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## Thermostabilization of the $\beta_1$ -adrenergic receptor Correlates with Increased Entropy of the Inactive State

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### Abstract

The dynamic nature of GPCRs is a major hurdle in their purification and crystallization, but thermostabilization can facilitate structure determination, as has been shown by the structure of the thermostabilized  $\beta_1$ -adrenergic receptor ( $\beta_1$ AR) mutant, m23- $\beta_1$ AR, which has been thermostabilized in the inactive state. However, it is unclear from the structure how the six thermostabilizing mutations in m23- $\beta_1$ AR affect receptor dynamics. We have used molecular dynamics simulations in explicit solvent to compare the conformational ensembles for both wild type  $\beta_1$ AR (wt- $\beta_1$ AR) and m23- $\beta_1$ AR. Thermostabilization results in an increase in the number of accessible microscopic conformational states within the inactive state ensemble, effectively increasing the side chain entropy of the inactive state at room temperature, while suppressing large-scale main chain conformational changes that lead to activation. We identified several diverse mechanisms of thermostabilization upon mutation. These include, decrease of long range correlated movement between residues in the G-protein coupling site to the extracellular region (Y227A<sup>5,58</sup>, F338M<sup>7,48</sup>), formation of new hydrogen bonds (R68S), and reduction of local stress (Y227<sup>5,58</sup>, F327<sup>7,37</sup> and F338<sup>7,48</sup>). This study provides insights into microscopic mechanisms underlying thermostability that leads to an understanding of the effect of these mutations on the structure of the receptor.

### Keywords

G-Protein Coupled Receptors; Molecular Dynamics; Thermostability; Entropy

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Supporting Information available: Details of MD simulations, RMSD based clustering, mutual information and entropy calculation, discussion of MD convergence; Discussion of the role of R68S in thermostabilization; tables for TM Definitions used for PCA analysis and RMSD based clustering and definitions of EC and IC domains; table for calculated reduction in entropy due to correlated movement (called Mutual Information) of residues; figures showing regions of the receptor that contribute significantly to the principal components 1 and 2; properties of conformational clusters by PC coordinates; fraction of variance captured by the PCs for wt and m23  $\beta_1$ AR; results of RMSD based clustering of wt and m23  $\beta_1$ AR conformations; schematic representation of the receptor population distribution in wt and m23  $\beta_1$ AR; distribution of protein potential energy over total MD trajectories; difference in orientation for Y219<sup>5,58</sup> and Y326<sup>7,53</sup> between inactive (pdb: 2RH1) and active state (pdb: 3SN6) of  $\beta_2$ AR; effect of R68S mutation on ICL1-helix 8 interaction; Mutual Information of all residues in the receptor with mutated residues M90V, A282L and F327A; population distributions of dot products of first four eigenvectors between pairs of individual trajectories of (a) WT and (b) m23  $\beta_1$ AR. This material is available free of charge via the Internet at <http://pubs.acs.org>.

## INTRODUCTION

G-protein-coupled receptors (GPCRs) are seven helical transmembrane (TM) proteins that activate G-proteins upon binding agonists, such as hormones or neurotransmitters. Their key role in signal transduction makes them useful targets for therapeutic treatment. Newly developed strategies for determining the structures of GPCRs<sup>1-3</sup> will have a dramatic impact on the development of new drugs with enhanced specificity and better efficacy<sup>4</sup>. However, determination of GPCR structures is still challenging due to their conformational flexibility, which plays an essential role in the activation of GPCRs. This conformational flexibility contributes significantly to the instability of GPCRs in detergents, hence decreasing the probability of obtaining well-ordered crystals<sup>5</sup>. One strategy for improving the likelihood of obtaining crystals using lipidic cubic phase technology has been to engineer GPCR chimeras with T4 lysozyme inserted into intracellular loop 3 (ICL3) or by co-crystallization with antibody fragments<sup>6-7</sup>. The other strategy is to conformationally thermostabilize GPCRs so that the receptors are preferentially in a single conformational state and, in addition, tolerant of short chain detergents ideal for crystallography<sup>8</sup>. Thermostabilized GPCRs are readily purified and ideal for biophysical studies such as analyses using surface plasmon resonance or for crystallization<sup>2,9-10</sup>. The structures of thermostable mutants of three class A GPCRs, the adenosine A<sub>2A</sub> receptor and the avian  $\beta_1$ -adrenergic receptor ( $\beta_1$ AR), and the rat neurotensin receptor have been determined bound to either antagonists, agonists or biased agonists<sup>11</sup>. However, the structural basis for the difference in stability of the mutant GPCRs compared to the wild type is not clear.  $\beta_1$ AR, was stabilized by the inclusion of six thermostabilizing mutations<sup>12</sup>. The thermostabilization procedure was based on Ala/Leu scanning mutagenesis where every residue was mutated to Ala (except for Ala residues that were mutated to Leu), each mutant was expressed and its thermostability was determined relative to the wild type receptor. Once thermostabilizing mutations were identified, they were mutated to other amino acid residues to see if further improvements in thermostability were obtained. The best thermostabilizing mutations were then combined to give the optimally stable mutant m23- $\beta_1$ AR. This is a time-consuming process, so if we understand the structural and kinetic basis for why the mutations were thermostabilizing, then it might be possible to predict which residues in a model of a GPCR structure are thermostabilizing, thus facilitating an accelerated route to thermostabilization and structure determination.

The structural basis of thermostability of soluble proteins has been studied extensively from the perspectives of structure, biophysics and thermodynamics<sup>13</sup>. In contrast, studies on membrane protein stability are less common, which is probably because they are far less tractable experimentally<sup>14-15</sup>. The physicochemical properties of membrane proteins dictate that detergents are essential for their solubilization from membranes and for their subsequent purification. The environment of the detergent micelle is not a perfect mimetic of a membrane, which often results in decreased stability of membrane proteins after purification<sup>5</sup>. In addition, the ideal scenario for a thermodynamic analysis of protein stability would be to have a reversible folding-unfolding equilibrium, but this has been achieved for only a handful of membrane proteins<sup>16</sup>. For example, a number of studies have been performed on the kinetics of unfolding and folding of bacteriorhodopsin<sup>17</sup>, light harvesting complex<sup>18</sup>, and bacterial porins<sup>19</sup>, but it has not yet been possible to do this on GPCRs. Thus, to study the role of the thermostabilizing mutations in GPCRs, the only viable option currently available is to use computational approaches.

Molecular dynamics (MD) simulations of GPCRs have been used extensively to understand conformational changes and to provide insights into the mechanism of activation in GPCRs<sup>20-24</sup>. MD simulations have been used in conjunction with NMR data to explore the mechanism of activation of rhodopsin<sup>25</sup> and they have also been used to sample the

ensemble of conformations in a given potential energy minimum for GPCRs<sup>26–27</sup>. The role of thermostable mutations has been attributed to formation or breaking of inter-residue contacts at the site of mutation that stabilizes one selective receptor conformation over the others<sup>8</sup>. However not all mutations that lead to thermostability conform to this concept. In this paper we aim to analyze the effect of mutations on the global scale by calculating the differences in the ensemble of conformations sampled by the wild type and the m23- $\beta_1$ AR. We have used in total 2.10  $\mu$ s all-atom MD simulations combined for both wild type  $\beta_1$ AR (wt- $\beta_1$ AR) and the thermostable mutant m23- $\beta_1$ AR, in explicit lipid and water to provide a rationale for how the mutations lead to thermostabilization of m23- $\beta_1$ AR compared to wt- $\beta_1$ AR.

