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Oxidized Fatty Acid Analysis by Charge Switch Derivatization, Selected Reaction Monitoring and Accurate Mass Quantification

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Abstract

A highly sensitive, specific and robust method for the analysis of oxidized metabolites of linoleic acid (LA), arachidonic acid (AA) and docosahexaenoic acid (DHA) was developed using charge-switch derivatization, LC-ESI MS/MS with selected reaction monitoring (SRM) and quantitation by high mass accuracy analysis of product ions thereby minimizing interferences from contaminating ions. Charge-switch derivatization of LA, AA and DHA metabolites with N-(4-aminomethylphenyl)-pyridinium resulted in a 10-30 fold increase in ionization efficiency. Improved quantitation was accompanied by decreased false positive interferences through accurate mass measurements of diagnostic product ions during SRM transitions by ratiometric comparisons with stable isotope internal standards. The limits of quantification (LOQ) were between 0.05 to 6.0 pg with a dynamic range of 3-4 orders of magnitude (correlation coefficient $r^2 > 0.99$). This approach was utilized to quantitate the levels of representative fatty acid metabolites from WT and iPLA₂^{-/-} mouse liver identifying the role of iPLA₂ in hepatic lipid 2nd messenger production. Collectively, these results demonstrate the utility of high mass accuracy product ion analysis in conjunction with charge-switch derivatization for the highly specific quantitation of diminutive amounts of LA, AA and DHA metabolites in biologic systems.

Keywords

Eicosanoids; Docosanoids; Metabolite Determination; LC-MS/MS; Phospholipase; iPLA₂

Introduction

Polyunsaturated fatty acids (PUFAs) such as arachidonic acid (AA, 20:4n-6), docosahexaenoic acid (DHA, 22:6n-3) and linoleic acid (LA, 18:2n-6) are essential components of mammalian membranes that serve multiple roles in cellular function [1-5].

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Typically, these polyunsaturated fatty acids are stored in an inactive form esterified to membrane phospholipids and are released during cellular activation by a wide variety of phospholipases to facilitate cellular responses to external perturbations [6-10]. The precise complement of lipid 2nd messengers in each cell type is determined, in large part, by the integrated array of cell type-specific phospholipases and the kinetic flux through specific downstream oxidative pathways tailored to the function of each cell type to generate a highly specific array of polyunsaturated oxidized signaling metabolites. Prominent examples of oxidative enzymes leading to the generation of signaling fatty acids include the cyclooxygenases [11, 12], lipoxygenases [13-15] and P450 monooxygenases [16, 17]. Many hundreds to thousands of oxidized fatty acid metabolites are generated in mammalian cells that coordinate physiologic signaling processes and cellular adaptation in health, but are subject to maladaptive alterations in disease processes promoting disease progression [18-20]. Accordingly, a major goal of lipid 2nd messenger research is the identification of the diversity of oxidized fatty acids in biologic systems, quantitation of alterations in their amounts during physiologic and pathologic perturbations, and mechanistic identification of their roles in disease processes [e.g., 21-29].

Recent approaches for the identification and quantification of the diversity of oxidized fatty acids in biologic systems have largely relied on liquid chromatographic separation followed by electrospray ionization in the negative ion mode and selected reaction monitoring (SRM) [30-38]. However, ionization in the negative ion mode is inefficient in comparison to ionization in the positive ion mode. Although oxidized fatty acids in inflammatory cells, abscesses, exudates or inflammatory fluids can be measured using negative mode ionization [39-51], the much lower abundance of oxidized fatty acid signaling metabolites that regulate cellular function in muscle, liver, brain and other solid organs are typically 20 - 2000 fold lower than that present in inflammatory cells or fluids. Thus, the substantially lower ionization efficiency in the negative ion mode has placed severe limitations on the accurate identification and quantitation of oxidized fatty acids in non-inflammatory cells and tissues.

The inherent limitations in sensitivity in the analysis of oxidized fatty acids in the negative ion mode have previously been appreciated. Accordingly, multiple charge-switch chemical derivatization procedures to improve the ionization efficiency for detection of oxidized fatty acid signaling metabolites have been developed using derivatization and positive ion mode ionization [e.g., 52-59]. Examples of charge switch derivatization of the carboxylic acid group include derivatization with tris(2,4,6-trimethoxyphenyl)phosphonium propylamine (TMPP) bromide [52], cholamine [53], 2-bromo-1-methylpyridinium iodide (BMP) and 3-carbinol-1-methylpyridinium iodide (CMP) [54], p-dimethylaminophenacyl bromide (DmPA-Br) [55], 2-Hydrazinopyridine (HP), 2-picolylamine (PA) and others [56-58]. However, these approaches are suboptimal for analysis of oxidized fatty acid moieties using ESI due to the undesirable fragmentation of the charge switch tag in many cases. This weakness was recognized by Gelb *et al.* who utilized molecular orbital calculations to develop a novel charge-switch reagent, N-(4-aminomethylphenyl)-pyridinium (AMPP) [59], which possessed the desired charge switch mediated increase in S/N and markedly improved the generation of informative fragment ions. Through using high mass accuracy product ion analysis in conjunction with AMPP charge-switch derivatization we demonstrate the sensitivity and specificity of this method for the detection and quantitation of oxidized linoleic, arachidonic, and docosahexaenoic acid metabolites in the high attomole and low fmol range. Furthermore, high mass accuracy analysis of diagnostic product ions collected during SRM dramatically decreases false positive interferences which compromise identification and quantitation of the targeted analytes that would otherwise go undetected or over estimated using traditional isobaric SRM analyses. The utility of this approach in biological tissues was examined in hepatic tissue from wild-type and iPLA₂^{-/-} mice we previously generated and characterized [60-62]. Collectively, this approach provides a facile

method to dramatically increase the sensitivity and accuracy of analysis of diminutive amounts of oxidized fatty acids in biologic samples while simultaneously decreasing false positive interferences through the use of high mass accuracy product ion analysis during monitored SRM transitions.

Materials and Methods

Materials

The following standards from Cayman Chemical (Ann Arbor, MI) were used: thromboxane B₂ (TXB₂), TXB₂-d₄, 6-keto prostaglandin F₁ (6keto-PGF₁), prostaglandin F₂ (PGF₂), prostaglandin F₁ (PGF₁), prostaglandin E₂ (PGE₂), PGE₂-d₄, prostaglandin E₁ (PGE₁), prostaglandin D₂ (PGD₂), leukotriene B₄ (LTB₄), LTB₄-d₄, 5-hydroxy-6E,8Z,11Z,14Z-eicosatetraenoic acid (5-HETE), 8-hydroxy-5Z,9E,11Z,14Z-eicosatetraenoic acid (8-HETE), 11-hydroxy-5Z,8Z,12E,14Z-eicosatetraenoic acid (11-HETE), 12-hydroxy-5Z,8Z,10E,14Z-eicosatetraenoic acid (12-HETE), 12-HETE-d₈, 15-hydroxy-5Z,8Z,11Z,13E-eicosatetraenoic acid (15-HETE), 20-hydroxy-5Z,8Z,11Z,14Z-eicosatetraenoic acid (20-HETE), 5(6)-epoxy-8Z, 11Z,14Z-eicosatrienoic acid (5,6-EET), 8(9)-epoxy-5Z,11Z,14Z-eicosatrienoic acid (8,9-EET), 11(12)-epoxy-5Z,8Z,14Z-eicosatrienoic acid (11,12-EET), 14(15)-epoxy-5Z,8Z,11Z-eicosatrienoic acid (14,15-EET), arachidonic acid (AA), 9-hydroxy-10E,12Z-octadecadienoic acid (9-HODE), 13-hydroxy-9Z,11E-octadecadienoic acid (13-HODE), 13-HODE-d₄, 9-oxo-10E,12Z-octadecadienoic acid (9-OxoODE), 13-oxo-9Z,11E-octadecadienoic acid (13-OxoODE), 9(10)-dihydroxy-12Z-octadecenoic acid (9,10-DiHOME), 9,10-DiHOME-d₄, 12,13-dihydroxy-9Z-octadecenoic acid (12,13-DiHOME), 12,13-DiHOME-d₄, 9(10)-epoxy-12Z-octadecenoic acid (9(10)-EpOME), 12(13)epoxy-9Z-octadecenoic acid (12(13)-EpOME), linoleic acid (LA), 4-hydroxy-5E,7Z,10Z,13Z,16Z,19Z-docosahexaenoic acid (4-HDoHE), 7-hydroxy-4Z,8E,10Z,13Z,16Z,19Z-docosahexaenoic acid (7-HDoHE), 8-hydroxy-4Z,6E,10Z, 13Z,16Z,19Z-docosahexaenoic acid (8-HDoHE), 10-hydroxy-4Z,7Z,11E,13Z,16Z,19Z-docosahexaenoic acid (10-HDoHE), 11-hydroxy-4Z,7Z,9E,13Z,16Z,19Z-docosahexaenoic acid (11-HDoHE), 13-hydroxy-4Z,7Z,10Z,14E,16Z,19Z-docosahexaenoic acid (13-HDoHE), 14-hydroxy-4Z,7Z,10Z,12E,16Z,19Z-docosahexaenoic acid (14-HDoHE), 16-hydroxy-4Z,7Z,10Z, 13Z,17E,19Z-docosahexaenoic acid (16-HDoHE), 19,20-dihydroxy-4Z,7Z,10Z,13Z,16Z-docosapentaenoic acid (19,20-DiHDPa), 10,17-dihydroxy-4Z,7Z,11E,13Z,15E,19Z-docosahexaenoic acid (10,17-DiHDoHE), 7S,8R,17S-trihydroxy-4Z,9E,11E,13Z,15E,19Z-docosahexaenoic acid (RvD1), 7S,16R,17S-trihydroxy-4Z,8E,10Z,12E,14E,19Z-docosahexaenoic acid (RvD2), and docosahexaenoic acid (DHA). AMPP was obtained from Alchem Laboratories Corporation (Alachua, FL). HPLC-grade acetonitrile, methanol, chloroform, and water were purchased from Burdick & Jackson (Muskegon, MI). Glacial acetic acid and N, N-dimethylformamide were obtained from Sigma-Aldrich (St Louis, MO) while N-hydroxybenzotriazole and 3-(dimethylamino)propyl-ethyl carbodiimide hydrochloride were purchased from Advanced Chem Tech (Louisville, KY) and TCI America (Portland, OR), respectively. Ascentis Express C18 reverse phase high performance liquid chromatography (HPLC) columns (15cm × 2.1mm) were obtained from Supelco (Bellefonte, PA) while solid phase extraction Strata-X columns were purchased from Phenomenex (Torrance, CA).

