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## **Extra-matrix Mg2+ limits Ca2+ uptake and modulates Ca2+ uptake– independent respiration and redox state in cardiac isolated mitochondria**

**Age D. Boelens**1, **Ranjan K. Pradhan**2,3, **Christoph A. Blomeyer**1, **Amadou K. S. Camara**1,4, **Ranjan K. Dash**2,3, and **David F. Stowe**1,2,4,5,6

<sup>1</sup>Department of Anesthesiology, Medical College of Wisconsin, Milwaukee, Wisconsin 53226, USA

<sup>2</sup>Department of Physiology, Medical College of Wisconsin, Milwaukee, Wisconsin 53226, USA

<sup>3</sup>Biotechnology and Bioengineering Center, Medical College of Wisconsin, Milwaukee, Wisconsin 53226, USA

<sup>4</sup>Cardiovascular Research Center, Medical College of Wisconsin, Milwaukee, Wisconsin 53226, USA

<sup>5</sup>Research Service, Zablocki Veterans Affairs Medical Center, Milwaukee, Wisconsin, 53295, USA

<sup>6</sup>Department of Biomedical Engineering, Marquette University, Milwaukee, Wisconsin, 53201, USA

## **Abstract**

Cardiac mitochondrial matrix (m) free  $Ca^{2+}$  ([Ca<sup>2+</sup>]<sub>m</sub>) increases primarily by Ca<sup>2+</sup> uptake through the Ca<sup>2+</sup> uniporter (CU). Ca<sup>2+</sup> uptake via the CU is attenuated by extra-matrix (e) Mg<sup>2+</sup> ([Mg<sup>2+</sup>]<sub>e</sub>). How [Ca<sup>2+</sup>]<sub>m</sub> is dynamically modulated by interacting physiological levels of [Ca<sup>2+</sup>]<sub>e</sub> and  $[Mg^{2+}]_e$  and how this interaction alters bioenergetics is not well understood. We postulated that as  $[Mg^{2+}]_e$  modulates  $Ca^{2+}$  uptake via the CU, it also alters bioenergetics in a matrix  $Ca^{2+}$ induced and matrix Ca<sup>2+</sup>-independent manner. To test this, we measured changes in  $[Ca^{2+}]_e$ ,  $[Ca^{2+}]_{m}$ ,  $[Mg^{2+}]_{e}$  and  $[Mg^{2+}]_{m}$  spectrofluorometrically in guinea pig cardiac mitochondria in response to added CaCl<sub>2</sub> (0–0.6 mM; 1 mM EGTA buffer) with/without added MgCl<sub>2</sub> (0–2 mM). In parallel, we assessed effects of added  $CaCl<sub>2</sub>$  and  $MgCl<sub>2</sub>$  on NADH, membrane potential  $(\Delta \Psi_m)$ , and respiration. We found that 0.125 mM MgCl<sub>2</sub> significantly attenuated CU-mediated Ca<sup>2+</sup> uptake and  $[Ca^{2+}]_{m}$ . Incremental  $[Mg^{2+}]_{e}$  did not reduce initial Ca<sup>2+</sup>uptake but attenuated the subsequent slower  $Ca^{2+}$  uptake, so that  $[Ca^{2+}]$ <sub>m</sub> remained unaltered over time. Adding CaCl<sub>2</sub> without MgCl<sub>2</sub> to attain a  $\text{[Ca}^{2+}\text{]}_{\text{m}}$  from 46 to 221 nM enhanced state 3 NADH oxidation and increased respiration by 15%; up to 868 nM  $[Ca<sup>2+</sup>]_{m}$  did not additionally enhance NADH oxidation or respiration. Adding MgCl<sub>2</sub> did not increase  $[Mg^{2+}]_m$  but it altered bioenergetics by its direct effect to decrease  $Ca^{2+}$  uptake. However, at a given  $[Ca^{2+}]_m$ , state 3 respiration was incrementally attenuated, and state 4 respiration enhanced, by higher [Mg<sup>2+</sup>]<sub>e</sub>. Thus, [Mg<sup>2+</sup>]<sub>e</sub> without a change in  $[Mg^{2+}]_m$  can modulate bioenergetics independently of CU-mediated Ca<sup>2+</sup> transport.

Present address of ADB: Academic Medical Center, University of Amsterdam, Amsterdam, The Netherlands **Disclosures**

Corresponding author: D.F. Stowe, M4240, 8701 Watertown Plank Rd, Medical College of Wisconsin, Milwaukee, WI 53226 (dfstowe@mcw.edu).

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#### **Keywords**

Cardiac mitochondria; Inner mitochondrial membrane; Bioenergetics;  $Ca^{2+}$  uniporter;  $Ca^{2+}$ uptake;  $Mg^{2+}$  inhibition

#### **Introduction**

Mitochondria are not only the major source of ATP production but are also vital to cellular homeostasis and function. Mitochondria must be able to adjust ATP production to meet the cell's ever changing energy demand. A widely held hypothesis is that mitochondrial matrix free Ca<sup>2+</sup> ([Ca<sup>2+</sup>]<sub>m</sub>) modulates ATP production; this is because an increase in [Ca<sup>2+</sup>]<sub>m</sub> was reported to differentially stimulate the activities of three mitochondrial dehydrogenases in the tricarboxylic acid (TCA) cycle (Denton 2009, McCormack et al 1990). In cardiomyocytes, where an increase in extra-matrix free  $Ca^{2+}$  ([Ca<sup>2+</sup>]<sub>e</sub>) is linked to muscle contraction during the process of  $Ca^{2+}$ –induced  $Ca^{2+}$  release by the sarcoplasmic reticulum (SR), an increase in  $\left[Ca^{2+}\right]_m$  would provide an elegant feed-forward signal for mitochondrial bioenergetics to adapt to increased workload. More recent studies also suggest that  $F_1F_0$ -ATPase activity is post-translationally regulated by  $\left[Ca^{2+}\right]_{m}$  (Hubbard et al 1996, Territo et al 2001), although there are no reported binding sites for  $Ca^{2+}$  on  $F_1F_0$ -ATPase (Abrahams et al 1994), and  $Ca^{2+}$  does not directly stimulate  $F_1F_0$ -ATPase (Balaban 2009).

There are other proposed mechanisms that may control mitochondrial energy production, such as changes in ADP/ATP and P<sub>i</sub>, based on computational evaluation (Beard 2006) of original data (Katz et al 1989), as well as a change in [Ca $^{2+}$ ]<sub>e</sub> via Ca $^{2+}$  binding domains on regulatory proteins extending into the intermembrane space (IMS) (Gellerich et al 2010, Satrustegui et al 2007). These mechanisms are not likely mutually exclusive. Because oxidative phosphorylation changes rapidly depending on energy demand, it must be highly regulated; thus it is possible that these several mechanisms are synergistic, with distinct mechanisms dominating under different metabolic conditions.

The main route for matrix  $Ca^{2+}$  influx is the ruthenium red (RR) sensitive  $Ca^{2+}$  uniporter (CU) (Bernardi 1999). Another accessory pathway for  $Ca^{2+}$  uptake has been proposed, i.e. the rapid mode (RaM) of  $Ca^{2+}$  uptake (Buntinas et al 2001, Sparagna et al 1995), which may occur at a different site, or be due to a conformational change in the CU. The structural identity of the CU was recently reported (Baughman et al 2011, De Stefani et al 2011). The CU is a 40 kDa protein that forms oligomers in the inner mitochondrial membrane (IMM) and resides within a large molecular weight complex; the CU has two predicted transmembrane helices. The protein MICU1 is an EF-hand-containing protein (Baughman et al 2011, Perocchi et al 2010) that physically interacts with the CU and serves as a putative sensor of external  $Ca^{2+}$  for its uptake by the CU.

The transport of Ca<sup>2+</sup> via the CU is differently affected by other divalent cations, e.g.  $Mg^{2+}$ ,  $Mn^{2+}$ , as well as by the IMM pH gradient, either directly or indirectly.  $Mg^{2+}$  is an essential and abundant intracellular cation involved in many cellular processes.  $Mg^{2+}$  is well known to antagonize  $Ca^{2+}$  uptake (Denton et al 1980, McCormack et al 1980, Panov et al 1996), specifically by impeding  $Ca^{2+}$  uptake via the CU (Favaron et al 1985). A change in extramatrix Mg<sup>2+</sup> ([Mg<sup>2+</sup>]<sub>e</sub>) might also alter matrix Mg<sup>2+</sup> ([Mg<sup>2+</sup>]<sub>m</sub>) to modulate mitochondrial bioenergetics. Like  $[Ca^{2+}]_{m}$ ,  $[Mg^{2+}]_{m}$  may modulate TCA cycle activity directly, e.g. by effects on mitochondrial dehydrogenases (Rodriguez-Zavala et al 1998); alternatively, this effect of  $Mg^{2+}$  could be due to an indirect effect on  $Ca^{2+}$  uptake. Because the uptake of  $Mg^{2+}$  into mitochondria is normally very slow (Brierley et al 1987), it is unclear how it could have any rapid effects on TCA cycle activity. But  $Mg^{2+}$ –induced regulation of Ca<sup>2+</sup>

uptake may help to attenuate  $Ca^{2+}$  overload and to reduce its pathological consequences (Romani et al 1990). Although  $Mg^{2+}$  can modulate mitochondrial activity indirectly by altering  $[Ca^{2+}]_{m}$  due to its direct effect to restrict  $Ca^{2+}$  uptake via the CU, it is possible that a change in  $[Mg^{2+}]_e$  itself may alter mitochondrial activity in a way that is unrelated to restricting  $Ca^{2+}$  up-take.

