> This chemistry is biocompatible and selective, and therefore ideal for protein conjugation and has been successfully used for conjugation of polymers to Cys-terminated peptides,<sup>74,75</sup> genetically engineered growth factors with an added Cys,<sup>76,77</sup> and other proteins.<sup>78,79</sup>

Homo- and hetero-bifunctional crosslinkers, despite their inherent nonspecific reactivity, are very useful for the crosslinking of polymers and proteins. Thus, in addition to linkers that have been applied for decades, such as glutaraldehyde, new crosslinkers (Table 1) are being developed with further improvements over existing technologies. Main advantages of the novel crosslinkers include lack of toxicity to cells, as well as improved versatility by incorporation of specific moieties, such as acrylates or disulfides, that permit controlled protein release, gelation strategies and kinetics, or gel degradation. For example, genipin, a naturally occurring crosslinker, has toxicity levels 10,000-fold lower than glutaraldehyde.<sup>80</sup> Further, the acrylate functionality, which is responsible for photopolymerization of PEG-diacrylate gels, is utilized for several protein-hydrogel crosslinking reactions, including cleavable protein hydrogel crosslinkers (2-(2-pyridin-2-yl)disulfanyl)ethyl 2-(methacrylamido)acetate in Table 1)<sup>81,82</sup> and degradable hydrogels, where redox poly-

merization strategies allows for *in-situ* cell encapsulation.<sup>83</sup> HA has been similarly modified with methacrylate to obtain gels that encompass the range of physiologically relevant mechanical properties, while maintaining HA cell receptors.<sup>84,85</sup>

### Tissue Engineering Applications of Protein-Hydrogel Interactions

Protein-hydrogel interactions are desirable in a variety of tissue engineering applications, including sustained, localized, or targeted protein delivery, induction of specific cell signaling, and control of stem cell fate. We focus here on the two most common applications: immobilization of growth factors and polymer-protein conjugates for use as 3D scaffolds for cell growth, where the protein presents cell binding domains or degradation sites.

#### Immobilization of growth factors

Growth factors, which are involved in cellular proliferation, differentiation, and tissue regeneration, can be supplied exogenously via solution injection or by immobilization onto scaffolds.<sup>40</sup> The latter method is generally more efficient in promoting the desired cell outcome, while preserving bioactivity and stability, which in turn prolongs growth factor

signaling and minimizes costs of growth factor therapeutics. Superior growth factor performance upon immobilization has been demonstrated by Chiu and Radisic, where VEGF tethered onto collagen scaffolds was associated with a higher endothelial cell proliferation versus treatment with VEGF in a solution.<sup>86</sup> In addition, the immobilization method itself is important for growth factor performance, as demonstrated by Park *et al.*, who determined that BMP-2 immobilized covalently onto chitosan nanofibers had higher bioactivity, promoted greater cell proliferation, alkaline phosphatase activity, and calcium deposition when compared with nanofibers with absorbed BMP-2.<sup>87</sup>

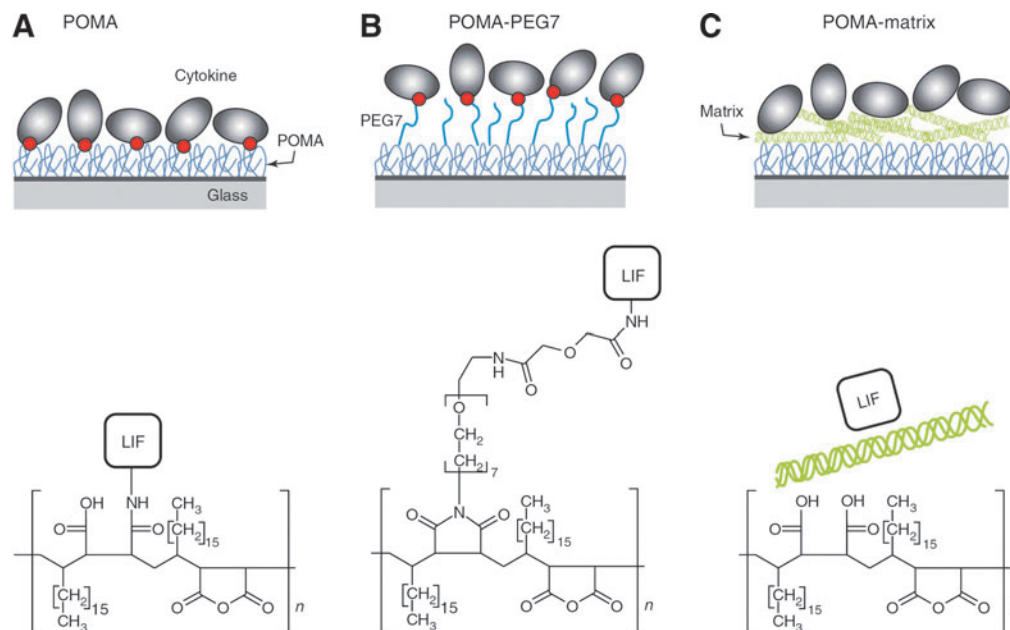
The concentration of the growth factor influences cell response, and when varying growth factor concentration is presented as a gradient, it has the potential to guide directional cell responses. For example, Davis *et al.* demonstrated that increased concentration of BMP-2 adsorbed onto polymer scaffolds enhanced the osteogenic response of human mesenchymal stem cells,<sup>4</sup> while Odedra *et al.* found that VEGF gradients on collagen scaffolds promoted directional endothelial cell migration.<sup>48</sup> Further, Moore *et al.* revealed that neurotrophin gradients on synthetic gels promoted guided neurite outgrowth from chick dorsal root ganglia neurons.<sup>88</sup>

