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3-Hydroxypyridin-2-thione as Novel Zinc Binding Group for Selective Histone Deacetylase Inhibition

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Abstract

Small molecules bearing hydroxamic acid as the zinc binding group (ZBG) have been the most effective histone deacetylase inhibitor (HDACi) to date. However, concerns about the pharmacokinetic liabilities of the hydroxamic acid moiety have stimulated research efforts aimed at finding alternative non-hydroxamate ZBGs. We have identified 3-hydroxypyridin-2-thione (3- HPT) as a novel ZBG that is compatible with HDAC inhibition. 3-HPT inhibits HDAC 6 and HDAC 8 with an IC₅₀ of 681 nM and 3675 nM respectively. Remarkably, 3-HPT gives no inhibition of HDAC 1. Subsequent optimization led to several novel 3HPT-based HDACi that are selective for HDAC 6 and HDAC 8. Furthermore, a subset of these inhibitors induces apoptosis in various cancer cell lines.

Introduction

Eukaryotic DNA is wrapped around nucleosomes comprised of histone proteins that are subjected to various post-translational modifications including acetylation, phosphorylation, sumoylation and methylation. These post-translational modifications function to regulate transcription.1, 2 Histone acetylation/deactylation, which have been the most studied covalent modifications, are mediated by the histone acetyl transferases (HATs) and the histone deacetylases (HDACs) respectively.^{3, 4} We now know that a significant fraction of cellular proteins are also substrates for HDAC and HAT enzymes, extending their role beyond that of transcriptional regulation.⁵ Presumably due to their involvement in repressing transcription, various HDAC isoforms are overexpressed in different cancers and as such are valid targets for cancer treatment.⁶ In fact, two histone deacetylase inhibitors (HDACi) – suberoylanilide hydroxamic acid (SAHA) and cyclic peptide FK228– are approved for the treatment of cutaneous T-cell lymphoma (CTCL).⁴ Other pathological conditions where targeting HDAC constitute a plausible therapeutic option include inflammatory diseases, parasitic infections, hemoglobinopathies and neurodegenerative diseases.7, 8, 9, 10

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Supporting Information Available: 1H NMR and 13C NMR spectral information, solubility data and molecular modeling outputs. This material is available free of charge via the Internet at<http://pubs.acs.org>.

The classic pharmacophoric model of HDACi consists of a zinc binding group (ZBG) that chelates the active site Zn^{2+} ion, a linker and a surface recognition cap group that interacts with the amino acid residues present at the surface of the HDAC (Figure 1).¹¹

Chelation of the Zn²⁺ ion has proven crucial for HDAC inhibition.¹² The hydroxamic acid has been the preferred ZBG due to its strong Zn^{2+} ion chelation.^{13, 14} Yet the hydroxamic acid could present metabolic and pharmacokinetic challenges, including a short half-life and poor bioavailability.^{1516, 17} The hydroxamate also chelates other biologically relevant metals, including Fe²⁺ and Cu²⁺ with affinities that can exceed that of Zn^{2+} ion.^{18, 19} Extensive reports have aimed to improve the HDAC inhibition profile by manipulating the surface recognition cap group and linker region while retaining the hydroxamic acid as ZBG. Indeed, these efforts have resulted in highly potent and, in some cases, isoformselective compounds.^{20, 21}

Several efforts have replaced the hydroxamic acid with alternative chemical moieties.²² For example, MS-275 is a class I selective HDACi having a benzamide ZBG.23 It has been suggested that the benzamide ZBG exploits the difference in the region adjacent to the active site to achieve its isoform selectivity.24 Further, subtle differences in the active sites of various HDAC isoforms have been exploited to design compounds having other ZBGs, including thiols, α-ketoesters, electrophilic ketones, mercaptoamides and phosphonates.11, 25 However, most of these analogues had reduced potency.

Non-hydroxamate chemotypes that chelate Zn^{2+} ion have been well studied in the context of inhibitors of the matrix metalloproteins (MMPs) (Figure 2). This work has revealed that bidentate heterocyclic ZBGs are stronger metal chelators than are the monodentate analogs.26, 27 Furthermore, the bidentate heterocyclic ZBGs are resistant to hydrolysis and are effective at inhibiting the proteinase activities of various MMP isoforms.²⁷ We therefore borrowed the bidentate heterocyclic ZBGs to evaluate a new class of HDACi that may be devoid of many of the liabilities of the hydroxamate moiety. We herein report that 3 hydroxypyridin-2-thione (3-HPT) is a bidentate heterocyclic ZBG that is compatible with HDAC inhibition. 3-HPT inhibits the deacetylase activities of HDAC 6 and HDAC 8 with IC50 of 680nM and 3700nM respectively. Remarkably, 3-HPT is inactive against HDAC 1. Subsequent optimization led to several novel 3-HPT-based HDACi that are selective for HDAC 6 and HDAC 8. Furthermore, a subset of these inhibitors induces apoptosis in various cancer cell lines.

RESULTS AND DISCUSSION

Initial Molecular Docking Studies

We first performed molecular docking analyses on selected bidentate heterocyclic ZBG fragments against three HDAC isoforms – HDAC 1, HDAC 6 and HDAC 8. Our choice of bidendate ZBG fragments is informed by their reported Zn^{2+} ion chelation affinity and the ease with which subsequent modification could be introduced to enhance potency.^{26, 27} The bidendate ZBG fragments that met these criteria, 3-hydroxypyridin-2-one; 3 hydroxypyridin-2-thione; 3-hydroxypyridin-4-thione and 1-hydroxypyridin-2-thione were selected for initial studies (Figure 3). We performed docking analyses against the crystal structure of HDAC 8 (PDB code: 1VKG),²⁹ histone deacetylase-like protein (HDLP), a HDAC 1 homolog, 12 and the homology models of HDAC 1 and HDAC 6 built respectively from human HDAC 2 (PDB code: 3MAX) and HDAC 8 (PDB code: 3FOR) 21 using a validated molecular docking program (AutoDock 4.2) as described previously.^{30, 31}

Preliminary docking analyses revealed that 3-hydroxypyridin-2-one (3-HP), 3 hydroxypyridin-2-thione (3-HPT), 3-hydroxypyridin-4-thione and 1-hydroxypyridin-2-

thione chelate the Zn^{2+} ion at active site of HDLP, HDACs 1, 6 and 8, suggesting that these bidentate heterocycles could provide the critical ZBG in the HDACi pharmacophoric model. A closer inspection of the docked poses in HDACs 1, 6 and 8 revealed that the N-1 position of 3-hydroxypyridin-4-thione is oriented towards the base of the active site pocket as opposed to the N-1 positions of 3-HP and 3-HPT which are oriented towards the surface of the channel that the linker region of prototypical HDACi occupies to present the surface recognition cap group to the enzyme outer rim. Conversely, the docked poses of 3-HP and 3- HPT at HDLP active site that retained zinc chelation are those in which their N1-positions are oriented towards the base of the active site pocket in similar manner to the N-1 position of 3-hydroxypyridin-4-thione (Supplementary Information, Figures S1–S3). In HDLP, the N-1 position of 3-HPT engages in hydrogen bonding with GLY129 which stabilizes its orientation, however, in HDAC 1 homology model the outwards orientation of the 3-HPT is probably influenced by the prospect of hydrogen bonding of its N-1 with TYR303 phenol group and GLY149 backbone (Figure S1C).

The docked poses adopted by 3-HP and 3-HPT, particularly on HDACs 6 and 8, should enable facile introduction of linkers and surface recognition cap groups through the N-1 position in order to further enhance HDAC inhibition potency. Such modifications, however, could compromise inhibition activity of the 3-hydroxypyridin-4-thione fragment as its N-1 position participates in potentially stabilizing H-bonding interactions at the base of the enzyme active sites. Although, 1-hydroxypyridin-2-thione shows Zn^{2+} ion chelation at HDAC active sites (Supplementary Information, Figure S1, S2, S3), this ZBG is similar to hydroxamic acid and it is synthetically less tractable relative to 3-HP and 3-HPT. Based on these observations, we directed our efforts on the 3-HP and 3-HPT fragments. To test the validity of our in silico predictions, we investigated the effects of 3-HPT **2** and 3-HP **3** on the deacetylase activities of HDAC 1, HDAC 6 and HDAC 8. 3-HPT **2** was synthesized from the commercially available 3-hydroxypyridin-2-one as previously reported.³²

In vitro HDAC Inhibition activity of 3-HP and 3-HPT fragments

We used SAMDI mass spectrometry as previously described to assay inhibitor potency against the HDAC isoforms 1, 6 and 8.^{30, 33, 34} 3-HPT 2 inhibited HDAC6 with an IC₅₀ of 680 nM and inhibited HDAC8 with an IC_{50} of 3.7 μ M. This compound had no activity against HDAC 1 (Table 1). The analogue 3-HP **3** was inactive against each of the three HDAC isoforms. The disparity in the HDAC inhibition activity of 3-HP and 3-HPT could be due to the thiophilicity of zinc which favors 3-HPT.³⁵

The inactivity of 3-HPT against HDAC 1 may not be surprising giving the ambivalence of its interaction with the HDAC 1 (homology model) and HDLP active sites (Supplementary Info, Fig. S1). This data suggests that the docked structure of 3-HPT on HDLP may closely mimic its binding with HDAC 1. Alternatively, it is plausible that the narrowness of HDAC 1 tunnel, relative to those of HDAC 6 and HDAC 8, could constitute a challenge toward proper accommodation of 3-HPT at HDAC 1 active site. HDAC 1 activity or lack thereof could then be influenced by conformational changes which are not captured by docking analysis such as AutoDock.

Structure Activity Relationship (SAR) Studies

We next performed docking analyses of the N- 1 methyl, benzyl and biphenyl derivatives of 3-HP and 3-HPT and found that the compounds were predicted by AutoDock to adopt docking poses similar to 3-HP and 3-HPT (See Supplementary Information Figure S4 and S5 for docking results on N-1 methyl derivative). We then synthesized representative examples of these compounds - N-1 methyl (**5a**), benzyl (**5b** and **7b**) and biphenyl (**5c** and **7c**) derivatives (Scheme 1), and tested their HDAC inhibition activities against HDACs 1, 6

and 8. As observed with the 3-HP/3-HPT pair, carbonyl-compounds **5a**, **5b** and **5c** failed to inhibit HDAC activity (data not shown). The thione-compounds **7b** and **7c** inhibited HDAC 6 with mid-nanomolar IC_{50} s and less potent activity against HDAC 8 (Table 2). Similar to our observation with 3-HPT, neither **7b** nor **7c** possess any measurable activity against HDAC 1 at maximum tested concentration of 10 μM. The anti-HDAC 6 and 8 activities of **7b** and **7c** mirrored those of 3-HPT, however, **7b** is somewhat more active against both HDAC isoforms. That the biphenyl moiety of **7c** did not increase affinity of this compound for the HDAC may reflect a lack of interactions of this hydrophobic group with the hydrogen-bonding surface of the protein channel.

We reasoned that the diaryl moiety of **7c** could be tuned to introduce interactions with the enzyme active site channel residues. We substituted the proximal phenyl group with a 1,2,3 triazole ring to give analgoues **15a** and **17a**. In silico interrogation against HDACs 6 and 8 again revealed that **15a** and **17a** adopted poses whereby the active site Zn^{2+} ion chelation is maintained in both isoforms. Interestingly, the triazole ring of either compound is positioned close to the HDAC 8 Tyr306 OH-group for possible H-bonding. The analogous interaction is missing in HDAC 6 (Figure 4, note that only the docked poses of the 3-HPT diaryl **17a** are shown for clarity).

We next synthesized the diaryl compounds **15a** and **17a** and their analogs (discussed later) using the route shown in Scheme 3 and tested their HDAC inhibition activities against HDACs 1, 6 and 8. Consistent with the observations described above, the 3-HP compound **15a** is inactive against the HDAC isoforms tested. Surprisingly, 3-HPT diaryl **17a** is only marginally inhibitory to HDAC 6 at a concentration of 10μ M, quite in contrast to the effect of the analogous compound **7c**. We also found that **17a** is about 2 to 3 fold more active against HDAC 8 relative to **7c** (Table 2). This result is consistent with our in silico observation which suggested that the triazole ring could be beneficial to HDAC 8 inhibition (Figure 4).

