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## Approaches for Enhancing Oral Bioavailability of Peptides and Proteins

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### Abstract

Oral delivery of peptide and protein drugs faces immense challenge partially due to the gastrointestinal (GI) environment. In spite of considerable efforts by industrial and academic laboratories, no major breakthrough in the effective oral delivery of polypeptides and proteins has been accomplished. Upon oral administration, gastrointestinal epithelium acts as a physical and biochemical barrier for absorption of proteins resulting in low bioavailability (typically less than 1–2%). An ideal oral drug delivery system should be capable of a) maintaining the integrity of protein molecules until it reaches the site of absorption, b) releasing the drug at the target absorption site, where the delivery system appends to that site by virtue of specific interaction, and c) retaining inside the gastrointestinal tract irrespective of its transitory constraints. Various technologies have been explored to overcome the problems associated with the oral delivery of macromolecules such as insulin, gonadotropin-releasing hormones, calcitonin, human growth factor, vaccines, enkephalins, and interferons, all of which met with limited success. This review article intends to summarize the physiological barriers to oral delivery of peptides and proteins and novel pharmaceutical approaches to circumvent these barriers and enhance oral bioavailability of these macromolecules.

### Keywords

oral delivery; proteins; peptides; nanoparticles; microparticles; GI tract

## 1. Introduction

Delivery via oral route remains the most preferred mode of drug administration. More than 60% of conventional small molecule drug products available in the market are administered

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via oral route (DeVane, 2004). However, this does not hold true for protein and peptide drugs. Effective delivery of peptides and proteins via oral route still remains unaccomplished. The journey of protein drugs has started in 1920, with the use of bovine or porcine insulin for the treatment of diabetes. Recent advancement in the field of molecular biology and biotechnology led to the development of several large molecular weight therapeutic protein and peptides. At present there are more than 40 peptide and protein drugs in the market worldwide with approximately 270 peptides in clinical phase and 400 peptides in advanced preclinical phase testing (Marx, 2005). The 2008 PhRMA report on "Biotechnology Medicines" identified 633 biotechnology products in developmental stages, of which 59 for autoimmune diseases, 254 for cancer, 34 for HIV/AIDS and related conditions, 162 for infectious diseases, 25 for cardiovascular disease, and 19 for diabetes and related conditions.

Typically, a peptide consists of a chain of amino acids with amide bonds. When the peptide chain folds into a three-dimensional configuration, it is called as protein (Lehninger et al., 2005; Sheppard, 1981). Peptides and proteins offer several advantages as compared to conventional drugs. These include high activity, high specificity, low toxicity, and minimal nonspecific and drug-drug interactions. Apart from these peptides and proteins have become the drugs of choice in certain disease states such as enzyme deficiency, genetic and degenerative disease and protein-dysfunction. Developments in the field of biotechnology resulted in large-scale production of peptides, hormones and vaccines in an economical manner (Fix, 1996). Most of the commercially available protein formulations are delivered via traditional routes such as intramuscular (IM), subcutaneous (SC), or intravenous (IV) injections because of their poor oral bioavailability. Oral route of administration has several advantages which include: patient compliance, ease of administration and reasonably low cost of production. Low oral bioavailability of macromolecular drugs stems mainly from pre-systemic enzymatic degradation and poor penetration across the intestinal membrane (Hamman et al., 2005). Development of a viable oral drug delivery system for proteins and peptides requires a careful consideration of their physicochemical properties (molecular weight, pH stability, hydrophobicity, molecular size, and ionization constant) and biological barriers (proteolysis in stomach, variable pH, poor permeation and membrane efflux) that restrict absorption from the gastrointestinal (GI) tract (Mahato et al., 2003). Despite these challenges some polypeptide drugs such as desmopressin and cyclosporin A have been developed as oral dosage forms (Mahato et al., 2003). Major pharmaceutical companies and academic institutions have intensified their research towards oral peptide and protein delivery during the past few decades (Shaji and Patole, 2008). This review article summarizes the physiological barriers to oral delivery of peptides and proteins and provides an insight into novel pharmaceutical approaches to improve oral bioavailability of therapeutic proteins.

## 2. Barriers to oral delivery of peptide and protein drugs

Following oral administration gastrointestinal epithelium acts as a physical and biochemical barrier to absorption of permeants. The physical barrier is mainly represented by impermeable gastrointestinal (GI) epithelium, while the biochemical barrier comprises of enzymatic degradation by peptidases. Indeed it is necessary to understand these barriers for achieving optimal delivery of proteins and peptides (Catnach et al., 1994). GI tract exhibits site specific absorption based on the nature of drugs and regional differences such as pH, enzyme activity, thickness of mucosa, residence time and surface area (Kompella and Lee, 2001). The length of GI tract is ~20 feet and it is broadly divided into two segments. The upper segment mainly consists of mouth, pharynx, esophagus and stomach while the lower segment consists of small intestine (duodenum, jejunum and ileum) large intestine (cecum, colon and rectum) and anus. It consists of 4 concentric layers:

*Mucosa*: It is the innermost, mucus secreting layer which contains many projections (villi) responsible for absorption of food and drug substances. This layer is further divided into epithelium, lamina propria and muscularis mucosa. These cells mainly secrete pepsinogen, hydrochloric acid, and gastric lipase.

*Submucosa*: It consists of a connective tissue with large blood vessels, lymphatics, and nerves branching into the mucosa and muscularis externa.

*Muscularis externa*: It is made up of longitudinal and circular muscle fibers. The longitudinal fibers shorten the tract, while the circular fibers prevent food from traveling backward and propel the balled-up food through the GI tract.

*Serosa*: It is also known as adventitia. This consists of several epithelial layers and forms an external protective coat.

Bioavailability of protein and peptide molecules depends on their ability to cross the intestinal mucosa and reach the systemic circulation (Johnson, 1994; Kwan, 1997).

The pH of GI tract varies from 1–7, with stomach pH between 1–3, duodenum pH between 6.0–6.5, and large intestine pH from 5.5–7.0 (Van de Graaff, 1986). Protein absorption through the stomach is limited by several factors such as low surface area, action of pepsin and harsh degradative acidic environment (Kompella and Lee, 2001). Intestinal epithelium is made up of phospholipid bilayer membrane and cholesterol. Upon oral administration drug molecules must traverse through this lipoidal membrane before entering into systemic circulation. Small intestine is responsible for absorption of more than 90% of nutrients (carbohydrates, proteins, lipids, water, vitamins and minerals), while the rest are absorbed in the stomach and large intestine. The microvilli present on the absorptive mucosal cells of small intestine provide extended surface area for nutrient absorption following which they enter the bloodstream or lymphatic circulation (Tortora and Grabowski, 1996). However, capillary drug absorption eventually results in first-pass metabolism by the hepatic enzymes. Therefore, absorption through Peyer's patches in the ileum that consists of lymph nodes can be explored as a potential alternative for protein and peptide drugs (Mahato et al., 2003; Shakweh et al., 2004). Compounds absorbed through the lymphatic system enter the blood circulation via thoracic duct. By this approach, first pass metabolism by the liver can be mostly eliminated.

The inner wall of small intestine is made of mucosa which consists of ~1  $\mu\text{m}$  long projections or evaginations called microvilli, mucus secreting goblet cells, secretin secreting enteroendocrine cells and lysozyme secreting Paneth cells. Most of the nutrients (lipids, proteins, and carbohydrates) undergo digestion and absorption from the small intestine and hence can be considered as a potential absorptive site for protein and peptide drugs. Moreover, Paneth cells are phagocytic in nature and can aid in the uptake of particulate peptides (Repassy and Lapis, 1979).

Besides goblet cells and enteroendocrine cells, enterocytes and M cells are also important for intestinal transport (Yun et al., 2012). Enterocytes line the gastrointestinal tract and M cells are primarily located within the epithelium of Peyer's patches. M cells represent only about 5% of the human follicle-associated epithelium. These cells are capable of delivering proteins and peptides from the lumen to the underlying lymphoid tissues and induce immune responses. On the other hand, M cells are also exploited by some pathogens as a means of host invasion. Moreover, the high endocytotic ability of M cells enables oral delivery of proteins and peptides. The high transcytotic capability of M cells allows transport of a wide variety of substances, including nanoparticles, microparticles etc (Yun et al., 2012). Macromolecules, particles and microorganisms are taken up by M cells through adsorptive

endocytosis via clathrin-coated pits and vesicles, phagocytosis and fluid phase endocytosis (Buda et al., 2005).

The large intestine consisting of the cecum and colon differs from the small intestine. The wall of the large intestine consists of simple columnar epithelium and mucus secreting goblet cells. Large intestine houses more than 400 species of bacteria that help in digestion of polysaccharides and reduction of azo and nitro compounds which in turn are absorbed by passive diffusion (Chien, 1992; Rafii et al., 1990). The bacteria also generate vitamins such as vitamin K, thiamine, riboflavin, and biotin for absorption into the circulation. The colon acts as a suitable absorption site for protein and peptide drugs due to the absence of digestive enzymes and proteolytic activity besides providing longer residence time (Sinha and Kumria, 2003). Vitamin influx receptors can also be exploited for delivery of macromolecules using surface modified particulate systems. Unique bacterial colonization in the colon also offers a platform for oral protein delivery. Intact protein and peptide molecules can be delivered to the colon with pH sensitive, mucoadhesive or azo polymers that can be degraded by the bacteria in the colon (Chourasia and Jain, 2003).

### 3. Intestinal drug transport mechanisms

Drug transport across the intestinal epithelium is mediated by active or passive transport processes (Fig. 1). Mechanism of transport depends mainly on the physicochemical properties of drug molecule. Active transport involves the movement of drug molecules against concentration gradient (i.e. from low to high concentration) by transmembrane proteins with expenditure of ATP molecules. Passive transport involves the diffusion of drug molecules in the direction of concentration gradient (Gibaldi, 1991). The rate of drug transfer is governed by Fick's law of diffusion (Eq. 1).

$$dQ/dt = DKA(C_1 - C_2)/h \quad (\text{Eq. 1})$$

$dQ/dt$  = rate of diffusion

$D$  = diffusion coefficient

$K$  = oil/water partition coefficient of drug

$A$  = surface area of the membrane across which drug transfer occurs

$h$  = thickness of the membrane through which diffusion occurs

$(C_1 - C_2)$  = difference in drug concentrations in area 1 and 2 respectively

Passive diffusion of peptides and proteins can be described by a combination of two processes:

- a. *Paracellular transport*: This process involves the transport of molecules via water filled pores/channels between cells. Approximately 0.01–0.1% of the total intestinal surface area consists of water filled pores. Taking into consideration that the intestinal epithelium has a surface area of  $\sim 2 \times 10^6 \text{ cm}^2$  (Fasano, 1998), paracellular route corresponds to  $\sim 200$  to  $2000 \text{ cm}^2$ . This surface area is sufficient for the absorption of small quantities (pM–nM range) of a protein adequate to exert their biological activity (Salamat-Miller and Johnston, 2005). This route is preferred by low molecular weight hydrophilic compounds such as small peptide fragments generated from the breakdown of proteins. Peptide and protein molecules are hydrophilic in nature with  $\log P$  value  $< 0$ . These molecules enter cells mostly via paracellular route (Pappenheimer and Reiss, 1987). However, the presence of tight junctions or zonula occludens between the epithelial cell layer of

GIT severely limit penetration ability of polar macromolecules (Stella, 2007). The diffusion of polypeptides via paracellular route depends on their physicochemical properties, molecular dimension and overall ionic charge (Pauletti et al., 1997; Salamat-Miller and Johnston, 2005). The bioavailability of drugs decrease rapidly with increase in molecular weight beyond 700 Da (Antosova et al., 2009). Unfortunately, most of the therapeutic proteins have molecular weight much greater than 700 Da and hence exhibit low bioavailability.

The tight epithelial junctions of colon are impermeable to molecules with radii larger than 8–9 Å. However, in case of polypeptides with high conformational flexibility it is possible that even larger molecules can diffuse through the tight junctions (Tomita et al., 1988). Chittchang et al. (Chittchang et al., 2002) studied the effect of secondary structure on the aqueous diffusion of a model peptide poly(L-lysine) through a microporous membrane. This study concluded that the change in secondary structure of poly(L-lysine) from the random coil to the  $\alpha$ -helix did not alter apparent permeability ( $P_{app}$ ) and intrinsic diffusion coefficient ( $D_{aq}$ ). However, the  $\beta$ -sheet conformer significantly lowered  $P_{app}$  and  $D_{aq}$  values. This result was attributed to higher solution viscosity and extended  $\beta$ -sheet structure of poly(L-lysine). In another study, Chittchang et al. (Chittchang et al., 2007) examined the effect of secondary structure and charge of a model polypeptide, poly(D-glutamic acid) on its permeability through negatively charged pores of synthetic porous membranes and Caco-2 cell monolayers. Poly(D-glutamic acid) exists as a highly negatively charged random coil conformer at neutral pH and below pH 5.0 it changes to  $\alpha$ -helix conformer. Transport studies across Caco-2 cell monolayers revealed higher permeability of poly(D-glutamic acid) at pH of 7.4 (Fig. 2), while a completely opposite trend was observed in the moderately hindered diffusion case (Fig. 3). This observation may be due to the effect of electric field that plays a significant role in the permeation of solutes which are small relative to the pores.

However, for large molecules sieving through the pores is dependent mainly on the molecular size which dominates the influence of electric field. This study concluded that charge and secondary structure of polypeptides play a significant role in determining the rate of aqueous diffusion in a hindered diffusion model. Dadoo et al. (Dadoo et al., 2000) studied the permeability of 14 synthetic model peptides labeled with an amino acid fluorophore on rat alveolar cell monolayers cultured on permeable supports. The results indicated that the peptides entered cells primarily via paracellular route and  $P_{app}$  values were inversely proportional to the molecular size. Scientists have investigated the role of paracellular route in the absorption of peptides such as potent analogs of vasopressin octreotide (Jaehde et al., 1994), thyrotropin-releasing hormone (TRH) (Thwaites et al., 1993), salmon calcitonin (Guggi et al., 2003; Lee and Sinko, 2000) and peptidomimetic renin inhibitors (Walter et al., 1995). Novel strategies such as modification of drug molecule and modulation of tight junctions associated with the paracellular pathway were investigated to increase the penetration of macromolecules (Lane and Corrigan, 2006).

- b.** *Transcellular transport:* This process involves the diffusion of drug molecules through the apical and basolateral membranes. This route is ideal for lipophilic drugs which express relatively high affinity for the lipid bilayer of cell membrane. Many theoretical models based on molecular size, charge, hydrogen bonding, conformation and lipophilicity have been developed to study transcellular transport of drugs molecules (Rautio et al., 2008). Since cell membrane consists of lipid bilayer, it is widely accepted that lipophilicity plays an important role in

determining the transport mechanism. However, early *in vivo* studies concluded that the intestinal absorption diminishes when lipophilicity is very high (usually  $\log P > 5$ ) (Catnach et al., 1994). Burton et al. (Burton et al., 1996) studied the effect of lipophilicity, chain length and number of polar groups on the transport of model peptides in Caco-2 cell monolayers. Interestingly it was observed that the permeability of peptide depends on the number of polar groups that require desolvation before diffusion of peptide into the cell membrane rather than lipophilicity as observed in small organic molecules. Corandi et al. (Conradi et al., 1991) studied the relationship between structure and permeability of neutral and zwitterionic peptides prepared from D-phenylalanine and glycine across Caco-2 cell monolayers. The lipophilicity ( $\log P$ ) of peptides varied from  $-2.2$  to  $+2.8$ . The results indicated no apparent correlation between the apparent lipophilicity and observed flux. Moreover, a strong correlation was noted for the flux of neutral series and the total number of possible hydrogen bonds of the peptide with water molecules. These results clearly indicate that the passive transcellular absorption of a peptide depends on the energy required to break water-peptide hydrogen bonds so the molecules can enter the cell membrane.

*Carrier mediated transport:* This mechanism involves the movement of small molecules, or macromolecules via membrane proteins (transporters). This is also known as facilitated diffusion or active transport process. It has been well established that intestinal absorption of di- and tri-peptides occurs via carrier mediated peptide transport systems. The presence of peptide transport system in mammalian gut was first reported by Newey and Smyth in 1959 (Newey and Smyth, 1959). These oligopeptide transporters also help in the absorption of peptidomimetics such as amino- $\beta$ -lactam antibiotics, renin-inhibitors, and angiotensin converting enzyme inhibitors (Bai and Amidon, 1992). Detailed understanding about the structural features of the peptide is required to target these transporters for protein delivery.

