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Pharmacology of Iron Transport

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Abstract

Elucidating the molecular basis for the regulation of iron uptake, storage, and distribution is necessary to understand iron homeostasis. Pharmacological tools are emerging to identify and distinguish among different iron transport pathways. Stimulatory or inhibitory small molecules with effects on iron uptake can help characterize the mechanistic elements of iron transport and the roles of the transporters involved in these processes. In particular, iron chelators can serve as potential pharmacological tools to alleviate diseases of iron overload. This review focuses on the pharmacology of iron transport, introducing iron transport membrane proteins and known inhibitors.

Keywords

hepcidin; hemochromatosis; transferrin; DMT1; NTBI

INTRODUCTION

Iron is central to DNA and ATP synthesis as well as oxygen transport and a variety of other critical enzymatic and nonenzymatic processes. Iron's catalytic utility arises from its two oxidation states, which interconvert through one-electron oxidation-reduction reactions. However, ferric (Fe^{3+}) iron is virtually insoluble in aqueous solution at neutral pH (concentrations cannot exceed 10^{-17} M), and ferrous (Fe^{2+}) iron is highly toxic. If left free in solution, Fe^{2+} reacts with molecular oxygen and forms the potent hydroxyl radical (OH^{\bullet}) through Fenton chemistry (1, 2). The reactive oxygen species (ROS) generated can then cause extensive damage to DNA, proteins, and lipids (3). Because of its redox-active nature and insolubility under physiological conditions, the transport of iron is complex. Proteins or small-molecule chelators are often necessary to shuttle iron within cells and between tissues of the human body.

Both surplus and deficiency of this transition metal lead to human pathologies associated with oxidative stress, iron deficiency anemia, or impaired immune function. Maintenance of systemic iron balance in the body requires tight regulation of iron homeostasis (3, 4). There does not appear to be a regulated physiologic pathway to excrete excess iron, and the primary mechanism of controlling iron levels in the body is thought to be the regulation of its intake (5). Elucidating the molecular basis for the regulation of mammalian iron uptake, storage, and distribution is necessary to understand iron homeostasis. Unfortunately, the lack

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of pharmacological tools has hampered progress in identifying the pathways, regulatory mechanisms, and connections among iron transport networks (Figure 1). Nonetheless, emerging technologies such as chemical genetic screening have led to the discovery of small-molecule compounds that modulate different pathways of cellular iron uptake. This review focuses on the pharmacology of iron transport in mammalian systems, introducing the membrane proteins involved and some known transport inhibitors.

IRON TRANSPORTERS AND MECHANISMS OF IRON UPTAKE

Transferrin and Transferrin Receptors

Iron is acquired by most mammalian cells through receptor-mediated endocytosis of the serum protein transferrin (Tf). Diferric Tf binds to the Tf receptor-1 (TfR1) on the cell surface at the neutral pH of the blood. TfR1 is a homodimer of 90-kDa subunits in a complex with the major histocompatibility complex (MHC) class I-like molecule HFE and β_2 -microglobulin (6). HFE defects are associated with hereditary hemochromatosis (HH) (7). Whereas TfR1 is ubiquitously expressed, its closely related homolog TfR2 has a more restricted distribution and predominates in the liver (8, 9). Defects in TfR2 are also associated with HH (10). Once internalized, TfR1-Tf is delivered to endosomes, wherein iron is released owing to the low pH of these endosomal compartments. The acidic environment stabilizes the binding of apotransferrin (apo-Tf) to its receptor (11), and the ligand-receptor complex recycles back to the cell surface, where apo-Tf dissociates from the receptor at neutral pH (12). Mechanistically, transport of iron released from Tf across endosomal membranes involves the reduction of Fe^{3+} to Fe^{2+} (13–15), a function provided by the ferrireductase Steap3 (16). In the so-called kiss-and-run hypothesis, endosomes bearing Tf directly deliver cargo iron to mitochondria for heme biosynthesis (17). Still other membrane transporters may transfer iron out of the endosomal-lysosomal system directly into the labile iron pool of the cytosol, including divalent metal transporter-1 (DMT1) (18), ZRT/IRT-like protein 14 (Zip14) (19), and transient receptor potential mucolipin 1 (TRPML1) (20).

Divalent Metal Transporter-1

DMT1 (Slc11a2) was first identified to be involved in iron assimilation because defects in its gene produce microcytic anemia in the *mk* mouse, a genetic model with defective dietary iron absorption (21). DMT1, a member of the Nramp family of metal transporters (22), is a 62-kDa protein that has 12 predicted membrane-spanning domains with intracellular N and C termini (23). The functional activity of DMT1 has been best characterized using exogenous expression in *Xenopus* oocytes (24). DMT1 is a proton-coupled symporter with a stoichiometry of 1 Fe^{2+} :1 H^+ and with apparent affinities of 6 and 1–2 μM , respectively (24). Membrane-spanning domains 1 and 6 play a key role in proton-dependent metal transport (25–27). The transporter interacts with several different divalent metals, and it influences both manganese and iron metabolism (28). DMT1 protein and mRNA are most abundant in the duodenum, consistent with a function in intestinal iron absorption (24, 29). Reduction of Fe^{3+} at the brush border by duodenal cytochrome B precedes translocation of Fe^{2+} by DMT1 across the apical membrane (30).