Growth factor tethering is especially favored for *in-vivo* scaffold applications, where it offers the additional advantage of localizing the effect of the growth factor within the scaffold itself rather than allow it to freely diffuse into the surrounding tissue. Further, immobilized growth factors have a physiological significance, as both bound and free growth factors are present *in vivo*. It is thought that cells first

respond to the soluble growth factor, migrate toward its source, and then respond to the higher concentration of bound protein at the source. For example, Miyagi *et al.* covalently immobilized VEGF onto collagen scaffolds to promote neovascularization *in vivo*.<sup>89</sup> The authors showed that scaffolds containing VEGF had greater blood vessel density than the control scaffolds with no VEGF, due to increased endothelial cell infiltration and proliferation.<sup>89</sup> Further, Choi *et al.* demonstrated that the epithelial growth factor covalently immobilized onto polycaprolactone-PEG scaffolds promoted keratinocyte differentiation and resulted in superior wound healing *in vivo* than control groups with the soluble growth factor.<sup>44</sup> Alberti and coworkers compared various methods of immobilizing the leukemia inhibitory factor onto hydrogels, including covalent attachment, covalent attachment via a flexible PEG spacer arm, and non-covalent binding (Fig. 5). They determined that the spacer arm increased growth factor accessibility and resulted in greater mouse embryonic stem cell pluripotency.<sup>90</sup>

#### Protein–hydrogel copolymers for use as 3D cell scaffolds

In tissue engineering applications, cell scaffolds ideally should mimic the natural structure of the tissue they are replacing, while providing a structural support for the cells. Polymer–protein conjugates are especially promising for use as tissue engineering scaffolds for a variety of reasons, including the addition of specific biological activity to already biocompatible substrates.



**FIG. 5.** (A–C) A cytokine, leukemia inhibitory factor (LIF), was immobilized by covalent attachment to a hydrogel composed of poly(octadecene-alt-maleic anhydride) (POMA) (A), covalent attachment to flexible poly(ethylene glycol) (PEG) spacer arms tethered to POMA (B) and noncovalent binding to extracellular matrix coating deposited on top of hydrolyzed POMA (C). Red circles indicate covalent bond. Note that POMA was covalently bound to amino-functionalized glass substrates and is the key component of the immobilization platform. POMA and POMA-PEG7 surfaces were coated with matrix for cell culture experiments (omitted from the figure for clarity). Only one covalent bond between the protein and the surface is shown, although more bonds are possible. Chemical structures depict the POMA layer and the immobilization mode of LIF. Image reproduced from Alberti *et al.*<sup>90</sup> Color images available online at [www.liebertpub.com/teb](http://www.liebertpub.com/teb)

Conjugates of proteins and synthetic hydrogels. Synthetic biomaterials are widely used in tissue engineering because they offer precise control over a wide range of scaffold mechanical properties.<sup>85,91,92</sup> Protein modification is especially advantageous for synthetic materials because proteins may provide intrinsic biological activity, such as cell adhesion or proteolytically degradable sites to typically inert polymers.