Encouraged by these results, we focused our SAR studies on **7c** and **17a** as leads, introducing substituents at the distal phenyl group (relative to the ZBG). We prepared a series of aromatic and heteroaromatic derivatives to investigate the effect of the substitution pattern and the nature of substituents on inhibition activity. Because we used a Suzuki cross coupling36 to join the 3-HP phenyl bromide **8** with a phenyl boronic acid, we could readily prepare several analogous biphenyl compounds (**9a–i)**. Subsequent deprotection of the Omethoxy group with boron tribromide gave 3-HP biphenyl compounds **11a–i**. Alternatively, the reaction of **9a–i** with Lawesson's reagent resulted in O-methoxy protected 3-HPT biphenyl compounds **10a–i** which upon treatment with boron tribromide furnished the desired 3-HPT biphenyl compounds **12a–i** (Scheme 2).

Consistent with the pattern seen with the previous analogs, all 3-HP biphenyl compounds **11a–i** lacked HDAC inhibition activity against HDACs 1, 6 and 8 (data not shown). Similarly, all the 3-HPT biphenyl compounds **12a–i** are inactive against HDAC 1 while the majority possess substituent dependent inhibitory activities against HDACs 6 and 8 with varying degrees of selectivity for either isoform (Table 2). Compound **12a**, an analog of **7c** with a cyano-substituent at the *para*-position, inhibits HDAC 6 at a level nearly identical to that of **7c** while it is about 2- fold better HDAC 8 inhibitor relative to **7c**. A shift of the cyano-substitution to the meta-position in compound **12b** resulted in the abrogation of HDAC 6 inhibition while a slight enhancement of HDAC 8 inhibition was observed. Orthoplacement of the cyano-substituent in compound **12c** restored HDAC 6 inhibition with higher potency relative to compounds **7b** and **7c** while HDAC 8 inhibitory activity remained largely unperturbed. Interestingly, methyl substituted compounds **12d–f** are strongly inhibitory to HDAC 6 irrespective of the position of substitution. In fact compound **12f** is

the strongest HDAC 6 inhibitor in this series of 3-HPT biphenyl compounds with IC_{50} of 306 nM (Table 2). Conversely, the ability of compounds **12d–f** to inhibit HDAC 8 is very sensitive to the placement of the methyl-substituent. The para-methyl derivative **12d** is more active, with IC₅₀ of 800 nM, than the corresponding *meta*-methyl and *ortho*-methyl analogs, **12e** and **12f**, respectively. Placement of the electron donating N,N-dimethylamino group at the para-position in compound **12g**, analogous to the substitution pattern on the surface recognition group of simple HDACi Trichostatin A (TSA), resulted in loss of HDAC 6 inhibition while HDAC 8 inhibition activity at micromolar IC_{50} is about 2 fold enhanced relative to the lead compound **7c**. The 4-pyridyl substitution in compound **12i** partially restored HDAC 6 inhibition lost in **12g** without any added benefit to HDAC 8 inhibition. Interestingly, the combination of pyridine and N,N-dimethylamine substitutions as seen in compound **12h** resulted in the loss of HDAC 8 inhibition and a low micromolar HDAC 6 activity, a near mirror image of the HDAC inhibition preference of **12g**. Finally, we found that the incorporation of N,N-dimethylamino group at the para-position of triazole containing 3-HPT **17a**, to afford compound **17b**, did not improve HDAC 8 selectivity. Instead, it conferred HDAC 6 inhibition activity to **17b**. A similar substitution on the biphenyl congeners improved HDAC 8 inhibition but resulted in the loss of HDAC 6 inhibition (Table 2, comparing **12g** and **17b**). This observation suggests a possible divergence in the SARs of the triazole and biphenyl 3-HPT compounds. We describe the furtherance of our SAR studies on the triazole 3-HPT compounds in the next manuscript.

We then performed further docking analyses on representative compounds against HDAC 6 to gain a better understanding of the molecular basis of their stronger binding interaction with this HDAC isoform. We docked compounds **12a–f** which showed a clear dependence of substitution pattern on HDAC inhibition activities (Table 2). The docked poses adopted by these compounds gave insights into the preference for ortho-substituents in the analogues (Figure 5). The biphenyl moieties of the ortho-substituted compounds **12c** and **12f** adopted a nearly indistinguishable orientation in which the cyano and methyl groups, respectively, are neatly nestled within a hydrophobic pocket at the enzyme outer rim where they could engage in potentially stabilizing hydrophobic interactions with amino acid residues including Tyr782 and Phe620. In contrast the cyano and methyl groups of compounds **12a**, **12b**, **12d** and **12e** are not oriented towards the same hydrophobic pocket. Instead, they are solvent exposed thus missing the potentially stabilizing hydrophobic interactions enjoyed by their ortho-substituted congeners. This observation might explain the enhanced potency of **12c** and **12f** against HDAC 6 compared to **12a**, **12b**, **12d** and **12e** (Table 2).

To further probe the structural basis for the hindrance of the proper accommodation of these 3HPT-based compounds at HDAC 1 active site, we docked compound **12d**, analog with potent HDAC 6 and 8 inhibition activities but lacking the potentially obstructive substitution ortho/meta substitution, against HDAC 1 homology model and HDLP. We observed that **12**d was incapable of entering HDAC 1 pocket to chelate the zinc at the active site, instead it was bound to the residues at the surface of the binding pocket. While **12d** bound to mutiple spots on HDLP, none of its conformation was able to present the 3HPT ZBG for chelation to the active site zinc ion (Figure S6). This *in silico* observation may further explain the lack of activity of this class of compounds against HDAC 1.

In Vitro **Cell Growth Inhibition**

We tested several of the compounds in proliferation assays of cancer cell lines DU-145 (androgen independent prostate cancer), LNCaP (androgen dependent prostate cancer) and the T-cell leukemia cell line Jurkat. The compounds included a selective HDAC8 inhibitor (**12g**), compounds that inhibited both HDAC6 and 8 (**12c** and **12f)** and a triazole-based inhibitor selective for HDAC8 (**17a**).

Although 3-HPT **2** is active against HDAC6 and to some extend HDAC8, it is inactive against the three cancer cell lines tested (Table 3). The discrepancy between the HDAC inhibition profile and the anti-proliferative activity of 3-HPT may be due to solubility problems that limit its diffusion across the cell membrane.³⁷ Against DU-145, a prostate cancer line known to be responsive to HDACi such as SAHA or TSA, only **12f** exhibited a weak activity; all other compounds were inactive at the highest tested concentration of 20 μ M (Table 3). SAHA, which was used as a positive control, had IC₅₀ value comparable to that reported in the literature.³⁸

The compounds were more active on the LNCaP cell line (Table 3). Compound **12f** is the most potent of the biphenyls, followed by **12c** and **12g**. Interestingly, the cell growth inhibition activity of the triazolyl compound **17a** is comparable to that of **12c** although **17a** is a weak HDAC 6 inhibitor (only 41% inhibition at 10μ M, Table 2). With the exception of **17a**, the observed trend of the cell growth inhibition activity mirrored that of the HDAC 6 inhibition seen in Table 2. This may be due to the fact that LNCaP cells are sensitive to the acetylation state of HSP90. Hyperacetylation of HSP90, induced by the tested 3-HPT derived HDACi, attenuates its interactions with proteins such as androgen receptors, which are key to the survival of LNCaP cells.^{39,40}

Because T-cell-derived leukemia cell line Jurkat has been previously shown to be sensitive to selective HDAC 8 inhibitor, 41 we investigated the effects of our compounds on the viability of Jurkat cells. Overall, all the compounds, excluding 3-HPT, inhibit Jurkat cell growth with single digit micromolar IC50 values (Table 3). Compound **12f**, which inhibited both HDAC 6 and HDAC 8, is the most potent against the Jurkat cells (IC₅₀ = 3.19 μ M) in similar manner to its activity against LNCaP cells. . The HDAC 8 selective compounds **12g** and **17a** are cytotoxic to the Jurkat cells with **17a**, the more potent HDAC 8 inhibitor of the two, being twice as potent. The enhanced cytotoxicity of **17a** relative to **12g** could suggest a causal relationship between HDAC 8 inhibition and the apoptotic effect against Jurkat cells.

The apoptotic pathway, which is mediated by the activity of phospholipase C γ 1 (PLC γ 1) in Jurkat cells, is believed to be initiated by an HDAC 8 specific enzyme.41 Balasubramanian and coworkers reported that an HDAC 8 specific inhibitor, PCI-34051, lost its ability to induce apoptosis in Jurkat J.gamma1, a mutant Jurkat cell line which has no detectable PLC γ 1 activity.⁴¹ To further delineate the contribution of HDAC 8 to apoptosis induction, we probed the effect of compounds **12f**, **12g**, **17a** and SAHA on the viability of Jurkat J.gamma1 cells. As anticipated for HDAC 8 selective inhibitors, compounds **12g** and **17a** are devoid of anti-proliferative activity against Jurkat J.gamma1 (Table 3). Quite unexpectedly, **12f** (which inhibited HDACs 6 and 8) and SAHA (a much broader HDACi) are non-cytotoxic to this mutant Jurkat cell line. The inactivity of **12g** and **17a** against Jurkat J.gamma1 confirmed HDAC 8 contribution to their bioactivity. In the Jurkat cell line, Ca^{2+} released through PLCγ1 triggers cascade of events resulting in mitochondrial cytochrome c release and caspase-dependent apoptosis. The defect in intracellular Ca^{2+} mobilization in the Jurkat J.gamma1 mutant may also explain the inactivity of **12f** and SAHA as this prevents the increase in cytochrome c release observed in the wild type Jurkat.41 Alternatively, the observed dependence of the cytotoxicity activity of **12f** and SAHA on PLCγ1 activity may be due to the attenuation of caspase-dependent apoptosis which is crucial for the cytotoxicity of HDACi.

Based on the foregoing, there is a discrepancy between HDAC inhibition and the whole cell activities of some of the tested compounds. This is to be expected as it is not unusual to observe such a lack of correlation between the two experiments. It is however possible that compound solubility could play a significant role in cell penetration which may perturb the whole cell activities. Therefore, we estimated the solubility of all compounds tested in cell

growth inhibition assay to probe for the effect of solubility on anti-proliferative activity. Among these compounds, 3HPT and **12g** had solubility less than 65μg/mL (supplementary info Fig. S7 and Table S1). For these two compounds, poor solubility may explain the lack of cellular activity.^{42, 43} However, the solubilities of the other compounds tested are beyond the suggested threshold that limits cellular activity.^{42, 43} Hence, the reduced efficacy of many of these compounds may be attributable to factors other than solubility issues.

Intracellular Target Validation

Of the compounds tested for anti-cancer activity, **12f** is approximately 10 fold selective toward HDAC 6 compared to HDAC8. To probe the contribution of HDAC 6 inhibition to the cytotoxic activity of **12f**, we determined the level of tubulin acetylation, a common marker for intracellular HDAC 6 inhibition, in LNCaP cells through Western blot. We observed that $12f$ led to an increase in tubulin acetylation in LNCaP cells at IC_{50} concentration and at 20μM (Figure 6). While SAHA, used as a positive control, showed concentration dependent tubulin hyperacetylation as observed before.44 This data provides evidence for the involvement of intracellular HDAC 6 inhibition as part of the mechanisms of anti-proliferative activity of **12f**.

Conclusion

We report that 3-hydroxypyridin-2-thione (3-HPT) is a novel ZBG that is compatible with HDAC inhibition. All of the 3-HPT-derived compounds reported herein are inactive against HDAC 1 but many possessed varying degrees of activity in inhibiting HDAC 6 and HDAC 8. Additionally, a subset of these compounds is cytotoxic to various cancer cell lines. The pattern of changes to key intracellular markers induced by representative members of these compounds confirmed the contribution of HDACs 6 and 8 to their bioactivity. To the best of our knowledge, this study described the first use of 3-HPT as ZBG for HDAC inhibition. Due to their anticipated immunity to many of the metabolic and pharmacokinetic liabilities that has beleaguered the hydroxamate ZBG, these 3-HPT-derived HDACi may display improved in vivo activity.

