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Application of nanotechnology in improving bioavailability and bioactivity of diet-derived phytochemicals

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Abstract

Nanotechnology is an innovative approach that has potential applications in nutraceutical research. Phytochemicals have promising potential for maintaining and promoting health, as well as preventing and potentially treating some diseases. However, the generally low solubility, stability, bioavailability and target specificity, together with the side-effects seen when used at high levels, have limited their application. Indeed, nanoparticles can increase solubility and stability of phytochemicals, enhance their absorption, protect them from premature degradation in the body, and prolong their circulation time. Moreover, these nanoparticles exhibit high differential uptake efficiency in the target cells (or tissue) over normal cells (or tissue)through preventing them from prematurely interacting with the biological environment, enhanced permeation and retention effect in disease tissues, and improving their cellular uptake, resulting in decreased toxicity, In this review we outline the commonly used biocompatible and biodegradable nanoparticles including liposomes, emulsions, solid lipid nanoparticles, nanostructured lipid carriers, micelles and poly (lactic-co-glycolic acid) (PLGA) nanoparticles. We then summarize studies that have used these nanoparticles as carriers for EGCG, quercetin, resveratrol and curcuminadministration to enhance their aqueous solubility, stability, bioavailability, target specificity, and bioactivities.

Keywords

Nanotechnology; Phytochemicals; Nanoparticles; Bioactivities; Biocompatible and biodegradable

1. Introduction

Nanotechnology is the study of the control of matter generally in the size range of 100 nm or smaller (1). As a comparison, an H atom has a size of 0.1 nm in diameter, a lysosome is between 200 to 500 nm, an *E. Coli* bacterium is about 2 µm in length, and most of eukaryotic cells have a size between 8 and 30 µmin diameter or larger [1]. The size of proteins is in a range between 3 and 90 nm, therefore, many enzymes, signaling molecules

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and receptors are in the nanoscalerange [1]. Since most of the biological processes occur at the nanoscale, nanoparticulate technology has a promising future in developing novel preventive, diagnostic, and therapeutic agents [2]. Such an application, often called the nanomedicine, has recently gained tremendous attention in pharmaceutical sciences[3]. In contrast, the application of nanotechnology in nutraceutics is far behind. Many nutrients, phytochemicals, and other natural compounds can be loaded into biocompatible and biodegradable nanoparticles, which will improve their aqueous solubility, stability, bioavailability, circulation time and target specificity, i.e., more nanoparticles enter disease tissues, due to leaky vasculature, but less to normal tissues[4].

2. Biocompatible and biodegradable nanoparticles

The common biocompatible and biodegradable nanoparticles include nanoliposomes, nanoemulsions, lipid nanocarries, micelles and poly(lactic-co-glycolic acid) (PLGA) nanoparticles.

2.1. Liposomes

Liposomes have lipid bilayed membrane structures composed of phospholipids, which have hydrophilic heads and hydrophobic fatty acid tails(Figure 1A). Initially, they were used to study biological membranes in the mid-1960s[5–7]. Since then, their application has been extended to a variety of areas such as in drug delivery, cosmetic formulations, diagnostic agents, and food industry[6,8–10]. Some liposome-based drugs have been approved by Food and Drug Administration (FDA) and they are available in the market for treating different diseases[11]. Due to its biphasic character, liposomes can serve as carriers for both hydrophilic(in the central aqueous compartment) and hydrophobic (in lipid bilayers)compounds[8].

The term nanoliposome has been introduced recently to exclusively refer to nanometricsize of liposomes[12]. Although, in a broad sense, liposomes and nanoliposomes have the same chemical, structural and thermodynamic properties, the smaller size of nanoliposomes could produce larger interfacial area of encapsulated compounds with biological tissues and thus provide higher potential to increase the bioavailability of encapsulated compounds[12]. Especially for solid tumor treatment, nanoliposomes can accumulate more in tumors because of the enhanced permeation and retention (EPR) effect[12,13]. Higher energy input is required to produce nanoliposomes in the aqueous solution[9]. The commonly used methods for nanoliposome synthesis include sonication, extrusion, freeze-thawing, ether injection, and microfluidization. Sonication and extrusion are widely used in the laboratory scale[9,14]. High power, a long period and small pore size of the extruder filtration can generate small size of nanoliposomes. Microfluidization method is a commonly used technique for industrial manufacturers, which involves high pressure and high force technologies using a device called a microfluidizer to produce a flow stream passing through a fine orifice in order to reduce particle sizes of liposomes[9,14]. The notable advantages of this method are the adjustable size, high reproducibility for large scale of nanoliposome preparation, and noexposure to toxic organic solvent [14].

Nanoliposomes can be administered parenterally, orally, topically, or nasally [12,15,16]. Nanoliposomesin the circulatory system are recognized as foreign particles and are rapidly cleared by the reticuloendothelial system (RES)[17]. Additionally, electrostatic, hydrophobic, and van der Waals forces can disintegrate nanoliposomes[18,19]. Therefore, steric stabilization is required and can be achieved by coating the nanoliposomes with inert polymers[20,21]. The polymer coating reduces adsorption of opsonins and avoids rapid RES clearance[20]. Poly(ethylene glycol) (PEG)or poloxamer can form a sterically stabilized corona on nanoliposomes[17]. This "STEALTH" technology increases circulation time of

nanoliposomes[20]. In 1995, the FDA approved the first liposomal drug, a PEG-lated liposomal formulation of doxorubicin (Doxil in the U.S. and Caelyx outside the U.S.), for the treatment of Kaposi's sarcoma[11]. Doxil is a liquid suspension of 80-100nm sterically stabilized nanoliposomes containing doxorubicin HCl at 2mg/ml[11]. PEG-lated liposomes significantly decrease doxorubicin'scardiotoxicity and increase the circulation half-life of doxorubicin from several minutes to more than 20hours[22,23]. Due to the success of liposomal doxorubicin, many liposomal formulations have been developed and are currently under test in clinical trials.

