

NIH Public Access

Author Manuscript

Org Biomol Chem. Author manuscript; available in PMC 2014 November 21.

Published in final edited form as:

Org Biomol Chem. 2013 November 21; 11(43): . doi:10.1039/c3ob41702f.

Chemoenzymatic synthesis and *in situ* **application of Sadenosyl-L-methionine analogs**

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Abstract

Analogs of *S*-adenosyl-L-methionine (SAM) are increasingly applied to the methyltransferase (MT) catalysed modification of biomolecules including proteins, nucleic acids, and small molecules. However, SAM and analogs suffer from an inherent instability, and their chemical synthesis is challenged by low yields and difficulties in stereoisomer isolation and inhibition. Here we report the chemoenzymatic synthesis of a series of SAM analogs using wild-type (wt) and point mutants of two recently identified halogenases, SalL and FDAS. Molecular modelling studies are used to guide the rational design of mutants, and the enzymatic conversion of L-Met and other analogs into SAM analogs is demonstrated. We also apply this *in situ* enzymatic synthesis to the modification of a small peptide substrate by protein arginine methyltransferase 1 (PRMT1). This technique offers an attractive alternative to chemical synthesis and can be applied *in situ* to overcome stability and activity issues.

> *S*-adenosyl-L-methionine (SAM, **1a**) is the most common enzyme substrate in nature after ATP.¹ It serves as a substrate for methylation of proteins, nucleic acids, and small molecules, but is also involved in polyamine biosynthesis and radical reactions.² There has been considerable interest in using analogs of SAM as selective inhibitors, for exploring methyltransferases (MTases), and for profiling MTase substrates.³ The latter approach is highly interesting as it is speculated that many substrates are still to be identified.^{2b, 4}

> Two types of enzymatically active SAM analogs have been reported.⁵ Aziridinium-based SAM analogs have been employed with DNA and protein MTases resulting in bi-substrate products in which the SAM analog is covalently linked to the MTase substrate.^{3b,3c,6} However, when using protein arginine MTase 1 (PRMT1) the bi-substrate product has been observed to cause product inhibition indicating that the scope of these analogs is limited.^{3c} Recently, SAM analogs have been applied where the methyl group in L-Met of SAM is exchanged with a new reactive functional group (e.g. terminal alkynyl, keto, or amino group). Similar to the methyl group in SAM, the new chemical moiety is transferred to MTase substrates by the respective MTases. This method offers a compelling strategy, as it

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[†]Electronic Supplementary Information (ESI) available. See DOI: 10.1039/b000000x/

allows for subsequent selective bioorthogonal conjugation of the MTase product with functional tags such as fluorescent or affinity probes.^{3a, 3d-i}

Currently, SAM analogs are synthesized chemically from *S*-adenosyl-L-homocysteine (SAH, **2**) with low yields and in approximately a 1:1 ratio of the sulfonium ion diastereomers. Usually only the (*S,S*) diastereomer is biologically active, whereas the (*R,S*) diastereomer has been demonstrated to inhibit MTases.⁷ The impractical separation of these diastereomers typically leads to application of diastereomeric mixtures for the MT transformations.^{3a, 3f–3i} Additionally, caution must be applied when storing and handling SAM and SAM analogs, as these molecules are inherently unstable, particularly at neutral and alkaline pH.⁸ An *in situ* enzymatic synthesis of SAM analogs would therefore be of high value if diastereoselective and able to overcome instability problems by producing SAM analogs.

We recently demonstrated the enzymatic synthesis of SAM and the direct coupling to the enzymatic methylation of the antibiotic teicoplanin by MtfA MTase and of DNA by HhaI MTase.⁹ The applied SAM producing enzyme was a recently discovered chlorinase SalL from *Salinispora tropica*10 and this enzyme is homologous to a fluorinase FDAS from *Streptomyces cattleya*. ¹¹ *In vivo* these halogenases catalyse the breakdown of SAM by a nucleophilic attack of chloride or fluoride at C5′ of the ribosyl moiety of SAM producing L-Met (**3a**) and 5′-chloro or fluoro-5′-deoxyadenosine (ClDA or FDA, **4**, X = Cl or F) (Figure 1). FDAS can also utilize chloride as a nucleophile, and for both enzymes it has been shown that *in vitro* equilibrium favors the synthesis of SAM from ClDA and L-Met.^{10, 12} Herein, we report the successful use of wt SalL, designed SalL point mutants, and FDAS for the enzymatic diastereoselective synthesis of SAM analogs from ClDA and readily prepared L-Met analogs (**3b–j**) (Figure 1).