A remarkable link between intestinal iron import and Tf iron assimilation resulted from the discovery that DMT1 is mutated in the Belgrade rat (18). Defects in Tf-bound iron uptake were well characterized for this animal model of anemia. Although Tf is internalized by Belgrade reticulocytes, iron fails to be captured because of a loss of transmembrane transport activity in the endosome (31–34). The observation that the pH-dependent DMT1 transporter localizes to acidic endosomal compartments supports its role in this process (35, 36). Notably, inactivating mutations of the *mk* mouse and Belgrade rat reside at the same

residue (G185R). Numerous loss-of-function mutations also have been identified in human patients (R416C, G1221V, V114, G75R) (37–40).

Zip14 and Zip Family Members

Although there is strong evidence for the role of DMT1 in endocytic iron transport, mice that are deficient in Steap3 (41) or DMT1 (42) remain able to import liver iron. Moreover, certain cells also import non-transferrin-bound iron (NTBI) directly across the plasma membrane (43). This pathway contributes to the pathology of hemochromatosis, as evidenced by NTBI that is found in serum from patients (44). In addition, NTBI is found in infant cord blood and plasma (45), in patients with hepatic failure (46), in patients undergoing chemotherapy (47), in arthritic patient synovial fluid (48), and in cerebral spinal fluid (49). Recent studies suggest that Zip14 (Slc39a14) is a divalent metal transporter that not only facilitates iron uptake from Tf but also functions as an NTBI importer.

Zip family members were first identified as zinc transporters in yeast (50, 51) and iron-regulated metal transporters in plants (52). Zip14 functions in both zinc and iron transport (53, 54), and several Zip family members also interact with metals other than zinc (55). Zip proteins have eight predicted membrane-spanning domains. Zip14 belongs to the LIV-1 subfamily, although the initial histidine in conserved membrane-spanning domain 4 (HEXPHEXGD) is replaced with a glutamic acid residue (EEXPHEXGD); this substitution may contribute to Zip14's broader metal transport range. Mechanistically, iron and zinc are recognized and translocated by Zip14 through different mechanisms (54). Exogenous expression of Zip14 in HepG2 cells not only stimulates NTBI uptake but also increases acquisition of iron from Tf (19). Conversely, knockdown of Zip14 by siRNA decreases Tf-bound iron uptake (19). Thus, both DMT1 and Zip14 may play redundant roles in the cellular uptake of Tf-bound and NTBI uptake pathways.

Iron-Transporting Channels

Calcium uptake and, by extension, calcium channels are of interest in the context of iron uptake. From a nutritional point of view, dietary calcium reduces iron bioavailability (56). Altered cellular calcium levels suppress both NTBI uptake pathways (57) and Tf-bound pathways (58). L-type calcium channels have been implicated in iron entry under conditions of overload in cardiomyocytes (59, 60), whereas Fe^{2+} , in turn, can block voltage-gated calcium currents (61). More recently, Shawki & Mackenzie (62) examined the interaction between calcium and DMT1 in *Xenopus* oocytes and hypothesized that although human DMT1 does not mediate calcium uptake, Ca^{2+} is a low-affinity noncompetitive inhibitor of this iron transporter. Interestingly, the G185R mutation in the DMT1 gene, which leads to defective iron assimilation in Belgrade (*b*) rats and microcytic *mk* mice, confers calcium transport activity to DMT1 (63).

Members of the TRPML superfamily transport Fe^{2+} and are located in endolysosomes. TRPML has six predicted membrane-spanning domains with intracellular (cytosolic) N and C termini (64). In particular, a role for TRPML1 as a channel for iron release across late endosomes and lysosomes has been described (20). Mutations in this transient receptor potential channel are associated with mucopolipidosis, and patients carrying these disease alleles display not only neurodegenerative impairments but also iron deficiency anemia. TRPML1 likely enhances the efficiency of iron uptake from the Tf-bound pathway, either working in conjunction with or in a pathway that is redundant with that of DMT1 and/or Zip14 (Figure 1). Ineffective iron transport caused by disease mutations results in lower cytosolic iron levels and increased lysosomal iron pools (20).

Iron Export by Ferroportin

Although multiple import pathways have been identified, only one iron export pathway is known. The body's conservation of iron suggests that export would be restricted. Three independent studies seeking to elucidate the molecular basis for iron export led to the identification of the same factor: Ferroportin was found through positional cloning of a defective gene in anemic zebrafish (65), iron-regulated protein-1 (Ireg1) was characterized through subtractive hybridization of mRNAs induced in *hpx* (hypotransferrinemic) mice (66), and metal transport protein-1 (MTP1) was cloned by construction of a library enriched for iron-responsive protein-1 (IRP-1) binding (67). There is some agreement that this transporter—also known by its mouse and human gene names, *Slc39A1* (68) and *SLC11A3* (69)—is best described by the name ferroportin.

