

# Purified inositol hexakisphosphate kinase is an ATP synthase: Diphosphoinositol pentakisphosphate as a high-energy phosphate donor

(phosphotransferase/pyrophosphate)

SUSAN M. VOGLMAIER\*, MICHAEL E. BEMBENEK†, ADAM I. KAPLIN\*, GYÖRGY DORMÁN‡, JOHN D. OLSZEWSKI‡, GLENN D. PRESTWICH‡, AND SOLOMON H. SNYDER\*§

\*Departments of Neuroscience, Pharmacology and Molecular Sciences, and Psychiatry, The Johns Hopkins University School of Medicine, Baltimore, MD 21205; †NEN Medical Products Department, E.I. DuPont de Nemours and Co., Inc., Boston, MA 02118; and ‡Department of Chemistry, The State University of New York, Stony Brook, NY 11794

Contributed by Solomon H. Snyder, January 5, 1996

**ABSTRACT** Diphosphoinositol pentakisphosphate (PP-IP<sub>5</sub>) and bis(diphospho)inositol tetrakisphosphate (bis-PP-IP<sub>4</sub>) are recently identified inositol phosphates that possess pyrophosphate bonds. We have purified an inositol hexakisphosphate (IP<sub>6</sub>) kinase from rat brain supernatants. The pure protein, a monomer of 54 kDa, displays high affinity ( $K_m = 0.7 \mu\text{M}$ ) and selectivity for inositol hexakisphosphate as substrate. It can be dissociated from bis(diphospho)inositol tetrakisphosphate synthetic activity. The purified enzyme transfers a phosphate from PP-IP<sub>5</sub> to ADP to form ATP. This ATP synthase activity indicates the high phosphate group transfer potential of PP-IP<sub>5</sub> and may represent a physiological role for PP-IP<sub>5</sub>.

Inositol 1,4,5-trisphosphate (1,4,5-IP<sub>3</sub>) is one of the major intracellular messenger molecules in signal transduction and is associated with the regulation of intracellular calcium (1). The importance of 1,4,5-IP<sub>3</sub> in cellular signaling has led to the examination of the numerous other *myo*-inositol phosphates in cells, whose functions are less clear. Of these inositol metabolites, inositol hexakisphosphate (IP<sub>6</sub>) is the most abundant, with concentrations in mammalian tissues 10–100 times greater than those of IP<sub>3</sub>. The high concentration of IP<sub>6</sub> (15–60  $\mu\text{M}$ ) (2, 3) is presumed to be compartmentalized within the cell (4, 5) with at least 80% of IP<sub>6</sub> in releasable cytosolic pools (6).

In plants, IP<sub>6</sub> acts as an antioxidant (7) and phosphate store (8). In humans, dietary IP<sub>6</sub> from cereals and legumes is associated with decreased iron absorption (9) and lowered cancer rates (10). In cells, IP<sub>6</sub> may act as a siderophore (11, 12) and thus block iron-mediated oxidative damage (13). IP<sub>6</sub> also interacts with several proteins that regulate endocytosis (14–18), synaptic vesicle trafficking (19–21), and receptor desensitization (22).

IP<sub>6</sub> was thought to be the metabolic endpoint of *myo*-inositol metabolism, but recently inositol pyrophosphates that have seven and eight phosphates associated with the inositol ring have been identified and are termed diphosphoinositol pentakisphosphate (PP-IP<sub>5</sub>) and bis(diphospho)inositol tetrakisphosphate (bis-PP-IP<sub>4</sub>) (23–25). In PP-IP<sub>5</sub>, the pyrophosphate occurs at the 1 position, and in bis-PP-IP<sub>4</sub>, the two pyrophosphates are at the 1 and 2 or 1 and 4 positions (25). The functions of the inositol pyrophosphates have been difficult to demonstrate. We now report purification to homogeneity of an IP<sub>6</sub> kinase and demonstration that this enzyme has an ATP synthase activity, implying that the inositol pyrophosphates function as high-energy phosphate donors.

The publication costs of this article were defrayed in part by page charge payment. This article must therefore be hereby marked "advertisement" in accordance with 18 U.S.C. §1734 solely to indicate this fact.

## MATERIALS AND METHODS

**Materials.** [<sup>3</sup>H]IP<sub>6</sub>, [ $\gamma$ -<sup>32</sup>P]ATP, [<sup>3</sup>H]PP-IP<sub>5</sub>, PP-IP<sub>5</sub>, and bis-PP-IP<sub>4</sub> were supplied by DuPont/NEN. [<sup>32</sup>P]PP-IP<sub>5</sub> and [<sup>32</sup>P]bis-PP-IP<sub>4</sub> were synthesized as described (26) and supplied by DuPont/NEN. KH<sub>2</sub>PO<sub>4</sub>, K<sub>2</sub>HPO<sub>4</sub>, and KCl were from Aldrich. Mono Q HR 10/10 and Superdex 75 chromatography columns and gel filtration standards were from Pharmacia. Leupeptin, pepstatin, aprotinin, chymostatin, and antipain were purchased from Chemicon. Okadaic acid, 1,4,5-IP<sub>3</sub>, and inositol 1,3,4,5-tetrakisphosphate (IP<sub>4</sub>) were obtained from LC Services (Woburn, MA). Inositol 1,3,4,5,6-pentakisphosphate (1,3,4,5,6-IP<sub>5</sub>) and IP<sub>6</sub> were from Calbiochem. 1,2,4,5,6-IP<sub>5</sub> was a kind gift of James F. Maracek (State University of New York, Stony Brook). IP<sub>6</sub>-Affi-Prep resin was prepared as described (27). Polyethyleneimine (PEI)-cellulose TLC plates were obtained from EM Separations (Merck). Hepes buffer was from Research Organics. Nonradioactive nucleotides, nucleotide analogs, and nucleoside-diphosphate kinase (EC 2.7.4.6) were obtained from Boehringer Mannheim. SDS/PAGE gels and molecular mass standards were from NOVEX (San Diego). Protein assay dye reagents were from Pierce. Heparin-agarose, phosphocreatine, creatine kinase (EC 2.7.3.2), and all other reagents were from Sigma.

