

# Simultaneous Steady-state and Dynamic $^{13}\text{C}$ NMR Can Differentiate Alternative Routes of Pyruvate Metabolism in Living Cancer Cells<sup>\*[5]</sup>

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**Background:**  $^{13}\text{C}$  hyperpolarization sensitively and non-destructively detects pyruvate-lactate exchanges in cancer cells.

**Results:** Combining  $^{13}\text{C}$  hyperpolarization with isotopomer analysis allowed many pyruvate-dependent pathways to be quantified simultaneously.

**Conclusion:** Monitoring  $\text{H}[^{13}\text{C}]\text{O}_3^-$  production from hyperpolarized  $[1-^{13}\text{C}]\text{pyruvate}$  yielded a quantitative readout of onco-gene-regulated pyruvate dehydrogenase activity.

**Significance:** This approach might enable a broader quantitative assessment of metabolic activity in tumors.

Metabolic reprogramming facilitates cancer cell growth, so quantitative metabolic flux measurements could produce useful biomarkers. However, current methods to analyze flux *in vivo* provide either a steady-state overview of relative activities (infusion of  $^{13}\text{C}$  and analysis of extracted metabolites) or a dynamic view of a few reactions (hyperpolarized  $^{13}\text{C}$  spectroscopy). Moreover, although hyperpolarization has successfully quantified pyruvate-lactate exchanges, its ability to assess mitochondrial pyruvate metabolism is unproven in cancer. Here, we combined  $^{13}\text{C}$  hyperpolarization and isotopomer analysis to quantify multiple fates of pyruvate simultaneously. Two cancer cell lines with divergent pyruvate metabolism were incubated with thermally polarized  $[3-^{13}\text{C}]\text{pyruvate}$  for several hours, then briefly exposed to hyperpolarized  $[1-^{13}\text{C}]\text{pyruvate}$  during acquisition of NMR spectra using selective excitation to maximize detection of  $\text{H}[^{13}\text{C}]\text{O}_3^-$  and  $[1-^{13}\text{C}]\text{lactate}$ . Metabolites were then extracted and subjected to isotopomer analysis to determine relative rates of pathways involving  $[3-^{13}\text{C}]\text{pyruvate}$ . Quantitation of hyperpolarized  $\text{H}[^{13}\text{C}]\text{O}_3^-$  provided a single definitive metabolic rate, which was then used to convert relative rates derived from isotopomer analysis into quantitative fluxes. This revealed that  $\text{H}[^{13}\text{C}]\text{O}_3^-$  appearance reflects activity of pyruvate dehydrogenase rather

than pyruvate carboxylation followed by subsequent decarboxylation reactions. Glucose substantially altered  $[1-^{13}\text{C}]\text{pyruvate}$  metabolism, enhancing exchanges with  $[1-^{13}\text{C}]\text{lactate}$  and suppressing  $\text{H}[^{13}\text{C}]\text{O}_3^-$  formation. Furthermore, inhibiting Akt, an oncogenic kinase that stimulates glycolysis, reversed these effects, indicating that metabolism of pyruvate by both LDH and pyruvate dehydrogenase is subject to the acute effects of oncogenic signaling on glycolysis. The data suggest that combining  $^{13}\text{C}$  isotopomer analyses and dynamic hyperpolarized  $^{13}\text{C}$  spectroscopy may enable quantitative flux measurements in living tumors.

Cancer cells display metabolic properties that distinguish them from surrounding, normal tissue. Tumor metabolism is regulated by oncogenes and tumor suppressors (1–4), suggesting that mutations in these genes orchestrate metabolic activities to support an environment conducive to malignancy (5, 6). In accordance with this view, mutations in a subset of metabolic enzymes, including fumarate hydratase, the succinate dehydrogenase complex, and isocitrate dehydrogenases-1 and -2 cause significant perturbations in intermediary metabolism and contribute directly to cancer (7–11). Metabolism is also highly sensitive to the effects of the tumor microenvironment, particularly hypoxia, which correlates with chemotherapeutic resistance and poor patient survival (12). Thus, the ability to measure metabolic flux in live tumors would potentially report clinically and prognostically valuable information.

Imaging of hyperpolarized  $^{13}\text{C}$ -enriched substrates is emerging as a promising technology for cancer diagnosis and monitoring because it enables the real-time detection of enzyme-catalyzed metabolic activities in tissues (13–18). Recently, imaging has also been shown to be capable of detecting malignant lesions in the human prostate, even diagnosing masses that were not positively identified by other methods (19). These studies revealed that exchanges between hyperpolarized pyru-

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vate and lactate are readily detected in malignant cells and tend to be enhanced in tumors relative to nonmalignant tissue. Hyperpolarization involves the temporary redistribution of the populations of the available energy levels into a non-equilibrium state, enabling a massive gain in magnetic resonance signal. For  $^{13}\text{C}$ , this gain can exceed 10,000-fold (20), enabling detection of both a  $^{13}\text{C}$ -enriched substrate and some downstream metabolites with a temporal resolution of seconds. Hyperpolarized  $[1-^{13}\text{C}]$ pyruvate is widely used to interrogate cancer metabolism, in part because the long  $T_1$  of this carboxyl carbon allows the hyperpolarized state to persist for multiple metabolic steps and also because pyruvate is located in a pivotal intersection of intermediary metabolism involving lactate, alanine, and oxidation in the tricarboxylic acid (TCA) cycle.

Cancer cells often display high rates of glycolysis and lactate secretion even when oxygen is plentiful (the Warburg effect). This oncogene-driven phenomenon contributes to the utility of tumor detection by uptake of the glucose analog  $[^{18}\text{F}]$ fluoro-2-deoxyglucose (3). Abundant expression of lactate dehydrogenase (LDH) and the presence of a large lactate pool in tumor cells allows for rapid exchange between hyperpolarized  $[1-^{13}\text{C}]$ pyruvate and  $[1-^{13}\text{C}]$ lactate, making detection of  $[1-^{13}\text{C}]$ lactate appealing for *in vivo* detection of cancer and for monitoring response to therapy (15, 21). However, cancer cells also oxidize pyruvate in the mitochondria, producing both energy and macromolecular precursors for cell growth (23). This is of particular interest because lung tumors, gliomas, and metastatic brain tumors have all been demonstrated to oxidize pyruvate *in vivo* in humans and mice (24–28). Therefore, assessment of both pyruvate/lactate exchanges and pyruvate oxidation in the mitochondria would provide a much more comprehensive view of cancer cell metabolism than lactate formation alone.

We previously used conventional  $^{13}\text{C}$  NMR spectroscopy to evaluate fluxes through competing metabolic pathways supplied by pyruvate, including LDH and the TCA cycle, in cultured cancer cells (29, 30). These same activities were detected in mouse and human tumors by infusing  $^{13}\text{C}$ -enriched glucose before surgery, extracting metabolites from surgically resected tumor tissue, and analyzing  $^{13}\text{C}$  enrichment patterns by NMR (26, 28). We also used hyperpolarized  $[1-^{13}\text{C}]$ pyruvate to quantify flux into lactate (31). Here, we combined these methods to study two metabolically distinct cancer cell lines. First, we incubated cancer cells with thermally polarized  $[3-^{13}\text{C}]$ pyruvate for several hours to produce steady-state labeling of metabolic intermediates. Next, using a selective excitation pulse to maximize detection of  $\text{H}[^{13}\text{C}]\text{O}_3^-$  and  $[1-^{13}\text{C}]$ lactate, we subjected cells to hyperpolarized  $[1-^{13}\text{C}]$ pyruvate to measure flux into lactate and the TCA cycle. Combining the rate of pyruvate decarboxylation with steady-state isotopomer data provided a method to evaluate absolute flux rates through a variety of reactions associated with the TCA cycle.

## EXPERIMENTAL PROCEDURES

**Cell Culture Reagents and Basic Metabolism Experiments**—Two cell lines, SF188-derived glioblastoma cells overexpressing human Bcl-xL (SFxL) and Huh-7 hepatocellular carcinoma cells were maintained in culture as described previously (30, 32, 33). Metabolic experiments were performed in Dulbecco's

modified Eagle's medium (DMEM) prepared from powder lacking glucose, glutamine, phenol red, sodium pyruvate, and sodium bicarbonate. This basal medium was supplemented with 4 mmol/liter L-glutamine, 10% dialyzed fetal calf serum, 42.5 mmol/liter sodium bicarbonate, 25 mmol/liter HEPES, 10 units/ml penicillin, and 10  $\mu\text{g}/\text{ml}$  streptomycin. Glucose and pyruvate were added as indicated for each experiment. To measure the rates of metabolite consumption/excretion in the medium, glucose, lactate, glutamine, and glutamate were measured using a BioProfile Basic 4 analyzer (NOVA Biomedical), and ammonia was measured using a spectrophotometric assay (Megazyme). For oxygen consumption assays, cells were harvested by trypsinization, suspended in fresh medium at a concentration of  $10^8$  cells/ml, and transferred to an Oxygraph water-jacketed oxygen electrode (Hansatech). The Akt inhibitor was Akt Inhibitor VIII (Calbiochem).

