

Published in final edited form as:

Structure. 2013 December 3; 21(12): 2221–2227. doi:10.1016/j.str.2013.09.017.

Preferential and specific binding of human α B-crystallin to cataract-related variant of γ S-crystallin

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Summary

Transparency in the eye lens is maintained via specific, functional interactions among the structural $\beta\gamma$ - and chaperone α -crystallins. Here we report the structure and α -crystallin binding interface of the G18V variant of human γ S-crystallin (γ S-G18V), which is linked to hereditary childhood-onset cortical cataract. Comparison of the solution NMR structures of wild-type and G18V γ S-crystallin, both presented here, reveal that the increased aggregation propensity of γ S-G18V results from neither global misfolding nor the solvent exposure of a hydrophobic residue, but instead involves backbone rearrangement within the N-terminal domain. α B-crystallin binds more strongly to the variant, via a well-defined interaction surface that represents the first such interface directly observed between a variant structural crystallin and α -crystallin. In the context of the α B-crystallin structure and the finding that it forms heterogeneous multimers, our structural studies suggest a potential mechanism for cataract formation via the depletion of the finite α B-crystallin population of the lens.

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C. N. K. and W. D. B. prepared protein samples, performed NMR and DLS experiments, analyzed NMR and DLS data, and wrote the manuscript. These authors contributed equally. S. M., A.D., and A. J. B. prepared protein samples. H. O. and R. W. M. designed the experiments, analyzed data, and wrote the manuscript.

Accession Numbers

The solution NMR structures of γ S-WT and γ S-G18V have been deposited in the Protein Data Bank, with PDBIDs 2M3T and 2M3U, respectively.

Supplemental Information

In this appendix we describe the calculations used to determine the NMR structures of γ S-WT and γ S-G18V and report the 20 lowest-energy structures (Supplementary Figures S1 and S2). Additional structure calculations to confirm the position of valine-18 in the γ S-G18V structure are also reported here (Supplementary Figure S3). DLS data showing the particles sizes of the NMR samples are also presented here to show that both γ S-WT and γ S-G18V are monomeric under the NMR sample conditions (Supplementary Figure S4). Supplementary Figure S5 contains the full DLS data of α B, γ S-WT, γ S-G18V, and mixtures thereof as determined by dynamic light scattering.

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Introduction

The crystallins are the primary protein components of the eye lens, reaching concentrations higher than 400 mg/mL in humans (Tardieu et al., 1992). Short-range ordered interactions between crystallins at high concentrations are thought to maintain transparency while providing the refractive index gradient required to focus light on the retina (Delaye and Tardieu, 1983; Ponce et al., 2006). Perturbations to the inter-crystallin interactions concomitant with the formation of high molecular weight aggregates can lead to lens opacification during aging and cataractogenesis. Characterizing the interactions between members of the two crystallin superfamilies, α -, and $\beta\gamma$ -, is critical to understanding cataract formation, because insoluble aggregates of crystallins from both have been found in cataractous lenses (Takemoto and Sorensen, 2008). The α -crystallins (αA and αB) act as holdase chaperones, binding to but not refolding structural $\beta\gamma$ -crystallins for which solubility is compromised due to damage or mutation. Because of the extremely low protein turnover in lens fiber cells, the inter-crystallin interactions are thus the first line of defense against aggregation. Mutations in either the α -, or $\beta\gamma$ crystallin genes can alter these critical attractive interactions (Fu and Liang, 2003). In the case of γC -crystallin, α -crystallins do not recognize all known disease-related variants (Moreau and King, 2012). Likewise, in γD -crystallin, both the E107A and R76S variants are implicated in early-onset cataract, but the former exhibits increased α -crystallin attraction (Banerjee et al., 2011), while α -crystallin interactions with the latter remain unchanged (Ji et al., 2012). Here we focus on interactions between αB - and γS -crystallins; αB is the more versatile chaperone, abundantly expressed in tissues outside the eye lens (Iwaki et al., 1990), up-regulated by various stressors (Klemenz et al., 1991), and implicated in several neuropathological diseases (Iwaki et al., 1989; van Noort et al., 1995; Ousman et al., 2007), whereas γS is the most abundant of the structural $\beta\gamma$ -crystallins in the human lens cortex and is highly conserved across several species (Chang and Chang, 1987; Quax-Jeuken et al., 1985; van Rens et al., 1991; van Rens et al., 1989).