**Assay of IP<sub>6</sub> Kinase Activity.** Enzyme activity during the different purification steps was assayed in 10  $\mu\text{l}$  of reaction mixture containing 20 mM Hepes (pH 6.8), 1 mM dithiothreitol, 6 mM MgCl<sub>2</sub>, 5 mM Na<sub>2</sub>ATP, 10 mM phosphocreatine, 40 units/ml creatine kinase, 5 mM NaF, 5  $\mu\text{M}$  IP<sub>6</sub>, and 40–80 nM [<sup>3</sup>H]IP<sub>6</sub> and incubated at 37°C for 10–30 min under zero-order kinetics. Reactions were terminated either by addition of 1  $\mu\text{l}$  of 1 M HCl or by immersion in an ice water bath. We adapted methods used to separate IP<sub>6</sub>, IP<sub>5</sub>, IP<sub>4</sub>, IP<sub>3</sub>, ATP, and P<sub>i</sub> using PEI-cellulose TLC (28, 29) to separate IP<sub>6</sub>, PP-IP<sub>5</sub>, bis-PP-IP<sub>4</sub>, and ATP. The entire reaction mixture was spotted in 2.5- $\mu\text{l}$  aliquots onto a PEI-cellulose TLC plate, which was developed in 1.1 M KH<sub>2</sub>PO<sub>4</sub>/0.8 M K<sub>2</sub>HPO<sub>4</sub>/2.3 M HCl. The lanes were cut into 1-cm strips, and radioactivity was measured with 15 ml of DuPont Formula 963 scintillation cocktail. Approximately 25% of the added <sup>3</sup>H applied to the plates was recovered by this method. Alternatively, the PEI-cellulose TLC strips were first shaken with 3 ml of concentrated HCl, and radioactivity was measured with 5 ml of H<sub>2</sub>O and 10 ml of scintillation cocktail. Up to 80% of the added <sup>3</sup>H was recovered by this method. [<sup>3</sup>H]IP<sub>6</sub>, [<sup>3</sup>H]PP-IP<sub>5</sub>, and [<sup>3</sup>H]bis-PP-IP<sub>4</sub> migrated with  $R_f$  values of approximately 0.75, 0.45, and 0.3, respectively, and

Abbreviations: IP<sub>6</sub>, inositol hexakisphosphate; PP-IP<sub>5</sub>, diphosphoinositol pentakisphosphate; bis-PP-IP<sub>4</sub>, bis(diphospho)inositol tetrakisphosphate; IP<sub>3</sub>, inositol trisphosphate; IP<sub>4</sub>, inositol 1,3,4,5-tetrakisphosphate; IP<sub>5</sub>, inositol pentakisphosphate; PEI, polyethyleneimine. §To whom reprint requests should be addressed.

comigrated with standard preparations of [<sup>3</sup>H]IP<sub>6</sub>, [<sup>3</sup>H]PP-IP<sub>5</sub>, and [<sup>3</sup>H]bis-PP-IP<sub>4</sub>. No PP-IP<sub>5</sub> or bis-PP-IP<sub>4</sub> was formed under the following conditions: at 0°C; in zero time incubations; in the absence of brain extract or partially purified enzyme; with boiled (5 min) brain extract or partially purified enzyme; when partially purified enzyme was replaced with creatine kinase (40 units/ml) or nucleoside-diphosphate kinase (0.5 mg/ml); in the presence of 0.1 M HCl or 0.1 M NaOH; in the absence of ATP; or when ATP was replaced by 5 mM CTP, GTP, phosphocreatine, phosphoenolpyruvate, acetyl phosphate, or glucose 6-phosphate (data not shown). For *K<sub>m</sub>* determinations, done under initial rate conditions (<10% product formation), MgATP was used at the indicated concentrations, and 5 mM MgCl<sub>2</sub> was also included in the assay.

**Assay of ATP Synthase Activity.** The ability of the enzyme to synthesize ATP was assayed in 10 μl of reaction mixture containing 20 mM Hepes (pH 6.8), 1 mM dithiothreitol, 5 mM Na<sub>2</sub>ADP, 5 mM MgCl<sub>2</sub>, and 50 nM [<sup>32</sup>P]PP-IP<sub>5</sub> [20–100 Ci/mmol (1 Ci = 37 GBq)] and incubated at 37°C for 10–30 min. All solutions were adjusted to pH 6.8. The reactions were terminated as above and loaded onto PEI-cellulose TLC plates developed either as above, or in 1 M KH<sub>2</sub>PO<sub>4</sub>. In 1 M KH<sub>2</sub>PO<sub>4</sub>, [<sup>32</sup>P]PP-IP<sub>5</sub> and [<sup>32</sup>P]ATP migrated with *R<sub>f</sub>* values of approximately 0.05 and 0.55, respectively, and comigrated with standards (see Fig. 3). No [<sup>32</sup>P]ATP was formed from [<sup>32</sup>P]PP-IP<sub>5</sub> under the following conditions: at 0°C; in zero time incubations; in the absence of brain extract or partially purified enzyme; when partially purified enzyme was replaced with creatine kinase (40 units/ml) or nucleoside-diphosphate kinase (0.5 mg/ml); with boiled (5 min) brain extract or partially purified enzyme; or in the absence of ADP. No decrease in [<sup>32</sup>P]PP-IP<sub>5</sub> was observed when 5 mM ADP was replaced by 5 mM CDP, GDP, UDP, AMP, creatine, glucose, or 3-phosphoglycerate as phosphate acceptors. Replacing ADP with adenosine 5'-[β-thio]diphosphate (ADP[β-S]) resulted in the formation of a <sup>32</sup>P-labeled product of lower mobility than ATP (presumably ATP[β-S]) with 30% efficiency (data not shown). For *K<sub>m</sub>* determinations, done under initial rate conditions (<10% product formation), MgCl<sub>2</sub> was added in a 1:1 ratio with Na<sub>2</sub>ADP.