**Pyruvate Decarboxylation Assay**—Decarboxylation of  $[1-^{14}\text{C}]$ pyruvate was measured essentially as described (34). Micro-bridges (Hampton Research) were placed into wells of a 24-well plate with one piece of  $0.6 \times 1 \text{ cm}^2$  chromatography paper in each. Assay medium was prepared by supplementing DMEM (containing 10% fetal calf serum, 4 mM glutamine, and 6 mM sodium pyruvate) with 2.2  $\mu\text{Ci}$  of  $[1-^{14}\text{C}]$ pyruvate. This medium was warmed to  $37^\circ\text{C}$  and incubated for 2 h to remove any  $^{14}\text{CO}_2$  produced from spontaneous decarboxylation, then an aliquot was used to quantify radioactivity on a scintillation counter. This value was used to determine the specific activity of pyruvate, assuming a total pyruvate concentration of 6 mM. The specific activity ranged from 50 to 120 cpm/nmol of pyruvate. One million cells per well were then suspended in 370  $\mu\text{l}$  of assay medium on ice. Each micro-bridge was moistened with 30  $\mu\text{l}$  of 2 N NaOH, and the plate was sealed with adhesive film. Pyruvate metabolism was initiated by transferring the plate to a  $37^\circ\text{C}$  water bath. After 15 min, metabolism was terminated by adding 50  $\mu\text{l}$  of 20% trichloroacetic acid. The plate was re-sealed with adhesive film and incubated at  $37^\circ\text{C}$  for another 60 min to release  $^{14}\text{CO}_2$  completely. Then the  $^{14}\text{CO}_2$ -containing chromatography papers were collected for scintillation counts. Pyruvate oxidation rates were determined from the total specific activity in the reaction and were reported as nmol/ $10^6$  cells/h. Wells containing culture medium but no cells were used to establish background levels of  $^{14}\text{CO}_2$ . This level was subtracted from cell-containing wells.

**Gas Chromatography/Mass Spectrometry**—Mass spectrometry experiments were performed essentially as described previously (30). Cultures of 80–90% confluent cells grown in 60-mm dishes were rinsed twice in phosphate-buffered saline then overlaid with DMEM-based medium containing 10% dialyzed fetal calf serum, 4 mM glutamine, and 6 mM pyruvate (either  $[1-^{13}\text{C}]$ pyruvate or  $[3-^{13}\text{C}]$ pyruvate; Cambridge Isotope Laboratories). At the end of the culture period, the medium was removed, and the cells were briefly rinsed in ice-cold normal saline. A cold solution of 50% methanol, 50% water was added, and the cells were lysed using repeated freeze-thaw cycles. After centrifugation to remove debris, the samples were evaporated and derivatized by trimethylsilylation (Tri-Sil HTP reagent; Thermo). An aliquot of 1–3  $\mu\text{l}$  was injected onto an Agilent 6970 gas chromatograph networked to an Agilent 5973 Mass

## Noninvasive Quantitation of Metabolic Flux in Cancer Cells

Selective Detector. Fragment ions  $m/z$  334–338 and 465–471 were used to monitor enrichment in aspartate and citrate, respectively (30). The measured distribution of mass isotopomers was corrected for natural abundance of  $^{13}\text{C}$  as described (35).

**NMR Spectroscopy of Extracted Metabolites**—Cells were cultured to 80–90% confluence in eight 150-mm dishes. Freshly prepared DMEM with 10% dialyzed fetal calf serum, 0 mM glucose, 4 mM glutamine, standard concentrations of other amino acids, and 6 mM  $[1-^{13}\text{C}]$ pyruvate was added to each dish, and the cells were cultured for 6 h, then either harvested immediately for extraction or collected for hyperpolarization experiments. For hyperpolarization, the cells were trypsinized, pelleted by centrifugation, and resuspended at a concentration of  $1 \times 10^8$  cells/ml in fresh DMEM lacking glucose and pyruvate but containing 10% dialyzed fetal calf serum, 4 mM glutamine, and standard concentrations of other amino acids. This suspension was mixed thoroughly, then transferred to a 10-mm NMR tube that already contained hyperpolarized  $[1-^{13}\text{C}]$ pyruvate so that the final concentration was 6 mM, precisely mimicking the nutrient availability of the steady-state labeling experiment. Acquisition of spectra was initiated immediately and proceeded for ~3.5 min (see “Hyperpolarization, Shaped Pulse, and NMR Spectroscopy in Intact Cells” below for details on cell transfer and conditions for the hyperpolarization experiment). We previously reported that cells subjected to trypsinization and hyperpolarized  $[1-^{13}\text{C}]$ pyruvate maintain rates of LDH flux similar to rates of adherent cells (31). To protect against depletion of oxygen from the medium during trypsinization and preparation of the cell suspension, the medium was mixed frequently, including immediately before transfer of the cells to the NMR tube, which occurred within a few seconds of introduction of  $[1-^{13}\text{C}]$ pyruvate. Immediately after the hyperpolarization experiment, the cells were pelleted by centrifugation and frozen in liquid nitrogen. Frozen cell pellets were homogenized by sonication in 4 ml of 4% ice-cold perchloric acid. After centrifuging to remove debris, the acid-soluble material was neutralized with 8 N potassium hydroxide and centrifuged. Supernatants were subjected to lyophilization, reconstituted in deuterium oxide, and titrated to pH 7 for NMR analysis in a 3-mm tube.

NMR spectroscopy was performed on a Varian INOVA 14.1 T spectrometer (Agilent Instruments, Walnut Creek, CA) equipped with a 3-mm broadband probe with the observe coil tuned to  $^{13}\text{C}$  (150 MHz). Proton decoupling was performed using a standard WALTZ-16 pulse sequence. Carbon spectra were acquired under the following conditions: pulse flip angle  $45^\circ$ , repetition time 1.5 s, spectral width 35 kHz, number of data points 104,986, and number of scans 7,000–30,000, requiring 6–25 h. Free induction decays were zero-filled to 131,072 points and apodized with exponential multiplication. Peak areas were determined using ACD Labs SpecManager (Advanced Chemistry Development, Toronto, Canada). Glutamate isotopomer analysis to calculate  $F_{\text{C}_2}$  (the fraction of acetyl-CoA labeled in position-2 with  $^{13}\text{C}$ ) in Fig. 3 was performed using equations described in Malloy *et al.* (36). Relative flux values for the combined steady-state/hyperpolarization experiments in Table 1, including  $F_{\text{C}_2}$

and  $y_s$  (the rate of anaplerosis relative to citrate synthase flux), were calculated by tcaCALC (37).

**Hyperpolarization, Shaped Pulses, and NMR Spectroscopy of Intact Cells**—A solution of  $[1-^{13}\text{C}]$ pyruvic acid containing 15 mmol/liter OX63 radical was prepared for polarization. 8.6  $\mu\text{l}$  of the solution was polarized for 2 h in an Oxford HyperSense dynamic nuclear polarization system (Oxford Instruments Molecular Biotech Ltd, Oxfordshire, UK) at 1.4 K with a microwave irradiation frequency of 94.125 GHz at 100 milliwatts. The frozen sample was dissolved with 4 ml of 15.3 mM sodium bicarbonate (heated to  $190^\circ\text{C}$  and pressurized to 10 bar) in less than 10 s. An aliquot of the resulting solution was added to a 10-mm NMR tube (400  $\mu\text{l}$  for 2 ml total volume or 200  $\mu\text{l}$  for 1 ml total volume). The tube was placed in a Varian 10-mm broadband probe in a 9.4 tesla magnet equipped with a VNMR5 console (Agilent). The NMR tube was directly attached by a thin tube to a syringe containing the suspension of cells. At time zero of the acquisition, either 1.6 or 0.8 ml of the cell suspension (for final total volumes of 2 or 1 ml, respectively) was added to the hyperpolarized solution, resulting in a final pyruvate concentration of 6 mM and a cell concentration of  $1.0 \times 10^8$  cells/ml.

Frequency selective pulses were utilized to avoid depolarization of the pyruvate C1 resonance. Pulses were created in the PBox software provided with VNMRJ (Agilent). The pulse profile was designed to excite the pyruvate C2 resonance (205.9 ppm), lactate C1 (183.3 ppm), and  $\text{H}[^{13}\text{C}]\text{O}_3^-$  (160.1 ppm) with a Gaussian shape of narrow bandwidth (350–500 Hz). Profiles were fit to a triple Gaussian function,

$$\text{Signal} = A \left[ \exp\left(\frac{-(x - b_1)^2}{2c_1^2}\right) + \exp\left(\frac{-(x - b_2)^2}{2c_2^2}\right) + \exp\left(\frac{-(x - b_3)^2}{2c_3^2}\right) \right] + d \quad (\text{Eq. 1})$$

where  $A$  is the amplitude of excitation,  $b$  is the frequency location of each Gaussian shape,  $c$  is related to the width of the Gaussian shape such that the width at half-maximum height is  $2(2\ln(2))^{1/2}c$ , and  $d$  is the signal amplitude in the absence of excitation; the maximum flip angle due to excitation is, therefore, inverse cosine( $A + d$ ). Selective pulses were applied every 2 s for a total of 100  $^{13}\text{C}$  spectra with an acquisition time of 1 s and a 1-s delay. Spectra were collected with a 32-kHz bandwidth and zero-filled before apodization and Fourier transformation. Using Bayesian Analysis software (38–41), the full set of 100 free induction decays were simultaneously fit to a five-resonance model with uncorrelated phase. Normalization of the peak areas was done using the pyruvate C2 resonance, and the lactate and bicarbonate areas were further adjusted to account for small differences in cell number among experiments.