At present, there are four known cataractogenic mutations in human γS -crystallin: the γS -V42M variant, which distorts the compact β -sheet packing in the core of the N-terminal domain and causes severe congenital cataract in children (Vendra et al., 2013); the Coppock cataract-associated γS -D26G variant, which leads to decreased protein stability but apparently has little effect on the overall molecular architecture (Karri et al., 2013); the γS -S39C variant, which is linked to microcornea and cataract (Devi et al., 2008) and which is hypothesized to have an exposed cysteine available for disulfide crosslinkage and aggregation; and the γS -G18V variant, which is implicated in childhood-onset cortical cataract (Sun et al., 2005). Although its decreased thermodynamic stability relative to wild-type (γS -WT) (Ma et al., 2009) has been established, additional experiments indicated that γS -G18V is aggregation-prone well below its unfolding temperature, suggesting an aggregation mechanism more complex than simple denaturation (Brubaker et al., 2011). In order to better understand how structural changes in the cataract-related G18V variant of γS -crystallin lead to altered intermolecular interactions, we have solved the solution NMR structures of human wild-type and γS -G18V and elucidated the binding interface between αB -crystallin and γS -G18V.

Results

The G18V Mutation Causes Structural Perturbation

Like the highly homologous murine protein (Wu et al., 2005) and other mammalian structural crystallins, γS -WT has a double Greek key fold. A comparison of the γS -WT and γS -G18V structures reveals local shifts in the backbone but little change in the overall fold. The average heavy-atom RMSD between the two structures is 1.62 Å for the N-terminal

domain and 1.13 Å for the C-terminal domain. (Figure 1, Table 1, Supplementary Figures S1, S2). Because G18 is located on a surface-exposed loop, solvent exposure of the valine sidechain might be expected to provide a plausible mechanism for the solubility impact of this mutation; however the structural data indicate that it is buried, with the backbone occupying an unusual conformation at this position. Many examples of residues stabilized by hydrogen bonds in unfavorable Ramachandran angles (Jia et al., 1993; Gunasekaran et al., 1996) have been found in the context of enzyme active sites in which the conformation is required for activity (Jia et al., 1993; Pal and Chakrabarti, 2002). Here, the configuration of V18 is stabilized in part by the pi-stacking interactions between R20 and F16 (Figure 1c). The dihedral angles shift from $\phi = 79.3^\circ$, $\psi = -146.4^\circ$ for G18 to $\phi = 103.0^\circ$, $\psi = -134.0^\circ$ for V18. Presumably because of steric clashes with sidechains on the opposing side of the affected loop, the methyls of V18 are angled toward the C-terminal end of the polypeptide chain, locking the R19 amide proton into place centered between the V18 methyls (Figure 1b). Additional structural calculations indicate that the V18 side chain remains buried even upon exclusion of all distance restraints to V18, and the adoption of a favorable backbone dihedral angle configuration would require the elimination of restraints from several surrounding residues, producing an extensive structural disruption to the surrounding loop region that is not supported by the NMR data (Supplementary Figure S3). The effect propagates down the polypeptide chain through more than half of the N-terminal domain of γ S-crystallin. Despite these local differences, the overall folds of both structures are very similar (Figure 1a), consistent with previous CD and UV-fluorescence data. Both γ S-WT and γ S-G18V were monomeric under the conditions used for structural NMR (Supplementary Figure S4).

α B-crystallin and γ S-G18V Interact to Form Large Complexes

Because γ S-G18V is implicated in early-onset cataract formation and has an altered structure in solution, we hypothesized that the molecular chaperone α B would interact more strongly with the disease-related variant than with wild-type γ S. In order to assess the extent of binding, dynamic light scattering (DLS) measurements were performed at 25 °C on samples of α B, γ S-WT and γ S-G18V individually and as mixtures at pH 6.9. The DLS data are shown in Supplementary Figure S5. These results are consistent with previous findings; human α B-crystallin spontaneously forms spherical multimers 80 to 180 Å in diameter with a variable number (~24–32) of subunits (Haley et al., 1998; Jehle et al., 2010). γ S-WT has an average hydrodynamic diameter of 50.40 ± 0.28 Å, which is in agreement with reported values for other monomeric γ -crystallins (Liu et al., 1998). γ S-G18V forms large multimers with diameters up to 289.2 ± 8.8 Å at pH 6.9. Mixtures of α B and γ S-WT or γ S-G18V have apparent hydrodynamic diameters up to 155.88 ± 0.46 and 478.2 ± 2.3 Å, respectively.

The size of the particles in the α B + γ S-WT mixture is similar to that of α B alone, likely because there are only weak interactions between the two proteins, and the DLS size of the mixture thus reflects the much larger α B complexes. Conversely, the particle size of the α B + γ S-G18V mixture is much larger than either that of γ S-G18V or α B alone. This result indicates that α B is interacting more strongly with γ S-G18V than with γ S-WT and is consistent with the conclusions drawn by Abgar et al. (Abgar et al., 2001) that α -crystallin binds destabilized proteins to prevent non-specific aggregation and that the resulting complex reorganizes into large particles in order to remain soluble.

