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## Food-compatible method for the efficient extraction and stabilization of cranberry pomace polyphenols

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### Abstract

Cranberry pomace is a byproduct of cranberry processing and is comprised of seeds, skins and stems of the cranberry fruit. While cranberry pomace contains beneficial polyphenols, including proanthocyanidins and anthocyanins, it is not a palatable source of these compounds and is typically discarded. In this study, we have developed and optimized a method to extract polyphenols from cranberry pomace using aqueous ethanol, a food grade solvent. Biochemical characterization of the pomace extract showed the presence of a broad range of polyphenols also present in cranberry juice concentrate. By co-drying cranberry pomace extract with a protein-rich food matrix, such as soy protein isolate (SPI), we have developed a method to produce a cranberry polyphenol-SPI complex (CBP-SPI) containing 10% cranberry polyphenols. Unlike dried cranberry pomace extract alone, proanthocyanidins, anthocyanins and total polyphenols were found to be highly stable at 37 °C in the CBP-SPI powder. The extraction and stabilization of cranberry pomace polyphenols using SPI provides an innovative approach for utilizing pomace in the development of novel food ingredients.

### Keywords

cranberry pomace; polyphenols; proanthocyanidins; anthocyanins; soy protein isolate

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## 1. Introduction

Cranberry fruits (*Vaccinium macrocarpon* Ait) are a rich source of type A proanthocyanidins, anthocyanins, flavonols, phenolic acids, benzoates, hydroxycinnamic acids, terpenes and organic acids (Pappas & Schaich, 2009). Research has associated several health benefits with the consumption of cranberry. Cranberries components are antimicrobial against *Staphylococcus*, *Salmonella* and *Escherichia*, which are responsible for food-borne illness and human disease (Puupponen-Pimia, Nohynek, Alakomi & Oksman-Caldentey, 2005). Cranberry polyphenols elicit antimicrobial activity via mechanisms independent of the acidic pH (Lacombe, Wu, Tyler & Edwards, 2010). Cranberry is used as a prophylactic for urinary tract infection (UTIs) (Barbosa-Cesnik, Brown, Buxton, Zhang, DeBusscher & Foxman, 2011; Epp et al., 2010; Jepson & Craig, 2008) and the presence of A-type proanthocyanidins has been associated with preventing the adhesion of uropathogenic *E.coli* to uroepithelial cells (Howell, Reed, Krueger, Winterbottom, Cunningham & Leahy, 2005; Sobota, 1984). *In vitro* studies have also shown that cranberry can prevent the adhesion of *H. pylori* to gastric mucosal cells (Burger, Ofek, Tabak, Weiss, Sharon & Neeman, 2000; Burger, Weiss, Sharon, Tabak, Neeman & Ofek, 2002; Shmueli et al., 2004) and in a clinical study the daily consumption of cranberry juice cocktail for 90 days was shown to decrease the *H. pylori* infection in adults compared to placebo (Zhang, Ma, Pan, Go, Chen & You, 2005).

Cranberries are commonly consumed as dried fruit or juice adulterated with large amounts of sugar. The sugar used to mask the tartness of cranberry adds excess calories and can counteract the beneficial effects of the fruit. Earlier, we developed a technology that uses protein-rich legume flours or powders, such as soy protein isolate (SPI), to sorb, concentrate and stabilize mid-polarity range cranberry phytochemicals from cranberry juice concentrate while separating them from excess water, sugar and lipid components (Roopchand et al., 2012). Cranberry pomace, a byproduct of the cranberry processing industry, is composed of dietary fiber from skins, seeds and stems (White, Howard & Prior, 2010) and is typically discarded. Due to its acidity and low protein content it has limited use as animal feed and disposal in landfills poses an environmental problem due to low pH. Cranberry pomace still contains an assortment of beneficial phytochemicals, such as polyphenols, which have a natural affinity for proteins. Soy protein isolate (SPI) made from soybeans is an inexpensive readily available protein that is frequently incorporated into foods and supplements. Compared to animal proteins such as casein, SPI was shown to be hypocholesterolemic (Nagata, Ishiwaki & Sugano, 1982), decrease the activity of lipogenic enzymes (Iritani, Nagashima, Fukuda, Katsurada & Tanaka, 1986), and reduce body fat in diet-induced obese rats and mice (Aoyama, Fukui, Takamatsu, Hashimoto & Yamamoto, 2000). Therefore we have developed a food-compatible method for the extraction of cranberry pomace and subsequent stabilization of cranberry phytochemicals via complexation to SPI to produce a novel phytochemical-enriched protein that can be used as a food ingredient.