The kinetic mechanisms associated with CU–mediated  $Ca^{2+}$  influx and  $Mg^{2+}$  inhibition of CU function have been characterized using both initial velocity studies (Bragadin et al 1979, Crompton et al 1976, Scarpa et al 1973, Vinogradov et al 1973) and mathematical models (Pradhan et al 2011). Experimentally, the presence of  $MgCl<sub>2</sub>$  was shown to alter the kinetics of Ca<sup>2+</sup> uptake with added CaCl<sub>2</sub> from a hyperbolic relationship in the absence of Mg<sup>2+</sup> to a sigmoidal relationship in the presence of  $Mg^{2+}$ , with an increase in sigmoidicity at higher [Mg<sup>2+</sup>]<sub>e</sub> (Bragadin et al 1979, Crompton et al 1976, Scarpa et al 1973, Vinogradov et al 1973). However, these studies in isolated mitochondria used rather high  $\text{[Ca}^{\frac{1}{2}+}\text{]}_e$  (up to 400  $\mu$ M) and [Mg<sup>2+</sup>]<sub>e</sub> (up to 5 mM), which are above the physiological ranges of 0.1–1  $\mu$ M for [Ca<sup>2+</sup>]<sub>e</sub> and 0.25–1 mM for [Mg<sup>2+</sup>]<sub>e</sub> (Rotevatn et al 1989). Thus the effect of [Mg<sup>2+</sup>]<sub>e</sub> on matrix Ca<sup>2+</sup> uptake is not well characterized at physiological levels of  $[Ca^{2+}]_e$  or  $[Mg^{2+}]_e$ . Although  $[Mg^{2+}]_{m}$  and  $[Ca^{2+}]_{m}$  have been proposed to alter mitochondrial function independently (Panov et al 1996, Rodriguez-Zavala et al 1998), the effect of  $[Mg^{2+}]_e$  on dynamic responses of both  $[Ca^{2+}]_e$  and  $[Ca^{2+}]_m$  was not investigated. Also it is not known if  $[Mg^{2+}]_e$  alters mitochondrial bioenergetics in a manner that is not due to its direct effect to reduce matrix  $Ca^{2+}$  uptake via the CU at physiological levels of these ions.

We hypothesized a) that  $Ca^{2+}$  uptake via the CU at physiological levels of  $[Ca^{2+}]_e$  is significantly impeded at even low  $[Mg^{2+}]_e$ , and b) that an increase in  $[Mg^{2+}]_e$  can alter the bioenergetic state independent of its effect to decrease  $\left[Ca^{2+}\right]_{m}$  via the CU. Thus the aim of this study was two-fold: first, to dynamically characterize the extent that  $MgCl<sub>2</sub>$  impedes  $Ca^{2+}$  uptake via the CU when adding physiological levels of CaCl<sub>2</sub>, and second, to assess the extent that changes in  $[Mg^{2+}]_e$  and  $[Ca^{2+}]_e$ , directly  $([Ca^{2+}]_m$ -dependent), or indirectly  $([Ca<sup>2+</sup>]<sub>m</sub>$ –independent), modulate bioenergetics. Specifically, we determined if an increase in  $[Mg^{2+}]_e$  can also modulate the bioenergetic state of mitochondria at a given  $[Ca^{2+}]_m$ . To carry out these aims, we used fluorescence spectrophotometry to monitor dynamic changes in  $[Ca^{2+}]_e$ ,  $[Ca^{2+}]_m$ ,  $[Mg^{2+}]_e$ ,  $[Mg^{2+}]_m$ , NADH and membrane potential  $(\Delta \Psi_m)$  in isolated mitochondria from guinea pig hearts during extra-matrix addition of a range of physiological [CaCl<sub>2</sub>] and [MgCl<sub>2</sub>]; in parallel experiments, we measured  $O_2$  consumption with and without added ADP at the same ranges of  $[CaCl<sub>2</sub>]$  and  $[MgCl<sub>2</sub>]$ . To prevent  $Ca<sup>2+</sup>$  extrusion via Na<sup>+</sup>–Ca<sup>2+</sup> exchange (NCX<sub>m</sub>) we used only Na<sup>+</sup>–free solutions and compounds.

## **Materials and methods**

#### **Mitochondrial isolation**

All experiments conformed to the Guide for the Care and Use of Laboratory Animals and were approved by the Institutional Animal Care and Utilization Committee. Guinea pig heart mitochondria were isolated as described before (Blomeyer et al 2012, Haumann et al 2010, Riess et al 2005). Guinea pigs (250–400 g) were anesthetized by intra-peritoneal injection of 30 mg ketamine along with 700 units of the anticoagulant heparin to prevent clotting. Each heart was excised and minced to approximately 1 mm<sup>3</sup> pieces in ice-cold isolation buffer containing in mM: mannitol 200, sucrose 50,  $KH_2PO_4$  5, 3-(N-morpholino) propanesulfonic acid (MOPS) 5, EGTA 1, BSA 0.1%, at pH 7.15 (adjusted with KOH). The buffer was decanted and the minced heart was suspended in 2.65 ml buffer with 5U/ml protease (Bacillus licheniformis), and was homogenized at low speed using a CAT X homogenizer for 30 s; next 17 ml isolation buffer was added and this suspension was again homogenized for 30 s and then was centrifuged at 8000  $g$  for 10 min. The supernatant was discarded and

the pellet was resuspended in 25 ml isolation buffer and centrifuged at 900  $g$  for 10 min. The supernatant was centrifuged once more at 8000  $g$  to yield the final mitochondrial pellet, which was suspended in 0.5 ml isolation buffer and kept on ice until used. Mitochondrial protein concentration was measured using the Bradford method (Bradford 1976). The suspension volume was adjusted to obtain a concentration of 12.5 mg protein/ml isolation buffer. All experiments were conducted at room temperature ( $25^{\circ}$ C) in a Na<sup>+</sup>–free respiration buffer (0.5 mg protein/ml) containing in mM: KCl 130, K<sub>2</sub>HPO<sub>4</sub> 5, MOPS 20, EGTA 1, BSA 0.1% and at pH 7.15 (adjusted with KOH). Trial experiments were conducted in the presence of 25  $\mu$ M CGP 37157 (Tocris Bioscience), a NCX inhibitor, to verify that  $Na<sup>+</sup>$  was not present in the respiration buffer.

## **Measurements of**  $[Ca^{2+}]$ **<sub>m</sub> and**  $[Ca^{2+}]$ **<sub>e</sub>**

Fluorescence spectrophotometry (Qm-8, Photon Technology International) was used to measure  $[Ca^{2+}]$ <sub>m</sub> and  $[Ca^{2+}]_e$ . To measure  $[Ca^{2+}]_m$ , isolated mitochondria (5 mg/ml) were incubated with indo-1 acetoxymethyl (indo-1-AM) (Invitrogen) (5 μM in DMSO) for 20 min at room temperature (25  $^{\circ}$ C), followed by addition of 25 ml ice-cold isolation buffer and repeated centrifugation at 8000  $g$ . The AM form of indo-1 is taken up into the matrix where it is de-esterified and retained. The dye-loaded pellet was re-suspended in 0.5 ml isolation buffer, and protein concentration was measured again and diluted to 12.5 mg mitochondrial protein/ml. For  $\left[Ca^{2+}\right]_m$  measurements, the dye loaded mitochondria were suspended in the Na<sup>+</sup>–free respiration buffer (0.5 mg protein/ml).

The emission wavelengths ( $\lambda_{em}$ ) of indo-1 at an excitation wavelength of 350 nm ( $\lambda_{ex}$ ) were 456 nm and 390 nm. The ratio (R) between the two  $\lambda_{em}$ 's relative to the ratios obtained when all indo-1 is bound to  $Ca^{2+} (R_{max})$  and when no  $Ca^{2+}$  is bound to indo-1  $(R_{min})$  corresponds to  $[Ca^{2+}]$ <sub>m</sub> and also serves to correct for variations that affect both signals equally, e.g. differences in the amount of dye taken up into mitochondria and small fluctuations in excitation intensity. For each preparation  $R_{\text{max}}$  was measured in the presence of 500 nM cyclosporine A plus 10 mM CaCl<sub>2</sub>, and  $R_{min}$  was measured after adding A23187  $(Ca<sup>2+</sup>-ionophore)$  plus 1 mM EGTA. Because there may be considerable background autofluorescence (AF) at these wavelengths, which is largely attributable to NADH, mitochondria AF was measured in mitochondria not loaded with indo-1-AM but with its vehicle DMSO, using the same procedure as described above; these AF signals were then subtracted from the indo-1 signals before R was calculated.  $[Ca^{2+}]_{m}$  was then calculated using the equation (Grynkiewicz et al 1985):

$$
[Ca^{2+}]_{m} (nM)=K_d \bullet (S_{f2}/S_{b2}) \bullet (R-R_{min})/(R_{max}-R)
$$

The  $K_d$  value for indo-1-AM binding to Ca<sup>2+</sup> under our conditions was determined as 326 nM (see Figs. S1 and S2 of Supplemental Materials).  $S_{f2}$  is the signal intensity of free indo-1 measured at 456 nm;  $S_{b2}$  is the signal intensity of Ca<sup>2+</sup>-saturated indo-1 measured at 456 nm. Each fluorescence signals was measured every second.