α B-crystallin Interacts More Strongly With γ S-G18V

To localize the regions of γ S-crystallin involved in interactions with α B, heteronuclear single quantum coherence (HSQC) NMR spectroscopy was performed on mixtures of 15 N-labeled γ S-WT and γ S-G18V with α B at pH 6.9. Solution NMR is sensitive to small changes in the electronic environment of the detected nuclei, therefore binding interactions

between α B and γ S should result in shifts or disappearances of relevant cross peaks. The addition of α B to ^{15}N -labeled γ S-WT leads only to minor chemical shift changes in the ^{15}N - ^1H HSQC spectrum recorded at 25 °C (Figure 2a). Residues corresponding to cross peaks that undergo minor chemical shift changes include S35, W47, E66, G92, F122, and H123; all surface residues. Furthermore, the lack of cross peak disappearance in the presence of the molecular chaperone supports the conclusion that α B only weakly interacts with native γ S in dilute solution, consistent with the results of a past ^1H NMR spectroscopic study of bovine α - and γ S-crystallin (Cooper et al., 1994). In contrast, upon addition of α B to ^{15}N -labeled γ S-G18V, nearly 50% of the cross peaks broadened below the noise threshold of the ^{15}N - ^1H HSQC, presumably due to the formation of large α B/ γ S-G18V complexes in solution (expected to be > 300 kDa). Using ^{15}N - ^1H TROSY-HSQC spectroscopy, we acquired a 2D spectrum of the α B/ γ S-G18V mixture. Analysis of the resulting TROSY-HSQC of γ S-G18V (Figure 2b) reveals the disappearance of several N-H cross peaks; T9, Y11, D13, N15, F16, R19, Y21, C23, C25, C27, Y33*, L34, S35*, R36, C37, N38, I40, W47 ϵ , G65, Y67, S82, S85, and G91. These residues are located in the N-terminal domain except for G91, which is situated in the linker between the two domains. Interestingly, the majority of cross peaks that lose signal intensity correspond to γ -G18V residues that occupy different positions (and whose cross peaks therefore shift) with respect to γ S-WT. Y33* and S35* correspond to cross peaks from a minor alternate conformation of γ S-G18V, also previously assigned (Brubaker and Martin, 2011). There are several cross peaks that slightly shift in the γ S-G18V/ α B mixture, including Y33, A56, G57, W73 ϵ , S105, E110, I118, Q121, M124, G147, I161, W163 ϵ , A165, and V170. These peaks are distributed throughout the protein and may shift due to altered surface interactions with α B and γ S or due to changes in the local chemical environment resulting from the binding of α B and the associated formation of larger complexes.

Thermally Stressed γ S-WT Does Not Recruit α B-crystallin

Previous studies investigating the solubilizing capabilities of α -crystallin using γ -crystallin mixtures purified from bovine lenses have shown that α -crystallin can prevent the thermal aggregation of γ -crystallins (Horwitz, 1992). A similar study noted that the rate of protein aggregation was dependent on α -crystallin concentration and that with a 3:2 α -: γ -crystallin ratio, using combinations of bovine α -crystallin and bovine γ A-D, aggregation of the mixture occurred after heating at 72 °C (Wang and Spector, 1995).

In light of these results and the finding that only weak binding was observed between human α B and γ S-WT at 22 °C, we set out to characterize the interactions between α B and γ -WT upon heating. An ^{15}N - ^1H HSQC spectrum was taken every 5 °C as γ S-WT was heated between 22 – 47 °C, with and without α B. Sample precipitation and loss of signal occurred at temperatures above 47 °C, consistent with our previous observation that wild-type γ S-crystallin forms aggregates well below its unfolding temperature (Brubaker et al., 2011). In the spectra obtained for both γ S-WT and the mixture of γ S-WT + α B, (shown overlaid in Figure 3) several cross-peaks of γ S-WT shift due to temperature change. A direct comparison between the spectra of γ S-WT in the presence and absence of α B (Figure 3a) reveals slight shifts and significant line broadening of the cross peaks, which decrease with increasing temperature. However, no disappearance of cross peaks is observed. The line broadening is due to weak transient interactions between α B and γ S-WT, and its decrease with increasing temperature is attributed to accelerated molecular tumbling. Our observations of protein precipitation in the γ S-WT/ α B mixture at higher temperatures and the lack of disappearing cross peaks in the HSQCs indicates that α B-crystallin does not recognize thermally unfolded γ S-crystallin. Human α B crystallin retains its native secondary structure up to 70 °C, although its chaperone activity is greatest near physiological temperatures (van Boekel et al., 1999; Reddy et al., 2000; Datta and Rao, 1999). This

finding also suggests that thermal denaturation of native γ S-crystallin is not a realistic model for cataract formation in this system.