Preparation of Standard Solutions and Quantitation

Stock solutions of reference standards were prepared at a concentration of 10-100 pg/μl in ethanol and stored in a brown glass vial under nitrogen at -80°C. The stock solutions were further diluted by ethanol to prepare standard working solutions. Appropriate amounts of internal standards of oxidized fatty acids in ethanol were injected into samples (e.g., biologic tissues and fluids) for quantitative analysis by ratiometric comparisons of the high mass

accuracy diagnostic product ion areas to authentic standards. All samples were flash frozen and stored at -180°C prior to polytron homogenization at $0-4^{\circ}\text{C}$.

Hepatic Tissue Samples Preparation

Oxidized fatty acids were analyzed in hepatic tissue obtained from either control C57BL/6 mice or in C57BL/6 mice in which iPLA₂ was globally deleted by removal of the active site in exon 5 by homologous recombination as described previously [60]. All animal breeding, care and sacrifice were carried out by protocols approved by the Animal Studies Committee at Washington University School of Medicine. Following sacrifice, livers were immediately removed, quickly rinsed twice in ice-cold PBS (pH7.4) solution, blotted, snap-frozen in liquid nitrogen and stored at -180°C until homogenization and extraction. Samples were polytron homogenized at $0-4^{\circ}\text{C}$ prior to extraction using a modified Bligh and Dyer procedure. Briefly, ~20 mg of hepatic tissue was extracted by addition of 2 ml of ice-cold MeOH/CHCl₃ (1:1 v/v containing 2% HAc) followed directly by addition of 2 μl of an antioxidant mixture (0.2 mg/ml BHT, 0.2 mg/ml EDTA, 2 mg/ triphenylphosphine, and 2 mg/ml indomethacin in a solution of 2:1:1 methanol/ ethanol/H₂O) and stable isotope internal standards (e.g., TXB₂-d₄, LTB₄-d₄, 13-HODE-d₄, 9,10-DiHOME-d₄, 12,13-DiHOME-d₄, PGE₂-d₄, and 12-HETE-d₈). Next, the samples were polytron homogenized (at $0-4^{\circ}\text{C}$ as described above), 1 ml of ice-cold H₂O was added and the resultant mixture was centrifuged at 15,000g for 15 min to separate the organic and aqueous phases. The CHCl₃ layer was transferred to a new tube while the aqueous layer was re-extracted by subsequent addition of 1 ml CHCl₃, vortexed and layers were separated by centrifugation at 15,000g for 15 min. The CHCl₃ extracts were combined, evaporated under a nitrogen stream and resuspended in 1 ml of 10% methanol/water for solid phase extraction.

Solid Phase Extraction

The sample extracts were immediately applied to a Strata-X solid phase extraction cartridge (30 mg/1ml) that had been previously preconditioned with 1 ml of methanol followed by sequential washing with 1 ml of 10% methanol/90% H₂O and 2 \times 1 ml of 5% methanol/95% H₂O. The oxidized fatty acid metabolites were eluted with 2 \times 0.5 ml washes with methanol containing 1% glacial acetic acid. The combined methanol washes were evaporated to dryness using a SpeedVac concentrator prior to charge-switch derivatization with N-(4-aminomethylphenyl)-pyridinium (AMPP).

Derivatization Reactions

Derivatization with AMPP was performed as previously described in detail by Gelb *et al.* [59]. In brief, 15 μl of ice-cold acetonitrile/N,N-dimethylformamide (4:1, v:v) and 15 μl of ice-cold 640 mM (3-(dimethylamino)propyl)ethyl carbodiimide hydrochloride in HPLC grade water were added to the residue in the sample vial. The vial was briefly vortexed and 30 μl of 5 mM N-hydroxybenzotriazole /15 mM AMPP in acetonitrile were added, vortexed and placed in a 60°C water bath for 30 min. Samples were then cooled down and kept in the autosampler rack at 4°C while queued for injection.

LC-ESI MS/MS Analysis

LC-ESI MS/MS analysis was performed using an LTQ-Orbitrap mass spectrometer (Thermo Scientific, San Jose, CA) equipped with a Surveyor HPLC system (Thermo Scientific, San Jose, CA). Briefly, the derivatized metabolites were separated on a C18 reversed phase column (Ascentis Express, 2.7 μm particles, 150×2.1 mm) at 23°C using a linear gradient of solvent A (0.1% glacial acetic acid in water) and solvent B (0.1% glacial acetic acid in acetonitrile) at a flow rate of 0.2 ml/min. The following solvent gradient program was used: 0.0-5.0 min, 25% B; 5.0-7.0 min, 25-35% B; 7.0-20.0 min, 35-60% B; 20.0-20.1 min,

60-100% B; 20.1-24.0 min, 100% B; 24.0-25.0 min, 100% B to 25% B and 10 min isocratic hold at 25% B. The sample injection volume was 10 μ l and the autosampler tray temperature was kept at 4°C throughout the analysis. Diagnostic product ion transitions were determined by full mass scans of targeted metabolites to identify informative transitions to monitor during chromatography. Quantitative analysis of the derivatized fatty acids metabolites was performed using LC-MS/MS by ratiometric comparisons of the high mass accuracy (< 5ppm) diagnostic product ions in samples to that of stable isotope internal standards.

Mass spectrometric analyses were carried out on an LTQ-Orbitrap mass spectrometer (Thermo Scientific, San Jose, CA). The LTQ ion source was operated in positive ion mode at sheath, auxiliary, and sweep gas flows (arbitrary units) of 40, 5, and 1, respectively. The capillary temperature was set to 275°C and the electrospray voltage was 4.0 kV. Capillary voltage and the tube lens voltage were set to 2 and 100 V, respectively. Resolving powers (at m/z 400 Th) of 30,000 in full scan mode and 7500 in MS/MS mode were used. The instrument was calibrated externally following the manufacturer's instructions and mass accuracy was within 3 ppm at mass values from m/z 130 to 2000. For MS/MS analyses, a normalized collision energy of 30% was applied, the activation time was set at 30 ms with an activation parameter $q = 0.25$. Collision induced dissociation of derivatized oxidized fatty acid metabolites was initially performed by direct infusion of standards and analysis of full mass scan product ion spectra to identify diagnostic transitions. The identified informative precursor and product ion transitions were monitored by isolating the precursor and product ions within a range of ± 1 Th. Product ions were collected in the C trap prior to injection into the Orbitrap for accurate mass analysis and ions within 5ppm of the exact mass of the targeted compound were quantified. AGC target settings for FT full mass scan, FT MS2 scan and ion trap MS2 scan were 5×10^5 , 2×10^5 and 1×10^4 , respectively. Data acquisition was performed using the Xcalibur operating system, Version 2.1 (Thermo Scientific).

