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Leonurenones A–C: Labdane diterpenes from *Leonotis leonurus*

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Abstract

Labdanes, leonurenones A–C, two known labdanes, luteolin 7-*O*- β -glucoside and luteolin were isolated and characterized from a commercial source of *Leonotis leonurus*. Genetic methods allowed for identification of the plant material. The leonurenones contain an uncommon α,β -unsaturated enone moiety in ring B, and leonurenones A and B were evaluated in a competitive inhibition assay at the GABA A neuroreceptor site.

Keywords

Labdane; Diterpene; *Leonotis leonurus*; Leonurenone

1. Introduction

Leonotis leonurus R. Br. (Lamiaceae) is a shrub 2–5 m in height that is native to South Africa. In traditional medicine, decoctions have been used externally for dermatological problems (rashes, boils and eczema) and internally to treat coughs, fever, headaches and hypertension (Scott et al., 2004). Leaves of the plant are smoked for its anti-epileptic effects (Bienvenu et al., 2002). Aqueous leaf preparations have been reported to possess anticonvulsant, antinociceptive, anti-inflammatory and antidiabetic properties in rodents (Bienvenu et al., 2002; Ojewole, 2005; Oyedemi et al., 2011). Additionally, crude aqueous extracts have been shown to possess antihelminthic activity (Maphosa and Masika, 2012; Maphosa et al., 2010).

In addition to folkloric uses mentioned above, *L. leonurus* reportedly produces marijuana-like effects (Stafford et al., 2008). The plant is commercially available and is marketed largely for its psychoactive effects. Some internet websites claim that this activity is attributable to an alkaloid, leonurine. There is some doubt about the validity of this claim since leonurine has never been reported to occur in the plant (although its presence in related species is documented) (Chen and Kwan, 2001; Hayashi, 1962; Kong et al., 1976; Luo, 1985). Thus the component(s) responsible for the reputed psychoactive effects are scientifically unverified at this time. Prior phytochemical investigations of various extracts of the plant have uncovered a number of labdane diterpenes, an iridoid glycoside and phenolic compounds, primarily of the flavonoid class (Agnihotri et al., 2009; El-Ansari et al., 2009; Kaplan and Rivett, 1968; Kruger and Rivett, 1988; Laonigro et al., 1979; McKenzie et al., 2006; Naidoo et al., 2011; Obikeze et al., 2008; Piozzi et al., 2007; Rivett, 1964).

As part of our program to identify natural products with central nervous system (CNS) activity from *Leonotis* plants, an investigation of a commercially available source of *L. leonurus* was carried out. Verification of the taxonomic identity of the plant was achieved via molecular methods. Details of the taxonomic identification as well as our chemical, spectroscopic and biological studies are reported herein.

2. Results and discussion

Since the plant material procured was not of sufficient quality to permit positive morphological identification (i.e. as distinct from other similar *Leonotis* species), identification of the plant using genetic methods was done.

Molecular markers have proved a powerful tool in diagnostics of species and varieties of various commercial products in order to identify and delimiting closely related species and to ensure quality control. Both DNA fingerprinting techniques and DNA sequence information have been employed, even from degraded and unrecognizable plant material (Martellosi et al., 2005; Zerega et al., 2002). The application, sometimes referred to as “DNA barcoding”, can assist in the process of identifying unknown plant specimens to known species. (Hollingsworth et al., 2011) For example, this method has been suggested as useful for an accurate and rapid authentication of medicinal plant products and their adulterants (Chen et al., 2010).