## METHODS

### Preparation of Protein Structure for Simulations

The starting conformation of avian m23- $\beta_1$ AR for the MD simulations was obtained from the protein databank (PDB ID: 2Y04). Initial conformations for the wild type structure simulations were obtained by mutating the residues in the crystal structure of the mutant back to the wild type using Maestro9.2 [Schrodinger Inc]. Residues within 5Å of the sites of mutation were minimized using MacroModel [Schrodinger Inc.] with position restraints on all backbone atoms and all residues further than 5Å from the site of mutation.

### Details of the Molecular Dynamics Simulations

The receptor structures were solvated in explicit palmitoyl oleoyl phosphatidyl choline (POPC) lipid and water. The lipid was packed around the protein using inflategro package in GROMACS<sup>28</sup>. SPC water molecules<sup>29</sup> were added on both sides of the lipid bilayer, and 11 chloride ions were added for both m23- $\beta_1$ AR and wt- $\beta_1$ AR. Ten simulations with different starting velocities on wt- $\beta_1$ AR and m23- $\beta_1$ AR in a POPC lipid bilayer were performed using GROMACS 4.0.7<sup>28</sup> and the GROMOS96 53a6 force field<sup>30</sup>. After the systems were equilibrated at the correct temperature and pressure, MD simulations of 110 ns were performed for each of the 10 unique initial conformations, using a NVT ensemble with a NoséHoover thermostat<sup>31–32</sup>. A total of 1.05 $\mu$ s of simulation was performed for each wt- $\beta_1$ AR and m23- $\beta_1$ AR. This is sufficient since our goal is to analyze the conformational ensembles sampled near the inactive state of the respective receptors. More details of the simulation conditions and a discussion on the convergence of individual MD trajectories is given in the supporting text S1.

### Conformational Entropy and Mutual Information calculations

The use of internal coordinates for conformational entropy calculations has previously been shown to be a more successful strategy than using Cartesian coordinates<sup>33</sup>. We have neglected the contributions from bond and angle degrees of freedom since they tend to be relatively small. First order conformational entropy was calculated using the Gibbs Entropy<sup>33</sup> for each torsion angle. We used 35 bins in the torsion angle distribution, that choice was based on convergence of the calculated entropy values. A correction for under sampling was applied as has been done previously<sup>34–35</sup>. For the equations used please refer to the supporting text S1.

### Force Distribution Analysis

Forces on each residue were calculated using both the bonded and non-bonded components of the GROMOS96 53a6 force field. The force on each residue was calculated by averaging the force exerted by all the other residues within 3Å and omitting the residues that are directly bonded to the residue. The averaging was done for all the snapshots from the MD

simulations. The forces were calculated between each atom pair following the procedure described by Stacklies et al.<sup>36</sup>. The residue wise forces were then binned and the population density at each bin was calculated.

### Principal Component Analysis

The most dominant motions sampled during MD simulations can be described using only a few principal modes<sup>37</sup>. We used the Principal Component Analysis (PCA) to understand the complex multidimensional dynamics in reduced space. PCA analysis was performed using GROMACS, on the consolidated 1 $\mu$ s trajectory obtained by merging the ten different simulations for each wild type and mutant. Only C $\alpha$  atoms in TM1 to TM7 (Table S1) were included in the PCA to reduce noise by the highly flexible loops and the N terminal end of TM1. Eigenvalues and eigenvectors were calculated from the covariance matrix, using the `g_covar` command in GROMACS. We performed PCA on the MD trajectories of wt- $\beta_1$ AR and m23- $\beta_1$ AR receptors separately.

## RESULTS

### Conformational ensembles of wt- $\beta_1$ AR and m23- $\beta_1$ AR in the inactive state

Class A GPCRs have seven helical transmembrane (TM) regions connected by extracellular loops (ECLs) and intracellular loops (ICLs) and they usually contain a short amphipathic helix called helix 8 (H8) parallel to the membrane surface after TM7. The m23- $\beta_1$ AR is a thermostable mutant (thermostable by  $\sim 20^\circ$  C compared to the wt- $\beta_1$ AR) with six point mutations at the positions: R68S<sup>1.59</sup>, M90V<sup>2.53</sup>, Y227A<sup>5.58</sup>, A282L<sup>ICL3</sup>, F327A<sup>7.37</sup> and F338M<sup>7.48</sup>. The superscripts refer to the Ballesteros-Weinstein number of the residues<sup>38</sup>, with the ICL3 superscript referring to residues in intracellular loop 3.. To compare the ensemble of conformations sampled during the 1.05 $\mu$ s each of dynamics for wt- $\beta_1$ AR and m23- $\beta_1$ AR, we performed Principal Component Analysis (PCA) on the C $\alpha$  atoms of the TM domain (TM domains for  $\beta_1$ AR are defined in Table S1). We have used PCA to identify the dominant collective motions of the TM domain in both mutant and wild type receptors. The dominant motions in both receptors are a global pulsating motion involving the extracellular region of the TM domains, particularly TM5 and TM6 (referred to as PC1) and, fluctuations near the extracellular interface of TM2 and TM3 (referred to as PC2) (Fig. S1A). The distance between the intracellular regions of TM3 and TM6 in both m23- $\beta_1$ AR and wt- $\beta_1$ AR was determined to identify any movement related to receptor activation<sup>39-40</sup>, but it was found to be consistently close to the inactive state of the receptor (Fig. S1). This is in contrast to previous multiscale dynamics study on ligand free wild type  $\beta_2$ AR, where both inactive state and an active-like state were observed<sup>20</sup>. The dominant motion sampled during MD simulations can be described using the first few principal modes as shown by the fraction of variance captured by each of the PCs shown in Figure S3.

The microscopic sub-states sampled within the inactive state conformation for wt- $\beta_1$ AR and m23- $\beta_1$ AR were projected on to PC1 and PC2 (Fig. 1). One broad and densely populated conformational cluster was observed during the MD simulation of m23- $\beta_1$ AR that showed an RMSD of  $\sim 1.5\text{\AA}$  from the crystal structure<sup>41</sup>. Within this broad cluster there are three sub-clusters or microscopic sub-states (Fig. S2). In comparison, wt- $\beta_1$ AR has one densely populated conformational cluster and another relatively sparsely populated conformational cluster that shows partial disorder in TM1 and TM7 (Fig. S2). The RMSD of all the conformational clusters from the crystal structure (pdb: 2Y04), their calculated population, and their potential energies are shown in Fig. S2. However an increase in the number of clusters (clustered by PC) does not necessarily mean higher entropy unless we take into account the size of these clusters and the conformational diversity within individual clusters.

Therefore to make the conformational diversity uniform among clusters, we clustered the conformations by RMSD in coordinates.