Calibration Standards and Quantification

Data analysis was performed using Xcalibur Version 2.1 software. The linearity of the developed method was determined using calibration curves constructed from each of the authentic standards of the targeted metabolites. Metabolite quantitation was performed from ratiometric comparisons of accurate mass measurements corresponding to the accurate mass peak area of the metabolite to the accurate mass peak areas of stable isotope internal standards. Linear regression analysis determined r^2 values of 0.99 or greater for each metabolite.

Statistical Analysis

Statistical analysis was performed using the two-tailed Student's t test. P values of less than 0.05 were considered to be statistically significant. All data are reported as the mean \pm SEM unless otherwise noted.

Results and Discussion

Development of LC-ESI MS/MS Methods for the Analysis of Oxidized Metabolites of Linoleic Acid, Arachidonic Acid and Docosahexaenoic Acid

First, authentic standards of LA, AA and DHA metabolites were extracted, derivatized with AMPP and prepared for mass spectrometric analysis by solid phase extraction as described in "Materials and Methods." The RPHPLC separation of the AMPP-derivatized metabolites of LA, AA and DHA demonstrated excellent chromatographic characteristics during separation on a C18 reversed phase column at 23°C (Fig. 1). Next, each of the targeted oxidized fatty acid metabolites were analyzed by SRM in the LTQ and product ions resulting from diagnostic transitions were collected in the C trap. Product ions from each

monitored transition were injected into the Orbitrap for accurate mass determination (<5ppm) to minimize false positive interferences. Examination of each of the representative oxidized derivatized fatty acid metabolites demonstrated full mass product ion fragmentation patterns in the LTQ Orbitrap that were similar to those using QqQ instruments [59], but differed substantially in the relative abundance of each of the product ions produced. This result is consistent with the differing modes of fragmentation produced in ion traps which induce bond cleavage by RF excitation vs. fragmentation in QqQ instruments which induce fragmentation by collisional activation with inert gases.

The separation of AMPP charge-switch derivatized oxidized fatty acid products was optimized using a customized mobile phase gradient (described in “Materials and Methods”) which resulted in baseline resolution of the majority of targeted metabolites (Fig. 1). Diagnostic fragmentation products of the authentic standards of the targeted analytes were first identified using direct infusion and subsequently substantiated by fragmentation of column chromatographic eluents. Each of the oxidized fatty acids examined were either baseline resolved by RPHPLC or analytes that possessed overlapping chromatographic retention times were discriminated based on diagnostic SRM transitions. Ions corresponding to the monitored transitions were collected in the C trap and injected into the Orbitrap for accurate mass determination to minimize false positive interferences that are not related to the analyte of interest.

As previously discussed by Gelb *et al.* [59] the utility of the AMPP derivatization lies not only in its salutary effects of increased ionization efficiency from charge-switch derivatization that result in increased S/N, but moreover facilitates the accrual of multiple informative fragment ions that emanate from dual mechanisms of fragmentation of the AMPP tag. Specifically, the AMPP derivatization tag yielded markers for metabolites of derivatized fatty acids through monitoring the high mass accuracy fragmentation of the diagnostic ions present at m/z 183.0917 (benzenemethanimine pyridinium ion) and m/z 169.0886 (tolylpyridinium distonic radical ion) that can physically map the elution profile of candidate derivatized fatty acid metabolites as well as other metabolites containing a carboxylic group present during extraction (Scheme 1).

The salutary design of AMPP tag not only delivers markedly enhanced sensitivity but also results in concomitant fragmentations in both the analyte and tag portions as was envisioned using molecular orbital calculations performed by Gelb *et al.* [59]. As anticipated, this strategy works extremely well with multiple other oxidized fatty acid metabolites of great biologic significance. In some cases, chromatographic elution results in the incomplete resolution of metabolites with similar structures. To circumvent this difficulty, we identified diagnostic transitions that can be utilized to discriminate metabolites that are incompletely resolved. For example, 16- and 17-HDoHE as well as 10- and 11-HDoHE possessed overlapping chromatographic elution profiles which were discriminated by high mass accuracy determination of diagnostic product ions. Specifically, the unique transition in 16-HDoHE (m/z 511 m/z 401, Fig. 2A) was not present in the coeluting 17-HDoHE (m/z 511 m/z 441, Fig. 2B) thus allowing discrimination between each coeluting metabolite. As proposed in Scheme 2, formation of the ion at m/z 401.2587 likely results from the loss of 2, 4-heptadienal ($C_7H_{10}O$) from 16-HDoHE-AMPP).

Similarly, the product ion at m/z 441.2537 was likely generated from the loss of 2-pentene (C_5H_{10}) from 17-HDoHE-AMPP. The unique transition of m/z 511 m/z 321 for 10-HDoHE (Fig. 3A) was selected to discriminate the coeluting metabolite 11-HDoHE for which the identified diagnostic transition of m/z 511 m/z 361 was employed (Fig. 3B). The proposed fragmentation pathways for formation of the m/z 321.1961 and m/z 361.1911

ions from 10- and 11-HDoHE AMPP amides, due to loss of $C_{13}H_{18}O$ and $C_{11}H_{18}$, respectively, are shown in Scheme 3.

Accurate mass determination of product ions derived from SRM can substantively reduce false positive identifications. We and others [37] have identified difficulties in metabolite identification and quantitation using precursor and product ions with 1Th selection windows which have the potential to contain multiple different isobaric analytes/interferences in addition to the analyte of interest. In many cases, utilization of SRM results in relatively pure product ions as assessed by accurate mass determination. For example over 99% of murine 12-HETE in serum was authenticated by product ion accurate mass analysis thereby substantiating the integrity of the measured analyte using conventional procedures (Fig. 4A and B). In contrast, when a sample of mouse serum was analyzed for PGE_1 using the identified SRM transition of m/z 521 m/z 493 eluting with retention time 9.3 min (Fig. 4C and D), the resulting product ions from SRM demonstrated that although the majority of the product ions at the monitored transition were likely authentic PGE_1 (exact mass=493.3425), two contaminating ions were also present at m/z 492.8650 and 492.5438 (each markedly outside of the 5 ppm accurate mass selection window). Calculations of the combined areas of the two contaminating peaks to that of authentic PGE_1 demonstrated that an error of 27% was introduced by the presence of the contaminating ions. Accurate mass from SRM product ion determination eliminates the interfering ions at m/z 492.8650 and 492.5438, thereby providing more accurate quantitative results. Another example of the utility of accurate mass product ion analysis is demonstrated in the false positive identification of LTB_4 from mesenteric arteries (Fig. 4E and 4F). Using accurate mass analysis of the product ions generated during the monitored SRM transition (m/z 503 m/z 391) clearly eliminated the interference of the contaminating ion present at m/z 391.3361. This erroneous assignment of this ion contributes 44% of the total peak area that would be obtained using unit mass resolution during conventional isobaric SRM analysis. These results were not due to alterations resulting from the changes in the mode of ionization (negative to positive) or the derivatization reaction since false positive identification of 13,14-dihydro15keto PGF_2 were previously documented in underivatized samples using RPHPLC and LC-MS/MS with SRM in the negative ion mode. Specifically, Volmer *et al.* reported that the ion peak at m/z 113 which was identified as 13,14-dihydro15keto PGF_2 in chronic lymphocytic leukemia cells using SRM (m/z 353 m/z 113) with unit mass resolution represented a false positive identification (Fig. 2;[37];). We have observed multiple similar errors in identification and quantitation using conventional SRM analysis with isobaric product ion determination which can be deconvoluted using product ion accurate mass analysis. In many cases this problem is more significant, affecting not only quantitation, but leading to false identification of the observed ion peak. For example (Fig. 5), when a sample of a murine mesenteric artery was analyzed for LTB_4 (m/z 503 m/z 391, retention time 10.45min), the results demonstrated abundant product ions resulting from this transition leading to the conclusion that LTB_4 was present in mesenteric arteries. However, the accurate mass of the monitored product ion does not correspond to that of the product ion of LTB_4 standard (i.e. m/z 391.2016) while the spiked authentic deuterated internal standard LTB_4 was present in the anticipated amount as assessed by accurate mass (product ion accurate mass 393.2142 resulting from the transition of tetra deuterated LTB_4 m/z 507 m/z 393) (spectrum not shown). Thus, high mass accuracy determination eliminates the false positive identification of LTB_4 that cannot be revealed using SRM with isobaric mass resolution. Thus, one approach to minimize false positive interferences is the use of high mass accuracy analysis of product ions collected during the monitored SRM transitions. Product ions which do not match the accurate mass of their isobarically assigned analytes do not represent the moiety of interest and instead may be generated in a variety of ways from different known and unknown metabolites, their fragmentation products or

products from sequential fragmentations to give rise to unanticipated product ions during SRM analysis that do not result from the analyte of interest.

