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## **Sites of superoxide and hydrogen peroxide production during fatty acid oxidation in rat skeletal muscle mitochondria**

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## **Abstract**

H2O2 production by skeletal muscle mitochondria oxidizing palmitoylcarnitine was examined under two conditions: the absence of respiratory chain inhibitors and the presence of myxothiazol to inhibit complex III. Without inhibitors, respiration and  $H_2O_2$  production were low unless carnitine or malate was added to limit acetyl-CoA accumulation. With palmitoylcarnitine alone,  $H_2O_2$  production was dominated by complex II (44% from site II<sub>F</sub> in the forward reaction); the remainder was mostly from complex I (34%, superoxide from site I<sub>F</sub>). With added carnitine,  $H_2O_2$ production was about equally shared between complexes I, II, and III. With added malate, it was 75% from complex III (superoxide from site  $III<sub>Q0</sub>$ ) and 25% from site I<sub>F</sub>. Thus complex II (site II<sub>F</sub> in the forward reaction) is a major source of  $H_2O_2$  production during oxidation of palmitoylcarnitine  $\pm$  carnitine. Under the second condition (myxothiazol present to keep ubiquinone reduced), the rates of  $H_2O_2$  production were highest in the presence of palmitoylcarnitine  $\pm$  carnitine and were dominated by complex II (site II<sub>F</sub> in the reverse reaction). About half the rest was from site I<sub>F</sub>, but a significant portion, ~40 pmol  $H_2O_2 \cdot min^{-1} \cdot mg$ protein<sup>-1</sup>, was not from complex I, II, or III and was attributed to the proteins of β-oxidation (electron-transferring flavoprotein (ETF) and ETF-ubiquinone oxidoreductase). The maximum rate from the ETF system was ~200 pmol  $H_2O_2 \cdot min^{-1} \sim mg$  protein<sup>-1</sup> under conditions of compromised antioxidant defense and reduced ubiqui-none pool. Thus complex II and the ETF system both contribute to  $H_2O_2$  production during fatty acid oxidation under appropriate conditions.

#### **Keywords**

Palmitate; Palmitoylcarnitine; ROS; Complex I; Complex II; Succinate dehydrogenase; Complex III; ETF; Electron-transferring flavoprotein; ETFQOR; Electron transferring flavoprotein– ubiquinone oxidoreductase; Free radicals

> Fatty acid β-oxidation by skeletal muscle mitochondria is a major source of ATP under physiological conditions [1,2], and impaired lipid metabolism is associated with several pathological conditions, such as multi-acyl-CoA dehydrogenase deficiency, obesity-related

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insulin resistance, and type 2 diabetes [3]. Changes in reactive oxygen species  $(ROS)^1$ production in these pathological conditions have been reported [4–8].

Palmitoylcarnitine is a major substrate for β-oxidation. It enters mitochondria on the carnitine–acylcarnitine translocase and is converted to palmitoyl-CoA. The fatty acid βoxidation spiral involves four sequential enzymes: acyl-CoA dehydrogenase, 2-enoyl-CoA hydratase, L-3-hydroxyacyl-CoA dehydrogenase, and 3-ketoacyl-CoA thiolase [9]. In contrast to peroxisomal β-oxidation, in which molecular oxygen serves as the electron acceptor of acyl-CoA oxidases, mitochondrial acyl-CoA dehydrogenases transfer single electrons to electron-transferring flavoprotein (ETF) [10]. Singly reduced ETF− is then oxidized by ETF-ubiquinone oxido-reductase (ETFQOR), which donates electrons directly to the ubiquinone (Q) pool in the mitochondrial inner membrane  $[11-13]$ , to be passed to complex III of the respiratory chain, cytochrome *c,* complex IV, and finally molecular oxygen. The second dehydrogenation reaction in the β-oxidation spiral is catalyzed by hydroxyacyl-CoA dehydrogenase, which uses NAD<sup>+</sup> as its electron acceptor. The reduced NADH is then oxidized by complex I, reducing the Q pool. The end product of β-oxidation, acetyl-CoA, condenses with oxaloacetate to form citrate, which is then oxidized by the Krebs cycle. Thus fatty acid oxidation in mitochondria is linked to oxidative phosphorylation and ATP production.

In contrast to many other substrates, such as malate, glutamate, succinate, or glycerol 3 phosphate, oxidation of fatty acids requires four enzymatic reactions and donates electrons at multiple points in the electron transport chain: complex I, ETFQOR, and complex II (via formation of succinate in the Krebs cycle). This makes fatty acid oxidation a good candidate for high rates of superoxide or  $H_2O_2$  formation due to possible leaks of electrons to molecular oxygen at several different sites.

There are several sites of superoxide or  $H_2O_2$  production in the Krebs cycle and electron transport chain [14–16]. In order of maximum capacity in skeletal muscle mitochondria, they are the ubiquinol-oxidizing site of complex III (site  $III<sub>00</sub>$ ), the ubiquinone-reducing site of complex I (site I<sub>O</sub>) and the flavin site of complex II (site II<sub>F</sub>), the flavin site of complex I (site  $I_F$ ), and the Q-binding site of glycerol 3-phosphate dehydrogenase (GPDH) [17,18]. Other sites include the dihydrolipoate moieties of 2-oxoglutarate dehydrogenase and pyruvate dehydrogenase, and the ETF/ETFQOR system, but their maximum rates have not been established. The native rates from various sites in the absence of inhibitors have been measured only during oxidation of glutamate and malate, when sites  $I_F$  and  $III_{OO}$  dominate [19]; oxidation of succinate, when site  $I<sub>O</sub>$  dominates [20,21]; and oxidation of glycerol 3phosphate, when site  $I_{\text{O}}$  dominates and sites  $II_{\text{F}}$  and GPDH also contribute [17,18].

The sites of  $H_2O_2$  production during oxidation of palmitoylcarnitine by skeletal muscle mitochondria have been addressed in earlier studies in isolated mitochondria [20,22–25]. Although mitochondria isolated from the complex cellular environment lose some inputs of metabolic control, this widely used model still provides a detailed understanding of mechanisms of metabolic regulation, including those related to mitochondrial ROS production. These insights then feed forward our understanding of more complex systems such as intact cells, tissues, or animals. Prior studies in isolated mitochondria have identified

several different components of the respiratory chain and β-oxidation pathway as sources of  $H<sub>2</sub>O<sub>2</sub>$  during palmitoylcarnitine oxidation by isolated mitochondria: site I<sub>O</sub> [20,23], site  $III<sub>00</sub>$  [20,22–24], ETF/ETFQOR [20,22–24], and acyl-CoA dehydrogenase [22,25]. One possible source of disagreement is the use of site-specific inhibitors of the respiratory chain. Such inhibitors are great tools for understanding the capacities and mechanisms of  $H_2O_2$ production at sites of interest. However, inhibition of one center disrupts normal electron flow and may lead to changes in the reduction states of other centers far from the site of inhibition, with consequent changes in their production of superoxide or  $H_2O_2$ . In the presence of inhibitors it becomes impossible to measure changes in native rates of  $H_2O_2$ production in different metabolic states or disease models, which is crucial for the investigation of ROS-related physiology and pathology. Therefore it is of high importance to resolve the contradicting conclusions and perform measurements in the absence of respiratory chain inhibitors (native rates) that more closely match physiological conditions in vivo.

The disadvantages of site-specific inhibitors can be partially avoided by using endogenous reporters to predict rates of superoxide formation from sites  $I_F$  and  $III_{O<sub>0</sub>}[19]$ . This approach assumes a unique relationship between the reduction state of the superoxide-producing species within the site and its rate of reaction with oxygen to generate superoxide (or  $H_2O_2$ ). The reduction state of the superoxide producer can be detected by measuring the reduction state of an endogenous "reporter," a species that is close to equilibrium with the superoxideproducing moiety. NADH was established as a reporter of rates of superoxide production at site I<sub>F</sub> (plus any other sites that respond to NADH reduction state) and cytochrome  $b_{566}$  as a reporter of superoxide production at site  $III<sub>Qo</sub>[19]$ .