In a separate set of experiments the decay of the signal due to radiofrequency depolarization was determined in the absence of cells using the selective pulse on a sample of hyperpolarized pyruvate. Each experiment began with 15  $1^\circ$  excitations with a 2-s repetition time followed by 15 selective pulses. The effective  $T_1$  was determined from the first set of pulses, and this value

$$\frac{d}{dt} \begin{bmatrix} \text{Pyr} \\ \text{Lac} \\ \text{MPyr} \\ \text{Bic} \end{bmatrix} = \begin{bmatrix} -k_{PL} - k_{PM} - \rho_P & k_{LP} & k_{MP} & 0 \\ k_{PL} & -k_{LP} - \rho_L & 0 & 0 \\ k_{PM} & 0 & -k_{MP} - k_{MB} - \rho_P & 0 \\ 0 & 0 & -k_{MB} & -\rho_B \end{bmatrix} \begin{bmatrix} \text{Pyr} \\ \text{Lac} \\ \text{MPyr} \\ \text{Bic} \end{bmatrix}$$

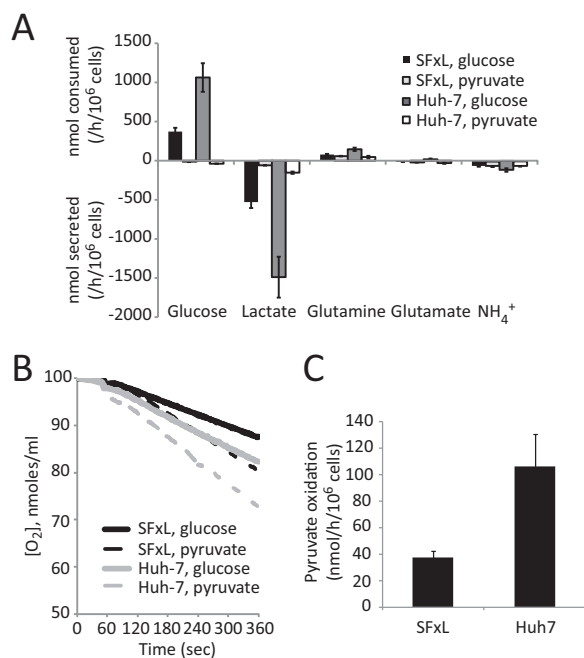
SCHEME 1

was used to extract the additional decay due to radiofrequency depolarization in the second set of 15 pulses. Both pulse profiles were tested with their corresponding sample volumes.

**Modeling**—The matrix shown in Scheme 1 was used to model flux as detected with hyperpolarized  $[1-^{13}\text{C}]$ pyruvate, where  $\rho_i = 1/T_{1i}$ . Pyr represents both extracellular and cytosolic  $[1-^{13}\text{C}]$ pyruvate, Lac represents total  $[1-^{13}\text{C}]$ lactate, MPyr is mitochondrial  $[1-^{13}\text{C}]$ pyruvate, and Bic is total  $\text{H}[^{13}\text{C}]\text{O}_3^-$ . Because the cells were injected into a waiting sample of HP pyruvate, they experience an instantaneous increase in pyruvate concentration, and there was no need for a time-dependent delivery term. Export of lactate from the cells is not included because previous experiments showed this to be unnecessary to obtain accurate flux values (31). Inclusion of a mitochondrial pool of pyruvate was required to adequately model the initial delay detected in the  $\text{H}[^{13}\text{C}]\text{O}_3^-$  time course. The model fits were done using 1000 sets of initial values as described previously (31). Radiofrequency depolarization rates were measured in cell-free phantom experiments, and  $T_1$  values were fit by the program.

To establish absolute fluxes through other pathways associated with the TCA cycle, the rate of appearance of  $\text{H}[^{13}\text{C}]\text{O}_3^-$  was used to reference all other TCA cycle fluxes. In SFxL cells, PDH<sup>4</sup> flux was taken to be equivalent to total  $\text{H}[^{13}\text{C}]\text{O}_3^-$  appearance as pyruvate cycling (PK) was inactive, and the contribution from oxidative reactions in the TCA cycle was assumed to be negligible (see “Results”). Because an isotopomer analysis yields metabolic fluxes relative to citrate synthase flux, absolute citrate synthase flux can be calculated as (absolute PDH flux)/(relative PDH flux). Remaining fluxes – the rate of anaplerosis via PC ( $y_{PC}$ ), the rate of entry of other unlabeled anaplerotic substrates into the cycle ( $y_S$ ), and flux through PK or other undefined pyruvate cycling pathways were obtained by multiplying the absolute citrate synthase flux by relative rates of these activities as derived from isotopomer analysis. Absolute cataplerosis was calculated by summing  $y_{PC}$  and  $y_S$  under steady state. Given that the parameter  $F_{C2}$  from isotopomer analysis in this experiment defines the fraction of acetyl-CoA that is derived from labeled pyruvate, the unlabeled fraction ( $F_{C0}$ ) from endogenous substrates is given by  $1 - F_{C2}$ . Thus, absolute fluxes from unlabeled substrates into the acetyl-CoA pool were calculated as  $(1 - F_{C2}) \times$  (absolute citrate synthase flux). In Huh-7 cells, a similar approach was taken, but additional calculations were used to account for the fact that PK was active. Due to symmetrization of  $^{13}\text{C}$ -labeled TCA cycle intermediates, the maximal amount of  $\text{H}[^{13}\text{C}]\text{O}_3^-$  that can be produced from pyruvate cycling is half the total flux through that pathway. Thus, total  $\text{H}[^{13}\text{C}]\text{O}_3^-$  production is equivalent to the sum of PDH flux and half of PK flux. The absolute PDH flux was

<sup>4</sup> The abbreviations used are: PDH, pyruvate dehydrogenase; PK, pyruvate kinase; OAA, oxaloacetate; PC, pyruvate carboxylase.



**FIGURE 1. Core metabolic activities in cancer cell lines with different routes of pyruvate oxidation.** A, consumption and secretion of major nutrients in SFxL glioblastoma and Huh-7 hepatocellular carcinoma cells. Cells were cultured in complete medium containing 10 mM glucose and no pyruvate or no glucose and 6 mM pyruvate. Data are the average and S.D. of three cultures. B, oxygen consumption in both cell lines. C, oxidation of  $[1-^{14}\text{C}]$ pyruvate to  $^{14}\text{CO}_2$  in both cell lines. Data are the average and S.D. of three cultures.

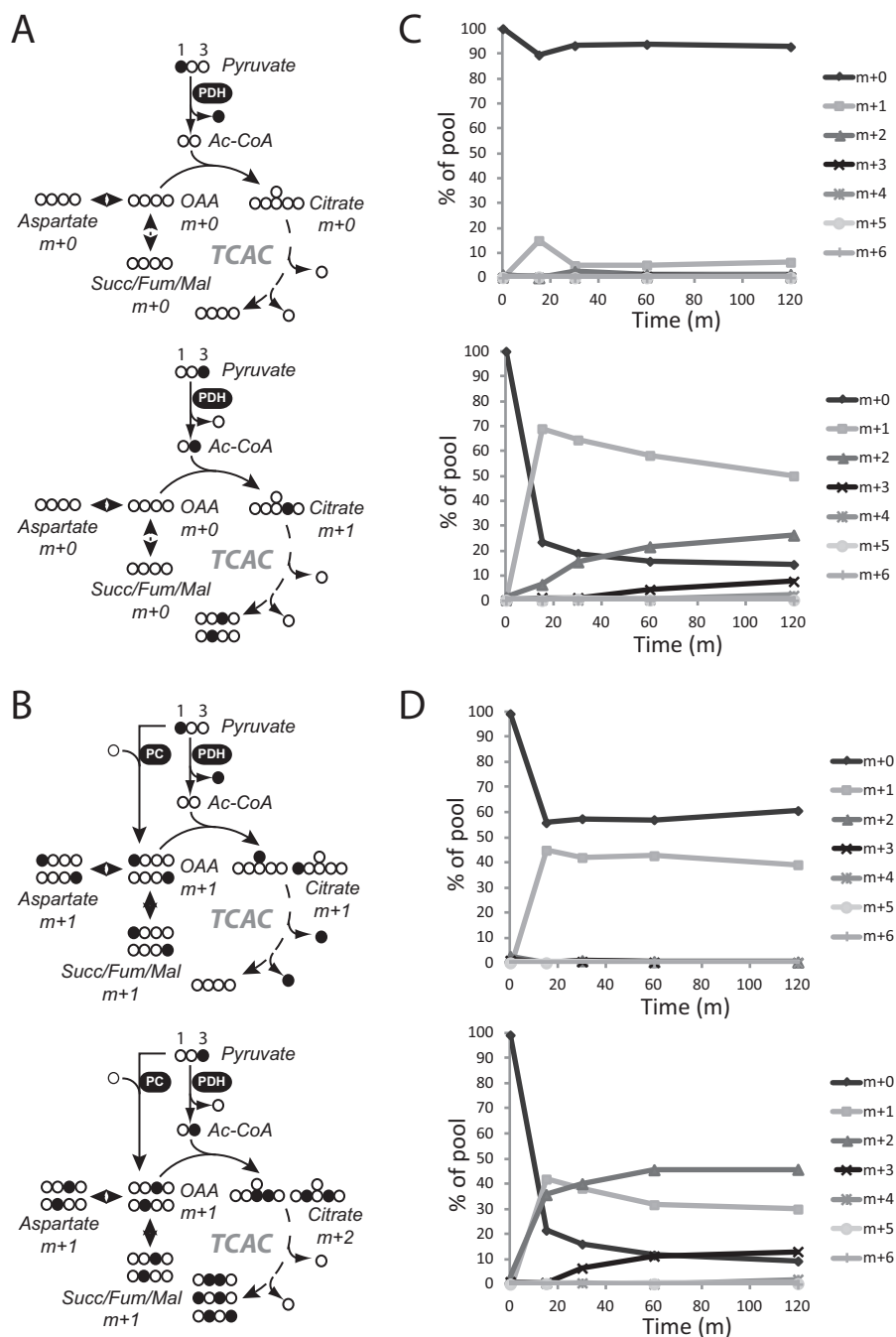
calculated as (relative PDH flux from isotopomer analysis)/(total  $\text{H}[^{13}\text{C}]\text{O}_3^-$  production). Other absolute fluxes were calculated as in SFxL cells.