## 4. Formulation strategies for increasing the oral bioavailability of peptides and proteins

### 4.1. Absorption enhancers

Absorption enhancers are substances that enhance or promote absorption of drugs for enhancing oral bioavailability (Aungst, 2012; Checkoway et al., 2012; Jitendra et al., 2011; Williams and Barry, 2004). Peptide and protein drugs are hydrophilic in nature with large molecular weight, hence absorption through transcellular and paracellular routes is severely limited (Shaji and Patole, 2008). Absorption enhancers' act by several mechanisms: a) temporarily disrupting the structural integrity of the intestinal barrier, b) decreasing the mucus viscosity, c) opening the tight junctions and d) increasing the membrane fluidity (Liu et al., 1999). These compounds allow therapeutic agents to permeate across biological membranes into systemic circulation and reach the site of action to exert pharmacological effect (Shaji and Patole, 2008). However, the selection of absorption enhancer and its efficacy depends on the physicochemical properties of the protein and peptide drugs, regional differences in intestinal membrane, nature of the vehicle and other excipients. Permeation enhancers should be safe and non toxic, pharmacologically and chemically inert, non-irritant, and non-allergenic (Jitendra et al., 2011; Senel and Hincal, 2001). Various absorption enhancers have been investigated for the enhancement of protein and peptide absorption through the intestinal membrane. These can be grouped into surfactants, chelating agents, bile salts, cationic and anionic polymers, acylcarnitines, fatty acids and their derivatives (Table 1). These enhancers in combinations show synergistic effect than the individual enhancers, particularly with specific combinations such as mixture of sodium salicylate and peanut oil (Yang, 1999). Chelators generally act by complex formation with calcium ions and facilitate permeation of proteins through paracellular opening. Protein and

peptide delivery can be enhanced by micellar protection or membrane disruption by solubilization of membrane phospholipids. Absorption of human calcitonin in rats was significantly increased by co-administration with surfactants, such as sodium myristate, sodium deoxycholate, sodium taurodeoxycholate, sodium lauryl sulfate, quillaja saponin, and sugar esters. These compounds can be used as absorption enhancers for oral peptide and protein absorption (Nakada et al., 1988). However, absorption enhancers are not protein specific, and risk of toxin or allergen import along with the proteins might be possible. It may result in unwanted side effects. For example, calcium chelators can lead to  $\text{Ca}^{2+}$  depletion, which can disrupt actin filaments, modify adherent junctions and reduce cell adhesion (Sood and Panchagnula, 2001).

Chitosans have also been studied for their enhanced permeation effect and their effect depends on the charge density and structural features of chitosan salts such as *N*-trimethyl chitosan chloride (Thanou et al., 2001). More recently, poly unsaturated fatty acids were added to increase cortisol transport in a BBB model through membrane fluidization (Navarro et al., 2011). Peroral delivery of therapeutics such as salmon calcitonin, interferon  $\alpha$ , insulin, and recombinant human growth hormone have been shown to be improved by the use of *N*-acetylated, non- $\alpha$  aromatic amino acids and *N*-acylated,  $\alpha$ -amino acids. This effect has been attributed to non-covalent interaction between peptide and protein drugs and permeation enhancers (Leone-Bay et al., 2001; Sood and Panchagnula, 2001). Prolonged use of salicylates, chelators, bile salts and surfactants may cause local irritation and mucosal damage (Sithigorngul et al., 1983; Swenson et al., 1994; Yamamoto et al., 1992). Chitosans (Artursson et al., 1994; Schipper et al., 1996; Schipper et al., 1999), acylcarnitines (Lecluyse et al., 1993) and dihydrofusidates (Hurni et al., 1993) appear to be relatively safer than other absorption enhancers. Since the oral route of administration is the most convenient and acceptable method of chronic drug administration, studies have been performed to examine insulin absorption from small intestine. Mixed micellar solutions were examined as a means for enhancing the fluidity of the mucosal membrane and subsequent increase in insulin absorption. Potential role of bile salt/fatty acid mixed micelle in gastrointestinal absorption was also explored. A mixture of sodium glycocholate and linoleic acid in the ratio of 3:1 was used in the preparation of mixed micelles. Use of mixed micellar solutions significantly increased the insulin permeability through the everted gut sacs in the duodenum, medial and distal jejunum but there was insignificant change in the ileum. Insulin absorption was significantly enhanced in the duodenum and jejunum, with the highest overall apparent permeability observed in the jejunum, where addition of mixed micelles to the buffer solution containing insulin increased  $P_{app}$  from about  $7 \times 10^{-7}$  to  $(1.5-1.8) \times 10^{-6}$  cm/s (Fig. 4) (Schilling and Mitra, 1990). Interestingly, experiments with intact gut sacs showed no significant degradation of insulin exposed to either the mucosal or serosal tissues. Mixed micelle solutions, employed as absorption enhancers significantly enhanced cumulative amount of insulin transported across the intestinal mucosa.

A recent study investigated the combined effect of absorption enhancers and electrical assistance on transbuccal delivery of salmon calcitonin (sCT) using fresh swine buccal tissue (Oh et al., 2011). The effect of enhancers on sCT buccal permeation was investigated in the presence of 10% of ethanol, various concentrations (1%, 2% and 5%) of *N*-acetyl-L-cysteine (NAC), sodium deoxyglycocholate (SDGC), and a mixture of enhancers. The absorptive flux of sCT in the ethanol group was enhanced compared to the control group, although not statistically significant. Presence of NAC enhanced sCT permeation, and the enhancing effect was not significant but apparently NAC concentration dependent. Further, the optimal concentration of SDGC for transbuccal sCT delivery was reported to be about 1% SDGC. The combination of 1% SDGC with 10% ethanol and 5% NAC with 10% ethanol showed the maximum drug permeation.

It is well known that oral delivery of proteins and peptides generally involves the coadministration of enzyme inhibitors and/or penetration enhancers, to inhibit proteolytic enzymes and promote the macromolecule permeation, respectively (Sinha et al., 2007). Controlling the release of enzyme inhibitors and/or penetration enhancers prior to the release of a macromolecule might enhance the effectiveness of these adjuvants and establish a more favorable environment. Therefore, a two-pulse delivery system was recently reported which could facilitate adjuvant(s) release prior to the drug release and reduce susceptibility to enzymatic degradation and/or improve mucosal permeation (Del Curto et al., 2011). This delivery platform consisted of (a) drug in a fast disintegrating core, (b) an inner low viscosity HPMC coating, (c) an intermediate layer composed of enzyme inhibitor/absorption enhancer, and (d) an outer additional HPMC coating (Fig. 5). An external gastroresistant film would also be envisaged to overcome the highly variable stomach residence and facilitate time-dependent colon delivery. The outer HPMC coat assists in delaying the adjuvant release from the intermediate layer offering the *in vivo* lag phase of 3–4 hr in colon, while the inner low viscosity HPMC coating is expected to delay the release of macromolecule with respect to that of the adjuvant.

#### 4.2. Enzyme Inhibitors

Co-administration of protease inhibitors can lower the enzymatic barrier and prevent degradation of proteins and peptides in the GI tract thereby facilitating intestinal absorption (Mahato et al., 2003; Malik et al., 2007; Werle et al., 2009). Structure of the molecule and their specificity towards enzymes is crucial to evaluate the stability of protein and peptide drugs (Bernkop-Schnurch, 1998). Inhibitors such as aprotinin (trypsin/chymotrypsin inhibitor), amastatin, bestatin, boroleucine, and puromycin (aminopeptidase inhibitors) have been widely employed. Naturally occurring inhibitors are comparatively non-toxic and among them aprotinin has been widely used for oral peptide delivery. Co-administration of insulin along with trypsin and chymotrypsin inhibitors enhanced oral bioavailability of insulin (Danforth and Moore, 1959; Kidron et al., 1982; Laskowski et al., 1958; Yamamoto et al., 1994). Yamamoto et al. studied the effect of enzyme inhibitors such as sodium glycocholate, camostat mesilate, bacitracin, soybean trypsin inhibitor, and aprotinin on the intestinal metabolism of insulin in rats. Sodium glycocholate, camostat mesilate, and bacitracin were found to be more efficient in improving the physiological availability of insulin in the large intestine. A profound increase (22-fold) in the oral bioavailability of pentapeptide enkephalin YAGFL [Tyr-Ala-Gly-Phe-Leu], with D-conformation of alanine and leucine amino acids was shown in the presence of peptidase inhibitor amastatin (Lee and Amidon, 2002). Most natural inhibitors have to be co-administrated excessively in large amounts because these compounds are susceptible to enzymatic degradation in gut. Even large amounts of these inhibitors may not be adequate to reduce protease activity which may necessitate the encapsulation of proteins and peptides in nano-drug delivery systems. On the other hand, chronic and prolonged usage of these inhibitors may result in high toxicity. It may also affect the absorption of other proteins that normally would be degraded. A major drawback associated with these inhibitors is that the non-site specific activity of such compounds will noticeably alter the metabolic pattern in the GI tract and intestinal membrane primarily due to reduced digestion of food proteins (Watanabe et al., 1992).

#### 4.3. Hydrogels

Hydrogels are three-dimensional mesh like networks containing hydrophilic polymers that imbibe large amounts of water and form a gel like matrix as a result of physical or chemical cross-linking of individual polymer chains. Synthetic hydrogels present a possibly efficient and convenient way to administer peptides and proteins (Ichikawa and Peppas, 2003; Peppas et al., 2000; Ridgley and Wilkins, 1991). Hydrogels can be broadly classified into i) neutral hydrogels, and ii) ionic hydrogels. Ionic hydrogels are composed of pendant groups that



ionize in response to variation in surroundings, making the hydrogel network bulge as it becomes more hydrophilic. Hydrogels that are capable of responding to surrounding stimuli such as temperature, ionic strength variation and pH alteration are known as physiologically-responsive hydrogels (Lowman et al., 1999; Peppas et al., 2000). Hydrogels made from natural polymers are biodegradable and biocompatible. However, these materials do not have adequate mechanical strength and may contain pathogens which may elicit auto-immune response. Synthetic hydrogels do not exhibit these intrinsic bioactive properties and can be tailored to yield hydrogels with desired degradability and functionality. There are two distinct groups of hydrogels, viz. performed and in-situ forming gels. Performed hydrogels are simple viscous solutions which do not undergo any alterations in their structure or properties after administration. While, *in situ* forming gels undergo gelation on administration due to changes in physicochemical properties depending on the environment (Gratieri et al., 2010; Nanjawade et al., 2007). The polymer network can be either homopolymers or copolymers, with the chemical structure determining the properties of hydrogel. Monomers widely used in preparation of hydrogels for protein delivery are 2-hydroxyethyl methacrylate, ethylene glycol dimethacrylate, N-isopropyl acrylamide, acrylic acid and methacrylic acid. Poly(ethylene glycol) (PEG) and poly(vinyl alcohol) are two other polymers that have been employed in hydrogels. One important parameter to be considered is the mesh size which is the distance between cross-links in the hydrogel network. Any change in the mesh size may modify the diffusion pattern of a therapeutic protein from the hydrogel matrix (Morishita et al., 2002; Torres-Lugo et al., 2002). Ichikawa et al. (Ichikawa and Peppas, 2003) designed poly [methacrylic acid-grafted-poly (ethylene glycol)] [P(MAA-g-EG)], a complexation hydrogel for oral delivery of insulin. This study examined cytotoxicity and insulin-transport enhancing effect of P(MAA-g-EG) hydrogel microparticles on intestinal epithelial cells. Results from these studies revealed that the P(MAA-g-EG) hydrogel microparticles are cytocompatible and have transport-enhancing effect of insulin on intestinal epithelial cell.

Recent study by Kamei et al. (Kamei et al., 2009a) on complexation of hydrogels P(MAA-g-EG) have shown high insulin encapsulation efficiency and rapid insulin release in the intestine in a pH-dependent manner. Further studies were carried out with other protein drugs such as calcitonin and interferon  $\beta$ . Incorporation efficiency of hydrogels were high with insulin, calcitonin, and interferon  $\beta$ . Moreover, polymeric microparticles loaded with calcitonin and interferon  $\beta$  displayed complexation/decomplexation and pH-sensitive release pattern. A dose-dependent augmentation of plasma interferon  $\beta$  levels and drastic reduction in plasma calcium levels accompanied with calcium absorption were observed after administration of particles loaded with interferon  $\beta$  or calcitonin into closed rat ileal segments. These findings imply that P(MAA-g-EG) hydrogels can be utilized as potential carriers for oral delivery of peptides and proteins.

#### 4.4. Mucoadhesive systems

These systems comprise of synthetic or natural polymers that can bind (adhere) to biological substrates such as mucosal membranes. The phenomenon of bioadhesion allows a greater amount of drug to be available at the target site resulting in desired therapeutic effect. The ability of the mucoadhesive polymers to adhere to mucin layer on the mucosal epithelium can improve oral bioavailability of protein and peptide therapeutics. Drug delivery systems comprising bioadhesive polymers are known to reduce the rate of clearance of drug molecules from the absorption site, thus prolonging the time available for absorption. Bioadhesive drug delivery systems also offer a controlled release of drugs and thus can reduce the frequency of drug administration. Increased oral bioavailability via delayed gastrointestinal transit induced by bioadhesive polymers was shown for the first time by

Longer et al (Longer et al., 1985). A mucoadhesive polymer should possess ideal characteristics (Guo and Cooklock, 1996; Peh and Wong, 1999; Shaikh et al., 2011):

- a. It should be hydrophilic in nature and be able to form strong adhesive bonds with mucosal membranes because of the presence of large amounts of water in the mucus layer.
- b. Polymers with a high molecular weight are desirable because they provide more bonding sites.
- c. It should possess optimum surface tension which can enable spreading of polymers onto mucosal/ epithelial cell layer.
- d. It should contain adequate hydrogen bond – forming groups such as -OH and -COOH groups that provide strong adhesive bonds between the entangled polymer chains.
- e. It should be non-irritant, non-toxic and non-allergenic in nature.
- f. It should be chemically inert and may not react with the oral epithelium or the protein/peptide drugs.
- g. The cost of the polymer should not be high, so that the final product remains competitive in the market place.

Mucoadhesive polymers were also found to inhibit proteolytic enzymes and/or modulate the permeability of tight epithelial tissue barriers (Lehr, 1996). Bioadhesive polymers are generally classified into synthetic or semi-natural. Synthetic bioadhesive polymers are either polyacrylic acid or cellulose derivatives. Polyacrylic acid-based polymers include carbopol, polycarbophil, polyacrylic acid, polyacrylate, poly(methylvinylether-*co*-methacrylic acid), poly(2-hydroxyethyl methacrylate), poly(methacrylate), poly(alkylcyanoacrylate), poly(isohexylcyanoacrylate) and poly(isobutylcyanoacrylate). Examples of cellulose derivatives are carboxymethyl cellulose, hydroxyethyl cellulose, hydroxypropyl cellulose, sodium carboxymethyl cellulose, methyl cellulose, and methylhydroxyethyl cellulose. Chitosan and various gums such as guar, xanthan, crylamide-acrylate polymer (PHPA), poly(vinylpyrrolidone), and poly(vinyl alcohol) constitute semi-natural bioadhesive polymers. A wide range of bioadhesive formulations have been investigated for the oral cavity. For instance, luminal degradation of insulin, calcitonin, and insulin-like growth factor-I (IGF-I) by trypsin and chymotrypsin was inhibited by employing carbopol polymers (Bai et al., 1996). Site specific targeting was achieved with lectins, that possess high affinity for carbohydrate binding with  $K_d$  values of  $10^4$ – $10^6$ , high diffusion coefficients, and high resistance to proteolytic breakdown. Lectins prefer binding to receptors on the cell surface rather than mucosal gel layer (Haas and Lehr, 2002). A significant enhancement in intestinal absorption of 9-desglycinamide, 8-arginine vasopressin (DGAVP) was observed in rats using the weakly cross-linked poly(acrylate) derivative (polycarbophil) dispersed in physiological saline (Haas and Lehr, 2002). Surface conjugation of the bioadhesive agent, tomato lectin demonstrated higher intestinal uptake of orally administered inert nanoparticles in rats (Hussain et al., 1997). Peptide and protein drugs formulated with chitosan–EDTA conjugates inhibited peptide and protein drugs from enzymatic degradation across the GI tract and greatly enhanced their oral bioavailability (Bernkop-Schnurch and Krajicek, 1998). Binding patterns associated with wheat germ agglutinin (WGA) and peanut agglutinin (PNA) to glycoproteins in human and rodent colon were examined in gastrointestinal diseases. The authors also investigated the feasibility of utilizing N-(2-hydroxypropyl)methacrylamide (HPMA) copolymer-lectin-drug conjugates to deliver therapeutic agents. This report suggested that HPMA copolymer-lectin-drug conjugates provide site-specific treatment of conditions such as colitis and Barrett’s esophagus (Wroblewski et al., 2000).