Recovery of Metabolites during Solid Phase Extraction, Linearity of Quantitation and the Limits of Quantitation Using High Mass Accuracy Product Ion Analysis

Recoveries of metabolites from Strata-X cartridges were measured following the extraction of samples of each metabolite by radiometric comparison to high mass accuracy measurements of stable isotope internal standard. Extraction yields were obtained by comparing the LC-ESI MS/MS peak integrals to those obtained from AMPP derivatized samples that were directly injected onto the LC column. The extraction recoveries for each metabolite were excellent (Table 1). Using the optimized conditions in “Materials and Methods” as described above, the sensitivity, linearity and accuracy of the developed method were examined. Table 1 summarizes the linearity of quantitation, lower limit of quantitation, dynamic range and accuracy of quantitation. The limits of quantitation for LA and AA oxidized metabolites were between 0.05 to 1.0 pg and linear calibration ranges up 300 to 1000 pg were observed. The limits of quantitation for DHA oxidized metabolites were between 0.5 to 6.0 pg with linear calibration ranges up 500 to 1000 pg. Linear regression analysis determined correlation coefficient r^2 values of 0.99 or greater for each analyte. Analytes were quantified using standard curves created through high mass accuracy (<5 ppm) determinations of peak areas of standards by radiometric comparisons to high mass accuracy product ion determinations of metabolites. The accuracy was calculated as the percentage of measured value to the expected values in quality control samples. Table 1 lists the figures of merit for the developed high mass accuracy charge switch approach for oxidized fatty acid metabolites.

Analysis of the Oxidized Metabolites of LA, AA and DHA in Hepatic Tissues from Wide-Type and iPLA₂ Genetic Knockout Mice

To demonstrate the utility of this approach in biologic samples, oxidized metabolites of LA, AA and DHA in hepatic tissues from wild-type mice and iPLA₂^{-/-} mice were examined. Using the developed method, we were able to identify and quantitate 37 targeted metabolites derived from LA, AA and DHA in murine liver (Fig. 6). The metabolites from linoleic acid (Fig. 6A) revealed significant decreases for 9(10)- and 12(13)-EpOME as well as their epoxide hydrolase products 9,10-DiHOME and 12,13-DiHOME ($p < 0.05$). In contrast, the hepatic content of 9- and 13-HODE and their allylic hydroxyl oxidation products (9- and 13-oxoODE) were unchanged in WT vs. iPLA₂^{-/-} mice. Analysis DHA metabolites (Fig. 6B) demonstrated specific alterations in hepatic docosanoids after genetic ablation of iPLA₂. These included a 43% decrease of 19,20-DiHDPA which was the major DHA metabolite in liver, a 68% decrease in RvD2 and 17-HDoHE, a 52% decrease in 14-HDoHE, and a 55% decrease in 4-HDoHE. The other measured DHA oxidized molecular species were statistically unchanged in comparisons between iPLA₂^{-/-} mice and their WT littermates. Additionally, oxidized AA metabolites revealed significant decreases in the content of 5-, 8-, 11-, 12- and 15-HETEs in the iPLA₂^{-/-} mouse. In contrast, the content of EETs was unchanged in hepatic tissue of the iPLA₂^{-/-} mouse (Fig. 6C). Significant changes were also found in prostaglandins PGD₂, TXB₂, PGF₂, and 6keto-PGF₁. We also measured the content of non-esterified AA, LA and DHA in WT murine liver (AA, 14.1 ± 2.5 ng/mg; LA, 21.7 ± 4.2 ng/mg; and DHA, 34.8 ± 2.1 ng/mg) as well as liver from iPLA₂^{-/-} mice (AA, 10.9 ± 1.3 ng/mg; LA, 12.9 ± 1.1 ng/mg; and DHA, 24.8 ± 1.8 ng/mg). The results demonstrated the decreased polyunsaturated fatty acids in iPLA₂^{-/-} liver and indicate the important role of iPLA₂ in murine liver lipid signaling and metabolism.

Conclusions

The identification and quantitation of diminutive amounts of oxidized fatty acid signaling metabolites in biologic tissues is essential to understanding the roles of lipid 2nd messengers in biologic systems. However, the identification and quantitation of the diminutive amounts of oxidized non-esterified fatty acids in non-inflammatory cells or exudates is challenging task. In this study, we have utilized a high mass accuracy charge-switch approach to gain deep penetrance into the extremely low abundance regime of oxidized fatty acids that enables their identification, improves quantitation and can be effectively multiplexed as the acquisition rate of tandem mass spectra in next generation instrumentation increases. Through the exquisite sensitivity of charge-switch derivatization in conjunction with the judicious use of informative transitions whose integrity is substantiated by high mass accuracy product ions analyses, new insights into the roles of oxidized fatty acids in cellular biology can be accrued.

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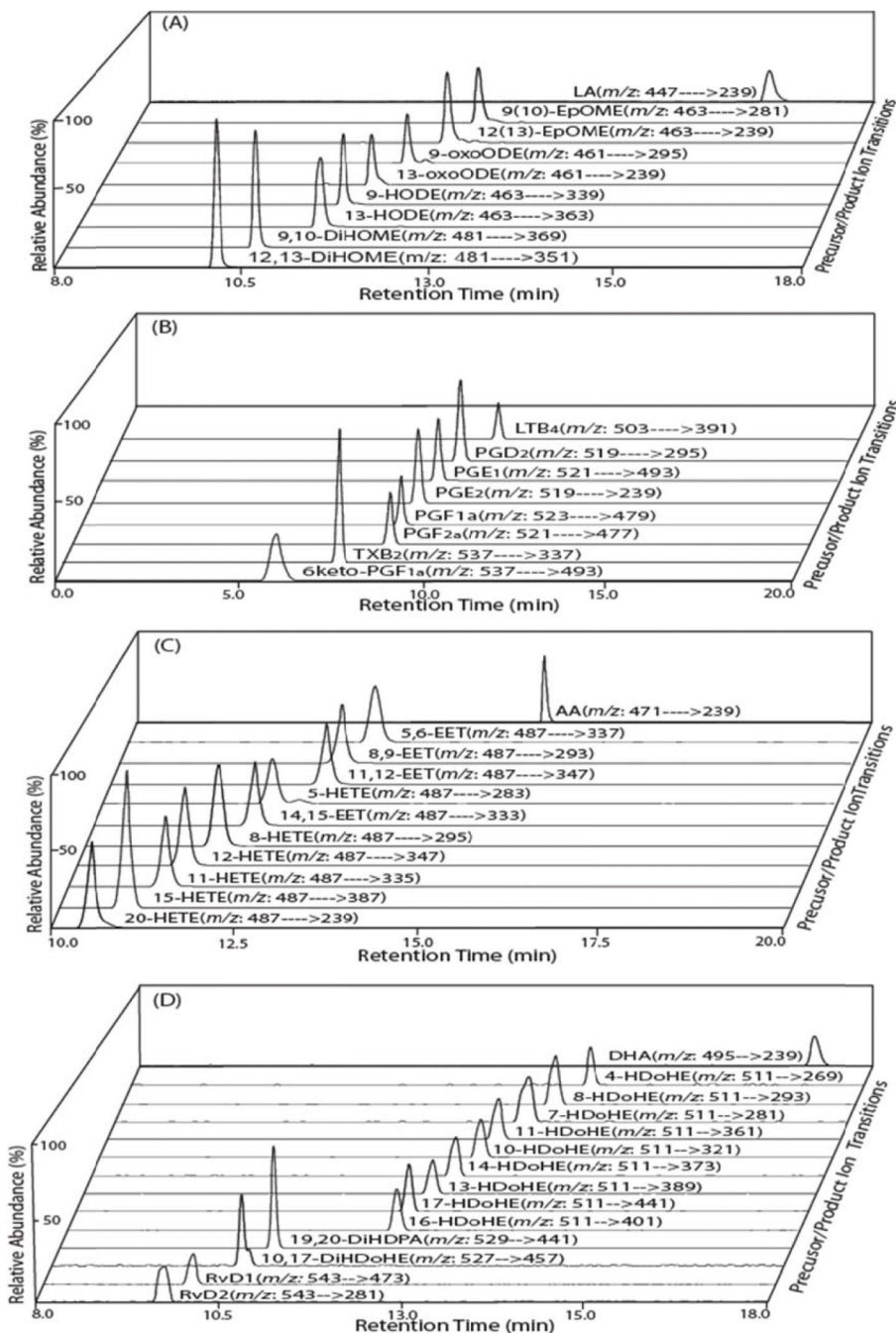