## RESULTS

**Two Cancer Cell Lines with Distinct Pathways of Pyruvate Oxidation**—We chose SFxL glioblastoma cells and Huh-7 hepatocellular carcinoma cells for this work. Both cell lines grow rapidly in culture (doubling time  $22 \pm 0.5$  h in SFxL and  $29 \pm 0.1$  h in Huh-7). Under the non-physiological conditions commonly used in mammalian cell culture (here, 10% fetal calf serum, 10 mM glucose, 4 mM glutamine, and room-air levels of oxygen), both cell lines consume glucose and glutamine rapidly and secrete lactate, glutamate, and ammonia into the extracellular space (Fig. 1A). Oxygen consumption was evident under these conditions and persisted when glucose was replaced with 6 mM pyruvate, the same concentration used in experiments with hyperpolarized  $[1-^{13}\text{C}]$ pyruvate (Fig. 1B). To determine the extent to which these cells could decarboxylate the C-1 position of extracellular pyruvate to  $\text{CO}_2$ , cells were cultured with  $[1-^{14}\text{C}]$ pyruvate, and  $^{14}\text{CO}_2$  was measured by scintillation counting. Both cell lines generated significant amounts of  $^{14}\text{CO}_2$  by this method, with Huh-7 cells producing it at a rate approximately three times higher than SFxL cells (Fig. 1C).

There are multiple mitochondrial pathways by which the carboxyl group of pyruvate can be released as  $\text{CO}_2$ . First, PDH converts pyruvate to acetyl-CoA, releasing  $\text{CO}_2$  in the process (Fig. 2A). Alternatively, pyruvate can be carboxylated via PC to form oxaloacetate (OAA), and subsequent decarboxylation reactions involving either pyruvate cycling (42) or a forward

## Noninvasive Quantitation of Metabolic Flux in Cancer Cells



**FIGURE 2. Pyruvate rapidly supplies the TCA cycle in both SFxL and Huh-7 cells.** *A*, predicted labeling of TCA cycle intermediates when PDH is active and PC is inactive in cells labeled with  $[1-^{13}\text{C}]$ pyruvate (top) or  $[3-^{13}\text{C}]$ pyruvate (bottom). Filled symbols are  $^{13}\text{C}$ . *B*, predicted labeling of TCA cycle intermediates when both PDH and PC are active in cells labeled with  $[1-^{13}\text{C}]$ pyruvate (top) or  $[3-^{13}\text{C}]$ pyruvate (bottom). *C*, time course of citrate mass isotopomers in SFxL cells labeled with  $[1-^{13}\text{C}]$ pyruvate (top) or  $[3-^{13}\text{C}]$ pyruvate (bottom). *D*, time course of citrate mass isotopomers in Huh-7 cells labeled with  $[1-^{13}\text{C}]$ pyruvate (top) or  $[3-^{13}\text{C}]$ pyruvate (bottom). Abbreviations: TCAC, TCA cycle; Succ/Mal/Fum, succinate/malate/fumarate; OAA, oxaloacetate.

turn of the cycle will release  $\text{CO}_2$  (Fig. 2*B*). To maximize detection of the various possible routes of pyruvate metabolism in SFxL and Huh-7 cells, each cell line was cultured in a large excess of pyruvate, with either  $[1-^{13}\text{C}]$ pyruvate or  $[3-^{13}\text{C}]$ pyruvate at a concentration of 6 mM in glucose-free medium. At several time points, the cells were freeze-clamped, metabolites were extracted, and mass spectrometry was used to determine  $^{13}\text{C}$  enrichment in TCA cycle intermediates. If PDH is active, but PC is not, citrate and other TCA cycle intermediates will be labeled by  $[3-^{13}\text{C}]$ pyruvate but not by  $[1-^{13}\text{C}]$ pyruvate (Fig. 2*A*).

If both enzymes are active, then TCA cycle intermediates are labeled by both  $[3-^{13}\text{C}]$ pyruvate and  $[1-^{13}\text{C}]$ pyruvate (Fig. 2*B*). When cultured in  $[1-^{13}\text{C}]$ pyruvate, Huh-7 cells demonstrated rapid and persistent  $^{13}\text{C}$  labeling in citrate and aspartate, but SFxL cells did not, indicating that only Huh-7 cells had high levels of PC activity (Fig. 2, *C* and *D*, and supplemental Fig. 1). When cultured in  $[3-^{13}\text{C}]$ pyruvate, both cell lines transferred  $^{13}\text{C}$  to the TCA cycle (Fig. 2, *C* and *D*, and supplemental Fig. 1). However, citrate in SFxL cells was found to be labeled predominantly with one  $^{13}\text{C}$  ( $m+1$ ), whereas  $m+2$  and  $m+3$  appeared

only after a delay due to progression of citrate  $m+1$  through complete turns of the cycle to acquire additional label from acetyl-CoA. In contrast, Huh-7 cells rapidly produced citrate  $m+2$  and higher order labeling. These data demonstrate that SFxL cells possess PDH activity, whereas Huh-7 cells use both PDH and PC concurrently. These data are consistent with results from longer term cultures in which glucose instead of pyruvate was the source of  $^{13}\text{C}$  (29, 30).

To establish conditions that would support glutamate isotopomer analysis for determining relative flux values of pyruvate-dependent metabolic pathways, we cultured large populations of SFxL and Huh-7 cells in a DMEM-based medium containing 6 mM  $[3-^{13}\text{C}]$ pyruvate and 4 mM glutamine and 10% fetal calf serum. After 6 h of culture, the cells were flash-frozen, and metabolites were extracted for NMR analysis. The resulting  $^{13}\text{C}$  NMR spectra had excellent signal-to-noise and showed well-resolved multiplets in glutamate carbons 2, 3, and 4 (Fig. 3), indicating that a complete isotopomer analysis would be supported by these culture conditions. In this labeling scheme, the large singlet in glutamate C4 reflects PDH activity, and the additional multiplets in carbons 2, 3, and 4 indicate that OAA was labeled with  $^{13}\text{C}$ . Isotopomer analysis of spectra from each cell line revealed that under these conditions,  $[3-^{13}\text{C}]$ pyruvate was a major source of the acetyl-CoA used in the TCA cycle of both cell lines (70% in SFxL and nearly 60% in Huh-7).

**A Selective Pulse Technique to Maximize Detection of Pyruvate Oxidation**—Most experiments involving hyperpolarized  $[1-^{13}\text{C}]$ pyruvate in cancer cells report  $^{13}\text{C}$  exchanges between pyruvate and lactate but do not report production of carbon dioxide or bicarbonate. Having determined that SFxL and Huh-7 cells oxidize extracellular pyruvate within a few minutes, we tested whether this activity could be detected by  $^{13}\text{C}$  NMR using hyperpolarized pyruvate. A bolus of 6 mM hyperpolarized  $[1-^{13}\text{C}]$ pyruvate was introduced to a suspension of SFxL cells, and  $^{13}\text{C}$  NMR spectra were continuously acquired for several minutes. When a non-selective excitation pulse was applied, a summation of spectra revealed labeling of lactate C1 (Fig. 4A), as previously reported for these cells (31). Labeling in alanine C1 and in C1 of pyruvate hydrate was also detected, but  $\text{H}^{13}\text{C}[\text{O}_3^-]$  was not detected (Fig. 4A). Next, a frequency-selective excitation pulse was developed to optimize detection of  $\text{H}^{13}\text{C}[\text{O}_3^-]$ . This shaped pulse selectively excites  $[1-^{13}\text{C}]$ lactate and  $\text{H}^{13}\text{C}[\text{O}_3^-]$  while avoiding excitation of  $[1-^{13}\text{C}]$ pyruvate, thereby minimizing destruction of the hyperpolarized precursor (43). The pulse also excites  $[2-^{13}\text{C}]$ pyruvate, which appears as a small doublet at 206 ppm because of natural abundance  $^{13}\text{C}$  at C2 (Fig. 5A). This resonance was used to monitor the abundance of hyperpolarized pyruvate in the absence of C1 excitation. When cells were analyzed under identical conditions using the new frequency-selective pulse, both  $[1-^{13}\text{C}]$ lactate and  $\text{H}^{13}\text{C}[\text{O}_3^-]$  were detected in individual scans as well as in summed spectra (Fig. 4B). This allowed a comparison of the dynamic labeling of  $[1-^{13}\text{C}]$ lactate and  $\text{H}^{13}\text{C}[\text{O}_3^-]$  (Fig. 4C). The time-dependent appearance of  $\text{H}^{13}\text{C}[\text{O}_3^-]$  lagged behind the appearance of  $[1-^{13}\text{C}]$ lactate, consistent with the expected faster kinetics of LDH relative to the various decarboxylation pathways, all of which require transport of pyruvate into the mitochondria.