Experimental

Materials and methods

Bromoalkanoic acid, benzyl bromide, 4-bromobenzylbromide, 4-(bromomethyl)-1,1' biphenyl, 3-methoxy-2(1H)-pyridone, propargyl bromide, phenylacetylene and representative boronic acids were purchased from either Sigma–Aldrich or Alfa-Aesar. Anhydrous solvents and other reagents were purchased and used without further purification. Analtech silica gel plates $(60 F₂₅₄)$ were used for analytical TLC, and Analtech preparative TLC plates (UV 254, 2000 μ m) were used for purification. UV light was used to examine the spots. Silica gel (200–400 Mesh) was used in column chromatography. NMR spectra were recorded on a Varian-Gemini 400 magnetic resonance spectrometer. ¹H NMR spectra were recorded in parts per million (ppm) relative to the peak of CDCl₃, $(7.24$ ppm), CD₃OD (3.31 ppm), or DMSO- d_6 (2.49 ppm). ¹³C spectra were recorded relative to the central peak of the CDCl₃ triplet (77.0 ppm), CD₃OD (49.0 ppm), or the DMSO-d₆ septet (39.7 ppm), and were recorded with complete heterodecoupling. Multiplicities are described using the abbreviation s, singlet; d, doublet, t, triplet; q, quartet; m, multiplet; and app, apparent. High-resolution mass spectra were recorded at the Georgia Institute of Technology mass spectrometry facility in Atlanta. All final 3HPT-based compounds were established to be > 95% pure using HPLC. These HPLC analyses were done on a Beckman Coulter instrument with a Phenomenex RP C-18 column $(250 \text{ mm} \times 4.6 \text{ mm})$, using water (solvent A) and acetonitrile (solvent B) gradient, starting from 40% to 80% of B over 20 min. The

flow rate was 1mL/min and detection was at 379 nm. Phenyl azide and 4-azido-N,Ndimethylaniline were synthesized as previously reported.45, 46 DU-145, LNCaP, Jurkat J.gamma1 were obtained from ATCC (Manassas, VA, USA), Jurkat E6-1 cell line was kindly donated by Dr. John McDonald and grown on recommended medium supplemented with 10% fetal bovine serum (Global Cell Solutions, Charlottesville, VA, USA) at 37°C in an incubator with 5% $CO₂$. Mouse anti-acetylated α-Tubulin antibody was obtained from Invitrogen (Life Technologies, Grand Island, NY, USA), rabbit anti-actin and rabbit antitubulin α antibodies were purchased from Sigma-Aldrich (St. Louis, MO, USA). Secondary antibodies, goat anti-rabbit conjugated to IRDye680 and goat anti-mouse conjugated to IRDye800 were purchased from LI-COR Biosciences (Lincoln, NE, USA). The CellTiter 96 AQueous One Solution Cell Proliferation assay (MTS) kit was purchased from Promega (Madison, WI, USA).

Histone deacetylase inhibition

The HDAC activity in presence of various compounds was assessed using the SAMDI mass spectrometry. As a label-free technique, SAMDI is compatible with a broad range of native peptide substrates without requiring potentially disruptive fluorophores. To obtain IC_{50} values, we incubated isoform-optimized substrates (50μM) with enzyme (250nM) and inhibitor (at concentrations ranging from 10nM to 1.0mM) in 96-well microtiter plates (60 min, 37°C). Solution-phase deacetylation reactions were quenched with trichostatin A (TSA) and transferred to SAMDI plates to immobilize the substrate components. SAMDI plates were composed of an array of self-assembled monolayers (SAMs) presenting maleimide in standard 384-well format for high-throughput handling capability. Following immobilization, plates were washed to remove buffer constituents, enzyme, inhibitor, and any unbound substrate and analyzed by MALDI mass spectrometry using automated protocols.34 Deacetylation yields in each triplicate sample were determined from the integrated peak intensities of the molecular ions for the substrate and the deacetylated product ion by taking the ratio of the former over the sum of both. Yields were plotted with respect to inhibitor concentration and fitted to obtain IC_{50} values for each isoform-inhibitor pair.

Cell viability assay

DU-145 cells were maintained in EMEM supplemented with 10 % FBS while all other cell lines were maintained in RPMI 1640 supplemented with 10 % FBS. DU-145 and LNCaP cells were incubated on a 96-wells plate for 24 hours prior to the drug treatment while Jurkat and Jurkat J.gamma1 cells were incubated in media containing the various compounds for 72 hours. Cell viability was measured using the MTS assay according to manufacturer protocol. The DMSO concentration in the cell media during the cell viability experiment was maintained at 0.1%.

Western blot analysis fortubulin acetylation

LNCaP cells were plated for 24 hours and treated with various concentrations of compounds for 4 hours. The cells were washed with PBS buffer and resuspended in CelLytic™M buffer containing a cocktail of protease inhibitor (Sigma-Aldrich, St. Louis, MO, USA). Following quantification through a Bradford protein assay, equal amount of protein was loaded onto an SDS-page gel (Bio-Rad, Hercules, CA, USA) and resolved by electrophoresis at a constant voltage of 100V for 2 hours. The gel was transfer onto a nitrocellulose membrane and probed for acetylated tubulin, tubulin and actin as loading control.

Statistical analysis

The values reported as mean \pm standard deviation from at least 2 independent triplicate experiments. A student's t-test was performed in Excel, and results with p value less than 5% were considered statistically different.

1-Methyl-3-methoxypyridin-2-one (4a)

To a stirring mixture of 3-methoxypyridin-2-one (0.20 g, 1.6 mmol) and KOH (0.18 g, 3.2 mmol) in methanol was added MeI (0.68 g, 4.8 mmol) dropwise in a round bottom flask. Stirring continued overnight at room temperature during which a quantitative consumption of starting materials was observed. The reaction mixture was diluted with water (35 mL) and $CHCl₃$ (40 mL) and the two layers were separated. The organic layer was washed with water (35 mL), brine (30 mL) and dried over $Na₂SO₄$. Solvent was evaporated off in vacuo to yield **4a** (0.20 g, 89%) as a colorless oil. 1H NMR (400 MHz, CDCl3) δ 6.66 (m, 1H), 6.35 (d, $J = 7.4$ Hz, 1H), 5.83 (t, $J = 7.1$ Hz, 1H), 3.52 (s, 3H), 3.28 (s, 3H). ¹³C NMR (100 MHz, CDCl3) δ 158.0, 149. 6, 128.8, 111.9, 104.3, 55.44, 37.1.

1-Benzyl-3-methoxypyridin-2-one (4b)

To a stirring mixture of 3-methoxypyridin-2-one (0.20 g, 1.6 mmol) and K_2CO_3 (0.66 g, 4.8) mmol) in DMF (8 mL) was added benzyl bromide (0.33 g, 1.92 mmol) dropwise in a condenser equipped round bottom flask. Reaction mixture was heated at 100 °C overnight. The reaction mixture was cooled down, diluted with water (40 mL) and CHCl₃ (50 mL) and the two layers were separated. The organic layer was washed with water $(3 \times 40 \text{ mL})$, brine (30 mL) and dried over Na_2SO_4 . Solvent was evaporated off *in vacuo* to yield **4b** (0.26 g, 75%) as a colorless oil without any further purification needed. ¹H NMR (400 MHz, CDCl₃) δ 7.26 (m, 5H), 6.85 (dd, J = 6.9, 1.7 Hz, 1H), 6.54 (dd, J = 7.5, 1.6 Hz, 1H), 6.03 (m, 1H), 5.13 (s, 2H), 3.76 (s, 3H). 13C NMR (100 MHz, CDCl3) δ 157.9, 150.1, 136.3, 128. 6, 128.0, 127.7, 111.8, 104.8, 104.8, 55.6, 51.6.

1-(1,1'-Biphenylmethyl)-3-methoxypyridin-2-one (4c)

The reaction of 3-methoxypyridin-2-one $(0.20 \text{ g}, 1.6 \text{ mmol})$, $K_2CO_3 (0.66 \text{ g}, 4.8 \text{ mmol})$, (4bromomethyl)-1,1'-biphenyl (0.47 g, 1.92 mmol) in DMF (8 mL) according to method described for the synthesis of $4b$ followed by column chromatography in CH_2Cl_2 with acetone $(0-15\%$ gradient) followed by CH₂Cl₂ with MeOH $(0-5\%$ gradient) afforded 4c (0.35 g, 76%) as a colorless oil. ¹H NMR (400 MHz, CDCl₃) δ 7.48 (dd, *J* = 25.9, 20.6 Hz, 4H), 7.32 (m, 5H), 6.88 (m, 1H), 6.54 (d, $J = 7.2$ Hz, 1H), 6.04 (t, $J = 7.1$ Hz, 1H), 5.16 (s, 2H), 3.76 (s, 3H). 13C NMR (100 MHz, CDCl3) δ 157.9, 150.0, 140.5, 140.2, 135.3, 128.5, 128.4, 127.7, 127.2, 127.1, 126.7, 111.8, 104.8, 55. 6, 51.4.

1-Methyl-3-hydroxypyridin-2-one (5a)

To a solution of $4a$ (0.10 g, 0.68 mmol) in dry CH₂Cl₂ (8 mL) was slowly added 1M BBr₃ (0.82 mL) at −30 °C under inert atmosphere. The reaction mixture was stirred for 48 h at room temperature. The mixture was again cooled to −30 °C and MeOH (5 mL) was slowly added to the mixture. After evaporation of solvent, the residue was adjusted to pH 7 with 1M NaOH and then extracted with CHCl₃ (3×30 mL). The combined organic layer was dried over $Na₂SO₄$ and solvent was evaporated in vacuo. The residue was purified by prep-TLC, eluting with CH_2Cl_2 : Acetone: MeOH (10:1:0.2), to give $5a$ (61 mg, 72%) as an off white solid. ¹H NMR (400 MHz, CDCl₃) δ 7.92 (s, 1H), 6.78 (dd, J = 11.1, 7.1 Hz, 2H), 6.10 (t, $J = 7.0$ Hz, 1H), 3.57 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 158.8, 146. 7, 127.6, 114.4, 106.7, 37.3. HRMS (EI) calcd for $C_6H_7NO_2$ [M]⁺ 125.0477 found 125.0477.

1-Benzyl-3-hydroxypyridin-2-one (5b)

The reaction of **4b** (0.15 g, 0.55 mmol) with $1M$ BBr₃ 0.66 mL) in dry CH₂Cl₂ (5 mL) within 48 h as described for **5a** afforded **5b** (0.13 g, 90%) as slightly brownish solid. ¹H NMR (400 MHz, CDCl₃) δ 7.31 (m, 5H), 6.83 (m, 3H), 6.14 (t, J = 7.1 Hz, 1H), 5.19 (s, 2H). 13C NMR (100 MHz, CDCl3) δ 158.6, 146. 8, 135.8, 128.7, 128.0, 126.5, 113.9, 107.0, 52.2. HRMS (EI) calcd for $C_{12}H_{11}NO_2$ [M]⁺ 201.0792 found 201.0790.

1-(1,1'-Biphenylmethyl)-3-hydroxypyridin-2-one (5c)

The reaction of $4c$ (0.13 g, 0.43 mmol) with $1M$ BBr₃ (0.51 mL) in dry CH₂Cl₂ (7 mL) within 48 h as described for **5a** afforded **5c** (0.10 g, 82%) as slightly brownish solid. ¹H NMR (400 MHz, DMSO- d_6) δ 9.06 (s, 1H), 7.61 (d, J = 7.5 Hz, 4H), 7.36 (m, 6H), 6.70 (d, $J = 6.5$ Hz, 1H), 6.12 (t, $J = 6.9$ Hz, 1H), 5.16 (s, 2H). ¹³C NMR (100 MHz, DMSO- d_6) δ 147.0, 139.8, 139.5, 136.6, 128.9, 128.4, 128.2, 127.5, 126.9, 126.7, 114.8, 105.6, 51.1. HRMS (EI) calcd for $C_{18}H_{15}NO_2$ [M]⁺ 277.1101 found 277.1103.