Discussion

Several three-dimensional structures of crystallins have been reported in the last decade, revealing the double Greek key fold as a common feature of $\beta\gamma$ -crystallins (Jaenicke and Slingsby, 2001; Mills et al., 2007). The high stability and solubility of the crystallins is critical to their function because protein degradation and synthesis do not occur in mature differentiated lens fiber cells. Our results indicate that the relatively minor structural changes in γ S-G18V result in a perturbation to the delicately balanced set of weak interactions between crystallins. The large body of structural investigations and interaction studies on β - and γ -crystallins suggests that specific interactions between them are functionally important (Slingsby et al., 1991). Weak interactions of individual $\beta\gamma$ -clusters with α -crystallins may be relevant at the high protein concentrations in the eye lens, coupling these clusters to the structural dynamics of the polydisperse α -crystallin oligomers and preventing the formation of insoluble aggregates. Our observation of weak interactions between wild-type γ S-crystallin and α B supports this hypothesis, because interactions should be weak even at millimolar concentrations to ensure an optimal level of exchange dynamics in the eye lens.

The spectra of the α B/ γ S-WT sample exhibited only minor chemical shift changes, and the affected residues are widely distributed throughout the structure (Figure 4a and b), confirming the presence of non-specific interactions. In this context, we expect the protein interfaces to exhibit very weak binding at the relatively low concentrations investigated here. Conversely, α B interacts more strongly with the disease-related variant γ S-G18V as evidenced by loss of cross peak intensities and shifts due to the interactions with the chaperone. More quantitative comparison of the binding affinities presents experimental challenges, because binding to wild-type γ S-crystallin is mainly characterized by line broadening rather than chemical shift differences. Furthermore, because α B-crystallin forms polydisperse oligomers alone and upon binding to aggregation-prone substrates, its binding to wild-type and G18V γ S-crystallin cannot be correctly described by a single dissociation constant. However, based on the NMR data the interaction surface of γ S-G18V that is recognized by α B-crystallin can be determined and is shown in Figure 4c and d. Residues of γ S-G18V whose cross peaks lose intensity are localized to the N-terminal domain, coincident with the greatest structural changes due to the V18 substitution. The additional weak interactions may play an important role in maintaining the solubility of the larger α B/ γ S-G18V complex.

In recognition of the results obtained with heat-denatured γ S as well as recent mass spectrometry analyses (Lampi et al., 2012), it appears that during the lifetime of the organism, the chaperone activity of α -crystallins is required to account for accumulated post-translational modifications, such as deamidation or oxidation, rather than protein unfolding per se. Naturally occurring disease-related mutations represent a good model for this type of aggregation. Here, α B-crystallin interacts only weakly with γ S-WT but more strongly and specifically with γ S-G18V as indicated by the differences in the NMR spectra between the two mixtures.

The stronger binding of α B to the variant is consistent with the hypothesis that the mechanism of cataractogenesis from the G18V mutation in γ S-crystallin may be related to the depletion of the finite amount of α -crystallin in the eye lens. Given the deoptimization of binding strength in the γ S-WT interactions, we speculate that many disease-related modifications or mutations cause the formation of tighter complexes with α -crystallins. In this way, α -crystallins perform a holdase chaperone function, preventing unfavorable

interactions among β - and γ -crystallins. Additionally, disease onset may be accelerated due to both larger γ S-G18V and α B- γ S-G18V particles which may be more prone to aggregation and precipitation. Similar effects were observed by others, reporting an increase in particle size after mixtures of α -crystallin and γ -crystallins from *D. mawsoni*, *T. obesus*, and *B. taurus* were heated extensively (Kiss et al., 2004). The exact mechanism that leads to this increase is yet unknown, but recruiting larger amounts of α B to the 'sick' protein is likely, since an aggregation-prone or stability-relevant area requires additional protection by α -crystallin on top of any normal interactions. In this model, one role α -crystallins may play in the eye lens is contributing a polydispersity principle, preventing formation of larger, more regular, and ultimately insoluble aggregates.

