

Chem Theory Comput. Author manuscript; available in PMC 2014 July 09.

Published in final edited form as:

J Chem Theory Comput. 2013 July 9; 9(7): 3151–3164. doi:10.1021/ct400104x.

Accurate calculation of mutational effects on the thermodynamics of inhibitor binding to p38α MAP kinase: a combined computational and experimental study

Shun Zhu, Sue M. Travis, and Adrian H. Elcock
Department of Biochemistry, University of Iowa, Iowa City, IA 52242

Abstract

A major current challenge for drug design efforts focused on protein kinases is the development of drug resistance caused by spontaneous mutations in the kinase catalytic domain. The ubiquity of this problem means that it would be advantageous to develop fast, effective computational methods that could be used to determine the effects of potential resistance-causing mutations before they arise in a clinical setting. With this long-term goal in mind, we have conducted a combined experimental and computational study of the thermodynamic effects of active-site mutations on a well-characterized and high-affinity interaction between a protein kinase and a small-molecule inhibitor. Specifically, we developed a fluorescence-based assay to measure the binding free energy of the small-molecule inhibitor, SB203580, to the p38a MAP kinase and used it measure the inhibitor's affinity for five different kinase mutants involving two residues (Val38 and Ala51) that contact the inhibitor in the crystal structure of the inhibitor-kinase complex. We then conducted long, explicit-solvent thermodynamic integration (TI) simulations in an attempt to reproduce the experimental relative binding affinities of the inhibitor for the five mutants; in total, a combined simulation time of 18.5 \(\mu\)s was obtained. Two widely used force fields – OPLS-AA/L and Amber ff99SB-ILDN - were tested in the TI simulations. Both force fields produced excellent agreement with experiment for three of the five mutants; simulations performed with the OPLS-AA/L force field, however, produced qualitatively incorrect results for the constructs that contained an A51V mutation. Interestingly, the discrepancies with the OPLS-AA/L force field could be rectified by the imposition of position restraints on the atoms of the protein backbone and the inhibitor without destroying the agreement for other mutations; the ability to reproduce experiment depended, however, upon the strength of the restraints' force constant. Imposition of position restraints in corresponding simulations that used the Amber ff99SB-ILDN force field had little effect on their ability to match experiment. Overall, the study shows that both force fields can

Correspondence to: Adrian H. Elcock.

adrian-elcock@uiowa.edu.

Supporting Information

Principal moieties of the inhibitor SB203580; probability distribution function of the key hydrogen bond observed in simulations of the wild-type p38 α -SB203580 complex; scheme for performing alchemical transformations of p38 α mutants; quadratic fits of the normalized fluorescence data obtained from titrations of p38 α protein with SB203580; time dependence of root mean square deviations (RMSDs) for selected groups of atoms in harmonically restrained simulations of wild-type p38 α in complex with SB203580; plots of $dH/d\lambda$ versus λ for unrestrained and restrained simulations of p38 α mutants using the OPLS-AA/L force field; root mean square deviation (RMSD) versus force constant for 12 ns restrained simulations of wild-type p38 α kinase in complex with SB203580 using the OPLS-AA/L force field; time dependence of $\Delta\Delta$ G values calculated from simulations using the OPLS-AA/L force field; time dependence of Δ angles at residue 38 in simulations of the A51V mutant; time dependence of Δ angles at residue 38 in simulations of the V38I/A51V mutant; time dependence of Δ angles at residue 51 in simulations of the V38I/A51V mutant; dihedral distributions for inter-ring angles in the inhibitor sampled during unrestrained simulations using the OPLS-AA/L and Amber ff99SB-ILDN force fields; three partial charge sets for the inhibitor SB203580; additional dihedral restraints applied to SB203580 to maintain its "propeller-like" shape. This information is available free of charge via the Internet at http://pubs.acs.org.

work well for predicting the effects of active-site mutations on small molecule binding affinities and demonstrates how a direct combination of experiment and computation can be a powerful strategy for developing an understanding of protein-inhibitor interactions.

Introduction

Protein kinases are of significant current interest as drug targets because aberrantly regulated kinase activities are associated with a wide variety of human diseases ranging from rheumatoid arthritis to cancer. 1-3 While there have been significant successes in developing small-molecule inhibitors to target selected kinases – most notably the development of imatinib to treat chronic myelogenous leukemia⁴ – it has been recognized for a number of years that a major challenge to overcome with such inhibitors is the development of drug resistance. 5,6 Resistance can occur by a number of mechanisms, 7,8 but the most important in the case of kinases appears to be spontaneous mutations in the ATP-binding domain that diminish inhibitor binding while preserving the kinase's catalytic function. 9,10 Given the apparent ubiquity of resistance as a problem in the development of kinase inhibitors, it would be helpful to have methods that enable potential resistance-causing mutations to be predicted or identified before they are encountered clinically.

One way to achieve this is to develop sophisticated experimental genetic screens. 11 An alternative, and the basis for the present study, is to use computational methods to explicitly calculate the binding thermodynamics of an inhibitor with potential mutants of the kinase target: if such calculations could be performed accurately and rapidly enough it would be possible, in principle, to preemptively screen all likely mutations that might occur within the active site of a kinase and to begin design efforts to deal with those mutations predicted to cause resistance. A key requirement of such an approach, of course, is the ability to accurately compute the thermodynamic effects of active-site mutations on the binding affinities of small-molecule inhibitors for proteins. While there are a number of approaches that might be used for this purpose, the most accurate predictions of ligand-protein binding affinities are, in principle, to be obtained from thermodynamic integration (TI) or free energy perturbation (FEP) techniques that explicitly compute absolute or relative free energies of binding; a number of excellent reviews of the use of these methods to compute ligand-protein binding affinities have recently been published. 12-14 Methods such as TI have the advantage of allowing binding thermodynamics to be computed from molecular dynamics (MD) simulations that explicitly include solvent and that directly account for contributions made by changes in conformational flexibilities of the inhibitor and the protein. While these advantages are to be weighed against the significant computational expense associated with such calculations, the continuing increase in computer power means that calculations can now be performed for much longer periods of time than was originally possible. 15,16 One purpose of the present study, therefore, is to explore the extent to which explicit-solvent free energy calculations, combined with MD simulations and using modern computing resources, can accurately predict the experimentally-observed effects of singleresidue mutations on the affinity of an inhibitor for a protein kinase.

As a well-characterized model system for conducting a direct comparison of simulation and experiment we have chosen to study the interaction between the p38a MAP kinase and the competitive small-molecule inhibitor SB203580, which binds with nanomolar affinity in the kinase's ATP-binding pocket. ^{17,18} Since many resistance-causing mutations involve relatively conservative changes in amino acid, we have selected mutations with the deliberate intention of causing only modest perturbations of the ATP-binding pocket. Our goal in each case has been: (a) to present the inhibitor with mild challenges to binding without abolishing binding completely, (b) to experimentally measure the inhibitor's binding

affinity for the mutant kinase, and (c) to then attempt to reproduce the measured binding affinity using free energy calculations. Five mutants of p38a were selected for study (Figure 1): A51V, V38I, A51G, V38A, and A51V/V38I. The first two of these mutations explore the effects of decreasing the size of the binding pocket (by increasing the size of sidechains that contact the inhibitor), the next two explore the effects of increasing the size of the binding pocket, and the final mutation explores the combined actions of the first two mutations. Although all of these mutations have been selected primarily as vehicles for exploring the effects of modest structural changes on the binding affinity of a kinase inhibitor it is perhaps worth noting that (a) the A51V mutation has been found in somatic mutations associated with stomach cancer, ¹⁹ and (b) the so-called glycine-rich loop of the kinase, of which V38 is a part, is one of the most common locations for resistance-causing mutations to occur in the BCR-ABL kinase. ¹⁰

This paper is organized as follows. In Methods we outline (a) the development of an experimental assay that utilizes changes in the fluorescence intensity of the SB203580 inhibitor to directly measure protein-inhibitor binding affinities, and (b) the simulation protocols used to compute the binding free energy of the inhibitor for each mutant kinase relative to that of the wild type protein. In Results we compare the experimental relative binding free energies of each mutant with calculated relative binding free energies obtained using combined molecular dynamics simulations and thermodynamic integration calculations (MD/TI). We compare results obtained with two different simulation force fields – OPLS-AA/L²⁰ and Amber ff99SB-ILDN^{21,22} – and show that although both force fields perform well for most of the mutants, simulations performed with the former force field produces surprisingly poor estimates for mutants that contain the A51V mutation. We then show that these apparent deficiencies can be corrected by applying harmonic restraints to the positions of atoms of the protein backbone and the inhibitor; we also show, however, that the results obtained with this approach are sensitive to the strength of the harmonic restraints' force constant. Overall, the results indicate that in the right circumstances MD/TI simulations can compute the effects of active-site mutations on protein-inhibitor interaction thermodynamics with reasonable accuracy. If similar accuracy can also be achieved in calculations that focus instead on the binding of the kinase's true substrate (ATP), it should be possible to use MD/TI simulations to predict potential resistance-causing mutations in protein kinases prior to their occurrence in the clinic.

Experimental Section

Mutagenesis, expression, and purification of p38α proteins

The bacterial vector pET21a-6xHIS-rTEV-p38a encoding wild-type p38a (GenBank ID: CAQ52036.1) was a generous gift from Professor Ernesto J. Fuentes (University of Iowa). The mutants A51V, A51G, V38A, V38I, and A51V/V38I, were made using the QuikChange site-directed Mutagenesis Kit (Stratagene) and primers were synthesized by Integrated DNA Technologies. All mutant vectors were sequenced to confirm the DNA sequence (DNA core facility, College of Medicine, University of Iowa). All p38a proteins were expressed in E. Coli. strain Rosetta 2 (Novagen). Bacterial cultures were grown at 37 °C until the optical density (OD) reached 0.6 before being induced with 1 mM (final concentration) IPTG and incubated at 30 °C for 6 hours. Following centrifugation at 6000 rpm for 25 minutes (Beckman Coulter, model J2-21M), pellets were re-suspended in buffer containing 50 mM Tris base, 500 mM NaCl, 25 mM imidazole, 5% glycerol and 1 mM DTT at pH 8.0. Cells were then lysed by sonication (Misonix Sonicator, model W-385) using three 2-minute rounds of sonication, with each round consisting of 40 pulses (pulse on for 1 second and then off for 2 seconds). The lysed cells were then centrifuged at 14000 rpm for 30 minutes (Beckman Coulter, model J2-21M). The supernatant was loaded onto a HiTrap Ni²⁺ affinity column (GE Healthcare) and elution was performed using a linear gradient of imidazole (25

to 300 mM); p38 α eluted at around 220 mM imidazole. Finally, the pooled fractions were loaded onto a Superdex 75 (GE Healthcare) gel filtration column and protein was eluted with buffer containing 5 mM HEPES, 150 mM NaCl, 20 mM MgCl₂-6H₂O and 2 mM TCEP at pH 7.4.

