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## Growth Inhibitory Activity of a Bis-benzimidazole-Bridged Arene Ruthenium Metalla-Rectangle and Prism

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### Abstract

Two new supramolecular coordination complexes (SCCs), were obtained from the self-assembly of a new bis-benzimidazole bridged Ru acceptor, **4**, with dipyridyl and tripyridyl donors, respectively. As part of a growing library of anticancer-active Ru-based SCCs, metalla-prism **6** selectively showed high cytotoxicities relative to cisplatin for a series of cancer cell lines, with IC<sub>50</sub> values as low as 8.41 μM for MCF7 cells, as determined from MTS assays.

Mainly, molecular clips are a specific class of dinuclear building blocks which could be comprised of two metal centers bridged by organic spacers, wherein two substitutionally labile ligands are oriented to give parallel coordination vectors. When such species are used in coordination-driven self-assembly reactions they represent a 0° ditopic building block, where the angle indicates the relative orientation of the two labile sites. As such, molecular clips are suitable for constructing a variety of 2D and 3D SCCs that contain a 90°-spacer-90° motif, such as squares, rectangles and prisms.<sup>1</sup> These designs have historically favoured Pt and Re-based metal acceptors, owing to their predictable coordination geometries and well-established chemistry.<sup>2</sup> However, the desire for functional SCCs has motivated the use of alternative acceptors based on Ir, Rh and Ru, due to the novel physiochemical properties that such metals can impart to their resulting structures.<sup>3</sup> Recently, a variety of arene-Ru molecular clips bridged by O,O-chelating ligands have been reported for the self-assembly of metalla-rectangles and prisms.<sup>4</sup> In this context, ruthenium is not simply a structural element; the metal fulfils a functional role as the impetus for antitumor activity.<sup>5</sup>

One of the strengths of coordination-driven self-assembly is the modularity and tunability of the building blocks, which can often be modified without significant synthetic redesign. Along these lines, we explore here Ru complexes wherein the O,O-bridging moieties are replaced by an N,N-chelating species, prompted by observations of cisplatin-levels of cytotoxicity in analogous small molecule Ru complexes.<sup>6</sup> Specifically, we demonstrate the use of a bis-benzimidazole bridging ligand in the formation of Ru-based SCCs with inherent biological activities.<sup>7</sup>

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**Supporting Information Available:** Synthetic details, spectral data and crystallographic data for the molecular clip **4**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The *p*-cymene complex  $[(p\text{-cymene})\text{RuCl}_2]_2$  (**1**) reacts with bis-benzimidazole (**2**) and sodium acetate in 1:1:2 molar ratio to furnish the dimeric species **3**, which subsequently converts into molecular clip **4** upon treatment with silver triflate in methanol (Scheme 1). The pure product, fully characterized by  $^1\text{H}$  and  $^{13}\text{C}$  NMR and HR-ESI-MS spectrometry, is isolated as a yellowish brown solid upon addition of diethyl ether. The  $^1\text{H}$  NMR spectrum exhibited two multiplets at  $\delta = 8.00$  and  $7.50$  ppm, corresponding to bis-benzimidazole protons. The *p*-cymene protons resulted in two doublets at  $\delta = 6.43$  and  $6.30$  ppm. HR-ESI-MS analysis of molecular clip **4** showed a peak at 852.0 for  $[\mathbf{4} - \text{O}_3\text{SCF}_3^-]^+$  with an isotopic distribution consistent with its theoretical pattern. Single crystals of **4** suitable for X-ray structural studies were grown by vapour diffusion of diethyl ether into a methanol solution, confirming the molecular clip nature of the compound with two labile sites, occupied by MeOH (Figure 1). Neither NMR nor ESI-MS experiments indicated MeOH coordination, suggesting that these sites are readily exchanged and labile in solution, a requirement for efficient SCC formation. A similar reaction of  $[(p\text{-cymene})\text{RuCl}_2]_2$  with bis-benzimidazole was attempted by Carmona et al.<sup>7g</sup> afforded the mixture of mononuclear complex  $[(p\text{-cymene})\text{Ru}(\text{H}_2\text{Bbzim})\text{Cl}]\text{Cl}$  along with dinuclear derivative  $[\{(p\text{-cymene})\text{RuCl}\}]_2\mu\text{-Bbzim}$  and failed to obtain a discrete product as described in this report.

