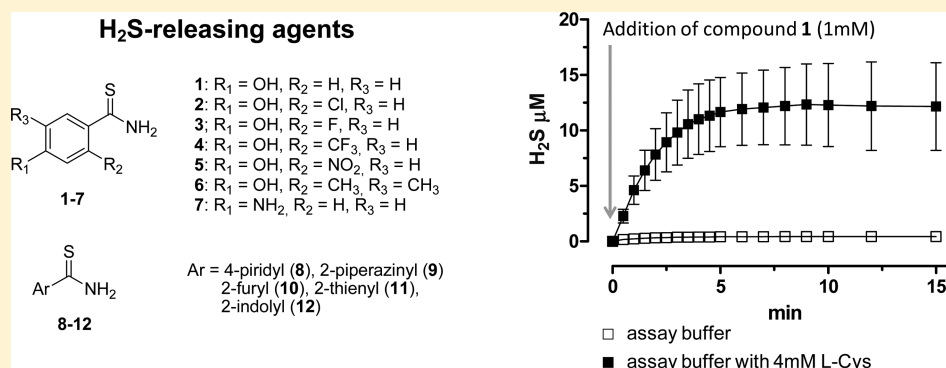


# Arylthioamides as H<sub>2</sub>S Donors: L-Cysteine-Activated Releasing Properties and Vascular Effects in Vitro and in Vivo

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## Supporting Information



**ABSTRACT:** A small library of arylthioamides 1–12 was easily synthesized, and their H<sub>2</sub>S-releasing properties were evaluated both in the absence or in the presence of an organic thiol such as L-cysteine. A number of arylthioamides (1–3 and 7) showed a slow and L-cysteine-dependent H<sub>2</sub>S-releasing mechanism, similar to that exhibited by the reference slow H<sub>2</sub>S-releasing agents, such as diallyl disulfide (DADS) and the phosphinodithioate derivative GYY 4137. Compound 1 strongly abolished the noradrenaline-induced vasoconstriction in isolated rat aortic rings and hyperpolarized the membranes of human vascular smooth muscle cells in a concentration-dependent fashion. Finally, a significant reduction of the systolic blood pressure of anesthetized normotensive rats was observed after its oral administration. Altogether these results highlighted the potential of arylthioamides 1–3 and 7 as H<sub>2</sub>S-donors for basic studies, and for the rational design/development of promising pharmacotherapeutic agents to treat cardiovascular diseases.

**KEYWORDS:** Hydrogen sulphide, H<sub>2</sub>S-releasing drugs, cardiovascular system, hypertension, potassium channels, hybrid drugs, vascular smooth muscle, thioamide

Hydrogen sulfide (H<sub>2</sub>S) is emerging as a hot topic in the field of drug discovery. Indeed, it is recognized as an important physiological gasotransmitter in mammalian body. Particularly, it plays a key role in regulating the cardiovascular homeostasis by acting as a vasodilator; this effect is mainly due to the activation of vascular ATP-sensitive (K<sub>ATP</sub>) and voltage-gated (K<sub>v7</sub>) potassium channels.<sup>1–4</sup> In mice models, blunted levels of H<sub>2</sub>S in blood, heart, and aorta are associated with increased blood pressure and decreased endothelium-mediated vasorelaxant effects.<sup>5</sup> In addition, in experimental models of hypertension, it has been evidenced that exogenous H<sub>2</sub>S can effectively prevent the progression of the pathology<sup>6</sup> and decrease the blood pressure values.<sup>3</sup> Therefore, impaired production of endogenous H<sub>2</sub>S is likely to contribute to the pathogenesis of hypertension, highlighting the great potential usefulness of the pharmacological modulation of this gasotransmitter.<sup>7,8</sup> As the administration of gaseous H<sub>2</sub>S is greatly limited by the risk of a poor posology control and overdose, the use of appropriate chemicals, behaving as H<sub>2</sub>S-releasing agents, appears to be safer. Thus, compounds

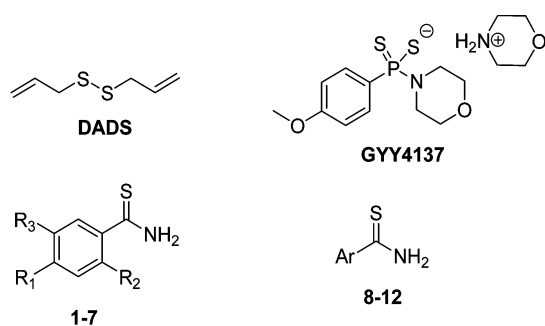
exhibiting the pharmacodynamic profile of H<sub>2</sub>S-releasing agents may be viewed as both powerful tools for basic studies and new potential pharmacotherapeutic agents for treatment of many cardiovascular diseases.