Ferroportin is highly conserved; human, mouse, and rat clones are 90–95% homologous at the protein level (70). The 62-kDa protein has nine predicted membrane-spanning domains and a cytoplasmically positioned N terminus. The protein localizes to the basolateral membrane of polarized cells (66), supporting the idea that it functions in the assimilation of dietary iron by exporting the metal from intestinal enterocytes. The copper-containing ferroxidase hephaestin assists by converting Fe^{2+} to Fe^{3+} , which is then bound by Tf (Figure 1). Furthermore, ferroportin is found in macrophages of the reticuloendothelial system, and it exports iron recycled after erythrophagocytosis (71). Type IV HH or ferroportin disease arises from mutations in this gene.

IRON TRANSPORT INHIBITORS AND STIMULATORS

Antioxidants: Ebselen and PDTC

Ebselen (2-phenyl-1,2-benzisoselenazol-3[2H]-one) is a lipid-soluble, selenium-containing compound that has neuroprotective effects against ischemic stroke (72, 73). Its pharmacological profile is due in large part to its antioxidant properties, and studies have shown that ebselen can (a) mimic glutathione peroxidase activity (74), (b) inhibit several inflammatory enzymes by thiol modification (74), and (c) reduce thioredoxin (75, 76). Interestingly, this selenazol decreased iron levels in the hearts of iron-overloaded mice (77), thus showing promise as an inhibitor of iron uptake. A fluorescence-based chemical genetic screen subsequently identified ebselen as an antagonist of DMT1 (78). Ebselen inhibits DMT1 $^{55}\text{Fe}^{2+}$ uptake activity with an IC_{50} of 0.22 μM , but it has no effect on ^{54}Mn uptake, despite the fact that Mn^{2+} is a known DMT1 transport substrate (24). Additionally, ebselen does not inhibit Tf-bound uptake; therefore, its effects appear specific to transport of ferrous iron. Pyrrolidinedithiocarbamate (PDTC) is a second antioxidant that inhibits DMT1 (IC_{50} = 1.54 μM). Both compounds affect cellular redox status because they increase cellular levels of glutathione in their reduced forms. Cytosolic iron levels are balanced between the ferric and ferrous forms (79), and the antioxidants perturb this equilibrium. Such compounds could promote a change in redox environment, potentially targeting ferrereductases involved in iron uptake pathways (80) (Table 1).

Calcium Channel Blockers

Effects of L-type calcium channel blockers on iron transport have been examined (81). Among antagonists tested, nifedipine, diltiazem, verapamil, and cinnarizine displayed little to no inhibition, whereas nifedipine [3,5-dimethyl 2,6-dimethyl-4-(2-nitrophenyl)-1,4-dihydropyridine-3,5-dicarboxylate] strongly stimulated NTBI uptake but not Tf-bound iron uptake. Nifedipine is light sensitive, and a biologically active photodegradation product was ultimately shown to be the active form as a nitrosophenylpyridine derivative (82). Because this compound also increased iron efflux, photodegraded nifedipine was hypothesized to act as a lipophilic iron ionophore that shuttles Fe^{2+} across the cell membrane.

Ludwiczek et al. (83) described new pharmacological effects of nifedipine and another L-type channel blocker, nifedipine, on DMT1. Mice treated with nifedipine had decreased serum iron, which correlates with a DMT1 genotype. Nifedipine-mediated DMT1 activation was proposed to mobilize excess iron from liver and to increase excretion by the kidneys and thus be of possible clinical benefit to patients with iron overload disorders. However, the effects were attributed to nifedipine and not its photodegradation products. Conflicting results were obtained when Mackenzie, Garrick, and colleagues (84) rigorously examined nifedipine effects on DMT1 in vitro. In their study, the parent compound did not increase DMT1-mediated transport of ^{59}Fe or ^{54}Mn . When photodegraded nifedipine was tested, it weakly stimulated iron uptake, but it had no effect on manganese uptake. Thus, the direct effect of nifedipine and/or its photodegradation products on DMT1 activity remains controversial.

NSC306711/Ferristatin

NSC306711 was also identified from a fluorescent cell-based assay in a search for NTBI uptake inhibitors (85). NSC306711 blocks DMT1 transport activity with $\text{IC}_{50} \sim 14.7 \mu\text{M}$ (86). This compound is a polysulfonated dye with two copper centers that do not influence drug activity. NSC306711 functions as a competitive and reversible inhibitor of DMT1 and does not perturb cellular redox status. Other polysulfonated compounds, structurally related to NSC306711 but lacking the copper centers, also interfere with DMT1 iron uptake, warranting additional work on their mechanism of action and potential use in remediating iron overload (86).