Figure 1. Selected reaction monitoring (SRM) total ion current chromatograms of the targeted oxidized metabolites of linoleic, arachidonic and docosahexaenoic acids

The indicated fatty acid metabolite standards were derivatized with N-(4-aminomethylphenyl)-pyridinium (AMPP) and measured by LC/MS/MS via SRM in the positive ion mode following chromatographic separation using a C18 reverse phase HPLC column as described in “Materials and Methods”. Representative SRM chromatograms for linoleic acid (LA) and the indicated oxidized LA metabolites (A); arachidonic acid (AA) and the indicated oxidized AA metabolites (B and C); and docosahexaenoic acid (DHA) and the indicated oxidized DHA metabolites (D) are shown with their retention times and the informative SRM transitions.

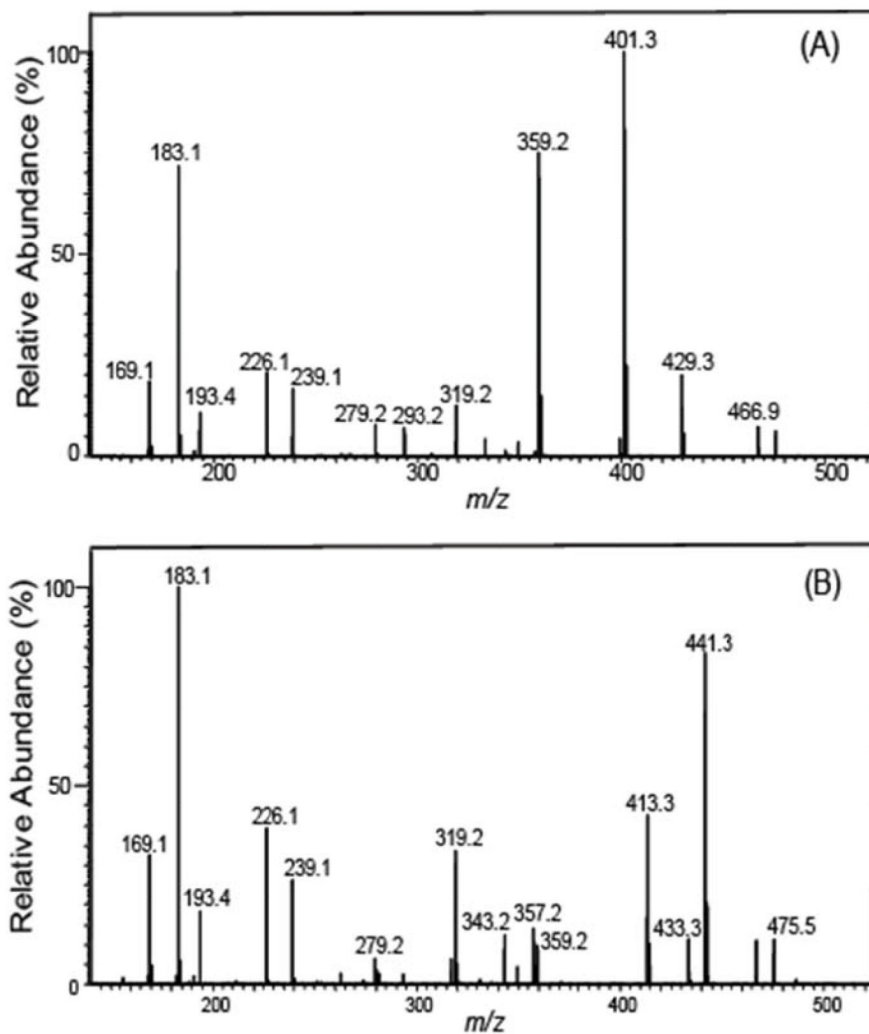


Figure 2. Full mass product ion scans used to discriminate 16- and 17-HDoHE derivatized with N-(4-aminomethylphenyl)-pyridinium (AMPP)

(A) Mass spectrum of the CID of 16-DHoHE-AMPP exhibiting a prominent diagnostic fragment ion at m/z 401 which was utilized for the SRM transition based identification and quantitation of 16-DHoHE. (B) Mass spectrum of the CID of 17-DHoHE-AMPP displaying an informative fragment ion at m/z 441 which was utilized for the SRM transition based identification and quantitation of 17-DHoHE.

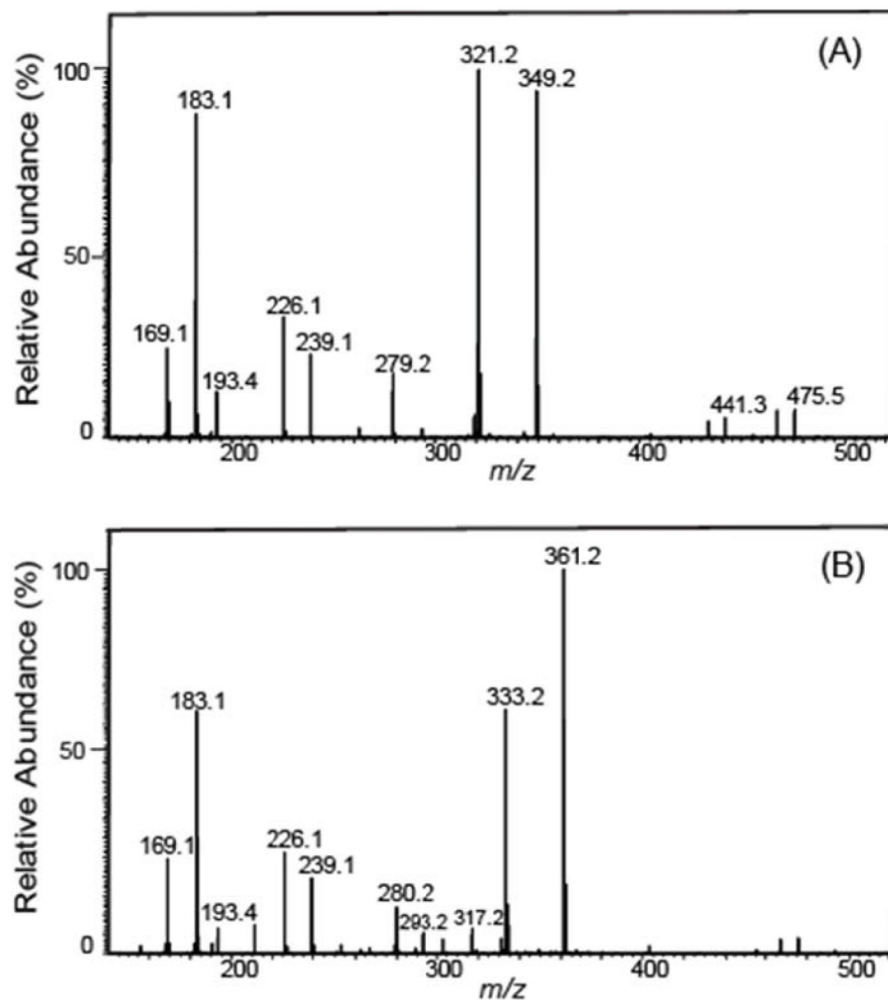


Figure 3. Full mass product ion scans used to discriminate 10- and 11-HDoHE derivatized with N-(4-aminomethylphenyl)-pyridinium (AMPP) (A) Mass spectrum of the CID of 10-HDoHE-AMPP exhibiting a diagnostic fragment ion at m/z 349 which was selected for the SRM transition based identification and quantitation of 10-HDoHE. (B) Mass spectrum of the CID of 11-HDoHE-AMPP displaying an informative fragment ion at m/z 361 which was selected for SRM transition based identification and quantitation of 11-HDoHE.