RMSD-based conformational clustering was performed to quantify the number of conformational clusters, the free energy difference between the conformational clusters and the number of transitions between these conformational clusters. We find that in the case of m23- $\beta_1$ AR there are 5 conformational clusters of similar population (within 1.5 kcal/mol) showing significant number of transitions to the inactive state cluster. The, wt- $\beta_1$ AR on the other hand has only 3 similarly populated clusters with direct transition to the inactive cluster (Fig. S4). Furthermore, the average difference in free energy calculated from the population, between the inactive state cluster and the neighboring clusters that show transitions is 1.83 kcal/mol for m23- $\beta_1$ AR, and 2.15 kcal/mol for wt- $\beta_1$ AR (Fig S4). This is also reflected in an increased number of transitions from and to the inactive cluster in the mutant (7.1/100 snapshots) as compared to the wild type (5.8/100 snapshots). Based on these data we have generated a schematic representation of the difference in the free energy landscape for the inactive state (Fig. 2) and the inactive state-active state transition (Fig S5) for both wt- $\beta_1$ AR and m23- $\beta_1$ AR. The population of the inactive state cluster is higher in m23- $\beta_1$ AR than in the wild type receptor. Thus, the dynamics of mutant m23- $\beta_1$ AR shows higher population of the inactive state compared to the wild type, and does so by generating multiple microscopic sub-states (Fig. 2) within the inactive state. We speculate that these microscopic states would dissipate the extra thermal energy at high temperatures thus retaining the overall fold of the inactive state. In contrast, wt- $\beta_1$ AR has lower population of the inactive state and fewer microscopic sub-states within the inactive state. The wild type receptor shows a population of other conformations well removed from the inactive state and are partially disordered and separated by large energy barriers. Therefore, at high temperatures wt- $\beta_1$ AR could possibly increase the population of disordered states leading to protein aggregation, as observed by Serrano-Vega et al.<sup>12</sup>. The multiple sub-states in m23- $\beta_1$ AR increases the localized entropy of the inactive state of the receptor, making the free energy of the inactive state more favorable. As shown in the section below, this increase in entropy comes from a reduction in correlated side chain movement between residues in EC region and the G-protein coupling region.

### Localized entropy differences between wt- $\beta_1$ AR and m23- $\beta_1$ AR stems from loss of correlated movement among residues in the mutant

To understand the basis for higher thermostability of the m23- $\beta_1$ AR mutant compared to wt- $\beta_1$ AR, we analyzed the enthalpic and entropic components of free energy of the ensemble of receptor conformations sampled in the MD simulations. There is no significant difference in the calculated average potential energy of the receptor when comparing the conformations sampled by wt- $\beta_1$ AR ( $-2167 \pm 130$  kcal/mol) compared to m23- $\beta_1$ AR ( $-2140 \pm 126$  kcal/mol). It should be noted that these potential energies were calculated in explicit lipid bilayer and water and not in detergents as done in experiments. The normalized potential energy distribution (Fig. S6A) shows a higher peak for the m23- $\beta_1$ AR compared to the wt- $\beta_1$ AR. This result, combined with the observation that the wt- $\beta_1$ AR samples less number of microscopic sub-states (Fig. 1), suggests that increased entropic contribution could be a major reason of thermostabilization for the m23 mutant.

To investigate which thermostabilizing mutations contribute to the increased entropy of m23- $\beta_1$ AR, we have calculated residue based torsional entropy as the sum of the entropy of the individual torsional degrees of freedom and the residue pairwise correlations that reduces the entropy<sup>33</sup>. Comparison of the residue entropy for wt- $\beta_1$ AR and m23- $\beta_1$ AR (Fig. 3) shows that the major differences in entropy is often found near the sites of mutation. To quantify this observation, we have reported the maximal change in entropy within 2 helical turns ( $\pm 4$  residues) of each thermostabilizing mutation (neighborhood entropy change). All

of the mutations show neighborhood entropy changes in excess of  $2.5\sigma$  ( $\sigma$ : standard deviation over all residues). The neighborhood entropy changes for individual mutations are 0.8 kcal/mol, z-score: 3.6 (R68S<sup>1.59</sup>), 0.57 kcal/mol, z-score: 2.6 (M90V<sup>2.53</sup>), 0.84 kcal/mol, z-score: 3.8 (Y227A<sup>5.58</sup>), 0.57 kcal/mol, z-score: 2.6 (A282L<sup>ICL3</sup>), 0.65 kcal/mol, z-score: 2.9 (F327A<sup>7.37</sup>) and 0.59 kcal/mol, z-score: 2.7 (F338M<sup>7.48</sup>). In addition, increased entropy in the mutant is observed in the extracellular half (0.95 kcal/mol more in the mutant) of the receptor (e.g. TM1, TM2 and TM7) while the entropy is reduced in the intracellular half of m23- $\beta_1$ AR (1.92 kcal/mol more in the wildtype), definitions of EC and IC half are given in Table S2. The reduced torsional entropy in the intracellular region of the m23- $\beta_1$ AR would be expected to lower the propensity for the mutant receptor to get activated. This was observed in the experiments of agonist induced activation and agonist binding affinity measurements by Serrano-Vega et al.<sup>12</sup>

### Thermostable mutant m23- $\beta_1$ AR loses correlated movement among residues involved in receptor activation

Certain residues in class A GPCRs have been shown to change their conformations upon activation and hence are proposed to act as 'molecular switches'<sup>20,23–24,42–43</sup>. To analyze if these residues are involved in activation leading to concerted motion between the ligand binding site and G-protein coupling site, we have analyzed the correlated movement of residue pairs during the dynamics of the wt- $\beta_1$ AR and m23- $\beta_1$ AR, using Mutual Information<sup>33–35</sup>. Correlation in movement between residues that are far removed in sequence leads to a reduction in entropy, because it reduces the number of unique conformational states sampled. Mutual Information has been used previously for the identification of allosteric communication pathways<sup>34,44</sup> and the quantification of conformational entropy<sup>33</sup>.

Several of the mutation sites show differences in cross correlation with other residues in the receptor when compared between wt- $\beta_1$ AR and m23- $\beta_1$ AR (Fig. 4). We observe that the residue Y227<sup>5.58</sup> in wt- $\beta_1$ AR shows many inter-residue correlations, the majority of which are lost upon mutation to alanine in m23- $\beta_1$ AR (Fig. 4A). Y227<sup>5.58</sup> also undergoes conformational change upon activation in the wt- $\beta_1$ AR and makes contact with Arg<sup>3.50</sup> on TM3 upon activation, and breaks the ionic lock between TM3 and TM6 (Fig. S7)<sup>39,45</sup>. Therefore mutation of this residue leads to preservation and stabilization of the ionic lock and hence the inactive state and a substantial increase in thermostability<sup>27</sup>. We observe a reduction in coupling (37% less than wild type) between the sites of the receptor correlated to Y227<sup>5.58</sup> and this contributes to the increased entropy in m23- $\beta_1$ AR. Thus the increased entropy resulting from the loss of inter-residue correlation as well as the strengthening of the ionic lock in the inactive state could explain the pronounced effect of the Y227<sup>5.58</sup> A mutation on the stability of the inactive state. Likewise, the R68S mutation leads to a loss in correlation in m23- $\beta_1$ AR (Fig 4B). We observed that the R68S mutation also leads to the formation of a hydrogen bond between TM1 and helix8 (H8) (Fig. S8). The mutated S68<sup>1.59</sup> forms a hydrogen bond with R355 on H8. Subsequent to our prediction, this hydrogen bond was also observed in a recent crystal structure<sup>46</sup> (additional discussion in SI Text). F327A mutation (Fig. S9C) leads to an increase in inter-residue correlation between this residue and other residues in the receptor. In the wildtype receptor F327<sup>7.37</sup> is pointing into the lipid bilayer and, upon mutation to alanine, this interaction is lost and the residue becomes more correlated with the rest of the protein. As we show later in the paper, the net stress on F327<sup>7.37</sup> is high and hence the mutation F327A relieves this stress conferring thermostability.