**Purification of the IP<sub>6</sub> Kinase.** Forebrains (70 g wet weight) from 70 adult male (175–300 g) Sprague–Dawley rats were homogenized in 150 ml of ice-cold buffer A (20 mM Hepes, pH 6.8/4 mM dithiothreitol/2 mM EGTA/0.75 mM EDTA/5 mM NaF/1.5 mM Na<sub>3</sub>VO<sub>4</sub>/0.5 mg/liter okadaic acid/4 mg/liter chymostatin/4 mg/liter pepstatin/4 mg/liter antipain/8 mg/liter leupeptin/8 mg/liter aprotinin/200 mg/liter phenylmethylsulfonyl fluoride). All subsequent procedures were performed at 0–4°C. The homogenate was centrifuged at 100,000 × *g* for 90 min. The supernatant was agitated with 70 ml of heparin-agarose in the presence of 1 mM MgCl<sub>2</sub> and 0.1% 3-[(3-cholamidopropyl)dimethyl ammonio]-1-propane sulfonate (CHAPS) for 60 min. The agarose was washed with 450 ml of 0.25 M KCl in buffer A with 1 mM MgCl<sub>2</sub> minus okadaic acid and was then eluted by agitation for 60 min with 75 ml of 0.5 M KCl at pH 7.4 with 1 mM EDTA in buffer A minus all phosphatase inhibitors, poured into a column, eluted, and then

eluted again with an additional 50 ml of the same buffer. The combined heparin column eluate was diluted to 1 liter in buffer B (20 mM Hepes, pH 6.8/1 mM dithiothreitol/1 mM EGTA/2 mg/liter chymostatin/2 mg/liter pepstatin/2 mg/liter antipain/4 mg/liter leupeptin/0.1% CHAPS) and applied to a Mono Q HR 10/10 fast protein liquid chromatography column at 4 ml/min, washed with 50 ml of 0.05 M KCl, and eluted at 0.5 ml/min with a 60-ml linear KCl gradient from 0.05 to 0.45 M KCl in buffer B plus 0.6% CHAPS. All further purification steps used this buffer. The pooled peak of IP<sub>6</sub> kinase activity was diluted to 0.05 M KCl and loaded onto a 1 × 8 cm IP<sub>6</sub>-Affi-Prep column at 0.5 ml/min, and the flow-through was passed over the column again. The column was then washed with 50 ml of 0.15 M KCl and eluted with 16 ml of 25 μM IP<sub>6</sub>. The peak fractions were diluted to 2 μM IP<sub>6</sub> and applied to a 0.5 × 4 cm heparin-agarose column, washed with 20 ml of 0.3 M KCl, and eluted at 0.2 ml/min with a 20-ml linear gradient from 0.3 to 0.6 M KCl. Peak fractions from the heparin column were concentrated in a Centriprep-30 concentrator (Amicon) to 100 μl and applied to a Superdex 75 gel filtration column, which was run at 0.2 ml/min. All columns were washed with 2.5 M KCl between runs.

**Other Methods.** 1,4,5-Trisphosphate 3-kinase activity was assayed as described (29). Protein assays were performed according to directions for Pierce Coomassie blue and Fluor-aldehyde protein assay reagents. SDS/PAGE was performed by the method of Laemmli (30).

## RESULTS

**Purification of IP<sub>6</sub> Kinase.** IP<sub>6</sub> kinase activity was monitored by measuring [<sup>3</sup>H]PP-IP<sub>5</sub> formed from [<sup>3</sup>H]IP<sub>6</sub>. <sup>3</sup>H-labeled IP<sub>6</sub>, PP-IP<sub>5</sub>, and bisPP-IP<sub>4</sub> were resolved by PEI-cellulose TLC, with no overlap between the inositol phosphates. In a screen of various tissues, the highest concentrations of enzyme activity are evident in the brain and testis. The thymus possesses about 60% as much activity as brain, whereas enzyme activity in heart, liver, and kidney are 5–25% of brain values. Because of the abundance of phosphatase activity in most tissues, these estimates may not reflect the absolute levels of the IP<sub>6</sub> kinase.

In brain extracts, virtually all IP<sub>6</sub> kinase activity is recovered in the soluble supernatant fraction after centrifugation at 100,000 × *g*. Enzyme activity adsorbs to a heparin column and is eluted with KCl, providing a 37-fold purification, and purification with the anion exchanger column Mono Q provides a further 4-fold enrichment of enzyme activity. The next greatest purification is obtained by adsorbing the enzyme to an IP<sub>6</sub> affinity column, affording a 13-fold increase in specific activity. Adsorption to a second heparin column with a gradient elution provides another 5- to 6-fold purification, and a gel filtration column affords another 3- to 4-fold purification. The final preparation is enriched 40,000-fold in enzyme activity and provides a 2.2% yield (Table 1).

SDS/PAGE of the purified enzyme reveals an apparent molecular mass of 54 kDa (Fig. 1). The elution profile of the enzyme from the gel filtration column is consistent with a

Table 1. Purification of IP<sub>6</sub> kinase

Fraction	Protein, μg	Specific activity, nmol/min/mg	Purification, fold	Recovery, %
100,000 × <i>g</i> supernatant	1,295,100	0.0365		
Heparin-agarose (batch)	31,000	1.37	37.5	90
Mono Q	4,050	5.48	151	47
IP <sub>6</sub> -Affi-Prep	73.5	72.9	1,998	11.4
Heparin-agarose	7.26	374.7	10,266	7.9
Superdex S75	0.662	1557.8	42,679	2.2

Enzyme was purified and fractions were assayed as described. Data presented are from a typical purification, which was repeated five times with similar results.