1-Benzyl-3-methoxypyridin-2-thione (6b)

A suspension of **4b** (0.06 g, 0.28 mmol) and Lawesson's reagent (0.07 g, 0.17 mmol) in toluene (10 mL) was heated at reflux overnight. The reaction mixture was cooled to room temperature and solvent was evaporated *in vacuo*. The crude solid was purified on prep-TLC, eluting with CH_2Cl_2 : Acetone: MeOH (5:1:0.2), to give **6b** (53 mg, 82%) as yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 7.30 (m, 6H), 6.65 (d, $J = 7.7$ Hz, 1H), 6.54 (t, $J = 7.0$ Hz, 1H), 5.89 (s, 2H), 3.88 (s, 3H). 13C NMR (100 MHz, CDCl3) δ 173.2, 159.0, 135.2, 131.7, 128.7, 128.0, 128.0, 111.6, 109.6, 58.8, 56.6. HRMS (EI) calcd for $C_{13}H_{13}NOS$ [M]⁺ 231.0718 found 231.0716.

1-(1,1'-Biphenylmethyl)-3-methoxypyridin-2-thione (6c)

The reaction of **4c** (0.13 g, 0.44 mmol) and Lawesson's reagent (0.11 g, 0.27 mmol) in toluene according to method described for the synthesis of **6b** afforded **6c** (122 mg, 88%) as yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 7.52 (m, 1H), 7.35 (m, 2H), 6.66 (dd, J = 7.8, 1.3 Hz, 1H), 6.55 (dd, $J = 7.8$, 6.6 Hz, 1H), 5.93 (s, 1H), 3.89 (s, 1H). ¹³C NMR (100 MHz, CDCl3) δ 173.1, 159.0, 140.8, 140.2, 134.1, 131.7, 128.6, 128.4, 127.4, 127.3, 126.8, 111.6, 109. 7, 104.8, 58.5, 56.6. HRMS (EI) calcd for C19H17NOS [M]+ 307.1031 found 307.1029.

1-Benzyl-3-hydroxypyridin-2-thione (7b)

The reaction of $6b$ (0.05 g, 0.21 mmol) with $1M$ BBr₃ (0.66 mL) in dry CH₂Cl₂ (5 mL) within 48 h as described for **5a** afforded **7b** (34 mg, 72%) as olive green solid. Retention time 21.27 min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 5 min) ¹H NMR (400 MHz, CD₃OD) δ 7.34 (m, 3H), 6.84 (m, 1H), 5.83 (d, J = 49.5 Hz, 1H). 13C NMR (100 MHz, CDCl3) δ 164.7, 134.2, 129.0, 128.5, 128.1, 127.8, 126.7, 118.8, 118.3, 62.2. HRMS (EI) calcd for $C_{12}H_{11}NOS$ [M]⁺ 217.0561 found 217.0762.

1-(1,1'-Biphenylmethyl)-3-hydroxypyridin-2-thione (7c)

The reaction of $6c$ (0.10 g, 0.32 mmol) with $1M$ BBr₃ (0.48 mL) in dry CH₂Cl₂ (5 mL) within 48 h as described for **5a** afforded **7c** (74 mg, 90%) as olive green solid. Retention time 15.37 min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 5 min) ¹H NMR (400 MHz, CD₃OD) δ 7.39 (m, 9H), 6.88 (m, 3H), 5.86 (s, 2H). ¹³C NMR (100 MHz, CDCl3) δ 169.9, 155.2 141.3, 140.2, 133.5, 130.9, 128.7, 128.5, 127.7, 127. 5, 126.9, 113.7, 111.9, 59.8. HRMS (EI) calcd for $C_{18}H_{15}NOS$ [M]⁺ 293.0874 found 293.0873.

1-(4-Bromobenzyl)-3-methoxypyridin-2(1H)-one (8)

To a stirring reaction mixture of 3-methoxypyridin-2-one (2.00 g, 16 mmol) and K_2CO_3 (4.42 g, 32 mmol) in THF was added 4-bromobenzyl bromide (5.20 g, 20.8 mmol) slowly. The reaction mixture was the heated at reflux overnight, cooled down and then partitioned between CH_2Cl_2 (120 mL) and water (60 mL). The organic layer was separated, washed with water (2×60 mL), brine (1×40 mL) and dried on Na₂SO₄. Solvent was evaporated in vacuo and the crude yellowish solid was triturated with hexanes to give **8** (3.49 g, 74%) as white solid without further purification. ¹H NMR (400 MHz, CDCl₃) δ 7.45 – 7.29 (m, 2H), 7.15 (dd, $J = 7.4$, 1.2 Hz, 2H), 6.84 (dd, $J = 6.9$, 1.7 Hz, 1H), 6.55 (dd, $J = 7.4$, 1.5 Hz, 1H), 6.21 – 5.79 (m, 1H), 5.06 (s, 2H), 3.76 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 158.0, 150.3, 135.5, 131.8, 129.9, 127.7, 121.9, 112.0, 105.2, 55.8, 51.41. HRMS (EI) calcd for $C_{13}H_{12}BrNO_2$ [M]⁺ 293.0051 found 293.0051.

Representative procedure for Suzuki coupling reactions for synthesis of 9 – 1-(4-Cyano- (1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-one (9a)

Compound **8** (0.26 g, 0.86 mmol), (4-cyanophenyl)boronic acid (0.14 g, 0.95 mmol), 2M aq. K_2CO_3 (0.24 g, 1.73 mmol), toluene (8 mL), EtOH (4mL) and water (4mL) were added into reaction flask equipped with magnetic stirrer and water condenser. The resulting suspension was degassed for 10 min by sparging with argon gas. $Pd(PPh₃)₄$ (2.5 mol%) was added and the reaction mixture was heated at reflux overnight under argon atmosphere. After cooling to room temperature, CH_2Cl_2 was added (50 mL) and the mixture was extracted with water (40 mL), brine (20 mL) and dried on Na_2SO_4 . Solvent was evaporated in vacuo and the residue was purified by column chromatography eluting with CHCl₃:Acetone:MeOH (first with $5\% - 20\%$ acetone gradient, no MeOH followed by $1\% - 8\%$ MeOH step gradient) to give **9a** (0.42 g, 78%) as off white solid. 1H NMR (CDCl3, 400 MHz) δ 7.60 (m, 4H), 7.40 $(m, 4H)$, 6.90 (dd, $J = 6.8$, 1.6 Hz, 1H), 6.55 (dd, $J = 7.6$, 1.6 Hz, 1H), 6.07 (t, $J = 7.2$ Hz, 1H), 5.16 (s, 2H), 3.75 (s, 3H). 13C NMR (100 MHz, CDCl3) δ 158.1, 150.3, 145.0, 138.6, 137.1, 132. 6, 132.0, 128.9, 128.6, 128.4, 128.0, 127.6, 127.5, 118.9, 112.1, 110.9, 105.2, 56.2, 51.8. HRMS (EI) calcd for $C_2OH_6N_2O_2$ [M]⁺ 316.1212 found 316.1210.

1-(3-Cyano-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-one (9b)

The reaction of **8** (0.25 g, 0.85 mmol), (3-cyanophenyl)boronic acid (0.14g, 0.93 mmol), 2M aq. K₂CO₃ (0.23 g, 1.69 mmol) and Pd(PPh₃)₄ (2.5 mol%) according to method described for the synthesis of **9a** within 18 h afforded **9b** (175 mg, 66 %) as white solid. ¹H NMR (400) MHz, CDCl₃) δ 7.70 (m, 2H), 7.48 (m, 6H), 6.92 (dd, $J = 6.9$, 1.6 Hz, 1H), 6.57 (dd, $J = 7.4$, 1.4 Hz, 1H), 6.08 (t, $J = 7.2$ Hz, 1H), 5.16 (s, 2H), 3.76 (s, 3H). ¹³C NMR (100 MHz, CDCl3) δ 158.1, 150.2, 141.7, 138.3, 136.8, 131.4, 130.8, 130. 5, 129.7, 128.9, 128.0, 127.4, 118.8, 112.8, 112.1, 105.3, 55.8, 51.7. HRMS (EI) calcd for C₂₀H₁₆N₂O₂ [M]⁺ 316.1212 found 332.1216.

1-(2-Cyano-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-one (9c)

The reaction of **8** (0.15 g, 0.50 mmol), (2-cyanophenyl)boronic acid (0.09 g, 0.60 mmol), 2M aq. K_2CO_3 (0.14 g, 1.69 mmol) and Pd(PPh₃)₄ (2.5 mol%) according to method described for the synthesis of **9a** within 18 h afforded **9c** (118 mg, 75 %) of white solid. ¹H NMR (400 MHz, CDCl₃) δ 7.72 (m, 1H), 7.60 (m, 1H), 7.45 (m, 2H), 6.94 (dd, $J = 6.9, 1.7$ Hz, 1H), 6.60 (dd, $J = 7.4$, 1.6 Hz, 1H), 6.10 (t, $J = 7.2$ Hz, 1H), 5.21 (s, 1H), 3.80 (s, 1H). 13C NMR (100 MHz, CDCl3) δ 158.0, 150.2, 144.7, 137. 6, 136.9, 133.6, 132. 8, 129.9, 129.0, 128.3, 127.9, 127.6, 118.5, 112.0, 110.9, 105.1, 55.7, 51.6. HRMS (EI) calcd for $C_{20}H_{16}N_2O_2$ [M]⁺ 316.1212 found 316.1201.

1-(4-Methyl-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-one (9d)

The reaction of **8** (0.25g, 0.85 mmol), p-tolylboronic acid (0.14g, 1.02 mmol), 2M aq. K_2CO_3 (0.23g, 1.69 mmol), Pd(PPh₃)₄ (2.5 mol%) according to method described for the synthesis of **9a** within 18 h afforded **9d** (259 mg, quantitative) as white solid. ¹H NMR (400 MHz, CDCl₃) δ 7.49 (d, J = 8.1 Hz, 2H), 7.42 (d, J = 8.1 Hz, 2H), 7.33 (d, J = 8.1 Hz, 2H), 7.19 (d, $J = 8.0$ Hz, 2H), 6.88 (dd, $J = 6.9$, 1.6 Hz, 1H), 6.54 (dd, $J = 7.4$, 1.4 Hz, 1H), 6.03 (t, $J = 7.2$ Hz, 1H), 5.15 (s, 2H), 3.76 (s, 3H), 2.34 (s, 3H), ¹³C NMR (100 MHz, CDCl₃) δ 157.8, 149.9, 140.4, 137.3, 136.9, 134.9, 129.2, 128.4, 127.7, 126.9, 126.5, 111.8, 104. 8, 55.5, 51.3, 20.8. HRMS (EI) calcd for $C_{20}H_{19}NO_2$ [M]⁺ 305.1416 found 305.1422.

1-(3-Methyl-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-one (9e)

The reaction of **8** (0.20 g, 0.68 mmol), m-tolylboronic acid (0.11 g, 0.82 mmol), 2M aq. K_2CO_3 (0.19 g, 1.36 mmol) and Pd(PPh₃)₄ (2.5 mol%) according to method described for the synthesis of **9a** within 18 h afforded **9e** (259 mg, quantitative) as white solid. ¹H NMR $(400 \text{ MHz}, \text{CDCl}_3)$ δ 7.49 (m, 1H), 7.29 (m, 2H), 7.11 (d, J = 7.2 Hz, 1H), 6.88 (dd, J = 6.9, 1.7 Hz, 1H), 6.53 (dd, $J = 7.4$, 1.6 Hz, 1H), 6.03 (t, $J = 7.2$ Hz, 1H), 5.15 (s, 1H), 3.75 (s, 1H), 2.35 (s, 1H). 13C NMR (100 MHz, CDCl3) δ 157.8, 149.9, 140.6, 140.2, 138.0, 135.1, 128.4, 128.3, 127.8, 127.7, 127.5, 127.1, 123.8, 111.8, 104.8, 55.5, 51.3, 21.2. HRMS (EI) calcd for $C_{20}H_{19}NO_2$ [M]⁺ 305.1416 found 305.1415.