In the dynamics of the wt- $\beta_1$ AR, we observed<sup>27</sup> two possible microstates for F338<sup>7.48</sup> -one in which it shows  $\pi$  stack interaction with Y343<sup>7.53</sup> which in turn  $\pi$  stacks with F349<sup>7.59</sup>; and the other in which there is no such  $\pi$  stacking with Y343<sup>7.53</sup>. Y343<sup>7.53</sup> is part of the

conserved NPxxY motif common to most class A GPCRs, and it is known to move upon activation in  $\beta_2$ AR<sup>39</sup>. However in the m23- $\beta_1$ AR, the F338M mutation breaks its  $\pi$  stack interaction with Y343<sup>7.53</sup>, resulting in strengthening of the  $\pi$  stacking between Y343<sup>7.53</sup> and F349<sup>7.59</sup> which is characteristic of the inactive receptor state<sup>10</sup>. Here we also observe a loss of correlation of M338<sup>7.48</sup> with Y343<sup>7.53</sup> in m23- $\beta_1$ AR (Fig. 4C) and this decouples M338<sup>7.48</sup> from the rest of the receptor (Fig. 4D), resulting in an increase in localized entropy and thereby stabilization. The other sites of mutation show less significant differences in mutual information (Fig. S9A–C). The stability of the two mutations M90V and A282L could not be explained by our studies. The mutation A282L is in the ICL3 which is truncated in the structure and hence shows large fluctuations.

Overall, the wt- $\beta_1$ AR shows more inter-residue correlations (Fig. S9D), leading to a higher order network of correlated residues. We have calculated the reduction in entropy due to correlated motion shown in Table S3. As expected, the correlation between residues in the IC half or between IC and EC half (Table S2) is higher in the wt- $\beta_1$ AR compared to m23- $\beta_1$ AR (Fig. S9D)(Table S3). This may explain the reduced potential for the m23- $\beta_1$ AR to attain an activate state conformation. m23- $\beta_1$ AR also shows increased inter-residue correlations localized in the EC half of the receptor (Fig. S9D)(Table S3). Thus the communication between residue pairs in the intracellular half of the receptor is reduced as a result of the thermostabilizing mutations, and several residues that are known to play a role in receptor activation lose correlation, resulting in a more inactive like receptor (Fig. 4).

### Dynamics of inter-residue contacts that are related to receptor activation suggest that m23 is less prone to get activated

We have measured the variation of several critical inter-residue distances over time from the MD simulations. Comparison of the distribution of these inter-residue contacts between wt- $\beta_1$ AR and m23- $\beta_1$ AR, shows that m23- $\beta_1$ AR has a stronger preference for the inactive state. Most noticeably, the distance distribution of Y343<sup>7.53</sup> to F349<sup>H8</sup> has two peaks for the wt- $\beta_1$ AR simulations, but only one peak corresponding to the inactive state in the m23- $\beta_1$ AR simulations (Fig. 5A). The salt bridge known as the “ionic lock” between R139<sup>3.50</sup> of the conserved DRY motif on TM3, and E285<sup>6.30</sup> on TM6 is more frequently formed in the m23- $\beta_1$ AR than in wt- $\beta_1$ AR simulations, although this interaction is very dynamic in our simulations (Fig. 5B). Therefore we conclude that this ionic lock is not an essential feature of the inactive state either in the wild type or m23- $\beta_1$ AR; in fact there is no ionic lock observed in the crystal structure of m23- $\beta_1$ AR in complex with cyanopindolol<sup>10</sup>. Fig. 5C shows the distance distribution between residues S212<sup>5.43</sup> to N310<sup>6.55</sup> that form a hydrogen bond in the agonist bound crystal structure of  $\beta_1$ AR (pdb ID: 2Y04), but not in the antagonist bound crystal structure (pdb ID: 2VT4). This suggests that the hydrogen bond may be important in agonist-induced activation of the receptor, which is corroborated by mutational analysis of Ser207<sup>5.43</sup> in  $\beta_2$ AR<sup>47</sup>. Simulations of wt- $\beta_1$ AR showed more sustained hydrogen bond formation than in m23- $\beta_1$ AR simulations (Fig. 5C), supporting the contention that m23- $\beta_1$ AR is preferentially in an inactive state. We have also calculated the distance between the C $\gamma$  atoms F299<sup>6.44</sup> and P219<sup>5.50</sup> since this distance increased upon activation of  $\beta_2$ AR<sup>39</sup>. However, in both m23- $\beta_1$ AR and wt- $\beta_1$ AR simulations we find F299<sup>6.44</sup> exclusively in the inactive conformation. This is in agreement with MD simulations that were performed starting from the active state conformation of  $\beta_2$ AR where F282<sup>6.44</sup> returns to its inactive conformation quickly in the absence of G-protein<sup>23</sup>, indicating that F299<sup>6.44</sup> conformational change may occur late in the activation pathway.

### Stress Distribution in $\beta_1$ AR correlates with the measured thermostability

The distribution of the intramolecular repulsive and attractive forces within a protein structure governs how the protein reacts to thermal stress, since non-optimal distribution of

internal stress can lead to structural destabilization at elevated temperatures<sup>48</sup>. Internal stress at certain regions/residues of the receptor is necessary for movement of those regions/residues during receptor activation<sup>36,49</sup>. Thus a balance of these internal forces is required to maintain receptor stability and activity. To understand the basis of thermostability, we have calculated the net force on each residue averaged over the snapshots of the MD simulations for both m23- $\beta_1$ AR and wt- $\beta_1$ AR. This average force is a measure of the intensity of localized stress in the structure. Fig. 6A compares the population distribution of stress at each residue within the wt- $\beta_1$ AR and m23- $\beta_1$ AR averaged over the trajectory of the MD simulations; positive and negative forces indicate repulsive and attractive forces, respectively. Although both wt- $\beta_1$ AR and m23- $\beta_1$ AR show a very similar force distribution, closer analysis shows that wt- $\beta_1$ AR has an extended high stress or repulsive force region (in excess of 9000 pN; Fig. 6). The relative population of these high stress regions in wt- $\beta_1$ AR is small and may not affect the folded structure at physiological temperature. However at high temperatures and in detergent, these high stress domains could form hot spots that lead to destabilization and possibly denaturation of the wild type receptor. The regions that show the highest stress in wt- $\beta_1$ AR are P176<sup>4,60</sup>, P305<sup>6,50</sup>, Y343<sup>7,53</sup> and F353<sup>H8</sup> (Fig. 6B). Three of the strong thermostable mutations<sup>12</sup> show high stress in the wild type that gets relieved upon mutation: Y227A<sup>5,58</sup> on TM5 (thermostability index: 252), F327A<sup>7,37</sup> on TM7 (thermostability index: 180) and F338M<sup>7,48</sup> on TM7 (thermostability index: 191).

Other regions that show significant stress in wt- $\beta_1$ AR are P305<sup>6,50</sup> on TM6 and F353 on helix 8. P305<sup>6,50</sup> forms the hinge for the kink in TM6 and modulation of the proline kink could be crucial for the outward motion of TM6 leading to activation<sup>25,43</sup>. Thus reduced stress at P305<sup>6,50</sup> could lead to increased stability of the inactive state in m23- $\beta_1$ AR. The residue F353<sup>H8</sup> forms  $\pi$  stacking interaction with Y343<sup>7,53</sup> and could be responsible for the stabilization of TM7 and helix 8 in the inactive state<sup>27</sup>. Comparison of the inactive and active state crystal structures of  $\beta_2$ AR (PDB: 2RH1 and 3SN6), shows that there is significant structural difference at the intracellular end of TM7 near the Y343<sup>7,53</sup>-F353<sup>H8</sup> interface (Fig. S8). Thus relaxation of stress near F353<sup>H8</sup> could stabilize the inactive conformation of TM7.