2.2. Emulsions

An emulsion is a mixture composed of two immiscible liquids. When oil is dispersed it will form into droplets through the aqueous phase; this is referred to as oil-in-water (O/W) emulsions (Figure 1B). On the contrary an aqueous solution dispersed in oil phase is referred to as water-in-oil (W/O) emulsions[24]. In order to disperse two immiscible liquids and to stabilize the emulsion structure, a surfactant or emulsifier is required, which has the amphiphilic structure with one fragment being hydrophilic and the other one being hydrophobic[25,26]. Emulsifiers can reduce the interfacial tension, create a film over one phase to repel the other phase, maintain and stabilize the emulsion structure, and increase the viscosity of the medium. Most of emulsions are of the O/W type, especially those designed for parenteral or oral administration[26]. In 1972, FDA approved the first intravenous fat emulsion, Intralipid®, which was composed of egg phospholipids, soy bean oil, and glycerin. Intralipid® is used to deliver essential fatty acids through intravenous injection for the patients who are unable to absorb those nutrients through diet[27]. The success of clinical application of this emulsion has paved the road for encapsulating and dissolving hydrophobic compounds into the internal oil core of emulsions for treating other diseases and disorders[28]. Several commercially available products, such as diazepam (Diazemuls®), propofol (Diprivan®), vitamin A,D,E,K (Vitalipid®), are developed using the O/W emulsion technique. Emulsion can be used to deliver many bioactive lipids and hydrophobic components including omega-3 fatty acids, carotenoids, phytosterols, flavonoids and other phytochemicals[26]. Nanoemulsions, having a size less than 100 nm in diameter, require more surfactants or co-surfactants and high energy input to lower the surface tension in order to be small and thermodynamically stable[29]. Sonication and homogenization are common methods for making nanoemulsions in the laboratory scale[30]. High-pressure homogenization and microfluidization can be used to produce nanoemulsions on a large scale [31].

Oral route is the easiest and most convenient route for the chronic delivery of preventive and therapeutic nutrients and dietary supplements, however, many phytochemicals have extremely low levels of oral bioavailability [32–35]. Novel self-emulsifying drug delivery systems (SEDDS) have received considerable attention due to their capability of improving oral absorption of highly hydrophobic compounds [36,37]. SEDDS formulations are isotropic mixture of oils, surfactants, nutrients (or drugs), usually with one or more of cosurfactants or co-emulsifiers [32,36]. When oral administration of SEDDS is in solid, liquid, or semi-liquid form, the mixture is dispersed into gastrointestinal fluids, yielding fine O/W emulsions containing hydrophobic compounds upon gentle agitation in gastrointestinal tract [36]. The SEDDS have small size and large surface areas, and thus enhance aqueous solubility of hydrophobic compounds, which in turn contributes to improved oral bioavailability [32].

2.3. Solid lipid nanoparticles (SLNs) and nanostructured lipid carriers (NLCs)

SLNs have a similar structure of the O/W nanoemulsion including a hydrophilic shell and a hydrophobic lipid core, which is solid at room temperature(Figure 1C)[38]. SLNs were

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developed in the early 1990s as an alternative novel carrier system to traditional nanocarriers such as polymeric nanoparticles, nanoemulsions, and nanoliposomes[39]. The hydrophobic compounds can be encapsulated into the solid lipid core, resulting in an increased stability, reduced degradation, sustained and prolonged release. This process can minimize toxicity, and improve target specificity of the compounds to disease tissues[40]. To achieve and to maintain a solid lipid nanocarrier structure, lipids in SLNs have relatively high melting temperature. SLNs are usually composed of solid lipids, surfactants, water, with or without co-surfactants[41]. The commonly used lipids include fatty acids(stearic acid), triglycerides (e.g. tristearine), waxes (e.g. cetylpalmitate), or a mixture of the above lipids[41,42]. Surfactants are used to stabilize the lipid dispersion. The typical surfactants used include bile salts, phospholipids, sorbitan esters, fatty acid ethoxylates, or a mixture of these components [41]. SLNs are favored because of their high biocompatibility, avoidance of organic solvent, excellent reproducibility even using different preparation methods, easily scaled-up synthesis processes [38]. Although SLNs are one of the most useful lipid based nanocarriers in nutraceutical and pharmaceutical research, some limitations exist, such as low compound loading capacity and leakage during storage[42]. These limitations are overcome by the recently developed NLCs, an improved, new generation of lipid nanocarriers[42]. NLCs are favorable because they are small in size, stable, biocompatible and biodegradable, and have high loading capacity [38]. In order to enhance the loading capacity, more complex lipid mixtures are employed in the hydrophobic core. For example, a mixture of mono-, di-, or triglycerides with fatty acids of different chain length forms less perfect crystals, which can accommodate more hydrophobic drugs, nutrients, phytochemicals, or other compounds to avoid expulsion[43,44]. NLCs represent a new delivery system for poorly soluble compounds, such as phytochemicals [45,46].

Hydrophobicphy to chemicals, such as quercetin and resveratrol, can be easily encapsulated into the lipid core and are stable in the lipid core [38,47]. The common methods for making SLNs and NLCs include high pressure homogenization, cold homogenization, hot homogenization/ultra sonication, phase inversion and solvent evaporation/ emulsification[44]. The most cost-effective and relatively simple way for the larger scale production is a high pressure homogenization method[48].