NaHS is the prototypical example of a H<sub>2</sub>S-generating agent: it is a rapid H<sub>2</sub>S donor and the most widely used H<sub>2</sub>S donor for experimental purposes. However, this salt is not suitable for clinical purposes, as the quick release of H<sub>2</sub>S may cause adverse effects, such as a rapid and excessive lowering of blood pressure.<sup>9</sup> For a safer and effective pharmacological administration, ideal H<sub>2</sub>S donors should generate H<sub>2</sub>S with slower releasing rates.<sup>10,11</sup> Organic polysulphides of garlic, such as diallyl disulfide (DADS, Chart 1), act as H<sub>2</sub>S-releasing compounds, with a relatively slow mechanism, requiring the presence of reduced glutathione.<sup>12</sup> Other examples of original synthetic H<sub>2</sub>S-releasing agents have been described in the

Received: December 21, 2012

Accepted: August 8, 2013

Published: August 8, 2013

Chart 1. Structures of Known and Newly Synthesized H<sub>2</sub>S-Releasing Agents 1–12

- 1: R<sub>1</sub> = OH, R<sub>2</sub> = H, R<sub>3</sub> = H  
 2: R<sub>1</sub> = OH, R<sub>2</sub> = Cl, R<sub>3</sub> = H  
 3: R<sub>1</sub> = OH, R<sub>2</sub> = F, R<sub>3</sub> = H  
 4: R<sub>1</sub> = OH, R<sub>2</sub> = CF<sub>3</sub>, R<sub>3</sub> = H  
 5: R<sub>1</sub> = OH, R<sub>2</sub> = NO<sub>2</sub>, R<sub>3</sub> = H  
 6: R<sub>1</sub> = OH, R<sub>2</sub> = CH<sub>3</sub>, R<sub>3</sub> = CH<sub>3</sub>  
 7: R<sub>1</sub> = NH<sub>2</sub>, R<sub>2</sub> = H, R<sub>3</sub> = H

- Ar = 4-piridyl (8), 2-piperazinyl (9)  
 2-furyl (10), 2-thienyl (11),  
 2-indolyl (12)

literature, including a number of aminothiols derivatives,<sup>13</sup> and the phosphinodithioate derivative GYY4137 (morpholin-4-ium 4-methoxyphenylmorpholinophosphinodithioate, Chart 1).<sup>14</sup> Furthermore, some H<sub>2</sub>S-releasing chemical moieties, such as the dithiolethione,<sup>15–18</sup> the thioamide, and the isothiocyanate,<sup>19,20</sup> have been used for synthesizing multifunctional drugs.

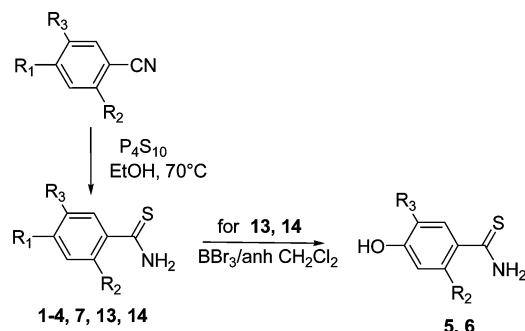
In this work, a small library of arylthioamides (compounds 1–12, Chart 1) was easily prepared through suitable synthetic routes and evaluated for their H<sub>2</sub>S-releasing properties. The *p*-hydroxybenzothioamide **1**<sup>19,20</sup> was selected as lead compound to build the library by inserting a number of modifications on the aromatic moiety: (i) introduction of different groups (Cl, F, CF<sub>3</sub>, NO<sub>2</sub>, and CH<sub>3</sub>) at the 2- and/or 5-position of the phenyl ring (2–6); (ii) replacement of the 4-hydroxy group with an amino moiety (7); and (iii) replacement of the phenyl ring with electron poor or electron rich heterocycles such as pyridine (8) and piperazine (9), or furane (10), thiophene (11), and indole (12), respectively (Chart 1). Finally, compound **1** was submitted to further experimental protocols aimed to evaluate the vasorelaxing effects on rat aortic rings, the membrane hyperpolarizing activity on human vascular smooth muscle (VSM) cells and the effects on the blood pressure of normotensive rats.