PEG, arguably the most widely used synthetic polymer in soft tissue engineering, has been conjugated to a variety of proteins to form biosynthetic hydrogel scaffolds. For example, Almany and Seliktar developed PEG-fibrinogen scaffolds as 3D environments for tissue regeneration therapies.<sup>78</sup> The fibrinogen provided cell attachment motifs as well as proteolytic sensitivity for degradation, while PEG was responsible for the superior hydrogel elastic modulus.<sup>78</sup> Comparing the properties of three different PEG-protein copolymers, Gonen-Wadmany *et al.* reported cell spreading and migration on PEG-fibrinogen and PEG-collagen gels, and not on PEG-bovine serum albumin gels due to lack of cell adhesion motifs on the bovine serum albumin backbone.<sup>93</sup> Interestingly, the composite hydrogels also showed different elasticity, swelling, and degradability based on the incorporated protein, underscoring the importance of the molecular relationship between the polymer and the protein.<sup>93</sup>

Composites of collagen and polyanhydrides, such as polylactic acid, polyglycolic acid, and polycaprolactone, have been synthesized as nanofibrous matrices,<sup>94-96</sup> grafted copolymers,<sup>97</sup> and composite gels<sup>98-100</sup> and are widely used in bone and cartilage tissue engineering, because of their similarity to natural bone or cartilage in main composition and hierarchical nanostructure. For example, these composite gels have been used in the form of a polylactic-glycolic acid sponge prepared with incorporated collagen microsponges in the pores, where the collagen facilitated homogeneous cell seeding and subsequent cell differentiation.<sup>101</sup> Electrospun collagen/polycaprolactone materials have also been used as nerve guide conduits *in vivo* where they were able to support the nerve regeneration through an 8-mm sciatic nerve gap in adult rats, thus approaching the effectiveness of the nerve autograft.<sup>94</sup> Collagen-poly(lactic acid-co-caprolactone) hybrid scaffolds have been used for bladder tissue regeneration, where the hybrid hydrogels exhibited a lower inflammatory reaction and degradation times than scaffolds without collagen.<sup>102</sup>

Peptides are commonly incorporated into hydrogels, including motifs derived from the ECM, such as the cell adhesive peptide RGD,<sup>83,103</sup> as well as peptides susceptible to matrix metalloproteinase cleavage that allow proteolytic gel degradation.<sup>32,104</sup> Methods to incorporate peptide ligands onto biomaterials have been reviewed extensively and, thus, can be found elsewhere.<sup>39,105</sup>

Conjugates of proteins and natural hydrogels. Because of their intrinsic biocompatibility, natural biomaterials are also very attractive for use in tissue engineering despite their relatively low mechanical strength. Usually, protein-hydrogel matrices are meant to either elicit a specific biological response from the cells (e.g., growth factor immobilization) or to improve the composite hydrogel mechanical, biochemical, or physical properties over the individual materials alone.<sup>106-108</sup> The most common natural materials used in

tissue engineering include HA, chitosan, collagen, fibrinogen, fibrin, and agarose.<sup>109</sup>

A variety of composites based on natural materials have been developed to address a major challenge in the construction of 3D scaffolds for tissue engineering; namely, the development of highly porous scaffolds with interconnected pores.<sup>110</sup> Various collagen/HA/chitosan scaffolds with highly porous interconnected structures were manufactured by varying ratios between the constituents to find the optimal conditions for tissue engineering applications.<sup>111</sup> Chitosan-HA-based hydrogels were created as injectable, biodegradable, and glucose-responsive hydrogels to deliver insulin *in vitro*.<sup>112</sup> Highly porous collagen/HA composite hydrogels have also been developed as a model of the mammary gland,<sup>113</sup> as a matrix for intervertebral disk repair,<sup>114,115</sup> and as a scaffold to study glioma cell invasion.<sup>116</sup> Further, collagen/fibrin composite matrices were used as vascular constructs because of their superior mechanical strength in comparison to each of the components alone.<sup>117</sup>

### Future Directions

While there is an array of well-established chemistries for physical, covalent, or affinity conjugation of proteins, one of the major areas in need of further development, especially in terms of growth factor conjugation, is the presentation of biomolecules. The majority of available conjugation chemistries are nonspecific, thus compromising molecule bioactivity by binding site unavailability due to orientation or steric hindrance. The nonspecific chemistries also raise issues of robustness, heterogeneities, as well as unwanted side interactions, such as competitive binding of media or serum proteins. These limitations are beginning to be addressed by adding tags to the biomolecule to ensure binding specificity or by adding spacer arms between the hydrogel and the protein to avoid steric hindrance. Examples of tags and linkers include Cys,<sup>118</sup> Fc,<sup>119</sup> polyhistidine,<sup>120,121</sup> as well as *de novo*-designed coil-tagged chimeras.<sup>122</sup> Overall, the development of more protein-hydrogel interactions, as well as the accumulation of knowledge about their effect on cell fate, will enable the design of engineered tissues that better mimic the *in-vivo* environment, as well as tissues that can be safely translated into the clinic.

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### Disclosure Statement

No competing financial interests exist.

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Address correspondence to:

Jennie B. Leach, PhD  
 Department of Chemical, Biochemical  
 and Environmental Engineering  
 University of Maryland, Baltimore County  
 Eng 314, 1000 Hilltop Circle  
 Baltimore, MD 21250

E-mail: jleach@umbc.edu

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