



## Report

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### Author for correspondence:

Balaji Sitharaman

e-mail: [balaji.sitharaman@stonybrook.edu](mailto:balaji.sitharaman@stonybrook.edu)

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# Quantification of single-cell nanoparticle concentrations and the distribution of these concentrations in cell population

Jason T. Rashkow<sup>1</sup>, Sunny C. Patel<sup>1</sup>, Ryan Tappero<sup>2</sup> and Balaji Sitharaman<sup>1</sup>

<sup>1</sup>Department of Biomedical Engineering, Stony Brook University, Stony Brook, NY 11794-5281, USA

<sup>2</sup>Department of Photon Sciences, National Synchrotron Light Source, Brookhaven National Lab, Upton, NY 11973-5000, USA

Quantification of nanoparticle uptake into cells is necessary for numerous applications in cellular imaging and therapy. Herein, synchrotron X-ray fluorescence (SXRF) microscopy, a promising tool to quantify elements in plant and animal cells, was employed to quantify and characterize the distribution of titanium dioxide (TiO<sub>2</sub>) nanosphere uptake in a population of single cells. These results were compared with average nanoparticle concentrations per cell obtained by widely used inductively coupled plasma mass spectrometry (ICP-MS). The results show that nanoparticle concentrations per cell quantified by SXRF were of one to two orders of magnitude greater compared with ICP-MS. The SXRF results also indicate a Gaussian distribution of the nanoparticle concentration per cell. The results suggest that issues relevant to the field of single-cell analysis, the limitation of methods to determine physical parameters from large population averages leading to potentially misleading information and the lack of any information about the cellular heterogeneity are equally relevant for quantification of nanoparticles in cell populations.

## 1. Introduction

Recent progress in nanoparticle technology allows the development of approaches to probe and/or manipulate the functions of cells in all their complexity down to the molecular level [1]. These approaches are essential for practical applications to improve health care, through advancements in molecular imaging and personalized molecular medicine [1]. At an early stage in the nanoparticle development, *in vitro* toxicity and efficacy studies of nanoparticles are performed on model cell lines. These studies provide average effects of the nanoparticle on these cells, and typically allow identification of therapeutic dosages. At those therapeutic dosages, the nanoparticles, if used as a probe or sensor, are assumed to be bioinert; provide information about the changes in the cellular anatomy or physiology without disturbing the cellular homeostasis. A nanoparticle used to manipulate the function of cells (e.g. as a therapeutic drug) is expected to be present in sufficient concentrations to disturb the cellular homeostasis. Additionally, in both the above scenarios, the nanoparticle should neither be present at low inefficacious doses nor at high toxic concentrations. Thus, direct quantification of nanoparticles per cell for the above studies is necessary and is routinely performed by random sampling of known number of cells and calculating the average nanoparticle concentration per cell [2]. In general, studies that use these average methods assume their accuracy to be acceptable. Additionally, these average methods do not provide any information of the distributions of nanoparticle in single-cell populations.

Direct quantification of nanoparticles in a single cell and the heterogeneity of the concentration distribution need to be investigated to help in understanding the variability in cell-to-cell nanoparticle concentration, and how these variations affect nanoparticle–cell interactions. These insights in turn could provide guiding principles to improve and allow nanoparticle-based personalized diagnostics and therapeutics. Microscopy and/or spectroscopy techniques and stereology have

been applied to track and follow the interactions of nanoparticles within single cells [3,4]. Recently, X-ray fluorescence (XRF) microscopy at synchrotron radiation light sources has enabled the characterization of nanoparticle–cell interactions at the sub-cellular level [5,6]. Synchrotron X-ray fluorescence (SXRF) microscopy is a versatile analytical tool widely used in biomedical research and employs hard X-rays to excite and detect characteristic  $K\alpha$  X-ray fluorescence of important elements in medicine [5,6]. The potential of SXRF to quantify a variety of elements in individual algae, fungal and *eukaryote* cells has also been reported [7–9]. To the best of our knowledge, a comparison of capabilities of SXRF to quantify nanoparticles in individual cells vis-à-vis a current gold standard and the examination of the heterogeneity of the nanoparticle's distribution in single-cell populations has not been investigated. Herein, using titanium dioxide ( $\text{TiO}_2$ ) nanospheres as model nanoparticles we have harnessed the capabilities of SXRF to quantify and analyse the distribution of nanoparticle concentration in a large population of individual cells. We compare these results with those obtained from inductively coupled plasma mass spectrometry (ICP-MS)—a current gold standard to determine average concentrations per cell of these nanoparticles.