## 2. Materials and methods

### 2.1. Optimization of method for cranberry pomace extraction

Frozen depectinized cranberry pomace (provided by BNK, Wisconsin Rapids, WI) was used in all analyses. Cranberry pomace (30 g wet wt.) was pureed in a Vitamix blender with 300 ml of water or 50% ethanol (10:1 extraction). Each puree was aliquoted into 40 ml volumes in 50 ml tubes and the pH was adjusted from 4 to 2, 3 or 5 with HCl or NaOH to investigate the effect of pH on extraction efficiency. The material was then extracted in loosely capped tubes in an 80 °C water bath for 2 h. The samples were centrifuged at 4,000 rpm for 10 min to separate solids from the liquid extract. The concentration of total polyphenols and proanthocyanidins in each extract was determined using the Folin-Ciocalteu (Singleton &

Rossi, 1965) and 4-(dimethylamino)cinnamaldehyde (DMAC) methods (Prior et al., 2010), respectively.

Three independent lots of frozen depectinized cranberry pomace were extracted to compare solvent to pomace ratio and percentage of ethanol on polyphenol extraction efficiency. A 10 g (wet wt.) sample was removed from each lot and the dry weight of each was determined by freeze drying. A 100 g (wet wt.) sample of each lot of cranberry pomace was pureed in a Vitamix blender with 1 l of 50% ethanol (10:1 extraction) or 500 ml of 50% ethanol (5:1 extraction). Each puree was adjusted to pH 2 with HCl and the material was then extracted in a rotary evaporation flask (without vacuum) in an 80 °C water bath for 2 h (Buchi, Switzerland). Any evaporated solvent collected in the catch flask was added back to the extraction flask. The samples were centrifuged at 4,000 rpm for 10 min to separate the solids from the extract, which was then filtered through miracloth (Calbiochem). Similarly, 100 g (wet wt.) of each lot of cranberry pomace was also extracted with 1 l of 75% ethanol (10:1 extraction) or 500 ml of 75% ethanol (5:1 extraction). The volumes of extract recovered were measured and the concentrations of total polyphenols and proanthocyanidins in each extract were quantified as described above. The dry weight of each extract was determined by vacuum drying 1 ml aliquots of liquid extract. These data were then used to calculate the percentage of total polyphenols and proanthocyanidins in the dried extract. Proanthocyanidins as a percentage of total polyphenols was also calculated.

## 2.2. Biochemical characterization of cranberry pomace extract

Cranberry pomace (50 g wet wt.) was blended with 250 ml of 50% ethanol, mixture was adjusted to pH 2 and then extracted in a flask at 80 °C for 2 h on a shaking water bath. The cranberry pomace extract was filtered (0.22 µm) prior to analysis. Samples were separated and analyzed with a Shimadzu LCMS-2010A high performance liquid chromatography/electrospray ionization/single quadrupole. The LCMS is equipped with two pumps (LC-10ADvp), controller (SCL-10Avp), autoinjector (SIL-10ADvp), column oven (CTO-10ACvp), photodiode array detector (SPD-M10Avp), and a single quadrupole analyzer. The HPLC column was a Widespore C5, 5 µm, 2.1×150 mm (Supelco, Bellefonte, PA). The mobile phase consisted of 0.1% formic acid in water (Solvent A) and 0.1% formic acid in acetonitrile (Solvent B) at a flow rate of 0.2 ml/min. The gradient used was: initially 90% A and 10% B using an isocratic flow for first 5 min; 85% A and 15% B over 15 min, followed by isocratic elution at 85% A and 15% B for 5 min; 100% A and 0% B over 2 min, followed by an isocratic elution at 100% A and 0% B for 5 min. There was a 13 min equilibration interval between injections.

For characterization of proanthocyanidins in the cranberry pomace extract, mass spectra were collected on a Bruker ULTRAFLEX<sup>®</sup>III MALDI TOF/TOF mass spectrometer (Billerica, MA, USA) equipped with delayed extraction and a SmartBeam<sup>®</sup> laser. All analyses were performed in positive linear mode. Spectra were the sum of 8–10 different locations in each well, accumulating a total of 400–500 shots to minimize intra-well variability and avoid heterogeneous co-crystallization spots. Threshold laser power was used to achieve optimal isotope patterns. The matrix was 2, 5-dihydroxybenzoic acid (DHB) at a concentration of 50 mg/ml in ethanol. FlexControl and FlexAnalysis (Bruker Daltonik GmbH, Bremen, Germany, version 3.0) were used for data acquisition and data processing, respectively. mMass (version 3.9.0) was used for spectra analysis. Based on previously described structures an equation was developed to predict the mass distribution of PAC in cranberry products (Reed, Krueger & Vestling, 2005). The equation is  $290 + 288a - 2b + 23$ , where 290 represents the molecular weight of the terminal catechin/epicatechin unit,  $a$  is the degree of polymerization of catechin/epicatechin units,  $b$  is the number of A-type interflavan bonds, and 23 is the molecular weight of sodium.