Recent molecular work in the Lamiaceae subfamily Lamioideae, to which *Leonotis* belongs, has produced a large database of chloroplast DNA sequences. (Scheen and Albert, 2009) A diagnostic assay was thus employed based on two chloroplast intron gene sequences (trnL intron and rps16 intron) that provide sufficient sequence variation to distinguish members of the group *Leonotis* belongs to, including closely related *Leonotis* and *Leucas* R.Br. species. After sequencing these gene regions from the procured *Leonotis* plant material several approaches were taken to ensure a correct identification. First, the sequences were analyzed using the BLAST search tool against sequences in the public database, GenBank, held at the National Center for Biotechnology Information (NCBI). Secondly, the DNA sequences were compared to the large data set of several hundred lamioide sequences, which clearly confirmed a large diagnostic “gap” in the rps16 sequence, which was shared with the *L. leonurus* sample in the data set. Finally, phylogenetic reconstruction was performed verifying that the obtained plant material grouped with the other *L. leonurus* in the data set. Together, these results clearly confirm the obtained material to belong to *Leonotis leonurus*.

The aqueous extracts of the aerial parts of the plant have not been subjected to phytochemical study and given the pharmacological activity attributed to this extract, it was considered worthy of such an investigation.

Repeated purification procedures (HPLC and flash column chromatography) on an aqueous extract of aerial parts of *L. leonurus* gave compounds **1**, **2**, **6** and **7** (Fig. 1). Similarly, repeated chromatography of an acetone extract afforded compounds **3–5**.

The initial diterpenoid nature of **1** came from its ^{13}C NMR spectrum where 20 carbons were observed. The HRESIMS spectrum showed a molecular ion peak at m/z 336.2380. This implied a molecular formula of $\text{C}_{20}\text{H}_{32}\text{O}_4$ in agreement with the ^{13}C NMR spectroscopic data. In the ^{13}C NMR spectrum, signals typical of the quaternary C-9 and C-13 positions of a spiroether labdane framework were observed (δ_{C} 95.9 and 85.7 ppm, respectively). Other low field quaternary signals were observed at δ_{C} 200.0 and 172.4 ppm. The former resonance was placed at position 7 of the labdane core based on HMBC correlations with the C-17 methyl protons (δ_{H} 1.21, d). The resonance at δ_{C} 172.4 showed HMBC cross-peaks to the C-18 and C-19 methyl protons as well as the C-20 methyl protons. This signal was thus

attributed to C-5 of the diterpene skeleton. A vinylic system appeared to be present based on the signal at δ_C 123.1 (CH). The proton attached to this vinylic carbon (δ_H 6.06) showed HMBC correlations to C-8 (δ_C 47.7) and C-10 (δ_C 45.2). On the basis of the preceding, it became evident that the resonances at δ_C 200.0, 172.4 and 123.1 were due to the presence of an α,β -unsaturated enone system in ring B. The presence of two oxymethylene carbons was inferred from DEPT-135 and HSQC data (δ_C 66.9 and 59.2). That at δ_C 66.9 was attached to mutually coupled doublets at δ_H 3.50 and 3.56, thereby placing this carbon at position 16. Further HMBC correlations from the proton at δ_H 3.56 to C-12, C-13 and the C-14 methylene carbon supported this assignment. The oxymethylene carbon at δ_C 59.2 was assigned to position 15. HMBC cross-peaks were observed for the H-15 protons to C-13 and C-14. The above established the gross structure of compound **1**. The relative stereochemical assignments for **1** were made on the basis of NOESY spectroscopic data. Here cross-peaks were seen for H-8 to the C-20 and C-11 protons, thereby establishing that these groups were on the same side of the labdane structure. H-20 in turn showed NOESY cross-peaks to H-16 while H-17 showed cross-peaks to H-14. The α,β -unsaturated ketone system in ring B seen in **1** is very rare in the *Leonotis* genus having only been reported once in the literature (Kaplan and Rivett, 1968).