Fluorescence binding assay

We developed a fluorescence-based assay to experimentally measure the binding affinities of wild-type and mutant p38a proteins to the inhibitor SB203580. The assay exploits the fact that the inhibitor's intrinsic fluorescence increases significantly in intensity and 'blue shifts' from 420 to 394 nm upon binding to the protein (Figure 2a). Discontinuous titration experiments performed with a range of inhibitor and protein concentrations can therefore be used as a sensitive measure of the binding affinity.

For each mutant, titrations were performed in 12 separate vials, each with a different concentration of protein and a constant concentration (50 nM) of the inhibitor. Protein concentrations were determined using a NanoDrop spectrophotometer (Thermo Scientific, model ND-1000) with molecular weight and extinction coefficient set to 43.8 kDa and 53000 M⁻¹cm⁻¹ respectively, both of which were estimated using the ProtParam web server (http://ca.expasy.org/tools/protparam.html). The concentration of SB203580 (LC Laboratories) was set by weighing out appropriate amounts into an appropriate volume of buffer. The emission signal at 394 nM following excitation at 320 nm was measured for each of the twelve titrations using a FluoroLog-3 spectrofluorometer (HORIBA Scientific). All experiments were performed at 25 °C in the following buffer: 5 mM HEPES, 150 mM NaCl, 20 mM MgCl₂-6H₂O, 2% DMSO and 2 mM TCEP at pH 7.4

For each mutant, the raw fluorescence intensities at 394 nm were first normalized by the highest fluorescence intensity obtained in the set of 12 titrations. The normalized fluorescence intensities were then fit to a quadratic binding function in SigmaPlot 10.0^{23} in order to obtain the dissociation constant, K_d :

$$f=a \times \left(\frac{\left([P]_{T}+[L]_{T}+K_{d}\right)-\sqrt{\left([P]_{T}+[L]_{T}+K_{d}\right)^{2}-4[P]_{T}[L]_{T}}}{2[L]_{T}}+b\right)$$

Here $[P]_T$ is the total protein concentration, $[L]_T$ is the total inhibitor concentration, K_d is the dissociation constant, and f is the fractional saturation. The two additional fitting parameters a and b allow, respectively, the relative fluorescence intensity at saturation and the relative fluorescence in the absence of protein to both float during the fit; the former parameter was found to be important in helping determine the K_d of the weaker-binding mutants. In all cases, fits to the above equation were excellent $(r^2>0.98)$. Error bars for all measurements were obtained from the standard deviation of three independent repeats of the titration experiments. Finally, experimental binding free energies, ΔG , were calculated from the K_d using the standard relation $\Delta G = -RT \ln{(1/K_d)}$. The relative binding free energies, $\Delta \Delta G$, of the mutants to wild-type p38 α were then obtained using $\Delta \Delta G = \Delta G$ (mutant) $-\Delta G$ (wild-type).

Simulation system preparation

Residues Val38 and Ala51 in p38 α were selected for mutation because of their proximity to the inhibitor in the crystal structure of the p38 α -SB203580 complex (Figure 1; Wang et al., 1998). For each of the five mutations studied experimentally – i.e., A51V, A51G, V38A, V38I, and A51V/V38I – separate free energy calculations were performed for both inhibitor-

bound and inhibitor-unbound states in order to obtain the relative binding free energy, $\Delta\Delta G$. For simulations of the bound state, the p38 α crystal structure with SB203580 bound (PDB code: 1A9U) was used as the initial structure. ¹⁸ Since this structure is of the human form of p38 α , two minor modifications, L48H and T263A were made to the structure using SCWRL 4.0^{24} to ensure that its sequence was identical to that of mouse form of p38 α (the form used in our fluorescence experiments; see above); both of these residues are distant from the inhibitor's binding site so we do not expect them to make a significant contribution to the binding free energy. For simulations of the *unbound* state, the apo p38 α crystal structure (PDB code: 1P38) was used as the initial structure; ²⁵ since this structure is of the mouse form no structural modification was required.

Molecular dynamics simulations

Explicit solvent molecular dynamics (MD) simulations were performed using GROMACS $4.5.1^{26,27}$ on Kraken, a Cray XT5 supercomputer hosted by the National Institute for Computational Sciences at the Oak Ridge National Laboratory. The simulation system was set up in a cubic box (each side 85 Å) solvated with ~13000 TIP3P water molecules;²⁸ all simulations contained a total of ~43000 atoms. Separate sets of simulations were performed with the OPLS-AA/L²⁰ and Amber ff99SB-ILDN²² force fields. The former force field is a modification of the original OPLS all-atom force field²⁹ refitted to high level quantum-mechanical (QM) data²⁰. The latter force field is based on the Amber ff99SB force field,²¹ which in turn was based on a modified Amber ff94 force field;³⁰ again, improvement was obtained by matching a number of backbone and side chain dihedrals to high level QM calculations.²²

Prior to deriving force field parameters for the SB203580 it was necessary to decide upon the likely protonation states of the (three) ionizable nitrogen atoms in its imidazole and pyridine rings. To this end, pKa values were predicted using the SPARC web server (http://sparc.chem.uga.edu/sparc/; 31 Hilal et al., 1995). The predicted pKa of the pyridine nitrogen was 5.77, suggesting that it should be unprotonated at neutral pH; this assignment is consistent with the fact that it accepts a hydrogen bond from the backbone NH of Met109 in the crystal structure of the complex with p38a. 18 The two imidazole nitrogen atoms were assigned pKas of 2.62 and 10.75 respectively (when viewed from top to bottom in Figure S1); these assignments were again found to be consistent with the arrangement seen in the crystal structure of the p38a-SB203580 complex: the predicted unprotonated nitrogen accepts a (charged) hydrogen bond from the side chain of Lys53.

Having decided on protonation states, it was possible to derive partial charge sets for the inhibitor for simulations with the two force fields. For simulations using the OPLS-AA/L force field, the partial charges for the inhibitor were initially obtained using the CM1A method³² (Table S1) as has in the past been used successfully by others.³³ However, in preliminary simulations performed with this charge set the key hydrogen bond between the pyridine ring of the inhibitor and the backbone NH group of Met109 was rapidly lost, suggesting that the charge set significantly underestimates the strength of this interaction (Figure S2). Using an approach that we described previously, ³⁴ therefore, we opted instead to derive partial charges for the inhibitor on the basis of analogy with functional groups already parameterized within the OPLS-AA/L force field. The inhibitor's structure can be usefully decomposed into five principal moieties: (1) imidazole; (2) fluorobenzene; (3) pyridine; (4) benzene; and (5) sulfoxide (Figure S1). Since each of these groups has already been parameterized in the OPLS-AA/L force field, ²⁰ it is straightforward to merge their partial charge sets to obtain a complete set of charges for the inhibitor. For simulations using the Amber ff99SB-ILDN force field, the partial charges of the inhibitor were obtained using the standard protocol³⁵ of performing restrained electrostatic potential (RESP) fits; these

calculations were performed using the RED web server 36 (http://q4mdforcefieldtools.org/REDS/).

For simulations using the OPLS-AA/L force field, the bonded and non-bonded parameters for SB203580 were also derived by analogy with functional groups already present in the OPLS-AA/L force field. For simulations using the Amber ff99SB-ILDN force field, the bonded and non-bonded parameters for SB203580 were parameterized using the generalized AMBER force field (GAFF) approach.³⁷ In order to maintain the correct "propeller-like" arrangement of the inhibitor's aromatic groups seen in its crystal structure with p38 α , additional dihedral restraints were applied to the ring atoms in all simulations (Table S2).

All MD simulations were performed in the NVT ensemble. The temperature was maintained at 298 K using the Langevin thermostat with a friction coefficient of 1 ps⁻¹. For van der Waals interactions and short-range electrostatic interactions, a 10 Å cutoff was used; for long-range electrostatic interactions, the smooth Particle Mesh Ewald (PME) method was used. St LINCS³⁹ was used to constrain all bonds, enabling a 2 fs time step to be used. For each simulation, energy minimization using the steepest descent algorithm was first carried out for 1000 steps, followed by 600 ps of 'equilibration' that involved a stepwise heating of the system from 50 K to 298 K (through 100 K, 150 K, 200 K, and 250 K). Following this period of equilibration, 'production' simulations were run for 12 ns with snapshots saved every 1 ps and energies saved every 0.1 ps for further analysis.

Free energy calculations

The use of thermodynamic integration (TI) methods allows us to calculate the free energy changes associated with mutations in both inhibitor-bound and unbound states of the protein, i.e., ΔG_{bound} and $\Delta G_{unbound}$, respectively. These free energy changes were calculated by performing a number of independent MD simulations at different values of the coupling parameter λ that interpolates between the wild-type and mutant forms of the kinase. In order to practically implement this, a number of new hybrid residue types were defined in GROMACS. As shown in Figure S3, for example, for the A51V mutation a hybrid residue type was constructed by combining the atoms of alanine and valine into a single topology file; atoms not present in one or other of the two residue types are represented by noninteracting 'dummy' atoms in such a way that the hybrid residue is effectively an alanine at $\lambda = 0$ (wild-type) and a valine at $\lambda = 1$. The free energy changes, ΔG_{bound} and $\Delta G_{unbound}$, were each found by integrating the ensemble average of $dH/d\lambda$ accumulated in independent MD simulations performed at each of the following eleven values of λ : 0.0, 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, 0.8, 0.9, 1.0. All integrations were performed using Simpson's rule as this has been shown by the Boresch group to be a considerably more robust method of integration than the more commonly used trapezoid method. 40,41 To avoid numerical instabilities at the end-points (i.e. near $\lambda = 0$ and $\lambda = 1$), soft-core potentials⁴² were applied to both van der Waals and Coulombic interactions with the coefficients α , ρ , and σ set to 0.5, 1.0, and 0.3, respectively (for a good tutorial on the selection of parameters see http:// www.dillgroup.ucsf.edu/group/wiki/index.php?title=Free_Energy:_Tutorial).

To estimate statistical uncertainties, three independent 12 ns MD simulations were performed at each λ value for each mutant, in each case starting with a different set of initial velocities assigned to the atoms in the system. If we assume that the three estimates of $dH/d\lambda$ obtained at each λ are truly independent, it is possible to obtain a total number of 3^{11} estimates of both ΔG_{bound} and $\Delta G_{unbound}$; we therefore chose to estimate the statistical error of ΔG_{bound} and $\Delta G_{unbound}$ from the standard deviation of these 3^{11} integrations of $dH/d\lambda$. Finally, following standard statistics, ⁴³ the standard deviation of $\Delta\Delta G$ was obtained as the square root of the sum of the squared standard deviations of ΔG_{bound} and $\Delta G_{unbound}$.

The total amount of simulation time accrued during this project can be computed as follows. Each $\Delta\Delta G$ estimate required 22×12 ns = 264 ns of MD simulation (11 λ values for the bound state, 11 λ values for the unbound state). For the OPLS-AA/L force field we obtained 40 $\Delta\Delta G$ estimates: 3 replicates of 5 mutants for unrestrained simulations, 3 replicates of 5 mutants for restrained simulations using a force constant of 0.239 kcal/mol/Ų (i.e. 100 kJ/mol/nm²; see Results), 1 replicate of 5 mutants using a force constant of 0.0239 kcal/mol/Ų and 1 replicate of 5 mutants using a force constant of 2.39 kcal/mol/Ų. For the Amber ff99SB-ILDN force field, we obtained 30 $\Delta\Delta G$ estimates since we did not perform position-restrained calculations using force constants of 0.0239 and 2.39 kcal/mol/Ų. The total simulation time used in this study therefore amounts to 70×264 ns = 18.48 μs .