 $[Ca<sup>2+</sup>]$ <sub>e</sub> was measured using the same procedure, but with indo-1 penta-potassium salt (indo-1-PP) instead of indo-1-AM; indo-1-PP is relatively impermeable to IMM. Mitochondria were isolated as above for the indo-1-AM experiments, but were incubated for 20 min at 25°C with an equivalent amount of the vehicle, DMSO, to mimic conditions of the indo-1-AM experiments. Indo-1-PP was present in the respiration buffer at a concentration of 1  $\mu$ M. The signal was corrected for AF and  $\left[Ca^{2+}\right]_e$  was calculated using the same formula as for  $[Ca^{2+}]_m$ . With 1 mM MgCl<sub>2</sub> in the respiration buffer  $[Ca^{2+}]_e$  was not altered after adding CaCl<sub>2</sub>, which verifies that  $Mg^{2+}$  does not interfere with the indo-1 fluorescence signal. The  $K_d$  value for indo-1-PP binding to  $Ca^{2+}$  was determined as 311 nM.

## Measurements of  $[Mg^{2+}]$ <sub>m</sub> and  $[Mg^{2+}]_e$

 $[Mg^{2+}]$ <sub>m</sub> was measured with Mag-fura-2-AM and  $[Mg^{2+}]_e$  was measured with Mag-fura-2 tetra-potassium salt (mag-fura-2-K) using fluorescence spectrophotometry. The  $\lambda_{ex}$ '<sub>s</sub> for Mag-fura-2 at  $\lambda_{em}$  490 nm were 385 and 340 nm. [Mg<sup>2+</sup>]<sub>m</sub> was calculated from the ratio between  $\lambda_{ex}$  385 and 340 nm using the same formula as for indo-1.  $R_{min}$  was generated by permeabilizing mitochondria with A23187 and 0.005% (v/v) Triton X-100 in respiratory buffer containing 1 mM EGTA (Rodriguez-Zavala et al 1998). R<sub>max</sub> was obtained after further addition of 100 mM MgCl<sub>2</sub>. For Mag-fura-2 binding to Mg<sup>2+</sup> the K<sub>d</sub> 2.9 mM (Howarth et al 1995) was used.

#### **Measurements of mitochondrial NADH and membrane potential**

NADH is a measure of reduction/oxidation (redox) potential. Unlike NAD+, NADH molecules have natural fluorescence properties. NADH AF was measured ratiometrically at  $\lambda_{\rm em}$  390 nm and  $\lambda_{\rm em}$  456 at  $\lambda_{\rm ex}$  350 nm after incubating with DMSO for 20 min. The ratio of the signals,  $\lambda_{em}$  456 nm/ $\lambda_{em}$  390 nm reflects an increase in the ratio of NADH to NAD<sup>+</sup>, i.e. a shift to a more reduced state (Brandes et al 1996, Haumann et al 2010).  $\Delta \Psi_m$  was measured using the lipophilic dye TMRM in a ratiometric excitation approach (Blomeyer et al 2012, Scaduto et al 1999). TMRM  $(1 \mu M)$ , dissolved in DMSO, was separately added to the experimental buffer. Fluorescence changes were detected by two  $\lambda_{ex}$  (546 and 573 nm) and one  $\lambda_{\text{em}}$  (590 nm). The calculated ratio of both excitation wavelengths (573/546) is proportional to  $\Delta \Psi_m$  and has the advantage of a broader dynamic range when compared to a single wavelength technique (Scaduto et al 1999). Measured ratios were scaled for each group to their average photon counts at  $t = 100$  s (for protocol and timeline see *Experimental* protocol).

#### **Measurement of mitochondrial respiration**

 $O<sub>2</sub>$  consumption rate was measured using a Clark-type polarographic  $O<sub>2</sub>$  sensor (Oxygraph-2K from OROBOROS Instruments, Austria) at 25 °C (Haumann et al 2010). Mitochondria (0.125 mg protein/ml) from 5 hearts were suspended in Na<sup>+</sup>-free respiration buffer. State 2 respiration was defined as  $O_2$  consumption in the presence of high substrate and low ADP levels (i.e. after addition of 0.5 mM pyruvic acid (PA)), state 3 respiration was defined as  $O_2$  consumption in the presence of high substrate and ADP levels (i.e. after adding ADP), and state 4 respiration was defined as  $O<sub>2</sub>$  consumption after all added ADP was phosphorylated to ATP; O<sub>2</sub> consumption was expressed as pmol/s/mg protein. Experiments were conducted in mitochondria that exhibited a respiratory control index (RCI; state 3 respiration/state 4 respiration in the absence of added  $Ca^{2+}$  and  $Mg^{2+}$ ) consistently greater than 16 with 0.5 mM PA and added ADP. At this RCI, mitochondria are tightly coupled with a fully polarized  $\Delta \Psi_m$ . At the end of each experiment, lasting up to 6 h, RCI's remained greater than 12.

#### **Experimental protocol**

All experiments followed the same timeline protocol as shown in Fig. 1. At  $t = -120$  s, the experiment was initiated with the Na<sup>+</sup>–free buffer containing 0 (de-ionized H<sub>2</sub>O), 0.125, 0.25, 0.5, 1 or 2 mM MgCl<sub>2</sub> and 1 mM EGTA. At t =  $-90$  s mitochondria were added to the  $Na<sup>+</sup>$ –free respiration buffer and at t = 0 s mitochondria were energized using 0.5 mM PA  $(Na^{+}$ –free). At t = 120 s, either 0 (control, de-ionized H<sub>2</sub>O), 0.25, 0.40, 0.50 or 0.60 mM CaCl<sub>2</sub> was added and responses were measured for another 120 s. Concentrations above 0.60 mM (i.e. 0.75 mM) CaCl<sub>2</sub> did not further increase  $\lbrack Ca^{2+} \rbrack_m$  but decreased NADH levels, indicating mCa<sup>2+</sup> overload (preliminary data). Finally, at t = 240 s ADP (250  $\mu$ M) was added. At  $t = 480$  s the experiment was concluded, or the uncoupler CCCP (carbonyl cyanide m-chlorophenyl-hydrazone,  $4 \mu M$ ) was added to fully depolarize the mitochondria.

Figures are presented starting at  $t = 0$  s. For some figures only the data with 0 mM (no added) MgCl<sub>2</sub>, or 0.25 and 1 mM added MgCl<sub>2</sub> are shown. The order of experiments was randomized during the course of each day. Data for comparison of groups were collected and averaged at the following time points:  $225-235$  s (state 2),  $245-255$  s (state 3), and 360– 370 s (state 4).

#### **Statistical analyses**

All data are presented as mean  $(\pm SEM)$ . ANOVA followed by a *post hoc* analysis using Student-Newman-Keuls' test was performed to determine statistically significant differences between and within groups using Sigmaplot 11 software (Systat Software, Inc., USA). A <sup>P</sup> value < 0.05 (two-tailed) was considered significant. Statistical comparisons are not shown for all time-collected data but are shown for key interrelationship summary data.

#### **Results**

## **Effect of extra-matrix MgCl2 on [Mg2+]e and [Mg2+]<sup>m</sup>**

 $[Mg^{2+}]_e$ , measured using mag-fura-2-K, was undetected prior to adding  $MgCl_2$  and proportional to the added MgCl<sub>2</sub> (Fig. 2A). [Mg<sup>2+</sup>]<sub>e</sub> rose rapidly but remained constant over 10 min.  $[Mg^{2+}]_{m}$  (Fig. 2B), measured using mag-fura-2-AM, was 0.35±0.09, 0.34±0.08 and  $0.34\pm0.09$  mM (state 2) in the 0.5, 1, and 2 mM MgCl<sub>2</sub> groups, respectively. There was no significant change in  $[Mg^{2+}]_{m}$  from these baseline values over 10 min indicating no  $Mg^{2+}$ uptake. Adding ADP at 240 s had no effect on either [Mg $^{2+}$ ]<sub>e</sub> or [Mg $^{2+}$ ]<sub>m</sub>. These data indicated that extra-matrix  $Mg^{2+}$  was not taken up into the matrix during this time so that any effects of  $Mg^{2+}$  on  $Ca^{2+}$  uptake or mitochondrial bioenergetics originated from the extra-matrix side.