Experimental Procedures

Sample preparations

γ S-WT, γ S-G18V (Brubaker et al., 2011; Brubaker and Martin, 2011), and α B-crystallin (Jehle et al., 2010) were produced in *E. coli* and purified as described previously. NMR samples for structural work were at protein concentrations of 2.11 mM and 1.50 mM, respectively, in 10 mM acetate buffer pH 4.5, 10% D₂O, 0.05% sodium azide, and 2 mM TMSP. Samples of γ S-WT and γ S-G18V under these conditions have remained stable and monomeric for over a year, showing no change in NMR spectra, when stored at 4 °C. Samples for NMR studies with α B were at protein concentrations of 1.5 mM for both γ S-WT and γ S-G18V in 10 mM phosphate buffer pH 6.9, 10% D₂O, 0.05% sodium azide. Mixed samples with α B-crystallin consisted of an α B: γ S molar ratio of 2:1. All experiments except for ¹⁵N IPAP spectra and the spectra observing the α B/ γ S thermal interactions were collected at 22 °C.

For RDC measurements the DIOTPC/DIOHPC bicelle system (Ottiger and Bax, 1999) at 10% (w/v) lipid concentration was used to align the protein samples. DIOTPC and DIOHPC dissolved in chloroform (Avanti Polar Lipids) were mixed at a molar ratio of 3:1 DIOTPC:DIOHPC and dried under a stream of nitrogen gas. The residual chloroform was removed by lyophilization. A 270 μ L protein sample in 10 mM acetate buffer pH 4.5, 10% D₂O, 0.05% sodium azide, and 2 mM TMSP was added to the lyophilized lipids. The sample was cycled several times between an ice bath and room temperature over the course of a few hours with gentle mixing between each incubation in order to fully rehydrate the lipids and mix the bicelle sample.

H-D exchange samples were prepared by concentrating the γ S samples to saturation in centrifugal concentrator columns (with γ S at a concentration of approximately 270 mg/mL at saturation) in a volume of approximately 60 μ L and adding 99.9% D₂O to a total volume of 300 μ L immediately before starting the data collection.

NMR experiments

NMR experiments were performed on a Varian UnityINOVA system operating at 800 MHz equipped with a ¹H/¹³C/¹⁵N 5 mm tri-axis PFG triple-resonance probe. Decoupling of ¹⁵N nuclei was performed by GARP sequences. ¹H shifts were referenced to TMSP, ¹³C and ¹⁵N shifts were referenced indirectly to TMSP. Heated samples of α B/ γ S mixtures were equilibrated for several minutes before data acquisition. NMR data were processed using NMRPipe and analyzed using Sparky. H-D exchange spectra were taken at intervals of approximately 30 minutes for the first 4 hours following the addition of D₂O, and then at intervals of 1 hour for the next 20 hours. Additional spectra were collected at intervals of several days. For each collection period a 1-dimensional proton spectrum was collected and

non-exchanging methyls in the protein were used to adjust for differences in shimming between data collections.

Restraints

NOE restraints were assembled by manually picking slices from ^{13}C -filtered and ^{15}N -filtered NOESY experiments corresponding to the ^{13}C and ^{15}N HSQC crosspeaks. The manually picked peaks were assigned in a “binned” fashion using three sets of chemical shift tolerances of decreasing stringency to minimize NOE crosspeak assignment ambiguity, and restraints were generated in CCPNMR Analysis. Duplicate and redundant restraints were eliminated from the exported restraint lists, and only unambiguous NOE restraints were used in the final structure calculations.

In-phase, anti-phase (IPAP) spectra of isotropic γS -WT and aligned γS -G18V were acquired at 32 °C and 30 °C. Each IPAP dataset was processed into two spectra each containing only one of the two doublet peaks so that all peaks could be easily resolved. For the RDCs, peaks were manually picked for all crosspeaks in the spectra and the difference between the J-splittings and the J+D splittings were computed for each resonance using a spreadsheet program. The starting error for all of the RDCs in the angular restraint table was set as the standard deviation of the measured J-splittings, 2.47 Hz and 3.01 Hz for γS -WT and γS -G18V, respectively. RDC experiments yielded 156 and 147 couplings for γS -WT and -G18V, respectively, including the sidechain tryptophan $\text{N}\epsilon$ protons, corresponding to nearly every visible peak in the ^1H - ^{15}N HSQC, which are included in the restraints deposited in the PDB. See Supplementary Tables 1 and 2.

H-D exchange experiments yielded a total of 46 and 90 hydrogen bonding restraints used to refine the final structures for γS -WT and -G18V, respectively.

Dihedral angle restraints were calculated using the TALOS+ program (Shen et al., 2009) for γS -WT and for both the major and minor chemical shift sets of γS -G18V.

3J HN-HA couplings were calculated from peaks in a 3D H-N-HA experiment and were used as restraints in structure calculations for the γS -WT and γS -G18V structures (Table 1).