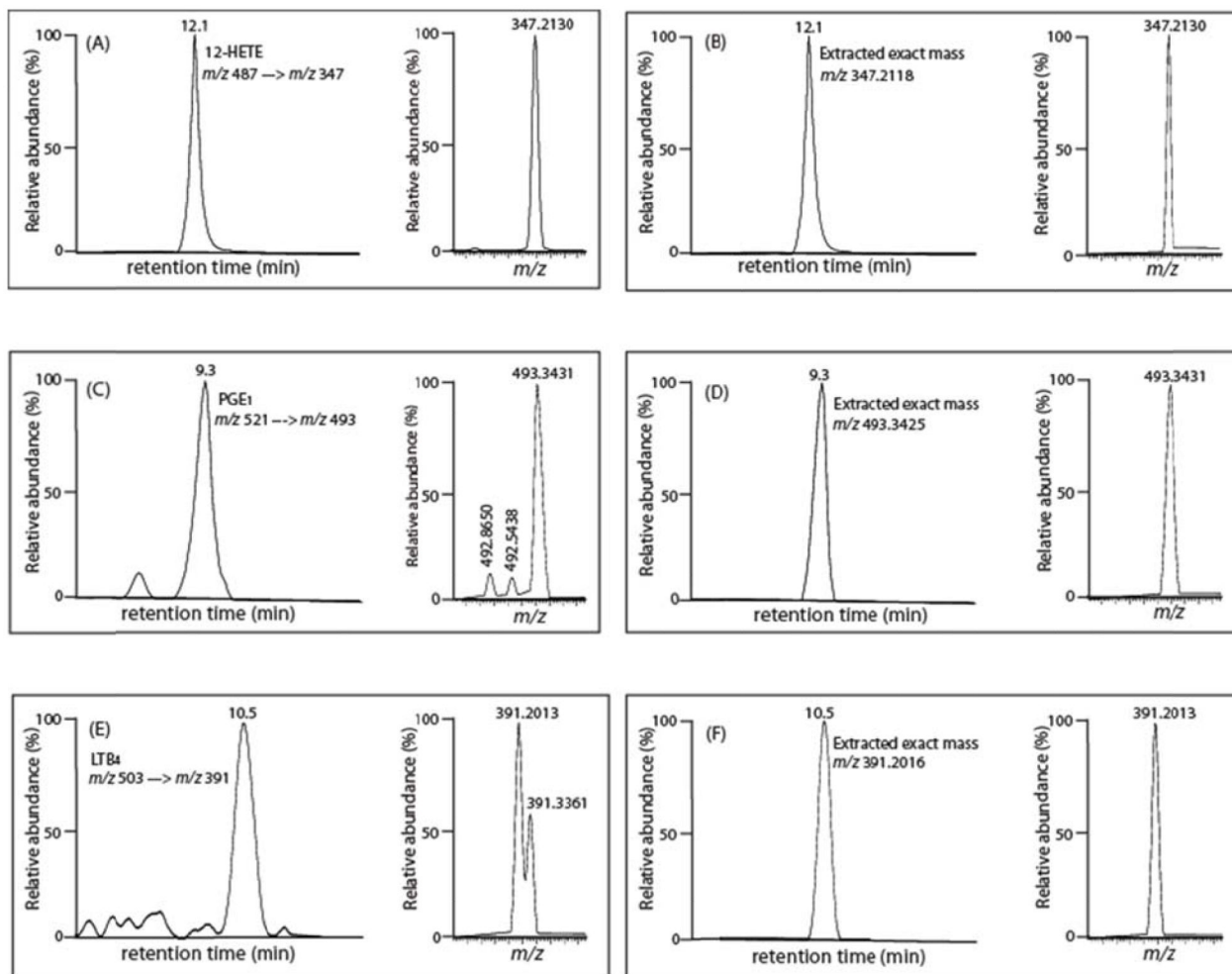


Figure 4. Comparison of quantitative methods for measuring selected eicosanoids in mouse serum using mass unit resolution of product ion SRM and accurate mass product ion SRM (A) Peak retention time profile and SRM transition (m/z 487 \rightarrow m/z 347) of 12-HETE-AMPP with mass unit resolution; (B) Peak retention time profile and high mass accuracy extracted ion chromatogram of the product ion at m/z 347.2188 (within 5 ppm); (C): Peak retention time profile and SRM transition (m/z 521 \rightarrow m/z 493) of PGE₁-AMPP with mass unit resolution; (D) Peak retention time profile and high mass accuracy extracted ion chromatogram of the product ion at m/z 493.3425 (within 5 ppm); (E) Peak retention time profile and SRM transition (m/z 503 \rightarrow m/z 391) of LTB₄-AMPP with mass unit resolution; (F) Peak retention time profile and high mass accuracy extracted ion chromatogram of the product ion at m/z 391.2016 (within 5 ppm).

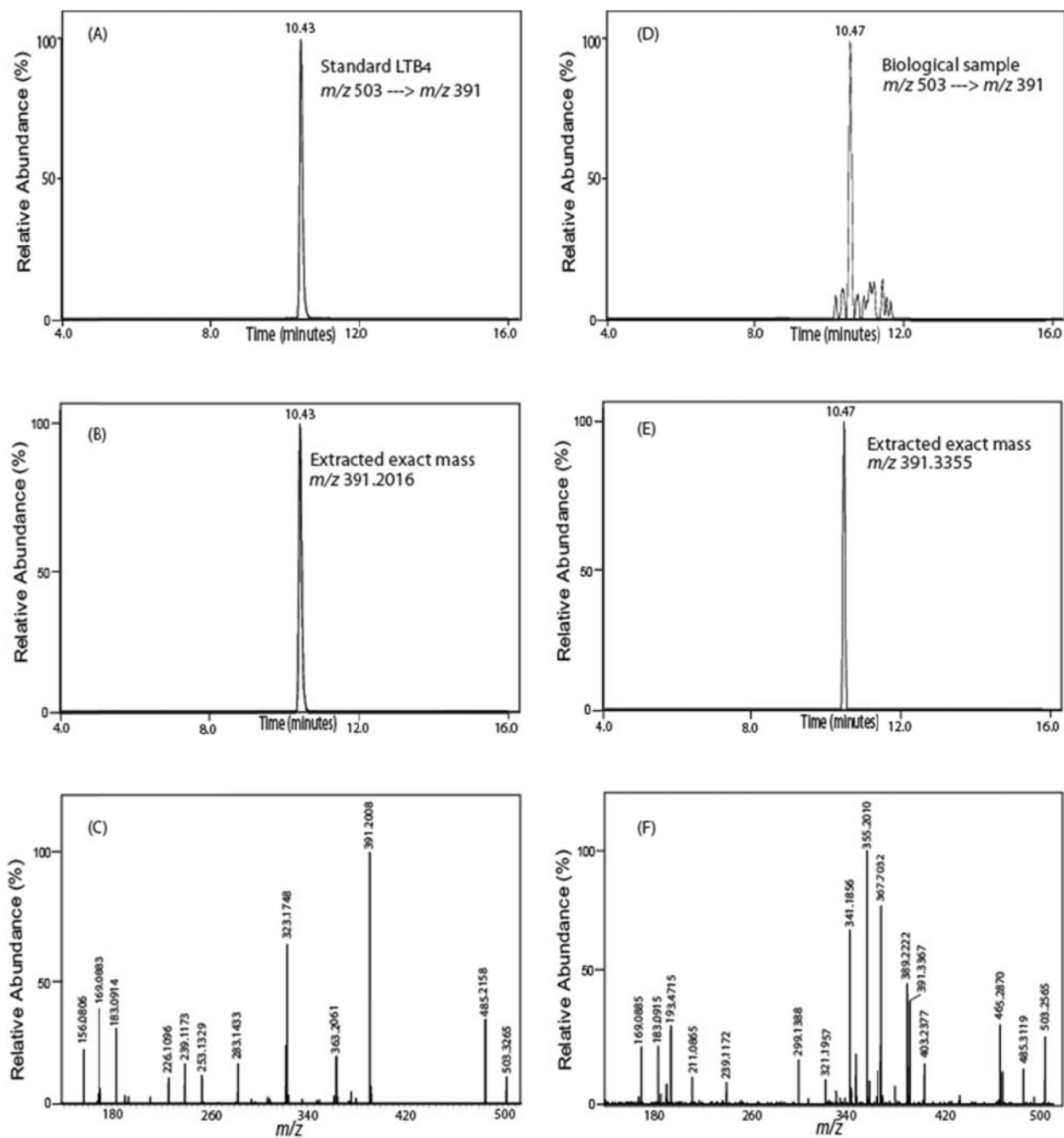


Figure 5. Determination of the false positive identification of LTB₄ in a biologic sample utilizing SRM with accurate product ion mass analyses