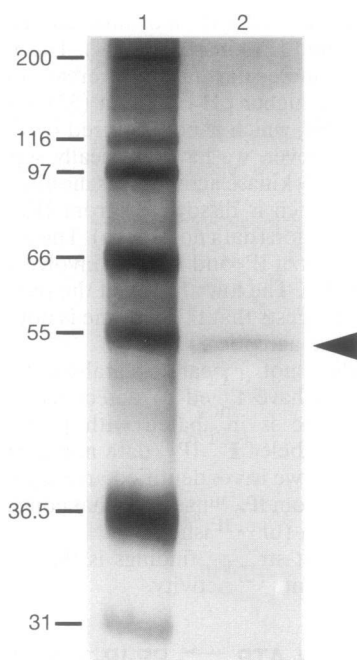


FIG. 1. SDS/PAGE analysis of purified IP<sub>6</sub> kinase. A 12% polyacrylamide gel was silver stained. Lane 1, molecular mass standards: myosin (200 kDa);  $\beta$ -galactosidase (116 kDa); phosphorylase *b* (97 kDa); bovine serum albumin (66 kDa); glutamate dehydrogenase (55 kDa); lactate dehydrogenase (36.5 kDa); carbonic anhydrase (31 kDa). Lane 2, 60 ng of purified IP<sub>6</sub> kinase (arrowhead).

molecular mass of about 60 kDa, which implies that IP<sub>6</sub> kinase is a monomer. The enzyme loses 50% of its activity when stored for 7 days at 4°C. Optimal stability is evident when the enzyme is stored with 20% glycerol at -70°C. Enzyme activity displays a broad pH optimum, with maximal activity at pH 6.8. A variety of buffers can be used, although Hepes is optimal.

The enzyme displays striking selectivity for IP<sub>6</sub> (Fig. 2*A* and Table 2). Unlabeled IP<sub>6</sub> inhibits [<sup>3</sup>H]PP-IP<sub>5</sub> formation by 50% at about 0.8  $\mu$ M. The product of the IP<sub>6</sub> kinase, PP-IP<sub>5</sub>, is 12% as potent as IP<sub>6</sub> as an inhibitor and is not a substrate for the kinase reaction (data not shown). The two isomers of IP<sub>5</sub> are only about 4% as potent. IP<sub>4</sub> is only 1.5% as potent as IP<sub>6</sub>, whereas both 1,3,4-IP<sub>3</sub> and 1,4,5-IP<sub>3</sub> fail to influence enzyme activity at concentrations of 100  $\mu$ M.

**IP<sub>6</sub> Kinase Has ATP Synthase Activity.** In the presence of [<sup>32</sup>P]PP-IP<sub>5</sub> and unlabeled ADP, [<sup>32</sup>P]ATP is formed (Fig. 3). The reaction is highly selective for ADP as GDP, CDP, and UDP are ineffective (Fig. 3). Also ineffective as PP-IP<sub>5</sub> phosphate acceptors are AMP, creatine, glucose, or 3-phosphoglycerate (data not shown). Unlabeled PP-IP<sub>5</sub> inhibits the ATP synthase activity of the enzyme by 50% at 1.8  $\mu$ M (Fig. 2*B* and Table 2). IP<sub>6</sub> has a similar IC<sub>50</sub> value, indicating that end-product inhibition may play a role in the regulation of this activity. 1,3,4,5,6-IP<sub>5</sub> is 14% as potent as PP-IP<sub>5</sub>, IP<sub>4</sub> is 5.5% as potent, and 1,4,5-IP<sub>3</sub> is 0.5% as potent at inhibiting enzyme activity.

Kinetic analysis reveals both the forward and reverse reactions to be random bireactant systems (Fig. 4 and Table 3). For the forward reaction, the *K<sub>m</sub>* values for IP<sub>6</sub> and ATP are 0.7  $\mu$ M and 1.35 mM, respectively. At concentrations of IP<sub>6</sub> > 50  $\mu$ M and ATP > 10 mM, substrate inhibition is observed (data not shown). For the reverse reaction, the *K<sub>m</sub>* values for PP-IP<sub>5</sub> and ADP are 1.97  $\mu$ M and 1.57 mM, respectively. The *V<sub>max</sub>* values for the forward and reverse reactions are 1.41 and 2.64  $\mu$ mol/min per mg of protein, respectively. Thus, the velocities of the two reactions appear to be similar. However, in the intact organism, the relative concentrations of the substrates are crucial determinants of the direction of an enzyme reac-

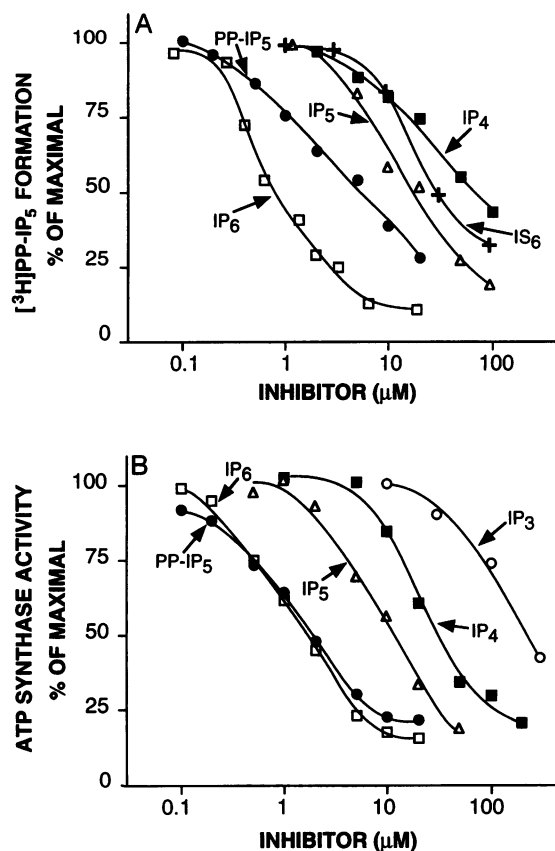


FIG. 2. Inositol phosphate specificity of IP<sub>6</sub> kinase forward and reverse reactions. (A) Inhibition of forward reaction ([<sup>3</sup>H]PP-IP<sub>5</sub> formation), assayed as described with 5 mM ATP and 41 nM [<sup>3</sup>H]IP<sub>6</sub>. IS<sub>6</sub>, inositol hexakisulfate. (B) Inhibition of reverse reaction (ATP synthase activity), assayed as described with 5 mM ADP and 50 nM [<sup>32</sup>P]PP-IP<sub>5</sub>. Peak fractions from the second heparin column were used (similar results were obtained with purified protein.) Data are means of triplicate determinations of three experiments.

tion. The concentration of PP-IP<sub>5</sub> in cells, 0.5–3.0  $\mu$ M (31), is similar to the observed *K<sub>m</sub>* value, whereas the concentration of IP<sub>6</sub> is considerably higher. Some compartmentalization or complexing of IP<sub>6</sub> is likely (4, 5), so that in cells, the enzyme may be exposed to a smaller effective concentration of IP<sub>6</sub>. The physiologic concentrations of ATP and ADP in mammalian brain, 2.7 and 0.7 mM, respectively (32), are similar enough to the observed *K<sub>m</sub>* values that both forward and reverse reactions are likely to take place *in vivo*, and the relative extent of the two may be determined by altered physiologic circumstances.