The target compounds 1–4 and 7 were easily obtained by reaction of the appropriate benzonitrile and P<sub>4</sub>S<sub>10</sub> in ethanol at 70 °C for 10 h (Scheme 1). The same procedure allowed the preparation of the methoxy-substituted intermediates **13** and **14**, which were demethylated with BBr<sub>3</sub> in dry dichloromethane to yield derivatives **5** and **6** (Scheme 1).

The heterocyclic compounds 8–12 were prepared from the corresponding amides by treatment with Lawesson's reagent in chlorobenzene at 130 °C for 12 h and in dry THF at room temperature for 12 h, for **11** and 8–10, **12**, respectively (Scheme 2). All the products were finally purified by recrystallization from toluene or flash chromatography (Supporting Information).

In order to characterize the potential H<sub>2</sub>S-releasing properties of arylthioamides 1–12, all the compounds were added at 1 mM concentration to the assay buffer (pH 7.4, 35–37 °C). The generation of H<sub>2</sub>S was evaluated by an amperometric approach, allowing to have a real-time determination of the H<sub>2</sub>S-release and thus to perform a qualitative/quantitative description of the

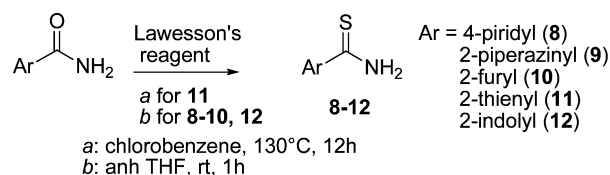
Scheme 1. Synthesis of 1–7, 13, and 14



- 1: R<sub>1</sub> = OH, R<sub>2</sub> = H, R<sub>3</sub> = H  
 2: R<sub>1</sub> = OH, R<sub>2</sub> = Cl, R<sub>3</sub> = H  
 3: R<sub>1</sub> = OH, R<sub>2</sub> = F, R<sub>3</sub> = H  
 4: R<sub>1</sub> = OH, R<sub>2</sub> = CF<sub>3</sub>, R<sub>3</sub> = H  
 7: R<sub>1</sub> = NH<sub>2</sub>, R<sub>2</sub> = H, R<sub>3</sub> = H  
 13: R<sub>1</sub> = OCH<sub>3</sub>, R<sub>2</sub> = NO<sub>2</sub>, R<sub>3</sub> = H  
 14: R<sub>1</sub> = OCH<sub>3</sub>, R<sub>2</sub> = CH<sub>3</sub>, R<sub>3</sub> = CH<sub>3</sub>

- 5: R<sub>2</sub> = NO<sub>2</sub>, R<sub>3</sub> = H  
 6: R<sub>2</sub> = CH<sub>3</sub>, R<sub>3</sub> = CH<sub>3</sub>

Scheme 2. Synthesis of 8–12

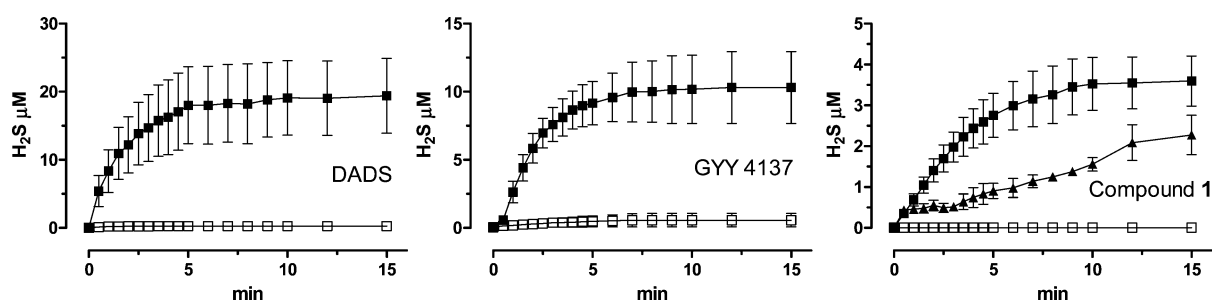


process. In addition, since the release of H<sub>2</sub>S from DADS is reported to be a process, which requires the presence of organic thiols such as reduced glutathione,<sup>12</sup> the experiments were also performed in the presence of 4 mM L-cysteine. NaHS, DADS, and GYY4137 were also assayed for comparison purposes. As expected, the addition of 1 mM NaHS at pH 7.4, both in the presence or in the absence of L-cysteine, was followed by an immediate formation of high concentration of H<sub>2</sub>S (about 200 μM). The concentration of H<sub>2</sub>S showed a rapid fall (by about 40%) in the first two minutes after the addition of NaHS, followed by a slower and apparently constant decline in the remaining time (data not shown).