### 2.3. Production of cranberry polyphenol-SPI complex

Cranberry pomace (300 g wet wt.) was extracted as described in Section 2.1 and 2.4 l of liquid extract was collected after separating the cranberry pomace solids; the concentrations of total polyphenols and proanthocyanidins were quantified in the extract. Three 1 ml aliquots of extract were dried in a speed vacuum and the average dry weight per ml was determined. A calculated amount of soy protein isolate (SPI; ADM, Decatur, IL) was added to the liquid cranberry pomace extract so that after solvent evaporation the result is a cranberry polyphenol-SPI complex (CBP-SPI) containing 10% total polyphenols. The dry weight of the 2.4 l of cranberry pomace extract was 19.1 g containing 3.04 g of total polyphenols. The final amount of CBP-SPI was therefore calculated to be 30.4 g to have the cranberry polyphenol content standardized to 10%. Therefore, the amount of SPI to be added to the cranberry extract was calculated as follows:  $19.1 \text{ g} + x = 30.4 \text{ g}$ ;  $x = 11.3 \text{ g}$  of SPI

After SPI was added to the cranberry pomace extract, ethanol was removed by rotary evaporation under vacuum (Büchi, Switzerland) with temperature set at 40 °C. The remaining liquid was removed by freeze-drying to yield CBP-SPI powder.

### 2.4. Temperature stability of polyphenols in CBP-SPI compared to cranberry pomace extract

Cranberry pomace extract was prepared as described in Section 2.1. A portion was used to make CBP-SPI containing 10% total polyphenols as described in Section 2.3 and a portion was dried to make cranberry pomace extract powder (CBP extract). The CBP-SPI and CBP extract were placed in a 37 °C incubator and aliquots of each sample were removed at indicated times. The dried cranberry pomace extract (10 mg) or CBP-SPI complex (100 mg) was eluted 3 times with 1 ml of 50% acetone. Proanthocyanidins were quantified using the DMAC method (Prior et al., 2010), anthocyanins were measured using the pH differential method (Lee, Durst & Wrolstad, 2005) and total polyphenols was measured by the Folin-Ciocalteu method (Singleton et al., 1965).

### 2.5. Statistics

Statistics were performed with *STATISTICA* v.10 (StatSoft). T-tests were performed between groups. One-way ANOVA was used to determine significance among three or more groups followed by the indicated post-hoc test.

## 3. Results

### 3.1. Optimization of cranberry pomace extraction using food-compatible solvents

The mean dry weight of three independent lots of frozen cranberry pomace was  $26.7 \pm 1.3 \text{ g}$  per 100 g of wet weight. To determine the effect of pH on extraction efficiency a series of extractions were performed at 80 °C for 2 h in water or 50% ethanol using a 10:1 extraction ratio (solvent:wet pomace) at different pH values. For water extracts, pH 2 gave the highest concentration of total polyphenols (266 mg/l) and proanthocyanidins (114 mg/l) while raising the pH to 5 resulted in the lowest concentrations (Table 1). Extraction with 50% ethanol was more efficient than water resulting in a 3.6 fold greater yield of total polyphenols (963 mg/l) and a 2.7 fold greater yield of proanthocyanidins (306 mg/l) at pH 2. Yield became progressively lower as the pH was increased to 5 (Table 1).

The extraction of polyphenols and proanthocyanidins was also compared using the three lots of cranberry pomace, each extracted with 50% or 75% aqueous ethanol using either a 10:1 or 5:1 (solvent:wet pomace) extraction ratio. Each mixture was adjusted to pH 2 with HCl and extractions were performed at 80 °C with continuous mixing for 2 h. In the case of 50%

ethanol a 10:1 solvent to solids ratio extracted a higher amount of total polyphenols ( $750 \pm 19$  mg) compared to the 5:1 ratio ( $507 \pm 58$  mg); however, there was no significant difference in the amount of proanthocyanidins extracted (Table 2). The 5:1 extractions performed with 50% ethanol resulted in the highest percentage of proanthocyanidins relative to total polyphenols (Table 2). The extractions performed with 75% ethanol resulted in higher amounts of proanthocyanidins when a 10:1 solvent ratio was used, but the amount of total polyphenols and the amount of proanthocyanidins as a percentage of total polyphenols was not significantly different between the 10:1 and 5:1 extractions (Table 2). The data indicated that optimal extraction conditions were a 5:1 (solvent:wet pomace) ratio of cranberry pomace in 50% ethanol adjusted to pH 2 incubated at 80 °C for 2 h with agitation. These conditions were best for obtaining an extract enriched in proanthocyanidins compared to total polyphenols while minimizing the volume of solvent therefore they were used for all subsequent extractions.