(A) Peak retention time profile of LTB₄-AMPP standard utilizing SRM (m/z 503 → m/z 391); (B) high mass accuracy extracted ion chromatogram of the product ion, m/z 391.2016 (within an error of 2 ppm) of the standard LTB₄-AMPP; (C) corresponding full MS/MS spectrum of the LTB₄-AMPP standard; (D) Peak retention time profile indicative of LTB₄-AMPP obtained from murine mesenteric arterial tissue utilizing the SRM transition (m/z 503 → m/z 391); (E) high mass accuracy extracted ion chromatogram of the product ion, m/z 391.3355, present in the biologic sample which does not correspond to the accurate mass of the product ion of LTB₄-AMPP thereby revealing a false positive identification if only mass

unit resolution is utilized (F). The MS/MS spectra of D and E confirm substantial differences between the LTB₄ standard and compounds in the biological sample.

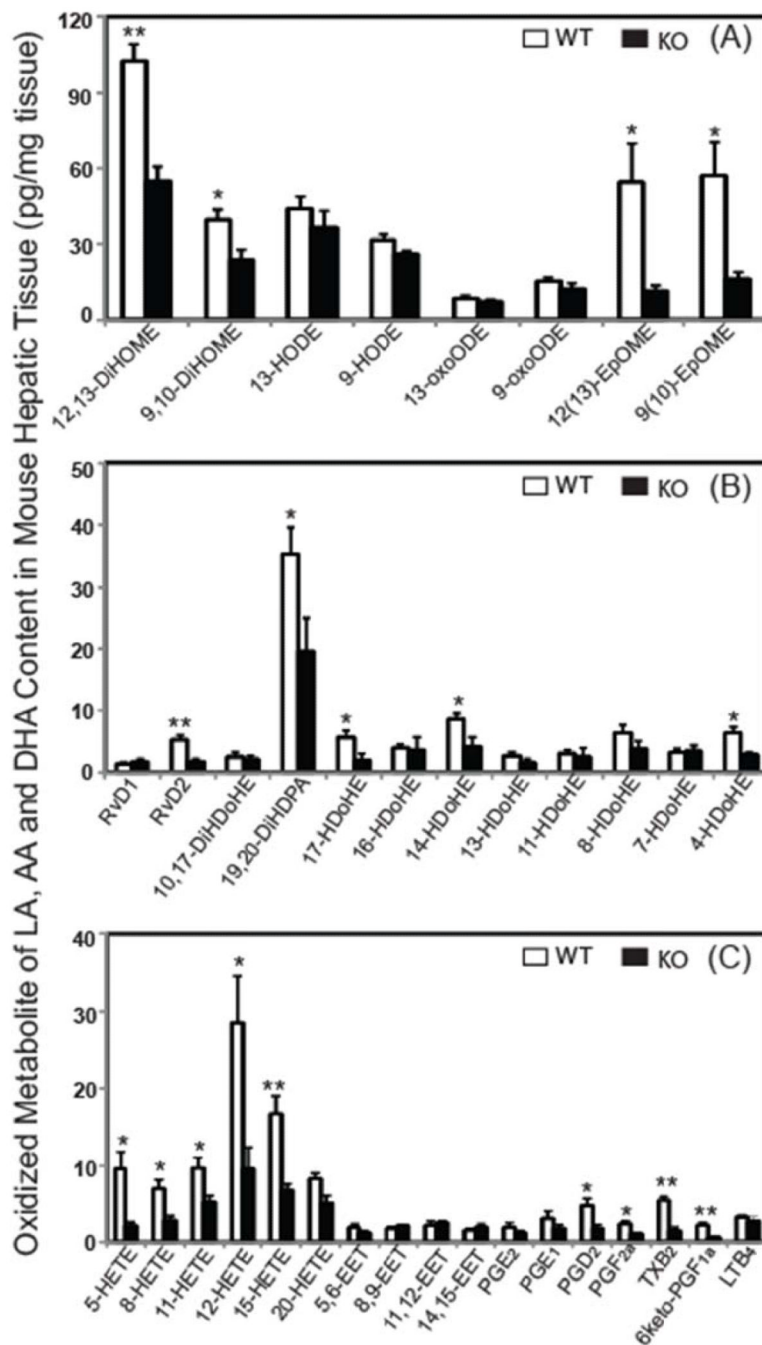
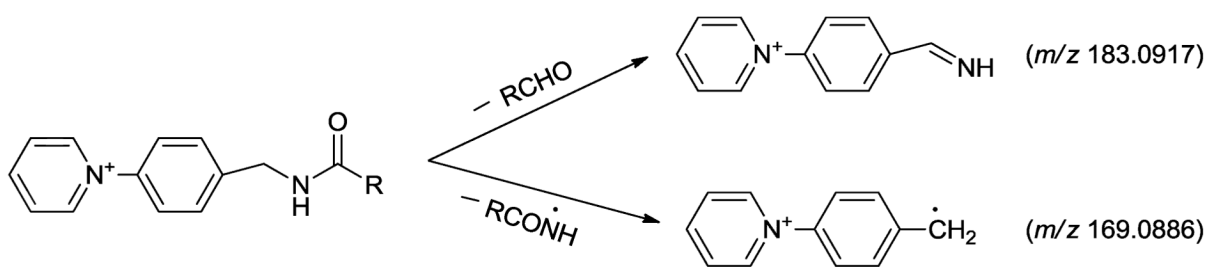
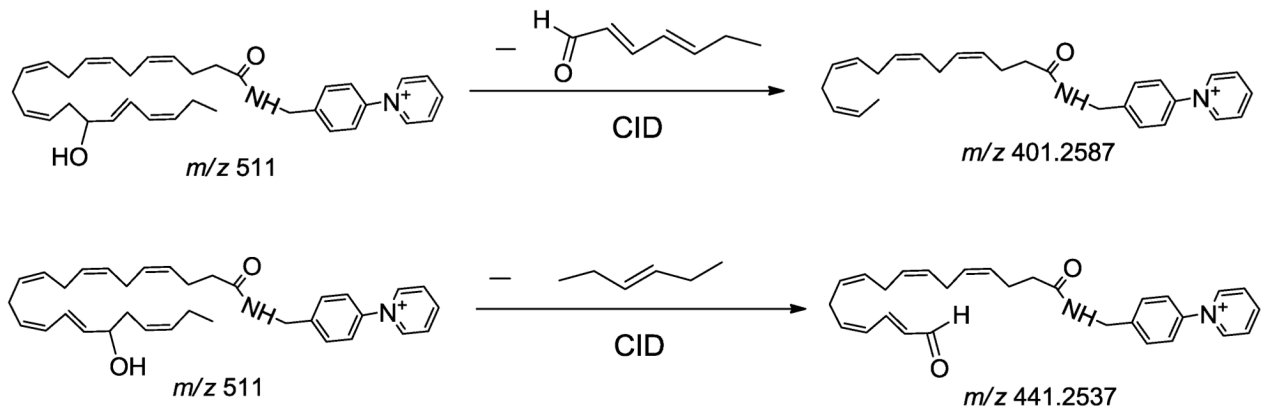


Figure 6. Oxidized metabolites of linoleic acid, arachidonic acid and docosahexaenoic acid in hepatic tissues from wild-type (WT) and iPLA₂ Knockout (KO) mice