In order to obtain a satisfactory description of the kinetics of those compounds generating H<sub>2</sub>S with a slow and progressive process, the parameters of C<sub>max</sub> (maximal concentration of H<sub>2</sub>S at the steady state) and t<sub>1/2</sub> (time required to reach a concentration = 50% of C<sub>max</sub>) were extrapolated from the curves H<sub>2</sub>S-release vs time. Figure 1 reports the curves H<sub>2</sub>S-release vs time for DADS, GYY4137, and **1**, taken as representative of the thioamide series. The values of C<sub>max</sub> and t<sub>1/2</sub> of the tested compounds are listed in Table 1.

The incubation of DADS and GYY4137 in the assay buffer led to a negligible formation of H<sub>2</sub>S, while in the presence of L-cysteine, a slow and significant release of H<sub>2</sub>S was observed. Analogously, in the assay buffer alone, *p*-hydroxybenzothioamide **1** did not show significant H<sub>2</sub>S-releasing properties. In contrast, **1** exhibited an L-cysteine-dependent release of H<sub>2</sub>S (Figure 1 and Table 1). In order to evaluate the possible influence of other organic thiols, the H<sub>2</sub>S-releasing properties of **1** were also tested in the presence of reduced glutathione (GHS). Like L-cysteine, even GSH triggered a significant, but slower, release of H<sub>2</sub>S from **1** (Figure 1 and Table 1).

As concerns the hydroxybenzothioamide derivatives 2–6 (Table 1), the incubation of **2** (2-Cl), **3** (2-F), and **6** [3,5-(CH<sub>3</sub>)<sub>2</sub>] led to a L-cysteine-dependent release of H<sub>2</sub>S. The C<sub>max</sub>



**Figure 1.** Curves describe the increase of  $\text{H}_2\text{S}$  concentration, with respect to time, following the incubation of DADS, GYY 4137, and **1** in the assay buffer, in the absence (white squares) or in the presence of L-cysteine (black squares) or glutathione (black triangles).  $\text{H}_2\text{S}$  was recorded by amperometry; the vertical bars indicate the SEM.

**Table 1. Parameters of  $C_{\text{max}}$  and  $t_{1/2}$  Relative to the  $\text{H}_2\text{S}$ -Releasing Effects Exhibited by the Arylthioamides 1–12 and Reference Drugs, after Their Incubation in the Assay Buffer at Physiological pH and Temperature, in the Absence or in Presence (+L-Cysteine) of 4 mM L-Cysteine; nc = Not Calculated**

compound	assay buffer		assay buffer + L-cysteine	
	$C_{\text{max}}$ ( $\mu\text{M}$ )	$t_{1/2}$ (min)	$C_{\text{max}}$ ( $\mu\text{M}$ )	$t_{1/2}$ (min)
1	<2	nc	$3.6 \pm 0.6$	$2.1 \pm 0.2$
2	<2	nc	$10.8 \pm 2.6$	$2.6 \pm 0.8$
3	<2	nc	$15.2 \pm 3.9$	$2.2 \pm 0.7$
4	<2	nc	<2	nc
5	<2	nc	<2	nc
6	$6.1 \pm 0.3$	$5.0 \pm 0.3$	$11.6 \pm 0.3$	$4.1 \pm 0.4$
7	<2	nc	$6.0 \pm 2.8$	$4.3 \pm 1.0$
8	$3.1 \pm 1.1$	$3.1 \pm 0.6$	$16.1 \pm 5.0$	$3.1 \pm 0.2$
9	$14.8 \pm 0.5$	$13.7 \pm 1.7$	$21.4 \pm 1.0$	$6.4 \pm 0.6$
10	$7.9 \pm 0.7$	$7.5 \pm 0.5$	$10.6 \pm 0.7$	$5.8 \pm 0.4$
11	$4.7 \pm 0.4$	$1.8 \pm 0.4$	$5.7 \pm 0.5$	$1.5 \pm 0.4$
12	$14.4 \pm 0.6$	$11.7 \pm 1.0$	$15.9 \pm 1.6$	$9.2 \pm 0.7$
DADS	<2	nc	$19.4 \pm 5.5$	$1.5 \pm 0.3$
GYY4137	<2	nc	$10.3 \pm 2.6$	$2.5 \pm 0.8$