## **Effect of extra-matrix CaCl2 and MgCl2 on Ca2+ uptake into mitochondria**

We first determined to what extent MgCl<sub>2</sub> alters  $Ca^{2+}$  uptake via the CU. Mitochondria were energized with the Na<sup>+</sup>–free substrate PA to prevent Ca<sup>2+</sup> extrusion via NCX<sub>m</sub>. At 0.25, 0.40, 0.50 or 0.60 mM CaCl<sub>2</sub>, with no added MgCl<sub>2</sub> and in the presence of 1 mM EGTA, [Ca<sup>2+</sup>]<sub>e</sub> increased rapidly to 60 $\pm$ 3, 120 $\pm$ 9, 173 $\pm$ 10 and 254 $\pm$ 36 nM, respectively, compared to the 0 mM (control) CaCl<sub>2</sub> value of  $6\pm1$  nM (Fig. 3). There were no further changes in  $[Ca<sup>2+</sup>]$ <sub>e</sub> over time among the corresponding CaCl<sub>2</sub> groups when adding any concentration of MgCl<sub>2</sub> to the buffer (data not shown). Although  $Ca^{2+}$  enters the mitochondria via the CU,  $[Ca^{2+}]_e$  changed little because of the high buffering effect of 1 mM EGTA and the much larger extra-matrix volume than matrix volume. However, at 0.50 and 0.60 mM CaCl<sub>2</sub>, there was a small initial transient decline in  $[Ca^{2+}]_e$  (Fig. 3), signifying a rapid increase in  $Ca^{2+}$ uptake. Test experiments conducted in the presence of CGP 37157 gave similar results thus verifying that the buffer was free of Na<sup>+</sup> and that  $Ca^{2+}$  extrusion via NCX<sub>m</sub> did not occur. Adding ADP at 240 s did not change  $\left[Ca^{2+}\right]_e$  (not shown).

The increases in  $[Ca^{2+}]_e$  resulted in a two-phase concentration-dependent increase in  $\left[\text{Ca}^{2+}\right]_{\text{m}}$  (Fig. 4); an initial faster uptake of Ca<sup>2+</sup> (t= 120–125 min) followed by a slower, more gradual uptake (t = 125–230 s) (Fig. 4A–C). Most  $Ca^{2+}$  uptake occurred during the initial faster (5–10 s) phase, particularly in the presence of added  $MgCl<sub>2</sub>$  which attenuated only the slower phase of Ca<sup>2+</sup> uptake. In the absence of added MgCl<sub>2</sub> (Fig. 4A), the increase in [Ca<sup>2+</sup>]<sub>e</sub> (above) in response to added CaCl<sub>2</sub> increased [Ca<sup>2+</sup>]<sub>m</sub> from 46±9 nM (no added CaCl<sub>2</sub>) to 221±24, 509±51, 775±27 and 868±25 nM (0.25, 0.40, 0.50, 0.60 mM CaCl<sub>2</sub>), respectively, by the end of state 2 respiration  $(t = 230 \text{ s})$ . In the presence of 0.125, 0.25, 0.5, 1 and 2 mM MgCl<sub>2</sub>, Ca<sup>2+</sup> uptake was significantly and equivalently reduced, which resulted in lower  $\left[Ca^{2+}\right]_{m}$  at the end of state 2 respiration (Fig. 4B–F). Interestingly, the decrease in

 $Ca^{2+}$  uptake was apparent with as little as 0.125 mM MgCl<sub>2</sub> and higher [MgCl<sub>2</sub>] did not appreciably enhance this decrease in  $Ca^{2+}$  uptake.

## Effect of extra-matrix MgCl<sub>2</sub> on  $[Ca^{2+}]$ <sub>m</sub> as a function of  $[Ca^{2+}]$ <sub>e</sub>

We next plotted  $[Ca^{2+}]_{m}$  as a function of  $[Ca^{2+}]_{e}$  with/without added MgCl<sub>2</sub> at the end of state 2 respiration (t = 230 s) (Fig. 5). [Ca<sup>2+</sup>]<sub>m</sub> rose up to 3.5 fold higher than [Ca<sup>2+</sup>]<sub>e</sub> in the absence of MgCl<sub>2</sub>. With no added MgCl<sub>2</sub>,  $[\text{Ca}^{2+}]_{\text{m}}$  reached a plateau level of about 868 nM at 0.50 mM CaCl<sub>2</sub> so that at 0.60 mM CaCl<sub>2</sub> (254 nM  $\left[ Ca^{2+} \right]_e$ ) there was no further increase in  $[Ca^{2+}]$ <sub>m</sub>. All MgCl<sub>2</sub> groups higher than 0 mM (0.125–2 mM) reduced  $[Ca^{2+}]$ <sub>m</sub> similarly at a given [Ca<sup>2+</sup>]<sub>e</sub>, such that each [MgCl<sub>2</sub>] from 0.125 mM and higher reduced [Ca<sup>2+</sup>]<sub>m</sub> between 45% and 55%, respectively, at 254 nM  $[Ca^{2+}]_e$ . Thus the strength of attenuation of  $Ca^{2+}$  uptake by  $Mg^{2+}$  was very steep, indicating that even adding very little  $MgCl<sub>2</sub> (0.125)$ mM) could reduce the maximal amount of  $Ca^{2+}$  uptake through the CU.

#### **Effects of [Ca2+]e and [Mg2+]e on mitochondrial NADH and membrane potential**

We next assessed changes in two important markers of bioenergetics, NADH and  $\Delta \Psi_{\text{m}}$ , that could be affected by increases in buffer  $[CaCl<sub>2</sub>]$  and  $[MgCl<sub>2</sub>]$ . NADH levels were measured (Fig. 6) following the same timeline as for the  $Ca^{2+}$  and  $Mg^{2+}$  experiments (Fig. 1) to correlate changes in  $\left[Ca^{2+}\right]_{m}$  to changes in mitochondrial bioenergetics in real time. Adding PA increased state 2 NADH markedly and adding CaCl<sub>2</sub> further increased NADH significantly ( $P<sub>0.05</sub>$  during state 2 at t = 230 s) in the absence of MgCl<sub>2</sub> (Fig. 6A). The presence of 0.25 and 1 mM  $MgCl<sub>2</sub>$  (Fig. 6B,C) attenuated the increases in state 2 NADH after adding CaCl<sub>2</sub> to values not significantly different ( $P$ >0.05) between 0 mM (control) and added  $CaCl<sub>2</sub>$  groups. Initiation of state 3 respiration by adding ADP led to a large transient oxidation of NADH (decreased NADH/NAD<sup>+</sup> ratio) as NADH was consumed by the electron transport chain to maintain the electrochemical gradient. There was no difference in NADH oxidation levels during state 3 respiration among the  $CaCl<sub>2</sub>$  groups  $(P>0.05)$  but state 3 duration decreased with increasing CaCl<sub>2</sub> added in the absence of MgCl<sub>2</sub> (Fig. 6A); this indicated faster electron transfer and ADP phosphorylation. After consumption of all ADP (state 4), and in the absence of  $MgCl<sub>2</sub>$ , NADH returned to pre-ADP levels (state 2 respiration) for all levels of CaCl<sub>2</sub>; NADH was higher after adding CaCl<sub>2</sub> compared to no CaCl<sub>2</sub> ( $P$ >0.05). At 0.25 or 1 mM MgCl<sub>2</sub>, NADH levels during state 4 respiration remained relatively more oxidized (decreased NADH) in all  $CaCl<sub>2</sub>$  groups (P<0.05) compared to pre-ADP levels (state 2 respiration). Unlike their effects to alter redox state (NADH), adding MgCl<sub>2</sub> or CaCl<sub>2</sub> had no appreciable effect to alter  $\Delta \Psi_m$  (Fig. 6D–F). Adding ADP caused a transient, submaximal depression of  $\Delta \Psi_{\text{m}}$ .

## **Effects of [Ca2+]e and [Mg2+]e on mitochondrial O2 consumption rates**

 $O<sub>2</sub>$  consumption (Fig. 7) was measured following the same timeline (Fig. 1) to correlate this data with changes in  $\text{[Ca}^{2+}\text{]}_{\text{m}}$ , NADH, and  $\Delta \Psi_{\text{m}}$  over time. Representative traces of O<sub>2</sub> concentration (Fig. 7A,B) and summarized  $O_2$  consumption rates (Fig. 7C,D) showed faster state 3 respiration at 0.60 mM CaCl<sub>2</sub> vs. no added CaCl<sub>2</sub> (control) with no added MgCl<sub>2</sub>. State 2 respiration did not differ among  $CaCl<sub>2</sub>$  groups in the presence or absence of  $MgCl<sub>2</sub>$ . Adding ADP induced state 3 respiration as shown by the marked increase in  $O_2$ consumption in all groups with or without added  $MgCl<sub>2</sub>$ ; moreover, adding CaCl<sub>2</sub> further enhanced state 3 respiration in a concentration-independent manner. Adding  $MgCl<sub>2</sub>$  reduced state 3 respiration (Fig. 7D) at all concentrations of added CaCl<sub>2</sub> ( $P<0.05$ ) compared to corresponding groups with no added MgCl<sub>2</sub> (Fig. 7D *vs*. C). In contrast, adding MgCl<sub>2</sub> resulted in faster state 4 respiration independent of added CaCl<sub>2</sub> (Fig. 7D *vs.* C). Rearranging part of these data to plot effects of added  $MgCl<sub>2</sub>$  at a constant CaCl<sub>2</sub> (0 or 0.6 mM  $CaCl<sub>2</sub>$ ) shows more readily how MgCl<sub>2</sub> reduced state 3 respiration while enhancing state 4 respiration (Fig. 8A–D).