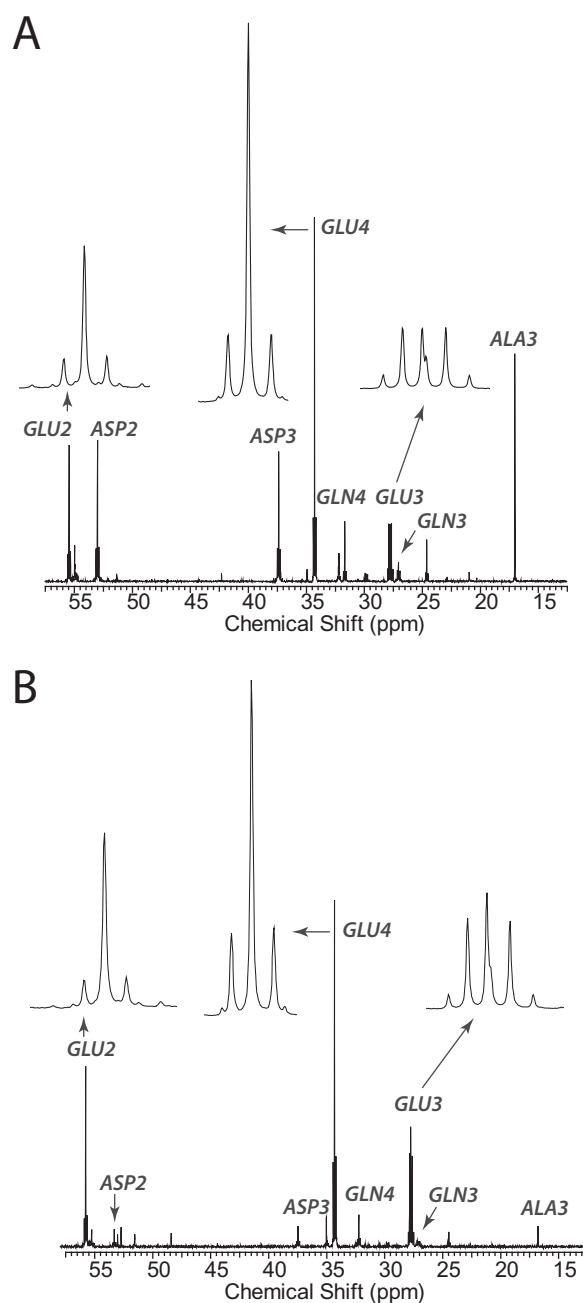
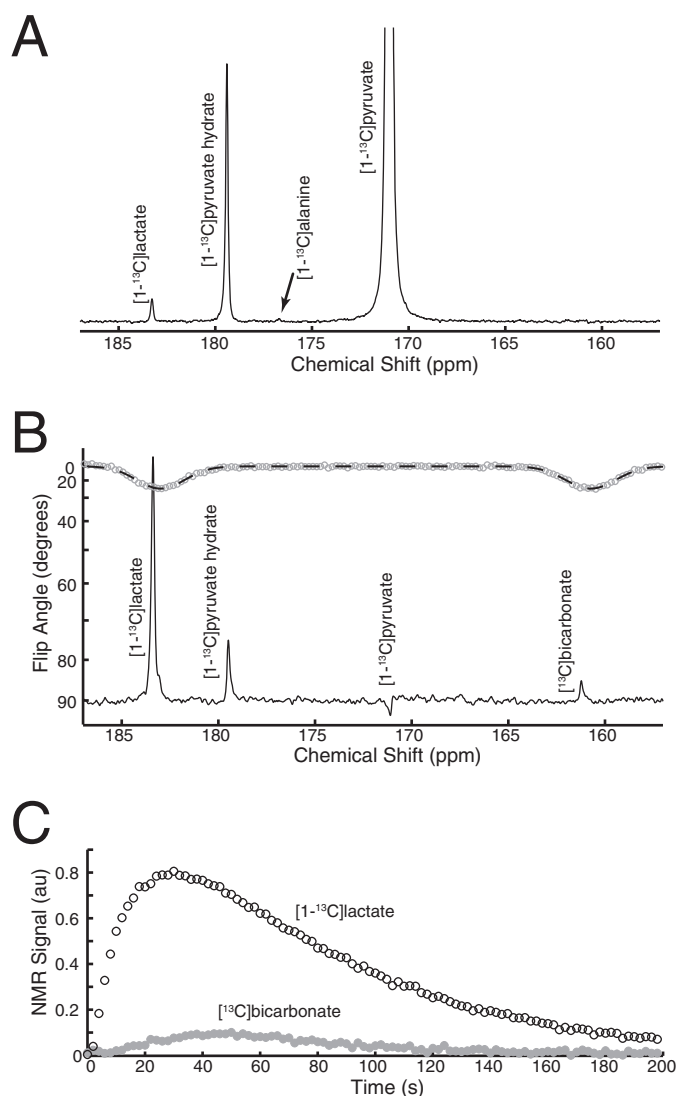


FIGURE 3.  $^{13}\text{C}$  NMR steady-state analysis of pyruvate metabolism. A, carbon 13 NMR spectrum of metabolites extracted from SFxL cells after 6 h of culture in  $[3-^{13}\text{C}]$ pyruvate. B, carbon 13 NMR spectrum of metabolites extracted from Huh-7 cells after 6 h of culture in  $[3-^{13}\text{C}]$ pyruvate. GLU, glutamate; ASP, aspartate; GLN, glutamine; ALA, alanine.

**Combining Dynamic Measures of Flux with Steady-state  $^{13}\text{C}$  Isotopomer Analysis**—If the sources of  $\text{H}^{13}\text{C}[\text{O}_3^-]$  are known on a fractional basis from  $^{13}\text{C}$  isotopomer data, then it should be possible to use the rate of appearance of hyperpolarized  $\text{H}^{13}\text{C}[\text{O}_3^-]$  to estimate fluxes through all pathways contributing to bicarbonate production. Because any decarboxylation reaction is essentially unidirectional, any ambiguities related to isotope exchange can be ignored. In the two cancer cell lines used here, hyperpolarized  $\text{H}^{13}\text{C}[\text{O}_3^-]$  would be predominantly generated by PDH in SFxL cells, whereas PDH and PC could both contribute to  $\text{H}^{13}\text{C}[\text{O}_3^-]$  production in Huh-7 cells. To obtain



**FIGURE 4. A selective pulse technique to detect oxidation of hyperpolarized pyruvate.** *A*, summed spectra of a suspension of SFxL cells cultured with hyperpolarized  $[1-^{13}\text{C}]$ pyruvate. NMR spectroscopy used a non-selective pulse. *B*, summed spectra of an identical suspension of SFxL cultured with hyperpolarized  $[1-^{13}\text{C}]$ pyruvate but analyzed with a shaped pulse designed to excite  $[1-^{13}\text{C}]$ lactate and  $[^{13}\text{C}]$ bicarbonate. This pulse also excites  $[2-^{13}\text{C}]$ pyruvate at 206 ppm. *C*, time course of  $[1-^{13}\text{C}]$ lactate and  $[^{13}\text{C}]$ bicarbonate signal in a suspension of SFxL cells analyzed using the shaped pulse.

quantitative flux data for the two cell lines, a two-stage labeling experiment that combined isotopomer analysis with dynamic hyperpolarization data was performed (Fig. 5A). First, each cell line was incubated in DMEM containing 6 mM  $[3-^{13}\text{C}]$ pyruvate, 4 mM glutamine, and 10% fetal calf serum for 6 h to achieve steady-state labeling of glutamate isotopomers. These same cells were then detached from culture dishes by brief exposure to trypsin and transferred to an NMR tube containing a solution of hyperpolarized  $[1-^{13}\text{C}]$ pyruvate, also at a pyruvate concentration of 6 mM. The cells were transferred in DMEM containing 4 mM glutamine and 10% fetal calf serum such that they were exposed to the same nutrient milieu experienced during the 6-h incubation. The NMR tube was prepositioned in a Varian 10-mm broadband probe tuned to  $^{13}\text{C}$ . Serial excitation with the triple-Gaussian pulse was immediately initiated, and  $^{13}\text{C}$  NMR spectra were acquired until disappearance of the

hyperpolarized  $^{13}\text{C}$  signal. At that point the cells were rapidly collected by centrifugation, frozen in liquid nitrogen, and extracted in acid to obtain the water soluble metabolites for isotopomer analysis. Dynamic labeling of both  $[1-^{13}\text{C}]$ lactate and  $\text{H}[^{13}\text{C}]\text{O}_3^-$  as well as the decay of signal from pyruvate C2 was apparent in both cell lines (Fig. 5, *B* and *C*).

According to a previously established kinetic model (31), the initial LDH flux in these experiments was  $190 \pm 40$  nmol/ $10^6$  cells/h in SFxL and  $380 \pm 166$  nmol/ $10^6$  cells/h in Huh-7 ( $n = 3$  for each cell line). Hyperpolarized  $\text{H}[^{13}\text{C}]\text{O}_3^-$  appeared more rapidly on average in the Huh-7 cells, although the difference did not approach the increase in  $^{14}\text{CO}_2$  release measured in Fig. 1C. To determine the rate of appearance of  $\text{H}[^{13}\text{C}]\text{O}_3^-$ , a multi-pool model was used to fit the hyperpolarization data. The model included exchanges between pyruvate and lactate, exchanges between cytosolic and mitochondrial pyruvate pools, and the conversion of mitochondrial pyruvate to  $\text{H}[^{13}\text{C}]\text{O}_3^-$ . The fitting protocol generated 200 sets of randomized  $k$  values as starting values. Each starting value was chosen at random in the range of 0–2, with no other restrictions. Within each experiment ( $n = 3$  for each cell line), the 50 best fits were essentially identical to each other. The top fit was used to calculate an overall rate of pyruvate decarboxylation (Fig. 6, *A* and *B*). When considering all six experiments, the rates of  $\text{H}[^{13}\text{C}]\text{O}_3^-$  appearance in SFxL and Huh-7 cells were  $38 \pm 11$  and  $49 \pm 20$  nmol/h/ $10^6$  cells, respectively. In SFxL cells, the averaged hyperpolarization data agreed closely with the  $^{14}\text{CO}_2$  experiment (39 versus 38 nmol/h/ $10^6$  cells). In contrast, despite excellent fits generated by the model, the rate of formation of hyperpolarized  $\text{H}[^{13}\text{C}]\text{O}_3^-$  in Huh-7 cells under-estimated the rate of  $^{14}\text{CO}_2$  measured in these cells (49 versus 100 nmol/h/ $10^6$  cells).

The data imply that hyperpolarization is unable to fully capture all decarboxylation reactions downstream of PC, possibly because of the time delay associated with these reactions and the rapid decay of the hyperpolarized signal. In contrast, the  $^{14}\text{C}$  experiments capture release of all  $^{14}\text{CO}_2$  from all reactions over the entire incubation period. To test the extent of PC-dependent  $^{14}\text{CO}_2$  formation, PC expression was silenced in SFxL and Huh-7 cells, and conversion of  $[1-^{14}\text{C}]$ pyruvate to  $^{14}\text{CO}_2$  was re-measured. As expected, it was possible to suppress  $^{14}\text{CO}_2$  production by partially silencing PC expression, and this effect was much greater in Huh-7 cells (supplemental Fig. 2). We conclude that production of hyperpolarized  $\text{H}[^{13}\text{C}]\text{O}_3^-$  from  $[1-^{13}\text{C}]$ pyruvate in cancer cells primarily reflects PDH flux.