### 3.2. Biochemical characterization of cranberry pomace extract

Cranberries contain three types of flavonoids: flavonols, anthocyanins and proanthocyanidins (PAC) (Puski & Francis, 1967; Hong & Wrolstad, 1990; Wang, Du & Francis, 1978). The flavonols in cranberries are glycosides of quercetin, myricetin and kaempferol (Hong & Wrolstad, 1990; Bilyk & Sapers, 1986). Anthocyanins present are glycosides of cyanidin and peonidin (Hong & Wrolstad, 1990). Proanthocyanidins are polymers of flavan-3-ols and flavans linked through A-type and B-type interflavan carbon bonds (Howell, Reed, Krueger, Winterbottom, Cunningham & Leahy, 2005; Porter, Krueger, Wiebe, Cunningham & Reed, 2001; Neto et al., 2006; Reed, Krueger, Vestling, 2005; Foo, Lu, Howell, Vorsa, 2000). Cranberry pomace was extracted using the optimal method, described in Section 3.1, for compound characterization by LC-MS and MALDI-TOF MS. Table 3 summarizes putatively identified compounds from cranberry pomace extract, which included anthocyanins, quercetin and quercetin glycosides. According to relative peak mass area, peonidin glycosides were the most abundant compounds detected, followed by cyanidin glycosides and quercetin. Quercetin was 10–17 times more abundant than quercetin glycosides. These findings are in agreement with previously reported data for cranberry pomace (White et al., 2010).

MALDI-TOF MS methods have been developed and applied to characterize structural features of proanthocyanidins and other tannins in fruits (Feliciano, Krueger, Shanmuganayagam, Vestling & Reed, 2012; Howell, Reed, Krueger, Winterbottom, Cunningham & Leahy, 2005; Reed, Krueger & Vestling 2005; Krueger, Dopke, Treichel, Folts & Reed, 2000; Krueger, Vestling & Reed, 2003; Krueger, Vestling & Reed, 2004, Afaq, Saleem, Krueger, Reed & Mukhtar, 2005; Neto et al 2006; Li et al 2010). MALDI-TOF MS is ideally suited for characterizing polydispersed oligomers and is considered the mass spectral method of choice for analysis of proanthocyanidins, which exhibit large structural heterogeneity (Reed, Krueger & Vestling, 2005; Hanton; 2001). MALDI-TOF MS produces a singly charged molecular ion for each parent molecule allowing precise detection of high mass (Montaudo, 2002) and baseline detection of a broad range of cranberry proanthocyanidin oligomers over the degree of polymerization (DP) range of 2 to 23 (Reed, Krueger, & Vestling, 2000). MALDI-TOF MS analysis of the cranberry pomace extract confirmed a series of masses corresponding to a proanthocyanidin series  $[M + Na]^+$  from three to fifteen degrees of polymerization ( $m/z$  887.9 to  $m/z$  4372.3), which have previously been described (Reed et al., 2005) in cranberry products (Figure 1). Therefore extraction of cranberry pomace using our described method efficiently captures a broad range of compounds representative of cranberry fruit.



### 3.3. Temperature stability of proanthocyanidins and anthocyanins in CBP-SPI

Polyphenol-protein complexes are known to form through hydrophobic and hydrogen bond interactions that depend on the structure of the protein and polyphenol molecules (Bandyopadhyay, Ghosh & Ghosh, 2012). We explored the utility of complexing cranberry polyphenols to SPI to create a novel food ingredient. Cranberry pomace was extracted as described in Section 3.1 and polyphenols in the cranberry pomace extract were complexed with SPI as described in Section 2.3 to produce cranberry polyphenol-SPI complex (CBP-SPI) containing 10% total polyphenols (Figure 2). Lyophilized cranberry pomace extract was also prepared, as described in Section 2.4. CBP-SPI and dried cranberry pomace extract were incubated at 37 °C and the stability of proanthocyanidins, anthocyanins and total polyphenols was evaluated over the course of 28 days. Prior to incubation at 37 °C (Day 0), the polyphenol concentrations were quantified in the dried pomace extract or CBP-SPI after eluting the powders three times with 50% acetone. Aliquots of dried pomace extract or CBP-SPI were removed after incubation at 37 °C on the indicated days and subjected to three rounds of elution with 50% acetone. Proanthocyanidins, anthocyanins and total polyphenols were measured as described in methods and expressed as a percentage of the original amounts that were eluted on day 0. The levels of proanthocyanidins, anthocyanins and total polyphenols in the CBP-SPI powder remained remarkably stable after the 28-day incubation period at 37 °C (Fig. 3). In contrast, on day 28 the levels of proanthocyanidins, anthocyanins and total polyphenols in the dried cranberry pomace extract declined significantly by 49%, 32% and 66%, respectively (Fig. 3). Anthocyanins are known to be highly unstable at elevated temperature (Buckow, Kastell, Terefe & Versteeg, 2010; Kechinski, Guimaraes, Norena, Tessaro & Marczak, 2010), therefore the data suggest that complexation with SPI protects them from degradation at 37 °C. Proanthocyanidins on the other hand are known to be stable (Grace, Massey, Mbeunkui, Yousef & Lila, 2012) therefore the decline in acetone-extractable proanthocyanidins was rather unexpected. One possible explanation is that at elevated temperature proanthocyanidins in the CBP extract rearrange to form insoluble complexes that cannot be measured by the colorimetric DMAC assay. In contrast proanthocyanidins in complex with SPI appear to be protected from such rearrangements at 37 °C and are thus more likely to retain their biological effects. Compared to CBP-SPI, the levels of all polyphenols decreased in the dried cranberry pomace extract at the elevated temperature (Fig. 3C), suggesting that complexation with SPI has a stabilizing effect on all polyphenols.