For the first time we provide a complete analysis and dissection of the specific sites of superoxide/ $H_2O_2$  production during fatty acid oxidation in the absence of respiratory chain inhibitors. We also investigate the contribution of the ETF/ETFQOR system to the rates of H2O2 production in the absence and presence of complex III inhibitors that lead to reduction of the Q pool and estimate the maximum rate of  $H_2O_2$  production from this system in skeletal muscle mitochondria. We identify complex II as a new site of superoxide and/or  $H_2O_2$  production that was not recognized in earlier studies and whose  $H_2O_2$  production was previously wrongly attributed to other sites.

#### **Experimental procedures**

#### **Animals, reagents, and mitochondrial preparation**

Female Wistar rats (Harlan Laboratories), age 5–8 weeks, were fed chow ad libitum with free access to water. Skeletal muscle mitochondria were isolated in Chappell–Perry buffer (100 mM KCl, 50 mM Tris, 2 mM EGTA, pH 7.1 at 25  $^{\circ}$ C) by standard procedures [26]. Protein concentration was determined by the biuret method. The animal protocol was approved by the Buck Institute Animal Care and Use Committee, in accordance with IACUC standards. All reagents were from Sigma (St. Louis, MO, USA) except for Amplex UltraRed (Invitrogen, Carlsbad, CA, USA) and atpenin A5 (Santa Cruz Biotechnology, Santa Cruz, CA, USA). The L-isomer of palmitoylcarnitine was used in all experiments.

#### **H2O2 production**

Rates of superoxide production were measured indirectly as rates of  $H_2O_2$  production after conversion of superoxide to  $H_2O_2$  by endogenous superoxide dismutase (SOD) in the matrix.  $H_2O_2$  was detected using horseradish peroxidase oxidizing Amplex UltraRed to its fluorescent resorufin product [19,27]. Exogenous SOD was added to convert any superoxide released in the medium. Mitochondria (0.3 mg protein · ml<sup>-1</sup>) were suspended in a medium containing 120 mM KCl, 5 mM Hepes, 5 mM  $K_2HPO_4$ , 1 mM EGTA, and 0.3% (w/v) bovine serum albumin (pH 7.0 at 37 °C), together with 5 U · ml−1 horseradish peroxidase, 25 U · ml−1 SOD, and 50 μM Amplex UltraRed. The fluorescence signal was recorded using a Varian Cary Eclipse spectrofluorimeter ( $\lambda_{ex}$ 560 nm,  $\lambda_{em}$ 590 nm) with constant stirring. Rates of fluorescence change were calibrated with known amounts of  $H_2O_2$  and normalized to the amount of protein [27]. The rates of  $H_2O_2$  production in the calibration curve in Fig. 6B (see below) were obtained by titrating rotenone in the presence of 4 μM FCCP and 5 mM malate [28].

#### **NAD(P) reduction state**

Experiments were performed using 0.3 mg mitochondrial protein · ml<sup>-1</sup> at 37 °C in the same medium as for  $H_2O_2$  measurements. The reduction state of endogenous NAD(P) was determined by autofluorescence [28] using a Shimadzu RF5301-PC spectrofluorimeter at  $\lambda_{\rm ex}$  365 nm,  $\lambda_{\rm em}$  450 nm. NAD(P) was assumed to be 0% reduced after 5 min without added substrate and 100% reduced with 5 mM malate and 4 μM rotenone (Fig. 2A). Intermediate values were determined as %NAD(P)H relative to the 0 and 100% values. Although this technique measures contribution from both mitochondrial NADH and mitochondrial NADPH, the content of NAD<sup>+</sup> plus NADH in skeletal muscle mitochondria is much greater than the combined NADP+ and NADPH [28,29]. Moreover the enhancement of NADH fluorescence in mitochondria is two- to fourfold greater than it is for mitochondrial NADPH [30]. The higher content and greater fluorescence enhancement of NADH make the autofluorescence signal predominantly a measure of NADH. The  $NAD(P)^+$  reduction states in the calibration curve in Fig. 6B (see below) were manipulated by varying the concentration of rotenone in the presence of 5 mM malate and 4 μM FCCP [28].

#### **Cytochrome b566 reduction state**

Experiments were performed at 1.5 mg mitochondrial protein  $\cdot$  ml<sup>-1</sup> in the same medium as for  $H_2O_2$  measurements. The reduction state of endogenous cytochrome  $b_{566}$  was measured with constant stirring at 37 °C in an Olis DW-2 dual-wavelength spectrophotometer as  $A_{566 \text{ nm}}-A_{575 \text{ nm}}$  [27]. The signal at this wavelength pair reports ~75% cytochrome  $b_{566}$  and  $\sim$ 25% cytochrome  $b_{562}$  [27,31]. Cytochrome  $b_{566}$  was assumed to be 0% reduced after 5 min without added substrate and 100% reduced with 5 mM succinate and 4 μM rotenone or with 15 μM palmitoylcarnitine plus 5 mM malate, both in the presence of 2 μM antimycin A (Fig. 2B, see below). Intermediate values were determined as  $\%b_{566}$  reduced relative to the 0 and 100% values. At least 15 data points were used to calculate the average percentage reduction under each condition.

#### **Respiration**

Experiments were performed in the same medium as  $H_2O_2$  measurements at 0.3 mg mitochondrial protein · ml−1. Oxygen consumption was recorded at 37 °C using a Clark electrode. State 2 respiration was in the presence of substrates only. Phosphorylating (State 3) respiration was initiated by addition of 0.5 mM ADP. Nonphosphorylating (State 4) respiration was recorded after addition of 1  $\mu$ g · ml<sup>-1</sup> oligomycin and uncoupled respiration was measured after subsequent addition of 1 μM FCCP.

#### **Complex II activity**

Experiments were performed strictly as in [17] but with the buffer used in the present study. Activity was measured as the rate of reduction of 50 μM dichlorophenolindophenol (DCPIP) in the presence of 1 mM phenazine methosulfate instigated by addition of 10 mM succinate at 15 °C to freeze the activation state.

#### **Endogenous reporters of superoxide production**

The method is described in [19]. Briefly, the reduction state of  $NAD(P)^+$  was used to predict superoxide production from site  $I_F$  from the calibration curve in Fig. 2A (inset, see below) for all data except Figs. 6C and 7 (ETF/ETFQOR system), which used the calibration in Fig. 6B (see below). The reduction state of cytochrome *b*566 was used to predict superoxide production from site  $III<sub>Qo</sub>$  using the calibration curve in Fig. 2B (inset, see below).

#### **Statistics and curve fitting**

Data are presented as means  $\pm$  SEM. Differences between groups were assessed by Student's *t* test using SigmaPlot version 11. One-way ANOVA was used to compare differences in  $H_2O_2$  production after antimycin A treatment (Fig. 4A and B). Calibration curves in Fig. 2A and B were single exponential fits [19]. The calibration curve in Fig. 6B was fit poorly by a single exponential, so it was fitted empirically using Eq. (1), which assumes that the reporter (NADH) and the superoxide producer  $(FADH<sub>2</sub>)$  have different potentials but are close to equilibrium:

$$
\ln\left(\frac{100-\frac{y}{k}}{\frac{y}{k}}\right) = a + \frac{1}{2}\ln\left(\frac{100-x}{x}\right), \quad (1)
$$

where *y* is the rate of H<sub>2</sub>O<sub>2</sub> production (pmol · min<sup>-1</sup> · mg protein<sup>-1</sup>); *x* is the reduction state of the NAD<sup>+</sup> pool (% NAD(P)H);  $k=0.0033 \cdot$  pmol · %<sup>-1</sup> · min<sup>-1</sup> · mg protein<sup>-1</sup>, a scaling factor expressing the rate constant of  $H_2O_2$  formation; and  $a = 2.05$ , a parameter proportional to the difference in midpoint potentials. The errors caused by the calibrations were calculated for each *x* value as the root mean square of the linearly interpolated SEM of the *x* value propagated through the calibration equation and linearly interpolated SEM of the *y* value.