Thiolated polymers (thiomers) have been widely employed as mucoadhesive polymers for enhancing oral delivery of hydrophilic macromolecular drugs (Bernkop-Schnurch et al., 2004). Unlike other mucoadhesive polymers, thiomers form covalent bonds with cysteine-rich subdomains of mucus glycoproteins via thiol/disulfide exchange reactions. These polymers form stronger covalent interactions than other non-covalent bonds such as hydrogen bonds, Van der Waal's forces and ionic interactions of polymer with anionic substructures of the mucus membranes (Bernkop-Schnurch et al., 2006; Iqbal et al., 2012). Furthermore, thiolated polymers act as enzyme inhibitors, permeation enhancers and efflux pump inhibitors. These polymers are also capable of protecting the incorporated peptides and protein drugs against enzymatic degradation in the intestine. However, stability of thiomers in solutions and gels is a major concern which may reduce the efficacy of thiomers. These polymers are susceptible to early oxidation (pH 5) unless protected under inert conditions. A recent study aimed at utilizing pre-activated thiol groups to facilitate better stability, prolong retention time of dosage forms, offer mucoadhesion in order to enhance uptake and oral bioavailability (Iqbal et al., 2012). Poly(acrylic acid)-cysteine-2-mercaptionicotinic acid (PAAcys-2MNA) conjugates were synthesized by the oxidative disulfide coupling of PAA-cys (100-, 250- and 450 kDa) with 2-mercaptionicotinic acid (2MNA). *In vitro* mucoadhesion studies revealed that immobilization of thiol groups on PAA (100, 250 and 450 kDa) exhibited 1.7-, 2.5- and 452-fold improvement in mucoadhesive properties, respectively. Tablets based on PAA-cys-2MNA (100, 250 and 450 kDa) conjugates displayed 5.0-, 5.4- and 960-fold improvement in the mucoadhesion time relative to corresponding unmodified PAAs (Fig. 6). Results from *in vitro* permeation studies displayed the permeation enhancement ability for preactivated thiomers and was ranked as follows: PAA(450)-Cys-2MNA (h) > PAA(250)-Cys-2MNA (h) > PAA(100)-Cys-2MNA (h) on both Caco-2 cells and rat intestinal mucosa. Also, the apparent permeability of sodium fluorescein was observed to be 5.08-fold higher in Caco-2 cells for PAA(450)-Cys-2MNA (h) and 2.46-fold higher on intestinal mucosa for PAA(450)-Cys-2MNA (m), respectively, relative to sodium fluorescein in buffer only (Wang et al., 2012). Such enhancement in permeability as well as better stability render preactivated thiomers as promising macromolecular permeation enhancers and mucoadhesive polymers and may be suitable for non-invasive drug administration.

#### 4.5. Liposomes

Liposomes are defined as microscopic vesicles composed of one or more phospholipid bilayer membranes with a diameter ranging from 0.0 – 10µm. These are lipid-based delivery systems which offer some degree of protection for peptides and proteins in GI tract (Li et al., 2010). Liposomes have been successfully applied in the delivery of wide range of therapeutics including nucleotides, proteins and plasmids (Kurz and Ciulla, 2002). Degradation of proteins and peptides can be prevented by encapsulation in liposomal bilayers. Depending on the size, liposomes are classified into small uni-lamellar vesicles (SUV) (10–100nm), large uni-lamellar vesicles (LUV) (100–300nm) and multi-lamellar vesicles (more than one bilayer). Liposome consists of aqueous inner core enclosed by a membrane, composed of phospholipids. Hydrophilic drugs can be encapsulated in the inner aqueous core while the hydrophobic drugs tend to stay in lipid bilayer (Kaur et al., 2004). Vesicles are made of naturally derived phospholipids such as egg phosphatidylethanolamine or dioleoylphosphatidylethanolamine (DOPE), phosphotidyl choline or phosphotidyl inositol (Dharma et al., 1986). Among various types of liposomes, dehydrated-rehydrated vesicles are generally used in protein drug delivery due to ease of preparation and low stress exerted on proteins (Kisel et al., 2001). Liposomes can be used for site specific delivery of peptides by decorating surface groups with targeting moieties such as antibodies. Various parameters such as composition of the liposomes, encapsulation efficiency, rate of drug release from lipid vesicles, size and surface charge are important for effective delivery of peptides from

liposomes. Though liposomes are generally considered to be stable carriers, some drawbacks exist such as: a) extensive leakage of water-soluble drugs during GIT passage, b) heterogeneity in the vesicular size, and c) instability of formulations. Scientists have come up with new liposomal formulations viz., archeosomes which could address these issues. Archeosomes are liposomal formulations that are prepared with archaeobacterial membrane lipids mainly composed of diether and/or tetraether lipids (Patel et al., 2000). These archaeobacterial lipids present unique features over conventional liposomes (Sprott, 1992). Archeosomes are proven to be substantially more stable against extreme pH and oxidation, and also against the action of bile salts and lipases (Sprott et al., 2003). Li et.al (Li et al., 2010) studied the potential of archeosomes as vehicles for oral delivery with insulin as a model peptide. Archeosomes were prepared from polar lipid fraction E (PLFE) purified from *Sulfolobus acidocaldarius*. The pharmacodynamics of insulin encapsulated archeosomes was evaluated in diabetic mice following oral administration. Archeosomes made of PLFE were comparatively stable in simulated GI tract conditions and also facilitated slow passage of fluorescently labeled peptide in the GI tract *in vivo*. Archeosomes controlled blood glucose levels more tightly relative to conventional liposomes. Hence archeosomes can be considered as better carriers for protein and peptide delivery due to higher stability. Mohanraj et.al (Mohanraj et al., 2010) developed hybrid silica-liposome nanocapsule (SNLC) system containing insulin. Capabilities of silica-liposome nanocapsule to shield against lipolytic degradation and prolong insulin release in simulated GI conditions were studied. These new carriers protected insulin from enzymatic degradation in presence of digestive enzymes. Liposomal drug release kinetics and stability of vesicles can be controlled by coating with a specifically tailored nanoparticle layer. SNLC is a promising candidate for storage and delivery of proteins and peptides. Thirawong et.al (Thirawong et al., 2008) prepared self-assembling pectin-liposome nanocomplexes (PLNs) by mixing cationic liposomes with pectin solution for improving intestinal absorption of calcitonin (eCT). The eCT-loaded PLNs exhibited good pharmacological action compared to eCT solution and eCT-loaded liposomes. The enhancing effect was attributed to the ability of pectin to adhere to the mucosal layer and extend the retention with the intestinal mucosa.

#### 4.6. Nanoparticles

Nanoparticles are colloidal carriers with size ranging between 1 to 1000 nm, nevertheless, particles with size greater than 200 nm are profoundly pursued (Kreuter, 1994; Yun et al., 2012). Nanoparticles can be broadly classified into two types, nanospheres and nanocapsules. Nanospheres are matrix systems in which drug is uniformly physically dispersed, whereas, nanocapsules are vesicles in which drug is encapsulated by a polymeric membrane. The physicochemical and drug release properties of nanoparticles vary with the preparation method (Singh and Lillard, 2009; Yun et al., 2012). Compared to other colloidal carriers such as liposomes and micelles, most of the nanoparticles are stable in the harsh GI environment. They can be tailor made to target certain tissue and achieve controlled drug release by altering the polymer features and surface chemistry (Panyam and Labhasetwar, 2003). Nanoparticles are taken up by cells via endocytosis process, which includes three subtypes phagocytosis, pinocytosis, and receptor-mediated endocytosis. Phagocytosis involves the assimilation of materials up to 10  $\mu\text{m}$  in diameter especially by macrophages, neutrophils, and dendritic cells. Pinocytosis is a cellular uptake mechanism, generally involves absorption of sub-micron material and substances in solution and its conducted by all cell types. Figure 7 explains the process of endocytosis and fate of nanoparticle after internalization into cell cytoplasm. First, the nanoparticles associate with the cell membrane and subsequently are endocytosed. Then nanoparticles escape from endosomes and degrade in the lysosome. Finally, therapeutic agent diffuses out from lysosome into cytoplasm and transport of therapeutic agent to target organelle takes place which is then followed by

exocytosis of nanoparticles (Faraji and Wipf, 2009). Captivating the benefit of varying pH in the GI tract, pH sensitive nanoparticles can be tailor-made to deliver peptides and proteins to different parts of the intestine. Such nanoparticles are essentially prepared with either polyanionic or polycationic polymer and their mixtures.. Mechanism of drug release from nanoparticles is mainly based on the drug dissolution property, swelling pattern of polymer or both of these at a particular pH. Nanoparticles can enhance drug stability, augment mucoadhesion, extend the residence time in GI tract, enhance intestinal permeability and improve the saturation solubility and dissolution rate. Most pH sensitive carriers have been widely used as enteric coating materials for a prolonged period of time and their safety has been approved. More recently, diethylene triamine pentaacetic acid (DTPA) is a complexing agent, known to disrupt intestinal tight junctions and prevent intestinal proteases by chelating divalent metal ions. Su et.al has made an attempt to incorporate DTPA in functionalized nanoparticles (NPs) for oral delivery of insulin. DTPA was conjugated to poly( $\gamma$ -glutamic acid) ( $\gamma$ PGA) to maintain the complexing agent concentrated on the intestinal mucosal surface, where enzyme inhibition and paracellular permeation enhancement are vital. NPs were prepared by mixing anionic  $\gamma$ PGA-DTPA conjugate and cationic chitosan (CS). The  $\gamma$ PGA-DTPA conjugate inhibited the activity of intestinal proteases significantly, and made a transient and reversable enrichment of paracellular permeability. The NPs were responsive to pH alterations, CS/ $\gamma$ PGA-DTPA NPs swelled with increasing pH and disintegrated above pH 7.0. Furthermore, the biodistribution of orally delivered insulin by CS/ $\gamma$ PGA-DTPA NPs in rats was observed by confocal microscopy and scintigraphy. Experimental results showed higher absorption of insulin from CS/ $\gamma$ PGA-DTPA NPs and absorbed insulin was evidently noticed in the kidney and bladder. CS/ $\gamma$ PGA-DTPA NPs have produced a prolonged reduction in blood glucose levels; the oral intake of enteric-coated capsule containing CS/ $\gamma$ PGA-DTPA NPs had shown maximum insulin levels at 4 hr after treatment. The relative oral bioavailability of insulin was approximately 20%. Results from this study clearly indicated the potential role of NPs in delivering insulin by oral route (Su et al., 2012).

Targeted drug delivery is a novel approach for augmenting the oral absorption and hypoglycemic activity of insulin by means of encapsulation in folate-(FA) coupled polyethylene glycol (PEG)ylated polylactide-co-glycolide (PLGA) nanoparticles (NPs; FA-PEG-PLGA NPs). FA-PEG-PLGA NPs (50 U/kg) demonstrated a two-fold surge in the oral bioavailability (twice hypoglycemia) without hypoglycemic shock when compared to subcutaneously administered regular insulin solution. Insulin NPs sustained the blood glucose levels for 24 hr, while, subcutaneous insulin exhibited a transient effect (<8 hr) with a severe hypoglycemic shock. This nanoformulation of insulin is suitable for once-daily administration and would be adequate to regulate blood glucose levels for at least 24 hr (Jain et al., 2012). In a different study, goblet cell-targeting nanoparticles were designed to enhance insulin oral absorption. The insulin loaded NPs were made using trimethyl chitosan chloride (TMC) surface decorated with a CSKSSDYQC (CSK) cell targeting peptide. Rather than unmodified nanoparticles, the CSK peptide on the surface facilitated the uptake process of nanoparticles in villi. Increase in drug permeation across the epithelium and higher internalization of drug was facilitated by clathrin and caveolae mediated endocytosis of goblet cell-like HT29-MTX cells. Orally administrated CSK peptide modified nanoparticles had shown a better hypoglycemic effect with a higher relative bioavailability of 1.5-fold compared to unmodified NPs. Over all, oral delivery of insulin by CSK peptide modified TMC nanoparticles was effective in targeting goblet cells (Jin et al., 2012; Rubinstein, 2012). Insulin loaded PLGA/HP55 nanoparticles were developed to improve the hypoglycemic effect of orally administered insulin. *In vivo* efficacy of nanoparticles was tested in diabetic rats. Upon oral administration (50 IU/kg) to diabetic rats, nanoparticles were able to decrease the blood glucose level rapidly with a maximal effect between 1 and 8 hr. The relative bioavailability of nanoparticles when compared to subcutaneous injection (5

IU/kg) in diabetic rats was  $11.3\% \pm 1.05\%$ . This study revealed that PLGA/HP55 nanoparticles might be used for oral delivery of insulin. Preactivated thiomers are biocompatible and improve mucoadhesion to a great extent. Thiomers nanoparticles are prepared by simple ionic gelation method. *In vivo* studies indicated enhanced bioavailability of protein-based drugs due to thiomers nanoparticulate formulations relative to formulations of non-thiolated polymers (Hauptstein and Bernkop-Schnurch, 2012). Numerous thiomers have been developed and studied in terms of nanoparticulate carrier systems. Considering the low bioavailability of protein and peptide based drugs when administered orally, very encouraging results have been reached with thiomers based nanoparticles. Table 2 shows the outcomes and best features of thiomers for insulin delivery.

Solid lipid nanoparticle (SLN) is another class of nanoparticles, also widely used in oral protein delivery (Almeida and Souto, 2007; Fonte et al., 2012; Yuan et al., 2009). SLN does not involve the use of toxic organic solvent and hence provide improved protein stability during formulation. SLN has demonstrated improved oral bioavailability of several therapeutic proteins such as insulin, calcitonin, and cyclosporine A. After oral administration of insulin loaded SLN to diabetic rats, Sarmiento et al. (Sarmiento et al., 2007) observed a substantial hypoglycemic effect during 24 hr. Relative bioavailability of insulin increased from 1.6% in oral solution to 5% when administered as loaded SLN. Garcia-Fuentes et al. (Garcia-Fuentes et al., 2005) evaluated ability of surface modified lipid nanostructures for oral delivery of salmon calcitonin (sCT) in rats. Following oral administration of sCT-loaded CS-coated nanoparticles, a significant and prolonged reduction in the serum calcium levels were obtained as compared to those obtained for sCT solution. Ability of nanoparticles to improve oral bioavailability of macromolecules by protection from harsh GI environment makes them a promising tool for oral protein delivery. In spite of encouraging results, requirement of high dose and lack of control over delivery hindered development of marketed nanoparticulate formulations.

#### 4.7. Microparticles

Oral bioavailability of peptides and proteins is extremely low due to extreme pH variation, enzymatic degradation by GI fluid and relatively impermeable intestinal epithelium. Various strategies have been employed to increase oral bioavailability of macromolecules. Among them particulate carrier systems such as microparticles have generated significant interest (Siegel and Langer, 1984). Encapsulation of macromolecules inside polymeric carriers not only protect them from enzymatic and pH degradation but also control their release and augment their absorption. Hagan et al. (O'Hagan et al., 1993; O'Hagan et al., 1991) developed PLGA microparticles loaded with cholera toxin B. Following oral administration in mice specific antibody-secreting cells were observed both in the mesenteric lymph nodes and spleen. Maculotti et al. (Maculotti et al., 2009) developed ovalbumin loaded microspheres using chondroitin sulphate/chitosan (CS/CH) polymers for oral delivery using new emulsion-complex coacervation method. Microspheres released ~30% of ovalbumin in 24 hr, while 100% release was observed in the presence of chondroitinase making them suitable for oral administration. Microspheres of mucin and sodium alginate polymers were developed using coacervation and diffusion filling method. Reduction in blood glucose levels were observed after oral administration to diabetic rabbits. Blood glucose lowering effect following oral administration of insulin-entrapped microparticles was identical to subcutaneously injected insulin solution (Builders et al., 2008). Cheng et al. (Cheng et al., 2006) developed responsive magnetic microparticles and evaluated the effect of prolonged GI transit on the bioavailability of insulin in presence of an external magnetic field. Up to 43.8% reduction in blood glucose levels were observed following administration of 100 U/kg of insulin-magnetite-PLGA microparticles in fasted mice. Based on glucose and ELISA assay, external magnetic field for 20 hr resulted in a bioavailability of  $2.77 \pm 0.46$  and

0.87  $\pm$  0.29%, respectively. In the absence of magnetic field, bioavailability values were 0.66  $\pm$  0.56 and 0.30  $\pm$  0.06%, for glucose and ELISA assay, respectively. A significant improvement in the hypoglycemic effect was noticed in mice that were orally administered with insulin loaded magnetite-PLGA microparticles in presence of an external magnetic field, implying that magnetic force can be used to enhance the effectiveness of orally delivered protein therapeutics. Kim et al. (Kim et al., 2005) developed biodegradable microparticles with alginate utilizing a piezoelectric ejection process. Then lectin (wheat germ agglutinin, WGA) was conjugated to alginate microparticles to obtain dual benefits of protective action and mucoadhesive effect from WGA for enhanced oral insulin delivery. The authors concluded that alginate-WGA microparticles enhanced the intestinal absorption of insulin to cause significant drop in blood glucose levels. Jones et al. (Jones et al., 1997) utilized biodegradable and biocompatible polymer, PLGA to encapsulate bacterial and viral proteins, synthetic peptides and plasmid DNA in microparticles. They compared the immune responses elicited following oral and parenteral administration in mice. Precise systemic and mucosal humoral immune responses as well as cell-mediated immune responses have confirmed the potential of this polymer as a vehicle for oral delivery of vaccines. Development of a clinically successful microparticulate formulation for long-term protein delivery systems requires improvement in loading efficiency, control of burst release and precise control over protein release kinetics.