Emulsification/ultra sonication technique is a common laboratory approach for making both SLNs and NLCs. To begin this procedure the solid lipid (for making SLNs) or solid-liquid lipid mixture(for making NLCs) is melted, and the hydrophobic nutrients or phytochemicals are dissolved in melted lipid phase. Next, the melted lipid and hydrophobic compound mixture is dispersed in a hot aqueous surfactant/stabilizer solution, and they are stirred at a high speed at the equivalent melting temperature. Lastly, obtained pre-emulsion is homogenized or sonicated yielding a hot O/W nanoemulsion. After cooling, SLNs or NLCs are formed[44]. SLNs and NLCs have been extensively investigated for application in pharmaceutics, cosmetics, food, and agricultural products[38,49]. They can be administrated through oral, parenteral, dermal, rectal, nasal, ocular and pulmonary routes[50]. Oral administration is the most preferable route for the application of SLNs and NLCs [50]. Both SLNs and NLCs are widely used to improve bioavailability and to achieve sustained release for hydrophobic nutrients and phytochemicals. The encapsulated hydrophobic compounds are absorbed into lacteals after oral administration of SLNs or NLCs[51]. Then, these compounds are transported from intestinal lymphatic vessels to the thoracic lymph duct, and eventually into the system circulation at the junction of the jugular and left subclavian vein[52]. The transport can avoid the first pass effect, which can metabolize the encapsulated compound in liver before it release into the circulation system, and therefore enhances bioavailability.

2.4. Micelles

Micelles belong to a group of amphiphilic colloids, which are composed of amphiphilic monomers including phospholipids and some polymers. Micelles usually have a size between 20 and 80 nm in diameter. The traditional micelles are lipid based[53]. When amphiphilic phospholipid concentrations reach the critical micelle concentration (CMC) and temperatures reach the critical micellization temperature, micelles are formed [53,54]. Hydrophilic heads of phospholipids form the shell of micelles, while fatty acid tails of phospholipids form a hydrophobic core, which can accommodate phytochemicals and other hydrophobic compounds (Figure 1D). More recently, polymeric micelles are of great interest to the investigators. Polymeric micelles are made of block-copolymers that consist of hydrophobic and hydrophilic monomer units [55]. Two monomers with different hydrophobicity can be conjugated and form a core-shell micelle structure, where the hydrophilic and hydrophobic blocks form the micelle shell and core, respectively (Figure 1E). The commonly used hydrophilic monomer is polyethylene glycol (PEG), and the widely used core-forming blocks are poly(propyleneoxide), poly(caprolactone), poly(D, Llactic acid) and poly(L-aspartic acid)[55,56]. Micelles have been used as various hydrophobic compound carriers for oral, nasal, topical, parenteral and ocular application [55]. Micelles can increase aqueous solubility of hydrophobic compounds, extend their blood circulation time, increase target specificity to disease tissues through enhanced EPR effect, and lower toxicity of compounds to normal tissues[57,58].

2.5. PLGA nanoparticles

PLGA is a widely used biocompatible and biodegradable polymers[59]. Due to minimal systemic toxicity, PLGA has been approved by FDA for developing therapeutic device[59]. PLGA can be hydrolyzed in the body and yield biodegradable lactic acid and glycolic acid. PLGA nanoparticles have been used as carriers for many phytochemicals such as quercetin and curcumin[60–62]. Those phytochemical can either entrapped inside or adsorbed on the surface of PLGA nanoparticle [63]. The solvent evaporation, the emulsification-diffusion, and the nanoprecipation methods can be used to synthesize PLGA nanoparticles [59,64,65].

2.6. Characteristics of nanoparticles

Characteristics of nanoparticles determine their functions. The major characteristics include, as follows: 1) nanoparticle size, which can be measured using a dynamic light scattering method, or a transmission electron microscope (TEM) or a scanning electron microscope (SEM); 2) Zeta potential, which indicates the surface charge of nanoparticles and can be measured using a Zeta potential analyzer; 3) polydispersity index, which indicates nanoparticle size distribution and can be measured by a dynamic light scattering method; 4) physical and chemical stability, which indicates the stability of nanoparticles and loaded compounds, respectively;5)encapsulation efficiency, which is determined as mass of encapsulated compound divided by mass of total compound \times 100%; 6) loading capacity, which is determined as mass of encapsulated compound divided by mass of nanoparticles $\times100\%$.

While in most cases, smaller size is preferable for enhancing absorption of encapsulated compounds into target tissues, it is not always true because the small nanoparticles also easily move in and out of the target tissues[66]. Additionally, smaller size requires a larger amount of surfactants, which may introduce toxicity, and may also hinder drug absorption under certain circumstances. However, the nanoscale is not well defined as many particles have a size larger than 100 nm in diameter and are still referred to as nanoparticles. Nanoparticles can be modified to improve their stability and functions. For example, an uptake enhancer can be coated on the surface of nanoparticles to increase uptake and bioavailability of loaded compounds [67,68]. Poly(ethylene glycerol) (PEG) can be

incorporated on the surface of nanoparticles to maintain their integrity and stability [69–71] and protect them from degradation by enzymes. PEG can also prolong the circulation of nanoparticles by stabilizing them against opsonization[2,3]. Target ligands can be incorporated on the surface of nanoparticles to increase the target specificity. Target ligands including antibodies, small peptides, and receptor binding compounds can be incorporated on the surface of nanoparticles [23,72]. Increased target specificity can improve bioactivities of encapsulated compounds, decrease their side-effects, and reduce administration dose and frequency [23,72].

3. Improvement of characteristics and bioactivities of phytochemicals by nanotechnology

Many natural compounds, especially phytochemicals, may have preventive and therapeutic potential for diseases. However, most of these compounds have low levels of solubility, stability, bioavailability, and target specificity in the body, which makes it unrealistic for these compounds to be present at their effective levels in the target tissues. This is particularly true for (−)-epigallocatechingallate (EGCG) found in green tea, resveratrol found in grapes, curcumin found in turmeric, and quercetin found in red onions, that are valuable for prevention and treatment of many diseases. Therefore, this represents an excellent opportunity for introducing nanoparticle technology to help resolve this issue.