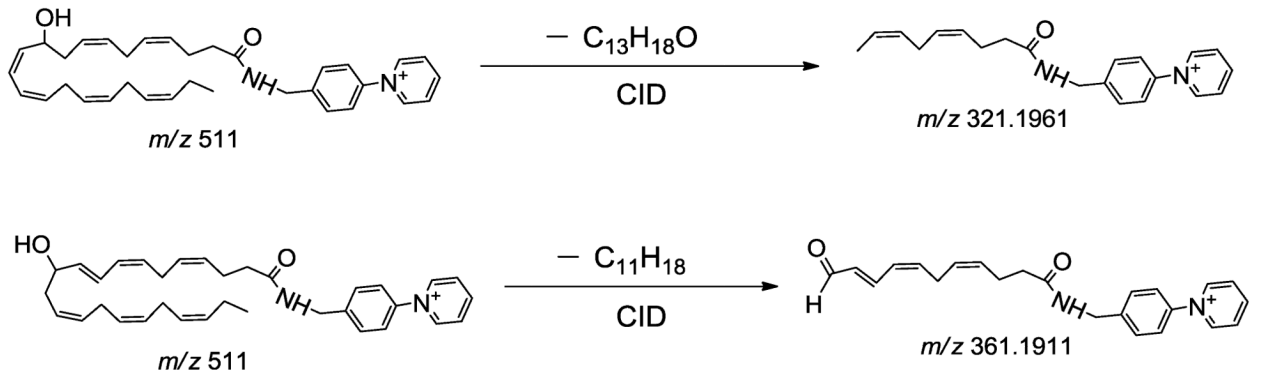
Oxidized fatty acid metabolites from murine hepatic tissue were extracted into solvent, isolated by solid phase extraction, derivatized with AMPP and quantitated by LC/MS/MS via selected reaction monitoring (SRM) in the positive ion mode following separation of molecular species using a reverse phase column as described in “Materials and Methods”. Significant decreases in the production of multiple identified oxidized metabolites of LA, DHA and AA as a result of genetic ablation of iPLA₂ are displayed in panels A, B and C, respectively. Values presented are the mean \pm S.E. and comparisons were made using Student’s *t*-test (n=4). *: P < 0.05 level, **: P < 0.01 level.



Scheme 1.



Scheme 2.



Scheme 3.

Table 1
**Relative response factor (R/R_0)^a, linearity of measurement, correlation coefficient (r^2),
 limits of quantitation (LOQs)^b, extraction yield, and method accuracy^c analyses**

Analytes	Extraction Yield (%)	Accuracy (% ± SD)	R/R ₀	LoQ (pg)	Calibrated range (pg)	Correlation coefficient (r^2)	Internal standard
9-HODE	98	95.0 ± 8.5	1.10 ± 0.11	0.05	0.05–300	0.9977	13-HODE-d ₄
13-HODE	90	105.0 ± 12.5	1.08 ± 0.08	0.07	0.07–300	0.9956	13-HODE-d ₄
9-OxoODE	84	86.2 ± 5.0	0.72 ± 0.05	0.5	0.5–1000	0.9990	13-HODE-d ₄
13-OxoODE	88	97.8.0 ± 7.3	0.88 ± 0.12	1.0	1.0–1000	0.9987	13-HODE-d ₄
9,10-DiHOME	80	103.0 ± 15.0	1.05 ± 0.15	0.08	0.08–300	0.9991	9,10-DiHOME-d ₄
12,13-DiHOME	77	94.2.0 ± 3.4	0.99 ± 0.10	0.04	0.04–300	0.9989	12,13-DiHOME-d ₄
9(10)-EpOME	76	91.5 ± 9.5	0.78 ± 0.10	0.5	0.5–1000	0.9999	9,10-DiHOME-d ₄
12(13)-EpOME	80	95.0 ± 8.5	0.75 ± 0.11	0.5	0.5–1000	0.9945	9,10-DiHOME-d ₄
Resolvin D1	75	94.1 ± 10.4	0.40 ± 0.10	5.0	5.0–1000	0.9998	PGE ₂ -d ₄
Resolvin D2	79	110.0 ± 14.8	0.55 ± 0.12	6.0	6.0–1000	0.9937	PGE ₂ -d ₄
10,17-DiHDoHEA	85	98.0 ± 10.3	0.40 ± 0.08	0.5	0.5–500	0.9960	PGE ₂ -d ₄
19,20-DiHDKPA	103	97.2 ± 7.8	5.80 ± 0.75	0.5	0.5–500	0.9948	PGE ₂ -d ₄
17-HDoHE	82	95.2 ± 10.1	1.20 ± 0.20	2.0	2.0–1000	0.9950	12-HETE-d ₈
16-HDoHE	78	99.0 ± 3.5	1.25 ± 0.16	2.0	2.0–1000	0.9984	12-HETE-d ₈
14-HDoHE	78	93.8 ± 8.8	1.18 ± 0.10	1.0	1.0–1000	0.9989	12-HETE-d ₈
13-HDoHE	83	105.0 ± 11.2	1.32 ± 0.20	2.0	2.0–1000	0.9995	12-HETE-d ₈
11-HDoHE	77	101.0 ± 8.5	1.15 ± 0.22	2.0	2.0–1000	0.9966	12-HETE-d ₈
10-HDoHE	81	88.0 ± 7.8	1.40 ± 0.08	2.0	2.0–1000	0.9984	12-HETE-d ₈
8-HDoHE	82	90.2 ± 9.9	1.05 ± 0.15	1.0	1.0–1000	0.9919	12-HETE-d ₈
7-HDoHE	79	102.0 ± 8.5	1.20 ± 0.11	1.0	1.0–1000	0.9963	12-HETE-d ₈
4-HDoHE	83	108.0 ± 4.5	1.00 ± 0.08	1.0	1.0–1000	0.9994	12-HETE-d ₈
6keto-PGF ₁	85	106.3 ± 8.5	0.95 ± 0.05	0.1	0.1–5000	0.9984	TXB ₂ -d ₄
TXB ₂	88	100.2 ± 3.5	1.03 ± 0.03	0.1	0.1–500	0.9990	TXB ₂ -d ₄
PGF ₂	78	98.1 ± 7.1	0.91 ± 0.03	0.4	0.4–500	0.9988	PGE ₂ -d ₄
PGF ₁	80	108.2 ± 5.2	2.10 ± 0.03	0.1	0.1–500	0.9917	PGE ₂ -d ₄
PGE ₂	77	99.1 ± 4.2	1.02 ± 0.03	0.3	0.3–500	0.9981	PGE ₂ -d ₄
PGE ₁	73	103.1 ± 7.1	1.42 ± 0.11	0.2	0.2–500	0.9972	PGE ₂ -d ₄
PGD ₂	73	92.1 ± 10.2	0.66 ± 0.07	0.5	0.5–1000	0.9981	PGE ₂ -d ₄
LTB ₄	65	98.8 ± 9.1	1.02 ± 0.02	0.7	0.7–1000	0.9992	LTB ₄ -d ₄
20-HETE	76	105.1 ± 3.1	2.20 ± 0.10	0.3	0.3–500	0.9977	12-HETE-d ₈
15-HETE	68	97.2 ± 9.2	1.10 ± 0.18	0.6	0.6–1000	0.9953	12-HETE-d ₈
12-HETE	66	99.9 ± 8.1	1.00 ± 0.05	0.7	0.7–1000	0.9969	12-HETE-d ₈

Analytes	Extraction Yield (%)	Accuracy (% \pm SD)	R/R ₀	LoQ (pg)	Calibrated range (pg)	Correlation coefficient (r ²)	Internal standard
11-HETE	73	97.9 \pm 4.2	1.32 \pm 0.13	0.5	0.5–1000	0.9973	12-HETE-d ₈
8-HETE	80	103.1 \pm 5.5	1.50 \pm 0.12	0.5	0.5–1000	0.9972	12-HETE-d ₈
5-HETE	70	108.4 \pm 7.2	0.82 \pm 0.08	0.8	0.8–1000	0.9999	12-HETE-d ₈
14,15-EET	75	94.3.1 \pm 4.4	0.91 \pm 0.11	0.8	0.8–1000	0.9989	12-HETE-d ₈
11,12-EET	80	98.9 \pm 6.7	0.80 \pm 0.08	0.8	0.8–1000	0.9982	12-HETE-d ₈
8,9-EET	79	101.0 \pm 9.9	0.75 \pm 0.14	0.8	0.8–1000	0.9991	12-HETE-d ₈
5,6-EET	85	106.2 \pm 7.7	0.95 \pm 0.08	0.5	0.5–1000	0.9990	12-HETE-d ₈

^aRelative response factor (R/R₀) was calculated as the ratio of an equivalent amount of analyte (R) to its internal standard (R₀) and expressed as the mean \pm standard deviation (N=3).

^bLOQ was defined as 3 \times the limit of detection which is expressed as pg injected on column resulting an signal to noise ratio of 3:1.

^cMethod accuracies were calculated as the percent deviation from the expected concentrations and expressed as the mean \pm standard deviation (n= 3 experiments).