#### 4.8. Cyclodextrins

Cyclodextrins (CD) are cyclic oligosaccharides containing at least 6 D-(+) glucopyranose units attached by  $\alpha$ -(1, 4) glucosidic bonds. These molecules contain hydrophobic inner cavities and hydrophilic outer surfaces (Challa et al., 2005; Kanwar et al., 2011). These cyclic oligosaccharides contain six ( $\alpha$ -CD), seven ( $\beta$ -CD), eight ( $\gamma$ -CD), nine ( $\delta$ -CD), ten ( $\epsilon$ -CD) or more ( $\alpha$ -1,4)-linked  $\alpha$ -D-glucopyranose units. Cyclodextrins containing less than 6 units cannot be formed due to steric hindrance. Higher homologs containing 9 or more glucose units are difficult to purify. However, several kinds of large ring CDs were isolated and purified to acquire relatively large amount of  $\delta$ -CD (cyclomaltonose) with 9 glucose units. These compounds are capable of interacting with a large variety of guest molecules to form non-covalent inclusion complexes. The chair conformation of the glucopyranose units shapes the cyclodextrins in the form of a truncated cone or torus rather than a perfect cylinder (Brewster and Loftsson, 2007). Commercially available cyclodextrins include hydroxypropyl derivatives of  $\beta$ -CD and  $\gamma$ -CD, randomly methylated  $\beta$ -cyclodextrin (RM- $\beta$ -CD), and sulfobutylether  $\beta$ -cyclodextrin sodium salt (SBE- $\beta$ -CD). The ability of CD to interact with biological membranes allow them to serve as potential carriers for delivery of large molecules such as proteins, peptides, and oligonucleotide drugs (Irie and Uekama, 1999). Efflux pumps such as P-gp serve as a barrier for non-specific uptake of peptides causing low peptide bioavailability. Hydrophobic peptides such as cyclosporin A (Augustijns et al., 1993), D, N-acetyl-leucyl-leucylnorleucinal (Burton et al., 1993), valinomycin (Ueda et al., 1993), gramicidin (Loe and Sharom, 1994), and ditekiren (Takahashi et al., 1997) are known substrates for P-gp. Also, P-gp mediated peptide efflux might play an important role in hindering availability to the central nervous system (Sharom et al., 1996). CD can also inhibit the efflux pumps i.e. P-gp and multidrug resistance associated proteins (MRP2). DM- $\beta$ -CD significantly impaired efflux function of P-gp and MRP2 in Caco-2 cell monolayers. However, cell viability and membrane integrity of Caco-2 cell monolayers remained unaltered (Arima et al., 2004). Cyclodextrins have been investigated for their potential to deliver glycosylated calcitonin (modified calcitonin) and somatostatin analog octapeptide octreotide (Sandostatin<sup>®</sup>). Inclusion of peptides in  $\beta$ -cyclodextrin ( $\beta$ -CD) and hydroxypropyl- $\beta$ -cyclodextrin (HP-  $\beta$ -CD) cavity resulted in increased chemical and enzymatic stability of these peptides besides improving absorption. The reason for improved absorption may be due to impairment of tight junctional integrity

which was observed through enhanced permeation of paracellular marker PEG-4000 (Haerberlin et al., 1996). Cyclodextrins have been shown to enhance chemical and physical stability of peptides and proteins through complexation. Thymopentin, a small peptide consisting of five amino acids, was stabilized in aqueous solutions by complexation with (2-hydroxypropyl)- $\alpha$ -cyclodextrin without any loss of pharmacological activity (Brown et al., 1993). Glutathione (GSH) was successfully encapsulated in Eudragit RS-100-based microparticles containing HP- $\beta$ -CD. Presence of CD resulted in enhanced GSH absorption and sustained release of tripeptide. The authors concluded that GSH-loaded Eudragit RS 100 microparticles containing HP- $\beta$ -CD may be suitable for sustained oral delivery of GSH (Trapani et al., 2007). Enzymatic degradation of basic fibroblast growth factor was minimized by employing water-soluble  $\beta$ -CD sulfate (Loftsson and Brewster, 1996).  $\beta$ -CD has been employed in the formulation of alginate microspheres containing insulin for oral delivery. Serum sugar and insulin levels following administration of microspheres (multiple oral doses) indicated absorption of insulin from the GI region. With this method, insulin absorption from optimized microspheres was found to take place from GI (Jerry et al., 2001). Shao et al. (Shao et al., 1994) studied the relative effectiveness of two  $\beta$ -CD derivatives, i.e., dimethyl- $\beta$ -cyclodextrin (DM- $\beta$ -CD) and HP- $\beta$ -CD, as mucosal absorption promoters. Insulin absorption was evaluated in the lower jejuna/upper ileal segments of the rat with an *in situ* closed loop method. Insulin alone and in the presence of 10 % w/v HP- $\beta$ -CD did not show any significant improvement in the systemic absorption. Figure 8, shows the plasma insulin concentration profiles plotted as a function of time.  $AUC_{0-\infty}$  values for insulin alone and in the presence of 10 % HP- $\beta$ -CD were reported to be  $3 \pm 2 \mu\text{U/ml}\cdot\text{hr}$ . Incorporation of 10% w/v DM- $\beta$ -CD to insulin solution showed a significant enhancement in insulin absorption as evidenced by an  $AUC_{0-\infty}$  of  $255 \pm 65 \mu\text{U/ml}\cdot\text{hr}$  and insulin bioavailability dramatically increased from 0.06% (insulin alone) to 5.63% (Fig. 8). Moreover, hypoglycemic effects of insulin were also evaluated in the presence of cyclodextrins. Enteral administration of insulin to lower jejuna/upper ileal segments of the rat resulted in a minimal glucose lowering effect. Addition of 10 % (w/v) HP- $\beta$ -CD to insulin solution did not result in significant hypoglycemic effect. However, incorporation of 10 % (w/v) DM- $\beta$ -CD significantly improved insulin hypoglycemic effect (Fig. 9) (Shao et al., 1994). In addition, histopathological evaluations displayed no observable destruction to the overall tissue integrity indicating high tolerance of the intestinal mucosa to HP- $\beta$ -CD and DM- $\beta$ -CD favoring their use as absorption promoters for oral delivery of proteins and peptides.

A recent investigation evaluated mechanistic aspects on the uptake and intracellular trafficking of novel CD transfection complexes by intestinal epithelial (Caco-2) cells (MJ et al., 2011). This study was directed at investigating the efficacy of a novel poly-6-cationic amphiphilic CD to transfect intestinal enterocytes; the endocytotic uptake pathways and the intracellular trafficking of the CD•DNA complexes, by changing the orientation of the lipid tail on CD. Inhibitors of clathrin- and caveolae-mediated endocytosis and macropinocytosis were employed to evaluate the mechanisms of CD•DNA uptake by both undifferentiated and differentiated Caco-2 cells. Significant levels of pDNA uptake and gene expression (comparable to PEI) was evident in both undifferentiated and differentiated Caco-2 cells via complexation of plasmid DNA (pDNA) with CD. Also, CDs were capable of transfecting the differentiated Caco-2 cells. The uptake of CD•DNA transfection complexes by both undifferentiated and differentiated Caco-2 cells was found to be mediated by macropinocytosis. On the other hand, heparan sulphate proteoglycans demonstrated binding of complexes to undifferentiated Caco-2 cells. This study provides an insight for the development of optimized CD based transfection complexes for intestinal delivery with enhanced cellular uptake and intracellular trafficking properties.



## 5. Other approaches to enhance the oral delivery of peptide and proteins

### 5.1. Site specific delivery to colon

High proteolytic degradation and poor absorption through epithelial barriers pose major challenges to successful oral delivery of protein and peptide therapeutics. One of the approaches to lessen degradation is to deliver protein molecules to a specific region of GI tract such as colon where proteolytic activity is comparatively low (Kumar and Mishra, 2008; Maroni et al., 2012). Though protein and peptides are inactivated in acidic environment of stomach and also by proteolytic enzymes in the small intestine, the colon offers an attractive target for delivery of protein therapeutics due to relatively low abundance of peptidases and presence of alkaline pH. Moreover, prolonged residence time and more responsiveness to absorption enhancer allow higher absorption of proteins through colonic epithelia (Patel, 2011). Colon targeted delivery has been exploited for several protein and peptide drug candidates such as vasopressin, insulin, calcitonin, glucagon and so on (Patel et al., 2007). Recent advances in pharmaceutical technology lead to the development of numerous colon-specific delivery approaches. These approaches include pH and time dependent delivery systems, pressure-induced drug delivery, microflora-activated systems, and particulate drug delivery systems (Gangurde H.H., April–June 2011; Gazzaniga et al., 1994; Hamman et al., 2005). Because of higher pH in the terminal ileum and colon, polymers such as methacrylic acid copolymers, Eudragit L100 and Eudragit S100, which disintegrate preferentially at high pH levels (6.5–7.0), are usually employed for colon-specific delivery. For instance, Touitou et al. (Touitou and Rubinstein, 1986) designed soft gelatin capsules coated with polyacrylic polymer (Eudragit) for colon-specific delivery of insulin. A significant rise in hypoglycemic effect was observed in rats after oral administration of capsule relative to control. In another study, Lowman et al. (Lowman et al., 1999) developed insulin loaded gel particles from pH responsive poly (methacrylic-*g*-ethylene glycol) hydrogels. In acidic environment, stomach prevents the swelling of gel particles protecting insulin from proteolytic degradation. However, in basic environment of small intestine the gel rapidly swells and dissociates to release the entrapped insulin. A strong hypoglycemic effect was observed in both healthy and diabetic rats within 2 hr of oral administration of insulin loaded gel particles (Lowman et al., 1999). Even though, a pH responsive system can protect protein in the stomach, and proximal small intestine, such systems suffer from poor site-specificity causing dissolution and drug release in the lower part of small intestine prior to the colon. Alternatively, polysaccharides and azo-polymers which are refractory in stomach and small intestine while degraded by the colonic microflora can be used. Polysaccharides such as chitosan, dextran, inulin, pectin, chondroitin sulfate have been exploited for colon-specific delivery (Gulbake and Jain, 2012; Shah et al., 2011). Tozaki et al. developed chitosan capsule for improving insulin absorption from the rat colon. To evaluate effectiveness of chitosan capsule in colon-specific delivery, carboxyfluorescein (CF) (a model water soluble compound) release from chitosan capsule was performed in three different dissolution media: artificial gastric juice (pH 1), an artificial intestinal juice (pH 6.8), and a suspension of rat cecal contents (pH 7.0). Figure 10 illustrates release profile of CF from chitosan capsule. When insulin loaded chitosan capsule was orally administered in rats, a strong hypoglycemic effect was observed relative to insulin solution. This effect was attributed to the protection of insulin from proteolytic degradation as well as the ability of chitosan to increase permeability of insulin through colonic epithelia (Tozaki et al., 1997).

Fetih et al. have evaluated applicability of chitosan for colon specific delivery of [Asu 1, 7]-eel calcitonin in rats (Fetih et al., 2006). In this study, [Asu 1,7]-eel calcitonin (ECT) loaded chitosan capsule was orally administered in rat and the % pharmacological availability (PA %) was calculated. Authors have also studied the effect of incorporation of various additives such as absorption enhancers (S-nitroso-N-acetyl-DL-penicillamine (SNAP) and sodium

glycocholate) and protease inhibitors (bacitracin and aprotinin) in the dosage forms on PA%. Following oral administration, PA% for ECT loaded chitosan capsule (PA%: 0.551%) was very high compared to ECT solution (PA %: 0.041%). In addition, the hypocalcemic effect and thereby PA% was further increased significantly by co-administration with various additives and the highest PA % (6.344%) was reached with chitosan capsule loaded with 20 IU ECT, 1.1mg SNAP, 2.5mg sodium glycocholate, 3.5mg bacitracin and 1mg aprotinin. It is evident from the results that chitosan capsule along with various additives significantly improved the efficacy of ECT. In another study, the suitability of chitosan hydrogel beads for colon-specific delivery of larger proteins was evaluated *in vitro* with FITC-BSA as a model protein. *In vitro* release studies in simulated intestinal fluid (SIF) (6 hr) followed by rat cecal and colonic enzyme medium (14 hr) revealed 20% release of BSA in SIF while approximately 50% release was observed in cecal and colonic enzyme medium (Zhang et al., 2002). Use of azopolymer is another widely used approach for colon specific delivery of proteins and peptides via oral route. Azopolymer contains an azo-aromatic group R-C<sub>6</sub>H<sub>4</sub>-N=N-C<sub>6</sub>H<sub>4</sub>-R' that is specifically reduced by microflora present in the colon which ultimately leads to polymer degradation and release of drug in the colon. The concept of azopolymer was first investigated by Saffran et al in 1986 for oral delivery of vasopressin and insulin (Roldo et al., 2007; Saffran et al., 1986). Later, Tozaki et al. developed azopolymer-coated pellets for colon specific delivery of insulin and ECT in rats. For both peptides, maximum biological effects correspond to the presence of pellet in the large intestine. However, some effects were observed prior to pellets reaching the colon indicating the leakage of drug from pellets. Moreover incorporation of protease inhibitors raised the bioavailability. Pellets containing insulin (12.5 IU) with camostat mesilate (24.2 mg) demonstrated approximately 30% reduction in glucose level as compared to pellets containing insulin alone. Similarly, in case of ECT, a greater hypocalcemic effect was observed after incorporation of camostat mesilate in pellets (Tozaki et al., 2001). Saffran et al. exploited this natural phenomenon by using copolymers of styrene and hydroxyl-ethylmethacrylate cross-linked with azo-bonds for oral insulin delivery. Oral administration of insulin containing azopolymer coated pellets to diabetic rats showed greater than 40% reduction in initial glucose levels, however, variable results were obtained after oral administration in normal rats. This may be due to the premature release of drug prior to reaching the colon (Saffran et al., 1986). Though the use of azopolymers looks more appealing for colon-specific delivery of proteins, further refinement in this approach is needed in terms of developing and testing other derivatives of azopolymers. Therefore, colon targeted drug delivery represents a novel concept for improving the systemic bioavailability of orally administered protein and peptide therapeutics. However, colon targeted delivery systems pose multiple challenges such as presence of lower surface area along with tight junctions which hinder drug absorption at the colonic site. In addition, certain disease conditions alter the enzymatic activity of microflora present in the colonic site and subsequently affect the drug release from cargo (Vinay Kumar K.V. et al., 2011).

## 5.2. Chemical modification

Chemical modifications of peptides and proteins have shown exciting results primarily due to enhanced enzymatic stability, intestinal permeability and lesser extent of immunogenicity. Generally therapeutic activity of protein drugs can be optimized up to certain extent by chemical modifications. These chemical modifications may include substitution of existing amino acids with new amino acids (analogue formation), acylation and PEGylation (Frokjaer and Otzen, 2005; Hamman et al., 2005). In analogue approach, modifications are carried out by interchanging of amino acids preferably by substitution of L-amino acid with D-amino acids or by replacement with different amino acids. Desmopressin, a vasopressin analogue, was developed by deamination of N-terminal amino acid and replacement of the C-terminal L649 arginine by D-arginine in vasopressin. Natural vasopressin is orally active

only at very large doses while desmopressin exhibits twice the activity of vasopressin only at 1/75 fraction of the dose, which is mainly attributed to enhanced permeability across intestinal epithelia and improved enzymatic stability (Rado et al., 1976; Vavra et al., 1974; Vilhardt and Lundin, 1986). Thyrotropin releasing hormone (THR) is a tripeptide (Glu-His-Pro), which is resistant to proteolytic degradation. However, oral bioavailability is poor due to high metabolic clearance. MK-771 is an analog of THR synthesized by substituting nitrogen atom with sulfur in pyrrolidine ring 2 of the proline residue of THR, which showed two hundred times higher CNS activity with less metabolic clearance. It was also equipotent to THR in causing release of TSH. However oral bioavailability was only 2% and it was attributed to inefficient membrane transport (Hichens, 1983).

