

Review

Matrix Effect on In-Source Decay Products of Peptides in Matrix-Assisted Laser Desorption/Ionization

Daiki Asakawa,^{*,1,2} Motoshi Sakakura,¹ and Mitsuo Takayama¹

¹Mass Spectrometry Laboratory, Graduate School in Nanobioscience, Yokohama City University, Yokohama, Japan

²General and Physical Chemistry Department, Mass Spectrometry Laboratory, University of Liège, Liège, Belgium

MALDI-ISD of peptides were studied using several salicylic acid derivatives, 2,5-dihydroxybenzoic acid (2,5-DHB), 5-aminosalicylic acid (5-ASA), 5-formylsalicylic acid (5-FSA), and 5-nitrosalicylic acid (5-NSA) as matrices. The difference in the nature of the functional group at the 5-position in the salicylic acid derivatives can dramatically affect the ISD products. The use of 2,5-DHB and 5-ASA leads to “hydrogen-abundant” peptide radicals and subsequent radical-induced N–C_α bond cleavage. N–C_α bond cleavage gave a c'/z' fragment pair and radical z'-series fragments gain a hydrogen radical or react with a matrix radical. In contrast, the use of 5-NSA resulted in the production of a “hydrogen-deficient” peptide radical that contained a radical site on the amide nitrogen in the peptide backbone. Subsequently, the radical site on the amide nitrogen induces C_α–C bond dissociation, leading to a'/x fragment pair. The a'-series ions undergo further hydrogen abstraction to form a-series ions after C_α–C bond cleavage. Since the Pro residue does not contain a nitrogen-centered radical site, C_α–C bond cleavage does not occur. Alternatively, the specific cleavage of CO–N bonds leads to a b'/y fragment pair at Xxx–Pro which occurs *via* hydrogen abstraction from the C_α–H in the Pro residue. The use of 5-FSA generated both a'/x- and c'/z'-series fragment pairs. An oxidizing matrix provides useful complementary information in MALDI-ISD compared to a reducing matrix for the analysis of amino acid sequencing and site localization in cases of phosphopeptides. MALDI-ISD, when used in conjunction with both reducing and oxidizing matrices is a potentially useful method for *de novo* peptide sequencing.

Keywords: MALDI in-source decay, hydrogen radical transfer, c'/z' and a'/x fragment pair, *de novo* peptide sequencing, site localization of post translational modifications

(Received February 2, 2012; Accepted May 7, 2012)

INTRODUCTION

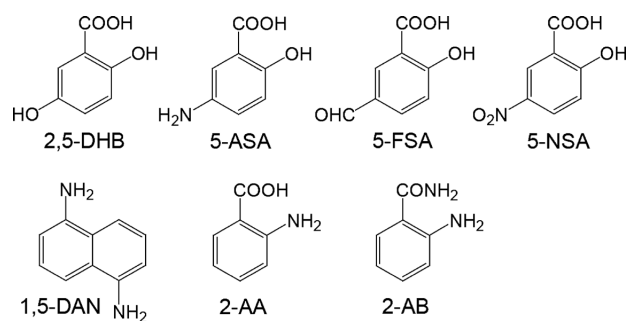
Mass spectrometry (MS) represents a powerful analytical tool that can be applied in a wide variety of scientific fields because of its high sensitivity and ease of use. Of the soft ionization methods, matrix-assisted laser desorption/ionization (MALDI)^{1–3} and electrospray ionization (ESI)^{4,5} are recognized as indispensable analytical methods for identifying peptides and proteins. It is noteworthy that peptide-mass fingerprinting (PMF) with MALDI-MS is now commonly used in the characterization of proteins,^{6,7} because MALDI occurs without abundant fragmentation.

MALDI often can cause fragmentation during the desorption/ionization process, and fragmentation is observed as either in-source decay (ISD)⁸ or post-source decay (PSD).^{9,10} Both ISD and PSD have been used for the amino acid sequencing of peptides.^{11,12} ISD is a type of fragmentation that occurs rapidly in the MALDI source, after the laser shot and before ion extraction, while PSD involves the fragmentation of metastable ions occurring in the field-free drift path in a time-of-flight mass spectrometer. PSD fragment ions can be revealed by reflectron time-of-flight mass spectrometer. The mechanisms for ISD and PSD are different from each other. In the case of PSD, a-, b- and y-series ions are mainly observed and can be explained by vibrational activation

processes.¹² During the desorption/ionization process, the excess energy deposited on peptide ions is converted into vibrational energy that is distributed over the entire molecule, leading fragmentations. Because of this, fragmentation efficiency is decreased for large peptides and proteins due to less energy received per degree of freedom. In addition, PSD of phosphopeptides results in a loss of phosphate groups (80 and/or 98 Da) from protonated molecules.¹³ The phosphate group in phosphopeptides is relatively labile to low-energy cleavage that competes with backbone fragmentation. The dominant loss of phosphoric acid(s) can be used as a specific marker for the identification of a phosphopeptide, while it is unfavorable for determining the location of phosphorylated modification sites.

In contrast to PSD, specific N–C_α bond cleavage at the peptide backbone is associated with the MALDI-ISD process.⁸ Backbone cleavage by MALDI-ISD methods is mechanistically similar to that by electrospray ionization-based radical-induced fragmentation methods, *i.e.*, electron-capture dissociation (ECD)¹⁴ and electron-transfer dissociation (ETD).¹⁵ Both ECD and ETD involve the association of electrons with multiply protonated analytes. Protons bound to the analyte are converted into hydrogen radicals *via* electron attachment, and the resulting hydrogen radicals are then transferred to backbone carbonyl oxygens with subsequent radical-induced cleavage at N–C_α bonds. In contrast to ECD and ETD, MALDI-ISD is mediated by the attachment of a hydrogen radical to a peptide.^{16,17} Such fragmentation methods lead to the formation of “hydrogen-abundant” peptide radicals and subsequent radical-induced N–C_α bond

* Correspondence to: Daiki Asakawa, Mass Spectrometry Laboratory, Graduate School in Nanobioscience, Yokohama City University, 22-2 Seto, Kanazawa-ku, Yokohama 236-0027, Japan, e-mail: dasakawa@yokohama-cu.ac.jp



Scheme 1. Matrix materials for MALDI-ISD.

cleavage.

Two major strategies, *i.e.*, “bottom-up” and “top-down” approaches rely the analysis of proteins by mass spectrometry. In the “bottom-up” approach, analyte proteins are subjected to enzymatic digestion. The resulting digested peptides are then analyzed by MALDI, followed by MS/MS with collision induced dissociation (CID) or MALDI-PSD for peptide sequencing. CID and MALDI-PSD are commonly available fragmentation methods that are applicable to closed shell ions. In contrast, “top-down” approach is used for the direct fragmentation of the intact protein in the mass spectrometer without any enzyme digestion. CID and MALDI-PSD are not applicable for a top-down approach, due to the low fragmentation efficiency of large proteins. Alternative fragmentation methods such as MALDI-ISD, ECD and ETD leading to the fragmentation of radical ions, can be used for the “top-down” approach for characterizing proteins. The advantages in the “top-down” approach are high throughput, straightforward methodology and the need for a small amount of sample.^{18,19} Additionally, they could be useful methods for *de novo* protein sequencing, including post-translational modifications.^{18,19}

In this review, we focus on the mechanisms of peptide backbone cleavage in MALDI-ISD. The usefulness of MALDI-ISD for the sequencing of phosphopeptides is also discussed. It was recently reported that the choice of matrix for MALDI-ISD can dramatically affect the observed ISD fragment ions and the quality of the mass spectrum. Matrices such as 2,5-dihydroxybenzoic acid (2,5-DHB),²⁰ 5-aminosalicylic acid (5-ASA),²¹ 5-formyl salicylic acid (5-FSA),²² 5-nitrosalicylic acid (5-NSA),²² 1,5-diaminonaphthalene (1,5-DAN),^{23,24} 2-aminobenzoic acid (2-AA),²⁵ and 2-aminobenzamide (2-AB)²⁵ were previously reported to be useful in efficiently inducing MALDI-ISD (the structures of some common matrices are shown in Scheme 1).