parameters of **2**, **3** and **6** were higher than that exhibited by **1** and were comparable to those of the reference drugs DADS and GYY4137. Moreover, it should be noted that in the absence of L-cysteine, the  $\text{H}_2\text{S}$ -release from **2** and **3** was negligible, whereas the release from **6** was lower than that observed in the presence of the aminoacid, although it was still significant. Compounds **4** and **5**, bearing strong electron-withdrawing groups, such as 2- $\text{CF}_3$  and 2- $\text{NO}_2$ , respectively, showed L-cysteine-dependent  $\text{H}_2\text{S}$ -releasing properties significantly lower with respect to the parent compound **1**. Finally, the replacement of the 4-OH group of **1** with a primary amino moiety (see compound **7**) led to a slight, albeit not significant, improvement of the  $\text{H}_2\text{S}$ -release.

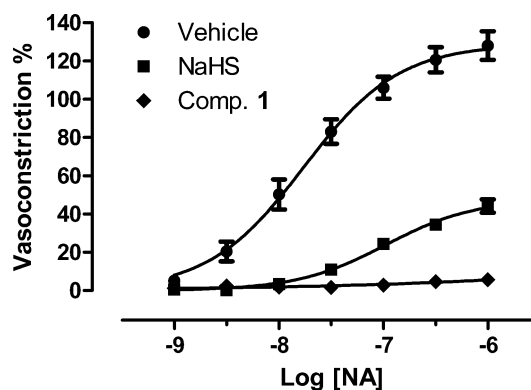
Mixed results are associated with the substitution of the phenyl ring with electron poor or electron rich heterocycles (**8**–**12**). In particular, the pyridylthioamide **8** (electron poor) and the thienylthioamide **11** (electron rich) both exhibited a modest but significant generation of  $\text{H}_2\text{S}$  in the absence of L-cysteine; whereas, in the presence of the aminoacid, compound **8** showed a dramatically increased  $C_{\text{max}}$ , a feature not possessed by compound **11**. The heterocyclic derivatives **9** (electron poor), **10**, and **12** (electron rich) exhibited high and L-cysteine-independent  $\text{H}_2\text{S}$ -releasing effects. The profile of L-cysteine-independent  $\text{H}_2\text{S}$  donor of derivatives **9**, **10**, and **12** is likely to

be due to an easy and spontaneous hydrolysis of the  $\text{H}_2\text{S}$ -releasing moiety.

This first phase of the experimental work led to identify a number of  $\text{H}_2\text{S}$ -releasing arylthioamides (**1**–**3** and **7**) that showed an L-cysteine-dependent mechanism. Such a profile has been suggested to be convenient for the development of a potential drug, as this feature may provide a selective  $\text{H}_2\text{S}$ -release in a biological environment (i.e., in the presence of endogenous organic thiols such as L-cysteine, reduced glutathione, etc.).

Compound **1** was selected for further pharmacological studies on the basis of the following issues: (i) **1** showed a thiol-dependent  $\text{H}_2\text{S}$ -releasing profile; (ii) although it has been already reported as a  $\text{H}_2\text{S}$ -releasing side-chain in naproxen analogues endowed with reduced toxicity,<sup>19,20</sup> its  $\text{H}_2\text{S}$ -mediated cardiovascular effects have not been yet specifically investigated.