3.1. (−**)-Epigallocatechingallate (EGCG)**

Green tea (*Camellia sinensis*) is a popular beverage. Green tea is prepared by drying the leaves under hot steam and air to inactivate the polyphenol oxidases, which prevents fermentation and gives green tea its distinctive color compared to black and oolong tea[73,74]. Green teacatechins constitute about 8–15% of total dry tea weight [74]. The most abundant, and also most bioactive, catechin is EGCG, which accounts for25–55% of total catechins. One cup of green tea made using a 2.5 g tea bag contains about 100 mg of EGCG [75]. Consumption of EGCG has been reported to have several health benefits including antioxidant, anti-inflammatory, antitumorigenic, and antiangiogenic properties [76].

EGCG is stable in an acidic solution, but is rapidly degraded in body fluids (at pH 7.4) [77]. Depending on the type of nanoparticles, incorporated EGCG can be partially or completely sequestered in the nanoparticles, resulting in high stability. We loaded EGCG into liposomes and chitosan-coated liposomes, with size less than 100 nm in diameter[78]. Nanoliposomes dramatically enhanced the EGCG stability in both $1 \times$ phosphate buffered saline (PBS) and Eagle's minimum essential (EME) cell culture medium [78]. At 4°C, 0.5 mM of free EGCG in EME medium was completely degraded after eight days. However, EGCG loaded into liposomes and chitosan-coated liposomes was degraded 62% and 38%, respectively, at the same conditions and initial concentrations[78]. After 1-hour incubation at 37°C, the EGCG degradation rates of 0.5 mM of free EGCG, EGCG loaded liposomes and chitosan-coated EGCG loaded liposomes in EME medium were 100%, 46% and 32%, respectively[78]. Barras A et al. demonstrated that free EGCG and EGCG loaded SLNs in water exhibited 100% degradation within four hours and over four weeks, respectively [79]. In addition, free and nano-EGCG exhibit burst and sustained release properties, respectively [80,81]. Sustained release creates a steady EGCG release pattern resulting in prolonged EGCG availability after administration. The advantages of sustained release include reduction in administration frequency, doses and side effects, and improvement of compliance[78,82].

EGCG is not readily absorbed in humans and research animals [83–85]. Scientists have conducted pharmacokinetic and bioavailability studies of green tea catechins in rats [86]. The blood peak concentrations of green tea catechins appear at 2 to 4 hours after oral

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administration. The absolute bioavailability of EGCG after intra gastric administration of decaffeinated green tea is about 0.1% in rats [86]. Consistent with this result, the bioavailability of EGCG is 0.14% in men and women after drinking tea containing 400 mg catechins throughout the day [84]. The peak plasma EGCG concentration is around $0.15 \mu M$ after drinking 2 cups of green tea [85]. A majority of published cell culture studies have used EGCG at physiologically irrelevant concentrations in the range of 10 to 200 µM[87,88]. Since EGCG at lower, and physiological relevant (achievable by oral intake) concentrations has little or very limited effect, it is important to increase EGCG bioavailability, and the nanotechnology appears to be an appropriate approach to meet this need. In fact, studies along this line have shown that nanoencapsulation significantly increases EGCG stability and improves its sustained release, which may partially contribute to the increased cellular uptake of EGCG [80,89]. Chitosan nanoparticles can significantly enhance EGCG bioavailability [90,91]. Since chitosan, a biocompatible polysaccharide, confers a positive charge to the surface of nanoparticles, it has been used as an absorption enhancer [92,93]. Hu et al. encapsulated EGCG into food grade peptide/chitosan nanoparticles, and they found the apparent permeation rate across the Caco-2 monolayers was increased more than two-fold by nanoencapsulation[90]. Dube A et al. compared the intestinal absorption of free EGCG and nano-EGCG (chitosan nanoparticles) using excised mouse jejunum[91]. They added 50 µM of freeornano-EGCG in the mucosal chambers and collected the transported EGCG in the serosal chambers over a three-hour period, and found that nano-EGCG had about two-fold higher accumulative transported amounts than free EGCG[91]. Enhanced EGCG stability and trans cellular transport process by nanoparticles, but not paracellulartransport process, may partially contribute to the enhanced intestinal absorption[91]. After reaching the apical membrane of intestinal epithelial cells, most nanoparticles cross enterocytes via transcellular transport[91]. Nanoparticles are internalized into enterocytes and then transported across enterocytes[91]. Finally, nanoparticles are moved out of the basolateral membrane through exocytosis to enter the bloodstream or lymphatic vessels. Particles less than 500 nm in diameter are internalized through both clathrin- and caveolae-mediated endocytosis [94]. The process of endocytosis can be enhanced by modifying nanoparticles, e.g., by adding PEG or positive charges on the surface of nanoparticles[95]. Moreover, coating nanoparticles with cationic chitosan or ions protects them from endolysosomal degradation in enterocytes[96]. Therefore, chitosan nanoparticles are capable of enhancing EGCG oral bioavailability. Dube A et al. further measured the plasma concentrations of EGCG in mice after oral administration of either free EGCG or EGCG encapsulated chitosan nanoparticles[89]. Compared to free EGCG, EGCG loaded chitosan nanoparticles increased plasma EGCG concentrations by a factor of 1.5 [89]. Consistently, EGCG nanolipidic particles increased its oral bioavailability by more than two-fold compared to free EGCG in rats [97] (Table 1).