## DISCUSSION

Conformational thermostabilization is a technique that results in conformational homogeneity in receptor mutants (8). This approach has now successfully been used for the structure determination of three GRPCs<sup>9-11,41,50</sup>. As the reasons for the increased stability of the receptor mutants were not obvious from the crystal structures, we have analyzed here the thermostabilized inactive-state m23- $\beta_1$ AR. Using all-atom MD simulations, we have investigated the dynamics of wt- $\beta_1$ AR and its thermostable mutant m23- $\beta_1$ AR to understand the role of the mutations in conferring thermostability. The backbone flexibility of the thermostable mutant and the wild type receptor are similar. m23- $\beta_1$ AR showed an increase in “localized flexibility” of the side chains within the inactive state (multiple microscopic states within the inactive state), effectively increasing the entropy of the inactive state. This aids in retention of the native functional state at higher temperatures by absorbing the extra thermal energy into localized fluctuations. Loss of correlated movement among residues could be a contributing factor behind the increase in the side chain movements.

In this study we present evidence that thermostabilization in  $\beta_1$ AR could be directly related to increased local entropy of the inactive state. Additionally, there are enthalpic contributions to stabilization by formation or breakage of localized residue networks through hydrogen bonds. Mutating residues that are points of high stress in the wild type receptor also leads to thermostabilization. The MD simulations in this study were done in explicit lipid bilayer and not in detergent solutions (as used in experiments), where there could be an



enthalpic difference. Another point to be noted is that we have not examined the effect of the mutations on the enthalpy and entropy of the active state of the receptor, since there is no crystal structure of the active state of  $\beta_1$ AR. It is possible that these mutations have effects on active state conformations that are not investigated here. We have compared the enthalpic and entropic differences of only the inactive state of the wt- $\beta_1$ AR and thermostable mutant m23- $\beta_1$ AR.

Entropic stabilization of proteins has been found previously for thermophilic soluble proteins. Hydrogen/Deuterium exchange experiments on the hyperthermophilic rubredoxin<sup>51</sup>, neutron scattering experiments on thermophilic and mesophilic dihydrofolate reductase<sup>52</sup> and  $\alpha$ -amylase<sup>53</sup>, Fourier Transform-IR studies on thermophilic esterase<sup>54</sup> and MD simulations on thermophilic homolog of subtilisin and esterases<sup>55–56</sup> all have shown that the thermophilic proteins exhibit greater microscopic flexibility than their corresponding mesophilic homologs. These studies propose that thermostability comes from the entropic contributions that flatten the energy surface similar to the data (Figs. 1 and 2) presented in this work. Although there are folding and unfolding kinetic measurements for membrane proteins, there are no direct measurements of the relationship between thermal stability and internal equilibrium fluctuations in the conformational dynamics.

## CONCLUSIONS

Our results provide valuable first insights into the one of the plausible mechanisms of thermostability in GPCRs. For instance, mutations that increase localized entropy while retaining the overall fold of the native state could have a higher propensity of being thermostable. Mutants that break networks of correlated residues that lead to activation have the potential to stabilize the inactive state of the receptor. Mutations that redistribute stress and/or form new favorable interaction networks can increase thermostability. Our study offers insight for four out of six mutations present in m23- $\beta_1$ AR. However we could not explain the stability of A282L<sup>ICL3</sup> since this is present in the truncated sequence of ICL3. The other mutation that we could not understand is M90V<sup>2,53</sup>. This residue is not involved in any inter-residue correlations in wt- $\beta_1$ AR and is not in a region of high stress. However, we did observe some changes in the flexibility of residues surrounding M90V (Fig 3). Thus although we can offer insights into the themostabilization effected by the majority of thermostabilizing mutations in m23- $\beta_1$ AR, there are clearly other factors that still need to be taken into account. For example, the role of the detergent micelle in stabilizing mutant receptors is unknown and this may lead to further insights into why some mutations that face the lipid bilayer are thermostabilizing. However, the results presented here provide a foundation for further studies on the stability of membrane proteins in a range of different environments.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

## Acknowledgments

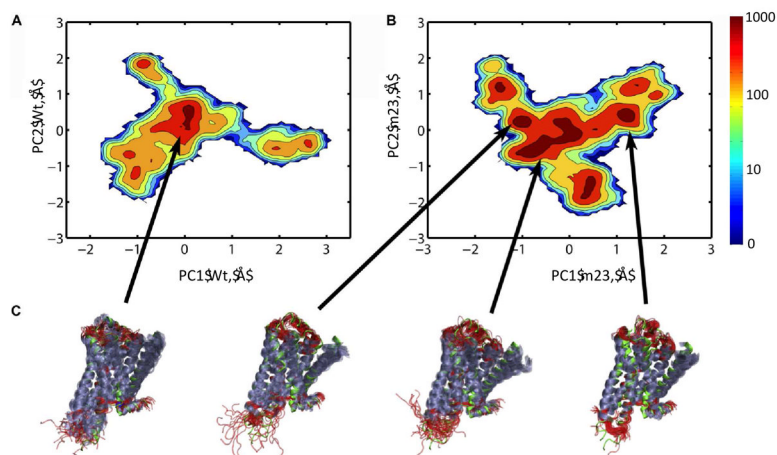
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## References

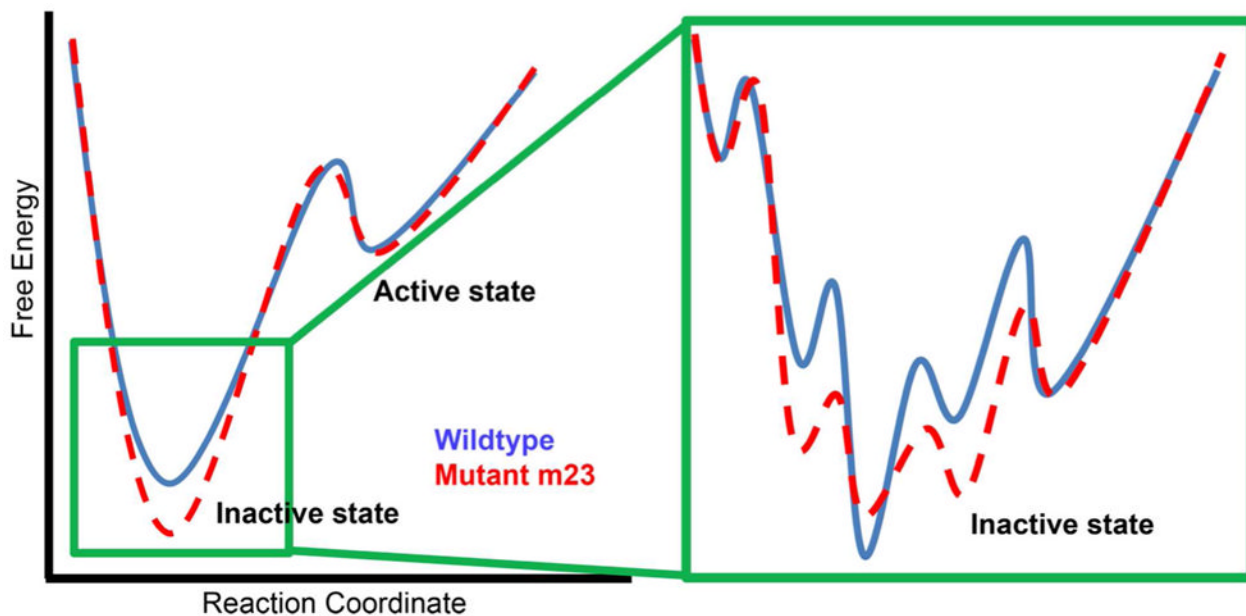
1. Hanson MA, Stevens RC. Structure. 2009; 17:8–14. [PubMed: 19141277]
2. Tate CG, Schertler GFX. Curr Opin Struc Biol. 2009; 19:386–395.
3. Steyaert J, Kobilka BK. Curr Opin Struc Biol. 2011; 21:567–572.