Inspecting molecular interaction field calculations of the SalL (PDB: 2Q6I) and FDAS (PDB: 1RQR) crystal structures containing L-Met, 10,11 two water molecules were found tightly bound in the active site of the enzymes where the L-Met analogs are assumed to propagate (Figure 2 and supporting information). This suggests that destabilization of these water molecules might facilitate the accommodation of larger substituents. When analysing refined structures of SalL and FDAS, the following amino acids were found to be involved in an extensive hydrogen bonding network with the two water molecules in SalL (S) and FDAS (F): $T128_S$ (T155_F), W190_S (W217_F), Y239_S (Y266_F), backbone amide of S242_S $(S269_F)$, and the carboxylate group of L-Met (Figure 2). A series of point mutants were generated for this T, W and Y in both SalL and FDAS with the aim of accommodating larger substituents through introduction of smaller amino acids and destabilizing the hydrogen bonding network (Figure 2).

For FDAS, all mutants expressed in sufficient yield whereas for SalL, only 6 out of the 12 mutants expressed properly (T128A and Y239H-A) (Supporting Information). The activities of these expressed mutants, as well as wt enzymes, were analyzed for activity with L-Met (**3a**) and the series of synthetic L-Met analogs (**3b–j**) of varying size and polarity, as illustrated in Figure 1 (see tables with activity data in Supporting information).

Activities were estimated by HPLC and reported in terms of initial velocities from the ratio of the product area against the total area of product and unreacted substrate (ClDA) relative to the applied enzyme concentrations. Identical concentrations (15 mM) of L-Met and analogs were used except for benzyl and phenethyl analogs (1.5 mM) due to low solubility. Negative control reactions without enzymes were performed in each series to verify enzyme activity. Consistent with earlier studies, the activity of L-Met was approximately 500-fold higher for SalL compared to FDAS.^{11, 12d} In general, activity with L-Met was lower for the

mutants compared to wt enzymes, except for $Y266F_F$ and $T155S_F$, that displayed a slightly increased activity compared to wt FDAS.

For wt SalL, activity was detected for all L-Met analogs except propargyl (**3f**), phenethyl (**3h**) and polar analogs (**3i,j**) (Figure 3 and Supporting Information). The formation of ethyl-, propyl, butyl-, allyl-, and benzyl-SAM was confirmed by LC MS (Figure 3 and Supporting Information). In general, activity decreased with increasing size of the L-Met analog. The rate of formation of ethyl-, propyl, butyl-, and allyl-SAM compared to SAM, were reduced 10-, 5000-, 15000- and 5000-fold, respectively.

For some mutants the decrease in activity relative to L-Met was much less pronounced (Figure 3 and Supporting Information). In particular, the Trp mutant series showed good activity. For $W190F_S$ activity to L-Met analogs decreased in a similar manner as observed for wt SalL. For W190H_S activity was observed using the propyl and butyl analogs of L-Met, but no activity was detected using either L-Met or the ethyl analog suggesting increased activity towards the larger L-Met analogs. The $W190A_S$ mutant accepted the same set of L-Met analogs as wt SalL, but the activities were not reduced as drastically as seen for SalL and the other mutants. Here, the activity of ethyl-, propyl, butyl-, and allyl-SAM formation compared to SAM formation was reduced 20-, 30-, 30-, and 60-fold, respectively. These results indicate that the activity of the ethyl analog of L-Met is reduced with similar magnitude compared to L-Met for both wt and W190A SalL, whereas the reduction in activity when using the larger substrates was $150-500$ -fold higher for wt SalL. W190A_S catalyzed the formation of butyl-SAM with a three-fold higher velocity than when using wt SalL (0.934 \pm 0.010 nM/min compared to 0.318 \pm 0.006 nM/min). Propyl-SAM was produced with a similar velocity as with wt SalL, whereas the formation of SAM was reduced 160-fold compared to wt SalL. Unlike wt SalL, $W190A_S$ showed no reduction in activity when exchanging the propyl analog of L-Met with the butyl analog. Only wt and W190A SalL were tested for activity towards the benzyl and phenethyl analogs of L-Met, and similar rates for the two enzymes were seen for the benzyl analog. This finding is intriguing with respect to further development of enzyme variants in order to discover even broader substrate specificities. From the Tyr mutant series, only $Y239F_S$ showed sufficient expression yield to study activity. This mutant showed a similar reduction in activity to wt SalL when increasing the size of the L-Met analogs. Finally, the Thr mutant series showed activity only for L-Met and the ethyl analog of L-Met. Unfortunately, the propargyl analog of L-Met was observed to be unstable, but also propargyl-SAM has been reported to highly unstable (see Supporting Information).3e,3f,3h

These data partially confirm our *in silico* structural analysis of SalL and FDAS, as we succeeded in modulating the active site to accept some L-Met analogs. Mutants designed to reduce the occupancy of water (W190F_S and Y239F_S) led to reduced activity of all analogs, presumably due to impaired interaction with the carboxylate group of L-Met and analogs. But for $W190A_S$, this loss was compensated by an increased size of the pocket, leading to activities comparable to wt for propyl, allyl, and benzyl L-Met analogs; and for the butyl analog a significant increase in activity was observed

Wt and mutant FDAS generally displayed much lower activities compared to SalL (Figure 4 and Supporting Information). However, two mutants (T155S_F and Y266F_F) showed a small but significant increase in the activity towards both L-Met and the ethyl analog of L-Met – ranging from an L-Met activity of 10.6 ± 1.5 nM/min for wt FDAS to 18.5 ± 0.1 and 15.6 ± 1.5 0.5 nM/min for the T155S_F and Y266F_F mutants, respectively. Activities of L-Met and the ethyl analog of L-Met were observed for all FDAS variants and the relative reductions in activity when exchanging L-Met with the ethyl analog were comparable with those found for wt SalL. Furthermore, HPLC analysis indicated formation of trace amounts of propyl-

SAM for Y266F, but this could not be confirmed by LC MS because the produced amount was below detection limit.