Treatment of molecular clip **4** with N, N-di(pyridine-4yl)oxalamide (**L1**) in 1:1 molar ratio afforded a new metalla-rectangle **5**. A similar treatment with 1, 3, 5-tris (4-pyridylethynyl) benzene (**L2**) in 3:2 molar ratio resulted in a self-assembled metalla-prism **6**, both with quantitative yields. The  $^1\text{H}$  NMR spectra of **5** and **6** show two doublets ( $\delta = 7.68$  and  $7.14$  ppm for **5**;  $\delta = 7.85$  and  $6.93$  ppm for **6**) for pyridyl protons and two multiplets ( $\delta = 8.03$  and  $7.58$  ppm for **5**;  $\delta = 8.08$  and  $7.64$  ppm for **6**) for the bis-benzimidazole protons. Additionally, two singlets at  $\delta = 9.57$  ppm (amidic NH proton for donor **L1**) and  $\delta = 7.29$  ppm (benzyl protons of donor **L2**) were observed for **5** and **6**, respectively. The *p*-cymene protons were observed as two doublets ( $\delta = 6.59$  and  $6.11$  ppm for **5**;  $\delta = 6.64$  and  $6.17$  ppm for **6**), significantly shifted from those in the spectrum of molecular clip **4** (Figure 2).

The formations of rectangle **5** and prism **6** were further supported by HR-ESI-MS analysis. Two charge states were observed at  $m/z = 1094.1$   $[\mathbf{5} - 2\text{O}_3\text{SCF}_3^-]^{2+}$  and  $679.8$   $[\mathbf{5} - 3\text{O}_3\text{SCF}_3^-]^{3+}$  for **5** and  $m/z = 1106.2$   $[\mathbf{6} - 3\text{O}_3\text{SCF}_3^-]^{3+}$  and  $604.1$   $[\mathbf{6} - 5\text{O}_3\text{SCF}_3^-]^{5+}$  for **6** (See SI). These peaks were isotopically resolved and showed good agreement with their theoretical distributions. A theoretical structure indicative the 2D rectangular nature of **5** is shown in Figure 1, as determined from a density functional theory (DFT) geometry optimization with the  $^i\text{Pr}$  and Me groups of the *p*-cymene omitted.

The absorption and emission spectra of molecular clip **4**, rectangle **5**, and prism **6** in methanol are shown in Figure 3. High energy bands were observed at  $\lambda_{\text{abs}} = 323$  nm for **4**;  $\lambda_{\text{abs}} = 316$  and  $289$  nm (shoulder) for **5** and  $\lambda_{\text{abs}} = 310$  nm for **6**. Low energy metal-ligand charge transfer bands are also seen at  $\lambda_{\text{abs}} = 449$  nm for all three species. Upon excitation at 290 nm (Figure 3, right), emission bands are observed at  $\lambda_{\text{em}} = 358$  nm for **4**;  $\lambda_{\text{em}} = 380$  nm for **5** and  $\lambda_{\text{em}} = 365$  nm for **6**. The relatively intense fluorescence of **6** is ascribed to the presence of ethynyl groups which result in extended  $\pi$  conjugation.<sup>1a</sup>

Due to the anticancer activity of existing O,O-bridged Ru SCCs and N,N-bridged Ru small molecules, the anti-proliferative activity of **4** and its self-assembled supramolecular derivatives **5** and **6** was investigated against various cancer cell lines such as Colo320 (colorectal cancer), A549 (lung cancer), MCF-7 (breast cancer) and H1299 (lung cancer). All cancer cells were exposed for 24 h to increasing concentrations of the compounds, and their activities were determined using a 3-(4,5-dimethylthiazol-2-yl)-5-(3-carboxymethoxyphenyl)-2-(4-sulfophenyl)-2H-tetrazolium (MTS) cell proliferation assay with the chemotherapeutic drug cisplatin used as a control. Based on the results of the MTS

assays, only one cell line, Colo320, was found to be sensitive to cisplatin while A549, MCF-7, and H1299, were resistant to it. The effects of **4–6** on these cell lines are summarized in Table 1. All four cell lines were resistant to **4**. Rectangle **5** showed activity only with the Colo320 cell line. Interestingly, the prism **6** was found to inhibit the proliferation of all four cell lines and their growths were effectively inhibited at a very low concentration (Table 1 and Figure 4).