4. Shoichet BK, Kobilka BK. *Trends Pharmacol Sci.* 2012; 33:268–272. [PubMed: 22503476]
5. Tate CG. *Methods Mol Biol.* 2010; 601:187–203. [PubMed: 20099147]
6. Rosenbaum DM, Cherezov V, Hanson MA, Rasmussen SGF, Thian FS, Kobilka TS, Choi HJ, Yao XJ, Weis WI, Stevens RC, Kobilka BK. *Science.* 2007; 318:1266–1273. [PubMed: 17962519]
7. Day PW, Rasmussen SG, Parnot C, Fung JJ, Masood A, Kobilka TS, Yao XJ, Choi HJ, Weis WI, Rohrer DK, Kobilka BK. *Nature methods.* 2007; 4:927–929. [PubMed: 17952087]
8. Tate, CG. *Trends in Biochemical Sciences.*
9. Dore AS, Robertson N, Errey JC, Ng I, Hollenstein K, Tehan B, Hurrell E, Bennett K, Congreve M, Magnani F, Tate CG, Weir M, Marshall FH. *Structure.* 2011; 19:1283–1293. [PubMed: 21885291]
10. Warne T, Serrano-Vega MJ, Baker JG, Moukhametzianov R, Edwards PC, Henderson R, Leslie AGW, Tate CG, Schertler GFX. *Nature.* 2008; 454:486–U482. [PubMed: 18594507]
11. White JF, Noinaj N, Shibata Y, Love J, Kloss B, Xu F, Gvozdenovic-Jeremic J, Shah P, Shiloach J, Tate CG, Grisshammer R. *Nature.* 2012; 490:508. [PubMed: 23051748]
12. Serrano-Vega MJ, Magnani F, Shibata Y, Tate CG. *Proceedings of the National Academy of Sciences of the United States of America.* 2008; 105:877–882. [PubMed: 18192400]
13. Booth PJ, Templer RH, Meijberg W, Allen SJ, Curran AR, Lorch M. *Crit Rev Biochem Mol.* 2001; 36:501–603.
14. Booth PJ, Curnow P. *Curr Opin Struc Biol.* 2009; 19:8–13.
15. Bowie JU. *Nature.* 2005; 438:581–589. [PubMed: 16319877]
16. Schleich JP, Cao Z, Bowie JU, Park C. *Protein Sci.* 2012; 21:97–106. [PubMed: 22095725]
17. Cao Z, Schleich JP, Park C, Bowie JU. *Biochimica et biophysica acta.* 2012; 1818:1049–1054. [PubMed: 21880269]
18. Santabarbara S, Horton P, Ruban AV. *Biophysical journal.* 2009; 97:1188–1197. [PubMed: 19686667]
19. Sugawara E, Nagano K, Nikaido H. *Febs J.* 2012; 279:910–918. [PubMed: 22240095]
20. Niesen MJM, Bhattacharya S, Vaidehi N. *J Am Chem Soc.* 2011; 133:13197–13204. [PubMed: 21766860]
21. Vaidehi N, Kenakin T. *Curr Opin Pharmacol.* 2010; 10:775–781. [PubMed: 20933468]
22. Provasi D, Artacho MC, Negri A, Mobarec JC, Filizola M. *Plos Comput Biol.* 2011; 7.
23. Dror RO, Arlow DH, Maragakis P, Mildorf TJ, Pan AC, Xu H, Borhani DW, Shaw DE. *Proceedings of the National Academy of Sciences of the United States of America.* 2011; 108:18684–18689. [PubMed: 22031696]
24. Khelashvili G, Grossfield A, Feller SE, Pitman MC, Weinstein H. *Proteins.* 2009; 76:403–417. [PubMed: 19173312]
25. Hornak V, Ahuja S, Eilers M, Goncalves JA, Sheves M, Reeves PJ, Smith SO. *Journal of molecular biology.* 2010; 396:510–527. [PubMed: 20004206]
26. Dror RO, Arlow DH, Borhani DW, Jensen MO, Piana S, Shaw DE. *Proceedings of the National Academy of Sciences of the United States of America.* 2009; 106:4689–4694. [PubMed: 19258456]
27. Balaraman GS, Bhattacharya S, Vaidehi N. *Biophysical journal.* 2010; 99:568–577. [PubMed: 20643076]
28. Hess B, Kutzner C, van der Spoel D, Lindahl E. *Journal of Chemical Theory and Computation.* 2008; 4:435–447.
29. Berendsen, HJC.; Postma, JP.; Van Gunsteren, WFJH. *Intermolecular Forces.* Pullman, B., editor. Dordrecht Reidel; The Netherlands: 1981.
30. Scott WRP, Hunenberger PH, Tironi IG, Mark AE, Billeter SR, Fennel J, Torda AE, Huber T, Kruger P, van Gunsteren WF. *J Phys Chem A.* 1999; 103:3596–3607.
31. Hoover WG. *Phys Rev A.* 1985; 31:1695–1697. [PubMed: 9895674]
32. Nose S. *J Chem Phys.* 1984; 81:511–519.
33. Killian BJ, Kravitz JY, Somani S, Dasgupta P, Pang YP, Gilson MK. *Journal of molecular biology.* 2009; 389:315–335. [PubMed: 19362095]
34. Pandini A, Fornili A, Fraternali F, Kleinjung J. *Faseb J.* 2012; 26:868–881. [PubMed: 22071506]