Conjugation of fatty acid with protein and peptides has also been attempted to improve their lipophilicity. Lipidization is reported to improve metabolic stability, membrane permeability, bioavailability, and changes pharmacokinetic and pharmacodynamic properties of several peptide therapeutics (Zhang and Bulaj, 2012). Compared to tetragastrin (TG), acyl, caproyl and lauroyl derivatives of TG have shown marked increase in gastric acid secretion following administration into the large intestinal loop. This increased activity is due to improved permeation of TG from the large intestine (Tenma et al., 1993). In another study, Wang et al. synthesized lipidized salmon calcitonin (REAL-sCT) conjugate by 'reversible aqueous lipidization' (REAL) technology using *N*-palmitoyl cysteinyl 2-pyridyl disulfide as a lipidizing reagent (Wang et al., 2003). The pharmacokinetic and pharmacodynamic aspects of REAL-sCT was studied in rats and compared with normal salmon calcitonin. Following oral administration, oral absorption of sCT in rats was minimal and only at 0.5 hr, plasma sCT levels were detectable with concentration of  $4.6 \pm 2.8$  pg/0.2 ml plasma (Fig. 11). On the other hand, lipidization enhanced absorption of sCT and greater than 7 pg/ml plasma level of the intact sCT was detected at 12 hr after the oral administration of REAL-sCT. The AUC of oral REAL-sCT was at least 19 times higher than oral sCT. Moreover, REAL-sCT also demonstrated anti-bone resorption activity in ovariectomized (OVX) rats following oral administration (Wang et al., 2003). Using REAL technology, Wang et al. have chemically modified opioid peptide leu-enkephalin (ENK) by conjugating with a novel amine-reacting lipophilic dimethylmaleic anhydride analog, 3,4-bis(decylthiomethyl)-2,5-furandione. The REAL-ENK demonstrated enhanced stability in mouse small intestinal mucosal homogenate with 12-fold increment in half-life over ENK. Moreover, after oral administration, significantly higher and sustained plasma peptide levels were detected upto 24 hr in normal mice. The peak concentration and area under the curve of REAL-ENK were 4.4 and 21 times higher relative to ENK. Most importantly, REAL-ENK also produced significant and sustained anti-nociception in a rodent inflammatory pain model (Wang et al., 2006).

Since decades, scientists have rationally designed and synthesized several derivatives of insulin in order to improve oral bioavailability by conjugation with either fatty acid or PEG. Hashizume et al. examined absorption of mono and dipalmitoyl insulin. Palmitoylation of insulin not only enhanced transport of insulin across the mucosal membrane of the large intestine but also improved stability against enzymatic degradation in intestine (Hashizume et al., 1992). Development of hexyl-insulin for oral delivery is one of the biggest achievements of the chemical modification approach. NOBEX Corporation has developed hexyl-insulin-monoconjugate-2 (HIM2) by covalent attachment of PEG-amphiphilic oligomer to free amino group on the lysine residue at 20th position in the  $\beta$ -chain of recombinant human insulin (Park et al., 2011). This modification has enhanced the stability of insulin in presence of gastrointestinal enzymes and facilitated greater absorption. Following oral administration of HIM2 (0.5 and 1.0mg/kg) in type-2 diabetes patients, 30 min prior to meal, lower postprandial plasma glucose levels were obtained during a 4 hr post

dose evaluation period compared to placebo. Even though lower peripheral insulin concentrations were attained after oral HIM2, control of post prandial plasma glucose levels during a 4 hr post dose evaluation period was comparable to subcutaneous insulin (8 units). This data is consistent with the hypothesis that oral delivery of insulin may lead to the development of a portal-to-peripheral insulin gradient similar to the state observed in individuals not administered subcutaneously (Kipnes et al., 2003). Following oral administration of HIM2 in type-1 diabetes patients, plasma glucose levels declined or remained stable for 30 min to 2 hr and were <150% of pre-dose levels throughout the post-dose evaluation period. Moreover, no episodes of symptomatic hypoglycemia were observed. Hence there may be less risk of severe hypoglycemia associated with HIM2 as compared to injectable insulin (Clement et al., 2002).

A recent study investigated the combined approach of chemical modification with application of pH responsive hydrogels for oral delivery of insulin (Tuesca et al., 2009). In this study, authors have developed PEG-insulin conjugate by PEGylation at the amino terminus of the B-chain. The biological activity of PEG-insulin conjugate was evaluated and compared with human insulin following intravenous and subcutaneous dosing in rats. The PEG-insulin conjugate retained complete biological activity as human insulin following intravenous administration in rats. On the other hand, the conjugate exhibited a relatively high hypoglycemic effect 127.8% than human insulin after subcutaneous dosing. Later, authors have loaded PEG-insulin conjugate and human insulin into pH responsive poly(methacrylic acid-g-ethylene glycol) (P(MAA-g-EG)) hydrogel and evaluated *in vitro* release profile. The fractional release of insulin and PEGylated insulin exhibited pH dependant release profile with low release at pH 4.7 and 5.6 which was increased at pH 6.2 and 7.4. However, in all cases, the release of PEGylated insulin was lower than human insulin which was suggested due to a stronger affinity of PEGylated insulin for the hydrogel. The authors suggested that this release pattern could be advantageous for oral insulin delivery because PEGylated insulin would maintain higher concentrations within the hydrogel as it moves down through the GI tract. This combined approach of PEGylated insulin and P(MAA-g-EG) hydrogels have significant promise for improving oral delivery of insulin (Tuesca et al., 2009).

Although PEGylation is considered a gold standard for chemical modifications of proteins drugs, safety and efficacy of newly developed PEGylated proteins need to be considered. PEG polymers are mixture of polymers of different molecular mass and in addition protein contains several accessible sites for PEGylation which allow product heterogeneity. Such product heterogeneities may alter the efficacy of the parent protein drugs. For example, PEGylation of epidermal growth factor at Lys28 and Lys 48 have shown reduced biological activity compared to its parent structure (Lee and Park, 2002).

### 5.3. Prodrug derivatization

Prodrugs are pharmacologically inactive entities; made by chemical modifications, and able to convert into their active forms by enzymatic or non-enzymatic transformation upon crossing barriers (Hsieh et al., 2009; Jana et al., 2010). An extensive research to improve solubility, stability, targetability and permeability of small molecules by prodrug approach has been carried out in recent years (Hsieh et al., 2009; Rautio et al., 2008). However, structural complexity, lack of methodology for synthesis and poor stability during synthesis limits application of this approach for peptides and proteins (R et al., 1997). In spite of these challenges recent studies have shown exciting results for several protein and peptide therapeutics. One of the promising prodrug approaches is bioreversible cyclization which has been utilized by Borchardt et al. (Borchardt, 1999) for improving oral absorption of opioid peptide. The phenylpropionic acid based cyclic prodrugs of (Leu<sup>5</sup>)-enkephaline have shown approximately 1680 fold higher permeability through Caco-2 cell monolayer

compared to (Leu<sup>5</sup>)-enkephaline. In the same study, the authors have synthesized coumarinic acid based cyclic prodrugs of (Leu<sup>5</sup>)-enkephaline which not only exhibited high permeability but also showed good stability in the presence of peptidases. Similar results were obtained with cyclization of kyotrophin. Cyclic kyotrophin showed improved stability and higher permeability on excised small intestine of rat (Mizuma et al., 2000). A recent study documented targeted lipid prodrug strategy which may help in higher absorption of hydrophilic molecules such as peptides and proteins. In this study, a lipid raft has been conjugated to the parent drug molecule to impart lipophilicity. Simultaneously a targeting moiety that can be recognized by a specific transporter on the cell membrane has also been tethered to the other terminal of lipid raft to facilitate active targeting. Results from this study demonstrated that both the targeting and the lipid moiety act synergistically toward cellular uptake. The authors concluded that this novel prodrug design strategy may help in higher absorption of hydrophilic therapeutics such as genes, siRNA, antisense RNA, DNA, oligonucleotides, peptides and proteins (Vadlapudi et al., 2012a).

#### 5.4. Membrane transporter and receptor targeting

Covalent conjugation of carrier molecules to protein and peptide represent a feasible approach for enhancing their intestinal absorption. Presence of a transport-carrier molecule enables recognition by a specific endogenous membrane transporter or a receptor. This strategy has been explored by many researchers to improve oral drug bioavailability. Various transporters have been identified on the intestinal epithelial cells (Kramer et al., 1997; Sood and Panchagnula, 2001; Tsuji and Tamai, 1996; Vadlapudi et al., 2012b). These include (i) amino acid and oligopeptide transporters, (ii) bile acid transporters, (iii) water-soluble vitamin transport systems, (iv) carbohydrate transporters, (v) monocarboxylic acid transporters, and (vi) phosphate transporters. The driving force for these carrier systems is provided through electrochemical gradients of Na<sup>+</sup> and H<sup>+</sup> ions. Inwardly directed ion gradients are a result of Na<sup>+</sup>/K<sup>+</sup>-ATPase (localized on the basolateral membrane) and Na<sup>+</sup>/H<sup>+</sup> exchanger (localized on luminal surface), which are responsible for lower intracellular Na<sup>+</sup> and H<sup>+</sup> ion concentrations in intestinal epithelial cells. Most transport systems are characterized to be sodium (amino acid, vitamin, bile acid, phosphate transporters) and proton (oligopeptide, short chain fatty acid transporters) dependent. For instance, protein or peptide therapeutics conjugated to an amino acid or a dipeptide carrier moiety can aid in recognition of the conjugates by an amino acid or peptide transporter which in turn facilitates enhanced oral absorption. Bile acid and oligopeptide transporter have shown to possess greater capacity (>10g/day), which can be utilized for delivery of protein or peptide therapeutics. However, biological barriers such as efflux transporters may limit the absorption of orally administered drugs including peptides. In such a case, simultaneous co-administration of efflux inhibitors may contribute to significant enhancement in the absorption of peptide drugs that are substrates of efflux proteins (Katragadda et al., 2005; Varma et al., 2003). Membrane transporters generally limit their ability to transport only small molecule drugs. On the other hand, receptors translocate drug molecules via receptor mediated endocytosis and do not limit their translocation based on the size of permeant. Receptor ligands such as lectins, toxins, viral haemagglutinins, invasins, transferrin and vitamins (vitamin B12, folate, riboflavin and biotin), can be conjugated or covalently attached to protein or peptide drug to achieve site specific targeting (Lim and Shen, 2005). Vitamin B12 absorption pathway has been shown to play a prominent role in the absorption of the cobalamin-protein conjugates (Russell-Jones, 2001). Transferrin receptor has been utilized for oral delivery of protein drugs such as insulin and granulocyte colony-stimulating factor (G-CSF) (Russell-Jones, 2004). An expression construct of granulocyte colony-stimulating factor (G-CSF)-transferrin fusion protein (G-CSF-Tf) was developed by fusing human cDNAs encoding G-CSF and transferrin. With the use of transferrin based recombinant fusion protein, the biological activity of orally administered protein was not

destroyed by proteolytic enzymes as compared to that of subcutaneously administered G-CSF. This technology may be valuable for the future development of orally effective biologicals (Bai et al., 2005).

### 5.5. Cell penetrating peptides

Recently, cell penetrating peptides (CPP) have gained a lot of attention in the delivery of small molecules as well as macromolecules, liposomes and nanoparticles into cells by chemical modifications (Chen et al., 2012; Khafagy el and Morishita, 2012; Koren and Torchilin, 2012). CPPs act by transporting their cargoes into the cytoplasm via perturbation of the lipid bilayer of the cell membrane or by endocytosis (Trehin and Merkle, 2004; Zorko and Langel, 2005). CPP linked to insulin (insulin-CPP hybrids) have demonstrated an increase in intestinal absorption efficiency of insulin compared to normal insulin across Caco-2 cell monolayers (Liang and Yang, 2005). Though this strategy lacks supportive *in vivo* data, it can be explored for enhancing therapeutic potential (Vives, 2005).

CPPs such as HIV-1 Tat, penetratin and oligoarginine are considered as a valuable tool for the intracellular delivery of therapeutic peptides and proteins (biodrugs) (Khafagy el and Morishita, 2012). Hence, CPPs may enhance the absorption of biodrugs through intestinal epithelium. CPPs are considered to be powerful tools for overcoming the low permeability of therapeutic macromolecules through the intestinal epithelial membrane, the major barrier to their oral delivery. This promising strategy appears to be advantageous in improving intestinal absorption especially when CPPs are coadministered with drugs. It is postulated that the intermolecular binding interactions between a drug and a CPP are important factors for enhancing the intestinal absorption of macromolecular drugs (Fig. 12). However, the mechanism of uptake of CPP-biodrug conjugate by epithelial cells and its subsequent enzymatic degradation and release into circulation are unclear. Further studies are warranted to delineate the complete mechanisms for the absorption-enhancing effect of CPPs (Khafagy el and Morishita, 2012).

The *in vivo* absorption of bioactive macromolecules conjugated to CPPs was evaluated in mice by Schwarze et al. (Schwarze et al., 1999). Intraperitoneal injection of  $\beta$ -galactosidase protein (120-kDa), fused to the protein transduction domain from the HIV TAT protein, facilitated delivery of the biologically active fusion protein to all tissues including the brain. Also, the potential of CPPs as a permeation enhancer to improve the penetrating ability of poorly permeable macromolecules have been explored (Kamei et al., 2008; Kamei et al., 2009b; Khafagy el et al., 2009a; Khafagy el et al., 2009b; Morishita et al., 2007). Previous reports have shown that CPPs were capable of enhancing peptide and protein delivery across mucosal membranes safely by non-covalent cargo interaction approach. Moreover, CPPs have successfully delivered the most challenging peptide and protein cargos through intestinal mucosa with greater bioavailability. The assembly of cargo/CPP interaction *via* non-covalent interaction strategy appears to be advantageous because it simplifies chemical conjugation protocols. Various cargos can be delivered by simply mixing the CPP with biomacromolecules without the need of optimizing individual synthetic process.

## 6. Conclusion

Oral delivery of peptides and proteins is currently a topic of intense research. Advancements in the biopharmaceutical industry have resulted in the development of several new protein and peptide based therapeutics. Oral administration is most preferred because of patient compliance and acceptability, but the physiological, enzymatic and chemical barriers for this route pose a significant challenge to the delivery of peptide and protein drugs. Better understanding of the routes of drug absorption (paracellular and transcellular), proteolytic enzyme activity, stability and degradation of macromolecules during the intestinal transport

process is important for the development of protein and peptide drug delivery systems. Several approaches including chemical modifications, use of absorption and penetration enhancers, use of mucoadhesive polymers, formulation design, and use of enzyme inhibitors have been investigated to improve their bioavailability. Although these strategies have marginally improved intestinal absorption of macromolecules, they often resulted in undesirable effects. For example, permeation enhancers may cause potential toxicity by modifying the phospholipid bilayers, leaching of proteins from the mucous membrane or destroying the epithelial cell membranes there by leading to irritation and inflammation. Also opening up the tight junctions may allow the entry of unwanted xenobiotics or pathogens. Therefore, there is an unmet need in the development of novel approaches or modification of existing approaches to achieve site specific and controlled delivery of protein and peptide drugs besides preserving their biological and therapeutic properties. Oral delivery of peptides and proteins can be facilitated by a thorough understanding of the primary, secondary and tertiary structures of macromolecules on a case-by-case basis.