Errors for predicted rates of  $H_2O_2$  production (Figs. 3, 5B, 5C, and 6C) were calculated as the root mean square of the calibration errors above and the propagated SEMs of the reporter level through the calibration equation. Exact equations are in [19]. Welch's *t* test was used to compare measured and predicted values in Figs. 3, 5B, 5C, and 6C.

## **Results**

#### **Oxygen consumption by skeletal muscle mitochondria oxidizing palmitoylcarnitine**

Table 1 summarizes the respiration rates of mitochondria on succinate plus rotenone and on three substrate combinations commonly used to assess β-oxidation. Respiration on succinate had appropriate high rates and respiratory control. Palmitoylcarnitine alone supported very slow respiration that was not stimulated by ADP or uncoupler, presumably because 3 ketoacyl thiolase, the last reaction of β-oxidation, was limited by the high acetyl-CoA/CoA ratio [32].

Addition of carnitine rescues this condition by accepting acetyl groups from acetyl-CoA to generate acetylcarnitine and CoA in a reaction catalyzed by carnitine acetyltransferase. The acetylcarnitine is then exported in exchange for incoming carnitine. Carnitine acetyltransferase activity is much higher in mitochondria from skeletal muscle (410 μmol · min<sup>-1</sup> · mg protein<sup>-1</sup>) than from liver (5 µmol · min<sup>-1</sup> · mg protein<sup>-1</sup>) [33], and there is correspondingly increased respiration on palmitoylcarnitine after addition of carnitine in mitochondria from skeletal muscle but not liver [24]. Table 1 shows that inclusion of carnitine stimulated oxidation of palmitoylcarnitine sufficiently to allow respiratory control. The combination of palmitoylcarnitine plus carnitine was attractive for this study because its metabolism may generate  $H_2O_2$  without overt involvement of Krebs cycle enzymes.

Addition of malate also rescues oxidation of palmitoylcarnitine, by generating oxaloacetate through malate dehydrogenase. Citrate synthase then condenses oxaloacetate and acetyl-CoA to form citrate, releasing CoA. Table 1 shows that palmitoylcarnitine plus malate supported substantial phosphorylating and uncoupled respiration and good respiratory control.

Thus, oxidation of palmitoylcarnitine, supplemented by either carnitine or malate, provides sufficient electron flux to establish these physiologically relevant substrate combinations as suitable for further investigation.

## **H2O2 production during oxidation of palmitoylcarnitine in the absence of respiratory chain inhibitors**

Fig. 1A shows that addition of palmitoylcarnitine at 5 min caused a measureable increase in the rate of Amplex UltraRed oxidation to resorufin; this was subsequently calibrated by additions of  $H_2O_2$  to give the rate of  $H_2O_2$  production. The open bars in Fig. 1B show the overall rates of  $H_2O_2$  production measured in this way during oxidation of each of the fatty acid substrate combinations in Table 1. Oxidation of palmitoylcarnitine alone produced  $H<sub>2</sub>O<sub>2</sub>$  at a low rate. Addition of malate or carnitine increased the rate several fold. The maximum rate was found with palmitoylcarnitine plus carnitine.

Next, we determined the individual sites of  $H_2O_2$  production with these substrate combinations. Even though respiration on palmitoylcarnitine alone was slow, we included it because, surprisingly, site  $II_F$  of complex II was the major  $H_2O_2$  producer under this condition (Figs. 3 and 5B below).

## **Complex II generates superoxide and/or H2O2 in the forward reaction during palmitoylcarnitine oxidation in the absence of respiratory inhibitors**

Addition of malonate, a specific inhibitor of succinate oxidation, significantly decreased the rates of  $H_2O_2$  production with palmitoylcarnitine (Fig. 1A and B) and with palmitoylcarnitine plus carnitine, but had little effect on the rate with palmitoylcarnitine plus malate (Fig. 1B, shaded bars). These observations can be explained by generation of superoxide or  $H_2O_2$  at complex II, which is sensitive to malonate during oxidation of succinate or glycerol3-phosphate in the presence of rotenone (to inhibit complex I) and myxothiazol (to inhibit complex III) in skeletal muscle mitochondria [17,18]. Malonate sensitivity of  $H_2O_2$ production during oxidation of palmitoylcarnitine has not been described before and is explored in more detail below.

We propose that under these conditions addition of palmitoylcarnitine ( $\pm$ carnitine) triggers superoxide or  $H_2O_2$  production from site  $II_F$  using electrons from endogenous succinate, but addition of palmitoylcarnitine plus malate does not. This hypothesis is supported by several observations.

First, measurements of complex II activity showed that palmitoylcarnitine± carnitine indirectly activated the enzyme, explaining the trigger mechanism. To calibrate, incubation of mitochondria with 5 mM succinate plus rotenone and myxothiazol at 37 °C gave high complex II activity (80±6 nmol DCPIP · min<sup>-1</sup> · mg · protein<sup>-1</sup>), and incubation without succinate gave low activity (40±4 units). Incubation with palmitoylcarnitine alone without rotenone or myxothiazol gave an activity of 64±8 units and incubation with palmitoylcarnitine plus carnitine gavean activity of 61±8 units (all values mean±SEM, *n*=3). This substantial activation of complex II by palmitoylcarnitine can be explained by removal of inhibitory oxaloacetate by condensation with acetyl-CoA formed by β-oxidation. In contrast, malate oxidation produces excess inhibitory oxaloacetate and prevented superoxide and/or  $H_2O_2$  production from site  $II_F$  when palmitoylcarnitine plus malate was used as substrate.

Second, atpenin A5, an inhibitor of the ubiquinone binding site of complex II [34], did not prevent  $H_2O_2$  production with palmitoylcarnitine as substrate, but stimulated it. It is known that H<sub>2</sub>O<sub>2</sub> production from site II<sub>F</sub> in the forward reaction from succinate is not prevented byatpenin A5 (unlike the reverse reaction from Q), but instead is stimulated as the flavin of complex II becomes more reduced [17]. Fig. 1A shows that addition of atpenin A5 increased the rate of  $H_2O_2$  production in the presence of palmitoylcarnitine alone, whereas malonate decreased the rate. The stimulated rate in the presence of atpenin A5 was abolished by subsequent addition of malonate (Fig. 1A), showing that it emanated from site  $II_F$  operating in the forward reaction from succinate (because the reverse reaction from  $QH<sub>2</sub>$  was inhibited by atpenin A5). The stimulation of  $H_2O_2$  production by subsequent addition of myxothiazol is addressed below (Fig. 5). Therefore, malonate but not atpenin A5 was used under native conditions below to inhibit  $H_2O_2$  production from site  $II_F$ .

We conclude from Fig. 1 that site  $II_F$  in the forward reaction produced  $\sim$ 25% of the observed  $H<sub>2</sub>O<sub>2</sub>$  during oxidation of palmitoylcarnitine plus carnitine and ~75% during oxidation of palmitoylcarnitine alone. When malate was present and the substrate-binding site of

complex II was inhibited by oxaloacetate, oxidation of palmitoylcarnitine did not result in significant H<sub>2</sub>O<sub>2</sub> production from site II<sub>F</sub>. After correction for small changes in H<sub>2</sub>O<sub>2</sub> production from sites  $I_F$  and  $III_{O}$  after addition of malonate (see Fig. 3 below), the contribution of site IIF become 36% for palmitoylcarnitine plus carnitine and 44% for palmitoylcarnitine alone.