First, the influence of **1** (or the relative vehicle) on the vasoconstrictor effects of noradrenaline (NA) was tested in isolated rat aortic rings and compared with that of NaHS (Figure 2).<sup>11</sup> NA produced a strong contractile response, with

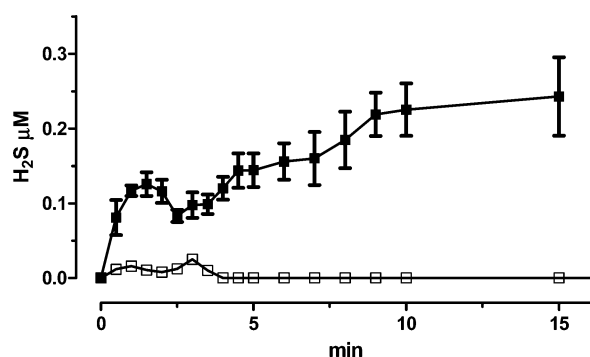


**Figure 2.** Concentration–response curves to NA, obtained on endothelium-denuded rat aortic rings preincubated with NaHS (squares), **1** (diamonds), or the corresponding vehicle (spots). The effects are expressed as a % of the contractile responses previously induced by the administration of 60 mM KCl. The vertical bars indicate the SEM.

an  $E_{\text{max}}$  of  $128 \pm 7$  and a potency value ( $\text{pEC}_{50}$ ) of  $7.76 \pm 0.08$ . In the presence of 1 mM NaHS, the vasoconstrictor effects of NA were strongly inhibited ( $E_{\text{max}}$   $44 \pm 3$ ;  $\text{pEC}_{50}$   $6.98 \pm 0.09$ ). It should be highlighted that, at pH 7.4 and 37 °C, 1 mM NaHS is expected to readily produce an initial and transient concentration of  $\text{H}_2\text{S}$  of about 200  $\mu\text{M}$ , due to the protonation of the hydrosulphide anion.<sup>21</sup> Such a theoretical hypothesis is

well confirmed by the above experimental data. Interestingly, the pretreatment of the aortic rings with compound **1** (1 mM) almost completely abolished the ability of NA to evoke any vasoconstriction (Figure 2).

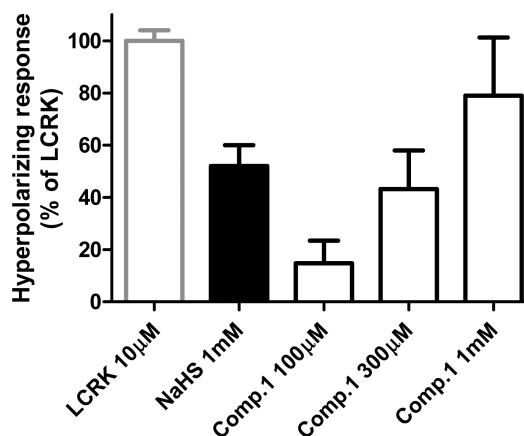
Noteworthy, the previous amperometric measurements indicated that **1** ensured a long-lasting release of relatively low concentrations of H<sub>2</sub>S, requiring the presence of L-cysteine or glutathion; however, in this experimental setup, exogenous L-cysteine or glutathion were not added. Therefore, the above experimental result suggests that (a) the amounts of organic thiols endogenously present in the biological sample (i.e., the aortic rings) can ensure the release of H<sub>2</sub>S from **1**; (b) a prolonged presence of relatively low concentrations of H<sub>2</sub>S seems to be highly effective in modulating the activity of vasoconstrictor agents, such as NA. In order to confirm the above hypotheses, compound **1** was also incubated in Tyrode buffer (i.e., in the same conditions of the functional experiments), in the presence or in the absence of rat aortic tissue. Neither L-cysteine nor glutathion were exogenously added. The formation of H<sub>2</sub>S was continuously evaluated by amperometry. In the presence of the vascular tissue (rat aortic rings), the addition of compound **1** led to a significant time-dependent formation of H<sub>2</sub>S. Fifteen minutes after the incubation, the concentration of H<sub>2</sub>S in the Tyrode buffer was  $0.24 \pm 0.05 \mu\text{M}$  (Figure 3). This evidence indicates that



**Figure 3.** Curves describe the increase of H<sub>2</sub>S concentration, with respect to time, following the incubation of **1** in Tyrode buffer, in the absence (white squares) or in the presence of aortic tissue (black squares). H<sub>2</sub>S was recorded by amperometry; the vertical bars indicate the SEM.

compound **1** is effectively transformed into H<sub>2</sub>S, by means of biomolecules (probably, organic thiols) endogenously present in the vascular tissue; in particular, effective H<sub>2</sub>S concentrations are likely to be produced in the intracellular compartment (i.e., where the levels of endogenous thiols are highly available). In fact, in the absence of the vascular tissue, no significant production of H<sub>2</sub>S was detected (Figure 3).