Calculations using the MM/GBSA method

An alternative to computing relative free energies of binding by performing MD simulations at a number of independent, intermediate λ values is to use a so-called 'end point' method that involves calculations only at the extreme λ values.⁴⁴ One such method is the molecular mechanics/generalized Born surface area (MM/GBSA) method; 45-47 a large number of studies have used this approach in ligand-receptor calculations. 48-55 The MM/GBSA method estimates the binding free energy, ΔG , as the difference in free energies of the ligandreceptor complex and its separated components, with each free energy being modeled as a sum of the following terms: $G = E_{int} + E_{elec} + E_{vdw} + G_{sol,p} + G_{sol,np} - TS$. E_{int} is the internal energy due to the traditional bonded terms of the force field, i.e. bond stretches, angles and dihedrals; Eelec and Evdw are the electrostatic and van der Waals nonbonded energies respectively; Gsol,p is the polar component of the solvation free energy, which is calculated here using the OBC generalized Born solvation model⁵⁶, with parameters $\alpha = 1$, $\beta = 0.8$ and $\gamma = 4.85$, as these parameters have been shown to work well in similar systems.⁵³ G_{sol,np} is the non-polar component of the solvation free energy, calculated using $G_{sol,np} = \gamma \cdot SASA$ where $\gamma = 0.0054$ kcal/mol⁵⁷ and SASA is the solvent accessible surface area. As is common in applications of the MM/GBSA method, the entropy term, TS, was omitted from the calculations reported here.

For each mutant, free energies were computed at the $\lambda=1$ end states using snapshots sampled at 10 ps intervals from 12 ns MD trajectories; since each trajectory was performed in triplicate, a total of 360 snapshots were used to obtain the binding free energy of each mutant. For the wild-type, free energies were computed at the $\lambda=0$ end state using snapshots sampled from all independent simulations; since five mutants were studied, and since each was simulated in triplicate, a total of $5\times 3\times 120=1800$ snapshots were available for calculating the binding free energy of the wild-type protein. The relative binding free energies, $\Delta\Delta G$, of all mutants were obtained by subtracting the binding free energy of the wild-type kinase. All calculations were performed using the 'rerun' option of the main MD engine, 'mdrun', in GROMACS. 26,27

Results

Experimental measurements of $\Delta\Delta G$ of p38 α mutants

In order to determine experimentally the binding affinities of p38 α mutants with the inhibitor SB203580, we developed a fluorescence-based binding assay; this assay exploits the fact that the fluorescence intensity of the inhibitor changes significantly when it binds to the protein (Figure 2a). To validate the assay, it was first applied to the wild-type p38 α so that comparison with previously reported binding affinities for SB203580 could be made. The dissociation constant (K_d) measured here for SB203580 with wild-type p38 α is 9.0 \pm 1.7 nM (Figure 2b). This agrees well with the previously reported values of 10 nM, ⁵⁸ 17 nM, ⁶⁰ and 4 nM; ⁶¹ the largest difference in free energy between the present value

and any of the other reported values, therefore, is 0.48 kcal/mol (i.e. a 4% difference). Having validated the assay for the wild-type protein, binding affinities of SB203580 for the five mutants examined here were determined using the same assay; the measured ΔG values are listed in Table 1 (see Figure S4 for fits of the experimental data for all mutants).

The two mutations that decrease the size of the sidechains contacting the inhibitor, V38A and A51G, decrease the binding affinity of the inhibitor by 2.5 and 0.8 kcal/mol respectively (Table 1). The former result is qualitatively consistent with a previous study showing that this mutation reduces cross-linking of the kinase to the similar ligand ¹²⁵I-SB206718 (identical to SB203580 except that the latter's fluorine is replaced by iodine⁶²), and both results are in line with our naïve expectations given that they entail the loss of hydrophobic contacts with the inhibitor. It is not immediately obvious why the V38A mutation should be so much more deleterious to binding than the A51G mutation since according to the Ligand-Protein Contacts web server⁶³ the two wild-type sidechains bury essentially the same amounts of surface area in binding to the inhibitor (22 and 20 Å² for V38 and A51 respectively). But this absence of an obvious explanation suggests that correctly reproducing the relative magnitudes of the effects of these two mutations should be a challenge for the computer simulations.

In contrast to what was seen with the above mutations, the two mutations that increase the size of the sidechains contacting the inhibitor, A51V and V38I, have qualitatively quite different effects on the binding affinity of the inhibitor: the A51V mutation decreases the binding affinity of the inhibitor by 2.0 kcal/mol, while the V38I mutation increases the binding affinity by 0.7 kcal/mol (Table 1). These two results can be at least qualitatively rationalized on the basis of simple molecular modeling performed on the crystal structure of the p38a-SB203580 complex (PDB code: 1A9U). When we use the sidechain modeling program SCWRL4²⁴ to model the A51V mutation (Figure 3), pronounced steric clashes are apparent between the valine sidechain and the fluorophenyl and pyridine rings of the inhibitor: the closest distance between heavy atoms is 3.15 Å. In contrast, when we carry out the same procedure with the V38I mutation (Figure 3), no clashes are apparent between the isoleucine sidechain and the inhibitor, and the sidechain appears much freer to rotate. That the V38I mutation introduces an additional CH2 group to contact the inhibitor without introducing any new steric clashes, therefore, provides a reasonable explanation for the finding that the mutation increases the binding affinity of the inhibitor. Interestingly, the idea that increasing the size of sidechains contacting the inhibitor can cause an increase in the binding affinity of the p38a-SB203580 complex is not without precedent: it has been shown previously that the A157V mutation (also in the active site of p38a) decreases the IC₅₀ for the inhibitor from 30 to 6 nM.⁶⁴ Evidently, therefore, despite the inhibitor's high affinity for p38a, there is room to improve its affinity by improving the packing of the protein's sidechains around it.

Since the two size-increasing single mutations, A51V and V38I have qualitatively different thermodynamic effects on SB203580's binding affinity, it was of interest to establish whether the two effects were additive, especially since the two sites are physically close to one another: the distance between the Ca atom of Val38 and Ala51 is 6.4 Å in the p38a-SB203580 complex crystal structure. We therefore measured the binding affinity of the double mutant, A51V/V38I, and found that the overall effect is to decrease the binding affinity by 1.0 kcal/mol; this is to be compared with the predicted combined effects of the two single mutations ($\Delta\Delta G = 2.0 - 0.7$ kcal/mol = 1.3 kcal/mol). The result is therefore somewhat equivocal: a very modest degree of cooperativity (~0.3 kcal/mol) may be present but it is not sufficiently large in magnitude to be clearly outside of the error in the experimental measurements.

Free energy calculations of $\Delta\Delta G$ of p38 α mutants

Having experimentally measured the binding free energy of SB203580 to a number of sitedirected mutants of the active site of p38a we next attempted to determine whether these data could be reproduced by explicit-solvent TI/MD simulations. Our initial round of free energy simulations used the OPLS-AA/L force field with parameters for the inhibitor devised to be as consistent as possible with the rest of the force field (see Methods). TI calculations with this force field produced very reasonable results for three of the mutants – A51G, V38A, and V38I (cyan, green, and yellow symbols in Figure 4a) – and in particular, were able to capture the finding that the V38A mutation is significantly more damaging to binding than is the A51G mutation (see above). But the TI results for the remaining two mutants – both of which involve the A51V substitution – were qualitatively incorrect (blue and red symbols in Figure 4a; Table 1): the simulations predict that both mutants should have an increased binding affinity for the inhibitor, when in fact their affinities are substantially decreased. We consider possible causes of these discrepancies in the Discussion. To explore whether similar results might be obtained with an alternative force field, we repeated the entire set of free energy calculations using the Amber ff99SB-ILDN force field (see Methods). In this case, the calculations performed considerably better (Figure 4b): qualitatively correct results were obtained for all five mutants, and although the calculations now overestimated the extent to which the A51V mutations disfavor inhibitor binding, the correspondence with the experimental data was markedly improved. Overall, the correlation coefficient increased from 0.49 (with OPLS-AA/L) to 0.76 (with Amber ff99SB-ILDN), and the average unsigned error in the computed $\Delta \Delta G$ values decreased from 1.1 to 0.7 kcal/mol.

Imposing harmonic position restraints can significantly improve calculated $\Delta\Delta G$ values

While there are several likely causes of the poor results obtained when the OPLS-AA/L force field is applied to the A51V-containing mutants (see below), one possible interpretation is that the force field may be allowing the protein-inhibitor complex to adopt conformations that, in reality, would not be sampled. If so, and assuming that the crystal structure of the wild-type kinase in complex with the inhibitor provides a good guide to what the A51V mutant *should* look like in complex with the inhibitor, then it might be expected that adding modest position restraints to limit the conformational freedom of the complex might lead to somewhat improved binding affinity predictions. To test this idea, we repeated the TI calculations with harmonic restraints applied to the backbone atoms of the protein and to the heavy atoms of the inhibitor; no harmonic restraints were imposed on the protein sidechains as it was felt important to allow them to freely sample alternative conformations (see Figure S5 for plots comparing the relative flexibilities of these groups).

Using an arbitrarily selected force constant of 0.239 kcal/mol/Ų (i.e. 100 kJ/mol/nm^2), the relative binding free energies of all five p38 α mutants were calculated in exactly the same way as for the unrestrained simulations. As expected, the primary effect of the position restraints was to affect the calculation of the free energy changes in the inhibitor-bound state (ΔG_{bound}) while making little difference to the calculation of the free energy changes in the inhibitor-free state ($\Delta G_{unbound}$) (see Figure S6). More importantly, however, the imposition of harmonic restraints hugely improved the calculated $\Delta\Delta G$ values for the A51V and A51V/V38I mutants without adversely affecting the calculated $\Delta\Delta G$ values for the other mutants (Figure 5a). In fact, the correlation coefficient for the OPLS-AA/L results improved from 0.49 without restraints to 0.95 with restraints; similarly, the average unsigned error improved from 1.1 to only 0.3 kcal/mol. Interestingly, we found that the same restraints could also be imposed in simulations with the Amber ff99SB-ILDN force field without causing a significant adverse change in the force field's ability to reproduce the experimental

data (Figure 5b): the correlation coefficient and average unsigned errors changed to 0.83 and 0.7 kcal/mol respectively when restraints were included.