EXPERIMENTAL

Analyte peptides were dissolved in water at a concentration of 20 pmol/ μ L. The matrices, 2,5-DHB, 1,5-DAN, and 5-NSA were dissolved in water–acetonitrile (1:1, v/v) with 0.1% TFA at concentration of 10 mg/ μ L. A volume of 0.5 μ L of analyte peptide solution was deposited onto a MALDI target, and 0.5 μ L of matrix solution was then added to the target. After extensive mixing, the mixture was allowed to dry in air at room temperature. 5-FSA was dissolved in acetone at a concentration of 10 mg/mL. A volume of 0.5 μ L of analyte solution was deposited onto a stainless-steel MALDI target and left to dry. After complete evaporation of the

solvent, 0.5 μ L of a 5-FSA solution in acetone was deposited on the dried peptides.

MALDI-ISD mass spectra were recorded using a MALDI time-of-flight (TOF) mass spectrometer, AXIMA-CFR (Shimadzu, Kyoto, Japan) equipped with a nitrogen laser (337 nm wavelength, 4 ns pulse width, 10 Hz pulse rate). Ions generated by MALDI were accelerated through a 20 kV potential with delayed extraction. The analyzer operated in the reflectron mode for standard peptides and in the linear mode for β -casein tryptic peptides. A total of 500 shots were accumulated for each mass spectrum.

Notation

We employed herein the unambiguous notation of Zubarev in the naming of fragment ions.²⁶ According to this notation, homolytic C _{α} –C bond cleavage yields the radical fragments a' and x' , and loss of a hydrogen radical from an a' or x' fragment produces an a or x fragment, respectively. The product of a hydrogen radical transfer to an a' or x' fragment is denoted as a' and x' , respectively. Thus, a and x fragments are 1.0078 Da smaller than a' and x' fragments, respectively, and a' and x' fragments are 1.0078 Da larger than a and x fragments. Unless otherwise noted, all assigned peaks represent singly protonated molecules $[M+H]^+$ in positive-ion mode and singly deprotonated molecules $[M-H]^-$ in negative-ion mode.

Matrix effect on the ISD fragment ions

The choice of matrix is essential for ISD. The hydroxybenzoic acid derivatives seem to be one of the best matrix candidate for the formation of c' - and z' -series ions in MALDI-ISD, and 2,5-DHB was far superior to other hydroxybenzoic acid derivatives.²⁰ The 5-hydroxyl group of 2,5-DHB plays an important role in the formation of MALDI-ISD fragment ions.^{16,20} Our focus was on the nature of the functional group at the 5-position in salicylic acid derivatives of the MALDI matrix.²² Figure 1 shows a comparison of positive-ion MALDI mass spectra of ACTH18–35 (RPVKVYPNGAEDESAEAF) obtained with four different matrices 2,5-DHB, 5-ASA, 5-FSA, and 5-NSA. The use of 2,5-DHB and 5-ASA generated c' -series ions accompanied by a -series ions with a weak intensity (Figs. 1a and 1b). By contrast, the use of 5-FSA generated both a - and c' -series ions, and the abundance of c' -series ions was less than that of the a -series ions (Fig. 1c). 5-NSA generated a -series ions with strong signal intensities and did not generate any c' -series ions (Fig. 1d). These findings suggest that the difference in the nature of functional groups at the 5-position in salicylic acid derivatives of the MALDI matrix can dramatically affect the ISD products that are produced.

Hydroxyl groups and amino groups have hydrogen-donating characteristics, while a nitro group has hydrogen-accepting characteristics. The formyl group has both hydrogen-donating and hydrogen-accepting characteristics. This suggests that both hydrogen-donating and -accepting properties are important factors for matrix, in terms of the production of MALDI-ISD fragments. Hydrogen-donating ability can be estimated by measuring the ability to reduce disulfide bonds^{21,23,24} (Scheme 2a). Figure 2 shows positive-ion MALDI mass spectra of $[\text{Arg}^8]$ -vasopressin (CYFQNCPRG-NH₂), which contains a disulfide bond between Cys1 and Cys6, using four different matrices, namely,

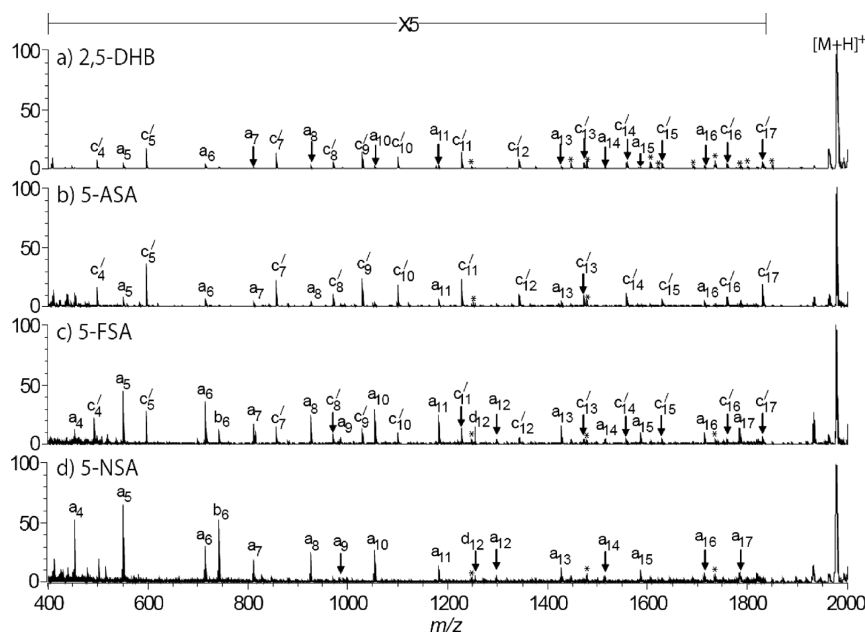


Fig. 1. MALDI mass spectra of ACTH18–35 obtained with (a) 2,5-DHB, (b) 5-ASA, (c) 5-FSA, and (d) 5-NSA. The asterisk indicates PSD signal. [Reproduced from ref. 22 with Copyright permission of Springer.]