Compound **2** was isolated as colorless oil and showed a molecular ion peak in its HRESIMS spectrum (m/z 332.1992) corresponding to the molecular formula $C_{20}H_{28}O_4$. The structural similarity to compound **1** was evident upon inspection of the ^{13}C NMR and 1H NMR spectroscopic data. The 1H NMR spectrum showed 3 methyl singlets and a methyl doublet as seen in **1**. Typical ^{13}C NMR signals for the α,β -unsaturated enone in ring B (δ_C 174.6, C-5; 122.3, C-6; 201.1, C-7) and the quaternary C-9 and C-13 positions (δ_C 96.5 and 86.6, respectively) were also observed. The main difference in the 1H and ^{13}C NMR spectra of the **2**, as compared to **1**, was the absence of additional signals corresponding to an oxymethylene group. This oxymethylene group was apparently replaced by an ester-like carbonyl functionality (δ_C 175.8). A pair of coupled oxymethylene doublets (δ_H 4.30 and 4.23; $J = 9.0$ Hz) was reminiscent of similar signals for C-16 of **1**. The resonance at δ_H 4.23 correlated to C-13 as well as δ_C 175.8. These data support the presence of a γ -butyrolactone moiety comprising C-13 to C-16. In the NOESY spectrum, correlations from H-8 to H-20 and H-11 were seen, placing these groups on the same face of the diterpene scaffold. Further NOESY cross-peaks between H-14 and H-17 and H-16 and H-12 established the 13*S* relative stereochemistry depicted. It is interesting that **2** being relatively non-polar was isolated from the aqueous extract; we cannot exclude the possibility that **2** is an artifact being formed via lactonization of a more polar C15 hydroxyl, C16 carboxylate intermediate during the isolation process.

The HRESIMS of compound **3** indicated a molecular formula of $C_{22}H_{34}O_5$ (m/z 378.2407). The previously described compounds **1** and **2** were obviously similar to **3** based on the NMR spectroscopic data accumulated. Assignments for C-1 to C-13 (particularly the now familiar α,β -unsaturated enone moiety) could readily be made by comparison of their 1H NMR and ^{13}C NMR spectra in tandem with HMBC analysis. Protons at δ_H 3.52 and 3.40 (both doublets, $J = 11.3$ Hz) were correlated to C-13 (δ_C 85.2) and C-12 in the HMBC spectrum. These protons were assigned to the oxymethylene group at C-16. HMBC cross-peaks were observed from protons at δ_H 1.91, 2.15, 4.06 and 4.16 to C-13. Signals at δ_H 4.06 and 4.16 were attached to δ_C 61.5. Their downfield chemical shift suggested attachment to an acyloxy group. Indeed, presence of an acetoxy group was established via resonances at δ_H 2.05 (3H, s); δ_C 21.0 (CH₃) and δ_C 171.1 (C). Thus the carbon at δ_C 61.5 was assigned to C-15 enabling complete assignments for the molecule and elucidation of the structure proposed. The NOESY data acquired suggested that **3** had the same relative configuration as seen in **1** and **2**. In that regard, key NOESY correlations were seen for H-8 to H-20 and H-11 and for H-17 to H-14.

In addition to the three new compounds above, the known labdane **4** was also isolated as an epimeric mixture, (Naidoo et al., 2011) nepetaefolin (**5**), (Hussein et al., 2003; Ohsaki et al., 2005; Von Dreele et al., 1975; White and Manchand, 1973) luteolin glycoside (**6**) (El-Ansari et al., 2009) and its aglycone (luteolin, **7**) (El-Ansari et al., 2009). The epimeric ratio of **4** was approximately 3:1 based on integration of the H-16 signals of the major and minor isomers – δ_{H} 4.75 and 5.32, respectively. NMR and physicochemical data for the known compounds are in close agreement with those previously reported (El-Ansari et al., 2009; Hussein et al., 2003; Naidoo et al., 2011). The presence of **4**, **6** and **7** in this species is known (El-Ansari et al., 2009; Naidoo et al., 2011). However, this is the first report of the presence of nepetaefolin (**5**) in *L. leonurus*. It is proposed that the absolute configurations of the new compounds **1–4** are in agreement with the absolute configurations of other known labdanes given the co-occurrence of **1–4** and **5** (for which the crystal structure has been reported) (Von Dreele et al., 1975).