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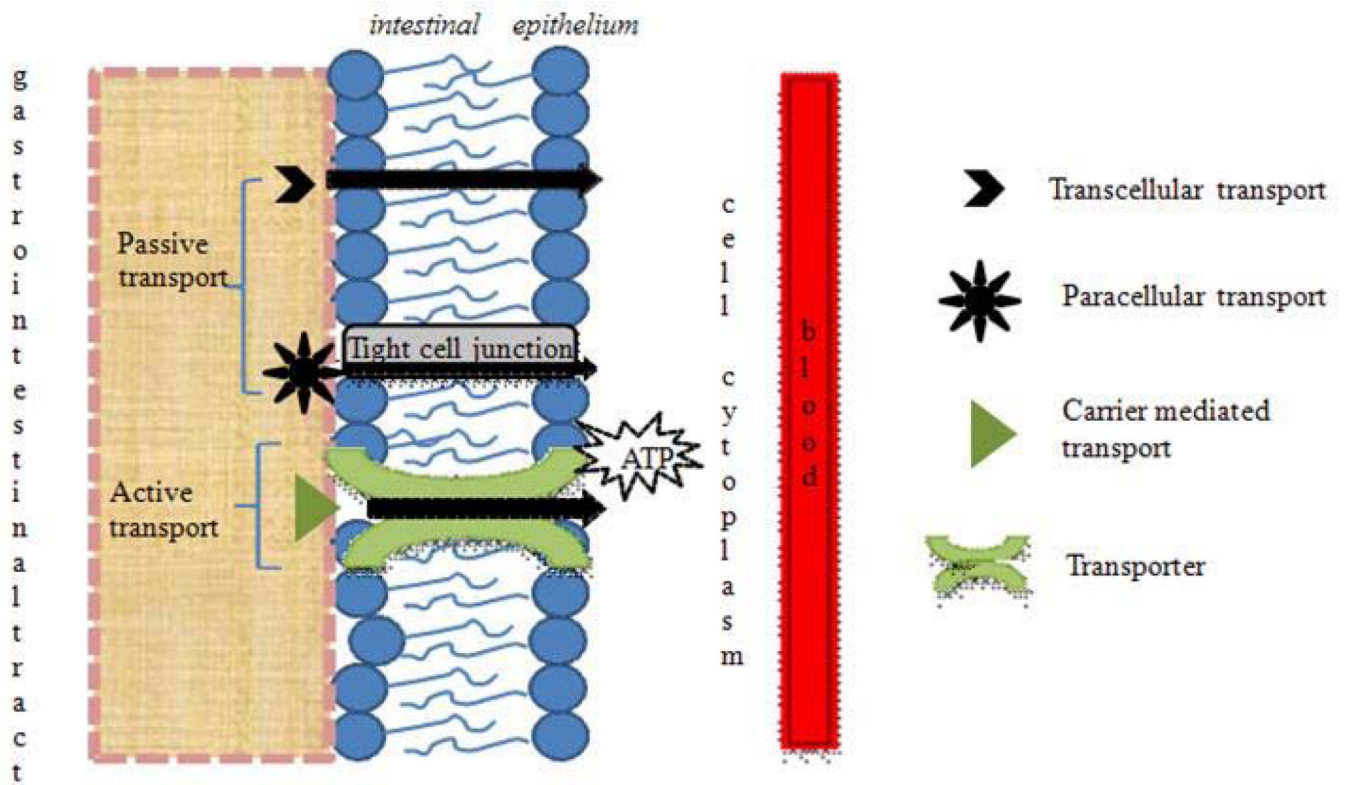
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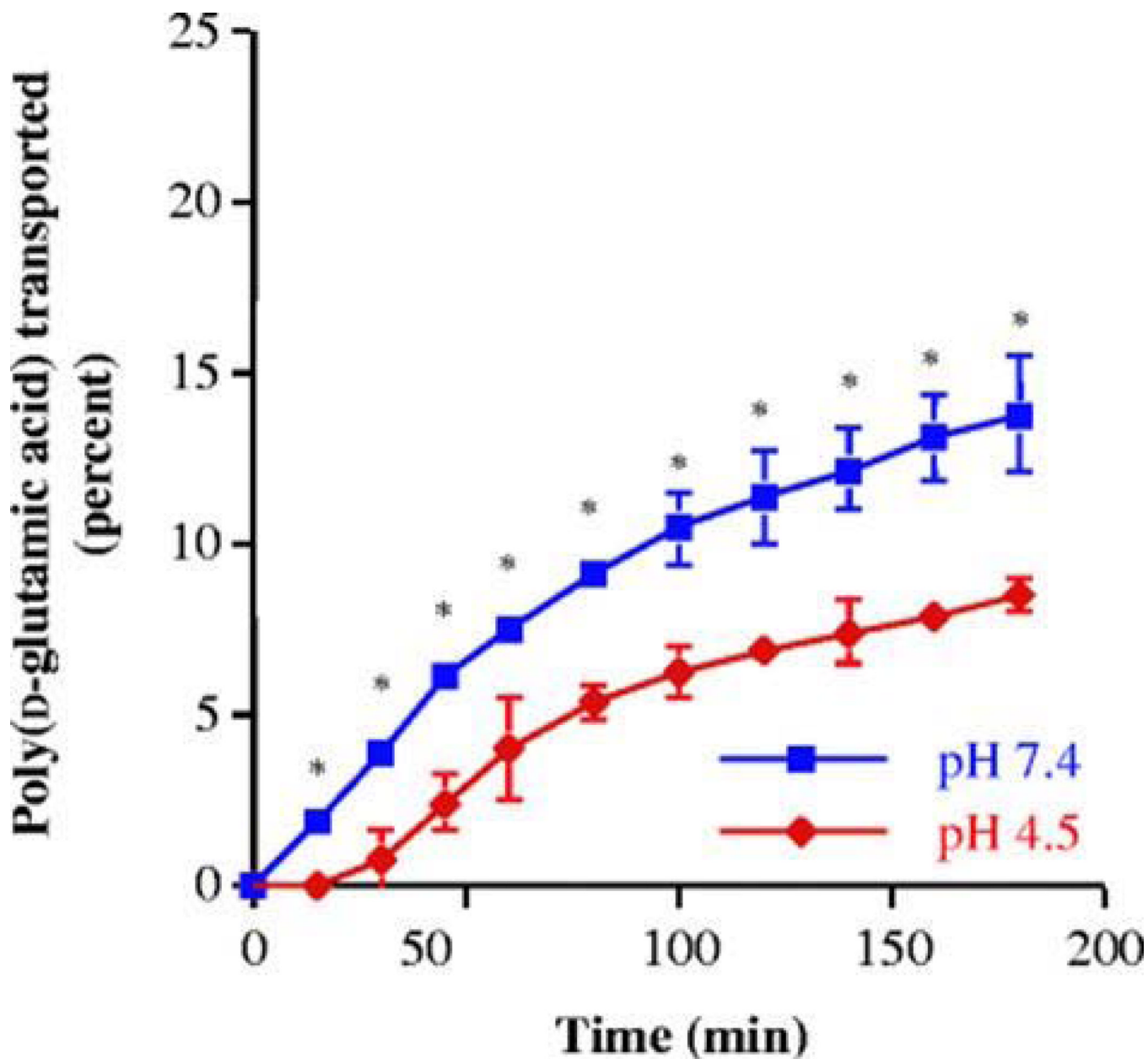
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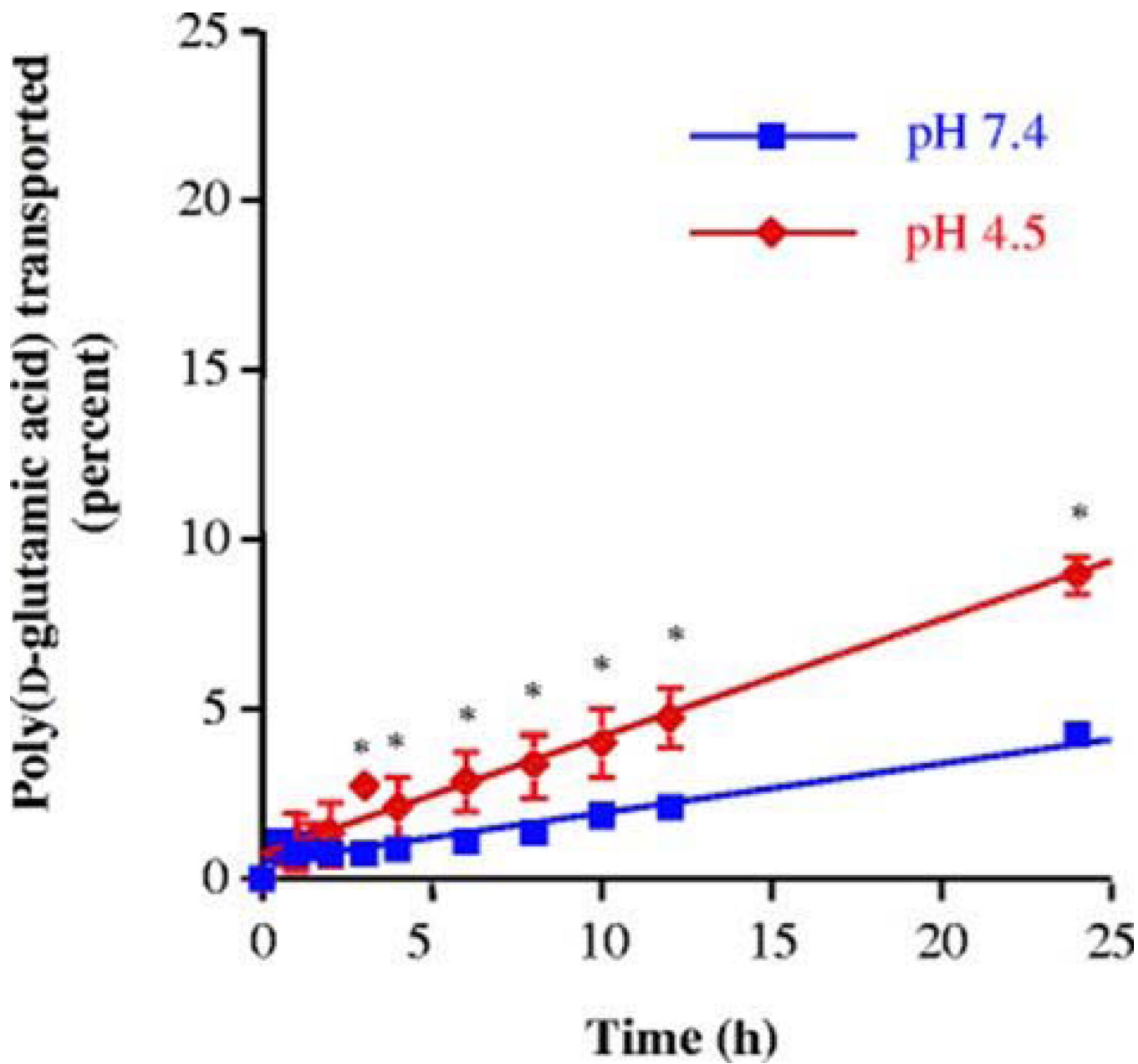
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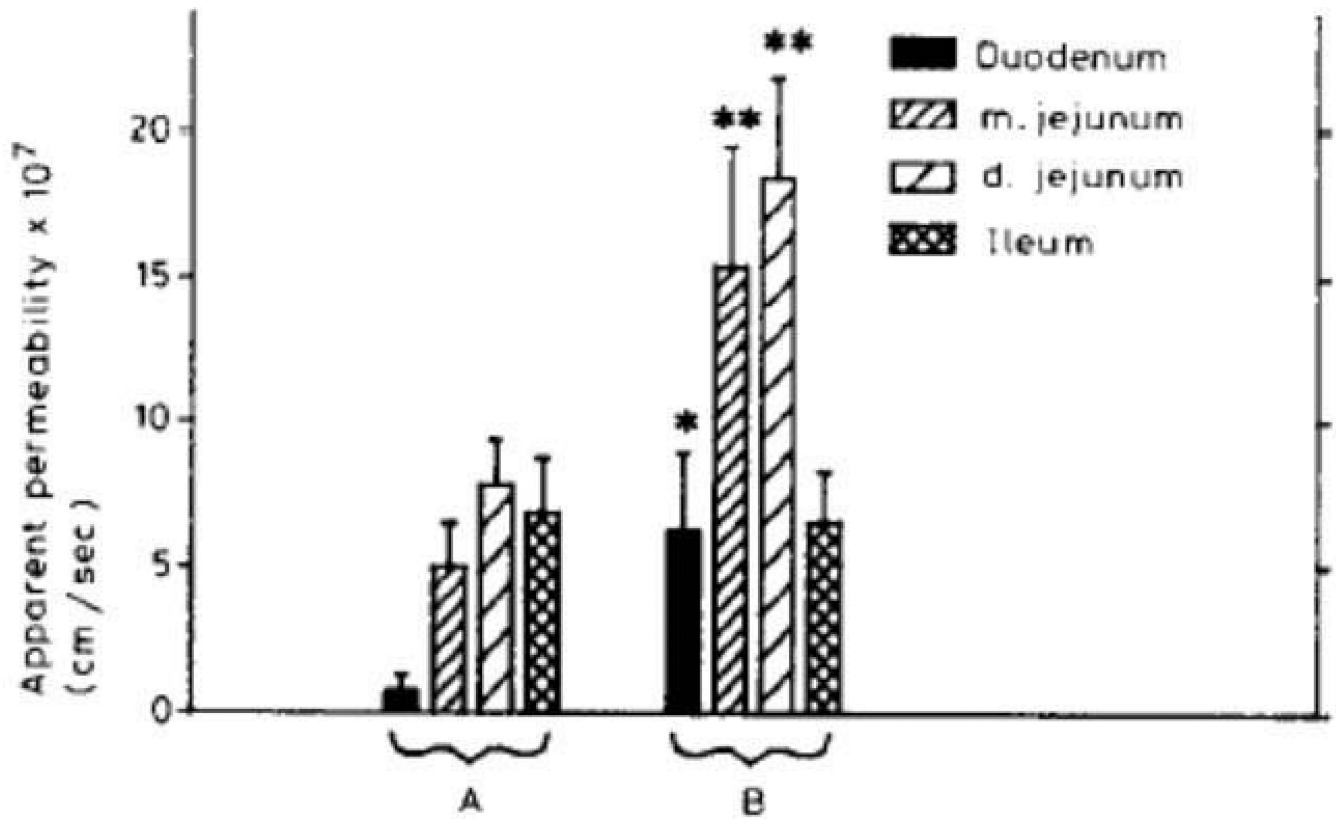
**Fig. 1.**  
Intestinal drug transport mechanisms.



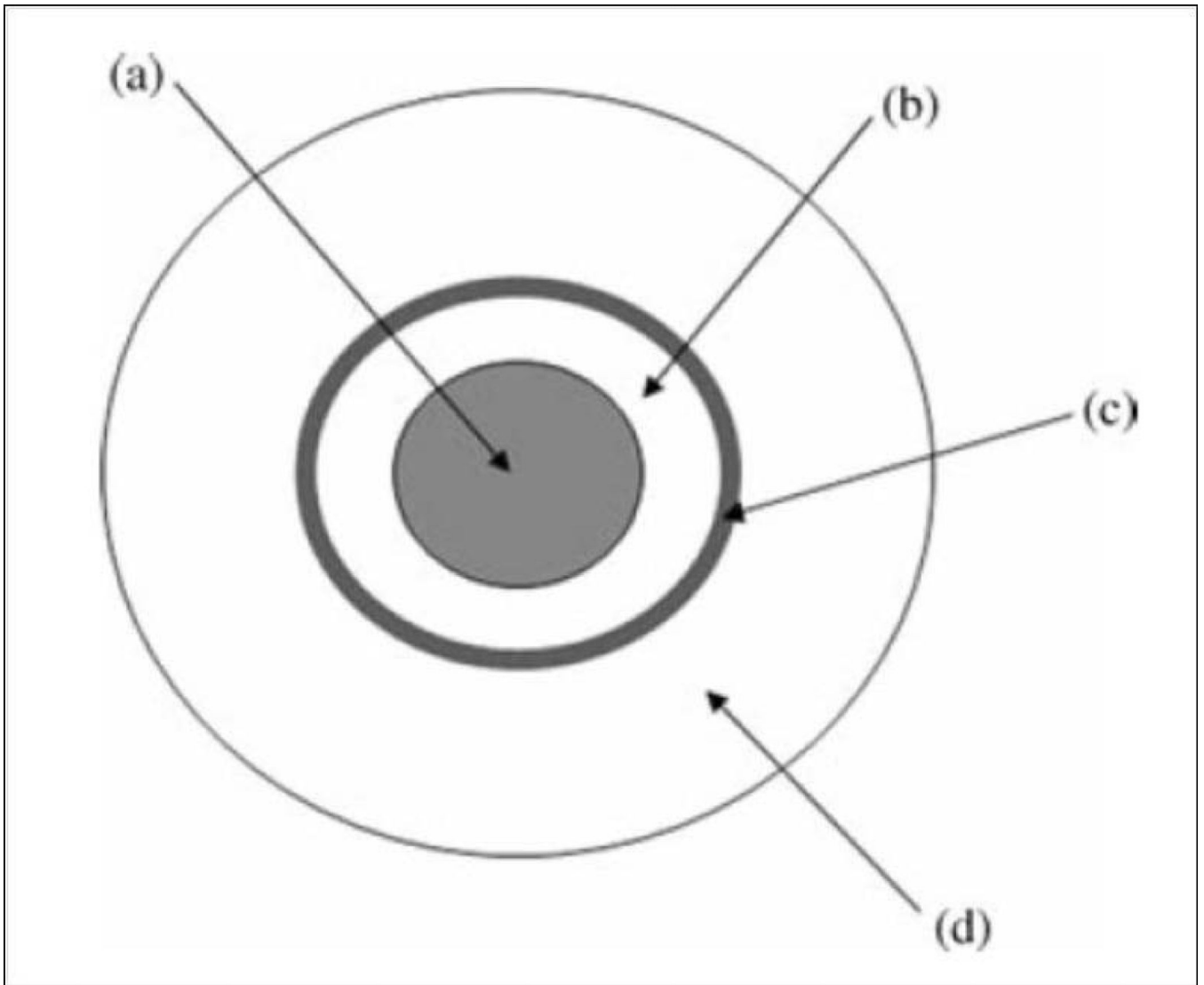
**Fig. 2.** Percentages of FITC-labeled poly(D-glutamic acid) transported across a Caco-2 cell monolayer at 37°C. Asterisks indicate a significant difference ( $P < 0.05$ ) compared to pH 4.5. The TEER values before and after the transport experiments of poly(D-glutamic acid) at pH 4.5 were not significantly different from each other (i.e.,  $189.7 \pm 13.8$  and  $185.6 \pm 11.1$   $\Omega$   $\text{cm}^2$ , respectively), which confirms that the integrity of Caco-2 cell monolayers was not affected after having been exposed to pH 4.5 for 3 hr (Reproduced with permission from reference (Chittchang et al., 2007)).