1-(2-Methyl-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-one (9f)

The reaction of $\frac{8}{0.20 \text{ g}}$, 0.68 mmol), *O*-tolylboronic acid (0.11 g, 0.82 mmol), 2M aq. K_2CO_3 (0.19 g, 1.36 mmol) and Pd(PPh₃)₄ (2.5 mol%) according to method described for the synthesis of **9a** within 18 h afforded **9f** (243, 98%) as white solid. ¹H NMR (400 MHz, CDCl₃) δ 7.31 (d, J = 7.9 Hz, 1H), 7.17 (m, 2H), 6.93 (m, 1H), 6.56 (dd, J = 7.4, 1.6 Hz, 1H), 6.06 (t, $J = 7.2$ Hz, 1H), 5.18 (s, 1H), 3.76 (s, 1H), 2.20 (s, 1H). ¹³C NMR (101 MHz, CDCl3) δ 158.1, 150.2, 141.5, 141.3, 135.2, 135.0, 130.3, 129.7, 129.5, 128.1, 127.9, 127.3, 125.8, 112.1, 105.1, 55.8, 51.8, 20. 5. HRMS (EI) calcd for $C_{20}H_{19}NO_2$ [M]⁺ 305.1416 found 305.1419.

1-(4-Dimethylamino-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-one (9g)

The reaction of **8** (0.25g, 0.85 mmol), (4-(dimethylamino)phenyl)boronic acid (0.17g, 1.02 mmol), 2M aq. K_2CO_3 (0.23g, 1.69 mmol), Pd(PPh₃)₄ (2.5 mol%) according to method described for the synthesis of **9a** within 18 h afforded **9g** (230 mg, 81 %) as white solid. ¹H NMR (400 MHz, CDCl₃) δ 7.46 (m, 4H), 7.29 (m, 2H), 6.86 (dd, J = 7.2, 2.0 Hz, 1H), 7.73 $(m, 2H)$, 6.52 (dd, J = 7.2, 1.6 Hz, 1H), 6.01 (t, J = 7.2 Hz, 1H), 5.13 (s, 2H), 3.76 (s, 3H), 2.93 (s, 6H). 13C NMR (100 MHz, CDCl3) δ 157.9, 149.9, 149.7, 140.6, 133.8, 128.4, 128.0, 127.7, 127.3, 126.2, 112.4, 111.8, 104.7, 55.5, 51.3, 40.2. HRMS (EI) calcd for $C_{21}H_{22}N_2O_2$ [M]⁺ 334.1681 found 334.1684.

1-(4-(6-(Dimethylamino)pyridin-3-yl)benzyl))-3-methoxyoxypyridin-2-one (9h)

The reaction of **8** (0.43g, 1.44 mmol), (6-(dimethylamino)pyridine-3-yl)boronic acid (0.2 g, 1.20 mmol), 2M aq. K_2CO_3 (0.33g, 2.41 mmol), Pd(PPh₃)₄ (2.5 mol%) according to method described for the synthesis of **9a** within 18 h afforded **9h** (335 mg, 83%) as white solid. 1H NMR (400 MHz, CDCl₃) δ 8.28 (d, *J* = 2.3 Hz, 1H), 7.51 (dd, *J* = 8.8, 2.5 Hz, 1H), 7.33 (d, $J = 8.2$ Hz, 2H), 7.22 (d, $J = 8.2$ Hz, 2H), 6.80 (dd, $J = 6.9$, 1.6 Hz, 1H), 6.44 (m, 2H), 5.94 (t, $J = 7.2$ Hz, 1H), 5.04 (s, 2H), 3.66 (s, 3H), 2.97 (s, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 158.1, 157.6, 149.7, 145.5, 137.8, 135.1, 134.2, 128.4, 127.5, 125.7, 123.2, 111.6, 105.2, 104.6, 55.4, 51.2, 37.7. HRMS (EI) calcd for $C_{20}H_{21}N_3O_2$ [M]⁺ 335.1634 found 335.1635.

1-(4-(Pyridin-4-yl)benzyl)-3-methoxy-pyridin-2-one (9i)

The reaction of **8** (0.25g, 0.85 mmol), pyrdin-4-ylboronic acid (0.12g, 1.02 mmol), 2M aq. K_2CO_3 (0.23g, 1.69 mmol), Pd(PPh₃)₄ (2.5 mol%) according to method described for the synthesis of **9a** within 18 h afforded **9i** (203 mg, 82%) as white solid. ¹H NMR (400 MHz, CDCl₃) δ 8.54 (dd, J = 4.5, 1.6 Hz, 2H), 7.48 (m, 2H), 7.36 (m, 4H), 6.87 (dd, J = 6.9, 1.7 Hz, 1H), 6.53 (dd, $J = 7.4$, 1.6 Hz, 1H), 6.03 (t, $J = 7.2$ Hz, 1H), 5.13 (s, 2H), 3.72 (s, 3H). 13C NMR (100 MHz, CDCl3) δ 157.8, 150.3, 150.0, 149.9, 147.4, 137.3, 137.3, 128.6, 127.7, 127.0, 121.2, 111.9, 105.0, 55.6, 51.4. HRMS (EI) calcd for $C_{18}H_{16}N_2O_2$ [M]⁺ 292.1212 found 292.1205.

Representative procedure for Thionation reaction. Synthesis of 10-1-(4-Cyano-(1,1' biphenylmethyl))-3-methoxyoxypyridin-2-thione (10a)

A suspension of **9a** (0.13 g, 0.42 mmol) and Lawesson's reagent (0.10 g, 0.25 mmol) in toluene (10 mL) was heated at reflux overnight. The reaction mixture was cooled to room temperature and solvent was evaporated in vacuo. The residue was purified on prep-TLC, eluting with CHCl3: Acetone: EtOH (12:1:0.2) to give **10a** (123 mg, 88%) as yellow solid. ¹H NMR (CDCl₃, 400 MHz) δ 7.60 (m, 4H), 7.40 (m, 4H), 6.90 (dd, $J = 6.8$, 1.6 Hz, 1H), 6.55 (dd, J = 7.6, 1.6 Hz, 1H), 6.07 (t, J = 7.2 Hz, 1H), 5.16 (s, 2H), 3.75 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 158.1, 150.3, 145.0, 138.6, 137.1, 132. 6, 132.0, 128.9, 128.6, 128.4, 128.0, 127.6, 127.5, 118.9, 112.1, 110.9, 105.2, 56.2, 51.8 HRMS (EI) calcd for $C_{20}H_{16}N_2OS$ [M]⁺ 332.0983 found 332.0987.

1-(3-Cyano-(1,1'-biphenylmethyl))-3-methoxypyridin-2-thione (10b)

The reaction of $9b$ (0.11 g, 0.36 mmol) and Lawesson's reagent (0.09 g, 0.22 mmol) in toluene according to method described for the synthesis of **10a** afforded **10b** (113 mg, 95%) as yellow solid. ¹H NMR (400 MHz, DMSO- d_{θ}) δ 7.92 (m, 3H), 7.84 (m, 2H), 7.71 (d, J = 8.4 Hz, 2H), 7.36 (d, $J = 8.8$ Hz, 2H), 7.00 (m, 1H), 6.80 (m, 1H), 5.95 (s, 2H), 3.78 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 173.4, 159.3, 144.8, 138.8, 135.8, 132.6, 131.8, 128.7, 127.6, 127.5, 118.8, 111.8, 111.0, 109.7, 58.6, 56.7. HRMS (EI) calcd for $C_{20}H_{16}N_2OS$ [M]+ 332.0983 found 332.0984.

1-(2-Cyano-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-thione (10c)

The reaction of **9c** (0.09 g, 0.28 mmol) and Lawesson's reagent (0.07 g, 0.17 mmol) in toluene according to method described for the synthesis of **10a** afforded **10c** (72 mg, 77%) of yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 7.74 (m, 1H), 7.63 (td, J = 7.7, 1.4 Hz, 1H), 7.45 (m, 2H), 6.71 (dd, $J = 7.8$, 1.3 Hz, 1H), 6.62 (dd, $J = 7.8$, 6.6 Hz, 1H), 6.00 (s, 1H), 5.28 (s, 1H), 3.93 (s, 1H). 13C NMR (101 MHz, CDCl3) δ 173.4, 159.2, 144.7, 137.8, 135.8, 133. 7, 132.9, 132.0, 130.0, 129.2, 128.3, 127.7, 118.6, 111.8, 111.0, 109.8, 58.6, 56.7. HRMS (EI) calcd for $C_{20}H_{16}N_2OS$ [M]⁺ 332.0983 found 332.0981.

1-(4-Methyl-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-thione (10d)

The reaction of **9d** (0.12 g, 0.39 mmol) and Lawesson's reagent (0.09 g, 0.23 mmol) in toluene according to method described for the synthesis of **10a** afforded **10d** (117 mg, 94%) as yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 7.51 (m, 2H), 7.42 (m, 2H), 7.34 (m, 3H), 7.21 (dd, $J = 8.4$, 0.6 Hz, 2H), 6.66 (dd, $J = 7.8$, 1.2 Hz, 1H), 6.55 (dd, $J = 7.8$, 6.6 Hz, 1H), 5.92 (s, 2H), 3.89 (s, 3H), 2.36 (s, 3H). 13C NMR (100 MHz, CDCl3) δ 173.1, 159.0, 140.7, 137.3, 137.1, 133.8, 131.7, 129.3, 128.4, 127.1, 126.6, 111.6, 109.7, 58.5, 56.6, 20.9. HRMS (EI) calcd for $C_{20}H_{19}NOS$ [M]⁺ 321.1187 found 321.1192.

1-(3-Methyl-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-thione (10e)

The reaction of **9e** (0.12 g, 0.37 mmol) and Lawesson's reagent (0.09 g, 0.23 mmol) in toluene according to method described for the synthesis of **10a** afforded **10e** (112 mg, 93%) as yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 7.53 (m, 1H), 7.33 (m, 2H), 7.14 (dd, J = 7.1, 0.6 Hz, 1H), 6.67 (dd, $J = 7.8$, 1.2 Hz, 1H), 6.56 (m, 1H), 5.94 (s, 1H), 3.91 (s, 1H), 2.39 (s, 1H). 13C NMR (101 MHz, CDCl3) δ 173.2, 159.1, 141.1, 140.3, 138.2, 134.1, 131.7, 128.6, 128.5, 128.1, 127.7, 127.4, 124.0, 111.6, 109.7, 58.6, 56.6, 21.4. HRMS (EI) calcd for $C_{20}H_{19}NOS$ [M]⁺ 305.1187 found 321.1188.

1-(2-Methyl-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-thione (10f)

The reaction of $9f(0.14 \text{ g}, 0.45 \text{ mmol})$ and Lawesson's reagent $(0.11 \text{ g}, 0.27 \text{ mmol})$ in toluene according to method described for the synthesis of **10a** afforded **10f** (118 mg, 86%) as yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 7.42 (dd, J = 6.6, 1.0 Hz, 1H), 7.22 (m, 3H), 6.68 (d, J = 7.3 Hz, 1H), 6.59 (m, 1H), 5.96 (s, 1H), 3.90 (s, 1H), 2.22 (s, 1H). ¹³C NMR (101 MHz, CDCl3) δ 159.0, 141.5, 140.9, 135.0, 133.6, 131.8, 130.1, 129.5, 129.4, 127.7, 127.2, 125.6, 111.6, 109. 7, 58. 6, 56.6, 20.3. HRMS (EI) calcd for $C_{20}H_{19}NOS$ [M]⁺ 305.1187 found 321.1189.

1-(4-Dimethylamino-(1,1'-biphenylmethyl))-3-methoxyoxypyridin-2-thione (10g)

The reaction of **9g** (0.22 g, 0.67 mmol) and Lawesson's reagent (0.16 g, 0.40 mmol) in toluene according to method described for the synthesis of **10a** afforded **10g** (172 mg, 73%) as yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 7.39 (m, 5H), 7.22 (m, 2H), 6.69 (m, 3H), 6.65 (m, 1H), 5.82 (s, 2H), 3.82 (s, 3H), 2.88 (s, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 172.4, 158.7, 149.9, 140. 9, 132.5, 131.8, 128.4, 128.2, 127.3, 126.3, 112.7, 112.0, 110.2, 58.7, 56.3, 40.2. HRMS (EI) calcd for $C_{21}H_{22}N_2OS$ [M]⁺ 350.1453 found 350.1451.