Given the known involvement of the GABA_A system in sedation, the aqueous extract of *L. leonurus* (1 g/mL) was screened in a binding assay at the GABA_A site using the services of Caliper Life Sciences (www.caliperls.com), and showed an 81% inhibition at this concentration. Compounds **1** and **2** (10 μM) were then submitted for CNS receptor binding assays at the GABA_A site. However, the compounds did not show any activity (<50% inhibition) in this assay. The paucity of the other compounds isolated did not permit these evaluations. Further studies are required to identify the GABA_A- active metabolites present in the crude aqueous extract.

3. Conclusions

There have been a number of phytochemical investigations on *L. leonurus* and interest in this plant continues to be high because of the reported biological activities. Previous phytochemical work on the leaves of *L. leonurus* have concentrated on organic extracts. However, the biological activities ascribed to the plant are attributable to consumption of aqueous preparations. An investigation of the aqueous extract was thus prudent at this time.

Our investigation has uncovered three new labdane diterpenes from an aqueous extract of a commercial source of the plant. Thus this work highlights the utility of commercially available plant sources as a viable option for phytochemical work. Clearly however, as carried out in this paper, the appropriate validation of the plant material must accompany any such investigation.

Results from our study, as well as a previous investigation on the anticonvulsant effects of *L. leonurus* (Bienvenu et al., 2002), suggest that the aqueous extract has GABA_A activity. However, compounds **1** and **2** isolated herein did not show any activity at this receptor, indicating that these compounds are not by themselves responsible for the activity.

4. Experimental

4.1. General experimental procedures

Optical rotations were performed on a Rudolph Autopol IV polarimeter with the Na 589 line. IR spectra were recorded on a Thermo Nicolet IR 100 spectrophotometer as thin films. UV spectra were recorded on a Varian Cary 1 Bio UV–Visible spectrophotometer in CH₂Cl₂ solution. NMR spectroscopic data were obtained with on a Bruker 500 MHz spectrometer in CDCl₃ as solvent unless otherwise stated and with TMS as internal standard. Chemical shift (δ) values are reported in ppm and coupling constants in Hertz (Hz). High Resolution Electrospray Ionization Mass Spectra (HRESIMS) were obtained using an Agilent 6520 Q-TOF instrument. TLC analysis was performed with Analtech Uniplate silica

gel G/UV 254 pre-coated plates (0.2 mm). TLC plates were visualized by UV (254 and 360 nm), and by staining with 10% sulfuric acid/vanillin reagent followed by heating. Flash column chromatography (FCC) was performed with Silica gel 60 (EMD Chemicals, 230–400 mesh, 0.04–0.063 μm particle size) and Diaion HP-20 (Supelco/1-3607/176255E). HPLC purifications were performed on an Agilent 1200 system equipped with variable wavelength detector and fraction collector with Partisil 10 (Whatman) or ZORBAX SB-C18, 21.2 \times 150 mm, 7 μm (Agilent). Lyophilization was performed with a Labconco Freezone 2.5 system.

4.2. 2. Plant material—Plant material (crushed leaves) was purchased from Bouncing Bear Botanicals in September 2009. A voucher specimen (voucher number BKL00104130) is deposited in the Herbarium at the Brooklyn Botanic Gardens, Brooklyn NY 11225. Authentication of the material was performed via genetic analysis as described below. Genetic sequencing data is accessible at www.ncbi.nlm.nih.gov (GenBank accessions JX073278 and JX073279).

4.3. Genetic analysis

Total DNA was extracted using the DNeasy Plant Mini Kit (Qiagen) following the manufacturer's instructions. Amplification of the chloroplast *trnL* and *rps16* introns was performed following the method outlined in (Lindqvist et al., 2010). Since the DNA was potentially of low quality, the internal primers *rpsLR* and *rpsLF* were used to amplify the *rps16* intron as two separate, shorter fragments (Lindqvist et al., 2010). The DNA was sequenced at the University of Washington High-Throughput Genomics Unit. The sequences were assembled and edited using Sequencher 4.1.4 (Gene Codes) before being aligned manually to an existing database of lamioide mint DNA sequences using BioEdit (Hall, 1999). Phylogenetic reconstruction was performed using the programs TNT and SplitsTree4. (Huson and Bryant, 2006). Accession numbers for GenBank accessions are JX073278 and JX073279.