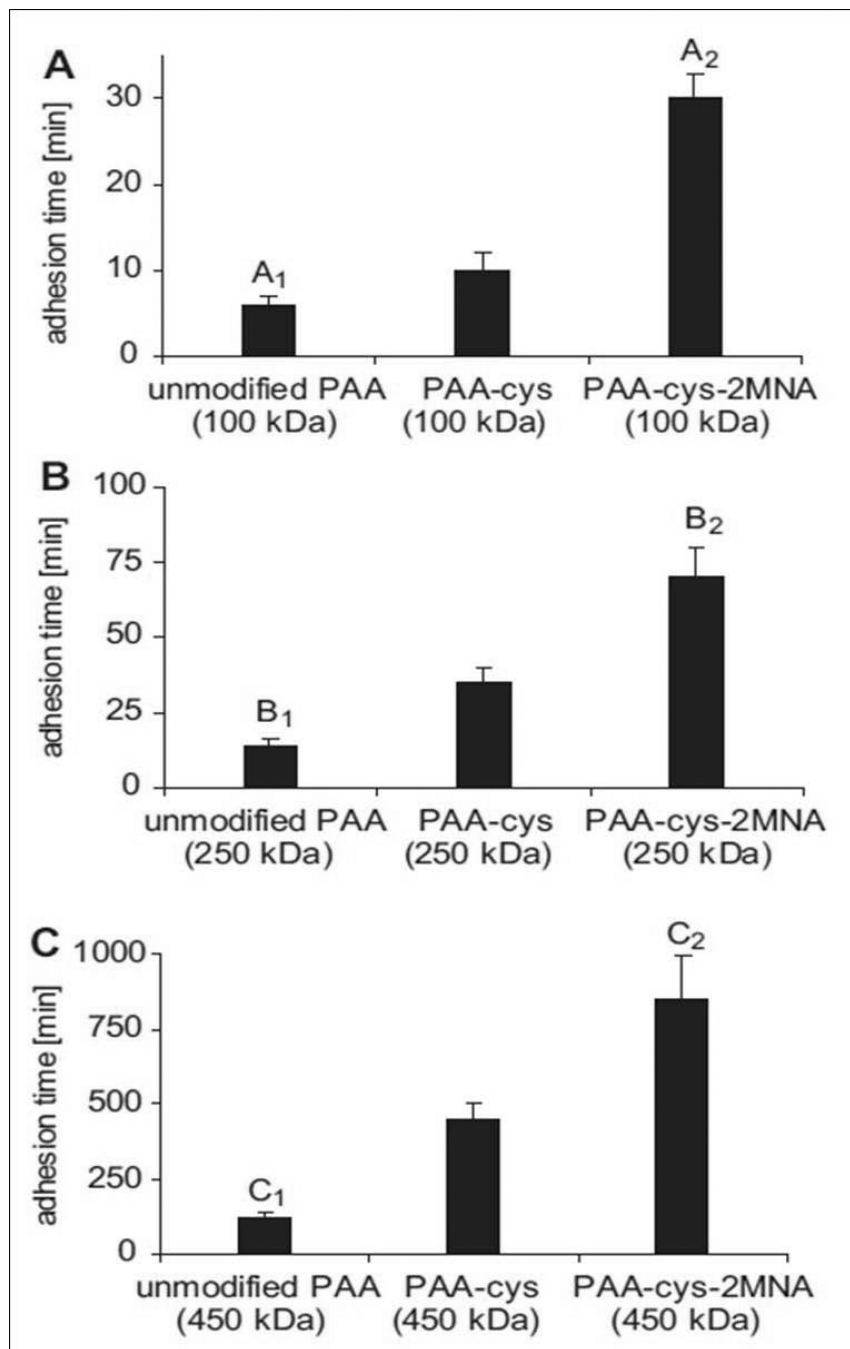
**Fig. 3.** Percentages of poly( $\alpha$ -glutamic acid) transported across a track-etched polycarbonate membrane with an average pore diameter of 0.015  $\mu\text{m}$  at 37°C. Asterisks indicate a significant difference ( $P < 0.05$ ) compared to pH 7.4 (Reproduced with permission from reference (Chittchang et al., 2007)).



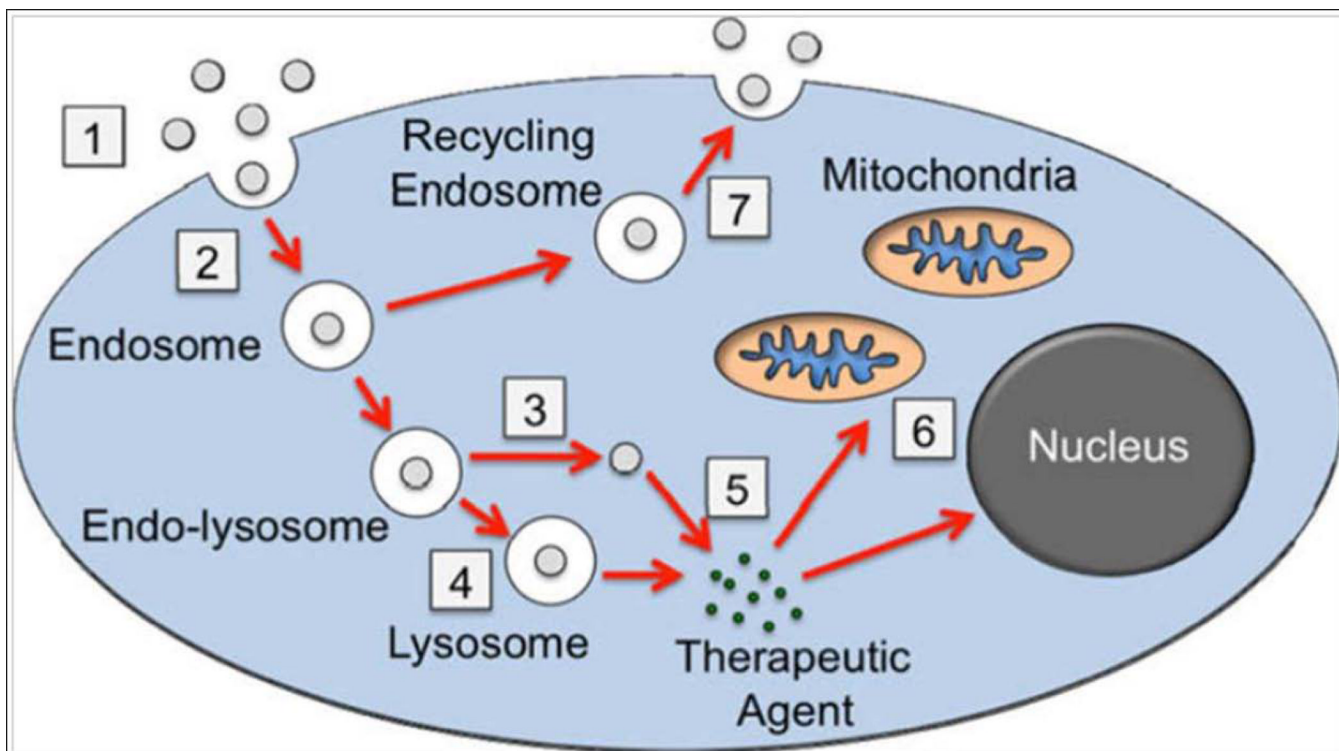
**Fig. 4.** Apparent permeability (mean+S.E.) of insulin across everted gut segments, with mucosal solutions of 83.3  $\mu$ M insulin in: A, modified buffer solution (n = 5); or B, modified buffer solution with 3:1 sodium glycocholate/linoleic acid mixed micelles (n = 4). For A versus B, \*p<0.1; \*\*p<0.05 (Reproduced with permission from reference (Schilling and Mitra, 1990).



**Fig. 5.** Schematic illustration of the two-pulse delivery system: (a) protein-containing core, (b) inner HPMC coating, (c) intermediate enzyme inhibitor/absorption enhancer, and (d) outer HPMC coating. Reproduced with permission from reference (Del Curto et al., 2011).

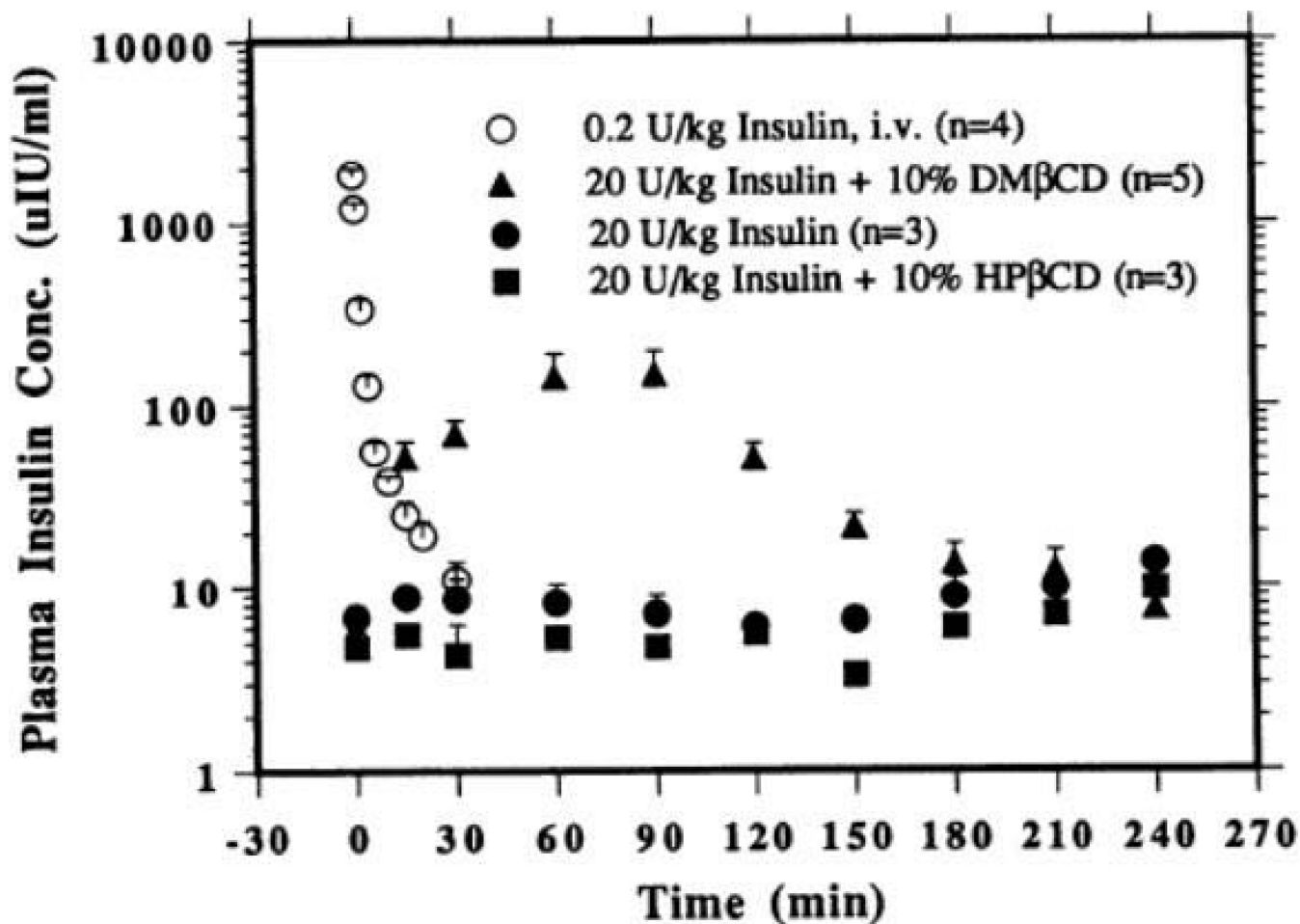


**Fig. 6.** Comparison of the mucoadhesive properties of unmodified, thiolated and preactivated poly acrylates with an average molecular mass of A: 100 kDa; B: 250 kDa and C: 450 kDa as determined by the rotating cylinder method. Reproduced with permission from reference (Iqbal et al., 2012).



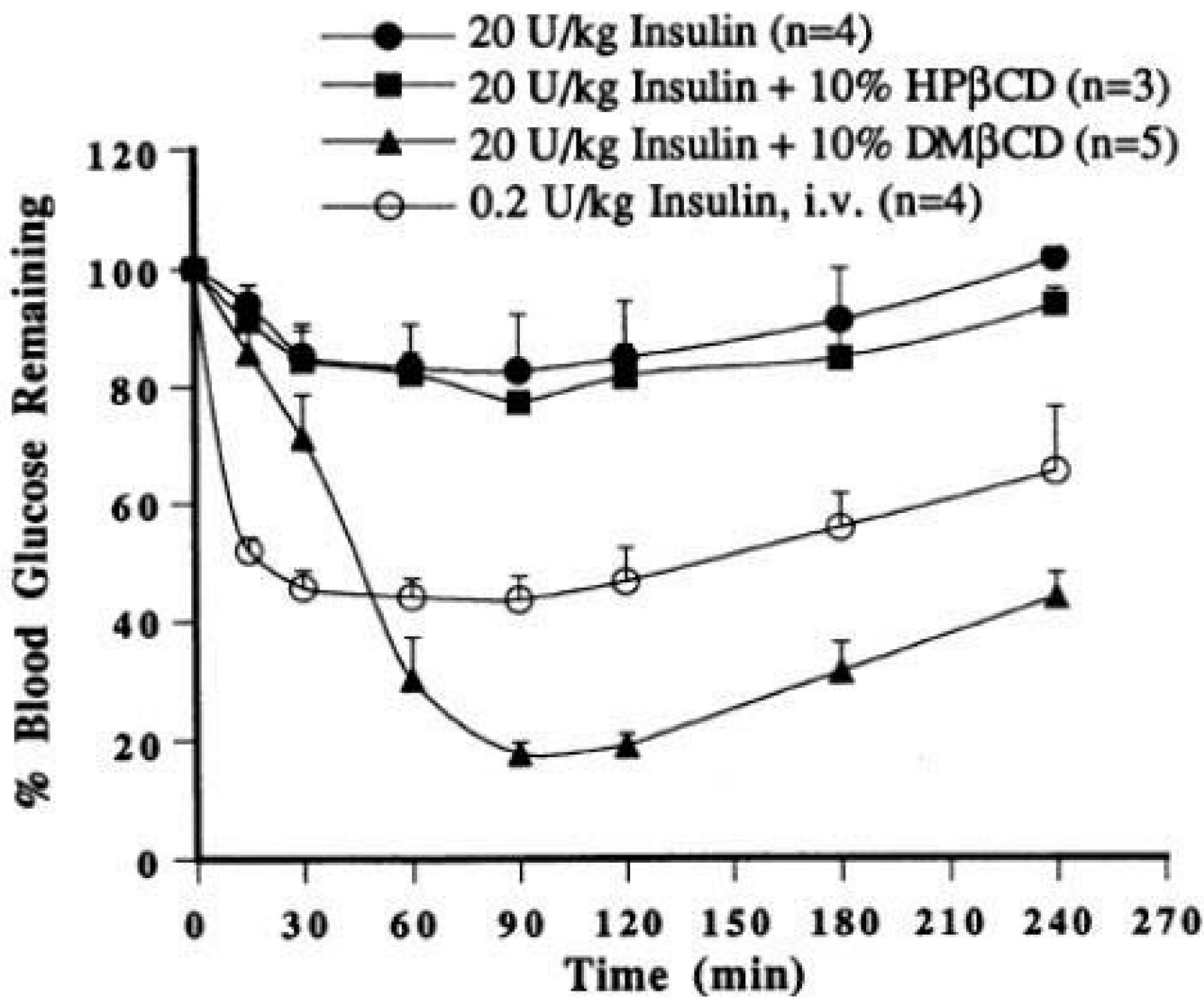
**Fig. 7.** Steps detailing the cytosolic delivery of therapeutic agents via nanoparticles carriers. (1) Association of nanoparticles with cell membrane, (2) internalization of nanoparticles by endocytosis, (3) escape of nanoparticles from endosomes, (4) degradation of nanoparticle in lysosome, (5) diffusion of therapeutic agent from lysosome into cytoplasm, (6) cytoplasmic transport of therapeutic agent to target organelle, (7) exocytosis of nanoparticles. Reproduced with permission from reference (Faraji and Wipf, 2009).