Lane et al. (87) used this compound to define how astrocytes import NTBI. Two independent mechanisms of astrocyte NTBI uptake are observed in the presence or absence of ascorbate. Ascorbate-depleted cells display a Fe^{3+} -selective pathway, whereas ascorbate-replete cells take up Fe^{2+} (88). Ascorbate loading of astrocytes markedly enhances iron accumulation, presumably due to reduction of Fe^{3+} to Fe^{2+} by ascorbate exported to the medium. NSC306711 blocks iron uptake by ascorbate-replete cells, and, on the basis of high expression of DMT1 in astrocytic endfeet (89, 90), Lane et al. (87) concluded that DMT1 is the transporter that facilitates NTBI import into ascorbate-replete astrocytes.

NSC306711 also blocks Tf-bound iron uptake with $\text{IC}_{50} \sim 20 \mu\text{M}$ (85). This is not altogether surprising because DMT1 is also involved in the release of Tf-bound iron from endosomes (18). However, the mechanism of inhibition was unexpected because NSC306711 induced internalization and degradation of TfR1. TfR1 trafficking has been well studied as a classic form of clathrin-mediated endocytosis (91), but siRNA knockdown of clathrin expression does not interfere with TfR1 downregulation by NSC306711. Instead, the action of this compound is sensitive to cholesterol depletion (92), and treatment of cells with nystatin and filipin (reagents that readily interact with cholesterol) blocks inhibition and degradation of TfR1 induced by NSC306711 (92). Because this compound affects TfR1 uptake through a cholesterol-dependent pathway, the name ferristatin was coined. However, a generalized effect on lipid rafts is not observed, and the stability of flotillin (93), a lipid raft resident protein, is not altered. Collectively, these findings revealed an unexpected role for lipid rafts in TfR1 trafficking.

Lysosomotropic Agents

Weak bases can penetrate and accumulate in acidic compartments of the cell such as lysosomes and endosomes, increase their pH, and disrupt Tf-bound iron uptake (91, 94–96). Lysosomotropic agents such as NH_4Cl , CH_3NH_2 , *n*-butyl amine, $(\text{CH}_3)_2\text{NH}$, and chloroquine interfere with Tf-bound iron uptake. Recently, chemical genetic screening resulted in the identification of a novel class of sulfonamides that also blocked endocytic

uptake of iron via the Tf-TfR pathway (97). These compounds elevate endosomal pH from 6.4 to 7.0 and inhibit dissociation of iron from Tf and recycling of receptors to the cell surface. The sulfonamides exhibited proton ionophore activity and blocked the V₁ domain of vacuolar ATPase, which maintains the pH by pumping protons into the intracellular organelle (98).

Hepcidin

Hepcidin, a peptide hormone that is secreted primarily by hepatocytes, is an important regulator of iron homeostasis (99, 100). Body iron stores, inflammation, erythropoiesis, bone morphogenic proteins, HFE, hemojuvelin, and TfR2 modulate its gene expression (101). Dysregulated hepcidin production contributes to hemochromatosis, hepatitis C, and iron-loading anemias (102–105). Bioactive hepcidin is a 25-residue-long cysteine-rich peptide that results from enzymatic cleavage of the 84-amino acid propeptide originally produced by the liver (106). This liver-derived peptide regulates systemic iron metabolism by functioning at two key stages: inhibiting iron absorption in the small intestine and inhibiting iron release from macrophages by binding to the iron exporter ferroportin. Hepcidin binds directly to ferroportin, which is located on the basolateral membrane of enterocytes and on the surface of macrophages, and induces phosphorylation, internalization, and degradation of the iron export protein (107). Treatment with hepcidin decreases ferroportin levels in both J774 macrophage cells and bone marrow macrophages (71, 108, 109). Thus, by downregulating ferroportin and, by extension, cellular iron efflux, hepcidin should function as an inhibitor of iron absorption. However, its mechanism of action in intestinal cells has been the focus of controversy. Several groups have reported that addition of hepcidin to Caco-2 cells and rat duodenal segments inhibits apical iron uptake and that this inhibition is associated with a significant decrease in DMT1 levels (110–112). Beaumont and coworkers (113) showed that hepcidin-mediated degradation of DMT1 occurs through a proteasomal pathway. Whereas hepcidin induced a decrease in DMT1 levels in these cells, levels of ferroportin remained unchanged. The effect of hepcidin, therefore, appears to be dependent on cell type: In macrophages, hepcidin inhibits iron export by ferroportin degradation; in intestinal cells, it inhibits iron absorption through DMT1 degradation.