## 2. Results and discussion

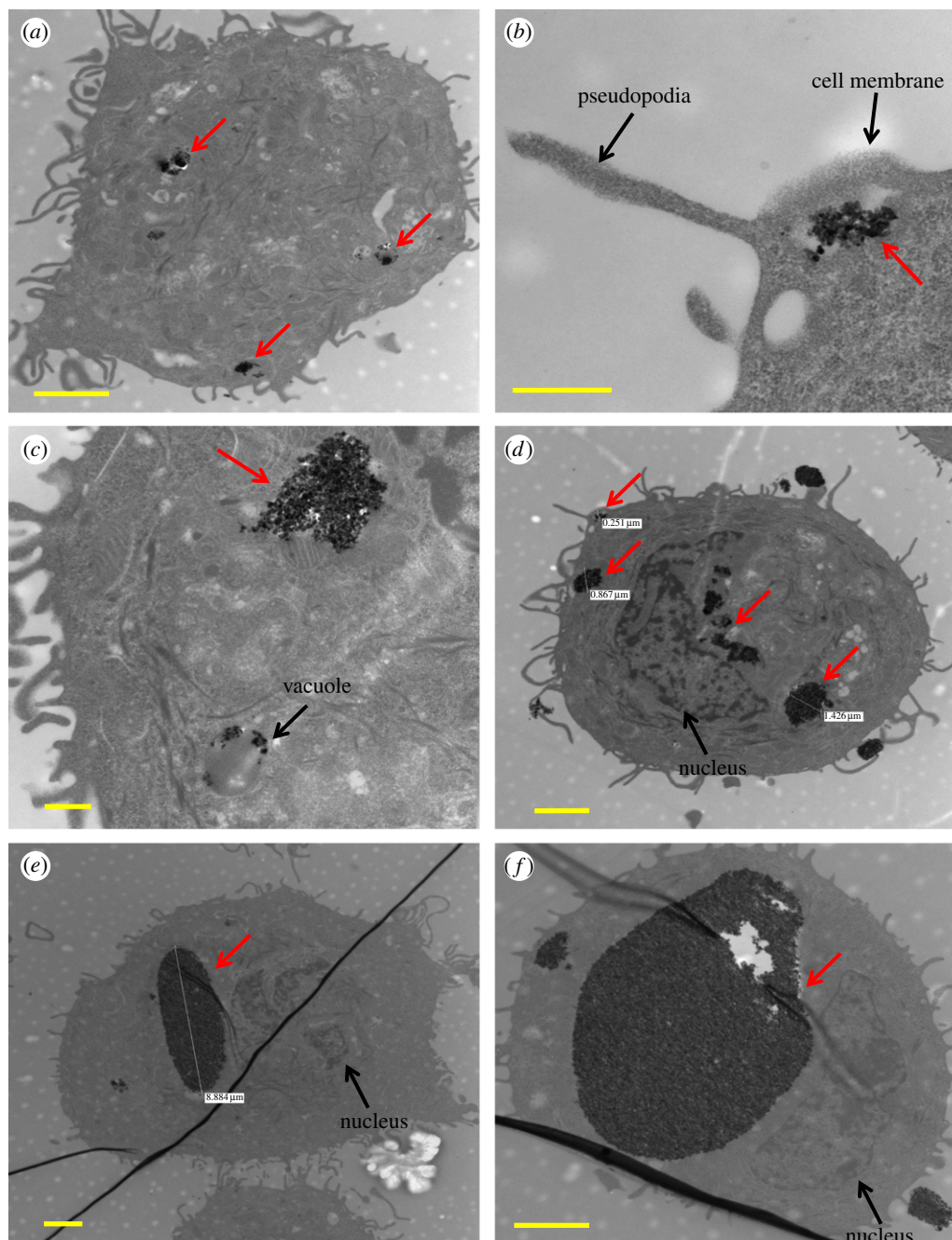
The SXRF microscope employed in the current study generates simultaneous elemental maps at a spatial resolution of 7–14  $\mu\text{m}$  [10].  $\text{TiO}_2$  nanoparticles were chosen for this study because they are one of the widely studied nanoparticles for various biomedical applications, such as gene therapy, biosensing, cancer therapeutics and bioimaging [11]. Commercially, they are manufactured in large amounts, and thus comprehensive studies on the biological and environmental impact of these nanomaterials have been reported [11]. The titanium (Ti) atom present in  $\text{TiO}_2$  exhibits excellent X-ray fluorescence, is not a constitutive element of cells and has been observed by SXRF microscopy after uptake by various cancer cell lines to determine the stability and targeting of  $\text{TiO}_2$  nanoparticles conjugated with DNA [6,9,12]. The electronic supplementary material, figure S1a–d, shows the characterization of the anatase  $\text{TiO}_2$  (99.5%) nanospheres and cells. The electronic supplementary material, S1a shows a representative TEM image of  $\text{TiO}_2$  nanospheres, which are smooth, spherical particles with diameters of  $\approx 20$ – $30$  nm. Raman spectra of the  $\text{TiO}_2$  nanospheres (electronic supplementary material, figure S1b) showed peaks for anatase  $\text{TiO}_2$  at 136, 166 and  $628\text{ cm}^{-1}$  corresponding to the  $E_g$  Raman mode,  $382\text{ cm}^{-1}$  signifying the  $B_{1g}$  mode, and  $514\text{ cm}^{-1}$ , which corresponds to  $A_{1g}$  and  $B_{1g}$  Raman modes. A small peak at  $454\text{ cm}^{-1}$ , which is the  $E_g$  Raman mode peak for rutile  $\text{TiO}_2$  was also observed [13]. Electronic supplementary material, S1c displays solubility of  $1\text{ mg ml}^{-1}$  concentration of  $\text{TiO}_2$  nanospheres in water over 24 h without aggregation. This formulation was used to treat the cells. The electronic supplementary material, S1d, shows a differential interference contrast (DIC) microscopy image of SK-BR-3 cells fixed onto Ultralene film for SXRF microscopy. The cells treated with  $\text{TiO}_2$  nanospheres were not functionalized or treated with additional transfecting agents.