1-(4-(6-(Dimethylamino)pyridin-3-yl)benzyl))-3-methoxyoxypyridin-2-thione (10h)

The reaction of **9h** (0.14 g, 0.42 mmol) and Lawesson's reagent (0.10 g, 0.25 mmol) in toluene according to method described for the synthesis of **10a** afforded **10h** (130 mg, 88%) as yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 8.32 (d, $J = 2.3$ Hz, 1H), 7.56 (dd, $J = 8.6$, 2.1 Hz, 1H), 7.36 (dd, $J = 19.8$, 7.3 Hz, 3H), 7.26 (d, $J = 8.0$ Hz, 2H), 6.61 (d, $J = 7.8$ Hz, 1H), 6.50 (m, 2H), 5.85 (s, 2H), 3.83 (s, 3H), 3.03 (s, 6H). 13C NMR (100 MHz, CDCl3) δ 172.7, 158.8, 158.3, 145.7, 138.1, 135.3, 133.1, 131.7, 128.5, 126.0, 123.2, 111.6, 109.6, 105.4, 58.4, 56.5, 37.9. HRMS (EI) calcd for C₂₀H₂₁N₃OS [M]⁺ 351.1405 found 351.1405.

1-(4-(Pyridin-4-yl)benzyl)-3-methoxypyridin-2-thione (10i)

The reaction of $9i$ (0.13 g, 0.45 mmol) and Lawesson's reagent (0.11 g, 0.27 mmol) in toluene according to method described for the synthesis of **10a** afforded **10i** (104 mg, 76%) as yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 8.59 (d, J = 5.9 Hz, 2H), 7.55 (d, J = 8.3 Hz, 2H), 7.39 (m, 5H), 6.67 (dd, J = 7.8, 1.1 Hz, 1H), 6.58 (dd, J = 7.7, 6.7 Hz, 1H), 5.95 (s, 2H), 3.89 (s, 3H). ¹³C NMR (100 MHz, CDCl₃) δ 173.3, 159.2, 150.1, 147.4, 137. 7, 136.2, 131. 8, 128.6, 127.3, 121.4, 111.7, 109. 7, 58.5, 56.7. HRMS (EI) calcd for $C_{18}H_{16}N_2OS$ [M]+ 308.0983 found 308.0975.

Representative procedure for deprotection of *O***-methyl group. Synthesis of 11-1-(4-Cyano- (1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-one (11a)**

To a solution of $9a$ (0.10 g, 0.32 mmol) in dry CH₂Cl₂ (8 mL) was slowly added 1M BBr₃ (0.35 mL) at −30 °C under argon atmosphere and the reaction mixture was stirred for 32 h at room temperature. The mixture was again cooled to −30 °C and then MeOH (5 mL) was slowly added to quench BBr_3 . Solvent was evaporated off, the residue was adjusted to pH 7

with aqueous 1M NaOH and then extracted with CHCl₃ (3×30 mL). The combined organic layer dried over $Na₂SO₄$ and solvent was evaporated *in vacuo*. The residue was purified by prep-TLC with CHCl3: Acetone: EtOH (10:1:0.2) to give **11a** (89 mg, 94%) as slightly brownish solid. ¹H NMR (400 MHz, DMSO) δ 9.08 (s, 1H), 7.87 (dd, $J = 24.8, 7.9$ Hz, 4H), 7.71 (d, $J = 7.6$ Hz, 2H), 7.40 (d, $J = 7.7$ Hz, 2H), 7.29 (d, $J = 6.7$ Hz, 1H), 6.70 (d, $J = 6.7$ Hz, 1H), 6.13 (t, J = 6.8 Hz, 1H), 5.18 (s, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 147.1, 145.1, 139.2, 136.8, 132.9, 129.0, 128.0, 127. 9, 127.1, 119.1, 114.2, 111.4, 107.6, 52.5, 29.9. HRMS (ESI) calcd for $C_{19}H_{15}N_2O_2$ [M+H]⁺ 303.1128 found 303.1124.

1-(3-Cyano-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-one (11b)

The reaction of **9b** (0.05g, 0.15 mmol) with 1M BBr₃ (0.26 mL) in dry CH₂Cl₂ within 48 h according to the procedure described for the synthesis of **11a** afforded **11b** (44 mg, quantitative) as brownish solid. ¹H NMR (400 MHz, DMSO-d₆) δ 9.08 (s, 1H), 8.11 (s, 1H), 7.98 (d, $J = 7.8$ Hz, 1H), 7.70 (m, 4H), 7.40 (d, $J = 7.7$ Hz, 2H), 7.28 (d, $J = 6.2$ Hz, 1H), 6.70 (d, $J = 6.8$ Hz, 1H), 6.13 (t, $J = 6.7$ Hz, 1H), 5.17 (s, 2H). ¹³C NMR (101 MHz, CDCl₃) δ 146. 8, 141.6, 138.6, 136.1, 131.3, 130.8, 130.5, 129.6, 128.7, 127.5, 126.7, 118.7, 113.9, 113.0, 107.3, 52.3. HRMS (EI) calcd for $C_{19}H_{14}N_2O_2$ [M+H]⁺ 302.1055 found 302.1055.

1-(2-Cyano-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-one (11c)

The reaction of $9c$ (0.072g, 0.15 mmol) with 1M BBr₃ (0.34 mL) in dry CH₂Cl₂ within 48 h according to the procedure described for the synthesis of **11a** afforded **11c** (61 mg, 90%) as brownish solid. ¹H NMR (400 MHz, CDCl₃) δ 7.75 (d, J = 7.6 Hz, 1H), 7.63 (t, J = 7.7 Hz, 1H), 7.53 (d, $J = 7.3$ Hz, 1H), 7.43 (m, 2H), 7.12 (m, 1H), 6.86 (dd, $J = 28.2$, 6.6 Hz, 1H), 6.19 (t, $J = 6.1$ Hz, 1H), 5.25 (s, 1H), ¹³C NMR (101 MHz, CDCl₃) δ 144. 7, 137.9, 136.4, 133.7, 132.9, 130.0, 129.2, 128.2, 127.7, 118. 6, 111.1, 107.2, 52.1. HRMS (EI) calcd for $C_{19}H_{14}N_2O_2$ [M+H]⁺ 302.1055 found 302.1053.

1-(4-Methyl-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-one (11d)

The reaction of **9d** (0.06 g, 0.20 mmol) with $1M$ BBr₃ (0.30 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **11d** (57 mg, quantitative) as brownish solid.¹H NMR (400 MHz, DMSO- d_6) δ 9.07 (s, 1H), 7.57 (m, 2H), 7.50 (d, $J = 8.1$ Hz, 2H), 7.34 (d, $J = 6.8$ Hz, 2H), 7.25 (dd, $J = 14.8$, 7.1 Hz, 3H), 6.69 (d, $J = 7.4$ Hz, 1H), 6.11 (t, $J = 7.1$ Hz, 1H), 5.14 (s, 2H), 2.30 (s, 3H). ¹³C NMR (100 MHz, DMSO-^d6) δ 139.8, 137.3, 137.2, 136.7, 129.9, 128.8, 127.0, 126. 9, 51.7, 21.1. HRMS (EI) calcd for $C_{19}H_{17}NO_2$ [M]⁺ 291.1259 found 291.1250.

1-(3-Methyl-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-one (11e)

The reaction of $9e(0.064 g, 0.20$ mmol) with $1M$ BBr₃ (0.30 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **11e** (60 mg, quantitative) as brownish solid. ¹H NMR (400 MHz, CDCl₃) δ 7.56 (d, J = 7.5 Hz, 1H), 7.33 (m, 2H), 7.17 (d, $J = 7.2$ Hz, 1H), 6.85 (dd, $J = 22.2$, 6.4 Hz, 1H), 6.16 (t, $J = 6.5$ Hz, 1H), 5.23 (s, 1H), 2.42 (s, 1H). ¹³C NMR (101 MHz, CDCl₃) δ 146.7, 141.2, 140.4, 138.3, 134.7, 128.63, 128.4, 128.2, 127.8, 127. 6, 126.6, 124.1, 113.7, 107.0, 52.1, 21. 5. HRMS (EI) calcd for $C_{19}H_{17}NO_2$ [M]⁺ 291.1259 found 291.1263.

1-(2-Methyl-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-one (11f)

The reaction of $9f(0.077 g, 0.25 mmol)$ with 1M BBr₃ (0.38 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **11f** (65 mg, 90%) as brownish solid. ¹H NMR (400 MHz, CDCl₃) δ 7.28 (m, 3H), 6.87 (dd, J = 31.4, 6.8 Hz, 1H), 6.18 (t, J = 6.9 Hz, 1H), 5.27 (d, J = 16.7 Hz, 1H), 2.26 (s, 1H). ¹³C NMR (101 MHz, CDCl3) δ 146.7, 141.8, 141.1, 135.2, 134.3, 130.3, 129.7, 127.7, 127.4, 126.7,

125.8, 113.6, 107.0, 52.2, 20.4. HRMS (EI) calcd for C₁₉H₁₇NO₂ [M]⁺ 291.1259 found 291.1260.

1-(4-Dimethylamino-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-one (11g)

The reaction of $9g(0.10 \text{ g}, 0.30 \text{ mmol})$ with $1M \text{ BBr}_3(0.45 \text{ mL})$ in dry CH₂Cl₂ (8 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **11g** (84 mg, 87%) as brownish solid. ¹H NMR (400 MHz, DMSO- d_0) δ 9.06 (s, 1H), 7.52 (d, J = 8.0 Hz, 3H), 7.28 (m, 4H), 6.76 (m, 2H), 6.10 (t, $J = 5.8$ Hz, 1H), 5.12 (s, 2H), 2.90 (s, 6H). ¹³C NMR (100 MHz, CDCl3) δ 158.7, 150.0, 146.6, 141.1, 135.8, 133.3, 128.8, 128.5, 128.2, 128.1, 128.0, 127.6, 126.6, 113.5, 112.6, 106. 9, 52.1, 40.4. HRMS (EI) calcd for $C_{20}H_{20}N_2O_2$ [M]⁺ 320.1524 found 320.1512.

1-(4-(6-(Dimethylamino)pyridin-3-yl)benzyl))-3-hydroxyoxypyridin-2-one (11h)

The reaction of **9h** (0.10 g, 0.29 mmol) with 1M BBr₃ (0.45 mL) in dry CH₂Cl₂ (7 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **11h** (67 mg, 83%) as brownish solid. ¹H NMR (400 MHz, CDCl₃) δ 8.40 (s, 1H), 7.66 (d, $J = 8.0$) Hz, 1H), 7.49 (d, $J = 7.7$ Hz, 2H), 7.33 (d, $J = 7.7$ Hz, 2H), 6.84 (dd, $J = 27.3$, 6.8 Hz, 3H), 6.57 (d, $J = 8.7$ Hz, 1H), 6.15 (t, $J = 7.0$ Hz, 1H), 5.20 (s, 2H), 3.12 (s, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 158.5, 146. 7, 145.7, 138.3, 135.8, 133. 9, 132.0, 128.5, 126.8, 126.3, 123.6, 114.2, 107.2, 105.9, 52.1, 38.2. HRMS (EI) calcd for $C_{19}H_{19}N_3O_2$ [M]⁺ 321.1477 found 321.1479.