Siderophores and Siderocalin

Naturally occurring iron chelators, siderophores, are secreted by microorganisms to scavenge environmental iron into bioavailable complexes. Pathogenic microbes also utilize siderophores to scavenge iron from bound protein complexes (e.g., transferrin, lactoferrin, hemoglobin) of their mammalian hosts (114). Siderophores are typically grouped into three classes: hydroxamates, catecholates, and hydroxycarboxylates (114). One of the most well-studied and clinically used siderophores is deferoxamine (DFO; see below). Hydroxybenzoates and hydroxybenzenes are often found in mammalian serum and urine, suggesting that a mammalian siderophore might function as a small-molecule iron chelator to traffic iron within cells when it is not bound to transport and storage proteins or incorporated into protein functional groups. The mammalian protein siderocalin (also known as Scn-Ngal, lipocalin-2, and 24p3) is a member of the lipocalin family that binds such small organic molecules. Siderocalin binds and sequesters bacterial siderophores such as enterobactin and enterochelin, thereby halting bacterial growth (115). Recently, two groups identified mammalian siderophores that bind to siderocalin and sequester the bound iron. Bao et al. (116) discovered that iron traffics as a siderocalin-catechol complex. Crystal structures show catechol-Fe bound in the calyx of siderocalin, and *in vivo* studies show that a siderocalin-catechol-Fe complex circulates to provide iron to tissues (most notably, the kidney). Devireddy et al. (117) suggest that siderocalin binds and delivers iron when bound to 2,5-dihydroxybenzoic acid, a mammalian siderophore similar to 2,3-dihydroxybenzoic

acid, a bacterial siderophore of enterobactin. More needs to be learned about mammalian siderophores and siderophore-binding complexes, but these discoveries suggest that endogenous inhibitors that chelate iron can inhibit transport, target delivery, and regulate metabolism of the host when it is infected.

IRON PHARMACOLOGY AND DISEASE

Chelation Therapy for Iron Loading

Several iron chelators have been tested as therapeutic agents to alleviate iron overload (Table 2). The most well studied is DFO, which has been the gold standard for iron overload treatment for >40 years. A hexadentate iron chelator (produced by *Streptomyces pilosus*), DFO binds Fe³⁺ with high affinity and renders it inactive and unable to produce ROS. DFO also decreases oxidative stress often associated with disease (118, 119). However, owing to its hydrophilicity, DFO is poorly absorbed by the gastrointestinal tract and has an extremely short half-life in the plasma of ~12 min (120). DFO also exhibits antiproliferative properties against aggressive cancers such as neuroblastoma and leukemia. Owing to its lack of membrane permeability and short half-life, improvements are necessary to make this a viable treatment option for both iron overload and cancer patients (119).

Deferiprone, a bidentate hydroxypyridinone that was synthesized and tested in the United Kingdom in the 1980s, is hydrophilic and relatively nontoxic in a variety of cell lines tested (121). A recent review summarizes the clinical trials that employed deferiprone as an iron-chelating agent to reduce iron overload (121). However, the safety and/or toxicity of deferiprone alone has been a topic of debate, prompting Motekaitis & Martell (122) to investigate the redox activity of preformed deferiprone-iron complexes. In vitro studies showed an increase in oxygen radical production at deferiprone:iron ratios less than 3:1. To reduce these toxic byproducts, investigators have explored combination therapy of DFO and deferiprone (119, 121). The basis of such therapy is that deferiprone can access intracellular compartments, chelating iron and making it available to extracellular DFO, which cannot cross cell membranes. The stable DFO-iron complex is then excreted in the urine. Currently, combination therapy has shown promising results, but long-term safety data are needed.

Deferasirox (ICL670A) is a bis-hydroxyphenyl-triazole tridentate ferric iron chelator that is orally available (123). It is highly lipophilic and mobilizes iron from liver and heart tissue efficiently. In vivo studies in rats and primates concluded that the drug was four to five times better at removing iron than was DFO (124). Clinically, deferasirox has been examined in the largest development program of any iron chelator, incorporating five clinical trials to assess efficacy, safety, and tolerability across several transfusion-dependent anemias (125–129). This drug is currently approved in the United States, the European Union, Switzerland, and elsewhere to treat transfusion-induced iron overload in both pediatric and adult patients and is in clinical trials for treatment of α -thalassemia, sickle cell disease, myelodysplastic syndromes, Diamond-Blackfan anemia, and other rare anemias.

Two other oral iron chelators are in various stages of clinical trials. Deferitrin is a tridentate chelator that has a half-life of 3–8 h and is excreted in the feces. Current data suggest that deferitrin is less effective than DFO and that—although evaluations of long-term and combination studies are necessary—it is not likely to suffice as a monotherapy (121). LINA11, a member of the β -keto-hydroxypyridine group of chelators that is currently in Phase I development, is a potent bidentate iron chelator capable of mobilizing iron from transferrin, ferritin, and hemosiderin (130). Increased iron excretion compared with deferiprone and DFO was shown in iron-loaded mice, rats, and rabbits (131). In a subacute toxicity study in rats, 45% mortality was reported in normal rats and no deaths were

recorded for iron-loaded rats, suggesting that LINA11 mobilized iron to toxic low levels in normal rats but offered protection in the iron-loaded rats (131).