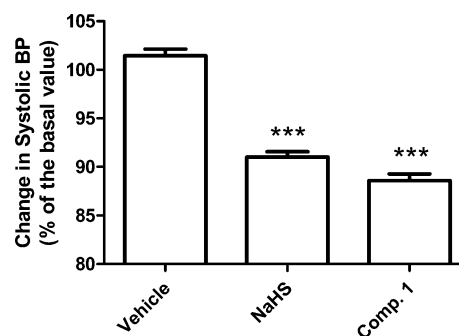
As the vasodilator effects of H<sub>2</sub>S are largely mediated by the activation of K<sub>ATP</sub> channels and consequent membrane hyperpolarization of VSM cells,<sup>2–4</sup> it was thought interesting to evaluate the effects of **1** and NaHS on the membrane potential of cultured human VSM cells (HASMCs). Furthermore, the K<sub>ATP</sub>-activator levcromakalim (LCRK) was selected as reference hyperpolarizing agent. As shown in Figure 4, the administration of NaHS caused a significant membrane hyperpolarization of HASMCs; the maximal hyperpolarization induced by 1 mM NaHS was  $52 \pm 8\%$  of that induced by LCRK (10 μM). Compound **1** hyperpolarized the membranes of VSM cells in a concentration-dependent fashion (Figures 3



**Figure 4.** Hyperpolarizing effects induced by LCRK, NaHS, and increasing concentrations of **1** in HASMCs. Data are calculated as changes in relative fluorescence and then expressed as a % of the maximal effect induced by LCRK. The vertical bars indicate the SEM.

and 4), showing a  $79 \pm 22\%$  hyperpolarizing effect (expressed as a % of the effect produced by LCRK at 10 μM) at the highest concentration (1 mM). As already observed in rat aortic rings, these results indicate that the release of H<sub>2</sub>S from **1** can be promoted by the organic thiols endogenously present in the HASMCs and that the long-lasting presence of relatively low concentrations of H<sub>2</sub>S is more effective than a high and transient peak concentration.

Finally, the influence of **1** and NaHS on the blood pressure of normotensive rats was tested. The anesthetized normotensive rats showed a basal level of systolic pressure of  $134 \pm 2$  mmHg. After oral administration of the vehicle, no significant change of the systolic blood pressure was observed ( $102 \pm 1\%$  of the basal levels), whereas the systolic blood pressure of the animals was significantly reduced after oral administration of 0.1 mg/kg NaHS ( $91 \pm 1\%$  of the basal levels) and equimolar doses of **1** ( $89 \pm 1\%$  of the basal levels); Figure 5.



**Figure 5.** Histograms represent the mean levels of systolic pressure, recorded in normotensive rats, 20 min after the oral administration of NaHS (0.1 mg/kg), compound **1** (equimolar dose), or their vehicle. Data are calculated as % of the basal systolic blood pressure, recorded before the administration. The vertical bars indicate the SEM; \*\*\* = significantly different ( $P < 0.001$ ) from vehicle.

In conclusion, this work furnished a small library of thioamides showing different H<sub>2</sub>S-releasing rates. Some derivatives (1–3 and 7) exhibited smart H<sub>2</sub>S-releasing properties triggered by the presence of organic thiols, while other derivatives (6 and 8–12) were thiol-independent H<sub>2</sub>S donors. The *p*-hydroxybenzothioamide **1** produced typical

vascular effects of H<sub>2</sub>S, both in in vitro and in vivo experiments. Specifically, compound **1**, which was selected as representative for investigating the potential vascular effects of all the analogues of this series, inhibited the NA-induced vasoconstriction in isolated rat aortic rings, hyperpolarized the membranes of human HASMCs, and reduced the systolic blood pressure after oral administration. These thioamide derivatives might represent useful tools for the rational development of promising H<sub>2</sub>S-releasing agents addressed to the therapeutic treatment of cardiovascular diseases.

## ■ ASSOCIATED CONTENT

### ■ Supporting Information

Synthesis, characterization, and pharmacological studies of compounds **1–12**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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### Notes

The authors declare no competing financial interest.

## ■ ACKNOWLEDGMENTS

This article has been supported by the “Regional Health Research Program 2009” of Regione Toscana, Italy.

## ■ ABBREVIATIONS

DADS, diallyl disulfide; GYY4137, morpholin-4-ium 4-methoxyphenyl-morpholino-phosphinodithioate; NA, noradrenaline; SEM, standard error of mean; VSM, vascular smooth muscle

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