The uptake of the  $\text{TiO}_2$  nanoparticles by SK-BR-3 cells was confirmed by TEM as displayed in figure 1.  $\text{TiO}_2$  nanoparticles (red arrows) were within the cytoplasm (figure 1a–f)

and vesicular structures (figure 1a,e,f), but not in the nucleus. Cell pseudopodia engulfing the nanoparticles were also observed (figure 1b,d). Analysis of TEM images of many individual cells including those in figure 1a–f indicated that most of the nanoparticles were internalized by cells with a small fraction present on the cell membranes. Qualitatively, cells either showed no uptake of nanoparticles, or small (figure 1a,b), medium (figure 1c,d) and large (figure 1e,f) nanoparticle concentrations and aggregate sizes.

SXRF was used to determine the concentration of nanoparticles in single cells. Initially, SXRF ‘flyscans’ (figure 2a), a rapid-scanning technique using a dwell time of 100 ms per pixel, were used to create an elemental map of an area of about  $1$ – $3\text{ mm}^2$ . The flyscans of approximately 30% cells (out of approx. 1700 cells) showed only Ca (i.e. no Ti uptake). Step scans, shown in figure 2b, were performed with a longer dwell time (5 s per pixel) on 112 individual cells (complete cell selection criteria can be found in experimental section 7 in the electronic supplementary information). Individual scans were processed for spectral summing (figure 2c) by creating an outline of the cell using interactive data language (IDL)-based beamline software which outputs a summed energy-dispersive spectra (figure 2d) for the highlighted area and contains the total counts for Ti  $K\alpha$  in the cell. Using the information of Ti counts per cell, a known and measured National Institute of Standards and Technology (NIST) standard 1833 for Ti, the concentration of  $\text{TiO}_2$  per cell, was quantified. The abundance of nanoparticles internalized per cell was heterogeneous with a range between 37 and  $84\text{ ng cell}^{-1}$  or  $773$ – $1751\text{ pmol per cell}$  (atomic weight of  $\text{Ti} = 47.9$ ). Stoichiometrically, one mole of Ti gives one mole of  $\text{TiO}_2$ . Thus, the concentration of  $\text{TiO}_2$  also ranges between  $773$  and  $1751\text{ pmol per cell}$  (figure 2e). The nanoparticle per cell concentration plot exhibits a Gaussian distribution.