Another advantage is that nanoencapsulation can increase EGCG bioactivities, in particular its antioxidant, antitumorigenic, and antiangiogenic properties. Hu B et al. reported that treating HepG2 cells with 26 ∼ 37 µM of nano-EGCG, resulted in a higher cellular antioxidant activity compared to free EGCG at the same concentrations [80]. Siddiqui IA et al. demonstrated that, compared to free EGCG, nano-EGCG (PLGA nanoparticles) exhibited more than ten-fold dose advantage in inducing apoptosis, decreasing viability, and inhibiting colony formation of prostate cancer cells[98]. The IC_{50} values of free and nano-EGCG are 43.6 and 3.74 µM, respectively. They also gave tumor xenograft mice either 100 µg of nano-EGCG or 1 mg of free EGCG three times per week through intra peritoneal injection and found that nano-EGCG, even at a dose ten-fold lower, significantly reduced prostate tumor size [98]. Moreover, when incorporating target ligands on the surface of nanoparticles, to specifically target an antigen on prostate cancer cells, the EGCG nanoparticles can reduce the viability of prostate cancer cells to a significantly larger degree compared to EGCG nanoparticles without target ligands [81] (Table 1).

3.2. Quercetin

Quercetin (3,3',4',5'-7-pentahydroxy flavone) is a plant-derived flavonol and it is abundant in caper, berries, buckwheat, black and green tea leaves, apple, onion, broccoli and other leafy green vegetables[99]. Studies have demonstrated that quercetin has antioxidant, antiviral, anti-inflammatory and antitumorigenic properties[100–102]. Quercetin has low aqueous solubility and bioavailability and is quickly metabolized in the body, which may reduce its efficacy as an application in preventing or treating diseases [103]. Quercetin is a hydrophobic compound and its solubility in an aqueous solution varies from 0.00215 g/L at 25 °C to 0.665 g/L at 140 °C [104]. Quercetin has a high solubility in organic solvents such as ethanol, dimethylsulfoxide (DMSO), and dimethyl form amide (DMF). The solubility of quercetin is approximately 2 g/L in ethanol and 30 g/L in DMSO at 25 °C [104]. Using nanomicelles, the aqueous solubility of quercetin can increase by 110fold [105]. We have successfully synthesized quercetin encapsulated NLCs and found that the aqueous solubility of quercetin increased about 1000-fold (Figure 2). Additionally, a sustained release pattern was observed in quercetin loaded nanoparticles including poly(lactic acid) or PLGA nanoparticles [60,106,107], SLNs and NLCs [108,109].

Consumption of quercetin rich food or supplements can increase the plasma quercetin concentrations up to low μ M levels [33,110]. Quercetin in human bodies is quickly metabolized by enzymes in the liver as well as the other organs or tissues. Hong Y et al. found that at least 21 metabolites of quercetin are detected in human urine after ingestion of quercetin glycosides from onions[111]. Blood quercetin includes not only free quercetin (low concentrations), but also its conjugated forms(high concentrations) including glucuronide or sulfate forms of quercetin, and O-methylated forms, among others[112]. Encapsulating quercetin into biodegradable and biocompatible nanoparticles may help delay its metabolism and maintain free quercetin levels in blood and other tissues for a prolonged period. Li W et al. developed a SNEDDS of persimmon leaf extract rich in flavonoids, especially quercetin. Persimmon leaf extract tablets and persimmon leaf extract-loaded SNEDDS at 5.3 mg/kg body weight were given to two groups of beagle dogs through oral administration. Compared to the tablets, persimmon leaf extract-loaded SNEDDS increased the bioavailability (area under curve) of plasma quercetin by 1.5 fold [113]. Li H et al. made quercetin encapsulated SLNs, which exhibited sustained release. They gave male Wistar rats free or nanoencapsulated quercetin solution at a dose of 50 mg/ kg body weight and collected blood at different time points for up to 48 hours. They found that the quercetin encapsulated SLNs increased oral bioavailability by a factor of five compared to free quercetin. They also demonstrated that the absorption sites were primarily ileum and colon, but not stomach and duodenum [109]. Sprague-Dawley rats are widely used animal model for pharmacokinetic study. After intravenous (10 mg/ kg body weight) or oral (50 mg/ kg body weight) administration of free and nano-quercetin (quercetin-NLCs) to male Sprague-Dawley rats, we found that compared to free quercetin solution, quercetin-NLCs increased bioavailability (area under the curve) of plasma total quercetin concentrations by 2.8 and 2.0 fold after intravenous and oral administration, respectively (unpublished data). We also measured blood concentrations of unmodified (free) quercetin after intravenous administration. Quercetin-NLCs significantly increased the blood unmodified (free)quercetin concentrations and prolonged its circulation time (Figure 3).

The increased solubility and bioavailability and improved sustained release by nanoencapsulation may elevate the bioactivities of quercetin. Some experimental results suggest that quercetin may have an anti-cancer potential[114,115]. However, there is no consistent clinical evidence to substantiate this proposed benefit[116,117]. One of the reasons may be that the tissue quercetin concentrations after consumption of quercetin may not be adequate enough to allow it to exert its effect to the fullest. Utilizing nanoparticles to increase the bioavailability and biopotency may open an area of research toward this