Supplementary Material

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Acknowledgments

We thank Wytze van der Veer for assistance with optical data collection and Doug Tobias and Melanie Cocco for helpful discussions. This work was supported by NIH grant 1R01EY021514 (to R.W.M) and by a DFG grant within the frame of the SFB 740 (to H.O).

Abbreviations

αB	human αB -crystallin
γS	human γS -crystallin
DLS	dynamic light scattering
NMR	nuclear magnetic resonance
HSQC	heteronuclear single quantum coherence spectroscopy
TROSY	transverse relaxation-optimized spectroscopy

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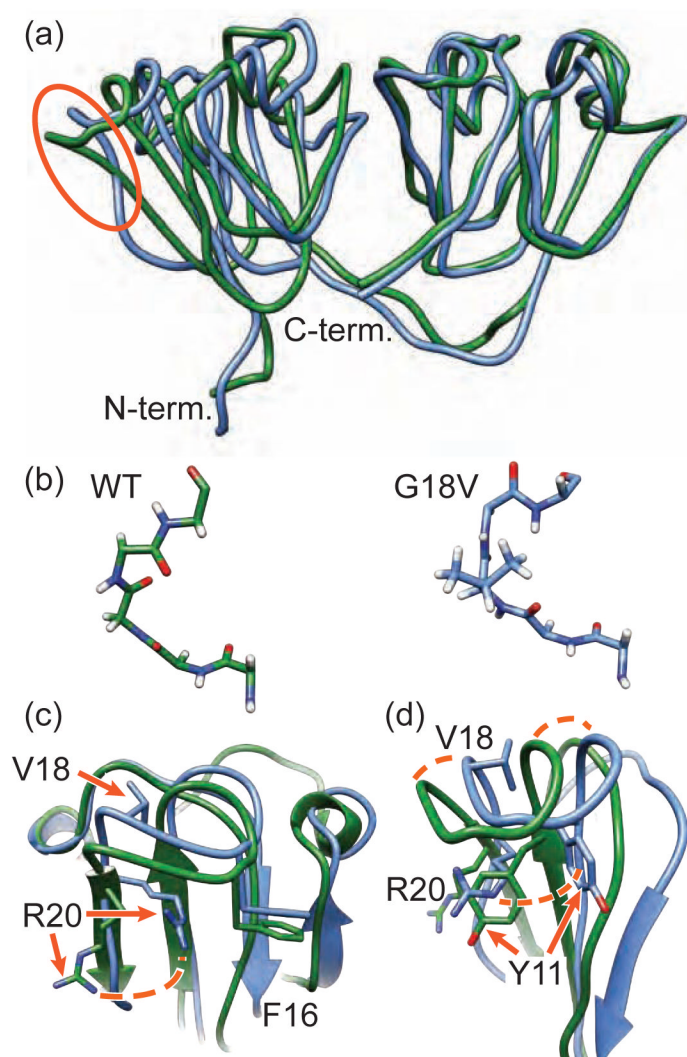


Figure 1.

Structural detail of wild-type and variant γ -S crystallin. (a) A licorice depiction of the average solution NMR structures of γ S-WT (green) and γ S-G18V (blue). Both proteins are highly structured, with the double Greek key fold typical of structural crystallins, although the variant protein displays structural changes in the N-terminal domain (left) relative to wild-type. The red circle indicates the region that is shown in greater detail in Panel (b). Here the affected loop remains essentially intact; in γ S-WT (left) the α proton from G18 is angled slightly askew from the R19 amide proton. In γ S-G18V the orientation of the V18 methyls forces the amide proton into alignment with the valine sidechain, altering the V18 ψ angle. Panel (c) shows the overlaid structures of γ S-WT (green) and γ S-G18V (blue) with the sidechains from F16, V18, and R20 shown, indicating the details of selected structural changes, particularly the dramatic shift in the position of R20. (d) The addition of V18 and its effect on the backbone angles the former β -strand outward and twists it, moving R20 inward, where it displaces Y11 and forces the tyrosine away from F16, placing it flat against the surface of the first Greek key motif. Although each of these structural changes is minor and has primarily local impact, taken in aggregate they result in significant perturbations to the N-terminal domain, potentially providing sites for altered intermolecular interactions and recognition by α B-crystallin. See also Figures S1–S4 and Supplementary Tables 1 and 2.