The SXRF results were compared to ICP-MS; a standard method of quantifying average concentration per cell [14]. For this analysis, 30 000 cells ( $n = 5$ ) were treated in the same manner as described for the SXRF experiments. After 24 h, 15 000 cells were counted by flow cytometry. These cells were digested and prepared for ICP-MS (see experimental section 8 in the electronic supplementary material for details). For the five samples, the total Ti abundance in 15 000 cells, determined by ICP-MS, was between 16 and  $24\text{ }\mu\text{g}$  and Ti abundance per cell was calculated to be  $1.1$  to  $1.6\text{ ng cell}^{-1}$ . The average concentration of Ti and hence  $\text{TiO}_2$  in moles was  $23$ – $33\text{ pmol per cell}$  (table 1). This average value does not provide any additional information about the distribution of the nanoparticle per cell concentration in the 15 000 cells. The difference in the nanoparticle per cell concentrations obtained by SXRF and ICP-MS methods was significant and taken together indicate that  $\text{TiO}_2$  nanospheres were present in single cells at concentrations of one to two orders of magnitude greater than the average concentration per cell values. The lower average nanoparticle concentrations per cell by ICP-MS could be due to systematic errors induced during sample preparation [15], which involves multiple steps (see the experimental section in the electronic supplementary material for these steps). During the sample preparation, cells are digested, involving transfer of the solution a number of times, as well as filtration and dilution to be within the linear range of detection for the instrument. Even though these and other sample preparation steps are undertaken with utmost caution, each of these steps is

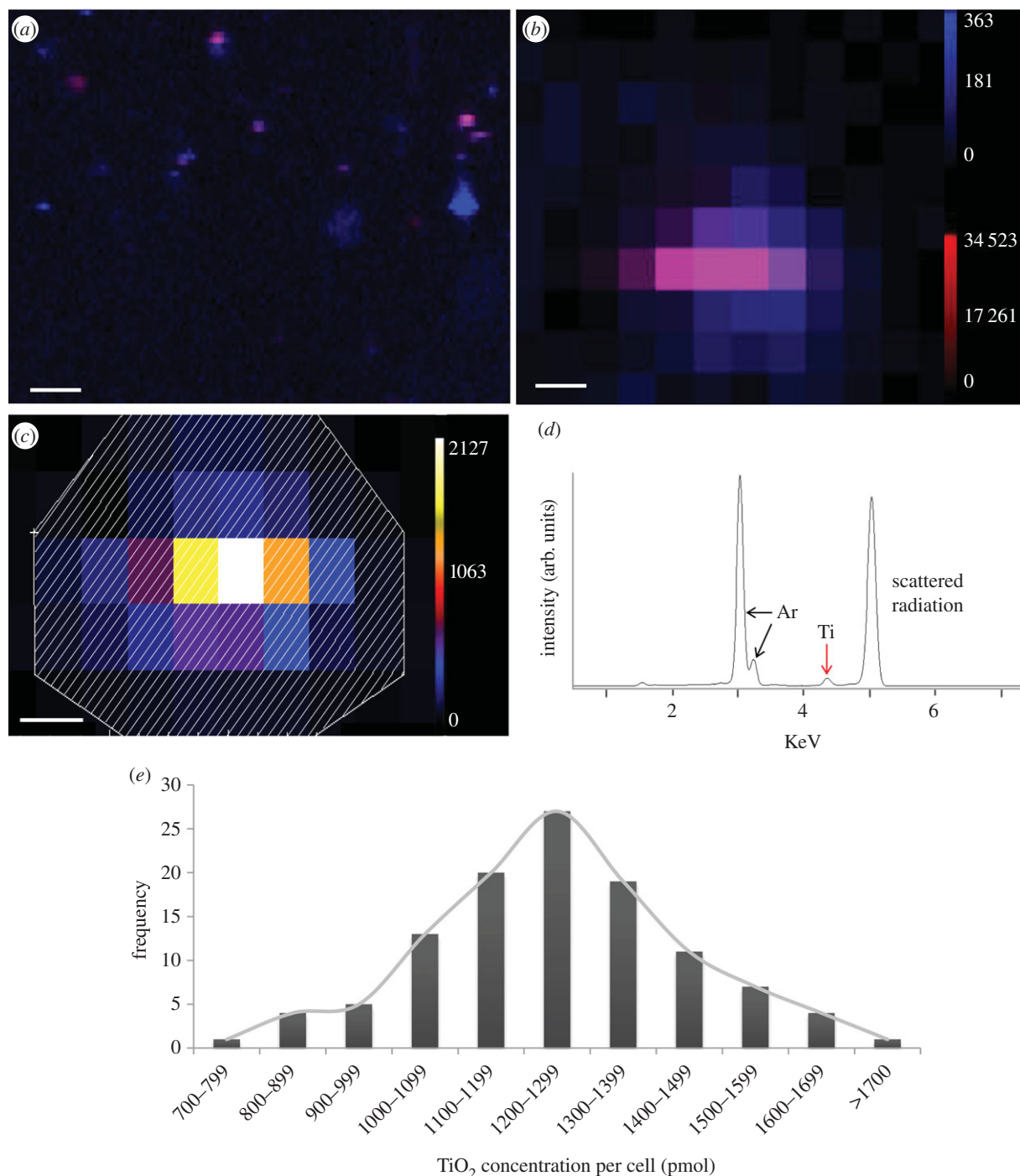


**Figure 1.**  $\text{TiO}_2$  nanoparticles (red arrows) within the cytoplasm (*a–f*). Nanoparticle aggregates within vesicular structures (*a,e,f*). Cell pseudopodia engulfing the nanoparticles (*b,d*). No nanoparticles were seen in the nucleus. (*a,d,e,f*) Scale bars,  $2\ \mu\text{m}$ ; (*b,c*) scale bars,  $500\ \text{nm}$ .