Table 2. Specificity of IP<sub>6</sub> kinase forward (IP<sub>6</sub> kinase) and reverse (ATP synthase) reactions

Inhibitor	IC <sub>50</sub> , $\mu$ M	
	IP <sub>6</sub> kinase	ATP synthase
IP <sub>6</sub>	0.8	1.8
PP-IP <sub>5</sub>	6.6	1.8
1,3,4,5,6-IP <sub>5</sub>	22	12.8
1,2,4,5,6-IP <sub>5</sub>	22*	ND
1,3,4,5-IP <sub>4</sub>	61	32.5
1,4,5-IP <sub>3</sub>	>100*	253
1,3,4-IP <sub>3</sub>	>100*	ND
IS <sub>6</sub>	30	ND

Enzyme activity was assayed as described in *Materials and Methods* and Fig. 2. ND, not determined.

\*Data not shown.

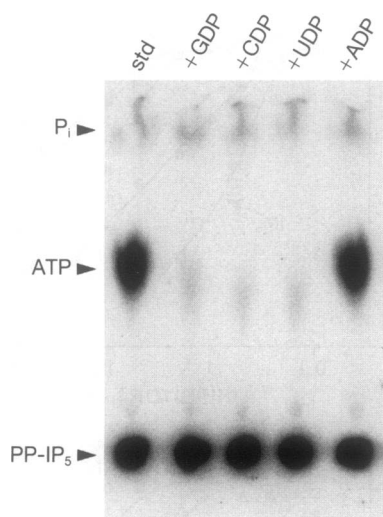


FIG. 3. Nucleoside diphosphate specificity of IP<sub>6</sub> kinase reverse reaction (ATP synthase activity). Lanes: std, approximately 15,000 cpm each of [ $\gamma$ -<sup>32</sup>P]ATP and [<sup>32</sup>P]PP-IP<sub>5</sub> were applied at the origin of the PEI-cellulose TLC plate; +GDP, +CDP, +UDP, and +ADP, the ATP synthase activity of the purified enzyme was assayed as described with the indicated nucleoside diphosphate at 5 mM. The TLC plate was exposed to a phosphor screen and scanned by a Molecular Dynamics PhosphorImager.

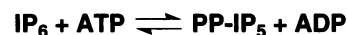
## DISCUSSION

The IP<sub>6</sub> kinase we have purified appears to be the enzyme responsible for the biosynthesis of PP-IP<sub>5</sub>, as it displays selectivity and high affinity for IP<sub>6</sub> as a substrate. The low affinity

for both 1,4,5-IP<sub>3</sub> and 1,3,4-IP<sub>3</sub> indicates that the enzyme is not identical to purified IP<sub>3</sub> kinases (33, 34). The 1,3,4-IP<sub>3</sub> kinase has a different molecular mass, whereas 1,4,5-IP<sub>3</sub> kinase activity displays a higher pH optimum (33) and is stimulated by calmodulin (29), which is not required by IP<sub>6</sub> kinase (data not shown). Moreover, we have physically separated the IP<sub>6</sub> kinase and 1,4,5-IP<sub>3</sub> kinase activities, as monitored by 1,4,5-IP<sub>3</sub> kinase activity, which is dissociated from IP<sub>6</sub> kinase activity during the purification (data not shown). The exact mechanism for the biosynthesis of IP<sub>5</sub> and IP<sub>6</sub> may involve specific IP<sub>4</sub> and IP<sub>5</sub> kinases (35, 36). The low affinity of the purified IP<sub>6</sub> kinase for IP<sub>4</sub> and IP<sub>5</sub> suggests that this enzyme is not responsible for these other kinase activities.

IP<sub>6</sub> kinase does not appear to catalyze the formation of bis-PP-IP<sub>4</sub>, as we have failed to detect its formation when purified IP<sub>6</sub> kinase is incubated with [<sup>3</sup>H]PP-IP<sub>5</sub> or with [<sup>32</sup>P]ATP and unlabeled PP-IP<sub>5</sub> (data not shown). In preliminary experiments, we have identified an enzyme fraction that can be separated from IP<sub>6</sub> kinase and that converts [<sup>3</sup>H]PP-IP<sub>5</sub> into [<sup>3</sup>H]bis-PP-IP<sub>4</sub> (unpublished).

One of our most striking findings is that the purified IP<sub>6</sub> kinase has ATP synthase activity.



Structure 1

We have demonstrated that the [<sup>32</sup>P]ATP formed by IP<sub>6</sub> kinase can phosphorylate protein substrates (data not shown). The enthalpy change ( $\Delta H$ ) of the hydrolysis of a pyrophosphate bond is on the order of  $-3$  kcal/mol (37) (1 kcal = 4.18 kJ). The pyrophosphate of PP-IP<sub>5</sub> formed by an ATP-dependent kinase has been assigned to the C-1 of the inositol ring in