## 4. Discussion

Our method of extracting phytochemicals from cranberry pomace and stabilizing extracted compounds via complexation with SPI offers a novel strategy to capture, stabilize and deliver valuable phytochemicals that are usually wasted in the cranberry juicing process. Extraction of cranberry pomace with 50% ethanol likely releases more high molecular weight oligomeric tannins than would be expected in water-extracted or juiced cranberries. The resulting CBP-SPI matrix is a phytochemically-enriched protein ingredient that can be incorporated into foods and dietary supplements.

Work on polyphenol-protein interactions have been recently reviewed (Bandyopadhyay et al., 2012; Santos-Buelga & Scalbert, 2000). In addition to solution parameters such as pH and temperature, polyphenol-protein complexation depends on the three dimensional distributions of different amino acid residues on protein surfaces or cavities and the size of the cavity to accommodate large or small polyphenol molecules. In general interactions are mediated by flexible proline-rich proteins that possess an open conformation allowing hydrophobic interactions with polyphenols, which can be reinforced by hydrogen bonds between phenolic hydroxyl groups (proton donors) and carbonyl groups of peptide bonds (proton acceptor) (Hagerman & Butler, 1980; Hagerman & Butler, 1981). Polyphenols-

protein interactions can cause either unfolding of a protein chain or induce proteins to go from an unfolded to folded state (Bandyopadhyay et al., 2012). Non-polar tannins (eg. pentagalloylglucose; PGG) were found to precipitate bovine serum albumin (BSA) by forming a hydrophobic coat around the protein while polar tannins (eg. epicatechin<sub>16</sub>(4–8) catechin; EC<sub>16</sub>-C) formed hydrogen bonds between protein molecules (Hagerman, Rice & Ritchard, 1998). Ethanol suppressed the interaction between BSA and non-polar PGG, but not the interaction with more polar EC<sub>16</sub>-C (Hagerman et al., 1998). In our process, the 50% ethanol used for extraction is evaporated after addition of SPI therefore polyphenol-protein complexes should form without disruption.

The stabilizing effect of SPI on cranberry pomace polyphenols at 37 °C is an advantage that will also have to be further evaluated at higher temperatures and in finished food products that incorporate the CBP-SPI ingredient. Based on our preliminary taste assessments we observed that complexation of CBP polyphenols with SPI masks the astringent taste associated with tannins as they are not available to interact with proteins in saliva; however, this remains to be formally evaluated in a sensory panel.

We have previously reported that defatted soybean flour (DSF) can be used to sorb polyphenols from cranberry juice concentrate while leaving behind excess sugars/lipids in the supernatant (Roopchand et al., 2012). However, the sorption of polyphenols from cranberry concentrate is limited by the sorption capacity of the DSF or other protein-rich matrix and is inversely related to the amount of DSF added to a given dilution of cranberry concentrate. For example, mixing 50 g/L protein-rich DSF matrix with cranberry concentrate (50° Brix) diluted 2 times with water will sorb 49% of the proanthocyanidins from the juice to produce a cranberry polyphenol-enriched DSF (CB-DSF) matrix containing 3% proanthocyanidins (29 mg/g) while adding 5 g/L of DSF to similarly diluted cranberry concentrate will sorb 17% of proanthocyanidins from the juice, but produce a CB-DSF matrix containing 10% proanthocyanidins (100 mg/g) (Roopchand et al., 2012). While the lower ratio of DSF to juice produces a more concentrated CB-DSF product, the scale is not favorable for manufacturing large quantities of product. In the process described in this study, the co-drying of cranberry pomace extract with the protein matrix allows production of a CBP-SPI product containing up to 10% total polyphenols by simply adjusting the amount of protein matrix added to the liquid extract prior to co-drying. Our food-compatible method of extracting polyphenols from fruit and vegetable byproducts and creating polyphenol-enriched protein or flour ingredients may be a practical way of utilizing agricultural byproducts and increasing the level of beneficial polyphenols in the diet.

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We thank Brandon Mirda for technical assistance. DER designed experiments, performed statistical analysis and wrote the manuscript. CGK did MALDI-TOF and LC-MS analysis. KM performed stability study. MAL and BF were consultants to the work. IR is the principal investigator of the laboratory where the research was performed. All authors read and approved the final manuscript.