Although it is encouraging that the mixed performance of the OPLS-AA/L force field can be improved by adding position restraints, it is important from a practical perspective to ask how strong these restraints must be in order to have a 'correcting' effect. To address this question, we again repeated our TI calculations with the OPLS-AA/L force field, but using instead force constants that were either an order of magnitude weaker or stronger (i.e. 0.0239 and 2.39 kcal/mol/Å² respectively); for reference, a plot showing the effects of increasing force constant on the flexibilities of backbone, sidechain and inhibitor atoms is shown in Figure S7. Interestingly, the computed $\Delta\Delta G$ values of the two A51V-containing mutants show a very significant dependence on the magnitude of the force constant, increasing linearly with the logarithm of the force constant (Figure 6). In contrast, the three non-A51V-containing mutants, A51G, V38A, and V38I, show much less dependence on the force constant, although the thermodynamically favorable V38I mutation is somewhat more poorly reproduced as the strength of the position restraints increases. The latter result suggests that correctly reproducing the favorable effect of this particular mutation requires a degree of conformational adjustment on the part of either (or both) the protein and the inhibitor. Overall, however, the results indicate that while the imposition of position restraints can overcome apparent limitations in a simulation force field, the strength of the restraints must be carefully chosen.

One final question that we have addressed is whether similar results could be obtained using a simpler computational approach that avoids the need to simulate intermediate λ values in the calculation of relative binding free energies. To this end, we have used the MM/GBSA method, ⁴⁴⁻⁴⁷ in which the energies of MD snapshots sampled at $\lambda = 0$ and 1 are calculated as a sum of a molecular mechanics component and an implicit solvent component that combines continuum electrostatic and surface area-dependent terms (see Methods). The results obtained when the MM/GBSA method is used to compute energies for snapshots sampled during the unrestrained simulations with both force fields are shown in Figure 7; corresponding results obtained using snapshots sampled during the simulations restrained with a force constant of 0.239 kcal/mol/Å² are shown in Figure 8. Interestingly, in the absence of position restraints, the correlation between the MM/GBSA calculations and experiment is quite poor with both force fields: correlation coefficients of 0.56 and 0.35 are obtained for the OPLS-AA/L and Amber ff99SB-ILDN force fields respectively. When position restraints are included, however, both force fields produce much better agreement with experiment: the correlation coefficients become 0.79 and 0.75 for the OPLS-AA/L and Amber ff99SB-ILDN force fields respectively, which are comparable to but somewhat lower than those obtained using the much more expensive TI calculations.

Discussion

The potential utility of free energy calculation methods such as FEP and TI for applications such as computer-aided drug design was first suggested and demonstrated in the mid-1980s⁶⁵⁻⁷⁰ and such methods continue to be widely used as tools for studying the thermodynamics of biomolecular interactions. In recent years, considerable interest has been focused on understanding the *absolute* thermodynamics of ligand binding to a variety of model protein systems such as T4 lysozyme, ⁷¹⁻⁷⁸ the FK-506 binding protein (FKBP), ⁷⁹⁻⁸¹ and cytochrome P450. ⁸² There also remains, however, significant interest in using such methods to compute *relative* binding free energies, for example, of closely related drug analogs binding to the same protein. ⁸³⁻⁸⁹

Conceptually, the present study belongs with other computational studies exploring the effects of mutations on a protein binding to a common ligand. Previous studies along these lines have included cases where the ligand is another protein, ^{90,91} a sugar⁹² or an enzyme substrate. 93-96 But of most relevance to the current work are those computational studies that have focused specifically on reproducing mutational effects on the binding thermodynamics of small molecules to proteins. 97-102 In comparison with all but the most recent of these studies the simulations reported here involve much longer simulation times; nevertheless, the overall accuracy of the results that we have obtained is broadly in line with that obtained in these previous studies. For example, a study of the effects of five mutations on the binding of the enzyme fructose 1,6-biphosphatase to adenosine monophosphate obtained excellent agreement with experiment: despite the use of comparatively short production periods of 306 ps, the error in the computed $\Delta\Delta G$ values in that study ranged from only 0.2 to 0.4 kcal/mol. ¹⁰¹ As a second example, in a study examining the interaction between protease inhibitors and three resistance-causing mutants of the Hepatitis C virus protease, errors ranging from 0.3 to 0.9 kcal/mol were obtained; interestingly, that study also used position restraints similar to those applied in the present study, although in that case the CHARMM force field was used. 100

In the present work we have used both unrestrained and restrained simulations in order to calculate $\Delta\Delta G$ values. We discuss the latter in detail below, but first we consider the results obtained from unrestrained simulations. One important point to note concerns the magnitudes of the discrepancies between the computed $\Delta\Delta G$ values and our experimentally measured values. As shown in Table 1, in the unrestrained calculations these discrepancies appear to scale with the size of the mutation – larger changes to sidechains appear more difficult for the calculations to quantitatively reproduce – and this trend holds true regardless of which force field is used in the simulations. Specifically, the mutations that involve the addition or loss of only a single carbon (A51G and V38I) are both very accurately reproduced: the unsigned error for both mutations with both force fields is only 0.3 ± 0.1 kcal/mol. In contrast, the mutants that involve the addition or loss of two carbons (A51V and V38A) are much less accurately reproduced: the unsigned error for these mutations with the two force fields is 1.3 ± 0.6 kcal/mol. We have already noted that the results for A51V with OPLS-AA/L are poor (Table 1), but at a quantitative level this mutation appears also to be difficult for Amber ff99SB-ILDN since it overestimates the change in binding affinity by 1.1 kcal/mol. And, while both force fields also qualitatively reproduce V38A's lower binding affinity, the magnitude of the effect is underestimated by 0.9 and 1.1 kcal/mol with OPLS-AA/L and Amber ff99SB-ILDN respectively. The fact that the thermodynamic effects of both of these mutations are difficult to reproduce indicates that mutations that introduce steric insults to inhibitor binding (A51V) may not be any more challenging than mutations that involve removal of interactions (V38A), at least from the point of view of producing quantitative agreement with experiment. This finding may have implications for attempts to model the effects of mutations on protein stability where, due to difficulties in modeling in sidechains, attention has previously been focused on those mutations that involve deletion of atoms. 103,104

A second result of the unrestrained simulations is that we find that increasing the 'production' period of the simulations (i.e. the period of time over which data are recorded) has equivocal effects on the calculated $\Delta\Delta G$ values. Most of the computed $\Delta\Delta G$ values are, perhaps surprisingly, relatively insensitive to changes in the duration of the production period (Figures S8 & S9). In the case of unrestrained simulations using the OPLS-AA/L force field (Figure S8a), for example, the most noticeable effect is on the computed $\Delta\Delta G$ value for the A51V mutant which decreases from 1.0 kcal/mol to -0.3 kcal/mol as the production period increases from 1 to 12 ns. Given that our experimentally determined value is 2.0 kcal/mol (Table 1) it is clear that the calculated result deviates further from the

experimental value as the simulation production period increases. In the case of unrestrained simulations using the Amber ff99SB-ILDN force field, on the other hand (Figure S9a), the most noticeable changes are to the $\Delta\Delta G$ values of the V38I and A51V/V38I mutants, both of which become closer to the experimental value over time.

While the dependence of the results on the duration of the production period is generally more muted in the simulations that include position restraints on the backbone atoms (Figures S8b and S9b), in certain cases (e.g. for the A51V/V38I mutant with the Amber ff99SB-ILDN force field) a clear time dependence remains. One likely cause of this remaining time dependence, and a significant possible source of discrepancies with experiment, is the relatively infrequent sampling of alternative rotameric states of sidechains that contact the inhibitor. In fact, infrequent sampling of sidechain rotamers appears to be an issue in all of our simulations, regardless of the force field used and regardless of whether or not position restraints are applied to the inhibitor and the protein backbone. As shown in Figures S10 and S11, for example, at position 38 we see very few transitions in the χ 1 dihedral over the course of the 12 ns production period in simulations of A51V-containing mutants; sampling is slightly better at position 51 in the same mutants (Figures S12 and S13) but remains infrequent. Such slow sampling of alternative sidechain conformations is not necessarily unrealistic since significant energy barriers to sidechain rotations are common 105 and since in seven different crystal structures of SB203580 in complex with p38 kinases (pdb codes 1A9U; 1PME; 2EWA; 3GCP; 3OBG; 3MPA; 3ZS5), the sidechain of V38 is always in the trans conformation. Nevertheless, it makes it difficult to be confident in the convergence of the computed free energy values.

This is especially a concern given that others have shown that proper consideration of alternative sidechain rotamers can be an important factor in obtaining accurate absolute binding free energies in ligand-receptor systems. For example, a number of studies have shown that inadequate sampling of the $\chi 1$ angle at Val111 – for which conformational rearrangements are slow - can have a critical effect on the computed binding free energies of aromatic molecules to T4 lysozyme. 73,75,78,82,106 A variety of different ways to correct for this problem have been proposed and have been explicitly demonstrated in the same system. One such method is the 'confine-and-release' protocol developed by Mobley, Chodera and Dill⁷⁶ in which the free energy contribution from a sidechain that is kinetically trapped within a single rotameric state is explicitly computed. An alternative comes the Roux group⁷⁸ showing that it is possible to accelerate convergence in absolute binding free energy calculations by using Hamiltonian exchange methods in combination with biasing potentials that are explicitly devised to improve sampling of the $\chi 1$ dihedral angles. Finally, the Berne and Friesner groups have shown that free energies that are independent of the initial rotameric states of sidechains can be achieved in simulations of only ~2 ns duration using a replica exchange solute-tempering method¹⁰⁶ that, in effect, simulates the ligand and its immediate binding site at a higher temperature; this same method has already been applied to a kinase system.⁵⁵ Any or all of these methods could be especially helpful for the present system.

It is perhaps because of the infrequent sampling of sidechain conformations, therefore, that we do not see any clear relationship between the length of the production period in the simulations and the accuracy with which the experimental results are reproduced. Because of this, therefore, we cannot conclusively rule out the possibility that increasing the production period of the simulations to a much longer timescale (e.g. microseconds) might change the results, either for better or for worse. The ability to perform much longer simulations thanks to the advent of superfast MD engines such as the purpose-built supercomputer Anton created by the Shaw group 107,108 should make it possible to eliminate incomplete sampling as a major concern in the near future. In the meantime, however, and

for the general user of simulations who does not have access to such machines, it appears worthwhile considering methods such as those outlined above, and others, that improve sampling in the computation of binding free energies. ^{76,78,106,109}

While the above discussion indicates that we should be cautious about the convergence of our computed $\Delta\Delta G$ values, it remains reasonably clear that the results we obtain for the A51V-containing mutants from unrestrained simulations with the OPLS-AA/L force field are qualitatively incorrect while those obtained with the Amber ff99SB-ILDN force field are qualitatively correct (Table 1). Given the highly interconnected nature of the parameters in current energy functions it can be extremely difficult to cleanly identify problem areas in a given force field. One potential contributor that we can probably rule out is differences in the conformational preferences of the inhibitor since with both force fields we have imposed the same set of dihedral restraints in order to ensure that the relative orientations of the fluorophenyl, pyridine and imidazole rings are preserved; as a result, the sampled distributions of these dihedral angles are essentially identical with the two force fields (Figure S14). Another possibility is the description of the kinase's conformational preferences, and in particular, those of the sidechains that contact the inhibitor: the OPLS and AMBER force fields have been shown by others, for example, to produce somewhat different $\chi 1$ rotamer distributions for residue types such as valine. ¹¹⁰ The infrequent sampling of these rotamers noted above – which is only one aspect of the more general problem of sampling that has been emphasized recently 111 – makes it difficult to determine if this is a major issue but, in any case, it should be noted that we can get excellent agreement with experiment from simulations that include backbone position restraints despite the fact that any errors in rotamer preferences would still be present in such simulations (restraints were not applied to sidechain atoms).