#### **Quantitative contributions of sites IF, IIIQo, and IIF to superoxide and/or H2O2 production during oxidation of palmitoylcarnitine**

Endogenous reporters can predict the rates of superoxide production from sites  $I_F$  and  $III<sub>Oo</sub>$ in the absence of site-specific respiratory inhibitors (e.g., rotenone, antimycin A, myxothiazol, or stigmatellin) [19]. We used this technique in this work. We measured the reduction state of endogenous NAD(P) (Fig. 2A), to report superoxide production from site  $I_F$  (plus any other sites that respond to NADH reduction state), and cytochrome  $b_{566}$  (Fig. 2B), to report superoxide production from site  $III<sub>Qo</sub>$ . The corresponding calibration curves are shown as insets in Fig. 2A and B. Table 2 presents the reduction states of NAD(P) and cytochrome  $b_{566}$ under the conditions of Fig. 1B and the rates of  $H_2O_2$  production from sites  $I_F$  and  $III<sub>O<sub>O</sub></sub>$  predicted from the calibration curves.

The effect of malonate addition on the reduction of the reporters and the predicted rates of  $H_2O_2$  production is also presented in Table 2. Malonate had no discernible effect on the predicted rate from site I<sub>F</sub>, but it lowered the predicted rate from site  $III<sub>OO</sub>$  with palmitoylcarnitine alone and raised it with palmitoylcarnitine plus carnitine (although these effects were not statistically significant). These data allowed the raw decrease in  $H_2O_2$ production rate at site  $II_F$  caused by addition of malonate (Fig. 1B) to be corrected for consequent changes in  $H_2O_2$  production rate at sites I<sub>F</sub> and  $III_{O_0}$ . The corrected values for site  $II_F$  are shown in Table 2.

For each substrate combination, Fig. 3 shows the predicted rates of  $H_2O_2$  production from sites I<sub>F</sub>, III<sub>Oo</sub>, and II<sub>F</sub> from Table 2 (black, gray, and striped bars) and compares them to the empirical rates (open bars) from Fig. 1B. During oxidation of palmitoylcarnitine alone, site II<sub>F</sub> was the dominant site, responsible for 44% of the total predicted rate of  $H_2O_2$ production. Site I<sub>F</sub> produced 34% and site  $III<sub>Oo</sub>$  produced 22% of the total predicted rate. When palmitoylcarnitine was oxidized with carnitine supplementation, the contributions from sites I<sub>F</sub> (29%), III<sub>Oo</sub> (35%), and II<sub>F</sub> (36%) were approximately equal. When palmitoylcarnitine plus malate was used as substrate, site  $II_F$  did not contribute (because it was inhibited by the added malate). Site  $III<sub>Oo</sub>$  was the major superoxide producer (75%), with the remaining 25% contributed by site I<sub>F</sub>.

In each case, the predicted rates fully accounted within experimental error for the empirical rates, demonstrating that essentially all the  $H_2O_2$  produced during oxidation of palmitoylcarnitine in the absence of respiratory chain inhibitors came from these three sites, and any contribution from other sites was below the detection limit of this methodology.

Thus, we have identified and quantified the native sites of superoxide or  $H_2O_2$  production during fatty acid oxidation by rat skeletal muscle mitochondria in the absence of respiratory chain inhibitors and absence of ATP synthesis. Site  $II<sub>F</sub>$  of complex II was identified as one

of the major contributors. The ETF/ETFQOR system, other enzymes of the β-oxidation spiral, and site  $I_Q$  were not shown to be major sources of  $H_2O_2$  under any condition examined, because there were no significant differences between the observed rates and the sum of the rates predicted by the endogenous reporters or prevented by inhibition of site  $II_F$ .

#### **Rates of H2O2 production with palmitoylcarnitine plus antimycin A**

We now switch from conditions with the Q pool relatively oxidized to studies with complex III inhibited to keep the  $QH<sub>2</sub>/Q$  ratio high. This condition was used in earlier studies and revealed very high rates of  $H_2O_2$  production from unknown sites. It was proposed to originate from complexes upstream of the Q pool such as ETF, ETFQOR, and acyl-CoA dehydrogenase [20,22,24]. We did not find that these enzymes produced measureable superoxide or  $H_2O_2$  in the absence of respiratory inhibitors (above). Here we explore if reduction of the Q pool can cause high rates of  $H_2O_2$  production from these or other upstream centers to reveal the mechanism of superoxide and/or  $H_2O_2$  production from fatty acid oxidation-specific enzymes.

Fig. 4A shows that addition of antimycin A, an inhibitor of the  $Q_i$  site of complex III, caused very rapid  $H_2O_2$  production with palmitoylcarnitine plus carnitine as substrate. Rates with palmitoylcarnitine alone were similar, but rates with palmitoylcarnitine plus malate were lower (Fig. 4B, open bars).

We estimated the contributions of sites  $III<sub>Oo</sub>$  and  $II<sub>F</sub>$  in each case. Antimycin A inhibition of the  $Q_i$  site enhances superoxide production from site  $III_{Qo}$ , which is prevented by  $Q_o$  site inhibitors, including myxothiazol [27,35–37]. Fig. 4 shows that addition of myxothiazol strongly inhibited  $H_2O_2$  production with all three substrate combinations, showing that site IIIQo was a major contributor under these conditions, as expected. However, with palmitoylcarnitine  $\pm$  carnitine, myxothiazol inhibition was incomplete, suggesting that sites other than  $III<sub>O</sub>$  were also generating  $H<sub>2</sub>O<sub>2</sub>$ . The contribution of site II<sub>F</sub> to the residual rates was estimated by adding malonate. This largely abolished  $H_2O_2$  production (Fig. 4), suggesting that most of the myxothiazol-insensitive rate originated not from the ETF system but from complex II.

Thus, the majority of  $H_2O_2$  production by muscle mitochondria incubated with palmitoylcarnitine  $\pm$  carnitine in the presence of antimycin A arose from sites III<sub>Oo</sub> and II<sub>F</sub>. The important contribution from site  $II_F$  explains why there was a lower overall rate and a greater proportional contribution from site  $III<sub>O<sub>0</sub></sub>$  in the presence of malate (Fig. 4B): malate completely inhibited this contribution of site IIF.

To quantify fully the contributions of each site in the presence of antimycin A, any effects of myxothiazol and malonate on the reduction state of site  $I_F$  should be measured using the changes in NAD(P)H and corrected for, as in Table 2 and Fig. 3 (complex III stays fully reduced under these conditions, so its contribution to the signal does not change when malonate is added). Unfortunately, the strong fluorescence of antimycin A at 450 nm overlaps NAD(P)H autofluorescence, making it hard to correct for changes at site I<sub>F</sub> accurately with antimycin A present. Therefore, we used myxothiazol alone in all following experiments.

#### **Rates of H2O2 production with palmitoylcarnitine plus myxothiazol**

Fig. 5 shows the rates of  $H_2O_2$  production with myxothiazol alone (to prevent superoxide formation at site  $III<sub>Oo</sub>$  yet still allow correction for any changes in superoxide production at site  $I_F$  after addition of inhibitors). These rates were slower than those with antimycin A plus myxothiazol (Fig. 4), which can be explained by more complete inhibition of complex III and higher  $QH<sub>2</sub>/Q$  ratio when both inhibitors were present. Addition of malonate very strongly inhibited the rates with palmitoylcarnitine  $\pm$  carnitine, but not with palmitoylcarnitine plus malate. Measurement of  $NAD(P)^+$  reduction state allowed prediction of the rates from site  $I_F$  under each condition using the calibration curve in Fig. 2A (inset) and correction of the rates from site  $II_F$  for the small changes in the rate from site  $I_F$  on addition of malonate.

Fig. 5B reports the contributions of sites I<sub>F</sub> and II<sub>F</sub> to H<sub>2</sub>O<sub>2</sub> production in the presence of myxothiazol. With palmitoylcarnitine $\pm$  carnitine,  $H_2O_2$  production was dominated by site  $II_F$ , which accounted for ~70% of the total measured rates. Site  $I_F$  produced only a small amount,~4% of the totals. When palmitoylcarnitine plus malate was used as substrate, site  $II_F$  did not contribute (because it was inhibited by the added malate), and site  $I_F$  was the major superoxide producer (68% of the total measured rate).