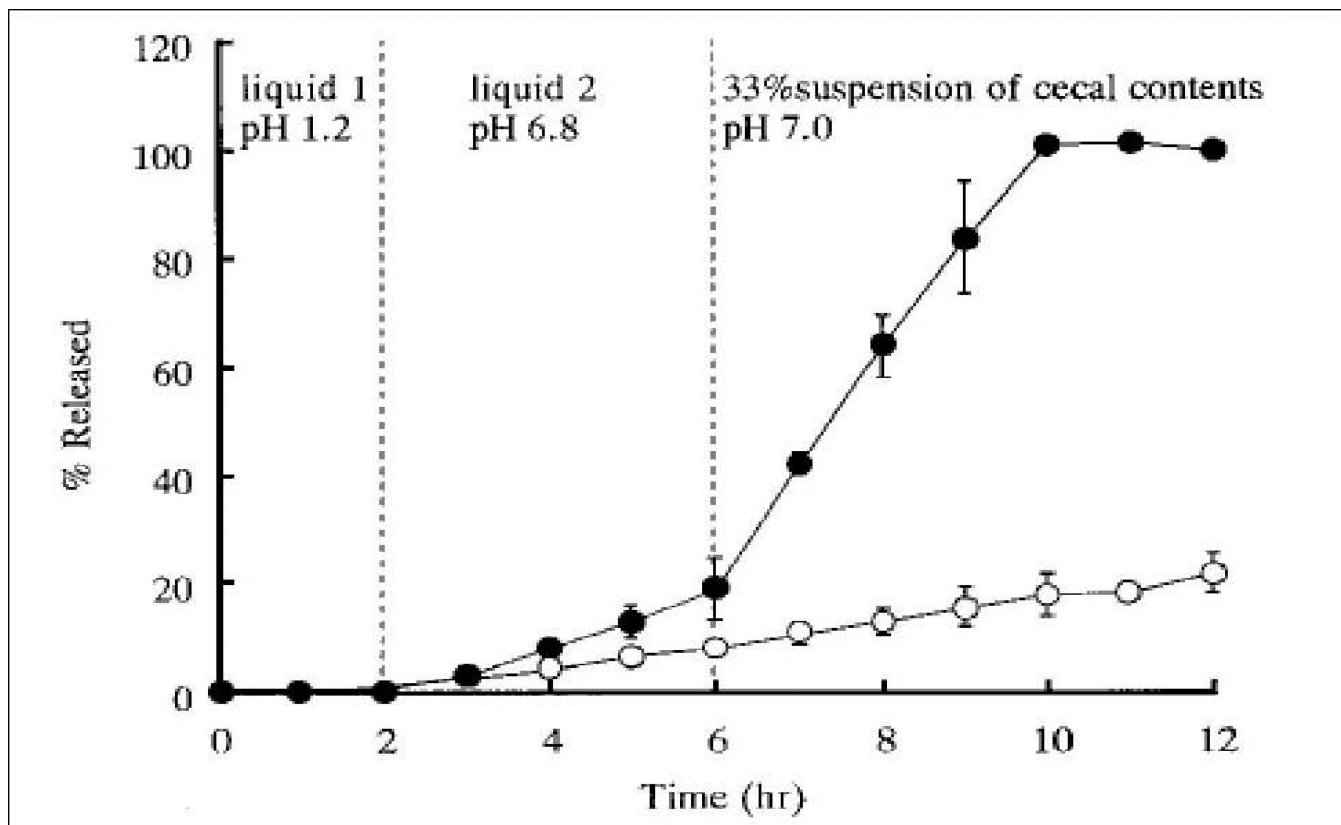


**Fig. 8.**

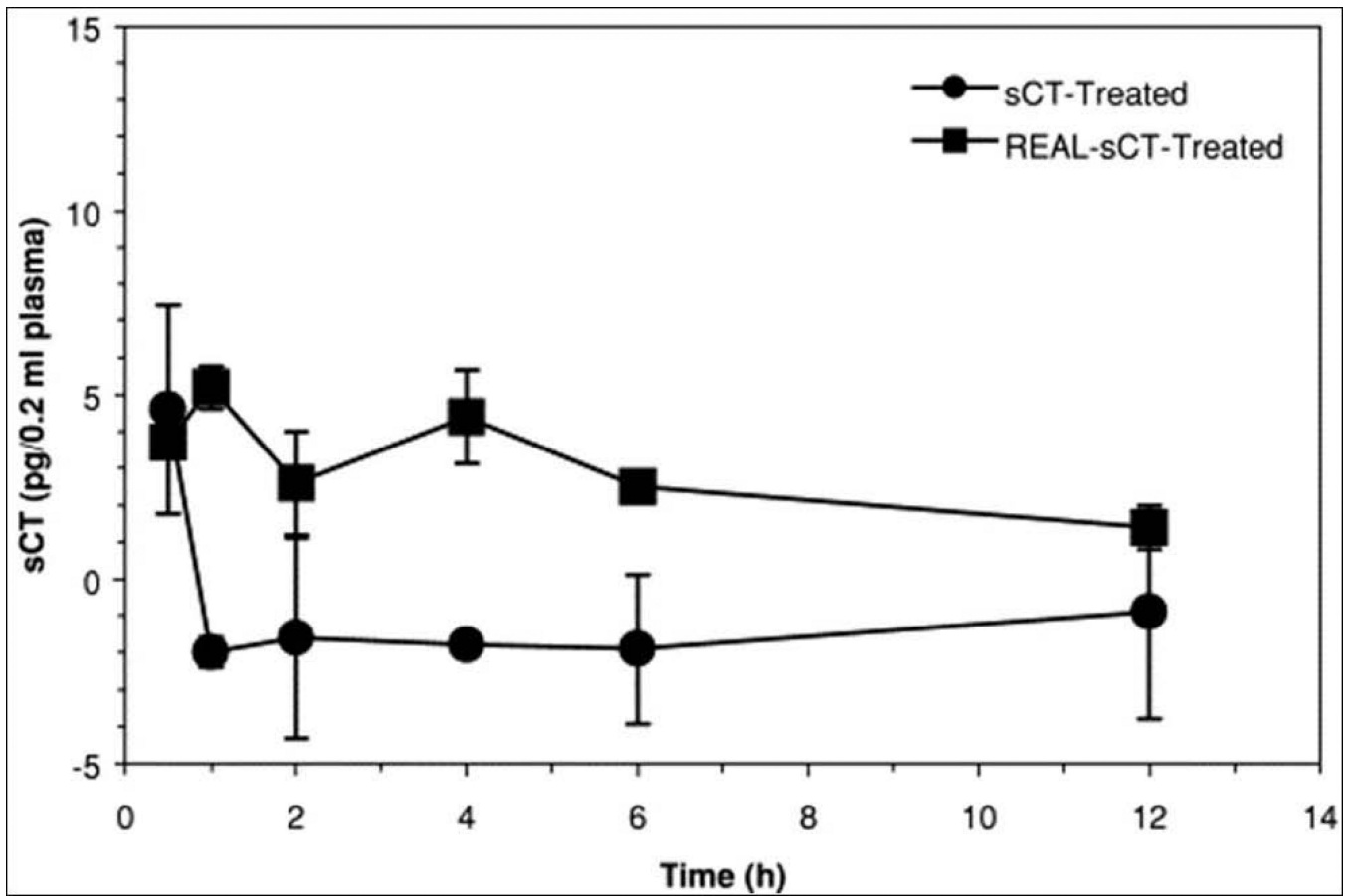
Plots of plasma insulin concentrations versus time following intravenous administration of 0.2 U/kg insulin, enteral administration of 20 U/kg insulin alone, enteral administration of 20 U/kg insulin with 10 % (w/v) HP-β-CD, and enteral administration of 20 U/kg insulin with 10 % (w/v) DM-β-CD. Values denote means ± SE. Reproduced with permission from reference (Shao et al., 1994).



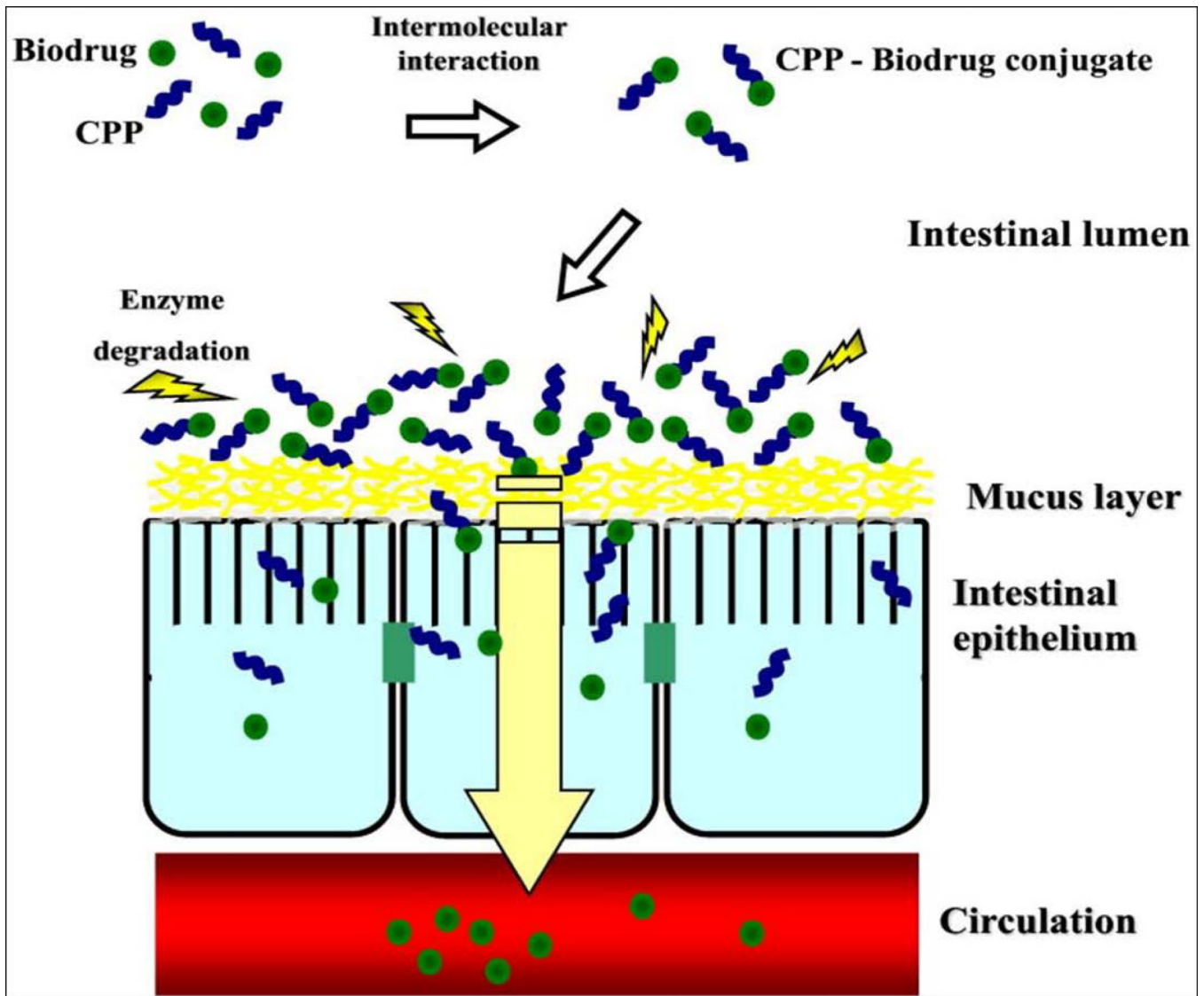
**Fig. 9.** Pharmacodynamic response following enteral administration of 20 U/kg porcine-zinc insulin in 0.01 M phosphate buffered saline in the absence and presence of 10 %  $\beta$ -cyclodextrin derivatives. Data from intravenous administration of 0.2 U/kg insulin have also been included. Values represent means  $\pm$  SE. Reproduced with permission from reference (Shao et al., 1994).

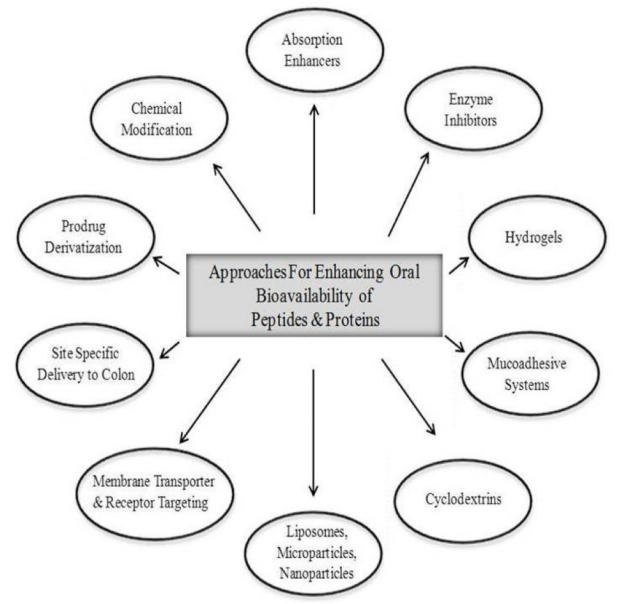
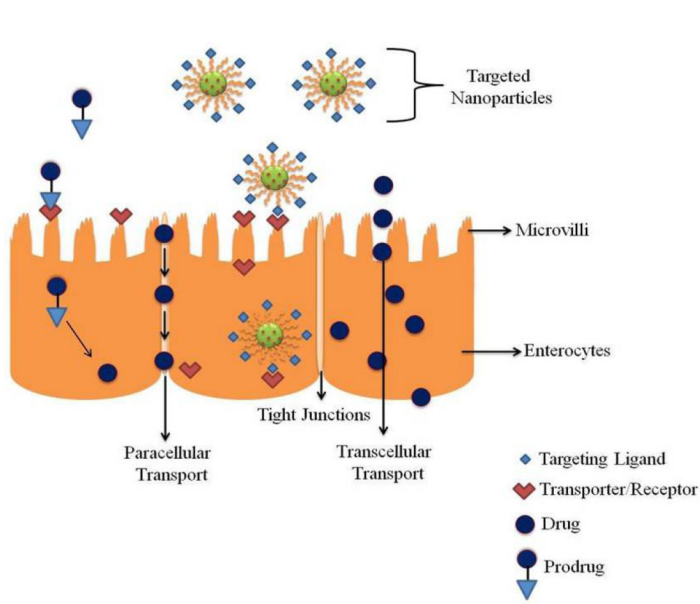


**Fig. 10.** Release of CF from chitosan capsules, determined by the Japanese Pharmacopoeia (J. P.) rotating basket method. (●) artificial gastric juice → an artificial intestinal juice → a suspension of rat cecal contents (O) phosphate buffered saline (pH 6.0). Reproduced with permission from reference (Tozaki et al., 1997).



**Fig. 11.** Plasma level–time profile of sCT in rats orally administered with 500 µg/kg of either sCT or REAL–sCT. The concentration of sCT in the plasma was determined by using a commercial RIA kit. Reproduced with permission from reference (Wang et al., 2003).





**Fig. 12.** Mechanism of intermolecular interaction between drug and CPP. Reproduced with permission from reference (Khafagy el and Morishita, 2012).

**Table 1**

Commonly used absorption enhancers and their mechanisms of action.

Category	Examples	Mechanism of action
Bile salts	Sodium Deoxycholate, Sodium Taurocholate, Sodium Glycodeoxycholate, Sodium Taurodihydrofusidate, Sodium Glycodihydrofusidate	Form reverse micelles and disrupt membrane, open up tight junctions, enzyme inhibition and mucolytic activity.
Chelators	EDTA, Citric acid, Salicylates	Interferes with calcium ions, chelation disrupts intracellular junctions and decreases transepithelial electrical resistance.
Surfactants	Sodium Lauryl Sulfate, Laureth-9, Sodium Dodecylsulfate, Sodium Taurodihydrofusidate, Poly Oxyethylene Ethers	Perturbation of intercellular lipids, lipid order, orientation and fluidity. Inhibition of efflux mechanisms.
Fatty acids and Derivatives	Oleic Acid, Linoleic Acid, Caprylic Acid, Capric Acid, Acyl Carnitines, Mono and Di-Glycerides	Increase fluidity of phospholipid membranes, contraction of actin myofilaments, opening of tight junctions
Cationic Polymers	Chitosan and its Derivatives	Combined effect of mucoadhesion and opening of tight junctions via ionic interactions with the cell membrane.
Anionic Polymers	Carbopol and Polyacrylic Acid Derivatives	Combined effect of enzyme inhibition and opening of tight junctions through removal of extracellular calcium ions.
	N-Acetyl Cysteine	Reduce the viscosity of mucus layer by breaking down disulfide bonds.
Acylcarnitines	Lauroyl-L-Carnitine Chloride, Palmitoylcarnitine Chloride	Membrane disruption, Opening of tight junctions with a calcium independent mechanism

**Table 2**

Salient features of thiomers nanoparticles in insulin delivery.

Thiomer	Drug/model drug	Outcome	Reference
PAA-Cys	Insulin	<i>In vitro</i> degradation studies, nanoparticles protected 44.47% of the initial insulin amount from trypsin, 21.33% from chymotrypsin, 45.01% from elastase compared to insulin solutions	(Perera et al., 2009)
Chitosan-6MNAcid	Insulin	AUC <i>in vivo</i> after oral administration to rats fourfold improved compared to unmodified chitosan nanoparticles	(Millotti et al., 2011)
TMC-Cys	Insulin	Oral and ileal application in rats blood glucose depression of 35% for oral administration and 70% for ileal application <i>in vivo</i> , hypoglycemic effect higher and longer-lasting compared to TMC-insulin nanoparticles	(Yin et al., 2009)