2,5-DHB, 5-ASA, 5-FSA, and 5-NSA. The use of 2,5-DHB or 5-ASA gave a high yield of the reduced ion $[M+2H+H]^+$, while 5-FSA or 5-NSA gave moderate or low yields of this species. The presence of a 5-hydroxyl group in 2,5-DHB and a 5-amino group in 5-ASA appears to be advantageous for the intermolecular hydrogen transfer from the matrix to the analyte peptide. The order of hydrogen-donating ability was $5\text{-ASA} > 2,5\text{-DHB} > 5\text{-FSA} > 5\text{-NSA}$, suggesting the hydrogen-donating property of the matrix is an important factor in terms of the formation of c' - and z' -series ions. The outstanding characteristics of the MALDI-MSD spectrum obtained with 5-ASA compared with other matrices were the high quality separation of isotope peaks of $[M+H]^+$ and ISD ions, as shown in Fig. 2. The peak broadening originates from the initial velocity dispersion of analyte and ISD ions. The sharpness in the ion peaks in the MALDI-MSD spectra with 5-ASA was maintained, even at higher laser fluence which are suitable for the appearance of ISD ion peaks. Therefore, using 5-ASA, the ISD ions could be clearly assigned, due to the decreased interference peak and the sharpness of the ISD ion peaks in the MALDI-MSD spectrum.²¹⁾

In contrast, as shown in Fig. 2, the use of 5-FSA and 5-NSA gave dehydrogenated or oxidized $[\text{Arg}^8]\text{-vasopressin}$ $[M-H+H]^+$. The oxidized product $[M-H+H]^+$ was also observed in the MALDI mass spectra of other peptides. The oxidized product $[M-2H+H]^+$ was formed by hydrogen transfer from peptide molecules to the 5-formyl group in 5-FSA and the 5-nitro group in 5-NSA. The abstraction of hydrogen from peptides to the matrix results in the formation of oxidized peptides bearing a radical site on the amide nitrogen and subsequent radical-induced cleavage at $C_\alpha\text{-H}$ bonds, leading to the formation of $[M-H+H]^+$ (Scheme 2b). The order of the ascertained hydrogen-accepting ability was $5\text{-NSA} > 5\text{-FSA} > 2,5\text{-DHB} \approx 5\text{-ASA} \approx 0$. These findings, therefore, suggest that 5-FSA and 5-NSA form a - and x -series

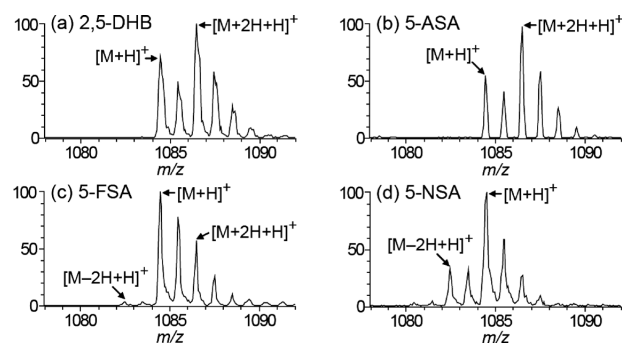
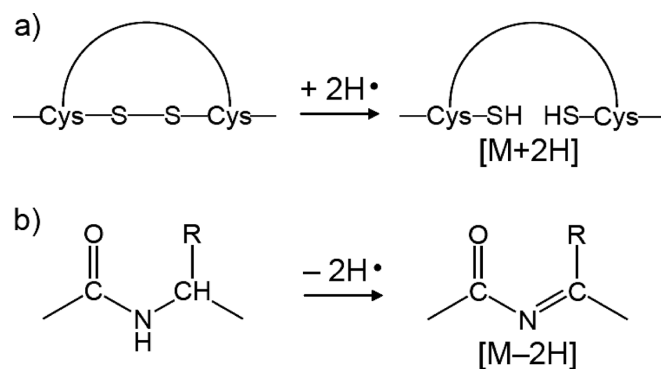
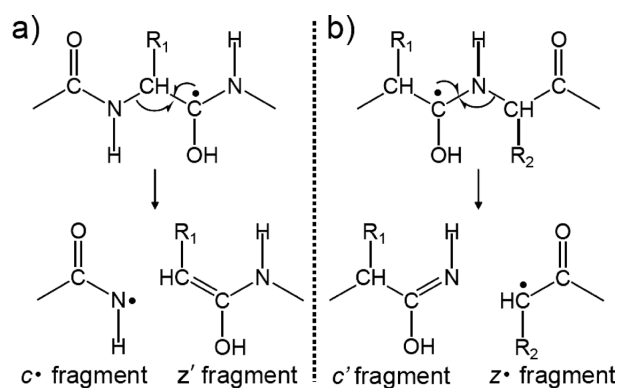


Fig. 2. Partial MALDI mass spectra of $[\text{Arg}^8]\text{-vasopressin}$ obtained with (a) 2,5-DHB, (b) 5-ASA, (c) 5-FSA, and (d) 5-NSA. [Reproduced from ref. 22 with Copyright permission of Springer.]

ions *via* hydrogen abstraction.

MALDI-MSD *via* hydrogen attachment

In MALDI-MSD with 2,5-DHB and 5-ASA, N-C_α bonds are preferentially cleaved, leading to the formation of c' - and z' -series ions *via* the attachment of hydrogen radicals to carbonyl oxygens on the peptide backbone. The source of the hydrogen radical involved in the MALDI-MSD was investigated by using peptides and matrix that were both labeled with deuterium.^{16,17)} The results suggest that intermolecular hydrogen abstraction from the matrix to analyte peptides occurred, resulting in the formation of c' -series ions. Consequently, MALDI-MSD is initiated by the transfer of a hydrogen radical from an excited matrix molecule to the carbonyl group on the peptide backbone, leading to a "hydrogen-abundant" peptide.^{16,17)} The N-C_α bond of the peptide backbone is subsequently cleaved. Upon forming this "hydrogen-abundant" peptide, two N-C_α bond cleavage pathways giving either c'/z' or c'/z' fragment pairs are theoretically possible (Scheme 3). However, all ISD fragment ions

Scheme 2. Mechanism for the formation of (a) $[\text{M}+2\text{H}]$ and (b) $[\text{M}-2\text{H}]$.Scheme 3. The formation of (a) c'/z' and (b) c'/z' fragment pairs originated from the cleavage of N-C_α bonds.

contain even-numbered electron, *i.e.*, c' - and z' -series ions. Therefore, it is important to determine the mechanism of cleavage that gives rise to the c' - and z' -series ions.

To address the most probable pathway for the formation of c' - and z' -series ions originating from the cleavage of N-C_α bond, MALDI-MSD experiments were conducted using ACTH18-35 (RPVKVYPNGAEDESAEAF) and $[\text{Arg}^{18}]$ -ACTH19-36 (PVKVYPNGAEDESAEAFR). The only difference between these peptides is the position of the Arg residue. It has been previously reported that peptides containing an Arg residue near the N-terminus preferentially gave c' -series ions, while the presence of an Arg residue at the C-terminal favored the formation of z' -series ions.^{27,28} Figure 3 shows the comparison of positive-ion MALDI-MSD spectra of ACTH18-35 and $[\text{Arg}^{18}]$ -ACTH19-36 obtained with 5-ASA. The MALDI-MSD of ACTH18-35 generates exclusively c' -series ions accompanied by a -series ions with weak intensity. In contrast, MALDI-MSD spectrum of $[\text{Arg}^{18}]$ -ACTH19-36 showed z' - and y' -series ions. Interestingly, the $[z+5\text{-ASA}]$ -series ions were generated by the recombination of z' -series ions with $[\text{5-ASA-H}]^+$, whereas the matrix adducts bound to c -series ions were not observed in the MALDI-MSD spectrum of ACTH18-35. It has also been reported that the presence of matrix adducts bound to z -series ions can be seen in the MALDI-MSD spectra when 2,5-DHB¹⁷ or 1,5-diaminonaphthalene (1,5-DAN)²⁸ was used as a matrix. Therefore, we speculate that, in the MALDI-MSD process, a c'/z' fragment pair is formed and subsequently the z' -series ions gain a hydrogen radical or react with a matrix radical (Scheme 4). The w -series ions were also observed in the MALDI-MSD spectrum of $[\text{Arg}^{18}]$ -ACTH19-36. The