In the presence of myxothiazol,  $H_2O_2$  production from site II<sub>F</sub> with palmitoylcarnitine was depressed by addition of carnitine (Fig. 5B). This can be explained by a measured decrease in the activity of complex II from 58.2±6.1 to 45.6±6.5 nmol DCPIP min−1 mg protein−1(*n*=3) under these conditions. This was presumably because decreased flux through the ETF pathway in the presence of myxothiazol stalled β-oxidation and lowered acetyl-CoA production, allowing removal of acetyl-CoA by carnitine and accumulation of inhibitory oxaloacetate.

From Fig. 1A we concluded that site  $II_F$  produced  $H_2O_2$  in the forward reaction from succinate during palmitoylcarnitine oxidation under native conditions, because the rate was inhibited by malonate (which inhibits at the flavin site) but stimulated by atpenin A5 (which inhibits at the Q-binding site of complex II). However, the situation was different in the presence of antimycin A (Fig. 4A) or myxothiazol (Fig. 5A), when the Qpool was much more reduced. In each case, atpenin A5 inhibited  $H_2O_2$  production as effectively as malonate, showing that under these conditions the route of electrons into site  $II_F$  was by the reverse reaction, from the Q pool.

Importantly, unlike the native condition (Fig. 3), in the presence of myxothiazol there was a clear and significant difference between the empirical rate and the sum of the predicted rates of H<sub>2</sub>O<sub>2</sub> production, with palmitoylcarnitine plus either carnitine or malate as substrate (Fig. 5B). This  $H_2O_2$  production was not from site  $I_F$ ,  $I_O$ ,  $II_F$ , or  $III_{O_0}$ , because these sites were either accounted for or inhibited directly (or, for site  $I<sub>O</sub>$ , inhibited indirectly by abolishing proton-motive force).

To investigate this unassigned rate more closely, in Fig. 5C we analyzed the residual rate after myxothiazol addition when complex II was inhibited from the start of the experiment (i.e., malonate added together with substrate). This avoided errors introduced in correcting

for contributions from other sites and led to a more significant difference between the empirical and the predicted rates, even though NAD(P) was more reduced after myxothiazol addition (because the escape of electrons to complex II was prevented), and superoxide production from site  $I_F$  was increased (Fig. 5C).

Thus, complex II produces superoxide and/or  $H_2O_2$  at very high rates by the reverse reaction at site II<sub>F</sub> using electrons from the Q pool when mitochondria oxidize palmitoylcarnitine  $\pm$ carnitine in the presence of antimycin A or myxothiazol. This novel finding has to be taken into account in interpreting other studies using fatty acids as substrate. In the presence of palmitoylcarnitine plus carnitine, malonate, and myxothiazol there was a highly significant rate of  $H_2O_2$  production that was not accounted for by site I<sub>F</sub>, I<sub>O</sub>, II<sub>F</sub>, or III<sub>Oo</sub> (Fig. 5C); we argue below that it originated in the ETF/ETFQOR system.

#### **Maximum rate of H2O2 production by the putative ETF/ETFQOR system**

We searched for conditions that maximized  $H_2O_2$  production dependent on the presence of palmitoylcarnitine that was unattributed to characterized sites. High rates were obtained in the presence of the uncoupler FCCP. This approach had two benefits. (i) Uncoupling oxidized the electron transport chain and therefore enzymes involved in electron supply were not inhibited by their products. Subsequent addition of myxothiazol caused a strong reduction of the Q pool and upstream redox centers  $(NAD(P)^+$ , ETFQOR, ETF). (ii) Uncoupling compromised the antioxidant defense systems by decreasing NAD(P)H formation by energy-dependent transhydrogenase and reduction of the glutathione pool [38].

In uncoupled mitochondria respiring on palmitoylcarnitine plus carnitine in the presence of malonate to inhibit site  $II_F$ , myxothiazol addition caused a high initial rate of  $H_2O_2$ production. This decreased to a lower rate after a few minutes (Fig. 6A). The possible mechanism of this biphasic kinetics is discussed below. Calculation of the contribution of site  $I_F$  required a calibration curve obtained under the same conditions as a compromised antioxidant defense system. Fig. 6B shows the dependence of  $H_2O_2$  production from site I<sub>F</sub> on NAD(P)+ reduction state in the presence of FCCP. The calibration curve superimposed that obtained after treatment of mitochondria with1-chloro-2,4-dintrobenzene (CDNB) to deplete glutathione [19], consistent with the interpretation that FCCP also compromised the glutathione pool.

Fig. 6C shows the empirical rates in the fast and slow phases and the predicted contributions of site  $I_F$ , the only characterized site that was active under these conditions. The final nonpredicted rate of H<sub>2</sub>O<sub>2</sub> production in the fast phase was 210±11 pmol · min<sup>-1</sup> · mg protein<sup>-1</sup>, which probably originated from the ETF/ETFQOR system.

Fig. 7 compares the rate that we have assigned to ETF/ETFQOR to the maximum rates of superoxide/ $H_2O_2$  production from the other characterized sites. The maximum rate from ETF/ETFQOR was much less than the maximum capacities of sites  $III<sub>00</sub>$ ,  $I<sub>0</sub>$ , and  $II<sub>F</sub>$ , but comparable to the maximum capacities of site  $I_F$  and glycerol 3-phosphate dehydrogenase.

## **Discussion**

We have characterized the rates and sites of origin of  $H_2O_2$  production in isolated rat skeletal muscle mitochondria oxidizing substrates of β-oxidation under two conditions: the absence of respiratory inhibitors (Fig. 3) and with complex III inhibited to reduce the Q pool

(Figs. 4 and 5B).

In contrast to previous studies of fatty acid oxidation (see the introduction) we used endogenous reporters in the absence of respiratory inhibitors to pinpoint sites of superoxide and/or  $H_2O_2$  production under native conditions. We found that the flavin site of complex I (site I<sub>F</sub>) (plus any other sites that respond to NADH reduction state) and the  $Q_0$  site of complex III (site  $III<sub>OO</sub>$ ) produced much of the superoxide during oxidation of palmitoylcarnitine. For the first time, the flavin site of complex II (site  $II_F$ ) was found to produce superoxide and/or  $H_2O_2$  in the forward reaction at substantial rates during oxidation of palmitoylcarnitine  $\pm$  carnitine (Fig. 3). There was no significant contribution by any other site.

With complex III inhibited using antimycin A, site  $III<sub>Oo</sub>$  was the major site, and site  $II<sub>F</sub>$ produced superoxide and/or  $H_2O_2$  at a high rate in the reverse reaction from the reduced Q pool (unless it was inhibited by added malate; Fig. 4B). When complex III was inhibited by myxothiazol to eliminate site  $III<sub>Oo</sub>$ , site  $II<sub>F</sub>$  was dominant, with only a small contribution from site  $I_F$  (Fig. 5B). Another site, probably the ETF/ETFQOR system, contributed significantly to  $H_2O_2$  production under these conditions, with a maximum rate of 210 pmol  $\cdot$ min−1· mg protein−1under optimal conditions (Fig. 6C).

Now we will consider each site in turn:

#### **Site I<sub>F</sub>**

Superoxide production by complex I (plus any other sites that respond to NADH reduction state) increases at higher NADH/NAD<sup>+</sup> ratio [19,28,39–41]. During β-oxidation, hydroxyacyl-CoA dehydrogenase converts one NAD<sup>+</sup> to NADH per turn of the cycle, making site  $I_F$  an appealing superoxide-producing candidate during fatty acid oxidation. However, oxidation of palmitoylcarnitine alone reduces  $NAD<sup>+</sup>$  to a much lower extent than that of NAD-linked substrates such as glutamate plus malate, so the contribution of site  $I_F$ has been thought to be negligible [24]. We found similar results  $(15\% \text{ NAD(P)}^+)$  reduction with palmitoylcarnitine (Table 2) and  $85\%$  NAD(P)<sup>+</sup> reduction with glutamate plus malate [19]), presumably because oxidation of palmitoylcarnitine was restricted by a high acetyl-CoA/CoA ratio, which inhibits the last enzyme of β-oxidation, 3-ketoacyl thiolase [32]. Nonetheless, Fig. 3 shows that site  $I_F$  was a major contributor to the low rate of superoxide production with palmitoylcarnitine alone as substrate.