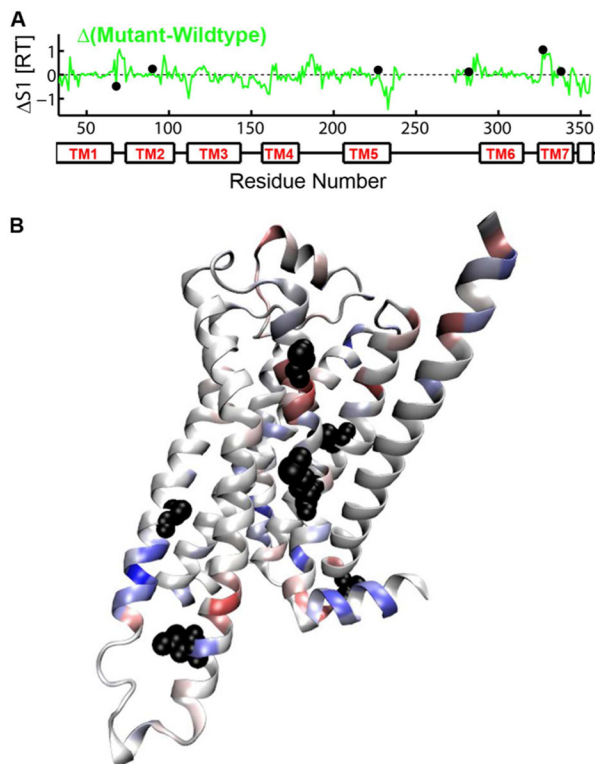
35. Steuer R, Kurths J, Daub CO, Weise J, Selbig J. *Bioinformatics*. 2002; 18:S231–S240. [PubMed: 12386007]
36. Stacklies W, Seifert C, Graeter F. *Bmc Bioinformatics*. 2011;12. [PubMed: 21219653]
37. Yang LW, Eyal E, Chennubhotla C, Jee J, Gronenborn AM, Bahar I. *Structure*. 2007; 15:741–749. [PubMed: 17562320]
38. Ballesteros, JA.; Weinstein, H. *Methods in Neurosciences*. Stuart, CS., editor. Vol. 25. Academic Press; 1995. p. 366-428.
39. Rasmussen SGF, DeVree BT, Zou YZ, Kruse AC, Chung KY, Kobilka TS, Thian FS, Chae PS, Pardon E, Calinski D, Mathiesen JM, Shah STA, Lyons JA, Caffrey M, Gellman SH, Steyaert J, Skiniotis G, Weis WI, Sunahara RK, Kobilka BK. *Nature*. 2011; 477:549–U311. [PubMed: 21772288]
40. Rasmussen SGF, Choi HJ, Fung JJ, Pardon E, Casarosa P, Chae PS, DeVree BT, Rosenbaum DM, Thian FS, Kobilka TS, Schnapp A, Konetzi I, Sunahara RK, Gellman SH, Pautsch A, Steyaert J, Weis WI, Kobilka BK. *Nature*. 2011; 469:175–180. [PubMed: 21228869]
41. Warne T, Moukhametzianov R, Baker JG, Nehme R, Edwards PC, Leslie AGW, Schertler GFX, Tate CG. *Nature*. 2011; 469:241–244. [PubMed: 21228877]
42. Ahuja S, Smith SO. *Trends Pharmacol Sci*. 2009; 30:494–502. [PubMed: 19732972]
43. Bhattacharya S, Hall SE, Vaidehi N. *Journal of molecular biology*. 2008; 382:539–555. [PubMed: 18638482]
44. McClendon CL, Friedland G, Mobley DL, Amirkhani H, Jacobson MP. *Journal of Chemical Theory and Computation*. 2009; 5:2486–2502. [PubMed: 20161451]
45. Scheerer P, Park JH, Hildebrand PW, Kim YJ, Krauss N, Choe HW, Hofmann KP, Ernst OP. *Nature*. 2008; 455:497–U430. [PubMed: 18818650]
46. Warne T, Edwards PC, Leslie AGW, Tate CG. *Structure*. 2012; 20:841–849. [PubMed: 22579251]
47. Sato T, Kobayashi H, Nagao T, Kurose H. *Brit J Pharmacol*. 1999; 128:272–274. [PubMed: 10510435]
48. Stadler AM, Garvey CJ, Bocahut A, Sacquin-Mora S, Digel I, Schneider GJ, Natali F, Artmann GM, Zaccai G. *Journal of The Royal Society Interface*. 2012
49. Stacklies W, Xia F, Grater F. *Plos Comput Biol*. 2009; 5
50. Lebon G, Warne T, Edwards PC, Bennett K, Langmead CJ, Leslie AGW, Tate CG. *Nature*. 2011; 474:521–U154. [PubMed: 21593763]
51. Hernandez G, Jenney FE, Adams MWW, LeMaster DM. *Proceedings of the National Academy of Sciences of the United States of America*. 2000; 97:3166–3170. [PubMed: 10716696]
52. Meinhold L, Clement D, Tehei M, Daniel R, Finney JL, Smith JC. *Biophysical journal*. 2008; 94:4812–4818. [PubMed: 18310248]
53. Fitter J, Heberle J. *Biophysical journal*. 2000; 79:1629–1636. [PubMed: 10969023]
54. D’Auria S, Herman P, Lakowicz JR, Tanfani F, Bertoli E, Manco G, Rossi M. *Proteins-Structure Function and Genetics*. 2000; 40:473–481.
55. Wintrode PL, Zhang DQ, Vaidehi N, Arnold FH, Goddard WA. *Journal of molecular biology*. 2003; 327:745–757. [PubMed: 12634066]
56. Colombo G, Merz KM. *J Am Chem Soc*. 1999; 121:6895–6903.



**Figure 1.** Comparison of the major motions observed in MD simulations of wt- $\beta_1$ AR and thermostabilized m23- $\beta_1$ AR. Conformational clusters, based on PC coordinates, were calculated from the MD simulations and are represented in 2-dimensions, projected on their respective principal components, PC1 and PC2: **A.** wt- $\beta_1$ AR and **B.** the m23- $\beta_1$ AR. Densely populated regions are colored in red, while sparsely populated regions are shown in deep blue. **C.** Representative ensemble of structures from each conformational cluster are depicted in cartoon representation ( $\alpha$ -helices purple, non-helical regions red) aligned with the crystal structure of m23- $\beta_1$ AR (green, pdb: 2Y04).

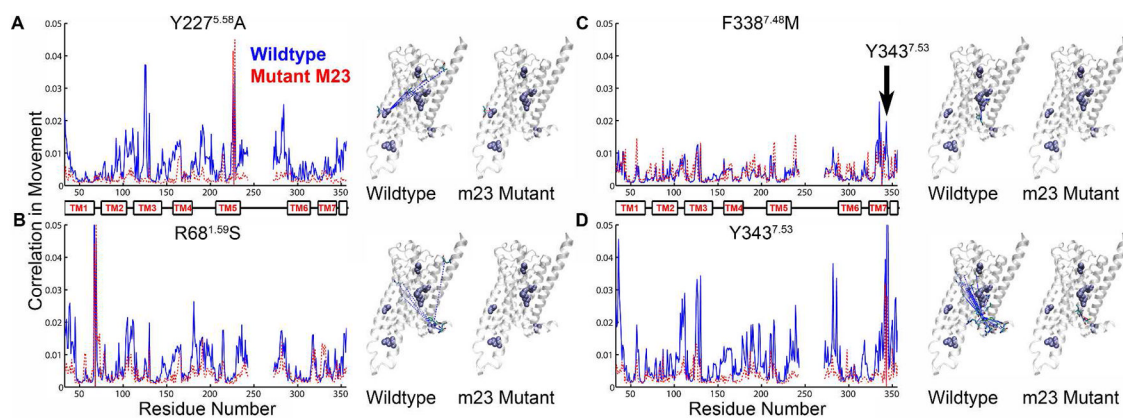


**Figure 2.** Schematic representation of the free energy landscape of the inactive state of  $\beta_1$ AR. The blue line represents the free energy profile for wt- $\beta_1$ AR and the dashed red line is for the thermostable mutant m23- $\beta_1$ AR. The schematic energy landscape is based on the data on the ensemble of conformations sampled in the MD simulations of these two receptors (Figure 1 and Figure S3).