dependent on the previous one and has potential to introduce compounding error propagation owing to sample loss. Although systematic errors do not affect the precision of the results, they can significantly affect accuracy [16]. Another reason for observed discrepancy in nanoparticle concentration per cell by ICP-MS measurements could be its dependence on the nanoparticle's cell labelling efficiency. Increase in the number of cells with nanoparticle concentrations below the detection limit of ICP-MS for Ti due to poor labelling efficiency could skew the average concentration of  $\text{TiO}_2$  per cell to lower values (see the electronic supplementary material for representative examples). It should also be stressed that methods such as ICP, *per se*, are precise (table 1). Additionally, similar limitations, such as underestimation of nanoparticle per cell concentrations and inability to characterize the concentration distribution in cell population, would apply to other methods (e.g. optical

fluorescence methods [17]) that provide average concentration information. While SXRF does not allow detection as low as ICP-MS, its detection limit is on par with ICP-optical emission spectrometry, which is in the microgram per millilitre range. Importantly, the nanoparticle concentration per cell values does not change depending on the efficiency of uptake by cells and furthermore, provides potentially valuable information on the distribution characteristics of nanoparticle concentration in a population of cells; hitherto not possible by ICP-based methods. The sample preparation for SXRF avoids the propagation of systematic errors. However, systematic errors could be generated by the technique including variable spatial regions of interest around the cells to determine the elemental abundance and scans of clusters of cells rather than individual cells. However, these user-dependent errors can be avoided by implementing proper cell selection and scanning area criteria.





**Figure 2.** (a) Representative SXRf ‘flyscan’ map showing areas of calcium (blue) and titanium (red), scale bar, 100  $\mu\text{m}$ . (b) SXRf scan of an individual cell showing calcium (blue) and titanium (red) counts, scale bar, 10  $\mu\text{m}$ . (c) SXRf scan of an individual cell showing titanium, scale bar, 10  $\mu\text{m}$ . The shaded area is the region of interest for the spectral sum of titanium counts. (d) Spectral sum of titanium counts found in a single cell. (e) Bar graph showing heterogeneity of concentrations of TiO<sub>2</sub> nanospheres internalized by SK-BR-3 cells.

**Table 1.** ICP-MS results for SK-BR-3 cells treated with 100  $\mu\text{g ml}^{-1}$  TiO<sub>2</sub> nanospheres.

sample number	total titanium in sample ( $\mu\text{g}$ )	TiO <sub>2</sub> per cell (ng)	Ti per cell (pmol)
1	16	1.1	23
2	17	1.1	23
3	24	1.6	33
4	20	1.3	28
5	17	1.2	24

Recent advancements in single-cell analysis clearly show heterogeneity in cell populations previously assumed to be identical [18]. There is an emerging consensus that experimental methods that provide information about average population-level cellular characteristics are insufficient, and sometimes potentially misleading [18]. Novel techniques have been explored to address the challenges associated with single-cell analysis [19]. The above results indicate that the concentration of nanoparticles in single cells could be larger than the values estimated by traditional methods, and suggest that similar challenges also exist while quantifying nanoparticles internalized by cells. A critical component of all single-cell analysis, imaging or therapy involving

nanoparticles would include accurate determination of the nanoparticle uptake concentration per cell and the distribution of this concentration in cell populations; this information will affect the safety and efficacy of the particular application [17]. Our study indicates that SXRF may be a suitable technique to determine nanoparticle concentration distribution in single-cell populations. It can easily be expanded for use with nanoparticles synthesized using other low or high Z elements [5] and provide fundamental insights into the concentration distribution modes (e.g. monomodal, bimodal), and heterogeneity and the dependence of these parameters on the structure (e.g. spherical, rod-like) and composition (e.g. metallic, ceramic). Furthermore, the ability of SXRF to map changes in elements integral to cellular

homeostasis [20], even at the sub-cellular level [5], could provide integrated understanding of nanoparticle-single-cell population interactions.

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