#### **Effects of MgCl2 on O2 consumption as a function of extra-matrix and matrix [Ca2+]**

Lastly, we plotted changes in states 3 (Fig. 9A,C) and 4 (Fig. 9B,D)  $O_2$  consumption rates as a function of end state 2 [Ca<sup>2+</sup>]<sub>e</sub> and [Ca<sup>2+</sup>]<sub>m</sub> in the presence and absence of added MgCl<sub>2</sub> to illustrate their inter-relationships. In the absence of added MgCl<sub>2</sub>, state 3 respiration was faster with an increase in  $[Ca^{2+}]_e$  or  $[Ca^{2+}]_m$  compared to after adding MgCl<sub>2</sub> (A,C). Adding as little as 0.25 mM CaCl<sub>2</sub> ( $\approx$  221 nM [Ca<sup>2+</sup>]<sub>m</sub>) maximally enhanced respiration in the absence of MgCl<sub>2</sub>, compared to the absence of CaCl<sub>2</sub>. Importantly, at a given  $[Ca^{2+}]_e$  or  $[Ca^{2+}]<sub>m</sub>$ , each added  $[MgCl<sub>2</sub>]$  had a concentration-dependent effect to slow the rate of state 3 respiration. At the highest  $\left[Ca^{2+}\right]_e$  or  $\left[Ca^{2+}\right]_m$ , 1 mM MgCl<sub>2</sub> significantly depressed the rate of state 3 respiration. These reductions in states 3 respiratory rate by added  $MgCl<sub>2</sub>$  at a given  $[Ca^{2+}]_e$  or  $[Ca^{2+}]_m$  indicated that an increase in  $[Mg^{2+}]_e$ , but not an increase in  $[Mg^{2+}]_{m}$  (as it did not change), had an effect to reduce state 3 respiration. Unlike during state 3, state 4 respiration was not affected by an increase in  $[Ca^{2+}]_e$  or  $[Ca^{2+}]_m$  in the absence or presence of  $MgCl_2$  (Fig. 9B,D). But adding as little as 0.25 mM  $MgCl_2$ maximally increased state 4 respiration by over two fold at any given  $[Ca^{2+}]_e$  or  $[Ca^{2+}]_{m}$ . Note that the presence of MgCl<sub>2</sub> increased state 4 respiration with no change in  $\Delta \Psi_{\text{m}}$  (Fig. 6D–F), indicating there was enhanced proton conductance. (see Supplemental Material (Figs. S5–7) for changes in respiration in the presence of ruthenium red to block CU.)

#### **Discussion**

An increase in cytosolic  $[Ca^{2+}]_e$  leads to mitochondrial  $Ca^{2+}$  uptake via the CU, particularly if  $\Delta \Psi_m$  is high; much of this Ca<sup>2+</sup> is buffered in the matrix; some is extruded via the NCX<sub>m</sub>. Cytosolic Ca<sup>2+</sup> over-load, as occurs in ischemia-reperfusion injury, can disrupt mitochondrial function and induce apoptotic mechanisms (Bernardi 1999, Camara et al 2010, Duchen 2000). Because Mg<sup>2+</sup> is a modulator of CU –mediated Ca<sup>2+</sup> uptake, its impact on  $[Ca^{2+}]$ <sub>m</sub> is therefore important. Moreover, Mg<sup>2+</sup> may have effects on bioenergetics unrelated to its modulation of  $[Ca^{2+}]$ <sub>m</sub>.

Our aims were: a) to directly compare effects of increasing  $[Mg^{2+}]_e$  and  $[Ca^{2+}]_e$  on  $[Ca^{2+}]_m$ dynamics over a physiologic range, and b) to assess the effects of these cations on three bioenergetics markers,  $\Delta \Psi_m$ , NADH redox state, and  $O_2$  consumption. We found that very small increases in  $[Mg^{2+}]_e$  near maximally attenuated the CU-mediated increase in  $[Ca^{2+}]_m$ for the range of  $[Ca^{2+}]_e$  examined (0–0.6 mM CaCl<sub>2</sub> with 1 mM EGTA); further increases in [Mg<sup>2+</sup>]<sub>e</sub> (from 0.125 to 2 mM) did not significantly alter the dynamics of Ca<sup>2+</sup> uptake. Also, a CaCl<sub>2</sub>–induced increase in  $\text{[Ca}^{2+}\text{]}_{\text{m}}$  from 46 to 221 nM was associated with a 5% increase in state 2 NADH (redox state) and a 15% increase in state 3 respiration in the absence of added MgCl<sub>2</sub>; greater than 221 nM [Ca<sup>2+</sup>]<sub>m</sub> had no additional effect on NADH or O<sub>2</sub> consumption. The presence of MgCl<sub>2</sub> resulted in a concentration–dependent fall in state 3 NADH and  $O_2$  consumption due in part to the concomitant decrease in  $[Ca^{2+}]$ <sub>m</sub>. However, at a given  $\left[Ca^{2+}\right]_{m}$  (e.g. 221 nM), 1 mM MgCl<sub>2</sub> nearly abolished the increases in NADH and  $O_2$  consumption due to the increase in  $[Ca^{2+}]$ <sub>m</sub> from 46 to 221 nM. In contrast, the presence of Mg<sup>2+</sup> markedly enhanced state 4 respiration at any given  $[Ca^{2+}]_{m}$ .

These results suggest that the modulatory effects of  $Mg^{2+}$  on states 3 and 4 bioenergetics appear to be dependent not only on  $[Mg^{2+}]_e$  to reduce  $Ca^{2+}$  uptake and decrease  $[Ca^{2+}]_m$ , but also on an increase in  $[Mg^{2+}]_e$  by matrix  $Ca^{2+}$ –independent mechanisms. Indeed, bioenergetic function was reduced foremost because  $Mg^{2+}$  attenuated  $Ca^{2+}$  uptake, even at higher  $[Ca^{2+}]_e$  or  $[Ca^{2+}]_{m}$ . However, because neither adding MgCl<sub>2</sub> nor ADP increased  $[Mg^{2+}]_{m}$ , Mg<sup>2+</sup> had to have an indirect, i.e. extra-matrix, effect to attenuate state 3 respiration and NADH, while it enhanced state 4 respiration. Because adding  $MgCl<sub>2</sub>$  or CaCl<sub>2</sub> had no effect on  $\Delta \Psi_m$ , these cations likely had no direct effect to alter proton conductance ("H+ leak"). Therefore, the slower state 3 respiration and faster state 4

respiration with increasing [Mg<sup>2+</sup>]<sub>e</sub> was due not only to an effect of Mg<sup>2+</sup> to reduce [Ca<sup>2+</sup>]<sub>m</sub> but, by exclusion, also due to effects mediated outside the matrix.

## **Mg2+reduces maximal mitochondrial Ca2+ uptake induced by increasing external Ca2+**

The main route for  $Ca^{2+}$  uptake is likely the CU, which can be considered a low affinity influx pathway (Bragadin et al 1979, Crompton et al 1976, Scarpa et al 1973, Vinogradov et al 1973). Patch clamp experiments show that the CU is a highly selective ion channel (Kirichok et al 2004), and its protein structure is now known (Baughman et al 2011, De Stefani et al 2011); an associated protein, MICU1, is required for  $Ca^{2+}$  uptake (Perocchi et al 2010). Though the CU is the major route for  $Ca^{2+}$  uptake, the rate of uptake may be regulated by a different distinct channel or a different conformational state of the CU (Bazil et al 2011, Buntinas et al 2001, Sparagna et al 1995). Our experiments were designed to dynamically measure extra-matrix and matrix free  $[Ca^{2+}]$  and  $[Mg^{2+}]$  to better assess how  $[Mg^{2+}]$  affects movement of Ca<sup>2+</sup> across the IMM via CU. Earlier studies (Denton et al 1980, McCormack et al 1990, Wan et al 1989) examined only static levels of  $Ca^{2+}$  and correlated them with added extra-matrix CaCl<sub>2</sub>. We show here how the maximal uptake of  $Ca^{2+}$  was significantly affected by  $[Mg^{2+}]_e$  at physiological levels of  $[Ca^{2+}]_e$ .

The range of  $[\text{Ca}^{2+}]_{\text{m}}$  we examined lies within a physiological range of 100 to 900 nM (up to 0.60 mM added CaCl<sub>2</sub>); this range of added CaCl<sub>2</sub> slightly enhanced respiration but did not decrease NADH. Most ionized  $Ca^{2+}$  entering the matrix is buffered by soluble phosphate complexes (Bataille et al 1994, Morgan et al 2006, Starkov 2010) that are influenced by proteins (e.g. annexins) and lipids (e.g. cardiolipin), so total loading of  $Ca^{2+}$  into the matrix from the added  $CaCl<sub>2</sub>$  is much greater than that seen from the less than 4 fold higher  $\text{[Ca}^{2+}\text{]}_{\text{m}}$  than  $\text{[Ca}^{2+}\text{]}_{\text{e}}$ ; a  $\text{[Ca}^{2+}\text{]}_{\text{m}}$  between 700 and 1100 nM, as measured in our intact heart model (Aldakkak et al 2011, Camara et al 2007, Rhodes et al 2012), may indicate excess  $Ca^{2+}$  loading and a nearly exhausted  $Ca^{2+}$  buffering capacity, so that mitochondrial permeability transition pore (mPTP) opening and apoptosis may occur. Furthermore, above a certain  $\left[Ca^{2+}\right]_m$  the increase in respiration might actually be due to the uncoupling effects of excess mCa2+.