NMR spectroscopy of  $^{13}\text{C}$ -labeled metabolites extracted from each sample produced well resolved spectra, similar to those in Fig. 3. Analysis of glutamate isotopomers in these spectra revealed that essentially all of the acetyl-CoA interacting with the TCA cycle was either  $[2-^{13}\text{C}]$ acetyl-CoA derived from  $[3-^{13}\text{C}]$ pyruvate or was unlabeled (44). Therefore, the introduction of hyperpolarized  $[1-^{13}\text{C}]$ pyruvate had a negligible effect on glutamate isotopomer labeling, as expected for the brief exposure of the cells to the hyperpolarized material. Isotopomer analysis was also used to calculate relative flux rates of numerous metabolic pathways interacting with the TCA cycle (36). Each pathway is normalized to citrate synthase flux given an arbitrary value of 1.0. As shown in Table 1, many of these

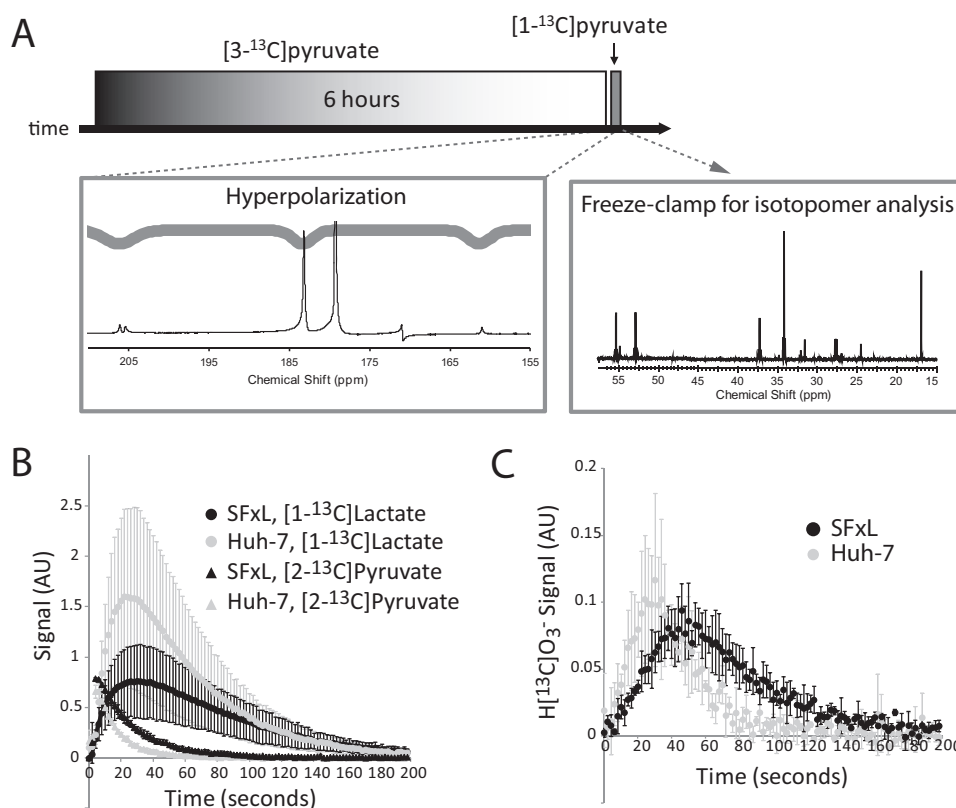


FIGURE 5. **Two-stage analysis of pyruvate metabolism in cancer cells.** *A*, overview of experimental scheme, detailed under "Results." An example spectrum (sum of 50 free induction decays, approximately 3 min of culture time) and triple-Gaussian pulse Mz-profile corresponding to the hyperpolarized  $[1-^{13}\text{C}]$ pyruvate experiment are shown on the *bottom left*. A representative  $^{13}\text{C}$  NMR spectrum from extracted metabolites, showing the effect of the 6-h culture in  $[3-^{13}\text{C}]$ pyruvate is shown in the *bottom right*. *B*, average and S.D. for hyperpolarization experiments ( $n = 3$ ) show the time course of detection of  $[1-^{13}\text{C}]$ lactate and the natural abundance signal from carbon 2 of pyruvate in SFxL cells and Huh-7 cells. *C*, time course of the average and S.D. ( $n = 3$ ) for  $\text{H}[^{13}\text{C}]\text{O}_3^-$  signal in SFxL and Huh-7. AU, arbitrary units.

activities were active in both cell lines. Relative PDH flux was similar in the two cell lines, consistent with the rapid labeling of citrate from  $[3-^{13}\text{C}]$ pyruvate observed by mass spectrometry (Fig. 2). Anaplerotic flux from unlabeled sources (given the symbol,  $y_S$ ) was active in both cell lines, consistent with the previous report that both cell lines use glutamine as an anaplerotic precursor (30). However, flux through pyruvate carboxylase ( $y_{PC}$ ) was much higher in Huh-7 than SFxL. Pyruvate cycling (PK), another pathway that may produce  $\text{CO}_2$  from pyruvate C1 via either malic enzyme or phosphoenolpyruvate carboxykinase, was essentially inactive in SFxL cells but modestly active in Huh-7 cells. The high PDH flux and low  $y_{PC}$  and pyruvate cycling (PK) in SFxL cells show that the great majority of pyruvate decarboxylation is PDH-dependent in these cells. On the other hand, PDH,  $y_{PC}$ , and PK likely all contribute to bicarbonate production in Huh-7 cells.

**Quantitative Models of Metabolic Flux from Single Extractions**—Finally, we used information from both hyperpolarization and isotopomer analysis to develop quantitative models of metabolic flux in these two cell lines. Quantitative determination of one or more fluxes should make it possible to reference other reactions from the isotopomer analysis to it, thereby determining flux values throughout the network. We considered the potential routes by which  $\text{H}[^{13}\text{C}]\text{O}_3^-$  could be formed from  $[1-^{13}\text{C}]$ pyruvate. Each molecule of  $[1-^{13}\text{C}]$ pyruvate metabolized by PDH generates one  $\text{H}[^{13}\text{C}]\text{O}_3^-$ . By contrast, only a fraction of PC-dependent

metabolism of  $[1-^{13}\text{C}]$ pyruvate generates  $\text{H}[^{13}\text{C}]\text{O}_3^-$ . PC converts  $[1-^{13}\text{C}]$ pyruvate to  $[1-^{13}\text{C}]\text{OAA}$ , and this may equilibrate with the malate, fumarate, and succinate pools, distributing up to half of the label to  $[4-^{13}\text{C}]\text{OAA}$ . Citrate synthase will then transfer approximately half of the  $^{13}\text{C}$  originating on  $[1-^{13}\text{C}]$ pyruvate to  $[6-^{13}\text{C}]\text{citrate}$  and half to  $[1-^{13}\text{C}]\text{citrate}$ . Release of  $^{13}\text{C}$  from  $[6-^{13}\text{C}]\text{citrate}$  occurs at the isocitrate dehydrogenase step, but release from  $[1-^{13}\text{C}]\text{citrate}$  does not occur until  $\alpha$ -ketoglutarate dehydrogenase. Importantly, when a non-selective pulse is applied to either of these cell lines in the presence of hyperpolarized  $[1-^{13}\text{C}]$ pyruvate, neither  $[6-^{13}\text{C}]\text{citrate}$  nor  $[1-^{13}\text{C}]\text{citrate}$  was observed (data not shown). Considering that both cell lines have appreciable citrate pools (Fig. 2), the lack of observable  $[^{13}\text{C}]\text{citrate}$  suggests that forward cycling does not contribute significantly to the  $\text{H}[^{13}\text{C}]\text{O}_3^-$  signal. Even the perfused liver, which contains abundant PC activity, produces only minimal amounts of observable hyperpolarized  $[6-^{13}\text{C}]\text{citrate}$  (45). Pyruvate cycling may also contribute to decarboxylation of pyruvate at C1. These pathways involve the return of OAA to the pyruvate pool either through decarboxylation by phosphoenolpyruvate carboxykinase or by conversion to malate and decarboxylation by malic enzyme. Again, because of randomization of the  $^{13}\text{C}$  label between the 1 and 4 positions of OAA and malate, only a fraction of  $^{13}\text{C}$  can be released as  $\text{H}[^{13}\text{C}]\text{O}_3^-$  on the first turn.



## Noninvasive Quantitation of Metabolic Flux in Cancer Cells

In SFxL cells, PC is much less active than PDH, and pyruvate cycling is inactive (Fig. 2 and Table 1). Thus, the rate of hyperpolarized  $\text{H}^{13}\text{C}[\text{O}_3^-]$  formation in SFxL cells ( $38 \pm 11 \text{ nmol}/10^6 \text{ cells}/\text{h}$ ) is equivalent to the PDH flux. Using the relative flux data derived from isotopomer analysis, PDH flux was used to derive quantitative flux values for PC, anaplerosis/cataplerosis,

citrate synthase/TCA cycle turnover, and the rates at which substrates other than labeled pyruvate feed the acetyl-CoA pool (Fig. 7A). In Huh-7 cells, PC and pyruvate cycling are active in addition to PDH. Assuming that both PDH and PC/pyruvate cycling can produce hyperpolarized  $\text{H}^{13}\text{C}[\text{O}_3^-]$ , we used data from the isotopomer analysis to calculate the fraction of  $\text{H}^{13}\text{C}[\text{O}_3^-]$  generated by PDH, converted this to a quantitative flux, and used this rate to determine the remaining rates in this network (Fig. 7B).