H1299 cells were treated with 10  $\mu\text{M}$  of both **6** and cisplatin for 12 h, after which they were stained with TUNEL and examined by FACS analysis. Cisplatin and metalla-prism **6** caused apoptosis in 25% and 18% of H1299 cell population, respectively, suggesting that the inhibitory effect of the metalla-prism **6** was not obtained from the induction of apoptosis..

This result indicated that metalla-prism **6** was much more effective than cisplatin in the inhibition of cancer cell growth and could be a candidate for the development of chemotherapeutic drug against cisplatin-resistant cancer cells. As such, a careful examination of the anti-proliferative activity of the prism **6** is warranted to guide future Ru-based drug design.

In conclusion, we have reported the synthesis and characterization of a novel N, N-bridged Ru acceptor and its subsequent coordination-driven self-assembly chemistry in the formation of both 2D and 3D species. The antitumor action of this suite of compounds was evaluated, revealing that bis-benzimidazole bridged SCCs have potential to act as potent anticancer agents, particularly in cell lines which are resistant to Pt-based molecules. Further studies are under progress to investigate the biological mechanism of these derivatives.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

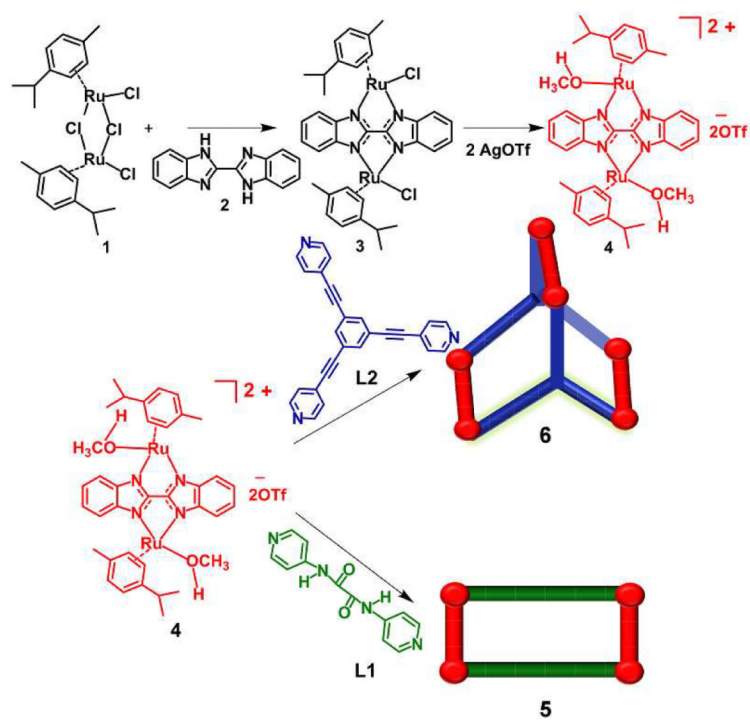
## Acknowledgments

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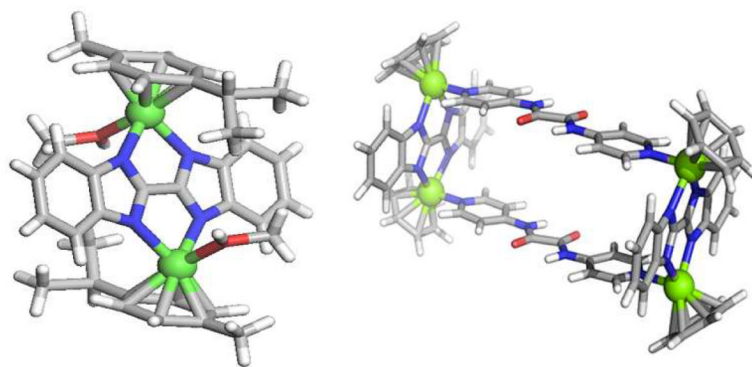
## References