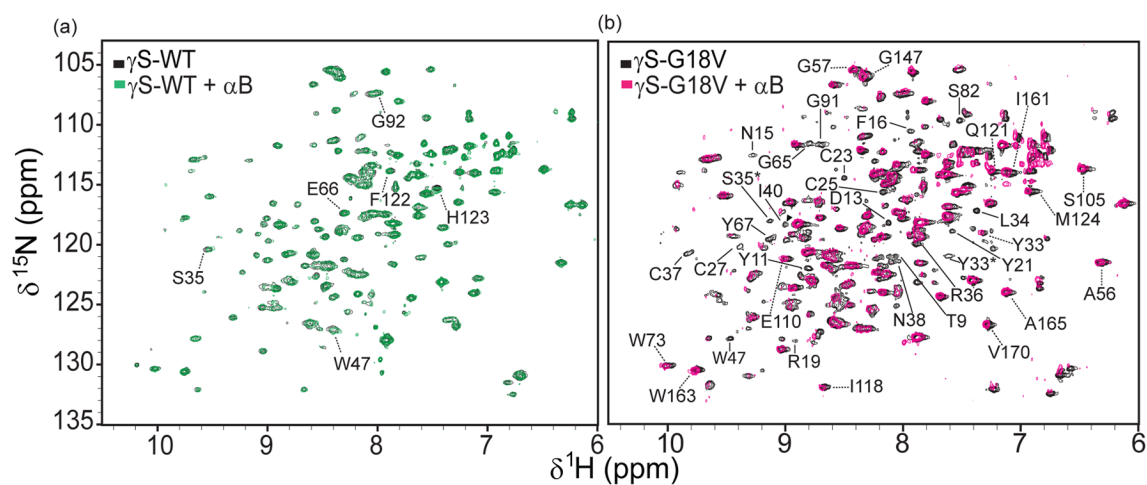


Figure 2. ^{15}N - ^1H HSQCs of (a) ^{15}N -labeled γS -WT with (green) and without (black) αB and (b) ^{15}N -labeled γS -G18V with (pink) and without (black) αB . Dashed lines indicate cross peaks that shift while solid lines indicate cross peaks that disappear with respect to the samples without αB . * Indicate cross peaks that belong to the alternate structure of γS -G18V. (Brubaker and Martin, 2011) See also Table S1.