*Akt-mediated Glycolytic Flux Alters the Fate of Hyperpolarized  $[1-^{13}\text{C}]$ Pyruvate*—Because cancer cells typically have high rates of glycolysis, we next determined the effects of glucose on

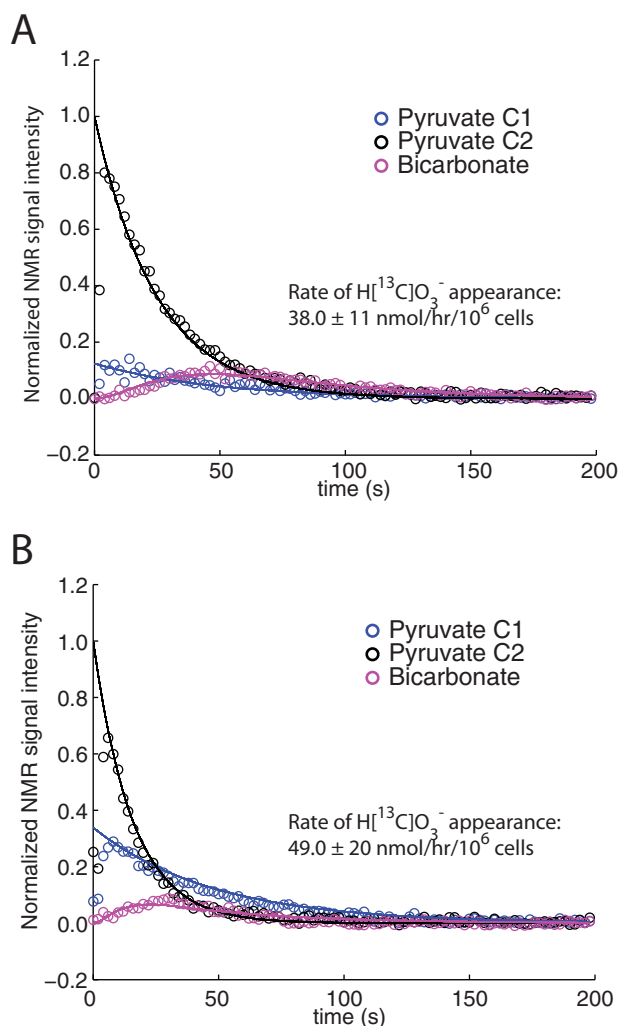


FIGURE 6. **Kinetic analysis of pyruvate decarboxylation.** Time courses for detection of signal for pyruvate carbons 1 and 2 (C1, C2), and  $\text{H}^{13}\text{C}[\text{O}_3^-]$  in suspensions of cells treated with 6 mM hyperpolarized  $[1-^{13}\text{C}]$ pyruvate. Results are shown for a typical experiment in SFxL cells (A) and Huh-7 cells (B). Rates of  $\text{H}^{13}\text{C}[\text{O}_3^-]$  formation are derived from three independent experiments.

TABLE 1

Relative flux through various pathways in two different cancer cell lines as reported by a  $^{13}\text{C}$  isotopomer analysis of tissue glutamate

All fluxes are relative to a tricarboxylic acid cycle (TCAC) flux = 1.

	Fraction of acetyl-CoA derived from pyruvate ( $F_{C2}$ )	PDH flux	Anaplerotic flux into the TCAC via PC ( $\gamma_{PC}$ )	PK flux	Anaplerotic flux of unlabeled substrates into the TCAC ( $\gamma_S$ )
SF <sub>x</sub> L cells	0.58	0.93	0.25	0	0.85
	0.58	0.94	0.21	0	0.80
	0.93	0.56	0	0	0.45
Average	0.70	0.81	0.15	0	0.70
S.D.	0.20	0.22	0.13	0	0.22
Huh-7 cells	0.61	1	1.18	0.27	0.33
	0.62	0.96	0.90	0.07	0.32
	0.83	0.90	1.21	0.62	0.53
Average	0.69	0.95	1.10	0.32	0.39
S.D.	0.12	0.05	0.17	0.28	0.12
p value	0.95	0.33	0.002	0.12	0.10

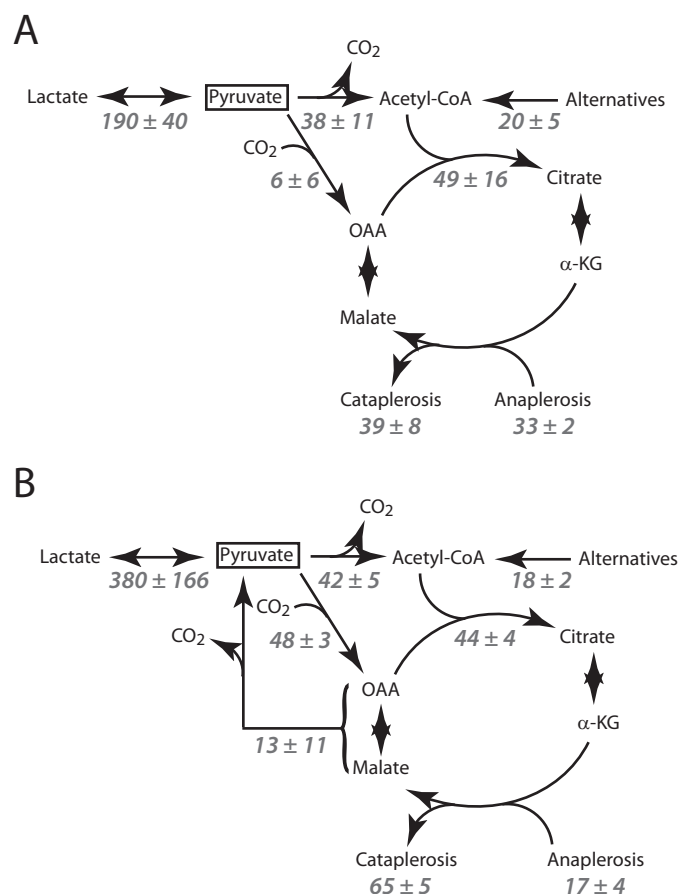
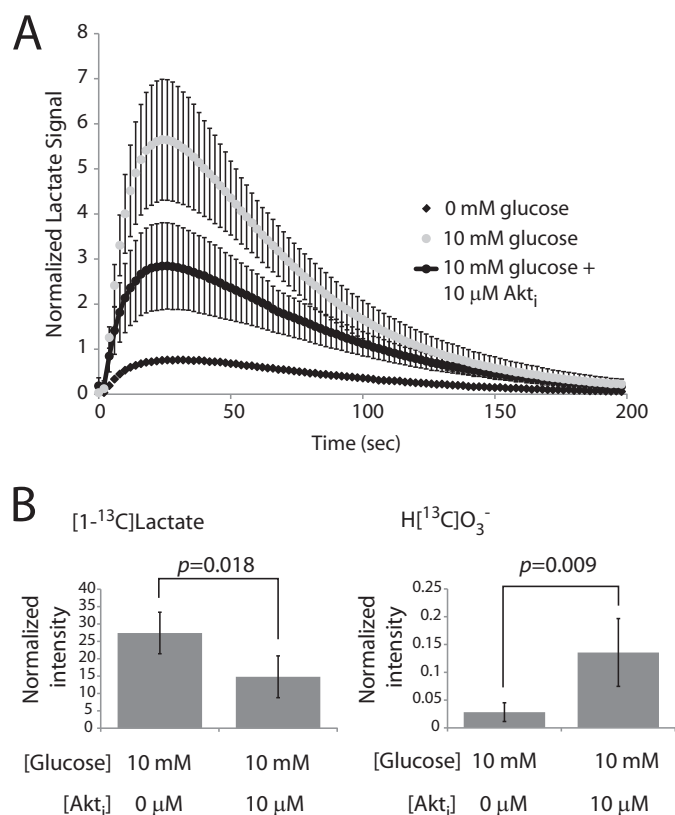


FIGURE 7. **Quantitative models of pyruvate metabolism in SFxL cells (A) and Huh-7 cells (B).** All flux values are in  $\text{nmol}/10^6/\text{h}$ . Each value is the average of three independent experiments  $\pm$  S.D.  $\alpha$ -KG,  $\alpha$ -ketoglutarate.



**FIGURE 8. Akt-stimulated glycolysis acutely alters LDH and PDH flux.** *A*, time evolution of  $[1-^{13}\text{C}]$ lactate appearance in SFxL cells cultured in DMEM supplemented with 6 mM hyperpolarized  $[1-^{13}\text{C}]$ pyruvate and 0, 10, or 10 mM glucose with 10  $\mu\text{M}$  Akt inhibitor. The 0 mM glucose time course is the average of the experiments from Fig. 5*B* and is included here for reference. The other time courses are the average and S.D. for three independent experiments. *B*, normalized intensity for  $[1-^{13}\text{C}]$ lactate and  $\text{H}[^{13}\text{C}]\text{O}_3^-$  in 10 mM glucose with and without the Akt inhibitor. Data are the average and S.D. for several independent experiments ( $n = 5$  for cultures containing glucose;  $n = 6$  for cultures containing glucose and the Akt inhibitor).

metabolism of  $[1-^{13}\text{C}]$ pyruvate. We reasoned that hyperpolarization would provide a unique and dynamic view of the effects of glucose on multiple fates of pyruvate simultaneously. The presence of an active glycolytic flux was expected to alter pyruvate metabolism in at least two ways. First, glycolysis provides NADH that would result in a more physiological redox ratio than the glucose-free, high pyruvate conditions used in previous experiments and might stimulate the reduction of pyruvate to lactate by LDH. Second, the availability of a lactate pool would increase the opportunity for exchange reactions by which hyperpolarized  $^{13}\text{C}$  could be transferred from pyruvate to lactate (15). As expected, culture of SFxL cells with 10 mM glucose before and during exposure to 6 mM hyperpolarized  $[1-^{13}\text{C}]$ pyruvate substantially increased the appearance of  $[1-^{13}\text{C}]$ lactate (Fig. 8*A*). Peak lactate signal was  $\sim 10$  times higher in cultures containing glucose than in those lacking it (Fig. 7*A*). Unexpectedly, providing glucose also suppressed the appearance of  $\text{H}[^{13}\text{C}]\text{O}_3^-$ . Although a small amount of this metabolite could be detected in summations of all spectra acquired over the course of a hyperpolarization experiment, time evolutions like those in Fig. 5*C* were no longer visible. The lack of time-dependent  $\text{H}[^{13}\text{C}]\text{O}_3^-$  data made it impossible to determine specific flux rates as in the glucose-deprived cells.