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## ABBREVIATIONS

<b>SPI</b>	soy protein isolate
<b>CBP-SPI</b>	cranberry polyphenol-SPI complex

ACN	anthocyanin
PAC	proanthocyanidin

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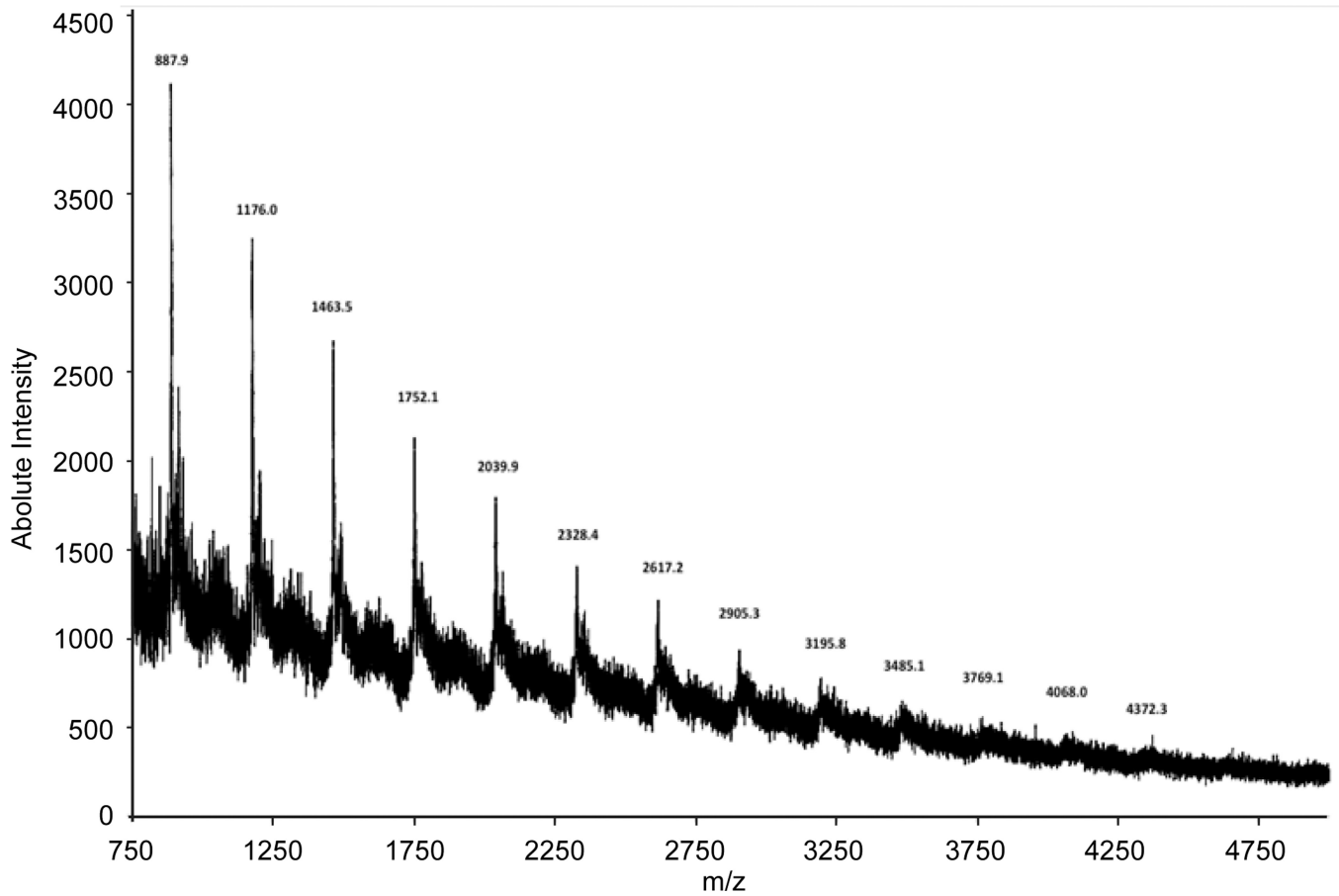
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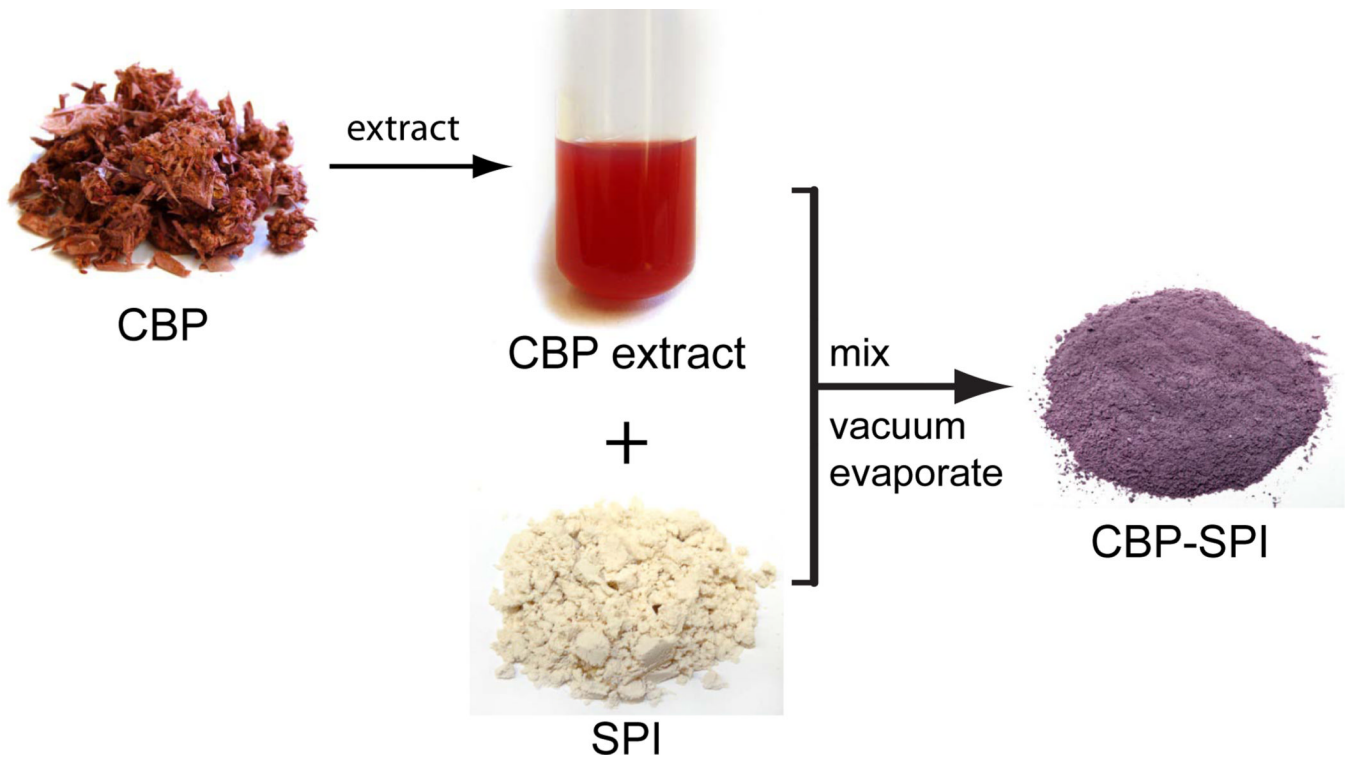
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### Highlights