Still another possible contribution comes from the partial charges assigned to the inhibitor. In Methods we note that in preliminary MD simulations with the OPLS-AA/L force field, the use of CM1A charges for the inhibitor led to a rapid loss of the key hydrogen bonding interaction between the pyridine ring of the inhibitor and the amide hydrogen of the backbone of Met109. This led us to instead derive partial charges for the inhibitor by analogy with functional groups already parameterized in the OPLS-AA/L force field, and while these analogy-derived charges improved the stability of the hydrogen bond significantly, its stability in 12 ns MD simulations of the wild-type kinase is still clearly somewhat lower than that obtained with the Amber ff99SB-ILDN forcefield (Figure S2). Notably, the A51V mutation is the only one considered here that introduces a clear steric clash into the system and, given the proximity to the pyridine ring, it is quite possible that adjustments to the hydrogen bond might be required in order to accommodate it (Figure 3). If the simulation force field's description of the hydrogen bond interaction is unrealistically weak, it might not correctly account for the fact that there should be a significant energetic cost to altering the geometry of the hydrogen bond and might, therefore, underestimate how unfavorable the mutation is for binding of the inhibitor. Arguing against this possible explanation, however, is the fact that in additional simulations that we have performed using the OPLS-AA/L force field but with the Amber charges assigned to the inhibitor, we obtain a $\Delta\Delta G$ for the A51V mutant of 0.3 ± 0.4 kcal/mol, which is only marginally better than the result obtained when the analogy-derived OPLS-AA/L charges are used ($\Delta\Delta G = -0.3 \pm 0.4$ kcal/mol; Table 1).

Although the exact cause of the OPLS-AA/L force field's problems with the A51V-containing mutations remains difficult to establish unambiguously, it is important to note that these problems can be made to disappear when position restraints are applied to the protein backbone and the inhibitor: in fact, when set appropriately, restraints allow average unsigned errors as low as 0.3 kcal/mol to be achieved. The addition of position restraints to

simulations with the Amber ff99SB-ILDN force field made little difference to what was already a rather good agreement with experiment, although there was a slight decrease in the correlation. Despite this, there was one important respect in which the Amber ff99SB-ILDN results were also markedly improved by the addition of restraints: this was in the reproduction of the cooperativity of the V38I/A51V double mutant. Without restraints, the 'coupling energy' between the two mutation sites, defined as $\Delta\Delta G_{V38I/A51V}-\Delta\Delta G_{V38I}-\Delta\Delta G_{A51V}$, was computed to be -1.4 kcal/mol; with restraints the value reduces to -0.6 kcal/mol, which is much closer to the experimental value of -0.3 kcal/mol. Similarly, with OPLS-AA/L, the computed coupling energy is +0.2 kcal/mol (i.e. qualitatively incorrect) in the absence of restraints, and -0.6 kcal/mol in the presence of restraints. Interestingly, the imposition of position restraints also markedly improved the results obtained with the faster but more approximate MM/GBSA method for *both* force fields (compare Figures 7 and 8).

The fact that imposing position restraints can lead to better calculations of mutational effects echoes a previous result from our lab. In work aimed at developing rapid computational methods for screening the entire complement of human protein kinases we showed that it was possible to use homology models of protein kinases, together with a simple physical energy function, to qualitatively predict which would be strongly bound by a chosen smallmolecule inhibitor.³⁴ Importantly, two key features of that approach were: (a) to assume that all kinases use the exact same backbone conformation to bind the putative inhibitor, and (b) to assume that the inhibitor binds in an identical orientation to all kinases. Notably, when either of these apparently drastic and unrealistic restrictions was removed, significantly poorer results were obtained. In comparison with the explicit-solvent TI/MD simulations used here, the previous approach was obviously highly simplified. Despite that, the key result of that study was essentially the same as that obtained here: in both cases, better results can be obtained when a simulation force field is denied the opportunity to lead the modeled protein-inhibitor complex into different, and possibly wrong, conformations. Obviously, in an ideal situation, position restraints would not be necessary at all, and as force fields continue to improve the advantages of including them will likely disappear; as a temporary stop-gap measure, however, they appear to provide a reasonable solution when the force field has clearly identified limitations. Importantly, the conclusion that position restraints can improve sampling results in MD also emerges from a recent study by the Shaw group reporting very long timescale MD simulations of homology-modeled protein structures. 112 It is also important to note, however, that there are clear cases where the addition of position restraints will likely worsen rather than improve results. Here again, the T4 lysozyme system offers a good example, as ligands binding to the T4 lysozyme system can adopt a number of different orientations and correctly accounting for them can be important for correctly reproducing experimental binding free energies. 113 An additional point to note is that, as shown here, the best choice of force constant to use is likely to depend on the specific system and force field under study.

Finally, it is worth noting that another factor that might affect the ability to produce good agreement with experiment is the presence of multiple conformational equilibria within the kinase. It has been known for a number of years, for example, that a flipping of the D168-F169-G170 motif in p38a from a so-called 'DFG-in' to a 'DFG-out' conformation, in which a cryptic hydrophobic binding site is exposed, can be exploited by so-called class II inhibitors. ¹¹⁴ Kinetic data ¹¹⁵ have shown that the binding of such inhibitors is much slower than that of the more conventional class I (i.e. ATP-binding site) inhibitors such as SB203580, suggesting that the DFG-in and DFG-out conformations interconvert slowly; NMR data indicate that in the ligand-unbound state this exchange occurs on millisecond timescale. ¹¹⁶ Interestingly, NMR studies have indicated that the binding of SB203580 does not affect the equilibrium between the DFG-in and DFG-out conformations, ¹¹⁶ and crystallographic studies have shown that SB203580 can in fact bind to both

conformations. 116,117 Evidence for additional conformational dynamics within p38 α comes from another NMR study 118 showing that significant line broadening of a number of resonances occurs when SB203580 binds; notably, these include a number of amino acids that are close to the inhibitor (e.g. G33-V38; I84-L86). Interestingly, residual dipolar coupling (RDC) data reported in the same study also indicate that the backbone conformation of unbound p38 α in solution might differ from that observed in the crystal structure at residues G110-A111. Still another feature to potentially consider is that in its physiological state, p38 α is activated by phosphorylation at T180 and Y182; 119 NMR and H/D exchange studies have shown that this double phosphorylation has effects distributed throughout the protein and causes the activation loop to switch from the intermediate to the fast exchange regime. 120 It is to be noted that phosphorylation is not an issue in the present study as both the experiments and simulations have been performed in conditions in which p38 α is unphosphorylated; in any case, it has been shown that SB203580 binds with similar affinities to both forms of the enzyme. 121

A number of computational studies have already considered some of the above issues. The transition between the DFG-in and DFG-out conformations of p38 α , for example, has been directly observed in artificially accelerated MD simulations performed by our group ¹²² and by the Menziani group. ¹²³ Very long, unforced MD simulations of the same transition in other kinases have also been reported by the Shaw group ^{124,125} and a number of attempts to model the DFG-out conformations of other kinases for use in virtual screening applications have also been reported. ¹²⁶⁻¹²⁸ The finding that SB203580 binds with equal affinities to the DFG-in and DFG-out conformations of p38 α has been reproduced by the Chang group in recent computer simulations using the 'mining minima' method ¹²⁹ in conjunction with a continuum solvent model. ¹³⁰ Finally, a number of other computational studies focusing on p38 α as a model system for virtual screening applications have also been reported. ^{128,131-134}

Given the significant conformational transitions that occur within p38 α , and the timescale over which they occur, it is perhaps surprising that comparatively short simulations coupled with position restraints can yield relative binding free energies that are in good agreement with experiment. Presumably this is because the mutations studied here do not significantly affect either p38 α 's conformational dynamics or its conformational free energy landscape; the effects of mutations that are closer to the site of action (e.g. mutations of the DFG motif itself¹³⁵) might well be much more difficult to reproduce. For the future, therefore, experimental measurements of the binding thermodynamics of SB203580 to a wider range of mutants might enable the p38 α -SB203580 system to serve as a very challenging model for testing both enhanced conformational sampling methods and simulation force fields.

In summary, we have reported here a combined experimental and computational approach to study the effects of active-site mutations on the binding free energy of p38a MAP kinase to the inhibitor SB203580. We have shown that, depending on the site of the mutation, increasing the size of a sidechain contacting the inhibitor can either increase or decrease the binding affinity of a small-molecule inhibitor. We have also shown that, under the right circumstances, these changes in binding affinity can be quite accurately predicted by TI/MD simulations. Given that the conservative mutations studied here are quite common in cases of drug resistance, the results obtained here suggest that TI/MD simulations could eventually find use in predicting potential resistance-causing mutations prior to their occurrence in the clinic; the results reported by others in such applications 100,102 support this notion. However, to do this in a comprehensive fashion, at least for protein kinases, it will be important also to determine the effects of the mutations on the binding of the kinase's substrate ATP. This is especially true given that resistance-causing mutations have already

been identified that exert their effects by increasing the affinity for ATP without affecting the affinity for the inhibitor. 136

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

This work was made possible by the support of NIH R01 GM068648 to AHE. This work used the Extreme Science and Engineering Discovery Environment (XSEDE), which is supported by National Science Foundation grant number OCI-1053575; generous allocations of time on the supercomputers Kraken and Ranger were awarded to AHE under project TG-MCB090131.