direction. Tan BJ et al. synthesized quercetin loaded nanomicelles, and found that they were stable in gastric and intestinal fluids, and had no toxic effects on Caco-2 cells. Both free quercetin and quercetin loaded nanomicelles decreased the viability of A549 lung cancer cells *in vitro*, but 100 µM of free and nano-quercetin decreased the cell viability to 60% and 100%, respectively. Oral administration of 30 mg/kg of free and nano-quercetin three times per week at week 3, 4, 5 post tumor inoculation to A549 lung tumor xenograft mice decreased the tumor size 20% and 40%, respectively, but they did not change animal body weight [105]. Wang G et al. demonstrated that quercetin nanoliposomes decreased the viability of C6 glioma cells, and induced necrotic death of those cells [118]. Dhawan S et al. found that the intravenous administration of quercetin improved memory retention and increased brain antioxidant capacity in rats with aluminium-induced dementia, and quercetin encapsulated SLNs significantly improved those beneficial effects by more than two-fold [119]. Using quercetin liposomes to prevent Arsenic-induced acute liver toxicity, Ghosh D et al. injected either free or nano-quercetin at 2.71 mg/ kg body weight to rats via tail veinsone hour after oral administration of Arsenic salt. Compared to control, free and nanoquercetin decreased liver Arsenic concentrations by 20% and 40%, increased liver quercetin concentrations by 20 and 40 fold, enhanced antioxidant capacity by 1.2 and 1.6 fold, and decreased blood aspartate aminotransferase concentrations by10% and 33%, respectively [120]. Nano-quercetin(PLGA nanoparticles) increased anti-oxidant capacity and decreased inflammatory responses in the stomach, which might contribute to the high preventive efficacy of gastric ulcer by a factor of 20 in rats compared to free quercetin[61]. Quercetin nanoparticles have also been used for topical delivery and treating skin problems. Chen-Yu G et al. demonstrated that quercetin encapsulated NLCs resulted in a two-fold increase in quercetin concentrations in epidermis and dermis of Kunming mice [121]; they also showed that the quercetin encapsulated NLCs decreased inflammatory response in inflamed skin. Consistent with this, quercetin encapsulated NLCs are shown to improve quercetin skin concentrations by more than two-fold in an *in vitro* permeation study using human skin [108](Table 2). Together, these studies suggest that nanoencapsulation may increase quercetin aqueous solubility, improve its sustained release, prevent it from modification and metabolism, enhance its bioavailability and bioactivity, and lower its toxicity.

3.3. Resveratrol

Resveratrol (trans-3, 4',-5-trihydroxystilebene) is a type of natural polyphenol abundant in the skin of red grapes and other fruits such as berries. Resveratrol possess two structural isomers: *cis*- and *trans*-resveratrol. Under UV exposure, *tran*-resveratrol is converted into *cis*-resveratrol [122]. Nanoencapsulation protects *trans*-resveratrol against light-exposure degradation and hence increases its stability [123,124]. Detoni CB et al. compared photostability of different *trans*-resveratrol incorporated nanoparticles including liposomes, SLNs, nanospheres, and polymeric lipid-core nanocapsules[124]. All tested nanoparticles increased photostability of *trans*-resveratrol and liposomes maintained *trans*-resveratrol concentrations for the longest time [124]. After applying the free or nanoencapsulated*trans*resveratrol to the porcine skins and then exposed to UVA radiation, they found that nanoencapsulation resulted in a larger increase in the *trans*-resveratrol concentrations in epidermis and dermis compared to free *trans*-resveratrol solution [124]. Consistent with these findings, PLGA nanoparticles significantly increased the photostability of *trans*resveratrol when exposed to UVA for two hours [123]. Since *trans*-resveratrol is more stable than *cis*-resveratrol and thus has been extensively used in the studies, it is the subject of this review and is referred as resveratrol in the rest of article[122].

The aqueous solubility of resveratrol is extremely low. Sigma-Aldrich company reports that resveratrol solubility in water is about 3 mg/100 mL. Resveratrol is much more soluble in organic solvents such as ethanol and dimethyl sulfoxide (DMSO), with solubility at 50 mg/

mL and at least 16 mg/mL, respectively. Resveratrol can be encapsulated or incorporated in the lipid compartment of nanoparticles, especially SLNs and NLCs, resulting in enhanced aqueous solubility. We have successfully encapsulated resveratrol into the hydrophobic core of lipid nanocarriers. The resveratrol aqueous solubility was improved more than 100 times (unpublished data). Other nanoparticles including SLNs, NLCs and liposomes can also increase resveratrol aqueous solubility [124–126](Table 3).

Like EGCG and quercetin, circulating resveratrol is rapidly metabolized, and consequently, blood concentrations of free resveratrol are at low mM levels [34]. Different from free resveratrol, nanoencapsulated resveratrol exhibited a sustained release pattern [126–128]. Less than 10% of resveratrol was released from NLCs or SLNs after incubating them at 25°C and 37°C for four hours[126]. Frozza R et al. encapsulated resveratrol into lipid-core nanoparticles and gave male Wistar rats 5 mg/kg body weight of either free or nanoresveratrol daily for 14 days via oral administration [129]. Compared to free resveratrol, nano-resveratrol significantly increased the rat tissue (kidney, brain and liver, but blood was not measured) resveratrol concentrations by more than twofold [129]. In addition, resveratrol loaded lipid-core nanoparticles had higher gastrointestinal safety than free resveratrol [129](Table 3).

Resveratrol has antioxidant, anti-inflammatory, anticarcinogenic properties[34,130,131]. Compared to free resveratrol and void nanoparticles, nano-resveratrol can decrease reactive oxygen species production and increase antioxidant capacity in cell culture and research animal model [45,128,132]. Lee CW et al. orally administered control (no resveratrol), free resveratrol and nano-resveratrol to rats with CCl4-induced hepatotoxicity. They found that nano-resveratrol doubled the beneficial effect of free resveratrol on reducing hepatocyte death, decreasing oxidative stress, lowering inflammatory cytokine production [132]. Shao J et al. demonstrated that nano-resveratrol doubled the inhibitory effect of free resveratrol on the viability of rat C6-glioma cells *in vitro*[127]. Guo et al. administered saline, free resveratrol and resveratrol loaded bovine serum albumin nanoparticles to the implanted ovarian tumor-bearing mice via intra peritoneal injection once per week for four weeks[133]. The free and nano-resveratrol concentrations were 50, 100, and 200 mg/kg body weight. Compared to free resveratrol, resveratrol loaded bovine serum albumin nanoparticles increased resveratrol concentrations in ovary by 1.8 fold. Both free and nanoresveratrol decreased ovarian tumor weight in a dose dependent manner. The inhibition rates of tumor growth by free and nano-resveratrol at 200 mg/kg body weight were 46% and 62%, respectively. None of the treatments changed animal body weight [133]. In addition, nanoresveratrol has a potential in improving resveratrol's preventive effect on Alzheimer's disease [128,129](Table 3).