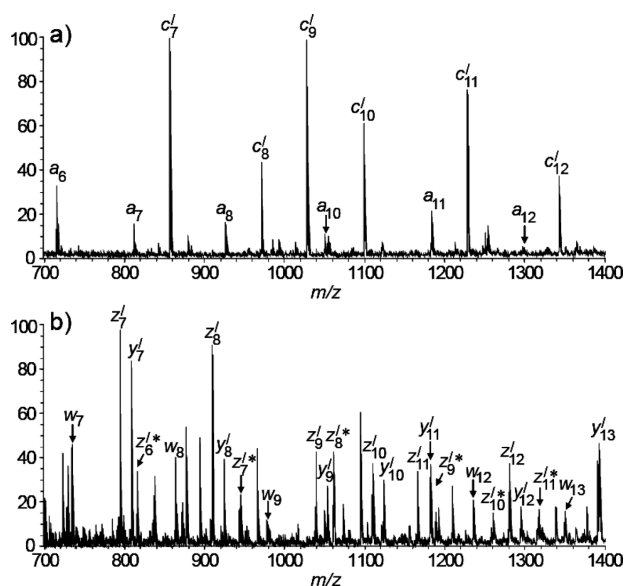
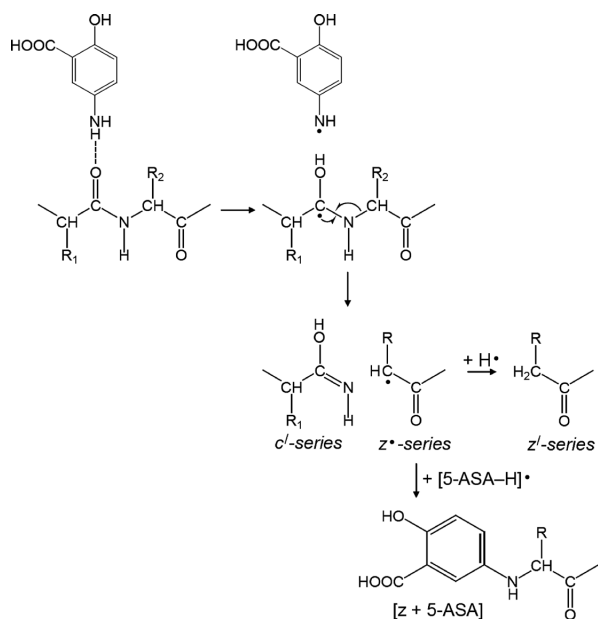
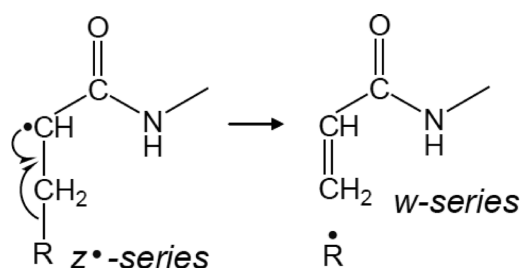
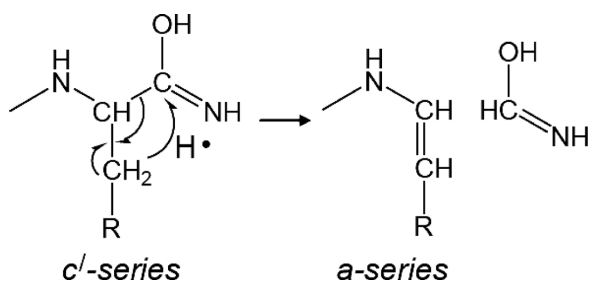


Fig. 3. Positive-ion MALDI-MSD mass spectra of (a) ACTH18-35 and (b) $[\text{Arg}^{18}]$ -ACTH19-36 with 5-ASA. The asterisk for the z -series ions denotes the matrix adduct of the z -series ions. [Modified from ref. 22 with Copyright permission of Springer.]

w -series ions are formed by the $\text{C}_\beta\text{-C}_\gamma$ bond cleavage at the side chain of the z' -series ions (Scheme 5). The w_{10} and w_{11} of $[\text{Arg}^{18}]$ -ACTH19-36 corresponding to Ala-Glu and Gly-Ala bond cleavage, respectively, were absent. Since Gly and Ala residues contain no $\text{C}_\beta\text{-C}_\gamma$ bonds, the formation of w fragments is impossible. The z' -series fragments can undergo radical reactions and subsequent degradation, because the reactivity of z' -series radical fragments is higher than that of c' -series fragments.¹⁷

In contrast, the MALDI-MSD spectrum of ACTH18-35 showed a -series ions, except for the a_9 ion corresponding to Gly-Ala bond cleavage. The c' -series ions may induce the dissociation of $\text{C}_\alpha\text{-N}$ bonds, leading to the production of a -series ions with hydrogen transfer from the β -carbon²¹ (Scheme 6). The a_9 ion was not observed in the cleavage at the Gly-Ala bond due to lack of a β -hydrogen in the Gly residue. The lack of an a ion at the C-terminal side of Gly residue can be seen in previously reported MALDI-MSD mass spectra.^{20,21}

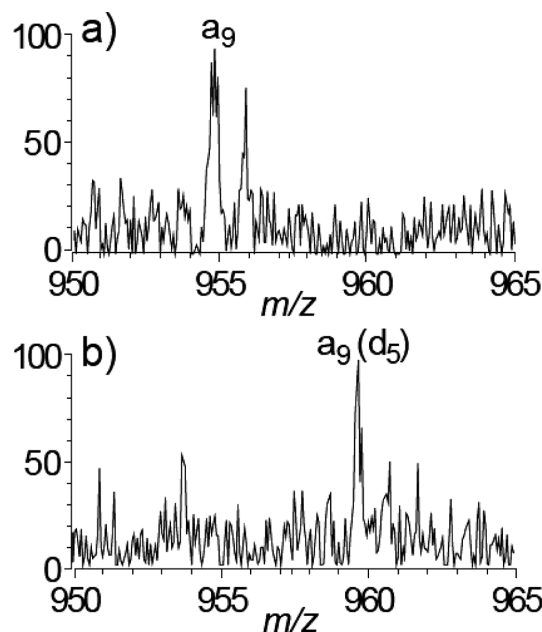
The c'_6 of ACTH18-35 and z'_{13} of $[\text{Arg}^{18}]$ -ACTH19-36 corresponding to the Tyr-Pro bond cleavage were absent.

Scheme 4. The mechanism of MALDI-ISD *via* hydrogen attachment.Scheme 5. Mechanism of formation of *w*-series fragments from *z'*-series fragments.Scheme 6. The formation mechanism of *a*-series fragments from *c'*-series fragments with hydrogen transfer from the β -carbon.

Since Pro has a cyclic structure, the formation of the *c'* and *z'* fragments originated from the cleavage of Xxx-Pro is impossible.