1. Cranberry pomace polyphenols are most efficiently extracted at pH 2.
2. Cranberry pomace extract contains compounds present in cranberry juice.
3. Cranberry polyphenols are stabilized when complexed to soy protein isolate.

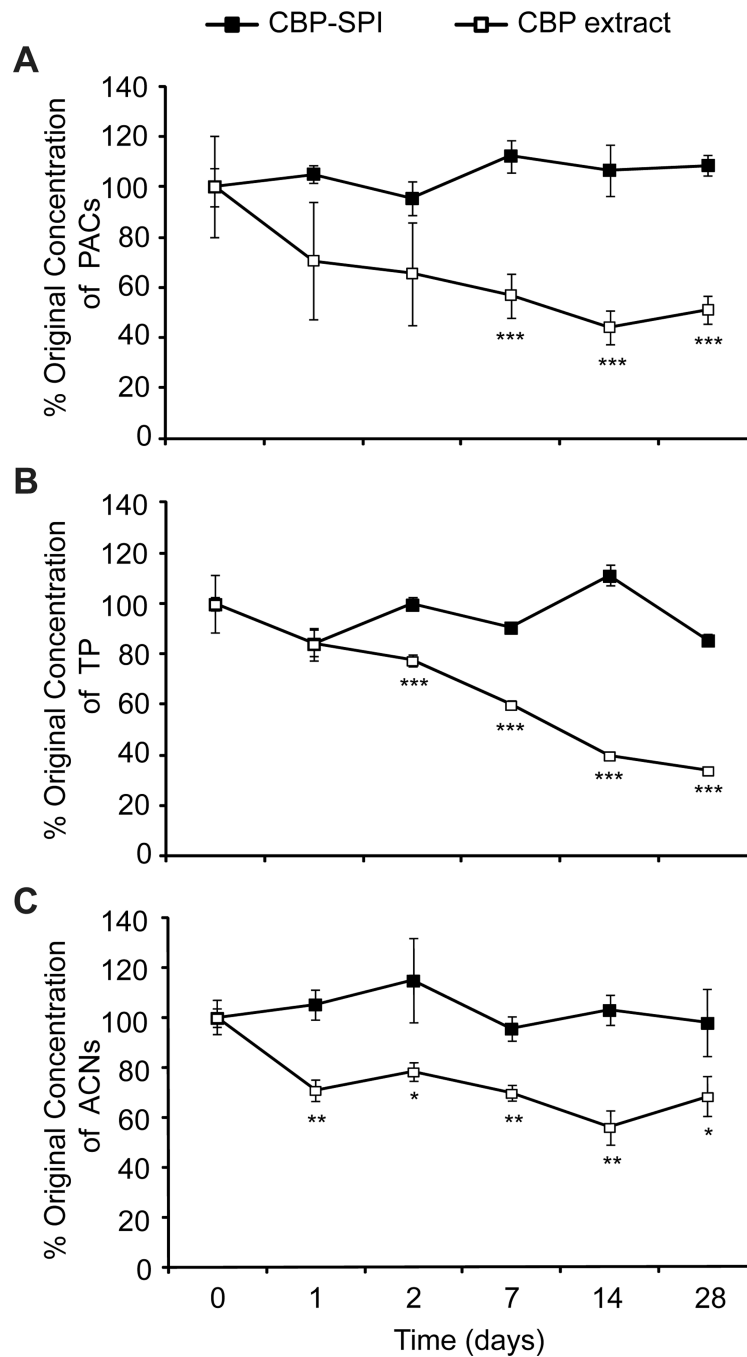


**Figure 1. MALDI-TOF mass spectrum of cranberry pomace extract**  
Data was generated in positive linear mode and shows a proanthocyanidin series  $[M + Na]^+$  from 3 degrees of polymerization (DP;  $m/z$  887.9) to DP 15 ( $m/z$  4372.3).



**Figure 2. Schematic for production of CBP-SPI complex**

Cranberry pomace was extracted in 50% ethanol at 80 °C at pH 2 for 2 h. After the total polyphenols in the cranberry pomace extract (CBP extract) was quantified, a calculated amount of soy protein isolate (SPI) is mixed into the extract to yield a final product containing up to 10% total polyphenols. Vacuum evaporation of 50% ethanol followed by freeze drying yielded the cranberry polyphenol-SPI complex (CBP-SPI).



**Figure 3. Stability of cranberry polyphenols bound and unbound to SPI**  
Cranberry (A) proanthocyanidins, (B) total polyphenols and (C) anthocyanins expressed as a percentage of the original concentration measured on day 0 (before incubation at 37 °C) and after 1, 2, 7, 14 and 28 days at 37 °C.



**Table 1**

pH dependence of polyphenols extracted from cranberry pomace in water or aq. ethanol (10:1, solvent:pomace)

Solvent	pH	Total Polyphenols (mg/l)	Proanthocyanidins (mg/l)
<b>H<sub>2</sub>O</b>	2	266	114
	3	104	47
	4	124	49
	5	78	25
<b>50% Ethanol</b>	2	964	306
	3	591	181
	4	469	154
	5	408	123

Total polyphenols were calculated as gallic acid and proanthocyanidins were calculated as proanthocyanidin A2 equivalents

**Table 2**

Comparison of 10:1 and 5:1 extraction of cranberry pomace with 50% and 75% aqueous ethanol

Solvent	Volume (ml)	TP (mg/l)	TP (mg)	% TP in dry extract	PACs (mg/l)	PACs (mg)	% PACs in dry extract	PACs as % of TP
50% EtOH (10:1)	793 ± 40	947 ± 73.1	750 ± 19 **	15.1 ± 1.6	495 ± 73	391 ± 39	7.9 ± 1.3	52 ± 4.1
50% EtOH (5:1)	328 ± 3	1545 ± 193	507 ± 58	12.6 ± 1.2	1170 ± 152	384 ± 47	9.6 ± 1.0	76 ± 7.0 **
75% EtOH (10:1)	878 ± 45	863 ± 114	761 ± 139	11.8 ± 1.3	531 ± 31	466 ± 4.2 **	7.3 ± 1.5	63 ± 11
75% EtOH (5:1)	418 ± 33	1610 ± 157	678 ± 118	15.7 ± 1.8	841 ± 111.5	352 ± 23	8.3 ± 1.5	53 ± 5.8

n=3 independent sample lots for each solvent condition.

T-test, \*\* p<0.01

TP = total polyphenols calculated as gallic acid equivalents. PACs = proanthocyanidins calculated as proanthocyanidin A2 equivalents

**Table 3**

Compounds extracted from cranberry pomac

Compounds	Peak mass area ( $\times 10^6$ )	MS ( $m/z$ ) ESI		Rt (min)
		[M] <sup>+</sup>	[M-H] <sup>+</sup>	
Cyanidin-3- <i>O</i> -galactoside	16.5	449		4.48
Cyanidin-3- <i>O</i> -arabinoside	19.7	419		4.91
Peonidin -3- <i>O</i> -galactoside	37.6	463		5.81
Peonidin-3- <i>O</i> -arabinoside	35.0	433		7.33
Quercetin	19.5		301	23.03
Quercetin-3- <i>O</i> -galactoside	1.46		463	12.3
Quercetin-3- <i>O</i> -arabinoside	1.98		433	15.03; 15.57
Quercetin-3- <i>O</i> -rhamnoside	1.13		447	17.63