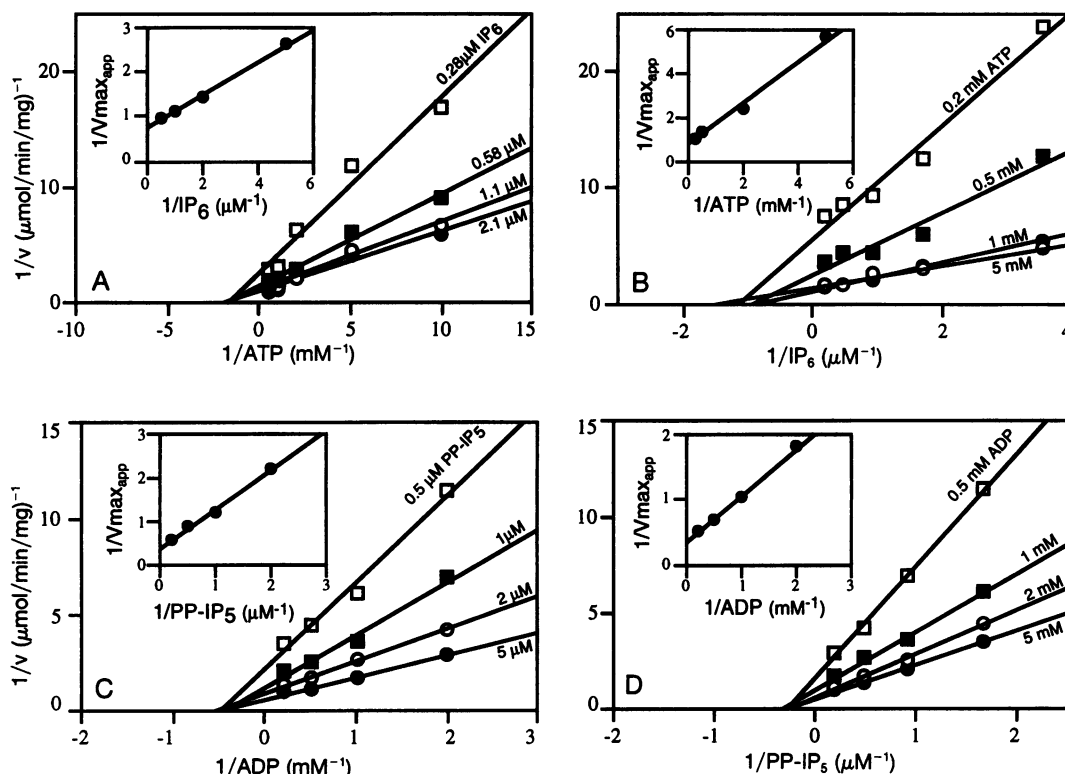


FIG. 4. Determination of IP<sub>6</sub> kinase kinetic constants and reaction mechanism. Enzyme was purified and assayed as described. Dependence of IP<sub>6</sub> kinase activity on the concentration of IP<sub>6</sub> (A) and ATP (B). Dependence of reverse reaction (ATP synthase activity) on the concentration of PP-IP<sub>5</sub> (C) and ADP (D). For A and B, velocity was measured as micromoles of PP-IP<sub>5</sub> formed per minute per milligram of protein. For C and D, velocity was measured as micromoles of ATP formed per minute per milligram of protein. The insets are plots of the reciprocals of the  $1/v$  intercepts ( $1/V_{\text{max,app}}$ ) vs. the reciprocals of substrate concentration. Data are means of triplicate determinations from a representative experiment, which was repeated twice with similar results.

Table 3. Properties of IP<sub>6</sub> kinase

Kinetic constant	Value
$K_m$ for IP <sub>6</sub>	0.70 ± 0.192 μM
$K_m$ for ATP	1.35 ± 0.106 mM
$V_{max}$ (forward)	1.41 ± 0.168 μmol/min per mg
$K_m$ for PP-IP <sub>5</sub>	1.97 ± 0.312 μM
$K_m$ for ADP	1.57 ± 0.243 mM
$V_{max}$ (reverse)	2.64 ± 0.113 μmol/min per mg

Purified enzyme was assayed as described. Values are means (±SD) of three determinations.

mammalian cells (25). Sequential hydrolysis of the phosphate esters of IP<sub>6</sub> has shown that the apparent  $\Delta H$  of the C-1 monophosphate ester of IP<sub>6</sub>, -4 kcal/mol, is higher than that of any of the other positions tested (38). [The C-2 position was not accessible but is thought to possess a higher energy because of charge repulsion (39)]. A pyrophosphate linkage at the C-1 position may possess an even higher  $\Delta H$ . The bond energy of the P—O—P linkage in PP-IP<sub>5</sub> has been predicted to be -6.6 kcal/mol, similar to ADP (25). The fact that the IP<sub>6</sub> kinase can readily transfer a phosphate group from PP-IP<sub>5</sub> to ADP to form ATP at 37°C suggests that the free energy of hydrolysis of the pyrophosphate bond of PP-IP<sub>5</sub> is similar to that of ATP. Further investigation will be required to determine whether the ATP synthase activity of this enzyme occurs *in vivo*. In liver homogenates incubated with [<sup>32</sup>P]PP-IP<sub>5</sub>, the labeled phosphate group is recovered only as free P<sub>i</sub> formed by the action of a specific phosphatase, although PP-IP<sub>5</sub> does accumulate in intact cells (31).

IP<sub>6</sub>, or phytate, has long been known to exist in high concentrations in plant seeds. It may be important as an antioxidant during dormancy (7) and may be the storage form for phosphate that can be converted into ATP during germination (8). A kinase from mung bean seeds that transfers a phosphate from IP<sub>6</sub> to GDP (41) was identified, and two kinases that can transfer a phosphate from IP<sub>6</sub> to ADP have been isolated from mung bean (42) and soybean (43) seeds. The phosphotransferase reaction of the soybean 1,3,4,5,6-IP<sub>5</sub> 2-kinase was observed in both directions *in vitro* and may proceed in both directions *in vivo* depending on substrate concentrations, including the ATP/ADP ratio of maturing versus germinating seeds (43).