References

- 1. McLean GW, Carragher NO, Avizienyte E, Evans J, Brunton VG, Frame MC. Nat. Rev. Cancer. 2005; 5:505–515. [PubMed: 16069815]
- 2. Shapiro GI. J. Clin. Oncol. 2006; 24:1770–1783. [PubMed: 16603719]
- 3. Schindler JF, Monahan JB, Smith WG. J. Dent. Res. 2007; 86:800–811. [PubMed: 17720847]
- 4. Deininger M, Buchdunger E, Druker BJ. Blood. 2005; 105:2640–2653. [PubMed: 15618470]
- 5. Daub H, Specht K, Ullrich A. Nat. Rev. Drug Discovery. 2004; 3:1001–1010.
- 6. Nardi V, Azam M, Daley GQ. Curr. Opin. Hematol. 2004; 11:35–43. [PubMed: 14676625]
- 7. Jänne PA, Gray N, Settleman J. Nat. Rev. Drug Discovery. 2009; 8:709–723.
- 8. Barouch-Bentov R, Sauer K. Expert Opin. Invest. Drugs. 2011; 20:153-208.
- Bikker JA, Brooijmans N, Wissner A, Mansour TS. J. Med. Chem. 2009; 52:1493–1509. [PubMed: 19239229]
- 10. Krishnamurty R, Maly DJ. ACS Chem. Biol. 2010; 5:121-138. [PubMed: 20044834]
- 11. Girdler F, Sessa F, Patercoli S, Villa F, Musacchio A, Taylor S. Chem. Biol. 2008; 15:552–562. [PubMed: 18559266]
- Aleksandrov A, Thompson D, Simonson T. J. Mol. Recognit. 2010; 23:117–127. [PubMed: 19693787]
- 13. Steinbrecher T, Labahn A. Curr. Med. Chem. 2010; 17:767–785. [PubMed: 20088755]
- 14. Michel J, Essex JW. J. Comput. Aid. Mol. Des. 2010; 24:639-658.
- 15. Straatsma TP, McCammon JA. Annu. Rev. Phys. Chem. 1992; 43:407–435.
- 16. Kollman P. Chem. Rev. 1993; 93:2395-2417.
- 17. Tong L, Pav S, White DM, Rogers S, Crane KM, Cywin CL, Brown ML, Pargellis CA. Nat. Struct. Biol. 1997; 4:311–316. [PubMed: 9095200]
- 18. Wang ZL, Canagarajah BJ, Boehm JC, Kassisa S, Cobb MH, Young PR, Abdel-Meguid S, Adams JL, Goldsmith EJ. Struct. Folding & Design. 1998; 6:1117–1128.
- 19. Stephens P, Hunter C, Bignell G, Edkins S, Davies H, Teague J, Stevens C, O'Meara S, Smith R, Parker A, Barthorpe A, Blow M, Brackenbury L, Butler A, Clarke O, Cole J, Dicks E, Dike A, Drozd A, Edwards K, Forbes S, Foster R, Gray K, Greenman C, Halliday K, Hills K, Kosmidou V, Lugg R, Menzies A, Perry J, Petty R, Raine K, Ratford L, Shepherd R, Small A, Stephens Y, Tofts C, Varian J, West S, Widaa S, Yates A, Brasseur F, Cooper CS, Flanagan AM, Knowles M, Leung SY, Louis DN, Looijenga LHJ, Malkowicz B, Pierotti MA, Teh B, Chenevix-Trench G, Weber BL, Yuen ST, Harris G, Goldstraw P, Nicholson AG, Futreal PA, Wooster R, Stratton MR. Nature. 2004; 431:525–526. [PubMed: 15457249]
- Kaminski GA, Friesner RA, Tirado-Rives J, Jorgensen WL. J. Phys. Chem. B. 2001; 105:6474–6487.
- 21. Hornak V, Abel R, Okur A, Strockbine B, Roitberg A, Simmerling C. Proteins-Struct., Funct., Bioinf. 2006; 65:712–725.

22. Lindorff-Larsen K, Piana S, Palmo K, Maragakis P, Klepeis JL, Dror RO, Shaw DE. Proteins-Struct., Funct., Bioinf. 2010; 78:1950–1958.

- 23. SigmaPlot V.10. Systat Software Inc.; Richmond, CA.:
- 24. Krivov GG, Shapovalov MV, Dunbrack RL. Proteins-Struct., Funct., Bioinf. 2009; 77:778–795.
- Wang ZL, Harkins PC, Ulevitch RJ, Han JH, Cobb MH, Goldsmith EJ. Proc. Natl. Acad. Sci. U. S. A. 1997; 94:2327–2332. [PubMed: 9122194]
- 26. Lindahl E, Hess B, van der Spoel D. J. Mol. Model. 2001; 7:306.
- 27. Hess B, Kutzner C, van der Spoel D, Lindahl EJ. Chem. Theory Comput. 2008; 4:435.
- 28. Jorgensen WL, Chandrasekhar J, Madura JD, Impey RW, Klein ML. J. Chem. Phys. 1983; 79:926.
- 29. Jorgensen WL, Maxwell DS, TiradoRives J. J. Am. Chem. Soc. 1996; 118:11225.
- 30. Cornell WD, Cieplak P, Bayly CI, Gould IR, Merz KM, Ferguson DM, Spellmeyer DC, Fox T, Caldwell JW, Kollman PA. J. Am. Chem. Soc. 1995; 117:5179.
- 31. Hilal SH, Karickhoff SW, Carreira LA. Quant. Struct. Act. Relat. 1995; 14:348.
- 32. Storer JW, Giesen DJ, Cramer CJ, Truhlar DG. J. Comput. Aid. Mol. Des. 1995; 9:87.
- 33. Tominaga Y, Jorgensen WL. J. Med. Chem. 2004; 47:2534. [PubMed: 15115396]
- 34. Rockey WM, Elcock AH. J. Med. Chem. 2005; 48:4138. [PubMed: 15943486]
- 35. Bayly CI, Cieplak P, Cornell WD, Kollman PA. J. Phys. Chem. 1993; 97:10269.
- 36. Vanquelef E, Simon S, Marquant G, Garcia E, Klimerak G, Delepine JC, Cieplak P, Dupradeau FY. Nucleic Acids Res. 2011; 39:W511. [PubMed: 21609950]
- 37. Wang JM, Wolf RM, Caldwell JW, Kollman PA, Case DA. J. Comput. Chem. 2004; 25:1157. [PubMed: 15116359]
- 38. Essmann U, Perera L, Berkowitz ML, Darden T, Lee H, Pedersen LG. J. Chem. Phys. 1995; 103:8577.
- 39. Hess B, Bekker H, Berendsen HJC, Fraaije J. J. Comput. Chem. 1997; 18:1463.
- 40. Bruckner S, Boresch S. J. Comput. Chem. 2011; 32:1320-1333. [PubMed: 21425289]
- 41. Bruckner S, Boresch S. J. Comput. Chem. 2011; 32:1303-1319. [PubMed: 21425288]
- 42. Beutler TC, Mark AE, Vanschaik RC, Gerber PR, van Gunsteren WF. Chem. Phys. Lett. 1994; 222:529.
- 43. Barlow, RJ. A Guide to the Use of Statistical Methods in the Physical Sciences. Wiley;
- 44. Hayes, JM.; Archontis, G. MM-GB(PB)SA calculations of protein-ligand binding free energies.. In: Wang, Lichang, editor. Molecular dynamics – studies of synthetic and biological macromolecules. InTech; Croatia: DOI:10.5772/37107
- 45. Srinivasan J, Cheatham TE III, Cieplak P, Kollman PA, Case DA. J. Am. Chem. Soc. 1998; 120:9401–9409.
- Kollman PA, Massova I, Reyes C, Kuhn B, Huo S, Chong L, Lee M, Lee T, Duan Y, Wang W, Donini O, Cieplak P, Srinivasan J, Case DA, Cheatham TE III. Acc. Chem. Res. 2000; 33:889–897. [PubMed: 11123888]
- 47. Massova I, Kollman PA. Perspect. Drug Discov. Des. 2000; 18:113-135.
- 48. Gohlke H, Kiel C, Case DA. J. Mol. Biol. 2003; 18:891–913. [PubMed: 12850155]
- 49. Gohlke H, Case DA. J. Comput. Chem. 2004; 25:238–250. [PubMed: 14648622]
- 50. Chachra R, Rizzo RC. J. Chem. Theory Comput. 2008; 4:1526–1540.
- Abel R, Young T, Farid R, Berne BJ, Friesner RA. J. Am. Chem. Soc. 2008; 130:2817–2831.
 [PubMed: 18266362]
- Ng CA, Oehme DP, Kato Y, Tanokura M, Brownlee RTC. J. Chem. Theory Comput. 2009;
 5:2886–2897.
- 53. Balius TE, Rizzo RC. Biochemistry. 2009; 48:8435–8448. [PubMed: 19627157]
- 54. Newhouse EI, Xu D, Markwick PRL, Amaro RE, Pao HC, Wu KJ, Alam M, McCammon JA, Li WW. J. Am. Chem. Soc. 2009; 131:17430–17442. [PubMed: 19891427]
- 55. Wang L, Deng Y, Knight JL, Wu Y, Kim B, Sherman W, Shelley JC, Lin T, Abel R. J. Chem. Theory Comput. 2013; 9:1282–1293.
- 56. Onufriev A, Bashford D, Case DA. Proteins: Struct. Funct. Bioinf. 2004; 55:383–394.

- 57. Sitkoff D, Sharp KA, Honig B. J. Phys. Chem. 1994; 98:1978-1988.
- 58. Regan J, Breitfelder S, Cirillo P, Gilmore T, Graham AG, Hickey E, Klaus B, Madwed J, Moriak M, Moss N, Pargellis C, Pav S, Proto A, Swinamer A, Tong L, Torcellini C. J. Med. Chem. 2002; 45:2994. [PubMed: 12086485]
- 59. Fabian MA, Biggs WH, Treiber DK, Atteridge CE, Azimioara MD, Benedetti MG, Carter TA, Ciceri P, Edeen PT, Floyd M, Ford JM, Galvin M, Gerlach JL, Grotzfeld RM, Herrgard S, Insko DE, Insko MA, Lai AG, Lelias JM, Mehta SA, Milanov ZV, Velasco AM, Wodicka LM, Patel HK, Zarrinkar PP, Lockhart DJ. Nat. Biotechnol. 2005; 23:329. [PubMed: 15711537]
- 60. Karaman MW, Herrgard S, Treiber DK, Gallant P, Atteridge CE, Campbell BT, Chan KW, Ciceri P, Davis MI, Edeen PT, Faraoni R, Floyd M, Hunt JP, Lockhart DJ, Milanov ZV, Morrison MJ, Pallares G, Patel HK, Pritchard S, Wodicka LM, Zarrinkar PP. Nat. Biotechnol. 2008; 26:127. [PubMed: 18183025]
- 61. Swann SL, Merta PJ, Kifle L, Groebe D, Sarris K, Hajduk PJ, Sun CH. Bioorg. Med. Chem. Lett. 2010; 20:5787. [PubMed: 20471255]
- 62. Gum RJ, McLaughlin MM, Kumar S, Wang ZL, Bower MJ, Lee JC, Adams JL, Livi GP, Goldsmith EJ, Young PR. J. Biol. Chem. 1998; 273:15605. [PubMed: 9624152]
- 63. Sobolev V, Sorokine A, Prilusky J, Abola EE, Edelman M. Bioinformatics. 1999; 15:327. [PubMed: 10320401]
- 64. Lisnock JM, Tebben A, Frantz B, O'Neill EA, Croft G, O'Keefe SJ, Li B, Hacker C, de Laszlo S, Smith A, Libby B, Liverton N, Hermes J, LoGrasso P. Biochemistry. 1998; 37:16573. [PubMed: 9843424]
- 65. Tembe BL, McCammon JA. Comput. Chem. 1984; 8:281.
- 66. Jorgensen WL, Ravimohan C. J. Chem. Phys. 1985; 83:3050.
- Lybrand TP, McCammon JA, Wipff G. Proc. Natl. Acad. Sci. U. S. A. 1986; 83:833. [PubMed: 3456569]
- 68. Wong CF, McCammon JA. J. Am. Chem. Soc. 1986; 108:3830.
- 69. Bash PA, Singh UC, Brown FK, Langridge R, Kollman PA. Science. 1987; 235:574. [PubMed: 3810157]
- 70. McCammon JA. Science. 1987; 238:486. [PubMed: 3310236]
- 71. Mann G, Hermans J. J. Mol. Biol. 2000; 302:979. [PubMed: 10993736]
- 72. Boresch S, Tettinger F, Leitgeb M, Karplus M. J. Phys. Chem. B. 2003; 107:9535.
- 73. Deng YQ, Roux B. J. Chem. Theory Comput. 2006; 2:1255.
- 74. Mobley DL, Chodera JD, Dill KA. J. Chem. Phys. 2006; 125:084902. [PubMed: 16965052]
- 75. Mobley DL, Graves AP, Chodera JD, McReynolds AC, Shoichet BK, Dill KA. J. Mol. Biol. 2007; 371:1118–1134. [PubMed: 17599350]
- 76. Mobley DL, Chodera JD, Dill KA. J. Chem Theory Comput. 2007; 3:1231. [PubMed: 18843379]
- 77. Boyce SE, Mobley DL, Rocklin GJ, Graves AP, Dill KA, Shoichet BK. J. Mol. Biol. 2009; 394:747. [PubMed: 19782087]
- 78. Jiang W, Roux B. J. Chem. Theory Comput. 2010; 6:2559–2565. [PubMed: 21857813]
- 79. Fujitani H, Tanida Y, Ito M, Jayachandran G, Snow CD, Shirts MR, Sorin EJ, Pande VS. J. Chem. Phys. 2005; 123:5.
- 80. Wang JY, Deng YQ, Roux B. Biophys. J. 2006; 91:2798. [PubMed: 16844742]
- 81. Jayachandran G, Shirts MR, Park S, Pande VS. J. Chem. Phys. 2006; 125:12.
- 82. Deng YQ, Roux B. J. Chem. Phys. 2008; 128:8.
- 83. Khavrutskii IV, Wallqvist A. J. Chem. Theory Comput. 2011; 7:3001. [PubMed: 22046108]
- 84. Reddy MR, Erion MD. J. Am. Chem. Soc. 2001; 123:6246. [PubMed: 11427047]
- 85. Erion MD, Dang Q, Reddy MR, Kasibhatla SR, Huang J, Lipscomb WN, van Poelje PD. J. Am. Chem. Soc. 2007; 129:15480. [PubMed: 18041833]
- 86. Reddy MR, Erion MD. J. Am. Chem. Soc. 2007; 129:9296. [PubMed: 17616196]
- 87. Zeevaart JG, Wang LG, Thakur VV, Leung CS, Tirado-Rives J, Bailey CM, Domaoal RA, Anderson KS, Jorgensen WL. J. Am. Chem. Soc. 2008; 130:9492. [PubMed: 18588301]