1-(4-(Pyridin-4-yl)benzyl)-3-hydroxypyridin-2-one (11i)

The reaction of $9i$ (0.10 g, 0.34 mmol) with 1M BBr₃ (0.51 mL) in dry CH₂Cl₂ within 48 h according to the procedure described for the synthesis of **11a** afforded **11i** (79 mg, 83%) as brownish solid. ¹H NMR (400 MHz, CDCl₃) δ 8.65 (s, 2H), 7.46 (s, 2H), 7.39 (d, $J = 7.9$ Hz, 2H), 6.83 (dd, $J = 22.5$, 7.0 Hz, 2H), 6.16 (t, $J = 7.0$ Hz, 1H), 5.22 (s, 2H). ¹³C NMR (100 MHz, CDCl3) δ 150.2, 147.6, 146.8, 138.0, 136.9, 128.7, 127.5, 126.6, 121.6, 113.7, 107.1, 52.2. HRMS (FAB) calcd for $C_{17}H_{15}N_2O_2$ [M+H]⁺ 279.1133 found 279.1147.

1-(4-Cyano-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-thione (12a)

The reaction of $10a$ (0.10 g, 0.30 mmol) with $1M$ BBr₃ (0.33 mL) in dry CH₂Cl₂ (8 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **12a** (84 mg, 88%) as green solid. Retention time 16.58 min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 5 min). ¹H NMR (400 MHz, DMSO-d₆) δ 7.87 (m, 5H), 7.74 (d, $J = 8.1$ Hz, 2H), 7.37 (d, $J = 8.1$ Hz, 2H), 7.04 (m, 1H), 6.90 (d, $J = 8.1$ Hz, 1H), 5.87 (s, 2H). 13C NMR (100 MHz, CDCl3) δ 164.5, 144.7, 139.2, 134.6, 132.6, 128. 6, 127.7, 127.6, 118.8, 111.2, 61.8, 29.6 .HRMS (EI) calcd for C₁₉H₁₅N₂OS [M]⁺ 318.0827 found 318.0828.

1-(3-Cyano-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-thione (12b)

The reaction of **10b** (0.07 g, 0.21 mmol) with $1M$ BBr₃ (0.32 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **12b** (53 mg, 79%) as green solid. Retention time 6.93 min (solvent gradient: 70 to 90% solvent B in 10 min and constant 90% of B for 15 min). ¹H NMR (400 MHz, CD₃OD) δ 7.89 (m, 2H), 7.65 (m, 5H), 7.44 (d, $J = 8.2$ Hz, 2H), 7.02 (d, $J = 7.7$ Hz, 2H), 6.77 (m, 1H), 5.90 (s, 2H). ¹³C NMR (100 MHz, CD₃OD) δ 143.0, 140.0, 132.8, 132.2, 131.7, 131.1, 130.0, 130.0, 129. 9, 128.7, 128.6, 119.8, 113.9, 106.4, 54. 8. HRMS (EI) calcd for $C_1 \rho H_{14} N_2 O_2$ [M+H]+ 318.0827 found 318.0827

1-(2-Cyano-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-thione (12c)

The reaction of $10c$ (0.042 g, 0.21 mmol) with $1M$ BBr₃ (0.19 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **12c** (31 mg, 78%) as green solid. Retention time 12.37 min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 5 min). ¹H NMR (400 MHz, CD₃OD) δ 7.77 (dd, J = 7.7, 0.9 Hz, 1H), 7.67 (m, 2H), 7.48 (m, 4H), 7.01 (d, $J = 13.1$ Hz, 1H), 6.70 (m, 1H), 5.90 (s, 1H). ¹³C NMR (101 MHz, CD₃OD) δ 145.3, 138.9, 134.4, 133.9, 130.8, 129.9, 128.9, 128.6, 119.2, 115.1, 111.4. HRMS (EI) calcd for $C_{19}H_{14}N_2O_2$ [M+H]⁺ 318.0821 found 318.0827

1-(4-Methyl-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-thione (12d)

The reaction of **10d** (0.06 g, 0.20 mmol) with $1M$ BBr₃ (0.30 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **12d** (45 mg, 73%) as green solid. Retention time 11.53 min (solvent gradient: 70 to 90% solvent B in 10 min and constant 90% of B for 15 min). ¹H NMR (400 MHz, CD₃OD) δ 7.70 (d, $J = 1.1$ Hz, 1H), 7.63 (d, $J = 6.3$ Hz, 1H), 7.55 (d, $J = 8.2$ Hz, 2H), 7.40 (m, 3H), 7.21 (d, $J = 8.1$ Hz, 2H), 7.03 (dd, J = 14.9, 7.8 Hz, 2H), 6.75 (t, J = 7.0 Hz, 1H), 5.87 (s, 2H), 2.35 (s, 3H). ¹³C NMR (100 MHz, CD3OD) δ 141.4, 137.15, 133.66, 131.83, 129.27, 128.43, 127.05, 126.59, 113.79, 112.26, 59.63, 20.32. HRMS (EI) calcd for $C_{19}H_{17}NOS$ [M]⁺ 307.1031 found 307.1022.

1-(3-Methyl-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-thione (12e)

The reaction of $10e$ (0.08 g, 0.20 mmol) with $1M$ BBr₃ (0.36 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **12e** (51 mg, 70%) as green solid. Retention time 10.52 min (solvent gradient: 70 to 90% solvent B in 10 min and constant 90% of B for 15 min). ¹H NMR (400 MHz, CD₃OD) δ 7.53 (m, 1H), 7.30 (m, 2H), 7.12 (d, $J = 7.2$ Hz, 1H), 6.97 (m, 1H), 6.68 (m, 1H), 5.81 (s, 1H), 2.36 (s, 1H). ¹³C NMR (101 MHz, CD₃OD) δ 141.5, 140.2, 138.3, 128.6, 128.5, 128.2, 127.7, 127.5, 124.0, 21.2. HRMS (EI) calcd for C19H17NOS [M]+ 307.1031 found 307.1031.

1-(2-Methyl-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-thione (12f)

The reaction of $10f(0.08 g, 0.20 mmol)$ with $1M BBr₃(0.36 mL)$ in dry $CH₂Cl₂(5 mL)$ within 48 h according to the procedure described for the synthesis of **11a** afforded **12f** (49 mg, 67%) as green solid. Retention time 17.67 min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 5 min). 1H NMR (400 MHz, CD3OD) δ 7.54 (m, 1H), 7.19 (m, 3H), 6.98 (m, 1H), 6.70 (m, 1H), 5.83 (s, 1H), 2.19 (s, 1H). 13C NMR (100 MHz, CD3OD) δ 142.1, 141.0, 135.1, 130.2, 129.6, 129.5, 127.8, 127.3, 125.7, 20.0. HRMS (EI) calcd for $C_{19}H_{17}NOS$ [M]⁺ 307.1031 found 307.1034.

1-(4-Dimethylamino-(1,1'-biphenylmethyl))-3-hydroxyoxypyridin-2-thione (12g)

The reaction of $10g$ (0.11 g, 0.33 mmol) with $1M$ BBr₃ (0.39 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **12g** (68 mg, 62%) as green solid. Retention time 10.42 min (solvent gradient: 70 to 90% solvent B in 10 min and constant 90% of B for 15 min). ¹H NMR (400 MHz, CDCl₃) δ 8.57 (d, $J = 8.5$ Hz, 1H), 7.53 (d, $J = 8.2$ Hz, 2H), 7.46 (m, 2H), 7.33 (d, $J = 8.0$ Hz, 3H), 6.97 (d, $J = 6.8$ Hz, 1H), 6.77 (d, J = 8.8 Hz, 2H), 6.62 (m, 1H), 5.79 (s, 2H), 2.98 (s, 6H). ¹³C NMR (100 MHz, CDCl3) δ 169.7, 155.1, 150.1, 141.5, 131.9, 130.8, 129.0, 128.8, 128.0, 127.6, 126.7, 113.6, 112.6, 111.9, 60.1, 40.5. HRMS (EI) calcd for $C_{20}H_{20}N_2OS$ [M]⁺ 336.1296 found 336.1295.

1-(4-(6-(Dimethylamino)pyridin-3-yl)benzyl))-3-hydroxyoxypyridin-2-thione (12h)

The reaction of **10h** (0.064 g, 0.18 mmol) with $1M$ BBr₃ (0.27 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **12h** (48 mg, 79%) as green solid. Retention time 27.43 min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 15 min). ¹H NMR (400 MHz, CDCl₃) δ 8.30 (s, 1H), 7.70 (d, $J = 8.9$ Hz, 1H), 7.46 (s, 2H), 7.31 (t, $J = 7.6$ Hz, 3H), 6.98 (d, $J = 7.8$ Hz, 1H), 6.67 (m, 2H), 5.75 (s, 2H), 3.11 (s, 8H). 13C NMR (100 MHz, CDCl3) δ 157.0, 143. 5, 142.2, 137.7, 137.0, 132.8, 128.8, 127. 6, 126.4, 123.7, 114.4, 107.0, 38.5, 29.5. HRMS (EI) calcd for $C_{19}H_{19}N_3Os [M]+337.1249$ found 337.1251.

1-(4-(Pyridin-4-yl)benzyl)-3-hydroxypyridin-2-thione (12i)

The reaction of **10i** (0.08 g, 0.25 mmol) with $1M$ BBr₃ (0.38 mL) in dry CH₂Cl₂ (5 mL) within 48 h according to the procedure described for the synthesis of **11a** afforded **12i** (65 mg, 88%) as green solid. Retention time 19.37min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 10 min). ¹H NMR (400 MHz, CDCl₃) δ 8.65 (d, J = 5.3 Hz, 2H), 8.55 (s, 1H), 7.62 (d, $J = 8.0$ Hz, 2H), 7.42 (m, 5H), 6.99 (d, $J = 7.5$ Hz, 1H), 6.67 (t, J $= 7.0$ Hz, 1H), 5.85 (s, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 170.1, 155.3, 150.3, 147.4, 138.2, 135.5, 130.9, 128.8, 127.6, 121.5, 113.7, 111.9, 59.9. HRMS (EI) calcd for $C_{17}H_{14}N_2OS$ [M]⁺ 294.0827 found 294.0823.

1-Propargyl-3-methoxypyridin-2-one (13)

To a stirring solution of 3-methoxy-2-pyridinone $(1.00 \text{ g}, 8.00 \text{ mmol})$ in DMF (25 mL) was added propargyl bromide (80 wt% solution in toluene, 1.5 equiv) and K_2CO_3 (3.313 g, 24 mmol). The reaction mixture was heated overnight at 100 °C and allowed to cool to room temperature. The reaction was partitioned between EtOAc (120 mL) and water (100 mL). The organic layer was separated, washed repeatedly with water (7×100 mL), brine (60 mL) and dried on $Na₂SO₄$. Solvent was evaporated *in vacuo* and the dark brown crude was purified by column chromatography using a gradient of $0-20\%$ acetone in CH₂Cl₂ to give **13** (0.79 g, 60%) as brownish solid. ¹H NMR (CDCl₃, 400 MHz) δ 2.36 (1H, t, $J = 2.8$), 3.67 (3H, m), 4.66 (1H, d, $J = 2.8$), 6.05 (1H, t, $J = 7.2$), 6.50 (1H, dd, $J = 1.6, 7.2$), 7.11 (1H, dd, $J = 1.6$, 6.8); ¹³C NMR (CDCl₃, 100 MHz) δ 37.2, 55.5, 74.6, 77.3, 104.8, 112.0, 126.3, 149.4, 157.1. HRMS (EI) calcd for C₉H₉NO₂ [M]⁺ 163.0633 found 163.0636

1-Phenyltriazolylmethyl-3-methoxypyridin-2-one (14a)

Compound 13 $(0.32 \text{ g}, 1.95 \text{ mmol})$ and phenylazide $(0.35 \text{ g}, 2.93 \text{ mmol})$ were dissolved in anhydrous THF (10 mL) and stirred under argon at room temperature. Copper (I) iodide (0.01 g, 0.07 mmol) and Hunig's base (0.1 mL) were added to the reaction mixture, and stirring continued for 4 h. The reaction mixture was diluted with CH_2Cl_2 (40 mL), washed with 1:4 NH₄OH/saturated NH₄Cl (3×30 mL) and saturated NH₄Cl (30 mL). The organic layer was dried over Na_2SO_4 and concentrated *in vacuo*. The crude product was triturated with hexanes to give **14a** (510 mg, 92%) as white solid.¹H NMR (CDCl₃, 400 MHz) δ 3.58 $(3H, s), 5.11$ $(2H, s), 5.93$ $(1H, t, J = 7.6), 6.42$ $(1H, d, J = 7.2), 7.07$ $(1H, d, J = 6.4), 7.16-$ 7.20 (1H, m), 7.26 (2H, t, $J = 7.6$), 7.49 (2H, d, $J = 8.0$); ¹³C NMR (CDCl₃, 100 MHz) δ 44.2, 55.2, 104.7, 112.0, 119.7, 121.8, 127.9, 128.1, 129.0, 136.2, 142.8, 149.3, 157.2. HRMS (EI) calcd for $C_{15}H_{14}N_4O_2$ [M]⁺ 282.1117 found 282.1115.