Our results suggest that PP-IP<sub>5</sub> constitutes an energy reservoir for mammalian cells analogous to IP<sub>6</sub> in plants or to phosphocreatine. Because endogenous concentrations of PP-IP<sub>5</sub> are substantially lower than those of phosphocreatine, PP-IP<sub>5</sub> is not likely to be a high-energy phosphate donor for general cellular functions. Instead, PP-IP<sub>5</sub> may have a more specialized role, either by generating ATP or by transferring its energy directly. One role may involve vesicle trafficking, especially in the brain, where IP<sub>6</sub> kinase activity is highest. The purified IP<sub>6</sub> receptor binding protein is identical to the clathrin assembly protein AP-2, which is a component of clathrin-coated pits and vesicles (14, 15). PP-IP<sub>5</sub> and IP<sub>6</sub> also bind with high affinity and specificity to the clathrin assembly protein AP-3 (16, 17), Golgi coatmer (18), and yeast coatmer (26). Although the inositol pyrophosphates have not been directly tested in each case, higher inositol polyphosphates also bind to the synaptic vesicle associated protein synaptotagmin (44),  $\beta$ -arrestin (22), and the cytoskeletal protein band 4.1 (45), indicating a possible role for these inositol compounds at the synapse.

Endocytosis and vesicle recycling are among the most dynamic processes of cellular regulation. Several specific stages of the endocytic process depend on unknown cytosolic factors as well as energy donors. Initiation or nucleation, membrane fission (46), and maintenance of an unassembled pool of coat proteins (47) are all energy-requiring steps of the endocytic process. ATP is required, and other energy sources

may be involved as well. Some endocytosis occurs in ATP-depleted cells (48), various methods of ATP depletion differentially affect measures of endocytosis, and neither levels of ATP nor rate of ATP depletion is directly correlated to loss of cell surface receptors (49).

Hydrolysis of the high-energy phosphate bonds of ATP and GTP also drive vesicle uncoating, docking, and fusion events, providing a directional regulatory switch (50). Misassembly of coated pits (51, 52) and functional differences in coat proteins isolated from assembled vs. unassembled pools (53) indicate a regulatory switch in coat assembly (50, 54). IP<sub>6</sub> inhibits AP-2 self-assembly, AP-2 binding to clathrin, and clathrin coat assembly (55). IP<sub>6</sub> and PP-IP<sub>5</sub> also inhibit AP-3 (16, 17) and coatmer-mediated clathrin coat assembly *in vitro* (26), indicating a role for IP<sub>6</sub> or PP-IP<sub>5</sub> in the regulation of coat assembly and disassembly in the cell. Another more specialized regulatory switch or "fusion clamp" (50, 56) may be involved in regulated fusion events like synaptic vesicle exocytosis and recycling or receptor down-regulation, which may involve cross-talk with second messenger systems. Higher inositol phosphates bind to synaptotagmin at the phospholipid binding domain CL2B (44), interfering with  $\beta$ -SNAP/synaptotagmin binding (21) and vesicle fusion (19). A regulatory mechanism could involve phosphate transfer or hydrolysis of PP-IP<sub>5</sub> bound at such domains. The high rate of turnover of PP-IP<sub>5</sub> (31) and its regulation by intracellular calcium (57) suggest that interconversion of IP<sub>6</sub> and PP-IP<sub>5</sub> and/or bisPP-IP<sub>4</sub> would be an excellent candidate as an energy source and/or regulatory switch for these processes.

We thank Jonathan N. Sachs and Adriana Izquierdo for technical assistance. This work was supported by U.S. Public Health Service Grant MH-18501 and Research Scientist Award DA-00074 to S.H.S., Training Grant GM-07309 to S.M.V., and U.S. Public Health Service Grant NS-29632 to G.D., J.D.O., and G.D.P.

- Berridge, M. J. & Irvine, R. F. (1989) *Nature (London)* **341**, 197-204.
- Szwergold, B. S., Graham, R. A. & Brown, T. R. (1987) *Biochem. Biophys. Res. Commun.* **149**, 874-881.
- Pittet, D., Schlegel, W., Lew, D. P., Monod, A. & Mayr, G. W. (1989) *J. Biol. Chem.* **264**, 18489-18493.
- Irvine, R. F., Moor, R. M., Pollock, W. K., Cmith, P. M. & Wregget, K. A. (1988) *Philos. Trans. R. Soc. London B* **320**, 281-298.
- Downes, C. P. (1988) *Trends Neurosci.* **11**, 336-339.
- Stuart, J. A., Anderson, K. L., French, P. J., Kirk, C. J. & Mitchell, R. H. (1994) *Biochem. J.* **303**, 517-525.
- Graf, E., Empson, K. L. & Eaton, J. W. (1987) *J. Biol. Chem.* **262**, 11647-11650.
- Morton, R. K. & Raison, J. K. (1963) *Nature (London)* **200**, 429-433.
- Hallberg, L., Brune, M. & Rossander, L. (1989) *Am. J. Clin. Nutr.* **49**, 140-144.
- Graf, E. & Eaton, J. W. (1993) *Nutr. Cancer* **19**, 9-11.
- Sasakawa, N., Sharif, M. & Hanley, M. R. (1995) *Biochem. Pharmacol.* **50**, 137-146.
- Poyner, D. R., Cooke, F., Hanley, M. R., Reynolds, D. J. & Hawkins, P. T. (1993) *J. Biol. Chem.* **268**, 1032-1038.
- Hawkins, P. T., Poyner, D. R., Jackson, T. R., Letcher, A. J., Lander, D. A. & Irvine, R. F. (1993) *Biochem. J.* **294**, 929-934.
- Voglmaier, S. M., Keen, J. H., Murphy, J., Ferris, C. D., Prestwich, G., Snyder, S. H. & Theibert, A. B. (1992) *Biochem. Biophys. Res. Commun.* **187**, 158-163.
- Timerman, A. P., Mayrlleitner, M. M., Lukas, T. J., Chadwick, C. C., Saito, A., Watterson, D. M., Schindler, H. & Fleischer, S. (1992) *Proc. Natl. Acad. Sci. USA* **89**, 8976-8980.
- Norris, F. A., Ungewickell, E. & Majerus, P. W. (1995) *J. Biol. Chem.* **270**, 214-217.
- Ye, W., Ali, N., Bembenek, M. E., Shears, S. B. & Lafer, E. M. (1995) *J. Biol. Chem.* **270**, 1564-1568.
- Fleischer, B., Xie, J., Mayrlleitner, M., Shears, S. B., Palmer, D. J. & Fleischer, S. (1994) *J. Biol. Chem.* **269**, 17826-17832.