Addition of carnitine or malate to mitochondria respiring on palmitoylcarnitine removes acetyl-CoA. This greatly increased NAD(P) reduction (Table 2) and thus superoxide production from site I<sub>F</sub>. However, the relative contribution of site I<sub>F</sub> was not much changed because of increased production at other sites as well (Fig. 3).

When complex III was inhibited, strong reduction of the Q pool caused other sites to increase superoxide or  $H_2O_2$  production greatly, and the relative contribution of site I<sub>F</sub> was correspondingly decreased (Figs. 4B, 5B, and 6C). However, when these other sites were inhibited by addition of myxothiazol and malate or malonate, site  $I_F$  was a major contributor to the remaining low rates (Figs. 5B and 5C).

#### **Site I<sup>Q</sup>**

Superoxide is produced at site  $I<sub>O</sub>$  a thigh rates when the proton-motive force and pH gradient are large, both during reverse electron flow from a reduced Q pool during oxidation of succinate [21,42] or glycerol3-phosphate [17,18,43]and during forward electron flow from NADH in the presence of certain  $I<sub>O</sub>$  inhibitors [42,44]. Because oxidation of fatty acids can reduce  $NAD<sup>+</sup>$  through hydroxyacyl-CoA dehydrogenase and the Q pool through acyl-CoA dehydrogenase and maintain a high proton-motive force and pH gradient, it has been speculated that oxidation of fatty acids smaysupport superoxide production at site I O [20,23,42]. We found no evidence for production of superoxide at site  $I<sub>O</sub>$  during oxidation of fatty acids, because there was no significant rate unattributable to sites  $I_F$ ,  $II_F$ , and  $III<sub>O<sub>O</sub></sub>$ under native conditions (Fig. 3). In the presence of complex III inhibitors the proton-motive force will have collapsed, so site  $I<sub>Q</sub>$  is not a candidate for the unattributed  $H<sub>2</sub>O<sub>2</sub>$  production rates in Figs. 5B, C, and 6C. Our results do not support the hypothesis that fatty acid oxidation generates superoxide at site  $I<sub>O</sub>$ , in agreement with the conclusions of others [24,45].

## **Site IIIQo**

Superoxide is produced during oxidation of  $QH<sub>2</sub>$  at the  $Q<sub>0</sub>$  site of complex III [46,47] at a rate that depends on the QH $_2$ /Q ratio [27]. During oxidation of fatty acids the Q pool can be reduced by complex I, complex II, and ETFQOR. Cytochrome  $b_{566}$ , which reports the reduction state of the Q pool [19,27], was more reduced with palmitoylcarnitine plus malate (48%; Table 2) than with glutamate plus malate (37%) [19], leading to high rates of superoxide production from site  $III<sub>Oo</sub>$ . In this study superoxide production from  $III<sub>Oo</sub>$  made the greatest individual contribution to total  $H_2O_2$  production under native conditions with palmitoylcarnitine plus carnitine or malate (Fig. 3) and when complex III was inhibited by antimycin A (Fig. 4B). Its high contribution had to be eliminated by addition of myxothiazol to allow analysis of the smaller rates from the ETF/ETFQOR system (Figs. 5 and 6). Site  $III<sub>Oo</sub>$  was also identified in previous studies using antimycin A [20,22–24].

#### **Site II<sub>F</sub>**

This is the first report of  $H_2O_2$  production from complex II during fatty acid oxidation. Under native conditions, inhibition by malonate was diagnostic of the contribution of complex II to the observed rate of  $H_2O_2$  production (Fig. 1). Site II<sub>F</sub> was a major contributor unless it was inhibited by addition of malate (Fig. 3). In contrast, the Q-binding-site inhibitor atpenin A5 increased  $H_2O_2$  production with palmitoylcarnitine. This  $H_2O_2$  production was abolished by subsequent addition of malonate (Fig. 1A), implicating forward electron flow into the flavin site of complex II. This effect can be explained by activation of complex II by the added palmitoylcarnitine generating acetyl-CoA and consuming endogenous inhibitory

oxaloacetate, allowing endogenous succinate to cause  $H_2O_2$  production at site II<sub>F</sub> in the forward reaction under our experimental conditions.

When complex III was inhibited, site  $II_F$  was a major contributor to the high observed rates of  $H_2O_2$  production (Fig. 4B) and the dominant contributor when site III<sub>Oo</sub> was inhibited by myxothiazol (Fig. 5B). Both malonate and atpenin A5 strongly inhibited myxothiazolinduced  $H_2O_2$  production (Figs. 4 and 5A), strongly supporting the idea that complex II can produce superoxide and/or  $H_2O_2$  at high rates in the reverse reaction from a reduced Q pool during fatty acid oxidation.

#### **Other sites**

When electron flow through complex III was abolished and sites  $I_F$ ,  $I_O$ ,  $II_F$ , and  $III_{O}$  were inhibited or fully accounted for, there remained a significant rate of  $H_2O_2$  production in the presence of palmitoylcarnitine plus carnitine or malate (Figs. 5B, 5C, and 6C). The possible candidates are other enzymes directly connected to the Q pool, including ETFQOR, glycerol 3-phosphate dehydrogenase, dihydroorotate dehydrogenase, sulfide ubiquinone oxidoreductase, and proline dehydrogenase, as well as enzymes of the Krebs cycle or βoxidation, including pyruvate and 2-oxoglutarate dehydrogenases, ETF, and acyl-CoA dehydrogenase.

The best candidate for the unassigned  $H_2O_2$  production is the ETF/ETFQOR system, for the following reasons. (i) There is a requirement for substrates of  $\beta$ -oxidation, and these will reduce ETF/ETFQOR but not the other candidates by forward electron flow. (ii) Most candidates that might use electrons from the Q pool are present at rather low concentrations in skeletal muscle mitochondria. This is not true for glycerol 3-phosphate dehydrogenase, but it is not known to use electrons from the Q pool to generate  $H_2O_2$  in the absence of glycerol 3-phosphate [18], making its participation unlikely. Skeletal muscle energy metabolism relies on β-oxidation, and therefore muscle mitochondria have high expression of the appropriate enzymes [48–50]. There is relatively high expression of ETFQOR (about 7% of total FAD-containing enzymes in the mitochondrial inner membrane [48]) and a larger ETF pool (63 pmol · mg protein−1 in skeletal muscle compared to brain mitochondria, which contain only 9 pmol · mg protein<sup>-1</sup> of ETF [49]). Altogether, flavoproteins of βoxidation account for 80% of the total content of mitochondrial flavoenzymes [50]. (iii) Compared to other complexes in the respiratory chain (complex I [51,52], complex II [53], glycerol 3-phosphate dehydrogenase [54,55], and dihydroorotate dehydrogenase [56,57]), ETFQOR has the most positive redox potentials (+27 mV for  $FAD_{1e}$ -/FAD<sub>ox</sub> and +47 mV for  $[FeS]^{2+}/[FeS]^{1+}$  [58,59]. Therefore it is thermodynamically favorable for a reduced Q pool to reduce ETFQOR and possibly ETF. (iv) Isolated ETFQOR is kinetically reversible [13] and submitochondrial particles catalyze the reduction of ETF by NADH and succinate under anaerobic conditions [60]. Thus ETFQOR can reduce ETF using electrons from the Qpool.

We now discuss the possible sites within the ETF/ETFQOR system.

#### **Acyl-CoA dehydrogenase**

Acyl-CoA dehydrogenases, in contrast to acyl-CoA oxidases, have low reactivity toward molecular oxygen [10], because of desolvation of the active site [61]. Moreover, the reduced medium-chain acyl-CoA dehydrogenase complexed with enoyl-CoA product is essentially devoid of oxygen reactivity [61,62]. Thus, acyl-CoA dehydrogenase is not a likely candidate.

