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Correction

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Correction to Rapid Determination of Ionization Constants (pK_a) by UV Spectroscopy Using 96-Well Microtiter Plates

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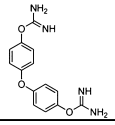
Four additional references have been added to the reference section, and the last column of Table 1 has been updated with the correct references.

Table 1. pK_a Values of Monoacidic, Monobasic, and Dibasic Compounds Determined by the 96-Well UV Spectrophotometric Method

Cpd	Structure	Solvent[a]	λ (nm)[b]	Measured[c] pK_a	Mean value ± SD[d]	Lit. value[e]	Ref.
1		H_2O	318/400	7.01 7.02 7.02	7.02 ± 0.01	7.16	18
		H_2O + 2% DMSO	318/400	6.84 6.88 6.90	6.87 ± 0.03	na	na
2		H_2O + 2% DMSO	268/280	6.20 6.27 6.21 6.20	6.22 ± 0.03	6.23	19
3		H_2O + 2% DMSO	297/354	9.22 9.17 9.20	9.20 ± 0.03	9.3	20
4		H_2O + 2% DMSO	244/312	8.15 8.24 8.05	8.14 ± 0.09	8.12 8.04	21
5		H_2O + 2% DMSO	250/390	10.63 10.72 10.75	10.70 ± 0.06	10.4	16

[a]The use of 2% v/v DMSO as cosolvent did not alter significantly the pK_a value of the test compounds. Working temperature = 30 °C. All pK_a were measured at constant ionic strength (I = 0.1 M) and concentration (C = 0.2 mM). [b]Analytical wavelengths are determined at the maximum and minimum absorption values in the spectral difference plot. [c]Experiments were repeated at least three times. [d]Standard deviation. [e]Experimental pK_a values at 25 °C in water.

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