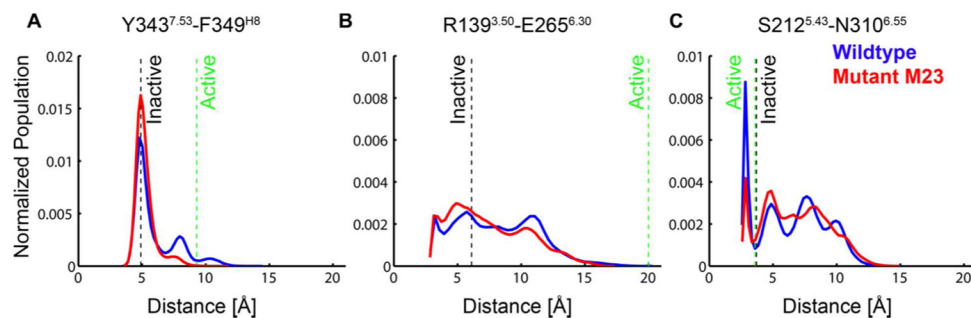
**Figure 3.**

The difference in the first order entropy ( $\Delta S^1$ ) for each amino acid in m23- $\beta_1$ AR and wt- $\beta_1$ AR. **A.**  $\Delta S^1$  plotted against the position in the amino acid sequence. Filled black circles show the location of the 6 thermostabilizing mutations in m23- $\beta_1$ AR. **B.** The difference in first order entropy between wt- $\beta_1$ AR and m23- $\beta_1$ AR is depicted on a cartoon of the m23- $\beta_1$ AR structure: regions in blue have lower entropy in m23- $\beta_1$ AR compared to wt- $\beta_1$ AR, while regions in red have higher entropy. The side chains of thermostabilizing residues are shown as black spheres.



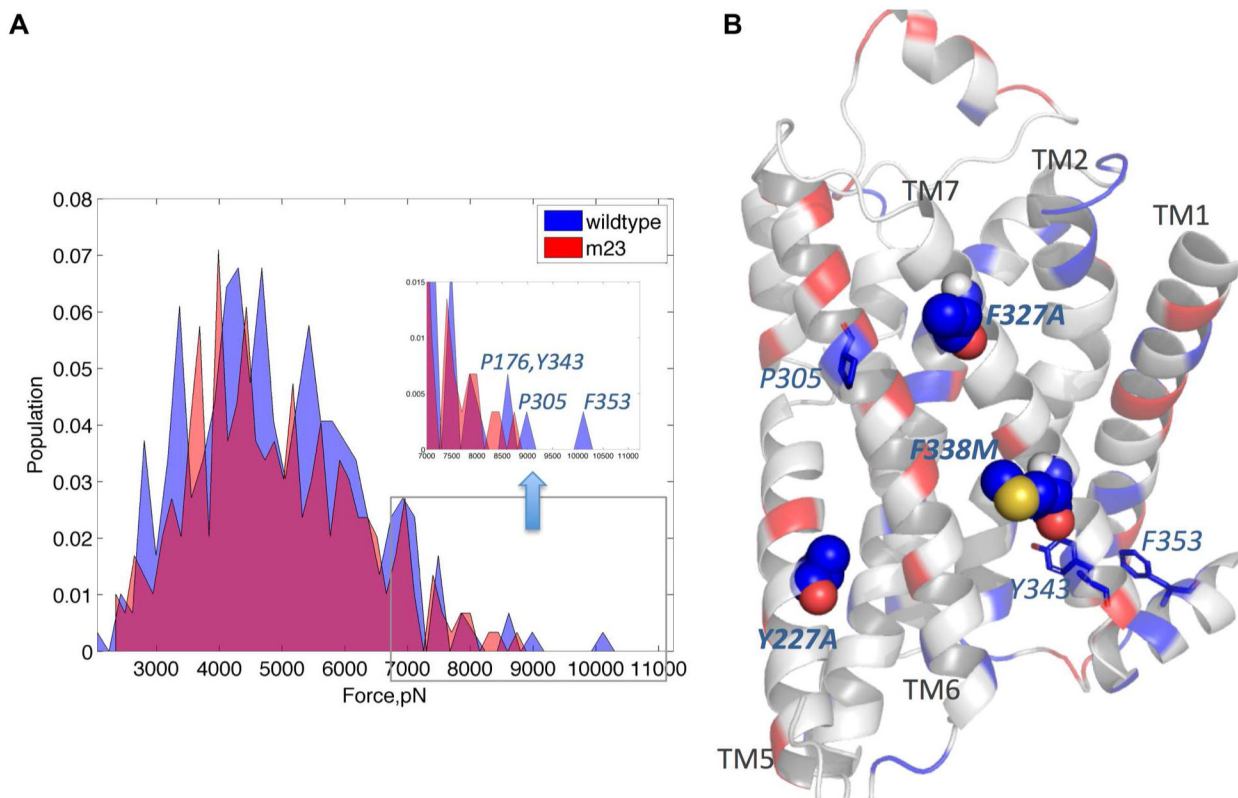
**Figure 4.**

Calculated correlated movement of thermostabilizing residues with all the residues in the receptor. Each graph shows the calculated correlation in movement versus the amino acid number for both wt- $\beta_1$ AR (solid blue line) and the thermostable mutant m23- $\beta_1$ AR (dashed red line). Associated to the right of each graph are two cartoons of the  $\beta_1$ AR structure showing the side chains of thermostabilizing amino acid residues (space filling models, purple) and amino acid residues that show correlated movement (sticks; C, cyan; O, red; N, blue). In each figure, dashed lines link the Ca atom of the thermostabilizing mutation with the amino acid residue showing correlated movement: wt- $\beta_1$ AR, blue; m23- $\beta_1$ AR, red. Both **A.** the Y227A and **B.** R68S thermostabilizing mutations reduce correlated movement between the thermostabilizing residue and many residues throughout the receptor. **C.** The major effect of the F338M mutation is to cause a loss in correlated movement between the residue and Y343<sup>7.53</sup>. **D.** The residue Y343<sup>7.53</sup>, which is affected by the neighboring mutation F338M, shows reduced correlated movement to many residues in the receptor.



**Figure 5.** m23- $\beta_1$ AR is more often in the inactive state as defined by distance distributions for inter-residue contacts related to receptor activation. The population of distances measured between specific atoms plotted; m23- $\beta_1$ AR, red; wt- $\beta_1$ AR, blue. Distances for the inactive state (black dashed line) were defined from the crystal structure of m23- $\beta_1$ AR (PDB ID: 2Y04) and the active state distance was estimated as the corresponding distance in the  $\beta_2$ AR-Gs bound crystal (PDB ID: 3SN6) (green dashed line) **A.** Distance between  $C_\gamma$  atoms of Y343<sup>7.53</sup>-F349<sup>H8</sup>; **B.** Minimum distance between R139<sup>3.50</sup>-E265<sup>6.30</sup> i.e. the 'ionic lock' at the intracellular end of TM3 and TM6. **C.** O $\gamma$  on S212<sup>5.43</sup> – N $\delta$  on N310<sup>6.55</sup>.





**Figure 6.**

**A.** Comparison of population density distribution of forces calculated on each residue. Forces are calculated by taking averages within 3 Å of each residue position, including both bonded and non-bonded force components; Populations (proportion of residues within each force range) are plotted with respect to the calculated force; wt- $\beta_1$ AR, blue; m23- $\beta_1$ AR, red; overlap, magenta. Regions of high stress in the wt- $\beta_1$ AR are shown in the inset and the residues that contribute most to the peaks with high force values are shown. **B.** Differences in forces calculated between m23- $\beta_1$ AR and wt- $\beta_1$ AR are depicted on the structure of m23- $\beta_1$ AR. Regions that show a change in force of more than 10 pN are highlighted. Regions of increased inter-residue forces in m23- $\beta_1$ AR compared to wt- $\beta_1$ AR, are shown in red and decreased stress are in blue. The thermostabilizing residues that show major stress reduction are shown as spheres and the other residues with significantly lower stress are shown as sticks.