To probe the application of these enzymes in an *in situ* enzymatic synthesis of SAM and analogs coupled to the modification of various substrates by MTs, we applied rat PRMT1 to explore the enzymatic modification of a small peptide sequence, the RGG peptide, that mimics the natural substrate fibrillarin (Figure 5). PRMT1 is the predominant PRMT in mammalian cells, estimated to be responsible for 85% of arginine methylation.¹³ As a type 1 PRMT, the product of PRMT1 is asymmetrically dimethylated arginine.¹⁴ First, we tested the use of the natural substrate SAM as well as the *in situ* SAM producing coupled assay using L-Met, ClDA, and wt SalL. The peptide was successfully dimethylated at all three arginines under both conditions (Supporting Information). Coupled assays were then performed using the ethyl, propyl, butyl, allyl, and benzyl analogs of L-Met, along with negative controls without rPRMT1.

The allyl L-Met analog transferred effectively, and both mono- and di-modified peptides were identified by LC MS. While allylation via synthetic analogs has been reported,¹⁴ this is the first time di-allylation has been seen using rPRMT1. This finding is likely the result of our technique providing only the biologically active (*S,S*) diastereomer and that no inhibitory (*R*,*S*) diastereomer contaminates the allyl-SAM preparation. Indeed, these findings strongly support the use of this chemoenzymatic method for a stereoselective analog preparation.

Transfer was also observed when using both the ethyl and benzyl analogs of L-Met. It is notable that that the bulky benzyl group is transferred and is probably due to the increased reactivity of this group. For the benzyl analog, MS indicated the formation of only monomodified peptide, whereas for the ethyl analog both mono-, di- and tri-modified RGG peptides were identified.

However, when employing the larger alkyl-SAM analogs (propyl and butyl), no activity was observed. Given that the rate of propyl- and butyl-SAM formation is approximately 500-fold lower when compared to ethyl-SAM, it remains unclear whether this lack of activity is simply due to reduced specificity or that the activity is abolished.

Conclusions

In conclusion, this study demonstrates the first published method for a chemoenzymatic synthesis of various SAM analogs using both wt and engineered halogenases. This method uses synthetically accessible L-Met analogs as precursors for the diastereoselective formation of SAM analogs. The method is useful for the enzymatic modification by MT's using *in situ* generated SAM analogs, hereby overcoming major issues of instability with SAM and related analogs. The utility of *in situ* coupled assays with other MTs becomes evident when considering the stability of these enzymes and their applicability in a range of conditions. Based on the *in silico* analysis of SalL and FDAS active sites, it is conceivable that future protein engineering studies can increase activities and further expand substrate specificity for additional SAM analogs. Recently, engineered MAT1 was applied for the production of a SAM analogue *in vivo* demonstrating the scope of such reactions. However, only one analogue was probed and no enzyme kinetic data was reported.15 Considering the involvement of SAM in metabolic processes and disease states, 2d the scope of these applications is highly promising.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgments

The authors thank Professor Bradley S. Moore and Professor James H. Naismith for the kind donations of the plasmids pAEM7 and pFLA_HT for the expression of SalL and FDAS, respectively. This work was financed by the Drug Research Academy Program and the Novo Nordisk STAR program.

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Fig. 2.

The active site of SalL overlaid with the molecular interaction field of water at −10.0 kcal/ mol (black surfaces) and listing of the series of mutations for T128_S (T155_F), W190_S (W217_F) and Y239_S (Y266_F).

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Fig. 3.

A) Enzyme activity of wt SalL and the W190 mutant series of SalL using L-Met analogs. B) HPLC chromatograms illustrating the formation of butyl-SAM using wt and W190A SalL. The synthesis of butyl-SAM was verified by LC MS. $M_{mono} =$ monoisotopic mass. *1.5 mM *S*-benzyl-L-Hcy. [∇]These peaks are impurities from the enzyme solutions.

Fig. 4.

The enzyme activity of wt and mutants of FDAS. Note that one y- axis is for L-Met and one is for the ethyl analog.

Fig. 5.

A) A schematic illustration of the coupling of the enzymatic SAM and analog formation and the modification of RGG peptide by rPRMT1. B) LCMS chromatogram of the coupled assay with the allyl analog of L-Met and the spectra of the RGG peptides. L-Met^X is an abbreviation for both L-Met and analogs.