MALDI-ISD *via* hydrogen abstraction

In the case of MALDI-ISD with 5-NSA and 5-FSA, C_{α} -C bonds are preferentially cleaved, leading to the formation of *a*- and *x*-series ions *via* the abstraction of a hydrogen radical from the peptide backbone to the matrix.²²⁾ These MALDI-ISD techniques share some similarities with negative-ion electrospray based fragmentation techniques, such as electron detachment dissociation (EDD)²⁹⁾ and negative electron

Fig. 4. Partial MALDI-ISD spectra of (a) synthetic peptide RLGNQWAVGDLAE and (b) deuterium-labeled peptide RLGNQWA(d_3)VG(d_2)DLAE with 5-NSA. [Reproduced from ref. 22 with Copyright permission of Springer.]

transfer dissociation (NETD).³⁰⁾ Both EDD and NETD involve electron detachment from multiply deprotonated analytes $[M-nH]^{n-}$, forming a charge-reduced peptide anion $[M-nH]^{(n-1)-}$ that contains a radical site on the carboxyl group of the side chain or at a C-terminal carboxyl group. Subsequently, a nitrogen-centered radical product is formed *via* hydrogen transfer from the backbone amide nitrogen to the radical site on the carboxyl group. The radical site on the amide nitrogen induces dissociation of the C_{α} -C bond.

To ascertain the most probable pathway for the formation of *a*-series ions in MALDI-ISD with 5-NSA, a synthetic peptide (RLGNQWAVGDLAE) and a deuterium-labeled peptide (RLGNQWA(d_3)VG(d_2)DLAE) were used with 5-NSA as the matrix.²²⁾ The deuterium labeled peptide contains Ala7 ($C_{\beta}D_3$) and Gly9 ($C_{\alpha}D_2$). As shown in the a_9 ion region of the mass spectra, a mass shift of 5 Da was observed in these a_9 products (Fig. 4). The mass shift is consistent with the number of deuterium labels in the Ala7 ($C_{\beta}D_3$) and Gly9 ($C_{\alpha}D_2$). No evidence of the formation of a a_9-d_4 as the result of the abstraction of a deuterium from the α -carbon ($C_{\alpha}D_2$) at Gly9 was observed. This indicates that the *a*-series ions in the MALDI-ISD spectra with 5-NSA are formed *via* the abstraction of the amide hydrogen on the peptide backbone, as shown in Scheme 7. The mechanism of C_{α} -C bond cleavage of peptide backbone in MALDI-ISD is described below.

To ascertain the most probable pathway for the formation of *a*- and *x*-series ions that originate from the C_{α} -C bond cleavage, a Pro-rich sequence peptide bradykinin potentiator B (Pyr-GLPPRPKIPP) was used in MALDI-ISD experiments.³¹⁾ The positive-ion MALDI-ISD spectrum of bradykinin potentiator B with 5-NSA included both *a*- and *x*-series ions (Fig. 5). However, the x_7 and x_8 ions derived from the cleavage of the C_{α} -C bonds at Xxx-Pro were absent, because Pro residue has no nitrogen-centered radical site. C_{α} -C bond cleavage at Xxx-Pro and Pro-Xxx bonds would lead to a/x' and a'/x fragment pairs, respectively (Scheme 8). The

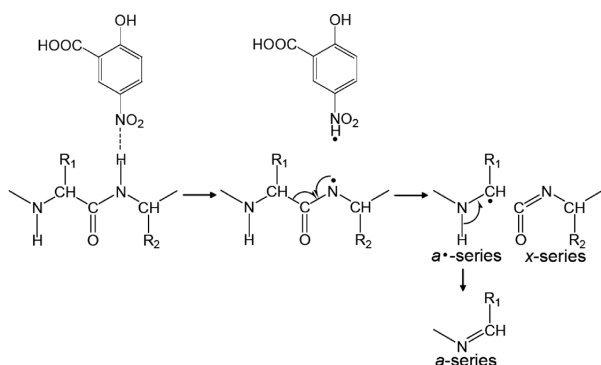
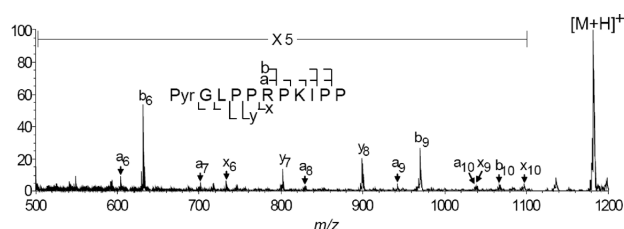
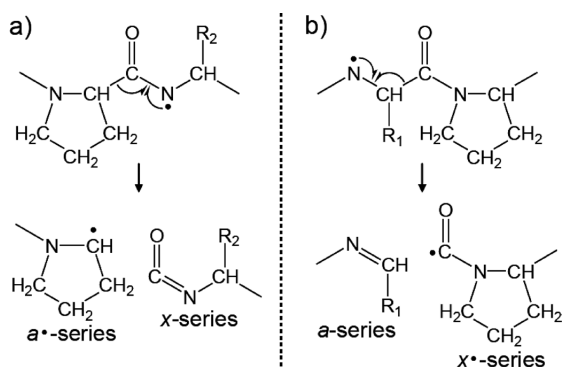
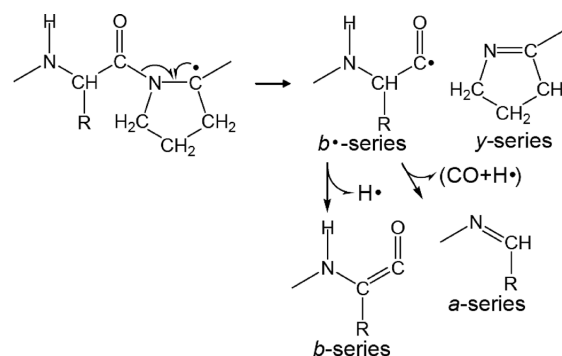
Scheme 7. The mechanism of MALDI-ISD *via* hydrogen abstraction.

Fig. 5. Positive-ion MALDI-ISD mass spectrum of bradykinin potentiator B with 5-NSA. [Reproduced from ref. 31 with Copyright permission of John Wiley and Sons.]

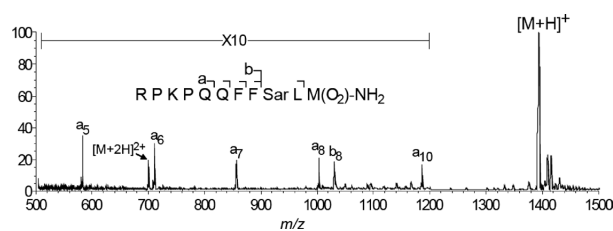
Scheme 8. The formation of (a) a'/x and (b) a/x' fragment pairs originated from the cleavage of $C_{\alpha}-C$ bonds at Pro residue.

absence of x -series ions from the cleavage of $C_{\alpha}-C$ bonds at Xxx-Pro in Fig. 5 indicates that fragmentation leading to an a/x' fragment pair does not occur (Scheme 8b). An *ab initio* calculation showed that the energy barrier for the formation of an a'/x fragment pair is lower and that formation of an a'/x fragment pair is more favorable than that of an a/x' fragment pair.^{32,33} Therefore, the proposed $C_{\alpha}-C$ bond cleavage mechanism (Scheme 7) is supported by the *ab initio* calculation. However, radical fragment a' -series ions were not observed in MALDI-ISD spectra when 5-NSA was used, and instead a -series ions were detected. It is likely that the amounts of exited 5-NSA molecules and 5-NSA radicals in the MALDI plume are sufficient to form a -series ions *via* the further hydrogen abstraction after the $C_{\alpha}-C$ bond cleavage.