4.4. Extraction and isolation

Ground leaves of *L. leonurus* (50.2 g) were extracted three times with H_2O (3×1 L) by percolating in distilled H_2O at room temperature overnight. The extracts were combined and dried by lyophilization to yield (4.5 g) a dried extract. A portion of the aqueous extract (3.3 g) was applied to a Diaion HP-20 (Supelco/1-3607/176255E) column (19.0×4.5 cm) and fractionated using a H_2O :MeOH gradient solvent system (100:0, 30:70, 0:100). Fractions were collected and pooled by TLC analysis to afford 5 combined fractions. From these combined fractions, the fraction eluted with 100% MeOH (0.12 g) was subjected to a silica gel CC (15.0×1.2 cm) using an EtOAc:hexanes gradient solvent elution (20:80, 40:60, 60:40, 80:20, 100:0) to give 12 combined fractions. The fraction eluted in EtOAc:hexanes (80:20) was purified by HPLC [Partisil 10 column; flow rate: 2.00 mL/min; injection volume: 15 μL ; UV detector wavelength: 254 nm] to provide 1 (4 mg).

Compound **2** (3 mg) was isolated following HPLC of the fraction eluted in EtOAc:hexanes (40:60) using the following chromatographic parameters. Column: Partisil 10; flow rate: 2.00 mL/min; injection volume: 20 μL ; UV detector wavelength: 254 nm.

Ground *L. leonurus* leaves (180.1 g) were defatted with hexanes (3×1 L) by percolation at room temperature overnight. Extraction of the marc of the hexanes extract with acetone provided an acetone extract (2.1 g) after removal of solvent. A portion of the acetone extract was subjected to repeated HPLC purification in MeOH: CH_2Cl_2 (1:99, v/v) to afford compound **3** (1 mg). The following parameters were used. Column: Partisil 10; flow rate: 5.00 mL/min; injection volume: 10 μL ; UV detector wavelength: 254 nm.

The fraction eluted in EtOAc:hexanes (20:80, v/v) was applied to on a silica gel column using an EtOAc:hexanes gradient solvent system (15:85, 20:80), to give compound **4** (1 mg).

A portion of the acetone extract prepared as described above, was purified by FCC on a silica gel column using an EtOAc:hexanes gradient solvent system (10:90, 20:80, 30:70, 40:60, 50:50). Repeated FCC of the fraction eluted in EtOAc:hexanes (50:50, v/v) gave compound **5** (1 mg).

A separate portion of *L. leonurus* aqueous extract (1.0 g) was subjected to HPLC with a ZORBAX SB-C18 column (eluted in H₂O and MeOH, respectively) yielding three fractions. Compounds **6** (1 mg) and **7** (1 mg) were obtained by repeated HPLC of the latter two fractions (eluted in 100% MeOH). The following parameters were used. Column: ZORBAX SB-C18, 21.2 × 150 mm, 7 μm; flow rate: 4.00 mL/min; injection volume: 55 μL; UV detector wavelength: 254 nm.

4.4.1. Leonurenone A (1)—Colorless oil; $[\alpha]_D^{20} -26.1$ (c 0.1, CH₂Cl₂); UV (MeOH) λ_{\max} (log ϵ) 236 (3.11), 269 (2.52) nm; IR (film) ν_{\max} 2933, 1664, 1464, 1360, 1170, 1043 cm⁻¹; for ¹H NMR (500 MHz) and ¹³C NMR (125 MHz) spectroscopic data, see Table 1. HRESIMS [M]⁺ m/z : 336.2308 (calculated for C₂₀H₃₂O₄ 336.2380).

4.4.2. Leonurenone B (2)—Colorless oil; $[\alpha]_D^{20} -8.0$ (c 0.1, CH₂Cl₂); UV (MeOH) λ_{\max} (log ϵ) 243 (2.90) nm; IR (film) ν_{\max} 2932, 1781, 1662, 1464, 1372, 1171 cm⁻¹; for ¹H NMR (500 MHz) and ¹³C NMR (125 MHz) spectroscopic data, see Table 1. HRESIMS [M]⁺ m/z : 332.1988 (calculated for C₂₀H₂₈O₄ 332.1976).