19. Llinas, R., Sugimori, M., Lang, E. J., Moeir, M., Fukuda, M., Niinobe, M. & Mikoshiba, K. (1994) *Proc. Natl. Acad. Sci. USA* **91**, 12990–12993.
20. Niinobe, A. U., Yamaguchi, Y., Fukuda, M. & Mikoshiba, K. (1994) *Biochem. Biophys. Res. Commun.* **205**, 1036–1042.
21. Schiavo, G., Gmachi, M. J. S., Stenbeck, G., Sollner, T. H. & Rothman, J. E. (1995) *Nature (London)* **378**, 733–734.
22. Palczewski, K., Pulvermuller, A., Buczylo, J., Gutmann, C. & Hofmann, K. P. (1991) *FEBS Lett.* **295**, 195–199.
23. Mayr, G. W., Radenberg, T., Thiel, U., Vogel, G. & Stephens, L. R. (1992) *Carbohydr. Res.* **234**, 247–262.
24. Menniti, F. S., Miller, R. N., Putney, J. W., Jr., & Shears, S. B. (1993) *J. Biol. Chem.* **268**, 3850–3856.
25. Stephens, L., Radenberg, T., Thiel, U., Vogel, G., Khoo, K. H., Dell, A., Jackson, T. R., Hawkins, P. T. & Mayr, G. W. (1993) *J. Biol. Chem.* **268**, 4009–4015.
26. Ali, N., Duden, R., Bembenek, M. E. & Shears, S. B. (1995) *Biochem. J.* **310**, 279–284.
27. Marecek, J. F. & Prestwich, G. D. (1991) *Tetrahedron Lett.* **32**, 1863–1866.
28. Spencer, C. E. L., Stephens, L. R. & Irvine, R. F. (1990) in *Methods in Inositide Research*, ed. Irvine, R. F. (Raven, New York), pp. 39–43.
29. Ryu, S. H., Lee, S. Y., Lee, K. Y. & Rhee, S. G. (1987) *FASEB J.* **1**, 388–393.
30. Laemmli, U. K. (1970) *Nature (London)* **227**, 680–685.
31. Shears, S. B., Craxton, A., Nawab, A. & Bembenek, M. E. (1995) *J. Biol. Chem.* **270**, 10489–10497.
32. Veech, R. L., Lawson, J. W. R., Cornell, N. W. & Krebs, H. A. (1979) *J. Biol. Chem.* **254**, 6538–6547.
33. Johanson, R. A., Hansen, C. A. & Williamson, J. R. (1988) *J. Biol. Chem.* **263**, 7465–7471.
34. Abdullah, M., Hughes, P. J., Craxton, A., Gigg, R., Desai, T., Marecek, J. F., Prestwich, G. D. & Shears, S. B. (1992) *J. Biol. Chem.* **267**, 22340–22345.
35. Menniti, F. S., Oliver, K. G., Putney, J. W., Jr., & Shears, S. B. (1993) *Trends Biol. Sci.* **18**, 53–56.
36. Stephens, L., Hawkins, P. T., Stanley, A. F., Moore, T., Poyner, D. R., Morris, J., Hanley, M. R., Kay, R. R. & Irvine, R. F. (1991) *Biochem. J.* **275**, 485–499.
37. Ging, N. S. & Sturtevant, J. M. (1954) *J. Am. Chem. Soc.* **76**, 2087–2091.
38. Raison, J. K. & Evans, W. J. (1968) *Biochim. Biophys. Acta* **170**, 448–451.
39. Atkinson, M. R. & Morton, R. K. (1960) in *Comparative Biochemistry*, eds. Morton, R. K. & Mason, H. (Academic, New York), pp. 1–95.
40. Ali, N., Craxton, A. & Shears, S. (1993) *J. Biol. Chem.* **268**, 6161–6167.
41. Biswas, S., Burman, S. & Biswas, B. B. (1975) *Photochemistry* **14**, 373–375.
42. Biswas, S., Maity, I. B., Chakrabarti, M. S. & Biswas, B. B. (1978) *Arch. Biochem.* **185**, 557–566.
43. Phillippy, B. Q., Ullah, A. H. & Ehrlich, K. C. (1994) *J. Biol. Chem.* **269**, 28393–28399.
44. Fukada, M., Aruga, J., Niinobe, M., Aimoto, S. & Mikoshiba, K. (1994) *J. Biol. Chem.* **269**, 29206–29211.
45. Ouggouti, S., Bournier, O., Boivin, P., Bertrand, O. & Dhermy, D. (1992) *Protein Expression Purif.* **3**, 488–496.
46. Schmid, S. L. & Smythe, E. (1991) *J. Cell Biol.* **114**, 869–880.
47. Schmid, S. L. (1992) *BioEssays* **14**, 589–596.
48. Smythe, E., Pypaert, M., Lucocq, J. & Warren, G. (1989) *J. Cell Biol.* **108**, 843–853.
49. Schmid, S. L. & Carter, L. L. (1990) *J. Cell Biol.* **111**, 2307–2318.
50. Rothman, J. E. (1994) *Nature (London)* **372**, 55–63.
51. Seaman, M. N. J., Ball, C. L. & Robinson, M. S. (1993) *J. Cell Biol.* **123**, 1093–1105.
52. Wang, L.-H., Rothberg, K. G. & Anderson, R. G. W. (1993) *J. Cell Biol.* **123**, 1107–1117.
53. Smythe, E., Carter, L. L. & Schmid, S. L. (1992) *J. Cell Biol.* **119**, 1163–1171.
54. Pfeffer, S. R. (1994) *Curr. Opin. Cell Biol.* **6**, 522–526.
55. Beck, K. & Keen, J. (1991) *J. Biol. Chem.* **266**, 4442–4447.
56. Sollner, T., Bennett, M. K., Whiteheart, S. W., Scheller, R. H. & Rothman, J. E. (1993) *Cell* **75**, 409–418.
57. Glennon, M. C. & Shears, S. B. (1993) *Biochem. J.* **293**, 583–590.