Iron Transport and Cancer

Removal of iron arrests cell proliferation at the G₁/S phase of the cell cycle (132–134); thus, iron chelation and transport inhibition are appealing prospects for cancer therapy. In general, chelation is more effective when iron is coordinated by “soft” electron donor groups, allowing for the so-called double-punch effect (135). The chelator not only sequesters iron from the tumor cell but also is redox active and can generate ROS, causing extensive cell damage that leads to apoptosis of cancer cells. Thiosemicarbazones are a class of strong antiproliferative iron chelators that bind iron with nitrogen and with sulfur donor groups that permit redox cycling and generation of ROS. 3-Aminopyridine-2-carboxaldehyde thiosemicarbazone (3-AP) entered clinical trials in 2007 (136–138), but many were terminated early owing to adverse side effects and no signs of improvement (137–141). A new series, the 2-hydroxyl-1-naphthylaldehyde-3-thiosemicarbazone (NT) chelators, was synthesized on the basis of information gained from the 3-AP studies (142–145), but compounds with the highest antiproliferative activity had lower iron chelation efficacy and vice versa (144). The di-2-pyridyl thiosemicarbazones (the DpT series) also showed promise against a variety of cancer cell lines. The derivative Dp44mT exhibited the most antiproliferative, iron-chelating, and redox activity, but cardiac fibrosis developed in nude mice when Dp44mT was administered in high dose (146). Another class of 2-benzoylpyridine thiosemicarbazones (the BpT series) appears to surpass the DpT series in its antiproliferative and redox properties in cancer cell line studies; this series thus warrants further study (147).

Small molecules that modulate iron transport have also been considered as potential anticancer agents. Because production of ROS by excess iron accrual ultimately results in increased oxidative damage and cell death, a stimulator of iron uptake named LS081, recently identified by chemical genetic screening (148), was tested in various cancer cell lines. LS081 plus 2–10 μM of iron increased cellular iron, decreased proliferation, decreased expression of hypoxia-inducible factor (HIF)-1 and HIF-2 (factors that can drive cancer growth), decreased clonogenicity, and stimulated production of ROS. Elevated ROS might make LS081-treated cells more sensitive to both radiation therapy and other drugs (148). In addition, high levels of HIF-1 and HIF-2 are indicators of poor prognosis, and increased intracellular iron could potentially enhance prolyl hydroxylase activity and proteasomal degradation of HIF proteins (148). The mechanism of LS081 action still remains to be determined, but its discovery represents an interesting new approach to cancer treatment.

The Tf-TfR1 endocytic pathway has been exploited as a means to deliver metals and conjugated drugs to cancer cells. Antibodies against TfR1 successfully inhibit human tumor growth in vitro and in vivo by blocking Tf-bound iron entry into cells (149, 150). Additionally, although primarily an iron-binding protein, Tf has a diverse metal-binding capacity that has been exploited to deliver to cells therapeutic metal ions such as gallium and indium (133). This approach has been used in both diagnostic and chemotherapeutic treatments of malignancies and neoplasms. Finally, diphtheria toxin conjugated to Tf (Tf-CRM107; TransMID™, Xenova Group) has been infused into malignant brain tumors, and Phase I and II trials have provided encouraging results: Antitumor effects were seen in patients resistant to conventional therapy (151).

Lysosomal iron chelation has also been proposed as means of protection against oxidative damage in cancer treatment and other modalities. Lysosomes have a large pool of redox-active labile iron, making them a target of initial oxidative damage. The large amount of iron present in lysosomes results from autophagy and autophagic degradation of ferritin and

other iron-containing proteins. Sensitivity to oxidative stress is influenced by ferritin autophagy (152), and use of the soluble iron chelator salicylaldehyde isonicotinoyl hydrazine has been proposed (152). As another example, amifostine has been explored in clinical trials for treatment of non-small-cell lung cancer and several other types of cancer (153–158). The use of this drug, which acts as a radioprotector and lysosomotropic iron chelator (159), has been approved for head and neck cancers to protect against radiation-induced xerostomia (160).

Iron Transport and Neurodegenerative Diseases

Mitochondrial iron chelation is another organelle-based means of protection for diseases such as Friedreich's ataxia (FA). FA is a severe neurodegenerative disorder that results in mitochondrial iron overload and disrupted iron-sulfur assembly (161). Sensitivity to the FA mutation is rescued with iron chelators (162). Clinically, chelation therapy is problematic because compounds have to specifically target mitochondrial iron to avoid nonspecific iron deficiency anemia. Pyridoxal isonicotinoyl hydrazone (PIH) is one iron chelator that can permeate the plasma membrane as well as the outer and inner mitochondrial membranes and can bind mitochondrial iron (163), and some work has been carried out to advance PIH derivatives to resolve problems with mitochondrial iron loading (reviewed in 164).