- (1). (a) Vajpayee V, Song YH, Cook TR, Kim H, Lee Y, Stang PJ, Chi K-W. *J. Am. Chem. Soc.* 2011; 133:19646–19649. [PubMed: 22085308] (b) Yan H, Süß-Fink G, Neels A, Stoeckli-Evans H. *J. Chem. Soc. Dalton Trans.* 1997:4345–4350. (c) Vajpayee V, Jung YJ, Kang SC, Kim H, Kim IS, Wang M, Cook TR, Stang PJ, Chi K-W. *Dalton Trans.* 2012; 41:3046–3052. [PubMed: 22278716] (d) Kaim W, Schwederski B, Dogan A, Fiedler J, Kuehl CJ, Stang PJ. *Inorg. Chem.* 2002; 41:4025–4028. [PubMed: 12132929] (e) Liao R-T, Yang W-C, Thanasekaran P, Tsai C-C, Sathiyendiran M, Liu Y-H, Rajendran T, Lin H-M, Tseng T-W, Lu K-L. *Chem. Commun.* 2008:3175–3177.
- (2). (a) Resendiz MJE, Noveron JC, Disteldorf H, Fischer S, Stang PJ. *Org. Lett.* 2004; 6:651–653. [PubMed: 14986941] (b) Kuehl C, Huang SD, Stang PJ. *J. Am. Chem. Soc.* 2001; 123:9634–9641. [PubMed: 11572685] (c) Kuehl C, Mayne CL, Arif AM, Stang PJ. *Org. Lett.* 2000; 2:3727–3729. [PubMed: 11073686] (d) Dinolfo PH, Williams ME, Stern CL, Hupp JT. *J. Am. Chem. Soc.* 2004; 126:12989–13001. [PubMed: 15469297] (e) Benkstein KD, Hupp JT, Stern CL. *Angew. Chem. Int. Ed.* 2000; 39:2891–2893. (f) Dinolfo PH, Hupp JT. *Chem. Mater.* 2001; 13:3113–3125.
- (3). (a) Zhang W-Z, Han Y-F, Lin Y-J, Jin G-X. *Dalton Trans.* 2009:8426–8431. [PubMed: 19789798] (b) Han Y-F, Fei Y, Jin G-X. *Dalton Trans.* 2010; 39:3976–3984. [PubMed: 20372723] (c) Vajpayee V, Song YH, Lee MH, Kim H, Wang M, Stang PJ, Chi K-W. *Chem. Eur. J.* 2011;