1-(4-Dimethylamino)phenylltriazolylmethyl-3-methoxypyridin-2-one (14b)

The reaction of 4-azido-N,N-dimethylaniline $(0.09 \text{ g}, 0.61 \text{ mmol})$ and **13** $(0.10 \text{ g}, 0.61 \text{ m})$ mmol) within 4 h as described for the synthesis of **14a** gave compound **14b** (0.13 g, 68 %) as white solid. ¹H NMR (400 MHz, CDCl₃) δ 8.04 (s, 1H), 7.41 (m, 2H), 7.15 (dd, J = 6.9, 1.7 Hz, 1H), 6.62 (m, 2H), 6.51 (dd, $J = 7.5$, 1.6 Hz, 1H), 6.03 (m, 1H), 5.19 (s, 2H), 3.70 (s, 3H), 2.89 (s, 6H). 13C NMR (100 MHz, CDCl3) δ 157.5, 150.2, 149.7, 142.6, 128.2, 126.2, 121.9, 121.4, 112.2, 111.8, 104.9, 55.6, 44.5, 40.1. HRMS (EI) calcd for $C_{17}H_{149}N_5O_2$ [M]⁺ 325.1539 found 325.1544.

1-Phenyltriazolylmethyl-3-hydroxypyridin-2-one (15a)

The reaction of **14a** (0.21 g, 0.75 mmol) and $1M$ BBr₃ in CH₂Cl₂ (1.5 equiv) withing 48 h as described for the synthesis of **11a** gave compound **15a** (0.123 g, 61%) as brown solid. ¹H NMR (DMSO, d_6 , 400 MHz) δ 5.27 (2H, s), 6.13 (1H, t, J = 6.4), 6.67 (1H, d, J = 7.2), 7.30 $(1H, d, J = 5.6), 7.45-7.49$ $(1H, m), 7.57$ $(2H, t, J = 7.6), 7.87$ $(2H, d, J = 8.0), 8.73$ $(1H, s),$ 9.08 (1H, s); ¹³C NMR (101 MHz, CDCl₃) δ 136.6, 129.6, 128.9, 122.3, 120.6, 108.0, 29.3. HRMS (EI) calcd for $C_{14}H_{12}N_4O_2$ [M]⁺ 268.0960 found 268.0967.

1-(4-Dimethylamino)phenylltriazolylmethyl-3-hydroxypyridin-2-one (15b)

The reaction of **14b** (0.045 g, 0.14 mmol) and $1M$ BBr₃ in CH₂Cl₂ (0.22 mL, 1.5 equiv) withing 48 h as described for the synthesis of **11a** gave compound **15b** (0.033 g, 77%) as brown solid. ¹H NMR (400 MHz, cdcl₃) δ 8.08 (s, 1H), 7.48 (dd, J = 19.9, 8.4 Hz, 2H), 7.18 $(d, J = 6.7 \text{ Hz}, 1\text{ H})$, 6.69 (ddd, $J = 25.8$, 24.8, 7.5 Hz, 4H), 6.16 (t, $J = 6.9 \text{ Hz}, 1\text{ H}$), 5.31 (s, 2H), 2.98 (m, 6H). 13C NMR (101 MHz, CDCl3) δ 158.4, 150.5, 146.7, 142.5, 127.1, 126.4, 122.1, 121.9, 121.8, 112.1, 107.1, 44.7, 40.3. HRMS (EI) calcd for $C_{16}H_{17}N_5O_2$ [M]⁺ 311.1382 found 311.1386.

1-Phenyltriazolylmethyl-3-methoxypyridin-2-thione (16a)

The reaction of $14a$ (0.30 g, 1.04 mmol) and Lawesson's reagent (0.25 g, 0.62 mmol) in toluene (15 mL) within 12 h as described for the synthesis of **10a** gave compound **16a** (0.29 g, 94%) as yellow solid. ¹H NMR (CDCl₃, 400 MHz) δ 3.90 (3H, s), 6.04 (2H, s), 6.62–6.68 $(2H, m)$, 7.38–7.42 (1H, m), 7.48 (2H, t, J = 7.2), 7.68 (2H, d, J = 8.0), 7.82 (1H, d, J = 4.8), 8.53 (1H, s); 13C NMR (CDCl3, 100 MHz) δ 50.9, 56.4, 110.0, 112.0, 120.2, 122.4, 128.5, 129.3, 132.2, 136.4, 141.9, 158.7, 171.5. HRMS (EI) calcd for C₁₅H₁₄N₄O₂ [M]⁺ 282.1117 found 282.1115. HRMS (EI) calcd for $C_{15}H_{14}N_4OS$ [M]⁺ 298.0888 found 298.0888.

1-(4-Dimethylamino)phenylltriazolylmethyl-3-methoxypyridin-2-thione (16b)

The reaction of **14b** (0.08 g, 0.25 mmol) and Lawesson's reagent (0.06 g, 0.15 mmol) in toluene (8 mL) within 12 h as described for the synthesis of **10a** gave compound **16b** (0.07 g, 84%) as yellow solid. ¹H NMR (400 MHz, cdcl₃) δ 8.36 (s, 1H), 7.81 (dd, J = 6.4, 1.5 Hz, 1H), 7.46 (m, 2H), 6.66 (m, 4H), 6.01 (s, 2H), 3.87 (s, 3H), 2.97 (s, 6H). 13C NMR (100 MHz, CDCl3) δ 171.8, 158.9, 150.4, 141.5, 132.4, 126.3, 122.5, 121.8, 112.1, 112.0, 110.1, 56.7, 51.2, 40.3. HRMS (EI) calcd for $C_{17}H_{19}N_5OS$ [M]⁺ 314.1310 found 314.1304.

1-Phenyltriazolylmethyl-3-hydroxypyridin-2-thione (17a)

The reaction of $16a$ (0.28 g, 0.93 mmol) and $1M$ BBr₃ in CH₂Cl₂ (1.5 equiv) within 48 h as described for the synthesis of **11a** gave compound **17a** (0.18 g, 67%) as green solid. Retention time 8.63 min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 5 min). ¹H NMR (400 MHz, CDCl₃) δ 8.43 (d, *J* = 16.6 Hz, 1H), 8.39 (s, 1H), 7.79 $\rm (dd, J = 6.6, 1.3 Hz, 1H), 7.69$ (m, 2H), 7.46 (m, 3H), 6.97 (dd, $J = 7.7, 1.2$ Hz, 1H), 6.69 (dd, $J = 7.6$, 6.8 Hz, 1H), 5.94 (s, 2H). ¹³C NMR (100 MHz, CDC₃) δ 168.6, 155.1, 141.6, 136.7, 131.7, 129.7, 129.0, 122.6, 120.6, 114.0, 112.5, 77.3, 77.0, 76. 7, 52.1. HRMS (EI) calcd for $C_{14}H_{12}N_4OS$ [M]⁺ 284.0730 found 284.0732.

1-(4-Dimethylamino)phenylltriazolylmethyl-3-hydroxypyridin-2-thione (17b)

The reaction of **16b** (0.07 g, 0.21 mmol) and $1M$ BBr₃ in CH₂Cl₂ (0.64 mL, 1.5 equiv) within 48 h as described for the synthesis of **11a** gave compound **17b** (0.03 g, 48%) as green solid. Retention time 8.43 min (solvent gradient: 40 to 80% solvent B in 20 min then constant 80% B for 5 min). ¹H NMR (400 MHz, CDCl₃) δ 8.42 (s, 1H), 8.29 (d, J = 11.5 Hz, 1H), 7.79 (d, $J = 6.4$ Hz, 1H), 7.50 (m, 2H), 6.98 (dd, $J = 36.5$, 28.9 Hz, 2H), 6.72 (m, 3H), 5.94 (s, 2H), 3.01 (d, $J = 2.7$ Hz, 6H). ¹³C NMR (100 MHz, CDCl₃) δ 155.1, 150.7, 141.1, 131.7, 127.7, 126.3, 122. 5, 122.0, 114.0, 112.5, 112.2, 52.2, 40.4. HRMS (EI) calcd for C_1 ₅H₁₇N₅OS [M]⁺ 327.1154 found 327.1154.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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(a) HDACi Pharmacophoric model (b) Representative examples of HDACi (Note color code highlights the three pharmacophores)

Representative examples of bidentate heterocyclic and monodentate non-hydroxamate ZBGs.26,28

Figure 3.

Docked bidendate ZBG fragments **a)** 1-hydroxypyridin-2-thione (**1a**); **b)** 3 hydroxypyridin-4-thione (**1b**); **c)** 3-hydroxypyridin-2-thione (3-HPT) (**2**); **d)** 3 hydroxypyridin-2-one (3,HP) (**3**).

Scheme 1.

Synthesis of the 3-HP and 3HPT based ZBG HDACi. **a**) R-CH₂Br, K₂CO₃, DMF, 100 °C. **b**) Lawesson's reagent, toluene, reflux. **c**) BBr₃, CH₂Cl₂. **d**) MeI, KOH, MeOH, rt.

Figure 4.

Docked structures of 3-HPT diaryl **17a** at the active site of (a) HDAC 6 and (b) HDAC 8. Note that the placement of the triazole ring of **17a** could facilitate H-bonding interaction with the OH-group of HDAC 8 Tyr306.

Scheme 2.

Synthesis of the biphenyl 3-HP and 3-HPT based HDACi. (a) 4-bromobenzylbromide, K₂CO, THF, reflux b) R-B(OH)₂, Pd(PPh₃)₄, K₂CO₃, toluene:ethanol:H₂O; 80 °C c) Lawesson's reagent, toluene, reflux; d) BBr_3 , CH_2Cl_2 .

Scheme 3.

Synthesis of the triazole 3-HP and 3-HPT based HDACi. (a) propargyl bromide, K_2CO_3 , THF, reflux b) azido intermediates (phenyl azide, 4-azido-N, N-dimethylaniline), CuI, DIPEA, THF; c) BBr₃, CH₂Cl₂; d) Lawessons' reagent, Toluene, reflux.

Figure 5.

a)

b)

Docked poses adopted by 3-HPT compounds **12a-f** at HDAC 6 active site. (**a**) Overlay of low energy conformations of **12a** (purple), **12b** (green) and **12c** (white), (i) surface view, (ii) ball and stick model showing crucial residues at enzyme outer rim hydrophobic pocket. (**b**) Overlay of low energy conformations of **12d** (orange), **12e** (pink) and **12f** (yellow), (i) surface view, (ii) ball and stick model showing crucial residues at enzyme outer rim hydrophobic pocket.

Figure 6.

Western blot analysis of tubulin acetylation (HDAC6 inhibition) in LNCaP cell line. Lanes - 1. SAHA (20 μM); 2. SAHA IC₅₀ (2.31 μM); 3. Control; 4. **12f** (20 μM); 5. **12f** IC₅₀ (7.75 μM).

Table 1

HDAC inhibition activities of 3-HP and 3-HPT*

* Activities determined by SAMDI analysis

 ϕ [†]No significant Inhibition (below 20% Inhibition) at maximum tested concentration of 10 μM

Table 2

In-vitro HDAC inhibition of phenyl and biphenyl HDACi.

* No significant Inhibition (below 20% Inhibition)

 ϕ inhibition of the compounds at 10 μM are given if the IC50 was above 10 μM

Table 3

Cell Viability Assay. IC_{50} of selected compounds against various cancer cell lines

NT = Not Tested

 $NI = No$ inhibition