88. Jorgensen WL, Bollini M, Thakur VV, Domaoal RA, Spasov KA, Anderson KS. J. Am. Chem. Soc. 2011; 133:15686. [PubMed: 21853995]

- 89. Genheden S, Nilsson I, Ryde U. J. Chem. Inf. Model. 2011; 51:947. [PubMed: 21417269]
- 90. Zhou RH, Das P, Royyuru AK. J. Phys. Chem. B. 2008; 112:15813. [PubMed: 19367871]
- 91. Park H, Jeon YH. J. Mol. Graphics Model. 2011; 29:643.
- 92. Das P, Li JY, Royyuru AK, Zhou RH. J. Comput. Chem. 2009; 30:1654. [PubMed: 19399777]
- 93. Zheng ZL, Ye MQ, Zuo ZY, Liu ZG, Tai KC, Zu GL. Biochem. J. 2006; 395:509. [PubMed: 16411898]
- 94. Pan YM, Gao DQ, Yang WC, Cho H, Zhan CG. J. Am. Chem. Soc. 2007; 129:13537. [PubMed: 17927177]
- 95. Yang WC, Pan YM, Zheng F, Cho H, Tai HH, Zhan CG. Biophys. J. 2009; 96:1931. [PubMed: 19254552]
- 96. Yang WC, Pan YM, Fang L, Gao DQ, Zheng F, Zhan CG. J. Phys. Chem. B. 2010; 114:10889. [PubMed: 20677742]
- 97. Rizzo RC, Wang DP, Tirado-Rives J, Jorgensen WL. J. Am. Chem. Soc. 2000; 122:12898.
- 98. Wang DP, Rizzo RC, Tirado-Rives J, Jorgensen WL. Bioorg. Med. Chem. Lett. 2001; 11:2799. [PubMed: 11597403]
- 99. Park H, Lee S. J. Comput. Aid. Mol. Des. 2005; 19:17.
- 100. Guo ZY, Prongay A, Tong X, Fischmann T, Bogen S, Velazquez F, Venkatraman S, Njoroge FG, Madison V. J. Chem. Theory Comput. 2006; 2:1657.
- 101. Mutyala R, Reddy RN, Sumakanth M, Reddanna P, Reddy MR. J. Comput. Chem. 2007; 28:932. [PubMed: 17253638]
- 102. Ripoll DR, Khavrutskii IV, Chaudhury S, Liu J, Kuschner RA, Wallqvist A, Reifman J. PLoS Comput. Biol. 2012; 8:e1002665. [PubMed: 22956900]
- 103. Guerois R, Nielsen JE, Serrano L. J. Mol. Biol. 2002; 320:369–387. [PubMed: 12079393]
- 104. Wickstrom L, Gallichio E, Levy RM. Proteins: Struct., Funct., Bioinf. 2011; 80:111–125.
- 105. Petrella RJ, Karplus M. J. Mol. Biol. 2001; 312:1161–1175. [PubMed: 11580256]
- 106. Wang L, Berne BJ, Friesner RA. Proc. Natl. Acad. Sci. USA. 2012; 109:1937–1942. [PubMed: 22308365]
- 107. Shaw, DE.; Dror, RO.; Salmon, JK.; Grossman, JP.; Mackenzie, KM.; Bank, JA.; Young, C.; Deneroff, MM.; Batson, B.; Bowers, KJ.; Chow, E.; Eastwood, MP.; Ierardi, DJ.; Klepeis, JL.; Kuskin, JS.; Larson, RH.; Lindorff-Larsen, K.; Maragakis, P.; Moraes, MA.; Piana, S.; Shan, Y.; Towles, B. Proceedings of the Conference on High Performance Computing, Networking, Storage and Analysis (SC09). ACM; New York: 2009.
- 108. Shaw DE, Maragakis P, Lindorff-Larsen K, Piana S, Dror RO, Eastwood MP, Bank JA, Jumper JM, Salmon JK, Shan YB, Wriggers W. Science. 2010; 330:341. [PubMed: 20947758]
- 109. Zheng L, Yang W. J. Chem. Theory Comput. 2012; 8:810.
- 110. Jiang F, Han W, Wu YD. J. Phys. Chem. B. 2010; 114:5840. [PubMed: 20392111]
- 111. Mobley DL. J. Comput. Aided Mol. Des. 2012; 26:93–95. [PubMed: 22113833]
- 112. Raval A, Piana S, Eastwood MP, Dror RO, Shaw DE. Proteins-Struct., Funct., Bioinf. 2012; 80:2071.
- 113. Gallicchio E, Lapelosa M, Levy RM. J. Chem. Theory Comput. 2010; 6:2961–2977. [PubMed: 21116484]
- 114. Pargellis C, Tong L, Churchill L, Cirillo PF, Gilmore T, Graham AG, Grob PM, Hickey ER, Moss N, Pav S. Regan, Nat. Struct. Biol. 2002; 9:268–272.
- 115. Regan J, Pargellis C, Cirillo PF, Gilmore T, Hickey ER, Peet GW, Proto A, Swinamer A, Moss A. Bioorg. Med. Chem. Lett. 2003; 13:3101–3104. [PubMed: 12941343]
- 116. Vogtherr M, Saxena K, Hoelder S, Grimme S, Betz M, Schieborr U, Pescatore B, Robin M, Delarbre L, Langer T, Wendt KU, Schwalbe H. Angew. Chem. Int. Ed. 2006; 45:993–997.
- 117. Simard JR, Getlik M, Grtter C, Pawar V, Wulfert S, Rabiller M, Rauh D. J. Am. Chem. Soc. 2009; 131:13286–13296. [PubMed: 19572644]

118. Honndorf VS, Coudevylle N, Laufer S, Becker S, Griesinger C. Angew. Chem. Int. Ed. 2008; 47:3548–3551.

- 119. Raingeaud J, Gupta S, Rogers JS, Dickens M, Han J, Ulevitch RJ, Davis RJ. J. Biol. Chem. 1995; 270:7420–7426. [PubMed: 7535770]
- 120. Nielsen G, Schwalbe H. Chem. Bio. Chem. 2011; 12:2599-2607.
- 121. Young PR, McLaughlin MM, Kumar S, Kassis S, Doyle ML, McNulty D, Gallagher TF, Fisher S, McDonnell PC, Carr SA, Huddleston MJ, Seibel G, Porter TG, Livi GP, Adams JL, Lee JC. J. Biol. Chem. 1997; 272:12116–12121. [PubMed: 9115281]
- 122. Frembgen-Kesner T, Elcock AH. J. Mol. Biol. 2006; 359:202–214. [PubMed: 16616932]
- 123. Filomia F, De Rienzo F, Menziani MC. Bioorg. Med. Chem. 2010; 18:6805–6812. [PubMed: 20724167]
- 124. Shan A, Seeliger M, Eastwood MP, Frank F, Xu H, Jensen MO, Dror RO, Kuriyan J, Shaw DE. Proc. Natl. Acad. Sci. USA. 2009; 106:139–144. [PubMed: 19109437]
- 125. Shan Y, Arkhipov A, Kim ET, Pan AC, Shaw DE. Proc. Natl. Acad. Sci. USA. 2013; 110:7270–7275. [PubMed: 23576739]
- 126. Kufareva I, Abagyan R. J. Med. Chem. 2008; 51:7921–7932. [PubMed: 19053777]
- 127. Xu M, Yu L, Wan B, Yu L, Huang Q. PLoS One. 2011; 6:e22644. [PubMed: 21818358]
- 128. Flick J, Tristram F, Wenzel W. J. Comput. Chem. 2012; 33:2504–2515. [PubMed: 22886372]
- 129. Chen W, Chang CE, Gilson MK. Biophys. J. 2004; 87:3035-3049. [PubMed: 15339804]
- Huang YH, Chen W, Potter MJ, Chang CA. Biophys. J. 2012; 103:342–351. [PubMed: 22853912]
- 131. Rao S, Sanschagrin PC, Greenwood JR, Repasky MP, Sherman W, Farid R. J. Comput. Aided Mol. Des. 2008; 22:621–627. [PubMed: 18253700]
- 132. Cheeseright TJ, holm M, Lehmann F, Luik S, Göttert M, Melville JL, Laufer S. J. Med. Chem. 2009; 52:4200–4209. [PubMed: 19489590]
- Armen RS, Chen J, Brooks CL III. J. Chem. Theory Comput. 2009; 13:2909–2923. [PubMed: 20160879]
- 134. Vinh NB, Simpson JS, Scammells PJ, Chalmers DK. J. Comput. Aided Mol. Des. 2012; 26:409–423. [PubMed: 22527960]
- 135. Bukhtiyarova M, Karpusas M, Northrop K, Namboodiri HVM, Springman EB. Biochemistry. 2007; 46:5687–5696. [PubMed: 17441692]
- 136. Yun CH, Mengwasser KE, Toms AV, Woo MS, Greulich H, Wong KK, Meyerson M, Eck MJ. Proc. Natl. Acad. Sci. U. S. A. 2008; 105:2070. [PubMed: 18227510]

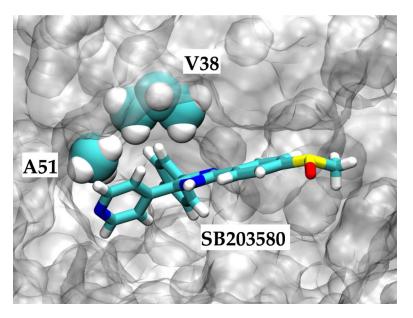


Figure 1. A close-up view of the p38 α -SB203580 binding site p38 α kinase (in gray) is shown in molecular surface representation (with a probe radius of 1.2 Å). The inhibitor SB203580 (in cyan) is shown in stick representation. The two mutation sites, Ala51 and Val38 are highlighted using space-fill representation.