The MALDI-ISD of bradykinin potentiator B did not produce x -series ions originating from the cleavage of Xxx-Pro bonds but, instead, y -series ions were produced (Fig. 5). Additionally, b -series ions originated from the cleavage at Xxx-Pro bonds were observed as well as a -series ions. The

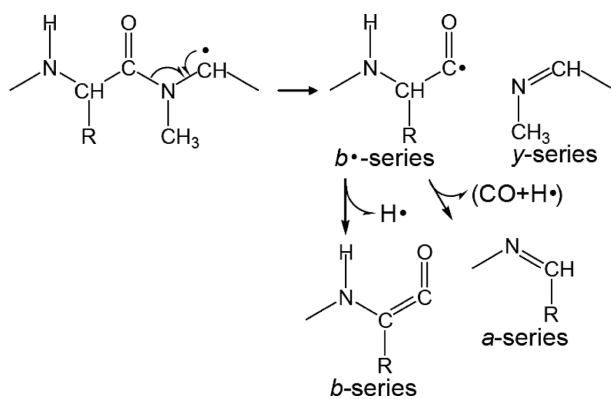


Scheme 9. Peptide bond cleavage at Xxx-Pro.

Fig. 6. Positive-ion MALDI mass spectrum of $[\text{Sar}^9, \text{Met}(\text{O}_2)^{11}]$ -substance P with 5-NSA. [Reproduced from ref. 31 with Copyright permission of John Wiley and Sons.]

cleavage of the CO-N bond at Xxx-Pro to form b - and y -series ions may occur with hydrogen abstraction from the $C_{\alpha}-\text{H}$ bond at a Pro residue (Scheme 9). In contrast, x -series ions arising from the cleavage at Xxx-Pro in the MALDI-ISD spectrum of bradykinin potentiator B were absent, whereas their counterpart a -series ions were observed (Fig. 5). These a -series ions may be derived from the cleavage of Xxx-Pro peptide bonds (Scheme 9).

To ascertain the most probable pathway for the formation of b - and y -series ions originating from the cleavage of Xxx-Pro bonds, $[\text{Sar}^9, \text{Met}(\text{O}_2)^{11}]$ -substance P was used for the MALDI-ISD with 5-NSA as the matrix.³¹ The Sar residue (*N*-methyl glycine residue) does not contain a hydrogen radical at the amide portion on the peptide backbone. The MALDI mass spectrum of $[\text{Sar}^9, \text{Met}(\text{O}_2)^{11}]$ -substance P also showed the preferential production of a -series ions, and a b_8 ion was observed as well (Fig. 6). This indicates that the b_8 ion of $[\text{Sar}^9, \text{Met}(\text{O}_2)^{11}]$ -substance P is formed *via* the abstraction of a hydrogen from the $C_{\alpha}-\text{H}$ in the Sar residue. The abstraction of hydrogen from the $C_{\alpha}-\text{H}$ in Pro and Sar residues results in the formation of a carbon-centered radical with subsequent radical-induced cleavage at CO-N bond, leading to the formation of b - and y -series ions (Scheme 10). Although a nitrogen-centered radical at Sar residue is not formed, the a_8 ion of $[\text{Sar}^9, \text{Met}(\text{O}_2)^{11}]$ -substance P is still observed. No evidence was found for the production of x -series ions originating from a cleavage at Xxx-Pro in the MALDI-ISD spectrum of bradykinin potentiator B, whereas the counterpart a -series ions were observed (Fig. 5). These results suggest that the a -series ions originating from the cleavage of Xxx-Pro and Xxx-Sar are formed by the further degradation of b' -series ions (Schemes 9 and 10), indicating the competitive formation of b - and a -series ions from b' -series ions during MALDI-ISD *via* hydrogen abstraction.



Scheme 10. Peptide bond cleavage at Xxx-Sar.

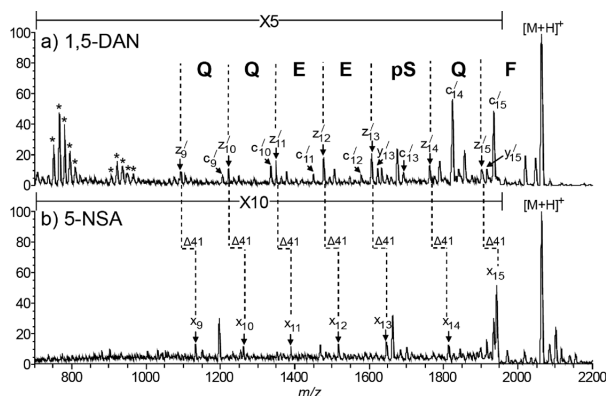


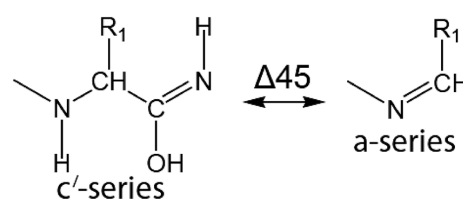
Fig. 7. Positive-ion MALDI-ISC spectra of mono-phosphopeptide (FQpSEEQQTEDELQDK) obtained with (a) 1,5-DAN and (b) 5-NSA. Asterisk indicates matrix peaks. [Reproduced from ref. 34 with Copyright permission of John Wiley and Sons.]

MALDI-ISC of phosphorylated peptides

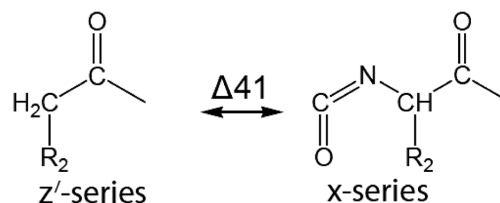
The utility of MALDI-ISC for the sequencing of phosphopeptides is discussed in this section.³⁴ We chose bovine β -casein tryptic peptides as model phosphopeptides, because they have frequently been used for evaluating the performance of MS instruments. In this experiment, the β -casein was digested with trypsin and the resulting phosphorylated tryptic peptides were isolated from the digest by TiO₂-based enrichment according to previously published protocol.³⁵ The tryptic digestion of β -casein produces two phosphopeptides, the 1–25 peptide containing four phosphate groups at Ser15, Ser17, Ser18, and Ser19 (RELEELN-VPGEIVepSLpSpSpSEESITR, monoisotopic mass=3121.3) and the 33–48 peptide containing one phosphate group at Ser35 (FQpSEEQQTEDELQDK, monoisotopic mass=2060.8). Subsequently, these phosphopeptides were isolated prior to MALDI-ISC analysis.

To determine the site of phosphorylation, we analyzed the phosphopeptides by MALDI-ISC using three different reducing matrices, 2,5-DHB, 5-ASA, and 1,5-DAN. 1,5-DAN was found to be a better matrix than 2,5-DHB and 5-ASA for the analysis of β -casein tryptic phosphopeptides. Therefore, 1,5-DAN was used as the reducing matrix for MALDI-ISC in this experiment. Figure 7 shows a comparison of positive-ion MALDI mass spectra of an isolated mono-phosphopeptide with different matrices 1,5-DAN and 5-NSA. MALDI-ISC occurs independently of the ionization

N-terminal information



C-terminal information



Scheme 11. Structure of ISD fragments.

processes,^{27,28} so that a charge site on the ISD fragments is necessary for them to be observed in the mass spectrum. The mono-phosphopeptide having a basic N-terminal amino group and Lys residue at the C-terminus would be expected to give both N- and C-terminal positive fragment ions in ISD experiments. The use of 1,5-DAN generated c' - and z' -series ions accompanied with y' -series ions with a weak intensity (Fig. 7a). In MALDI-ISC with 1,5-DAN, the peptides principally cleave at the N–C α bond on the peptide backbone without degradation of the phosphate group (Scheme 4), thereby allowing the location of the phosphorylation site to be determined. The mass difference of 167 Da between z'_{13} and z'_{14} can be assigned to a phosphorylated Ser residue. However, the resulting z' -series ions are difficult to identify, due to interference by c' - and y' -series ions and ions from contaminants. MALDI-ISC does not allow precursor ion selection, therefore the presence of contaminants severely interferes with the interpretation of the mass spectrum.