Chelation therapy has been explored as a means of neuroprotection, and the brain-permeable VK-28 series of iron chelators has been studied in models of Parkinson's disease (165). Alzheimer's disease is another neurodegenerative disorder that can be improved by chelation therapy (166). In vitro studies show that pretreatment of β -amyloid (a plaque-causing agent) with DFO reduced the cytotoxicity of β -amyloid in an iron-dependent manner. (167). Recently, the Alzheimer's disease β -amyloid precursor protein was shown to have a ferroxidase activity associated with ferroportin iron export (168) and to be regulated by iron through an atypical iron-responsive element in the 5' untranslated region of its mRNA (169). Both activities appear to be targets for disruption of iron transport (170, 171).

Use of Natural Products

Emerging information from the use of natural products has also begun to expand our understanding of iron transport. The anticancer properties of curcumin, an ingredient of the spice turmeric, include its biological activity as an iron chelator, and it induces iron depletion in vivo (172). Silybin, a polyphenol in milk thistle, reduces iron absorption in patients with HH (173). The antioxidant extract of *Ginkgo biloba* leaves (EGb761) prevents iron-dependent peroxidation of polyunsaturated fatty acids (174). These and other new developments in the pharmacology of iron transport hold great promise in combating a variety of diseases.

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Glossary

ROS	reactive oxygen species
Tf	transferrin
TfR1/TfR2	transferrin receptor-1 and -2
HH	hereditary hemochromatosis

DMT1	divalent metal transporter-1
Zip14	ZRT/IRT-like protein 14
TRPML1	transient receptor potential mucolipin 1
NTBI	non-transferrin-bound iron

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SUMMARY POINTS

1. Several membrane-spanning transporters, channels, and proteins are involved in iron uptake and export. In the duodenum, dietary iron is transported through DMT1 into enterocytes and exits the enterocyte through ferroportin to enter systemic circulation. Iron is transported to peripheral tissues by Tf and undergoes receptor-mediated endocytosis after binding to TfR1. Iron exits endosomes through DMT1. Alternatively, other factors including Zip14, calcium channels, and TRPML1 may help mediate delivery of iron to cells. The uptake, export, and usage of iron requires a series of oxidation-reduction steps, including the specific transport factors hephaestin and Steap3.
2. Iron transport is inhibited or stimulated by a variety of chemical and naturally occurring compounds. Antioxidant molecules (ebselen and PDTC) decrease iron levels by antagonizing NTBI uptake by DMT1. In contrast, NTBI uptake was stimulated by the presence of nifedipine (and its photodegradation product) through the enhancement of DMT1 activity, but the precise mechanism remains controversial. Tf-bound iron uptake is inhibited by NSC306711/ferristatin through degradation of TfR1. The naturally occurring peptide hormone hepcidin inhibits iron export by inducing degradation of ferroportin in macrophages. It may play a different role by inhibiting DMT1 function in intestinal cells.
3. Iron chelation therapy is being tested and used as a therapeutic for diseases including iron overload, several cancers, and neurodegenerative diseases. Treatment with DFO shows positive improvement in both iron overload and Alzheimer's disease. Through the chelation of iron, DFO treatment results in reduced levels of ROS in cells. Currently, deferasirox is approved for use in the treatment of transfusion-induced iron overload, with several clinical trials testing therapeutic value for a variety of anemias. Chelation therapy for cancer works via two mechanisms: Removal of iron arrests cell proliferation at the G₁/S phase of the cell cycle, and a redox-active chelator can induce generation of ROS that are toxic to cancer cells, leading to apoptosis.
4. The future study of iron transport is expanding through studies of natural products including curcumin, silybin, and *Ginkgo biloba*. These and other avenues provided by small-molecule screening help further the study of iron transport and possibly treat diseases.