#### **ETF and ETFQOR**

Identification of the species within the ETF/ETFQOR system that generates superoxide or  $H<sub>2</sub>O<sub>2</sub>$  is challenging. We were unable to obtain the most potent known inhibitor of ETFQOR, 2-(3-methylpentyl)-4,6-dinitrophenol [63], to test this question directly.

We hypothesize that ETF is the site of superoxide and/or  $H_2O_2$  production. Several observations support this hypothesis. (i) The one-electron reduction of ETF by acyl-CoA dehydrogenase is fast [12,64]. (ii) The half-reduced species of isolated ETF produces superoxide and  $H_2O_2$  at high rates [4]. (iii) The redox couple  $FAD_{1e}/FAD_{0x}$  in ETF has a more negative redox potential  $(+4 \text{ mV } [65])$  than the FAD in ETFQOR  $(+27 \text{ mV, see}$ above), making ETF the better reductant of oxygen to superoxide or  $H_2O_2$ . (iv) The crystal structure of ETFQOR predicts low reactivity towards oxygen at all redox centers. The hydrophobic Q-binding pocket of the enzyme allows Q to penetrate deeply. If electron transfer to Q involves semiquinone as a transient intermediate, the semiquinone is protected from molecular oxygen. The FAD and FeS clusters are deeply buried and therefore unlikely to be reactive [66].

The biphasic kinetics of  $H_2O_2$  production in Fig. 6A are probably explained by semiflavin ETF<sup> $-$ </sup> being the source of superoxide and  $H_2O_2$ . As ETF gradually becomes reduced in the presence of a high QH<sub>2</sub>/Q ratio, ETF<sup>-</sup> builds up, causing high initial rates of H<sub>2</sub>O<sub>2</sub> production. This is followed by an abrupt switch to a second phase of slower  $H_2O_2$ production, probably caused by fully reduced flavin ETF2− as the whole ETF pool becomes fully reduced by acyl-CoA dehydrogenase. Two-electron reduction of ETF may occur either by acyl-CoA dehydrogenase at a much lower rate [12,64] or by disproportionation catalyzed by ETFQOR [12,59]. However, further experiments are required to establish the superoxide/ $H_2O_2$ -producing species within the ETF/ETFQOR system.

#### **Conclusions**

We have identified multiple sites of  $H_2O_2$  and/or superoxide production during fatty acid oxidation in isolated skeletal muscle mitochondria. We conclude that when the Q pool is fairly oxidized, fatty acid oxidation in nonphosphorylating mitochondria produces superoxide and/or  $H_2O_2$  mainly at site I<sub>F</sub>, site II<sub>F</sub> in the forward reaction, and site III<sub>Oo</sub>, and the contribution of ETF/ETFQOR is negligible. However, reduction of the Q pool by inhibition of complex III causes high rates of superoxide and/or  $H_2O_2$  production from site  $III<sub>00</sub>$ , site  $II<sub>F</sub>$  in the reverse reaction, ETF/ETFQOR, and site I<sub>F</sub>. The contribution of each of these sites to superoxide and  $H_2O_2$  production in cells during fatty acid oxidation under physiological or pathological conditions remains to be determined.

The reduction state of the Q pool was found to be crucial for the identification of the sites and mechanisms of superoxide and/or  $H_2O_2$  production in isolated skeletal muscle mitochondria. However, it is difficult to correlate our in vitro conditions with in vivo studies because of the limited number of in vivo measurements of the reduction state of the Q pool in relevant systems. Direct measurements of flash-frozen samples of spontaneously contracting atria indicate that the Q pool is 40% oxidized under this condition [67]. The reduction states of cytochromes  $b_{1/h}$ , *c*, and  $aa_3$  were measured in isolated pulmonary artery smooth muscle cells under normoxia and hypoxia [68]. Under hypoxia, cytochrome *c* and cytochrome  $aa_3$  were reduced, whereas cytochrome  $b_{1/h}$  was oxidized. This is in a good agreement with a mechanism of ubiquinol oxidation by complex III [27] and suggests that the Q pool stays nearly fully reduced as it cannot be oxidized by complex III because of the lack of oxidized cytochrome *c.* Interestingly, under the hypoxia condition, higher rates of mitochondrial superoxide production were observed using the superoxide-sensitive dye MitoSOX [68]. Therefore, our novel in vitro findings can provide a firm foundation for future investigations of site-specific ROS production in muscle under physiological conditions (e.g., rest vs exercise) and in diseases featuring altered lipid metabolism.

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## **Abbreviations**



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#### **Fig. 1.**

Effects of complex II inhibitors, malonate and atpenin A5, on rates of  $H_2O_2$  production by rat skeletal muscle mitochondria oxidizing palmitoylcarnitine alone and in combination with carnitine or malate. (A) Typical traces of  $H_2O_2$  production during palmitoylcarnitine oxidation and effects of complex II inhibitors. Palmitoylcarnitine (PC) was added after 5 min incubation of mitochondria (0.3 mg protein · ml<sup>-1</sup>) with the Amplex UltraRed detection system plus exogenous SOD. The background rates in the absence of substrate were subtracted from all rates after substrate addition. Numbers by the traces represent rates of H<sub>2</sub>O<sub>2</sub> production in pmol · min<sup>-1</sup> · mg protein<sup>-1</sup> after calibration by addition of known amounts of H<sub>2</sub>O<sub>2</sub>. The addition of atpenin A5 increased the rate of H<sub>2</sub>O<sub>2</sub> production during oxidation of palmitoylcarnitine from  $27 \pm 1$  (*n*=6) to  $105 \pm 5$  (*n*=4) pmol · min<sup>-1</sup> · mg protein−1. After addition of malonate it decreased to 7±1 pmol · min−1 · mg protein−1 (*n*=4). Subsequent numbers are rates for the presented traces  $(n=1)$ . (B) Effects of malonate addition on the rates of  $H_2O_2$  production during oxidation of substrates from Table 1. Open bars, no malonate added; shaded bars, 0.5 mM malonate added after substrate. Values are means  $\pm$  SEM. *n*=4 (PC  $\pm$  malonate), *n*=6 (PC + carnitine  $\pm$  malonate), *n*=5 (PC + malate  $\pm$ malonate). *p* values were calculated using an unpaired Student *t* test.



#### **Fig. 2.**

Reduction states of NAD(P) and cytochrome  $b_{566}$  as reporters of superoxide production from sites I<sub>F</sub> (plus any other sites that respond to NADH reduction state) and  $III<sub>O<sub>0</sub></sub>$ , respectively. Typical traces of (A) NAD(P)H autofluorescence and (B) cytochrome  $b_{566}$ absorbance in suspensions of skeletal muscle mitochondria (0.3 mg protein · ml<sup>-1</sup> in (A) and 1.5 mg protein · ml<sup>-1</sup> in (B)). The oxidized signal was established after 5 min incubation in the absence of substrates. Substrates were then added to establish the reduction state to be measured (in the example illustrated, 15 μM palmitoylcarnitine (PC) and 5 mM malate), followed by the addition of inhibitors  $(2 \mu M$  myxothiazol in  $(A)$ ). 100% reduction was achieved by addition of 4  $\mu$ M rotenone (A) or 2  $\mu$ M antimycin A (B) in each run to calibrate the scale. 100% reduction in (B) was sometimes measured instead by adding 5 mM succinate plus 4 μM rotenone and 2 μM antimycin A and this value was used to calculate intermediate % reduction with other substrates during the experimental day. Insets are from [19]. Inset (A), calibration of the rate of H<sub>2</sub>O<sub>2</sub> production at site I<sub>F</sub> (pmol·min<sup>-1</sup>·mg) protein<sup>-1</sup>) as a function of % reduction of NAD(P). Inset (B), calibration of the rate of H<sub>2</sub>O<sub>2</sub> production at site  $III<sub>Qo</sub>(pmol · min<sup>-1</sup> · mg protein<sup>-1</sup>)$  as a function of % reduction of cytochrome  $b_{566}$ .