To avoid misinterpreting ISD fragment ions, we used the 5-NSA as a matrix in the MALDI-ISC experiment described below. The use of 5-NSA generated the a - and x -series ions originating from cleavage at the C α –C bond on the peptide backbone (Scheme 7). The MALDI-ISC spectrum showed x -series ions with phosphate groups intact (Fig. 7b). The x -series ions in MALDI-ISC with 5-NSA provide useful information that was complementary to MALDI-ISC with 1,5-DAN. The z' -series ions are 41 Da smaller than the x -series ions, as shown in Scheme 11. Therefore, the z' -series ions in the MALDI-ISC spectrum with 1,5-DAN could be identified by comparing the peaks with those for x -series ions (Fig. 7). The site of phosphorylation was determined by the ISD fragment z'_{13}/z'_{14} and x_{13}/x_{14} with 1,5-DAN and 5-NSA, respectively (Fig. 7). The MALDI-ISC spectra with 1,5-DAN and 5-NSA permitted Ser35 to be unambiguously identified as the phosphorylation site.

We next examined a tetra-phosphopeptide. The positive-ion MALDI-ISC spectra of an isolated tetra-phosphopeptide obtained with 1,5-DAN and 5-NSA are shown in Fig. 8. The tetra-phosphopeptide having Arg residues at both the N- and C-termini would give rise to N- and C-terminal side fragment ions in the positive-ion MALDI-ISC experiments. However, c' -series ions (Fig. 8a) and a -series ions (Fig. 8b) were found to be dominant. It is likely that the presence of

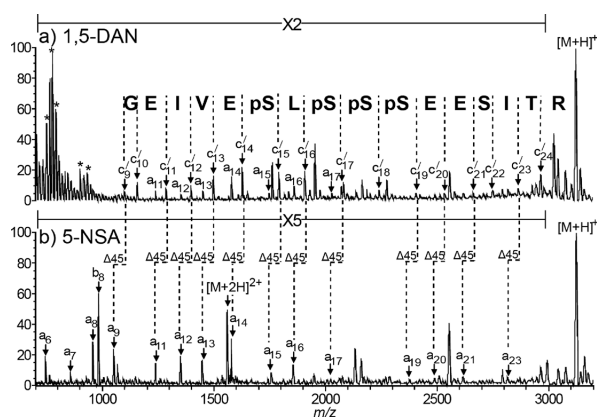


Fig. 8. Positive-ion MALDI-ISC spectra of tetra-phosphopeptide (RELEELNVPGEIVEpSLpSpSpSEESITR) obtained with (a) 1,5-DAN and (b) 5-NSA. Asterisk indicates matrix peaks. [Reproduced from ref. 34 with Copyright permission of John Wiley and Sons.]

an N-terminal Arg residue facilitates the protonation reaction compared with the C-terminal Arg residue.³⁶⁾ The use of 1,5-DAN generated c' -series ions with a small amount of a -series ions, which were derived from the degradation of c' -series ions *via* a hydrogen transfer from the β -carbon.²¹⁾ In contrast, the use of 5-NSA generated a -series ions, but the S/N ratio was found to be somewhat poor. Although all the c' -series ions (Fig. 8a) and a -series ions (Fig. 8b) containing modified residues retained all the four phosphate residues, identifying the sites of phosphorylation was difficult due to interference by contaminants. As described above for the mono-phosphopeptide, the use of 5-NSA provides useful complementary information to the MALDI-ISC with 1,5-DAN for the identification of ISC fragment ions. As shown in Fig. 8, the c' -series ions which are 45 Da larger than a -series ions are implicitly identified (Scheme 11). In particular, the a_{19} , a_{20} , and a_{21} ions which contain all the four phosphate residues provide complementary information to MALDI-ISC with 1,5-DAN for site determination, and thus, the precise determination of Ser15, Ser17, Ser18, and Ser19 as the sites of phosphorylation in β -casein were accomplished by combining information from both MALDI-ISC of the tetra-phosphopeptide with 1,5-DAN and 5-NSA. The use of MALDI-ISC with reducing and oxidizing matrices could be a useful method for the *de novo* sequencing of peptides.

ECD and EDD have been demonstrated to fragment the same phosphopeptides at the concentrations used in our experiments (5 pmol/ μ L).³⁷⁾ The ECD spectra provided sequence information and phosphorylated sites, while EDD provided very limited sequence information because of its low fragmentation efficiency.³⁷⁾ In a comparison of ECD and MALDI-ISC with a reducing matrix, ECD involves the addition of electrons to multiply protonated analytes, which are not often produced in the MALDI process. ECD spectra of the triply-protonated β -casein phosphopeptides shows both c' - and z' -series ions. The electron capture of $[M+3H]^{3+}$ results in a charge-reduced peptide cation $[M+3H]^{2+}$ and subsequently the radical-induced cleavage of N- C_{α} bonds. This suggests that the N- C_{α} bond cleavage of $[M+3H]^{2+}$ produced singly-charged c' - and z' -series ions. In the case of MALDI, singly-charged analytes are mainly observed and the presence of the basic amino acid residue contributes to

enhance the yields of protonated molecules.³⁶⁾ Therefore, a basic amino acid residue in an ISC fragment is required for them to be observed in the MALDI-ISC mass spectrum. ECD spectra usually gave more sequence information compared with MALDI-ISC with reducing matrix. By contrast, MALDI-ISC spectra were simpler than ECD spectra and ISC ions could be easily assigned. In a comparison of MALDI-ISC with 5-NSA and EDD, the MALDI-ISC spectra of β -casein phosphopeptides produced using 5-NSA as the matrix, compared with EDD spectra gave more sequence information and a higher S/N ratio of fragment ions. Therefore, the fragmentation efficiency of MALDI-ISC with 5-NSA was better than that of EDD.

CONCLUSION

MALDI-ISC of peptides was studied using several salicylic acid derivatives as matrices. The difference in the type of functional group at the 5-position in the salicylic acid derivatives can dramatically affect the ISC products. The hydrogen-donating ability of the matrix is a prominent factor in the generation of c' - and z' -series ions in MALDI-ISC. The presence of a 5-hydroxyl group in 2,5-DHB and a 5-amino group in 5-ASA likely enhances to their hydrogen-donating nature. MALDI-ISC with 2,5-DHB and 5-ASA is initiated by the transfer of hydrogen from excited matrix molecules to the carbonyl oxygen of the peptide backbone, leading to a "hydrogen-abundant" peptide. Subsequently, the c'/z' fragment pair is formed by the radical-induced cleavage at N- C_{α} bonds. The z' -series fragments can undergo radical reactions to form the z' -, $[z+matrix]$ - and w -series ions, because z' -series radical fragments are more reactive than c' -series fragments.