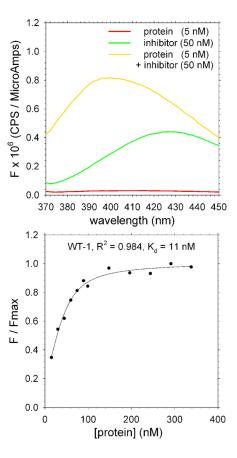


Figure 2. Fluorescence binding assay to measure the binding affinity of wild type $p38\alpha$ kinase to the inhibitor SB203580

(a) Fluorescence spectrum of the inhibitor, protein and a solution containing both the inhibitor and the protein; excitation was at 320 nm. (b) Typical titration of wild-type protein with the inhibitor; dots represent experimental data and lines represent fits to the quadratic function (see Methods).

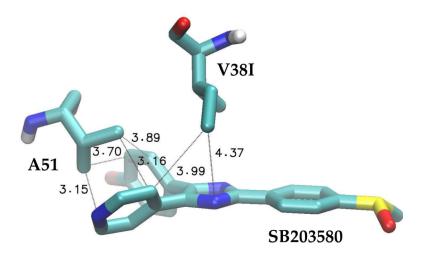


Figure 3. Putative close contacts between the mutation sites of p38a kinase and the inhibitor SB203580 $\,$

All distances are measured in Ångstroms.

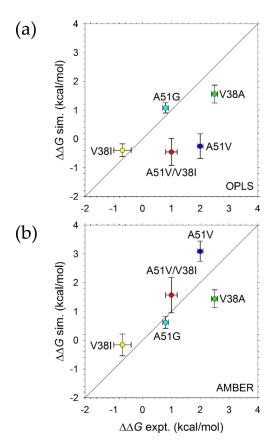


Figure 4. Correlation of the $\Delta\Delta G$ values calculated from unrestrained simulations with the experimentally measured $\Delta\Delta G$ values for p38a mutants: A51V, A51G, V38A, V38I and A51V/ V38I

The results of the simulations using the following force fields are shown: (a) OPLS-AA/L; (b) Amber ff99SB-ILDN. Each vertical error bar represents the standard deviation of 3^{11} independent $\Delta\Delta G$ estimates from three 12 ns simulations with different initial random velocities. Each horizontal error bar represents the standard deviation of $\Delta\Delta G$ from three independent fluorescence binding experiments. The diagonal line representing perfect correlation is also shown. The linear regressions obtained for OPLS and AMBER force fields are y=0.38x-0.12 ($R^2=0.26$) and y=0.75x+0.47 ($R^2=0.59$) respectively.

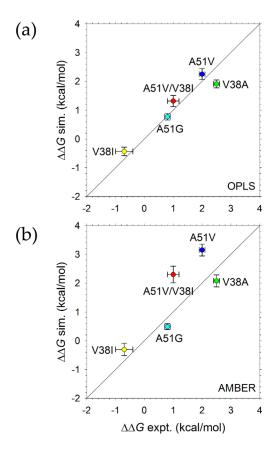


Figure 5. Correlation of the $\Delta\Delta G$ values calculated from restrained simulations with the experimentally measured $\Delta\Delta G$ values for p38a mutants: A51V, A51G, V38A, V38I and A51V/ V38I

The results of the simulations using the following force fields are shown: (a) OPLS-AA/L; (b) Amber ff99SB-ILDN. Position restraints with a force constant of 0.239 kcal/mol/Ų were applied to the backbone of the protein and heavy atoms of the drug. Each vertical error bar represents the standard deviation of 3^{11} independent $\Delta\Delta G$ estimates from three 12 ns simulations with different initial random velocities. Each horizontal error bar represents the standard deviation of $\Delta\Delta G$ from three independent fluorescence binding experiments. The diagonal line representing perfect correlation is also shown. The linear regressions obtained for OPLS and AMBER force fields are y = 0.82x + 0.25 ($R^2 = 0.92$) and y = 0.94x + 0.49 ($R^2 = 0.68$) respectively.

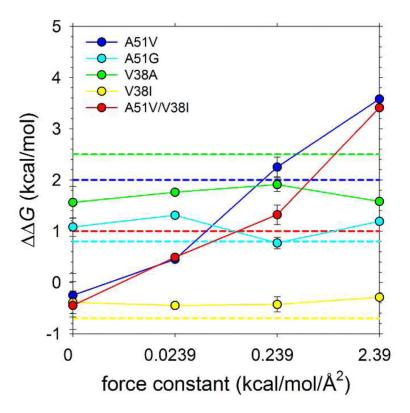


Figure 6. Force constant dependence of $\Delta\Delta G$ for p38a mutants obtained from simulations using OPLS-AA/L force field

Solid circles indicated the calculated $\Delta\Delta G$ values and dashed lines indicate experimentally measured $\Delta\Delta G$. The calculated $\Delta\Delta G$ s were obtained from three 12 ns simulations using the following four different force constants: (1) 0 kcal/mol/Å² (unrestrained); (2) 0.0239 kcal/mol/Å²; (3) 0.239 kcal/mol/Å²; (4) 2.39 kcal/mol/Å². The error bar for (1) and (3) were obtained from the standard deviation of three simulation replicates; the error bars for (2) and (4) are not calculated because only one simulation was performed.

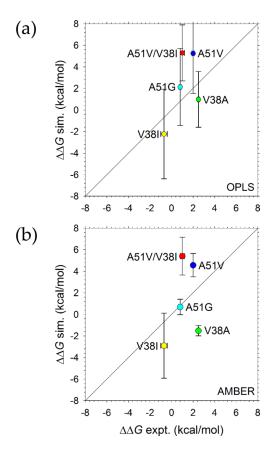


Figure 7. Correlation of the $\Delta\Delta G$ values obtained from MM/GBSA calculations for unrestrained simulations with the experimentally measured $\Delta\Delta G$ values for p38a mutants: A51V, A51G, V38A, V38I and A51V/V38I

The results of the simulations using the following force fields are shown: (a) OPLS-AA/L; (b) Amber ff99SB-ILDN. Each vertical error bar is obtained using the procedure as described in Methods section. Each horizontal error bar represents the standard deviation of $\Delta\Delta G$ from three independent fluorescence binding experiments. The diagonal line representing perfect correlation is also shown. The linear regressions obtained for OPLS and AMBER force fields are y=1.43x+0.68 ($R^2=0.31$) and y=1.04x+0.09 ($R^2=0.12$) respectively.

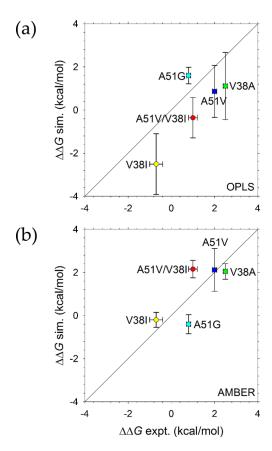


Figure 8. Correlation of the $\Delta\Delta G$ values obtained from MM/GBSA calculations for restrained simulations with the experimentally measured $\Delta\Delta G$ values for p38a mutants: A51V, A51G, V38A, V38I and A51V/V38I

The results of the simulations using the following force fields are shown: (a) OPLS-AA/L; (b) Amber ff99SB-ILDN. Position restraints with a force constant of $100 \text{ kJ.mol}^{-1}.\text{nm}^{-2}$ were applied to the backbone of the protein and heavy atoms of the drug. Each vertical error bar is obtained using the procedure as described in Methods section. Each horizontal error bar represents the standard deviation of $\Delta\Delta G$ from three independent fluorescence binding experiments. The diagonal line representing perfect correlation is also shown. The linear regressions obtained for OPLS and AMBER force fields are y = 1.05x - 1.03 ($R^2 = 0.62$) and y = 0.80x + 0.24 ($R^2 = 0.57$) respectively.

Table 1

Zhu et al.

Comparison of experimental and computational ΔG and $\Delta \Delta G$ values

| N/A | | AA-CAMBER (ncan/mor) | AAGOPLS (KCal/mol) | ΔG_{expt} (kcal/inot) $\Delta \Delta G_{\mathrm{expt}}$ (kcal/inot) $\Delta \Delta G_{\mathrm{OPLS}}$ (kcal/inot) $\Delta \Delta G_{\mathrm{OPLS}}$ (kcal/inot) $\Delta \Delta G_{\mathrm{OPLS}}$ (kcal/inot) |
|----------------|---------------------------------|----------------------|-------------------------------|--|
| | N/A | N/A | N/A | N/A |
| 2.0 ± 0.1 | -0.3 ± 0.4 | 3.1 ± 0.4 | 2.3 ± 0.2 | 3.2 ± 0.2 |
| 0.8 ± 0.1 | 0.6 ± 0.2 | 0.6 ± 0.2 | 0.8 ± 0.1 | 0.5 ± 0.1 |
| 2.5 ± 0.1 | 1.4 ± 0.3 | 1.4 ± 0.3 | 1.9 ± 0.1 | 2.1 ± 0.2 |
| -0.7 ± 0.3 | -0.4 ± 0.2 | -0.2 ± 0.4 | -0.4 ± 0.1 | -0.3 ± 0.2 |
| 1.0 ± 0.2 | -0.5 ± 0.5 | 1.6 ± 0.6 | 1.3 ± 0.2 | 2.3 ± 0.3 |
| | -0.7 ± 0.3 1.0 ± 0.2 | | -0.4 ± 0.2 -0.5 ± 0.5 | -0.4 ± 0.2 -0.2 ± 0.4 -0.5 ± 0.5 1.6 ± 0.6 |

a experimental absolute binding free energy.

Page 29

b experimental relative binding free energy.

c computational relative binding free energy obtained using OPLS-AA/L force field without position restraints.

 $[\]frac{d}{d}$ computational relative binding free energy using Amber ff99SB-ILDN force field without position restraints.

 $e^{-computational relative}$ binding free energy obtained using OPLS-AA/L force field with position restraints (0.239 kcal/mol/ 4 2).