## **Matrix Ca2+–induced changes in mitochondrial bioenergetics are sensitive but limited**

Several possible mechanisms by which matrix  $Ca^{2+}$  is believed to regulate bioenergetics have been reviewed (Balaban 2009, Griffiths et al 2009). The first proteins observed to be activated by  $Ca^{2+}$  were three dehydrogenases in the TCA cycle. Since then  $Ca^{2+}$  has been proposed to regulate a number of substrate transporters, as well as  $F_1F_0$ -ATPase and cytochrome c oxidase (Balaban 2009). We showed that an increase in  $[Ca^{2+}]$ <sub>m</sub> is accompanied by an immediate, but small, increase in NADH levels during state 2 respiration without an increase in  $O_2$  consumption. This effect was quite small and active only at very low  $\left[\text{Ca}^{2+}\right]_{\text{m}}$ . The small increase in redox potential, combined with the possible effects of  $[Ca^{2+}]$ <sub>m</sub> on some mitochondrial enzymes could underlie, in part, the higher state 3 respiration. However, the importance of  $[Ca^{2+}]$ <sub>m</sub> on mitochondrial dehydrogenase activities is strongly dependent on the substrate used. For example, it was reported that pyruvate dehydrogenase is activated at almost 100%, regardless of the  $[Ca^{2+}]<sub>m</sub>$ , and that  $\alpha$ ketoglutarate dehydrogenase is a more physiologically relevant target of  $Ca^{2+}$  regulation (Vinnakota et al 2011, Wan et al 1989). This means that the greatest effect of  $Ca^{2+}$  on bioenergetics would be with α-ketoglutarate as the substrate.

For this study, we chose PA as our substrate because it is effectively (through acetyl CoA) the two-carbon donor to oxaloacetate; when the six-carbon product isocitrate and other TCA cycle intermediates become oxidized this reduces NAD<sup>+</sup> to NADH used for proton pumping to maintain  $\Delta \Psi_m$ . This might explain in part why we observed only a small effect of Ca<sup>2+</sup> to

increase bioenergetic activity compared to previous studies using α-ketoglutarate and other TCA cycle substrates. The substrate glutamate+malate may provide a greater NADH response to increasing  $\lbrack Ca^{2+}\rbrack _e$  (Vinnakota et al 2011); moreover, higher concentrations of substrates to energize mitochondria may saturate ADP-independent TCA cycle flux. We used 0.5 mM PA, a lesser amount than used in many studies, which may have led to our observation of only a small, stimulated respiratory response to  $Ca^{2+}$  uptake. In contrast, an increase in cardiac work demand can increase cardiac  $O_2$  consumption by up to 8 fold, so it is doubtful that this is solely due to the reported modest increases in  $[\text{Ca}^{2+}]_{\text{m}}$ .

## **Mg2+–induced attenuation of mitochondrial Ca2+ uptake occurs at the Ca2+ entry site**

That Ca<sup>2+</sup> uptake is attenuated by  $[Mg^{2+}]_e$ , as shown by the fall in  $[Ca^{2+}]_m$  at a constant  $[Ca<sup>2+</sup>]$ <sub>e</sub> (Fig. 5A), agrees qualitatively with earlier findings (Favaron et al 1985, Litsky et al 1997). McCormack et al. (McCormack et al 1990) that adding 1-2 mM MgCl<sub>2</sub> reduced  $\lbrack Ca^{2+} \rbrack_m$  by about 8 fold at 250 nM  $\lbrack Ca^{2+} \rbrack_e$ . We show, however, that even at a very low physiologic range of  $[Mg^{2+}]_e$ , this effect is already close to a maximum but can be partially overcome by adding more CaCl<sub>2</sub> (Fig. 4B). This trend of  $[Mg^{2+}]_e$  to blunt, but not to completely inhibit, the increases in  $[Ca^{2+}]$ <sub>m</sub> was also observed recently (Wei et al 2012). As noted above, a temporal characteristic of Ca<sup>2+</sup> uptake is that at any [Ca<sup>2+</sup>]<sub>e</sub>, adding MgCl<sub>2</sub> reduced the net  $Ca^{2+}$  uptake, but not the initial uptake phase (Fig. 4). Interestingly, adding MgCl<sub>2</sub> did not at all increase  $[Mg^{2+}]_m$  over many minutes (Fig. 2). Thus our study indicates that  $Mg^{2+}$  inhibits CU-mediated Ca<sup>2+</sup> uptake by interfering with Ca<sup>2+</sup> only at the extramatrix side of the CU. Although it was reported that  $Mg^{2+}$  does not appreciably enter the matrix in the absence of ADP/ATP exchange (Chinopoulos et al 2009), we observed that adding ADP indeed did not cause a net movement of  $Mg^{2+}$  into the matrix.

 $Mg^{2+}$  is proposed to modulate mitochondrial cation and substrate transport systems; but the dynamic interaction of physiological concentrations of  $Mg^{2+}$  with  $Ca^{2+}$  inside or outside the mitochondrion has not been rigorously examined. Earlier  $Ca^{2+}$  uptake experiments in isolated mitochondria were conducted at very high  $\left[{\rm Ca^{2+}}\right]_e$  of up to 400  $\mu{\rm M}$ , and high  $[Mg^{2+}]_e$  of up to 5 mM (Bragadin et al 1979, Crompton et al 1976, Scarpa et al 1973, Vinogradov et al 1973), which are well above their physiologic ranges. Although these studies provided useful information on how  $Ca^{2+}$  and  $Mg^{2+}$  interacted at the putative CU, and on the type of  $Mg^{2+}$  inhibition (e.g. competitive, non-competitive or mixed-type), these effects were not examined within the lower physiological ranges of  $[Mg^{2+}]$  or  $[Ca^{2+}]$  and the dynamics of  $Mg^{2+}$  on the time course of  $Ca^{2+}$  uptake were not examined. Importantly, in these earlier studies  $[Mg^{2+}]_{m}$  was not directly measured during the duration of Ca<sup>2+</sup> uptake; interestingly, in our study  $[Mg^{2+}]_{m}$  did not change over 10 min at any added MgCl<sub>2</sub>. Therefore, it has remained largely unknown how  $[Mg^{2+}]_{m}$  could dynamically alter CUmediated Ca<sup>2+</sup> transport. Our study demonstrates clearly how normal levels of  ${\rm [Ca^{2+}]_e}$  and  $[Mg^{2+}]_e$  interact to modulate  $[Ca^{2+}]_m$  via the CU. A recent study in permeabilized HEK cells demonstrated that physiological changes in cytosolic  $Mg^{2+}$  attenuated  $Ca^{2+}$  uptake (Szanda et al 2009), which is consistent with our findings and those of others. But our study additionally provides detailed information on the characteristics of this  $Mg^{2+}$ –induced attenuation of  $Ca^{2+}$  uptake that can now be utilized to construct more accurate computational models of  $Ca^{2+}$  transport systems and their regulation.

## **External Mg2+ alters bioenergetics by both mCa2+–dependent and independent mechanisms**

The physiological range of cytoplasmic  $\lbrack Ca^{2+} \rbrack_e$  in cardiac cells fluctuates between about 100 and 400 nM with each heart beat and increases only moderately with inotropic stimulation (Rhodes et al 2006). In a normal cardiac cell, average  $[Ca^{2+}]$ <sub>m</sub> is about 150–200 nM (Riess et al 2002). An increase in [Ca<sup>2+</sup>]<sub>e</sub> to 250 nM led only to an increase in [Ca<sup>2+</sup>]<sub>m</sub>

of 870 nM at the end of state 2 respiration (Figs. 4A and 5A) at 0.60 mM CaCl<sub>2</sub> in the absence of MgCl<sub>2</sub>. We attempted to measure the dynamics of  $[Ca^{2+}]$ <sub>m</sub> in response to higher CaCl<sub>2</sub> (e.g. at 0.75 mM CaCl<sub>2</sub> equivalent to 1236 $\pm$ 218 nM [Ca<sup>2+</sup>]<sub>m</sub>) to increase [Ca<sup>2+</sup>]<sub>m</sub> further. However, we observed a detrimental effect of high  $[Ca^{2+}]$ <sub>m</sub> on state 3 respiration; i.e. state 3 respiration was slower, state 3 duration was longer, and in some experiments a faster state 3 respiration did not occur at all after adding ADP (data not shown). The finding that higher  $[Ca^{2+}]_e$  led to dysfunction is likely related to the absence of Na<sup>+</sup> in our buffer to extrude excess  $Ca^{2+}$  via NCX<sub>m</sub>. We showed previously that a lack of buffer Na<sup>+</sup> abolishes  $NCX<sub>m</sub>$  similar to inhibiting the  $NCX<sub>m</sub>$  with CGP 37157 (Agarwal et al 2012, Blomeyer et al 2012). Because NCX<sub>m</sub> is the major route for Ca<sup>2+</sup> extrusion in cmitochondria, the presence of extra-matrix Na<sup>+</sup> clearly exerts a diminishing effect on the net rise in  $\left[Ca^{2+}\right]_{m}$  (Maack et al 2006) when [Ca<sup>2+</sup>]<sub>e</sub> increases. Moreover, because excess Mg<sup>2+</sup> was reported to inhibit mPTP opening (Bernardi et al 1993), a lack of added  $MgCl<sub>2</sub>$  might contribute to earlier mPTP opening when  $Ca^{2+}$ uptake is large.