In contrast, the hydrogen-accepting nature of a matrix is an important factor for the generation of a - and x -series ions. The use of 5-NSA gave high ion yields of the oxidized ion $[M-2H+H]^+$. MALDI-ISC with 5-NSA is initiated by a hydrogen transfer from an amide nitrogen of the peptide backbone to the matrix molecule. The abstraction of hydrogen from peptides results in the formation of oxidized peptide molecules containing a radical site on the amide nitrogen with subsequent radical-induced cleavage at the C_{α} -C bonds, leading to the formation of an a'/x fragment pair. The a' -series ions undergo further hydrogen abstraction to form a -series ions after C_{α} -C bond cleavage. The specific cleavage of the CO-N bond at Xxx-Pro and Xxx-Sar was observed *via* hydrogen abstraction from the C_{α} -H in Pro and Sar residues. The CO-N bond cleavage leads to the formation of a b'/y fragment pair and the b' -series ions undergo further degradation to form b - and a -series ions after the CO-N bond cleavage.

MALDI-ISC with 5-FSA gave both a'/x and c'/z' fragment ions because 5-FSA has both hydrogen-donating and hydrogen-accepting properties. The use of the matrices described above gave ISC fragment ions without the loss of phosphoric groups in MALDI-ISC experiments of phosphopeptides. The use of oxidizing matrices can provide useful complementary information related to amino acid sequencing and site of post translational modifications in peptides. The MALDI-ISC with reducing and oxidizing matrices are a potentially useful method for the *de novo* sequencing of peptides.

Acknowledgments

D.A. acknowledges the research fellowship from the Japan Society for the Promotion of Science for Young Scientists (23-10272). M.T. acknowledges the supports from the Creation of Innovation Centers for Advanced Interdisciplinary Research Area in the Special Coordination Fund for Promoting Science and Technology, and Grant-in-Aid for Scientific Research (C) (23550101) from the Ministry of Education, Culture, Sports, Science and Technology of Japan.

REFERENCES

- 1) M. Karas, D. Bachmann, U. Bahr, F. Hillenkamp. *Int. J. Mass Spectrom. Ion Process.* 78: 53–68, 1987.
- 2) K. Tanaka, H. Waki, Y. Ido, S. Akita, Y. Yoshida, T. Yoshida, T. Matsuo. *Rapid Commun. Mass Spectrom.* 2: 151–153, 1988.
- 3) M. Karas, F. Hillenkamp. *Anal. Chem.* 60: 2299–2301, 1988.
- 4) C. M. Whitehouse, R. N. Dreyer, M. Yamashita, J. B. Fenn. *Anal. Chem.* 57: 675–679, 1985.
- 5) J. B. Fenn, M. Mann, C. K. Meng, S. F. Wong, C. M. Whitehouse. *Science* 246: 64–71, 1989.
- 6) W. J. Henzel, T. M. Billeci, J. T. Stults, S. C. Wong, C. Grimley, C. Watanabe. *Proc. Natl. Acad. Sci. U.S.A.* 90: 5011–5015, 1993.
- 7) D. J. C. Pappin, P. Hojrup, A. J. Bleasby. *Curr. Biol.* 3: 327–332, 1993.
- 8) R. S. Brown, J. J. Lennon. *Anal. Chem.* 67: 3990–3999, 1995.
- 9) B. Spengler, D. Kirsch, R. Kaufmann, E. Jaeger. *Rapid Commun. Mass Spectrom.* 6: 105–108, 1992.
- 10) R. Kaufmann, B. Spengler, F. Lützenkirchen. *Rapid Commun. Mass Spectrom.* 7: 902–910, 1993.
- 11) J. Hardouin. *Mass Spectrom. Rev.* 26: 672–682, 2007.
- 12) P. Chaurand, F. Luetzenkirchen, B. Spengler. *J. Am. Soc. Mass Spectrom.* 10: 91–103, 1999.
- 13) R. S. Annan, S. A. Carr. *Anal. Chem.* 68: 3413–3421, 1996.
- 14) R. A. Zubarev, N. A. Kruger, E. K. Fridriksson, M. A. Lewis, D. M. Horn, B. K. Carpenter, F. W. McLafferty. *J. Am. Chem. Soc.* 121: 2857–2862, 1999.
- 15) J. E. P. Syka, J. J. Coon, M. J. Schroeder, J. Shabanowitz, D. F. Hunt. *Proc. Natl. Acad. Sci. U.S.A.* 101: 9528–9533, 2004.
- 16) M. Takayama. *J. Am. Soc. Mass Spectrom.* 12: 1044–1049, 2001.
- 17) T. Köcher, Å. Engström, R. A. Zubarev. *Anal. Chem.* 77: 172–177, 2005.
- 18) N. L. Kelleher. *Anal. Chem.* 76: 196A–203A, 2004.
- 19) J. J. Coon. *Anal. Chem.* 81: 3208–3215, 2009.
- 20) M. Takayama. *J. Am. Soc. Mass Spectrom.* 12: 420–427, 2001.
- 21) M. Sakakura, M. Takayama. *J. Am. Soc. Mass Spectrom.* 21: 979–988, 2010.
- 22) D. Asakawa, M. Takayama. *J. Am. Soc. Mass Spectrom.* 22: 1224–1233, 2011.
- 23) Y. Fukuyama, S. Iwamoto, K. Tanaka. *J. Mass Spectrom.* 41: 191–201, 2006.
- 24) K. Demeure, L. Quinton, V. Gabelica, E. De Pauw. *Anal. Chem.* 79: 8678–8685, 2007.
- 25) N. Smargiasso, L. Quinton, E. De Pauw. *J. Am. Soc. Mass Spectrom.* 23: 469–474, 2012.
- 26) R. A. Zubarev. *Mass Spectrom. Rev.* 22: 57–77, 2003.
- 27) M. Takayama, A. Tsugita. *Int. J. Mass Spectrom.* 181: L1–L6, 1998.
- 28) K. Demeure, V. Gabelica, E. A. De Pauw. *J. Am. Soc. Mass Spectrom.* 21: 1906–1917, 2010.
- 29) B. A. Budnik, K. F. Haselmann, R. A. Zubarev. *Chem. Phys. Lett.* 342: 299–302, 2001.
- 30) J. J. Coon, J. Shabanowitz, D. F. Hunt, J. E. P. Syka. *J. Am. Soc. Mass Spectrom.* 16: 880–882, 2005.
- 31) D. Asakawa, M. Takayama. *Rapid Commun. Mass Spectrom.* 25: 2379–2383, 2011.
- 32) I. Anusiewicz, M. Jasionowski, P. Skurski, J. Simons. *J. Chem. Phys. A* 109: 11332–11337, 2005.
- 33) F. Kjeldsen, O. A. Silivra, I. A. Ivonin, K. F. Haselmann, M. Gorshkov, R. A. Zubarev. *Chem. Eur. J.* 11: 1803–1812, 2005.
- 34) D. Asakawa, M. Takayama. *J. Mass Spectrom.* 47: 180–187, 2012.
- 35) N. Sugiyama, T. Masuda, K. Shinoda, A. Nakamura, M. Tomita, Y. Ishihama. *Mol. Cell. Proteomics* 6: 1103–1109, 2007.
- 36) D. Asakawa, S. Moriguchi, M. Takayama. *J. Am. Soc. Mass Spectrom.* 23: 108–115, 2012.
- 37) H. K. Kweon, K. Håkansson. *J. Proteome Res.* 7: 749–755, 2008.