3.4. Curcumin

Curcumin is a hydrophobic curcuminoid present in the Indian spice turmeric extracted from the herb *Curcuma long a[134]*. Curcumin has been shown to have several beneficial effects including anti-inflammatory, antiangigenic, and antitumorigenic properties and may potentially help in preventing or even treating some chronic diseases, such as cancer, diabetes, and cardiovascular disease[134]. Despites these advantages, curcumin has poor aqueous solubility, low bioavailability, and is quickly metabolized by hepatic enzymes in humans and research animals [135]. Many biocompatible and biodegradable nanoparticles have been developed to overcome these limitations.

SLNs, nanoemulsions, and PLGA nanoparticles have successfully improved the aqueous solubility and chemical stability of curcumin[136–140]. Nano-curcumin exhibits a sustained release pattern [136,137,141,142]. The bioavailability of curcumin is low. After oral administration of 1 g/kg body weight and 2 g/kg body weight of curcumin to rats, the peak

blood concentrations detected at hour 0.8 were1.4 µM and 3.7 µM, respectively [35,143]. In human, consumption of less than 4 g of curcumin resulted in either undetectable or extremely low (less than 1.0 μ M) serum curcumin concentrations[144]. If humans consume 4–12 g of curcumin, the peak blood curcumin concentrations are increased, but less than 4.0 µM [35,144–146]. After encapsulating curcumin into SLNs, nanoemulsions and PLGA nanoparticles, the oral bioavailability can be enhanced more than two-fold [62,137,141,142,147,148]. After intravenous administration of nano-curcumin, blood curcumin concentrations and its circulation time are also significantly increased [139,149– 151](Table 4).

Nanoencapsulation increases curcumin bioactivities[138–140,150,152–154]. Researchers have investigated the anticancer activities of nanoencapsulated curcumin in a variety of cancer cells including HCT116, A2780CP, MDA-MB-231, KBM-5, PANC-1, MIA PaCa-2, K562, MCF 7, A549 cells[138–140,150]. Nano-curcumin significantly enhanced the inhibitory effect of curcumin on cancer cell viability, which is accompanied by an increase in curcumin uptake by cancer cells[136,138–140]. Importantly, the void nanoparticles including void nanoemulsions, PLGA nanoparticles and human serum albumin nanoparticles did not change viability of those cancer cells, indicating the safety of those void nanoparticles [136,139,140]. Punfa W et al. conjugated anti-P-glycoprotein (P-gp) to the surface of PLGA nanoparticles [155]. They found that targeting antibodies significantly increased the binding and targeting specificity of nanoparticles to cancer cells and further enhanced the uptake of nanoparticles by cancer cells [155]. Kim TH et al. inoculated HCT116 or MIA PaCa2 cells to Balb/c nu/nu male mice[138]. After inoculation for five days, they intravenously administered 10 mg/kg body weight of free or nanoencapsulated curcumin every other day for ten days[138]. Compared to control (saline), free and nanoencapsulated curcumin decreased the tumor volume by 20% and 45%, respectively. Importantly, they did not change body weight [138]. Curcumin-PLGA nanoparticles can decrease reactive oxygen species accumulation in neurons, improve neurons against oxidative damage *in vitro*, and increase curcumin accumulation in rat brain, which indicates its potential application in preventing Alzheimer's disease and other neuron degenerative diseases [149,152].

Wang W et al. gave 400 mg/kg body weight of free curcumin or nano-curcumin (curcumin-SLNs) to Balb/c mice via intra peritoneal injection [153]. They found that nano-curcumin dramatically increased curcumin concentrations in lung and doubled inhibitory effect of free curcumin on inflammatory responses in lung, which implies its application in asthma therapy [153]. In addition, curcumin liposomes can decrease inflammatory response and reverse insulin resistance in an animal model [154]. After intra peritoneal injection of curcumin-liposomes to insulin resistance (ob/ob) mice, blood fasting glucose and insulin levels and homeostasis model assessment-estimated insulin resistance index (HOMA-IR) were significantly decreased, inflammatory responses were reduced, and peripheral insulin sensitivity was improved [154](Table 4). Curcumin nanoparticle research is growing. When we used key words"curcumin nanoparticle" to search on Pub Med, we found 10, 19, 48, 79, 109 articles in 2008, 2009, 2010, 2011 and 2012, respectively.

4. Challenges and limitations

Nanomedicine is a very promising field especially when applied for disease prevention and/ or treatment using phytochemicals and other dietary supplements. However, this is a very new field and is still in its infancy, thus presenting many technical and translational challenges and limitations.