4.4.3. Leonurenone C (3)—Colorless oil; $[\alpha]_D^{20} -24.9^\circ$ (c 0.1, CH₂Cl₂); UV (MeOH) λ_{\max} (log ϵ) 231 (2.51) nm; IR (film) ν_{\max} 3449, 2929, 1737, 1663, 1466, 1365, 1242, 1042 cm⁻¹; for ¹H NMR (500 MHz) and ¹³C NMR (125 MHz), see Table 1. HRESIMS [M]⁺ m/z : 378.2407, (calculated for C₂₂H₃₄O₅ 378.2406).

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

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Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.phytochem.2012.07.014>.

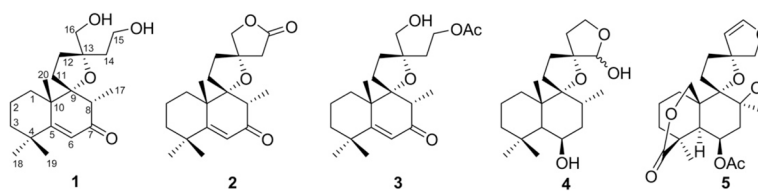


Fig. 1.
Structures of isolated compounds **1–5**.

Table 1

 ^{13}C NMR and ^1H NMR Spectroscopic Data for Compounds 1–3.

Position	Leonurenone A (1) ^a		Leonurenone B (2) ^b		Leonurenone C (3) ^d	
	δ_{C}	δ_{H}	δ_{C}	δ_{H}	δ_{C}	δ_{H}
1	29.4	1.92 (m) 2.18 (m)	30.4	1.62 (m) 1.73 (m)	30.7	1.66 (m)
2	17.9	1.72 (m) 1.85 (m)	17.2	1.72 (m) 1.93 (m)	17.4	1.74 (m) 1.87 (m)
3	39.2	1.44 (m) 1.60 (m)	38.8	1.47 (m) 1.61 (m)	39.3	1.58 (m)
4	37.1		36.6		37.1	
5	172.4		174.6		172.0	
6	123.1	6.06 (s)	122.3	6.00 (s)	123.2	6.05 (s)
7	200.0		201.1		199.8	
8	47.7	3.01 (q, 6.8)	47.1	3.05 (q, 6.8)	47.7	2.98 (q, 6.9)
9	95.9		96.5		95.7	
10	45.2		44.9		45.0	
11	30.8	1.67 (m) 1.71 (m)	29.3	2.10 (m) 2.40 (m)	29.7	1.91 (m) 2.18 (m)
12	35.4	1.87 (m) 1.88 (m)	36.0	2.21 (m)	34.3	1.78 (m) 1.91 (m)
13	85.7		86.6		85.2	
14	40.8	1.94 (m)	40.9	2.63 (d, 17.2) 2.80 (d, 17.2)	35.7	1.93 (m) 2.15 (m)
15	59.2	3.70 (m) 3.74 (m)	175.8		61.5	4.06 (m) 4.16 (m)
16	66.9	3.50 (d, 11.2) 3.56 (d, 11.2)	77.9	4.23 (d, 9.0) 4.30 (d, 9.0)	65.4	3.40 (d, 11.3) 3.52 (d, 11.3)
17	9.9	1.21 (d, 6.8)	8.6	1.17 (d, 7.8)	9.9	1.24 (d, 6.8)
18	31.7	1.20 (s)	30.6	1.19 (s)	31.7	1.20 (s)
19	30.6	1.21 (s)	29.8	1.26 (s)	30.6	1.21 (s)
20	24.0	1.38 (s)	23.0	1.43 (s)	24.1	1.38 (s)
CH_3COO					21.0	2.05 (s)
CH_2COO					171.1	

^a Measured in CDCI₃.

^b Measured in CD₃OD.