It would seem unclear how a change in [Mg<sup>2+</sup>]<sub>e</sub> (not [Mg<sup>2+</sup>]<sub>m</sub>) could exert any direct, i.e. not via changes in  $[Ca^{2+}]_{m}$ , effects on bioenergetics. However, a change in  $[Mg^{2+}]_{m}$  was reported to independently modulate oxidative phosphorylation (Klingenberg 2008, Panov et al 1996, Rodriguez-Zavala et al 1998) among other effects, because of actions on succinate and glutamate dehydrogenase (Romani 2007). Changes in  $[Mg^{2+}]_{m}$  were shown to occur in response to hormonal stimulation (Romani 2007) and altered metabolic state (Jung et al 1990). The process of very slow  $Mg^{2+}$  uptake into mitochondria is much less clearly defined than for Ca<sup>2+</sup> uptake and is more difficult to examine because trans-membrane Mg<sup>2+</sup> levels do not vary greatly.  $Mg^{2+}$  uptake was reported to be dependent on the activity of the ATP- $Mg^{2+}/P_i$  carrier, but this is many times lower than the activity of the adenine nucleotide transporter (ANT) (Chinopoulos et al 2009). A proposed channel, mrs2p, may be present in mitochondria to transport  $Mg^{2+}$  (Schindl et al 2007), but it is not believed to contribute significantly to  $Mg^{2+}$  transport (Chinopoulos et al 2009). Regardless, other reports (Brierley et al 1987, Jung et al 1990), like ours, show that  $Mg^{2+}$  flux through the IMM is very slow or negligible, at least at the physiological concentrations we used. In fact, in another study (Rodriguez-Zavala et al 1998) where the effects of changes in  $[Mg^{2+}]_m$  were observed, the ionophore A23187 was used to equilibrate trans-membrane  $Mg^{2+}$ .

From historical studies we expected that adding MgCl<sub>2</sub> would increase  $[Mg^{2+}]_{m}$  during the course of the experiments (Fig. 2). One report (Jung et al 1990) indicated that adding  $MgCl<sub>2</sub>$ increased  $[Mg^{2+}]_{m}$  and that adding ADP resulted in an even higher  $[Mg^{2+}]_{m}$ ; it was suggested that enhanced metabolism (state 3 respiration) was associated with  $Mg^{2+}$  uptake. However, it was reported later (Chinopoulos et al 2009) that added  $Mg^{2+}$  did not appreciably enter the matrix, except in the presence of ADP/ATP exchange; this group showed that when increasing amounts of ADP<sup>3−</sup> were exchanged with ATP<sup>4−</sup> via ANT,  $[Mg^{2+}]$ <sub>m</sub> decreased; conversely, an increase in  $[Mg^{2+}]$ <sub>m</sub> with added MgCl<sub>2</sub> caused a decrease in ADP/ATP exchange. However, our results support the notion that an increase in  $[Mg^{2+}]_e$ , rather than an increase in  $[Mg^{2+}]_m$ , indirectly attenuates electrogenic transport by the ADP/ATP carrier ANT (Gropp et al 1999, Klingenberg 2008) to retard mitochondrial respiration. Our preliminary work (see Fig. S3 of Supplementary Materials) supports an earlier finding (Kramer 1980) that extra-matrix  $Mg^{2+}$  attenuated state 3 ADP/ATP exchange rate mediated by the ANT per se; it is likely that this effect results from  $Mg^{2+}$  binding to nucleotides, which reduces the amount of free ADP<sup>3−</sup> for exchange with free ATP<sup>4−</sup> by ANT.

Thus, our results in part contradict the two prior reports (Chinopoulos et al 2009, Jung et al 1990) that adding either MgCl<sub>2</sub> or ADP causes an acute increase in  $[Mg^{2+}]_{m}$ . In our study, adding MgCl<sub>2</sub> did not increase  $[Mg^{2+}]_m$  from its baseline value over 10 min, irrespective of

Ca<sup>2+</sup> uptake; moreover, adding ADP neither decreased [Mg<sup>2+</sup>]<sub>e</sub> nor increased [Mg<sup>2+</sup>]<sub>m</sub>. First, the Mg<sup>2+</sup>–induced reduction of CU-mediated  $Ca^{2+}$  uptake occurred only at the extramatrix side of the CU. Second, it follows that the apparent attenuating (state 3) and enhancing (state 4) effects of  $Mg^{2+}$  on mitochondrial bioenergetics in our study was not exerted directly on matrix enzymes. In studies by Rodriguez-Zavala (Rodriguez-Zavala et al 1998) and Panov (Panov et al 1996), a higher  $[Mg^{2+}]_m$  was associated with faster oxidative phosphorylation. In contrast,  $[Mg^{2+}]_{m}$  did not change in our study, but rather an increase in [Mg<sup>2+</sup>]<sub>e</sub> attenuated state 3 and accentuated state 4 at any given, constant [Ca<sup>2+</sup>]<sub>m</sub> (Fig. 9).

## **External Mg2+ may alter non-CU Ca2+ binding sites to modulate bioenergetics**

Our data show that extra-matrix  $Mg^{2+}$  has a concentration-dependent effect to decrease state 3 respiration independently of induced changes in  $[Ca^{2+}]$ <sub>m</sub>. We suggest that one mechanism is an indirect effect of extra-matrix  $Mg^{2+}$ , rather than matrix  $Mg^{2+}$  (Chinopoulos et al 2009), to decrease ATP generation by attenuating ADP/ATP transport. Several groups (Gellerich et al 2010, Satrustegui et al 2007) have reviewed possible mechanisms by which  $[Ca^{2+}]_e$  (*vs.*  $[Ca^{2+}]_{m}$ ) might modulate bioenergetics. For example, extra-matrix  $Ca^{2+}$  may regulate glutamate-dependent state 3 respiration by modifying the supply of glutamate via aralar, a glutamate-aspartate carrier (Pardo et al 2006), thereby affecting oxidative phosphorylation indirectly via a change in extra-matrix  $Ca^{2+}$ .  $Ca^{2+}$  activation of aralar occurs by increasing  $V_{\text{max}}$  rather than by decreasing  $K_{\text{m}}$  (Contreras et al 2007). Moreover, some enzymes, like the ATP-Mg<sup>2+</sup>/P<sub>i</sub> carrier, have EF-hand Ca<sup>2+</sup>-binding motifs that are localized in the IMS where they sense extra-matrix  $[Ca^{2+}]_e$  (Satrustegui et al 2007). The ATP-Mg<sup>2+</sup>/P<sub>i</sub> carrier can slowly increase or decrease the matrix content of adenine nucleotides in an extra-matrix  $Ca<sup>2+</sup>$ -dependent manner (Traba et al 2008); but this is likely too slow to occur in our studies. These reports and others indicate that extra-matrix  $Ca^{2+}$  may alter oxidative phosphorylation via  $Ca^{2+}$ –sensing sites on several enzymes located in the IMS in a manner that is independent of the bioenergetics effects resulting from a change in  $\lbrack Ca^{2+}\rbrack_m$  (Gellerich et al 2010).

We also observed that extra-matrix  $Mg^{2+}$  decreased state 4 respiration independently of induced changes in  $\left[\text{Ca}^{2+}\right]_{\text{m}}$ . In preliminary experiments (see Fig. S4 of Supplementary Materials) we observed that the  $MgCl<sub>2</sub>$ -induced increase in state 4 respiration was reversed by the  $F_1F_0$  ATPsynthase(ase) inhibitor oligomycin. State 4 is generally faster than state 2 in part because of the presence of ATPase in the buffer that converts the generated ATP to ADP as substrate for the  $F_1F_0$  ATPsynthase, thus inducing an inward proton leak at a constant  $\Delta \Psi_{\text{m}}$ . Because Mg<sup>2+</sup> is required for ATPase activity (Pedersen et al 1987), the increase in extra-matrix Mg<sup>2+</sup> by adding MgCl<sub>2</sub> may enhance the amount of Mg<sup>2+</sup>-ATP available for hydrolysis so that more ADP becomes available. In light of our study, how extra-matrix  $[Mg^{2+}]$  retards state 3 respiration and enhances state 4 respiration by an effect on mitochondrial Ca<sup>2+</sup>-binding proteins, Mg<sup>2+</sup>-inhibited nucleotide transport, or Mg<sup>2+</sup>requiring ATPases, requires more investigation.

#### **Summary and conclusions**

A small increase in  $[\text{Ca}^{2+}]_{\text{m}}$ , but limited only to the low nM range, stimulates mitochondrial bioenergetics, most likely as part of a larger regulatory system to adapt ATP production to workload and ATP consumption. However, under physiological conditions, this role of matrix  $Ca^{2+}$  might be more sensitive, but weaker, and the role of cytosolic Mg<sup>2+</sup> stronger, but fixed, than previously believed. More work is needed to exactly define under which conditions this extra-matrix  $Mg^{2+}$ regulatory system plays a role. However,  $Mg^{2+}$  appears to be limited in its capacity to regulate matrix  $Ca^{2+}$  uptake via the CU, which itself should reduce bioenergetic activity, because its maximal effect already occurs at a low  $[Mg^{2+}]_e$ . Modulation of respiratory activity by [Mg $^{2+}$ ]<sub>e</sub>, at a given [Ca $^{2+}$ ]<sub>m</sub>, may be mediated

alternatively by  $Mg^{2+}$  to compete with  $Ca^{2+}$  to attenuate ADP/ATP transport (state 3) to enhance ATPase activity (state 4), or to compete at one or more extra-matrix  $Ca^{2+}$ -sensitive sites. Finally, attenuation of Ca<sup>2+</sup> uptake by physiological elevations of  $[Mg^{2+}]_e$  renders the  $Ca^{2+}$ -Mg<sup>2+</sup> interrelationship more complex and hence, incorporation of our collected data into computational models should yield clearer insights into how extra-matrix  $Mg^{2+}$ modulates  $\left[\text{Ca}^{2+}\right]_{\text{m}}$  as well as bioenergetic activity that is both independent and dependent on  $[Ca^{2+}]$ <sub>m</sub>.