The major challenge is potential toxicity of nanoparticles. Many components in nanoparticles, such as nucleic acids, antibody fragments, peptides and proteins, can function as antigens, resulting in increased immunotoxicity [160]. Moreover, if the encapsulation efficacy and loading capacity of nanoparticles are low, people would be consuming or receiving large amounts of nanocarriers containing surfactants and co-surfactants or emulsifiers, which may cause adverse effects. While clinical trials can be used to assess the short-term toxicity of nanoparticles, the possibility of long-term toxicity due to chronic exposure and accumulation needs to be carefully addressed. Currently, there are no good *in vivo* models, guidelines, and standardized safety test variables established to determine toxicity and adverse effects of nanoparticles. Incorporation of target ligands on the surface of nanoparticles can increase targeted delivery of encapsulated phytochemicals to targeted abnormal cells, which can be used to decrease toxicity and adverse effects [161]. However, many physical and biological barriers exist between nanoparticles and abnormal cells such as cancer cells [162]. These barriers include the blood vessel wall, blood-brain barrier, extra cellular matrix, interstitial fluid pressure gradients, among many others [163]. It is necessary for nanoparticles to overcome all these barriers before acting on the targeted cells.

Another relevant issue worth noting is the administration route of nanoparticles. In general, oral administration is the most practical and acceptable route for long-term administration of phytochemicals and dietary supplements. However, little is known about the absorption and metabolism of nanoparticles in the gastrointestinal tract, thus limited data exists about bioavailability of nanocarriers and their tissue specific pharmacokinetics [164]. Biocompatible and biodegradable nanocarriers, such as lipid nanoparticles, can be digested or degraded in the gastrointestinal tract. Even though the phytochemical encapsulated nanoparticles can be absorbed, the structure, characteristics and pharmacokinetics of nanoparticles may be changed after the digestion or degradation of nanocarriers.

Finally, the cost of applying this nanotechnology is another major limitation. Indeed, the synthesis of nanoparticles, especially multifunctional nanoparticles (e.g. incorporation of both preventive and therapeutic phytochemicals in nanoparticles), is an expensive and complicated process, which requires special ingredients, certain instruments and optimal conditions[160, 165]. Lowering the cost/benefit ratio is critical in the application of nanotechnology in nutrition research.

5. Conclusions

Many biocompatible and biodegradable nanoparticles are currently available for encapsulating bioactive compounds including phytochemicals. Each nanoparticle has its own advantages, disadvantages, and characteristics. We have demonstrated that nanoparticles can overcome some limitations in using phytochemicals for health promotion and prevention, and enhance their bioactivities. Given the wide use of dietary supplements (most of which are phytochemicals), and potential toxicity and safety concerns with some of these supplements, nanotechnology is a promising tool for limiting the dosage while increasing bioavailability and bioactivities. Even though nanotechnology offers promising approaches in nutraceutical applications, additional innovative research is needed to address the cost-effectiveness and long-term safety of those nanoparticles.

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Schematic structure of nanoparticles.

Figure 2.

Visual observation of void NLCs, and 20 mg of quercetin encapsulated in NLCs and free quercetin dissolved in 2 mL of phosphate buffered saline(PBS).

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Figure 3.

High performance liquid chromatography peaks of unmodifiedquercetin in blood isolated from male SD rats after single intravenous administration of 10 mg/kg body weight of free or nanoencapsulated quercetin in NLCs for 0.25 hours, 1 hour and 4 hours.

Table 1

The characteristics, functions and application of EGCG nanoparticles The characteristics, functions and application of EGCG nanoparticles

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EE: Encapsulation efficiency; N/A: not applicable; NLC: nanostructured lipid carrier; NP: Nanoparticle; PEG: polyethyleneglycol; PLGA: poly (D,L-lactide-co-glycolide); PSMA: Prostate-specific

membrane antigen; SZ: Size; ZP: Zeta-potential; ↑, increase; ↓, decrease.

EE: encapsulation efficiency; LC: loading capacity; N/A: not applicable; NP: nanoparticle; NLC: nanostructured lipid carriers; PLA: poly (D,L-lactide); PLGA: poly (D,L-lactide-co-glycolide); SLN: Solid
lipid nanoparticle; EE: encapsulation efficiency; LC: loading capacity; N/A: not applicable; NP: nanoparticle; NLC: nanoparticle; NLC: nanoparticle; NLA: poly (D,L-lactide-co-glycolide); SLN: Solide, Publicaling capacity; N/A: not applicable; lipid nanoparticle; SNEDDS: self-nanoemulsifying drug delivery system; SZ: size; ZP: Zeta-potential; ↑, increase; ↓, decrease.

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The characteristics, functions and application of resveratrol nanoparticles The characteristics, functions and application of resveratrol nanoparticles

EE: encapsulation efficiency; LC: loading capacity; mPEG-PCL: methoxy polyethylene glycol)- poly(caprolactone); NP: nanoparticle; NLC: nanostructured lipid carriers; PCL: poly-caprolactone; PEG:
polyethyleneglycol; PI: pol EE: encapsulation efficiency; LC: loading capacity; mPEG-PCL: methoxy polyethylene glycol)- poly(caprolactone); NP: nanoparticle; NLC: nanostructured lipid carriers; PCL: poly-caprolactone; PEG: polyethyleneglycol; PI: polydispersity index; SLN: solid lipid nanoparticle; SZ: size; ZP: Z-potential; ↑, increase; ↓, decrease.

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Table 4

The characteristics, functions and application of curcumin nanoparticles The characteristics, functions and application of curcumin nanoparticles

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EE: encapsulation efficiency; LC: loading capacity; mPEG-PA: methoxy poly(ethylene glycol)-Palmitate; NP: nanoparticle; PEG: polyethyleneglycol; PLGA: poly (D,L-lactide-co-glycolide); PVA:
Poly(vinyl alcohol); SLN: Solid l EE: encapsulation efficiency; LC: loading capacity; mPEG-PA: methoxy poly(ethylene glycol)-Palmitate; NP: nanoparticle; PEG: polyethyleneglycol; PLGA: poly (D,L-lactide-co-glycolide); PVA: Poly(vinyl alcohol); SLN: Solid lipid nanoparticle; SZ: size; TPGS: D-α-tocopheryl polyethylene glycol 1000 succinate