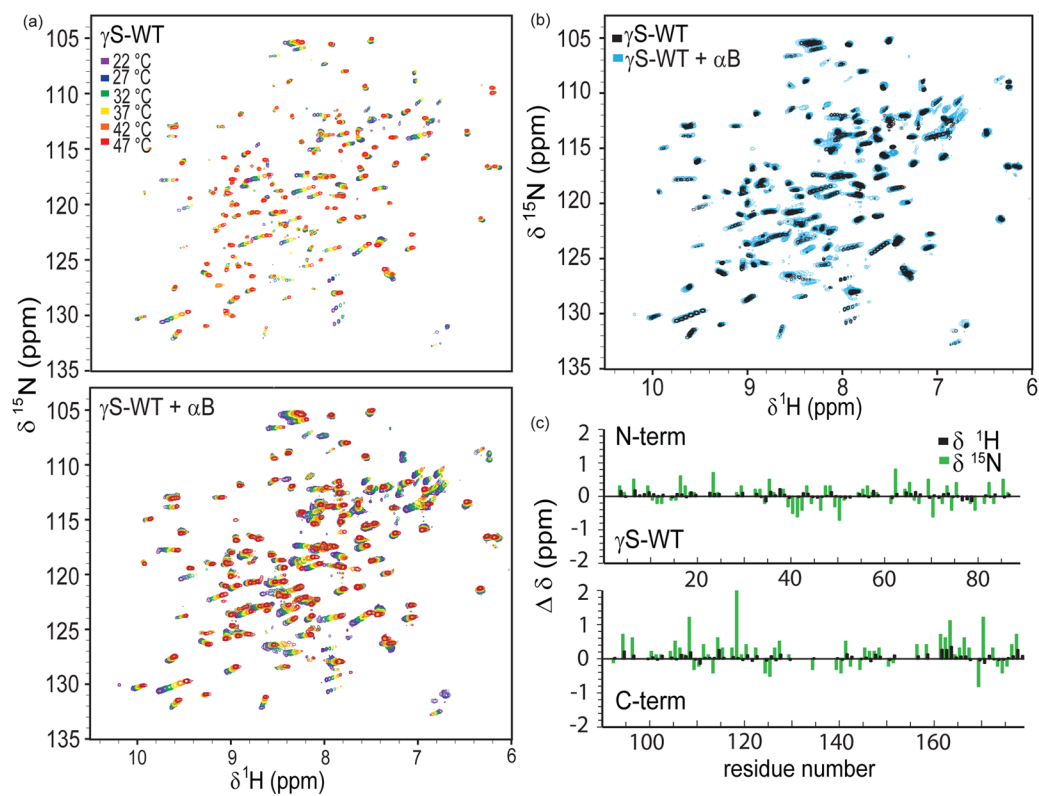


Figure 3. Solution NMR data indicating the minor conformational changes in γ S-WT upon heating, and its weak, non-specific interaction with α B-crystallin. (a) ^{15}N - ^1H HSQC temperature series from 22–47 °C of γ S-WT (top) and γ S-WT + α B (bottom). (b) Overlay of the γ S-WT (black) and γ S-WT + α B (blue) temperature series. (c) Change in ^{15}N and ^1H resonances by residue of γ S-WT between 22 and 47 °C.

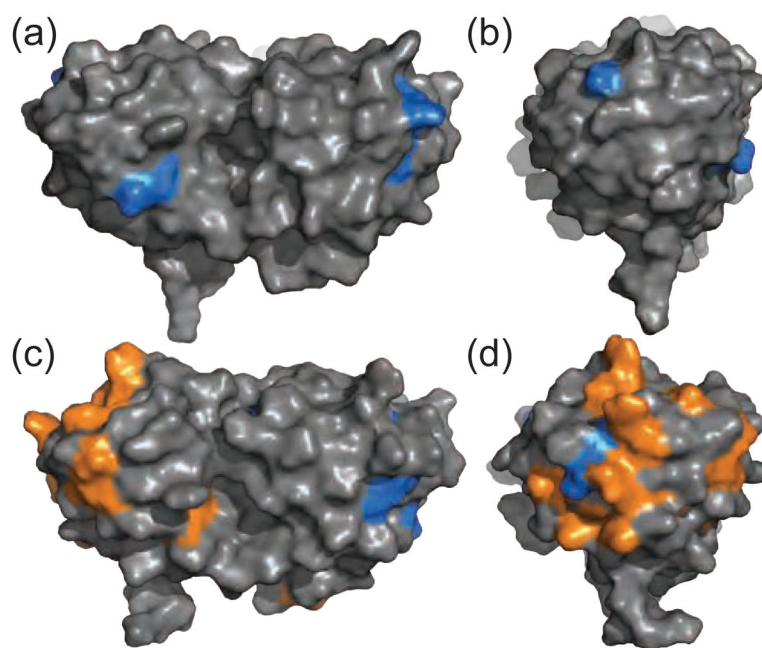


Figure 4. Residues of γ S-WT involved in weak transient interactions with α B are shown in blue on the surface of γ S in two views: (a) from the front and (b) with the N-terminal domain rotated forward. Residues of γ S-G18V involved in binding interactions (orange) and transient interactions (blue) with α B shown on the surface of γ S in two views: (c) from the front and (d) with the N-terminal domain rotated forward.

Table 1

A tabular summary of the NMR structures for γ S-WT and γ S-G18V structures, calculated using full restraint sets as described in the Methods section. See also Figures S1–S4 and Supplementary Tables 1 and 2.

Restraint Summary	γ S-WT	γ S-G18V
Total NOE restraints:	7444	4682
intraresidue:	1547	1329
interresidue:		
sequential ($ i - j = 1$):	1559	972
medium-range ($ i - j \leq 5$):	1286	709
long-range ($ i - j \geq 6$):	3052	1672
Total angular restraints:	408	450
RDCs:	156	147
dihedral ($3J$ coupling):	--	57
dihedral (TALOS+):	252	246
H-bond restraints:	46	90
Structure statistics (20 lowest-energy structures)		
Restraint violations		
NOE $> 0.3 \text{ \AA}$	0.6 ± 0.8	0.7 ± 1.1
dihedral $> 5^\circ$	1.1 ± 1.3	1.6 ± 0.9
RMSD from ideal covalent geometry		
bonds (\AA)	0.005 ± 0.000	0.004 ± 0.000
angles ($^\circ$)	0.582 ± 0.011	0.537 ± 0.016
impropers ($^\circ$)	0.503 ± 0.013	0.481 ± 0.016
Restraint RMSD		
NOE (\AA)	0.034 ± 0.025	0.024 ± 0.002
dihedral ($^\circ$)	0.914 ± 0.137	0.893 ± 0.114
RDC (Hz)	0.238 ± 0.021	0.188 ± 0.018
Average pairwise RMSD (residues 5–178)		
backbone RMSD (\AA)	0.4	0.5
heavy atom RMSD (\AA)	0.8	0.9
RDC Statistics		
R-factor (%)	0.642 ± 0.055	0.459 ± 0.046
R-factor (free) (%)	0.817 ± 0.073	0.556 ± 0.052