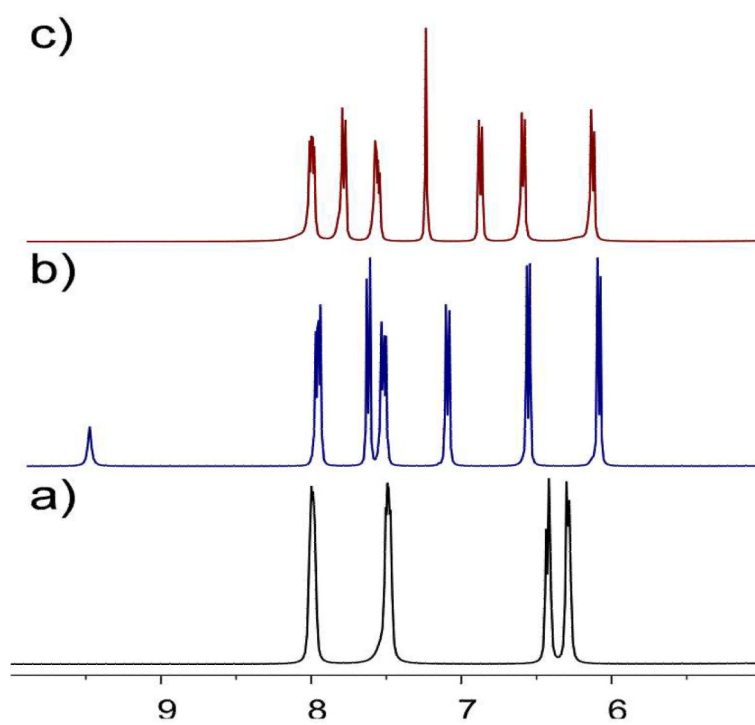
- 17:7837–7844. [PubMed: 21611989] (d) Furrer MA, Furrer J, Therrien B. *Organometallics*. 2012; 31:3149–3154. (e) Mishra A, Vajpayee V, Kim H, Lee MH, Jung H, Wang M, Stang PJ, Chi K–W. *Dalton Trans.* 2012; 41:1195–1201. [PubMed: 22116403]
- (4). (a) Wang M, Vajpayee V, Shanmugaraju S, Zheng YR, Zhao Z, Kim H, Mukherjee PS, Chi K–W, Stang PJ. *Inorg. Chem.* 2011; 50:1506–1512. [PubMed: 21214171] (b) Therrien, B. *Eur. J. Inorg. Chem.* 2009:2445–2453. (c) Barry NPE, Furrer J, Freudenreich J, Süß-Fink G, Therrien B. *Eur. J. Inorg. Chem.* 2010:725–728. (d) Vieille-Petit L, Therrien B, Süß-Fink G, Ward TRJ. *Organomet. Chem.* 2003; 684:117–123. (e) Ang WH, Grote Z, Scopelliti R, Juillierat-Jeanerret L, Severin K, Dyson PJJ. *Organomet. Chem.* 2009; 694:968–972. (f) Govender P, Renfrew AK, Clavel CM, Dyson PJ, Therrien, Smith BGS. *Dalton Trans.* 2011; 40:1158–1167. [PubMed: 21165516] (g) Vajpayee V, Lee S, Kim S-H, Kang SC, Cook TR, Kim H, Kim DW, Verma S, Lah MS, Kim IS, Wang M, Stang PJ, Chi K-W. *Dalton Trans.* 2012 DOI: 10.1039/c2dt31014g.
- (5). (a) Vajpayee V, Yang YJ, Kang SC, Kim H, Kim IS, Wang M, Stang PJ, Chi K-W. *Chem. Commun.* 2011; 47:5184–5186. (b) Barry NPE, Edefe F, Therrien B. *Dalton Trans.* 2011; 40:7172–7180. [PubMed: 21660364] (c) Barry NPE, Zava O, Furrer J, Dyson PJ, Therrien B. *Dalton Trans.* 2010; 39:5272–5277. [PubMed: 20442944] (d) Therrien B, Süß-Fink G, Govindaswamy P, Renfrew AK, Dyson PJ. *Angew. Chem., Int. Ed.* 2008; 47:3773–3776. (e) Zava O, Mattsson J, Therrien B, Dyson PJ. *Chem.–Eur. J.* 2010; 16:1428–1431. [PubMed: 20033971] (f) Barry NPE, Abd Karim NH, Vilar R, Therrien B. *Dalton Trans.* 2009:10717–10719. [PubMed: 20023899] (g) Vajpayee V, Song YH, Yang YJ, Kang SC, Cook TR, Kim DW, Lah MS, Kim IS, Wang M, Stang PJ, Chi K–W. *Organometallics.* 2011; 30:6482–6489. [PubMed: 22180698] (h) Mishra A, Jung H, Park JW, Kim HK, Kim H, Stang PJ, Chi K-W. *Organometallics.* 2012; 31:3519–3526. [PubMed: 22639481]
- (6). (a) Morris RE, Aird RE, Mudroch S, Chen HM, Cummings J, Hughes ND, Parsons S, Parkin A, Boyd G, Jodrel DI, Sadler PJ. *J. Med. Chem.* 2001; 44:3616–3621. [PubMed: 11606126] (b) Wang F, Bella J, Parkinson JA, Sadler PJ. *J. Biol. Inorg. Chem.* 2005; 10:147–155. [PubMed: 15735959] (c) Betanzos-Lara S, Salassa L, Habtemariam A, Novakova O, Pizarro AM, Clarkson GJ, Liskova B, Brabec V, Sadler PJ. *Organometallics.* 2012; 31:3466–3479.
- (7). (a) Haga M, Bond AM. *Inorg. Chem.* 1991; 30:475–480. (b) Mann J, Baron A, Opoku-Boahen Y, Johansson E, Parkinson G, Kelland LR, Neidle S. *J. Med. Chem.* 2001; 44:138–144. [PubMed: 11170623] (c) Mo H-J, Niu Y-L, Zhang M, Qiao Z-P, Ye B-H. *Dalton Trans.* 2011; 40:8218–8225. [PubMed: 21731958] (d) Saha D, Das S, Maity D, Dutta S, Baitalik S. *Inorg. Chem.* 2011; 50:46–61. [PubMed: 21114281] (e) Rau S, Büttner T, Temme C, Ruben M, Görls H, Walther D. *Inorg. Chem.* 2000; 39:1621–1624. [PubMed: 12526478] (f) Rau S, Ruben M, Büttner T, Temme C, Dautz S, Görls H, Rudolph M, Walther D, Brodkorb A, Duati M, O'Connor C, Vos JG. *J. Chem. Soc. Dalton Trans.* 2000:3649–3657. (g) Carmona D, Ferrer J, Mendoza A, Lahoz FJ, Oro LA, Viguri F, Reyes J. *Organometallics.* 1995; 14:2066–2080.