Cancer cell glycolysis is largely under the control of oncogenic signaling. In SFxL cells, the kinase Akt enables cells to maintain high rates of lactate production and secretion (33). We, therefore, tested whether inhibiting Akt would alter the metabolism of hyperpolarized  $[1-^{13}\text{C}]$ pyruvate. SFxL cells were cultured with a commercial inhibitor at doses that suppressed lactate secretion by  $\sim 50\%$  (33). This reduced the appearance of  $[1-^{13}\text{C}]$ lactate but enhanced the production of  $\text{H}[^{13}\text{C}]\text{O}_3^-$  (Fig. 8, *A* and *B*). These data demonstrate that metabolism of hyperpolarized  $[1-^{13}\text{C}]$ pyruvate by both LDH and PDH is acutely and dramatically responsive to Akt-mediated effects on glycolysis.

## DISCUSSION

We report here three advances in the quantitation of pyruvate metabolism in tumor cells. First, we developed an approach to maximize detection of bicarbonate in real time, making it possible to establish an accurate rate of PDH activity by monitoring formation of hyperpolarized  $\text{H}[^{13}\text{C}]\text{O}_3^-$ . Second, by performing simultaneous steady-state labeling and isotopomer analysis, we used the rate of hyperpolarized  $\text{H}[^{13}\text{C}]\text{O}_3^-$  formation to define a number of other rates related to pyruvate and TCA cycle metabolism. Third, we determined that Akt-driven effects on glycolysis suppress decarboxylation of hyperpolarized  $[1-^{13}\text{C}]$ pyruvate in cancer cells.

Hyperpolarization has the capacity to report quantitative fluxes of enzyme-catalyzed reactions in tumors and has been used successfully to detect oncogene-driven LDH-catalyzed exchanges between pyruvate and lactate *in vivo* (16, 18). However, cancer cells both in culture and in solid tumors metabolize pyruvate in the mitochondria (28, 29). Mitochondrial pyruvate metabolism is important, because it delivers carbon to pathways involved in both the assimilation of biomass in tumors and in the energy-producing reactions in the TCA cycle and oxidative phosphorylation (23, 46). Furthermore, intravenous administration of  $[^{13}\text{C}]$ glucose just before tumor resection has provided strong evidence that both PDH and PC are active in lung and brain tumors (24, 26, 28). Thus, hyperpolarization strategies designed to detect LDH without considering mitochondrial activity may miss many informative aspects of tumor metabolism.

Several important aspects of mitochondrial pyruvate metabolism are apparent from the data. First, combining isotopomer analysis with hyperpolarization revealed that essentially all decarboxylation of hyperpolarized  $[1-^{13}\text{C}]$ pyruvate was the result of PDH activity rather than the effects of pyruvate carboxylation and subsequent  $\text{CO}_2$ -releasing reactions. Second, the delayed formation of  $\text{H}[^{13}\text{C}]\text{O}_3^-$  relative to  $[1-^{13}\text{C}]$ lactate (e.g. Fig. 4*C*) is likely due to a delay associated with pyruvate transport, slower kinetics of PDH compared with LDH (47), and the greater number of steps needed to generate  $\text{H}[^{13}\text{C}]\text{O}_3^-$  after transit of  $[1-^{13}\text{C}]$ pyruvate through PC. However, the addition of more enzymatic steps to the model had no effect on the residual between the fit and the data, consistent with the idea that most of the hyperpolarized  $\text{H}[^{13}\text{C}]\text{O}_3^-$  is produced by PDH in both cell lines. Second, there was a delay in  $\text{H}[^{13}\text{C}]\text{O}_3^-$  formation in SFxL cells relative to Huh-7 cells. This delay was apparently due to multiple factors. The reduced initial LDH flux in SFxL is consistent with delayed entry of pyruvate via transporters on the cytoplasmic membrane. In addition, mod-

eling of the hyperpolarization data revealed a substantial reduction in the rate constant for entry of pyruvate into the mitochondria in SFxL cells (data not shown), suggesting that Huh-7 cells may also have a higher capacity for mitochondrial pyruvate import.

These experiments are an extension of combining oxygen consumption measurements with isotopomer analysis data in perfused hearts and livers (45, 48). In those cases, isotopomer analysis was used to evaluate the amount of  $\text{H}^{[13\text{C}]\text{O}_3^-}$  generated by PDH versus phosphoenolpyruvate carboxykinase. Simulating the data generated from a perfused organ *ex vivo* or *in vivo* is much more technically demanding than the cell suspensions used here. First, a pyruvate delivery function must be estimated to assess the starting concentration for the downstream reactions. This can be accomplished by using a non-metabolic marker such as  $^{13}\text{C}$  urea, which is co-injected with the pyruvate (49). The other complication is the presence of pyruvate in the vasculature that may not exchange with pyruvate in the target cells. In the methodology described here, cells were injected into the prepositioned hyperpolarized solution, enabling relatively simple kinetic models to be applied. Fitting the kinetic  $\text{H}^{[13\text{C}]\text{O}_3^-}$  signal did not require differentiating intracellular pyruvate from extracellular pyruvate, although cytosolic and intramitochondrial pyruvate were considered as separate pools.

The addition of glucose to the culture medium dramatically altered  $[1-^{13}\text{C}]$ pyruvate metabolism in SFxL cells. Exchanges between  $[1-^{13}\text{C}]$ pyruvate and  $[1-^{13}\text{C}]$ lactate were enhanced, whereas the appearance of  $\text{H}^{[13\text{C}]\text{O}_3^-}$  was markedly suppressed. Enhanced labeling of  $[1-^{13}\text{C}]$ lactate was expected, because glycolysis produces abundant lactate for participation in LDH-catalyzed exchanges with  $[1-^{13}\text{C}]$ pyruvate (15). Furthermore, flux through glyceraldehyde-3 phosphate dehydrogenase in glucose-replete cells provides NADH for the LDH reaction. The findings indicate that LDH activity may be underestimated in experiments with hyperpolarized  $[1-^{13}\text{C}]$ pyruvate unless glucose is provided. On the other hand, the suppression of pyruvate decarboxylation was unexpected. Because formation of hyperpolarized  $\text{H}^{[13\text{C}]\text{O}_3^-}$  occurred exclusively through PDH in these experiments, the finding indicates that glycolysis constrains flow of  $[1-^{13}\text{C}]$ pyruvate through the PDH reaction.

There are multiple potential explanations for this observation. One is that provision of glucose simply leads to substrate competition between pyruvate from glycolysis and hyperpolarized  $[1-^{13}\text{C}]$ pyruvate, suppressing the appearance of  $\text{H}^{[13\text{C}]\text{O}_3^-}$  from the latter. However, in these same cells glucose suppressed oxygen consumption relative to oxygen consumption on pyruvate (Fig. 1B). Thus, it is possible that suppressed  $\text{H}^{[13\text{C}]\text{O}_3^-}$  production reflects a more fundamental inhibitory effect of glucose on oxidative metabolism. This might have broader implications for cancer cell metabolism. The Warburg effect refers to the historical observation that cancer cells preferentially convert pyruvate to lactate in the presence of oxygen, supposedly because of an intrinsic bottleneck limiting pyruvate oxidation (50). In contrast, our data demonstrate that glycolysis itself may constrain the cell ability to steer pyruvate toward PDH, at least when that pyruvate is provided as an exogenous substrate. Under the conditions used here, both anaerobic and aerobic pyruvate metabolism were acutely and reversibly

impacted by the high glycolytic flux of SFxL cells. The lack of a fixed constraint on mitochondrial pyruvate metabolism recalls the Crabtree effect, in which a large glucose supply reversibly suppresses oxidative metabolism in yeast and cancer cells (22). Importantly, changes in pyruvate metabolism could be partially reversed in the presence of glucose by inhibiting Akt, the kinase that stimulates glycolysis in SFxL cells. This indicates that pyruvate handling is regulated by oncogenic signal transduction in cancer cells.

Translating approaches with  $^{13}\text{C}$  hyperpolarization to cancer patients poses significant challenges, including the need for appropriate pulse sequences, radiofrequency coils, and the ability to produce sufficient quantities of sterile hyperpolarized tracers for human studies (16). Most if not all of these challenges have been met, however, and successful proof-of-concept imaging to detect conversion of  $[1-^{13}\text{C}]$ pyruvate to  $[1-^{13}\text{C}]$ lactate has been demonstrated in human prostate cancer (19). Obtaining quantitative flux data from *in vivo* hyperpolarization, including quantitative rates of pyruvate decarboxylation, will require significantly more complicated modeling than what is used here. However, it may be possible to combine dynamic hyperpolarized  $^{13}\text{C}$  data with isotopomer analysis in solid tumors, because it is feasible and informative to obtain high quality  $^{13}\text{C}$  NMR spectra from tumors resected from patients infused intra-operatively with  $[^{13}\text{C}]$ glucose (26). Importantly, the ability to observe definitive changes in flux is an essential component of metabolic analysis in any system, including tumors. For example, although conventional, steady-state  $^{13}\text{C}$  NMR can readily be used to detect major changes in the choice of metabolic pathway (e.g. an induction of PC relative to PDH), this approach would likely miss changes in absolute flux unless they are associated with differences in which pathways are active. The method described here solves that problem by providing a quantitative reference flux (pyruvate decarboxylation) to establish accurate rates across a broad network.

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## Noninvasive Quantitation of Metabolic Flux in Cancer Cells

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