#### **Fig. 3.**

Comparison of the measured rate of  $H_2O_2$  production and the sum of the predicted rates from sites  $I_F$ ,  $II_F$ , and  $III<sub>O<sub>0</sub></sub>$  under native conditions. Measured rates are from Fig. 1B and predicted rates are from Table 2. *p* values were determined by Welch's *t* test. The contribution of site  $II_F$  was corrected for changes at the other sites using the changes in the reduction states of NAD(P)<sup>+</sup> and cytochrome  $b_{566}$  upon addition of malonate. PC, palmitoylcarnitine.



#### **Fig. 4.**

 $H<sub>2</sub>O<sub>2</sub>$  production during oxidation of palmitoylcarnitine (PC) in the presence of the complex  $III<sub>Oi</sub>$  inhibitor antimycin A. (A) Typical traces of  $H<sub>2</sub>O<sub>2</sub>$  production during oxidation of palmitoylcarnitine plus carnitine and effects of inhibitors. Substrates and inhibitors were added as shown. Numbers by the traces represent rates of  $\rm H_2O_2$  production in pmol  $\cdot$  min $^{-1}$   $\cdot$ mg protein−1 (*n*=6). (B) Total (open bars), myxothiazol-sensitive (gray bars), and malonatesensitive (cross-hatched bars) rates of  $H_2O_2$  production during oxidation of palmitoylcarnitine alone and in combination with carnitine or malate, all in the presence of antimycin A. Values are means  $\pm$  SEM ( $n=6$ ).



#### **Fig. 5.**

Measured and predicted rates of  $H_2O_2$  production during oxidation of palmitoylcarnitine (PC) in the presence of the complex  $III<sub>Qo</sub>$  inhibitor myxothiazol. (A) Typical traces of  $H<sub>2</sub>O<sub>2</sub>$ production during oxidation of palmitoylcarnitine plus carnitine and effects of inhibitors. Substrates and inhibitors were added as shown. Numbers by the traces represent rates of H<sub>2</sub>O<sub>2</sub> production in pmol · min<sup>-1</sup> · mg protein<sup>-1</sup> (*n*=6). (B) Total measured rates of H<sub>2</sub>O<sub>2</sub> production (open bars) and predicted rates from sites  $I_F$  (plus any other sites that respond to NADH reduction state) (black bars; predicted from the NAD(P) reduction state using the

calibration in Fig. 2A inset) and  $II_F$  (striped bars, predicted from the decrease in rate caused by addition of malonate after correction for changes in the predicted rate from site  $I_F$  after addition of malonate) during oxidation of palmitoylcarnitine alone and in combination with carnitine or malate, all in the presence of myxothiazol. Values are means  $\pm$  SEM ( $n=6$ ).  $p$ values were calculated by Welch's *t* test. (C) Total measured rates of H<sub>2</sub>O<sub>2</sub> production (open bar) and predicted rates from site  $I_F$  (black bar; predicted as in (B)) during oxidation of palmitoylcarnitine alone and in combination with carnitine or malate, all in the presence of myxothiazol and malonate. Values are means  $\pm$  SEM ( $n=6$ ).  $p$  values were calculated by Welch's *t* test.



#### **Fig. 6.**

H2O2 production during oxidation of palmitoylcarnitine (PC) plus carnitine plus malonate in the presence of the uncoupler FCCP. (A) Typical trace of  $H_2O_2$  production in the presence of 4 μM FCCP during oxidation of palmitoylcarnitine plus carnitine plus malonate and effect of addition of myxothiazol. The trace shows biphasic behavior: a first phase of high rate (labeled "fast") and a second phase of slower rate (labeled "slow"). Numbers by the trace represent rates of H2O2 production in pmol · min−1 · mg protein−1 (*n*=6). (B) Calibration of  $H_2O_2$  production at site I<sub>F</sub> (plus any other sites that respond to NADH reduction state) as a

function of reduction state of NAD(P)H in the presence of 4 μM FCCP (circles) (see Experimental procedures for details). The corresponding calibration in CDNB-treated mitochondria is also shown for comparison (triangles; data from [19]). (C) Total measured rates of  $H_2O_2$  production (open bars) and predicted rates from site I<sub>F</sub> (black bars; predicted using the calibration curve in (B)) during oxidation of palmitoylcarnitine plus carnitine, malonate, and myxothiazol. Values are means  $\pm$  SEM ( $n=6$ ).  $p$  values were calculated using Welch's *t* test.



#### **Fig. 7.**

Maximum rates of  $H_2O_2$  production from characterized sites of the mitochondrial respiratory chain. I<sub>F</sub>, flavin site of complex I (plus any other sites that respond to NADH reduction state);  $I<sub>O</sub>$ , ubiquinone binding site of complex I; II<sub>F</sub>, flavin site of complex II; IIIQo, Qo binding site of complex III; mGPDH, mitochondrial glycerol 3-phosphate dehydrogenase; ETF/ETFQOR, electron-transferring flavoprotein/electron-transferring flavoprotein–ubiquinone oxidoreductase system. Data are from [19] except that for ETF/ ETFQOR, which is from Fig. 6C.

#### **Table 1**

Oxygen consumption by rat skeletal muscle mitochondria.



Mitochondria were incubated at 0.3 mg protein · ml<sup>-1</sup> as described under Experimental procedures. State 2 respiration was measured after substrate addition. Subsequent additions were 0.5 mM ADP (to initiate State 3), 1 μg · ml<sup>−1</sup> oligomycin (State 4<sub>O</sub>), and 1 μM FCCP (uncoupled respiration). Values are means ±SEM for *n* = 3 independent experiments.

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# **Table 2**

Rates of  $H_2O_2$  production from sites I<sub>F</sub> (plus any other sites that respond to NADH reduction state),  $\text{III}_{\text{O}0}$ , and II<sub>F</sub> predicted from the reduction states of Rates of H<sub>2</sub>O<sub>2</sub> production from sites I<sub>F</sub> (plus any other sites that respond to NADH reduction state), III<sub>Qo</sub>, and II<sub>F</sub> predicted from the reduction states of NAD(P)H and cytochrome  $b_{566}$  and the inhibition by malonate in rat skeletal muscle mitochondria with various substrate combinations. NAD(P)H and cytochrome *b*566 and the inhibition by malonate in rat skeletal muscle mitochondria with various substrate combinations.



curves in Fig. 2A and B (insets). To calculate the corrected rates from site IIF, the measured decreases in H2O2 production rate caused by addition of malonate in Fig. 1B were corrected for the changes in rate from sites I<sub>F</sub> and IIIO<sub>O</sub> predicted from the changes in reduction state of the endogenous reporters ± malonate. Values are means ± SEM (n independent experiments). The SEM of the predicted rates of rate from sites IF and IIIQo predicted from the changes in reduction state of the endogenous reporters ± malonate. Values are means ± SEM (*n* independent experiments). The SEM of the predicted rates of curves in Fig. 2A and B (insets). To calculate the corrected rates from site IIF, the measured decreases in H2O2 production rate caused by addition of malonate in Fig. 1B were corrected for the changes in H2O2 production from sites IF and III Qo were calculated by propagation of calibration and measurement errors using the equations in [19]. The SEM of the corrected rate from site IIF was calculated as  $^2$ ), where SEM<sub>a</sub>, SEM<sub>b</sub>, and SEM<sub>C</sub> represent the errors of the predicted rates from sites I<sub>F</sub> and IIIQ<sub>0</sub> before and after malonate addition and the SEM of the rate from H2O2 production from sites IF and IIIQo were calculated by propagation of calibration and measurement errors using the equations in [19]. The SEM of the corrected rate from site IIF was calculated as The reduction states of NAD(P) and cytochrome *b*566 were measured under the conditions of Figs. 1 and 2. The rates of H2O2 production from sites IF and IIIQ<sub>O</sub> were predicted from the calibration º<br>∍ יע ביטב site IIF assessed as the malonate-sensitive rate in Fig. 1B.  $\frac{1}{200}$  $2 +$ SEM<sub>c</sub>  $2 + \text{SEM}_b$  $\texttt{sEM} = \texttt{(SEM}_\texttt{a}$  $\frac{1}{2}$ 

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