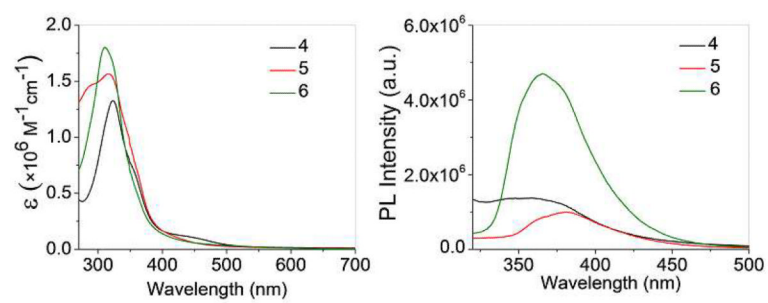
**Scheme 1.**  
Synthesis of acceptor **4**, metalla-rectangle **5** and metalla-prism **6**.



**Figure 1.** X-ray crystal structure of molecular clip **4** (*left*) and DFT-optimized computational structure of a model of rectangle **5** (*right*) (*i*Pr and Me groups were omitted for DFT calculation). Atom (color): Ru (green), O (red), N (blue), C (grey), H (white).

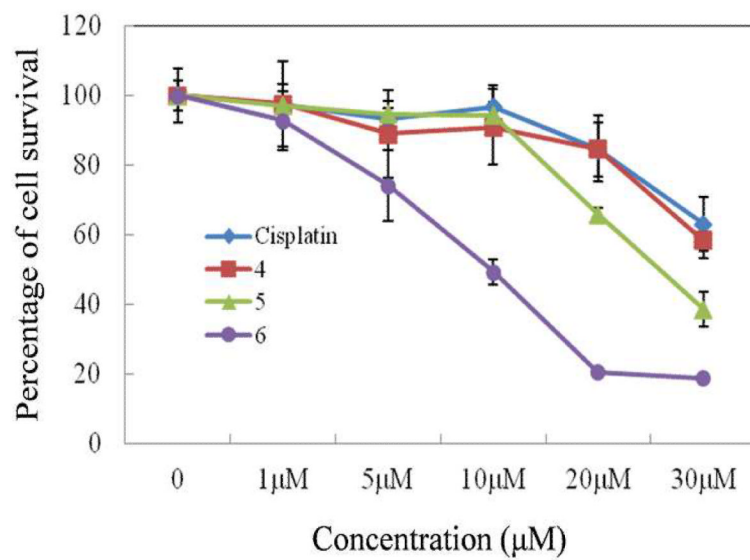


**Figure 2.**  
Partial  $^1\text{H}$  NMR spectra of **4(a)**, **5(b)** and **6(c)** in  $\text{CD}_3\text{NO}_2$ .



**Figure 3.** UV-vis. (*left*) and Fluorescence (*right*) spectra of **4**, **5** and **6** in Methanol.





**Figure 4.** Viability of H1299 lung cancer cells under treatment with 4– 6 and cisplatin.

**Table 1**

Cytotoxicity of the complexes in human cancer cells.

Compound	IC <sub>50</sub> $\mu\text{M}$ <sup>[a]</sup>			
	Colo320	A549	H1299	MCF7
<b>4</b>	>100	>100	>100	>100
<b>5</b>	13.94 $\pm$ 3.49	>100	>100	>100
<b>6</b>	78.86 $\pm$ 12.11 *	15.42 $\pm$ 4.73	15.65 $\pm$ 4.55	8.41 $\pm$ 2.19
<b>Cisplatin</b>	38.6 $\pm$ 6.26	>100	>100	>100

<sup>[a]</sup> drug concentration necessary for 50% inhibition of cell viability,

\* mean $\pm$ SD