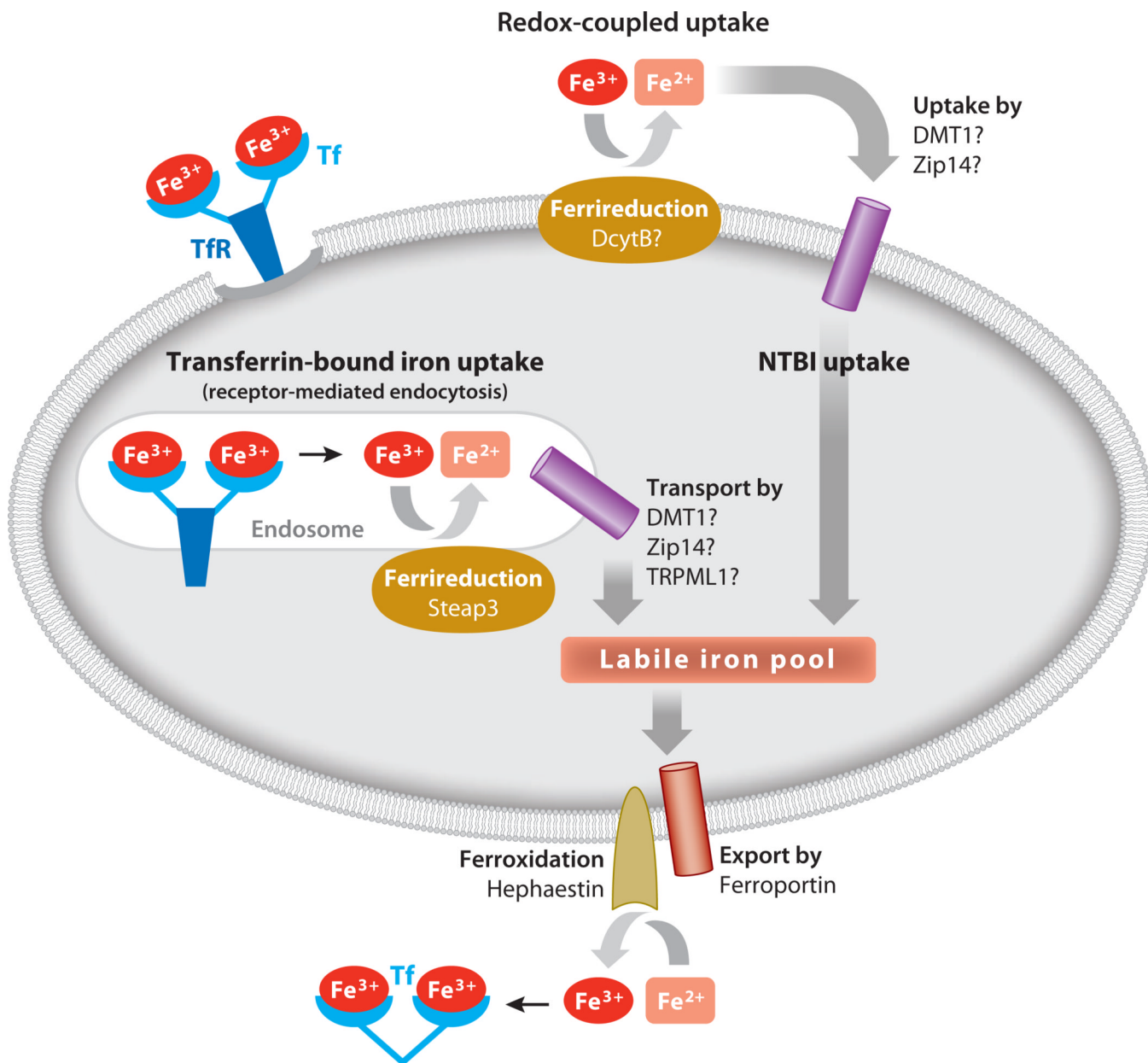


Figure 1.

Mechanisms of cellular iron import and export. Fe^{3+} -Tf binds to TfR on the cell surface and undergoes receptor-mediated endocytosis. The released Fe^{3+} is reduced to Fe^{2+} by Steap3 within the endosome. Iron exits the endosome and enters the labile iron pool. Alternatively, NTBI is reduced at the cell surface prior to cellular uptake. Cellular iron (Fe^{2+}) exits through ferroportin and is oxidized to Fe^{3+} and transported bound to Tf. Abbreviations: DcytB, duodenal cytochrome B; DMT1, divalent metal transporter-1; NTBI, non-transferrin-bound iron; Tf, transferrin; TfR, transferrin receptor; TRPML1, transient receptor potential mucolipin 1; Zip14, ZRT/IRT-like protein 14.

Table 1

Inhibitors and stimulators of iron transport pathways

Transporter	Transport pathway	Inhibitor/stimulator
TfR1/TfR2	Tf-bound endocytic uptake	Chloroquine, NH ₄ Cl, weak bases NSC306711/Ferristatin Sulfonamides
DMT1 (Slc11a2)	Intestinal uptake, Tf-bound endocytic uptake	Ebselen, PDTC, antioxidants Nifedipine? NSC306711/Ferristatin, polysulfonated dyes
Zip14 (Slc39a14)	NTBI uptake, Tf-bound endocytic uptake	??
L-type channels	NTBI uptake by cardiomyocytes	Nifedipine
TRPML1	Lysosome iron export	??
Ferroportin	Intestinal/macrophage iron export	Hepcidin Mini-hepcidin (175)

Abbreviations: NTBI, non-transferrin-bound iron; PDTC, pyrrolidinedithiocarbamate; Tf, transferrin; TfR, transferrin receptor; TRPML1, transient receptor potential mucolipin 1; Zip14, ZRT/IRT-like protein 14.

Table 2

Chemical properties and structures of iron chelators

Chelator	Electron-donating groups	Structure
DFO	Hexadentate (6) 1:1, chelator:iron	
Deferiprone	Bidentate (2) 3:1, chelator:iron	
Deferasirox	Tridentate (3) 2:1, chelator:iron	
Deferitran	Tridentate (3) 2:1, chelator:iron	
LINA11	Bidentate (2) 3:1, chelator:iron	