## **Supplementary Material**

Refer to Web version on PubMed Central for supplementary material.

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#### **Fig. 1.**

Timeline for adding substances to experimental buffer. Figures are presented from addition of PA (t=0 s); 500 μM PA, Na<sup>+</sup>-free pyruvic acid; 250 μM ADP, (adenosine diphosphate).



#### **Fig. 2.**

Changes in external free  $[Mg^{2+}]_e$  on addition of  $MgCl_2$  to the buffer containing isolated mitochondria (A);  $[Mg^{2+}]_e$  was slightly less than the amount of added  $MgCl_2$ . Measure of matrix  $[Mg^{2+}]_{\text{m}}$  on addition of MgCl<sub>2</sub> (B). Note that over 10 min  $[Mg^{2+}]_{\text{m}}$  did not increase with the increase in [Mg<sup>2+</sup>]<sub>e</sub>. Adding ADP had no effect to alter either [Mg<sup>2+</sup>]<sub>e</sub> or [Mg<sup>2+</sup>]<sub>m</sub>.



#### **Fig. 3.**

Changes in dynamics of  $[Ca^{2+}]_e$  in response to adding PA, CaCl<sub>2</sub> and ADP as shown in Fig. 1 timeline. Adding CaCl2 in the presence of 1 mM EGTA caused a rapid and sustained increase in  $[Ca^{2+}]_e$  that remained unchanged due to buffering by EGTA. Steady state  $[Ca^{2+}]_e$  ranged from 5±1 to 268±30 nM for 0 to 0.60 mM CaCl<sub>2</sub>, respectively. Adding ADP at 240 s did not alter  $[Ca^{2+}]_e$ .



#### **Fig. 4.**

Changes in dynamics of  $[Ca^{2+}]$ <sub>m</sub> after adding PA, CaCl<sub>2</sub> and ADP as shown in Fig. 1 timeline. Each left panel (A–F) shows five traces corresponding to added  $CaCl<sub>2</sub>$  in the presence of 0 mM (no added)  $MgCl_2$  (A) and added  $MgCl_2$  (B, 0.125; C, 0.25; D, 0.5; E, 1; F, 2 mM). There appeared to be two phases of  $Ca^{2+}$  uptake via the CU at these levels of MgCl<sub>2</sub>: a rapid uptake phase (5–10 s) and a slow uptake phase (1–2 min). See Fig 5 for statistics.



#### **Fig. 5.**

 $[Ca^{2+}]$ <sub>m</sub> as a function of  $[Ca^{2+}]_e$  at increasing  $[Mg^{2+}]_e$  during state 2 respiration (at t=230 s). In the absence of added MgCl<sub>2</sub>,  $[Ca^{2+}]$ <sub>m</sub> was 3.5 fold greater than  $[Ca^{2+}]_e$ ; in the presence of MgCl<sub>2</sub>,  $\text{[Ca}^{2+}\text{]}_{\text{m}}$  remained greater than  $\text{[Ca}^{2+}\text{]}_{\text{e}}$ . Note the near maximal effect of added MgCl<sub>2</sub> on reducing  $\left[\text{Ca}^{2+}\right]_{\text{m}}$ .\*P<0.05 each  $\left[\text{Ca}^{2+}\right]_{\text{m}}$  with added CaCl<sub>2</sub> vs. at 0 mM CaCl<sub>2</sub>;  $^{\#}P<0.05$  each  $\text{[Ca}^{2+}\text{]}_{\text{m}}$  with added MgCl<sub>2</sub> vs. 0 mM (no added) MgCl<sub>2</sub> at a given  $[Ca^{2+}]_e$ .



#### **Fig. 6.**

Changes in dynamics of mitochondrial NADH redox state (autofluorescence) and  $\Delta \Psi_{\text{m}}$ (ratio of TMRM signals) after adding PA, CaCl<sub>2</sub> and ADP as shown in the timeline scheme (Fig. 1). Each left panel (A,B,C) for NADH and each right panel (D,E,F) for TMRM shows traces corresponding to added CaCl<sub>2</sub> in the presence of 0 mM (no added) MgCl<sub>2</sub> (A,D) and 0.25 mM (B,E) and 1 MgCl<sub>2</sub> (C,F). Tracings show that NADH increased markedly after adding PA and less so after adding CaCl<sub>2</sub> (t=120 s). Between 0.25 and 0.6 mM CaCl<sub>2</sub> there was no significant increase in state 2 NADH for all added  $[MgCl_2]$ . Adding 1 mM  $MgCl_2$ (C) decreased the  $Ca^{2+}$ –induced rise in NADH compared to corresponding CaCl<sub>2</sub> groups without added  $MgCl<sub>2</sub>(A)$  at each respiratory state. NADH was similar during states 4 and 2 respiration in the 0 mM (no added) MgCl<sub>2</sub> group, but in the added MgCl<sub>2</sub> groups NADH was lower during state 4 than in state 2 respiration ( $P<0.05$ ). Adding ADP (state 3 respiration) transiently and similarly decreased NADH as it was consumed to maintain the proton gradient in each  $CaCl<sub>2</sub>$  and  $MgCl<sub>2</sub>$  group. Note that state 3 duration in the group without added  $CaCl<sub>2</sub>$  and  $MgCl<sub>2</sub>$  (A) was longer than in the groups with added  $CaCl<sub>2</sub>$  and MgCl<sub>2</sub>, reflecting slower state 3 respiration. Adding CaCl<sub>2</sub> or MgCl<sub>2</sub> had no effect on  $\Delta \Psi_{\text{m}}$ (D–F); only adding ADP caused a transient fall in  $\Delta \Psi_{m}$ . At 480 s CCCP was given to completely depolarize the membrane potential. Other statistics are given in text.



#### **Fig. 7.**

Representative traces of  $O_2$  concentration for 0 (solid black line) and 0.60 mM (solid blue line) CaCl<sub>2</sub> without added MgCl<sub>2</sub> (A) or with 1 mM MgCl<sub>2</sub> (B) in response to adding PA,  $CaCl<sub>2</sub>$ , and ADP as shown in the timeline scheme (Fig. 1). Dotted lines are the first derivatives showing the maximal rates of change during state 3. Calculated mean  $O_2$ consumption rates  $(C,D)$  increased in the presence 0.25 mM CaCl<sub>2</sub> during state 3 respiration; higher  $[CaCl<sub>2</sub>]$  did not increase state 3 respiration further. Adding  $MgCl<sub>2</sub>$ attenuated state 3 respiration. Adding MgCl<sub>2</sub> enhanced state 4 respiration at all [CaCl<sub>2</sub>], which was compatible with the decreased NADH during state 4 respiration (Fig. 6). See Fig. 9 for statistics.



#### **Fig. 8.**

Representative traces of  $O_2$  concentration for 0 (solid blue line) and 1 mM (solid black line)  $MgCl<sub>2</sub>$  without added CaCl<sub>2</sub> (A) or with 0.6 mM CaCl<sub>2</sub> (B) in response to adding PA, CaCl2, and ADP. Dotted lines are the first derivatives showing the maximal rates of change during state 3. Calculated mean  $O_2$  consumption rates (C,D) decreased in the presence 1 mM MgCl<sub>2</sub> during state 3 respiration (B); higher  $[CaCl<sub>2</sub>]$  did not increase state 3 respiration further. Adding CaCl<sub>2</sub> enhanced state 3 respiration. Adding MgCl<sub>2</sub> enhanced state 4 respiration at all  $[CaCl<sub>2</sub>]$ , which was compatible with decreased NADH during state 4 respiration (Fig. 5). See Fig. 9 for statistics.



#### **Fig. 9.**

State 3 (A,C) and state 4 (B,D) O<sub>2</sub> consumption rates as a function of  $[Ca^{2+}]_e$  (A,B) and  $[Ca^{2+}]$ <sub>m</sub> (C,D) in the absence and presence of 0.25 and 1 mM MgCl<sub>2</sub>. In the absence of MgCl<sub>2</sub> state 3 respiration was enhanced between 6 to 60 nM  $[Ca<sup>2+</sup>]_e$ , but not by higher  $[Ca^{2+}]_e$  up to 254 nM (A). At a given  $[Ca^{2+}]_e$  or  $[Ca^{2+}]_m$  at the end of state 2, increasing MgCl<sub>2</sub> (from 0 to 1 mM) reduced state 3 respiration  $(A, C)$  in a stepwise manner. State 4 respiration (B,D) was not affected by adding  $CaCl<sub>2</sub>$  in the absence or presence of MgCl<sub>2</sub> but was enhanced similarly by both 0.25 and 1 mM MgCl<sub>2</sub>. \* $P<0.05$  added CaCl<sub>2</sub> vs. no added CaCl<sub>2</sub> at a given MgCl<sub>2</sub>;  $^{\#}P<0.05$  added MgCl<sub>2</sub> vs. no added MgCl<sub>2</sub> at a given CaCl<sub>2</sub>.